

**Tunable magnetic exchange between rare-earth metal $5d$
and iron $3d$ states. A case study of the multiple magnetic
transitions in Gd_6FeBi_2 and the solid solutions
 $\text{Dy}_{6-x}\text{Gd}_x\text{FeBi}_2$ ($1 \leq x \leq 5$) with Curie temperatures in the
range 120–350 K**

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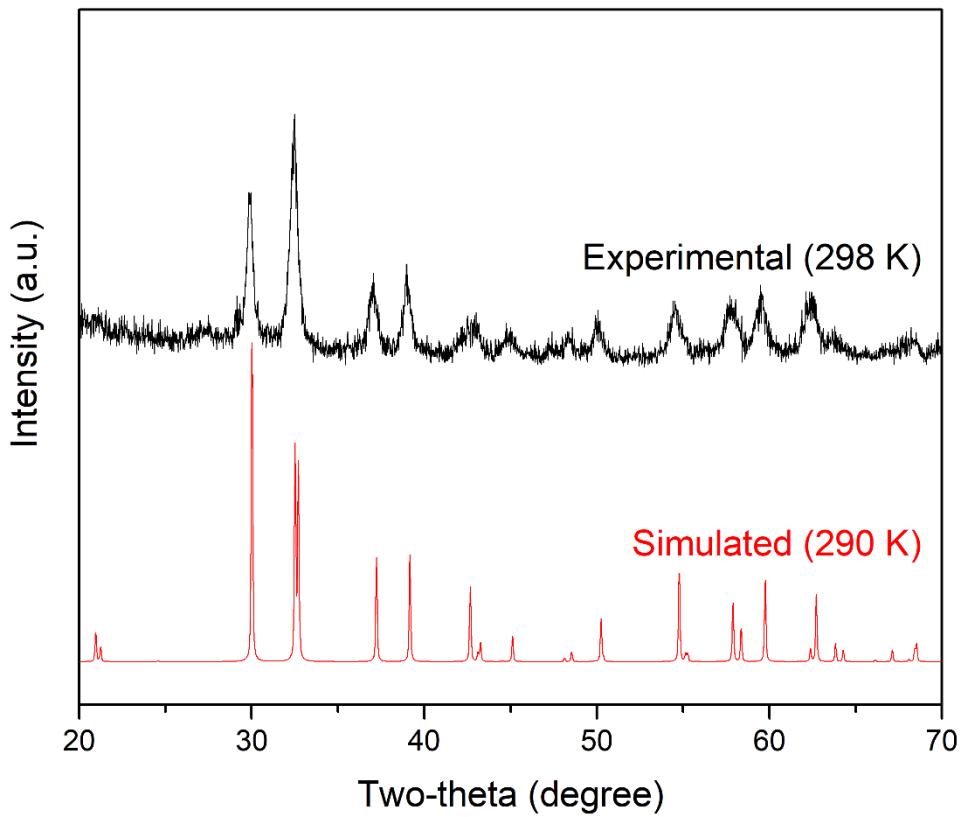


Figure S1. Experimental and simulated PXRD patterns of Gd_6FeBi_2 . The slight peak shifts of experimental pattern towards low angle are due to the high temperature compared with the simulated pattern.

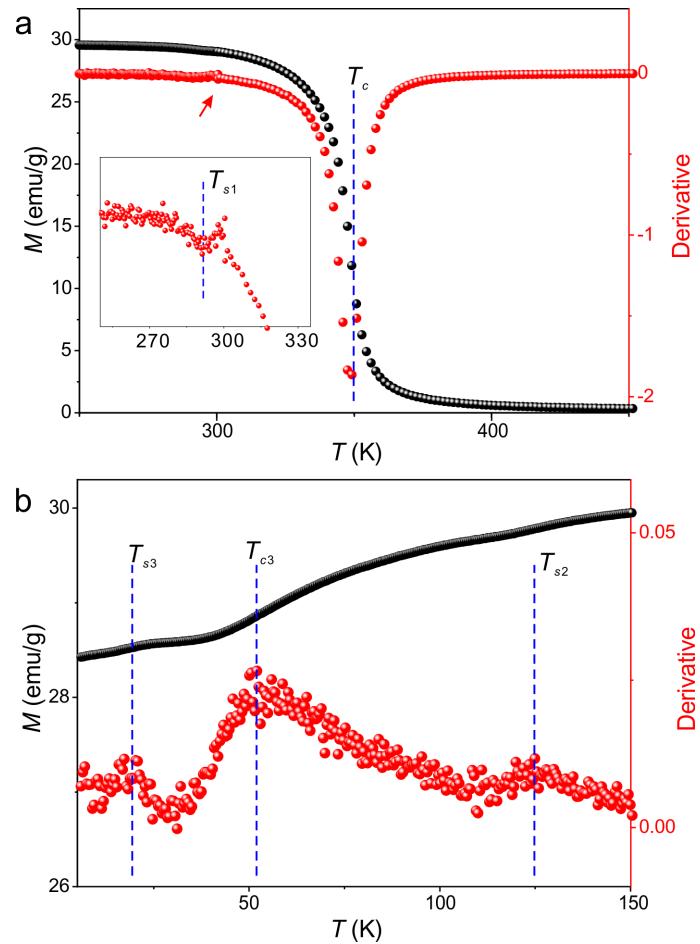


Figure S2. Temperature dependent magnetization M (a) and its derivative (b) for Gd_6FeBi_2 . In (a), the arrows mark the transitions and the insets show the determination of T_{c2} and T_{c3} . The blue dash lines in (b) label the transitions determined by peak values of the derivative.

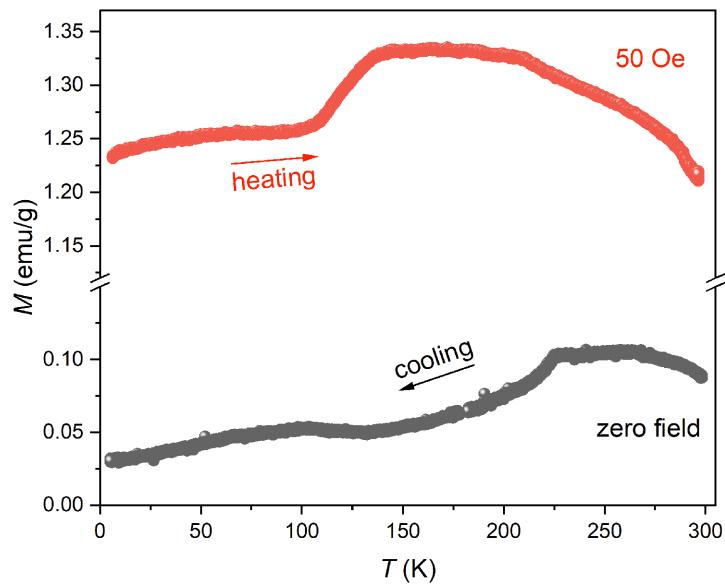


Figure S3. Temperature dependent magnetization (M) of Gd_6FeBi_2 in zero applied field (black), and under an applied field of 50 Oe (red).

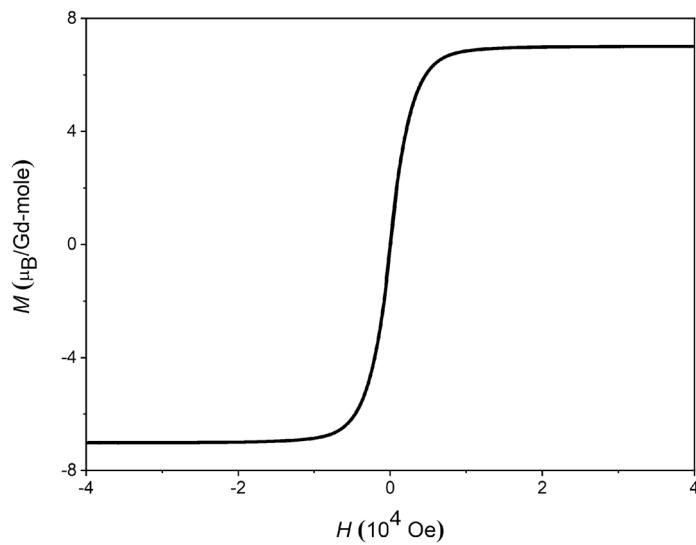


Figure S4. Field dependent magnetization (M) of Gd_6FeBi_2 at 5 K.

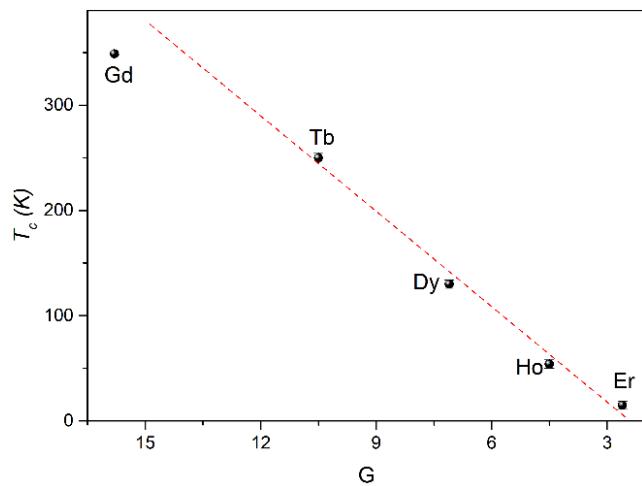


Figure S5. Curie temperatures of the RE_6FeBi_2 compounds, plotted as a function of the corresponding de Gennes factors.

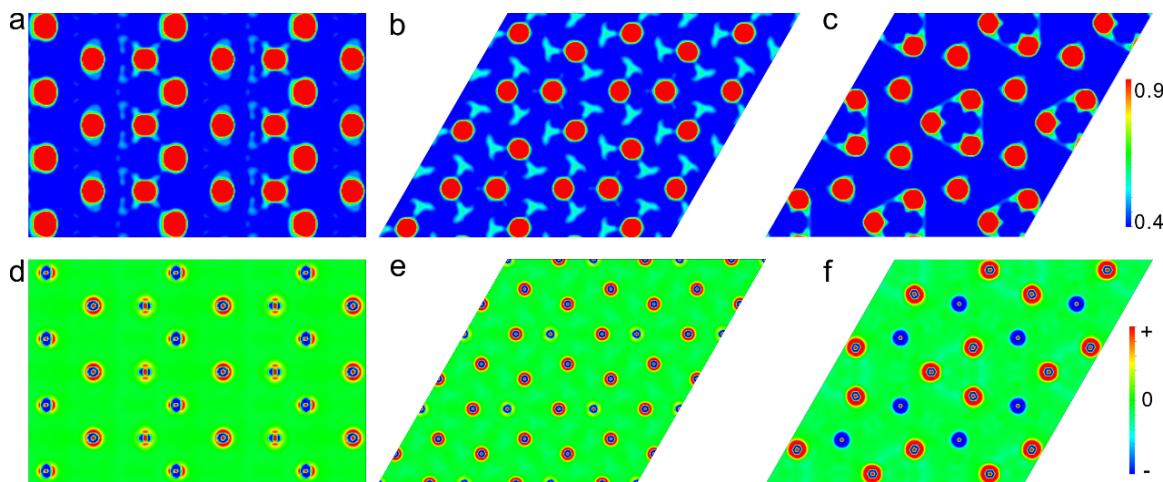


Figure S6. MEM electron density (a-c) and difference electron density maps (d-f) projected on lattice planes shown in Figure 5 (a, d for $x=0$; b, e for $z=0$; c, f for $z=0.5$). The unit in the legend for electron density is e/a_0^3 where a_0 is Bohr radius.

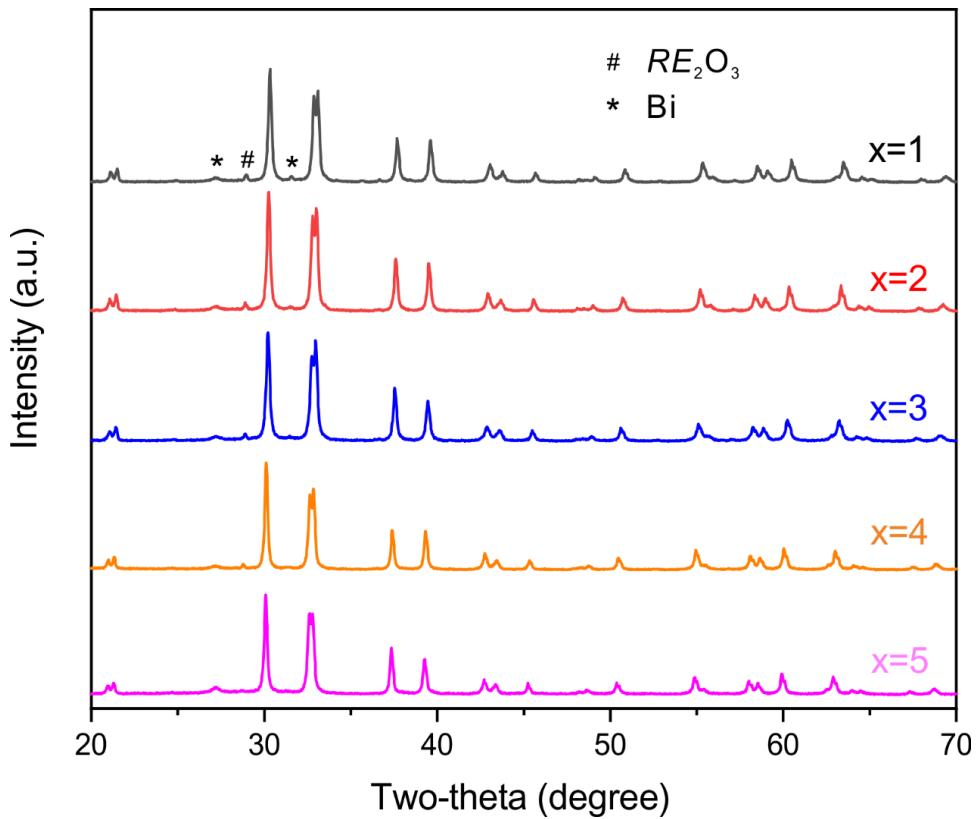


Figure S7. PXRD patterns of solid solutions $\text{Dy}_{6-x}\text{Gd}_x\text{FeBi}_2$ ($1 \leq x \leq 5$). The prepared samples appear highly-crystalline and homogenous, with only tiny impurities of $RE_2\text{O}_3$ and Bi observed (labelled), and other peaks belong to the Zr₆AlCo₂-type structure.

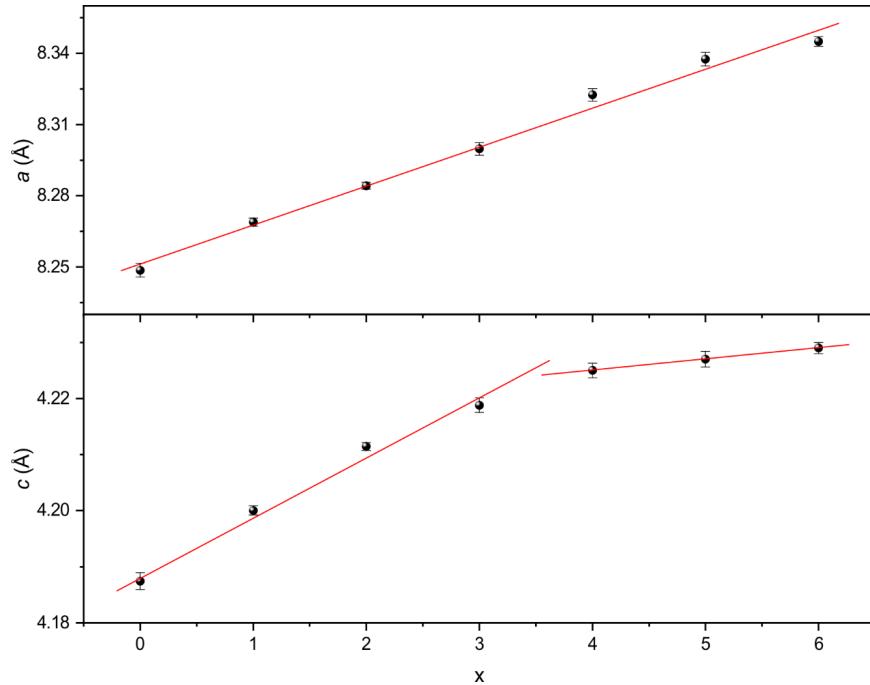


Figure S8. Variation of lattice parameters against Gd concentration in $\text{Dy}_{6-x}\text{Gd}_x\text{FeBi}_2$ ($1 \leq x \leq 5$) compounds. The solid lines show the linear data fitting of lattice parameters against Gd concentration.

Table S1. Selected Crystallographic Data for Gd₆FeBi₂ at different temperatures.

Temperature (K)	135	150	170	190	230	250	270
Radiation	Mo K α , $\lambda = 0.71073 \text{ \AA}$						
Space group	$P\bar{6}2m$ (No. 189), $Z = 1$						
a (\AA)	8.3343(16)	8.337(4)	8.3474(17)	8.349(2)	8.3512(14)	8.3561(16)	8.359(5)
c (\AA)	4.2279(16)	4.229(2)	4.2339(9)	4.2342(12)	4.2343(8)	4.2354(9)	4.231(3)
V (\AA^3)	254.33(12)	254.5(2)	255.48(9)	255.58(12)	255.75(8)	256.11(11)	256.0(3)
ρ_{cal} (g cm^{-3})	9.254	9.246	9.210	9.208	9.202	9.189	9.192
μ (cm^{-1})	742.70	742.10	739.15	739.06	738.58	737.53	737.76
Final R ₁ [$I > 2\sigma(I)$]	0.0260	0.0234	0.0249	0.0215	0.0203	0.0218	0.0249
Final wR ₂ [$I > 2\sigma(I)$]	0.0430	0.0392	0.0406	0.0372	0.0343	0.0433	0.0429

$R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$; $wR_2 = [\sum [w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)^2]]^{1/2}$, where $w = 1/[\alpha^2 F_o^2 + (AP)^2 + (BP)]$, and $P = (F_o^2 + 2F_c^2)/3$; A, B are the respective weight coefficients (see CIF).

Table S2. Atomic Coordinates and Equivalent Isotropic Displacement Parameters (U_{eq}) of Gd_6FeBi_2 at 210 K and 290 K.

Atom	site	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{eq}} (\text{\AA}^2)$
210 K					
Bi1	<i>2d</i>	2/3	1/3	1/2	0.0054(4)
Gd1	<i>3g</i>	0.2382(2)	0	1/2	0.0072(4)
Gd2	<i>3f</i>	0.5984(2)	0	0	0.0061(4)
Fe1	<i>1a</i>	0	0	0	0.0124(19)
290 K					
Bi1	<i>2d</i>	2/3	1/3	1/2	0.0111(3)
Gd1	<i>3g</i>	0.2375(2)	0	1/2	0.0135(4)
Gd2	<i>3f</i>	0.5986(2)	0	0	0.0131(4)
Fe1	<i>1a</i>	0	0	0	0.0189(15)

Table S3. Atomic displacement parameters (\AA^2) at 120 K, 210 K and 290 K.

Atom	U ¹¹	U ²²	U ³³	U ¹²	U ¹³	U ²³
120 K						
Bi1	0.0058(6)	0.0058(6)	0.0048(8)	0.0029(3)	0	0
Gd1	0.0055(7)	0.0047(9)	0.0112(8)	0.0023(5)	0	0
Gd2	0.0069(7)	0.0064(9)	0.0049(8)	0.0032(4)	0	0
Fe1	0.016(3)	0.016(3)	0.006(4)	0.0078(15)	0	0
210 K						
Bi1	0.0073(7)	0.0073(7)	0.0079(9)	0.0036(3)	0	0
Gd1	0.0081(9)	0.0083(11)	0.0142(10)	0.0042(5)	0	0
Gd2	0.0108(9)	0.0086(10)	0.0072(9)	0.0043(5)	0	0
Fe1	0.017(3)	0.017(3)	0.005(4)	0.0084(16)	0	0
290 K						
Bi1	0.0108(5)	0.0108(5)	0.0117(6)	0.0054(2)	0	0
Gd1	0.0099(6)	0.0099(6)	0.0214(7)	0.0055(7)	0	0
Gd2	0.0137(6)	0.0137(6)	0.0127(6)	0.0075(6)	0	0
Fe1	0.024(2)	0.024(2)	0.009(3)	0.0119(12)	0	0

Table S4. Selected bond lengths at 210 K and 290 K.

210 K			
Bi1–Gd1 ($\times 3$)	3.2614(8)	Gd1–Gd1 ($\times 2$)	3.441(2)
Bi1–Gd2 ($\times 6$)	3.3156(5)	Gd1–Gd2 ($\times 4$)	3.6117(10)
Fe1–Gd1 ($\times 6$)	2.9051(8)	Gd1–Gd2 ($\times 2$)	3.6860(15)
Fe1–Gd2 ($\times 3$)	3.3582(14)		
290 K			
Bi1–Gd1 ($\times 3$)	3.2645(11)	Gd1–Gd1 ($\times 2$)	3.441(2)
Bi1–Gd2 ($\times 6$)	3.3181(7)	Gd1–Gd2 ($\times 4$)	3.6111(13)
Fe1–Gd1 ($\times 6$)	2.9054(11)	Gd1–Gd2 ($\times 2$)	3.6921(19)
Fe1–Gd2 ($\times 3$)	3.3568(17)		

Table S5. Hirshfeld analysis from DFT calculations.

Atom	Hirshfeld charge (e)	Spin ($\hbar/2$)
Fe	-0.24	-2.03
Gd1	-0.01	7.96
Gd2	0.05	7.84
Bi	-0.08	0.11

Table S6. Bond population analysis from DFT calculations.

Bond	Population
Fe–Gd1	0.62
Fe–Gd2	0.35
Bi–Gd1	-1.15
Bi–Gd2	-1.08