

Supporting Information for Monte Carlo Second- and Third-Order Many-Body Green's Function Methods with Frequency-Dependent, Non-Diagonal Self-Energy

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1 Integrand of the third-order self-energy

The integrand of the third-order self-energy is

$$f_{pq}^{(3)}(\mathbf{r}_1, \mathbf{r}_2, \mathbf{r}_3, \mathbf{r}_4, \mathbf{r}_5, \mathbf{r}_6, \tau_1, \tau_2, \omega) = \sum_{n=1}^{84} f_{pq}^{(3:n)}(\mathbf{r}_1, \mathbf{r}_2, \mathbf{r}_3, \mathbf{r}_4, \mathbf{r}_5, \mathbf{r}_6, \tau_1, \tau_2, \omega), \quad (1)$$

where

$$f_{pq}^{(3:n)}(\mathbf{r}_1, \mathbf{r}_2, \mathbf{r}_3, \mathbf{r}_4, \mathbf{r}_5, \mathbf{r}_6, \tau_1, \tau_2, \omega) = \frac{\tilde{f}_{pq}^{(3:n)}(\mathbf{r}_1, \mathbf{r}_2, \mathbf{r}_3, \mathbf{r}_4, \mathbf{r}_5, \mathbf{r}_6, \tau_1, \tau_2, \omega)}{r_{12} r_{34} r_{56}} \quad (2)$$

and r_{12} is the distance between \mathbf{r}_1 and \mathbf{r}_2 . The terms, $\tilde{f}_{pq}^{(3:n)}$, with the arguments omitted for brevity are

$$\tilde{f}_{pq}^{(3:1)} = -G^+(\mathbf{r}_2, \mathbf{r}_6, \tau_1 + \tau_2) G^+(\mathbf{r}_4, \mathbf{r}_5, \tau_1) G^-(\mathbf{r}_3, \mathbf{r}_2, -\tau_2) G^-(\mathbf{r}_5, \mathbf{r}_1, -\tau_1 - \tau_2) G^-(\mathbf{r}_6, \mathbf{r}_4, -\tau_1) \phi_p^*(\mathbf{r}_3) \phi_q(\mathbf{r}_1) e^{-\omega \tau_2},$$

$$\bar{f}_{pq}^{(3:78)} = G^+(\mathbf{r}_1, \mathbf{r}_6, \tau_1 + \tau_2)G^+(\mathbf{r}_2, \mathbf{r}_4, \tau_2)G^+(\mathbf{r}_3, \mathbf{r}_5, \tau_1)G^-(\mathbf{r}_5, \mathbf{r}_1, -\tau_1 - \tau_2)G^-(\mathbf{r}_6, \mathbf{r}_2, -\tau_1 - \tau_2)\phi_p^*(\mathbf{r}_3)\phi_q(\mathbf{r}_4),$$

$$\bar{f}_{pq}^{(3:79)} = 2G^+(\mathbf{r}_1, \mathbf{r}_5, \tau_1 + \tau_2)G^+(\mathbf{r}_2, \mathbf{r}_6, \tau_1 + \tau_2)G^-(\mathbf{r}_3, \mathbf{r}_1, -\tau_2)G^-(\mathbf{r}_5, \mathbf{r}_4, -\tau_1)G^-(\mathbf{r}_6, \mathbf{r}_2, -\tau_1 - \tau_2)\phi_p^*(\mathbf{r}_3)\phi_q(\mathbf{r}_4),$$

$$\bar{f}_{pq}^{(3:80)} = -2G^+(\mathbf{r}_1, \mathbf{r}_4, \tau_2)G^+(\mathbf{r}_2, \mathbf{r}_6, \tau_1 + \tau_2)G^+(\mathbf{r}_3, \mathbf{r}_5, \tau_1)G^-(\mathbf{r}_5, \mathbf{r}_1, -\tau_1 - \tau_2)G^-(\mathbf{r}_6, \mathbf{r}_2, -\tau_1 - \tau_2)\phi_p^*(\mathbf{r}_3)\phi_q(\mathbf{r}_4),$$

$$\bar{f}_{pq}^{(3:81)} = G^+(\mathbf{r}_1, \mathbf{r}_3, \tau_2)G^+(\mathbf{r}_2, \mathbf{r}_4, \tau_2)G^+(\mathbf{r}_4, \mathbf{r}_6, \tau_1)G^-(\mathbf{r}_3, \mathbf{r}_2, -\tau_2)G^-(\mathbf{r}_5, \mathbf{r}_1, -\tau_1 - \tau_2)\phi_p^*(\mathbf{r}_5)\phi_q(\mathbf{r}_6),$$

$$\bar{f}_{pq}^{(3:82)} = -G^+(\mathbf{r}_1, \mathbf{r}_4, \tau_2)G^+(\mathbf{r}_2, \mathbf{r}_6, \tau_1 + \tau_2)G^-(\mathbf{r}_3, \mathbf{r}_1, -\tau_2)G^-(\mathbf{r}_4, \mathbf{r}_2, -\tau_2)G^-(\mathbf{r}_5, \mathbf{r}_3, -\tau_1)\phi_p^*(\mathbf{r}_5)\phi_q(\mathbf{r}_6),$$

$$\bar{f}_{pq}^{(3:83)} = -2G^+(\mathbf{r}_1, \mathbf{r}_3, \tau_2)G^+(\mathbf{r}_2, \mathbf{r}_4, \tau_2)G^+(\mathbf{r}_3, \mathbf{r}_6, \tau_1)G^-(\mathbf{r}_4, \mathbf{r}_2, -\tau_2)G^-(\mathbf{r}_5, \mathbf{r}_1, -\tau_1 - \tau_2)\phi_p^*(\mathbf{r}_5)\phi_q(\mathbf{r}_6),$$

$$\bar{f}_{pq}^{(3:84)} = 2G^+(\mathbf{r}_1, \mathbf{r}_3, \tau_2)G^+(\mathbf{r}_2, \mathbf{r}_6, \tau_1 + \tau_2)G^-(\mathbf{r}_3, \mathbf{r}_1, -\tau_2)G^-(\mathbf{r}_4, \mathbf{r}_2, -\tau_2)G^-(\mathbf{r}_5, \mathbf{r}_4, -\tau_1)\phi_p^*(\mathbf{r}_5)\phi_q(\mathbf{r}_6).$$

2 Data for Table 4 of the Main Text

Table 1: The number of Monte Carlo steps, variances (in E_h^2), and total simulation times (in seconds) of the MC-GF3 calculations used in Table 4.

τ -Integration ^a	Molecule	Formula	n_{ele} ^b	N_{MC} ^c	Time	σ^2 ^d
MC	Methane	CH_4	10	19393536	$1.393 \cdot 10^5$	$8.634 \cdot 10^{-1}$
MC	Propane	C_3H_8	26	1577984	$1.243 \cdot 10^4$	$5.536 \cdot 10^1$
MC	Benzene	C_6H_6	42	17617920	$1.400 \cdot 10^5$	$2.773 \cdot 10^2$
MC	Nonane	C_9H_{20}	74	1571840	$1.239 \cdot 10^4$	$5.189 \cdot 10^3$
MC	Decane	$\text{C}_{10}\text{H}_{22}$	82	1555456	$1.237 \cdot 10^4$	$6.280 \cdot 10^3$
MC	Pentacene	$\text{C}_{22}\text{H}_{14}$	146	17075200	$1.400 \cdot 10^5$	$8.417 \cdot 10^4$
Quadrature	Methane	CH_4	10	40960	$1.295 \cdot 10^5$	$4.442 \cdot 10^{-1}$
Quadrature	Propane	C_3H_8	26	35840	$1.175 \cdot 10^5$	$1.788 \cdot 10^1$
Quadrature	Benzene	C_6H_6	42	38912	$1.293 \cdot 10^5$	$8.336 \cdot 10^1$
Quadrature	Nonane	C_9H_{20}	74	56320	$1.804 \cdot 10^5$	$1.647 \cdot 10^3$
Quadrature	Decane	$\text{C}_{10}\text{H}_{22}$	82	71680	$2.351 \cdot 10^5$	$3.176 \cdot 10^3$
Quadrature	Pentacene	$\text{C}_{22}\text{H}_{14}$	146	56320	$1.879 \cdot 10^5$	$1.173 \cdot 10^4$

^a‘MC’ stands for the direct Monte Carlo integration, while ‘quadrature’ a 21-point Gauss–Kronrod quadrature.

^bThe number of electrons.

^cThe number of Monte Carlo steps.

^dVariance of the diagonal element of the third-order self-energy matrix associated with the HOMO.

3 Data for Figures 5 and 6 of the Main Text

Table 2: The variances (in E_h^2) of the MC-GF3/aug-cc-pVDZ calculations used in Figures 5 and 6.

Molecule	Formula	n_{ele}^a	N_{MC}^b	σ^2 (diagonal) ^c	σ^2 (non-diagonal) ^d
Methane	CH ₄	10	6745088	$8.679 \cdot 10^{-1}$	$1.477 \cdot 10^0$
Acetylene	C ₂ H ₂	14	6673408	$1.066 \cdot 10^0$	$1.417 \cdot 10^0$
Ethylene	C ₂ H ₄	16	6650880	$4.200 \cdot 10^0$	$4.502 \cdot 10^0$
Ethane	C ₂ H ₆	18	5740544	$1.442 \cdot 10^1$	$2.690 \cdot 10^1$
Butadiyne	C ₄ H ₂	26	5728256	$1.088 \cdot 10^1$	$1.722 \cdot 10^1$
Propane	C ₃ H ₈	26	5671936	$5.724 \cdot 10^1$	$6.691 \cdot 10^1$
Butadiene	C ₄ H ₆	30	5680128	$6.190 \cdot 10^1$	$9.359 \cdot 10^1$
Butane	C ₄ H ₁₀	34	5598208	$1.708 \cdot 10^2$	$2.052 \cdot 10^2$
Pentane	C ₅ H ₁₂	42	5531648	$4.223 \cdot 10^2$	$5.330 \cdot 10^2$
Benzene	C ₆ H ₆	42	5657600	$2.570 \cdot 10^2$	$8.888 \cdot 10^3$
Hexatriene	C ₆ H ₈	44	5564416	$4.189 \cdot 10^2$	$1.076 \cdot 10^3$
Hexane	C ₆ H ₁₄	50	6178816	$1.137 \cdot 10^3$	$2.096 \cdot 10^3$
Heptane	C ₇ H ₁₆	58	6102016	$1.796 \cdot 10^3$	$3.306 \cdot 10^3$
Octane	C ₈ H ₁₈	66	5965824	$3.173 \cdot 10^3$	$1.208 \cdot 10^4$
Naphthalene	C ₁₀ H ₈	68	6147072	$2.076 \cdot 10^3$	$7.139 \cdot 10^3$
Nonane	C ₉ H ₂₀	74	5856256	$5.153 \cdot 10^3$	$2.692 \cdot 10^4$
Decane	C ₁₀ H ₂₂	82	5769216	$9.502 \cdot 10^3$	$4.667 \cdot 10^4$
Anthracene	C ₁₄ H ₁₀	94	5892096	$9.582 \cdot 10^3$	$7.081 \cdot 10^4$
Naphthacene	C ₁₈ H ₁₂	120	5569536	$2.501 \cdot 10^4$	$3.293 \cdot 10^5$
Pentacene	C ₂₂ H ₁₄	146	5328896	$1.114 \cdot 10^5$	$7.458 \cdot 10^6$

^aThe number of electrons.

^bThe number of Monte Carlo steps.

^cVariance of the binding energy of the HOMO within the diagonal, frequency-independent approximation.

^dVariance of the binding energy of the HOMO within the frequency-independent approximation.

Table 3: The same as Table 2, but with the aug-cc-pVTZ basis set.

Molecule	Formula	n_{ele}^a	N_{MC}^b	σ^2 (diagonal) ^c	σ^2 (non-diagonal) ^d
Methane	CH ₄	10	5783552	$8.371 \cdot 10^0$	$2.230 \cdot 10^1$
Acetylene	C ₂ H ₂	14	7341056	$1.061 \cdot 10^1$	$1.406 \cdot 10^1$
Ethylene	C ₂ H ₄	16	7264256	$4.010 \cdot 10^1$	$4.812 \cdot 10^1$
Ethane	C ₂ H ₆	18	7151616	$1.006 \cdot 10^2$	$1.224 \cdot 10^2$
Butadiyne	C ₄ H ₂	26	7207936	$6.601 \cdot 10^1$	$3.517 \cdot 10^2$
Propane	C ₃ H ₈	26	6890496	$1.320 \cdot 10^3$	$2.871 \cdot 10^3$
Butadiene	C ₄ H ₆	30	6868992	$6.291 \cdot 10^2$	$1.055 \cdot 10^3$
Butane	C ₄ H ₁₀	34	6649856	$1.724 \cdot 10^3$	$3.744 \cdot 10^3$
n-Pentane	C ₅ H ₁₂	42	6409216	$7.221 \cdot 10^3$	$1.322 \cdot 10^4$
Benzene	C ₆ H ₆	42	5254144	$2.328 \cdot 10^3$	$6.264 \cdot 10^3$
Hexatriene	C ₆ H ₈	44	6598656	$2.745 \cdot 10^3$	$7.579 \cdot 10^3$
Hexane	C ₆ H ₁₄	50	6907904	$9.220 \cdot 10^3$	$1.126 \cdot 10^5$
n-Heptane	C ₇ H ₁₆	58	6601728	$1.434 \cdot 10^4$	$6.014 \cdot 10^4$
n-Octane	C ₈ H ₁₈	66	6304768	$2.988 \cdot 10^4$	$1.753 \cdot 10^5$
Naphthalene	C ₁₀ H ₈	68	6832128	$1.796 \cdot 10^4$	$1.421 \cdot 10^5$
Nonane	C ₉ H ₂₀	74	6037504	$4.470 \cdot 10^4$	$2.403 \cdot 10^5$
n-Decane	C ₁₀ H ₂₂	82	5585920	$5.422 \cdot 10^4$	$6.886 \cdot 10^5$
Anthracene	C ₁₄ H ₁₀	94	6159360	$9.449 \cdot 10^4$	$1.553 \cdot 10^6$
Naphthacene	C ₁₈ H ₁₂	120	5321728	$2.879 \cdot 10^5$	$1.048 \cdot 10^7$
Pentacene	C ₂₂ H ₁₄	146	3533824	$1.147 \cdot 10^6$	$6.014 \cdot 10^7$

^aThe number of electrons.

^bThe number of Monte Carlo steps.

^cVariance of the binding energy of the HOMO within the diagonal, frequency-independent approximation.

^dVariance of the binding energy of the HOMO within the frequency-independent approximation.