

## Supplementary Material

### Determination of surface energy parameters of hydrophilic porous membranes via a corrected contact angle approach

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#### Contents:

**Figure S1.** Illustration of the diminishing process of a liquid droplet on the porous membrane  
surface. Page S2

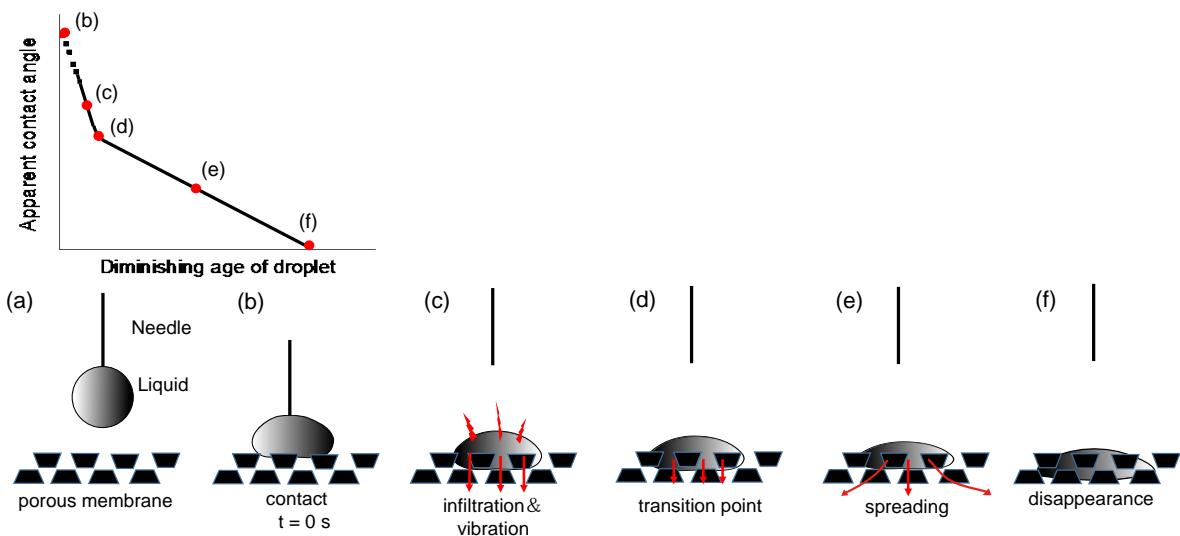
**Table S1.** Properties of the standard test liquids at 20 °C for the contact angle measurement.

Page S2

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**Figure S1.** Illustration of the diminishing process of a liquid droplet on the porous membrane surface.

**Table S1.** Properties of the standard test liquids at 20 °C for the contact angle measurement.

| Liquid                    | Label of polarity | Surface energies (mJ/m <sup>2</sup> ) |                   |              |              | Density (g/cm <sup>3</sup> ) <sup>b</sup> | Viscosity (mPa·s) <sup>b</sup> |
|---------------------------|-------------------|---------------------------------------|-------------------|--------------|--------------|---|--------------------------------|
|                           |                   | $\gamma_L$                            | $\gamma_L^{LW}$   | $\gamma_L^+$ | $\gamma_L^-$ |   |                                |
| Diiodomethane (DI)        | Nonpolar          | 50.8                                  | 50.8              | 0            | 0            | 3.33                                      | -                              |
| Water                     | Polar             | 72.8                                  | 21.8              | 25.5         | 25.5         | 1.00                                      | 1.00                           |
| Formamide (FO)            | Polar             | 58.0                                  | 39.0              | 2.28         | 39.6         | 1.13                                      | 4.32                           |
| Dimethyl sulfoxide (DMSO) | Polar             | 44.0                                  | 26.7 <sup>a</sup> | 3.10         | 24.0         | 1.10                                      | 2.47                           |

Note: <sup>a</sup> Estimated via  $\gamma_L^{LW} = \gamma_L - 2\sqrt{\gamma_L^+ \gamma_L^-}$ ; <sup>b</sup> Data obtained from PubChem.