

Supporting Information

Photo-assisted hydrothermal synthesis of IrO_x-TiO₂ for enhanced water oxidation

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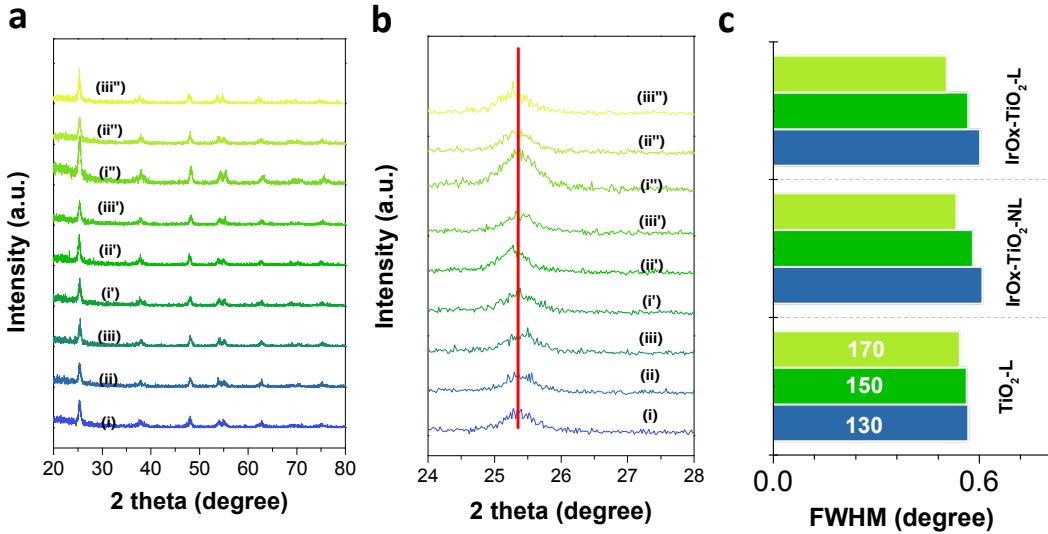


Figure S1. XRD results of the IrOx-TiO₂ samples obtained from different preparation conditions. (a) XRD patterns, i-iii: TiO₂-130-L, TiO₂-150-L, TiO₂-170-L; i'-iii'': IrOx-TiO₂-130-NL, IrOx-TiO₂-150-NL, IrOx-TiO₂-170-NL; i''-iii'': IrOx-TiO₂-130-L, IrOx-TiO₂-150-L, IrOx-TiO₂-170-L. (b) The enlarge region at 2 theta of 24 to 28 degree and (c) the corresponding full width at half maximum (FWHM) values of the peak (101) at 2 theta of 25.3 degree.

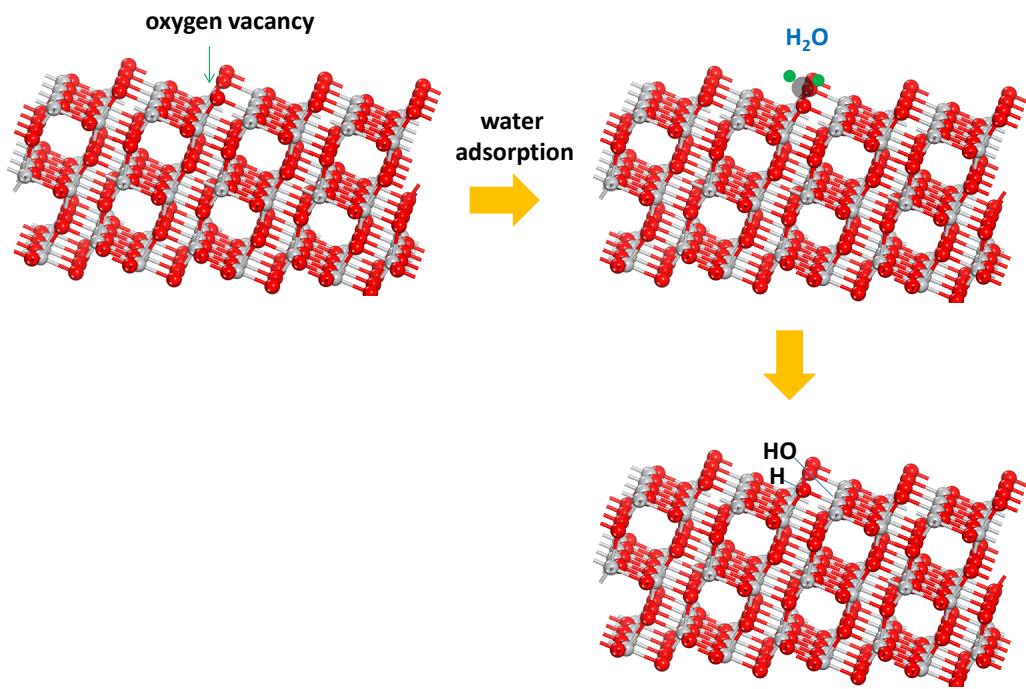


Figure S2. The formation process of two types of surface OH.

Table S1. The relative percentage of the three oxygen-based peaks (%).

Sample	Lattice O	OH group	Adsorbed water
IrOx-TiO₂-130-L	74.869	14.765	10.366
IrOx-TiO₂-150-L	51.891	46.233	1.877
IrOx-TiO₂-170-L	95.644	2.568	1.788

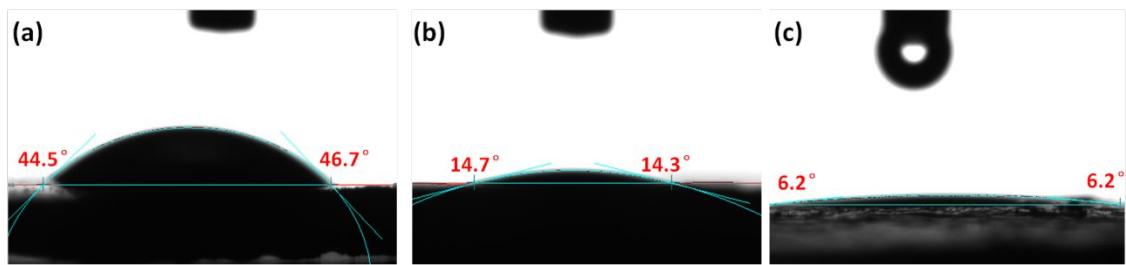


Figure S3. The contact angles of (a) TiO_2 -150-NL, (b) TiO_2 -150-L and (c) IrOx- TiO_2 -150-L.

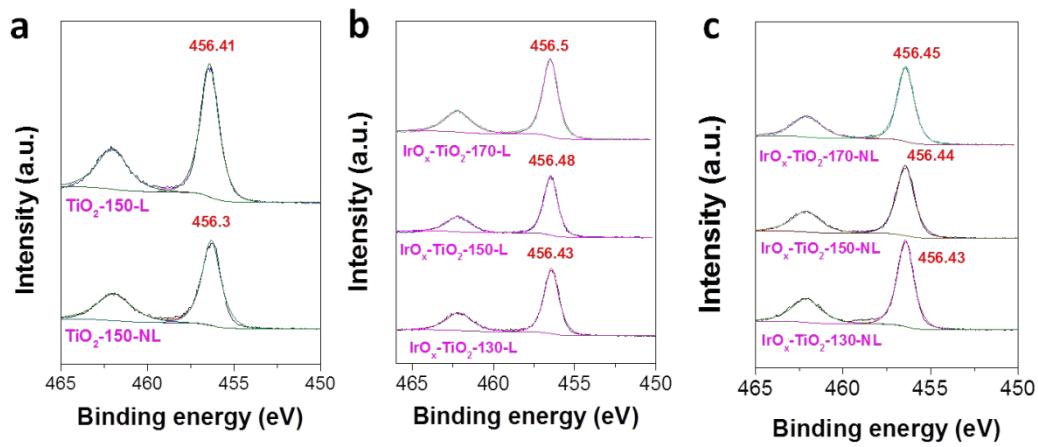


Figure S4. Ti 2p XPS results of the studied samples.

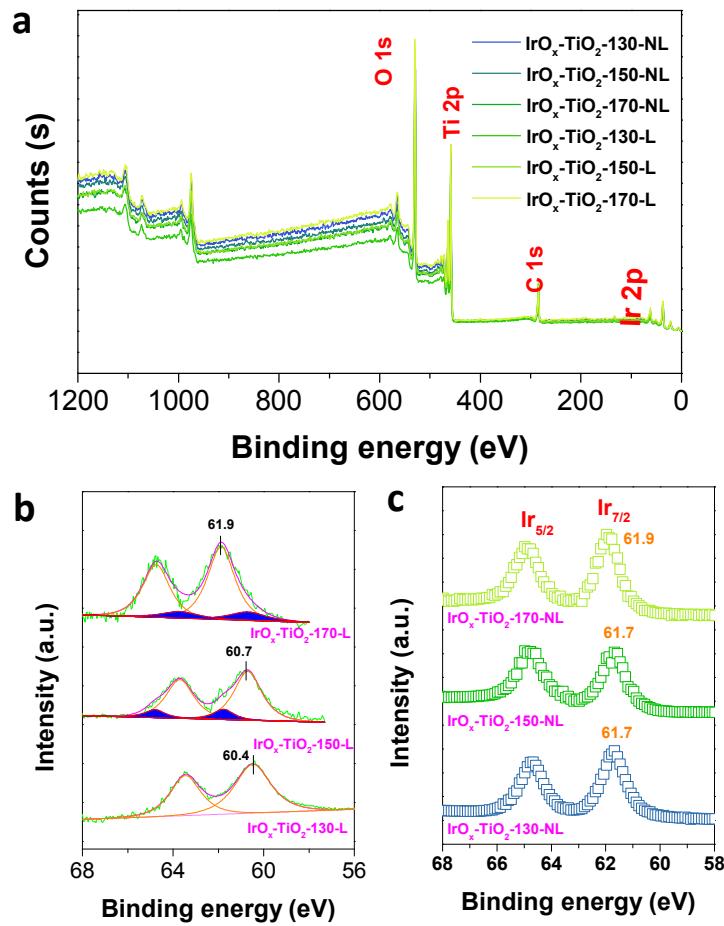


Figure S5. (a) The wide XPS results, (b)-(c) the Ir 4f results of IrO_x-TiO₂ based samples under illumination and dark conditions.

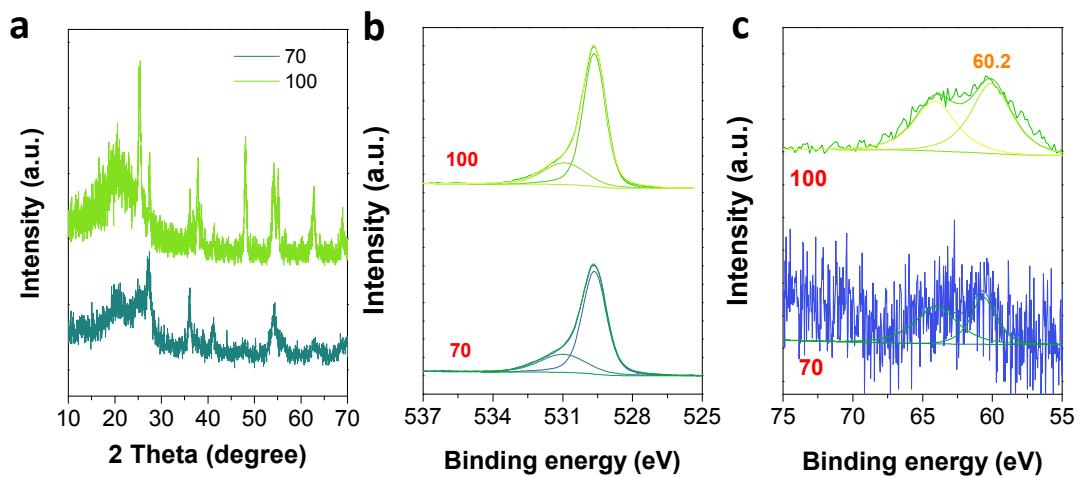


Figure S6. The physical properties of the two samples obtained from 70 and 100 °C. (a) XRD patterns, (b) O 1s XPS and (c) Ir 4f XPS results.

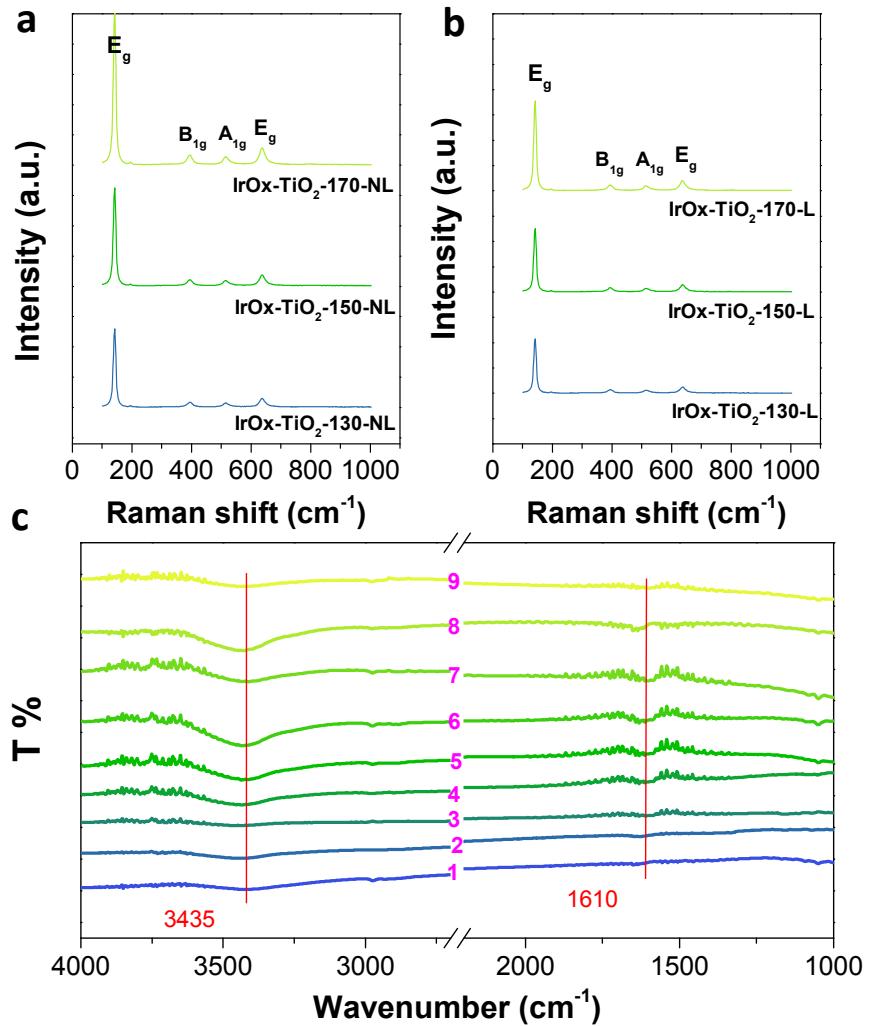


Figure S7. Raman spectra of (a) IrOx-TiO₂-NL samples and (b) IrOx-TiO₂-L samples. (c) FTIR spectrum of the studied samples: 1, TiO₂-170-L; 2, TiO₂-150-L; 3, TiO₂-150-NL; 4-6, IrOx-TiO₂-130-L, IrOx-TiO₂-150-L, IrOx-TiO₂-170-L; 7-9, IrOx-TiO₂-130-NL, IrOx-TiO₂-150-NL, IrOx-TiO₂-170-NL.

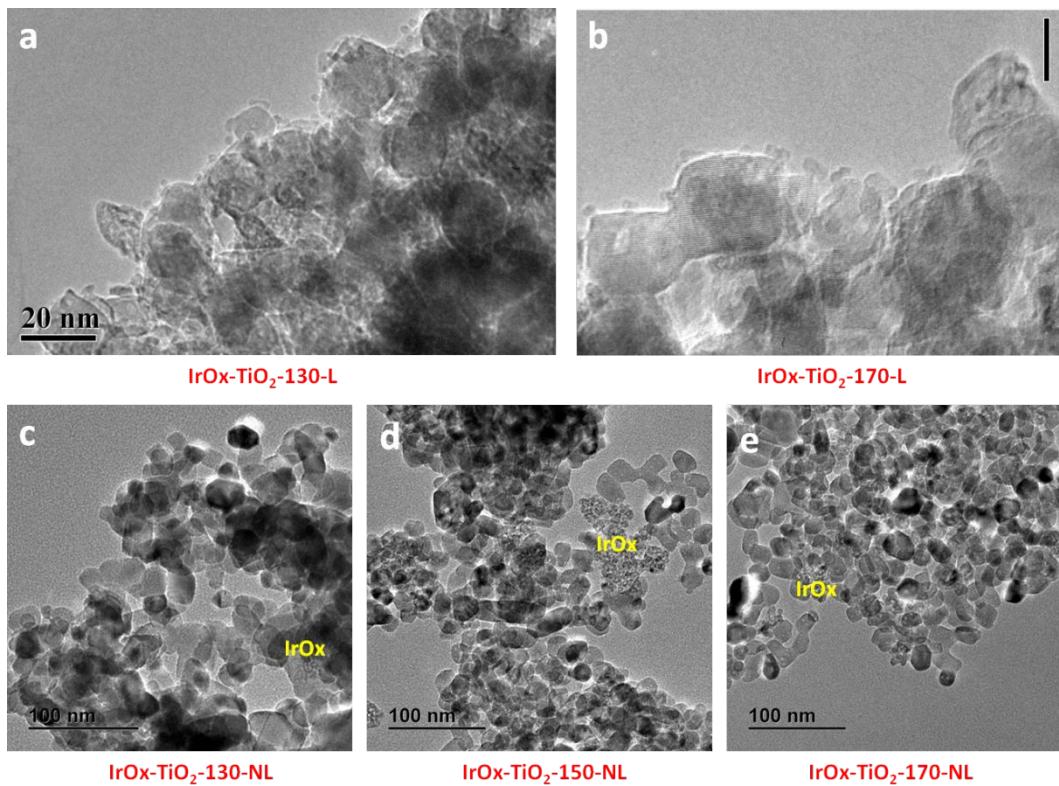


Figure S8. (a)-(b) TEM image of IrOx-TiO₂-130-NL and IrOx-TiO₂-170-L samples, (c)-(e) TEM images of IrOx-TiO₂-130-NL, IrOx-TiO₂-150-NL and IrOx-TiO₂-170-NL.

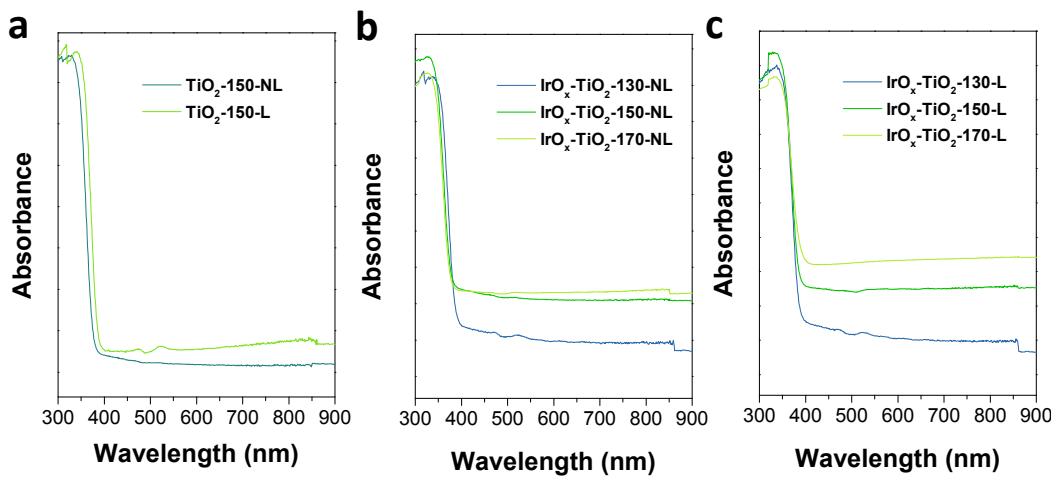


Figure S9. UV-vis absorption spectroscopy, (a) TiO₂-150-NL, TiO₂-150-L; (b) IrO_x-TiO₂-130-NL, IrO_x-TiO₂-150-NL and IrO_x-TiO₂-170-NL; (c) IrO_x-TiO₂-130-L, IrO_x-TiO₂-150-L and IrO_x-TiO₂-170-L.

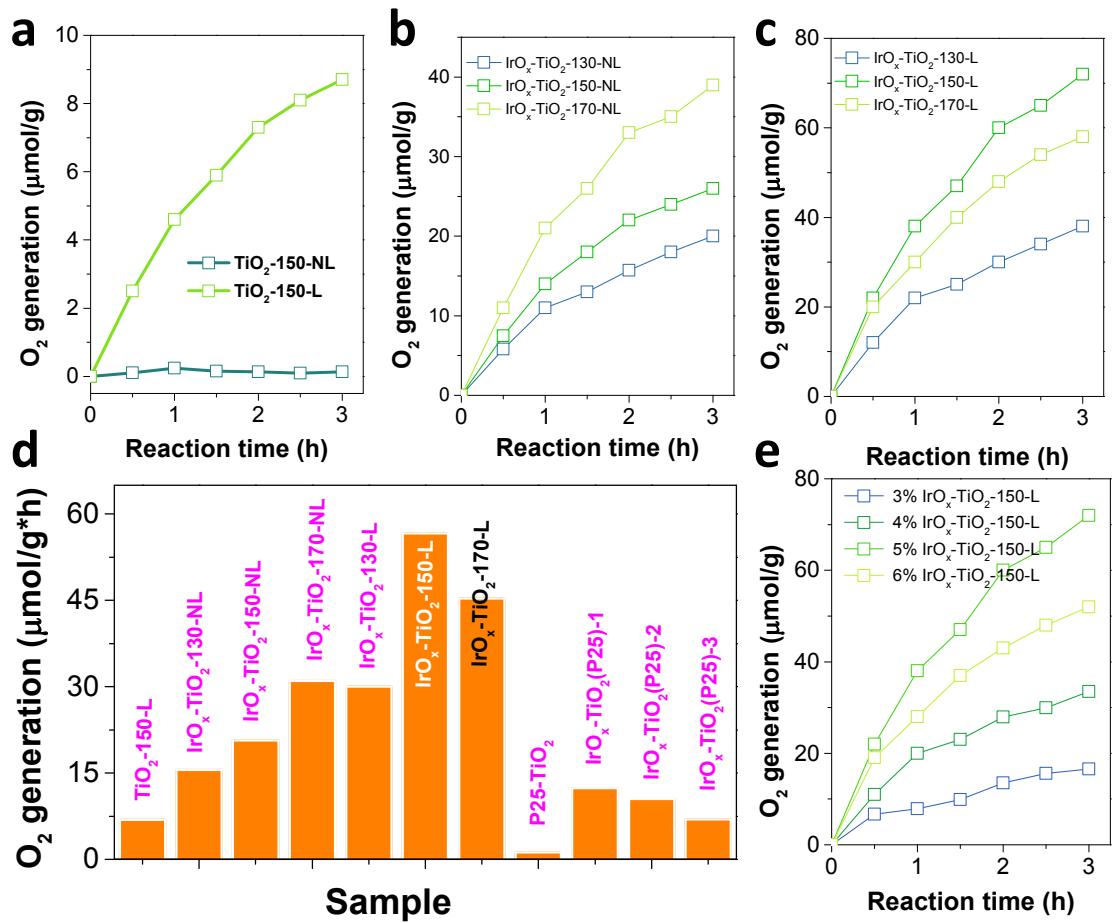


Figure S10. Photocatalytic water oxidation performance. (a) TiO₂-150-NL and TiO₂-150-L; (b) IrO_x-TiO₂-130-NL, IrO_x-TiO₂-150-NL and IrO_x-TiO₂-170-NL; (c) IrO_x-TiO₂-130-L, IrO_x-TiO₂-150-L and IrO_x-TiO₂-170-L; (d) oxygen generate rates of the studied samples and reference commercial P25-TiO₂ based samples; (e) the oxygen generation of IrO_x-TiO₂-150-L based samples with theoretical 7%, 8%, 9% and 10% IrO_x loading. Reaction conditions: 0.1g catalyst, 100 mL water solution, 1g AgNO₃, LED-365 lamp.

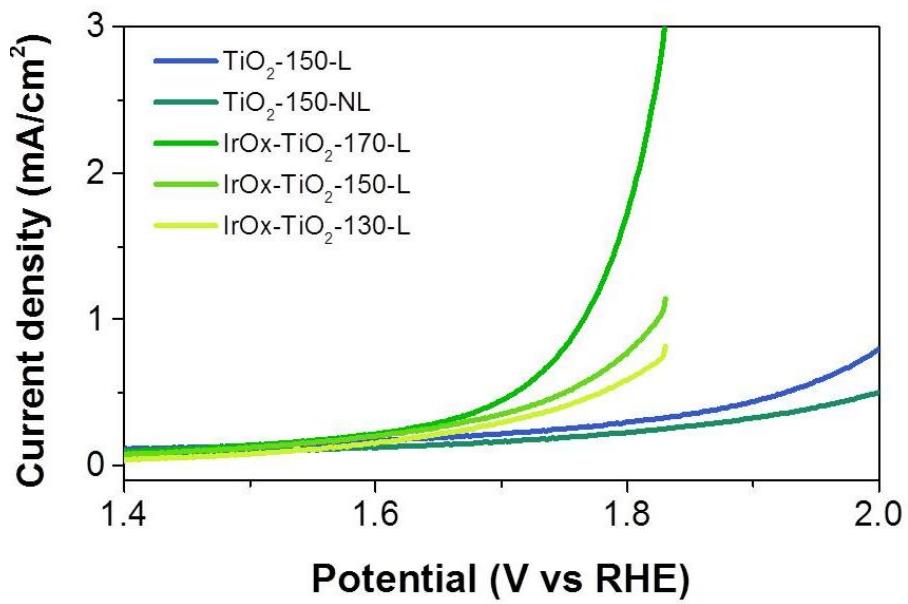


Figure S11. LSV plots of the samples under study. Conditions: 0.5 M Na_2SO_4 , Ag/AgCl and Pt wire are employed as the reference and counter electrodes.

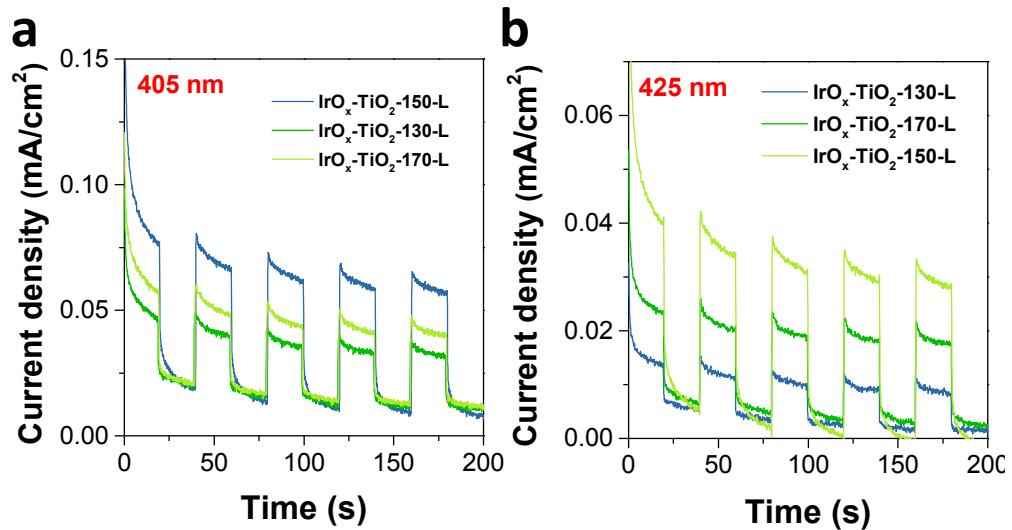


Figure S12. Photocurrent response under LED-405 and LED-425 nm lamps for the samples of $\text{IrO}_x\text{-TiO}_2\text{-130-L}$, $\text{IrO}_x\text{-TiO}_2\text{-150-L}$ and $\text{IrO}_x\text{-TiO}_2\text{-170-L}$, please note that a 1.2 V of bias voltage is used for the measurement.

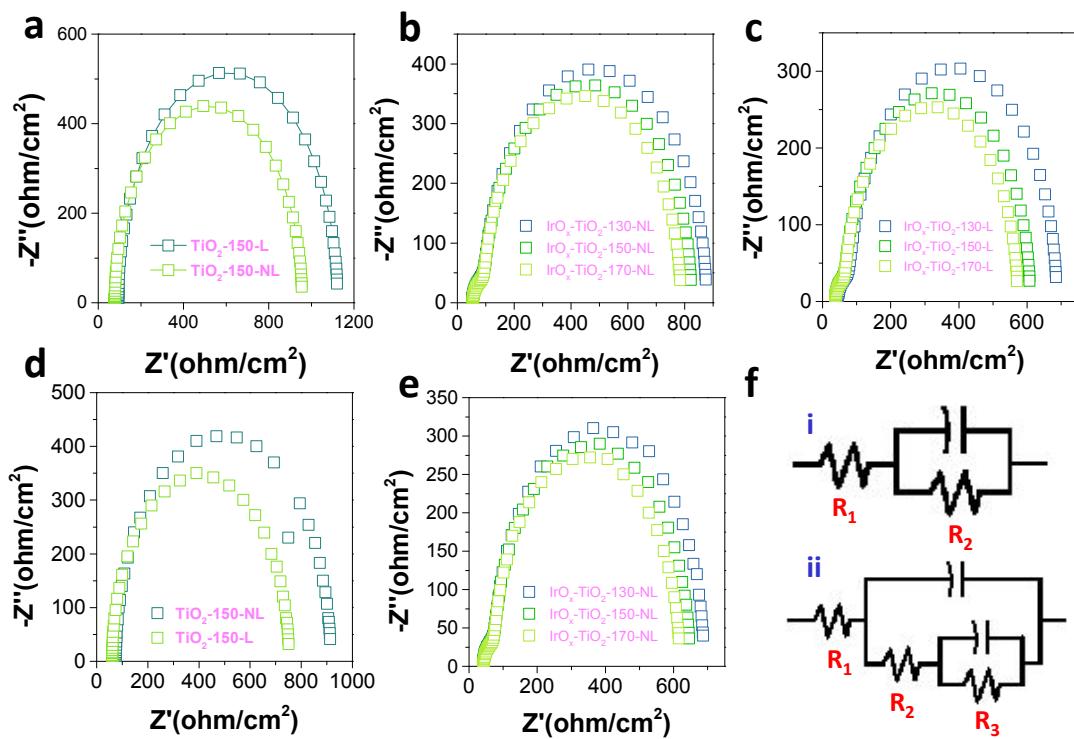


Figure S13. EIS plots. (a) and (d), the results of two samples of TiO_2 -150-L and TiO_2 -150-NL under dark and LED-365 illumination conditions; (b) and (e), the results of three samples of $\text{IrO}_x\text{-TiO}_2$ -130-NL, $\text{IrO}_x\text{-TiO}_2$ -150-NL and $\text{IrO}_x\text{-TiO}_2$ -170-NL under dark and LED-365 illumination conditions; (c) the results of the three samples of $\text{IrO}_x\text{-TiO}_2$ -130-L, $\text{IrO}_x\text{-TiO}_2$ -150-L and $\text{IrO}_x\text{-TiO}_2$ -170-L under dark condition.

Table S2. The fitted values of the corresponding impedance (ohm/cm^2).

Sample	Without illumination			Under illumination		
	R ₁	R ₂	R ₃	R ₁	R ₂	R ₃
TiO ₂ -150-NL	94.9	1029	-	75.3	839.6	-
TiO ₂ -150-L	78.2	879.2	-	60.9	691.4	-
IrO _x -TiO ₂ -130-NL	54.1	73.72	744	41.6	61.9	585.6
IrO _x -TiO ₂ -150-NL	53.4	64.83	696	42.7	55.1	520
IrO _x -TiO ₂ -170-NL	54.8	69.9	661	42.2	58.1	547.4
IrO _x -TiO ₂ -130-L	45.1	61.5	580	29.7	43.6	413.7
IrO _x -TiO ₂ -150-L	37.1	51.2	483.3	27.4	38.2	358.8
IrO _x -TiO ₂ -170-L	37.6	54.6	517	30.1	40.95	386.7

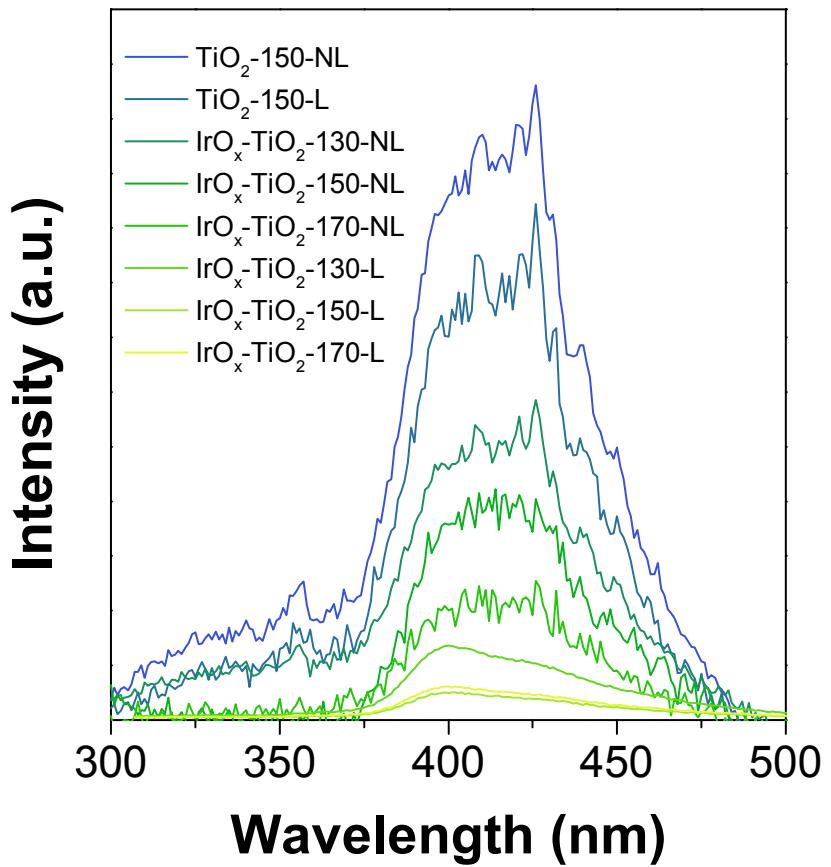


Figure S14. Room-temperature PL spectrum of the studied samples under the 295 nm incident light excitation.

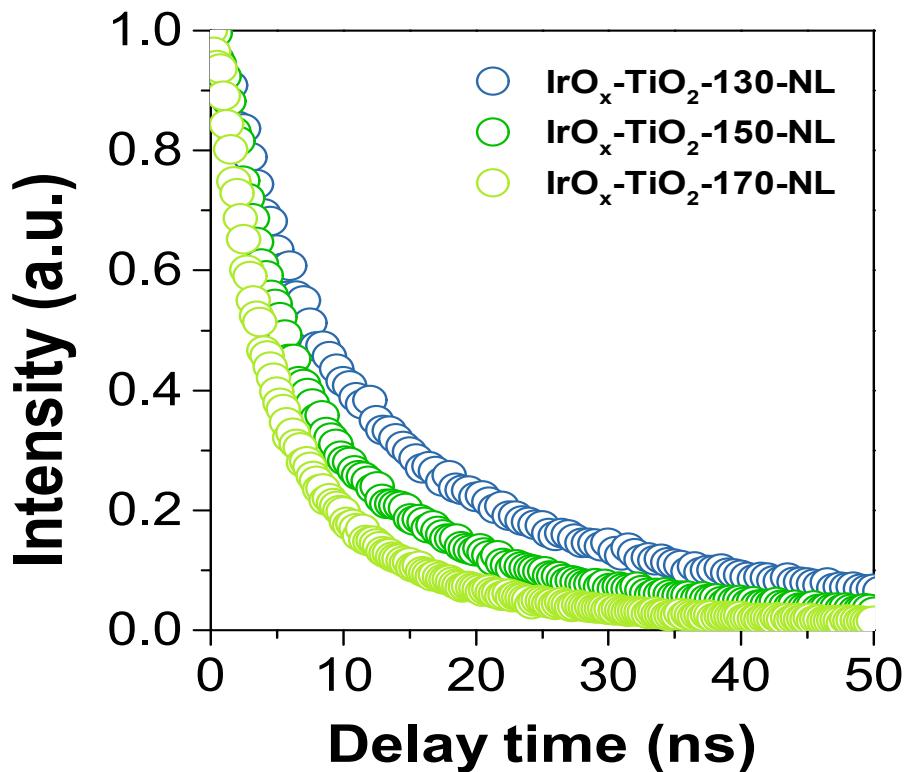


Figure S15. Time resolved fluorescence spectra of IrO_x-TiO₂-NL samples (Excitation: 295 nm, emission: 400 nm).

Table S3. The fitted values of the time parameters and the corresponding percents from the TRPL.

Sample	τ_1 (ns)	A_1 (%)	τ_2 (ns)	A_2 (%)	τ_{av}^* (ns)
TiO₂-150-NL	46.7	16	9.7	84	27.39
TiO₂-150-L	32.8	17.5	5.6	82.5	20.67
IrO_x-TiO₂-130-NL	29.7	28.5	8.1	71.5	20.93
IrO_x-TiO₂-150-NL	27.4	22.5	5.8	77.5	18.29
IrO_x-TiO₂-170-NL	16.4	19.4	4.2	80.6	10.11
IrO_x-TiO₂-130-L	6.7	85.9	33.7	14.1	18.91
IrO_x-TiO₂-150-L	5.2	87.7	24.6	12.3	12.94
IrO_x-TiO₂-170-L	6.3	84.6	23.1	15.4	13.02
$*\tau_{av} = (\tau_1 * \tau_1 * A_1 + \tau_2 * \tau_2 * A_2) / (\tau_1 * A_1 + \tau_2 * A_2)$					