

SUPPORTING INFORMATION

B-H Functionalization of Hydrogen-Rich $[(\text{Cp}^*\text{V})_2(\text{B}_2\text{H}_6)_2]$: Synthesis and Structures of $[(\text{Cp}^*\text{V})_2(\text{B}_2\text{X}_2)_2\text{H}_8]$ ($\text{X} = \text{Cl, SePh}$; $\text{Cp}^* = \eta^5\text{-C}_5\text{Me}_5$)

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I Experimental

I.1 Supplementary Data

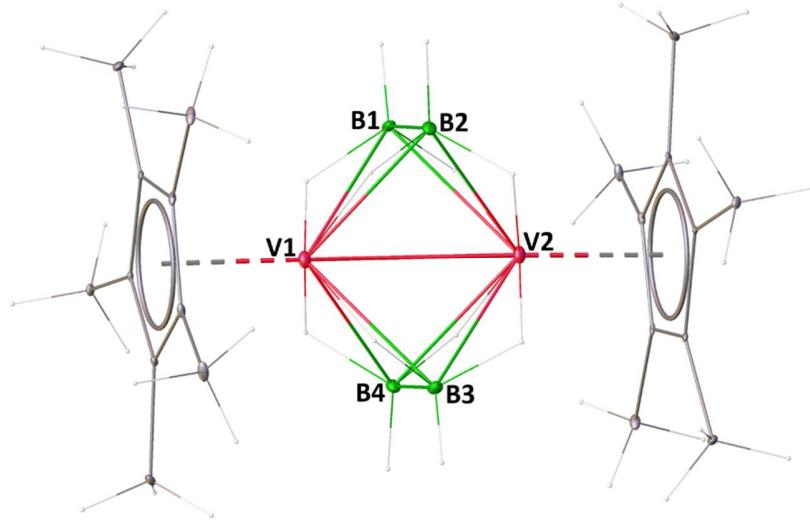


Figure S1. Molecular structure and labeling diagram for compound **1**. Selected bond lengths (\AA) and bond angles ($^{\circ}$): V1-V2 2.7820(9), V1-B1 2.300(4), V1-B2 2.295(4), V1-B3 2.301(4), V1-B4 2.307(4), B1-B2 1.752(7); B2-V1-V2 53.20(11), B2-V1-B1 44.81(16), B2-B1-V1 67.4(2), B2-V1-B1 44.81(16), B1-V1-B4 88.98(16).

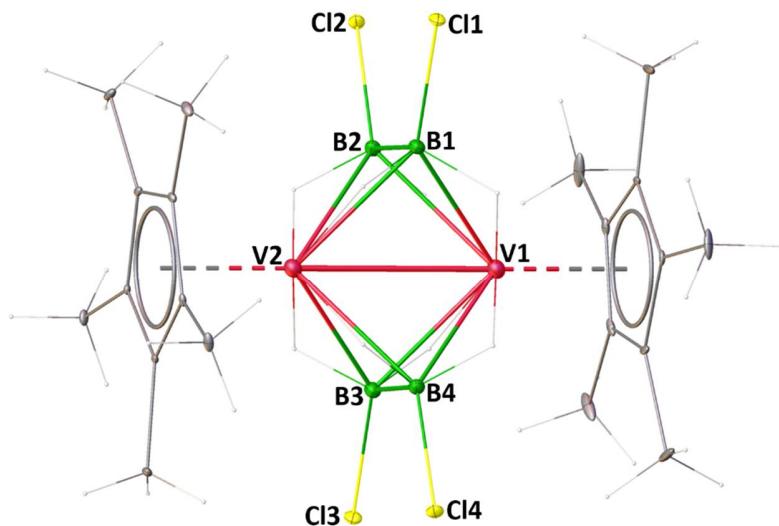


Figure S2. Molecular structure and labeling diagram for compound **2**. Selected bond lengths (\AA) and bond angles ($^{\circ}$): V1-V2 2.8389(6), V1-B1 2.302(3), V1-B2 2.302(3), V1-B3 2.306(3), B1-Cl1 1.844(3), B2-Cl2 1.844(3), B1-B2 1.755(5); B2-V1-V2 52.06(8), B1-V2-B2 44.64(12), B2-B1-V1 67.61(16), B2-V1-B1 44.80(12), B1-V1-B4 86.86(12), V1-B2-Cl2 140.52(18), V2-B2-Cl2 143.26(19), B2-B1-Cl1 115.7(2).

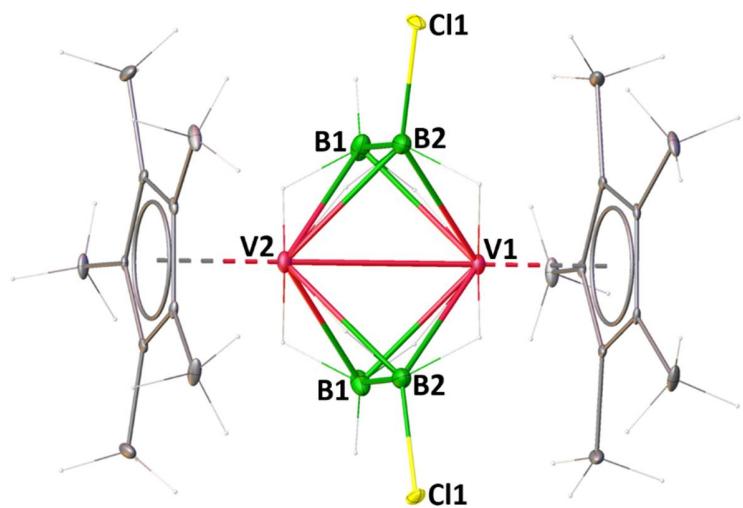


Figure S3. Molecular structure and labeling diagram for compound **3**. Selected bond lengths (\AA) and bond angles ($^{\circ}$): V1-V2 2.7946(7), V1-B1 2.304(3), V1-B2 2.293(3), B2-Cl1 1.791(3), B1-B2 1.746(4); B2-V1-V2 52.40(7), V1-B2-V2 75.14(9), B2-V2-B1 105.14(10), B2-B1-V1 67.61(16), B2-V1-B1 44.64(10), B2-V1-B1 105.02(10), B2-V1-B2 88.38(14), V1-B2-Cl1 143.38(15), V2-B2-Cl1 141.02(15), B1-B2-Cl1 113.63(19).

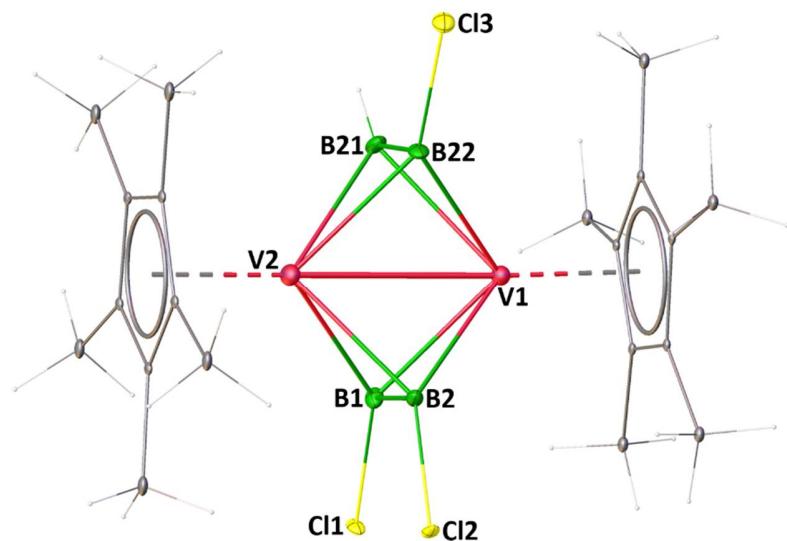


Figure S4. Molecular structure and labeling diagram for compound **4**. Selected bond lengths (\AA) and bond angles ($^{\circ}$): V1-V2 2.8105(15), V1-B2 2.268(9), V1-B22 2.295(7), B1-Cl1 1.825(9), V1-B21 2.328(9), B1-B2 1.732(13), B21-B22 1.749(15); V1-V2-B1 52.0(2), B1-V2-B21 88.6(4), B1-V2-B22 104.2(3), B21-V2-B22 44.3(5), B1-V2-B2 44.6(4), V2-B22-Cl3 143.8(5), V1-B22-Cl3 140.9(5), V2-B22-V1 75.3(2), V2-B21-V1 74.3(3), V2-B1-B2 66.8(4).

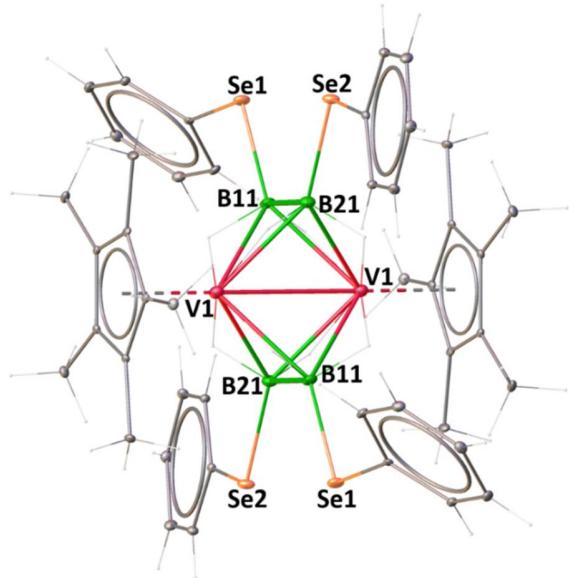


Figure S5. Molecular structure and labeling diagram for compound **5**. Selected bond lengths (\AA) and bond angles ($^{\circ}$): V1-V1 2.8595(10), V1-B11 2.331(3), V1-B21 2.333(3), B11-Se1 2.029(3), B11-B21 1.741(5); V1-V1-B11 52.28(8), B21-V1-B21 104.22(10), B21-V1-B11 87.56(12), B21-V1-B11 44.00(11), V1-B21-V1 75.78(10), V1-B21-Se2 145.44(16), V1-B21-B11 67.38(15).

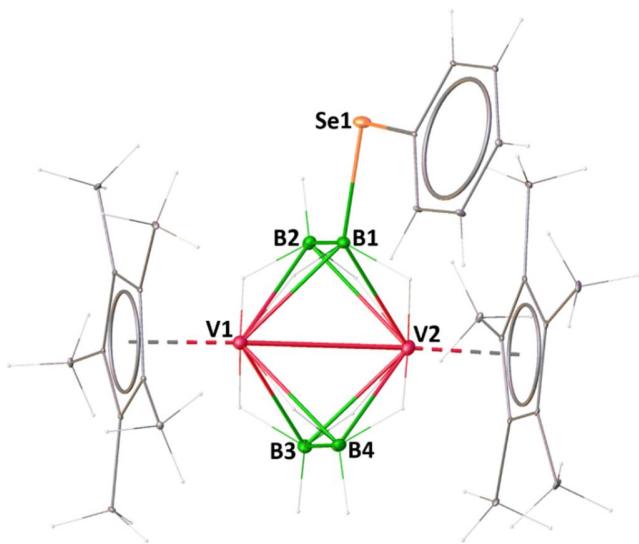


Figure S6. Molecular structure and labeling diagram for compound **6**. Selected bond lengths (\AA) and bond angles ($^{\circ}$): V1-V2 2.7996(5), V1-B2 2.320(2), V1-B1 2.310(2), B1-B2 1.745(3), B3-B4 1.759(3), B1-Se1 2.024(2); V2-V1-B1 52.78(5), V2-V1-B2 52.95(5), B3-V1-B2 88.58(8), B4-V1-B1 88.50(8), B2-V1-B1 44.29(8), B3-V1-B4 44.99(8), V1-B1-V2 74.55(6), B1-B2-V2 67.60(10), V1-B1-Se1 143.06(11).

I.2 Spectroscopic details

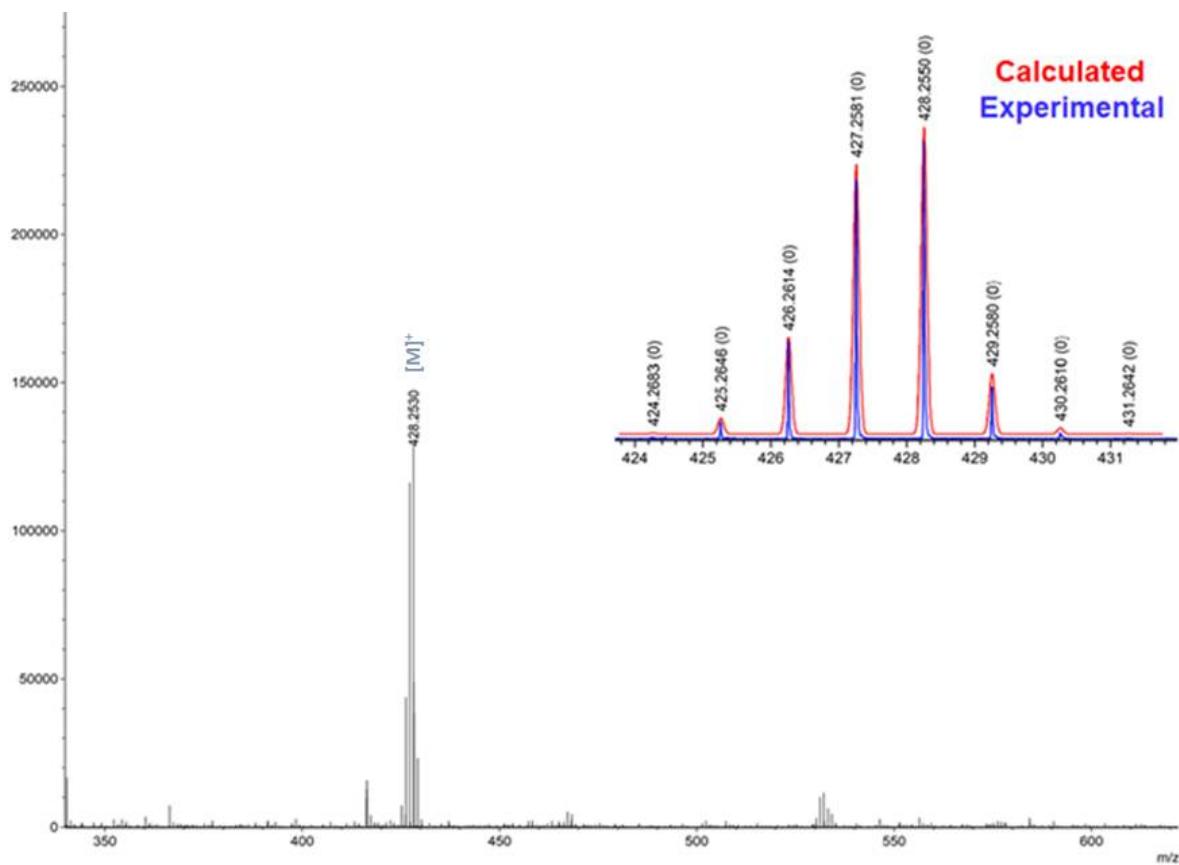


Figure S7. ESI-MS spectrum of compound 1

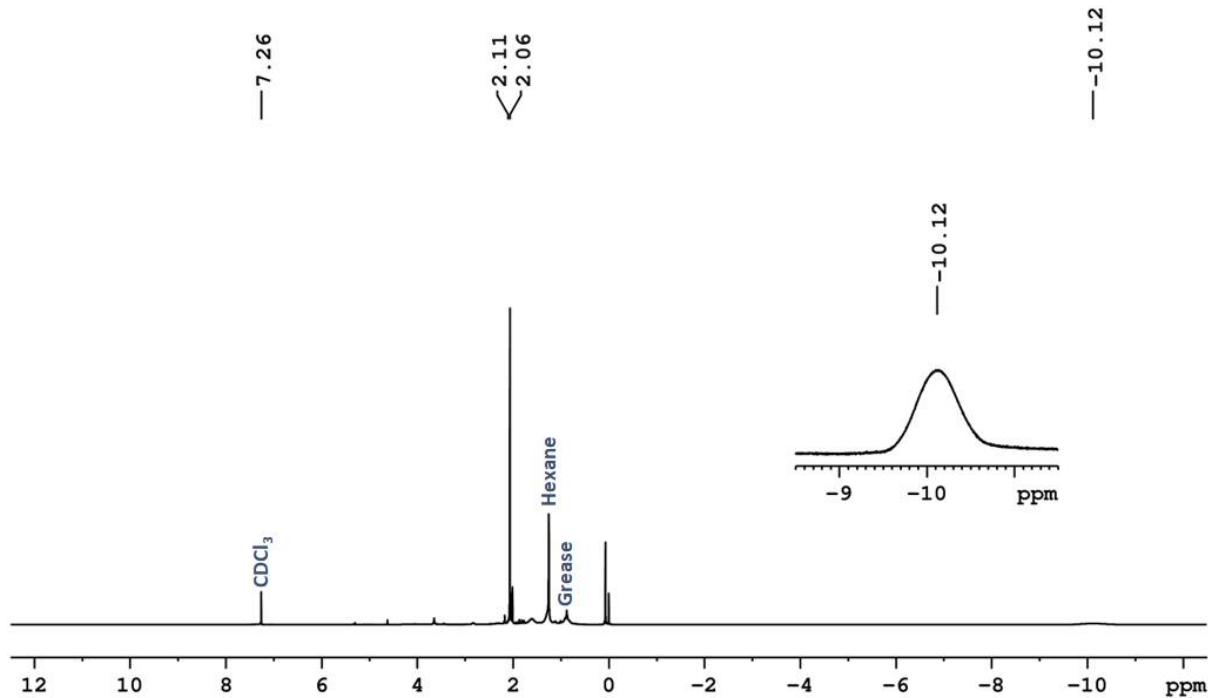


Figure S8. ^1H NMR spectrum of compound 1 in CDCl_3

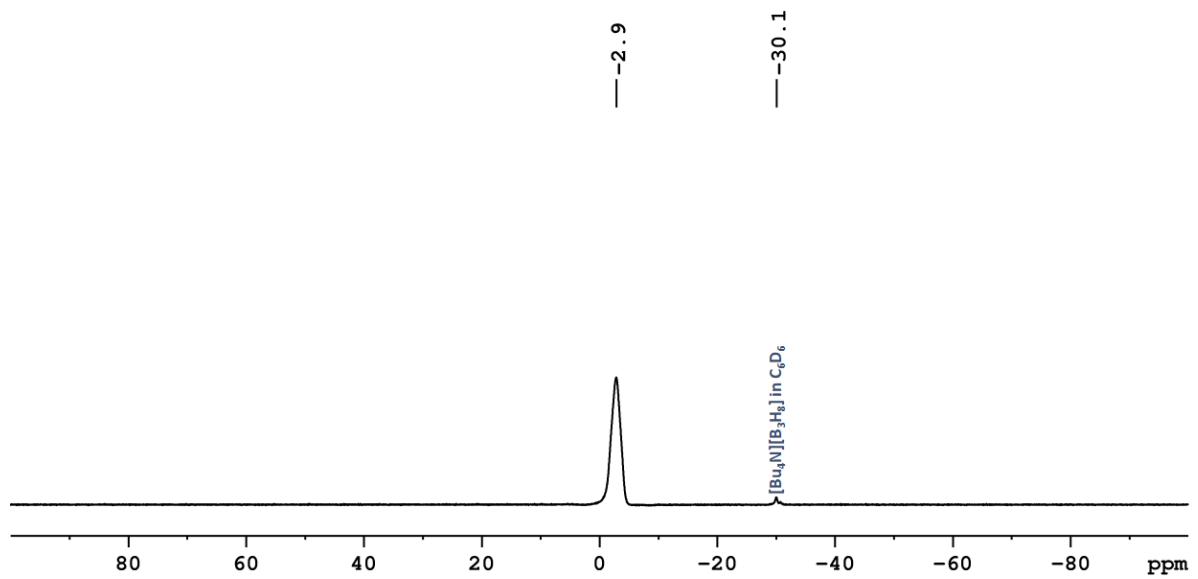


Figure S9. $^{11}\text{B}\{^1\text{H}\}$ NMR spectrum of compound **1** in CDCl_3

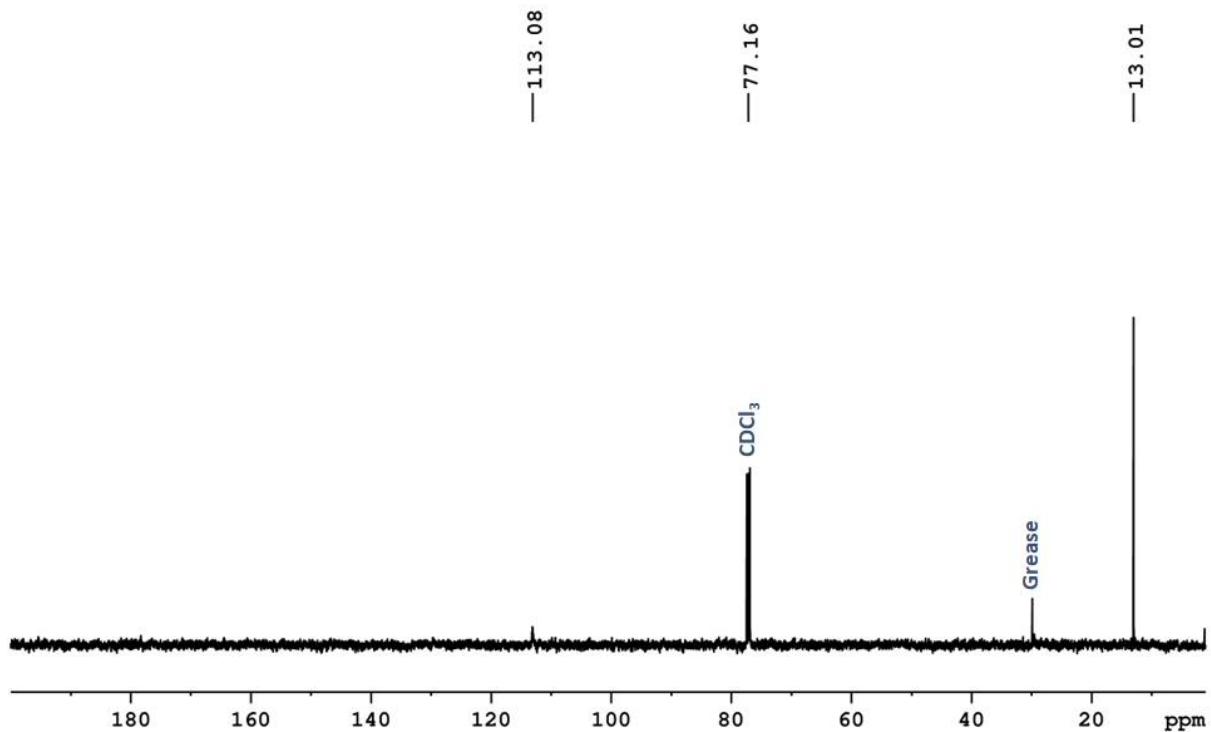


Figure S10. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of compound **1** in CDCl_3

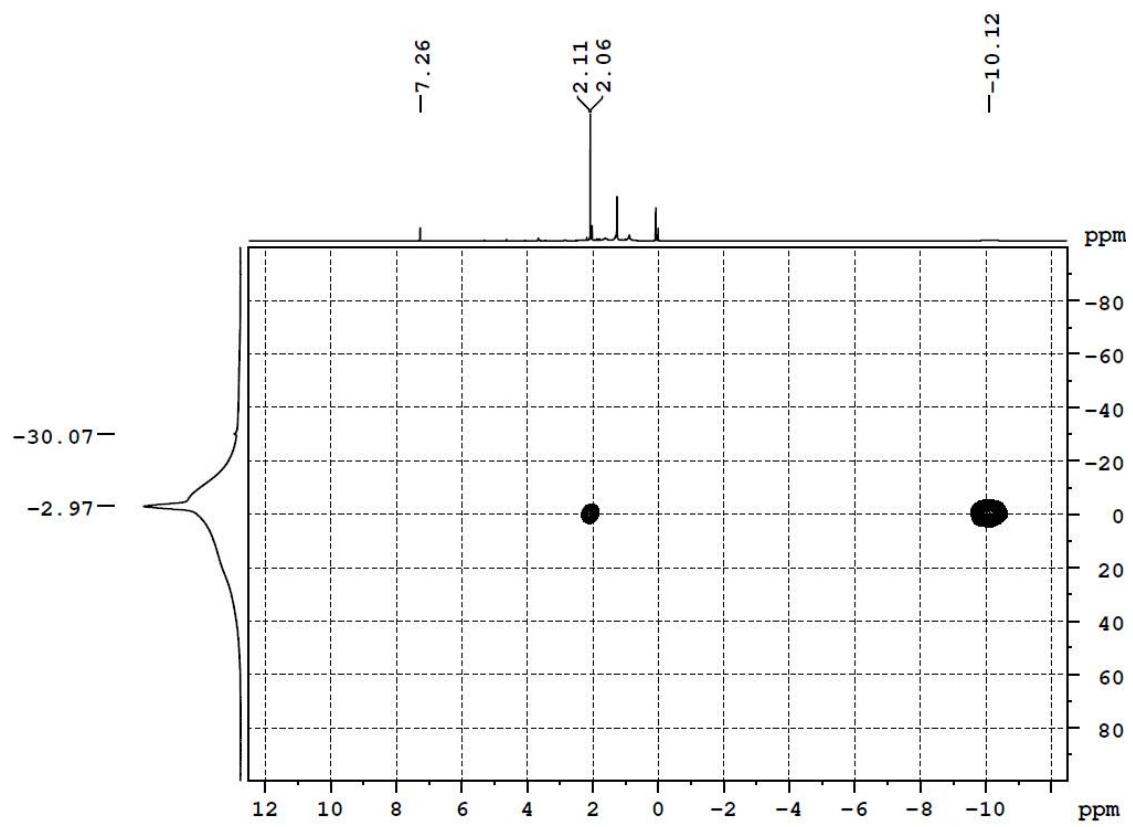


Figure S11. ^1H - ^{11}B HSQC NMR spectrum of compound **1** in CDCl_3

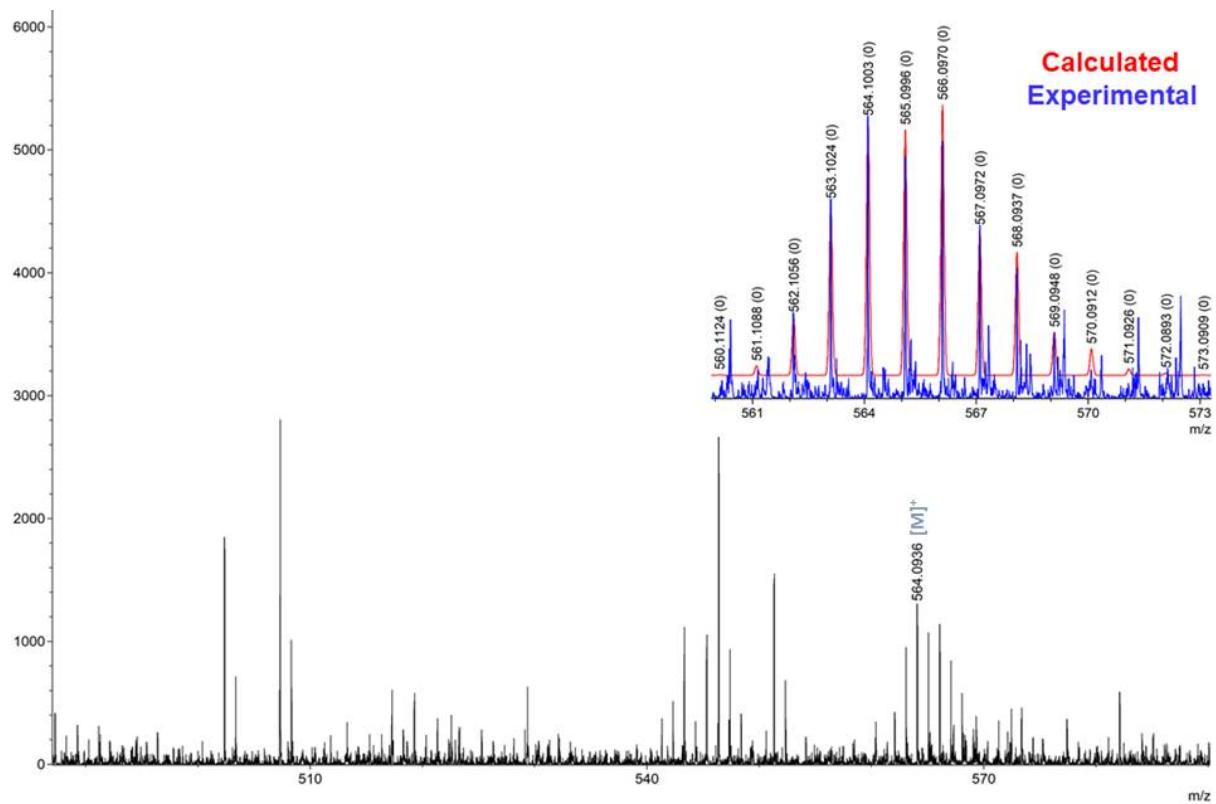


Figure S12. ESI-MS spectrum of compound **2**

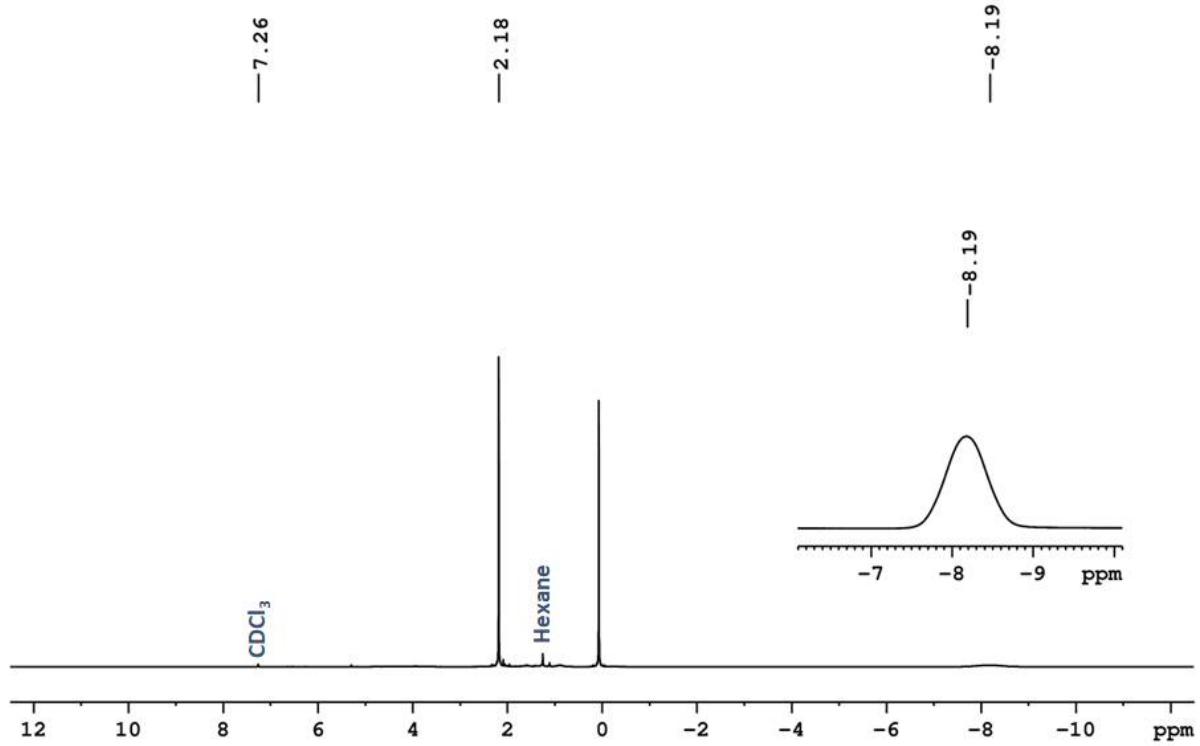


Figure S13. ¹H NMR spectrum of compound **2** in CDCl_3

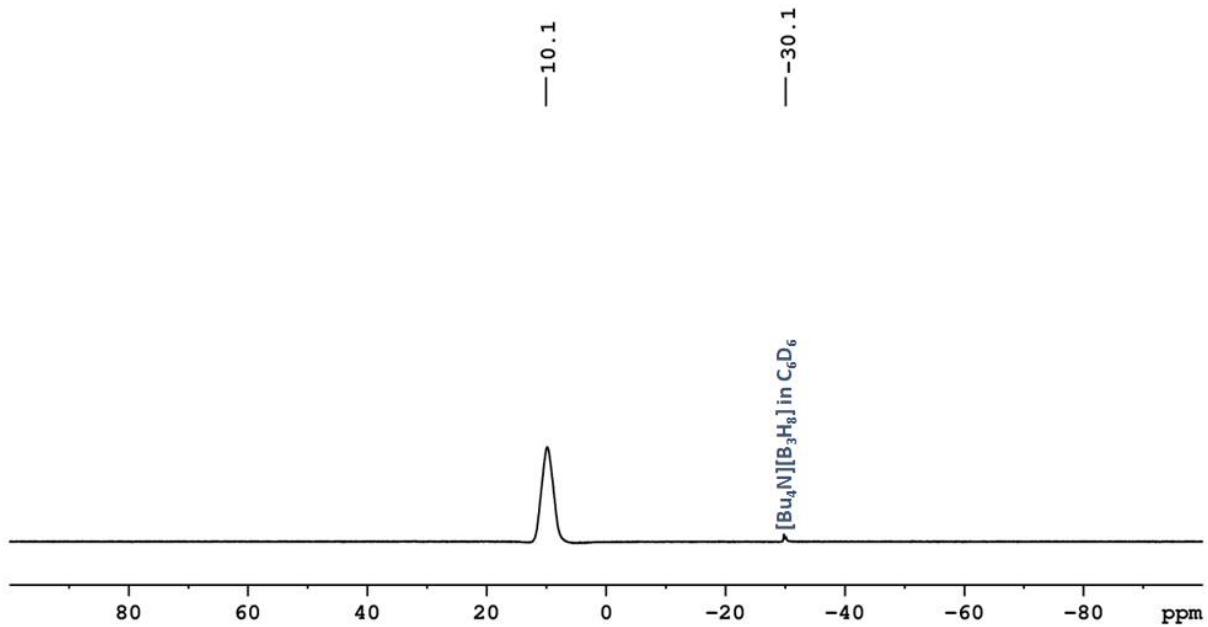


Figure S14. ¹¹B{¹H} NMR spectrum of compound **2** in CDCl_3

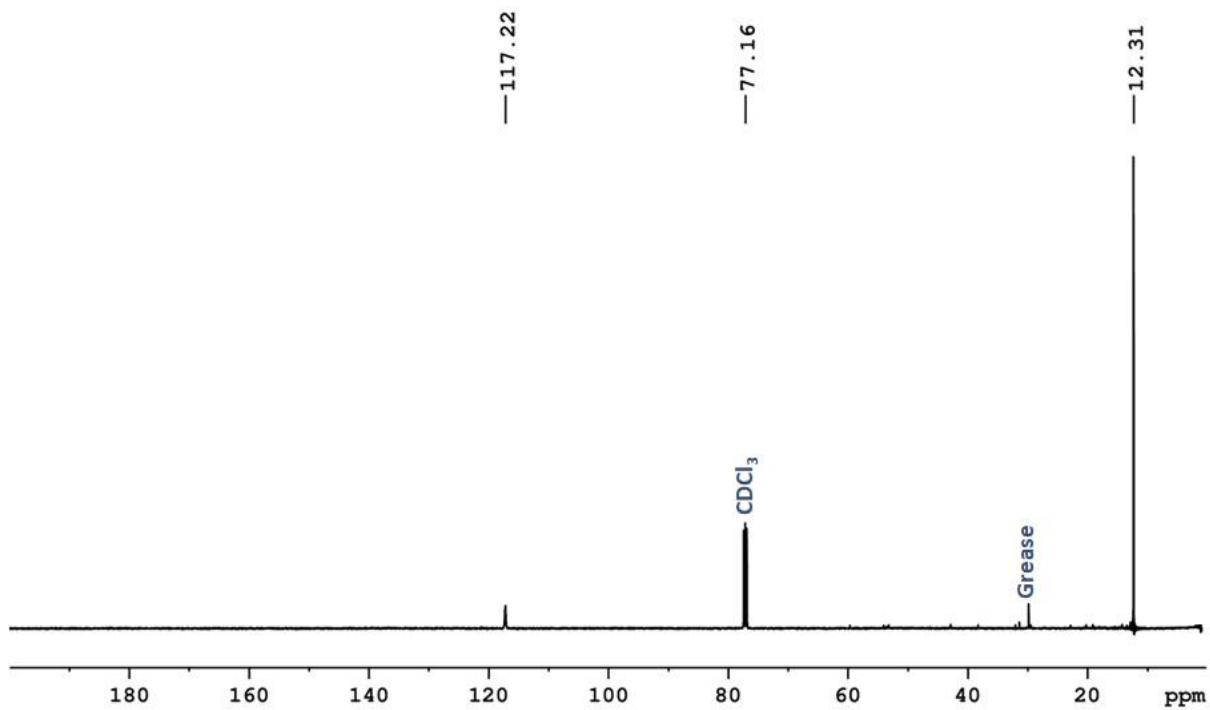


Figure S15. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of compound **2** in CDCl_3

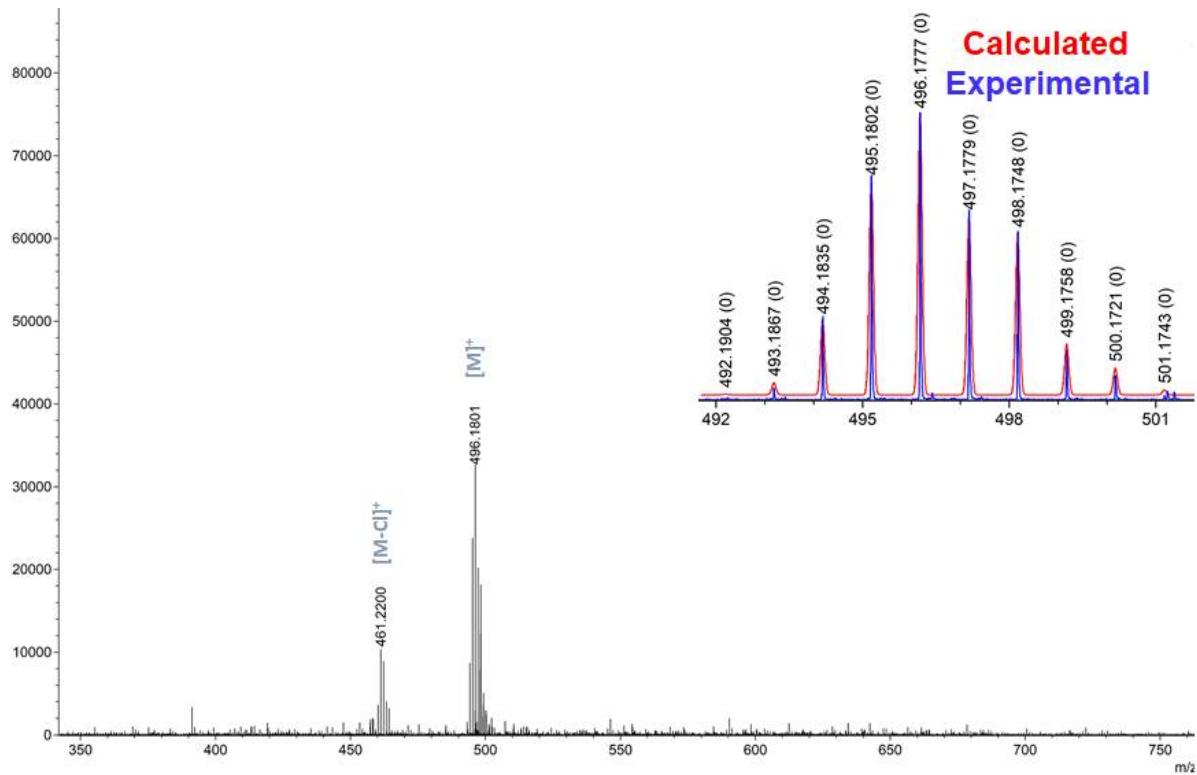


Figure S16. ESI-MS spectrum of compound **3**

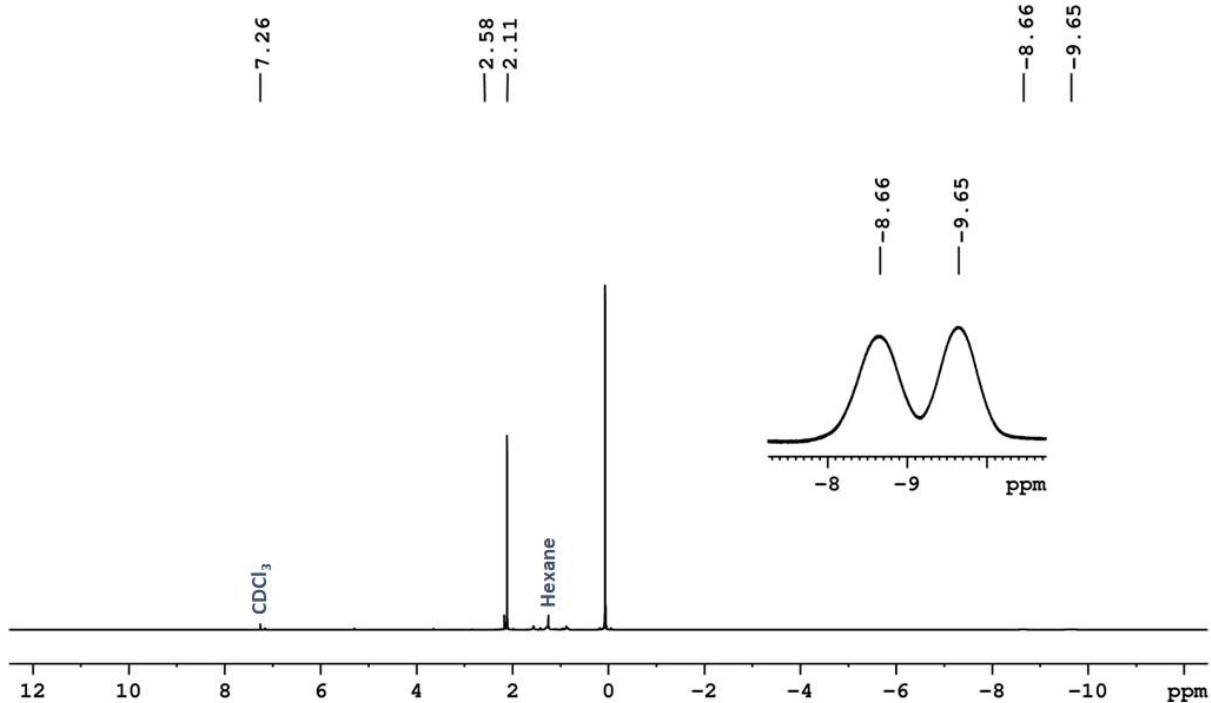


Figure S17. ^1H NMR spectrum of compound 3 in CDCl_3

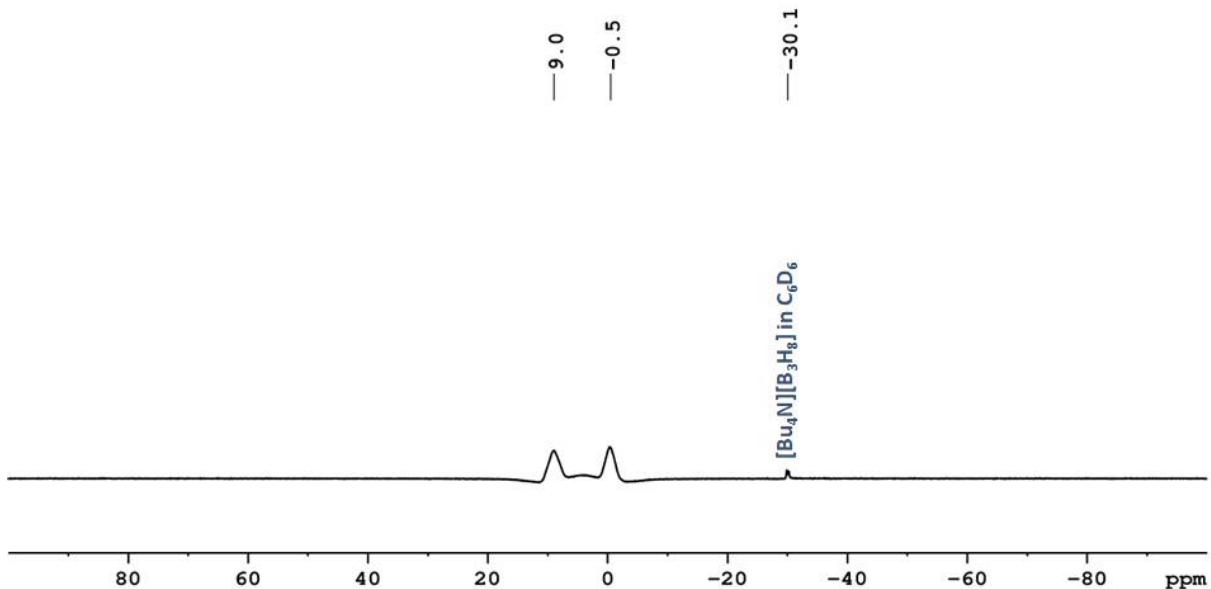


Figure S18. $^{11}\text{B}\{^1\text{H}\}$ NMR spectrum of compound 3 in CDCl_3

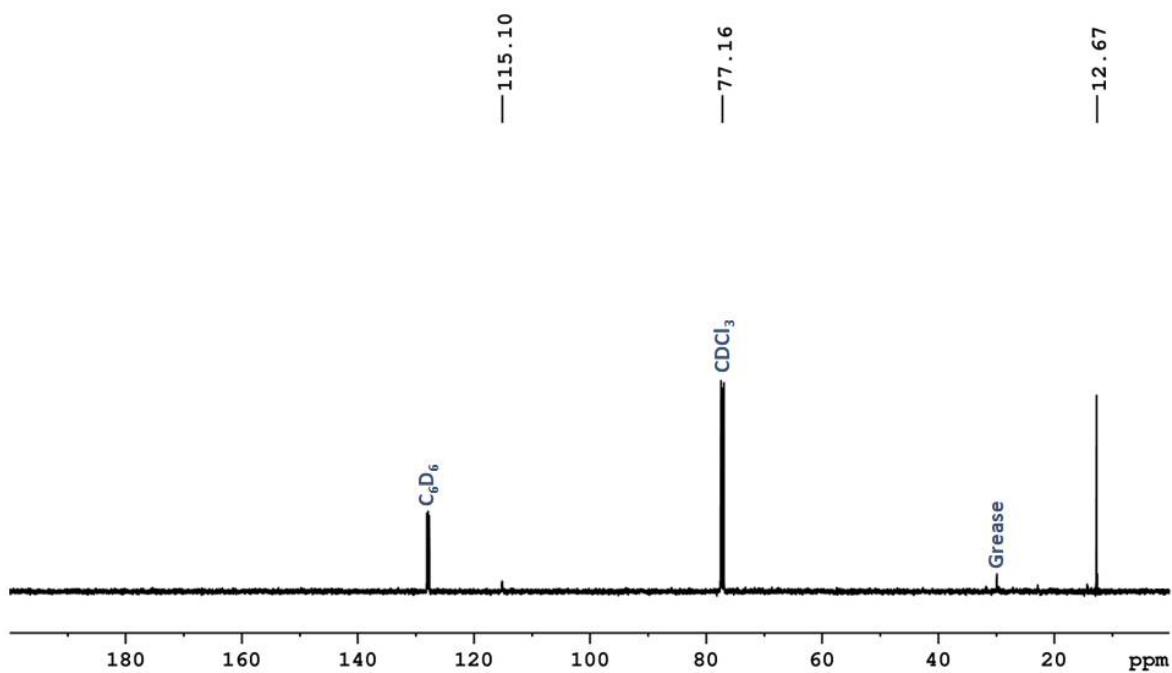


Figure S19. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of compound **3** in CDCl₃

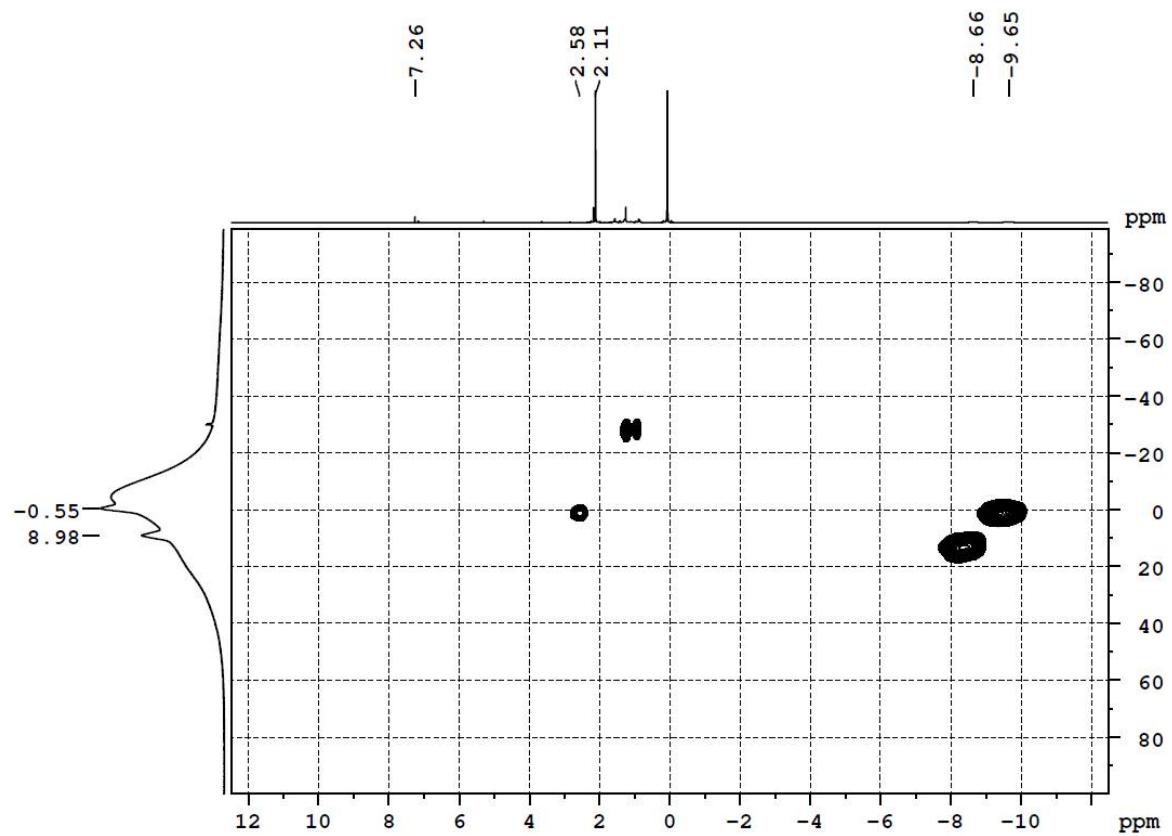


Figure S20. ^1H - ^{11}B HSQC NMR spectrum of compound **3** in CDCl₃

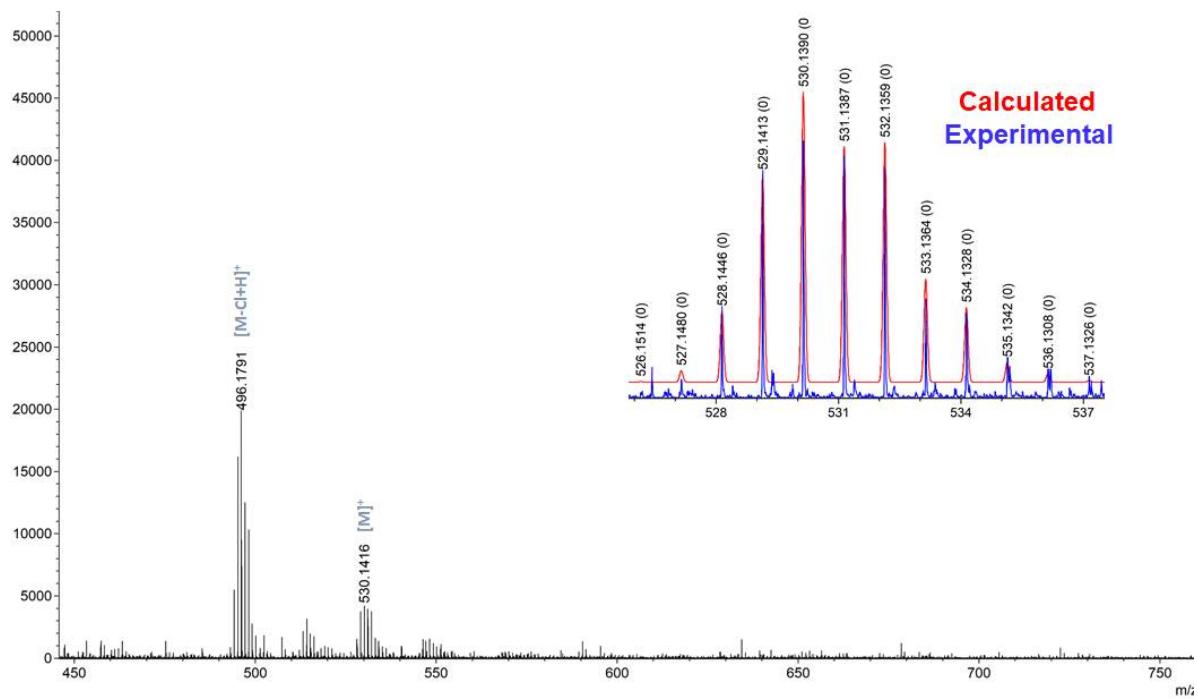


Figure S21. ESI-MS spectrum of compound 4

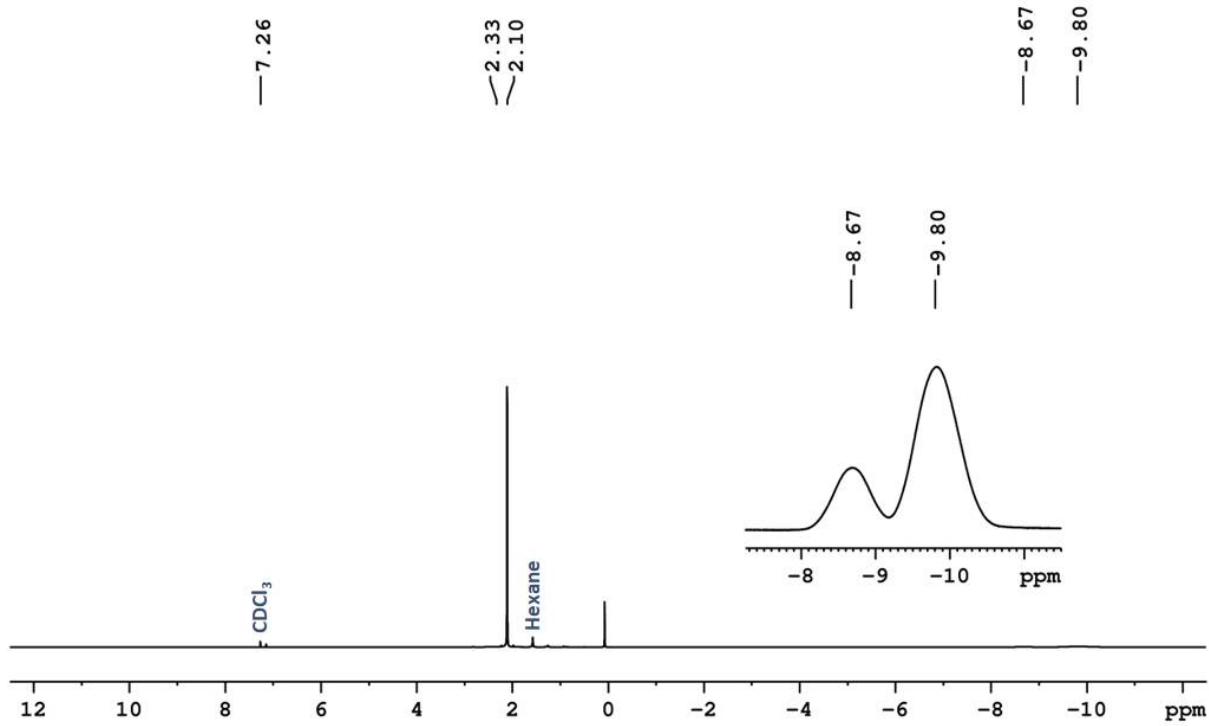


Figure S22. ^1H NMR spectrum of compound 4 in CDCl_3

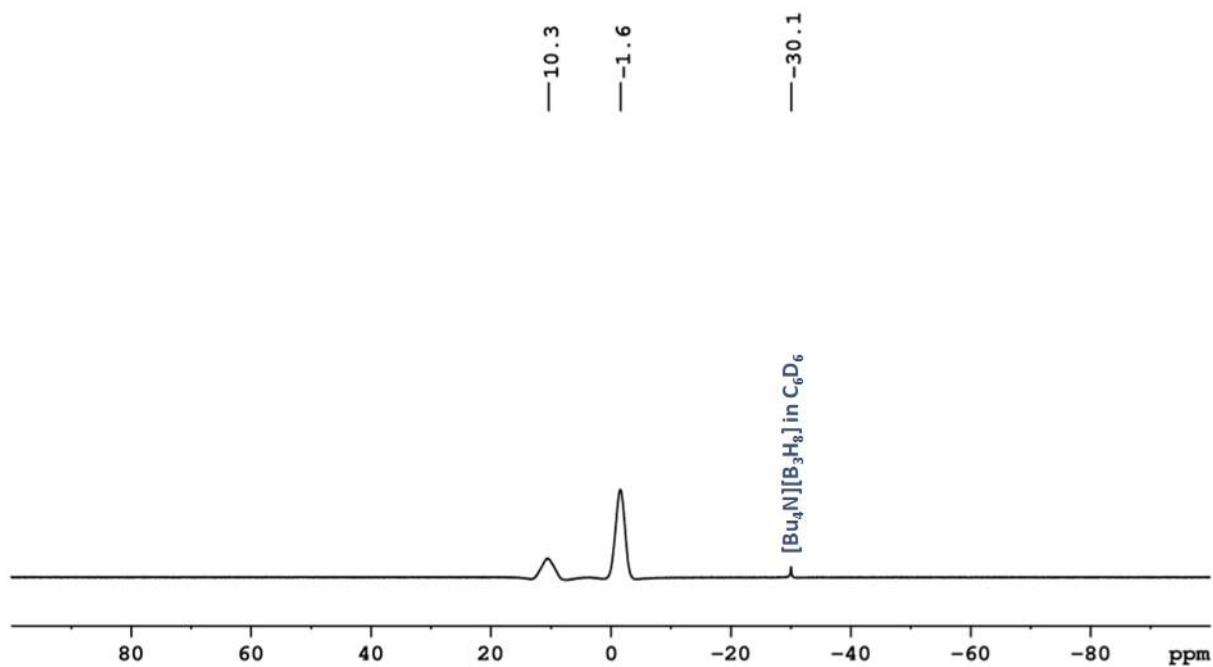


Figure S23. $^{11}\text{B}\{^1\text{H}\}$ NMR spectrum of compound **4** in CDCl_3

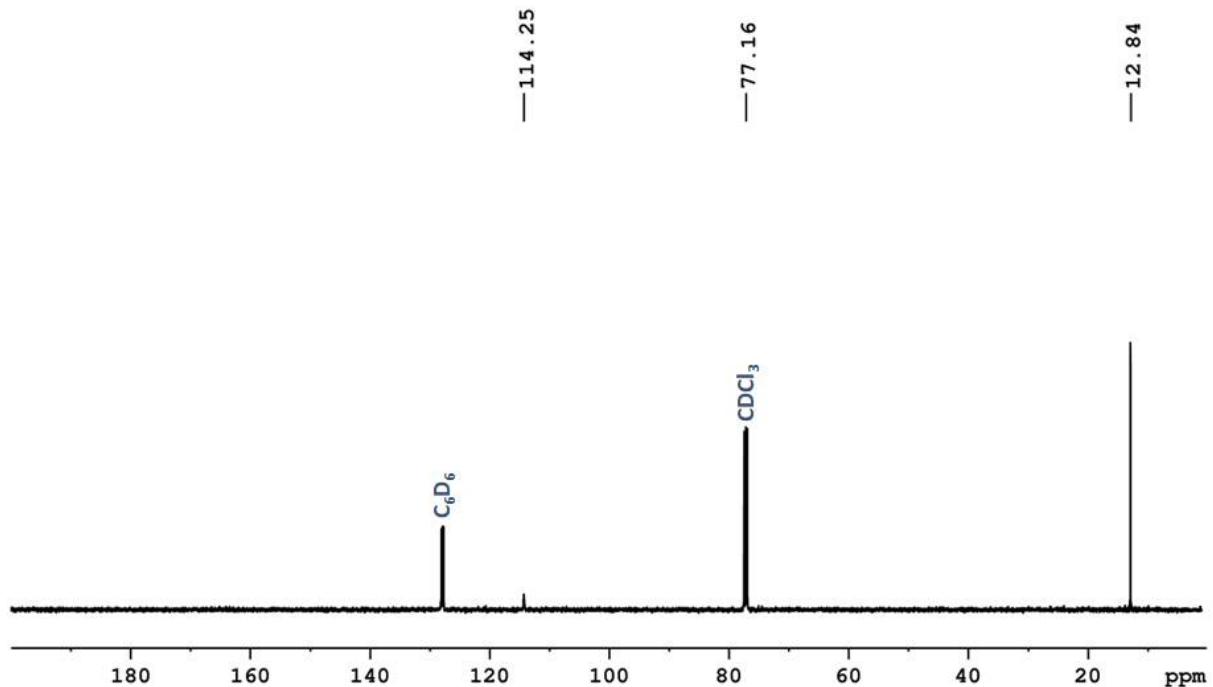


Figure S24. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of compound **4** in CDCl_3

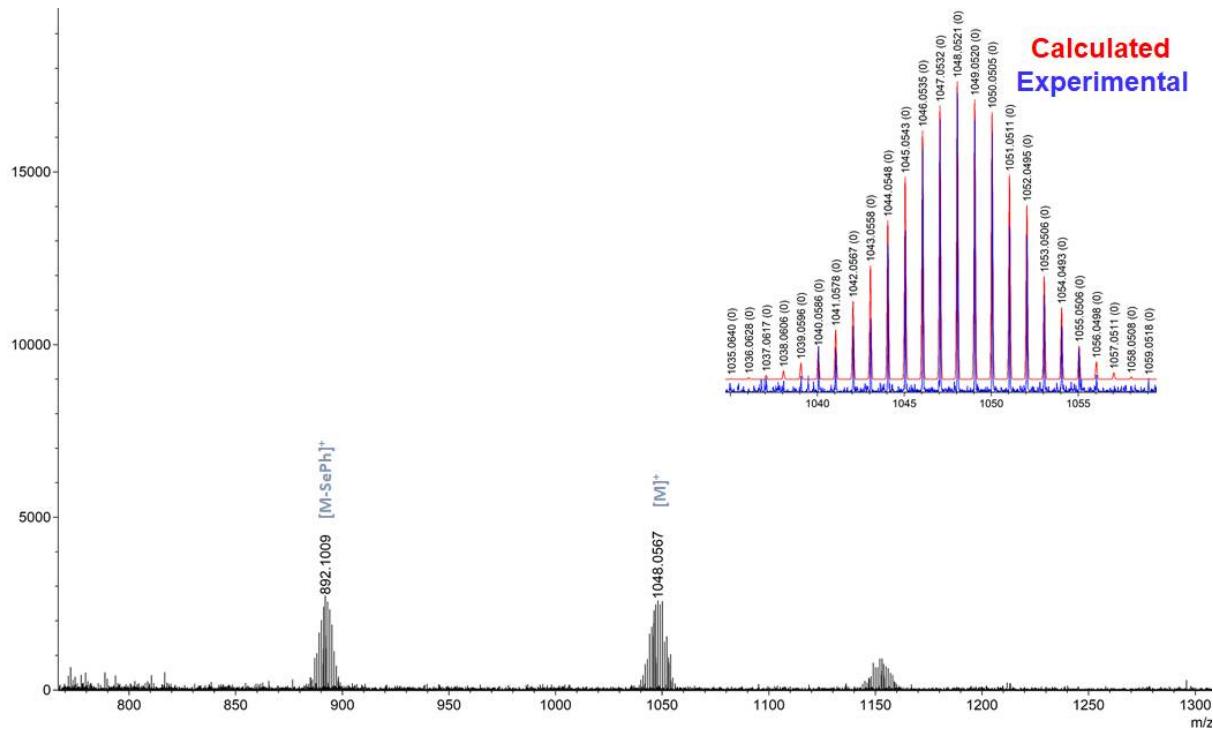


Figure S25. ESI-MS spectrum of compound 5

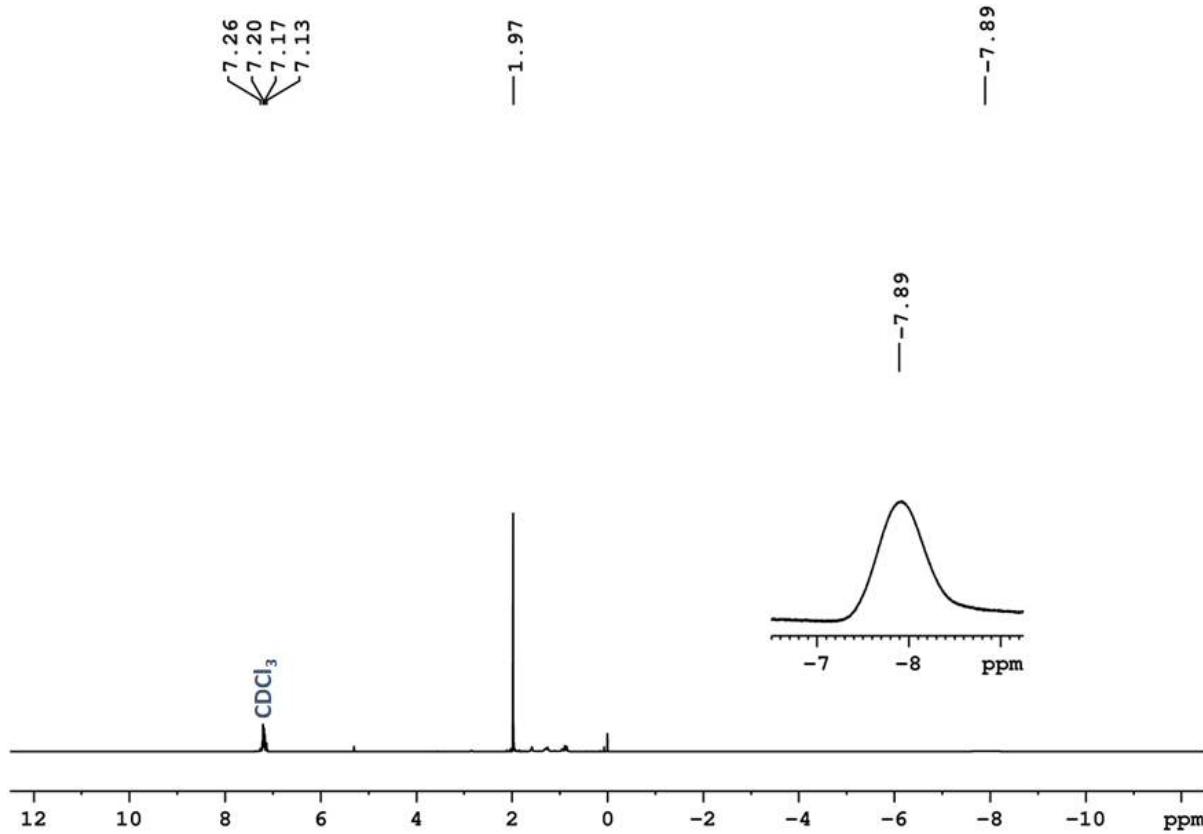


Figure S26. ^1H NMR spectrum of compound 5 in CDCl_3

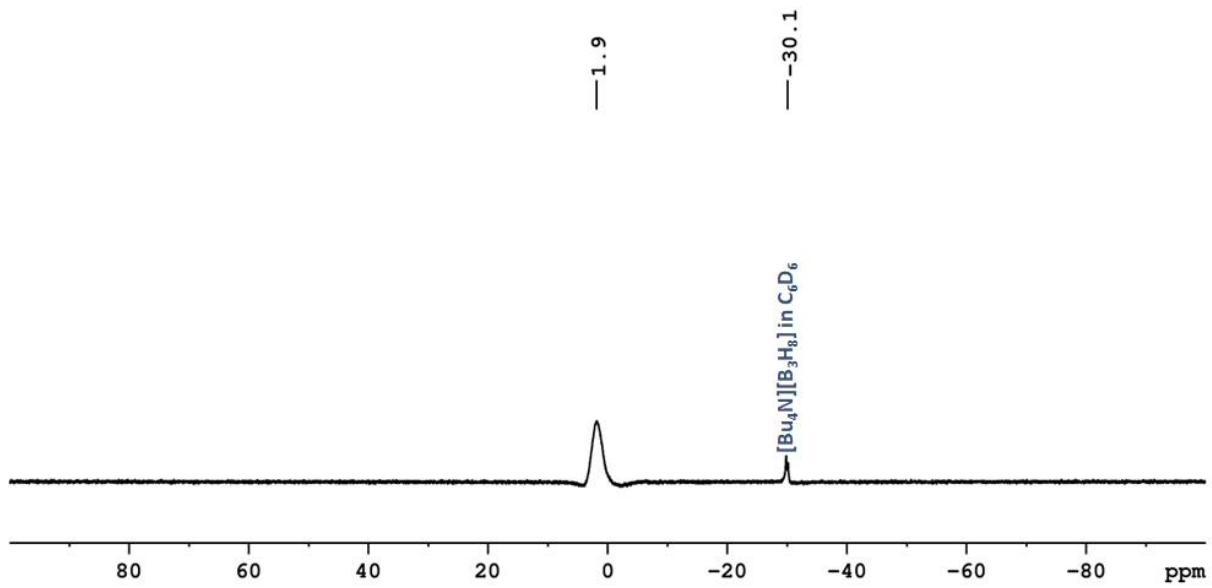


Figure S27. $^{11}\text{B}\{^1\text{H}\}$ NMR spectrum of compound 5 in CDCl_3

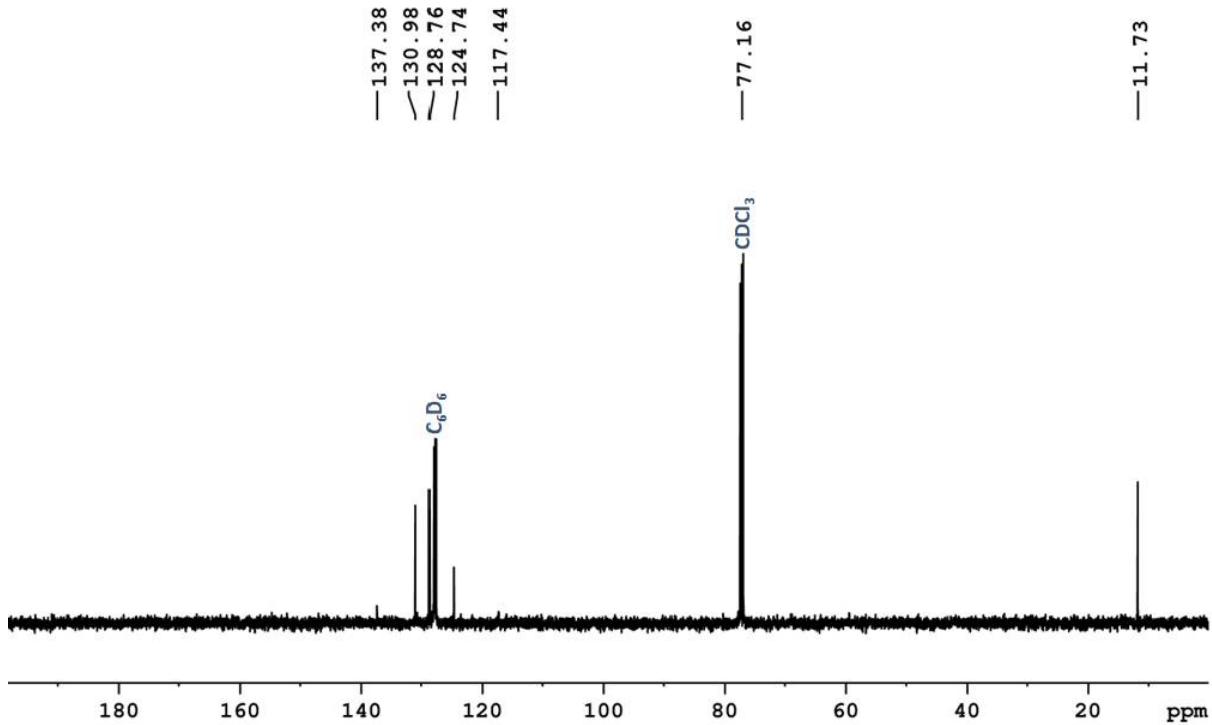


Figure S28. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of compound 5 in CDCl_3

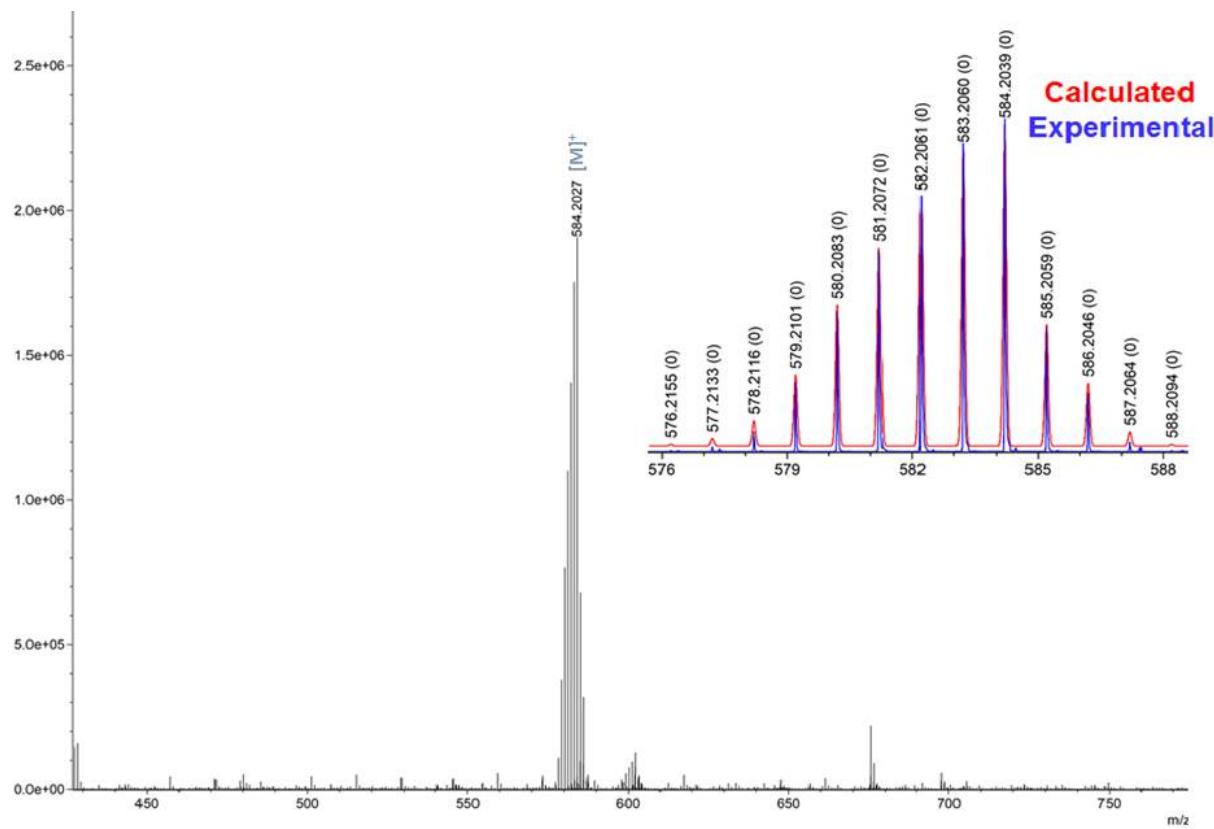


Figure S29. ESI-MS spectrum of compound **6** in CDCl_3

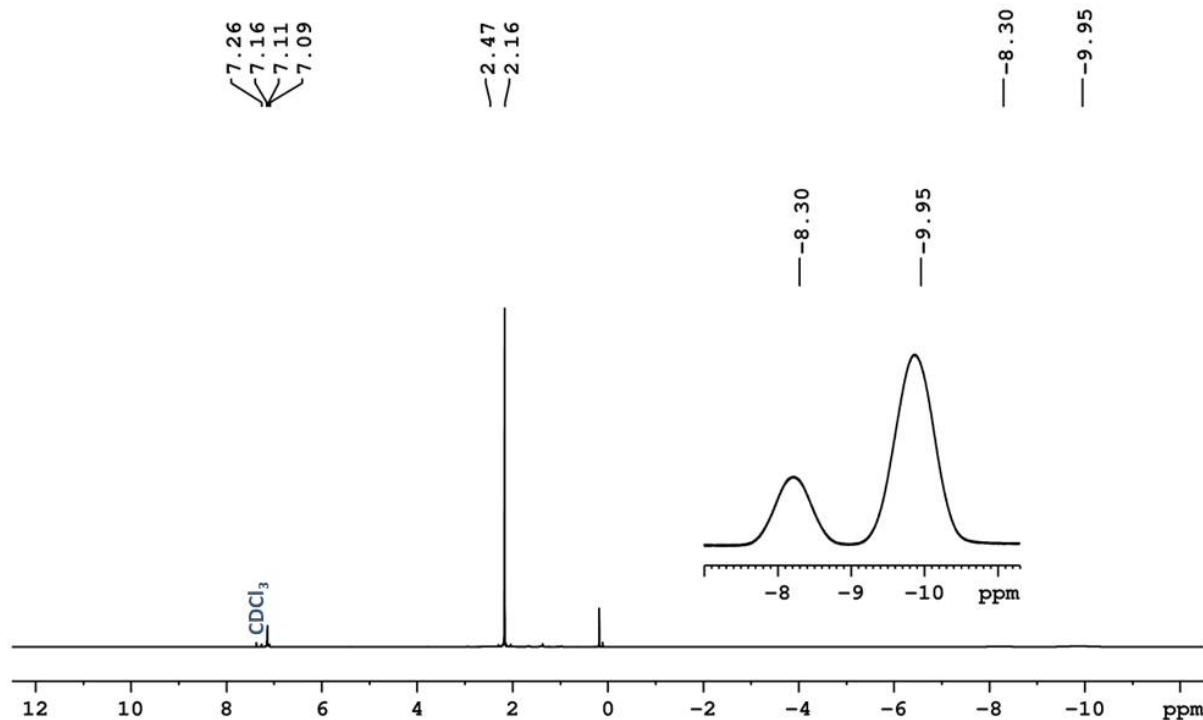


Figure S30. ^1H NMR spectrum of compound **6** in CDCl_3

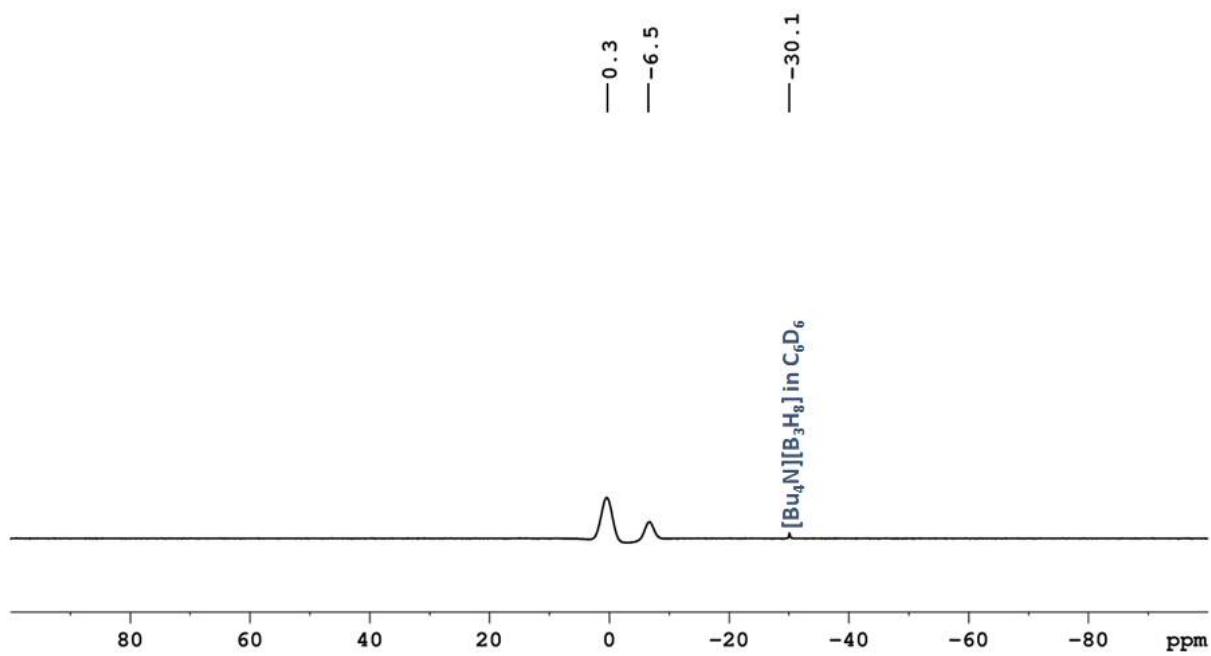


Figure S31. $^{11}\text{B}\{\text{H}\}$ NMR spectrum of compound **6** in CDCl_3

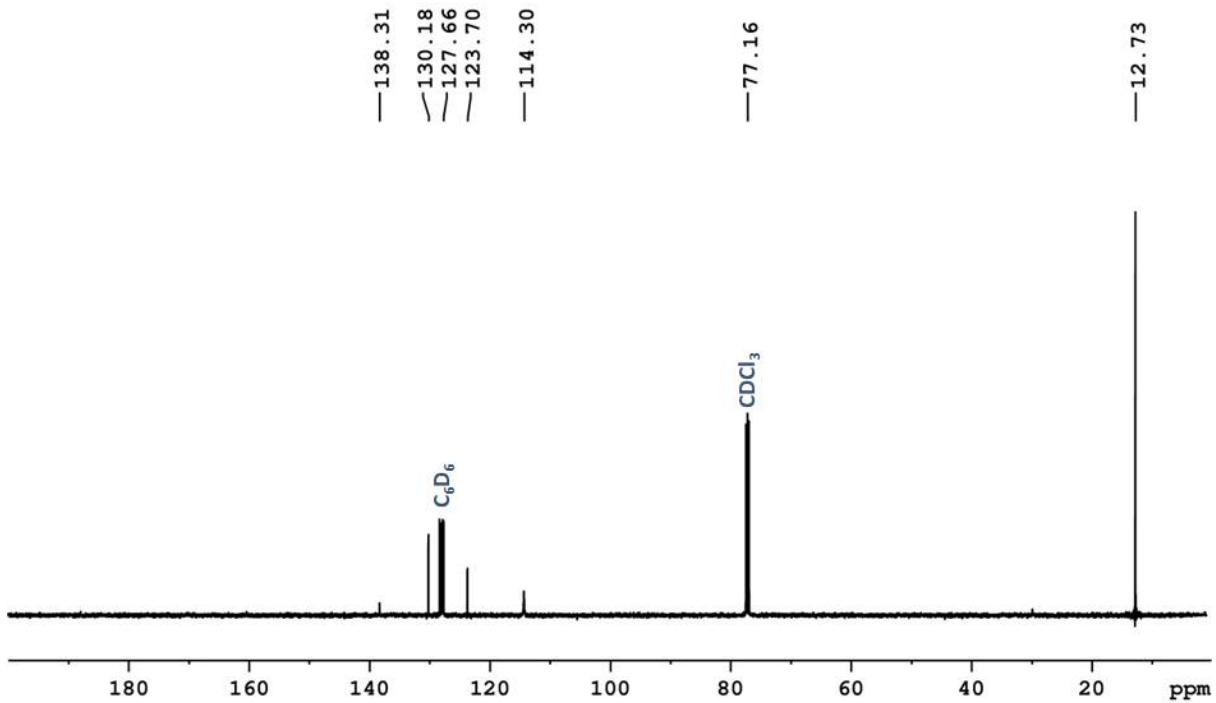


Figure S32. $^{13}\text{C}\{\text{H}\}$ NMR spectrum of compound **6** in CDCl_3

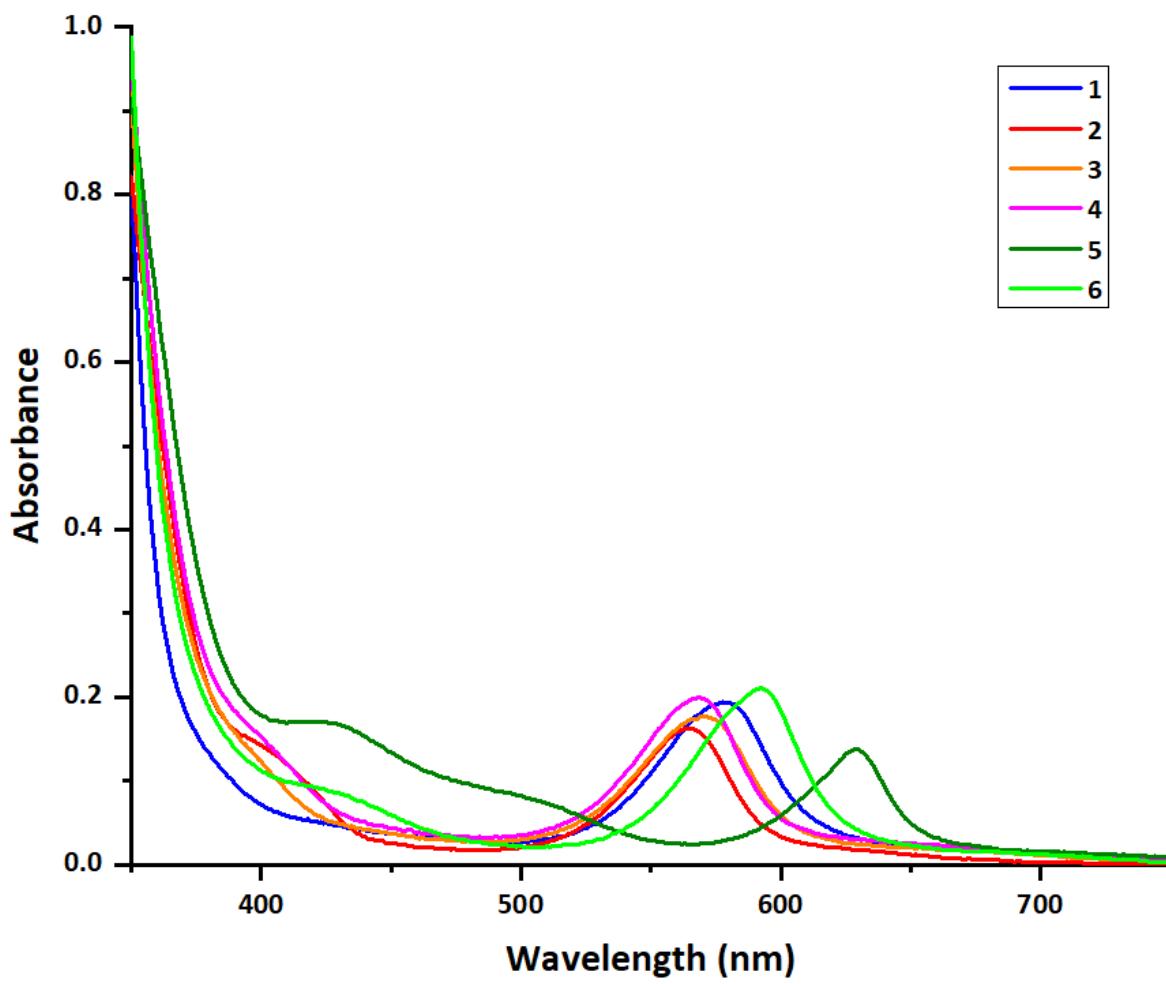


Figure S33. UV-Vis spectra of compounds **1-6** in CH_2Cl_2

II. Supplementary Data

Table S1. Experimental and calculated bond distances of **1-6** (**E** = Cl or Se).

| | d_{V-V} | | d_{V-B} | | d_{B-B} | | d_{B-E} | |
|----------|------------------------|------|------------------------|------|------------------------|------|------------------------|------|
| | Expt. | Cal. | Expt. | Cal. | Expt. | Cal. | Expt. | Cal. |
| 1 | 2.78 | 2.73 | 2.30 | 2.27 | 1.75 | 1.75 | - | - |
| 2 | 2.83 | 2.79 | 2.30 | 2.28 | 1.75 | 1.76 | 1.84 | 1.85 |
| 3 | 2.79 | 2.76 | 2.29 | 2.27 | 1.74 | 1.75 | 1.79 | 1.86 |
| 4 | 2.81 | 2.77 | 2.29 | 2.27 | 1.74 | 1.75 | 1.84 | 1.85 |
| 5 | 2.85 | 2.79 | 2.32 | 2.26 | 1.74 | 1.74 | 2.03 | 2.01 |
| 6 | 2.79 | 2.75 | 2.30 | 2.26 | 1.75 | 1.74 | 2.02 | 2.00 |

Table S2. Calculated natural charges (**q_V**, **q_B**, and **q_E**), natural valance population (**Pop**) and HOMO–LUMO gaps of **1-6** (**E** = Cl or Se).

| | q_V | q_B | q_E | Pop(V_{val}) | Pop(B_{val}) | Pop(E_{val}) | ΔE_{H-L} (ev) |
|----------|----------------------|----------------------|----------------------|-----------------------------|-----------------------------|-----------------------------|------------------------------|
| 1 | 0.527 | -0.428 | - | 4.480 | 3.408 | - | 2.72 |
| | 0.527 | -0.428 | - | 4.480 | 3.407 | - | |
| | | -0.428 | - | | 3.408 | - | |
| | | -0.428 | - | | 3.407 | - | |
| 2 | 0.487 | -0.241 | -0.191 | 4.519 | 3.208 | 7.180 | 2.76 |
| | 0.487 | -0.242 | -0.189 | 4.519 | 3.209 | 7.179 | |
| | | -0.231 | -0.190 | | 3.198 | 7.179 | |
| | | -0.231 | -0.191 | | 3.199 | 7.180 | |
| 3 | 0.510 | -0.452 | - | 4.497 | 3.432 | - | 2.74 |
| | 0.510 | -0.212 | -0.212 | 4.497 | 3.179 | 7.201 | |
| | | -0.452 | - | | 3.432 | - | |
| | | -0.212 | -0.212 | | 3.179 | 7.201 | |
| 4 | 0.496 | -0.463 | - | 4.510 | 3.443 | - | 2.76 |
| | 0.496 | -0.217 | -0.207 | 4.510 | 3.184 | 7.197 | |
| | | -0.226 | -0.193 | | 3.194 | 7.182 | |
| | | -0.229 | -0.194 | | 3.197 | 7.183 | |
| 5 | 0.547 | -0.456 | 0.247 | 4.457 | 3.435 | 5.734 | 2.32 |
| | 0.548 | -0.467 | 0.259 | 4.456 | 3.446 | 5.722 | |
| | | -0.457 | 0.249 | | 3.435 | 5.732 | |
| | | -0.467 | 0.259 | | 3.445 | 5.722 | |
| 6 | 0.474 | -0.515 | 0.269 | 4.534 | 3.494 | 5.712 | 1.39 |
| | 0.474 | -0.468 | - | 4.533 | 3.449 | - | |
| | | -0.462 | - | | 3.449 | - | |
| | | -0.464 | - | | 3.443 | - | |

Table S3. Electron density, $\rho(r)$, Laplacian of electron density, $\nabla^2\rho(r)$, total energy density, $H(r)$, potential energy density, $V(r)$ and kinetic energy density, $G(r)$ in a.u. of selected bond critical points (BCPs) of **1**, **2** and **5**.

| Compound | BCP | $\rho(r)$ | $\nabla^2\rho(r)$ | $H(r)$ | $G(r)$ | $V(r)$ |
|----------|-------|-----------|-------------------|---------|--------|---------|
| 1 | V1-V2 | 0.0487 | -0.0039 | -0.0173 | 0.0164 | -0.0337 |
| 2 | V1-V2 | 0.0464 | -0.0066 | -0.0157 | 0.0141 | -0.0299 |
| 5 | V1-V1 | 0.0461 | -0.0005 | -0.0152 | 0.0151 | -0.0304 |

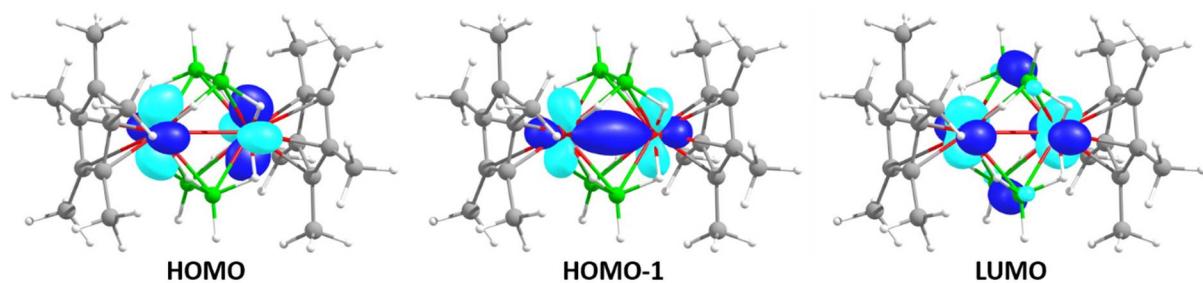


Figure S34. Selected frontier molecular orbitals of compound **1**.

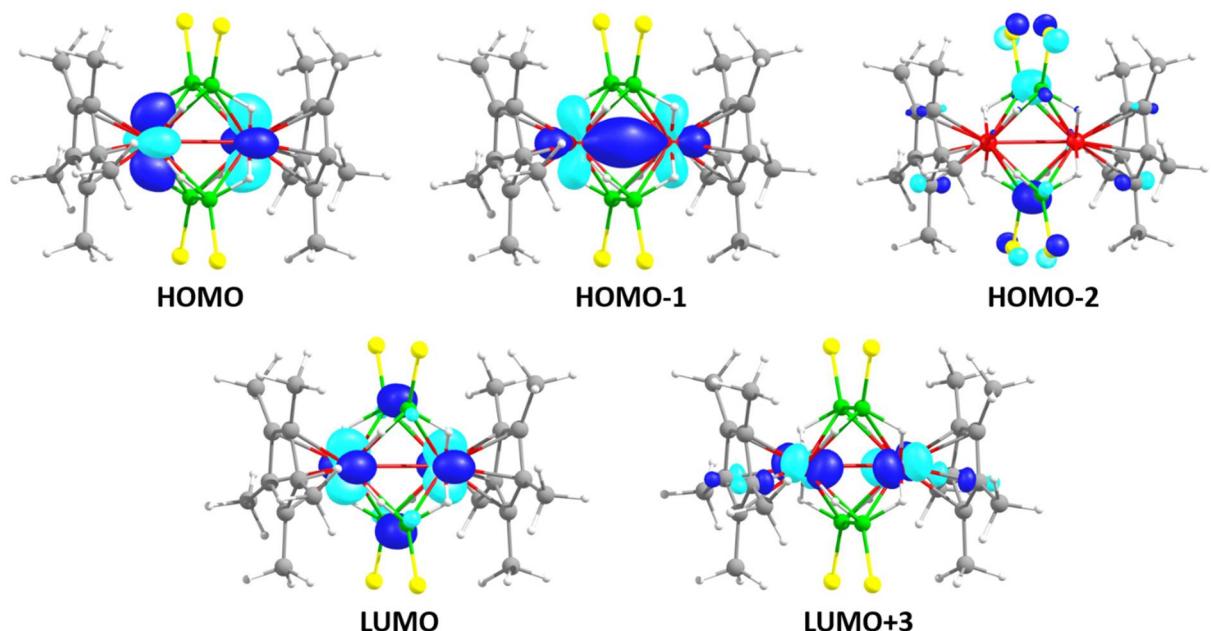


Figure S35. Selected frontier molecular orbitals of compound **2**.

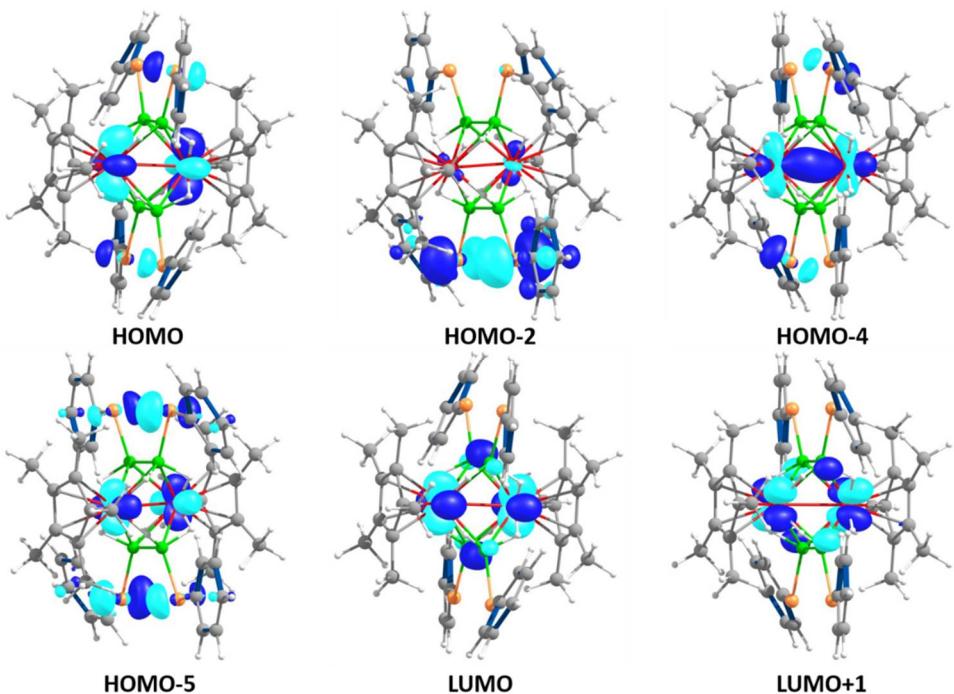


Figure S36. Selected frontier molecular orbitals of compound 5.

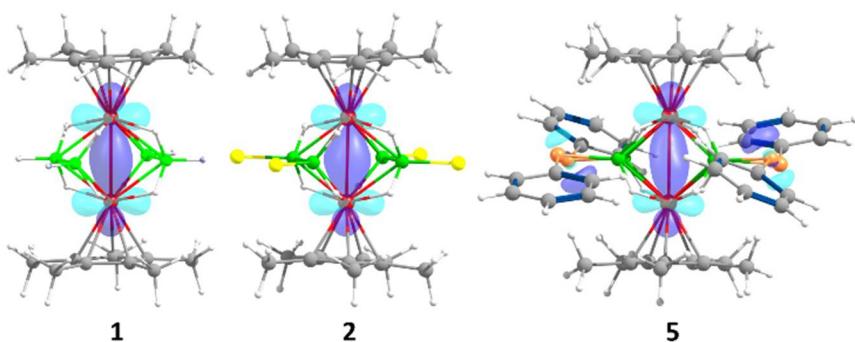


Figure S37. HOMO-1 of 1 and 2, HOMO-4 of 5 showing V-V interactions.

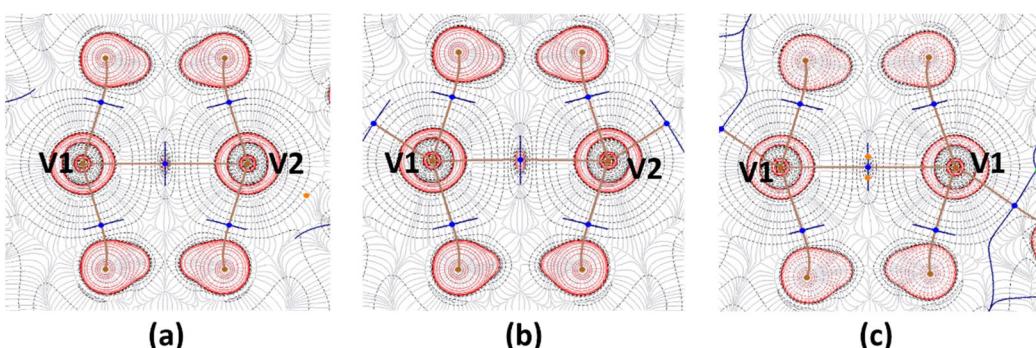


Figure S38. Contour line diagrams of the Laplacian of the electron density, $\nabla^2\rho(r)$ of V1-V2-H, V1-V2-H and V1-V1-H planes in **1** (a), **2** (b) and **5** (c) respectively. Solid red lines indicate areas of charge concentration ($\nabla^2\rho(r)<0$) while dashed black lines show areas of charge depletion ($\nabla^2\rho(r)>0$). Blue dots indicate BCPs.

Table S4. TD-DFT calculated electronic transition configuration for **1** along with their corresponding excitation energies, wavelength and oscillator strengths

HOMO is 114

| No | Transition configurations (%) ^[a] | Excitation Energy (ev) | Wavelength (nm) | | Osc, Strength |
|----|---|---------------------------|-----------------|-------|------------------|
| | | | Cal. | Expt. | |
| 1 | HOMO-1→LUMO(98) | 1.7797 | 696.6578 | | 0.0 |
| 2 | HOMO-1→LUMO+3(12) | 2.2491 | 551.2613 | 579 | 0.0598 |
| | HOMO→LUMO(89) | | | | |
| 3 | HOMO→LUMO+2(100) | 2.4513 | 505.7895 | | 0.0008 |
| 4 | HOMO→LUMO+1(97) | 2.5359 | 488.9159 | | 0.0 |
| 5 | HOMO-1→LUMO+1(94) | 2.6002 | 476.8256 | | 0.0001 |
| 6 | HOMO-1→LUMO+2(99) | 2.6908 | 460.7707 | | 0.0 |
| 7 | HOMO-1→LUMO+4(95) | 2.9858 | 415.2461 | | 0.0 |
| 8 | HOMO-2→LUMO(79) | 2.9896 | 414.7183 | | 0.0 |
| | HOMO→LUMO+4(18) | | | | |
| 9 | HOMO→LUMO+3(93) | 3.0186 | 410.7340 | | 0.0 |
| 10 | HOMO-3→LUMO(45) | 3.0706 | 403.7783 | | 0.0 |
| | HOMO-1→LUMO+5(53) | | | | |

^[a]components with greater than 10% contribution shown

Table S5. TD-DFT calculated electronic transition configuration for **2** along with their corresponding excitation energies, wavelength and oscillator strengths

HOMO is 146

| No | Transition configurations (%) ^[a] | Excitation Energy (ev) | Wavelength (nm) | | Osc, Strength |
|----|---|---------------------------|-----------------|-----------------------|------------------|
| | | | Cal. | Expt. | |
| 1 | HOMO-1→LUMO(95) | 2.0406 | 607.5869 | | 0.0001 |
| 2 | HOMO→LUMO+1(100) | 2.1822 | 568.1614 | | 0.0 |
| 3 | HOMO-1→LUMO+3(11) | 2.2947 | 540.3067 | 565 | 0.0591 |
| | HOMO→LUMO(89) | | | | |
| 4 | HOMO→LUMO+2(97) | 2.5959 | 477.6154 | | 0.0 |
| 5 | HOMO-1→LUMO+1(99) | 2.6093 | 475.1626 | | 0.0001 |
| 6 | HOMO-2→LUMO(89) | 2.7357 | 453.2082 | 410(S) ^[b] | 0.0168 |
| 7 | HOMO-1→LUMO+2(93) | 2.8448 | 435.8274 | | 0.0002 |
| 8 | HOMO-3→LUMO(27) | 2.8908 | 428.8923 | | 0.0 |
| | HOMO→LUMO+4(72) | | | | |
| 9 | HOMO→LUMO+3(92) | 2.9502 | 420.2569 | | 0.0 |
| 10 | HOMO-4→LUMO(95) | 2.9873 | 415.0376 | | 0.0 |

^[a]components with greater than 10% contribution shown, ^[b]shoulder peak

Table S6. TD-DFT calculated electronic transition configuration for **5** along with their corresponding excitation energies, wavelength and oscillator strengths

HOMO is 262

| No | Transition configurations (%) ^[a] | Excitation Energy (ev) | Wavelength (nm) | | Osc, Strength |
|----|---|---------------------------|-----------------|-----------------------|------------------|
| | | | Cal. | Expt. | |
| 1 | HOMO-5→LUMO(14) | 1.6763 | 739.6300 | | 0.0278 |
| | HOMO-2→LUMO(18) | | | | |
| | HOMO→LUMO(66) | | | | |
| 2 | HOMO-1→LUMO(89) | 1.6856 | 735.5493 | | 0.0001 |
| 3 | HOMO-2→LUMO(80) | 1.7349 | 714.6474 | 630 | 0.0304 |
| | HOMO→LUMO(13) | | | | |
| 4 | HOMO-3→LUMO(88) | 1.7578 | 705.3373 | | 0.0001 |
| 5 | HOMO-4→LUMO(91) | 1.8168 | 682.4317 | | 0.0001 |
| 6 | HOMO→LUMO+1(100) | 2.0281 | 611.3317 | | 0.0021 |
| 7 | HOMO-5→LUMO(62) | 2.2907 | 541.2502 | 508(s) ^[b] | 0.0326 |
| | HOMO-2→LUMO+1(26) | | | | |
| 8 | HOMO-3→LUMO+1(17) | 2.4002 | 516.5577 | | 0.0001 |
| | HOMO-1→LUMO+1(78) | | | | |
| 9 | HOMO→LUMO+2(94) | 2.4178 | 512.7975 | | 0.0 |
| 10 | HOMO-4→LUMO+1(13) | 2.4982 | 496.2941 | | 0.0 |
| | HOMO-3→LUMO+1(72) | | | | |

^[a]components with greater than 10% contribution shown, ^[b]shoulder peak