1 Supporting information for:

2 Prominent Contribution of Hydrogen Peroxide to Intracellular Reactive Oxygen Species

- 3 Generated upon Exposure to Naphthalene Secondary Organic Aerosols
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24 Section S1: Design and optimization of catalase assay

Determination of exposure time for catalase treatment of cells. Previous studies have shown 25 that catalase treatment of cells can mitigate the effects of redox insults and effectively reduce 26 ROS/RNS signals.¹⁻⁷ However, there are substantial variations in the employed methodologies 27 depending on cell lines used, such as time of treatment with catalase, fluorescent dyes, and 28 stimulants tested. In previous studies,⁵⁻⁷ cells were pretreated with catalase for a range of 0.5 to 4 29 h before exposing them to stimulants. However, these studies used specific features to facilitate 30 the intracellular access of catalase, a macromolecule, to cellular and subcellular regions.⁷ In this 31 study, to ensure that there was sufficient time for the catalase to interact with and/or be absorbed 32 33 by the cells, it was added to naphthalene SOA extracts at a specific concentration and incubated for 24 h. Note that H₂O₂ in naphthalene SOA extracts can be decomposed immediately after adding 34 catalase. The chosen time of exposure was based on the stability of catalase for 24 h at 37°C^{8,9} 35 and on the exposure time used in the intracellular ROS/RNS assay.¹⁰ 36

Optimization of catalase concentration. The optimized catalase concentration was determined 37 by exposing macrophage cells to positive controls (1 μ g/mL LPS and 200 μ M H₂O₂) with the 38 addition of 0, 50, and 200 U/mL catalase (Sigma-Aldrich, C-3515), which were concentrations 39 generally used in prior studies.^{1, 2} As shown in Fig. S1, LPS and H₂O₂ induced a response of 1.5-40 fold compared to control cells. The addition of 50 and 200 U/mL catalase decreased (p < 0.1) the 41 42 ROS/RNS signal induced by LPS and H₂O₂ to control values. This confirmed that catalase removed the species that mediated the oxidation of the probe compound and that the protocol can 43 44 be used to identify reactive species that drive ROS/RNS response. The difference in the response after treating the cells with 50 and 200 U/mL catalase was not significant (p = 0.9), therefore, the 45 chosen concentration of catalase was 50 U/mL. 46



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Figure S1. ROS/RNS response of cells exposed to positive controls (1 μ g/mL LPS, 200 μ M H₂O₂) without (grey) and with (orange) the addition of catalase (50 and 200 U/mL). ROS/RNS was measured using the fluorescent dye carboxy-H₂DCFDA. Values represent the fold of change over control cells. Data are presented as mean \pm SE of measurements carried out in triplicate. Statistically significant differences were determined with the *t*-test using a 90% confidence interval. * (p < 0.1) indicates significance compared to LPS or H₂O₂ samples without the addition of catalase.

Established catalase assay protocol. The cells were plated and exposed to the probe following
steps (1) to (3) of our intracellular ROS/RNS protocol (Fig. S2).¹⁰ Step (4) involved replacing the
ROS/RNS probe solution with naphthalene SOA containing 50 U/mL catalase. H₂O₂ (200 μM)

was used as a positive control and unstained cells exposed to probe solution were used as a negative control. Stained cells exposed to supplemented media or catalase (50 U/mL) only were used to correct for the background ROS/RNS signal. After 24 h of incubation, the medium was removed and replaced with phosphate buffer solution (PBS). Lastly, the plate was placed in a microplate reader (BioTek Synergy H4) to measure the fluorescence intensity at excitation of 485 nm and emission of 528 nm.



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Figure S2. Optimized protocol for catalase assay. Cells were plated and exposed to naphthalene
SOA containing 50 U/mL catalase following the methodology described in Tuet et al. ¹⁰.
Fluorescence intensity was measured after 24 h exposure with a microplate reader. Figure was
modified from Tuet et al.¹⁰

72 Section S2: List of experimental conditions in aerosol generation and elemental composition 73 of SOA

Table S1. Experimental Conditions and Elemental Composition of Naphthalene SOA Formed in the Presence of NO_x

	Initial	Relative	[SOA] ^a	Initial	Initial				
Sample	naphthalene	humidity		NO	NO_2	0:C	H:C	N:C	OSc
1		0/0	иа т ⁻³	nnh	nnh	ratio	ratio	ratio	USC
	ppv	70	$\mu_{\mathcal{S}} m$	PPU	PPU				
a	222	< 5%	213	229	392	0.30	0.96	0.01	-0.36
b	321	< 5%	430	245	471	0.28	0.97	0.01	-0.40
с	450	< 5%	596	225	333	0.29	0.98	0.01	-0.39
d	550	< 5%	492	236	401	0.25	0.98	0.01	-0.48

^{*a*} Average SOA mass concentration in the chamber during filter collection; The SOA mass concentration

was calculated by multiplying SMPS volume concentrations with a density of 1.48 g cm⁻³ based on prior
 experiments.¹¹ The total aerosol mass collected on filters was calculated by multiplying average SOA mass

79 concentration with the total volume of air collected, as described in Tuet et al. 12 .

81 Section S3: Results of intracellular ROS/RNS with and without catalase

82 Results of naphthalene SOA samples



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Figure S3. Dose-response curve of ROS/RNS produced by exposing to naphthalene SOA (samples 84 a-d). Cells were exposed to (A) naphthalene SOA only (red curves) and (B) naphthalene SOA + 85 50 U mL⁻¹ catalase (blue curves) for 24 h. ROS/RNS values represent the fold change in 86 fluorescence over control cells. Data points are presented as mean \pm SE of measurements carried 87 88 out in triplicate, generating 3 sets of response for each naphthalene SOA sample. Every 10-dilution set of response was fitted with a dose-response curve and used to calculate the area under the curve 89 (AUC) as described in Tuet et al.¹⁰ The reported AUC represents the mean \pm SE estimated from 3 90 AUCs for each measurement. 91

92 Results of laboratory-prepared tBOOH and H₂O₂ solutions



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Figure S4. ROS/RNS response by cells exposed to 50, 100, 200, and 400 μ M tBOOH solutions without (red) and with (blue) catalase. 200 μ M H₂O₂ was used as a positive control. Values represent the fold change in fluorescence over control cells. Data are presented as mean \pm SE of measurements carried out in triplicate. ** indicates p < 0.01.

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100 Section S4: Results of H₂O₂ quantification using Amplex red

101 Calibration curves for H₂O₂ quantification



Figure S5. Calibration curve of H_2O_2 (in PBS) for the quantification of H_2O_2 (A) in naphthalene SOA extracts and diffusion of H_2O_2 into cells; and (B) produced by cells using Amplex red. An additional calibration curve was made for quantifying H_2O_2 produced by cells to account for the lower concentration range.

Determination of H₂O_{2[SOA]} diffusion into cells. In order to investigate the diffusion of H₂O_{2[SOA]}
into cells, we separately determined H₂O_{2[SOA]} without cells and H₂O_{2[SOA]} in the presence of cells.
The decrease in H₂O₂ in the presence of cells compared with that in the absence of cells was
considered to be due to uptake into the cells.

As described in the main text, for each SOA sample (samples a-d), 50 μ L of working solution (100 μ M Amplex red reagent and 0.2 U/mL horseradish peroxidase (HRP) diluted in 1x reaction buffer) were added to 50 μ L of naphthalene SOA extracted in PBS and incubated for 30 min at room temperature based on the Amplex red protocol (Invitrogen). Fluorescence intensity was measured after 30 min incubation. Here, for selected samples (samples b and d), we also determined H₂O_{2[SOA]} at multiple time points (every 15 min over a period of 100 min after the 30 min incubation), as shown in Fig. S6 (blue circles).

To determine H₂O_{2[SOA]} in the presence of cells, cells were plated and stained with the probe 118 following steps (1) to (3) of our intracellular ROS/RNS protocol (Fig. S2).¹⁰ Step (4) involved 119 replacing the ROS/RNS probe solution with naphthalene SOA samples extracted in PBS. H₂O₂ in 120 stained cells exposed to naphthalene SOA was quantified every 15 minutes over 1 h. At each time 121 point, 50 µL of cellular medium (PBS) containing naphthalene SOA was transferred to a 96 well 122 123 plate. Then, the reaction was started by adding 50 μ L of working solution to each well with the samples. Fluorescence intensity was measured at multiple time points after 30 min incubation. 124 Results are shown in Fig. S6 (green circles). 125



Figure S6. Time evolution of H₂O₂ concentration in naphthalene SOA extracts (without cells) and
 H₂O₂ concentration in cell medium containing naphthalene SOA (cells stained with ROS probe).

- 129 Panels (A) and (B) correspond to data from SOA samples b and d, respectively. H_2O_2 was
- 130 quantified using Amplex red. Fluorescence signal was converted to H₂O₂ concentration based on
- 131 calibration curves in Fig. S5-A. Data are presented as mean \pm SE of measurements carried out in
- 132 triplicate.
- 133

134 Quantification of H₂O₂ produced by cells



Figure S7. Quantification of H₂O₂ produced by cells. The cells were first exposed to naphthalene 136 137 SOA samples a-d for 24 h (data in orange). Afterwards, the cell medium was replaced by 100 µL 138 of working solution (100 µM Amplex red reagent and 0.2 U/mL HRP diluted in 1x reaction buffer) and incubated for 10 min, and then the fluorescence signal was measured every 15 minutes over a 139 140 period of 100 min. Data in grey correspond to cells that were exposed to culture media only (no 141 naphthalene SOA) for 24 h. Values are presented as mean ± SE of experiments carried out in 142 triplicate. Fluorescence signal was converted to H₂O₂ concentration based on calibration curve in Fig. S5-B. 143

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