

Supporting Information

Probing the role of Anions in Influencing the Structure, Stability and Properties in Neutral N-donor Linker based Metal-Organic Frameworks

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Figures:

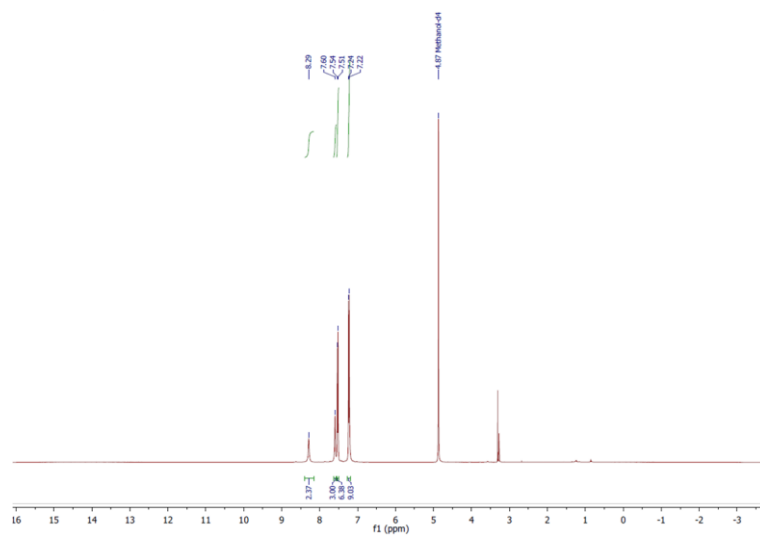


Figure S1: ¹H-NMR for ligand (L) recorded in CD₃OD.

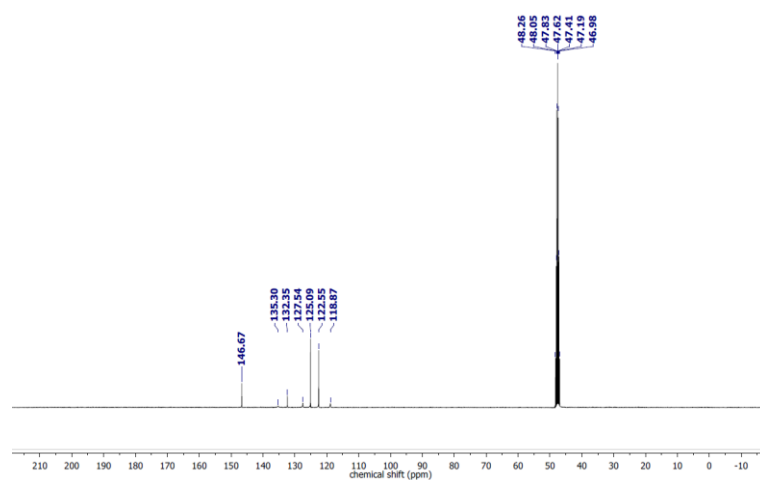


Figure S2: ¹³C-NMR for ligand (L) recorded in CD₃OD.

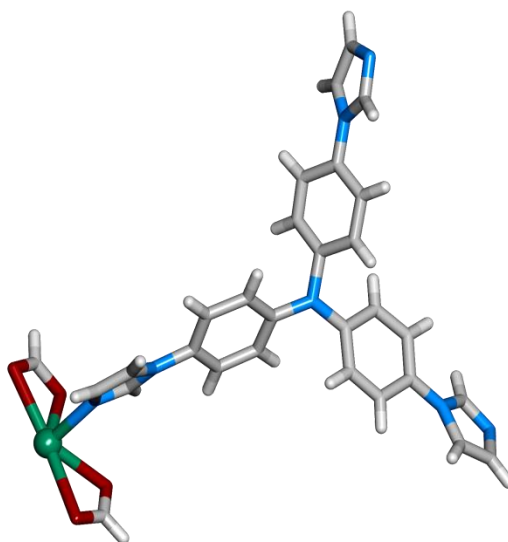


Figure S3: Asymmetric unit of **IPM-311**. (Colour code: C, grey; N, blue; Cd, green; O, dark red; H, white)

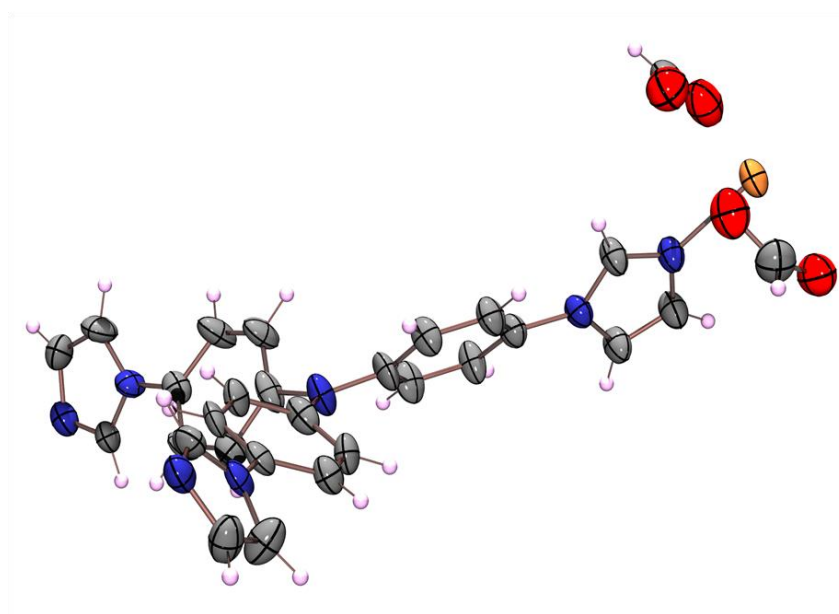


Figure S4: ORTEP diagram of **IPM-311** in thermal ellipsoids 50% probability. (Colour code: C, grey; N, blue; Cd, orange; O, dark red; H, pink)

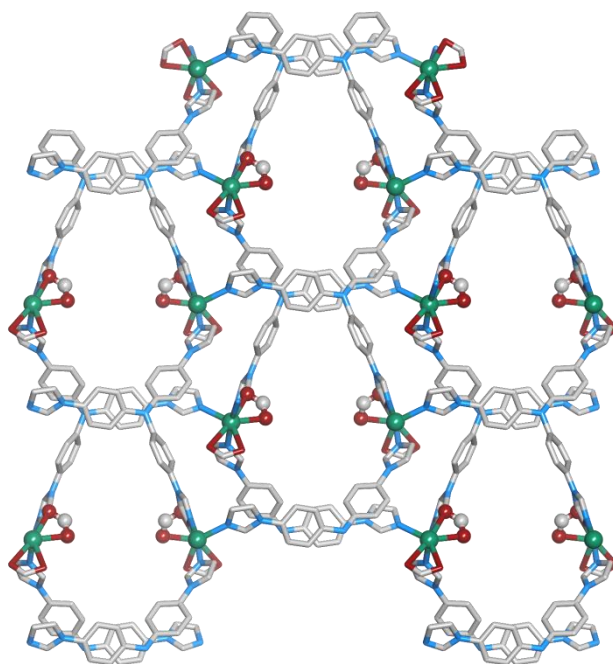


Figure S5: Figure showing packing of **IPM-311** along crystallographic *c*-axis, (Hydrogen atoms have been omitted for clarity. Colour code: C, grey; N, blue; Cd, green; O, dark red)

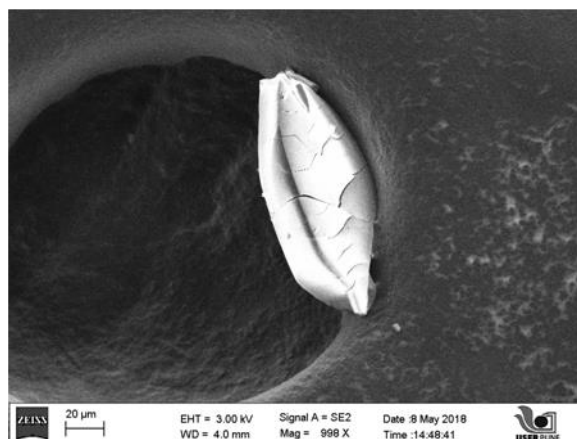
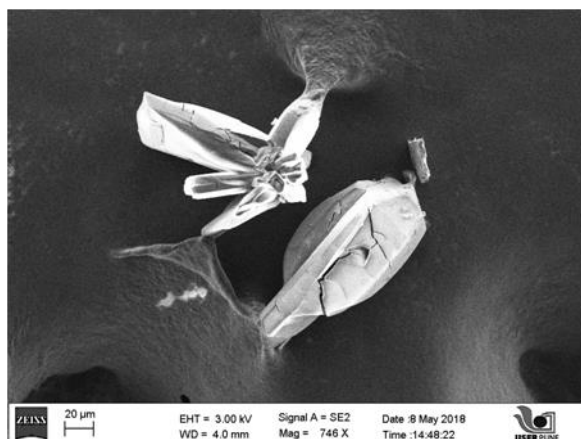


Figure S6: FESEM images for compound **IPM-311**. The images were recorded for the solid sample on carbon tape.

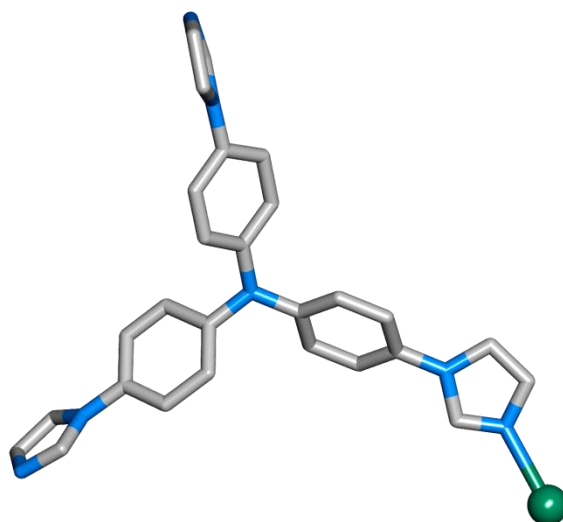


Figure S7: Asymmetric unit of **IPM-206**. (Colour code: C, grey; N, blue; Cd, green. H-atoms are omitted for clarity)

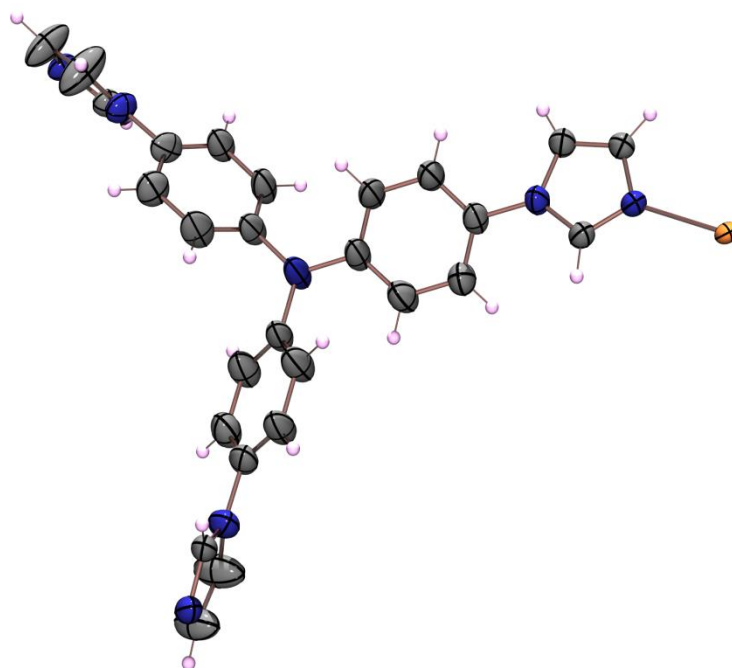


Figure S8: ORTEP diagram of **IPM-206** in thermal ellipsoids 50% probability. (Colour code: C, grey; N, blue; Cd, orange; H, pink)

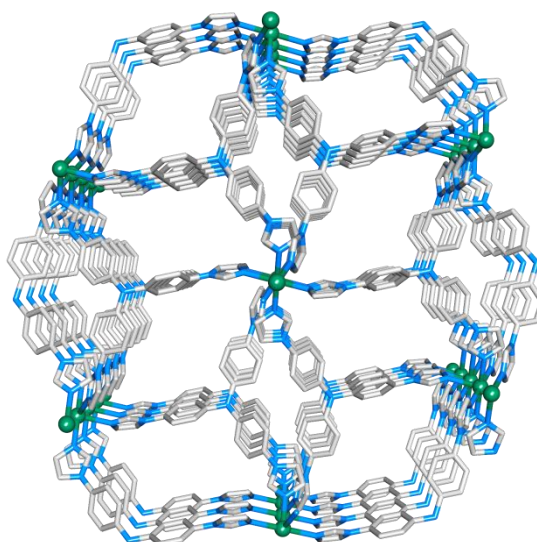


Figure S9: Figure showing packing of **IPM-206** along crystallographic *b*-axis (Hydrogen atoms have been omitted for clarity. Colour code: C, grey; N, blue; Cd, green)

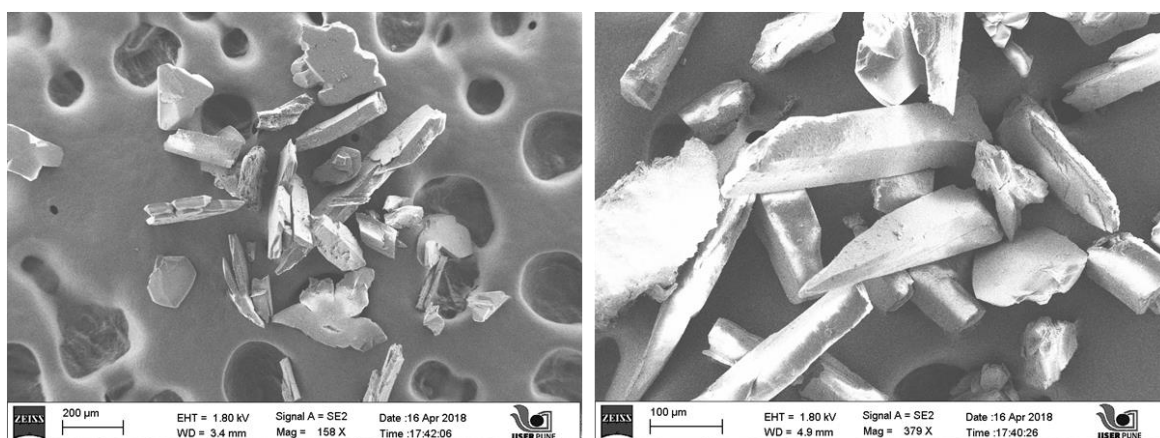


Figure S10: FESEM images for compound **IPM-206**. The images were recorded for the solid sample on carbon tape.

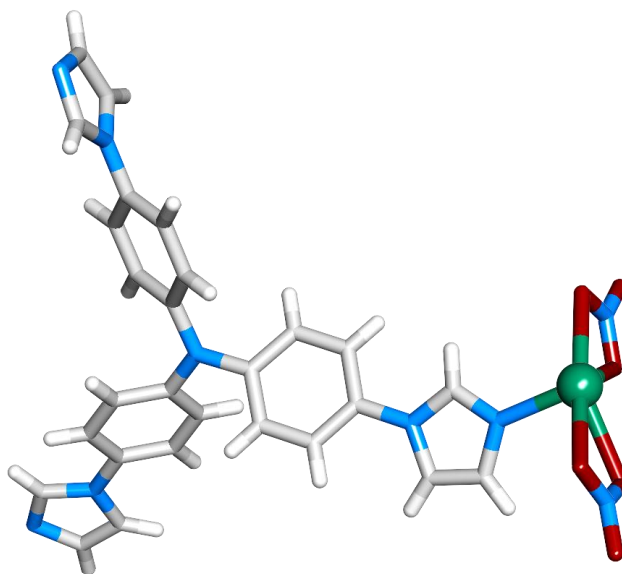


Figure S11: Asymmetric unit of **IPM-310**. (Colour code: C, grey; N, blue; Cd, green; O, dark red; H, white)

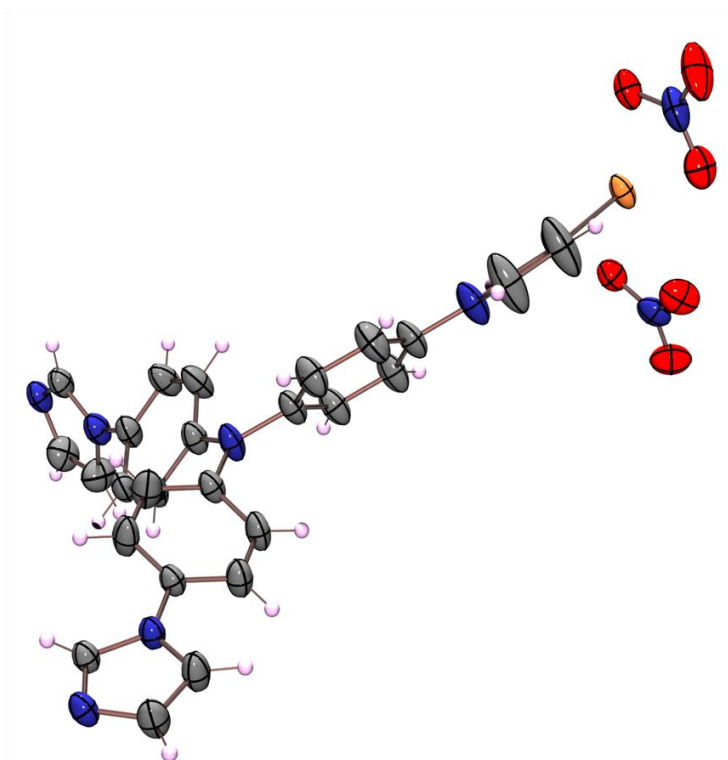


Figure S12: ORTEP diagram of **IPM-310** in thermal ellipsoids 50% probability. (Colour code: C, grey; N, blue; Cd, orange; O, red; H, pink)

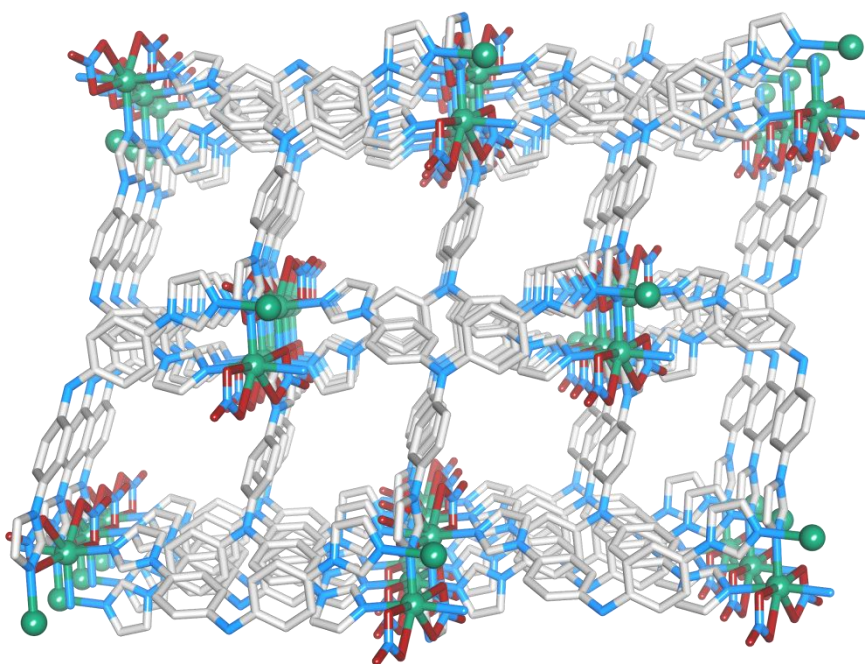


Figure S13: Figure showing packing of **IPM-310** along crystallographic *b*-axis. (Hydrogen atoms have been omitted for clarity. Colour code: C, grey; N, blue; Cd, green)

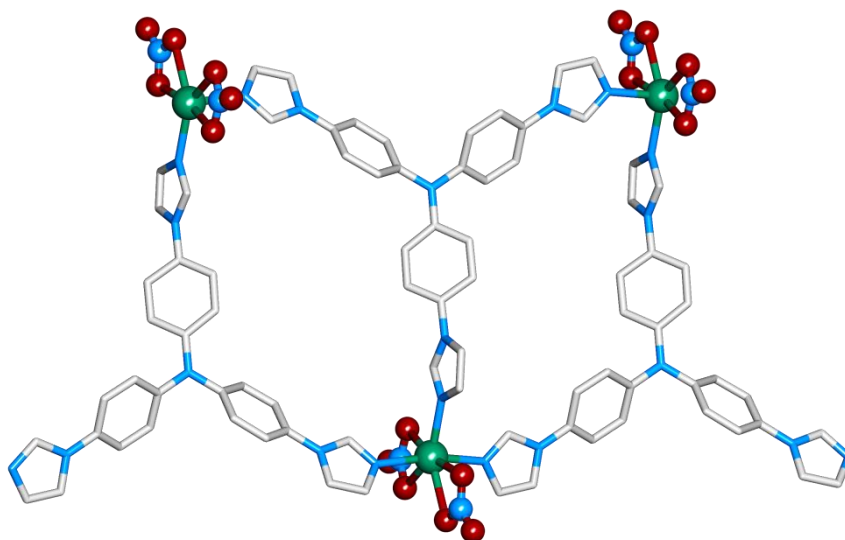


Figure S14: Figure showing one-dimensional packing of **IPM-310** showing coordinated nitrate anions. (Hydrogen atoms have been omitted for clarity. Colour code: C, grey; N, blue; Cd, green; O, red)

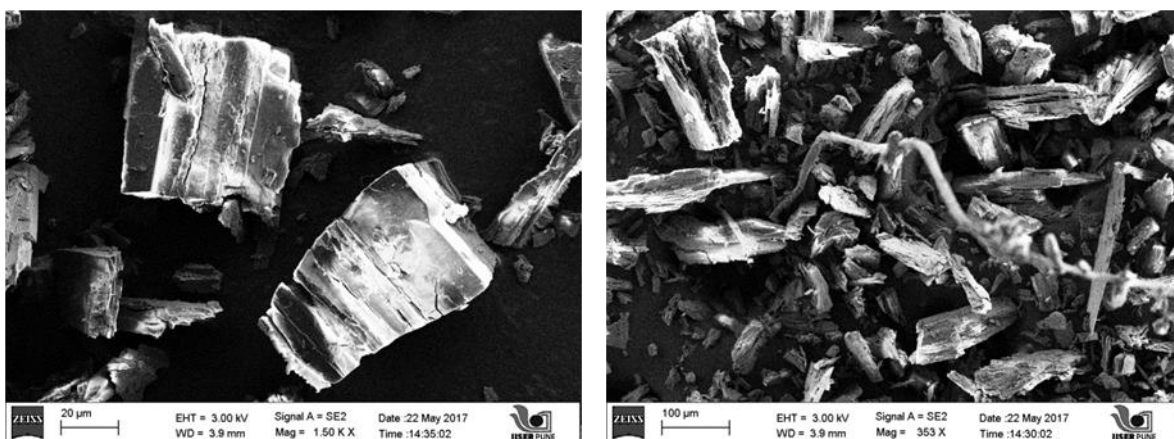


Figure S15: FESEM images for compound **IPM-310**. The images were recorded for the solid sample on carbon tape.

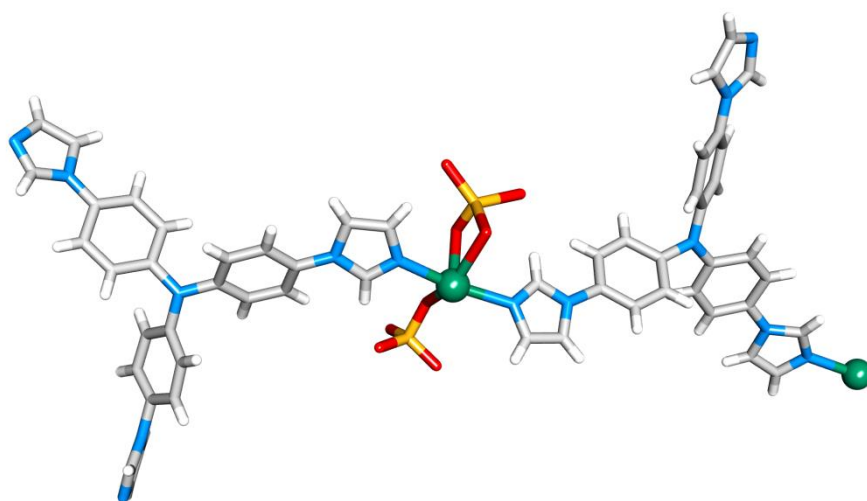


Figure S16: Asymmetric unit of **IPM-312**. (Colour code: C, grey; N, blue; Cd, green; O, dark red; H, white; S, yellow)

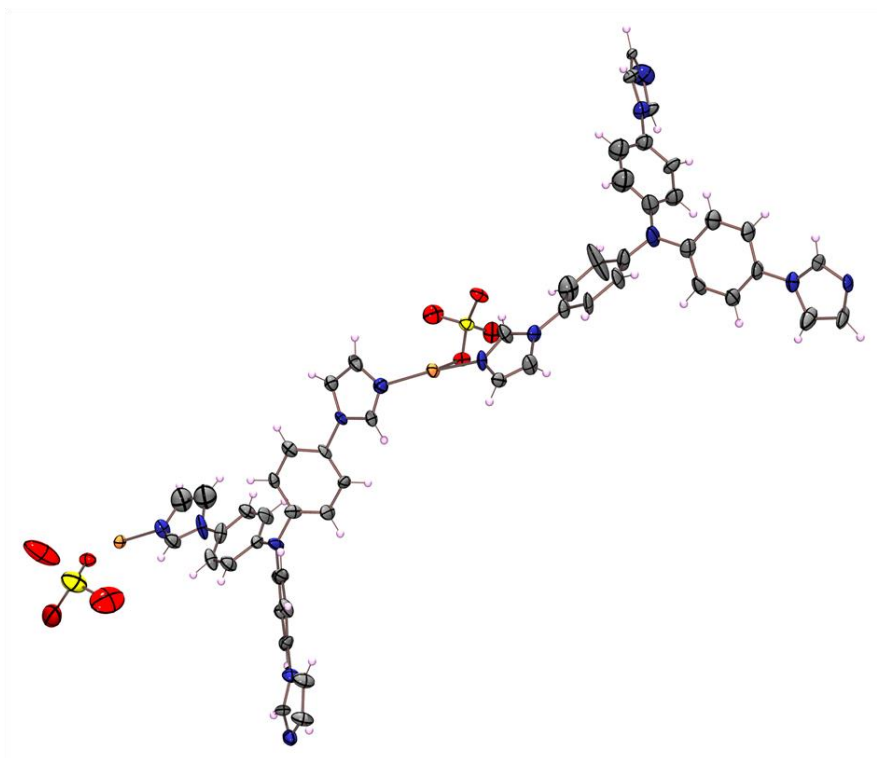


Figure S17: ORTEP diagram of **IPM-312** in thermal ellipsoids 50% probability. (Colour code: C, grey; N, blue; Cd, orange; O, red; S, yellow; H, pink)

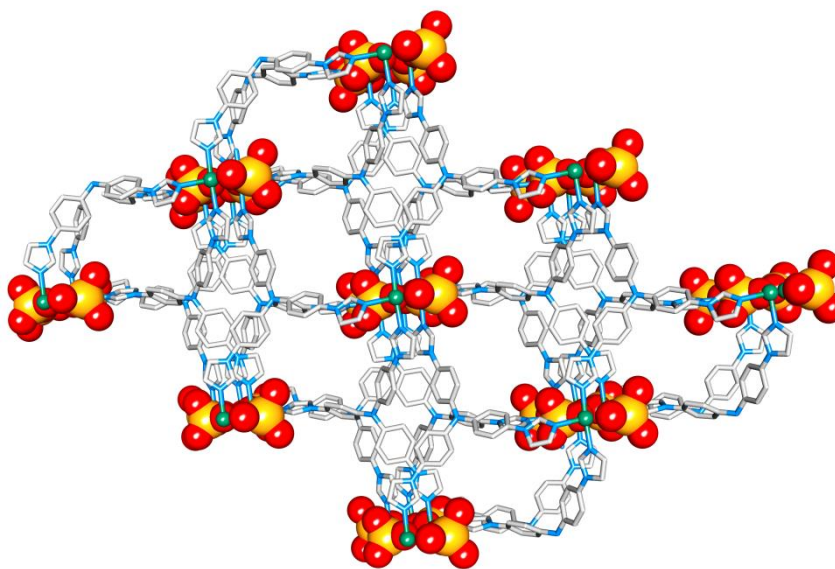


Figure S18: Figure showing packing of **IPM-312** along crystallographic *b*-axis, (Hydrogen atoms have been omitted for clarity. Colour code: C, grey; N, blue; Cd, green; O, red; S, yellow).

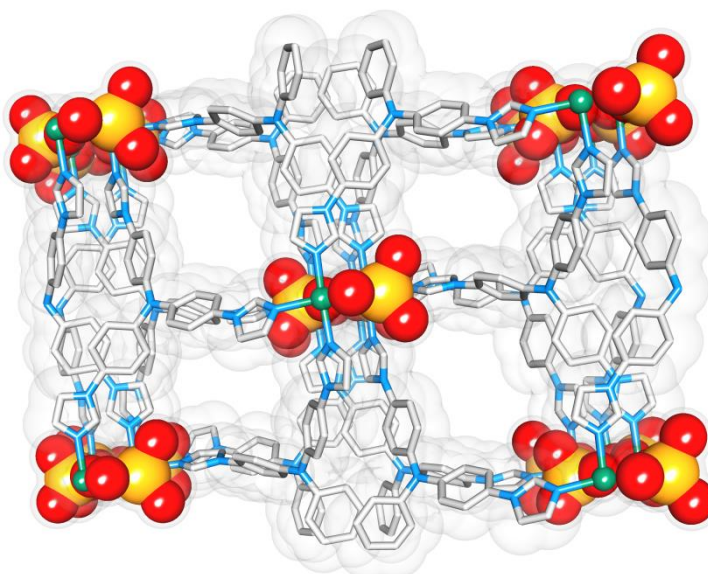


Figure S19: Figure showing packing of **IPM-312** along crystallographic *b*-axis, (Hydrogen atoms have been omitted for clarity. Colour code: C, grey; N, blue; Cd, green; O, red; S, yellow)

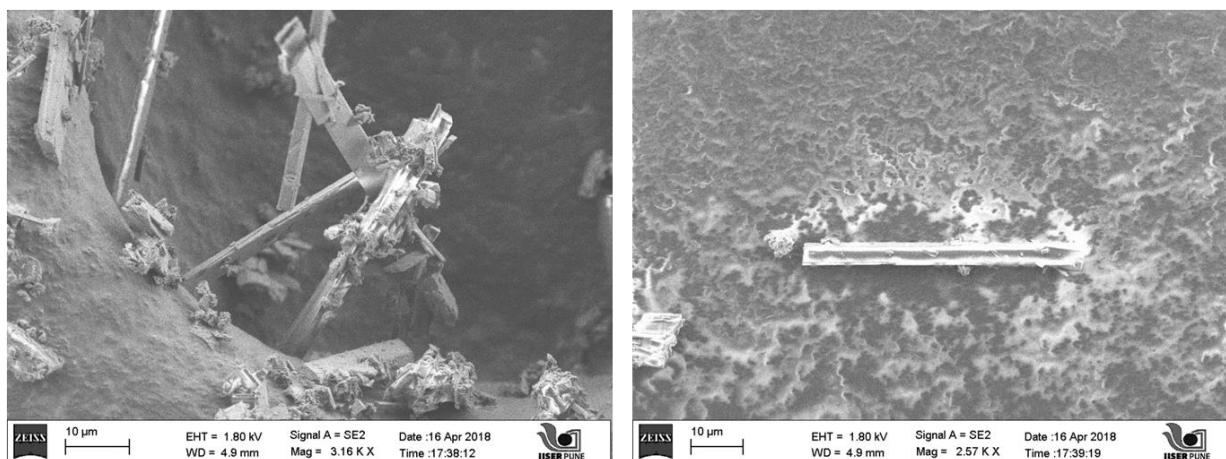


Figure S20: FESEM images for compound **IPM-312**. The images were recorded for the solid sample on carbon tape.

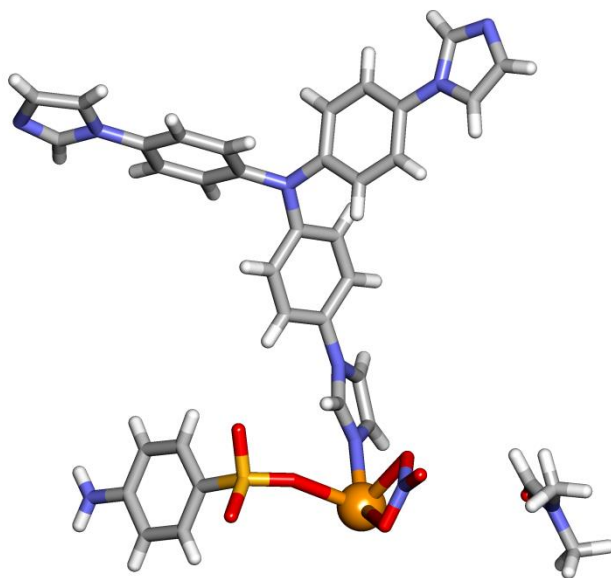


Figure S21: Asymmetric unit of **IPM-315**. (Colour code: C, grey; N, blue; Cd, orange; O, dark red; H, white; S, yellow)

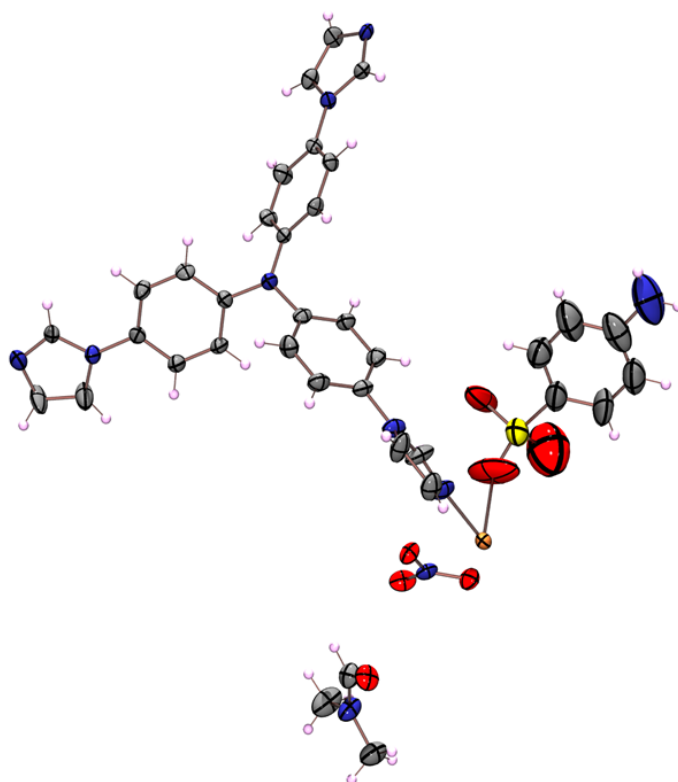


Figure S22: ORTEP diagram of **IPM-315** in thermal ellipsoids 50% probability. (Colour code: C, grey; N, blue; Cd, orange; O, red; H, pink)

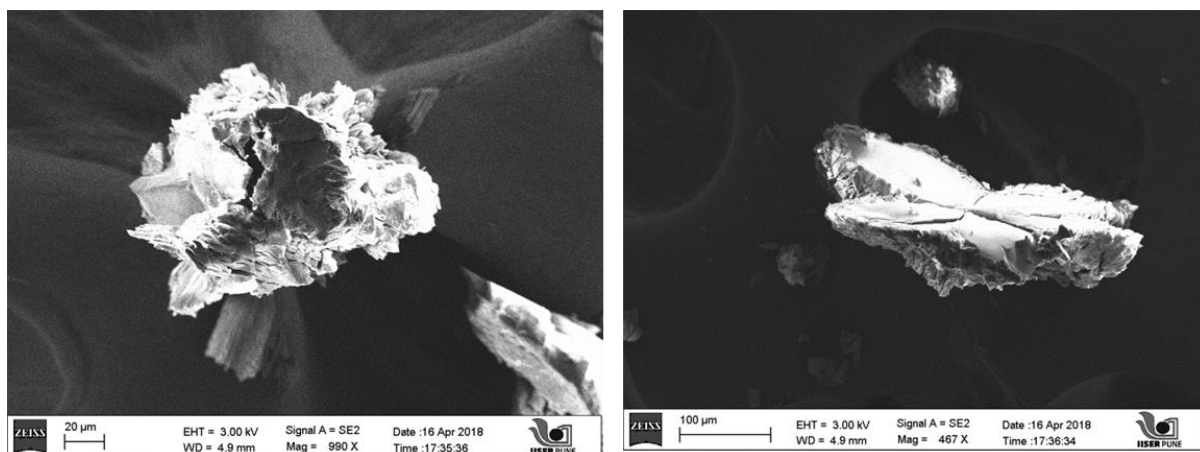


Figure S23: FESEM images for compound **IPM-315**. The images were recorded for the solid sample on carbon tape.

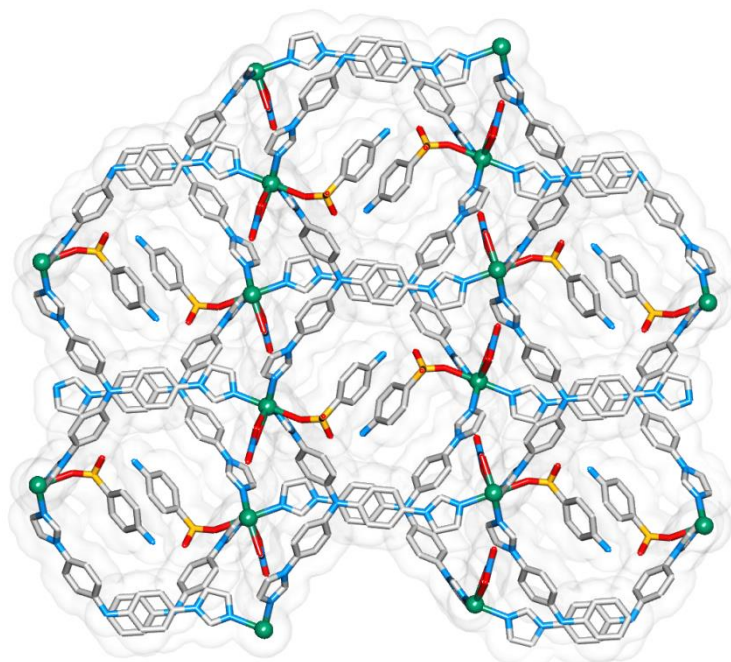


Figure S24: Figure showing packing of **IPM-315** along crystallographic *a*-axis. Guest solvent have been omitted for clarity. (Hydrogen atoms have been omitted for clarity. Colour code: C, grey; N, blue; Cd, green; O, red; S, yellow).

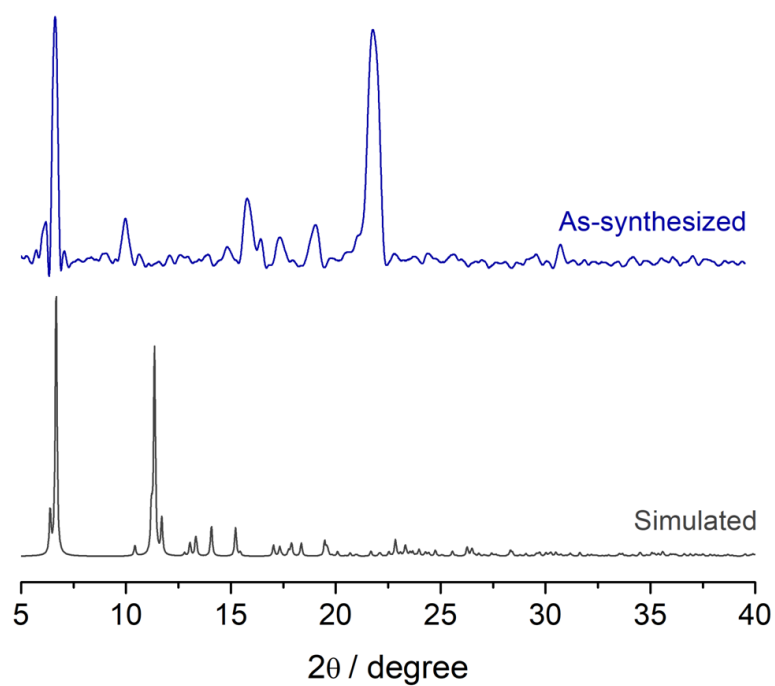


Figure S25: Powder X-ray diffraction patterns of **IPM-206**, simulated (grey), as-synthesized (blue).

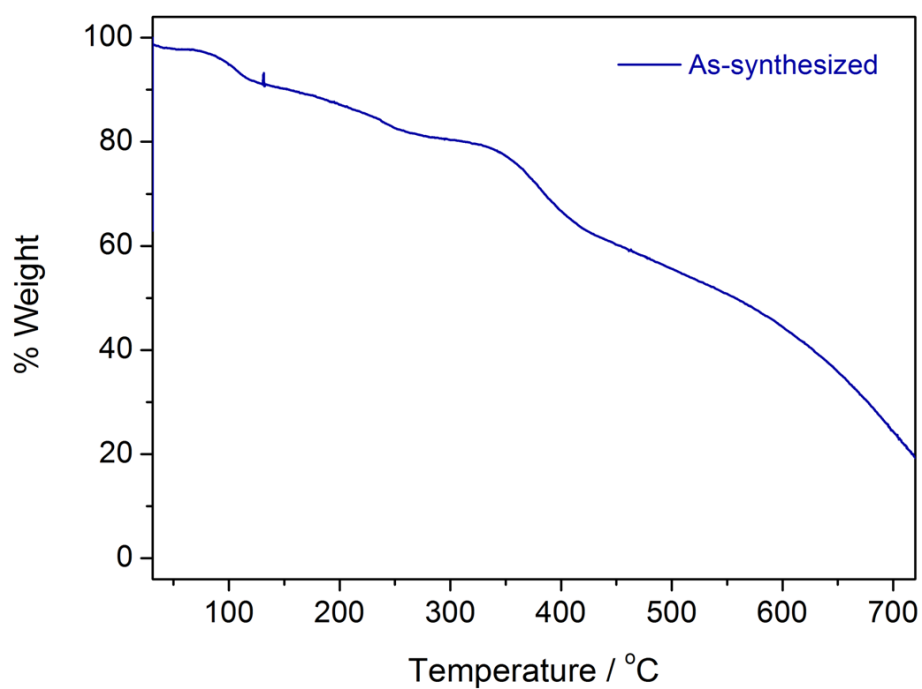


Figure S26: TGA profile for compound **IPM-206**.

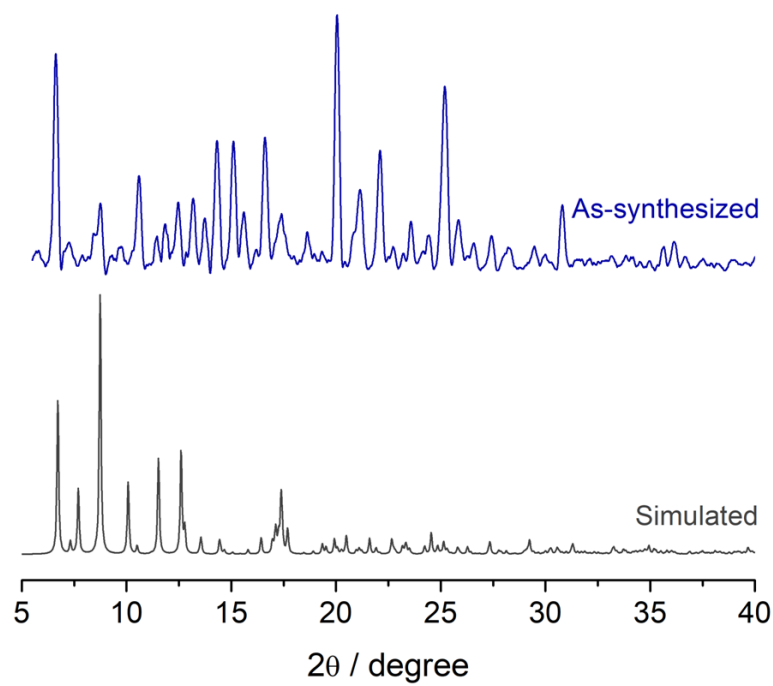


Figure S27: Powder X-ray diffraction patterns of **IPM-310**, simulated (grey), as-synthesized (blue).

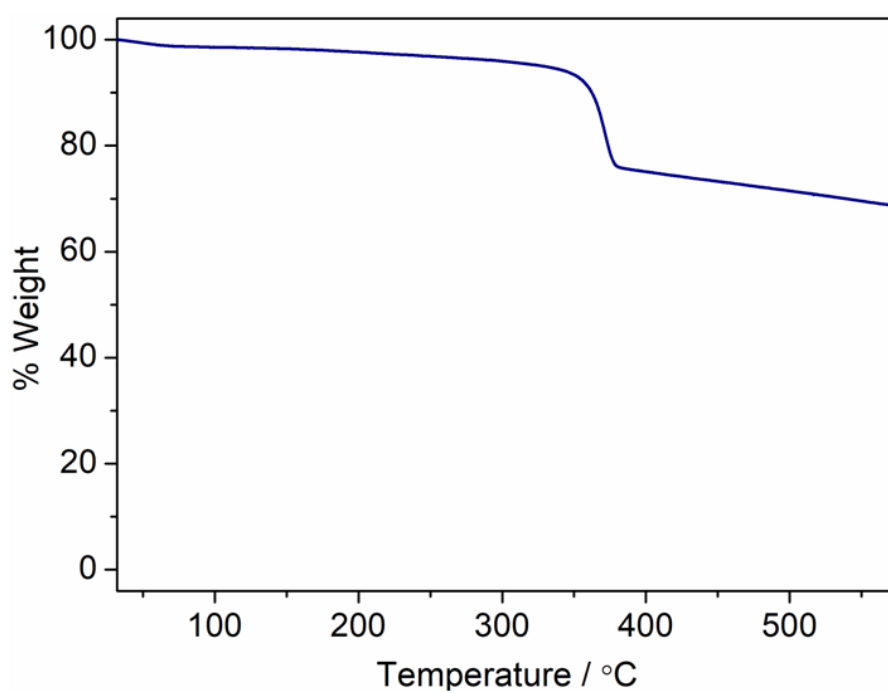


Figure S28: TGA profile for compound **IPM-310**.

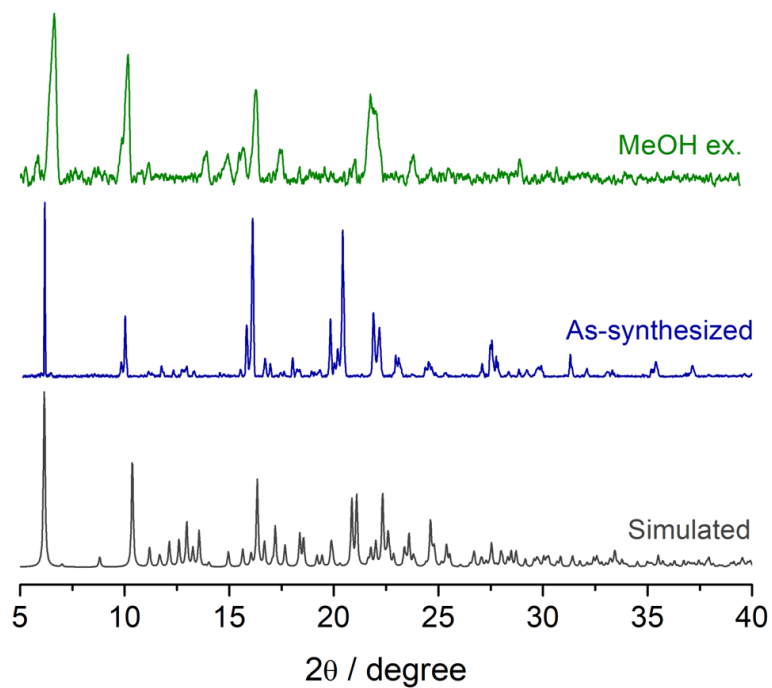


Figure S29: Powder X-ray diffraction patterns of **IPM-315**, simulated (grey), as-synthesized (blue), MeOH exchanged (green).

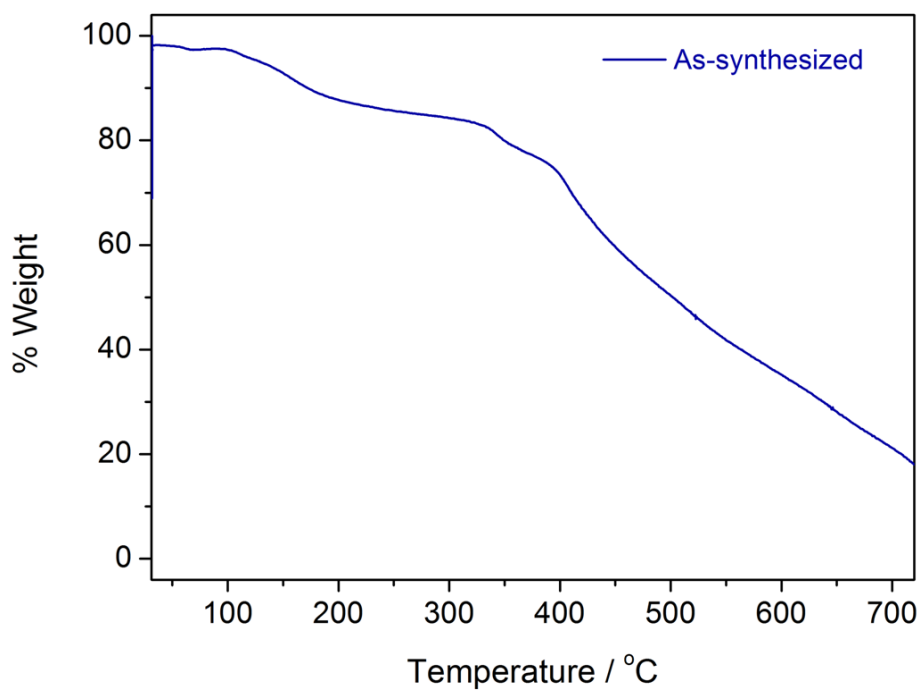


Figure S30: TGA profile for compound **IPM-315**.

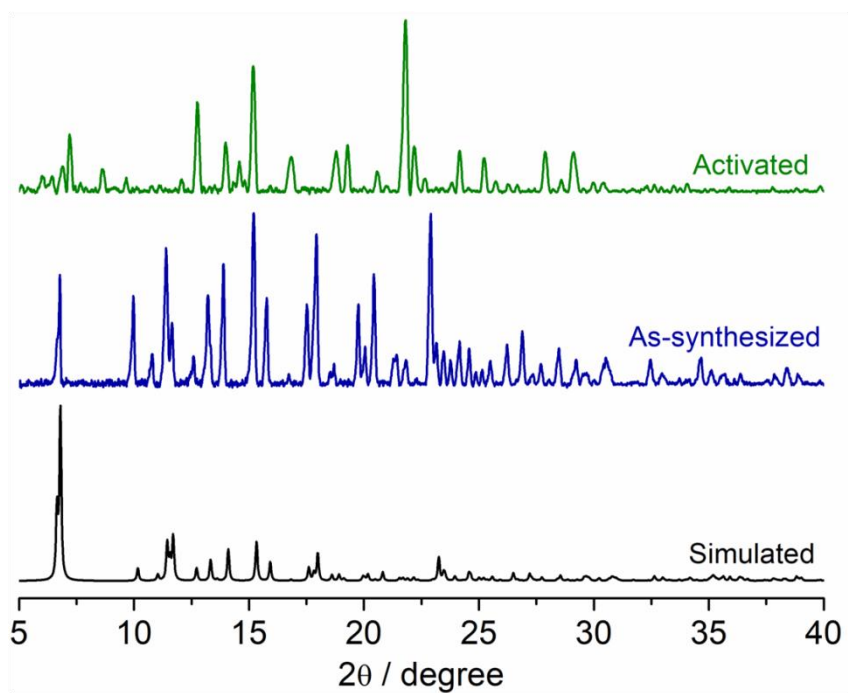


Figure S31: Powder X-ray diffraction patterns of **IPM-311**, simulated (grey), as-synthesized (blue), activated (green).

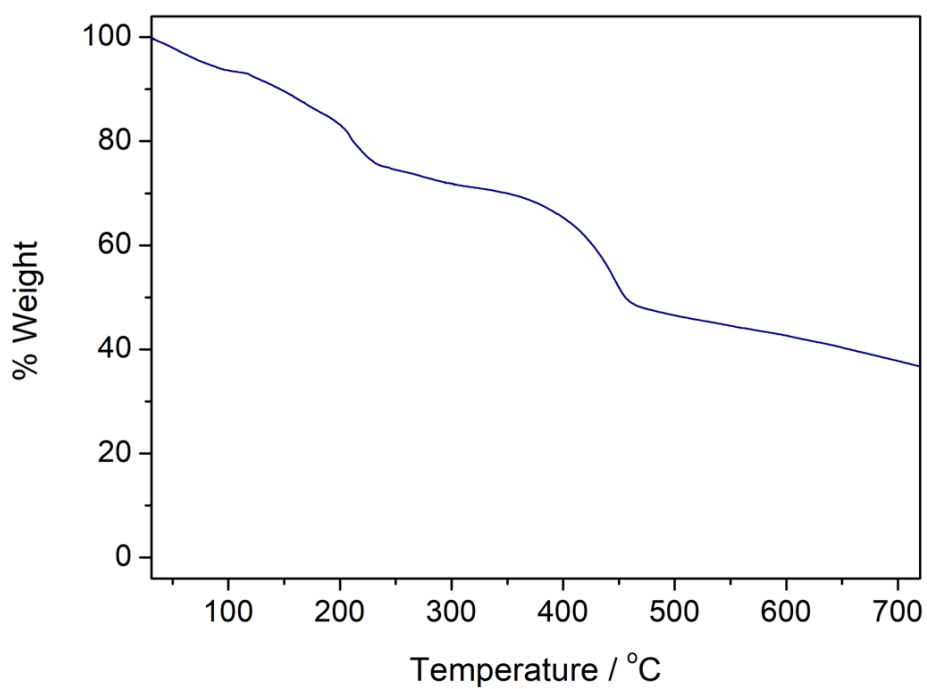


Figure S32: TGA profile for compound **IPM-311**.

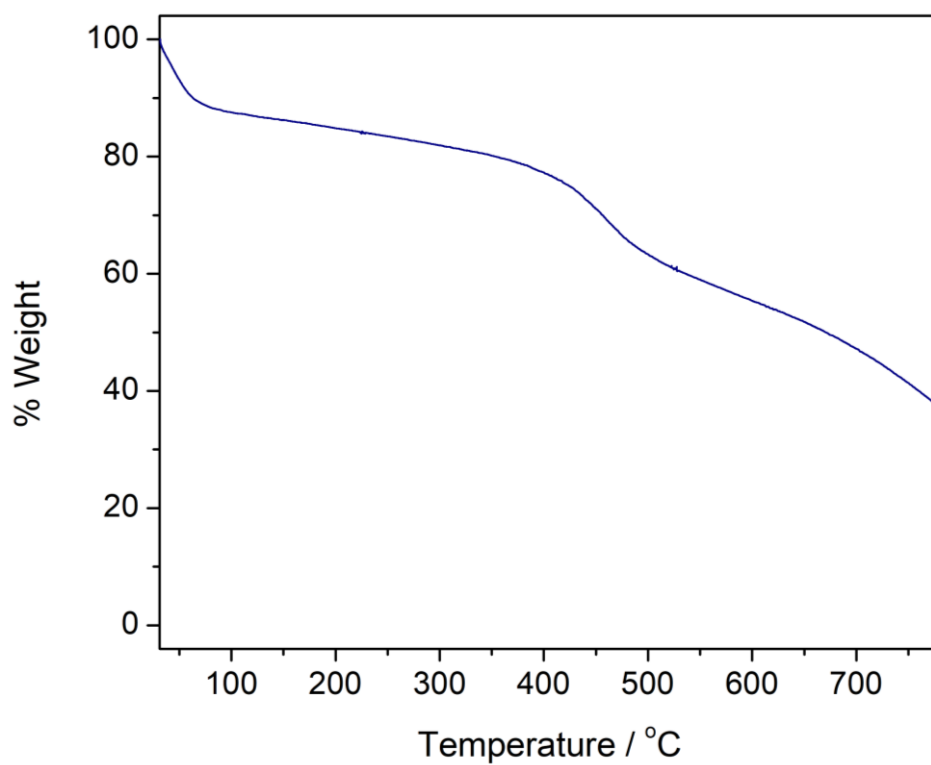


Figure S33: TGA profile for compound **IPM-312**.

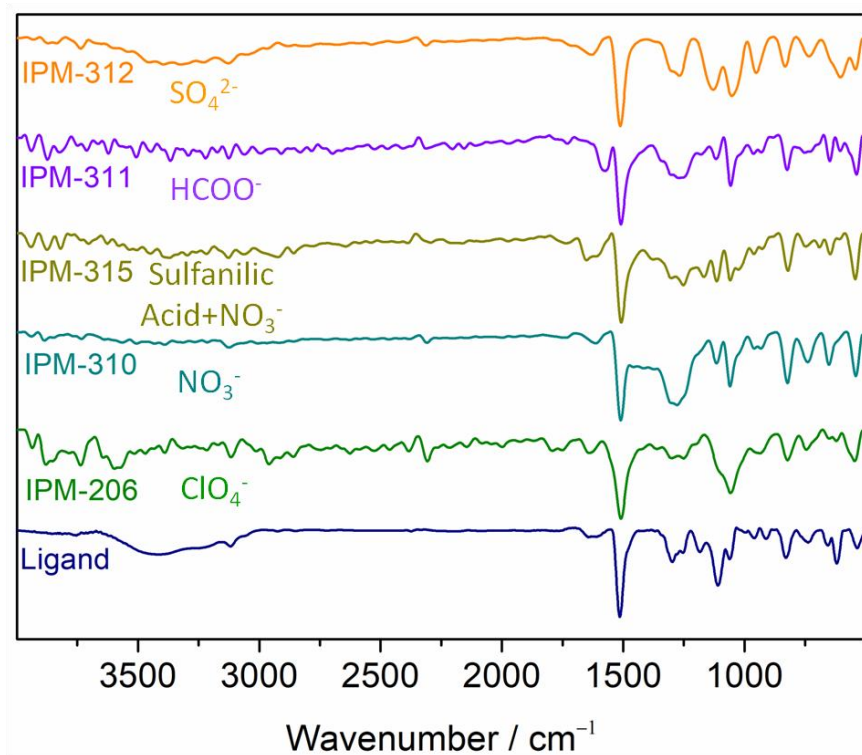


Figure S34: FT-IR profiles for ligand and all the compounds.

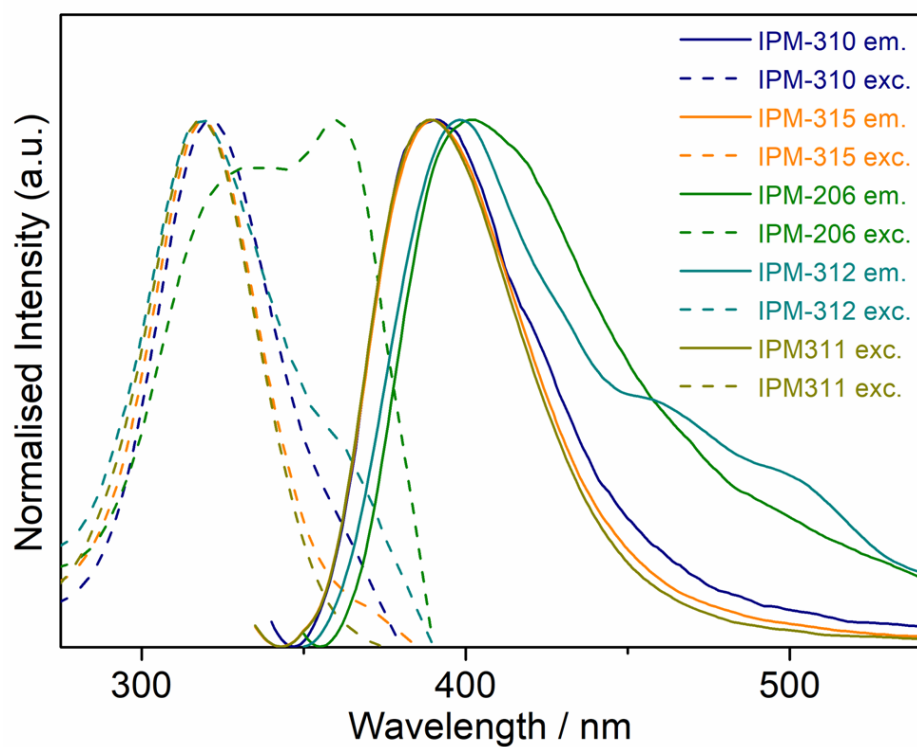


Figure S35: Excitation and emission profiles for various compounds.

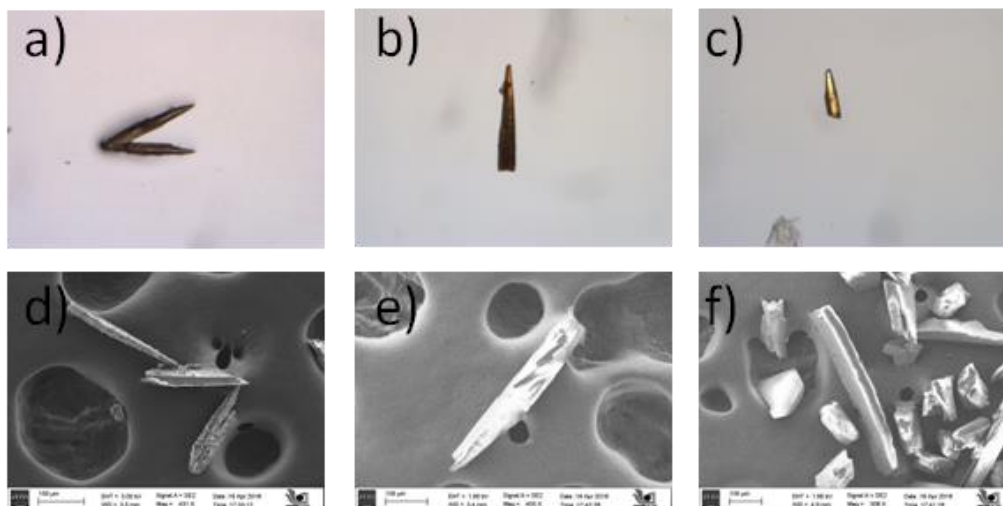


Figure S36: Naked eye images of **IPM-206**; a) As-synthesized, b) MnO_4^- exchanged, c) ReO_4^- exchanged, & SEM image of d) As-synthesized, e) MnO_4^- exchanged, f) ReO_4^- exchanged.

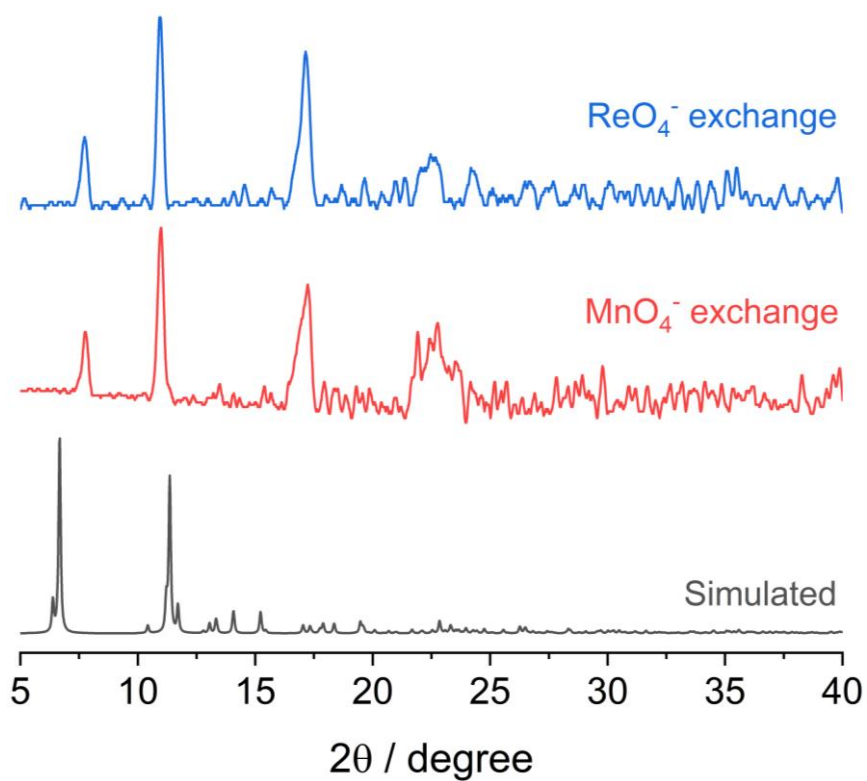


Figure S37: PXRD patterns for **IPM-206** - simulated, MnO₄⁻ exchanged, ReO₄⁻ exchanged.

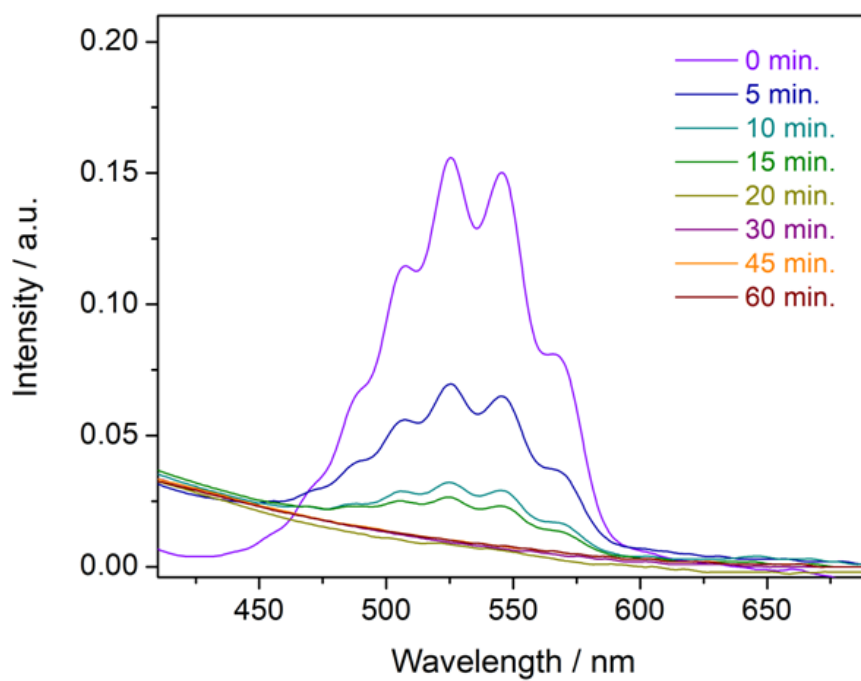


Figure S38: UV-Vis absorption spectra of MnO₄⁻ after addition of **IPM-206** at different intervals.

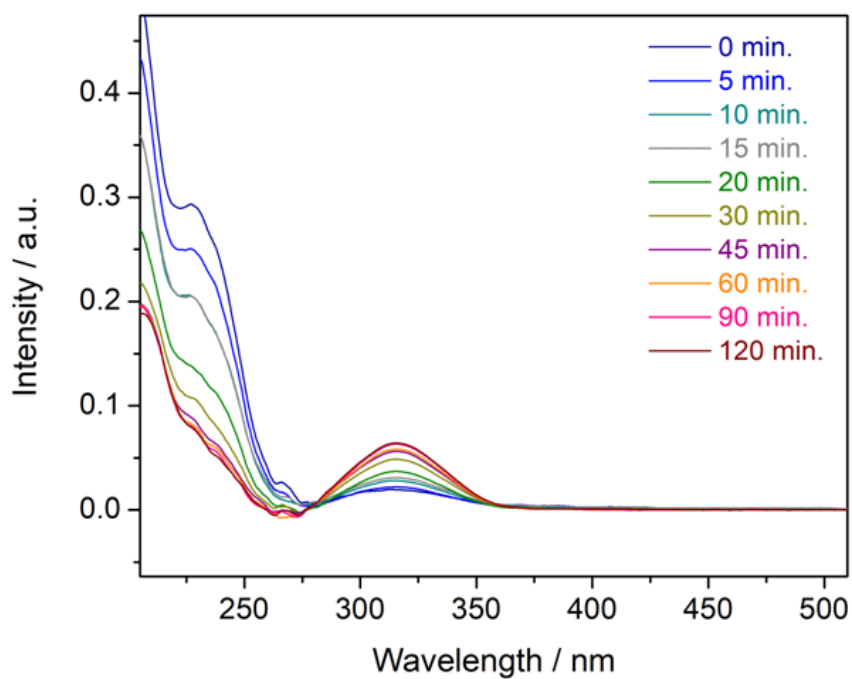


Figure S39: UV-Vis absorption spectra of ReO_4^- after addition of **IPM-206** at different intervals.

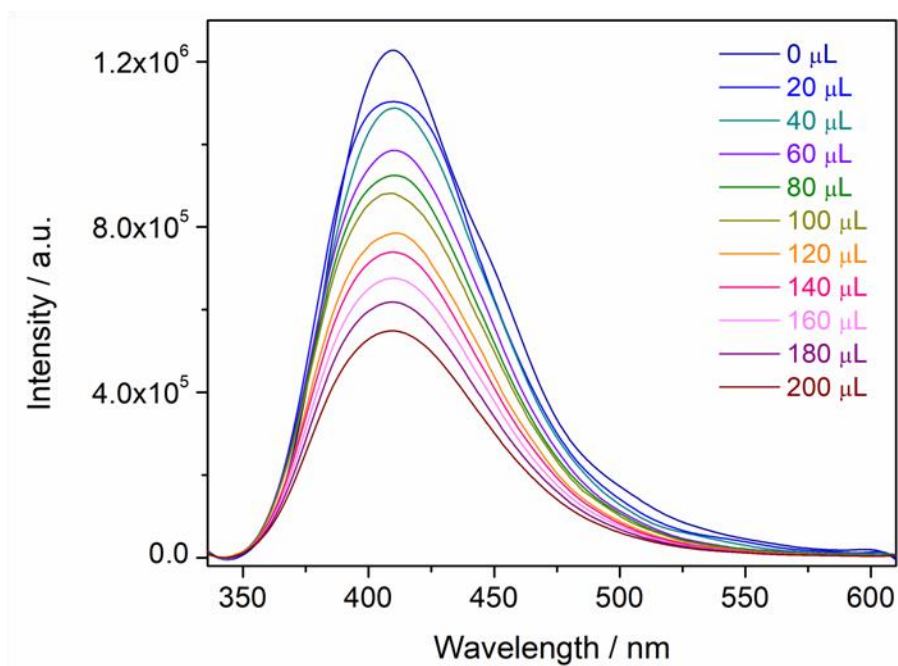


Figure S40: Fluorescence emission spectra of MnO_4^- after addition of **IPM-206** at different intervals.

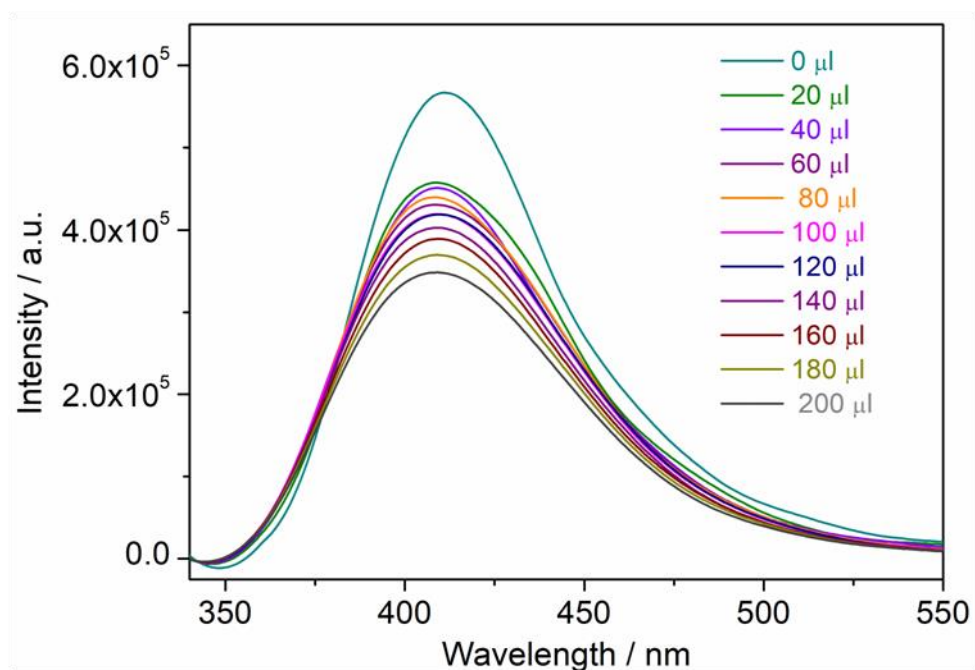


Figure S41: Fluorescence emission spectra of ReO_4^- after addition of IPM-206 at different intervals.

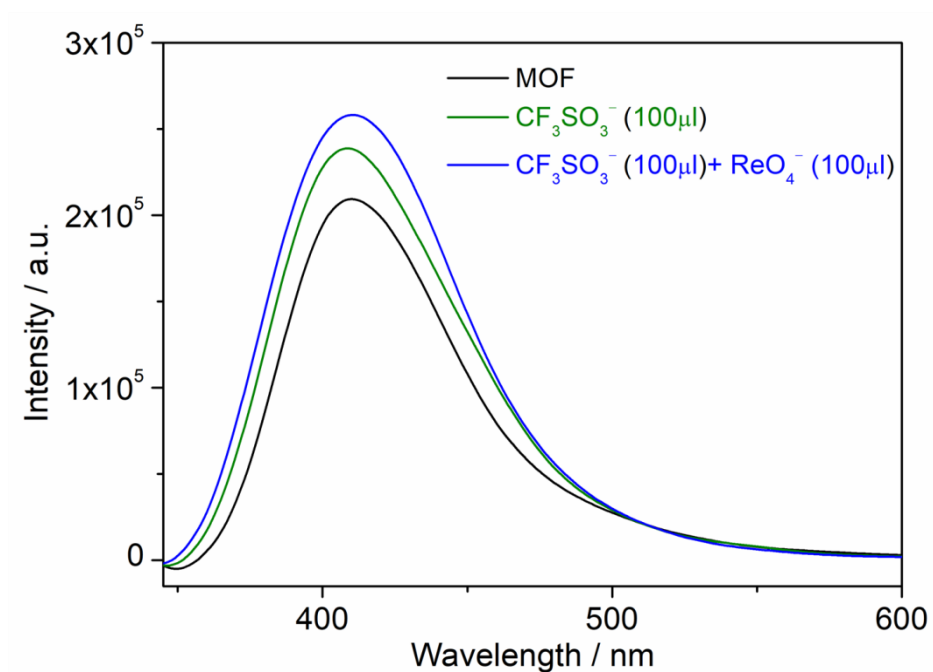


Figure S42: Fluorescence emission spectra of IPM-206 after addition of different analytes.

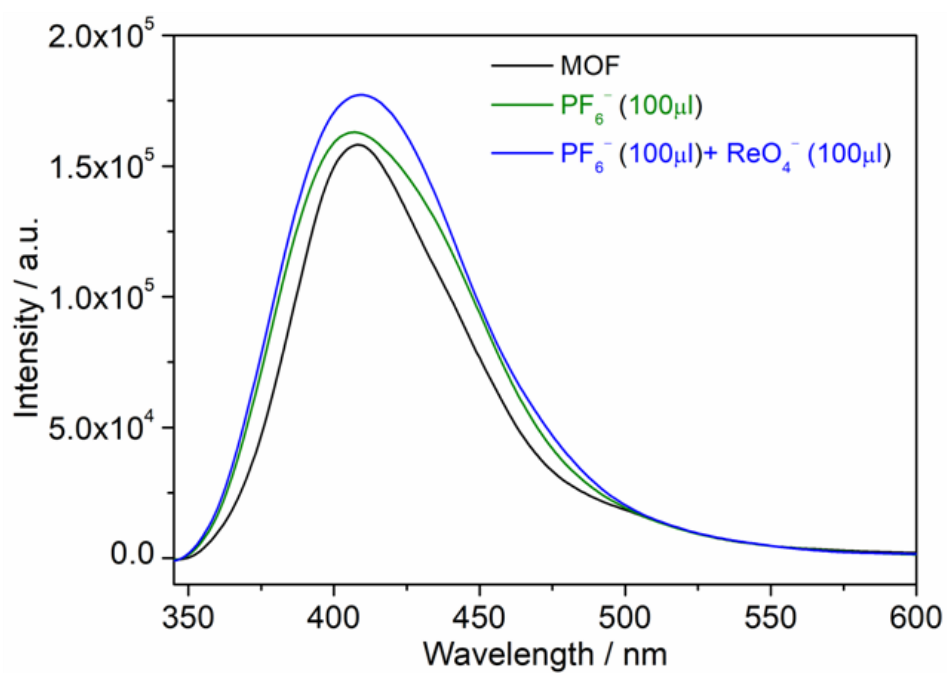


Figure S43: Fluorescence emission spectra of **IPM-206** after addition of different analytes.

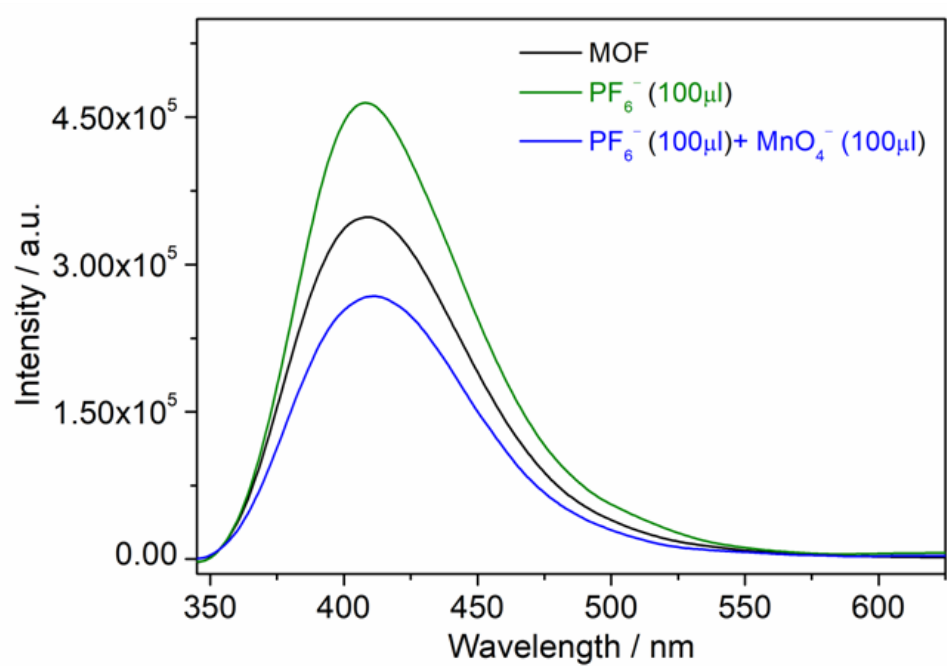


Figure S44: Fluorescence emission spectra of **IPM-206** after addition of different analytes.

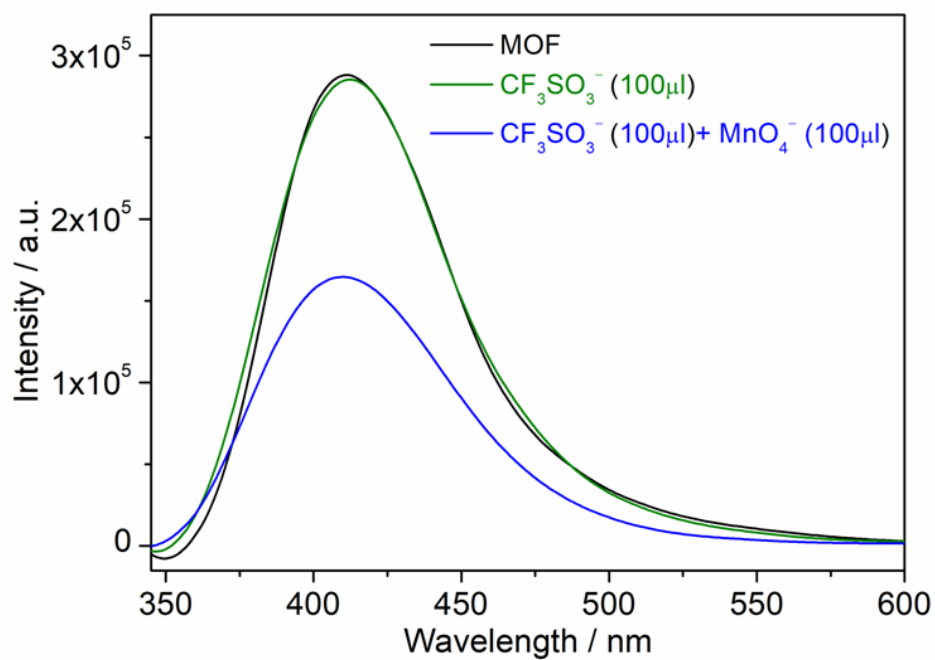


Figure S45: Fluorescence emission spectra of **IPM-206** after addition of different analytes.

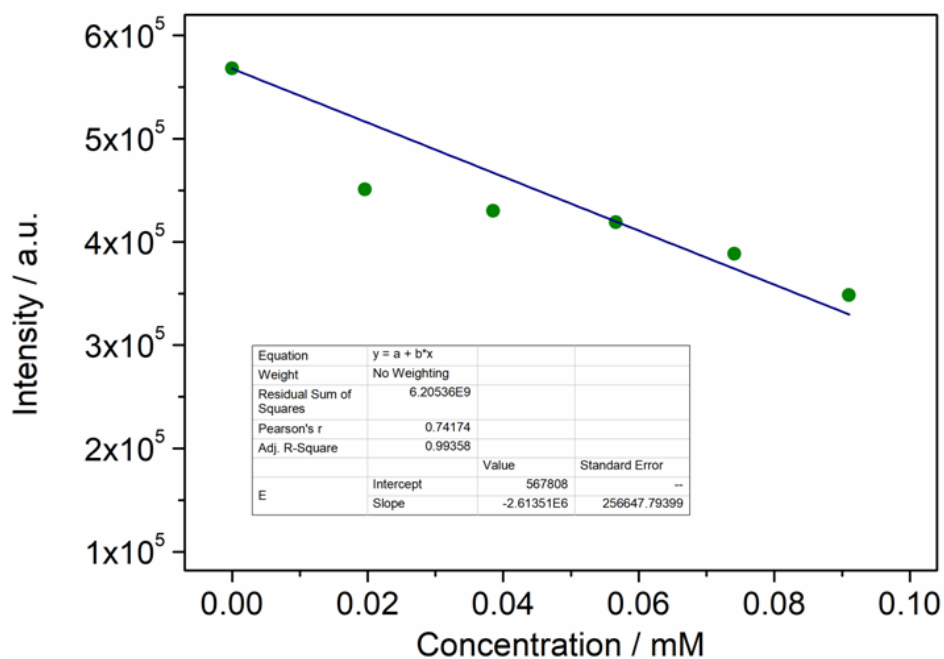


Figure S46: Linear region in the Fluorescence emission spectra after addition of ReO_4^- to **IPM-206** at $\lambda_{\text{em}} = 410\text{nm}$ ($\lambda_{\text{exc.}} = 325\text{nm}$) ($R^2 = 99.35\%$).

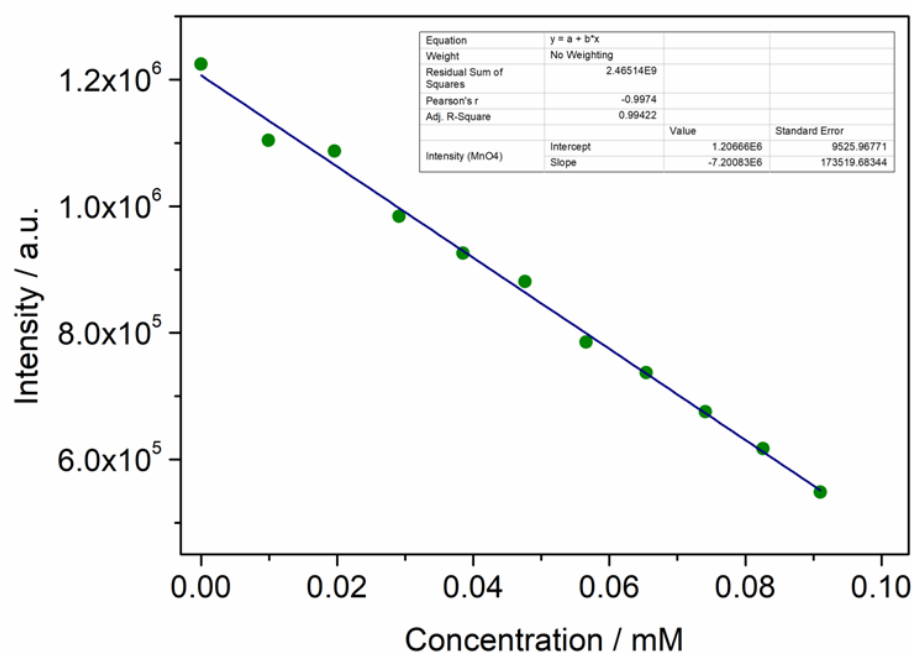


Figure S47: Linear region in the Fluorescence emission spectra after addition of MnO_4^- to **IPM-206** at $\lambda_{\text{em}} = 410\text{nm}$ ($\lambda_{\text{exc}} = 325\text{nm}$) ($R^2 = 99.42\%$).

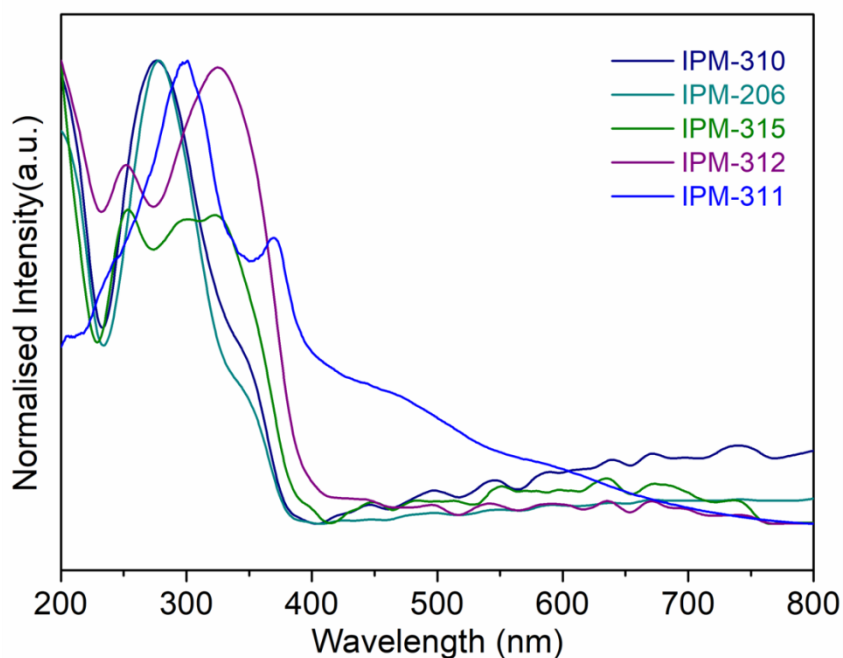


Figure S48: Solid State UV-Vis absorption profiles for **IPM-310** (Blue), **IPM-206** (cyan), **IPM-315** (green), **IPM-312** (violet) and **IPM-311** (bright blue).

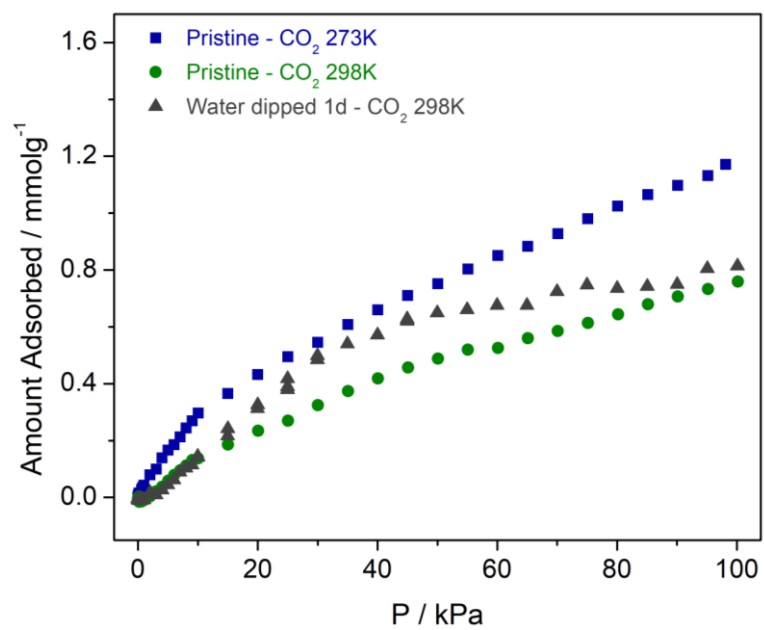


Figure S49: CO₂ adsorption isotherm for **IPM-312** after dipping in water.

Table 1. Crystal data and structure refinement for **IPM-206**.

Identification code	IPM-206 (CCDC 1846341)	
Empirical formula	$C_{54} H_{42} Cd N_{14}$	
Formula weight	999.41	
Temperature	150(2) K	
Wavelength	0.71073 Å	
Crystal system	Monoclinic	
Space group	$C2/c$	
Unit cell dimensions	$a = 30.2102(19)$ Å	$\alpha = 90^\circ$
	$b = 8.9486(6)$ Å	$\beta = 118.440(2)^\circ$
	$c = 31.4489(17)$ Å	$\gamma = 90^\circ$
Volume	7475.8(8) Å ³	
Z	4	
Density (calculated)	0.888 Mg/m ³	
Absorption coefficient	0.326 mm ⁻¹	
F(000)	2048	
Crystal size	0.14 x 0.12 x 0.10 mm ³	
Theta range for data collection	2.402 to 28.344°.	
Index ranges	-40 ≤ h ≤ 40, -11 ≤ k ≤ 11, -41 ≤ l ≤ 38	
Reflections collected	113594	
Independent reflections	9296 [R(int) = 0.0974]	
Completeness to theta = 28.42°	100.0 %	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	9296 / 0 / 313	
Goodness-of-fit on F ²	1.100	
Final R indices [I > 2σ(I)]	$R_1 = 0.0482$, $wR_2 = 0.1128$	
R indices (all data)	$R_1 = 0.0885$, $wR_2 = 0.1307$	
Largest diff. peak and hole	0.450 and -0.392 e.Å ⁻³	

Table 2. Crystal data and structure refinement for **IPM-310**.

Identification code	IPM-310 (CCDC 1846342)		
Empirical formula	C ₂₇ H ₂₁ N ₉ Cd O ₆		
Formula weight	679.93		
Temperature	150(2) K		
Wavelength	0.71073 Å		
Crystal system	Triclinic		
Space group	<i>P</i> -1		
Unit cell dimensions	<i>a</i> = 11.984(4) Å	α = 73.727(10)°	
	<i>b</i> = 12.573(4) Å	β = 73.273 (10)°	
	<i>c</i> = 14.220(5) Å	γ = 84.150 (12)°	
Volume	1969.1(11) Å ³		
<i>Z</i>	2		
Density (calculated)	1.147 Mg/m ³		
Absorption coefficient	0.597 mm ⁻¹		
<i>F</i> (000)	684		
Crystal size	0.12 x 0.11 x 0.10 mm ³		
Theta range for data collection	2.421 to 28.396°.		
Index ranges	-15<= <i>h</i> <=14, -16<= <i>k</i> <=16, -18<= <i>l</i> <=18		
Reflections collected	39207		
Independent reflections	9768 [<i>R</i> (int) = 0.0540]		
Completeness to theta = 28.42°	99.9 %		
Refinement method	Full-matrix least-squares on <i>F</i> ²		
Data / restraints / parameters	9768 / 0 / 388		
Goodness-of-fit on <i>F</i> ²	1.035		
Final <i>R</i> indices [<i>I</i> >2sigma(<i>I</i>)]	<i>R</i> ₁ = 0.0430, <i>wR</i> ₂ = 0.1054		
<i>R</i> indices (all data)	<i>R</i> ₁ = 0.0642, <i>wR</i> ₂ = 0.1160		
Largest diff. peak and hole	1.922 and -0.636 e.Å ⁻³		

Table 3. Crystal data and structure refinement for **IPM-315**.

Identification code	IPM-315 (CCDC 1846347)		
Empirical formula	$C_{36} H_{34} N_{10} Cd O_7 S$		
Formula weight	863.19		
Temperature	150(2) K		
Wavelength	0.71073 Å		
Crystal system	Orthorhombic		
Space group	$P2_12_12_1$		
Unit cell dimensions	$a = 8.9284(4)$ Å	$\alpha = 90^\circ$.	
	$b = 14.0416(7)$ Å	$\beta = 90^\circ$.	
	$c = 28.6683(14)$ Å	$\gamma = 90^\circ$.	
Volume	$3594.1(3)$ Å ³		
Z	4		
Density (calculated)	1.595 Mg/m ³		
Absorption coefficient	0.731 mm ⁻¹		
F(000)	1760		
Crystal size	0.12 x 0.10 x 0.08 mm ³		
Theta range for data collection	2.389 to 28.362°.		
Index ranges	$-11 \leq h \leq 11$, $-18 \leq k \leq 18$, $-38 \leq l \leq 38$		
Reflections collected	89134		
Independent reflections	8957 [R(int) = 0.1167]		
Completeness to theta = 28.42°	99.9 %		
Refinement method	Full-matrix least-squares on F ²		
Data / restraints / parameters	8957 / 6 / 498		
Goodness-of-fit on F ²	1.035		
Final R indices [I > 2sigma(I)]	$R_1 = 0.0616$, $wR_2 = 0.1437$		
R indices (all data)	$R_1 = 0.1019$, $wR_2 = 0.1608$		
Largest diff. peak and hole	1.713 and -1.366 e.Å ⁻³		

Table 4. Crystal data and structure refinement for **IPM-311**.

Identification code	IPM-311 (CCDC 1846344)	
Empirical formula	$C_{29} H_{23} N_7 Cd O_4$	
Formula weight	645.94	
Temperature	150(2) K	
Wavelength	0.71073 Å	
Crystal system	Monoclinic	
Space group	$C2$	
Unit cell dimensions	$a = 26.535(2)$ Å	$\alpha = 90^\circ$
	$b = 15.453(2)$ Å	$\beta = 101.627(4)^\circ$
	$c = 8.8765(9)$ Å	$\gamma = 90^\circ$
Volume	$3565(7)$ Å ³	
Z	4	
Density (calculated)	1.203 Mg/m ³	
Absorption coefficient	0.650 mm ⁻¹	
F(000)	1304	
Crystal size	0.18 x 0.16 x 0.14 mm ³	
Theta range for data collection	2.343 to 28.280°.	
Index ranges	$-34 \leq h \leq 35, -20 \leq k \leq 20, -10 \leq l \leq 11$	
Reflections collected	43878	
Independent reflections	8814 [$R(\text{int}) = 0.1057$]	
Completeness to $\theta = 28.42^\circ$	99.9 %	
Refinement method	Full-matrix least-squares on F^2	
Data / restraints / parameters	8814 / 7 / 371	
Goodness-of-fit on F^2	0.981	
Final R indices [$I > 2\sigma(I)$]	$R_1 = 0.0504, wR_2 = 0.0968$	
R indices (all data)	$R_1 = 0.1214, wR_2 = 0.1167$	
Largest diff. peak and hole	0.546 and -0.448 e.Å ⁻³	

Table 5. Crystal data and structure refinement for **IPM-312**.

Identification code	IPM-312 (CCDC 1846346)	
Empirical formula	$C_{54} H_{42} N_{14} Cd_2 O_8 S_2$	
Formula weight	1303.93	
Temperature	150(2) K	
Wavelength	1.54178 Å	
Crystal system	Triclinic	
Space group	<i>P</i> -1	
Unit cell dimensions	$a = 15.9617(11)$ Å	$\alpha = 113.108(4)^\circ$
	$b = 16.1208(11)$ Å	$\beta = 113.283(4)^\circ$
	$c = 16.3842(11)$ Å	$\gamma = 99.327(5)^\circ$
Volume	3301.3(4) Å ³	
Z	2	
Density (calculated)	1.312 Mg/m ³	
Absorption coefficient	6.220 mm ⁻¹	
F(000)	1312	
Crystal size	0.15 x 0.12 x 0.10 mm ³	
Theta range for data collection	3.216 to 67.393°.	
Index ranges	-18 ≤ <i>h</i> ≤ 19, -19 ≤ <i>k</i> ≤ 19, -16 ≤ <i>l</i> ≤ 19	
Reflections collected	24186	
Independent reflections	11522 [<i>R</i> (int) = 0.0704]	
Completeness to theta = 28.42°	97.0 %	
Refinement method	Full-matrix least-squares on <i>F</i> ²	
Data / restraints / parameters	11522 / 36 / 721	
Goodness-of-fit on <i>F</i> ²	0.936	
Final <i>R</i> indices [<i>I</i> > 2σ(<i>I</i>)]	<i>R</i> ₁ = 0.0800, <i>wR</i> ₂ = 0.2361	
<i>R</i> indices (all data)	<i>R</i> ₁ = 0.1170, <i>wR</i> ₂ = 0.2652	
Largest diff. peak and hole	3.778 and -2.110 e.Å ⁻³	