Supporting Information

Three-Dimensional PdPtCu Nanoalloys with Controllable Composition and Spiny Surface for the Enhancement of Ethanol Electrocatalytic Properties

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Samples	Atomic ratio of (Pd), Pt and Cu		
PtCu NPs	49.13:50.87		
Pd _{0.5} PtCu NAs	20.78:39.26:39.96		
PdPtCu NAs	34.89:28.95:36.16		
Pd ₂ PtCu NAs	51.85:27.13:21.02		

Table S1. The ICP-AES results of different components Pt-based NMs



Figure S1. The TEM images of NMs with different components: (A) PtCu NPs, (B) Pd_{0.5}PtCu NAs and (C) Pd₂PtCu NAs.



Figure S2. The XPS spectra of Pt 4f in PdPtCu NAs and Pt/C catalysts.



Figure S3. The TEM images of PdPtCu NPs: (A) without PF-127, (B) 0.06 g PF-127 and (C) 0.03 g PVP.



Figure S4. The CV curves of CO stripping test: (A) PtCu NPs, (B) Pd_{0.5}PtCu NAs, (C) PdPtCu NAs, (D) Pd₂PtCu NAs and (E) Pt/C catalysts; (F) The histogram of ECSA values of the corresponding catalysts.

Samples	ECSA	Peak current density	Mass activity	Specific activity
_	$(m^2 g^{-1})$	$(j/mA \ cm^{-2})$	$(j_{MA}/mA mg^{-1})$	$(j_{SA}/mA \text{ cm}^{-2})$
PtCu NPs	46.47	69.19	695.25	1.43
Pd _{0.5} PtCu NAs	49.18	105.70	847.68	2.62
PdPtCu NAs	89.59	225.87	1679.87	4.21
Pd ₂ PtCu NAs	70.73	143.07	849.67	3.53
Pt/C	45.21	10.87	484.72	1.09

Table S2. The electrochemical performance parameters of all samples toward EOR

medium				
Samples	Electrolyte condition Mass activity		Ref.	
		$(j_{MA}/mA mg^{-1})$		
Pt-Pd-Cu nanodendrites	0.1 M KOH +0.5 M ethanol	2588.00	20	
Pd-Pt-Ag nanosheets	0.1 M KOH +0.5 M ethanol	1387.00	50	
$Pt_1Ru_{0.5}Sn_{0.5}\text{-}RGO$	1.0 M KOH +1.0 M ethanol	1517.00	51	
PdPtCu NAs	1.0 M KOH +1.0 M ethanol	1679.87	This work	

Table S3. The electrocatalytic properties of ternary Pt-based nanocatalysts toward EOR in alkaline



Figure S5. The TEM images of PdPtCu NAs after EOR stability tests.



Figure S6. EDS spectrum of PdPtCu NAs after EOR stability tests.