

Direct Transformation of N-protected α,β -Unsaturated γ -Amino Amides into γ -Lactams through a Base Mediated Molecular Rearrangement

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Table of content

1	ORTEP diagrams	S2
2	Crystallographic information	S7
3	List of organic and inorganic bases used for the rearrangement	S12
4	¹ H and ¹³ C NMR of all compounds	S13

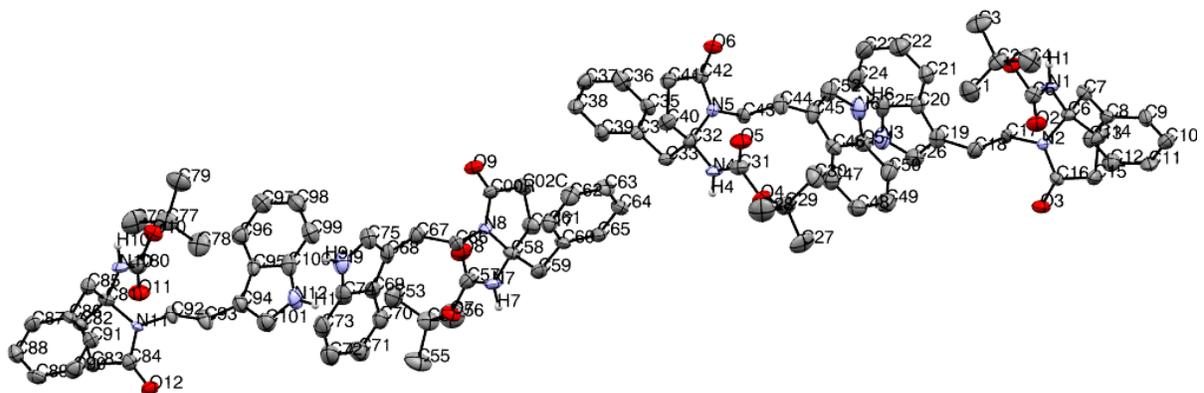


Fig S3: ORTEP diagram of compound **2e**. H-atoms are omitted for clarity. Ellipsoids are drawn at 50% probability (CCDC no 1498004).

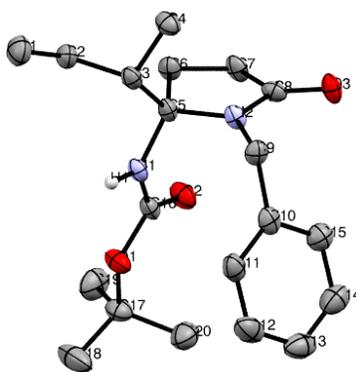


Fig S4: ORTEP diagram of compound **2f**. H-atoms are omitted for clarity. Ellipsoids are drawn at 50% probability (CCDC no 1498005).

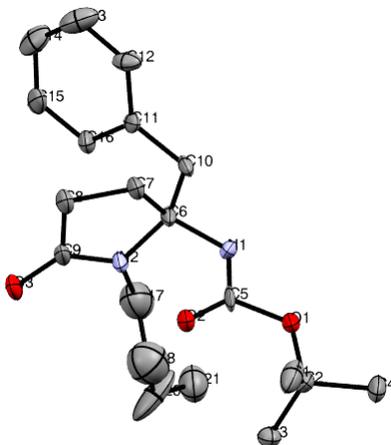


Fig S5: ORTEP diagram of compound **2h**. H-atoms are omitted for clarity. Ellipsoids are drawn at 50% probability (CCDC no 1498006).

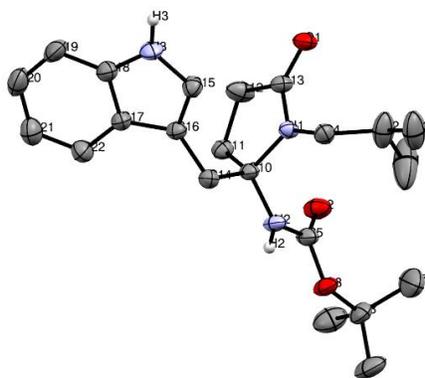


Fig S6: ORTEP diagram of compound **2i**. H-atoms are omitted for clarity. Ellipsoids are drawn at 50% probability (CCDC no 1498007).

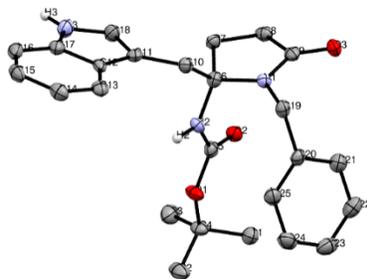


Fig S7: ORTEP diagram of compound **2j**. H-atoms are omitted for clarity. Ellipsoids are drawn at 50% probability (CCDC no 1498008).

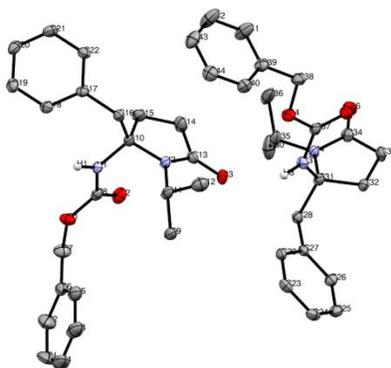


Fig S8: ORTEP diagram of compound **2m**. H-atoms are omitted for clarity. Ellipsoids are drawn at 50% probability (CCDC no 1498009).

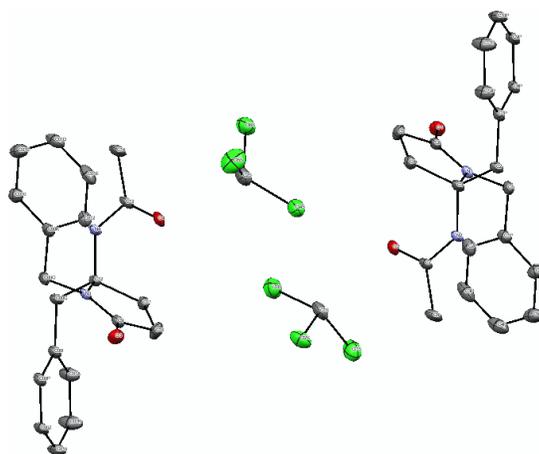


Fig S9: ORTEP diagram of compound **5c**. H-atoms are omitted for clarity. Ellipsoids are drawn at 50% probability (CCDC no1871180).

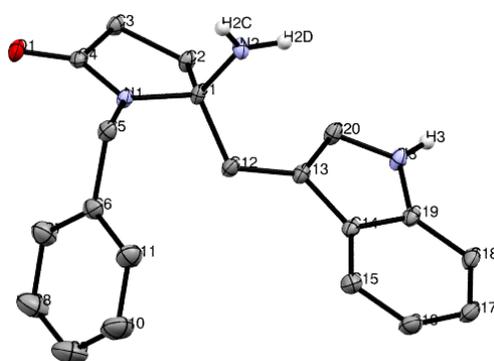


Fig S10: ORTEP diagram of compound **3j**. H-atoms are omitted for clarity. Ellipsoids are drawn at 50% probability (CCDC no 14980010).

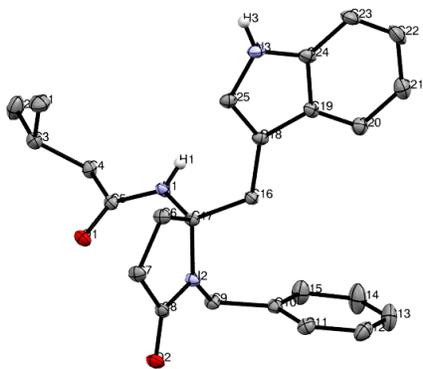


Fig S11: ORTEP diagram of compound **4j**. H-atoms are omitted for clarity. Ellipsoids are drawn at 50% probability (CCDC no 14980011).

2. Crystallographic Information:

Compound 2a: Crystals of **2a** were grown by slow evaporation from a solution of EtOAc. A single crystal ($0.2 \times 0.1 \times 0.08$ mm) was mounted on loop with a small amount of the paraffin oil. The X-ray data were collected at 100K temperature on a Bruker APEX(II) DUO CCD diffractometer using Mo K_{α} radiation ($\lambda = 0.71073 \text{ \AA}$), ω -scans ($2\theta = 57.04$), for a total of 12065 independent reflections. Space group P12(1)1, $a = 6.639 (3)$, $b = 8.413 (3)$, $c = 36.816 (16)$, $\beta = 90.173 (10)$, $V = 2065.3 (16) \text{ \AA}^3$, Monoclinic, $Z = 4$ for chemical formula $C_{23}H_{28}N_2O_3$, with two molecules in asymmetric unit; $\rho_{\text{calcd}} = 1.229 \text{ gcm}^{-3}$, $\mu = 0.081 \text{ mm}^{-1}$, $F(000) = 816$, $R_{\text{int}} = 0.0529$. The structure was obtained by intrinsic methods using SHELXS-97. The final R value was 0.0620 ($wR_2 = 0.1167$) 8305 observed reflections ($F_0 \geq 4\sigma(|F_0|)$) and 511 variables, $S = 0.973$.

Compound 2b: Crystals of **2b** were grown by slow evaporation from a solution of EtOAc. A single crystal ($0.15 \times 0.1 \times 0.05$ mm) was mounted on loop with a small amount of the paraffin oil. The X-ray data were collected at 100K temperature on a Bruker APEX(II) DUO CCD diffractometer using Mo K_{α} radiation ($\lambda = 0.71073 \text{ \AA}$), ω -scans ($2\theta = 66.65$), for a total of 20711

independent reflections. Space group C 2, $a = 14.9637(5)$, $b = 11.3042(4)$, $c = 21.2771(9)$, $\beta = 110.4360(10)$, $V = 3372.6(2) \text{ \AA}^3$, Monoclinic, $Z = 8$ for chemical formula $C_{17}H_{24}N_2O_3$, with two molecule in asymmetric unit; $\rho_{\text{calcd}} = 1.199 \text{ gcm}^{-3}$, $\mu = 0.665 \text{ mm}^{-1}$, $F(000) = 1312$, $R_{\text{int}} = 0.0243$. The structure was obtained by intrinsic methods using SHELXS-97. The final R value was 0.055 ($wR_2 = 0.2125$) 5783 observed reflections ($F_0 \geq 4\sigma(|F_0|)$) and 405 variables, $S = 2.070$.

Compound 2e: Crystals of **2e** were grown by slow evaporation from a solution of EtOAc. A single crystal ($0.1 \times 0.05 \times 0.03 \text{ mm}$) was mounted on loop with a small amount of the paraffin oil. The X-ray data were collected at 100K temperature on a Bruker APEX(II) DUO CCD diffractometer using Mo K_{α} radiation ($\lambda = 0.71073 \text{ \AA}$), ω -scans ($2\theta = 66.39$), for a total of 9955 independent reflections. Space group P1, $a = 6.8272(3)$, $b = 8.0949(3)$, $c = 42.0284(15)$, $\beta = 89.977(2)$, $V = 2322.72(16) \text{ \AA}^3$, triclinic, $Z = 4$ for chemical formula $C_{26}H_{31}N_3O_3$, with four molecule in asymmetric unit; $\rho_{\text{calcd}} = 1.240 \text{ gcm}^{-3}$, $\mu = 0.652 \text{ mm}^{-1}$, $F(000) = 1012$, $R_{\text{int}} = 0.0950$. The structure was obtained by intrinsic methods using SHELXS-97. The final R value was 0.0664 ($wR_2 = 0.1056$) 14782 observed reflections ($F_0 \geq 4\sigma(|F_0|)$) and 1165 variables, $S = 1.263$.

Compound 2f: Crystals of **2f** were grown by slow evaporation from a solution of aqueous methanol. A single crystal ($0.15 \times 0.1 \times 0.05 \text{ mm}$) was mounted on loop with a small amount of the paraffin oil. The X-ray data were collected at 100K temperature on a Bruker APEX(II) DUO CCD diffractometer using Mo K_{α} radiation ($\lambda = 0.71073 \text{ \AA}$), ω -scans ($2\theta = 50.48$), for a total of 15144 independent reflections. Space group P21, $a = 6.696(2)$, $b = 13.783(4)$, $c = 10.694(4)$, $\beta = 102.690(8)$, $V = 962.8(5) \text{ \AA}^3$, monoclinic, $Z = 2$ for chemical formula $C_{20}H_{30}N_2O_3$, with one molecule in asymmetric unit; $\rho_{\text{calcd}} = 1.195 \text{ gcm}^{-3}$, $\mu = 0.080 \text{ mm}^{-1}$, $F(000) = 376$, $R_{\text{int}} = 0.0551$. The structure was obtained by intrinsic methods using SHELXS-97. The final R value was 0.0599 ($wR_2 = 0.1470$) 4750 observed reflections ($F_0 \geq 4\sigma(|F_0|)$) and 231 variables, $S = 1.026$.

Compound 2h: Crystals of **2h** were grown by slow evaporation from a solution of EtOAc. A single crystal ($0.1 \times 0.07 \times 0.05$ mm) was mounted on loop with a small amount of the paraffin oil. The X-ray data were collected at 100K temperature on a Bruker APEX(II) DUO CCD diffractometer using Mo K_{α} radiation ($\lambda = 0.71073 \text{ \AA}$), ω -scans ($2\theta = 57.358$), for a total of 34755 independent reflections. Space group C 2/c, $a = 28.90(4)$, $b = 10.508(13)$, $c = 13.397(17)$, $\beta = 106.42(3)$, $V = 3902(9) \text{ \AA}^3$, monoclinic, $Z = 8$ for chemical formula $C_{20}H_{29}N_2O_3$, with one molecule in asymmetric unit; $\rho_{\text{calcd}} = 1.176 \text{ g cm}^{-3}$, $\mu = 0.079 \text{ mm}^{-1}$, $F(000) = 1480$, $R_{\text{int}} = 0.0719$. The structure was obtained by intrinsic methods using SHELXS-97. The final R value was 0.1511 ($wR_2 = 0.3044$) 4863 observed reflections ($F_0 \geq 4\sigma(|F_0|)$) and 230 variables, $S = 1.308$. Author comment on check CIF: The investigated single crystal was a small-sized, brittle and poorly diffracting. Numerous datasets were collected on single crystals from different batches, whereof the one of the highest quality is reported herein.

Compound 2i: Crystals of **2i** were grown by slow evaporation from a solution of EtOAc. A single crystal ($0.15 \times 0.1 \times 0.05$ mm) was mounted on loop with a small amount of the paraffin oil. The X-ray data were collected at 100K temperature on a Bruker APEX(II) DUO CCD diffractometer using Mo K_{α} radiation ($\lambda = 0.71073 \text{ \AA}$), ω -scans ($2\theta = 50.48$), for a total of 52510 independent reflections. Space group P -1, $a = 6.8063(18)$, $b = 13.002(4)$, $c = 13.629(4)$, $\beta = 92.523(7)$, $V = 1106.6(5) \text{ \AA}^3$, triclinic, $Z = 2$ for chemical formula $C_{22}H_{30}N_3O_3$, with one molecule in asymmetric unit; $\rho_{\text{calcd}} = 1.154 \text{ g cm}^{-3}$, $\mu = 0.077 \text{ mm}^{-1}$, $F(000) = 484$, $R_{\text{int}} = 0.2621$. The structure was obtained by intrinsic methods using SHELXS-97. The final R value was 0.0841 ($wR_2 = 0.1747$) 5554 observed reflections ($F_0 \geq 4\sigma(|F_0|)$) and 257 variables, $S = 1.017$. Author comment on check CIF: The investigated single crystal was a small-sized, brittle and poorly diffracting.

Numerous datasets were collected on single crystals from different batches, whereof the one of the highest quality is reported herein.

Compound 2j: Crystals of **2j** were grown by slow evaporation from a solution of aqueous methanol. A single crystal ($0.15 \times 0.1 \times 0.05$ mm) was mounted on loop with a small amount of the paraffin oil. The X-ray data were collected at 100K temperature on a Bruker APEX(II) DUO CCD diffractometer using Cu K_{α} radiation ($\lambda = 1.54178 \text{ \AA}$), ω -scans ($2\theta = 66.765$), for a total of 34755 independent reflections. Space group P 21/n, $a = 6.7044(10)$, $b = 28.659(4)$, $c = 11.4082(18)$, $\beta = 91.966(9)$, $V = 2190.7(6) \text{ \AA}^3$, monoclinic, $Z = 4$ for chemical formula $C_{25}H_{29}N_3O_3$, with one molecule in asymmetric unit; $\rho_{\text{calcd}} = 1.272 \text{ gcm}^{-3}$, $\mu = 0.675 \text{ mm}^{-1}$, $F(000) = 816$, $R_{\text{int}} = 0.0721$. The structure was obtained by intrinsic methods using SHELXS-97. The final R value was 0.0712 ($wR_2 = 0.1771$) 3756 observed reflections ($F_0 \geq 4\sigma(|F_0|)$) and 283 variables, $S = 1.090$.

Compound 2m: Crystals of **2m** were grown by slow evaporation from a solution of EtOAc. A single crystal ($0.12 \times 0.1 \times 0.06$ mm) was mounted on loop with a small amount of the paraffin oil. The X-ray data were collected at 100K temperature on a Bruker APEX(II) DUO CCD diffractometer using Mo K_{α} radiation ($\lambda = 0.71073 \text{ \AA}$), ω -scans ($2\theta = 50.48$), for a total of 47013 independent reflections. Space group P 21/c, $a = 12.367(5)$, $b = 16.811(7)$, $c = 19.201(8)$, $\beta = 90.00$, $V = 3992(3) \text{ \AA}^3$, monoclinic, $Z = 8$ for chemical formula $C_{22}H_{26}N_2O_3$, with two molecule in asymmetric unit; $\rho_{\text{calcd}} = 1.219 \text{ gcm}^{-3}$, $\mu = 0.081 \text{ mm}^{-1}$, $F(000) = 1706$, $R_{\text{int}} = 0.077$. The structure was obtained by intrinsic methods using SHELXS-97. The final R value was 0.0407 ($wR_2 = 0.1008$) 6229 observed reflections ($F_0 \geq 4\sigma(|F_0|)$) and 491 variables, $S = 0.665$.

Compound 5c: Crystals of **5c** were grown by slow evaporation from a solution of EtOAc. A single crystal ($0.13 \times 0.1 \times 0.05$ mm) was mounted on loop with a small amount of the paraffin oil. The X-ray data were collected at 100K temperature on a Bruker APEX(II) DUO CCD diffractometer using Mo K_{α} radiation ($\lambda = 0.71073 \text{ \AA}$), ω -scans ($2\theta = 57.172$), for a total of 11092 independent reflections. Space group P-1, $a = 6.6014(15)$, $b = 9.469(2)$, $c = 35.287(8)$ $\alpha = 93.454(5)$ $\beta = 93.826(5)$ $\gamma = 98.649(5)$, $V = 2170.3(8) \text{ \AA}^3$, monoclinic, $Z = 2$ for chemical formula $C_{20} H_{22} N_2 O_2$, $C H Cl_3$, with two molecule in asymmetric unit; ρ calcd = 1.352 g cm^{-3} , $\mu = 0.442 \text{ mm}^{-1}$, $F(000) = 918.0$. The structure was obtained by intrinsic methods using SHELXS-97. The final R value was 0.0913, $wR2 = 0.3006$ (11029) observed reflections ($F_0 \geq 4\sigma(|F_0|)$) and 507 variables, $S = 0.781$.

Compound 3j: Crystals of **3j** were grown by slow evaporation from a solution of EtOAc. A single crystal ($0.2 \times 0.1 \times 0.08$ mm) was mounted on loop with a small amount of the paraffin oil. The X-ray data were collected at 100K temperature on a Bruker APEX(II) DUO CCD diffractometer using Mo K_{α} radiation ($\lambda = 0.71073 \text{ \AA}$), ω -scans ($2\theta = 50.48$), for a total of 15583 independent reflections. Space group P -1, $a = 6.5787(11)$, $b = 9.2449(14)$, $c = 14.324(2)$, $\beta = 77.982(4)$, $V = 822.4(2) \text{ \AA}^3$, triclinic, $Z = 4$ for chemical formula $C_{20} H_{21} N_3 O$, with one molecule in asymmetric unit; ρ calcd = 1.290 g cm^{-3} , $\mu = 0.081 \text{ mm}^{-1}$, $F(000) = 364$, $R_{int} = 0.0273$. The structure was obtained by intrinsic methods using SHELXS-97. The final R value was 0.0523 ($wR2 = 0.1857$) 4106 observed reflections ($F_0 \geq 4\sigma(|F_0|)$) and 217 variables, $S = 1.563$. Author comment on check CIF: It is possible that shorter D-H...H-D distances are an artifact of refinement.

Compound 4j: Crystals of **4j** were grown by slow evaporation from a solution of EtOAc. A single crystal ($0.18 \times 0.1 \times 0.06$ mm) was mounted on loop with a small amount of the paraffin oil. The X-ray data were collected at 100K temperature on a Bruker APEX(II) DUO CCD diffractometer using Mo K_{α} radiation ($\lambda = 0.71073 \text{ \AA}$), ω -scans ($2\theta = 50.48$), for a total of 34755 independent

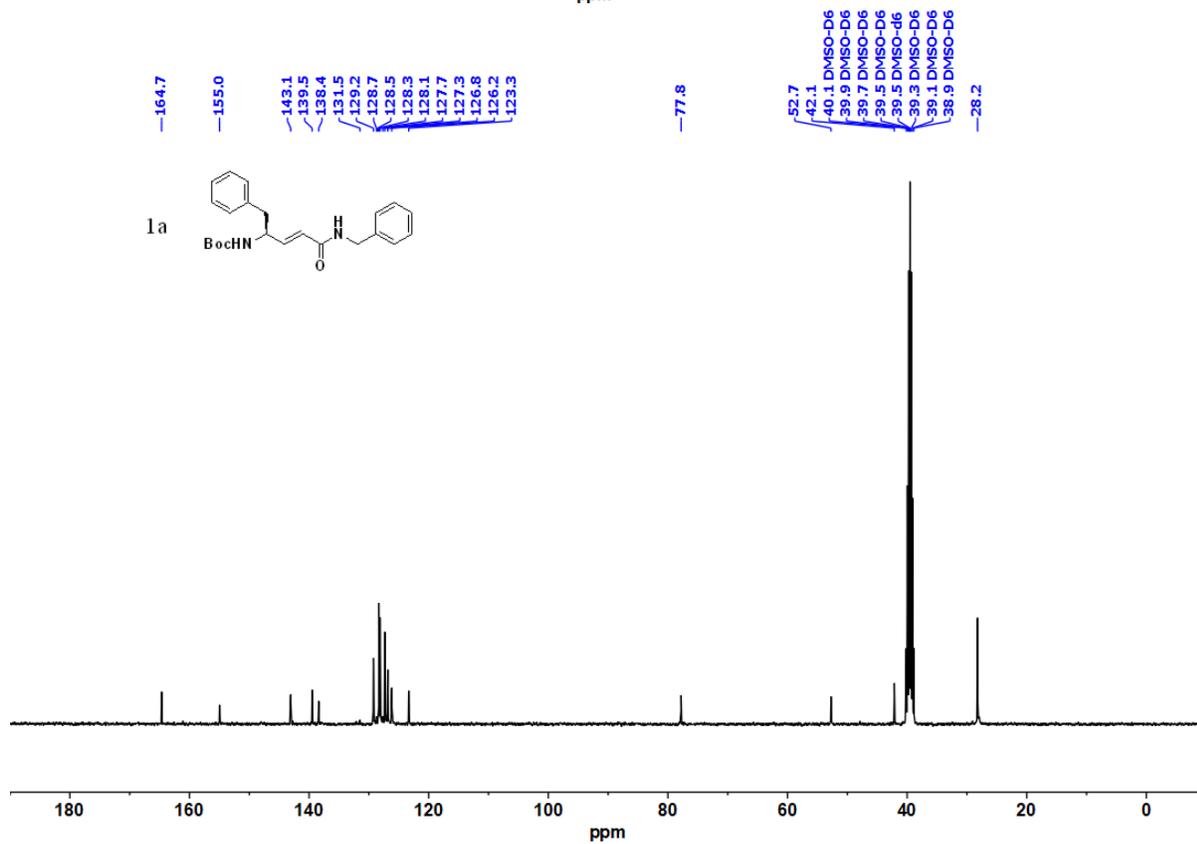
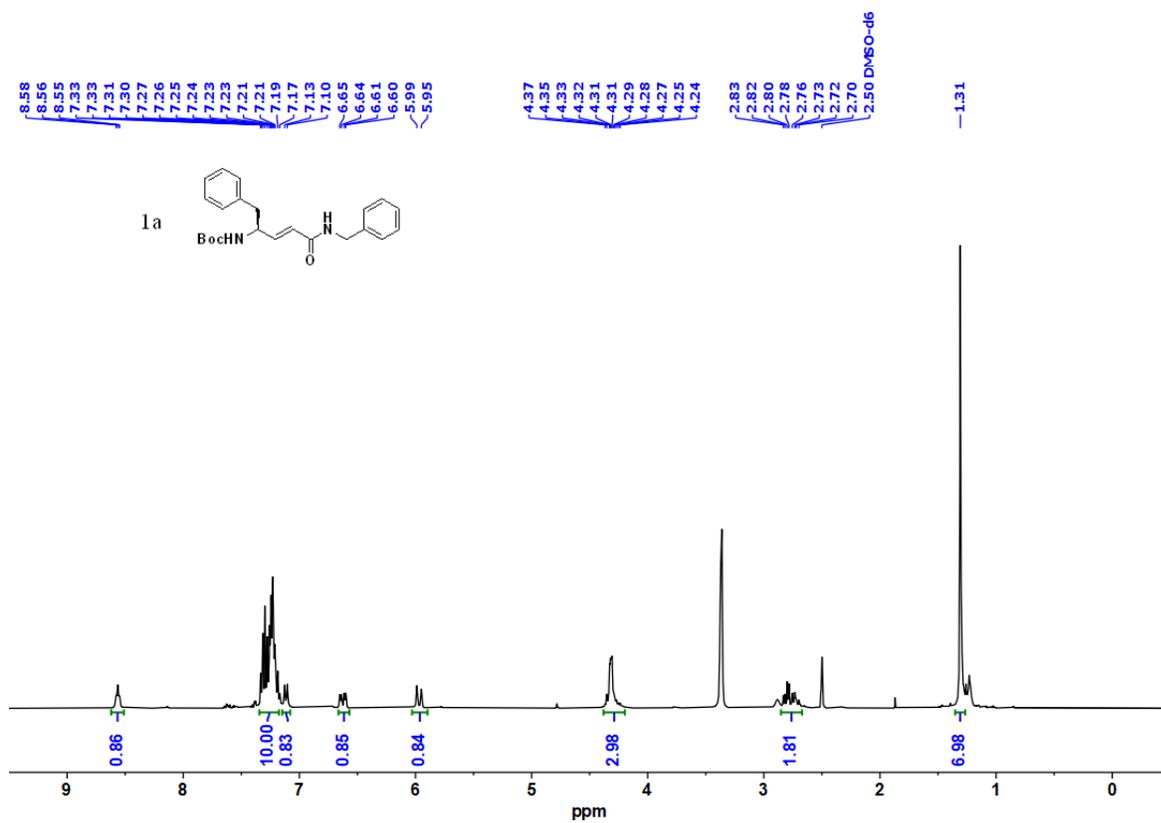
reflections. Space group Cc, $a = 6.9007(11)$, $b = 25.211(4)$, $c = 13.042(2)$, $\beta = 101.886(4)$, $V = 2220.2(6) \text{ \AA}^3$, monoclinic, $Z = 4$ for chemical formula $C_{25} H_{29} N_3 O_2$, with one molecule in asymmetric unit; $\rho_{\text{calcd}} = 1.207 \text{ g cm}^{-3}$, $\mu = 0.077 \text{ mm}^{-1}$, $F(000) = 968$, $R_{\text{int}} = 0.0412$. The structure was obtained by intrinsic methods using SHELXS-97. The final R value was 0.0361 ($wR_2 = 0.0975$) 11452 observed reflections ($F_0 \geq 4\sigma(|F_0|)$) and 635 variables, $S = 0.761$.

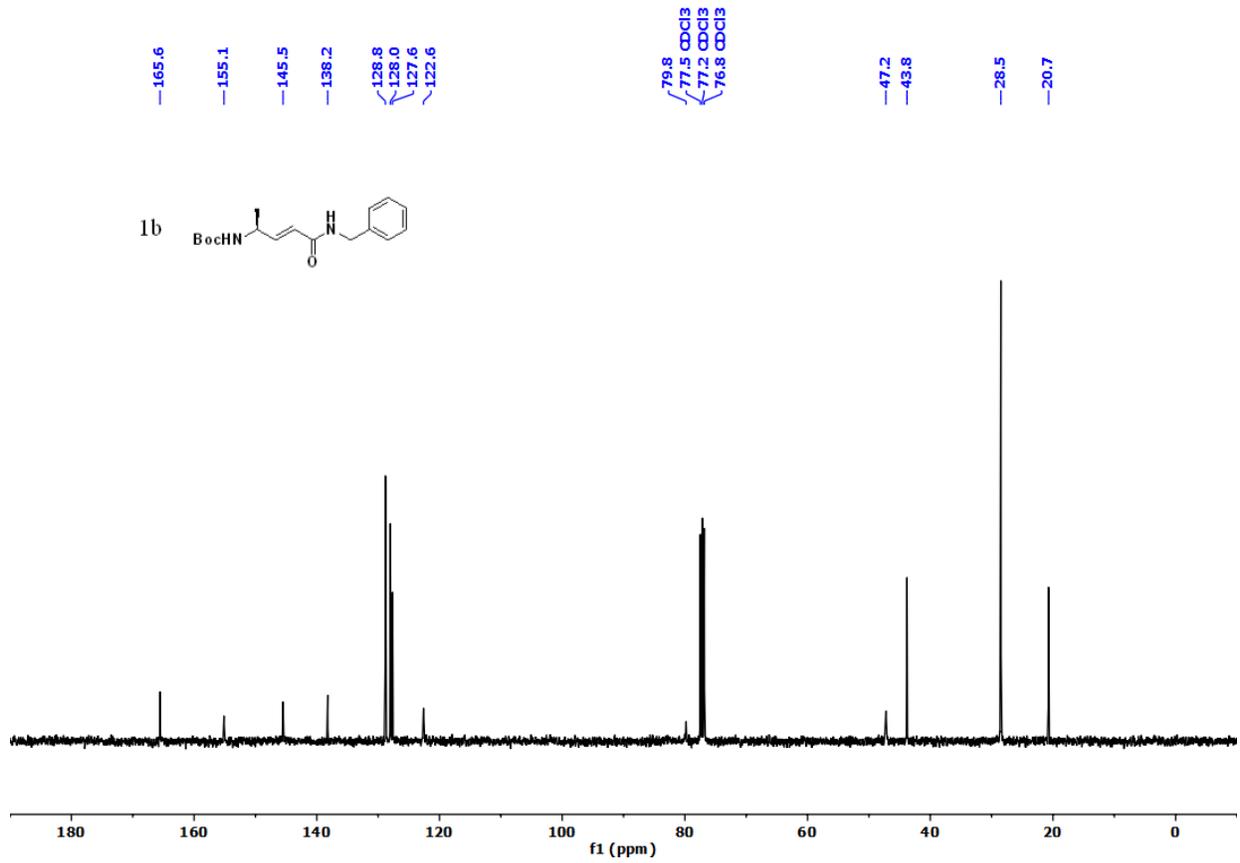
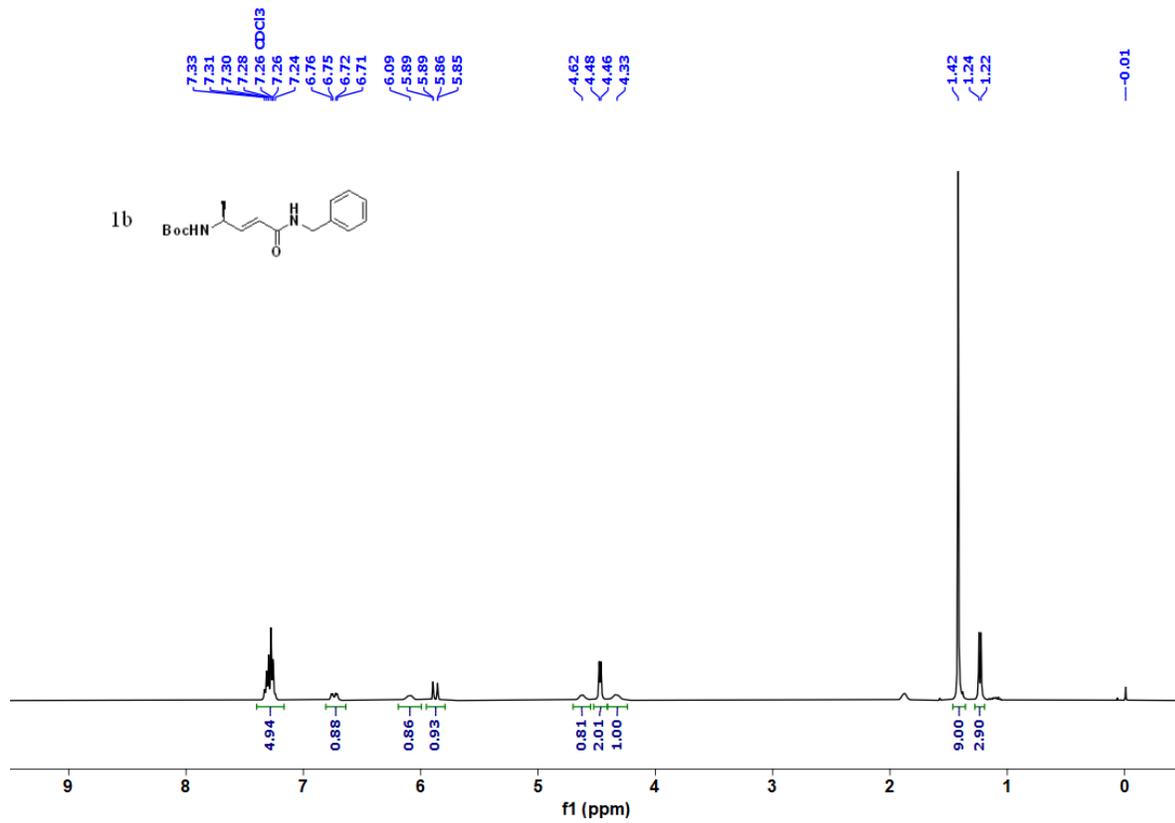
3) List of organic and inorganic bases used for the rearrangement

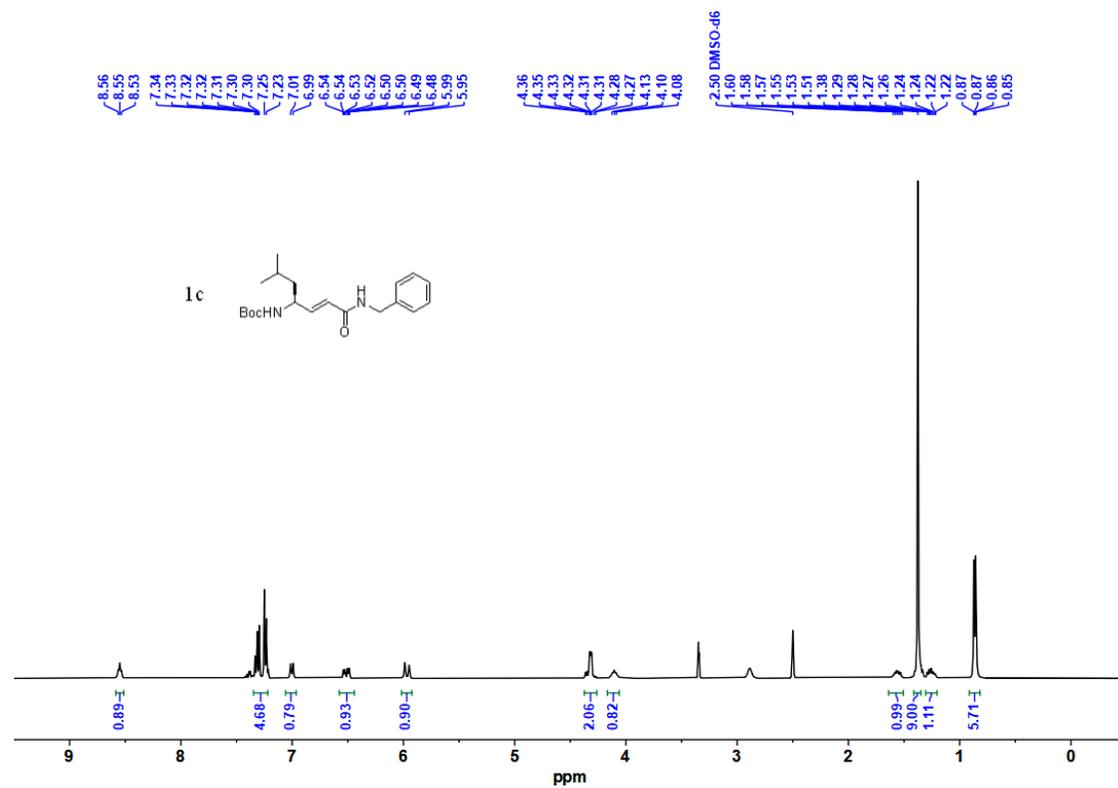
Table S1: List of organic and inorganic bases used for the rearrangement

Base	Solvent	Conversion	Time
LiOH (1M)	THF	-	2 d
NaOH (1M)	THF	40 %	2 d
CsOH (1M)	THF/DMF	50%	2 d
DBU(up to 3.0 equvi)	THF/DCM	-	2d
n-BuLi (1.0 equvi)	THF	-	1 d
KO ^t Bu (up to 3.0 equvi)	THF	100%	8 h

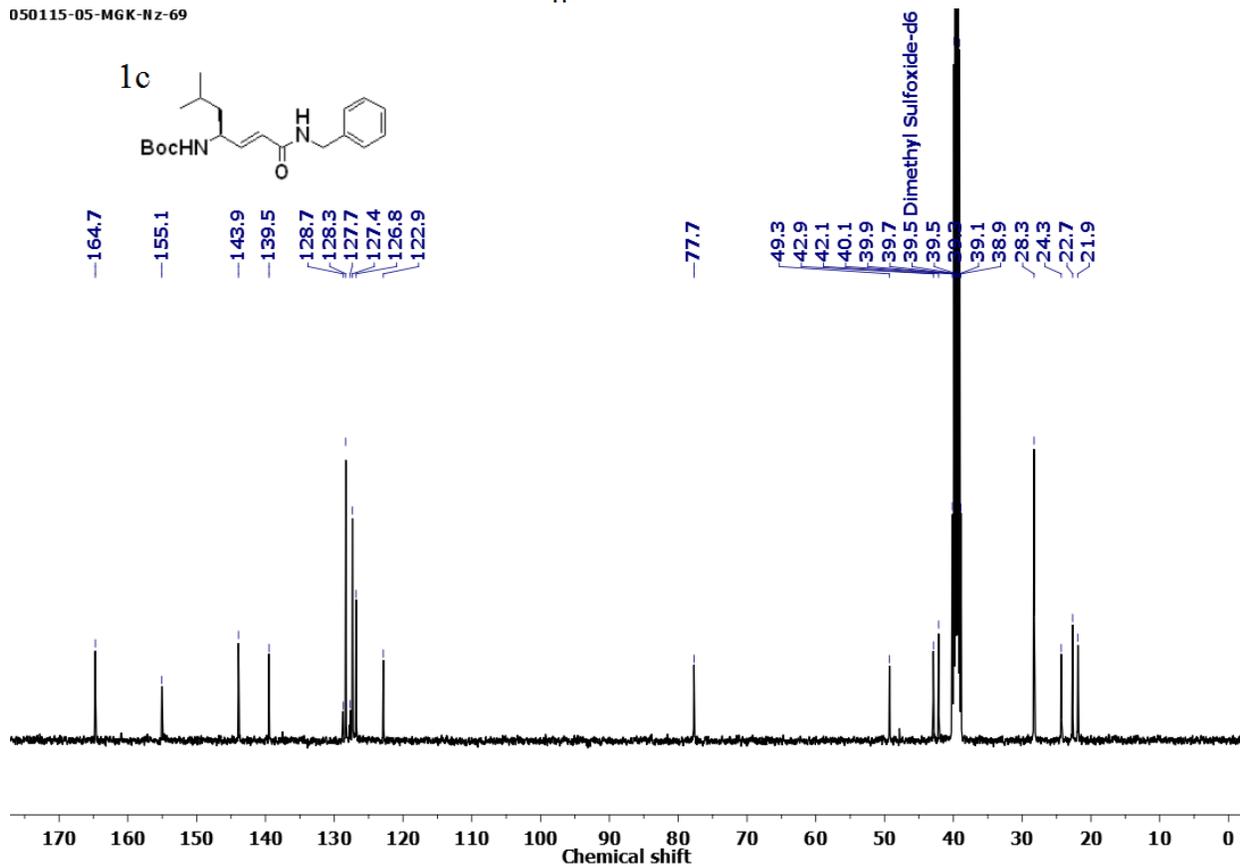
4) ¹H and ¹³C NMR of all compounds

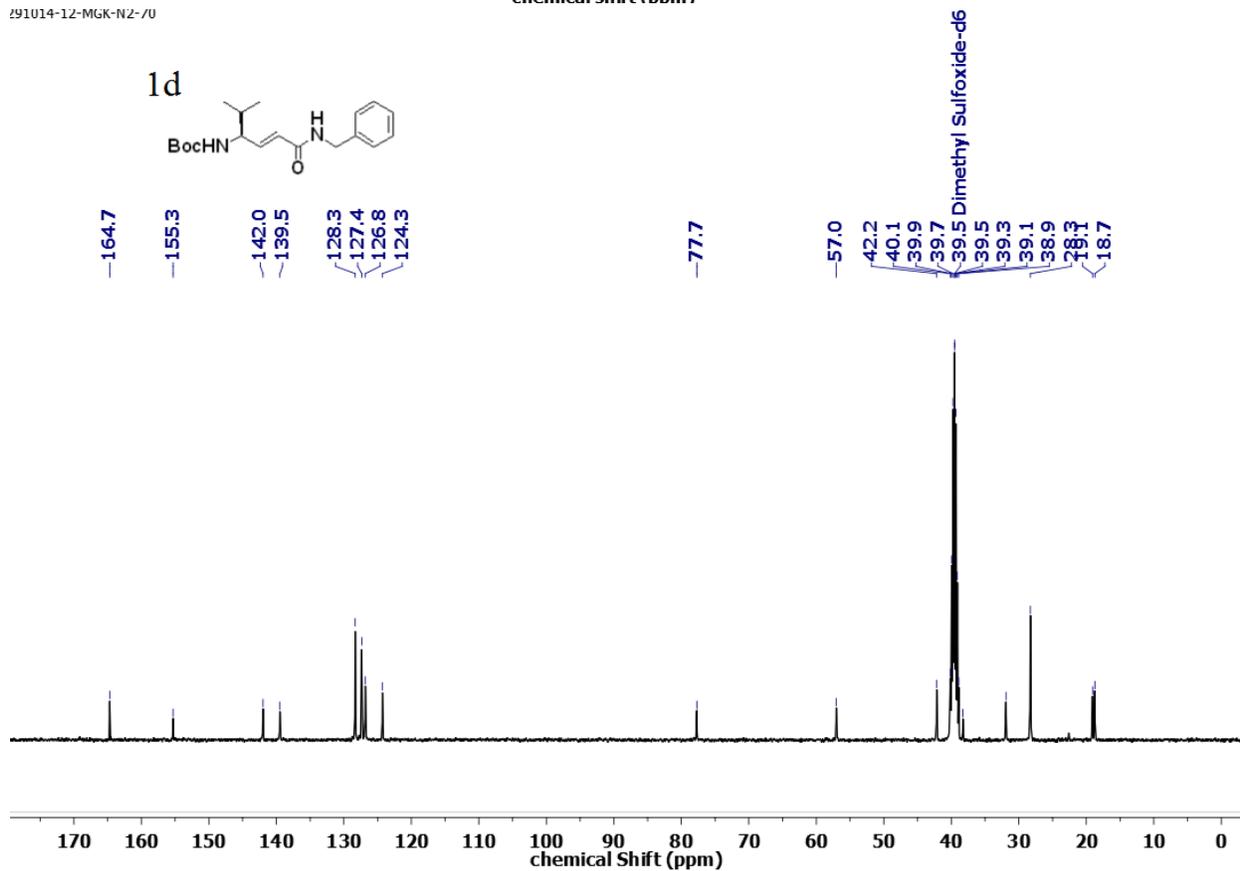
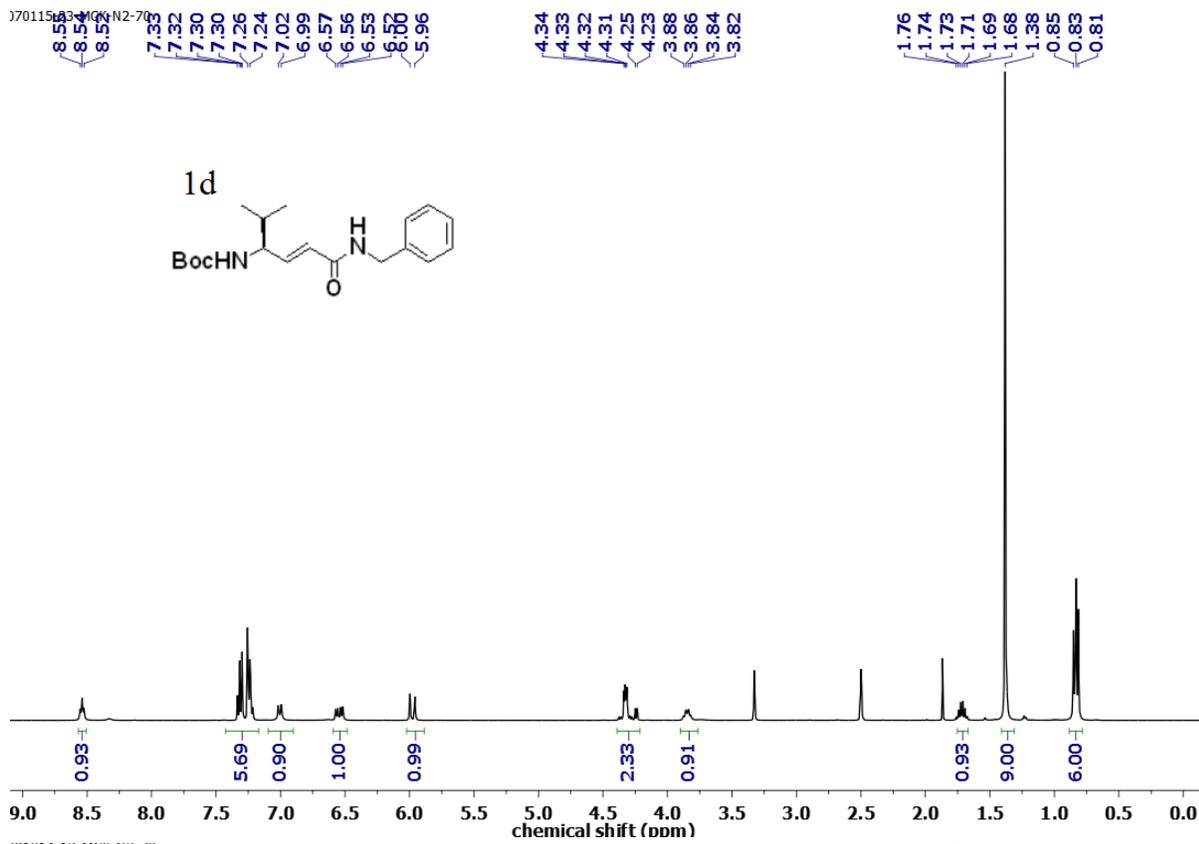






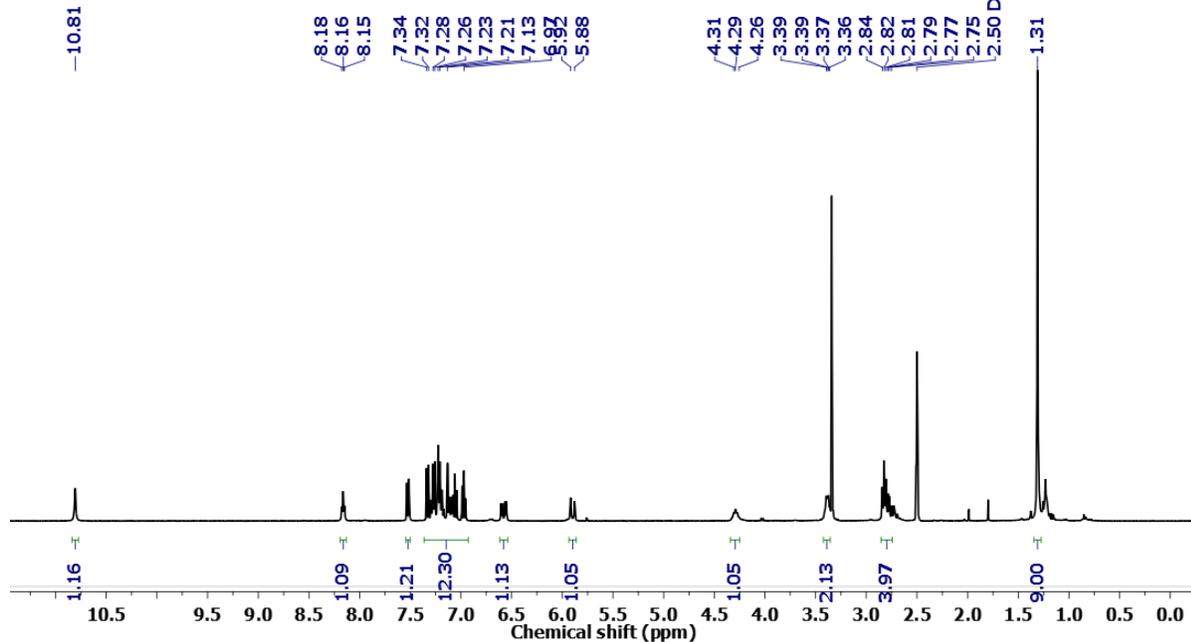
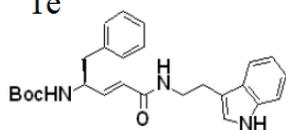
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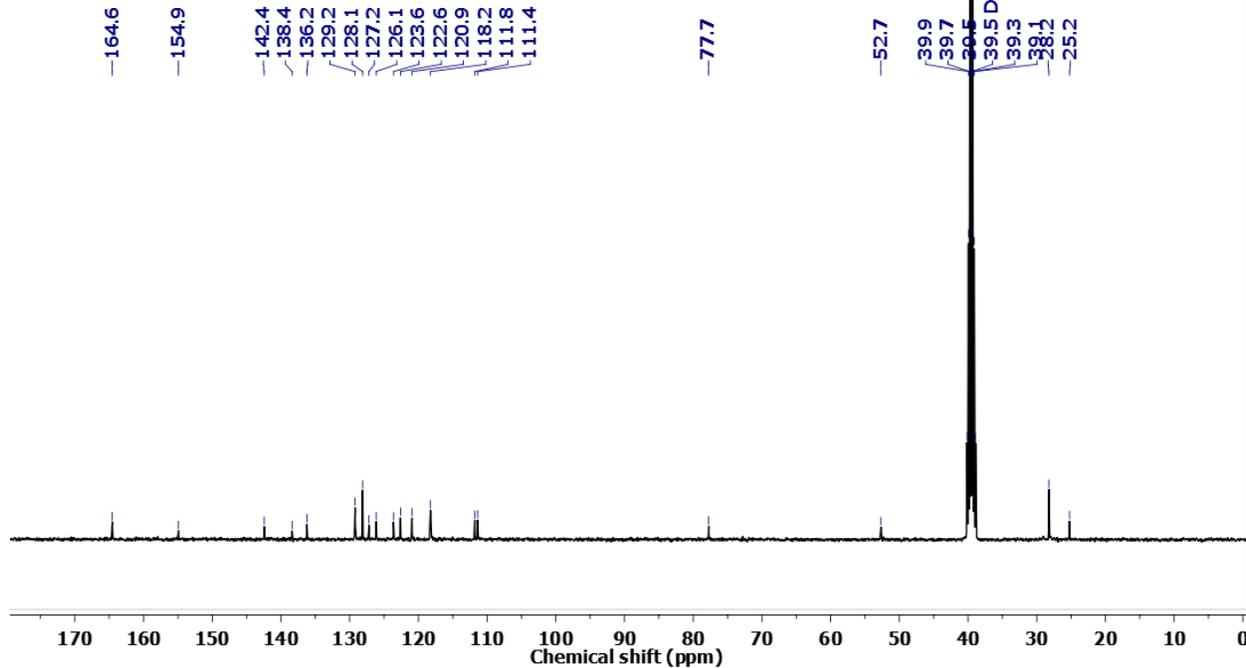
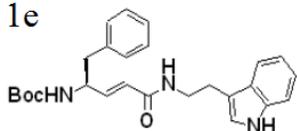
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1e

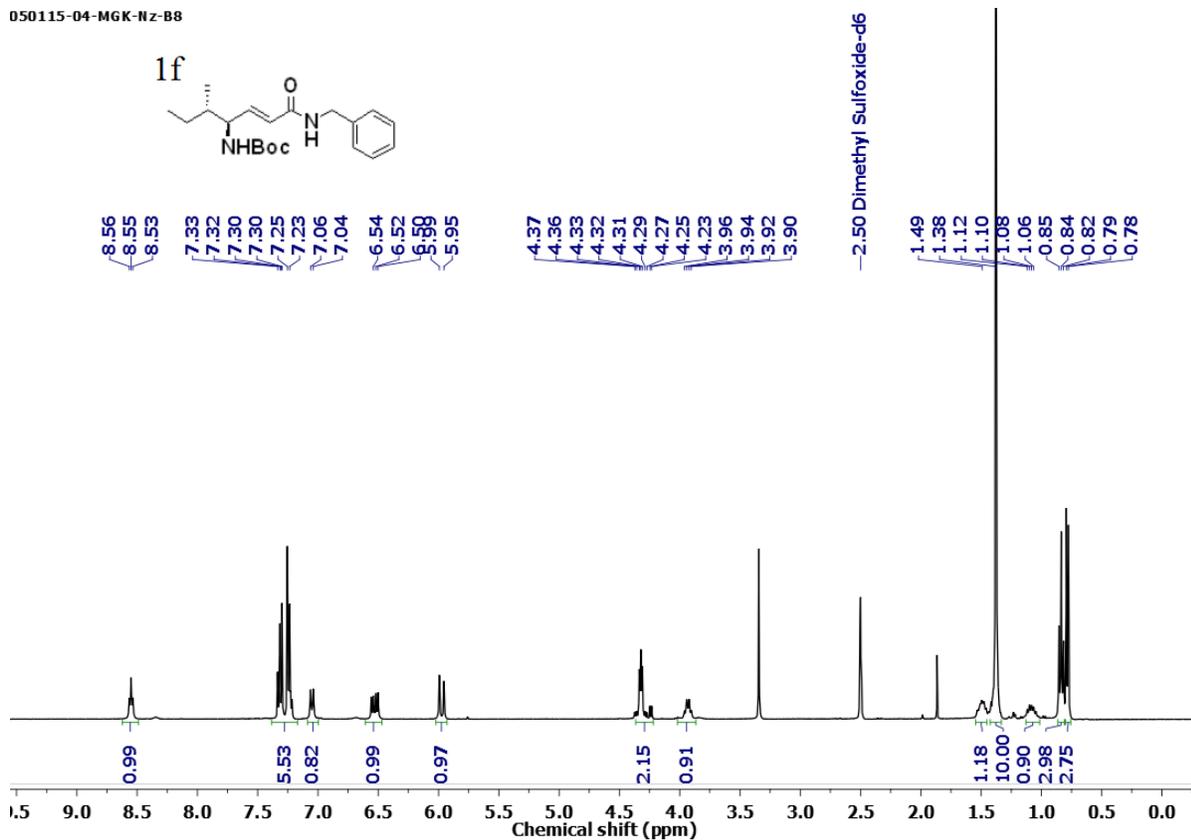


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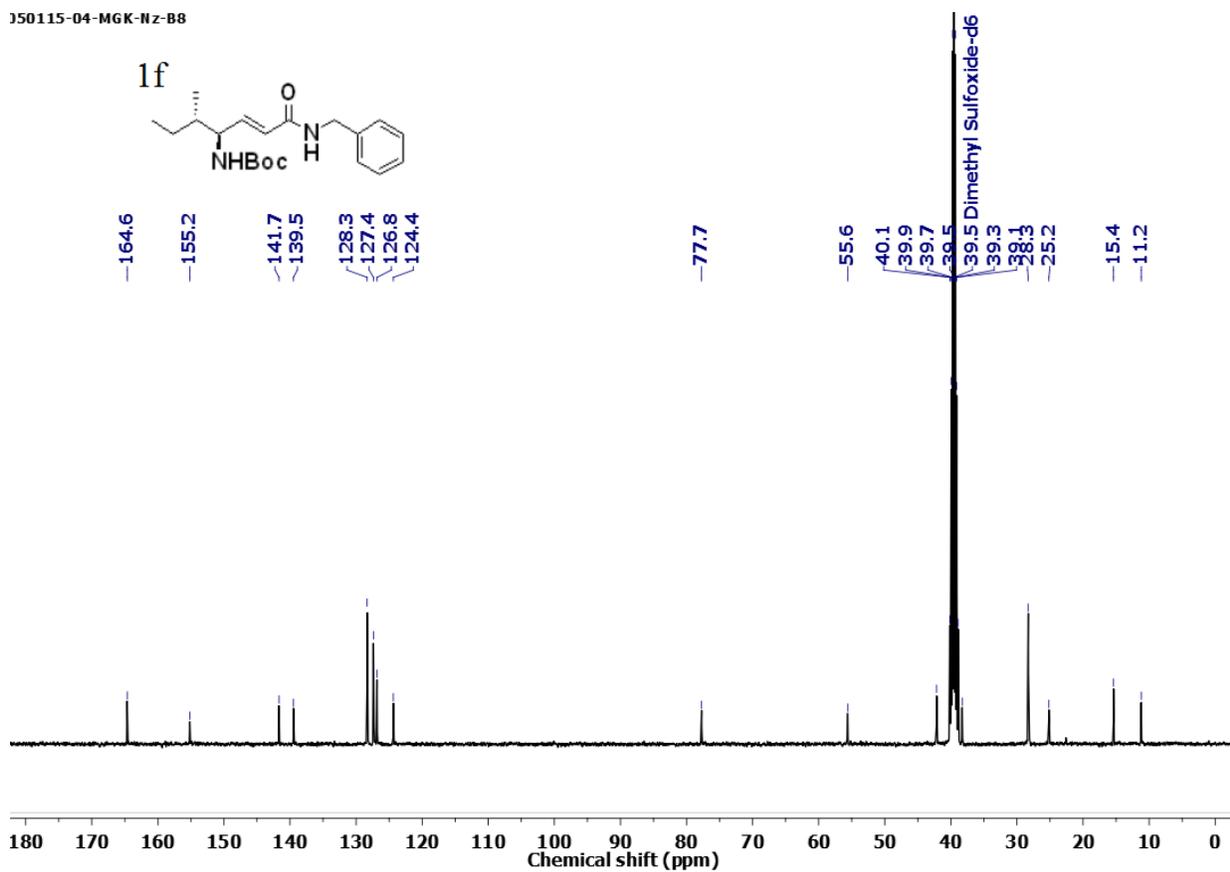
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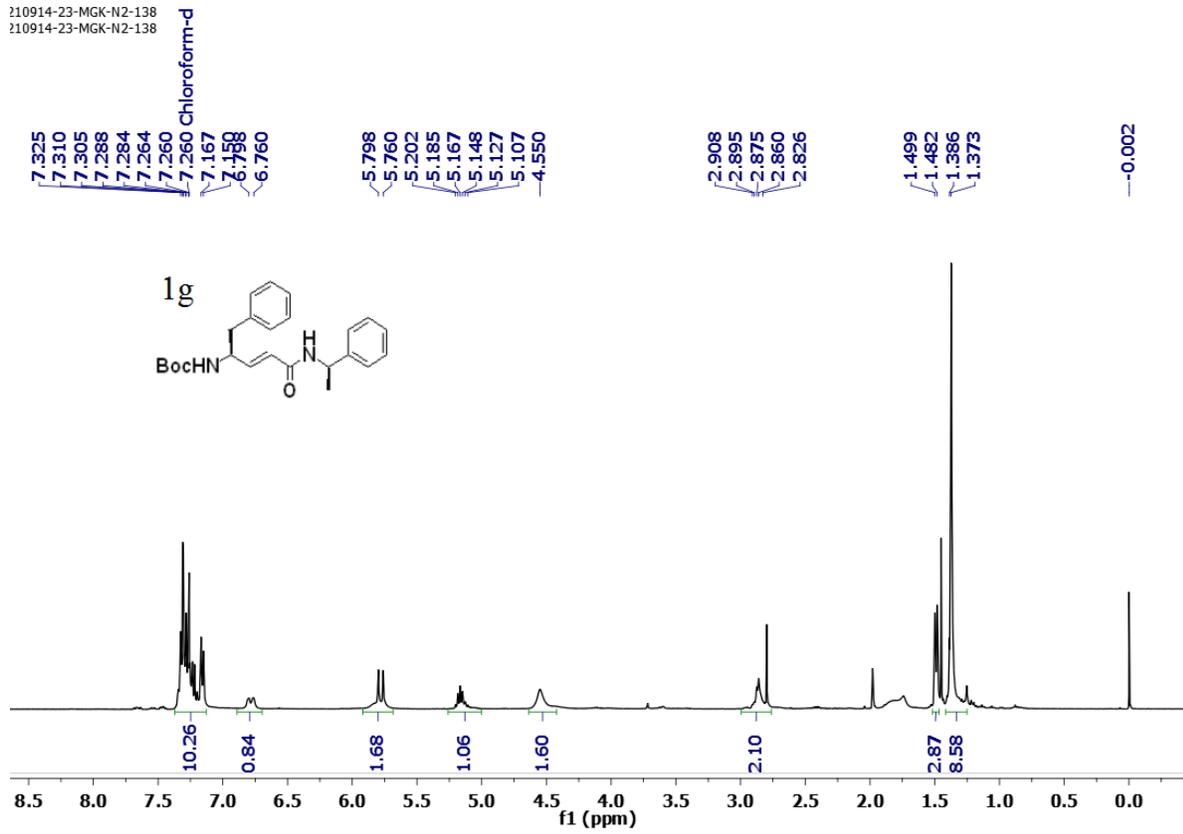
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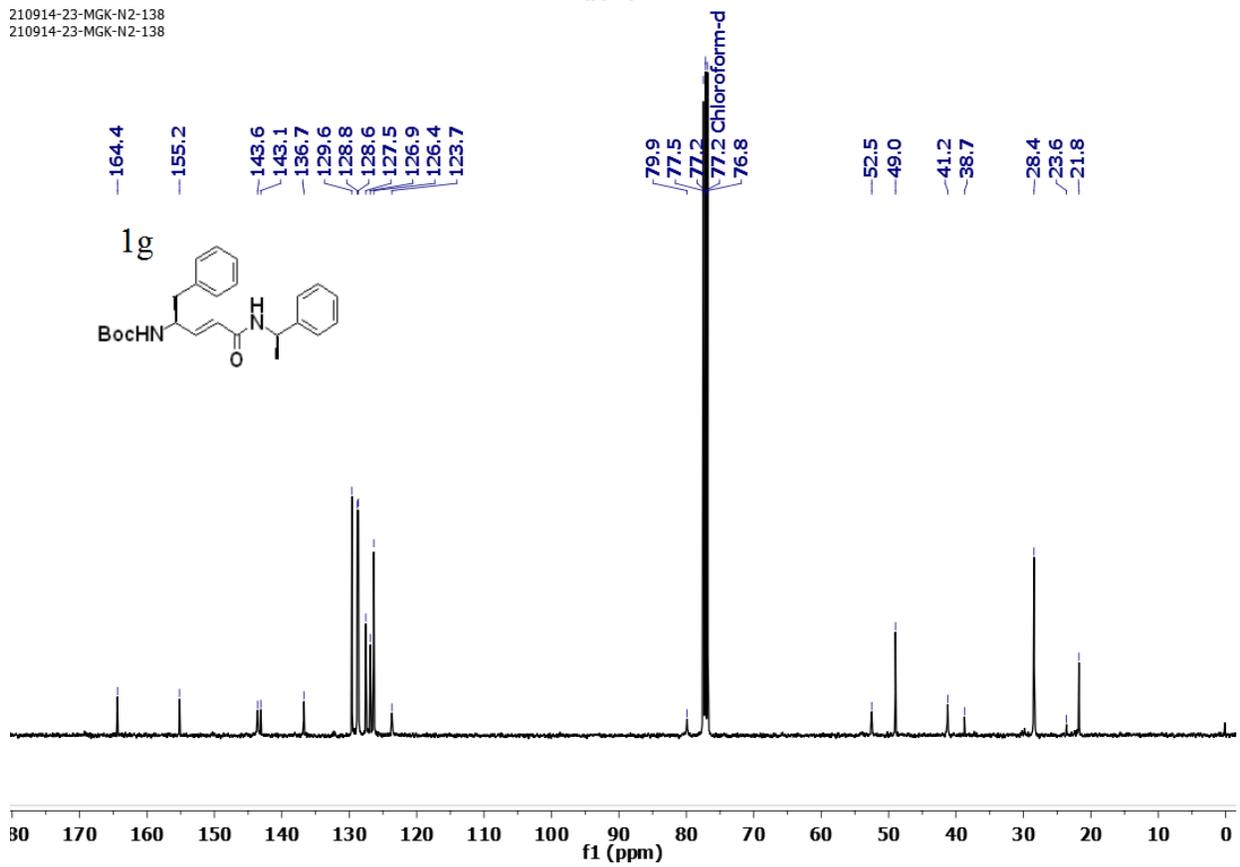
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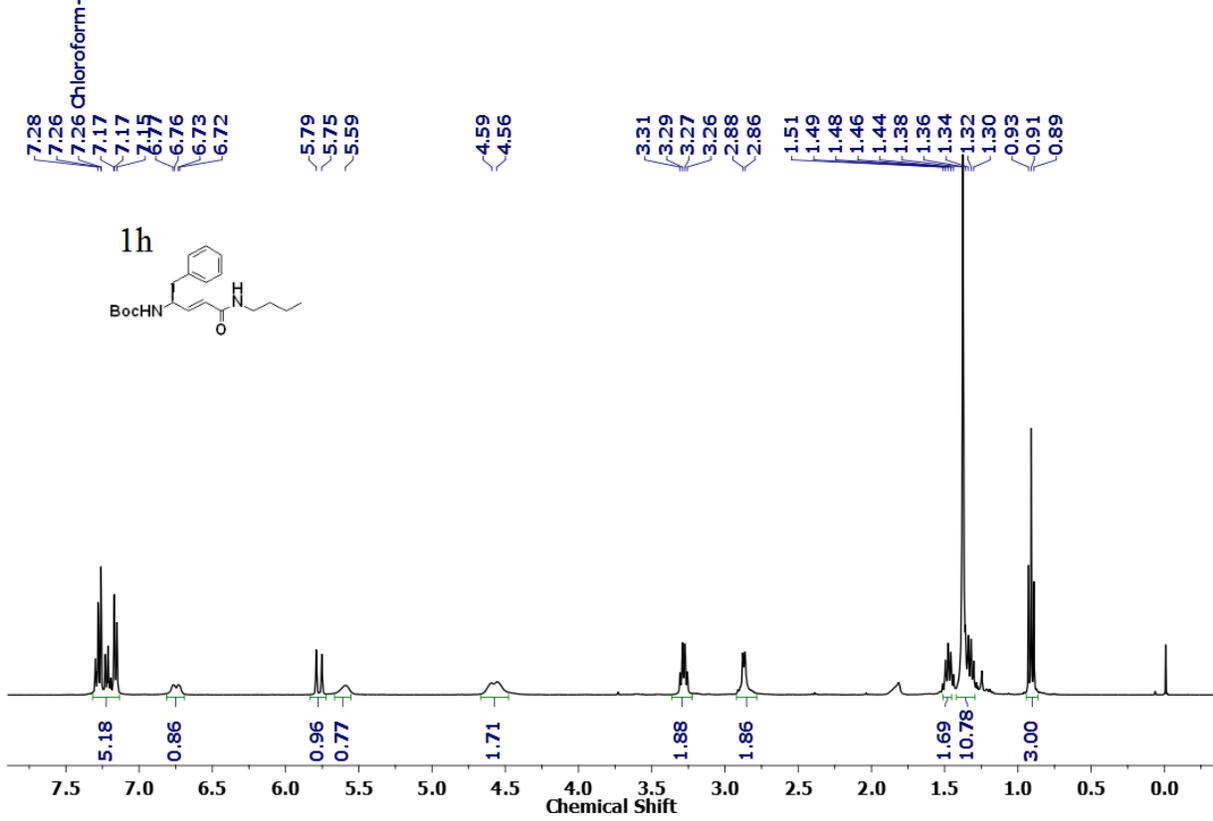
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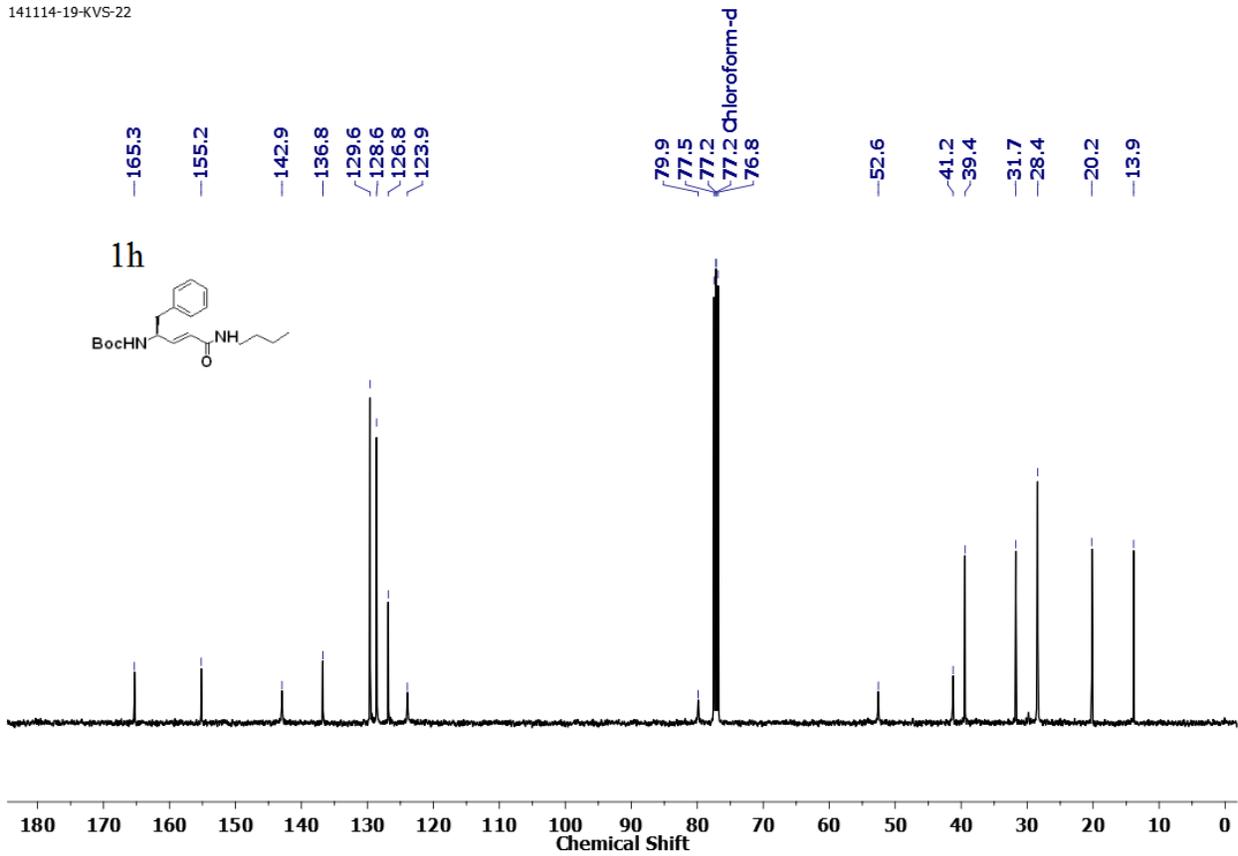
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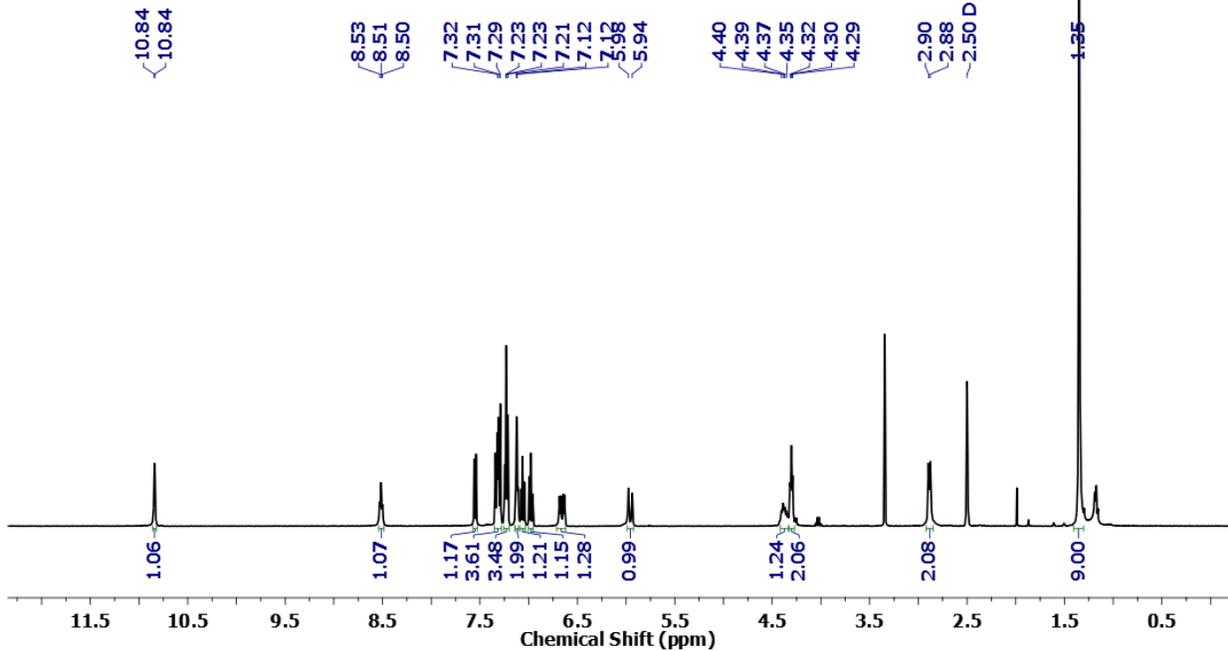
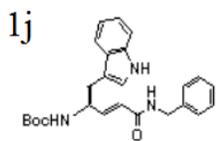
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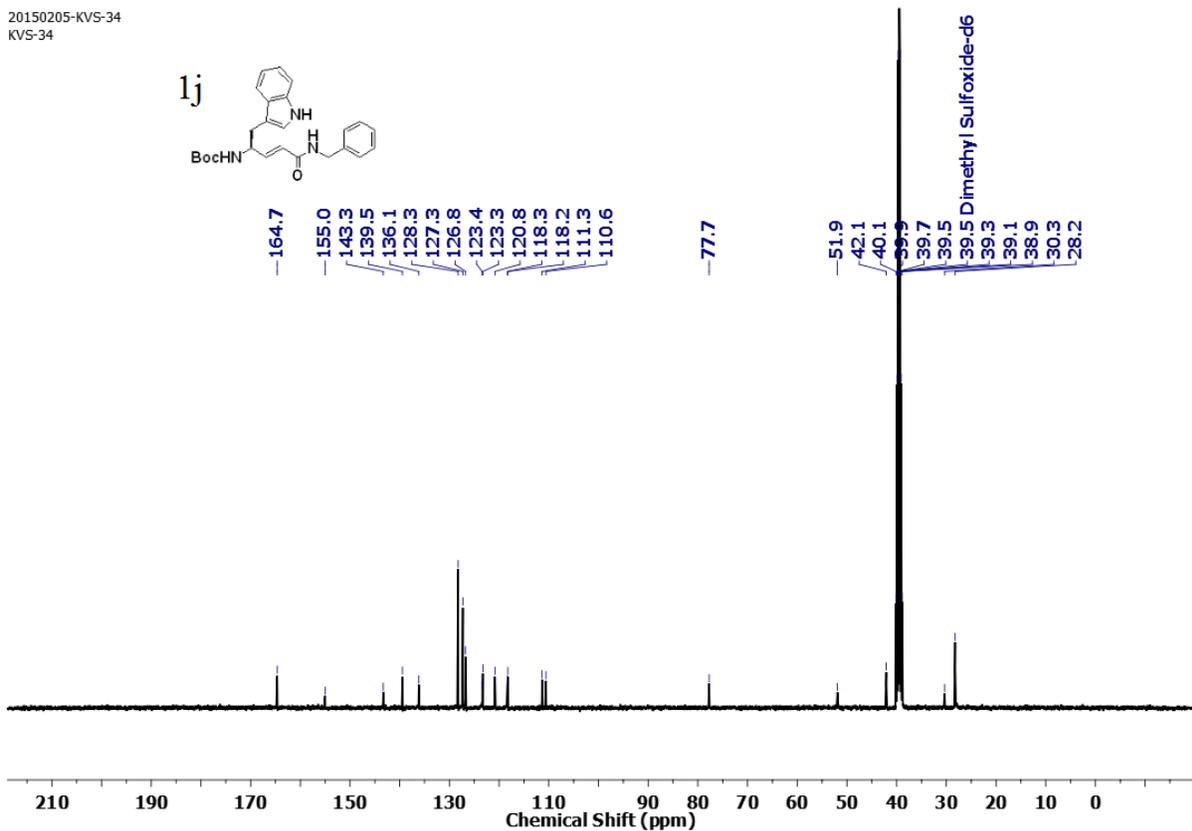
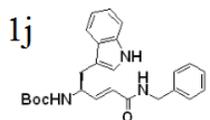
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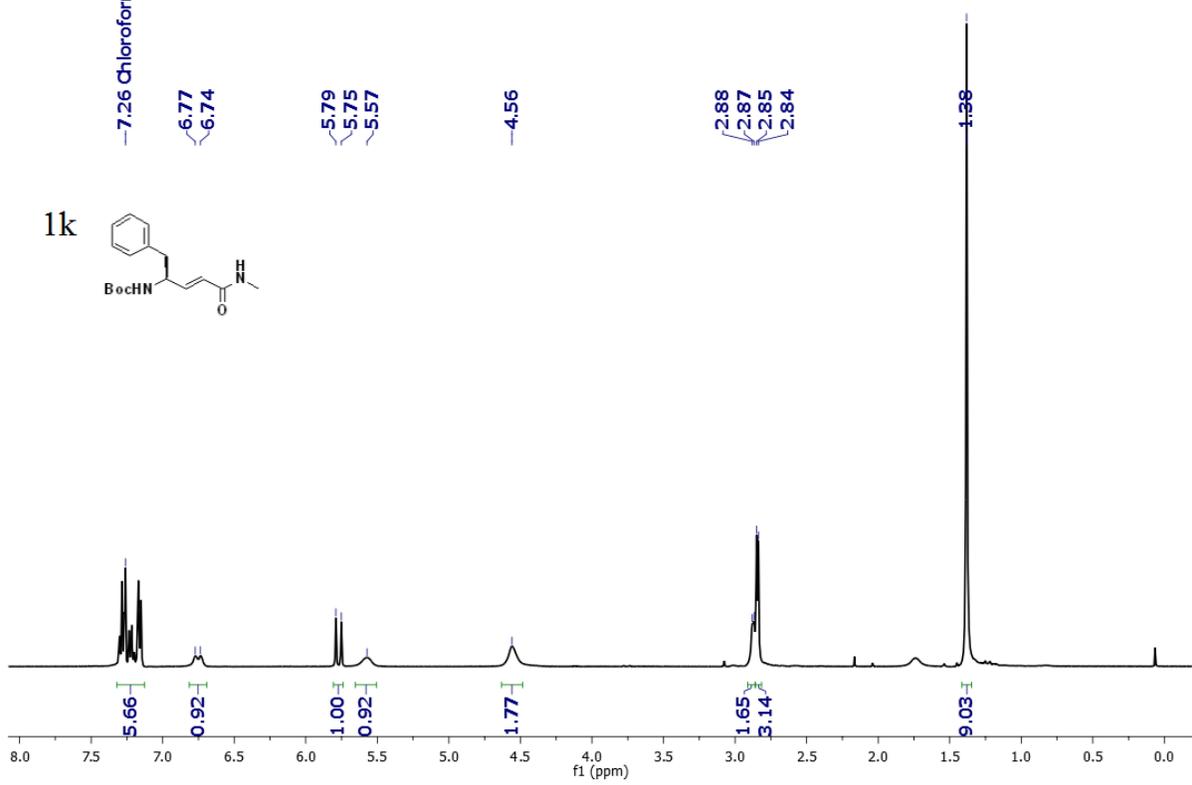
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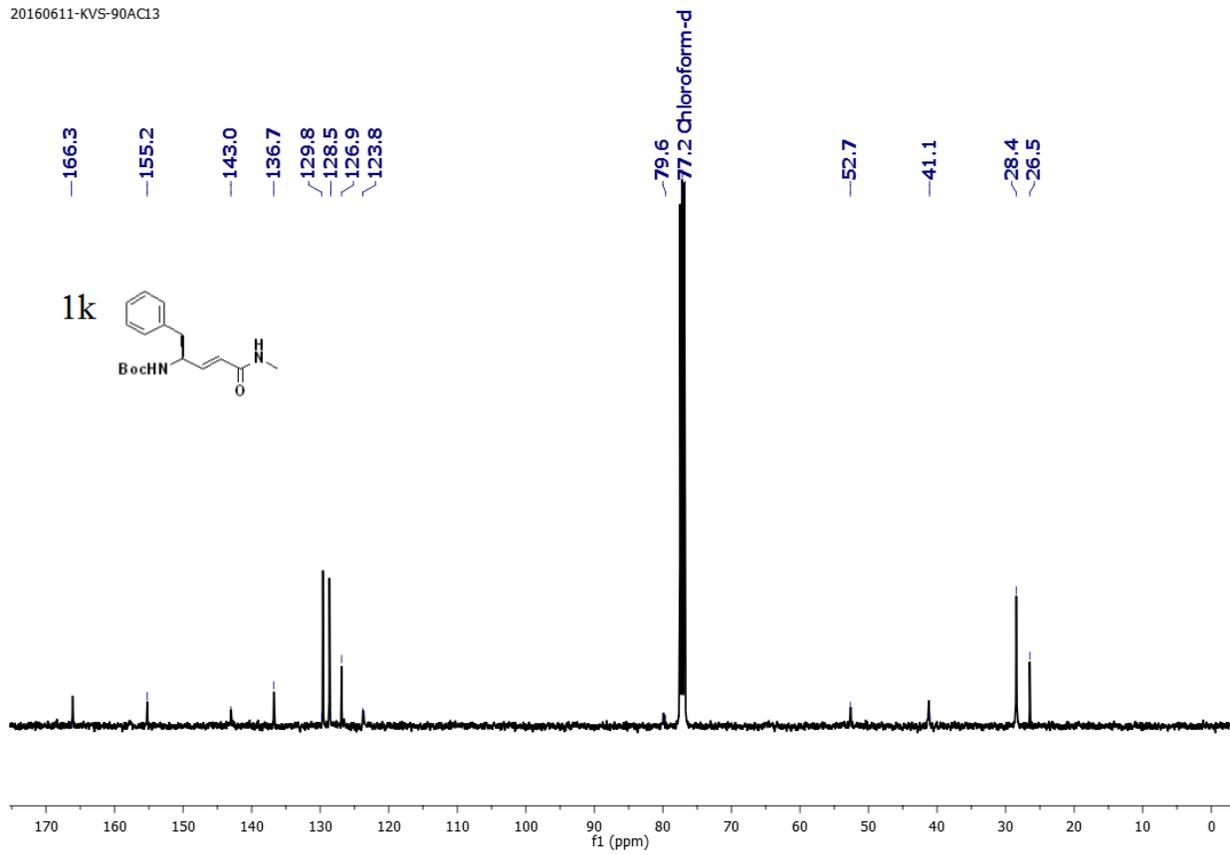
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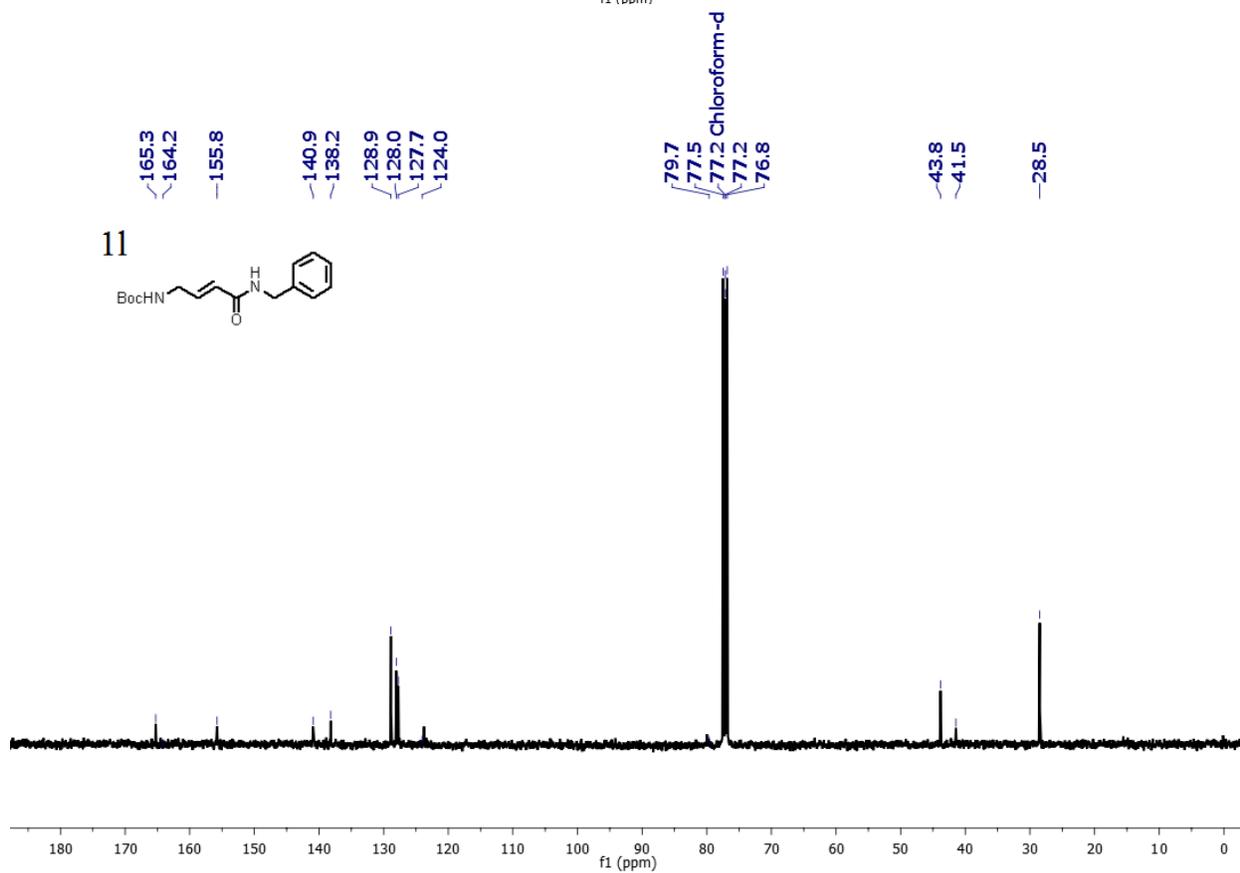
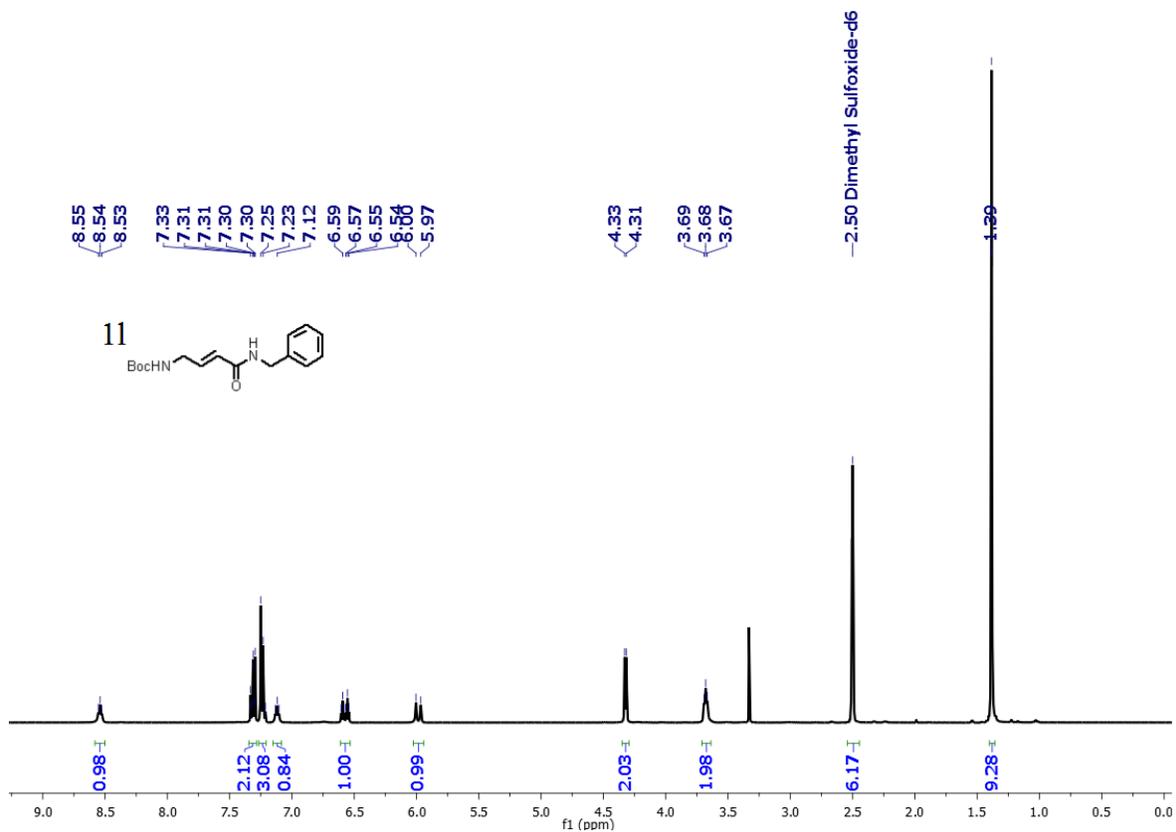


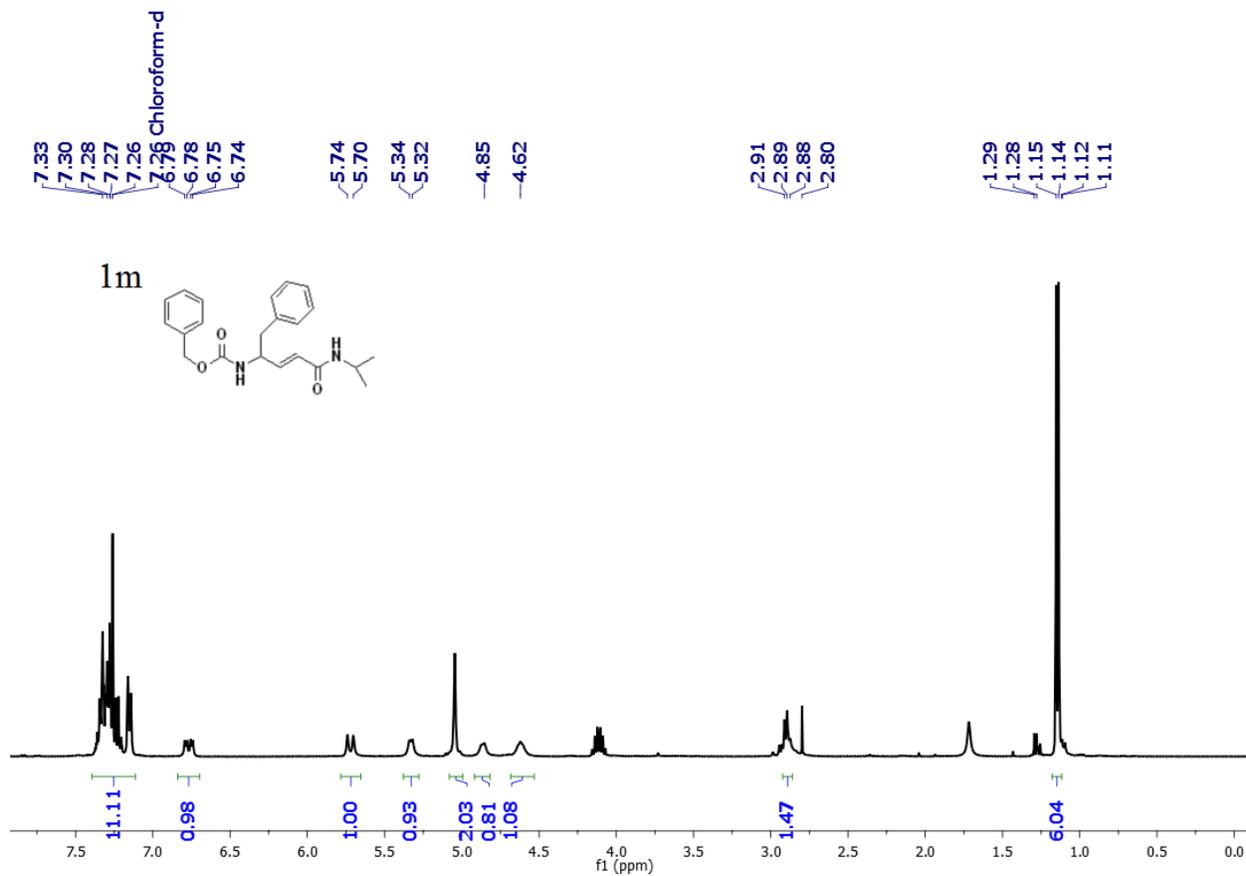
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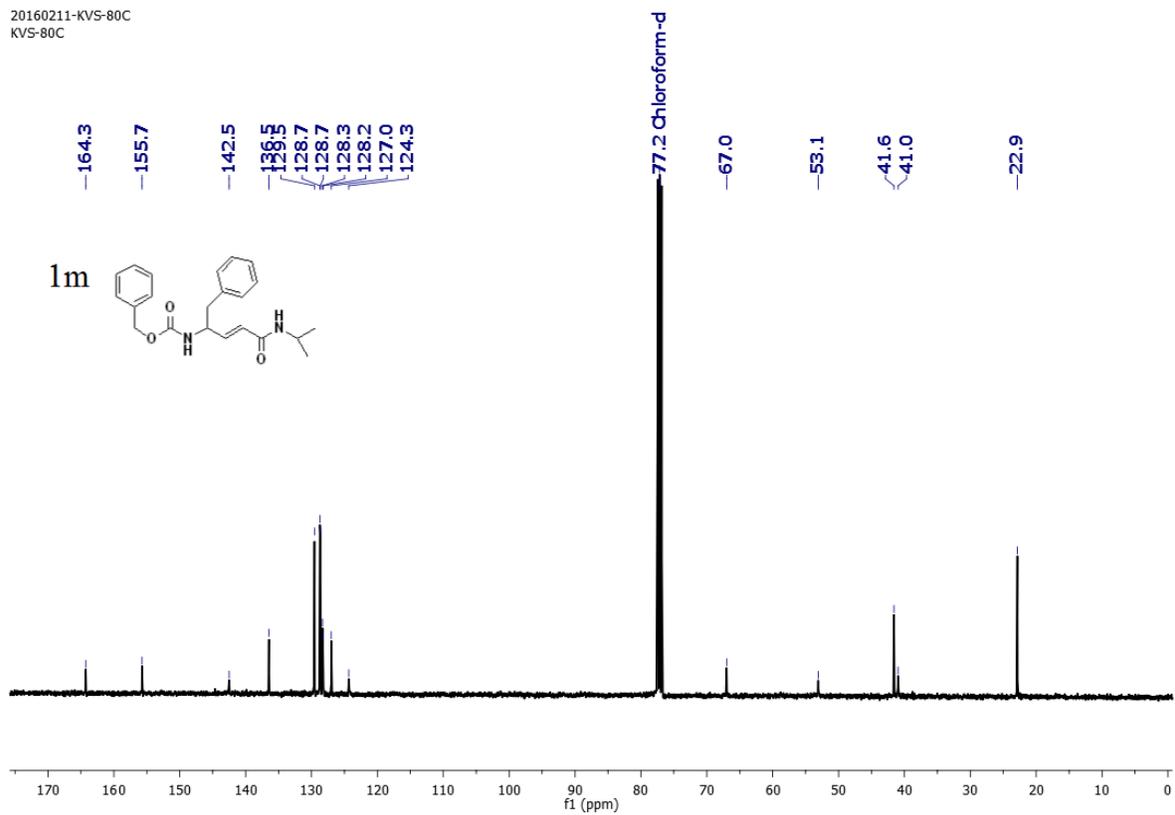
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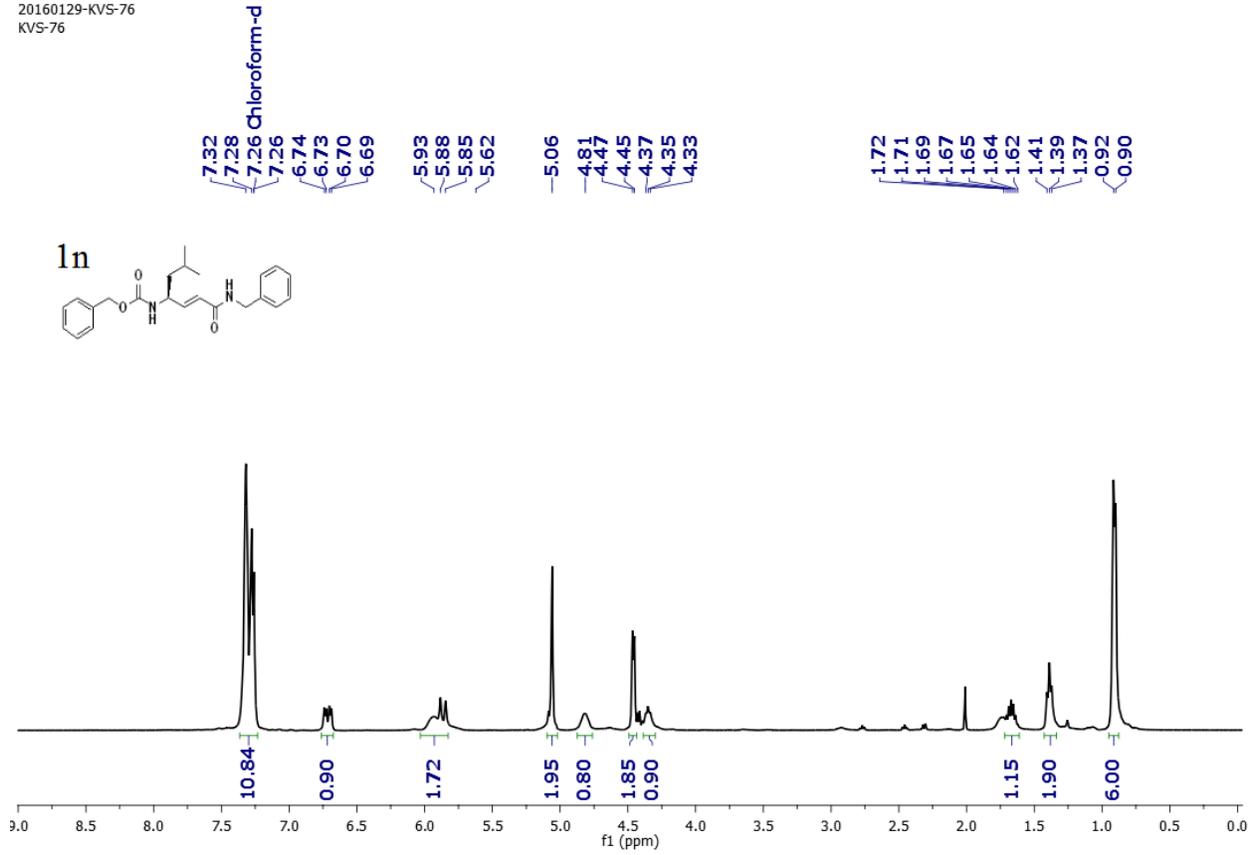




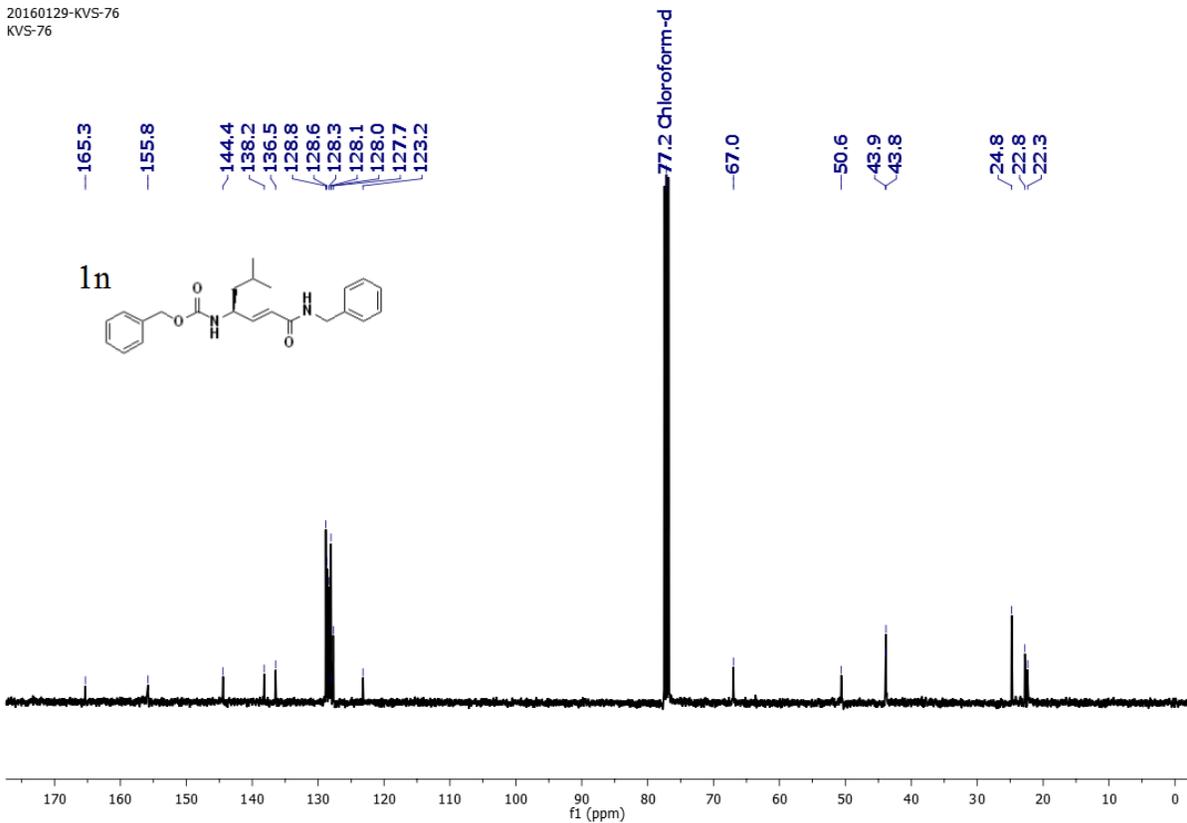
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KVS-80C



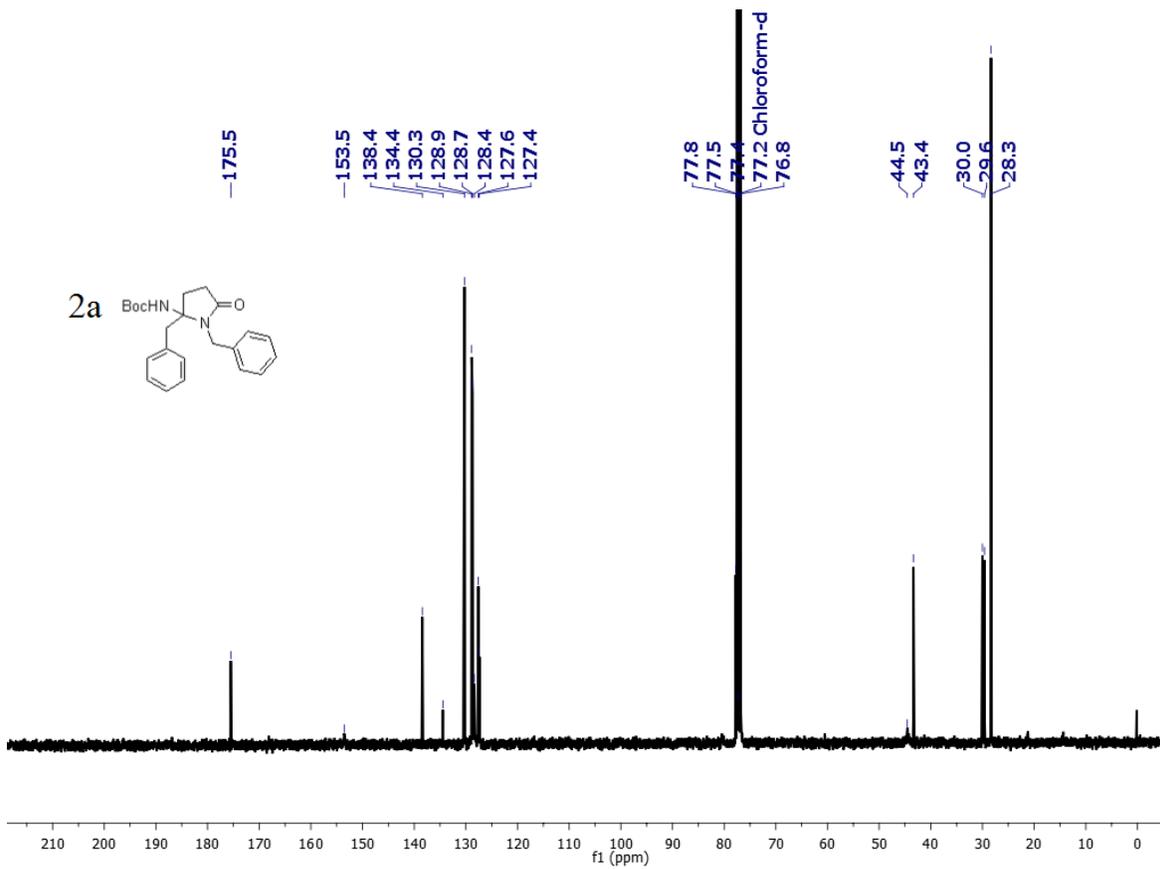
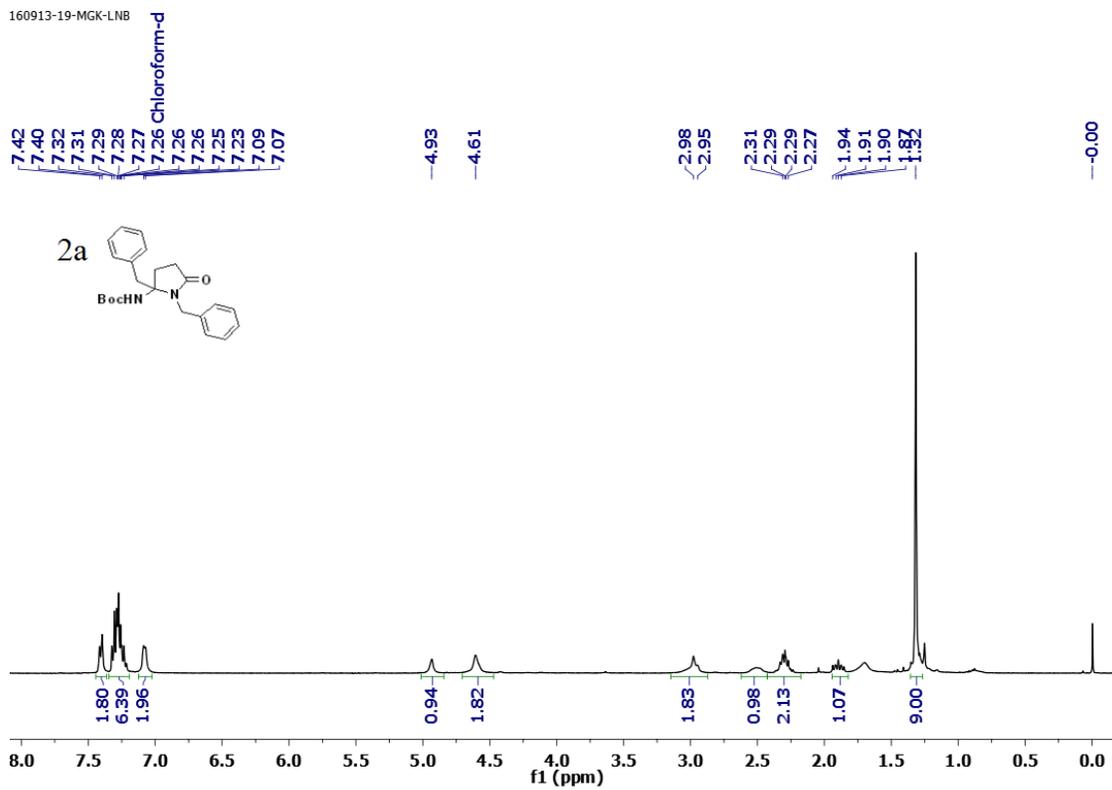
20160129-KVS-76
KVS-76

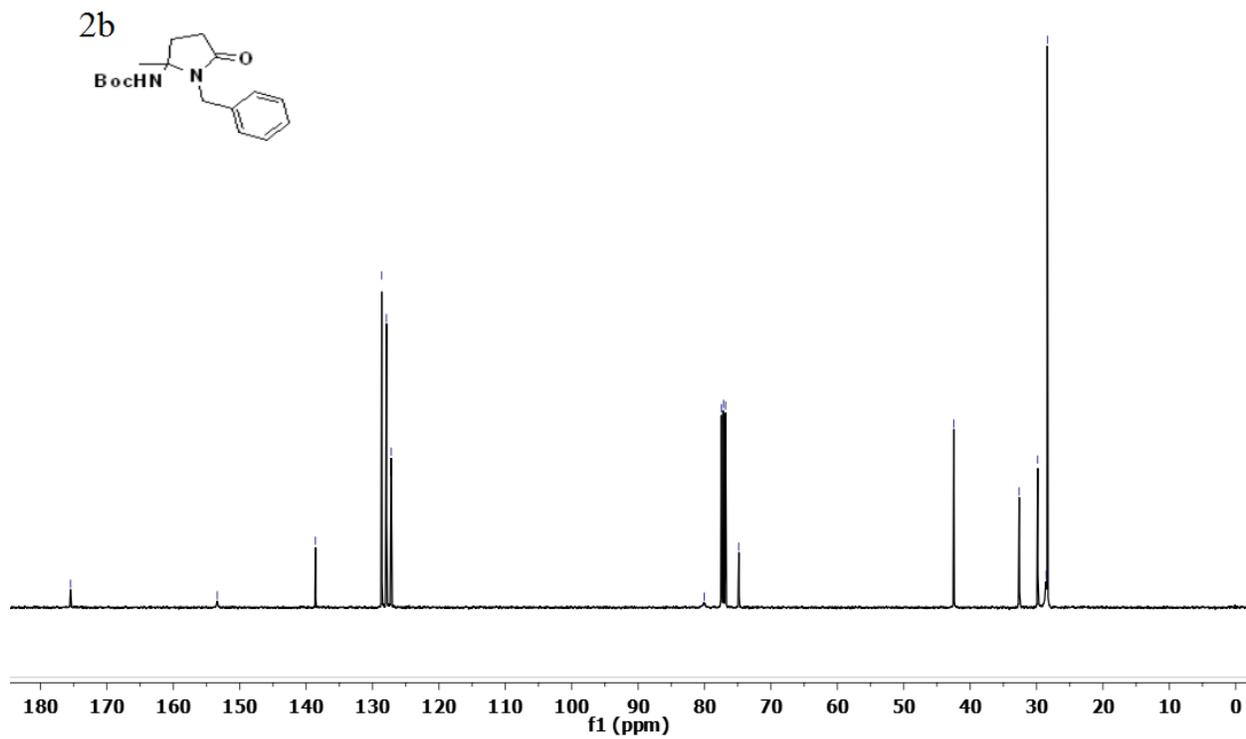
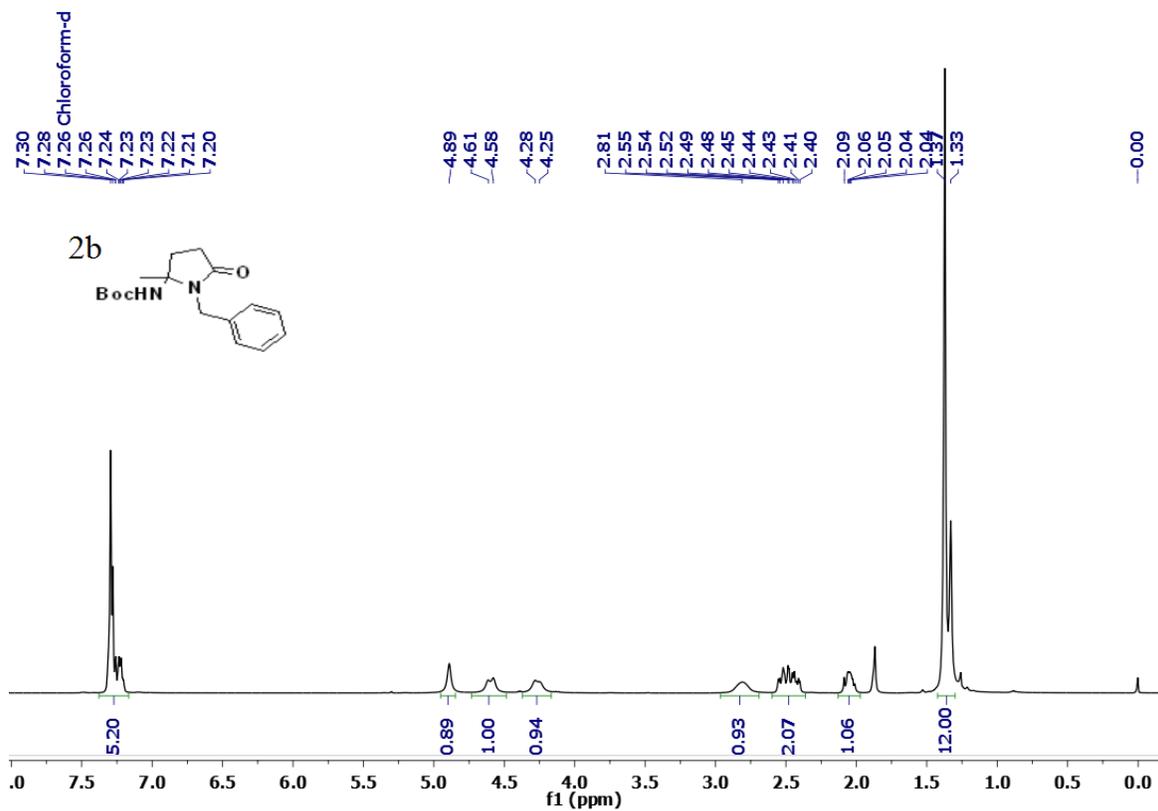


20160129-KVS-76
KVS-76

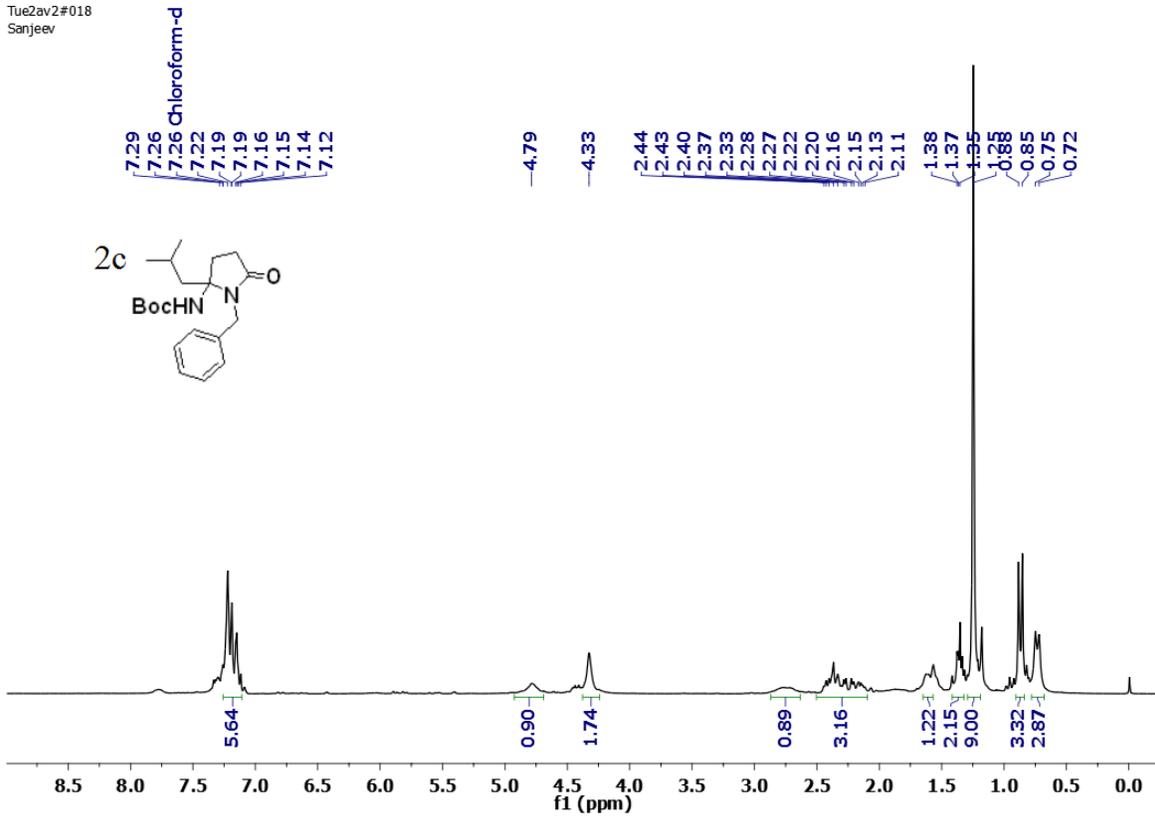


160913-19-MGK-LNB

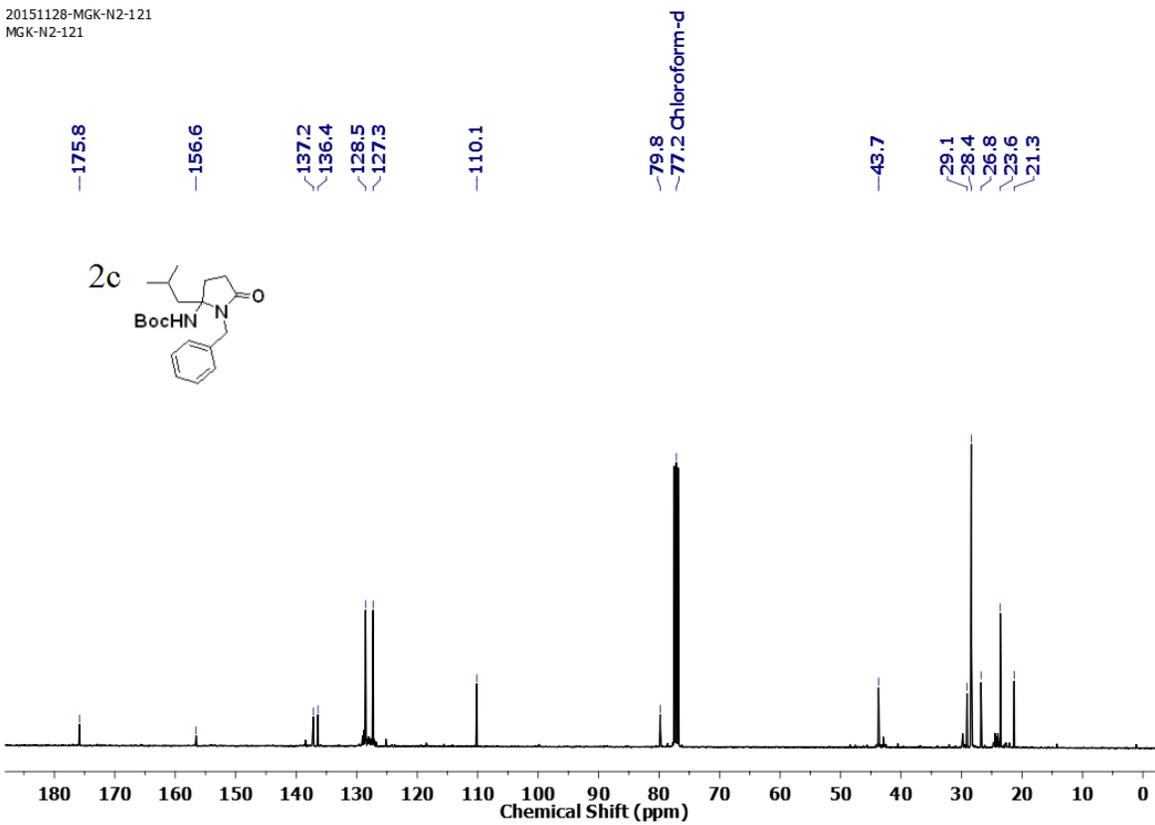




Tue2av2#018
Sanjeev



20151128-MGK-N2-121
MGK-N2-121



210914-17-MGK-N2-136
single_pulse

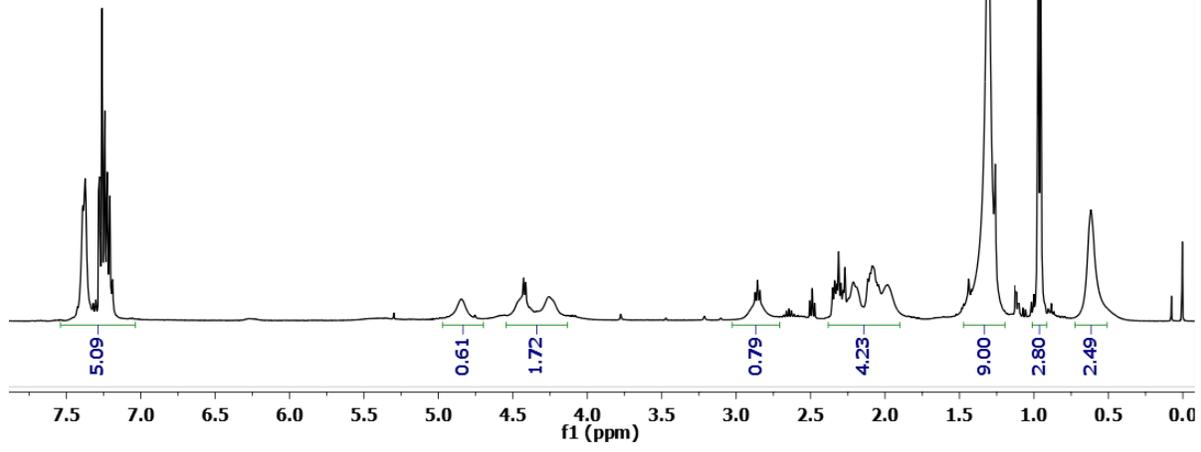
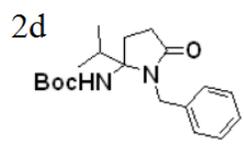
7.389
7.373
7.283
7.280
7.277
7.260 Chloroform-d
7.241
7.228
7.225
7.207

4.850
4.428
4.413
4.263
4.259
4.230

2.873
2.856
2.844
2.840

2.339
2.312
2.296
2.270
2.212
2.115
2.101
1.988

0.972
0.955
0.616

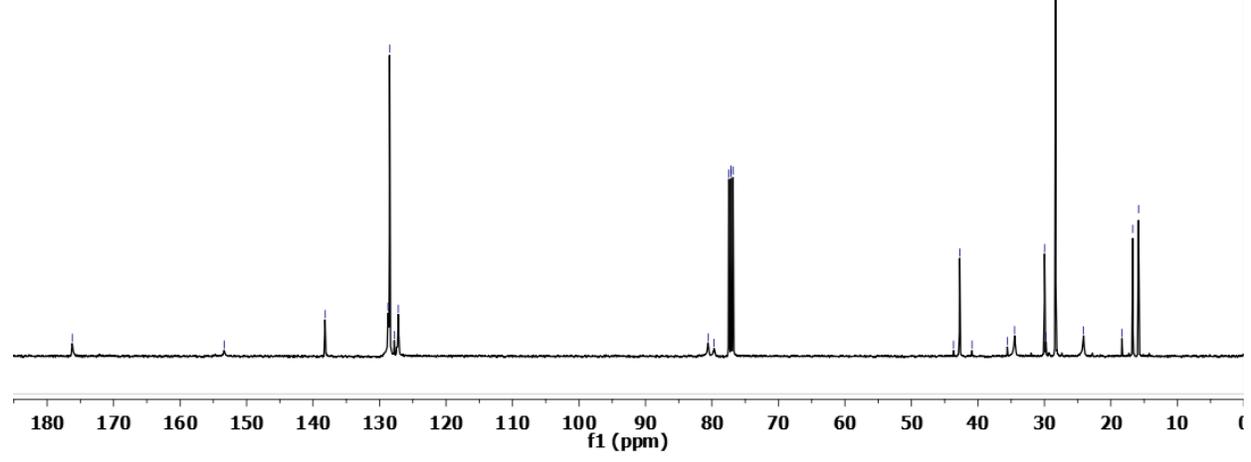
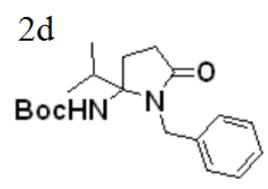


210914-17-MGK-N2-136
210914-17-MGK-N2-136

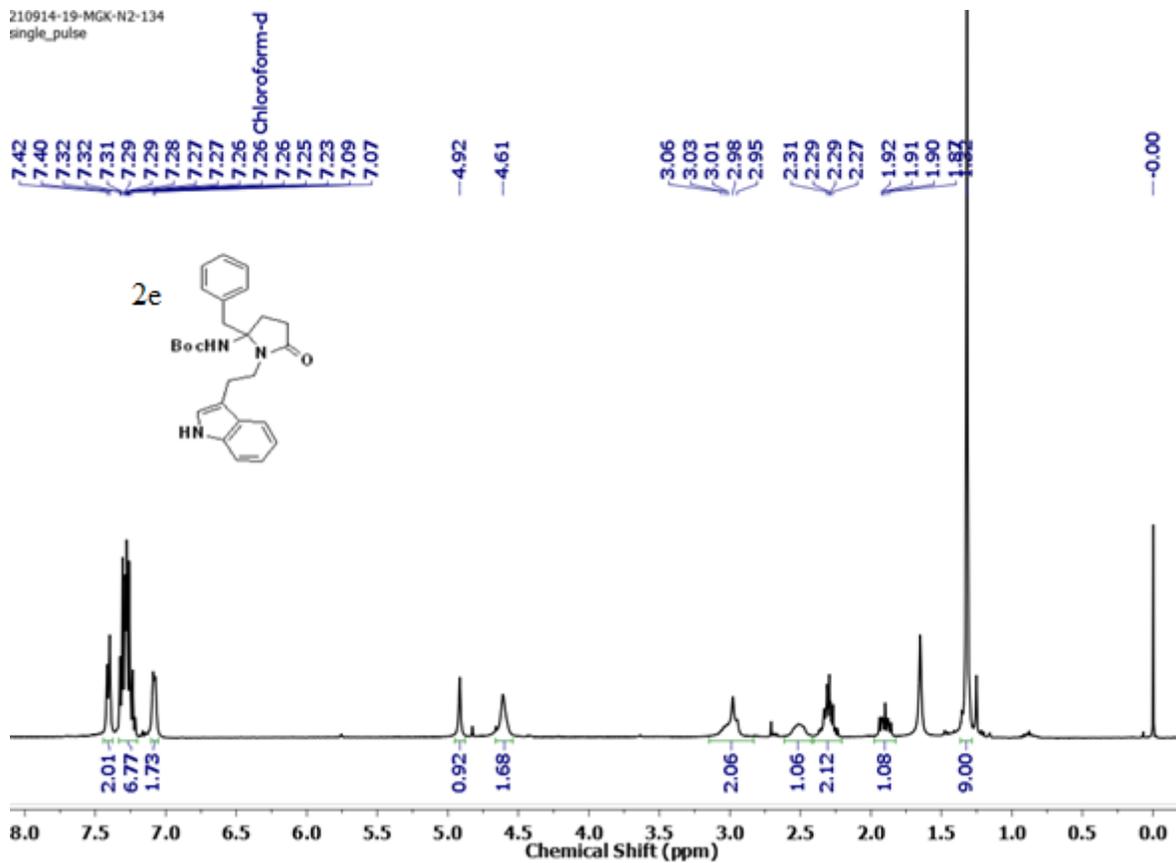
176.2
153.4
138.2
128.7
128.5
127.8
127.2

80.6
79.7
77.5
77.2 Chloroform-d
76.8

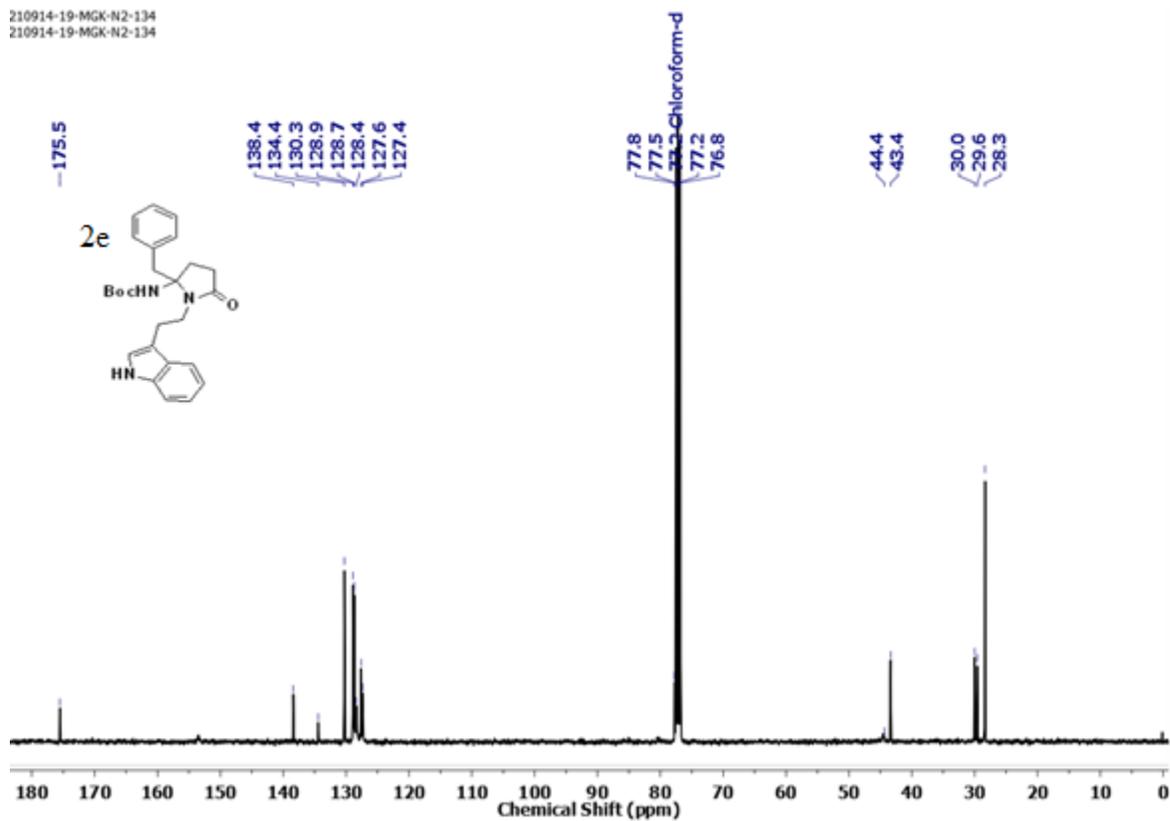
43.7
42.7
40.9
35.6
34.5
30.0
29.8
28.3
24.1
18.3
16.7
15.8



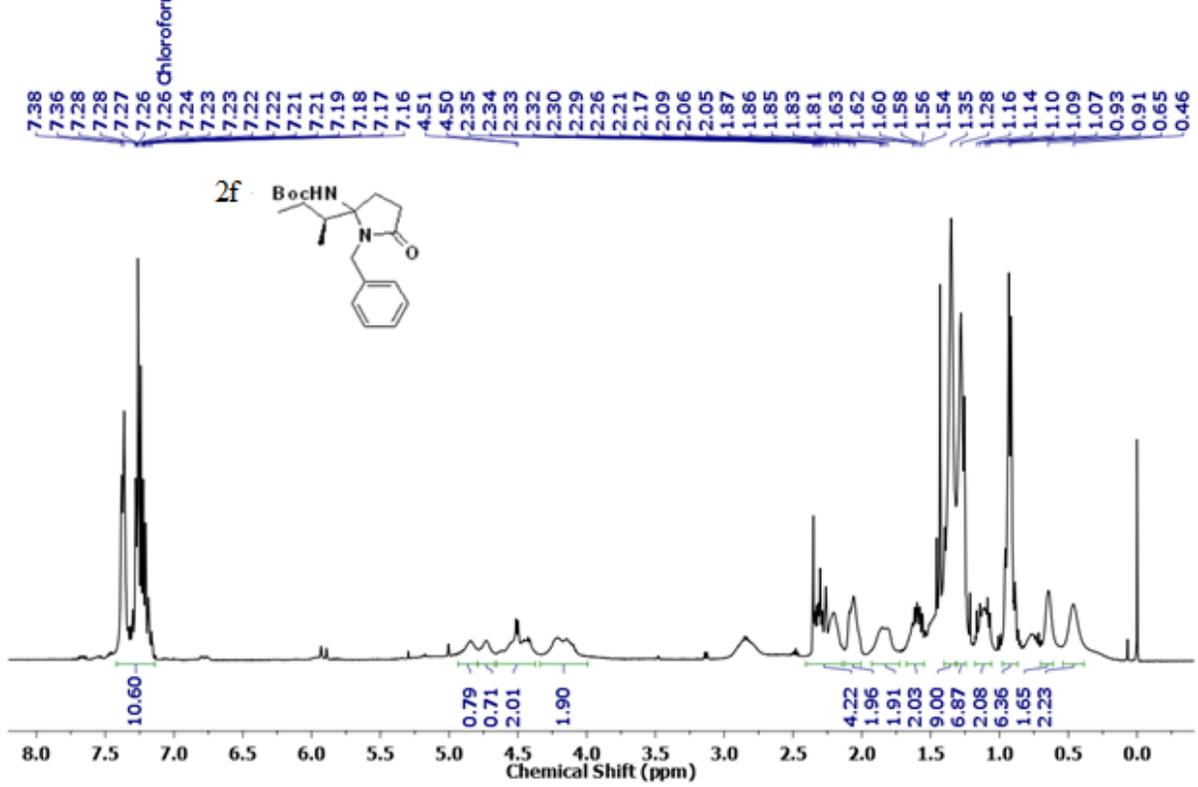
210914-19-MGK-N2-134
single_pulse



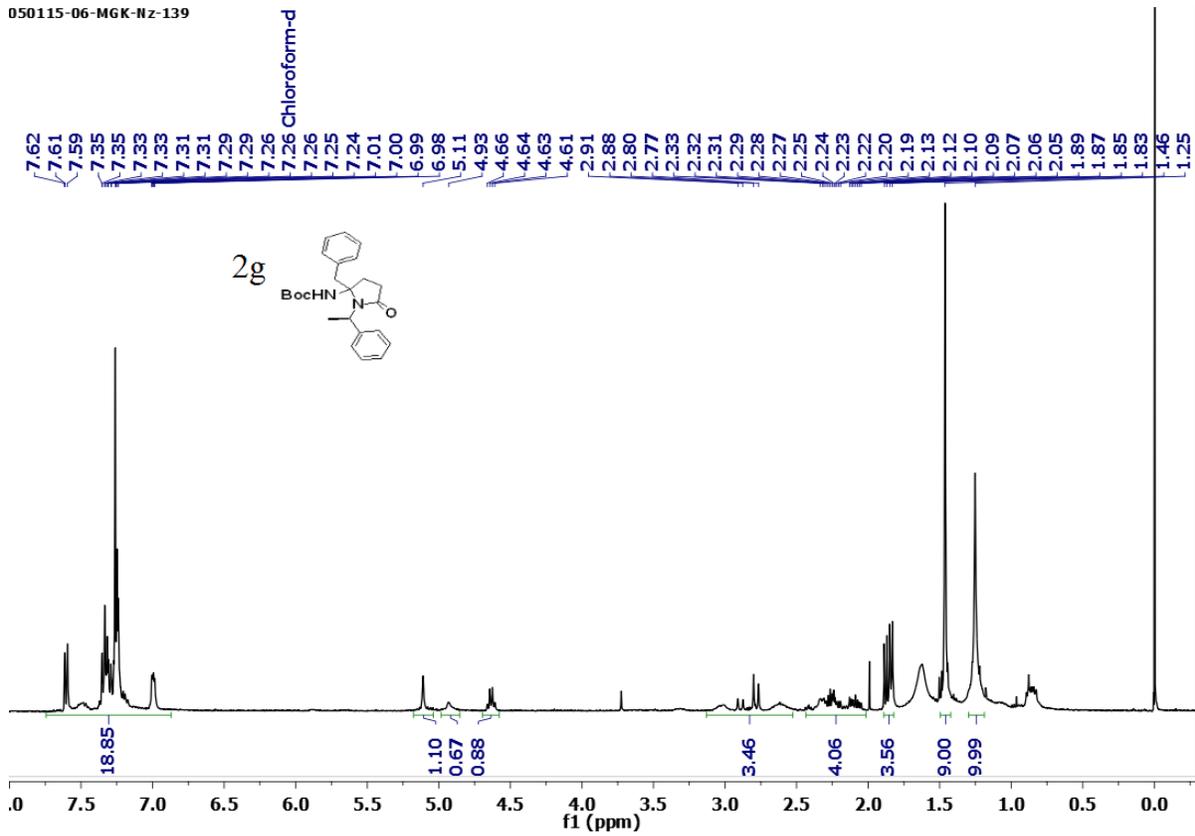
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210914-19-MGK-N2-134



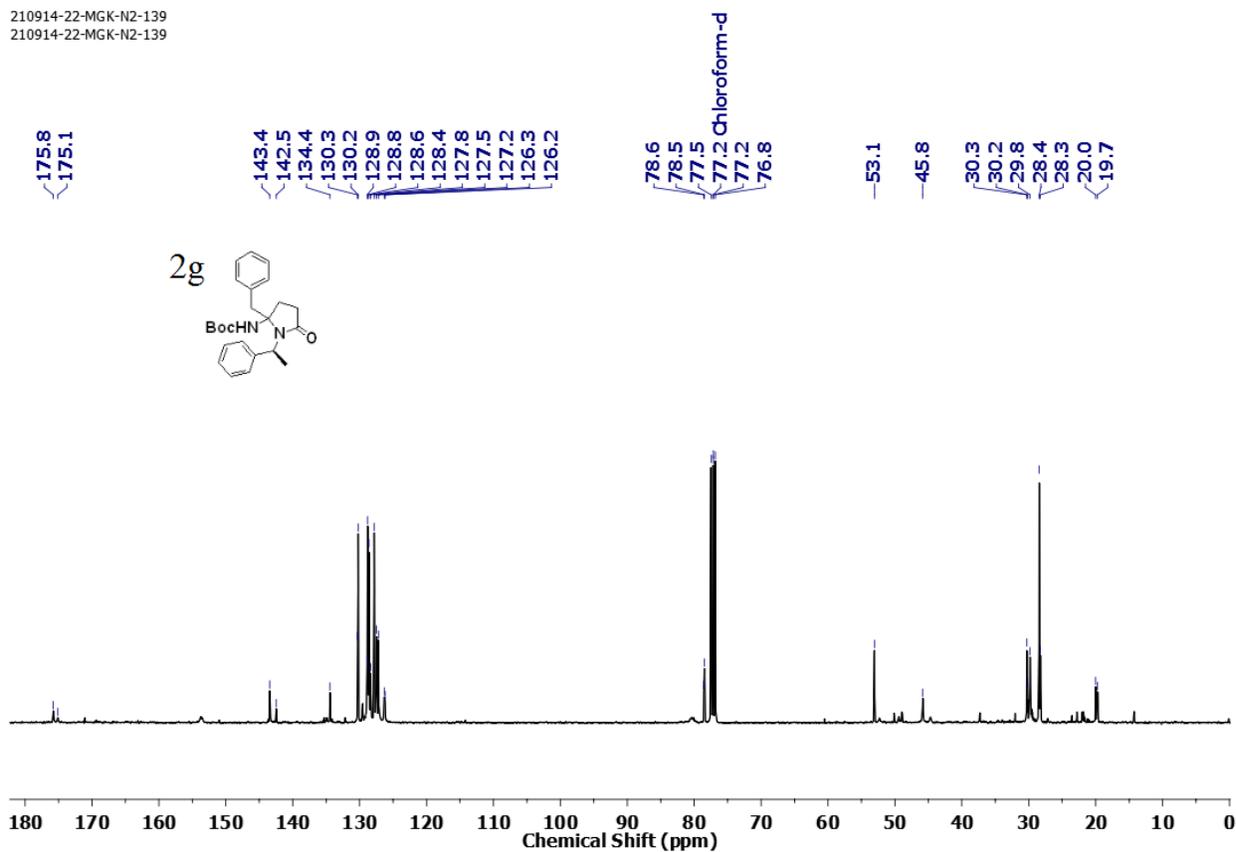
210914-24-MGK-N2-133
210914-24-MGK-N2-133



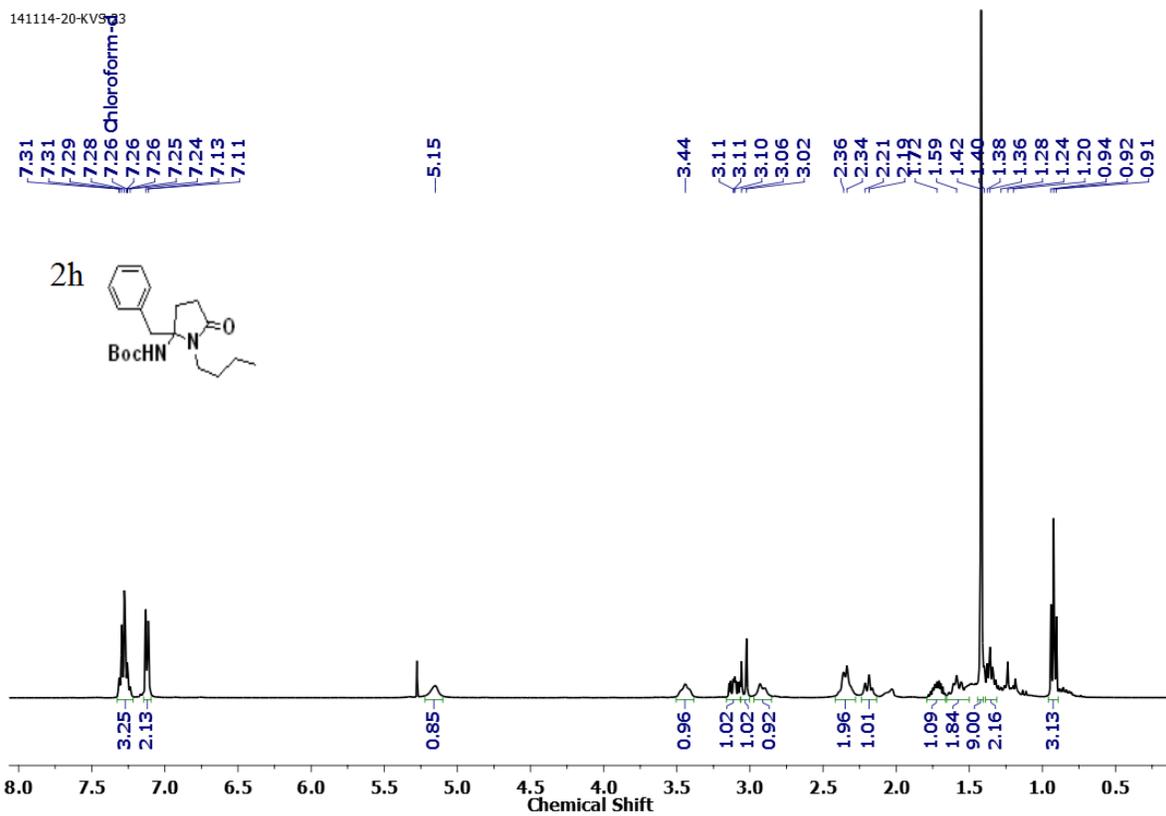
050115-06-MGK-Nz-139



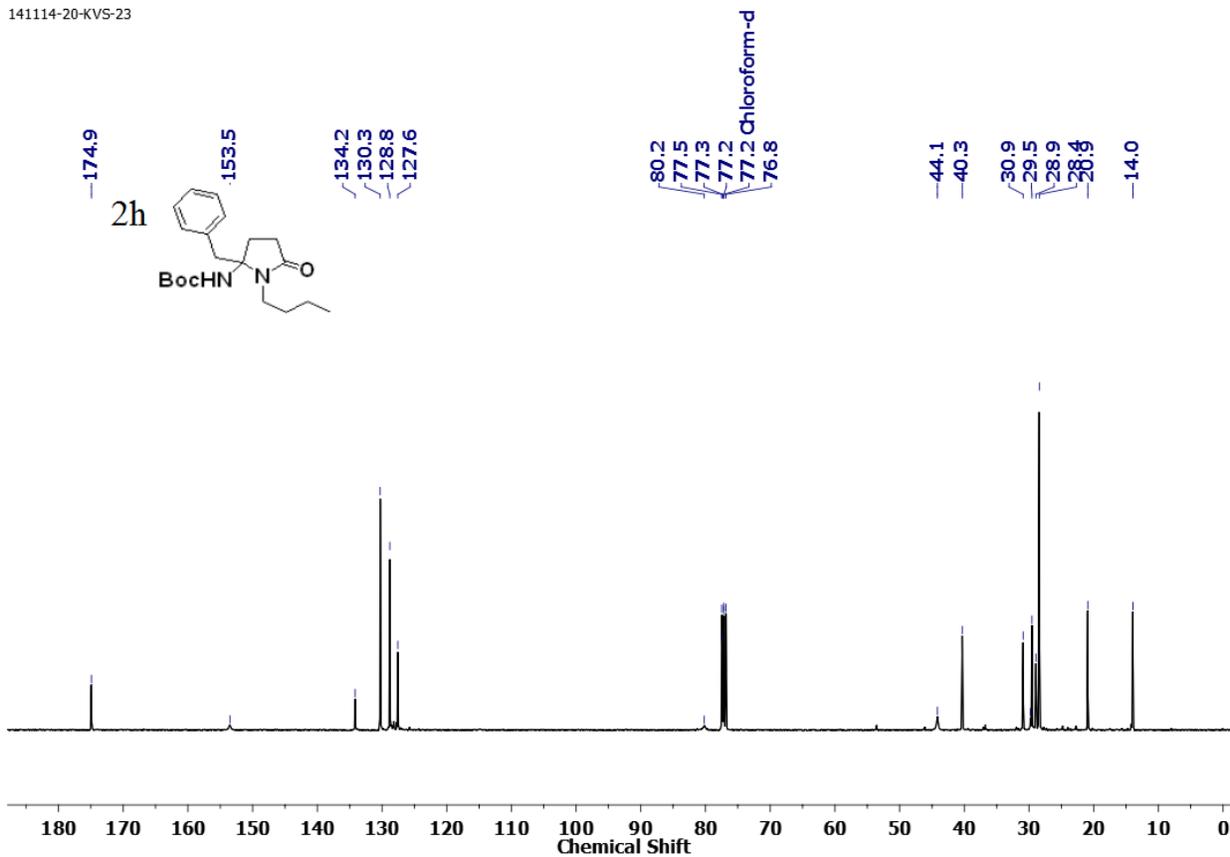
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210914-22-MGK-N2-139



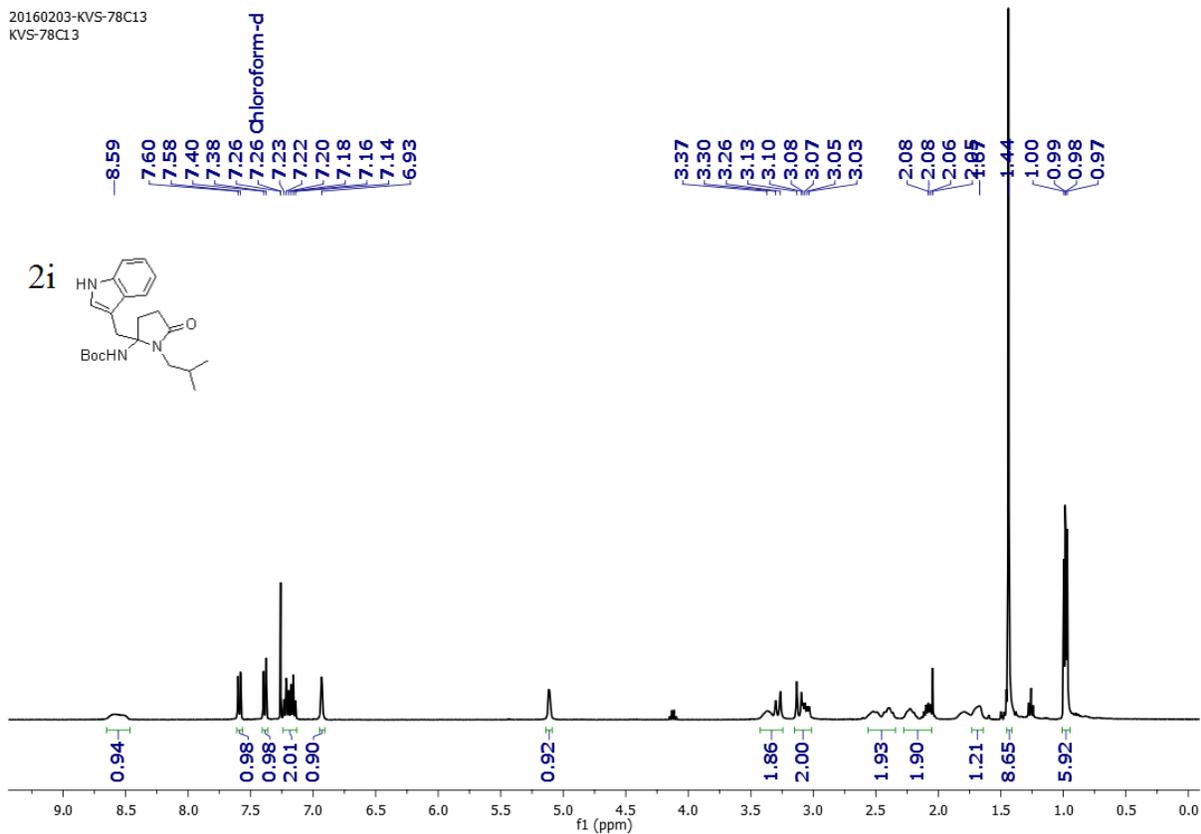
141114-20-KVS-93

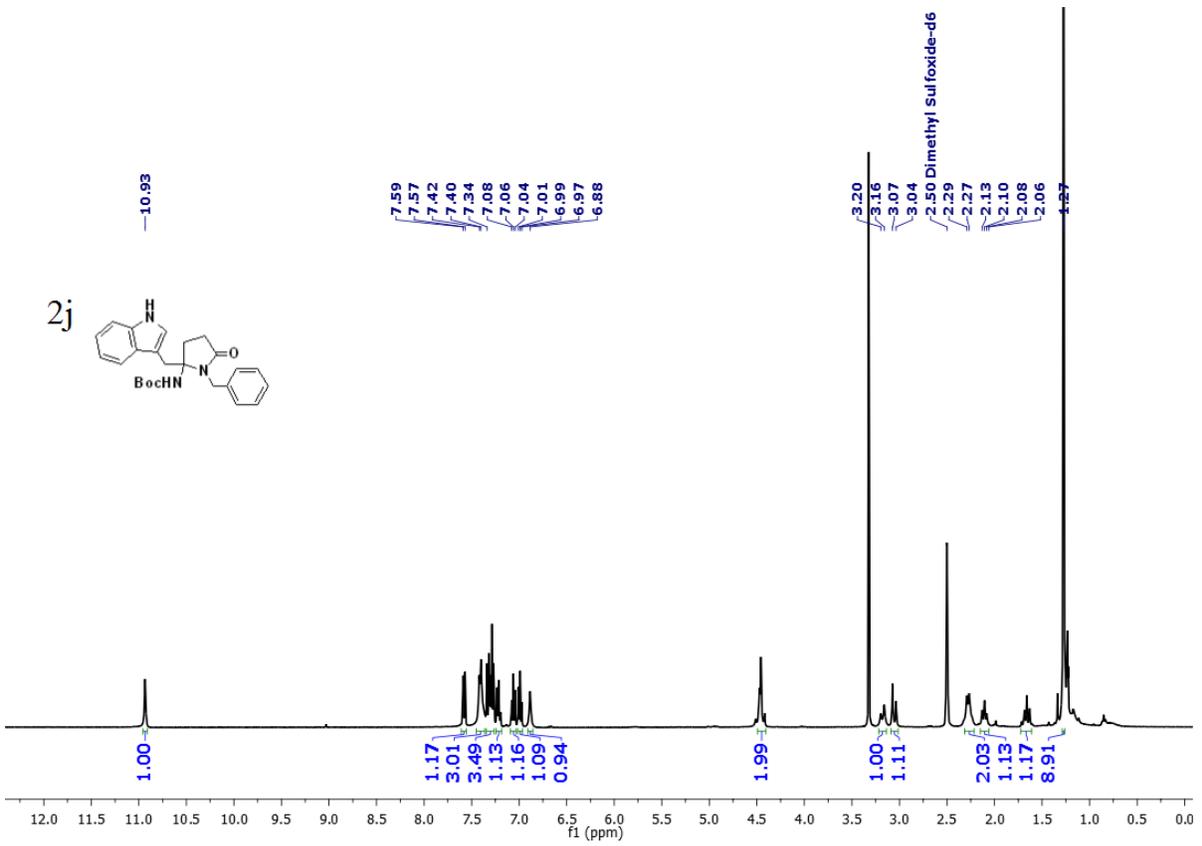
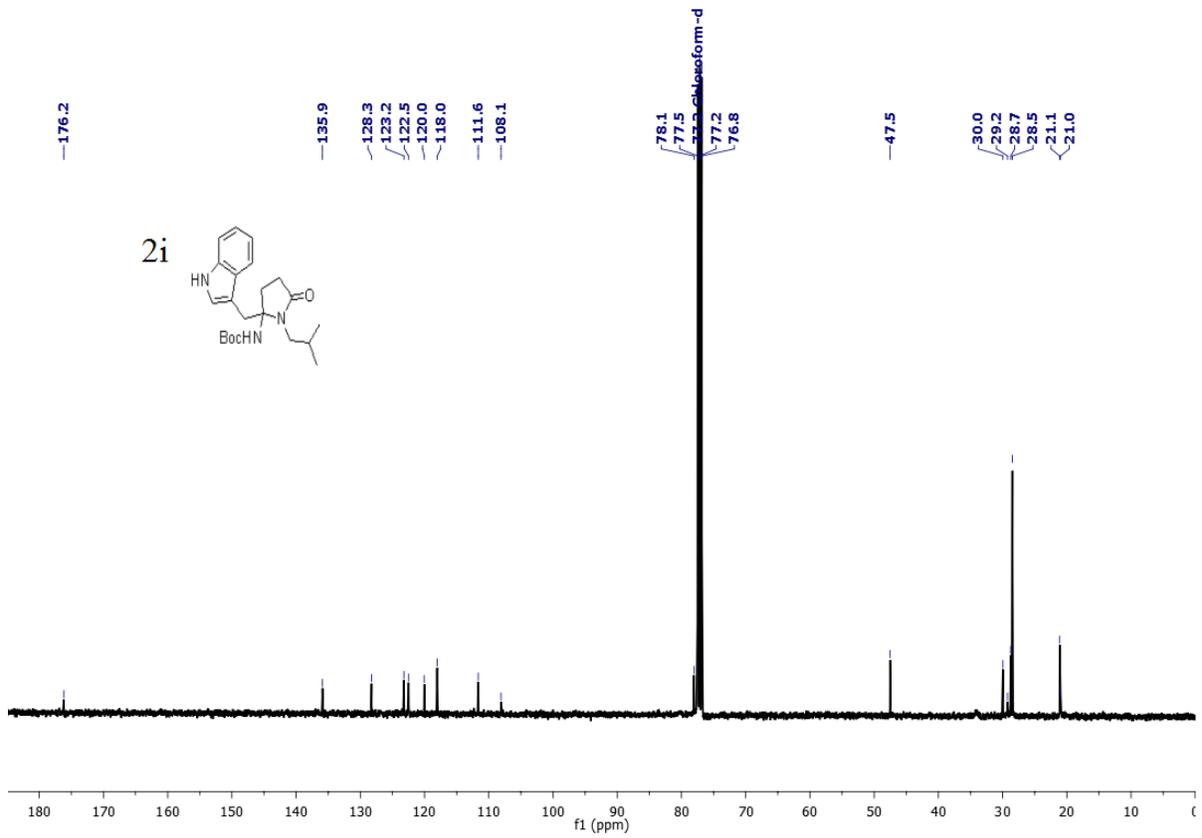


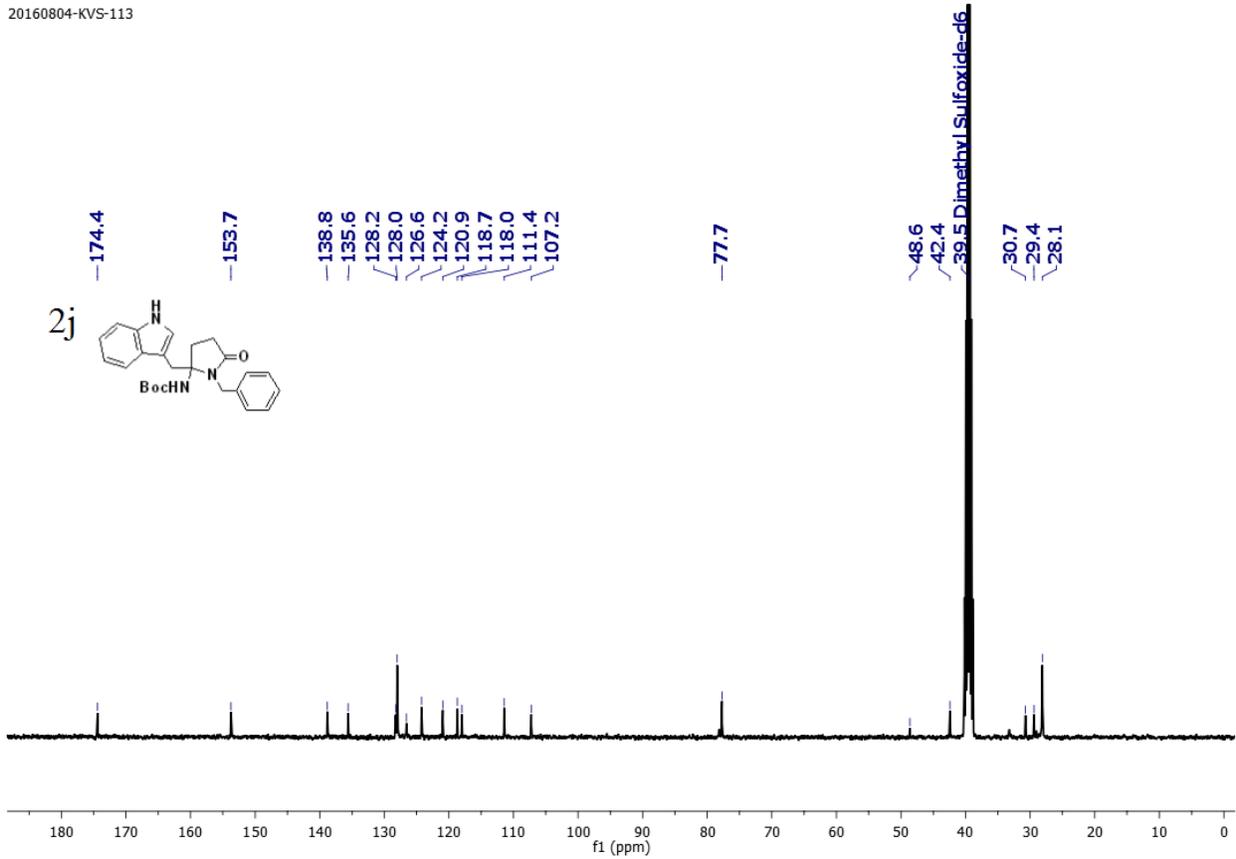
141114-20-KVS-23



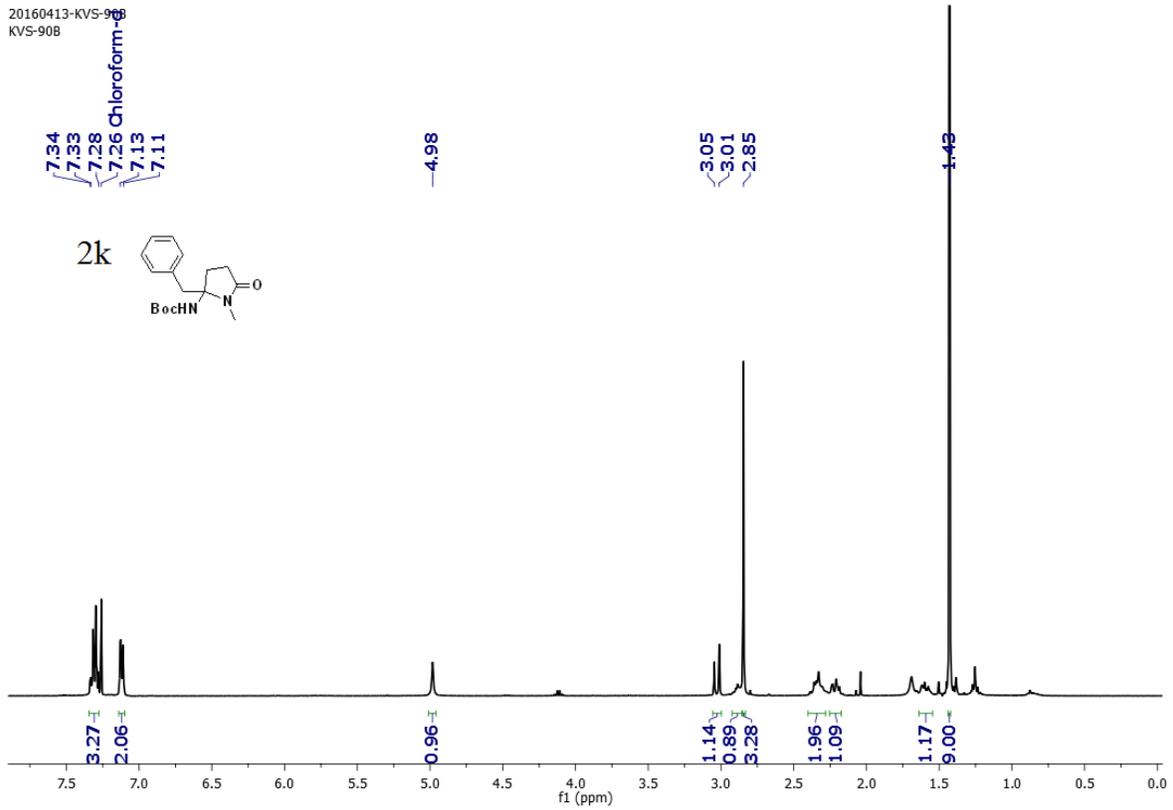
20160203-KVS-78C13
KVS-78C13

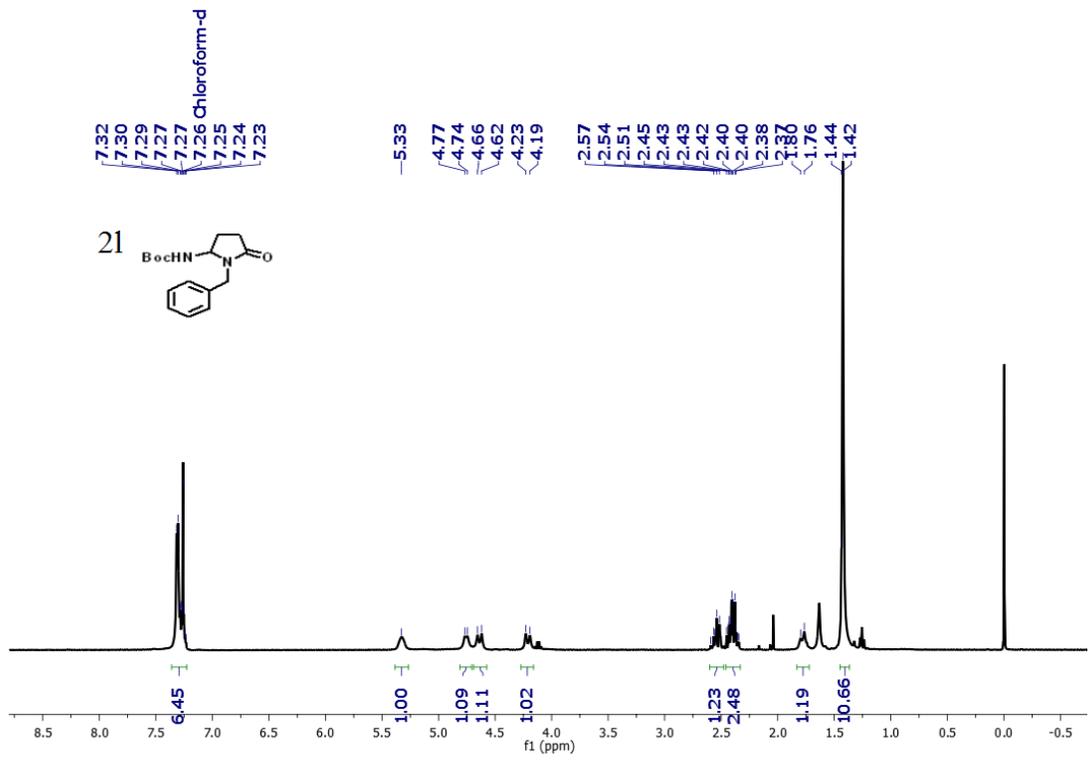
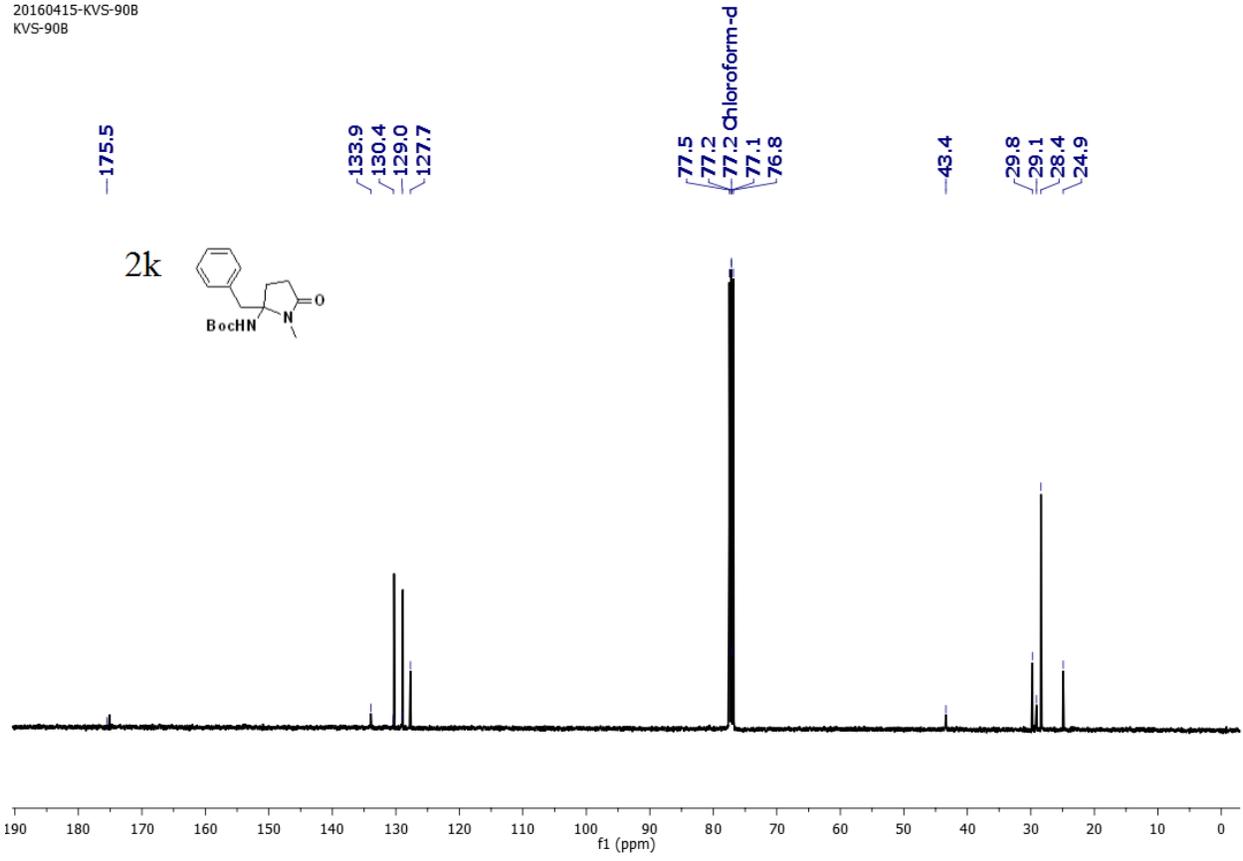


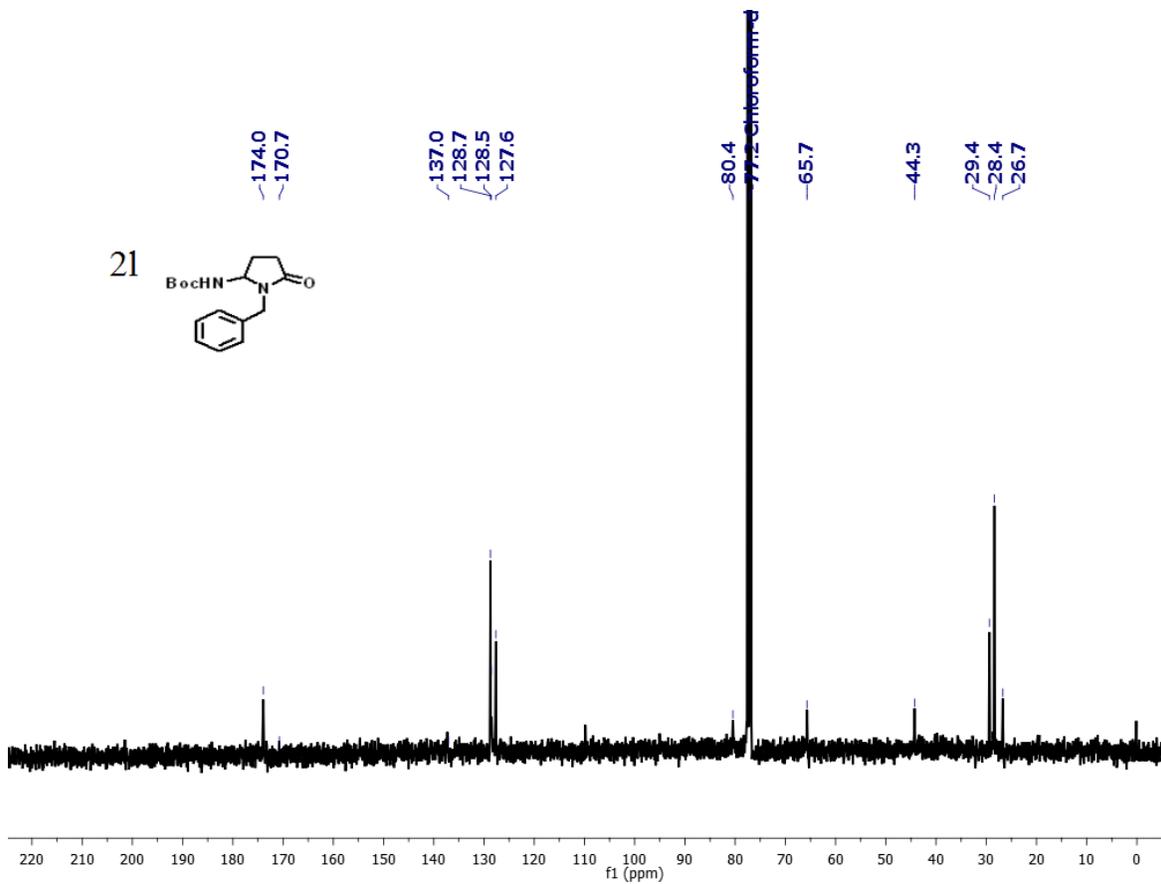




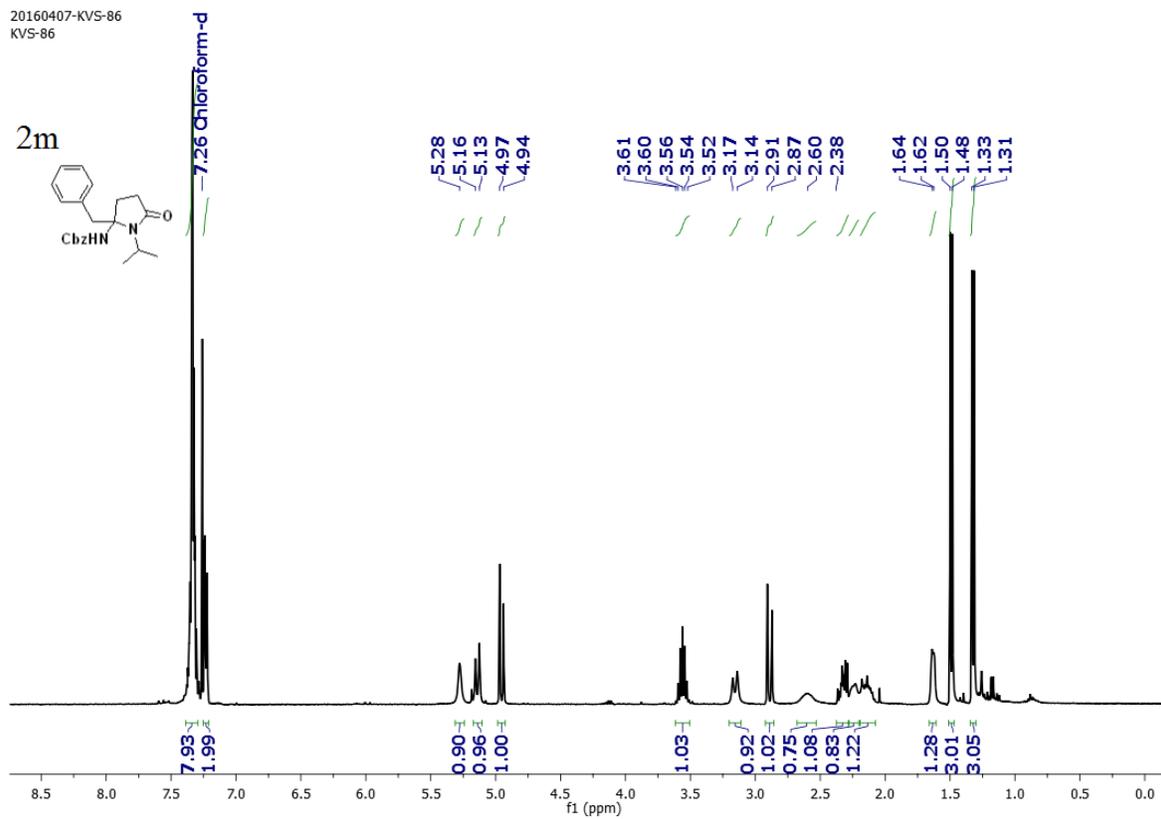
20160413-KVS-98
KVS-90B



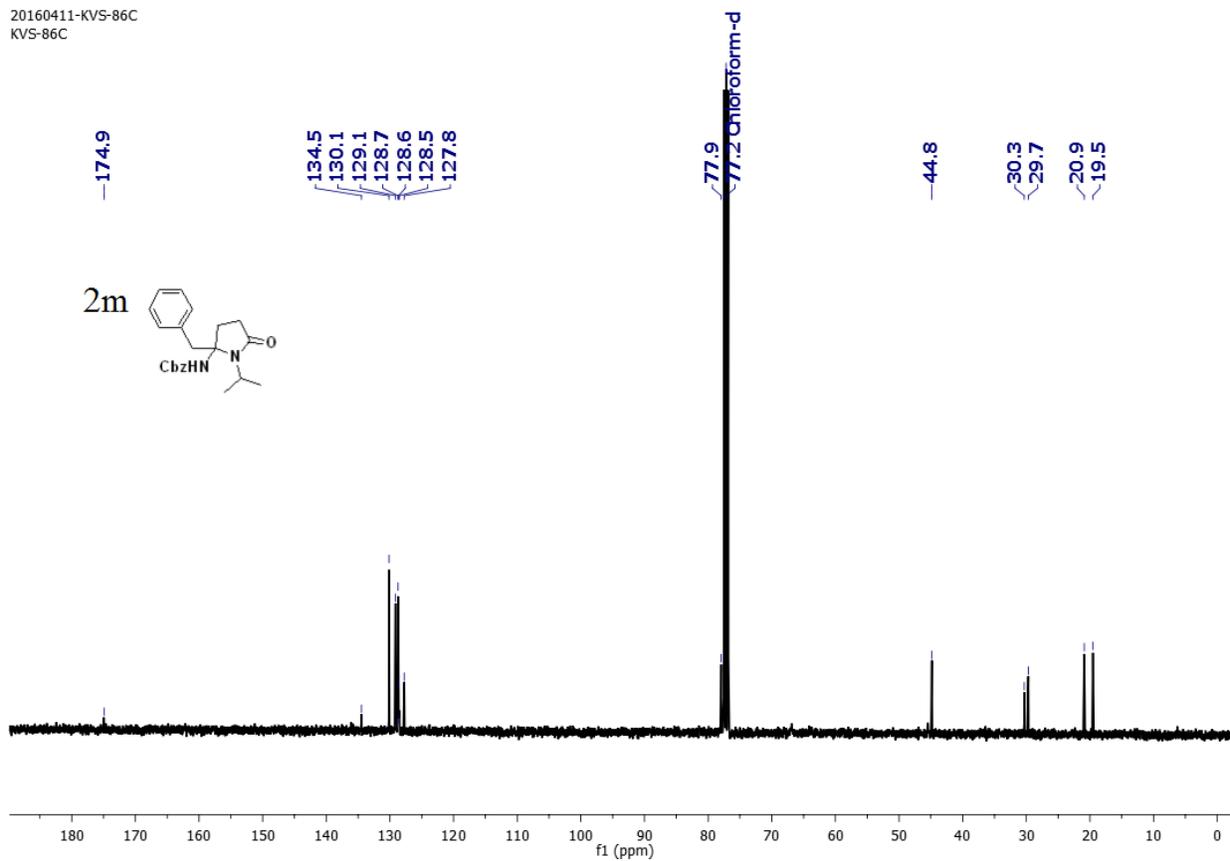




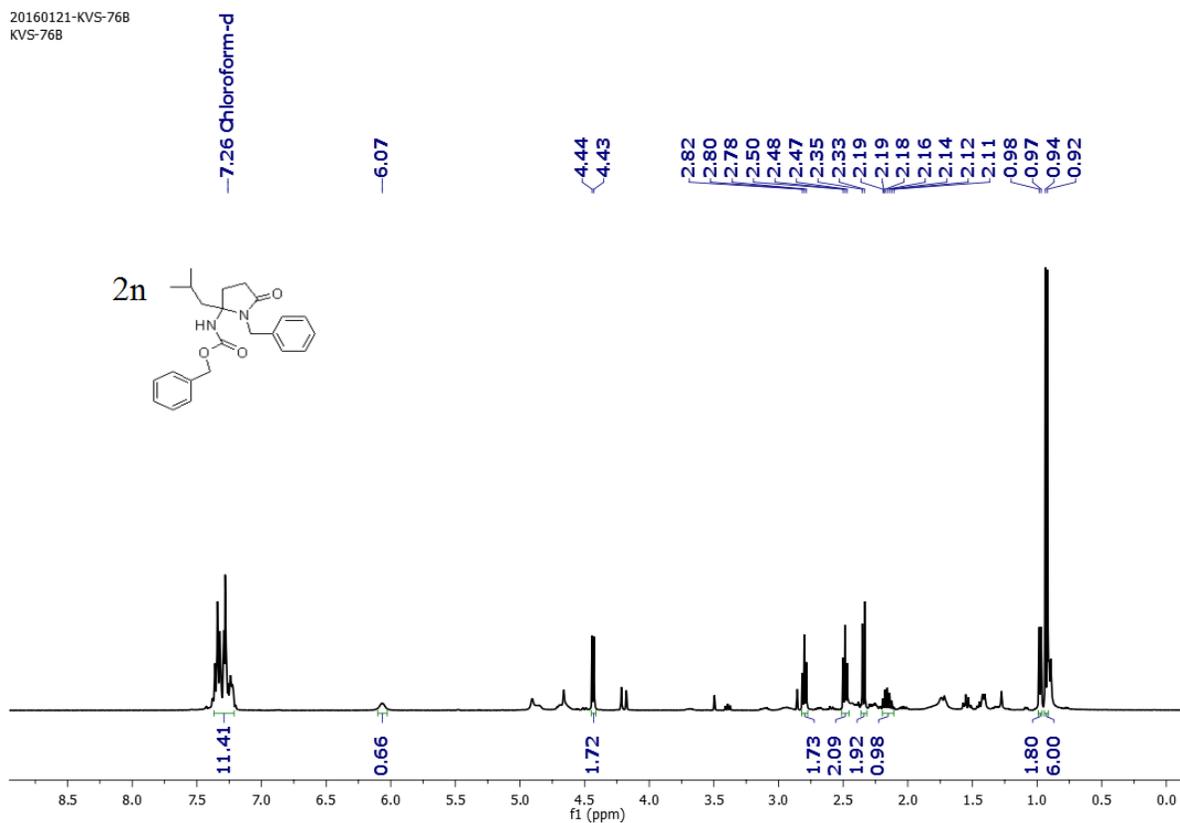
20160407-KVS-86
KVS-86



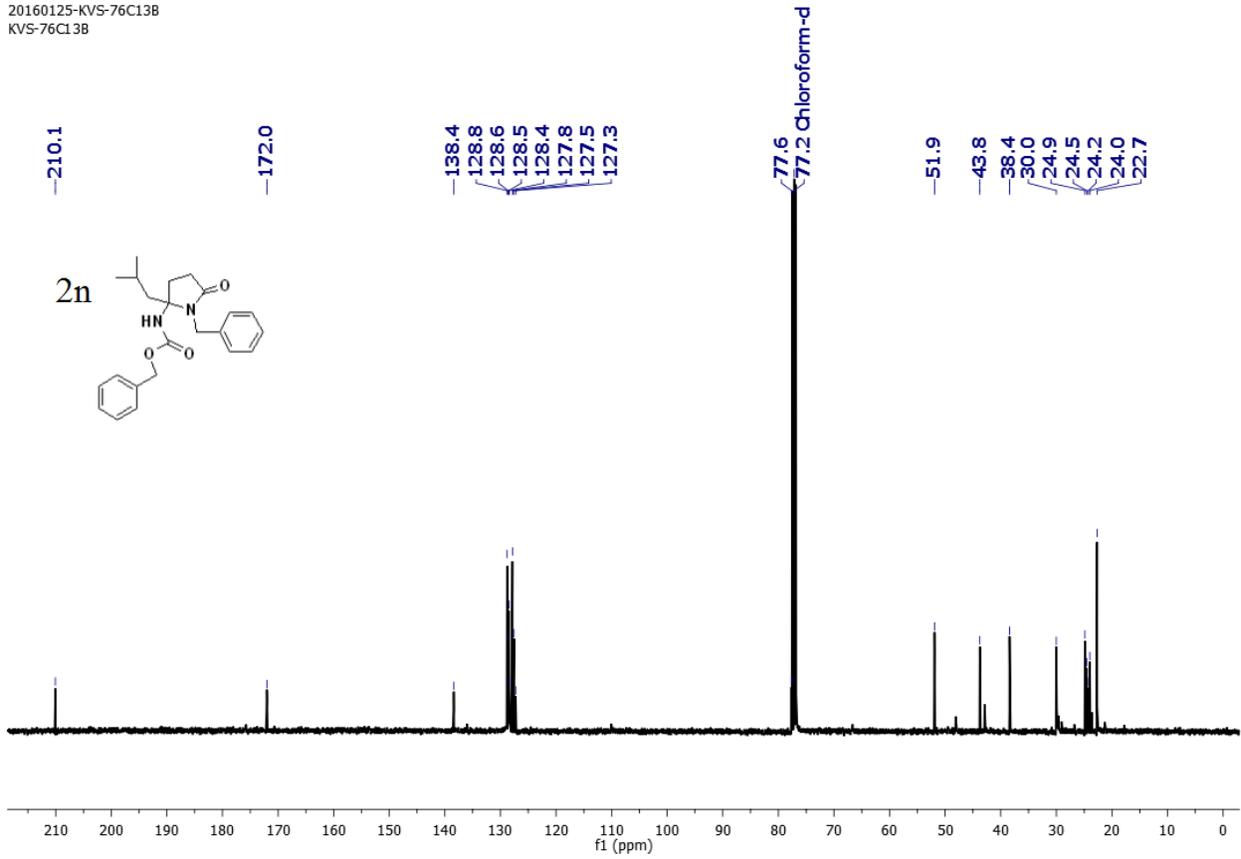
20160411-KVS-86C
KVS-86C



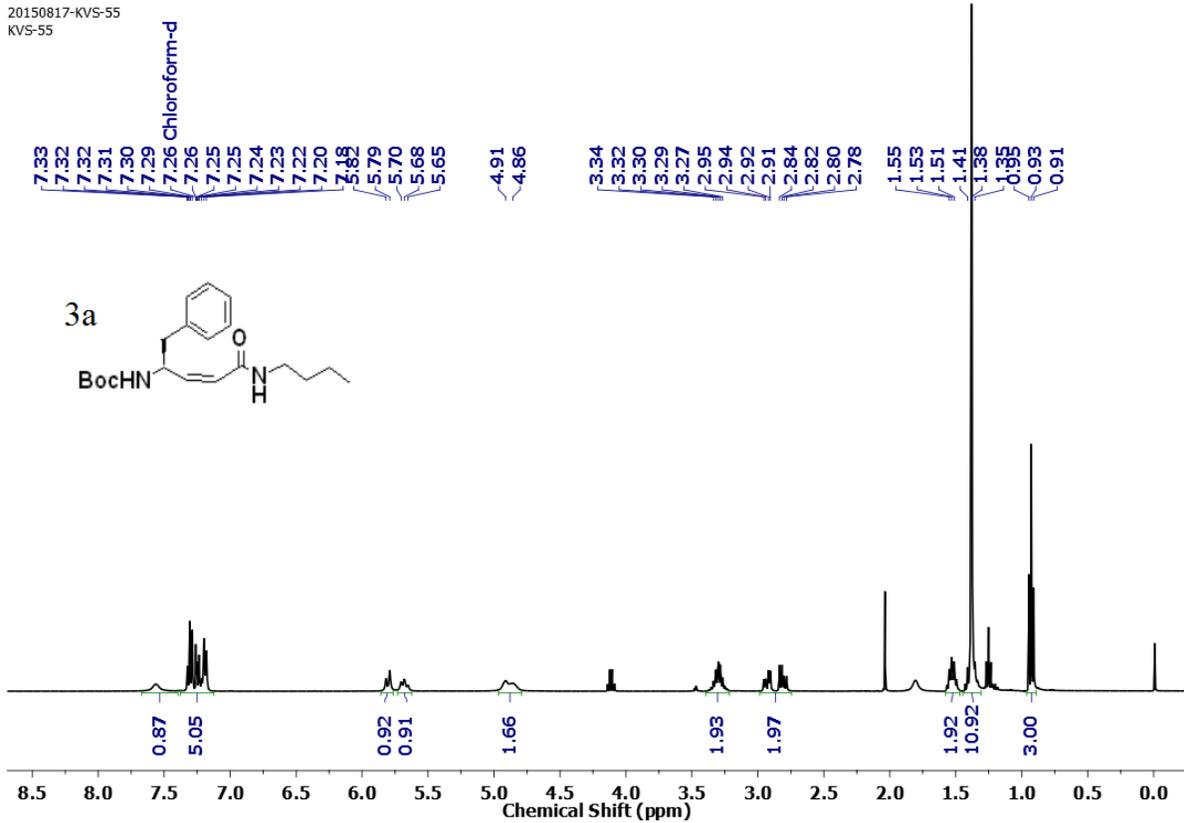
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KVS-76B

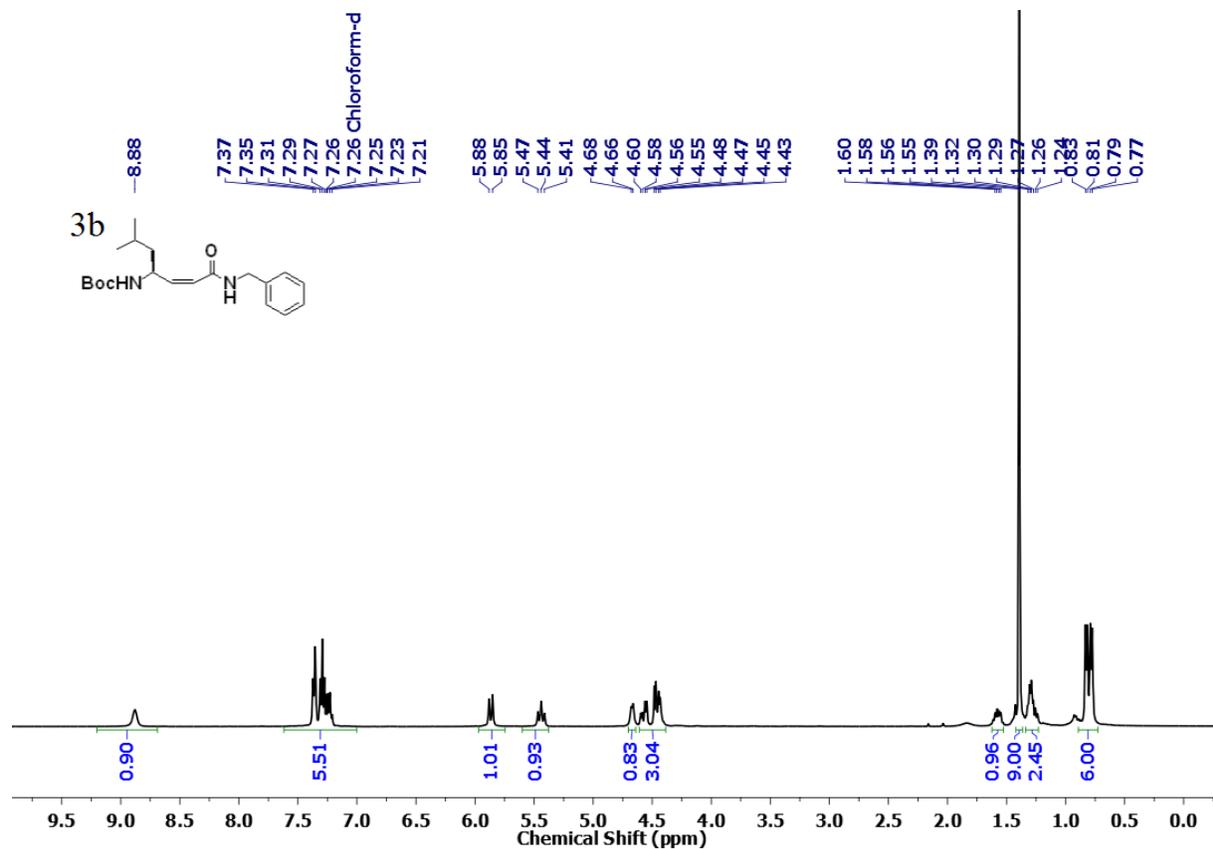
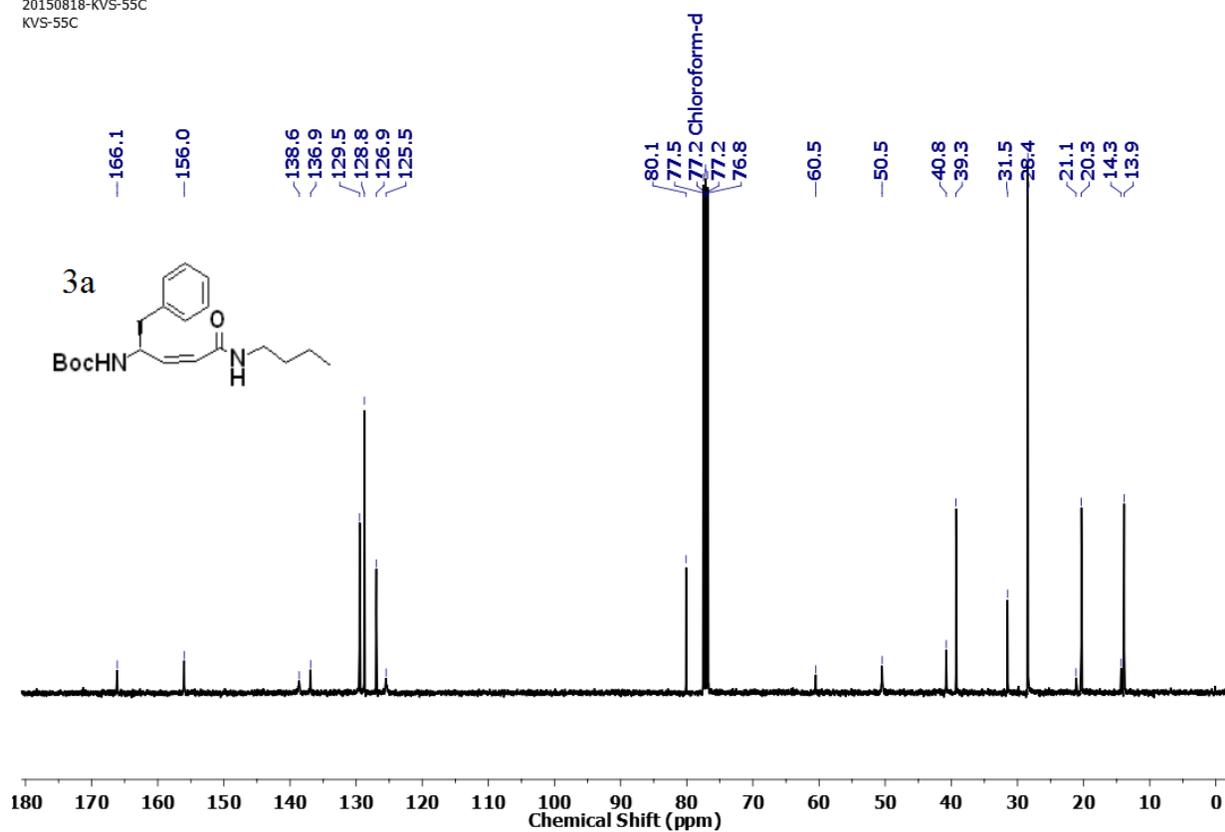


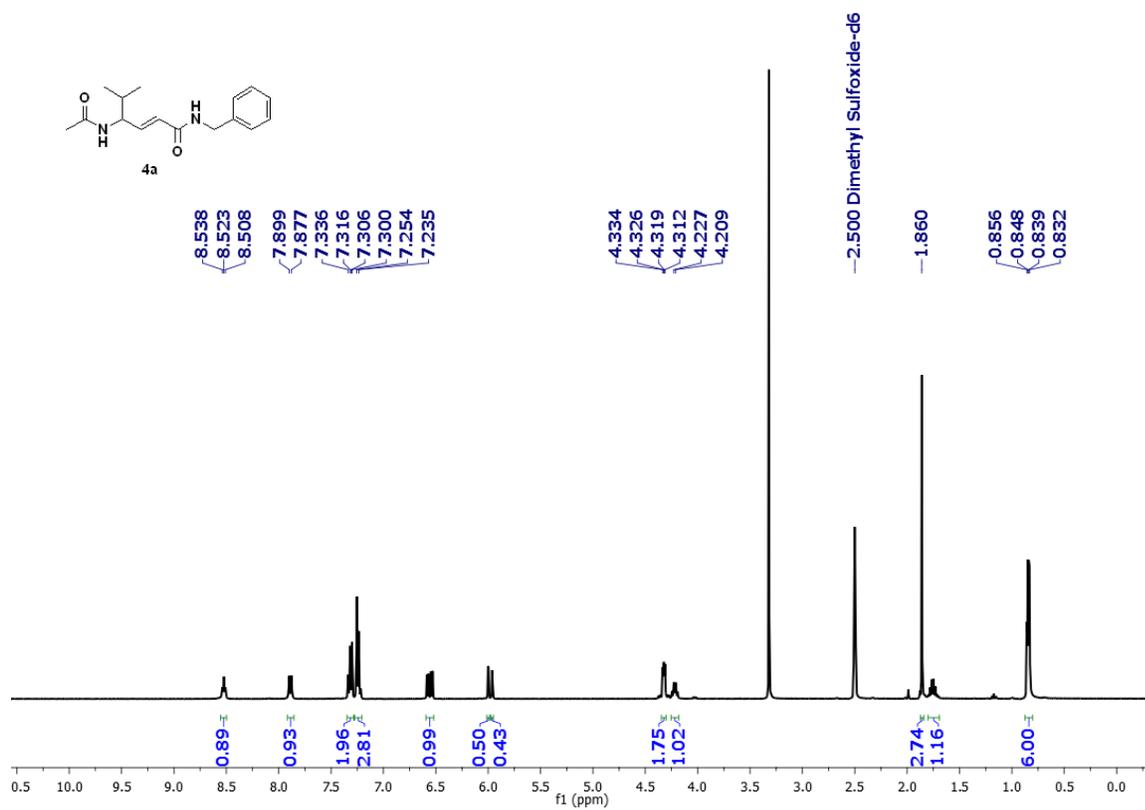
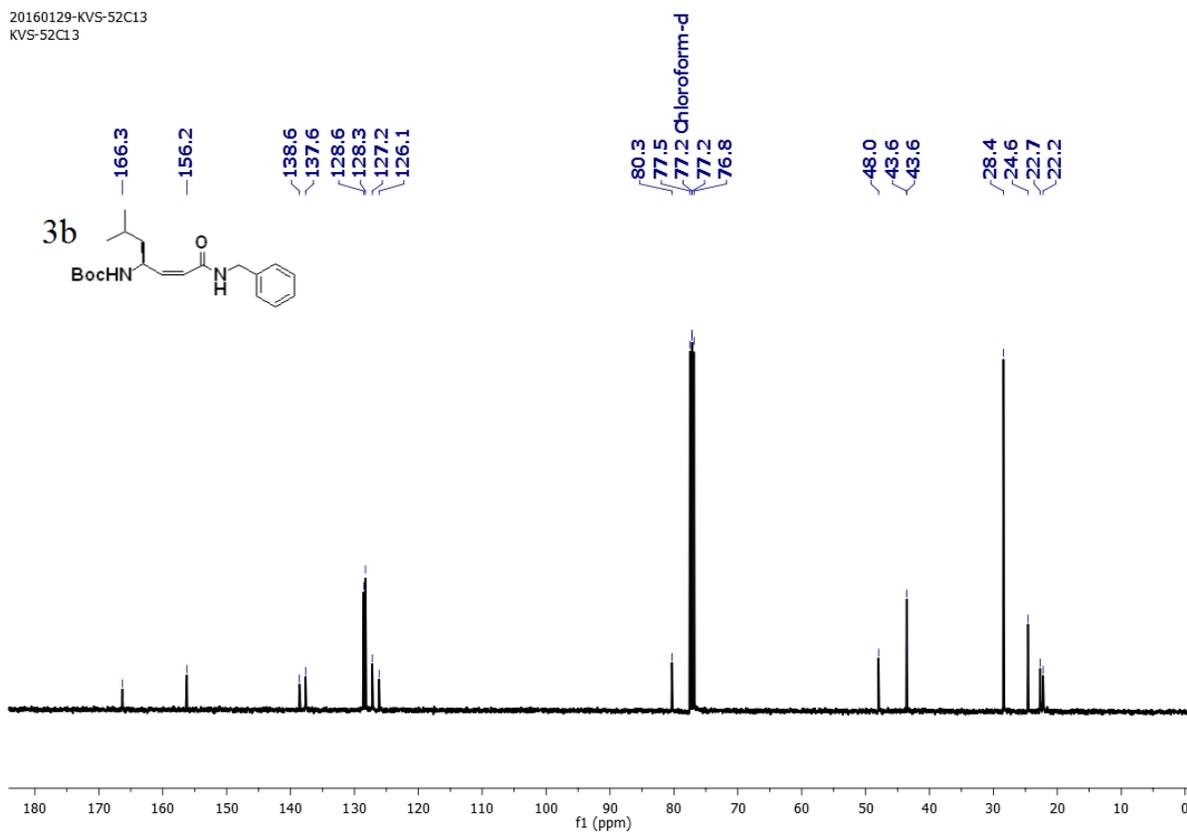
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KVS-76C13B

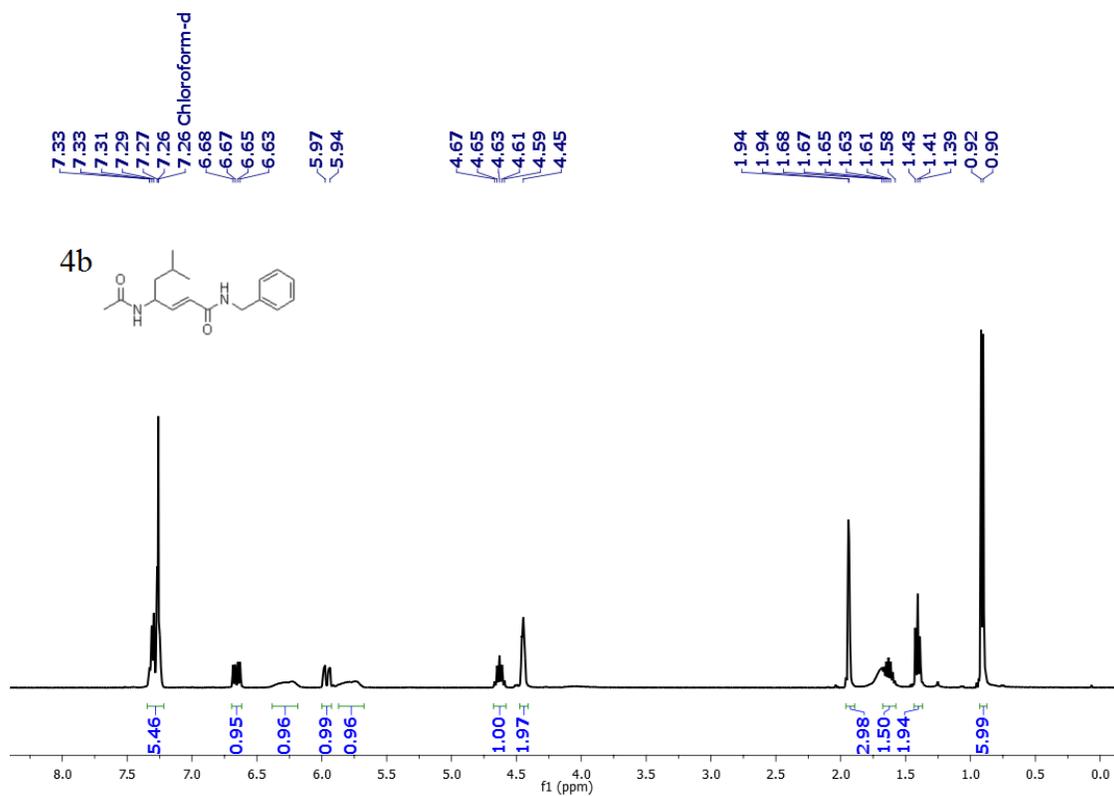
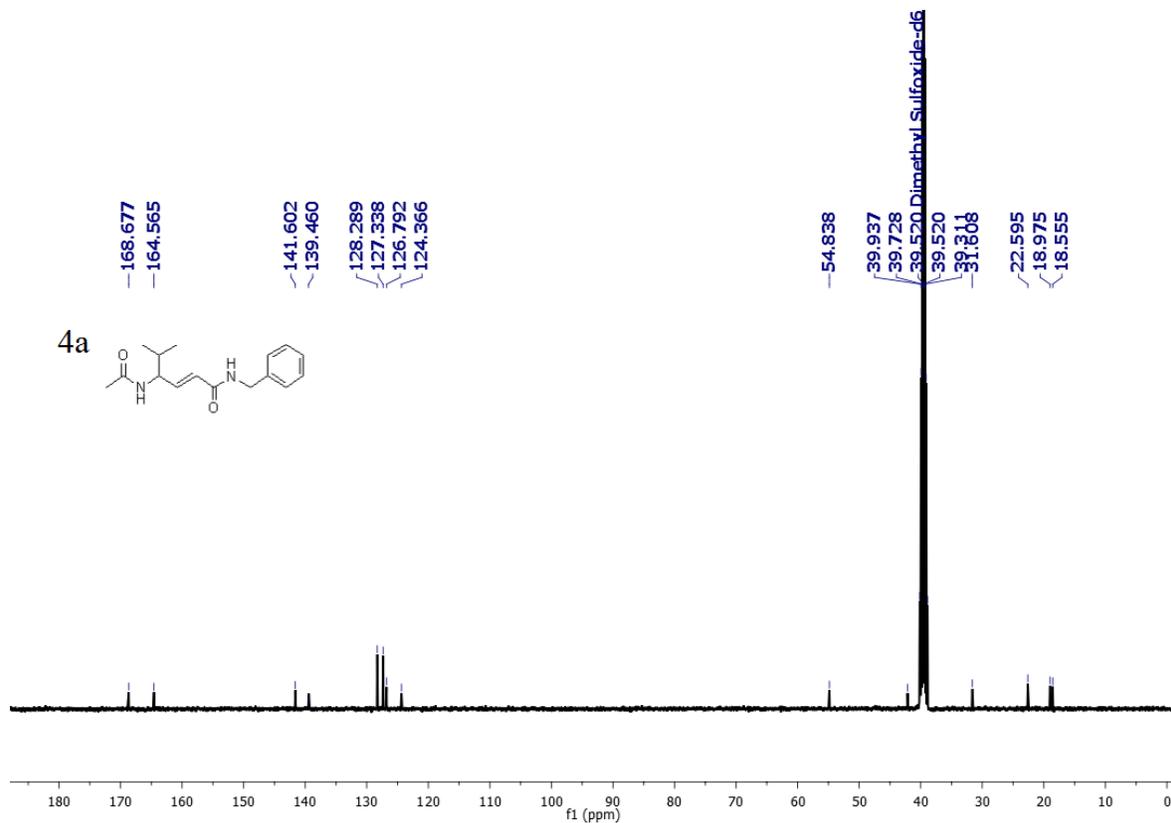


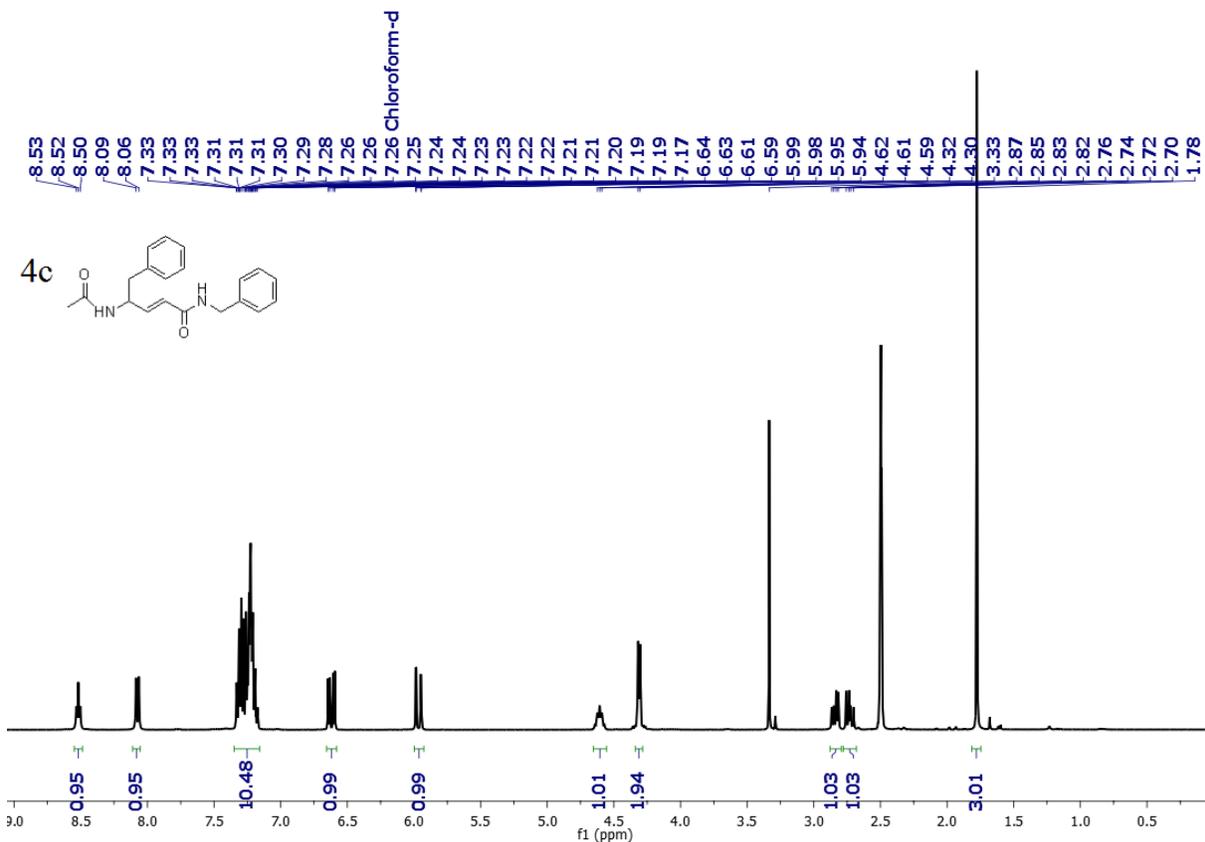
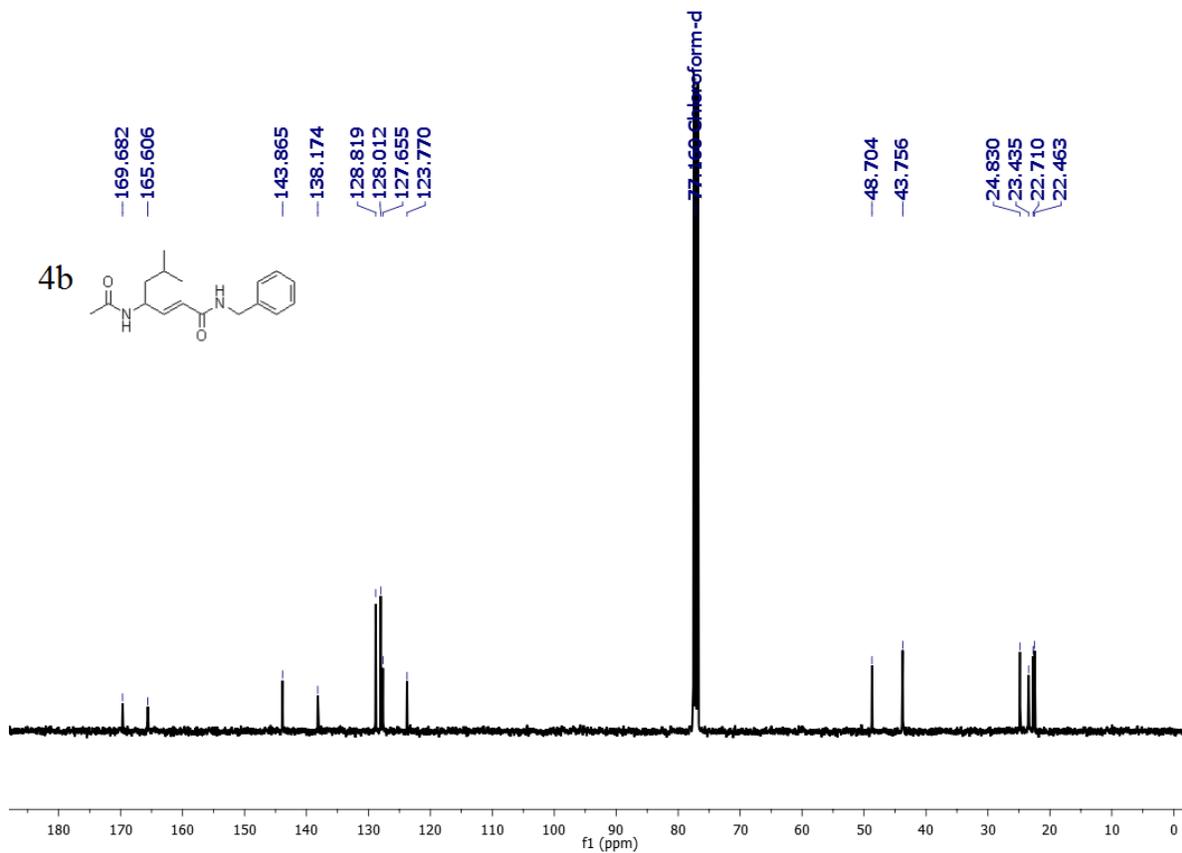
20150817-KVS-55
KVS-55

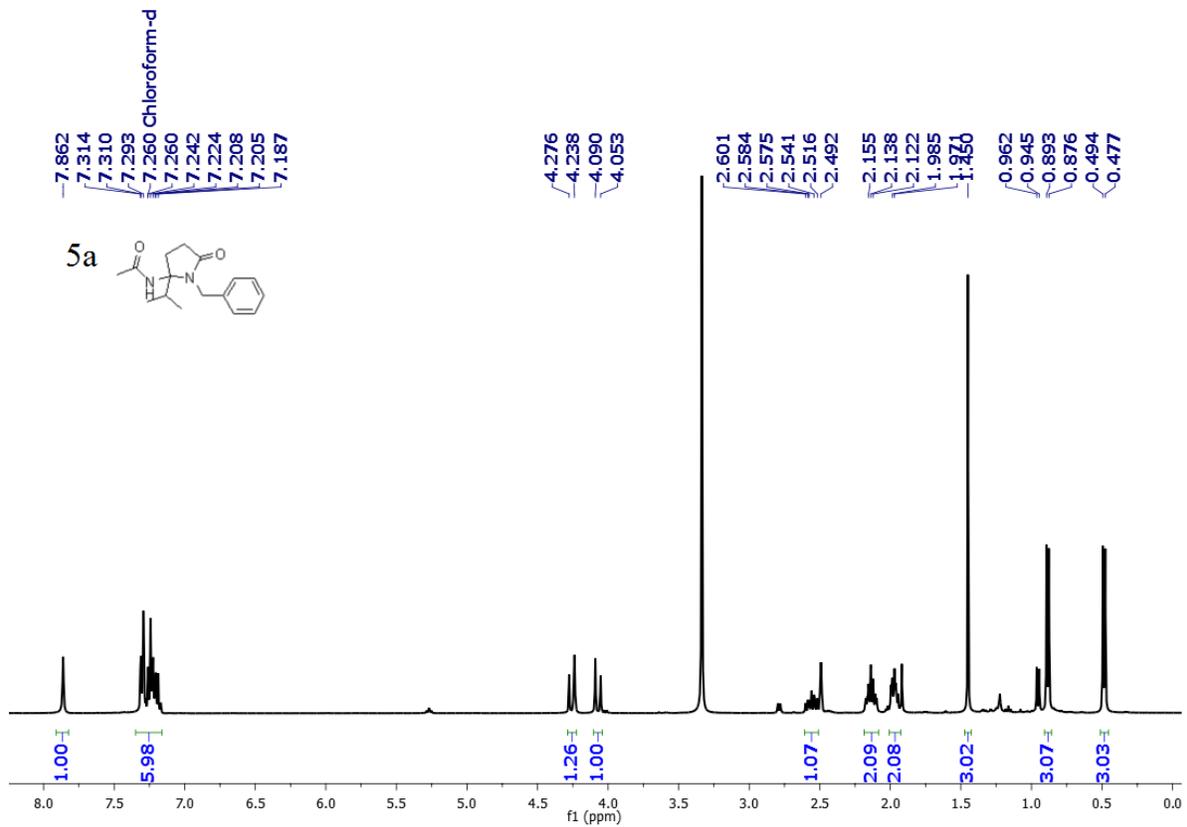
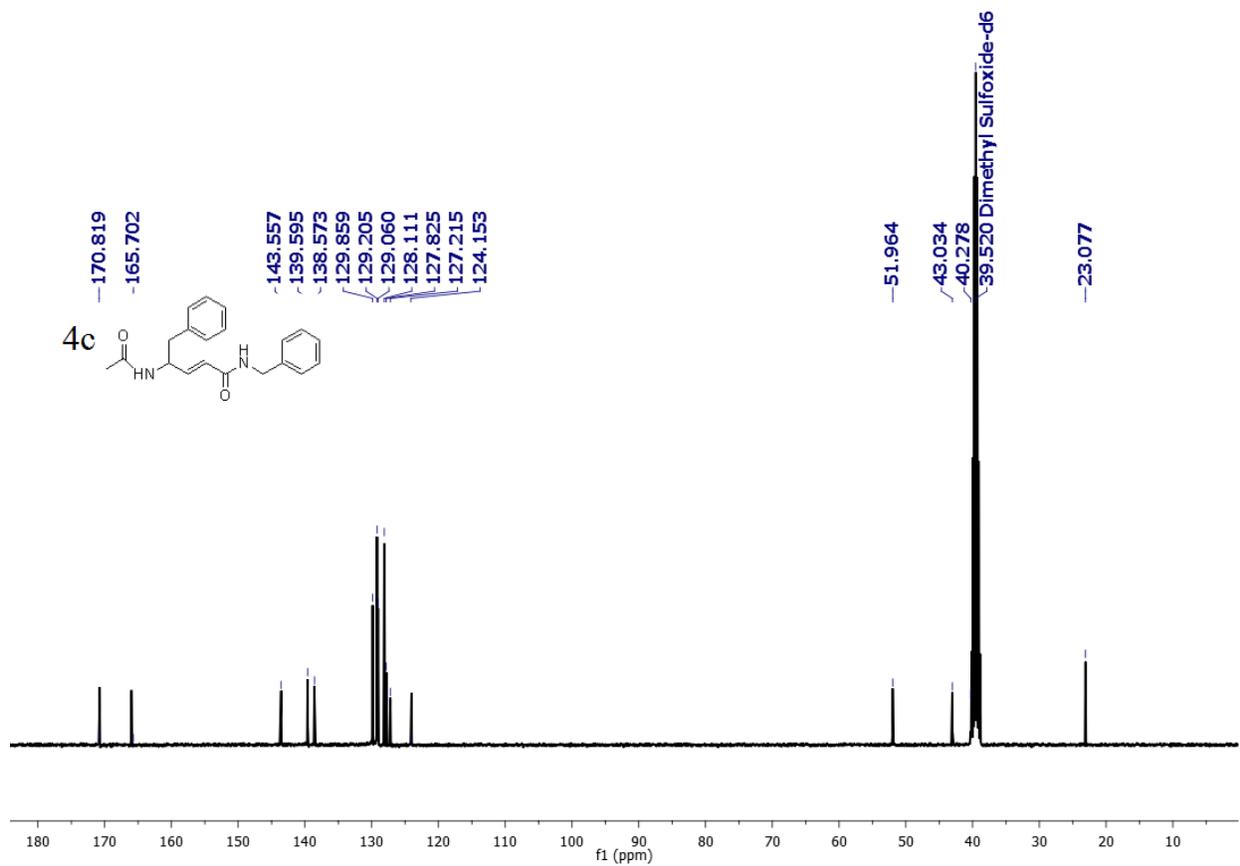


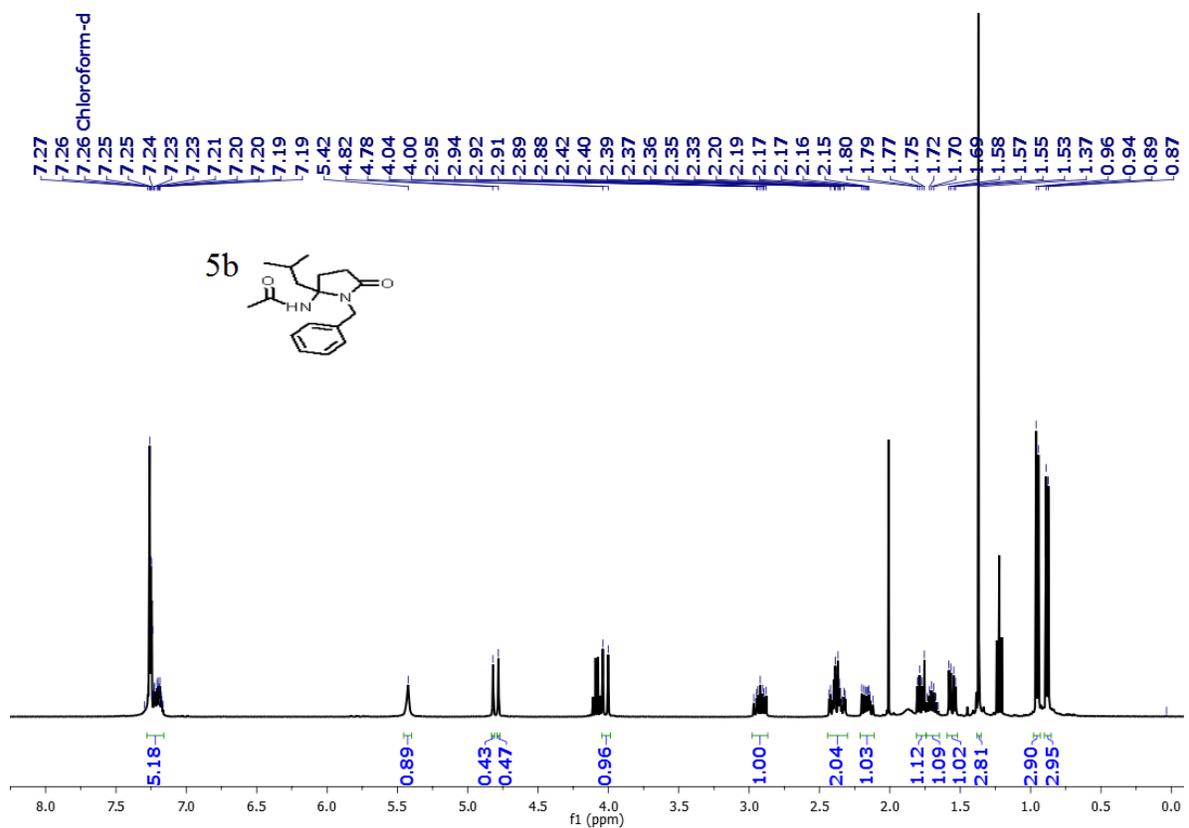
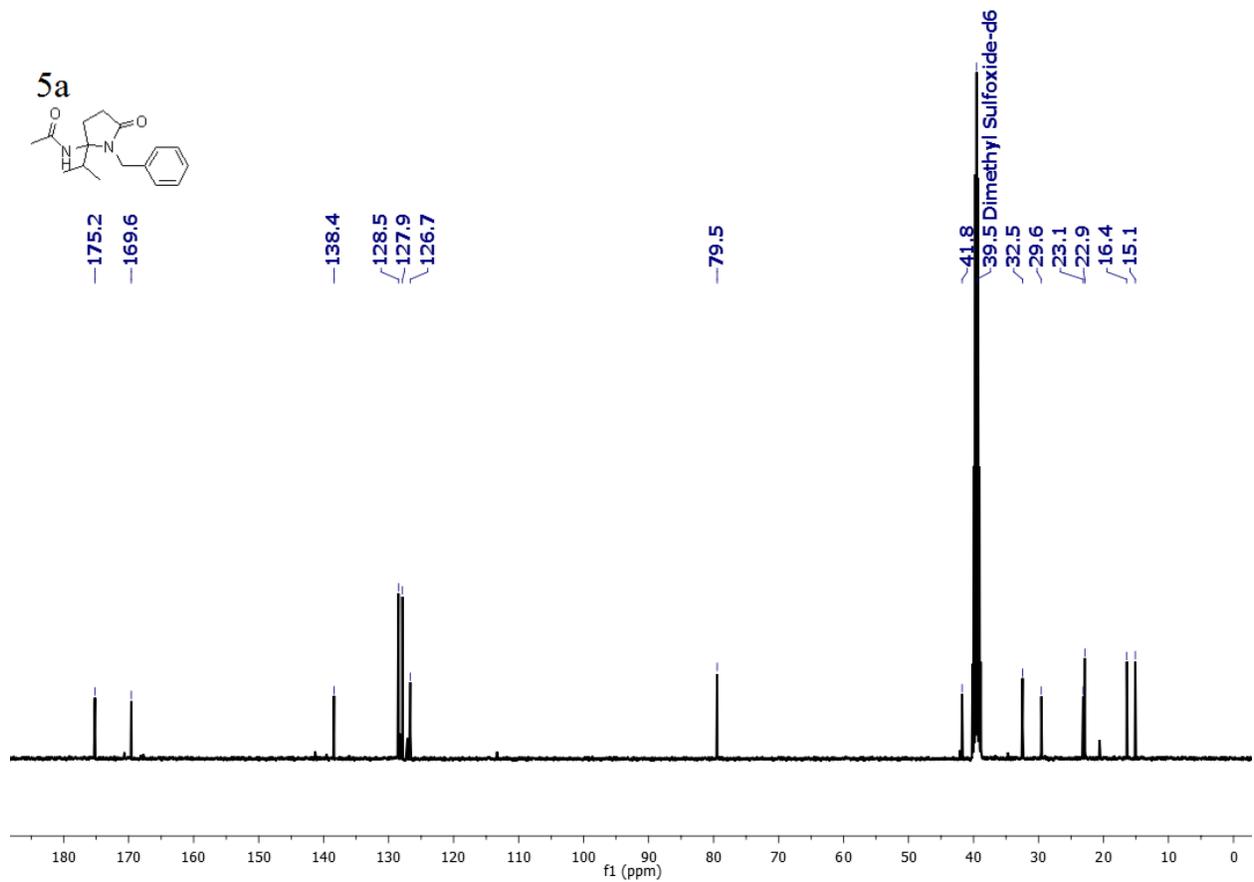


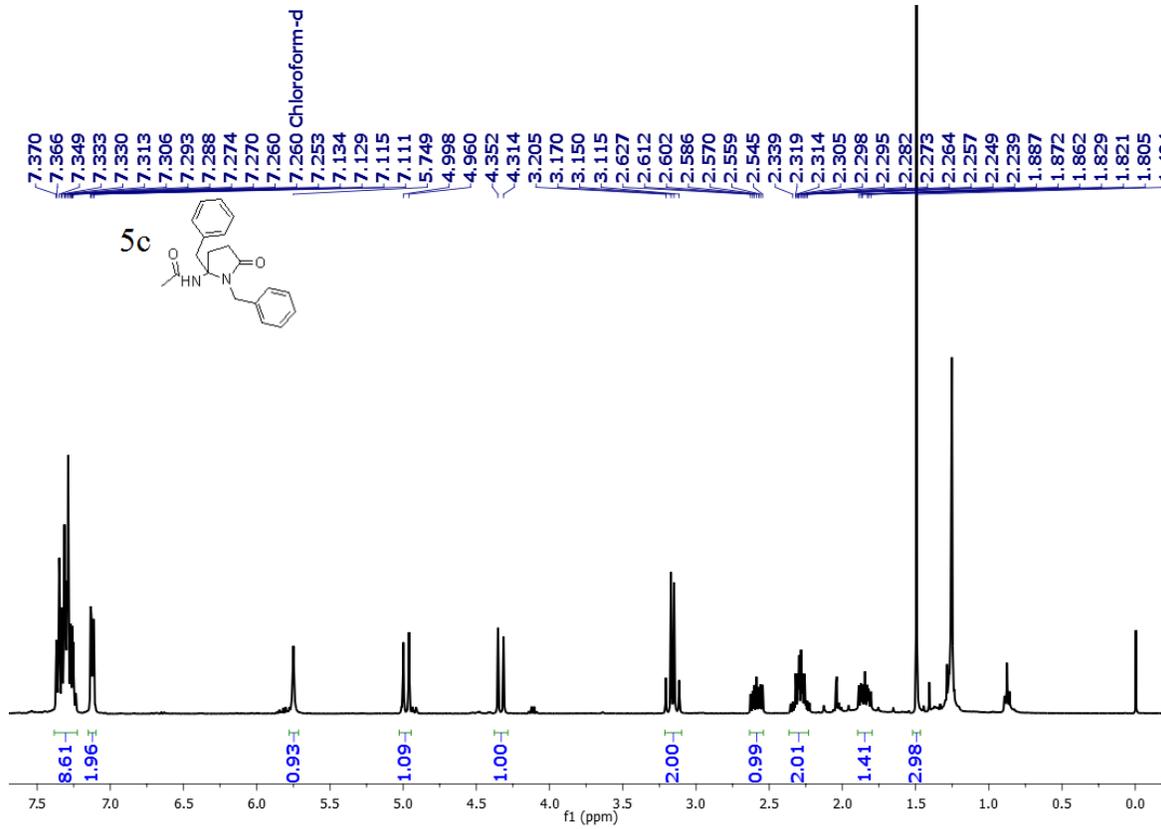
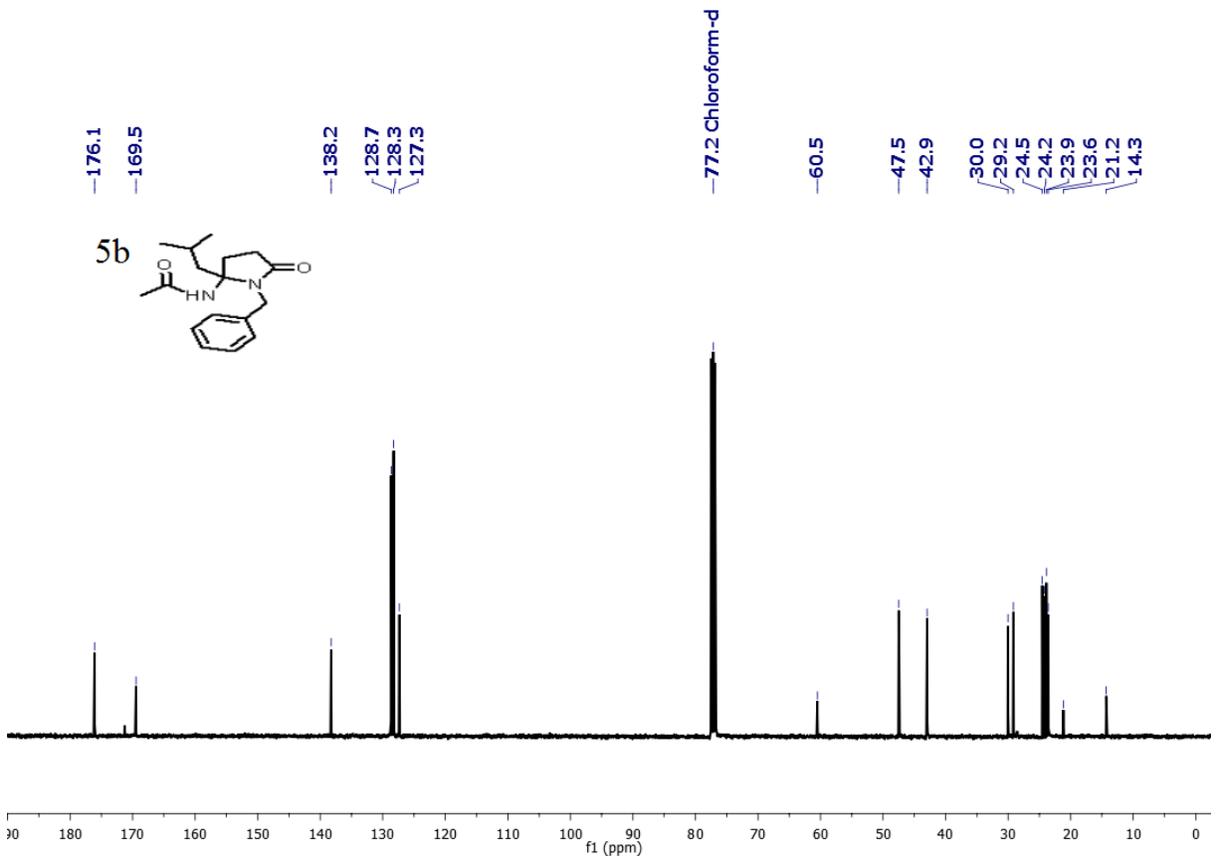


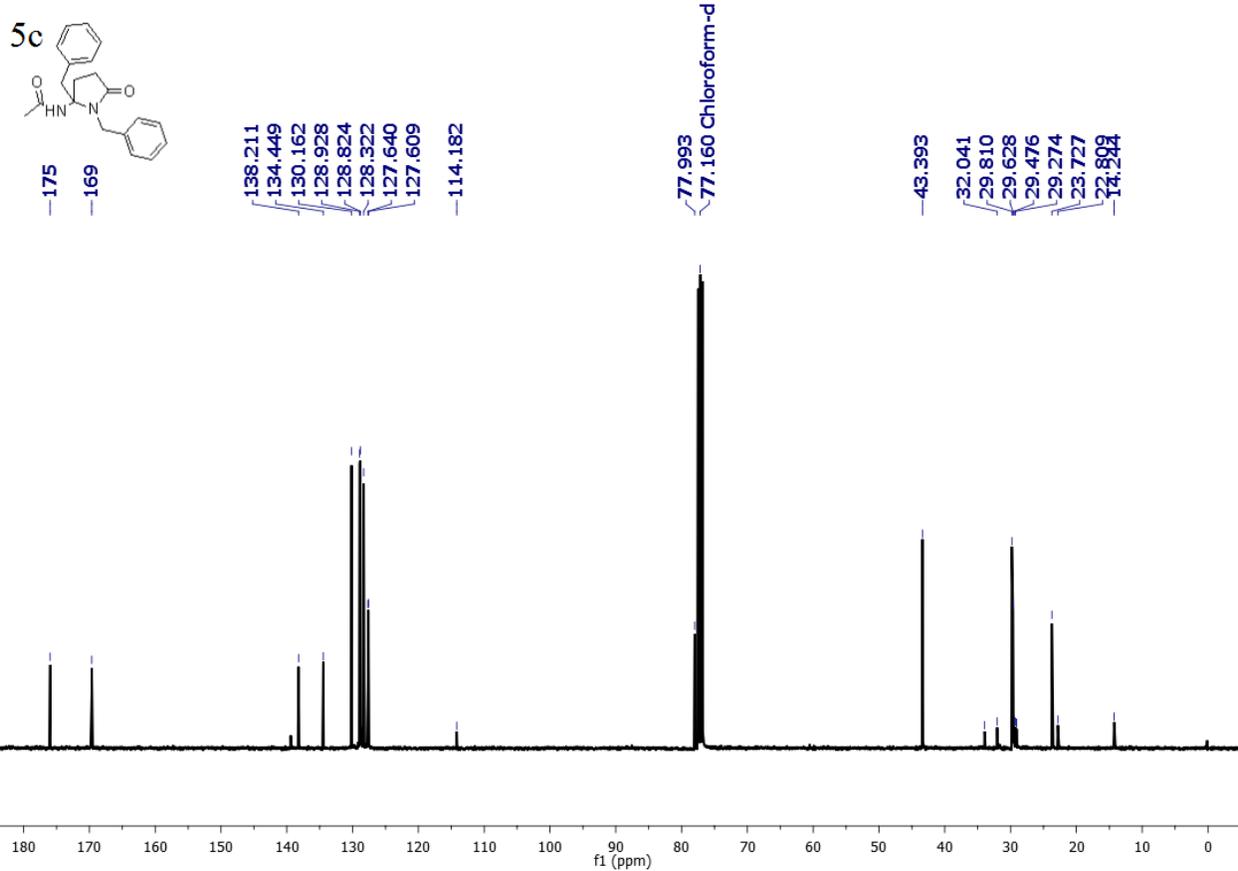




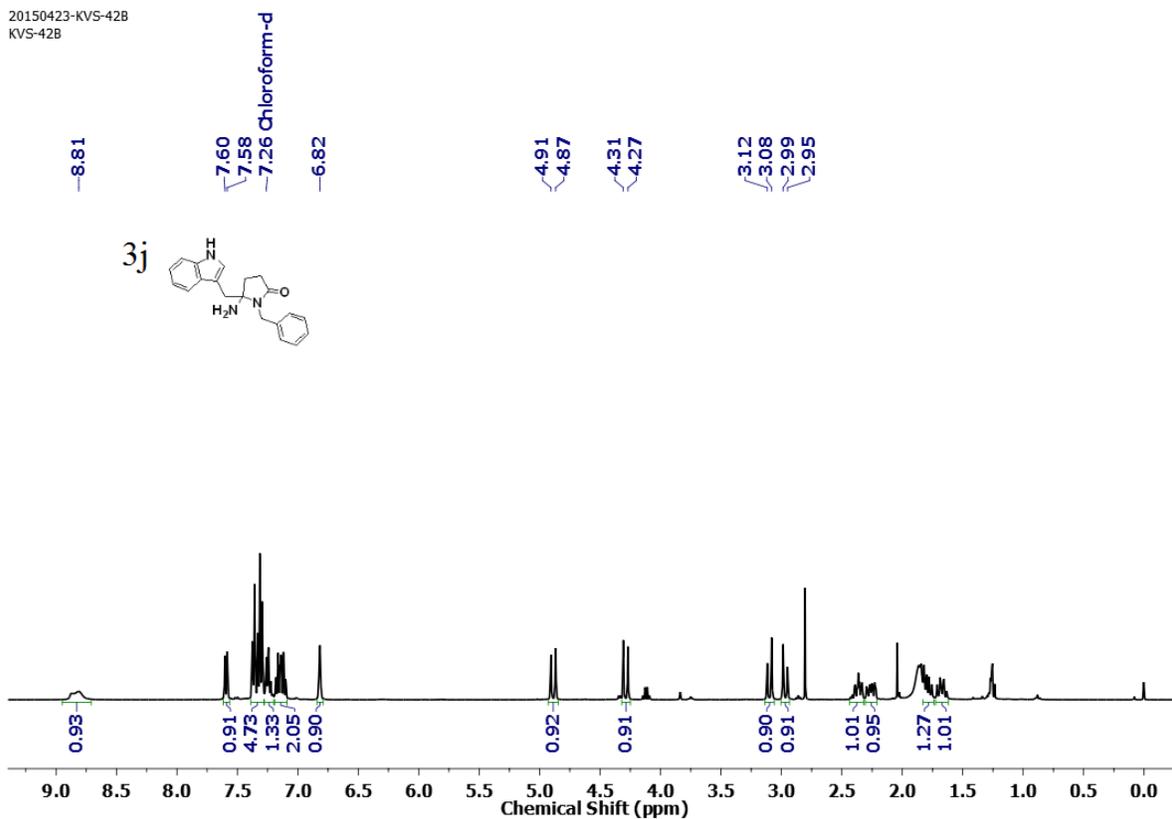




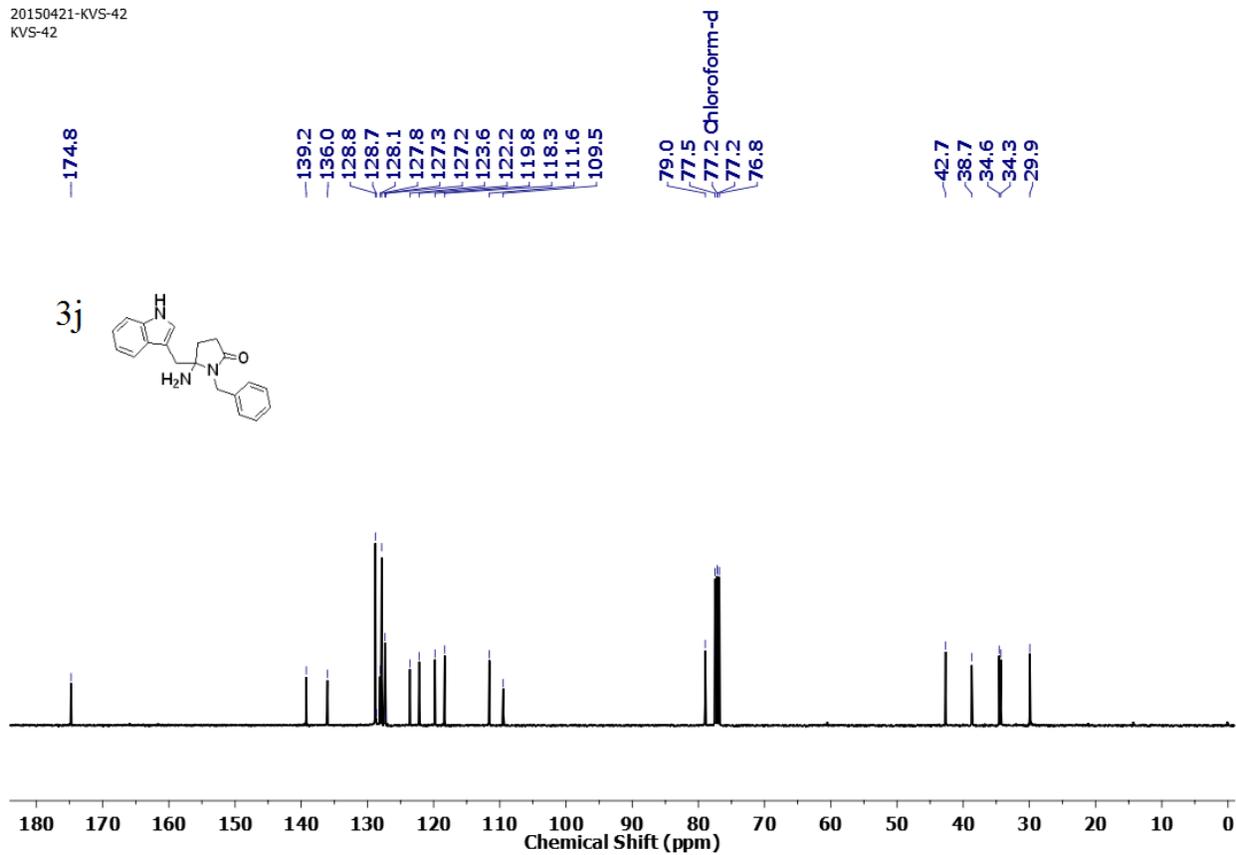




20150423-KVS-42B
KVS-42B



20150421-KVS-42
KVS-42



20150516-KVS-45
KVS-45

