

## **Supplementary Information**

### **Four New Luminescent Metal-Organic Frameworks as Multifunctional Sensors for Detecting $\text{Fe}^{3+}$ , $\text{Cr}_2\text{O}_7^{2-}$ and Nitromethane**

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Figure **S1**. IR spectrum of the L ligand.

Figure **S2**. IR spectrum of MOF **1**.

Figure **S3**. IR spectrum of MOF **2**.

Figure **S4**. IR spectrum of MOF **3**.

Figure **S5**. IR spectrum of MOF **4**.

Figure **S6**. The PXRD patterns of **1** before and after immersed in  $\text{Fe}^{3+}$ ,  $\text{Cr}_2\text{O}_7^{2-}$  and nitromethane for 3 days.

Figure **S7**. The PXRD patterns of **2** before and after immersed in  $\text{Fe}^{3+}$ ,  $\text{Cr}_2\text{O}_7^{2-}$  and nitromethane for 3 days.

Figure **S8**. The PXRD patterns of **3** before and after immersed in  $\text{Fe}^{3+}$ ,  $\text{Cr}_2\text{O}_7^{2-}$  and nitromethane for 3 days.

Figure **S9**. The PXRD patterns of **4** before and after immersed in  $\text{Fe}^{3+}$ ,  $\text{Cr}_2\text{O}_7^{2-}$  and nitromethane for 3 days.

Figure **S10**. The TGA diagram of MOF **1**.

Figure **S11**. The TGA diagram of MOF **2**.

Figure **S12**. The TGA diagram of MOF **3**.

Figure **S13**. The TGA diagram of MOF **4**.

Figure **S14**. Liquid UV/vis spectra of different metal cations in aqueous solutions.

Figure **S15**. Liquid UV/vis spectra of different anions in aqueous solutions.

Figure **S16**. Solid state UV/vis spectra of MOFs **1-4**.

Table **S1**. Crystal date for complexes **1-4**.

Table **S2**. Selected bond lengths and angles for MOFs **1-4**.

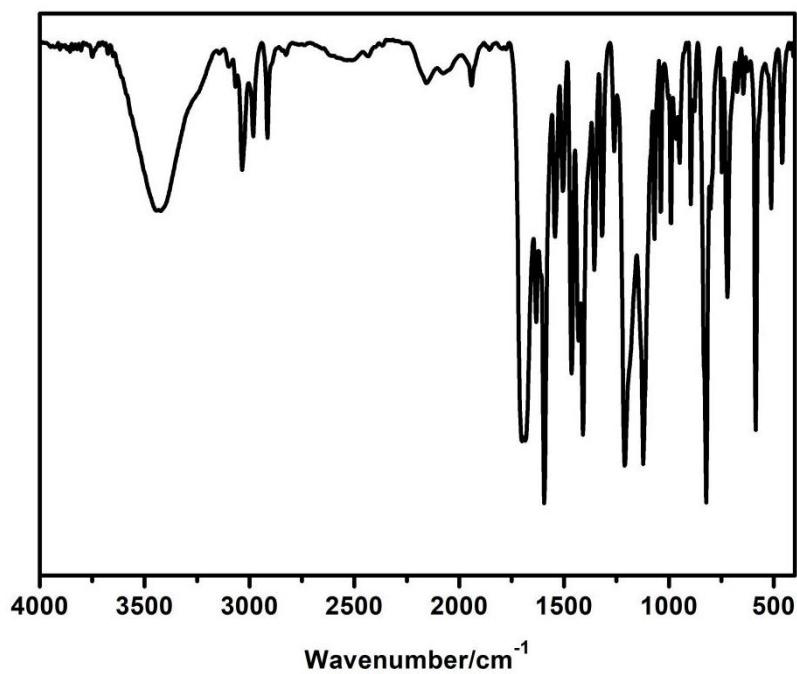


Figure S1 IR spectrum of the L ligand.

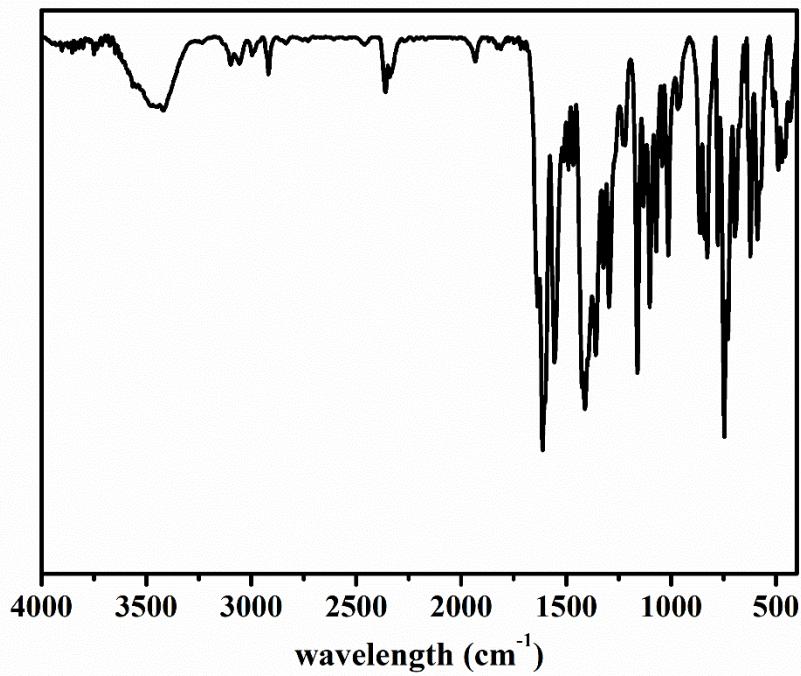


Figure S2 IR spectrum of MOF 1.

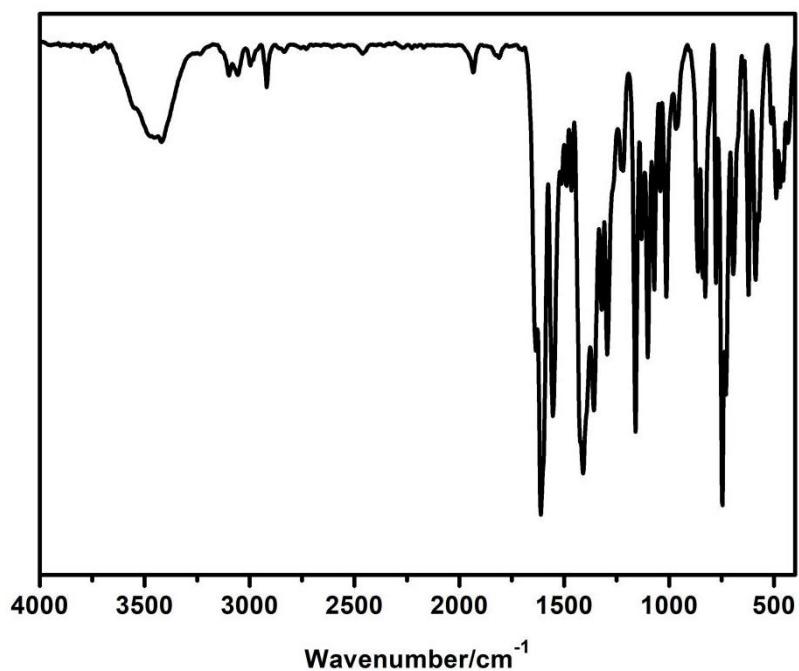


Figure S3 IR spectrum of MOF 2.

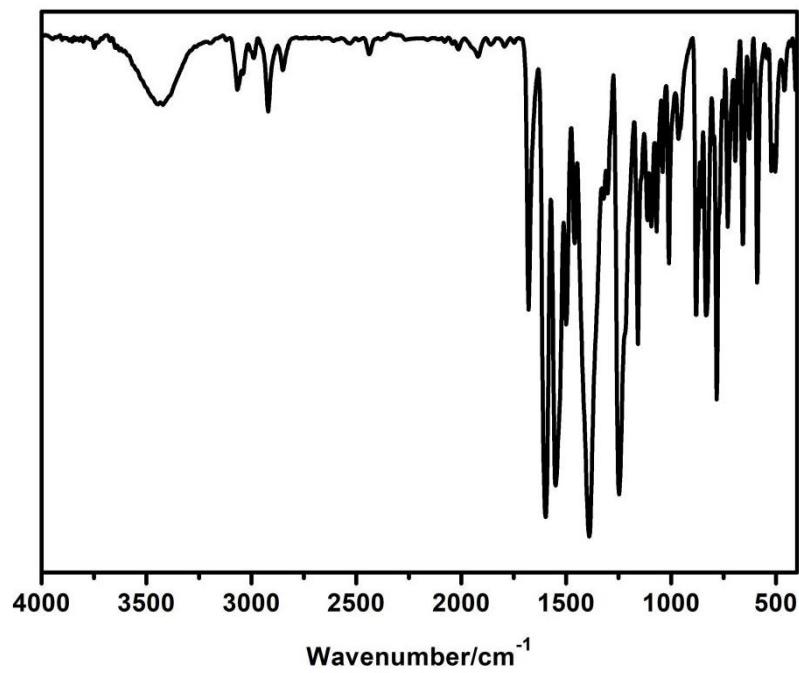


Figure S4 IR spectrum of MOF 3.

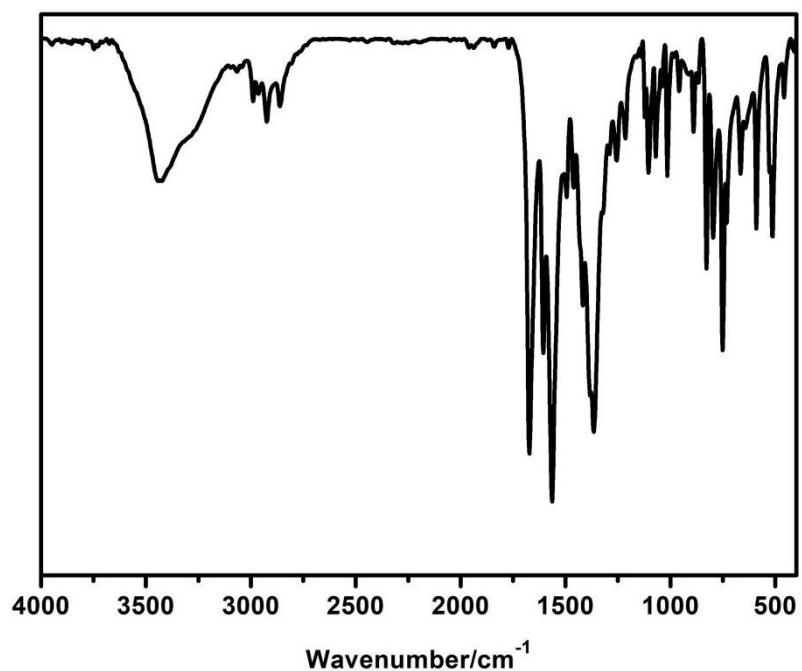


Figure S5 IR spectrum of MOF 4.

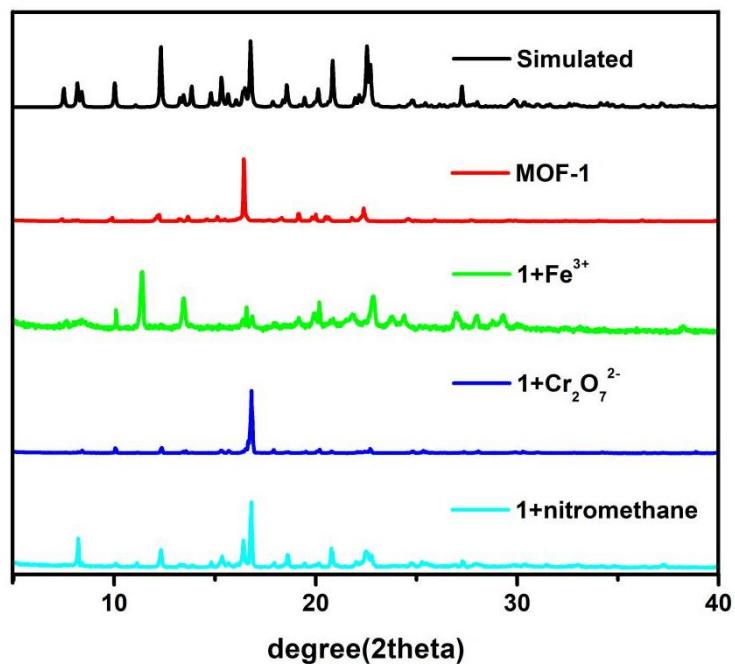


Figure S6 The PXRD patterns of **1** before and after immersed in Fe<sup>3+</sup>, Cr<sub>2</sub>O<sub>7</sub><sup>2-</sup> and nitromethane for 3 days.

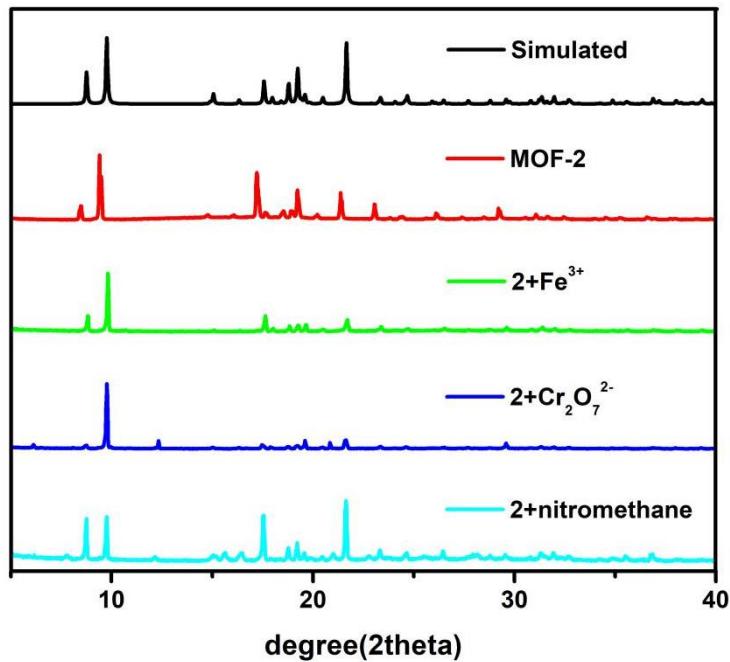


Figure S7 The PXRD patterns of **2** before and after immersed in Fe<sup>3+</sup>, Cr<sub>2</sub>O<sub>7</sub><sup>2-</sup> and nitromethane for 3 days.

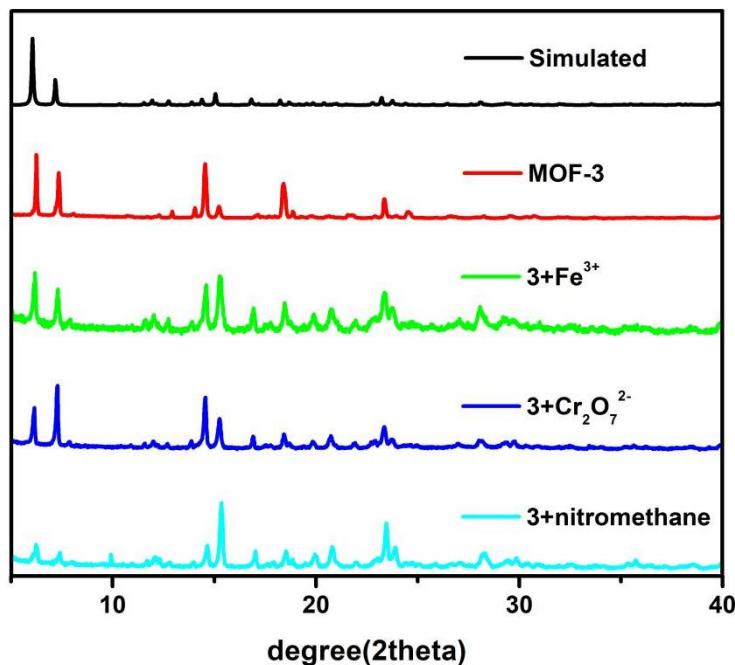


Figure S8 The PXRD patterns of **3** before and after immersed in Fe<sup>3+</sup>, Cr<sub>2</sub>O<sub>7</sub><sup>2-</sup> and nitromethane for 3 days.

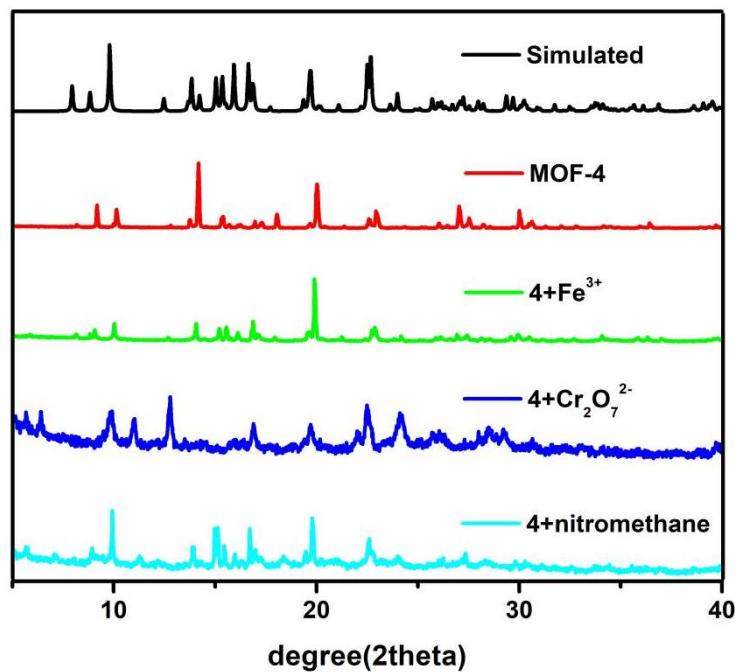


Figure S9 The PXRD patterns of 4 before and after immersed in  $\text{Fe}^{3+}$ ,  $\text{Cr}_2\text{O}_7^{2-}$  and nitromethane for 3 days.

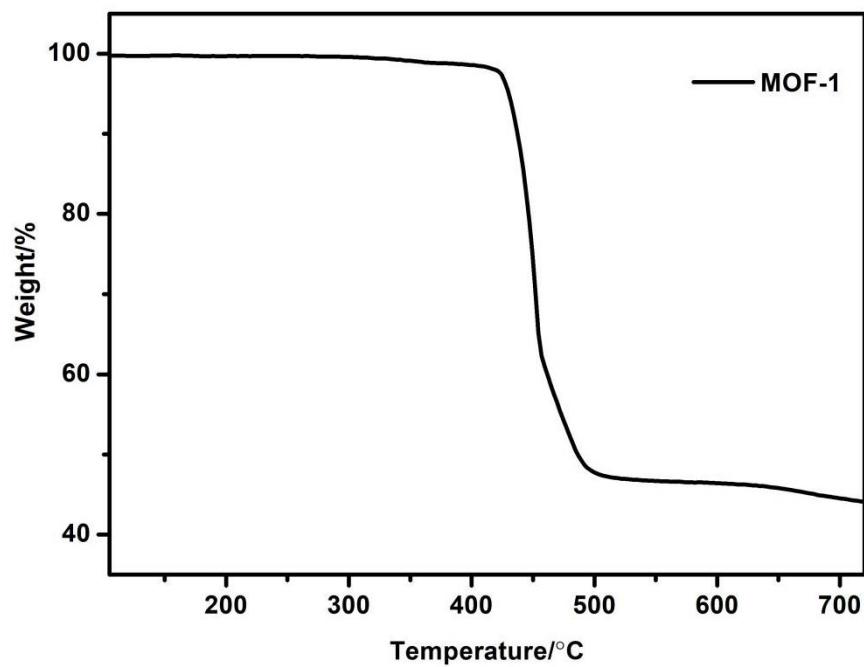


Figure S10 The TGA diagram of MOF 1.

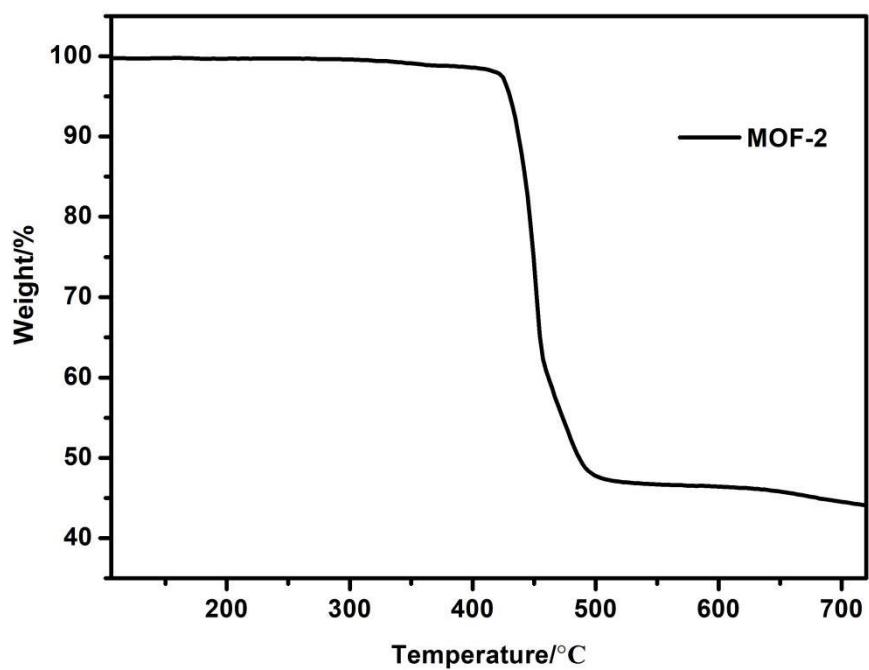


Figure S11 The TGA diagram of MOF 2.

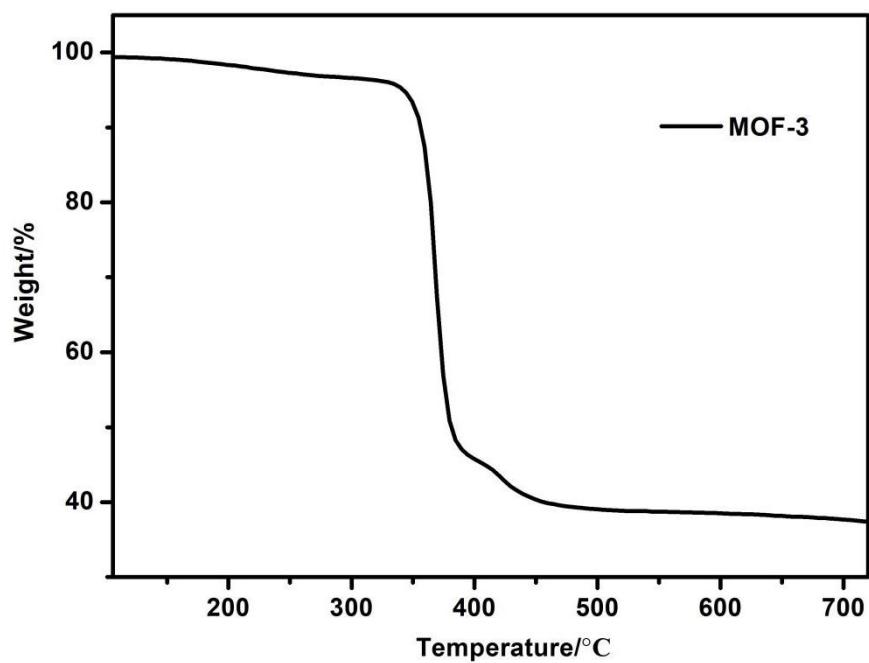


Figure S12 The TGA diagram of MOF 3.

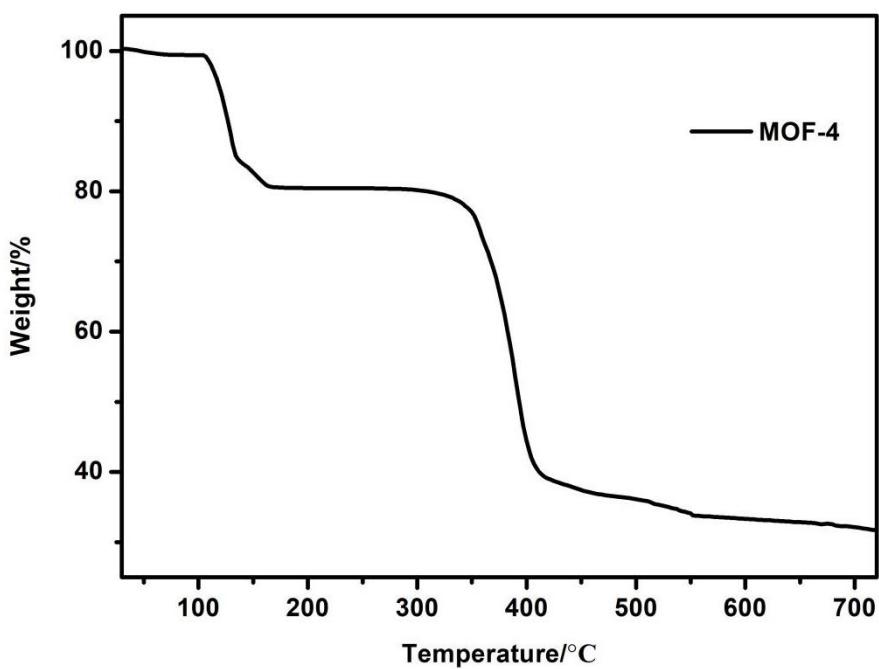


Figure S13 The TGA diagram of MOF 4.

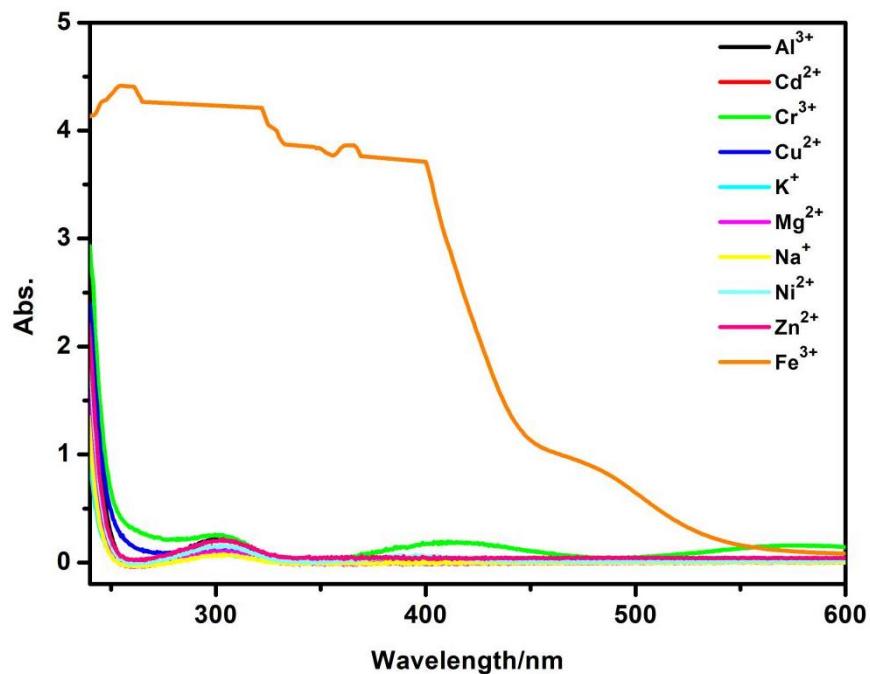


Figure S14 Liquid UV/vis spectra of different metal cations in aqueous solutions.

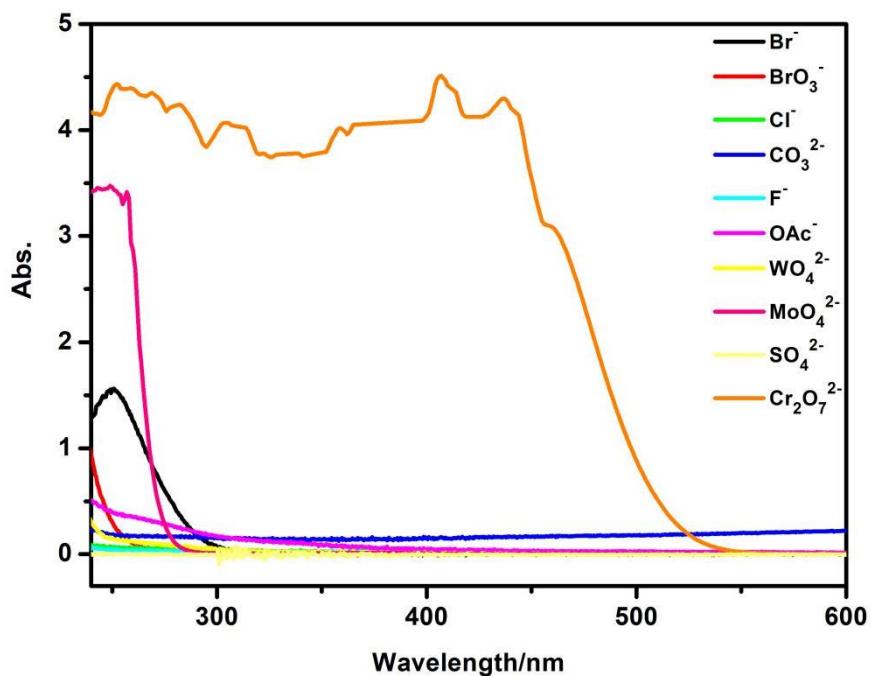


Figure S15 Liquid UV/vis spectra of different anions in aqueous solutions.

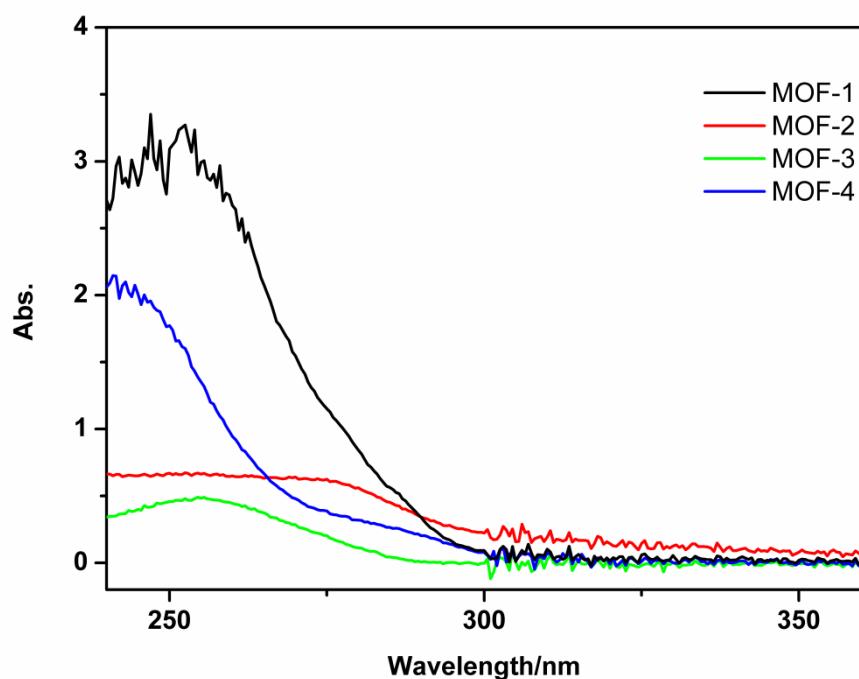


Figure S16 Solid state UV/vis spectra of MOFs 1-4.

**Table S1.** Crystal data and structure refinement for MOFs **1-4**.

| MOFs  | <b>1</b>   | <b>2</b>   | <b>3<sup>a</sup></b>   | <b>4</b>   |
|---|--|--|--|--|
| Formula   | C <sub>32</sub> H <sub>24</sub> ZnN <sub>2</sub> O <sub>6</sub> S <sub>3</sub> | C <sub>26</sub> H <sub>20</sub> ZnN <sub>2</sub> O <sub>4</sub> S <sub>2</sub> | C <sub>32</sub> H <sub>24</sub> CdN <sub>2</sub> O <sub>5</sub> S <sub>2</sub> | C <sub>32</sub> H <sub>38</sub> CdN <sub>4</sub> O <sub>8</sub> S <sub>2</sub> |
| Formula weight  | 694.08   | 553.93   | 693.05   | 783.18   |
| Crystal system  | Triclinic  | Monoclinic   | Triclinic  | Triclinic  |
| Space group   | <i>P</i>    | <i>C</i> 2/c   | <i>P</i>    | <i>P</i>    |
| <i>a</i> / Å  | 11.0591(19)  | 22.955(2)  | 9.5445(10)   | 7.2280(8)  |
| <i>b</i> / Å  | 12.268(2)  | 6.2161(6)  | 12.8573(14)  | 10.7330(11)  |
| <i>c</i> / Å  | 13.439(2)  | 19.744(2)  | 15.4179(17)  | 11.7457(13)  |
| $\alpha$ / °  | 116.419(3)   | 90   | 106.726(2)   | 108.220(2)   |
| $\beta$ / °   | 94.590(3)  | 118.4960(10)   | 98.616(2)  | 91.311(2)  |
| $\gamma$ / °  | 102.601(3)   | 90   | 90.113(2)  | 100.104(2)   |
| <i>V</i> / Å <sup>3</sup>   | 1560.5(5)  | 2476.0(4)  | 1789.5(3)  | 849.15(16)   |
| <i>Z</i>  | 2  | 4  | 2  | 1  |
| <i>D<sub>calcd</sub></i> /g·cm <sup>-3</sup>  | 1.477  | 1.486  | 1.286  | 1.532  |
| $\mu$ / mm <sup>-1</sup>  | 1.035  | 1.196  | 0.763  | 0.821  |
| <i>F</i> (000)  | 712  | 1136   | 700  | 402  |
| $\theta$ min-max / °  | 1.884, 27.473  | 2.019, 27.240  | 1.396, 27.160  | 1.832, 27.436  |
| Tot., uniq. data  | 10643, 7093  | 10294, 2731  | 15596, 7757  | 5759, 3842   |
| <i>R</i> (int)  | 0.0470   | 0.0369   | 0.036  | 0.0887   |
| Observed<br>data[ <i>I</i> >2σ( <i>I</i> )]   | 7093   | 2731   | 7757   | 3842   |
| N <sub>ref</sub> , N <sub>par</sub>   | 594, 490   | 0, 160   | 156, 418   | 225, 265   |
| GOF on F <sup>2</sup>   | 0.923  | 1.073  | 1.037  | 0.938  |
| <i>R</i> <sub>1</sub> , <i>wR</i> <sub>2</sub> [ <i>I</i> >2σ( <i>I</i> )] <sup>b</sup> | 0.0468, 0.1072   | 0.0348, 0.0920   | 0.0380, 0.1064   | 0.0415, 0.0902   |
| <i>R</i> <sub>1</sub> , <i>wR</i> <sub>2</sub> (all data)                               | 0.0788, 0.1188   | 0.0407, 0.0974   | 0.0471, 0.1131   | 0.0510, 0.0938   |

[a] The residual electron densities were flattened by the SQUEEZE routine in PLATON.

[b]  $R_1 = \Sigma |F_o| - |F_c| / \Sigma |F_o|$ ,  $wR_2 = [\Sigma w(F_o^2 - F_c^2)^2 / \Sigma wF_o^4]^{1/2}$ ; where  $w = 1/[\sigma^2(F_o^2) + (aP)^2 + bP]$ ,  $P = (F_o^2 + 2F_c^2)/3$ .

**Table S2.** Selected bond lengths [Å] and angles [°] for MOFs **1-4**.

| <b>1</b>            |            |                     |            |
|---------------------|------------|---------------------|------------|
| Zn(1)-O(4)#1        | 1.859(5)   | Zn(1)-O(2)          | 2.121(2)   |
| Zn(1)-N(1)          | 2.031(2)   | Zn(1)-O(1)          | 2.254(2)   |
| Zn(1)-N(2)          | 2.116(2)   |                     |            |
|                     |            |                     |            |
| O(4)#1-Zn(1)-N(1)   | 127.2(3)   | N(2)-Zn(1)-O(2)     | 97.63(9)   |
| O(4)#1-Zn(1)-N(2)   | 90.9(2)    | O(4)#1-Zn(1)-O(1)   | 98.6(2)    |
| N(1)-Zn(1)-N(2)     | 100.09(9)  | N(1)-Zn(1)-O(1)     | 90.93(9)   |
| O(4)#1-Zn(1)-O(2)   | 127.3(3)   | N(2)-Zn(1)-O(1)     | 156.86(10) |
| N(1)-Zn(1)-O(2)     | 102.40(9)  | O(2)-Zn(1)-O(1)     | 59.92(8)   |
| <b>2</b>            |            |                     |            |
| Zn(1)-O(1)#1        | 1.9384(16) | Zn(1)-N(1)#1        | 2.0505(19) |
| Zn(1)-O(1)          | 1.9384(16) | Zn(1)-N(1)          | 2.0505(19) |
|                     |            |                     |            |
| O(1)#1-Zn(1)-O(1)   | 100.81(11) | O(1)-Zn(1)-N(1)     | 124.28(8)  |
| O(1)-Zn(1)-N(1)#1   | 104.13(8)  | N(1)#1-Zn(1)-N(1)   | 101.41(11) |
| O(1)#1-Zn(1)-N(1)   | 104.13(8)  | O(1)#1-Zn(1)-N(1)#1 | 124.28(8)  |
| <b>3</b>            |            |                     |            |
| Cd(1)-O(2)          | 2.223(2)   | Cd(1)-N(2)#3        | 2.372(2)   |
| Cd(1)-O(1)#1        | 2.273(2)   | Cd(1)-O(3)#2        | 2.391(2)   |
| Cd(1)-N(1)          | 2.366(2)   | Cd(1)-O(4)#2        | 2.366(2)   |
|                     |            |                     |            |
| O(2)-Cd(1)-O(1)#1   | 102.35(9)  | N(1)-Cd(1)-N(2)#3   | 166.56(11) |
| O(2)-Cd(1)-N(1)     | 104.81(10) | O(2)-Cd(1)-O(3)#2   | 150.02(9)  |
| O(1)#1-Cd(1)-N(1)   | 90.60(9)   | O(1)#1-Cd(1)-O(3)#2 | 105.39(9)  |
| O(2)-Cd(1)-N(2)#3   | 88.27(10)  | N(1)-Cd(1)-O(3)#2   | 86.14(10)  |
| O(1)#1-Cd(1)-N(2)#3 | 83.46(8)   | N(2)#3-Cd(1)-O(3)#2 | 83.84(9)   |

|                     |          |                     |           |
|---------------------|----------|---------------------|-----------|
| O(2)-Cd(1)-O(4)#2   | 96.51(9) | O(1)#1-Cd(1)-O(4)#2 | 160.00(9) |
| O(4)#2-Cd(1)-N(2)#3 | 90.58(9) | O(4)#2-Cd(1)-O(3)#2 | 54.85(9)  |
| O(4)#2-Cd(1)-N(1)   | 90.97(9) |                     |           |

**4**

|                      |          |                      |           |
|----------------------|----------|----------------------|-----------|
| Cd(1)-O(1)#1         | 2.281(2) | Cd(1)-N(1)           | 2.340(2)  |
| Cd(1)-O(1)           | 2.281(2) | Cd(1)-O(1W)#1        | 2.356(3)  |
| Cd(1)-N(1)#1         | 2.340(2) | Cd(1)-O(1W)          | 2.356(3)  |
|                      |          |                      |           |
| O(1)#1-Cd(1)-O(1)    | 180.0    | N(1)#1-Cd(1)-O(1W)#1 | 84.69(9)  |
| O(1)#1-Cd(1)-N(1)#1  | 89.47(9) | N(1)-Cd(1)-O(1W)#1   | 95.31(9)  |
| O(1)-Cd(1)-N(1)#1    | 90.53(9) | O(1)#1-Cd(1)-O(1W)   | 89.34(8)  |
| O(1)#1-Cd(1)-N(1)    | 90.53(9) | O(1)-Cd(1)-O(1W)     | 90.66(8)  |
| O(1)-Cd(1)-N(1)      | 89.47(9) | N(1)#1-Cd(1)-O(1W)   | 95.31(9)  |
| N(1)#1-Cd(1)-N(1)    | 180.0    | N(1)-Cd(1)-O(1W)     | 84.69(9)  |
| O(1)#1-Cd(1)-O(1W)#1 | 90.65(8) | O(1W)#1-Cd(1)-O(1W)  | 180.00(9) |
| O(1)-Cd(1)-O(1W)#1   | 89.34(8) |                      |           |

Symmetry codes: #1:  $x, y+1, z+1$ ; #2:  $-x+3, -y+2, -z+2$ ; #3:  $-x+1, -y, -z+1$ ; #4:  $x, y-1, z-1$  for **1**; #1:  $-x, y, -z+1/2$ ; #2:  $-x+1/2, -y+3/2, -z+1$ ; #3:  $-x, -y-1, -z$  for **2**; #1:  $-x+1, -y+2, -z+2$ ; #2:  $x, y, z+1$ ; #3:  $x-1, y+1, z$ ; #4:  $x, y, z-1$ ; #5:  $x+1, y-1, z$  for **3**; #1:  $-x+1, -y+1, -z+1$ ; #2:  $-x, -y+2, -z+2$ ; #3:  $-x+2, -y+2, -z+1$  for **4**.