Supporting Information for

Activated Carbon by One-Step Calcination of Deoxygenated Agar for High Voltage Lithium Ion Supercapacitor

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1. Calculation

The Specific capacitance values are calculated via equation (1), where *C* [F g⁻¹] is the specific capacitance, *m* [mg] is the mass loading of AC, v [V s⁻¹] is the scan rate, V_{\min} [V vs RHE] is the minimum electrode potential, V_{\max} [V vs RHE] is the maximum potential and *i* [mA] is the current., the integral part is the area of CV curve at each scan rate.

$$C = \frac{1000}{mv(V_{\text{max}} - V_{\text{min}})} \int_{V_{\text{min}}}^{V_{\text{max}}} i(V) dV \qquad (1)$$

Specific energy was calculated by integrating the area beneath the galvanostatic discharge curve of hybrid lithium ion supercapacitor with equation (2), where E [Wh kg⁻¹] is the energy density, U [V] is the cell working voltage, i [mA] is the discharge current, m [mg] is the mass loading of AC, t_{min} [s] is the discharge starting time and t_{max} [s] is discharge ending time.

$$E = \frac{1000i}{3600m} \int_{t_{\min}}^{t_{\max}} U(t) dt \qquad (2)$$

Specific power was calculated with equation (3), where P [W kg⁻¹] is specific power, E [Wh kg⁻¹] is the specific energy and Δt [s] is the discharge time.

$$P = \frac{3600E}{\Delta t} \qquad (3)$$

2. Supporting Figures



Figure S1. Electrode potential and potential stability window for deoxygenated agar-derived AC electrode in 21 m LiTFSI "water-in-salt" at 25 °C and 101.325 kPa. The actually operating potential window of hybrid lithium-ion supercapacitor is between 2.14 V and 4.04 V.



Figure S2. Differential capacitance curves from the charge/discharge curves of activated carbon electrode in the hybrid lithium ion supercapacitor at various current rates from 0.05 to 1.0 mA cm^{-2} .



Figure S3. IR drops of hybrid lithium-ion supercapacitor at various current rates.



Figure S4. C1s XPS spectra of deoxygenated agar-derived AC before and after cycles.