

Supporting Information for

**Lanthanide-based Metal–Organic Frameworks Containing
“V-Shaped” Tetracarboxylate Ligands: Synthesis, Crystal
Structures, “Naked-Eye” Luminescent Detection and Catalytic
Properties**

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Figure S1 The chemical structure of 5,5'-methylenediisophthalic acid (H₄MDIA).

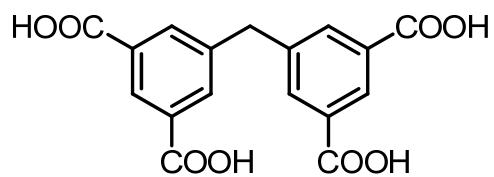


Figure S2 The asymmetric unit of Tb-MDIA, Tb = cyan, O = red, C = gray and H = white.

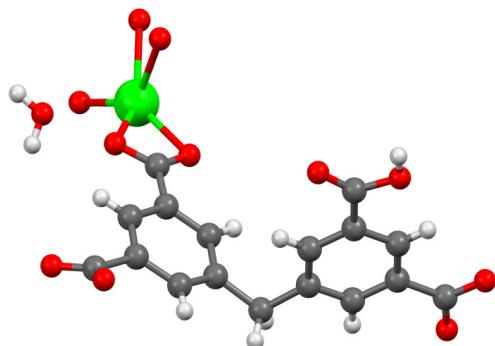


Figure S3 The coordination mode of HMDIA³⁻ ligands in Tb-MDIA, Tb = cyan, O = red, C = gray and H = white.

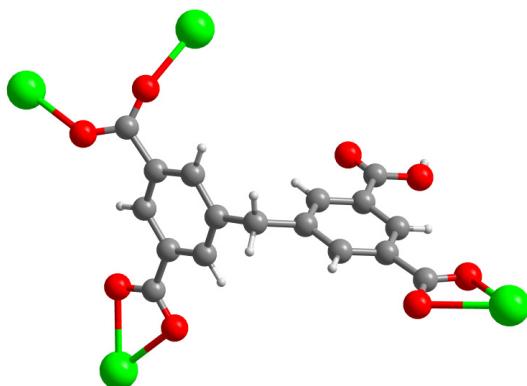


Figure S4 The asymmetric unit of Ho–MDIA, Ho = cyan, O = red, C = gray and H = white.

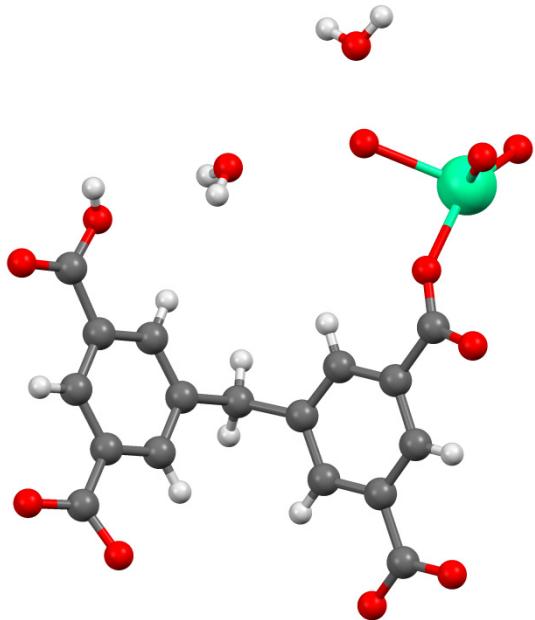


Figure S5 The coordination mode of HMDIA³⁻ ligands in Ho–MDIA, Ho = cyan, O = red, C = gray and H = white.

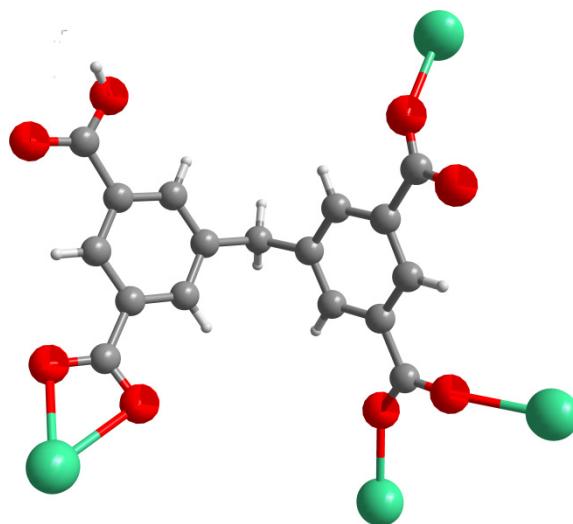


Figure S6 The asymmetric unit of Nd–**MDIA**, Nd = cyan, O = red, C = gray and H = white.

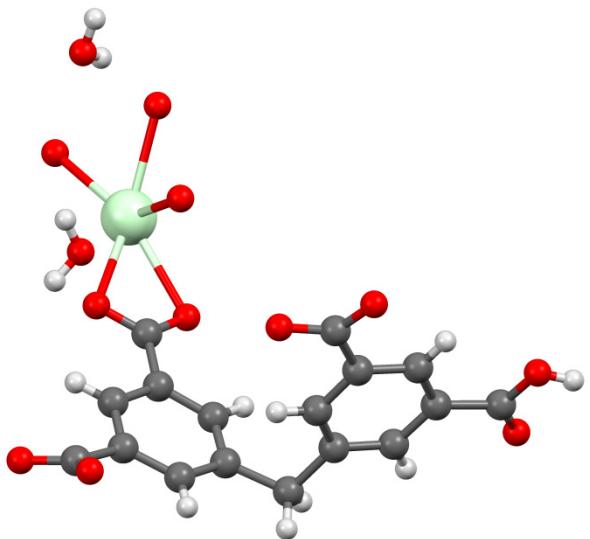


Figure S7 The coordination mode of HMDIA³⁻ ligands in Nd–**MDIA**, Nd = cyan, O = red, C = gray and H = white.

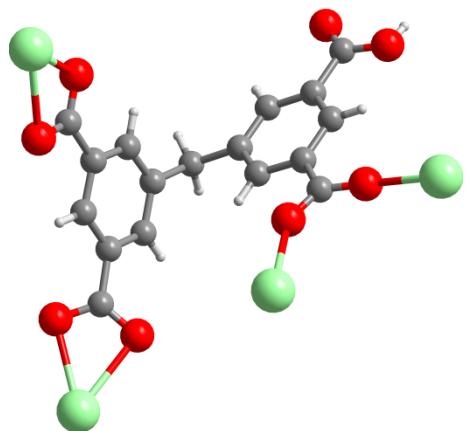


Table S1 Selective bond distance (\AA) and angle ($^\circ$) in Tb-MDIA.

Tb(1)-O(3A)	2.317(3)	Tb(1)-O(4B)	2.366(3)
Tb(1)-O(1)	2.407(3)	Tb(1)-O(6C)	2.413(3)
Tb(1)-O(2W)	2.453(3)	Tb(1)-O(3W)	2.455(3)
Tb(1)-O(1W)	2.490(3)	Tb(1)-O(2)	2.498(3)
Tb(1)-O(5C)	2.563(3)		
O(3A)-Tb(1)-O(4B)	95.29(11)	O(3A)-Tb(1)-O(1)	79.39(12)
O(4B)-Tb(1)-O(1)	125.01(11)	O(3A)-Tb(1)-O(6C)	77.74(11)
O(4B)-Tb(1)-O(6C)	146.80(11)	O(1)-Tb(1)-O(6C)	86.04(11)
O(3A)-Tb(1)-O(2W)	134.62(11)	O(4B)-Tb(1)-O(2W)	80.95(11)
O(1)-Tb(1)-O(2W)	138.74(11)	O(6C)-Tb(1)-O(2W)	81.44(11)
O(3A)-Tb(1)-O(3W)	148.94(12)	O(4B)-Tb(1)-O(3W)	77.59(11)
O(1)-Tb(1)-O(3W)	80.32(12)	O(6C)-Tb(1)-O(3W)	123.91(11)
O(2W)-Tb(1)-O(3W)	74.73(12)	O(3A)-Tb(1)-O1(W)	67.63(11)
O(4B)-Tb(1)-O(1W)	70.73(11)	O(1)-Tb(1)-O(1W)	145.21(11)
O(6C)-Tb(1)-O(1W)	76.69(10)	O(2W)-Tb(1)-O(1W)	68.54(11)
O(3W)-Tb(1)-O(1W)	134.29(11)	O(3A)-Tb(1)-O(2)	73.98(11)
O(4B)-Tb(1)-O(2)	72.69(11)	O(1)-Tb(1)-O(2)	53.07(10)
O(6C)-Tb(1)-O(2)	133.44(11)	O(2W)-Tb(1)-O(2)	143.34(11)
O(3W)-Tb(1)-O(2)	75.05(12)	O(1W)-Tb(1)-O(2)	122.97(11)
O(3A)-Tb(1)-O(5C)	121.09(10)	O(4B)-Tb(1)-O(5C)	143.52(10)
O(1)-Tb(1)-O(5C)	69.58(10)	O(6C)-Tb(1)-O(5C)	52.06(9)
O(2W)-Tb(1)-O(5C)	71.76(10)	O(3W)-Tb(1)-O(5C)	72.28(10)
O(1W)-Tb(1)-O(5C)	118.30(10)	O(2)-Tb(1)-O(5C)	117.28(10)

Symmetry code A: $x, 0.5-y, 0.5+z$; B: $-x, -0.5+y, 2.5-z$; C: $1-x, -0.5+y, 1.5-z$.

Table S2 Selective bond distance (\AA) and angle ($^\circ$) in Nd-MDIA.

Nd(1)-O(8A)	2.340(5)	Nd(1)-O(7B)	2.372(5)
Nd(1)-O(1C)	2.488(5)	Nd(1)-O(1W)	2.532(5)
Nd(1)-O(3W)	2.538(6)	Nd(1)-O(2W)	2.543(5)
Nd(1)-O(4)	2.555(5)	Nd(1)-O(2C)	2.593(5)
Nd(1)-O(3)	2.599(5)		
O(8A)-Nd(1)-O(7B)	100.1(2)	O(8A)-Nd(1)-O(1C)	76.8(2)
O(7B)-Nd(1)-O(1C)	142.6(2)	O(8A)-Nd(1)-O(1W)	145.4(2)
O(7B)-Nd(1)-O(1W)	81.8(2)	O(1C)-Nd(1)-O(1W)	81.2(2)
O(8A)-Nd(1)-O(3W)	145.0(2)	O(7B)-Nd(1)-O(3W)	76.1(2)
O(1C)-Nd(1)-O(3W)	127.0(2)	O(1W)-Nd(1)-O(3W)	69.3(2)
O(8A)-Nd(1)-O(2W)	76.3(2)	O(7B)-Nd(1)-O(2W)	72.3(2)
O(1C)-Nd(1)-O(2W)	70.7(2)	O(1W)-Nd(1)-O(2W)	71.3(2)
O(3W)-Nd(1)-O(2W)	132.1(2)	O(8A)-Nd(1)-O(4)	72.4(2)
O(7B)-Nd(1)-O(4)	69.32(18)	O(1C)-Nd(1)-O(4)	139.7(2)
O(1W)-Nd(1)-O(4)	137.6(2)	O(3W)-Nd(1)-O(4)	73.9(2)
O(2W)-Nd(1)-O(4)	124.17(18)	O(8A)-Nd(1)-O(2C)	109.79(19)
O(7B)-Nd(1)-O(2C)	150.12(19)	O(1C)-Nd(1)-O(2C)	51.03(17)
O(1W)-Nd(1)-O(2C)	74.57(19)	O(3W)-Nd(1)-O(2C)	78.6(2)
O(2W)-Nd(1)-O(2C)	115.56(18)	O(4)-Nd(1)-O(2C)	118.07(16)
O(8A)-Nd(1)-O(3)	78.2(2)	O(7B)-Nd(1)-O(3)	117.25(18)
O(1C)-Nd(1)-O(3)	98.83(19)	O(1W)-Nd(1)-O(3)	131.91(18)
O(3W)-Nd(1)-O(3)	73.3(2)	O(2W)-Nd(1)-O(3)	154.1(2)
O(4)-Nd(1)-O(3)	50.13(15)	O(2C)-Nd(1)-O(3)	69.23(16)

Symmetry code A: $x, -1+y, z$; B: $1-x, 2-y, 1-z$; C: $1+x, y, z$.

Table S3 Selective bond distance (\AA) and angle ($^\circ$) in Ho-MDIA.

Ho(1)-O(5A)	2.261(3)	Ho(1)-O(6B)	2.296(3)
Ho(1)-O(3)	2.326(3)	Ho(1)-O(3W)	2.354(3)
Ho(1)-O(1W)	2.363(3)	Ho(1)-O(8C)	2.387(3)
Ho(1)-O(7C)	2.455(3)	Ho(1)-O(2W)	2.469(3)
O(5A)-Ho(1)-O(6B)	96.81(10)	O(5A)-Ho(1)-O(3)	94.99(11)
O(6B)-Ho(1)-O(3)	140.28(10)	O(5A)-Ho(1)-O(3W)	153.71(13)
O(6B)-Ho(1)-O(3W)	72.95(11)	O(3)-Ho(1)-O(3W)	79.86(11)
O(5A)-Ho(1)-O(1W)	77.77(12)	O(6B)-Ho(1)-O(1W)	70.60(10)
O(3)-Ho(1)-O(1W)	149.12(11)	O(3W)-Ho(1)-O(1W)	119.11(12)
O(5A)-Ho(1)-O(8C)	78.18(10)	O(6B)-Ho(1)-O(8C)	144.24(10)
O(3)-Ho(1)-O(8C)	75.38(10)	O(3W)-Ho(1)-O(8C)	124.26(10)
O(1W)-Ho(1)-O(8C)	73.76(10)	O(5A)-Ho(1)-O(7C)	130.67(10)
O(6B)-Ho(1)-O(7C)	114.88(10)	O(3)-Ho(1)-O(7C)	84.18(11)
O(3W)-Ho(1)-O(7C)	74.82(11)	O(8C)-Ho(1)-O(7C)	53.79(9)
O(5A)-Ho(1)-O(2W)	75.39(11)	O(6B)-Ho(1)-O(2W)	74.36(10)
O(3)-Ho(1)-O(2W)	72.22(10)	O(3W)-Ho(1)-O(2W)	78.51(11)
O(1W)-Ho(1)-O(2W)	132.34(11)	O(8C)-Ho(1)-O(2W)	135.66(10)
O(7C)-Ho(1)-O(2W)	147.01(10)		

Symmetry code A: 2- x , 1- y , 2- z ; B: -1+ x , y , z ; C: 2- x , 0.5+ y , 1.5- z .

Figure S8 TGA traces of Tb–MDIA ranging from room temperature to 500 °C.

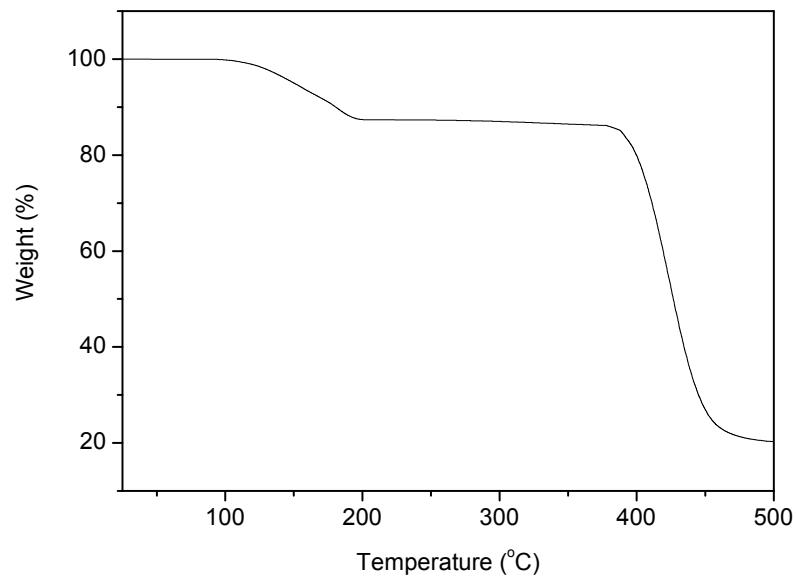


Figure S9 The Stern–Volmer plot of Tb–MDIA quenched by Fe^{3+} , where I_0 and I are the fluorescence intensity before and after Fe^{3+} incorporation, respectively.

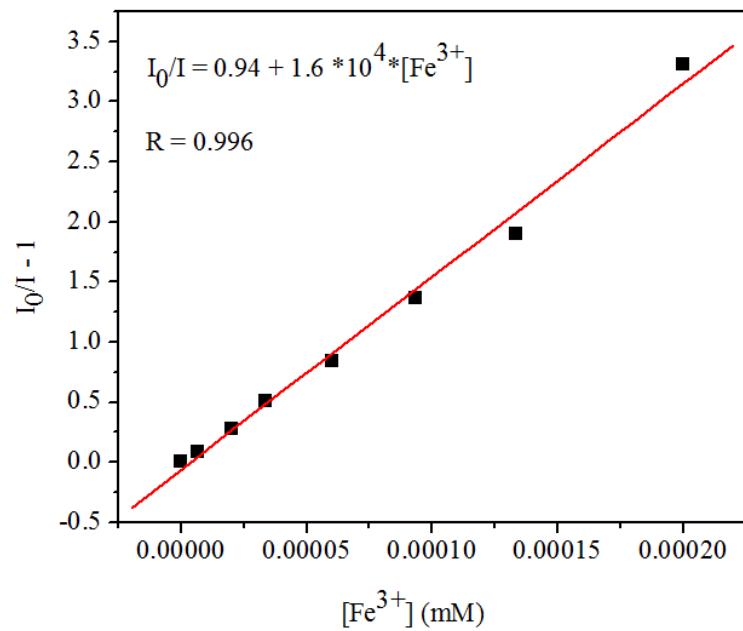
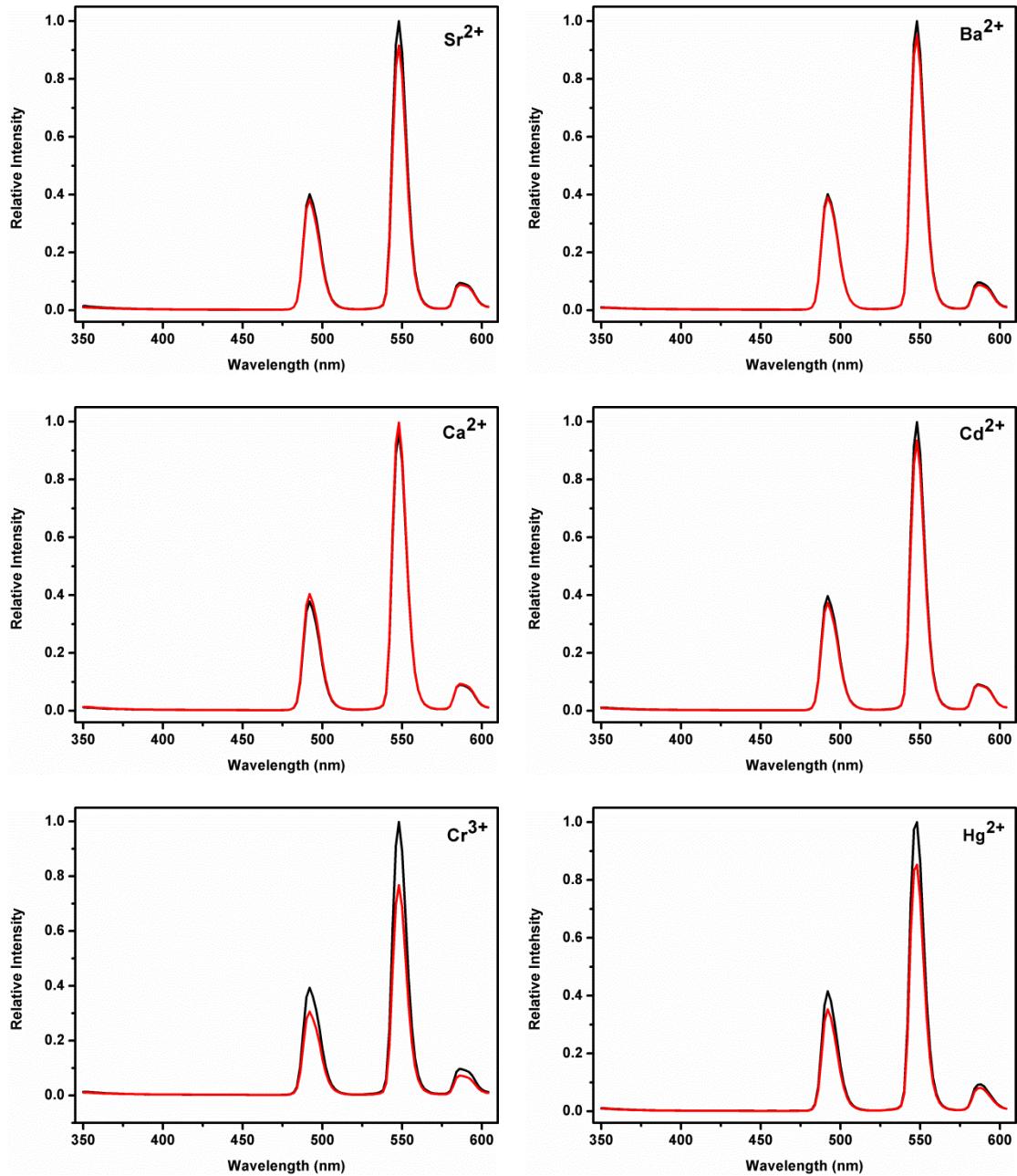
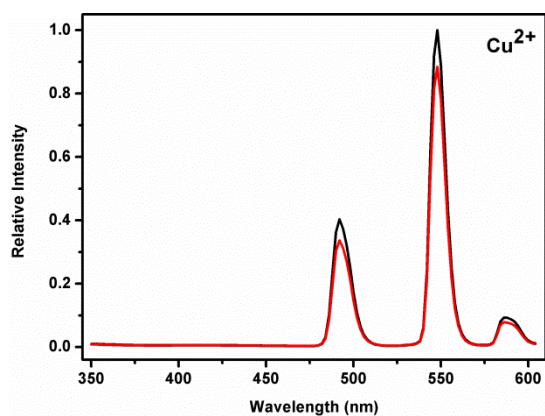
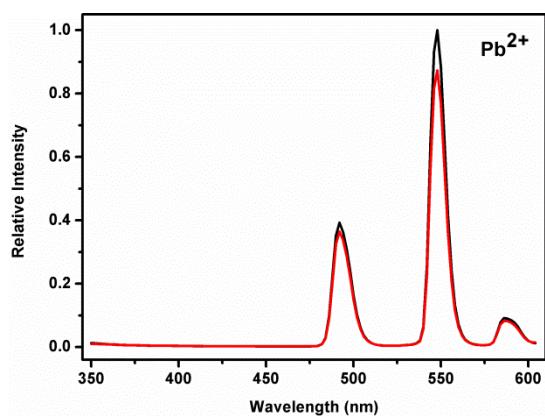
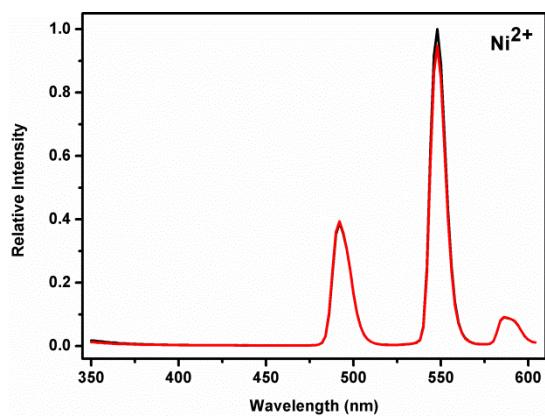
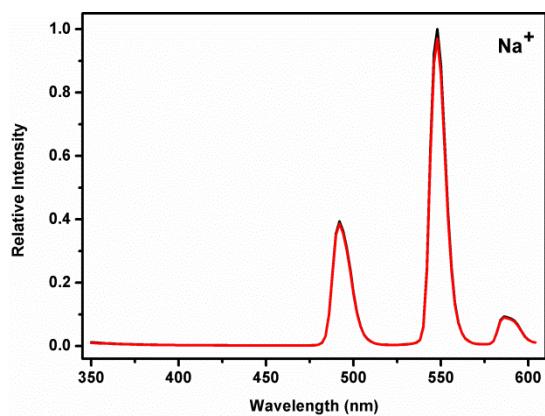
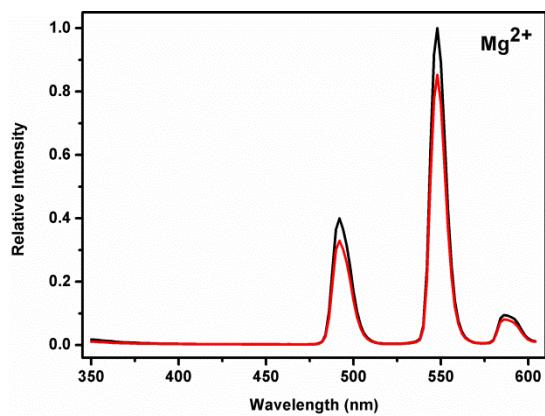
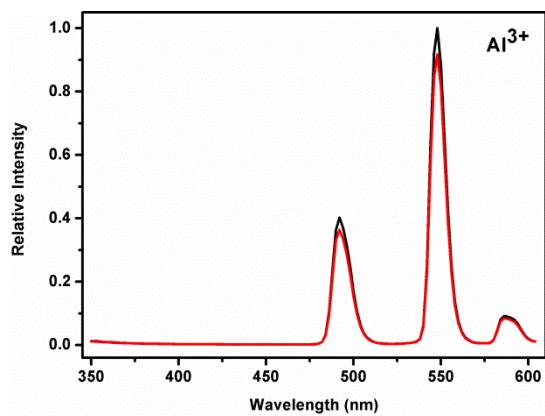
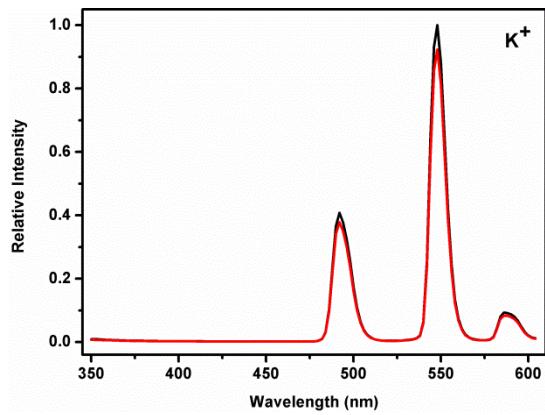
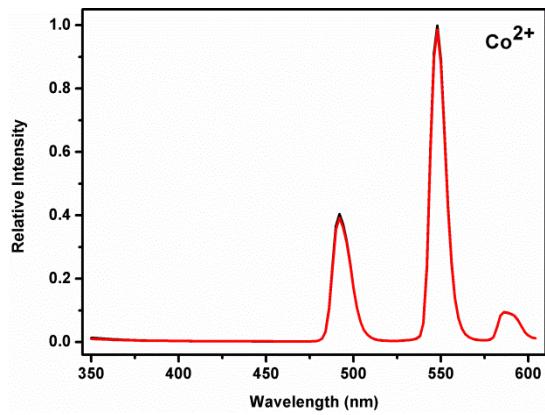


Figure S10 Families of various fluorescence spectra of Tb-MDIA (0.33 g/L) in water solution upon the addition of 0.4 mM of different selected analytes.





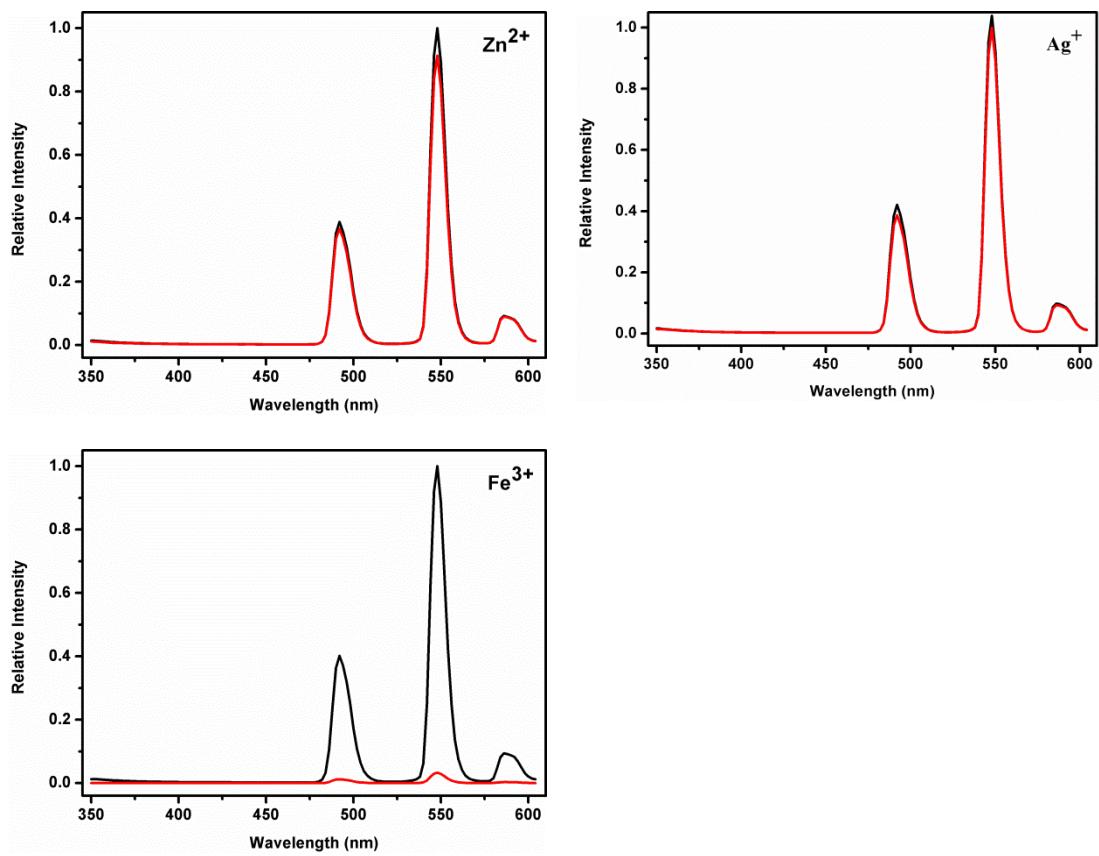


Figure S11 Fluorescence spectra of Tb-MDIA (0.33 g/L) with gradual addition of different concentrations of Fe^{3+} ions in HEPES aqueous solutions.

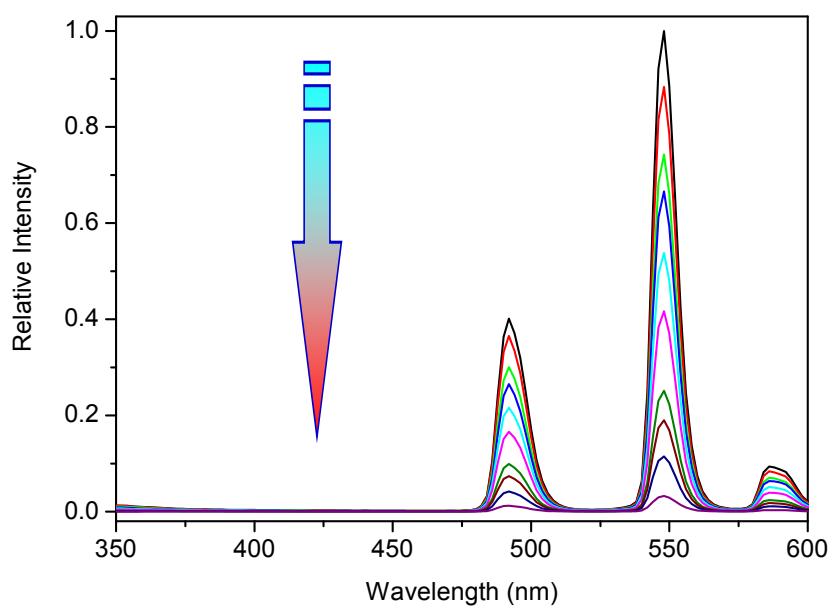


Figure S12 The Stern–Volmer plot of Tb–MDIA quenched by PA ethanol solution, where I_0 and I are the fluorescence intensity before and after PA incorporation, respectively.

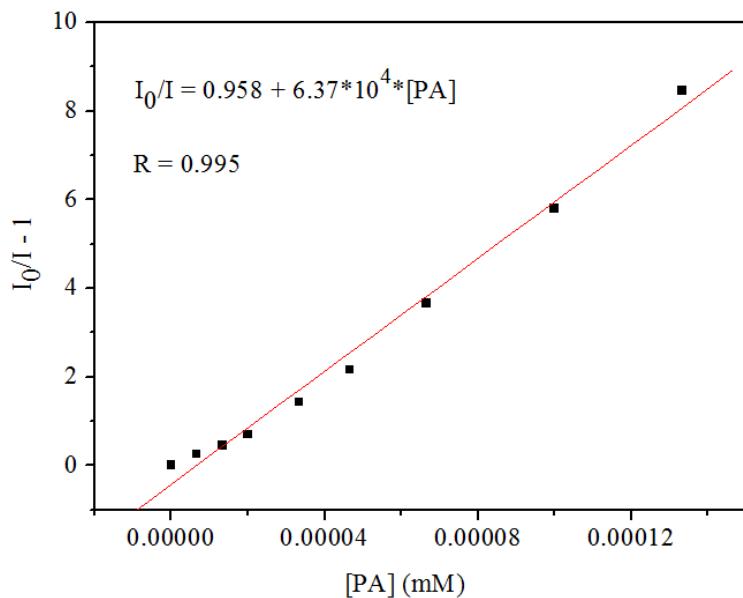


Figure S13 Powder X-ray diffraction (XRD) profiles for as-synthesized Tb–MDIA and Tb–MDIA samples after the PA detection.

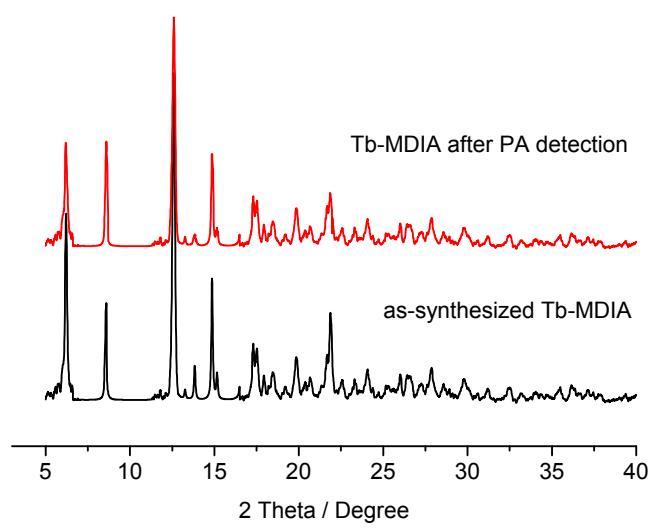
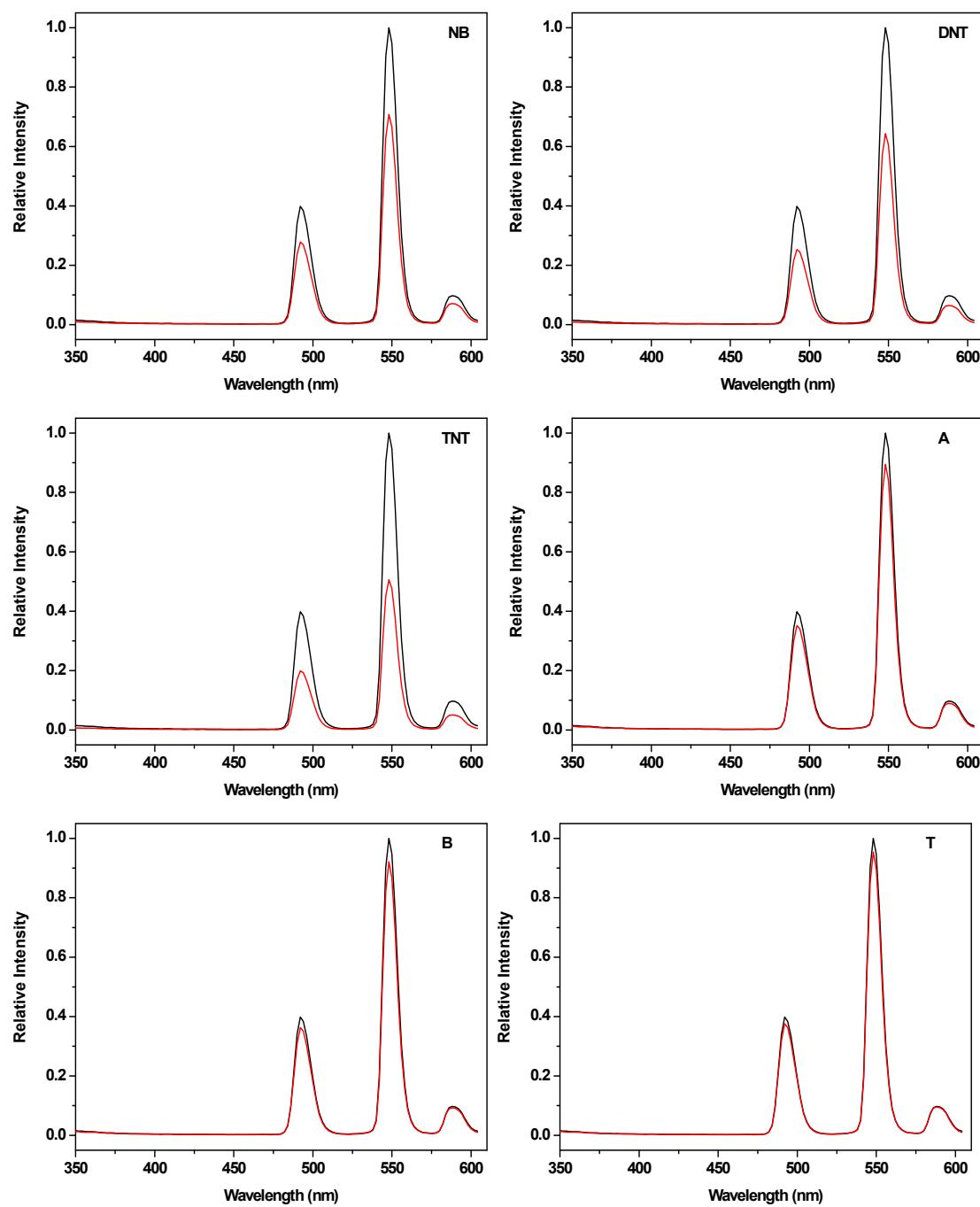
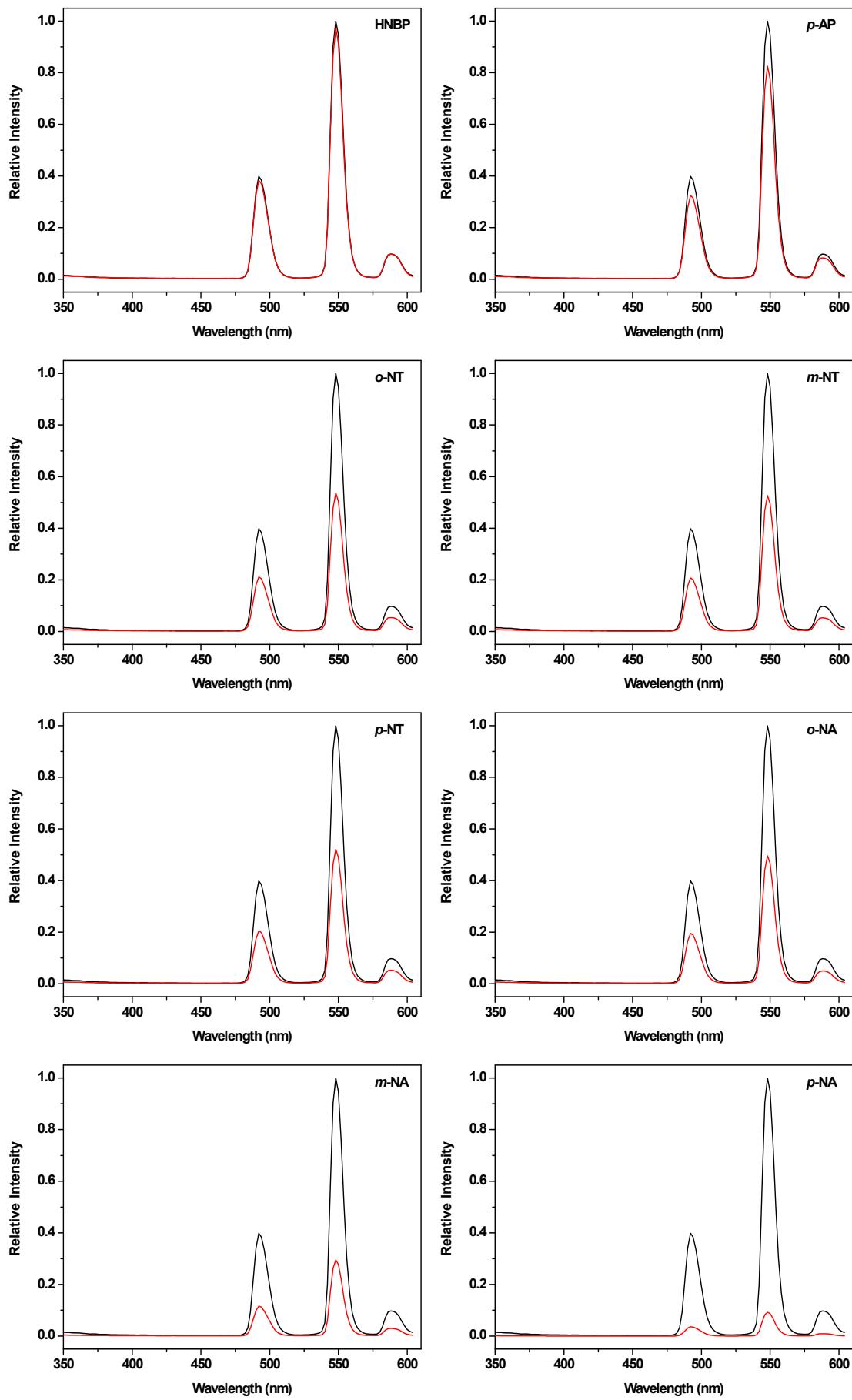


Figure S14 Families of various fluorescence spectra of Tb-MDIA (0.33 g/L) in ethanol solution upon the addition of 0.33 mM of different selected analytes.





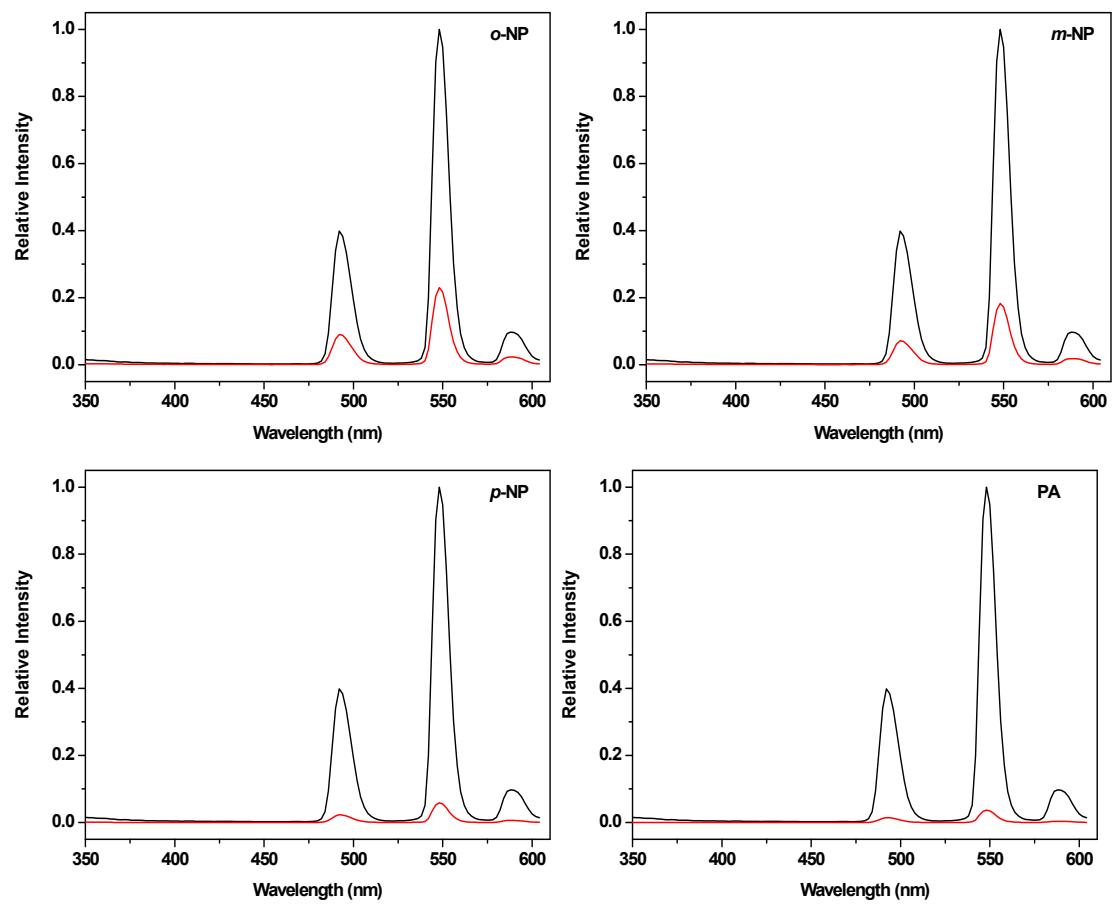


Figure S15 Yields of 2-phenyl-2-((trimethylsilyl)oxy)acetonitrile catalyzed by Ho–MDIA (violet) and without Ho–MDIA through filtration after 2 h of the reaction (green).

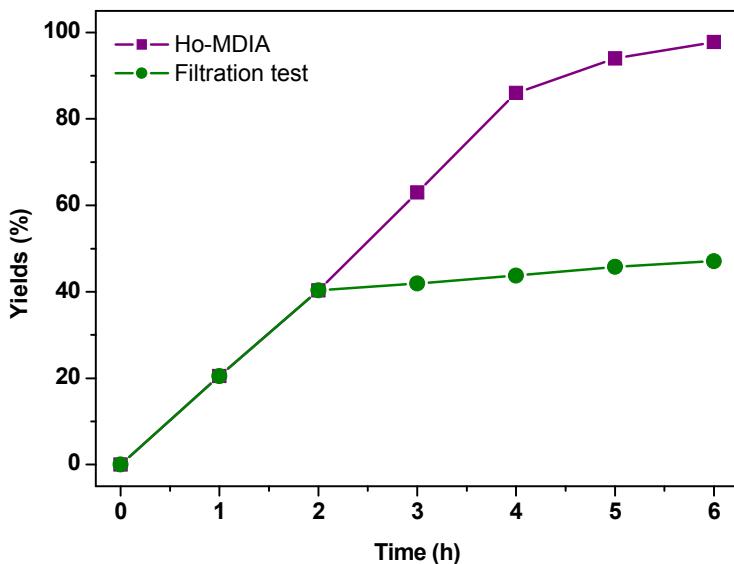


Figure S16 Study on recycling of catalyst Ho–MDIA for the heterogeneous cyanosilylation: $(CH_3)_3SiCN$: 1.2 mmol; benzaldehyde: 0.5 mmol; Ho–MDIA catalysts: 2 mol%, room temperature for 6 hours.

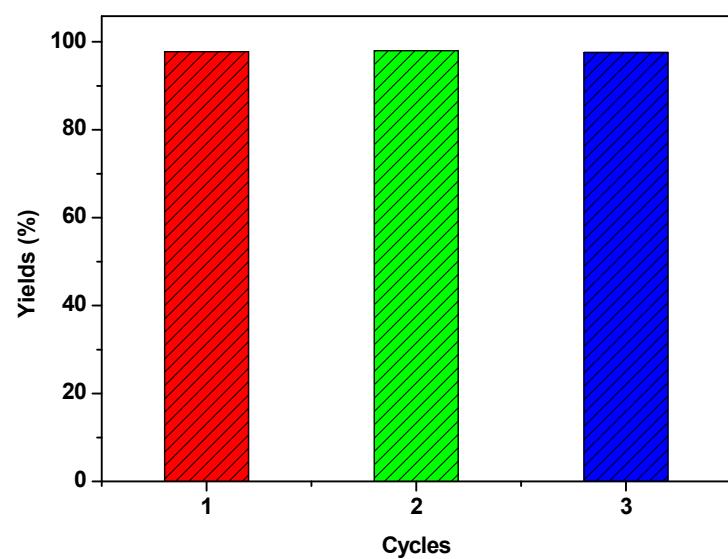


Figure S17 Powder X-ray diffraction (XRD) profiles for as-synthesized Ho-MDIA/Nd-MDIA and Ho-MDIA/Nd-MDIA samples after the third catalytic cycles.

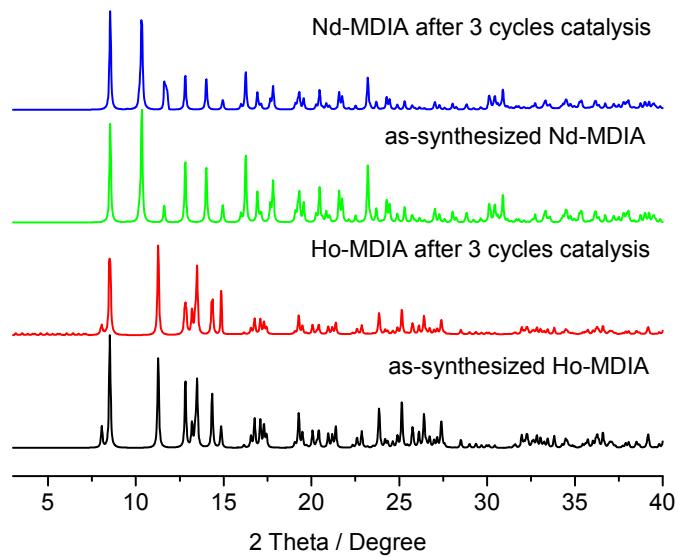


Table S4 Comparison with different MOF catalysts in the catalysis of cyanosilylation reaction of benzaldehyde with $(CH_3)_3SiCN$.

Entry	Catalyst	T (°C)	t (hr)	Yield(%)	Ref.
1	1·Cd	r.t.	18	94	[S1]
2	Cd-PBA	r.t.	8	99	[S2]
3	Ce-MDIP1	r.t.	24	93	[S3]
4	Ce-MDIP2	r.t.	24	94	[S3]
5	Eu-PDC	r.t.	3	93	[S4]
6	MIL-47 (V)	r.t.	3	46	[S5]
7	MIL-53 (Al)	r.t.	3	26	[S5]
8	MIL-101 (Cr)	r.t.	4	96	[S5]
9	Zn-MOF	r.t.	10	74	[S6]
10	Ce-MOF	r.t.	2	94	[S7]
11	Ps-CMOF	r.t.	48	93	[S8]
12	POMOF-1	r.t.	24	98	[S9]
13	UPC-15	r.t.	24	99	[S10]
14	UPC-16	r.t.	24	97	[S10]
15	Co-MOF	r.t.	12	98	[S11]
16	Cd-bpdc	r.t.	14	95	[S12]
17	Mn-MOF	r.t.	9	98	[S13]
18	Ho–MDIA	r.t.	6	97.8	This work

Reference

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