Structure-activity relationship-based optimization of small temporin-SHf analogs with potent antibacterial activity

Sonia André[†], Shannon K. Washington[§], Emily Darby[§], Marvin M. Vega[§], Ari D. Filip[§], Nathaniel S. Ash[§], Katy A. Muzikar[§], Christophe Piesse[&], Thierry Foulon^{†,‡}, Daniel J. O'Leary^{§,I,*} and Ali Ladram^{†,‡,I}

[†]Sorbonne Universités, UPMC Univ Paris 06, FR 3631, Institut de Biologie Paris Seine (IBPS), Biogenèse des Signaux Peptidiques (BIOSIPE), F-75005, Paris, France UPMC. [‡]CNRS, FR 3631, IBPS, BIOSIPE, F-75005, Paris, France UPMC.

[§] Department of Chemistry, Pomona College, 645 North College Avenue, Claremont, CA 91711, California, USA.

[&]Sorbonne Universités, UPMC Univ Paris 06, FR 3631, Institut de Biologie Paris Seine (IBPS), Plate-forme Ingénierie des Protéines et Synthèse Peptidique, F-75005 Paris, France.

Supporting Information

1.	Methods	S2-S10
2.	References	.S10-S11
3.	LC traces of synthetic peptides	.S12-S17
4.	400 MHz ¹ H NMR shift assignments	.S18-S29
5.	400 MHz ¹ H 2D-TOCSY and 2D-ROESY data	S30-S53

METHODS

Peptide synthesis. Temporin-SHf, [S²]SHf, [R⁵]SHf and [R⁵]SHf-CO₂H were synthesized using solid-phase standard FastMoc chemistry procedure on an Applied Biosystems 433A automated peptide synthesizer, as previously described⁴² (Peptide Synthesis Platform, FR 3631 UPMC-CNRS, IBPS, Paris, France). Peptides were purified by RP-HPLC using the same previously described procedure.¹ D-SHf, Retro-SHf and RI-SHf were purchased from RS Synthesis (Louisville, KY, USA).

Temporin analogs [HmS⁵]SHf (Ser⁵ replaced with an achiral hydroxymethyl serine residue), $\left[\alpha MeF^3\right]SHf$, $\left[p^{-t}BuF^1\right]SHf$, $\left[p^{-t}BuF^2\right]SHf$, $\left[p^{-t}BuF^3\right]SHf$, $\left[p^{-t}BuF^1\right]SHf$, $\left[p^{-t}BuF^1\right]SHf$, $\left[p^{-t}BuF^2\right]SHf$, $\left[p^{-t}BuF^2\right$ ^tBuF², R⁵]SHf and [p-^tBuF³, R⁵]SHf were synthesized manually using Fmoc-based SPPS protocols on Knorr Resin. Fmoc-a-Me-L-Phe-OH, Fmoc-L-4-tert-butyl-Phe-OH, and Fmoc-Arg-(Pbf)-OH were purchased from PepTech. Fmoc-L-Ile-OH, Fmoc-L-Leu-OH, Fmoc-L-Phe-OH, and Fmoc-O-tert-butyl-L-Ser-OH were purchased from Chemimpex. The N-Fmoc isopropylidene-protected variant of α -hydroxymethylserine (Fmoc-HmS(Ipr)-OH) was prepared according to literature procedures.²⁻⁵ Couplings involving C-H_□ amino acids used the appropriate Fmoc protected amino acid (6 equiv relative to resin) with HBTU as the activating agent (6 equiv) and DIEA (12 equiv) in DMF at room temperature. These couplings were monitored using the Kaiser test, and replicate couplings were employed until this test was negative. Couplings with or to resin-bound α -Me-Phe or HmS(Ipr) residues used the appropriate Fmoc protected amino acid (3 equiv relative to resin) with PyAOP/HOAT as the activating agent (3 equiv) and DIEA (8 equiv) in DMF at room temperature (HmS(Ipr)) or 85 °C (α -Me-Phe). Couplings involving resin-bound α -Me-Phe or HmS(Ipr) residues were monitored with the chloranil test, and replicate couplings were used until this test was negative. Fmoc deprotection was accomplished by resin treatment with piperidine in DMF (20% v/v) for 30 min.

Peptides $[\alpha MeF^3]SHf$, $[p-'BuF^1]SHf$, $[p-'BuF^2]SHf$, $[p-'BuF^3]SHf$, $[p-'BuF^1, R^5]SHf$, $[p-'BuF^2, R^5]SHf$ and $[p-'BuF^3, R^5]SHf$ were cleaved from resin by treatment with a freshly prepared solution of TFA (95%)/ H₂O(2.5%)/ triisopropyl silane (2.5%) v/v/v mixture for 2-3 h with mechanical stirring. Filtration, followed by concentration *in vacuo* overnight, afforded the crude peptides as white solids or brown oils.

Peptide [HmS⁵]SHf was cleaved from resin using the same cleavage solution for 3 h. The solvents were removed *in vacuo* and the residue was treated⁴ with acetonitrile (50%)/0.1 M aqueous sodium bicarbonate (50%) v/v mixture for 20 min. The solvents were removed by lyophilization, and the crude peptide was recovered after dissolution in minimal TFA and precipitation in diethyl ether at 4°C. The precipitate was recovered by centrifugation, washed 3 times with cold diethyl ether and dried under a stream of nitrogen. Trace solvents were removed *in vacuo*.

The crude peptides were purified by RP-HPLC on a Zorbax 300SB-C18 semipreparative column (7 μ m, 250 × 21.2 mm) and eluted (5 mL/min) with a 30-70% linear gradient of ACN (0.1% TFA) in 0.1% TFA/water (1% ACN/min), monitored with UV detection at 250 nm.

Finally, the identity of [S²]SHf and [R⁵]SHf-CO₂H was assessed by matrix-assisted laser desorption/ionization-time of flight (MALDI-TOF) mass spectrometry (Voyager DE-PRO Applied Biosystems, Mass Spectrometry and Proteomics Platform, FR 3631 UPMC-CNRS, IBPS, Paris, France), and HRMS ESI data was obtained for D-SHf, Retro-SHf, RI-SHf (Orbitrap XL, Mass Spectrometry and Proteomics Platform, FR 3631 UPMC-CNRS, IBPS, Paris, France) and the remaining peptides (JEOL JMS-600H, California Institute of Technology, Pasadena, CA). The homogeneity of all synthetic peptides was assessed by analytical RP-HPLC using an Aeris PEPTIDE XB-C18 column (3.6 μ m, 250 x 4.6 mm, Phenomenex) (peptide purity \geq 95%).

SHf: MS (ESI) $[M+H]^+$ calc 1075.6087, found 1075.6104; $[S^2]SHf$: MS (MALDI-TOF) calc 1015.670, found 1015.572; $[R^5]SHf$: MS (ESI) calc 1144.6778, found 1144.6776; $[R^5]SHf$ -CO₂H: MS (MALDI-TOF) calc 1145.662, found 1145.584; D-SHf: MS (ESI) calc 1075.6087, found 1075.6146; Retro-SHf: MS (ESI) calc 1075.6087, found 1075.6156; RI-SHf: MS (ESI) calc 1075.6087, found 1075.6142; $[HmS^5]SHf$: MS (ESI) calc 1105.6193, found 1105.6219; $[\alpha MeF^3]SHf$: MS (ESI) calc 1089.6244, found 1089.6257; $[p^{-t}BuF^3]SHf$: MS (ESI) calc 1131.6713, found 1131.6717; $[p^{-t}BuF^2]SHf$: MS (ESI) calc 1131.6713, found 1131.6754; $[p^{-t}BuF^1]SHf$: MS (ESI) calc 1131.6713, found 1131.6747; $[p^{-t}BuF^3, R^5]SHf$: MS (ESI) calc 1200.7404, found 1200.7397; $[p^{-t}BuF^2, R^5]SHf$: MS (ESI) calc 1200.7404, found 1200.7422.

Bacterial strains. The following strains were used: *Escherichia coli* ATCC 25922, *Pseudomonas aeruginosa* ATCC 27853, *Acinetobacter baumannii* ATCC 19606, *Klebsiella pneumoniae* ATCC 13883 (Gram-negative bacteria), *Staphylococcus aureus* ATCC 25923, multi-drug resistant *Staphylococcus aureus* ATCC BAA-44, *Streptococcus pyogenes* ATCC 19615, *Listeria ivanovii, Enterococcus faecalis* ATCC 29212 (Grampositive bacteria), *Candida albicans* ATCC 90028, *Candida parapsilosis* ATCC 22019, *Saccharomyces cerevisiae* (yeasts). Antimicrobial assay. Antimicrobial activity was determined by a liquid growth inhibition assay as described previously,⁶ with the following modification. Using 96-well microtitration plates, 50 μ L of the microorganism suspension (10⁶ cfu/mL) resuspended in MH broth (in LB broth for *Enterococcus faecalis* strain and in YPD medium for yeasts) was mixed with 50 μ L of 2-fold serial dilutions of synthetic peptide in sterile Milli-Q H₂O (100-1 μ M, final concentrations). After overnight incubation at 37 °C (30 °C for yeasts), the minimal inhibitory concentration (MIC), i.e. the lowest concentration of peptide that totally inhibited bacterial growth, was obtained by measuring absorbance at 630 nm. MIC values represent the average of three independent experiments, each performed in triplicate with negative (sterile Milli-Q H₂O) and positive (0.7% formaldehyde) controls.

Antimicrobial activities in the presence of salt and serum. MIC values were determined against two reference bacterial strains, the Gram-positive *S. aureus* ATCC 25923 and the Gram-negative *E. coli* ATCC 25922, according to the liquid growth inhibition assay described above and using MH broth containing either 150 mM NaCl (final concentration) or fetal bovine serum (FBS) at various percentages (final percentages: 10, 20 and 30%). MICs of D-SHf, RI-SHf, and retro-SHf (Table 1) were also determined against *E. coli* and *S. aureus*.

Cytotoxicity assay. Hemolytic experiments were performed using erythrocytes obtained from fresh blood samples of Wistar male rats (Charles River Laboratories, France) according to a previously described protocol.⁶ Briefly, synthetic peptides (500-6.25 μ M, final concentrations) were incubated (100 μ L, final volume) with washed rat erythrocytes (2 x 10⁷ cells) in Dulbecco's phosphate-buffered saline (pH 7.4) for 1 h at 37

°C. After centrifugation (12,000 x g, 15 s), the absorbance of the supernatant was measured at 450 nm. The LC_{50} value, which is the average concentration of peptide producing 50% hemolysis, was determined from three independent experiments carried out in triplicate with positive control (100% hemolysis) corresponding to 0.1% triton (v/v). The cytotoxicity was also assessed on THP-1-derived macrophages, obtained from the differentiation of the human leukemia monocyte cell line THP-1 (kindly provided by Dr. B. Oury, IRD, Montpellier, France), and also on the human embryonic kidney cell line HEK-293 (kindly provided by Dr. O. Jean-Jean, IBPS, UPMC, Paris, France). THP-1 monocytes were cultured in RPMI 1640 (Life Technologies) supplemented with 10% FBS, 25 mM HEPES, GlutaMax 1%, 1 mM sodium pyruvate, penicillin (100 U/mL) and streptomycin (100 μ g/mL). Cells were incubated at 37°C in a humidified atmosphere of 5% CO2. HEK-293 cells were cultured in Dulbecco's modified Eagle's medium (DMEM, Life Technologies) supplemented with 10% FBS, penicillin (100 U/mL) and streptomycin (100 μ g/mL), and incubated under the same conditions as THP-1 cells. THP-1 monocytes were differentiated into macrophages by adding 20 ng/mL PMA (phorbol 12-myristate 13-acetate) in the culture medium containing the cells (5 x 10^5 cells/mL). Cells were then seeded in 96-well plates (100 μ L/well) and incubated 48 h at 37°C. After washing (2 times) with RPMI 1640, macrophages were incubated for 24 h at 37°C with 100 µL of peptide (final concentrations: 12.5 to 200 µM) diluted in culture medium. HEK-293 cells were seeded in 96-well plates (100 μ L/well) at a density of 5 x 10⁵ cells/mL and incubated overnight at 37°C. The cells were washed with DMEM and then incubated for 24 h at 37°C with 100 µL of peptide (final concentrations: 12.5 to 200 μ M) diluted in culture medium. Cell viability was quantified after peptide incubation using a MTT (Methylthiazolyldiphenyl-tetrazolium bromide)-based microassay. Briefly, 10 μ L of 5 mg/mL MTT were added to the 96-well plate. After 4 h incubation at 37°C, in the dark, formazan crystals were dissolved with 100 μ L of a solubilization solution (40% dimethylformamide in 2% glacial acetic acid, 16% sodium dodecyl sulfate, pH 4.7), followed by 1 h incubation at 37°C under shaking (150 rpm). Finally, absorbance of formazan was measured at 570 nm (Asys Hitech UVM 340) using a reference wavelength of 630 nm. The 50% inhibitory concentration (IC₅₀) was determined with GraphPad Prism® 5.0 software. Results were expressed as the mean of three independent experiments performed in triplicate.

SYTOX Green uptake assay. Peptide-induced permeabilization of the bacterial cytoplasmic plasma membrane (*S. aureus* ATCC 25923 and *E. coli* ATCC 25922) was determined by fluorometric measurement of SYTOX Green influx.⁷ SG is impermeant to live cells. However, when the cell membrane is damaged, this high-affinity nucleic acid dye penetrates into the cell and binds to intracellular DNA, leading to an intense green fluorescence light. For SG uptake assay, exponentially growing bacteria (6×10^5 cfu/mL) were resuspended in PBS (1 M NaCl, 0.1 M Na₂HPO₄, pH 7.3) after centrifugation (1000 x g, 10 min, 4 °C) and washing steps. 792 µl of the bacterial suspension was preincubated with 8 µL of 100 µM SG during 30 min at 37 °C in the dark. After peptide addition (200 µL, final concentration 2-fold above the MIC), the fluorescence was measured during 1 h at 37 °C in a Varian Cary Eclipse fluorescence spectrophotometer, with excitation and emission wavelengths of 485 and 520 nm, respectively. Three independent experiments were performed and results correspond to a representative experiment with negative (PBS) and positive (melitin) controls.

Membrane Depolarization Assay. The peptide-induced depolarization of the bacterial cytoplasmic membrane (E. coli ATCC 25922 and S. aureus ATCC 25923) was analyzed using the membrane potential-sensitive cyanine dye $DiSC_3(5)$ (3.3'dipropylthiadicarbocyanine iodide). This cationic and lipophilic dye accumulates into the cytoplasmic membrane according to the magnitude of the membrane potential gradient and then aggregates, causing self-quenching of the fluorescence.⁸ Membranepermeabilizing agents generate membrane potential dissipation that provokes the release of $DiSC_3(5)$ into the medium, leading to an increase of fluorescence that can be monitored over time. For the experiment, exponentially growing bacteria were collected by centrifugation (1000 x g, 10 min, 4 °C), washed with PBS, and resuspended in the same buffer to an OD₆₃₀ of 0.1. 700 μ L of bacteria containing 1 μ M DiSC₃(5) were preincubated in the dark during 10 min at 37 °C to allow maximal dye uptake and fluorescence quenching. 100 µL of 1 mM KCl were then added to the mixture in order to equilibrate the cytoplasmic and external K⁺ concentrations. Changes in fluorescence due to the collapse of the cytoplasmic membrane potential were recorded after addition of the peptide (200 µL, final concentration: 2-fold above the MIC) at 37 °C during 20 min at an excitation wavelength of 622 nm and an emission wavelength of 670 nm (Varian Cary Eclipse fluorescence spectrophotometer). Three independent experiments were performed and results correspond to a representative experiment with negative (PBS) and positive (melittin) controls.

Time-Killing Assay. Time-kill kinetics of temporins were studied against the Gramnegative strain *E. coli* ATCC 25922, as described previously.⁹ Briefly, peptides at 2-fold concentration above the MIC were incubated with bacterial suspension (10^6 cfu/mL) resuspended in PBS buffer (10 mM phosphate, 100 mM NaCl, pH 7.3). Aliquots of 10 μ L were withdrawn at different times and spread onto LB agar plates for cell counting after overnight incubation at 37 °C. Controls were run without peptide. Three independent experiments were performed in triplicate. Results were expressed as the mean ± S.E.M. of a representative experiment.

Circular Dichroism Spectroscopy. Circular dichroism (CD) spectra were recorded (Jobin-Yvon CD6 spectropolarimeter) at 25°C in a 0.1 cm quartz cell over a wavelength range of 190 to 260 nm. Spectra were acquired with a spectral bandwidth of 2 nm and a step size of 0.5 nm. Peptides (100 μ M) were solubilized either in H₂O Milli-Q or 80 mM sodium dodecyl sulfate (SDS). Each spectrum was the average of four successive scans and solvent spectra (H₂O, 80 mM SDS) were subtracted, followed by baseline correction and smoothing. CD measurements were reported as the dichroic increment ($\Delta\epsilon$) per residue.

NMR Spectroscopy. Peptides (2 mg) were dissolved in 0.7 ml of a 9:1 ratio of H₂O to D₂O containing sodium dodecyl- d_{25} sulfate (17.5 mg, Cambridge Isotope Laboratories) and a trace amount of 4,4-dimethyl-4-silapentane-1-sulfonic acid (DSS) as a chemical shift calibration standard. Peptide concentrations, when adjusted for MW differences, ranged from 1.8 to 2.3 mM. For consistency with the CD spectroscopy experiments, the pH of the NMR samples was adjusted to 5.3-5.5 using a pH microelectrode (Thermo Scientific Orion 9826BN) and microliter additions of 0.1 M aqueous HCl or 0.1 M NaOH.

400 MHz ¹H NMR spectra were obtained on a 400 MHz NMR spectrometer equipped with a 5 mm dual ${}^{1}\text{H}/{}^{13}\text{C}$ Z-gradient probe whose temperature was maintained

at 40 °C. 1D ¹H spectra were acquired with a gradient-enhanced excitation sculpting water suppression sequence.¹⁰ The same water suppression method was used with 2D double quantum filtered COSY,¹¹ TOCSY,¹² and ROESY¹³ experiments. The TOCSY and ROESY spin lock periods were 60 ms and 100 ms, respectively.

ABBREVIATIONS

ACN, acetonitrile; AMP, antimicrobial peptide; CD, circular dichroism; DiSC3(5), 3,3dipropylthiadicarbocyanine iodide; LB, Luria Bertani; MH, Mueller-Hinton; MIC, minimal inhibitory concentration; NMR, nuclear magnetic resonance; OD, optical density; PBS, phosphate buffered solution; RP-HPLC, reversed-phase high performance liquid chromatography; SDS, sodium dodecyl sulfate; SG, SYTOX green; TFA, trifluoroacetic acid; YPD, Yeast Peptone Dextrose.

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Waters Breeze HPLC system Aeris PEPTIDE XB-C18 analytical column (3.6 µm, 250 x 4.6 mm) Linear gradient of acetonitrile (ACN): 20-80% (2% ACN/min) Flow rate: 1 mL/min Absorbance monitored at 220 (dark) and 260 nm (blue)



Figure S1. Background/blank injection.



Figure S2. Temporin-SHf (3.2 µg) LC trace.



Figure S3. $[S^2]$ SHf (3.0 µg) LC trace.



0,28 100,00 0,26 -90,00 0,24 -80,00 0,22 0,20 -70,00 0,18 60,00 0,16 % Compo: 50,00 ₹ 0,14 0,12 40,00 0,10 -30,00 0,08 0,06 20,00 0,04 -10,00 0,02-0,00 0,00 16,00 18,00 20,00 22,00 24,00 26,00 28,00 30,00 32,00 Minutes 2,00 4,00 6,00 8,00 10,00 12,00 14,00 0,00

Figure S5. $[HmS^5]SHf (3.3 \mu g) LC trace.$





Figure S8. $[p^{-t}BuF^2]SHf (3.4 \mu g) LC$ trace.



Figure S9. $[p^{-t}BuF^{1}]SHf (3.4 \mu g) LC$ trace.



Figure S10. $[p^{-t}BuF^3, R^5]SHf (3.6 \mu g) LC$ trace.



Figure S11. [p-^tBuF², R⁵]SHf (3.6 μ g) LC trace.



Figure S12. $[p^{-t}BuF^1, R^5]SHf (3.6 \ \mu g) LC$ trace.



Figure S13. [R⁵]SHf-CO₂H (3.4 µg) LC trace.



Figure S14. D-SHf (10.7 μ g) LC trace.



Figure S15. RI-SHf (10.7 $\mu g)$ LC trace



Figure S16. Retro-SHf (10.7 μ g) LC trace

Hydrogen	Chemical Shift	Hydrogen	Chemical Shift
	(ppm)		(ppm)
$Phe1 - H\alpha$		Ser5 – NH	7.90
Phe1 – H β 1		$Ser5 - H\alpha$	4.25
$Phe1 - H\beta1$		Ser5 – Hβ1	3.94
Phe2-NH		$Ser5 - H\beta2$	3.94
$Phe2 - H\alpha$		Arg6 – NH	7.72
$Phe2 - H\beta1$		Arg6 – Hα	4.17
$Phe2 - H\beta2$		Arg6 – Hβ1	1.84
Phe3 – NH	7.96	Arg6 – Hβ2	1.84
$Phe3 - H\alpha$	4.33	Arg6 – Ηδ11	3.13
$Phe3 - H\beta1$	3.10	Arg6 – Hδ12	3.13
Phe3 – H β 2	3.10	Ile7 – NH	7.43
Leu4 – NH	7.98	Ile7 – Ha	3.90
$Leu4 - H\alpha$	4.08	Ile7 – Hβ	1.54
$Leu4 - H\beta 1$	1.82	Ile7 – Hγ11	1.28
$Leu4 - H\beta2$	1.66	Ile7 – Hy12	1.00
Leu4 – Hδ11	0.96	Ile7 – H γ 21	0.37
$Leu4 - H\delta 12$	0.96	Ile7 – H γ 22	0.37
Leu4 – Hδ13	0.96	Ile7 – H γ 23	0.37
Leu4 – H δ 21	0.91	Ile7 – Hδ1	0.70
$Leu4 - H\delta 22$	0.91	Ile7 – H δ 2	0.70
$Leu4 - H\delta 23$	0.91	Ile7 – Hδ3	0.70
$Leu4 - H\gamma$	1.66	Phe8 – NH	7.72
		$Phe8 - H\alpha$	4.62
		$Phe8 - H\beta1$	3.30
		$Phe8 - H\beta2$	2.84

Table S1. 400 MHz 1 H Proton Chemical Shifts of SHf in 80 mM SDS- d_{25} (9:1
H₂O/D₂O), 40 °C.

Hydrogen	Chemical Shift	Hydrogen	Chemical Shift
	(ppm)		(ppm)
$Phe1 - H\alpha$		Ser5 – NH	7.95
$Phe1 - H\beta1$		$Ser5 - H\alpha$	4.26
$Phe1 - H\beta1$		$Ser5 - H\beta1$	3.96
Ser 2 – NH		$Ser5 - H\beta2$	3.96
Ser 2 – H α		Arg6 – NH	7.77
Ser 2 – H β 1		Arg6 – Ha	4.19
Ser 2 – H β 2		Arg6 – Hβ1	1.88
Phe3 – NH	7.96	Arg6 – Hβ2	1.88
Phe3 – Ha	4.40	Arg6 – Hδ11	3.18
$Phe3 - H\beta1$	3.18	Arg6 – Hδ12	3.18
$Phe3 - H\beta2$	3.08	Ile7 – NH	7.43
Leu4 – NH	7.96	$Ile7 - H\alpha$	3.89
$Leu4 - H\alpha$	4.05	$Ile7 - H\beta$	1.52
Leu4 – Hβ1	1.90	$Ile7 - H\gamma 11$	1.26
$Leu4 - H\beta2$	1.61	Ile7 – H γ 12	0.98
Leu4 – Hδ11	0.98	Ile7 – H γ 21	0.35
Leu4 – Hδ12	0.98	Ile7 – H γ 22	0.35
Leu4 – Hδ13	0.98	$Ile7 - H\gamma 23$	0.35
Leu4 – Hδ21	0.90	Ile7 – Hδ1	0.68
$Leu4 - H\delta 22$	0.90	Ile7 – Hδ2	0.68
$Leu4 - H\delta 23$	0.90	Ile7 – Hδ3	0.68
$Leu4 - H\gamma$	1.61	Phe8 – NH	7.75
		$Phe8 - H\alpha$	4.64
		$Phe8 - H\beta1$	3.32
		Phe8 – H β 2	2.86

Table S2. 400 MHz 1 H Proton Chemical Shifts of [S 2]SHf in 80 mM SDS- d_{25} (9:1
H $_{2}$ O/D $_{2}$ O), 40 °C.

Hydrogen	Chemical Shift	Hydrogen	Chemical Shift
	(ppm)		(ppm)
$Phe1 - H\alpha$		Arg5 – NH	7.78
$Phe1 - H\beta1$		Arg5 – Hα	4.10
$Phe1 - H\beta1$		Arg5 – Hβ1	1.84
Phe2 – NH		$Arg5 - H\beta2$	1.84
$Phe2 - H\alpha$		Arg5 – Hδ11	
$Phe2 - H\beta1$		Arg5 – Hδ12	
$Phe2 - H\beta2$		Arg6 – NH	7.72
Phe3 – NH	7.77	Arg6 – Hα	4.14
$Phe3 - H\alpha$	4.43	Arg6 – Hβ1	1.81
Phe3 – H β 1	3.09	Arg6 – Hβ2	1.81
$Phe3 - H\beta2$	3.09	Arg6 – Hδ11	3.17
Leu4 – NH	7.88	Arg6 – Hδ12	3.17
$Leu4 - H\alpha$	4.06	Ile7 – NH	7.53
$Leu4 - H\beta 1$	1.81	Ile7 – H α	3.85
$Leu4 - H\beta2$	1.59	Ile7 – H β	1.55
$Leu4 - H\delta 11$	0.98	Ile7 – Hγ11	1.26
$Leu4 - H\delta 12$	0.98	Ile7 – Hy12	1.01
Leu4 – Hδ13	0.98	Ile7 – Hγ21	0.37
$Leu4 - H\delta 21$	0.91	Ile7 – H γ 22	0.37
$Leu4 - H\delta 22$	0.91	Ile7 – Hy23	0.37
$Leu4 - H\delta 23$	0.91	Ile7 – Hδ1	0.68
$Leu4 - H\gamma$	1.69	$Ile7 - H\delta2$	0.68
		Ile7 – Hδ3	0.68
		Phe8 – NH	7.69
		$Phe8 - H\alpha$	4.60
		$Phe8 - H\beta1$	3.29
		$Phe8 - H\beta2$	2.84

Table S3. 400 MHz 1 H Proton Chemical Shifts of [R 5]SHf in 80 mM SDS- d_{25} (9:1 $H_{2}O/D_{2}O$), 40 °C.

Hydrogen	Chemical Shift	Hydrogen	Chemical Shift
	(ppm)		(ppm)
$Phe1 - H\alpha$		Arg5 – NH	7.72
$Phe1 - H\beta1$		Arg5 – Ha	4.23
$Phe1 - H\beta1$		Arg5 – Hβ1	1.74
Phe2 – NH		Arg5 – Hβ2	1.74
$Phe2 - H\alpha$		Arg5 – Hδ11	3.17
$Phe2 - H\beta1$		Arg5 – Hδ12	3.17
$Phe2 - H\beta2$		Arg6 – NH	7.79
Phe3 – NH	7.55	Arg6 – Hα	4.28
$Phe3 - H\alpha$	4.52	Arg6 – Hβ1	1.79
$Phe3 - H\beta1$	3.07	Arg6 – Hβ2	1.79
$Phe3 - H\beta2$	3.07	Arg6 – Ηδ11	3.17
Leu4 – NH	7.70	Arg6 – Hδ12	3.17
$Leu4 - H\alpha$	4.14	Ile7 – NH	7.59
$Leu4 - H\beta 1$	1.71	Ile7 – Ha	4.03
$Leu4 - H\beta2$	1.60	Ile7 – H β	1.70
Leu4 – Hδ11	0.96	Ile7 – Hγ11	
Leu4 – Hδ12	0.96	Ile7 – Hy12	1.04
Leu4 – Hδ13	0.96	Ile7 – Hy21	0.58
$Leu4 - H\delta 21$	0.89	Ile7 – H γ 22	0.58
$Leu4 - H\delta 22$	0.89	Ile7 – H γ 23	0.58
$Leu4 - H\delta 23$	0.89	Ile7 – H δ 1	0.73
Leu4 – Hy	1.60	Ile7 – Hδ2	0.73
		Ile7 – Hδ3	0.73
		Phe8-NH	7.54
		$Phe8 - H\alpha$	4.54
		$Phe8 - H\beta1$	3.23
		$Phe8 - H\beta2$	2.89

Table S4. 400 MHz 1 H Proton Chemical Shifts of [R 5]SHf-CO₂H in 80 mM SDS- d_{25} (9:1 H₂O/D₂O), 40 °C.

Hydrogen	Chemical Shift	Hydrogen	Chemical Shift
	(ppm)		(ppm)
$Phe1 - H\alpha$		Ser5 – NH	
$Phe1 - H\beta1$			
$Phe1 - H\beta1$		$Ser5 - H\beta 1/H\beta 2$	
Phe2 – NH		Ser5 – H β 1'/H β 2'	
Phe2 – Ha		Arg6 – NH	7.81
Phe2 – H β 1		Arg6 – Hα	4.11
$Phe2 - H\beta2$		Arg6 – Hβ1	1.85
Phe3 – NH	7.68	Arg6 – Hβ2	1.85
$Phe3 - H\alpha$	4.34	Arg6 – Hδ11	3.15
$Phe3 - H\beta1$	3.08	Arg6 – Hδ12	3.15
Phe3 – H β 2	3.08	Ile7 – NH	7.56
Leu4 – NH	8.00	$Ile7 - H\alpha$	3.89
$Leu4 - H\alpha$	4.05	$Ile7 - H\beta$	1.54
$Leu4 - H\beta 1$	1.88	Ile7 – Hγ11	1.29
$Leu4 - H\beta2$	1.65	Ile7 – Hy12	1.01
Leu4 – Hδ11	0.97	Ile7 – Hy21	0.36
Leu4 – Hδ12	0.97	Ile7 – H γ 22	0.36
Leu4 – Hδ13	0.97	Ile7 – H γ 23	0.36
$Leu4 - H\delta 21$	0.90	Ile7 – Hδ1	0.70
$Leu4 - H\delta 22$	0.90	Ile7 – H δ 2	0.70
$Leu4 - H\delta 23$	0.90	Ile7 – Hδ3	0.70
Leu4 – Hy	1.65	Phe8-NH	7.60
		$Phe8 - H\alpha$	4.65
		$Phe8 - H\beta1$	3.36
		$Phe8 - H\beta2$	2.76

Table S5. 400 MHz 1 H Proton Chemical Shifts of [HmS 5]SHf in 80 mM SDS- d_{25} (9:1H₂O/D₂O), 40 °C.

Hydrogen	Chemical Shift	Hydrogen	Chemical Shift
	(ppm)		(ppm)
Phe1 – Ha		Ser5 – NH	8.00
Phe1 – Hβ1		$Ser5 - H\alpha$	4.26
Phe1 – H β 1		$Ser5 - H\beta1$	3.96
Phe2-NH		$Ser5 - H\beta2$	3.96
$Phe2 - H\alpha$		Arg6 – NH	7.61
$Phe2 - H\beta1$		Arg6 – Hα	4.17
$Phe2 - H\beta2$		Arg6 – Hβ1	1.75
Phe3 – NH		Arg6 – Hβ2	1.75
Phe3 – Mea	1.21	Arg6 – Hδ11	3.12
Phe3 – H β 1	2.93	Arg6 – Hδ12	3.12
Phe3 – H β 2	3.79	Ile7 – NH	7.30
Leu4 – NH	7.94	Ile7 – Ha	3.83
$Leu4 - H\alpha$	3.97	Ile7 – H β	1.49
Leu4 – Hβ1	1.79	Ile7 – Hγ11	1.21
$Leu4 - H\beta2$	1.66	Ile7 – Hy12	0.94
Leu4 – Hδ11	1.03	Ile7 – Hy21	0.30
Leu4 – Hδ12	1.03	Ile7 – H γ 22	0.30
Leu4 – Hδ13	1.03	Ile7 – H γ 23	0.30
Leu4 – Hδ21	1.01	Ile7 – H δ 1	0.59
$Leu4 - H\delta 22$	1.01	Ile7 – Hδ2	0.59
$Leu4 - H\delta 23$	1.01	Ile7 – Hδ3	0.59
$Leu4 - H\gamma$	1.66	Phe8 – NH	7.75
		$Phe8 - H\alpha$	4.63
		$Phe8 - H\beta1$	3.32
		$Phe8 - H\beta2$	2.86

Table S6. 400 MHz ¹H Proton Chemical Shifts of $[\alpha MeF^3]$ SHf in 80 mM SDS- d_{25} (9:1 H₂O/D₂O), 40 °C.

Hydrogen	Chemical Shift	Hydrogen	Chemical Shift
	(ppm)		(ppm)
Phe1 – Ha		Ser5 – NH	7.88
Phe1 – Hβ1		$Ser5 - H\alpha$	4.27
Phe1 – H β 1		$Ser5 - H\beta1$	3.94
Phe2-NH		$Ser5 - H\beta2$	3.94
$Phe2 - H\alpha$		Arg6 – NH	7.72
$Phe2 - H\beta1$		Arg6 – Hα	4.18
$Phe2 - H\beta2$		Arg6 – Hβ1	1.85
Phe3 – NH	7.77	Arg6 – Hβ2	1.85
Phe3 – Ha	4.31	Arg6 – Ηδ11	3.15
Phe3 – H β 1	3.08	Arg6 – Hδ12	3.15
$Phe3 - H\beta2$	3.08	Ile7 – NH	7.44
$Phe3 - p-^{t}Bu$	1.18	Ile7 – Ha	3.90
Leu4 – NH	7.67	Ile7 – H β	1.55
Leu4 – Ha	4.07	Ile7 – Hγ11	1.34
Leu4 – Hβ1	1.79	Ile7 – Hy12	1.03
$Leu4 - H\beta2$	1.56	Ile7 – Hy21	0.37
$Leu4 - H\delta11$	0.95	Ile7 – H γ 22	0.37
$Leu4 - H\delta 12$	0.95	Ile7 – H γ 23	0.37
$Leu4 - H\delta 13$	0.95	Ile7 – H δ 1	0.73
$Leu4 - H\delta 21$	0.90	Ile7 – Hδ2	0.73
$Leu4 - H\delta 22$	0.90	Ile7 – Hδ3	0.73
$Leu4 - H\delta 23$	0.90	Phe8 – NH	7.74
$Leu4 - H\gamma$	1.56	$Phe8 - H\alpha$	4.64
		$Phe8 - H\beta1$	3.32
		$Phe8 - H\beta2$	2.86

Table S7. 400 MHz ¹H Proton Chemical Shifts of $[p^{-t}BuF^3]$ SHf in 80 mM SDS- d_{25} (9:1 H₂O/D₂O), 40 °C.

Hydrogen	Chemical Shift	Hydrogen	Chemical Shift
	(ppm)		(ppm)
$Phe1 - H\alpha$		Ser5 – NH	7.94
Phe1 – Hβ1		$Ser5 - H\alpha$	4.28
Phe1 – Hβ1		Ser5 – Hβ1	3.94
Phe2-NH		$Ser5 - H\beta2$	3.94
$Phe2 - H\alpha$		Arg6 – NH	7.76
$Phe2 - H\beta1$		Arg6 – Hα	4.22
$Phe2 - H\beta2$		Arg6 – Hβ1	1.85
$Phe2 - p-^{t}Bu$	1.28	Arg6 – Hβ2	1.85
Phe3 – NH		Arg6 – Hδ11	3.15
Phe3 – H α		Arg6 – Hδ12	3.15
Phe3 – H β 1		Ile7 – NH	7.47
Phe3 – H β 2		Ile7 – Ha	3.92
Leu4 – NH	7.87	Ile7 – H β	1.56
Leu4 – Ha	4.14	Ile7 – Hγ11	1.27
Leu4 – Hβ1	1.83	$Ile7 - H\gamma 12$	1.02
$Leu4 - H\beta2$	1.66	Ile7 – Hy21	0.40
Leu4 – Hδ11	0.98	Ile7 – H γ 22	0.40
$Leu4 - H\delta 12$	0.98	Ile7 – H γ 23	0.40
Leu4 – Hδ13	0.98	Ile7 – H δ 1	0.71
$Leu4 - H\delta 21$	0.92	Ile7 – H δ 2	0.71
$Leu4 - H\delta 22$	0.92	Ile7 – Hδ3	0.71
$Leu4 - H\delta 23$	0.92	Phe8-NH	7.71
$Leu4 - H\gamma$	1.66	$Phe8 - H\alpha$	4.65
		$Phe8 - H\beta1$	3.30
		$Phe8 - H\beta2$	2.86

Table S8. 400 MHz ¹H Proton Chemical Shifts of $[p^{-t}BuF^2]$ SHf in 80 mM SDS- d_{25} (9:1 H₂O/D₂O), 40 °C.

Hydrogen	Chemical Shift	Hydrogen	Chemical Shift
	(ppm)		(ppm)
$Phe1 - H\alpha$		Ser5 – NH	7.91
$Phe1 - H\beta1$		$Ser5 - H\alpha$	4.28
$Phe1 - H\beta1$		$Ser5 - H\beta1$	3.95
$Phe1 - p-^{t}Bu$	1.30	$Ser5 - H\beta2$	3.95
Phe2 – NH		Arg6 – NH	7.73
Phe2 – Ha		Arg6 – Hα	4.18
$Phe2 - H\beta1$		Arg6 – Hβ1	1.85
$Phe2 - H\beta2$		Arg6 – Hβ2	1.85
Phe3 – NH		Arg6 – Hδ11	3.13
Phe3 – Ha	4.33	Arg6 – Hδ12	3.13
$Phe3 - H\beta1$	3.08	Ile7 – NH	7.44
$Phe3 - H\beta2$	3.08	$Ile7 - H\alpha$	3.91
Leu4 – H	7.98	$Ile7 - H\beta$	1.55
$Leu4 - H\alpha$	4.10	$Ile7 - H\gamma 11$	1.29
Leu4 – Hβ1	1.84	$Ile7 - H\gamma 12$	1.00
$Leu4 - H\beta2$	1.67	Ile7 – Hy21	0.38
Leu4 – Hδ11	0.98	Ile7 – H γ 22	0.38
Leu4 – Hδ12	0.98	Ile7 – H γ 23	0.38
$Leu4 - H\delta 13$	0.98	Ile7 – H δ 1	0.70
$Leu4 - H\delta 21$	0.92	$Ile7 - H\delta2$	0.70
$Leu4 - H\delta 22$	0.92	Ile7 – Hδ3	0.70
$Leu4 - H\delta 23$	0.92	Phe8 – NH	7.73
$Leu4 - H\gamma$	1.67	$Phe8 - H\alpha$	4.64
		$Phe8 - H\beta1$	3.30
		$Phe8 - H\beta2$	2.85

Table S9. 400 MHz ¹H Proton Chemical Shifts of $[p^{-t}BuF^{1}]$ SHf in 80 mM SDS- d_{25} (9:1 H₂O/D₂O), 40 °C.

Hydrogen	Chemical Shift	Hydrogen	Chemical Shift
	(ppm)		(ppm)
$Phe1 - H\alpha$		Arg5 – NH	7.78
$Phe1 - H\beta1$		Arg5 – Ha	4.11
$Phe1 - H\beta1$		Arg5 – Hβ1	1.77
Phe2 – H		Arg5 – Hβ2	1.77
Phe2 – Ha		Arg5 – Hδ11	
$Phe2 - H\beta1$		Arg5 – Hδ12	
$Phe2 - H\beta2$		Arg6 – NH	7.73
Phe3 – NH	7.68	Arg6 – Hα	4.15
Phe3 – Ha	4.42	Arg6 – Hβ1	1.81
Phe3 – H β 1	3.07	Arg6 – Hβ2	1.81
Phe3 – H β 2	3.07	Arg6 – Hδ11	
$Phe3 - p-^{t}Bu$	1.22	Arg6 – Hδ12	
Leu4 – NH	7.74	Ile7 – NH	7.52
$Leu4 - H\alpha$	4.05	$Ile7 - H\alpha$	3.85
Leu4 – Hβ1	1.73	$Ile7 - H\beta$	1.55
$Leu4 - H\beta2$	1.53	Ile7 – Hγ11	1.31
Leu4 – Hδ11	0.94	Ile7 – H γ 12	1.02
Leu4 – Hδ12	0.94	Ile7 – Hy21	0.36
Leu4 – Hδ13	0.94	Ile7 – H γ 22	0.36
Leu4 – Hδ21	0.88	Ile7 – H γ 23	0.36
Leu4 – H δ 22	0.88	Ile7 – H δ 1	0.71
Leu4 – H δ 23	0.88	Ile7 – Hδ2	0.71
Leu4 – Hy	1.53	Ile7 – Hδ3	0.71
,		Phe8 – NH	7.71
		$Phe8 - H\alpha$	4.60
		Phe8 – Hβ1	3.28
		Phe8 – $H\beta2$	2.84

Table S10. 400 MHz ¹H Proton Chemical Shifts of $[p^{-t}BuF^3, R^5]SHf$ in 80 mM SDS- d_{25} (9:1 H₂O/D₂O), 40 °C.

Hydrogen	Chemical Shift	Hydrogen	Chemical Shift
	(ppm)		(ppm)
$Phe1 - H\alpha$		Arg5 – NH	7.80
$Phe1 - H\beta1$		Arg5 – Hα	4.14
$Phe1 - H\beta1$		$Arg5 - H\beta1$	1.85
Phe2 – NH		$Arg5 - H\beta2$	1.85
$Phe2 - H\alpha$		Arg5 – Hδ11	3.17
$Phe2 - H\beta1$		Arg5 – Hδ12	3.17
$Phe2 - H\beta2$		Arg6 – NH	7.78
$Phe2 - p-^{t}Bu$	1.29	Arg6 – Hα	4.21
Phe3 – NH		Arg6 – Hβ1	1.85
$Phe3 - H\alpha$		Arg6 – Hβ2	1.85
Phe3 – H β 1		Arg6 – Hδ11	3.19
$Phe3 - H\beta2$		Arg6 – Hδ12	3.17
Leu4 – NH	7.66	Ile7 – NH	7.59
$Leu4 - H\alpha$	4.13	Ile7 – Ha	3.89
$Leu4 - H\beta1$	1.71	Ile7 – Hβ	1.60
$Leu4 - H\beta2$	1.63	Ile7 – Hγ11	1.27
$Leu4 - H\delta 11$	0.98	Ile7 – Hy12	1.04
Leu4 – Hδ12	0.98	Ile7 – Hy21	0.42
Leu4 – Hδ13	0.98	Ile7 – H γ 22	0.42
$Leu4 - H\delta 21$	0.91	Ile7 – H γ 23	0.42
$Leu4 - H\delta 22$	0.91	Ile7 – H δ 1	0.72
$Leu4 - H\delta 23$	0.91	Ile7 – H δ 2	0.72
Leu4 – Hy	1.63	Ile7 – H δ 3	0.72
		Phe8-NH	7.69
		$Phe8 - H\alpha$	4.62
		$Phe8 - H\beta1$	3.29
		$Phe8 - H\beta2$	2.88

Table S11. 400 MHz ¹H Proton Chemical Shifts of $[p^{-t}BuF^2, R^5]$ SHf in 80 mM SDS- d_{25} (9:1 H₂O/D₂O), 40 °C.

Hydrogen	Chemical Shift	Hydrogen	Chemical Shift
	(ppm)		(ppm)
$Phe1 - H\alpha$		Arg5 – NH	7.73
$Phe1 - H\beta1$		Arg5 – Ha	4.25
Phe1 – H β 1		Arg5 – Hβ1	1.83
$Phe1 - p-^{t}Bu$	1.30	Arg5 – Hβ2	1.83
Phe2 – NH		Arg5 – Hδ11	3.18
$Phe2 - H\alpha$		Arg5 – Hδ12	3.18
$Phe2 - H\beta1$		Arg6 – NH	7.71
$Phe2 - H\beta2$		Arg6 – Ha	4.16
Phe3 – NH	7.82	Arg6 – Hβ1	1.83
$Phe3 - H\alpha$	4.28	Arg6 – Hβ2	1.83
Phe3 – H β 1		Arg6 – Hδ11	3.18
$Phe3 - H\beta2$		Arg6 – Hδ12	3.18
Leu4 – NH	7.61	Ile7 – NH	7.54
Leu4 – Ha	4.07	Ile7 – Ha	3.86
Leu4 – Hβ1	1.86	Ile7 – Hβ	1.56
$Leu4 - H\beta 2$	1.61	Ile7 – Hγ11	1.25
$Leu4 - H\delta 11$	0.99	Ile7 – Hy12	1.01
$Leu4 - H\delta 12$	0.99	Ile7 – Hγ21	0.38
Leu4 – Hδ13	0.99	Ile7 – H γ 22	0.38
$Leu4 - H\delta 21$	0.92	Ile7 – H γ 23	0.38
$Leu4 - H\delta 22$	0.92	Ile7 – Hδ1	0.68
$Leu4 - H\delta 23$	0.92	Ile7 – H δ 2	0.68
Leu4 – Hy	1.73	Ile7 – Hδ3	0.68
		Phe8 – NH	7.71
		Phe8 – Ha	4.61
		$Phe8 - H\beta1$	3.28
		Phe8 – H β 2	2.84

Table S12. 400 MHz ¹H Proton Chemical Shifts of $[p^{-t}BuF^1, R^5]$ SHf in 80 mM SDS- d_{25} (9:1 H₂O/D₂O), 40 °C.





2 mg [S²]SHf in 80 mM SDS- d_{25} (9:1 H₂O/D₂O), 40 °C, 400 MHz TOCSY.





file djo_2_277 5 2 mg [S²]SHf in 80 mM SDS- $d_{_{25}}$ (9:1 H₂O/D₂O), 40 °C, 400 MHz ROESY.

S-33

2 mg [R⁵]SHf in 80 mM SDS- d_{25} (9:1 H₂O/D₂O), 40 °C, 400 MHz TOCSY.







2 mg [R⁵]SHf-CO₂H in 80 mM SDS- d_{25} (9:1 H₂O/D₂O), 40 °C, 400 MHz ROESY.



2 mg [HmS⁵]SHf in 80 mM SDS- d_{25} (9:1 H₂O/D₂O), 40 °C, 400 MHz TOCSY.







2 mg [α MeF³]SHf in 80 mM SDS- d_{25} (9:1 H₂O/D₂O), 40 °C, 400 MHz ROESY.





2 mg [p-^tBuF³]SHf in 80 mM SDS- d_{25} (9:1 H₂O/D₂O), 40 °C, 400 MHz ROESY.













2 mg [p-^tBuF³, R⁵]SHf in 80 mM SDS- d_{25} (9:1 H₂O/D₂O), 40 °C, 400 MHz TOCSY.





2 mg [$p^{-t}BuF^2$, R⁵]SHf in 80 mM SDS- d_{25} (9:1 H₂O/D₂O), 40 °C, 400 MHz TOCSY.







