

Supporting Information

Coral-like TeO₂ microwires for rechargeable aluminum batteries

Jiguo Tu,[†] Mingyong Wang,^{*,†} Yiwa Luo,[†] and Shuqiang Jiao^{*,†}

[†] State Key Laboratory of Advanced Metallurgy, University of Science and
Technology Beijing, Beijing, 100083, PR China

***Corresponding Authors:** sjiao@ustb.edu.cn, mywang@ustb.edu.cn

The number of pages: 13; The number of figures: 9; The number of tables: 3

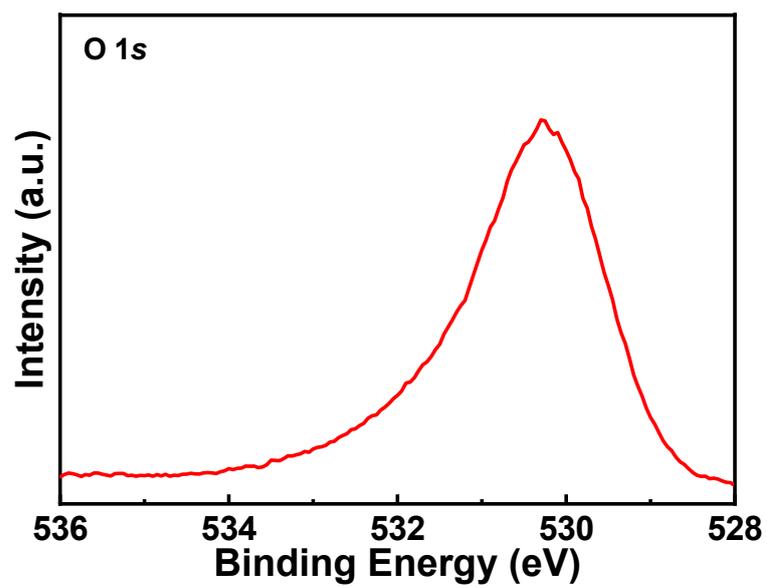


Figure S1. High-resolution XPS spectrum of the O 1s of the as-prepared sample.

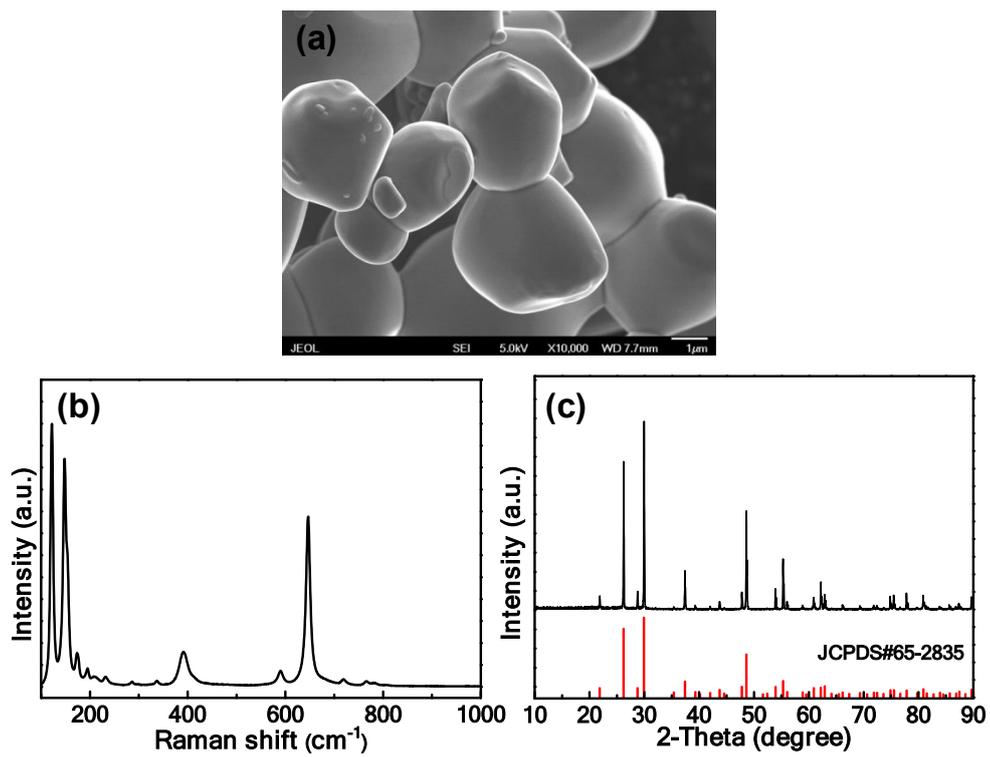


Figure S2. (a) SEM image, (b) Raman spectrum and (c) XRD pattern of commercial TeO₂ powder.

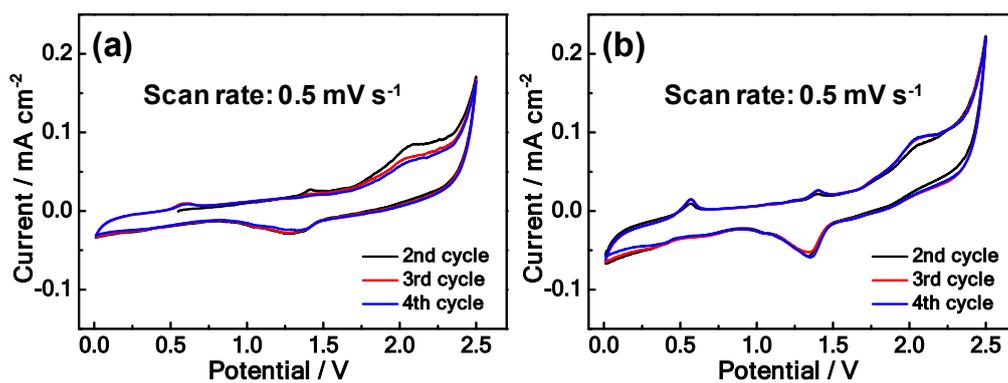


Figure S3. CV curves from 2nd to 4th cycle of (a) C- TeO_2 and (b) H- TeO_2 electrodes at a scan rate of 0.5 mV s^{-1} between 0.01 and 2.5 V vs. Al^{3+}/Al .

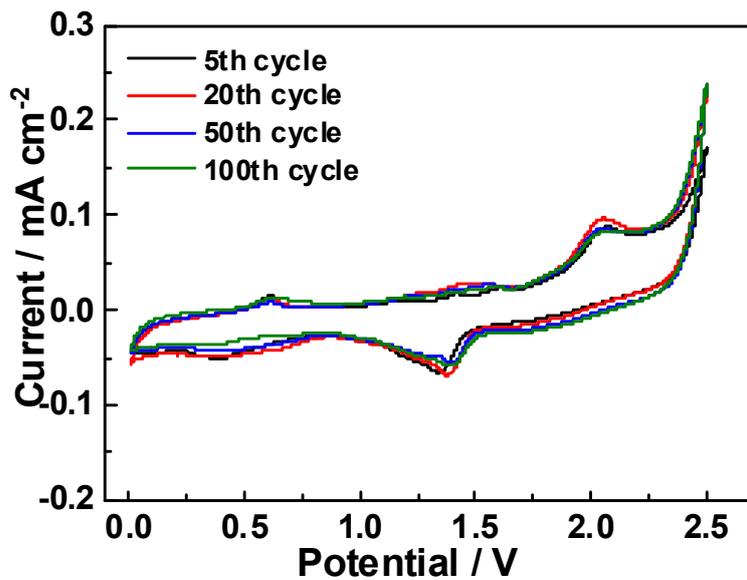


Figure S4. CV curves at different cycles of H-TeO₂ electrode at a scan rate of 0.5 mV s⁻¹ between 0.01 and 2.5 V vs. Al³⁺/Al.

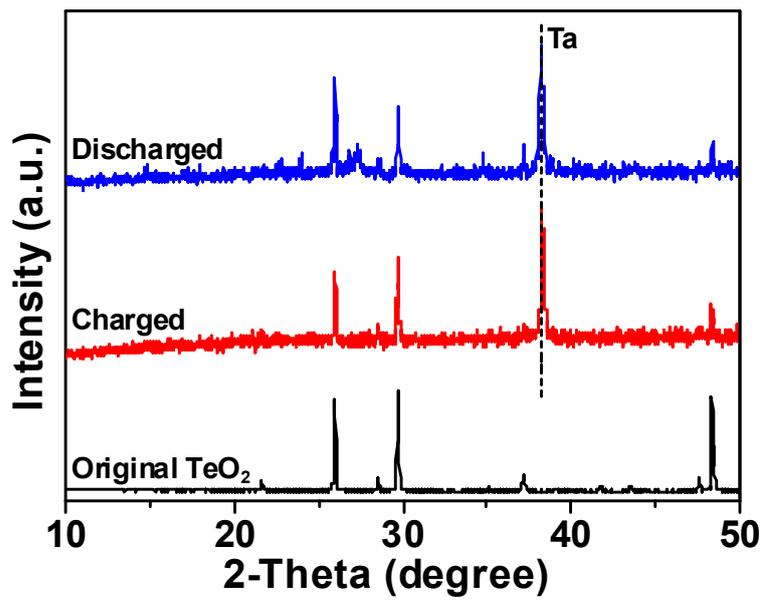


Figure S5. XRD patterns of TeO₂ after fully charged and discharged state.

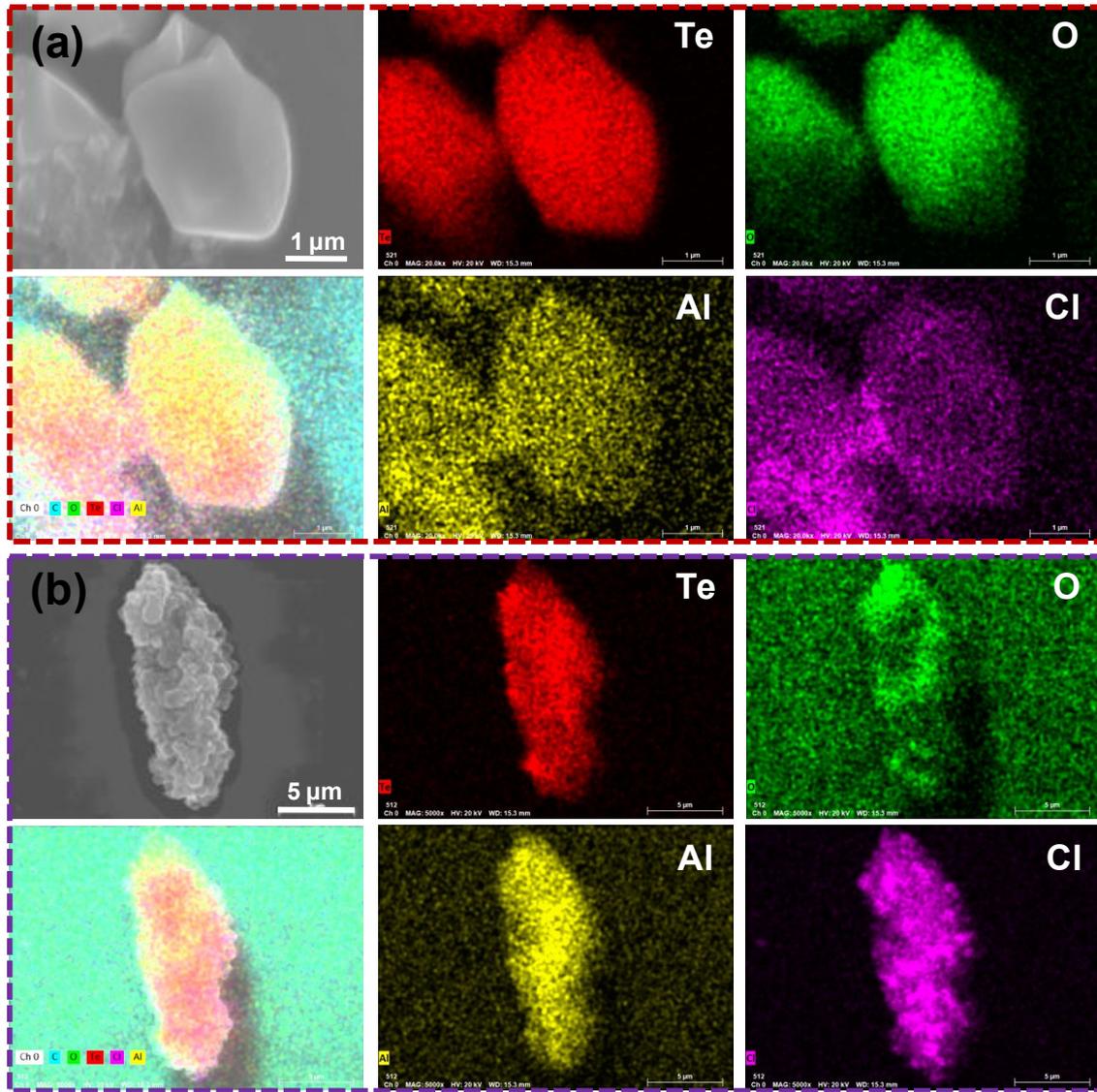


Figure S6. SEM images and element mapping images of Te, O, Al and Cl of the charged (a) and discharged (b) TeO_2 electrodes.

Table S1. EDX element analysis results of the fully charged TeO₂.

<i>Element</i>	<i>Weight/%</i>	<i>Atom/%</i>
<i>C</i>	69.84	82.50
<i>Te</i>	8.21	0.91
<i>O</i>	15.48	13.73
<i>Al</i>	2.13	1.12
<i>Cl</i>	4.33	1.73
<i>Total</i>	100.00	100.00

Table S2. EDX element analysis results of the fully discharged TeO₂.

<i>Element</i>	<i>Weight/%</i>	<i>Atom/%</i>
<i>C</i>	41.07	61.85
<i>Te</i>	18.82	2.67
<i>O</i>	22.80	25.78
<i>Al</i>	5.40	3.62
<i>Cl</i>	11.92	6.08
<i>Total</i>	100.00	100.00

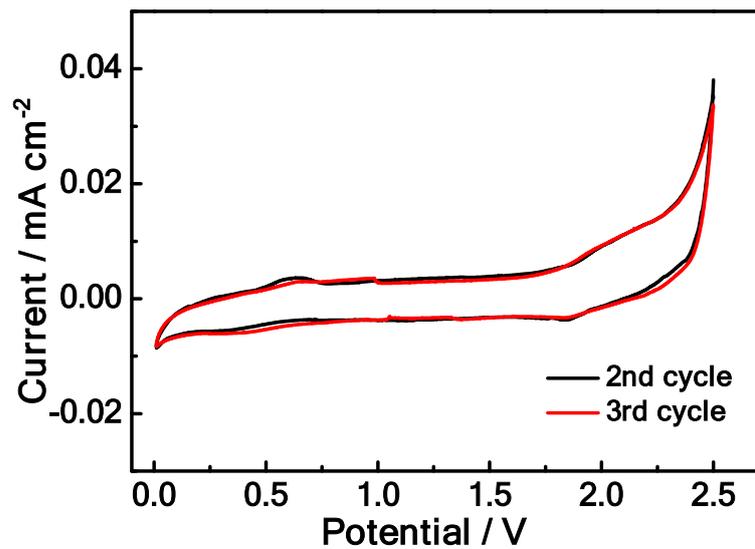


Figure S7. CV curves from 2nd to 3rd cycle of AB-PVDF-GF/A at a scan rate of 0.5 mV s⁻¹ between 0.01 and 2.5 V vs. Al³⁺/Al.

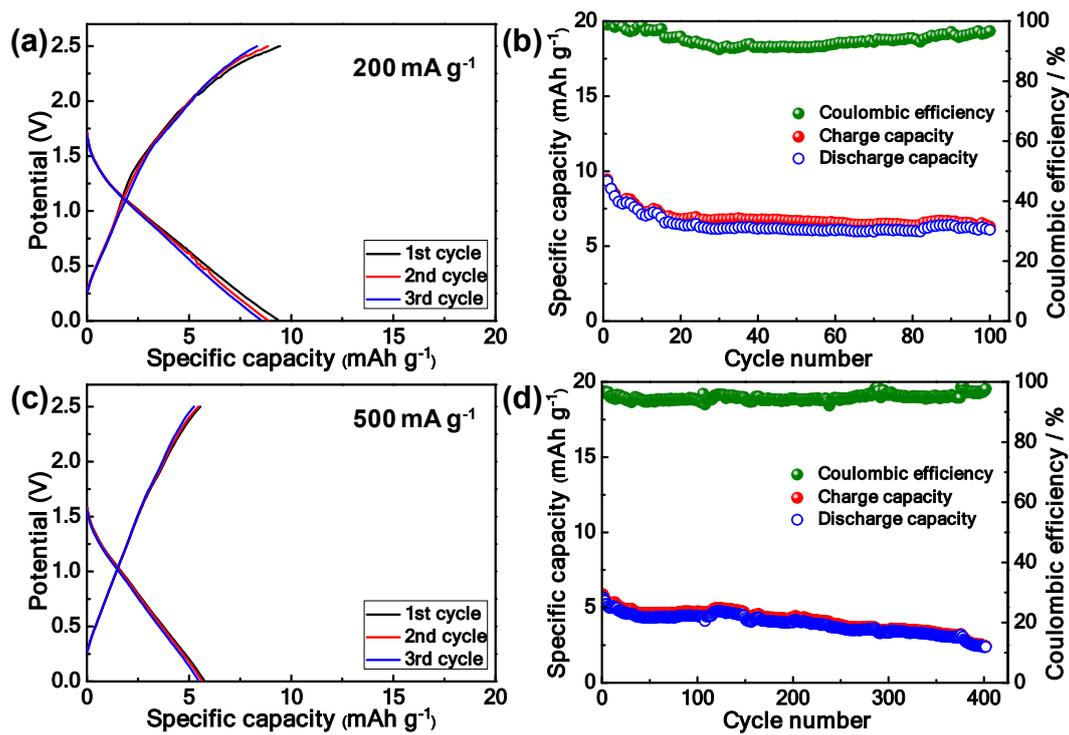


Figure S8. The first three charge-discharge curves for (a,c) the first three cycles and (b,d) cycling performance of AB-PVDF-GF/A at a current density of (a,b) 200 mA g⁻¹ and (c,d) 500 mA g⁻¹.

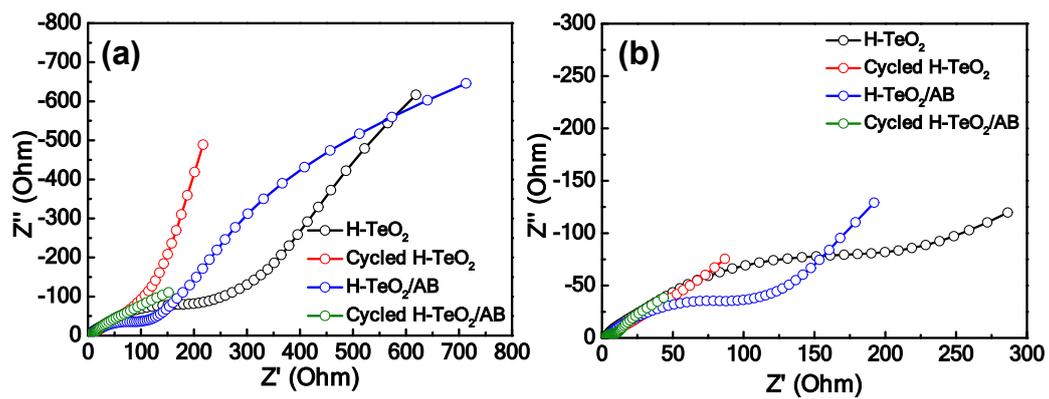


Figure S9. (a) The Nyquist plots of the H-TeO₂ and H-TeO₂/AB electrodes before and after cycling. (b) The partial enlargement of (a) in the high-medium frequency.

Table S3. Comparative electrochemical performance of TeO₂ with other cathode materials recently reported.

Cathode materials	Electrolyte	Discharge capacity (mAh g ⁻¹)		Cycle number	Current density (mA g ⁻¹)	Ref.
		1 st cycle	Last cycle			
H-TeO ₂ /AB	AlCl ₃ :[EMIm]Cl (1.3:1)	285.9	152.0	150	200	This work
Carbon paper	AlCl ₃ :[EMIm]Cl (1.3:1)	50	62.7	50	150	1
Pyrolytic graphite	AlCl ₃ :[EMIm]Cl (1.3:1)	60	66	200	66	2
Pyrolytic graphite	AlCl ₃ :[EMIm]Cl (1.5:1)	62	72	2000	75	3
Graphite	AlCl ₃ :Urea (1.3:1)	68	73	200	100	4
Polypyrrole	AlCl ₃ :[EMIm]Cl (1.5:1)	71	48	100	20	5
Ni ₃ S ₂	AlCl ₃ :[EMIm]Cl (1.3:1)	350	60	100	100	6
V ₂ O ₅	AlCl ₃ :[BMIm]Cl (1.1:1)	239	~180	5	44.2	7
SnO ₂ /C	AlCl ₃ :[EMIm]Cl (1.3:1)	~140	~140	1100	1000	8

Reference:

- (1) Sun, H.; Wang, W.; Yu, Z.; Yuan, Y.; Wang, S.; Jiao, S. A new aluminium-ion battery with high voltage, high safety and low cost. *Chem. Commun.* **2015**, *51*, 11892-11895.
- (2) Lin, M.; Gong, M.; Lu, B.; Wu, Y.; Wang, D.; Guan, M.; Angell, M.; Chen, C.; Yang, J.; Hwang, B. J.; *et. al.* An ultrafast rechargeable aluminium-ion battery. *Nature* **2015**, *520*, 324-328.
- (3) Elia, G. A.; Hasa, I.; Greco, G.; Diemant, T.; Marquardt, K.; Hoepfner, K.; Behm, R. J.; Hoell, A.; Passerini, S.; Hahn, R. Insights into the reversibility of aluminum graphite batteries. *J. Mater. Chem. A* **2017**, *5*, 9682-9690.
- (4) Angell, M.; Pan, C.-J.; Rong, Y.; Yuan, C.; Lin, M.-C.; Hwang, B.-J.; Dai, H. High Coulombic efficiency aluminum-ion battery using an AlCl₃-urea ionic liquid

- analog electrolyte, *P. Natl. Acad. Sci. USA* **2017**, *114*, 834-839.
- (5) Hudak, N. S. Chloroaluminate-doped conducting polymers as positive electrodes in rechargeable aluminum batteries. *J. Phys. Chem. C* **2014**, *118*, 5203-5215.
- (6) Wang, S.; Yu, Z.; Tu, J.; Wang, J.; Tian, D.; Liu, Y.; Jiao, S. A novel aluminum-ion battery: Al/AlCl₃-[EMIm]Cl/ Ni₃S₂@graphene. *Adv.Mater.* **2016**, *6*, 1606137.
- (7) Wang, H.; Bai, Y.; Chen, S.; Luo, X.; Wu, C.; Wu, F.; Lu, J.; Amine, K. Binder-free V₂O₅ cathode for greener rechargeable aluminum battery. *ACS Appl. Mater. Interfaces* **2015**, *7*, 80-84.
- (8) Lu, H.; Wan, Y.; Wang, T.; Jin, R.; Ding, P.; Wang, R.; Wang, Y.; Teng, C.; Li, L.; Wang, X.; *et. al.* A high performance SnO₂/C nanocomposite cathode for aluminum-ion batteries. *J. Mater. Chem. A* **2019**, *7*, 7213-7220.