## Supporting information for

## Barium-promoted ruthenium catalysts on yittria-stabilized zirconia supports for ammonia synthesis

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Figure S1: Normalized loading of catalyst/promoters as a function of normalized precursor concentration for Ru, Cs, K, and Ba displaying the nominally linear relationship between loading and precursor concentration.



Figure S2: Conversion as a function of the catalyst weight. Flowrate is adjusted to keep space velocity constant including 18,000, 36,000, and 72,000 mL gcat<sup>-1</sup> h<sup>-1</sup>, for 0.1, 0.2, and 0.3 g in catalyst weight, separately. At 18,000 and 36,000 the rate slowly increases with catalysts wright whereas at 72,000 mL gcat<sup>-1</sup> h<sup>-1</sup> the rate saturates with mass. Reaction conditions: P = 10 bar (gauge),  $H_2/N_2 = 3$ ,  $T = 450^{\circ}$ C.



Figure S3: Normalized NH<sub>3</sub> synthesis rate of Ru/YSZ over time at 450°C. Reaction conditions: p = 0 bar (gauge), H<sub>2</sub>/N<sub>2</sub> = 3, SV = 72,000 mL gcat<sup>-1</sup> h<sup>-1</sup>. The Ru/YSZ takes up to 60 hours of activation to approach its maximum rate but then remains stable

The supplementary data include required input files in Chemkin format. The files include:

1- surf.inp: The file contains the surface species, side densities and the elementary step reaction mechanism for Ba-Ru/YSZ catalysts.

The mechanism uses the units of cm, s, kJ, mol.

2- chem.inp: Contains the gas phase species and element lists

3- therm.dat: Contains NASA constants of gas-phase and surface species to calculate the thermodyamic properties.

The gas-phase species thermodynamic data are taken from Burcat (http://garfield.chem.elte.hu/Burcat/burcat.html). The surface species NASA constants are based on

Gibbs free energy calculations after the adjustment procedure.

4- Trans.dat: Gas-phase speices transport properties are listed.

The data are taken from CURRAN, H. J., GAFFURI, P., PITZ, W. J., AND WESTBROOK, C. K. "A COMPREHENSIVE MODELING STUDY OF ISO-OCTANE OXIDATION"

COMBUSTION AND FLAME 129:253-280 (2002)

```
SITE /Ru_surface/ SDEN/2.6079E-9/
Ru(s)
N(s)
H(s)
NH(s)
NH2(s)
NH3(s)
END
THERMO
 300.0
         1000.0
                 3000.0
Ru(s)
                      Ru 1H 0
                                        G 0300.00
                                                    5000.00 1000.00
                                                                           1
0.0000000E+00 0.0000000E+00-0.0000000E+00 0.0000000E+00-0.0000000E+00
                                                                           2
0.0000000E+00 0.0000000E+00-0.00000000E+00 0.0000000E+00-0.00000000E+00
                                                                           з
0.0000000E+00 0.0000000E+00-0.0000000E+00 0.0000000E+00
                                                                           4
N(s)
                      N
                         1H
                              ORU 1
                                         G 0300.00
                                                     5000.00 1000.00
                                                                           1
0.0000000E+00 0.0000000E+00-0.0000000E+00 0.0000000E+00-0.0000000E+00
                                                                           2
0.0000000E+00 0.0000000E+00-0.00000000E+00 0.0000000E+00-0.0000000E+00
                                                                           з
0.0000000E+00 0.0000000E+00-0.0000000E+00 0.0000000E+00
                                                                           4
H(s)
                      0
                         0H 1RU 1
                                        G 0300.00
                                                    5000.00 1000.00
                                                                           1
0.0000000E+00 0.0000000E+00-0.0000000E+00 0.0000000E+00-0.0000000E+00
                                                                           2
 0.0000000E+00 0.0000000E+00-0.00000000E+00 0.0000000E+00-0.0000000E+00
                                                                           з
0.0000000E+00 0.0000000E+00-0.0000000E+00 0.0000000E+00
                                                                           4
NH(s)
                      N 1H 1Ru 1
                                         G 0300.00
                                                     5000.00 1000.00
                                                                           1
0.0000000E+00 0.0000000E+00-0.0000000E+00 0.0000000E+00-0.0000000E+00
                                                                           2
 0.0000000E+00 0.0000000E+00-0.00000000E+00 0.0000000E+00-0.00000000E+00
                                                                           з
0.0000000E+00 0.0000000E+00-0.0000000E+00 0.0000000E+00
                                                                           4
NH2(s)
                      H 2N 1Ru 1
                                        G 0300.00
                                                    5000.00 1000.00
                                                                           1
0.0000000E+00 0.0000000E+00-0.0000000E+00 0.0000000E+00-0.0000000E+00
                                                                           2
0.0000000E+00 0.0000000E+00-0.0000000E+00 0.000000E+00-0.0000000E+00
                                                                           з
0.0000000E+00 0.0000000E+00-0.0000000E+00 0.0000000E+00
                                                                           4
NH3(s)
                      н
                          ЗN
                              1Ru 1
                                         G 0300.00
                                                     5000.00 1000.00
                                                                           1
0.0000000E+00 0.0000000E+00-0.0000000E+00 0.0000000E+00-0.0000000E+00
                                                                          2
0.0000000E+00 0.0000000E+00-0.00000000E+00 0.0000000E+00-0.0000000E+00
                                                                          з
0.0000000E+00 0.0000000E+00-0.0000000E+00 0.0000000E+00
                                                                          4
END
REACTIONS KJOULES/MOLE MWOFF
N2 + Ru(s) + Ru(s) = > N(s) + N(s)
                                       2.892E-06 0.000
                                                            38,949
   STICK
N(s) + N(s) = Ru(s) + Ru(s) + N2
                                       2.015E+17 -0.279
                                                            148.027
                                         0.000E+00 0.000
                                                              -14.000/
   COV/N(s)
H2 +Ru(s) +Ru(s) =>H(s) +H(s)
                                       4.007E-03 0.000
                                                            0.000
   STICK
H(s) + H(s) =>Ru(s) + Ru(s) + H2
                                       3.600E+20 0.658
                                                            91.948
                                         0.000E+00 0.000
   COV/H(s)
                                                              -2.000/
                                         1.247E-05 0.000
                                                              0.000
```

surf

NH3 +Ru(s) =>NH3(s) STICK

NH3(s) =>Ru(s) +NH3
H(s) +N(s) =>NH(s) +Ru(s)
COV/N(s)
NH(s) +Ru(s) =>H(s) +N(s)
COV/H(s)
NH(s) +H(s) =>NH2(s) +Ru(s)
NH2(s) + Ru(s) => NH(s) + H(s)
COV/H(s)
NH2(s) +H(s) =>NH3(s) +Ru(s)
NH3(s) + Ru(s) => NH2(s) + H(s)
COV/H(s)
END

surf		
2.235E+11	0.083	83.536
8.424E+20	0.000	83.620
0.000E+00	0.000	-7.000/
6.813E+19	0.207	30.972
0.000E+00	0.000	1.000/
4.949E+19	0.083	75.236
8.321E+19	-0.083	15.767
0.000E+00	0.000	1.000/
3.886E+19	0.083	17.036
1.478E+20	0.000	64.980
0.000E+00	0.000	1.000/

ELEMENTS H N Ru AR END SPECIES H2 NH3 N2 AR END REACTIONS KJOULES/MOLE END chem

therm2

!DetchemDatabase THERMO 0300.00 1000.00 5000.00 ! FROM GRI MECHANISM G 0300.00 5000.00 1000.00 AR 120186AR 1 1 0.02500000E+02 0.0000000E+00 0.0000000E+00 0.0000000E+00 0.0000000E+00 2 -0.07453750E+04 0.04366000E+02 0.02500000E+02 0.0000000E+00 0.0000000E+00 3 0.0000000E+00 0.0000000E+00-0.07453750E+04 0.04366000E+02 4 121386N 1H 3 G 0300.00 5000.00 1000.00 NH3 1 0.02461904E+02 0.06059166E-01-0.02004976E-04 0.03136003E-08-0.01938317E-12 2 -0.06493269E+05 0.07472097E+02 0.02204351E+02 0.10114765E-01-0.14652648E-04 3 0.14472350E-07-0.05328509E-10-0.06525488E+05 0.08127138E+02 4 N2 121286N 2 G 0300.00 5000.00 1000.00 1 0.02926640E+02 0.01487977E-01-0.05684761E-05 0.01009704E-08-0.06753351E-13 2 -0.09227977E+04 0.05980528E+02 0.03298677E+02 0.01408240E-01-0.03963222E-04 3 0.05641515E-07-0.02444855E-10-0.01020900E+05 0.03950372E+02 4 121286H 2 G 0300.00 5000.00 1000.00 1 H2 0.02991423E+02 0.07000644E-02-0.05633828E-06-0.09231578E-10 0.15827519E-14 2 -0.08350340E+04-0.13551101E+01 0.03298124E+02 0.08249441E-02-0.08143015E-05 3 -0.09475434E-09 0.04134872E-11-0.10125209E+04-0.03294094E+02 4 END

tran2 !Methyl Formate Oxidation: Speciation Data, Laminar Burning Velocities, Ignition Delay Times and a Validated Chemical Kinetic Model !S. Dooley, M.P. Burke, M. Chaos, Y. Stein, F.L. Dryer, V.P. Zhukov, O. Finch, J.M. Simmie and H.J. Curran. !Comments to Stephen Dooley, Princeton University, dooleys@princeton.edu ! Base transport parameters from, CURRAN, H. J., GAFFURI, P., PITZ, W. J., AND WESTBROOK, C. K. ! "A COMPREHENSIVE MODELING STUDY OF ISO-OCTANE OXIDATION" ! COMBUSTION AND FLAME 129:253-280 (2002). ! UCRL-WEB-204236 ! REVIEW AND RELEASE DATE: MAY 19, 2004. ! MANY OF THE TRANSPORT DATA ARE TAKEN FROM ! KEE, R.J., DIXON-LEWIS, G., WARNATZ, J., COLTRIN, M.E., AND MILLER, J.A, ! A FORTRAN COMPUTER CODE PACKAGE FOR THE EVALUATION OF GAS-PHASE MULTICOMPONENT ! TRANSPORT PROPERTIES, SANDIA REPORT SAND86-8246, 1986. ! NH3 2 481.000 2.920 1.470 0.000 10.000 0.000 0.000 AR 0 136.500 3.330 0.000 1 38.000 2.920 0.000 0.790 280.000 H2 N2 1 97.530 0.000 1.760 4.000 3.621 END