Supporting Information

Synthesis of Large Area two-dimensional MoS₂ films by Sulfurization of Atomic Layer Deposited MoO₃ Thin Film for Nanoelectronic applications

Roman I. Romanov¹, Maxim G. Kozodaev¹, Denis I. Myakota¹, Anna G. Chernikova¹, Sergey M. Novikov¹, Valentyn S. Volkov¹, Alexander S. Slavich¹, Sergey S. Zarubin¹, Pavel S. Chizhov^{1,2}, Roman R. Khakimov¹, Anastasya. A. Chouprik¹, Cheol S. Hwang^{3*}, and Andrey M. Markeev¹

¹Moscow Institute of Physics and Technology (national research university), Institutskii per. 9, 141701 Dolgoprudny, Moscow region, Russia

²Chemistry Department, Moscow State University, Leninskie Gory 1, 119992, Moscow, Russia

³Department of Materials Science and Engineering, and Inter-University Semiconductor Research Center, Seoul National University, Seoul 08826, Republic of Korea

*Corresponding author: cheolsh@snu.ac.kr

General process schematic (synthesis/fabrication)



Figure S1. Process schematic for the thin MoS₂ film fabrication by the MoO₃ sulfurization from a chemistry perspective

Development of the ALD process for MoO₃ growth

Thin MoO₃ films were deposited in a commercially available hot-wall Picosun R200adv reactor equipped with a remote RF inductively coupled plasma source, operated at a power of 2500 W using Mo(CO)₆ (molybdenum hexacarbonyl 99.9% from Sigma Aldrich) and O₂-plasma as the Mo-precursor and oxygen source, respectively. Nitrogen (99.999%) was used both as carrier and purge gas, while Ar (99.9999%) was used as a carrier in plasma source. Solid-State Mo(CO)₆ source was kept at 40°C during the ALD process to produce an adequate vapor pressure. The Mo(CO)₆ and O₂-plasma exposure times were fixed at 1.6 and 21.0 s, respectively. After each reactant exposure, the reactor was purged by 8.0 s. The substrate temperature was maintained at 165°C. The specific characteristics of the utilized equipment are extremely long (\approx 60 cm) distance from the plasma source to the substrate and high working pressure (\sim 1.0 mbar). Under such conditions, the kinetic energy of charged particles reaching the substrate is evaluated to be negligibly low, which was confirmed by additional measurements carried out directly on the described equipment.

At the described conditions the MoO₃ film growth rate was found to be ~0.45 Å/cycle which is comparable to the existing literature data. The utilized ALD process allowed obtaining thin MoO₃ films with a good uniformity ($1\sigma\approx3.5\%$) on a 4" silicon wafers with 10 nm-thick Al₂O₃ seed layer (Figure S2).



Figure S2. Photograph of the 4" silicon wafer with the ALD-grown MoO₃ film (400 ALD cycles) (a) and MoO₃ film thickness distribution across the wafer area, measured by spectral ellipsometry (b)

Angle-resolved XPS analysis for MoO₃ film thickness determination

Mo3d and Al2p XPS spectra collected from *as-grown* MoO₃ film at different electron emission angles (25, 45 and 65°) are presented in Figure S3.



Figure S3. Angle-resolved XPS spectra collected from as-grown thin MoO₃ film on sapphire

It becomes obvious that a MoO₃ film with a notable thickness was achieved since the Mo3d/Al2p ratio monotonically increased with the rise of collection angle (θ) while the Al2p signal completely vanished at θ =65°. The following equations were used for the film thickness (*h*) and continuity factor (*f*) determination:

$$I_{Mo} = I_{Mo}^{0} \times f \times \cos(\theta) \times \left(1 - exp\left(-\frac{h}{\lambda_{Mo3d}^{Mo03} \times \cos(\theta)}\right)\right)$$
$$I_{Al} = I_{Al}^{0} \times (1 - f) \times \cos(\theta) \times exp\left(-\frac{h}{\lambda_{Al2p}^{Mo03} \times \cos(\theta)}\right)$$

Here λ denotes the inelastic mean free path while I^0 denotes the signal from the uncovered thick (XPS non-transparent) layer for a given material. The λ values were calculated by the TPP-2M method (MoO₃ density was given as 4,69 g/cm³ and bandgap E_g as 2,88 eV) with the following result: $\lambda_{Mo3d}^{Mo03} = 20,2$ Å and $\lambda_{Al2p}^{Mo03} = 22,5$ Å. Next, the experimental I_{Mo}/I_{Al} values were fitted by the described model and the completely continuous film (*f*=1) was found in all cases while its thickness depended on ALD cycles number.

XPS analysis of as-grown MoO₃ film



The completely oxidized Mo⁶⁺ state in as-grown MoO₃ film is clearly seen in Figure S4:

Figure S4. XPS Mo*3d* spectra for the as-deposited MoO₃ film. The blue line represents the raw spectra, and other color lines represent the deconvoluted peaks, and the green line represents the background.

Raman spectra deconvolution

The result of the Raman spectra deconvolution for the films, sulfurized at different temperatures are presented in the table below:

mode	mode 500°C			700°C		900°C			1000°C			
	k,	Γ,	A	k,	Γ,	A	k,	Γ,	A	k,	Γ,	A
	(cm ⁻¹)	(cm ⁻¹)	(a.u.)	(cm ⁻¹)	(cm ⁻ 1)	(a.u.)	(cm ⁻¹)	(cm ⁻¹)	(a.u.)	(cm ⁻ 1)	(cm ⁻ 1)	(a.u.)
A _{1g} - LA(M)	181,9	36,5	2623	178,1	15,5	2320	177,8	10,3	2257	177,0	10,9	4282
TA(K)	188,7	6,3	391	188,3	6,2	555	188,2	9,8	877	188,0	13,2	2031
-	217,6	21,2	1438	212,4	30,5	1847	208,7	20,4	938	212,3	27,7	2833
LA(M)	229,5	13,2	2315	227,8	13,6	1811	228,2	22,5	1984	228,4	14,4	2389
TO(M)	349,7	40,8	2085	348,9	51,6	2000	346,6	49,8	2203	344,7	49,3	2241
LO(M)	377,5	17,0	2271	279,1	9,5	1992	377,9	8,2	1991	376,7	8,8	2325
E _{2g} ¹	383,2	7,6	714	383,2	6,2	1716	383,3	5,5	2934	383,3	4,7	4370
B _{1u}	404,1	5,1	372	404,2	5,4	908	403,8	6,3	1421	404,1	6,3	2637
A _{1g}	407,8	6,2	1025	407,7	4,5	1094	407,8	4,7	1134	407,7	4,8	1880
b	416,6	23,6	2881	417,2	18,9	6684	418,9	13,8	5298	419,0	13,4	9146
2LA(M)	450,6	28,4	2451	453,1	21,7	11266	456,6	17,1	9081	456,4	20,0	1995 6
A _{2u}	463,2	18,2	1726	464,7	10,2	2729	466,8	8,6	4620	467,0	8,3	6775
-							445,9	29,0	6219	444,6	40,0	1077 9

Table 1. Raman spectra deconvolution parameters

SAED images

In addition to the presented in the manuscript SAED images, similar ones were also collected with a smaller aperture size (200 nm), allowed to collect information from less number of grains (Figure S5).



Figure S5. SAED patterns collected with an aperture diameter of 0.2 μm

One can see that sulfurization temperature increase resulted not only in the crystalline size growth but also improvement of their quality since the diffraction rings firstly become sharper (700 °C) and, finally, look-like a finite spotty pattern (1000 °C).

Rocking curve analysis

The observed variation in the preferred crystallographic orientation of the grains after the sulfurization at the two temperatures (900 and 1000 °C) was confirmed by the X-ray rocking curve analysis (Figure S6):



Figure S6. (0002) and rocking curve collected from the 10-nm-thick MoS₂ films sulfurized at 900°C (a) and 1000°C (b), respectively

The observed X-ray rocking curve maximum at the non-zero Ψ angle indicates the presence of the grains along the direction deviated from the normal direction to the substrate surface by the Ψ angle. First, the significant intensity in the rocking curve near the $2\theta \sim 32.7^{\circ}$ coincides with the θ -2 θ result indicating the presence of the grains along the out-of-plane direction with a relatively large variation in their out-of-plane direction orientation for the case of the sulfurization at 900 °C (Figure S6). Another notable finding is that the [0001]-oriented grains have misoriented grains with a particular misorientation angle of ~0.3 °, in addition to the main precisely [0001]-oriented grains. These two findings suggest that the crystallographic alignment of the grain in this film is less perfect, compared to the MoS₂ film sulfurized at 1000 °C, which did not show any peculiarity in the [0001]-oriented grains and involvement of the (10-10)-oriented grains (Figure S6).

The volume ratio of fractions with different orientations (for the 10 nm MoO₃ film sulfurized at 900 °C) could be calculated using a simple uniaxial texture model (March-Dollase) and rocking curve maxima comparison:

Fraction	[0001] _{MoS2}	[0001] _{MoS2}	Misoriented
	$\uparrow \uparrow$	\perp	$(\angle_{MoS2/Al2O3} \approx 0.3^{\circ})$
	[0001] _{A12O3}	[0001] _{A12O3}	
Volume ratio	1	0.05	0.12

Table 2. Volume ratio of mactions with different orientations	Table 2.	Volume	ratio o	f fractions	with	different	orientations
---	----------	--------	---------	-------------	------	-----------	--------------

The model for the MoO_3 to MoS_2 conversion

According to the presented in the main manuscript experimental data, the following model can be suggested for the MoO_3 to MoS_2 conversion:



Figure S7. The possible model of MoO_3 to MoS_2 conversion at different temperatures.