

# Supporting Information

## **A Thermo-Electrochemical Converter using High Temperature Polybenzimidazole (PBI) Membranes for Harvesting Heat Energy**

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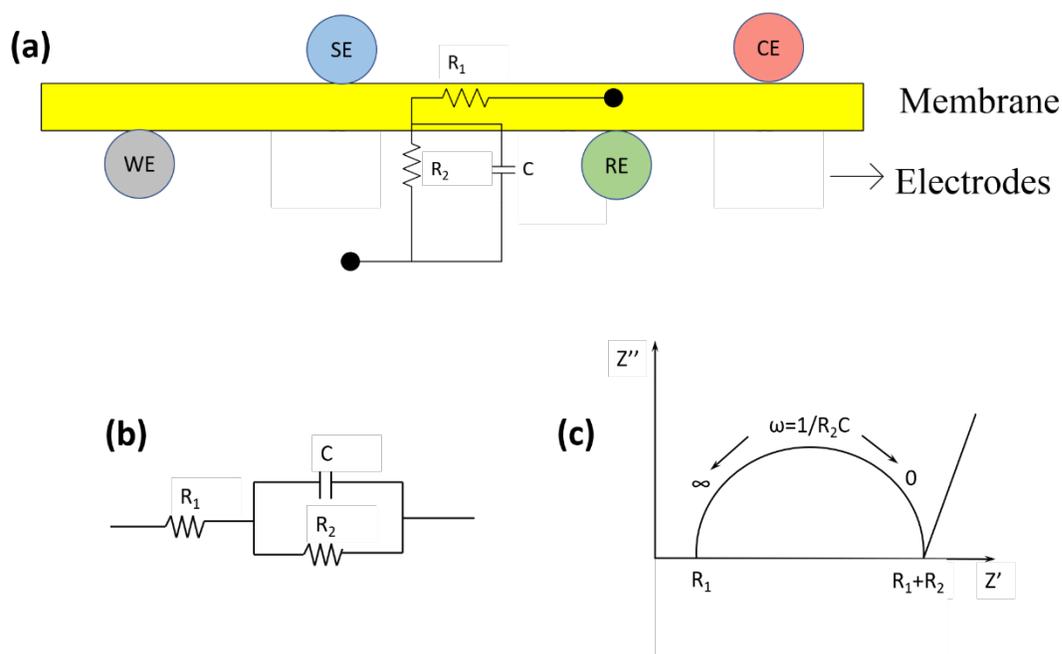
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## Proton Conductivity Measurement

The through-plane proton conductivities were measured by an AC Zahner IM6e electrochemical work station using four-probe impedance spectroscopy method over the frequency range from 1 Hz to 100 kHz with an amplitude of 5mV. A rectangular piece of membrane (3.5 cm×7.0 cm) was cut from the bulk membrane and placed between two Kapton support layers. Four platinum wire acting as four probes were set in a custom designed glass cell. The membrane ohmic resistance was determined by the model fitting according to the Nyquist plot based on a three-component equivalent circuit.<sup>1-2</sup>

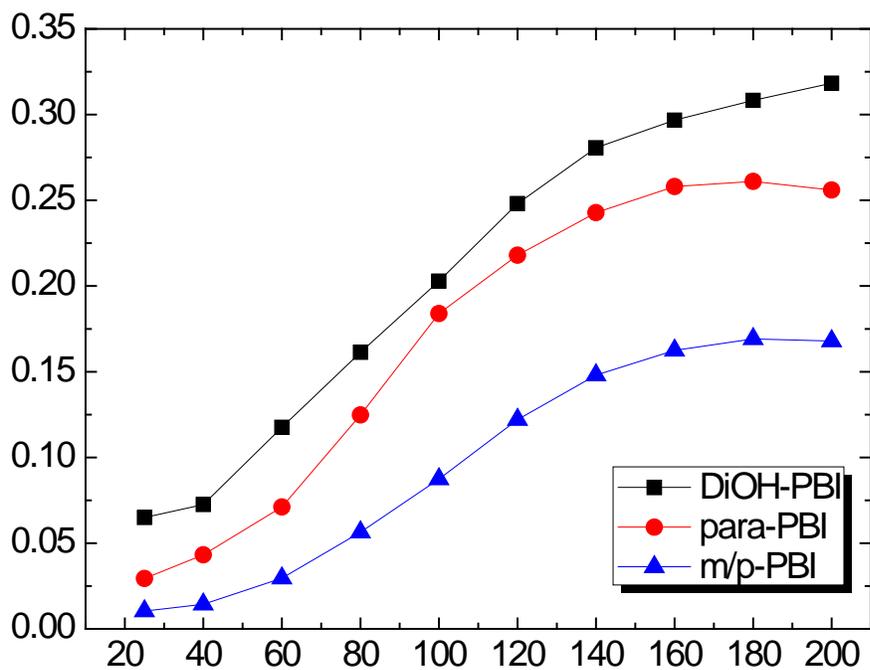


**Figure S1.** (a) Schematic four-probe proton conductivity testing setup, (b) Equivalent circuit model, and (c) A typical Nyquist plot used for membrane resistance fitting.

## Effect of the Membranes on the JTEC Performance

**Table S1.** PBI membrane composition

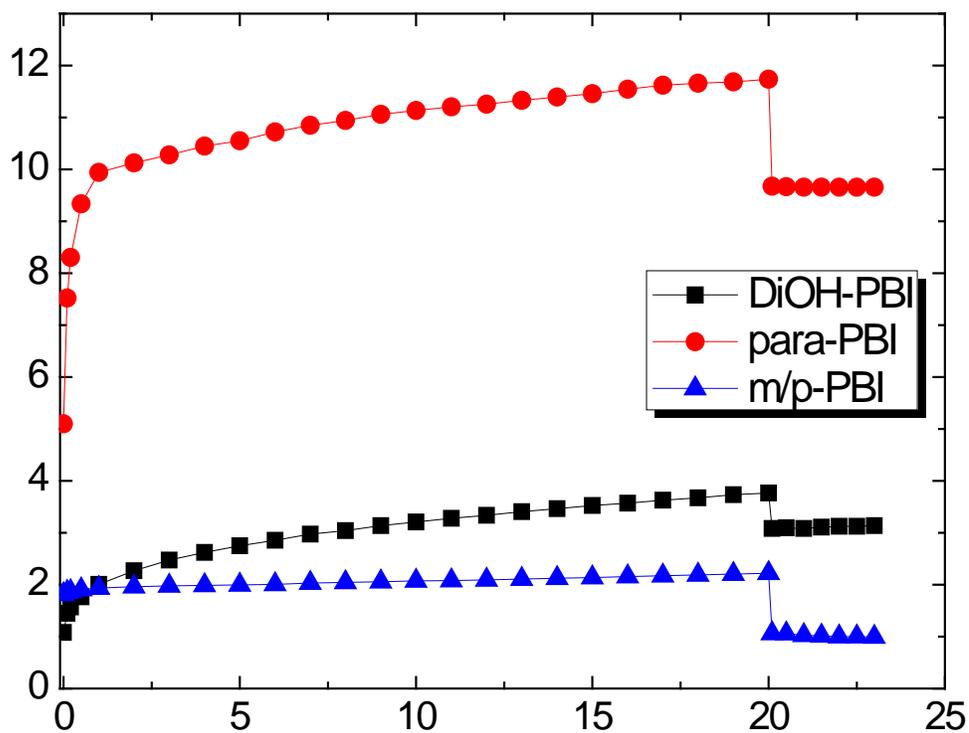
Membrane	Monomer Charge (wt%)	IV (dL/g)	Polymer Content (wt%)	PA content (wt%)	PA/PBI r.u. (molar ratio)
DiOH-PBI	3	/	6.6	56.9	27.3
para-PBI	2	3.3	5.6	57.3	32.3
m/p-PBI	10	1.8	16.8	57.6	10.8



**Figure S2.** Proton conductivities of DiOH-PBI, para-PBI, and m/p-PBI membranes as a function of temperature.

**Table S2.** Creep compliance test results of DiOH-PBI, para-PBI, and m/p-PBI membranes.

	DiOH-PBI	para-PBI	m/p-PBI
$J_s^0$ ( $10^{-6}\text{Pa}^{-1}$ )	2.5	10.3	1.9
creep rate ( $10^{-12}\text{Pa}^{-1}\text{s}^{-1}$ )	18.9	21.8	4.1



**Figure S3.** Creep compliance of the DiOH-PBI, para-PBI, and m/p-PBI copolymer membranes.

Membranes were preconditioned at 180°C for 24h and compressed at 0.1MPa at 180°C for 20h.

## References

- (1) Bard, A. J.; Faulkner, L. R. Fundamentals and applications. *Electrochemical Methods* **2001**, 2, 482.
- (2) Xie, Z.; Song, C.; Andraus, B.; Navessin, T.; Shi, Z.; Zhang, J.; Holdcroft, S. Discrepancies in the measurement of ionic conductivity of PEMs using two- and four-probe AC impedance spectroscopy. *J. Electrochem. Soc.* **2006**, 153, E173-E178.