Supporting information

Chemical looping reforming of glycerol for continuous H₂ *production by moving-bed reactors: Simulation and experiment*

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Supplementary text

Definitions of Ni/G, S/G, conversion, H_2 yield, and selectivity in experiments and simulations were shown as follows.

$$Ni / G = \frac{Ni \text{ mole flow in feedstock}}{Glycerol \text{ mole flow in feedstock}}$$
(S1)

$$S/G = \frac{Steam \text{ mole flow in feedstock}}{Glycerol \text{ mole flow in feedstock}}$$
(S2)

$$X_{gly} = \frac{F_{gly,in} - F_{gly,out}}{F_{gly,in}} \times 100\%$$
(S3)

where $F_{gly,in}$ and $F_{gly,out}$ represent the glycerol mole flow of inlet and outlet, respectively.

$$H_2 \text{ yield} = \frac{\text{mole flow rate of } H_2 \text{ produced}}{3 \times \text{mole flow rate of glycerol feed}}$$
(S4)

$$S_j = \frac{F_{out} \times c_j}{3 \times F_{gly,in}} \times 100\%$$
(S5)

where F_{out} is the total mole flow of the outlet and c_j represents the concentrations of CO₂, CO, and CH₄. C selectivity is calculated on the basis of C balance.

$$S_C = \left(1 - \sum S_j\right) \times 100\% \tag{S6}$$

$$S_{H_2} = \frac{3}{7} \times \frac{\text{mole flow rate of } H_2 \text{ produced}}{\text{mole flow rate of glycerol feed} \times X_{gly}} \times 100\%$$
(S7)

Carbon balance

The effluents were analyzed online by a gas chromatograph (GC, Agilent 7890B) with two detectors. One is a flame ionization detector with a Porapak-A column using N₂ as a carrier gas to analyze the organic species including C_2H_4 , C_2H_6 , CH_3OH , CH_3CHO , C_2H_5OH , CH_3COOH , and $C_3H_8O_3$. The other one is a thermal conductivity detector with a TDX-01 column using He as a carrier gas to detect incondensable gas species including H₂, CO, CH₄ and CO₂. The carbon balance for all runs was greater than 95%. Table S1shows a typical carbon balance calculation (S/G = 3, T= 650°C and Ni/G = 0.9 for the NiW-Al). We calculated the C balance according to the following equation:

Carbon balance =
$$\frac{C \text{ mole flow rates in the products}}{3 \times \text{glycerol mole flow rates} \times X_{gly}} \times 100\%$$
 (S8)

СО	CO ₂	CH_4	Other	C balance
44.0%	51.1%	4.5%	0.2%	99.8%

Table S1 Carbon balance in moving-bed reactors

In the simulation, the FR and AR were set at isothermal conditions. The solids and fuels entering the

reactors were preheated to the reactor temperatures, and the pressure of the system was set to 1 atm. Key simulation parameters are listed in Table S2. The reaction conditions corresponds to our experimental ones. The property method is based on the guidelines for choosing a property method in Aspen plus user guide.

Reactor model	RGibbs (all possible products)
Stream class	MIXCISLD, CISOLID
Property method	NRTL and STEAM-TA for free water
Calculation method	Sequential Modular
Databank	PURE28, AQUEOUS, SOLIDS, INOGANIC
Product	C ₂ H ₄ , C ₂ H ₆ , CH ₃ OH, CH ₃ CHO, C ₂ H ₅ OH, CH ₃ COOH, C ₃ H ₈ O ₃ , CH ₄ , CO,
Tiouuci	CO ₂ , H ₂ , H ₂ O, C, NiWO ₄ , Ni, W

 Table S2 Description of the Aspen Plus Model

Table 55 input conditions for Aspen Simulation									
Glycerol flow rate	100 kmol/hr								
NiWO ₄ flow rate	10 kmol/hr-150 kmol/hr								
Steam flow rate	150 kmol/hr-600 kmol/hr								
Block temperature	600°C-700°C								
Input stream temperature	650°C								

Table S3 Input Conditions for Aspen Simulation

Table S4 shows the fuelout flow rates of 5-stage and 10-stage systems. The results indicate that our 5-stage model converges.

	5-stage (kmol/h)	10-stage (kmol/h)
C ₃ H ₈ O ₃	6.9083E-26	6.9083E-26
H ₂	317.9918	317.9918
H ₂ O	145.4745	145.4745
СО	119.1185	119.1185
CO ₂	112.7035	112.7035
C ₂ H ₄	1.80499E-05	1.80499E-05
C ₂ H ₆	9.12719E-05	9.12719E-05
CH ₃ CHO	2.26272E-07	2.26272E-07
CH ₃ OH	2.01221E-06	2.01221E-06
CH ₃ COOH	2.07871E-08	2.07871E-08
CH ₄	18.26666	18.26666

Table S4 Flow rates of the last fuel reactor outlets of 5-stage and 10-stage systems ($S/G = 1$, $Ni/G = 1$)	=
0.9, FR temperature = 650° C, and AR temperature = 650° C).	

NiWO ₄	H ₂	СО	CO ₂	CH ₄	С	AR1	AR2	AR3	AR4	AR5	FR1	FR2	FR3	FR4	FR5
KMOL/HR	KMOL/HR	KMOL/HR	KMOL/HR	KMOL/HR	KMOL/HR	MJ/HR	MJ/HR	MJ/HR	MJ/HR	MJ/HR	MJ/HR	MJ/HR	MJ/HR	MJ/HR	MJ/HR
10.00	505.41	124.59	161.11	14.30	0.00	3540.31	0.00	0.00	0.00	0.00	529.13	0.00	0.00	17533.48	17533.48
20.00	499.38	122.42	164.43	13.15	0.00	7080.62	0.00	0.00	0.00	0.00	1042.85	0.00	0.00	17533.48	17533.48
30.00	493.12	120.19	167.73	12.07	0.00	10620.93	0.00	0.00	0.00	0.00	1541.71	0.00	0.00	17533.48	17533.48
40.00	486.64	117.92	171.02	11.06	0.00	14161.25	0.00	0.00	0.00	0.00	2026.29	0.00	0.00	17533.48	17533.48
50.00	479.95	115.61	174.28	10.11	0.00	17701.56	0.00	0.00	0.00	0.00	2497.17	0.00	0.00	17533.48	17533.48
60.00	473.04	113.26	177.52	9.23	0.00	21241.87	0.00	0.00	0.00	0.00	2954.93	0.00	0.00	17533.48	17533.48
70.00	465.94	110.87	180.73	8.40	0.00	24782.18	0.00	0.00	0.00	0.00	3400.15	0.00	0.00	17533.48	17533.48
80.00	458.63	108.44	183.92	7.63	0.00	28322.49	0.00	0.00	0.00	0.00	3833.43	0.00	0.00	17533.48	17533.48
90.00	451.14	105.99	187.09	6.92	0.00	31862.80	0.00	0.00	0.00	0.00	4255.36	0.00	0.00	17533.48	17533.48
100.00	443.46	103.52	190.23	6.26	0.00	35403.12	0.00	0.00	0.00	0.00	4666.54	0.00	0.00	17533.48	17533.48
110.00	435.61	101.02	193.34	5.64	0.00	38943.43	0.00	0.00	0.00	0.00	5067.55	0.00	0.00	17533.48	17533.48
120.00	427.58	98.50	196.42	5.08	0.00	42483.74	0.00	0.00	0.00	0.00	5458.97	0.00	0.00	17533.48	17533.48
130.00	419.40	95.97	199.47	4.56	0.00	46024.05	0.00	0.00	0.00	0.00	5841.39	0.00	0.00	17533.48	17533.48
140.00	411.05	93.42	202.49	4.08	0.00	49564.36	0.00	0.00	0.00	0.00	6215.36	0.00	0.00	17533.48	17533.48
150.00	402.56	90.87	205.49	3.64	0.00	53104.67	0.00	0.00	0.00	0.00	6581.45	0.00	0.00	17533.48	17533.48

Table S5 Material and energy stream table for different Ni/G ratios (FR = 650° C, AR = 650° C, and S/G = 4.5).

Steam	H ₂	СО	CO ₂	CH ₄	С	AR1	AR2	AR3	AR4	AR5	FR1	FR2	FR3	FR4	FR5
KMOL/HR	KMOL/HR	KMOL/HR	KMOL/HR	KMOL/HR	KMOL/HR	MJ/HR	MJ/HR	MJ/HR	MJ/HR	MJ/HR	MJ/HR	MJ/HR	MJ/HR	MJ/HR	MJ/HR
150.00	345.06	130.36	132.58	17.63	19.44	-39561.27	0.00	0.00	0.00	0.00	6649.78	0.00	0.00	0.00	12068.78
200.00	374.52	137.83	144.05	18.11	0.00	-31862.80	0.00	0.00	0.00	0.00	6434.63	0.00	0.00	0.00	14375.98
250.00	394.70	130.68	154.47	14.86	0.00	-31862.80	0.00	0.00	0.00	0.00	5837.64	0.00	0.00	0.00	15332.43
300.00	412.06	123.91	163.89	12.21	0.00	-31862.80	0.00	0.00	0.00	0.00	5330.85	0.00	0.00	0.00	16097.39
350.00	427.01	117.53	172.40	10.06	0.00	-31862.80	0.00	0.00	0.00	0.00	4904.51	0.00	0.00	0.00	16701.60
400.00	439.93	111.56	180.11	8.33	0.00	-31862.80	0.00	0.00	0.00	0.00	4549.10	0.00	0.00	0.00	17172.34
450.00	451.14	105.99	187.09	6.92	0.00	-31862.80	0.00	0.00	0.00	0.00	4255.36	0.00	0.00	0.00	17533.48
500.00	460.89	100.81	193.42	5.77	0.00	-31862.80	0.00	0.00	0.00	0.00	4014.44	0.00	0.00	0.00	17805.60
550.00	469.43	96.00	199.16	4.84	0.00	-31862.80	0.00	0.00	0.00	0.00	3818.15	0.00	0.00	0.00	18006.15
600.00	476.93	91.54	204.38	4.08	0.00	-31862.80	0.00	0.00	0.00	0.00	3659.14	0.00	0.00	0.00	18149.68

Table S6 Material and energy stream table for different S/G ratios (FR = 650° C, AR = 650° C, and Ni/G = 0.9).

Т	H2	СО	CO ₂	CH ₄	С	Ni	AR1	AR2	AR3	AR4	AR5	FR1	FR2	FR3	FR4	FR5
°C	KMOL/HR	KMOL/HR	KMOL/HR	KMOL/HR	KMOL/HR	KMOL/HR	MJ/HR	MJ/HR	MJ/HR	MJ/HR	MJ/HR	MJ/HR	MJ/HR	MJ/HR	MJ/HR	MJ/HR
600.00	416.26	78.75	198.80	22.45	0.00	90.00	31904.94	0.00	0.00	0.00	0.00	5761.36	0.00	0.00	9347.98	9347.98
650.00	451.14	105.99	187.09	6.92	0.00	90.00	31862.80	0.00	0.00	0.00	0.00	4255.36	0.00	0.00	17533.48	17533.48
700.00	454.04	123.70	174.53	1.76	0.00	90.00	31830.56	0.00	0.00	0.00	0.00	3634.77	0.00	0.00	22581.06	22581.06

Table S7 Material stream table for different FR temperatures (Ni/G = 0.9 and S/G = 4.5).

The XRD results show that only Ni, W, and Al_2O_3 phases existed in the samples, indicating that the glycerol feedstock is capable of completely reducing the NiWO₄ to its metallic states. This result is consistent with the analysis from the Ellingham diagram.



Fig. S1. XRD profiles of the OCs from the FR outlet (NiW-Al, S/G = 4.5, Ni/G = 0.9, FR temperature = 650°C, and AR temperature = 650°C).