## Supporting Information DFT-based Global Optimization Of Sub-nanometre Ni-Pd Clusters

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Table S1: Total energies (taken from MEGA-VASP calculations), spin magnetic moments (SMM), binding energies  $(E_b)$  and excess energies  $(\Delta)$  for 3-8-atom Ni, Pd and Ni-Pd clusters. Second differences in energy  $(\Delta_2 E)$  are also given for pure Ni and Pd clusters

Composition	$\rm E$ / $\rm eV$	${ m SMM}~/~\mu_{ m B}$	$E_b / \text{eV}$	$\Delta$ / eV	$\Delta_2 E / \text{eV}$
$Pd_3$	-8.32	2	1.30	0.00	-0.30
$Ni_1Pd_2$	-8.54	2	1.65	-0.46	
$Ni_2Pd_1$	-8.28	2	1.84	-0.44	
Ni <sub>3</sub>	-7.61	4	1.89	0.00	-0.38
$Pd_4$	-12.66	2	1.69	0.00	0.55
$Ni_1Pd_3$	-12.82	2	1.94	-0.48	
$Ni_2Pd_2$	-12.45	2	2.05	-0.44	
$Ni_3Pd_1$	-11.79	2	2.09	-0.10	
$Ni_4$	-11.37	2	2.20	0.00	-0.36
$Pd_5$	-16.46	2	1.82	0.00	-0.35
$\mathrm{Ni_1Pd_4}$	-16.71	2	2.03	-0.44	
$Ni_2Pd_3$	-16.47	2	2.15	-0.40	
$Ni_3Pd_2$	-16.44	2	2.31	-0.56	
$\mathrm{Ni}_4\mathrm{Pd}_1$	-16.01	2	2.39	-0.33	
$Ni_5$	-15.49	4	2.45	0.00	-0.33
$Pd_6$	-20.62	2	1.96	0.00	0.46
$Ni_1Pd_5$	-20.70	2	2.11	-0.19	
$Ni_2Pd_4$	-20.71	2	2.25	-0.32	
$Ni_3Pd_3$	-20.56	2	2.37	-0.28	
$Ni_4Pd_2$	-20.30	4	2.46	-0.13	
$Ni_5Pd_1$	-19.92	2	2.54	0.14	
$Ni_6$	-19.94	4	2.68	0.00	0.27
$Pd_7$	-24.31	2	2.00	0.00	-0.42
$Ni_1Pd_6$	-24.59	2	2.16	-0.31	
$Ni_2Pd_5$	-24.78	2	2.30	-0.52	
$Ni_3Pd_4$	-24.97	4	2.45	-0.74	
$Ni_4Pd_3$	-24.78	2	2.54	-0.58	
$Ni_5Pd_2$	-24.63	2	2.64	-0.45	
$Ni_6Pd_1$	-24.41	6	2.72	-0.26	
$Ni_7$	-24.13	6	2.80	0.00	-0.04
$Pd_8$	-28.43	4	2.08	0.00	-0.18
$\rm Ni_1Pd_7$	-28.78	2	2.23	-0.36	
$Ni_2Pd_6$	-29.24	2	2.39	-0.82	
$Ni_3Pd_5$	-29.27	2	2.50	-0.87	
$Ni_4Pd_4$	-29.30	4	2.60	-0.90	
$Ni_5Pd_3$	-29.11	4	2.68	-0.73	
$Ni_6Pd_2$	-29.00	6	2.77	-0.63	
$\rm Ni_7Pd_1$	-28.74	6	2.84	-0.38	
Ni <sub>8</sub>	-28.35	8	2.90	0.00	-0.14

Table S2: Total energies (taken from MEGA-VASP calculations), spin magnetic moments (SMM), binding energies  $(E_b)$  and excess energies  $(\Delta)$  for 9- and 10atom Ni, Pd and Ni-Pd clusters. Second differences in energy  $(\Delta_2 E)$  are also given for pure Ni and Pd clusters. \*Indicates that the spin magnetic moment is quasi-degenerate in energy with the quintet (SMM=4)

Composition	E / eV	SMM / $\mu_{ m B}$	$E_b / \mathrm{eV}$	$\Delta$ / eV	$\Delta_2 E / \text{eV}$
Pd <sub>9</sub>	-32.72	$2^{*}$	2.16	0.00	0.20
$Ni_1Pd_8$	-33.18	2	2.31	-0.46	
$Ni_2Pd_7$	-33.53	4	2.44	-0.81	
$\rm Ni_3Pd_6$	-33.85	2	2.56	-1.13	
$Ni_4Pd_5$	-33.67	2	2.64	-0.95	
$Ni_5Pd_4$	-33.70	2	2.73	-0.98	
$\rm Ni_6Pd_3$	-33.75	2	2.83	-1.03	
$\rm Ni_7Pd_2$	-33.50	6	2.89	-0.78	
$Ni_8Pd_1$	-33.18	4	2.95	-0.46	
$Ni_9$	-32.72	8	2.99	0.00	-0.12
Pd <sub>10</sub>	-36.82	4	2.21	0.00	
$\rm Ni_1Pd_9$	-37.27	2	2.34	-0.41	
$Ni_2Pd_8$	-37.79	4	2.47	-0.89	
$Ni_3Pd_7$	-38.12	6	2.59	-1.18	
$Ni_4Pd_6$	-38.33	4	2.69	-1.36	
$Ni_5Pd_5$	-38.45	4	2.78	-1.43	
$Ni_6Pd_4$	-38.65	8	2.89	-1.60	
$Ni_7Pd_3$	-38.36	4	2.94	-1.27	
$Ni_8Pd_2$	-38.02	2	2.99	-0.89	
$\rm Ni_9Pd_1$	-37.64	4	3.03	-0.47	
Ni <sub>10</sub>	-37.21	8	3.07	0.00	

Cluster	$\Delta_{\rm HL}$	I / eV	A / eV	$\chi$ / eV	$\mu$ / eV	$\eta$ / eV	$\omega$ / eV
N = 1	I	,	r		· ,	• /	,
Ni <sub>1</sub>	0.06	3.78	3.72	3.75	-3.75	0.03	218.63
Pd <sub>1</sub>	0.67	3.88	3.21	3.54	-3.54	0.33	18.76
N = 2	1	1			1		1
Ni <sub>2</sub>	0.21	3.641	3.43	3.54	-3.54	0.10	59.96
Ni <sub>1</sub> Pd <sub>1</sub>	0.04	4.65	4.62	4.64	-4.4	0.02	613.81
Pd <sub>2</sub>	0.37	4.42	4.05	4.23	-4.23	0.19	48.49
N = 3	1				1		
Ni <sub>3</sub>	0.14	3.198	3.06	3.13	-3.13	0.07	70.93
Ni <sub>2</sub> Pd <sub>1</sub>	0.30	3.51	3.21	3.36	-3.36	0.15	37.64
Ni <sub>1</sub> Pd <sub>2</sub>	0.19	4.17	3.98	4.07	-4.07	0.10	85.68
Pd <sub>3</sub>	0.14	4.57	4.43	4.50	-4.50	0.07	142.24
N = 4	I	<u> </u>	1	<u> </u>	I	<u> </u>	1
Ni <sub>4</sub>	0.56	3.65	3.10	3.37	-3.37	0.28	20.483
Ni <sub>3</sub> Pd <sub>1</sub>	0.05	2.87	2.82	2.85	-2.85	0.02	166.72
Ni <sub>2</sub> Pd <sub>2</sub>	0.28	3.07	2.79	2.93	-2.93	0.14	30.70
Ni <sub>1</sub> Pd <sub>3</sub>	0.28	3.40	3.12	3.26	-3.26	0.14	38.01
Pd <sub>4</sub>	0.04	3.61	3.57	3.59	-3.59	0.02	326.75
N = 5		1	1			1	1
Ni <sub>5</sub>	0.24	3.43	3.19	3.31	-3.31	0.12	45.44
Ni <sub>4</sub> Pd <sub>1</sub>	0.41	3.64	3.23	3.43	-3.43	0.20	28.88
Ni <sub>3</sub> Pd <sub>2</sub>	0.62	3.87	3.25	3.56	-3.56	0.31	20.41
Ni <sub>2</sub> Pd <sub>3</sub>	0.43	4.10	3.67	3.89	-3.89	0.22	34.77
$Ni_1Pd_4$	0.32	3.47	3.15	3.31	-3.31	0.16	34.25
Pd <sub>5</sub>	0.07	3.53	3.46	3.50	-3.50	0.04	165.47
N = 6	<u> </u>	1				1	1
Ni <sub>6</sub>	0.11	3.863	3.75	3.81	-3.81	0.06	128.10
Ni <sub>5</sub> Pd <sub>1</sub>	0.10	3.89	3.78	3.83	-3.83	0.05	141.97
Ni <sub>4</sub> Pd <sub>2</sub>	0.23	3.63	3.40	3.52	-3.52	0.12	53.32
Ni <sub>3</sub> Pd <sub>3</sub>	0.28	3.69	3.41	3.55	-3.55	0.14	45.04
Ni <sub>2</sub> Pd <sub>4</sub>	0.41	3.91	3.51	3.71	-3.71	0.20	33.84
Ni <sub>1</sub> Pd <sub>5</sub>	0.23	3.39	3.16	3.27	-3.27	0.12	46.31
Pd <sub>6</sub>	0.11	3.39	3.28	3.34	-3.34	0.06	100.51

Table S3: Calculated quantum molecular descriptors for free atoms, dimers and clusters,  ${\cal N}=3-6$ 

Cluster	$\Delta_{\rm HL}$	I / eV	A / eV	$\chi$ / eV	$\mu$ / eV	$\eta$ / eV	$\omega$ / eV
N = 7	1	, ,	,	,	. ,	• ,	,
Ni <sub>7</sub>	0.13	3.89	3.76	3.82	-3.82	0.07	111.24
$Ni_6Pd_1$	0.41	3.64	3.23	3.43	-3.43	0.20	28.88
Ni <sub>5</sub> Pd <sub>2</sub>	0.23	3.76	3.54	3.65	-3.65	0.12	58.01
Ni <sub>4</sub> Pd <sub>3</sub>	0.23	3.90	3.67	3.79	-3.79	0.12	60.77
Ni <sub>3</sub> Pd <sub>4</sub>	0.13	3.88	3.75	3.82	-3.82	0.07	110.56
$Ni_2Pd_5$	0.20	3.88	3.68	3.78	-3.78	0.10	71.94
$Ni_1Pd_6$	0.10	3.81	3.71	3.76	-3.76	0.05	141.29
Pd <sub>7</sub>	0.09	3.63	3.54	3.59	-3.59	0.04	145.73
N = 8							
Ni <sub>8</sub>	0.10	3.69	3.59	3.64	-3.64	0.05	133.97
$Ni_7Pd_1$	0.10	3.89	3.78	3.83	-3.83	0.05	141.97
$Ni_6Pd_2$	0.13	4.08	3.95	4.02	-4.02	0.07	119.56
Ni <sub>5</sub> Pd <sub>3</sub>	0.14	3.83	3.69	3.76	-3.76	0.07	101.95
$Ni_4Pd_4$	0.26	3.67	3.42	3.54	-3.54	0.13	49.10
Ni <sub>3</sub> Pd <sub>5</sub>	0.24	3.65	3.41	3.53	-3.53	0.12	52.23
$Ni_2Pd_6$	0.24	3.80	3.56	3.68	-3.68	0.12	56.34
$Ni_1Pd_7$	0.06	3.73	3.66	3.70	-3.70	0.03	221.26
$Pd_8$	0.15	3.68	3.53	3.61	-3.61	0.08	86.00
N = 9							
$Ni_9$	0.05	3.52	3.47	3.50	-3.50	0.03	237.08
$\mathrm{Ni}_8\mathrm{Pd}_1$	0.09	3.79	3.70	3.74	-3.74	0.04	158.81
$\operatorname{Ni_7Pd_2}$	0.07	3.81	3.74	3.78	-3.78	0.03	215.37
$Ni_6Pd_3$	0.07	3.96	3.89	3.93	-3.93	0.04	210.46
$Ni_5Pd_4$	0.10	3.96	3.86	3.91	-3.91	0.05	153.18
$Ni_4Pd_5$	0.10	3.80	3.69	3.74	-3.74	0.05	139.22
Ni <sub>3</sub> Pd <sub>6</sub>	0.24	3.67	3.44	3.55	-3.55	0.12	53.62
$Ni_2Pd_7$	0.06	3.51	3.45	3.48	-3.48	0.03	201.26
$Ni_1Pd_8$	0.28	3.96	3.68	3.82	-3.82	0.14	52.28
Pd <sub>9</sub>	0.08	3.84	3.75	3.79	-3.79	0.04	170.69
N = 10	ſ	I			1	1	
Ni <sub>10</sub>	0.30	3.60	3.30	3.45	-3.45	0.15	39.06
$Ni_9Pd_1$	0.27	3.66	3.39	3.52	-3.52	0.14	45.22
$Ni_8Pd_2$	0.26	3.78	3.52	3.65	-3.65	0.13	51.43
Ni <sub>7</sub> Pd <sub>3</sub>	0.26	3.95	3.69	3.82	-3.82	0.13	57.22
$Ni_6Pd_4$	0.27	3.93	3.65	3.79	-3.79	0.14	51.90
Ni <sub>5</sub> Pd <sub>5</sub>	0.26	3.89	3.63	3.76	-3.76	0.13	54.16
$Ni_4Pd_6$	0.19	3.87	3.67	3.77	-3.77	0.10	73.42
Ni <sub>3</sub> Pd <sub>7</sub>	0.12	3.86	3.75	3.80	-3.80	0.06	124.79
Ni <sub>2</sub> Pd <sub>8</sub>	0.06	3.84	3.78	3.81	-3.81	0.03	252.25
Ni <sub>1</sub> Pd <sub>9</sub>	0.10	3.88	3.78	3.83	-3.83	0.05	139.78
$Pd_{10}$	0.11	3.89	3.78	3.84	-3.84	0.06	134.96

Table S4: Calculated quantum molecular descriptors for free clusters, N = 7 - 10

Table S5: Bader charge distributions calculated for pure, monosubstituted and 50:50 composition for Ni-Pd nanoalloys, N=m+n=2-4

Composition	Numbering of	Type of	Bader	Structuro
Composition	the atoms	atom	charge	Sunctine
Nie	1	Ni	-0.05	
1112	2	Ni	0.05	
Ni Pd	1	Ni	0.06	
	2	Pd	-0.06	
Dd	1	Pd	-0.01	
1 02	2	Pd	0.01	•
	1	Ni	0.001	
Ni <sub>3</sub>	2	Ni	0.002	
	3	Ni	-0.003	2 3
	1	Ni	0.10	1
$Ni_2Pd_1$	2	Ni	0.09	
	1	Pd	-0.19	
	1	Ni	0.23	
$Ni_1Pd_2$	1	Pd	-0.12	<b>X</b>
	2	Pd	-0.10	
	1	Pd	-0.001	1
$Pd_3$	2	Pd	-0.002	
	3	Pd	0.003	2-3
	1	Ni	-0.01	
N;	2	Ni	0.004	
1114	3	Ni	-0.001	2 3
	4	Ni	0.005	
	1	Ni	0.02	•
N; DJ	2	Ni	0.10	
M <sub>3</sub> r u <sub>1</sub>	3	Ni	0.10	
	1	Pd	-0.22	
	1	Ni	0.17	2
N; Dd	2	Ni	0.17	
$M_2 \Gamma u_2$	1	Pd	-0.17	
	2	Pd	-0.17	
	1	Ni	0.26	
Ni Da	1	Pd	-0.09	
$1N1_1Pd_3$	2	Pd	-0.09	3
	3	Pd	-0.08	
	1	Pd	0.002	
L LA	2	Pd	-0.003	
1 4	3	Pd	0.001	
	4	Pd	-0.0002	

Composition	Numbering of	Type of	$\operatorname{Bader}$	Structuro
	the atoms	atom	$\operatorname{charge}$	
	1	Ni	0.09	4
	2	Ni	0.09	
Ni <sub>5</sub>	3	Ni	0.09	
	4	Ni	-0.13	
	5	Ni	-0.13	
	1	Ni	0.14	
	2	Ni	0.14	<b>A</b>
$Ni_4Pd_1$	3	Ni	0.15	2 1
	4	Ni	-0.12	
	1	Pd	-0.31	
	1	Ni	0.21	
	2	Ni	0.21	<u> </u>
$Ni_3Pd_2$	3	Ni	0.21	3
	1	Pd	-0.32	
	2	Pd	-0.32	
	1	Ni	0.27	
	2	Ni	0.27	<b>♀</b>
$Ni_2Pd_3$	1	Pd	-0.04	
	2	Pd	-0.25	
	3	Pd	-0.25	2
	1	Ni	0.30	_
	1	Pd	-0.05	<b>₽</b>
$Ni_1Pd_4$	2	Pd	-0.05	
	3	Pd	-0.10	
	4	Pd	-0.10	
	1	Pd	0.09	•
	2	Pd	0.09	
$\mathrm{Pd}_5$	3	Pd	0.09	
	4	Pd	-0.14	
	5	Pd	-0.14	

Table S6: Bader charge distributions calculated for pure, monosubstituted and 50:50 composition for Ni-Pd nanoalloys, N = m + n = 5

Composition	Numbering of	Type of	Bader	Structuro
	the atoms	atom	charge	Duraciare
	1	Ni	-0.001	
	2	Ni	-0.002	
N;	3	Ni	0.002	2 6 4
1116	4	Ni	-0.001	
	5	Ni	-0.001	3
	6	Ni	0.002	
	1	Ni	0.17	
	2	Ni	0.17	
N; Dd	3	Ni	-0.07	
$1151 u_1$	4	Ni	0.05	
	5	Ni	-0.03	
	1	Pd	-0.29	
	1	Ni	0.25	
	2	Ni	0.25	
N; DJ	3	Ni	0.15	
113F U3	1	Pd	-0.22	
	2	Pd	-0.17	
	3	Pd	-0.26	
	1	Ni	0.28	
	1	Pd	-0.06	
N: DJ	2	Pd	-0.07	
$M_1Pu_5$	3	Pd	-0.06	
	4	Pd	-0.07	5
	5	Pd	-0.02	
	1	Pd	-0.01	
	2	Pd	-0.01	
ЪЧ	3	Pd	-0.01	
Γu <sub>6</sub>	4	Pd	-0.01	
	5	Pd	0.02	3
	6	Pd	0.02	

Table S7: Bader charge distributions calculated for pure, monosubstituted and 50:50 composition for Ni-Pd nanoalloys, N=m+n=6

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Composition	Numbering of	Type of	Bader	Structure
	the atoms	atom	charge	Structure
	1	Ni	0.04	
	2	Ni	0.03	
	3	Ni	0.04	
Ni <sub>7</sub>	4	Ni	-0.01	4 7 1 3
	5	Ni	-0.06	
	6	Ni	-0.02	
	7	Ni	-0.01	
	1	Ni	0.08	
	2	Ni	0.10	
	3	Ni	0.11	
Ni <sub>6</sub> Pd <sub>1</sub>	4	Ni	0.00	6 5 2
	5	Ni	-0.02	
	6	Ni	-0.01	
	1	Pd	-0.26	
	1	Ni	0.26	
	2	Ni	0.26	1
	3	Ni	0.08	
$Ni_4Pd_3$	4	Ni	0.09	
	1	Pd	-0.25	
	2	Pd	-0.22	
	3	Pd	-0.22	
	1	Ni	0.25	
	2	Ni	0.25	4
	3	Ni	0.25	
Ni <sub>3</sub> Pd <sub>4</sub>	1	Pd	-0.13	3 3 2
	2	Pd	-0.14	
-	3	Pd	-0.14	<b>4</b>
	4	Pd	-0.34	
	1	Ni	0.35	
	1	Pd	0.004	
	2	Pd	-0.02	
Ni <sub>1</sub> Pd <sub>6</sub>	3	Pd	0.002	6521
	4	Pd	-0.07	
	5	Pd	-0.07	
	6	Pd Pd	-0.19	

Table S8: Bader charge distributions calculated for pure, monosubstituted and 50:50 composition for Ni-Pd nanoalloys, N=m+n=7

Table S9: Bader charge distributions calculated for pure, monosubstituted and 50:50 composition for Ni-Pd nanoalloys, N = m + n = 7 - 8

Composition	Numbering of	Type of	Bader	Structuro
Composition	the atoms	$\operatorname{atom}$	charge	Structure
	1	Pd	0.05	
	2	Pd	0.03	
	3	Pd	0.05	
Pd <sub>7</sub>	4	Pd	-0.06	0 3 2
	5	Pd	-0.04	
	6	Pd	-0.02	6
	7	Pd	-0.02	
	1	Ni	0.03	
	2	Ni	0.03	
	3	Ni	0.04	7 3
Nie	4	Ni	0.03	
1418	5	Ni	-0.03	
	6	Ni	-0.04	
	7	Ni	-0.03	
	8	Ni	-0.03	
	1	Ni	0.09	
	2	Ni	0.09	
	3	Ni	-0.03	
Ni <sub>7</sub> Pd <sub>1</sub>	4	Ni	0.02	4-5
	5	Ni	0.03	
	6	Ni	-0.03	
	7	Ni	0.08	_
	1	Pd	-0.25	
	1	Ni	0.22	_
	2	Ni	0.21	
	3	Ni	0.21	
Ni₄Pd₄	4	Ni	0.22	
1141 044	1	Pd	-0.22	
	2	Pd	-0.22	
	3	Pd	-0.22	
	4	Pd	-0.22	
	1	Ni	0.35	_
	1	Pd	-0.05	
	2	Pd	-0.02	
Ni <sub>1</sub> Pd <sub>7</sub>	3	Pd	-0.03	
	4	Pd	-0.10	
	5	Pd	0.04	
	6	Pd	-0.10	
	7	Pd	-0.09	

Composition	Numbering of	Type of	Bader	Structuro
Composition	the atoms	atom	charge	Structure
	1	Pd	0.03	
	2	Pd	-0.03	
	3	Pd	0.03	6
Pd.	4	Pd	-0.03	
1 48	5	Pd	-0.03	
	6	Pd	-0.03	
	7	Pd	0.03	
	8	Pd	0.04	
	1	Ni	0.02	
	2	Ni	0.02	
	3	Ni	0.03	3
	4	Ni	0.03	
Ni <sub>9</sub>	5	Ni	0.03	2 6
	6	Ni	0.03	
	7	Ni	-0.06	9
	8	Ni	-0.04	
	9	Ni	-0.05	
	1	Ni	0.08	
	2	Ni	0.08	
	3	Ni	0.02	3
	4	Ni	0.08	
Ni <sub>8</sub> Pd <sub>1</sub>	5	Ni	0.08	
	6	Ni	-0.03	
	7	Ni	-0.04	•
	8	Ni	0.02	
	1	Pd	-0.28	
	1	Ni	0.13	
	2	Ni	0.22	
	3	Ni	0.12	
	4	Ni	0.20	2
Ni <sub>5</sub> Pd <sub>4</sub>	5	Ni	0.20	
	1	Pd	-0.16	
	2	Pd	-0.23	
	3	Pd	-0.22	
	4	Pd	-0.27	

Table S10: Bader charge distributions calculated for pure, monosubstituted and 50:50 composition for Ni-Pd nanoalloys, N = m + n = 8 - 9

Table S11: Bader charge distributions calculated for pure, monosubstituted and 50:50 composition for Ni-Pd nanoalloys, N=m+n=9-10

Composition	Numbering of	Type of	Bader	Ctrusture
Composition	the atoms	atom	charge	Structure
	1	Ni	0.27	
	2	Ni	0.24	
	3	Ni	0.19	
	4	Ni	0.20	
$Ni_4Pd_5$	1	Pd	-0.13	
	2	Pd	-0.21	
	3	Pd	-0.20	<b>9</b>
	4	Pd	-0.15	
	5	Pd	-0.19	
	1	Ni	0.37	
	1	Pd	0.01	6
	2	Pd	0.003	8
	3	Pd	-0.04	6 6
$Ni_1Pd_8$	4	Pd	-0.08	
	5	Pd	-0.04	
	6	Pd	-0.08	
	7	Pd	-0.07	
	8	Pd	-0.08	
	1	Pd	0.02	
	2	Pd	0.02	
	3	Pd	-0.008	
	4	Pd	-0.01	
$Pd_9$	5	Pd	-0.01	
	6	Pd	-0.007	
	7	Pd	0.02	
	8	Pd	-0.01	
	9	Pd	-0.01	
	1	Ni	0.08	
	2	Ni	0.08	_
	3	Ni	0.08	- 8
	4	Ni	0.08	5
NI:	5	Ni	0.08	
<sup>1NI</sup> 10	6	Ni	-0.12	
	7	Ni	0.07	
	8	Ni	-0.12	
	9	Ni	-0.12	
	10	Ni	-0.12	

Table S12: Bader charge distributions calculated for pure, monosubstituted and 50:50 composition for Ni-Pd nanoalloys, N = m + n = 10

Composition	Numbering of	Type of	Bader	Structure
	the atoms	atom	charge	Structure
	1	Ni	0.13	
	2	Ni	0.14	
	3	Ni	0.14	
	4	Ni	0.08	
Ni-Pd.	5	Ni	0.08	
	6	Ni	0.08	
	7	Ni	-0.10	
	8	Ni	-0.11	
	9	Ni	-0.11	
	1	Pd	-0.33	
	1	Ni	0.22	
	2	Ni	0.22	
	3	Ni	0.21	5
	4	Ni	0.17	
N; Dd	5	Ni	0.17	
1151 U5	1	Pd	-0.17	
	2	Pd	-0.17	3-3-2
	3	Pd	-0.17	
	4	Pd	-0.17	
	5	Pd	-0.29	
	1	Ni	0.41	
	1	Pd	-0.07	
	2	Pd	-0.002	
	3	Pd	-0.07	82 50
Ni Da	4	Pd	0.007	
1111 119	5	Pd	0.003	
	6	Pd	-0.07	
	7	Pd	-0.06	
	8	Pd	-0.07	
	9	Pd	-0.07	
	1	Pd	0.09	
	2	Pd	0.02	
	3	Pd	0.02	6
	4	Pd	0.01	
Dd	5	Pd	-0.004	
1 u <sub>10</sub>	6	Pd	-0.008	
	7	Pd	-0.004	
	8	Pd	-0.04	
	9	Pd	-0.04	
	10	Pd	-0.04	