

Supporting Information

Effect of Holding Temperature on Growth of Ruby Crystal
Films via Molybdenum Trioxide Flux Evaporation–
Solubility of Aluminum Oxide, Growth Rate, and Material
Balance

Shunsuke Ayuzawa^{†,‡}, Sayaka Suzuki[§], Miki Hidaka[§], Shuji Oishi^{‡,§}, Katsuya Teshima^{,‡,§,||}*

[†]Department of Science and Technology, Graduate School of Medicine, Science and Technology, Shinshu University, 4-17-1 Wakasato, Nagano 380-8553, Japan

[‡]Nagano Prefecture Nanshin Institute of Technology, 8304-190 Minamiminowa, Nagano 399-4511, Japan

[§]Department of Materials Chemistry, Faculty of Engineering, Shinshu University, 4-17-1 Wakasato, Nagano 380-8553, Japan

^{||} Research Initiative for Supra-Materials, Shinshu University, 4-17-1 Wakasato, Nagano 380-8553, Japan

*Corresponding Author:

Katsuya Teshima

E-mail: teshima@shinshu-u.ac.jp

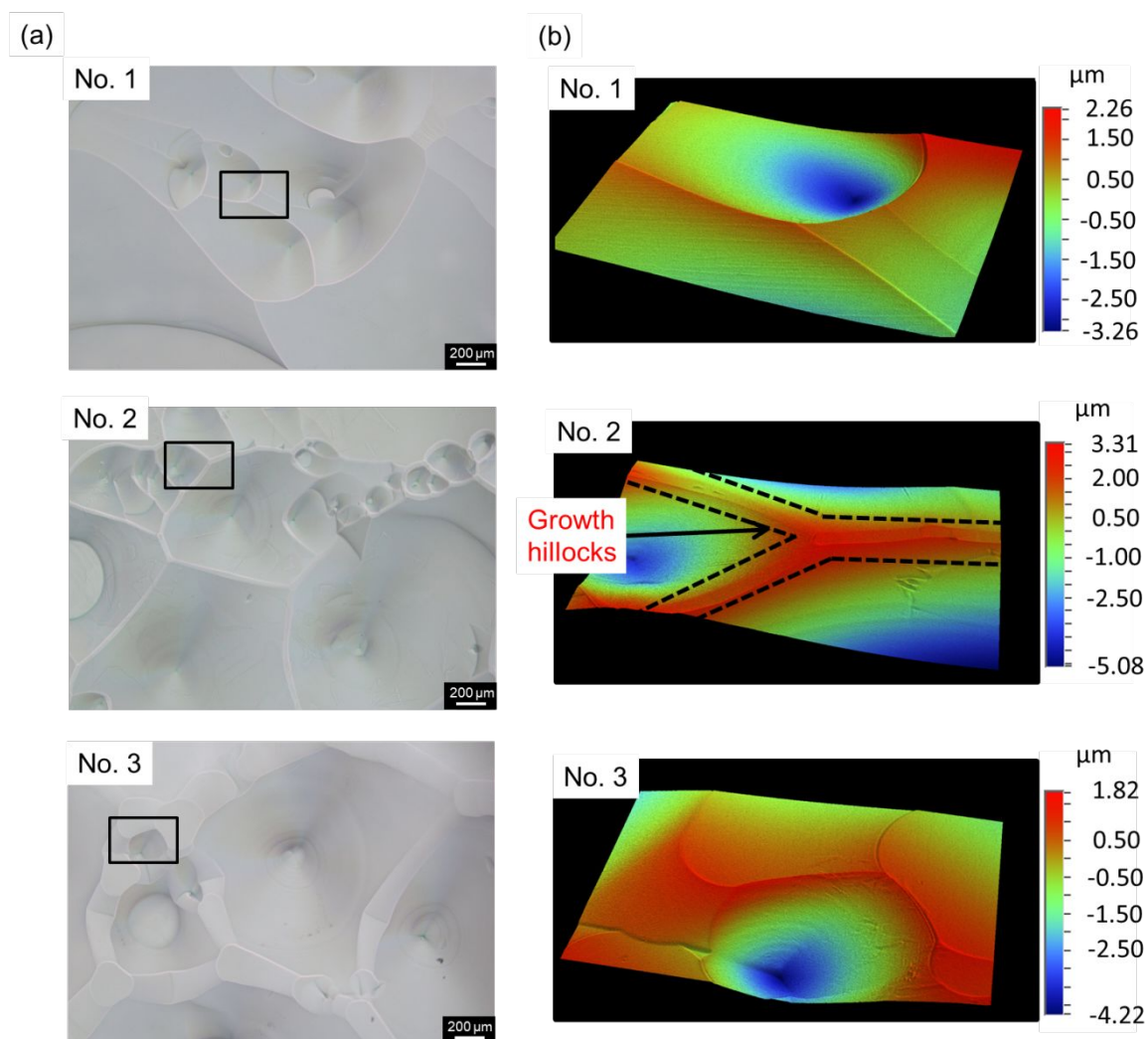


Figure S1. Photographs to judge the commencement of ruby crystal growth at 1050 °C. (a) Optical micrographs of the substrate surfaces obtained under the respective conditions. (b) 3D optical micrographs of the corresponding parts shown in (a).

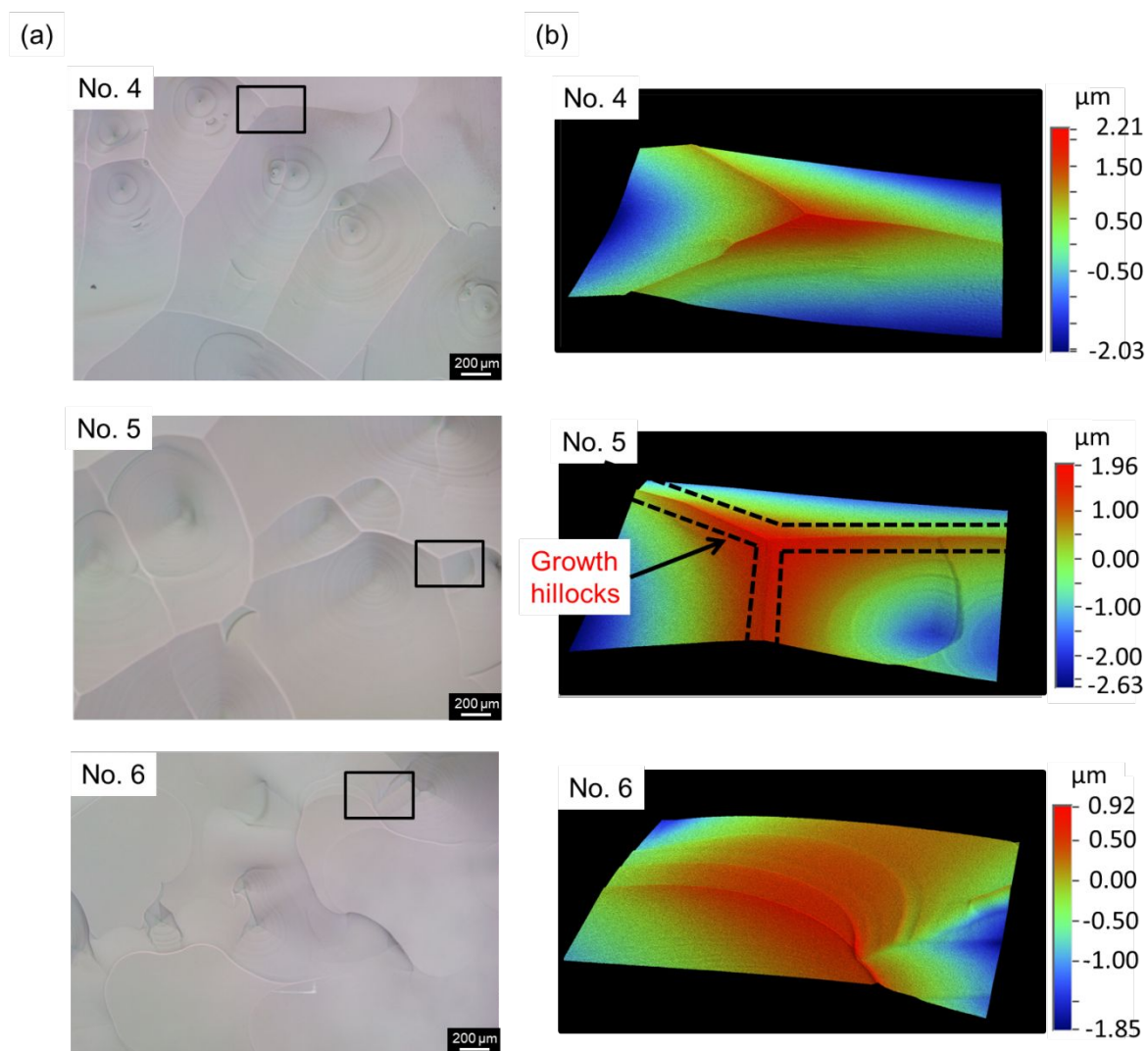


Figure S2. Photographs to judge the commencement of ruby crystal growth at 1150 °C. (a) Optical micrographs of the substrate surfaces obtained under the respective conditions. (b) 3D optical micrographs of the corresponding parts shown in (a).

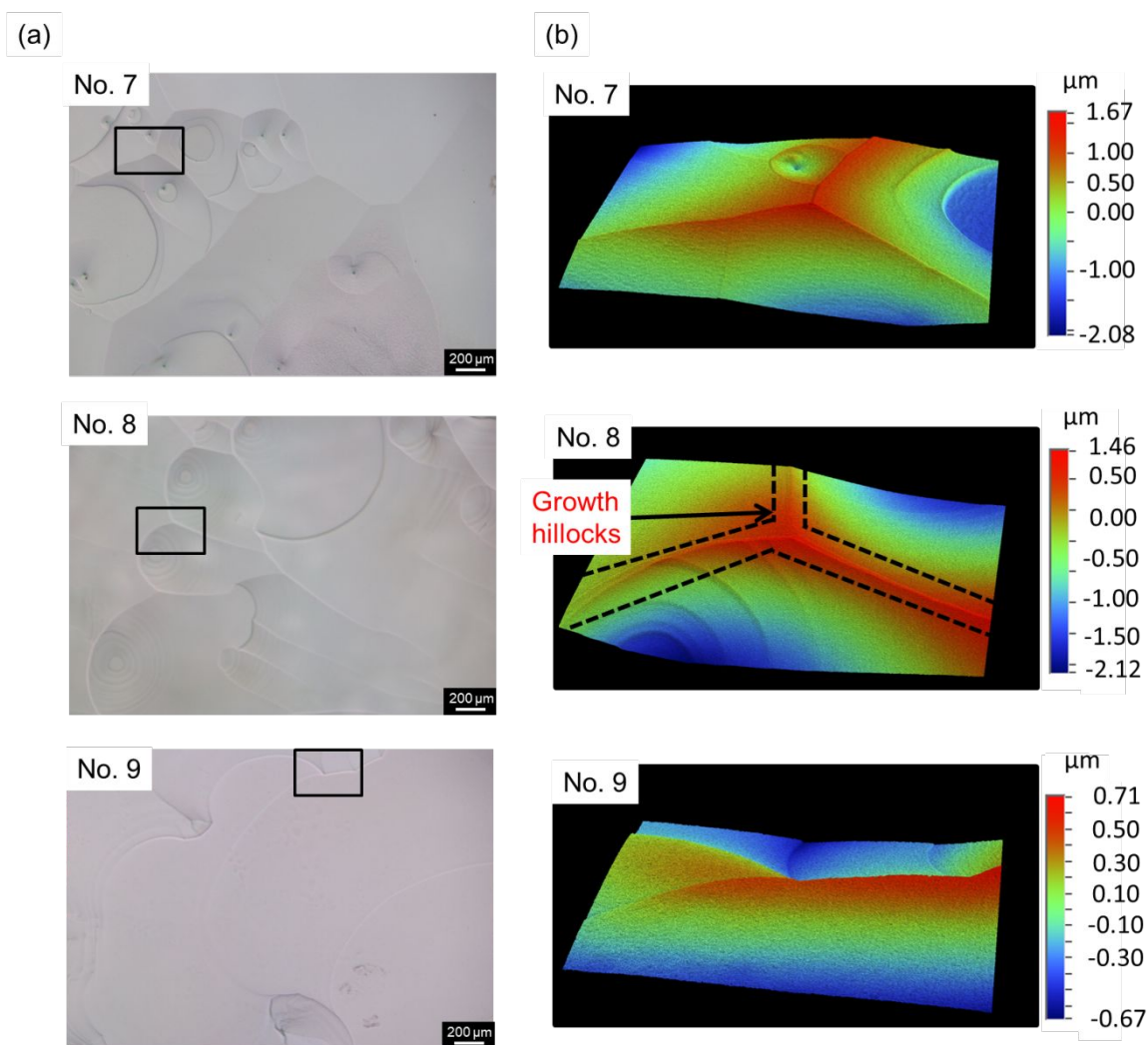


Figure S3. Photographs to judge the commencement of ruby crystal growth at 1200 °C. (a) Optical micrographs of the substrate surfaces obtained under the respective conditions. (b) 3D optical micrographs of the corresponding parts shown in (a).

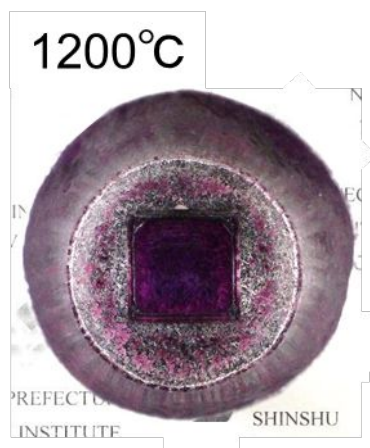


Figure S4. The inside of a crucible after the flux had evaporated completely at 1200 °C.