Supporting Information

1-Methyl-1,4-cyclohexadiene as a traceless reducing agent for the synthesis of catechols and hydroquinones

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Table S1: Representative results of the screening of solvents.^[a]

Entry	Solvent	4a Conv. ^[b] [%]
1	Tetrahydrofuran	0
2	Acetonitrile	5
3	Dichloromethane	98
4	Methanol	1
5	Dimethyl sulfoxide	0
6	Hexane	92
7	Cyclohexane	94
8	Toluene	>99

[a]Reaction conditions: **1a** (0.09 mmol), 1-methyl-1,4-cyclohexadiene (3.6 mmol), trifluoroacetic acid (3.6 mmol), solvent (1.0 mL), 30 °C, 24 h. [b]Conversion was determined by ¹H NMR spectroscopy.

Table S2: Representative results of the screening of MeCHD equivalents on product conversion.^[a]

Entry	MeCHD (equiv)	4a Conv. ^[b] [%]
1	0	0
2	2	20
3	5	40
4	10	52
5	20	85
6	30	>99
7	40	>99

[a]Reaction conditions: **1a** (0.09 mmol), trifluoroacetic acid (3.6 mmol), solvent (1.0 mL), 30 °C, 24 h. [b]Conversion was determined by ¹H NMR spectroscopy.

Table S3: MeCHD conversion to toluene in absence or presence of quinone 1a. [a]

Entry	Time (h)	1a	2a Conv. ^[b] [%]	Toluene:MeCHD ^[c] [%]
1	0	no	/	2.47
2	1	no	/	2.29
3	2.5	no	/	2.50
4	5.5	no	/	2.64
5	7	no	/	2.81
6	24	no	/	3.03
7	0	yes	0	2.83
8	1	yes	42	4.33
9	2.5	yes	78	5.42
10	5.5	yes	100	6.03
11	7	yes	100	5.65
12	24	yes	100	6.53

[a]Reaction conditions: Experiment performed in a NMR tube: **1a** (0.045 mmol), CD₂Cl₂ (0.75 mL), MeCHD (1.35 mmol), 30 °C. [b]Conversion was determined by ¹H NMR spectroscopy, [c] ¹H NMR signal: 2.38 ppm for Toluene, 1.70 ppm for MeCHD.

Table S4: Conversion of the *para*-quinone **3a** to the hydroquinone **4a** in the presence of the *ortho*-quinone **1a**.^[a]

Entry	MeCHD (equiv)	1a:3a	2a Conv. ^[b] [%]	4a Conv. ^[b] [%]
1	30	0:100	/	< 0.5
2	30	10:90	>99	< 1
3	30	50:50	>99	4.5
4	0	50:50	< 0.5	< 0.5
5 ^[c]	0	50:50	>99	< 0.5

[a]Reaction conditions: **1a+3a** (0.18 mmol), Toluene (2.0 mL), 30 °C, 24 h. [b]Conversion was determined by ¹H NMR spectroscopy, ^[c] **2a** was used instead of **1a**.

Table S5: Oxygen effect on the reduction reaction of 1a and 3a.[a]

Entry	Comp	Drad	t	TFA	Conv.[b]
Entry	Comp.	Plou.	[h]	[equiv]	[%]
1	1a	2a	1	/	55
2	1a	2a	1	/	71 ^[c]
3	3a	4a	5	40	45
4	3a	4a	5	40	$62^{[c]}$

[a]Reaction conditions: **1a** or **3a** (0.18 mmol), 1-methyl-1,4-cyclohexadiene (5.4 mmol), toluene (2.0 mL), 30 °C. [b]Conversion was determined by ¹H NMR spectroscopy, [c]Experiment performed under nitrogen atmosphere.

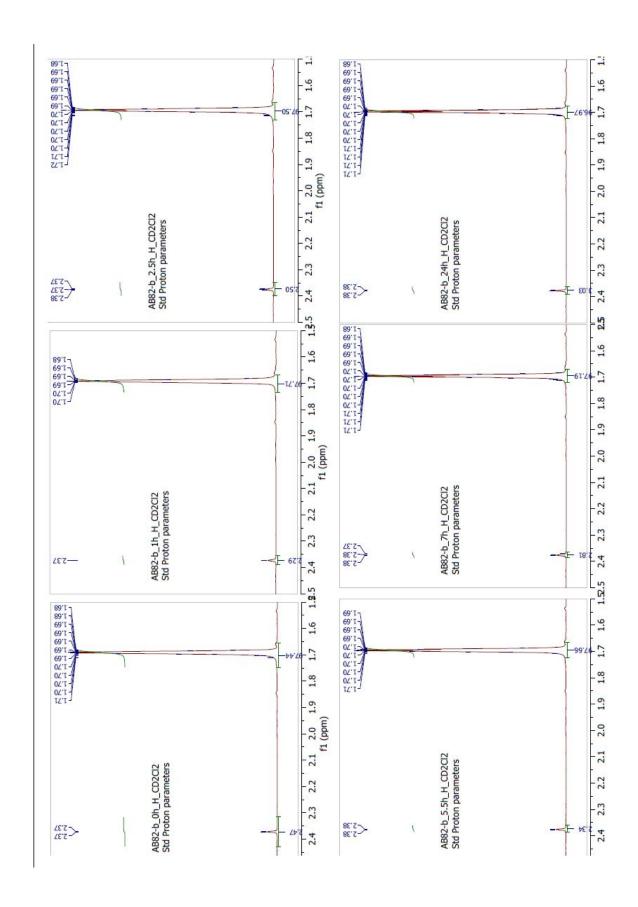


Figure S1. MeCHD conversion to toluene in absence of quinone 1a.

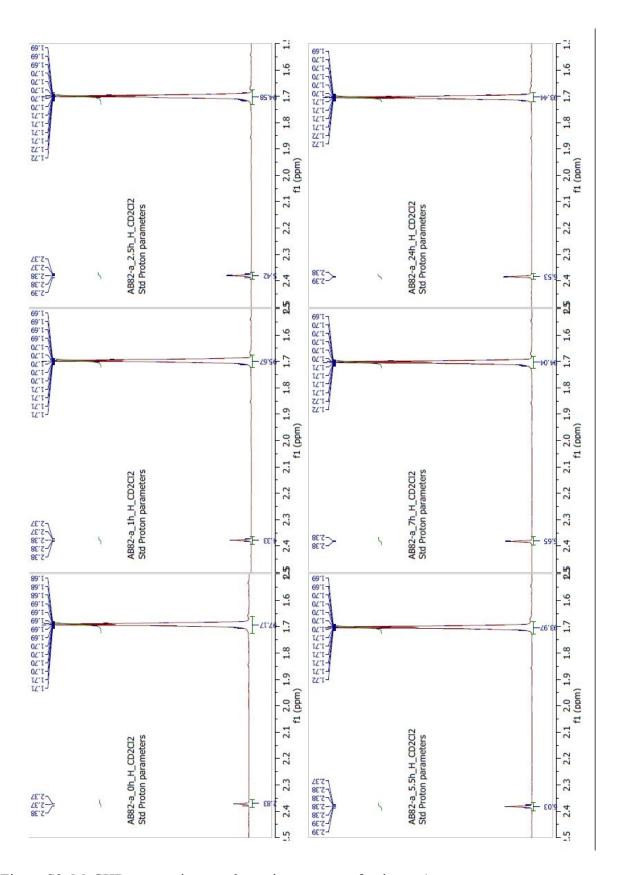


Figure S2. MeCHD conversion to toluene in presence of quinone 1a.

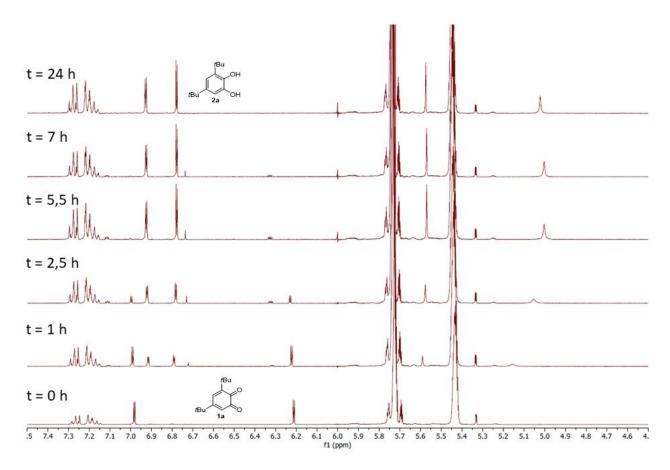


Figure S3. Conversion of compound ${\bf 1a}$ to catechol ${\bf 2a}$ in a NMR tube after 24 h. Experiment performed in CD_2Cl_2 as solvent.

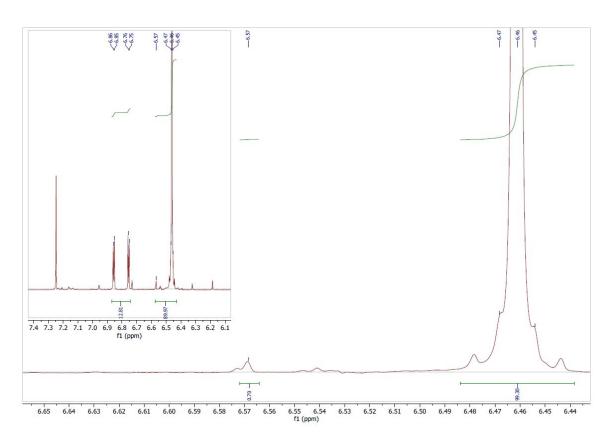


Figure S4. Conversion of compound 3a (6.46 ppm) to hydroquinone 4a (6.57 ppm) in the presence of 10% of 1a (6.86 – 6.75ppm).

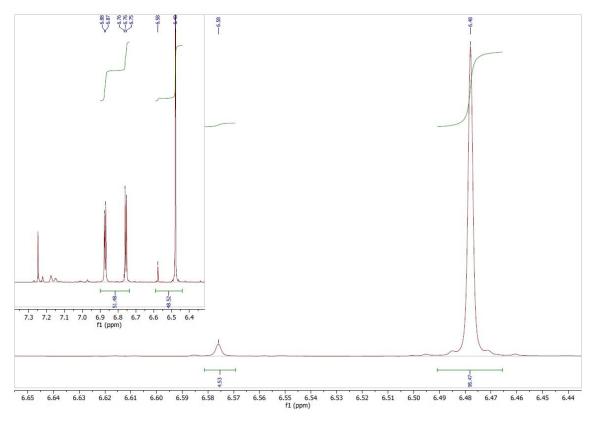
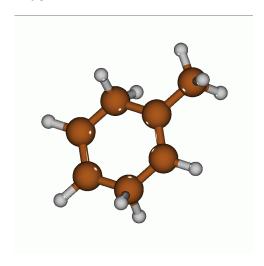


Figure S5. Conversion of compound 3a (6.48 ppm) to hydroquinone 4a (6.58 ppm) in the presence of 50% of 1a (6.88 – 6.75ppm).

Calculation results

MeCHD

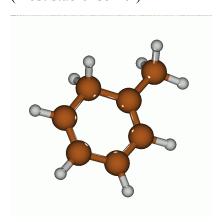


CBS-QB3 Enthalpy= -272.164977 CBS-QB3 Free Energy= -272.203348

No imaginary frequencies

C,0,-0.0787811878,-0.1034861486,-0

MeCHD alkyl radical (most stable isomer)



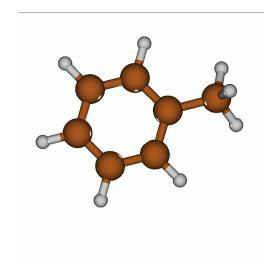
CBS-QB3 Enthalpy= -271.549665 CBS-QB3 Free Energy= -271.588237 (More stable of 1.23 kcal/mol than the isomer on C6)

No imaginary frequencies

C,0,-0.0046266383,0.0000025732,0.0515359345

 $\label{eq:continuous} $$ \c,0,0.0860672715,0.0000754893,1.4647901112\c,0,1.3575553274,-0.000024 $$ 7762,2.0890826192\c,0,2.5227396356,-0.0001434669,1.3753942072\c,0,2.49 $$ 23356129,-0.0002157599,-0.1322979838\c,0,1.1141062368,-0.0001235438,-0.726106327\c,0,3.8739575416,-0.0001879839,2.024533029\c,0,1.4064208173 $$ 0.0000273718,3.1746221411\c,0,3.066663178,0.8657233046,-0.5103970728\c,0,1.0320187869,-0.0001688812,-1.8079254068\c,0,0.9838283557,0.00004 $$ 35382,-0.4168421167\c,0.8663056125,-0.5102938156\c,0,4.4591154673,-0.8787481071, 1.7231470544\c,0,4.4591323923,0.8783900949,1.7232250611\c,0,3.79931563 $$ 14,-0.0002311139,3.1138140185$

Toluene



CBS-QB3 Enthalpy= -271.013230

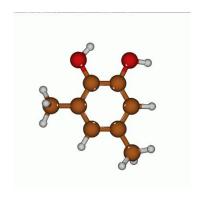
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No imaginary frequencies

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3,5-di-methylcatechol



CBS-QB3 Enthalpy= -460.541498

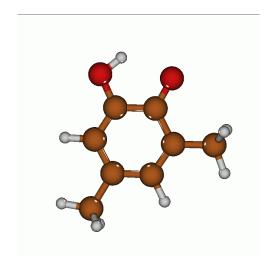
CBS-QB3 Free Energy= -460.589309

No imaginary frequencies

C,0,0.0357424462,-0.0007135469,0

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Phenoxyl radical from 3,5-di-methylcatechol (most stable isomer)



CBS-QB3 Enthalpy=

-459.923251

CBS-QB3 Free Energy=

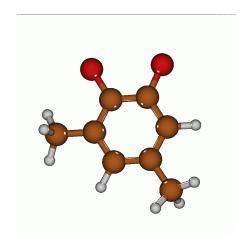
-459.969991

No imaginary frequencies

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070454,2.1533254866\C,0,2.4353138713,0.0923562347,1.4543160451\C,0,2.4
760968851,0.051828271,-0.0073688345\C,0,1.2129373983,-0.0234209015,-0.
7143542369\O,0,3.6212534048,0.1613617298,2.0604752924\O,0,3.6018082898
,0.084715193,-0.5589484708\C,0,1.2468786871,-0.0652839647,-2.212895231
4\H,0,-0.899013324,-0.1087833317,-0.4976807677\C,0,-1.2799536632,-0.04
82871869,2.1665149996\H,0,1.2524811487,0.0926048151,3.2364046091\H,0,4
.26827428,0.1681373074,1.3236119087\H,0,0.2408552873,-0.1210685192,-2.
6318785395\H,0,1.7479245906,0.8207513756,-2.6137361051\H,0,1.823776382
3,-0.9262340712,-2.5632199959\H,0,-1.9107459942,0.7963526075,1.8696492
312\H,0,-1.834233088,-0.9603604504,1.9208570286\H,0,-1.1496107779,-0.0
110709776,3.2487847848

3,5-di-methyl orthoquinone



CBS-QB3 Enthalpy= -459.313646

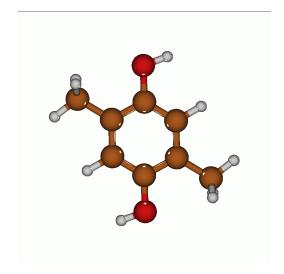
CBS-QB3 Free Energy= -459.359148

No imaginary frequencies

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2,5MeQH2



CBS-QB3 Enthalpy= -460.539744

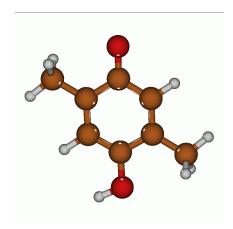
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No imaginary frequencies

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Radical from 2,5-dimethyl-hydroquinone



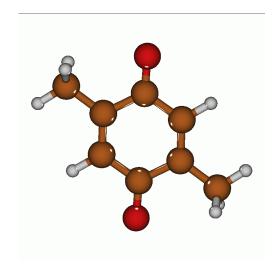
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CBS-QB3 Free Energy= -459.960006

No imaginary frequencies

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2,5-dimethyl-para benzoquinone



CBS-QB3 Enthalpy= -459.328177

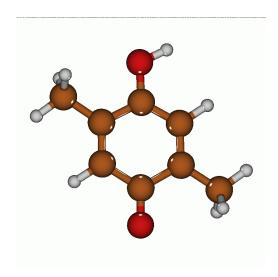
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No imaginary frequencies

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O-protonated 2,5-dimethyl-para benzoquinone cbs-qb3 geom=check scrf=(solvent=toluene)



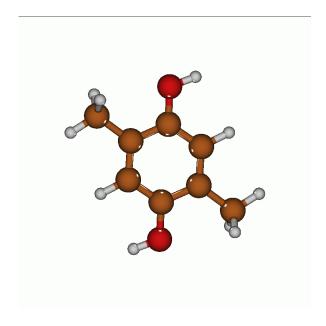
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2,5-dimethyl-para hydroquinone radical cation cbs-qb3 geom=check scrf=(solvent=toluene)



CBS-QB3 Enthalpy= -460.304174

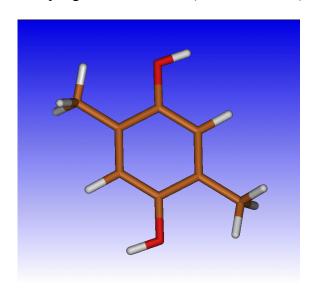
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79401,0.0178192026,-0.0026167622\C,0,1.1597897591,0.0036456155,-0.7573
306506\O,0,-1.1100857418,0.0442952752,2.0878473926\C,0,1.1829222501,-0
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650240026\H,0,0.6537872816,0.8772726807,3.9927521282\H,0,0.6324428875,
-0.8763914268,3.9839650187\H,0,2.1861729762,-0.0183346431,4.042623131\
H,0,0.2036897012,-0.0175346333,-2.6742677072\H,0,1.7367853074,0.876891
7553,-2.6244227394\H,0,1.756737003,-0.8767876772,-2.6155696163

2,5MeQH2 (in toluene)

cbs-qb3 geom=check scrf=(solvent=toluene)



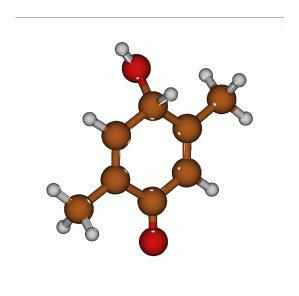
CBS-QB3 Enthalpy= -460.544148

CBS-QB3 Free Energy= -460.590657

C,0,0.0035608738,0.0102658609,-0.0082258884\C,0,-

 $0.005135956, 0.010424762, 1.3858349108 \setminus C, 0, 1.1957090876, 0.0102340337, 2.1054174335 \setminus C, 0, 2.3863118363, 0.0102472847, 1.3765896052 \setminus C, 0, 2.3950083266, 0.0104553172, -0.0174848957 \setminus C, 0, 1.1941779291, 0.0102794146, -0.7370598397 \setminus O, 0, -1.1725416845, 0.0103661697, 2.1134655544 \setminus C, 0, 1.1820402182, 0.009881 6509, 3.611557352 \setminus O, 0, 3.5624294222, 0.0103961133, -0.7450947742 \setminus C, 0, 1.207 8220507, 0.0099414709, -2.2432012541 \setminus H, 0, 3.3316254948, 0.0098161363, 1.913 4883363 \setminus H, 0, -0.9417430202, 0.0098806782, -0.5451412342 \setminus H, 0, -1.9208440608 +0.0098592774, 1.5075922211 \setminus H, 0, 4.3107200712, 0.0101589502, -0.139207076 \setminus H, 0, 0.6584408174, 0.887261276, 4.0030157994 \setminus H, 0, 0.6570260302, -0.86691007 06, 4.0025803558 \setminus H, 0, 2.1972132562, 0.0090401835, 4.0123883873 \setminus H, 0, 0.19264 13436, 0.0095769834, -2.6440142557 \setminus H, 0, 1.7318308189, 0.8870734869, -2.6346 567204 \setminus H, 0, 1.7324073507, -0.8670996367, -2.6342509761$

2,5MeQH2 tautomer hydride addition C1



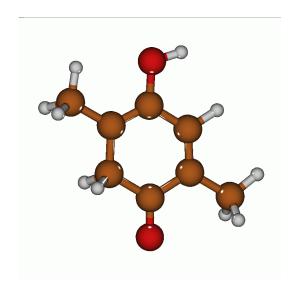
CBS-QB3 Enthalpy= -460.510604

CBS-QB3 Free Energy= -460.557201

C,0,0.0135926728,0.0095175641,-0.0064402233\C,0,0

 $.0052914373, 0.0113660578, 1.4659000551 \cdots \c$

2,5MeQH2 tautomer hydride addition C3

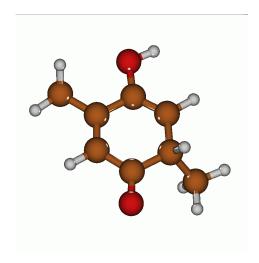


CBS-QB3 Enthalpy= -460.516286

CBS-QB3 Free Energy= -460.564729

 $0.0074566228, 0.2185189179, 1.5150037205 \colonormalco$

2,5MeQH2 tautomer hydride addition C5



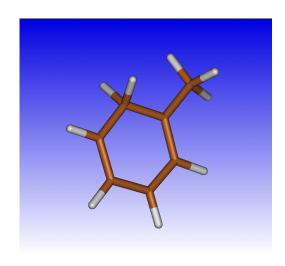
CBS-QB3 Enthalpy= -460.514455

CBS-QB3 Free Energy= -460.560755

 $C,0,-0.0354512093,-0.1714931408,0.0297474399 \capable C,0,\\ -0.0638056374,-0.4392526063,1.4660738032 \capable C,0,1.2775512969,-0.578263653\\ 9,2.196100776 \capable C,0,2.4880674336,-0.3717060141,1.3344468535 \capable C,0,2.405677\\ 322,-0.113089597,0.0203513301 \capable C,0,1.1118494312,-0.0099937849,-0.669441\\ 566 \capable C,0,-1.1171328529,-0.5563693072,2.0786474331 \capable C,0,1.2977448278,0.32\\ 04696889,3.4515526706 \capable C,0,3.4814347077,0.0802792299,-0.8054683209 \capable C,0,\\ 1.1173741707,0.2787587504,-2.1420519294 \capable H,0,3.4603428875,-0.4426980572\\ 1.8158750141 \capable H,0,-1.0007034025,-0.0886674107,-0.4575999081 \capable H,0,4.2932\\ 814855,0.0019915268,-0.2915450235 \capable H,0,1.3170650647,1.3760697381,3.1688\\ 663294 \capable H,0,0.4036112973,0.1350969073,4.0473072781 \capable H,0,2.1803945359,0.1\\ 124793784,4.0609472336 \capable H,0,0.1016131417,0.3313993622,-2.533672441 \capable H,0,1.2883288096,-1.6207256291,2.5496333718\\ -2.6837386917 \capable H,0,1.2883288096,-1.6207256291,2.5496333718\\ \capable C,0,0.1016131417,0.1.2883288096,-1.6207256291,2.5496333718\\ \capable C,0,0.1016131417,0.10161$

$MeCHD(-H^{+})$

cbs-qb3 geom=check scrf=(solvent=toluene)



CBS-QB3 Enthalpy= -271.348264

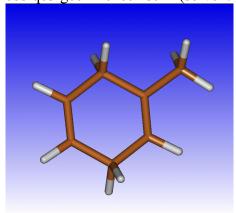
CBS-QB3 Free Energy= -271.386823

 $C,0,0.0009904163,0.0407302422,0.0006532001\C,0,0.$

 $0019610067, 0.0207256209, 1.3619109214 \cdots, 0, 1.2421932609, -0.0319213455, 2. \\0394712906 \cdots, 0, 2.4757736063, -0.0567330022, 1.380352316 \cdots, 0, 2.5307187415 \\0.0229975845, 0.0006838971 \cdots, 0, 1.2613609079, 0.0068485857, -0.760168706 \\4 \cdots, 0, 3.8194265189, 0.0066276779, -0.7414144422 \cdots, 0, 3.389353892, -0.09521 \\36968, 1.9603847253 \cdots, 0, 1.2788734564, 0.8353639676, -1.4887301909 \cdots, 0, 0.9264053784, 0.0711053268, -0.5595213745 \cdots, 0, 0.9203147706, 0.0406174325, 1.9270889892 \cdots, 0, 1.2368452007, -0.0533949833, 3.1242946403 \cdots, 0, 1.23524340 \\65, -0.8677168134, -1.4367044153 \cdots, 0, 3.7819069852, -0.6135420302, -1.63976 \\41558 \cdots, 0, 4.0157362399, 1.0338054341, -1.0761821887 \cdots, 0, 4.653807947, -0.3 \\053765173, -0.1146599997$

MeCHD

cbs-qb3 geom=check scrf=(solvent=toluene)



CBS-QB3 Enthalpy= -272.165916

CBS-QB3 Free Energy= -272.204271

 $C,0,-0.0793069454,-0.1035290474,-0.0874741998\C,0,-0.$

H CBS-OB3 Enthalpy=

CBS-QB3 Enthalpy= -0.497457 CBS-QB3 Free Energy=

-0.510472

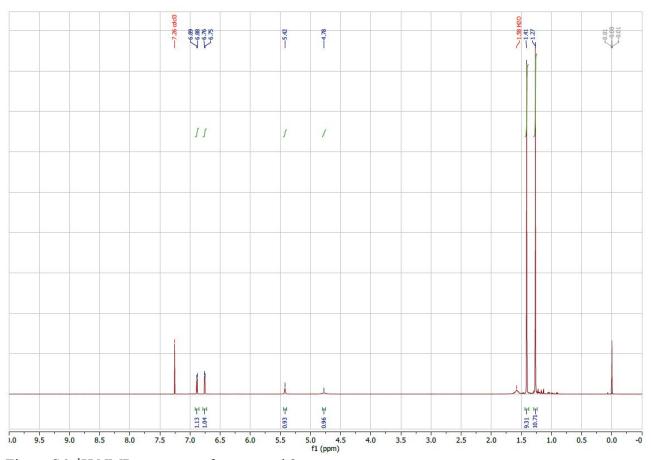


Figure S6. ¹H NMR spectrum of compound **2a**.

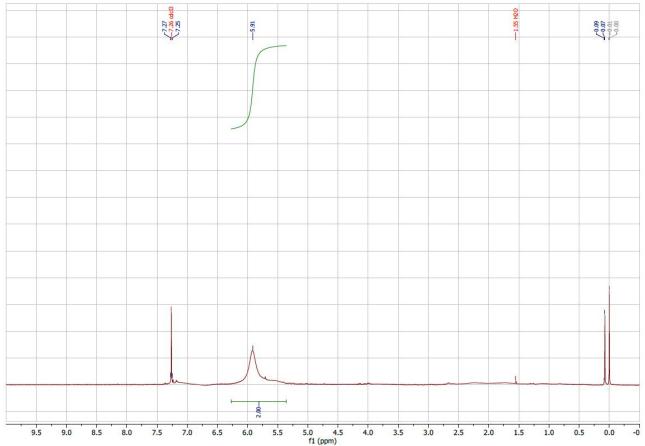


Figure S7. ¹H NMR spectrum of compound **2b**

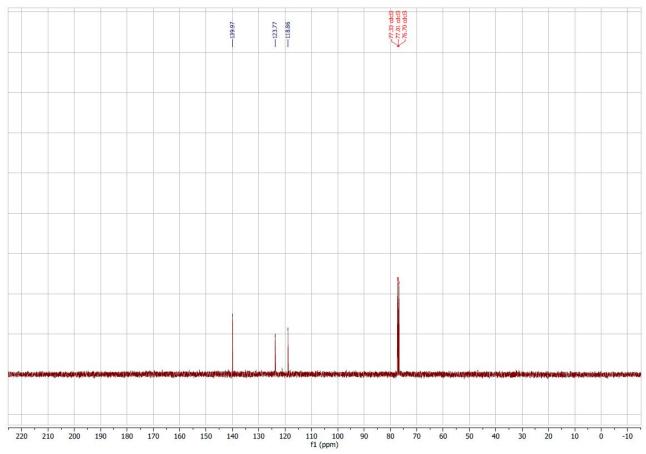


Figure S8. $^{13}C\{^{1}H\}$ NMR spectrum of compound **2b**.

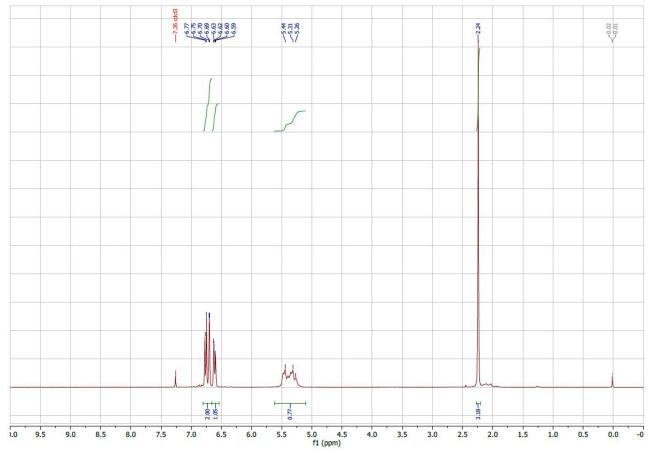


Figure S9. ^1H NMR spectrum of compound 2c.

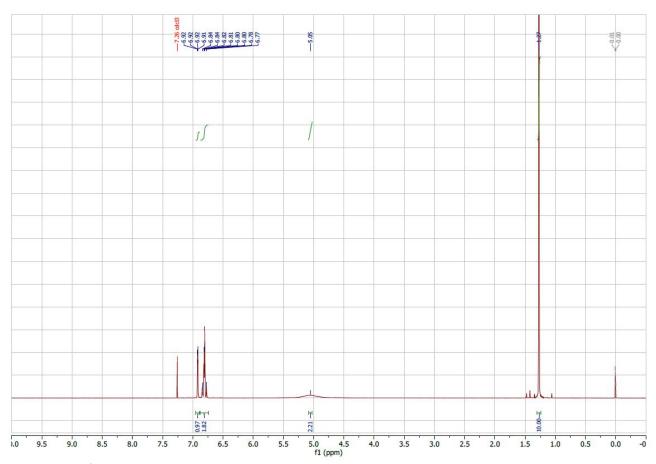


Figure S10. ¹H NMR spectrum of compound **2d**.

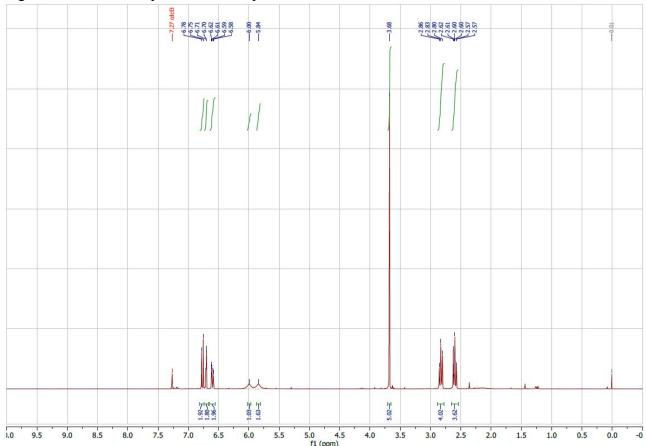


Figure S11. ¹H NMR spectrum of compound **2e**.

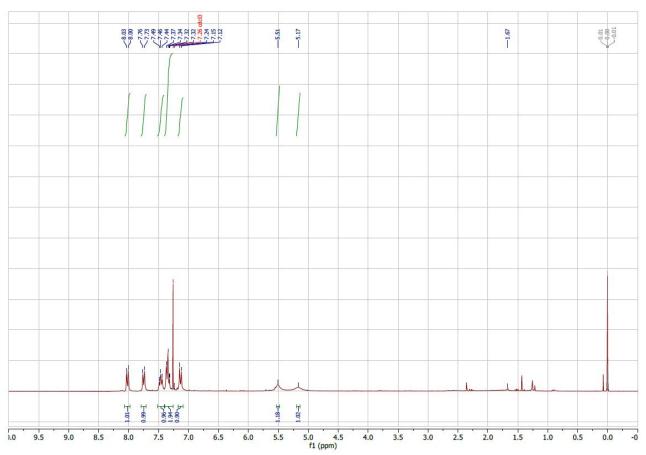


Figure S12. ¹H NMR spectrum of compound **2f**.

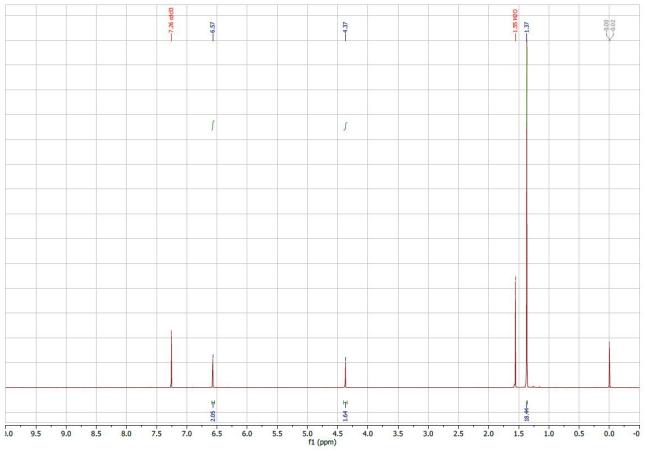


Figure S13. ¹H NMR spectrum of compound **4a**.

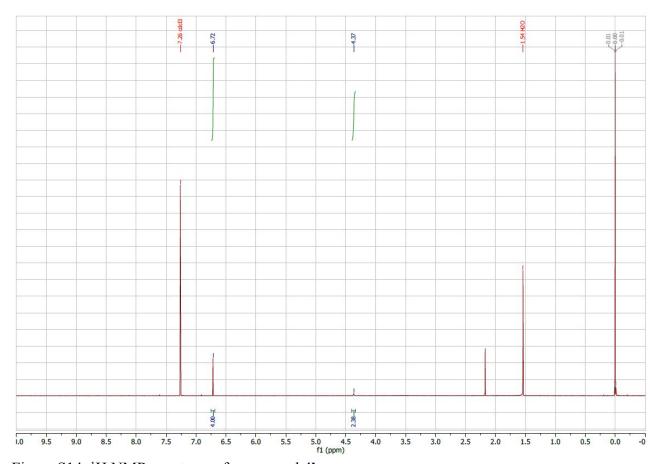


Figure S14. ¹H NMR spectrum of compound **4b**.

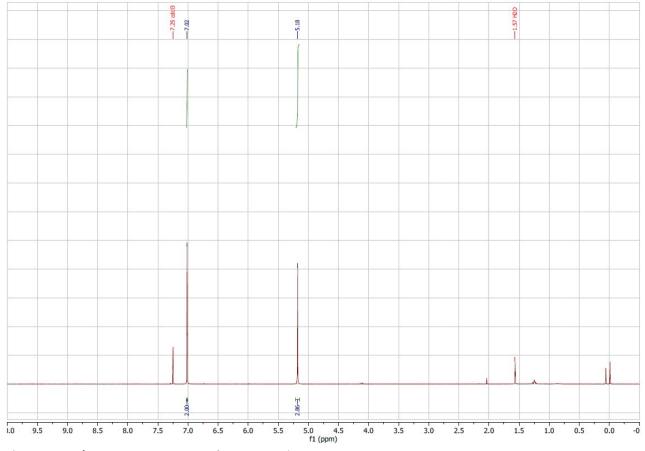


Figure S15. ¹H NMR spectrum of compound **4c**.

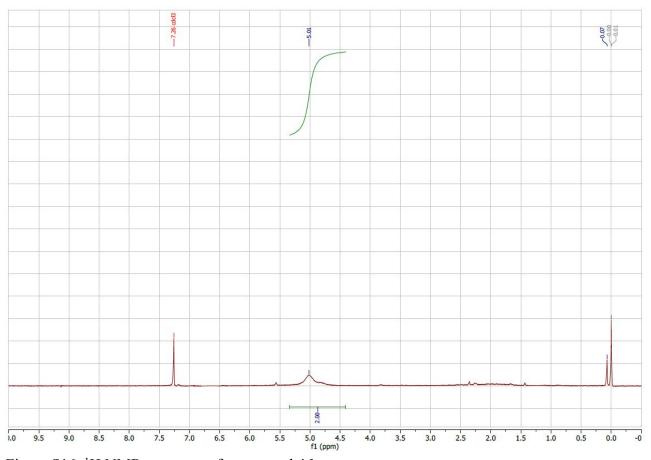


Figure S16. ¹H NMR spectrum of compound **4d**.

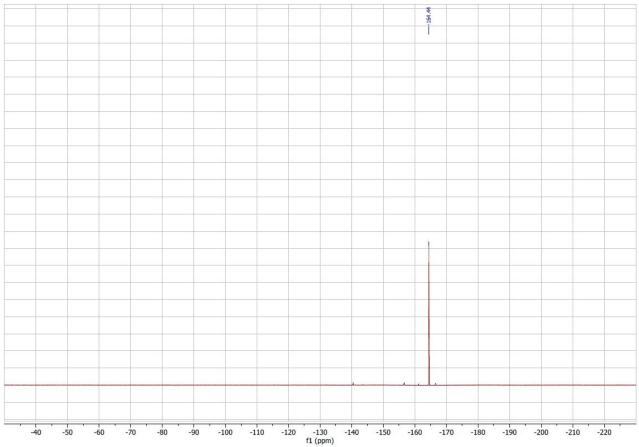


Figure S17. ¹⁹F NMR spectrum of compound **4d**.

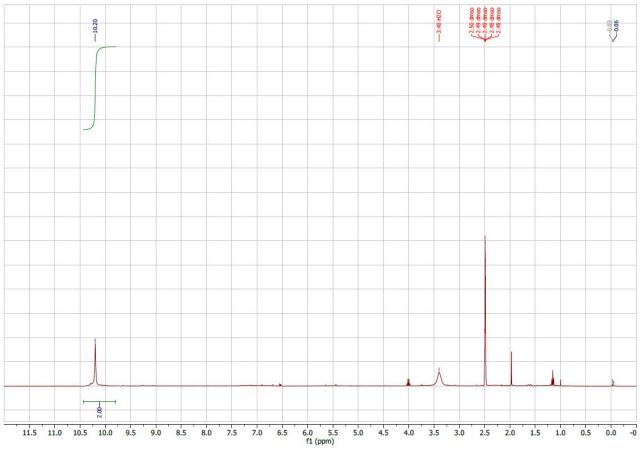


Figure S18. ¹H NMR spectrum of compound **4e**.

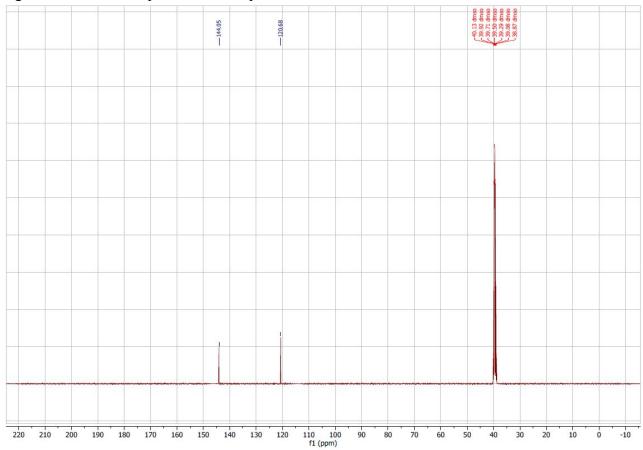


Figure S19. $^{13}C\{^{1}H\}$ NMR spectrum of compound 4e.

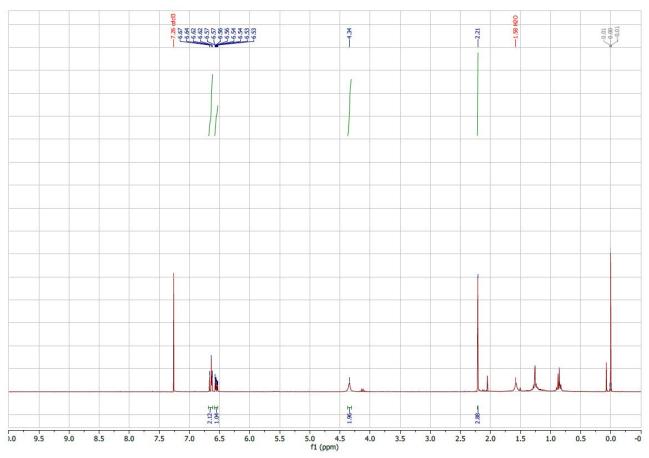


Figure S20. ¹H NMR spectrum of compound **4f**.

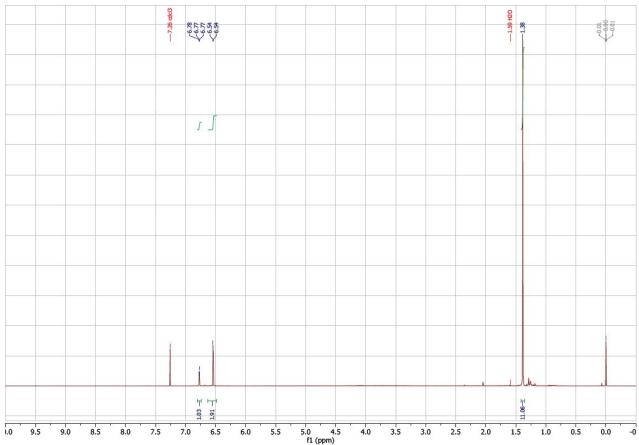


Figure S21. ¹H NMR spectrum of compound **4g**.

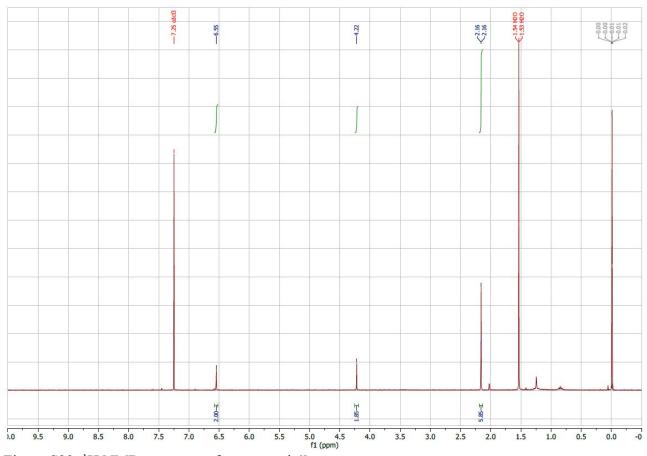


Figure S22. ¹H NMR spectrum of compound **4h**.

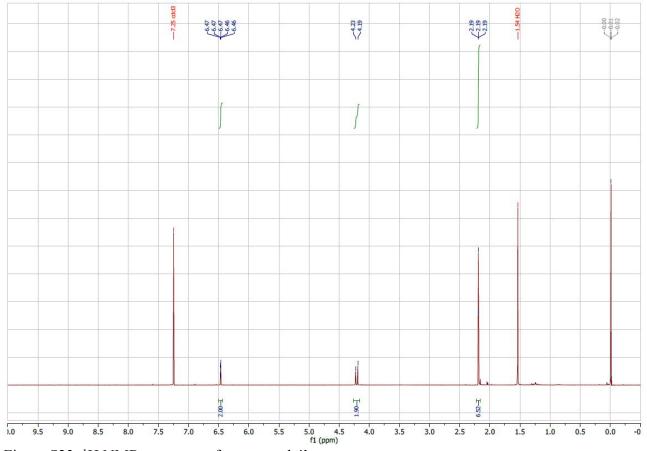
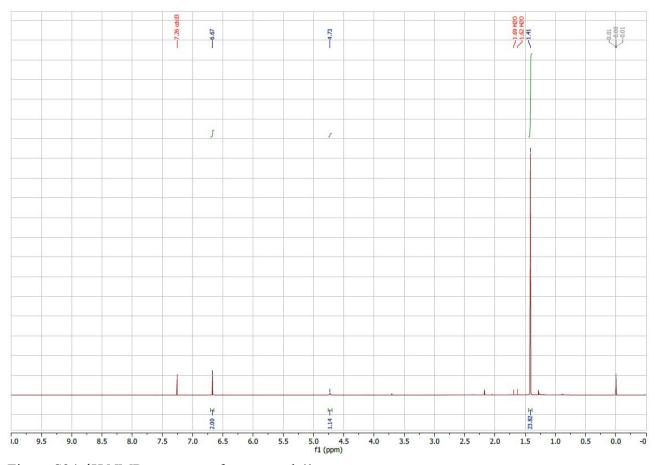
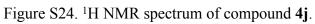


Figure S23. ¹H NMR spectrum of compound **4i**.





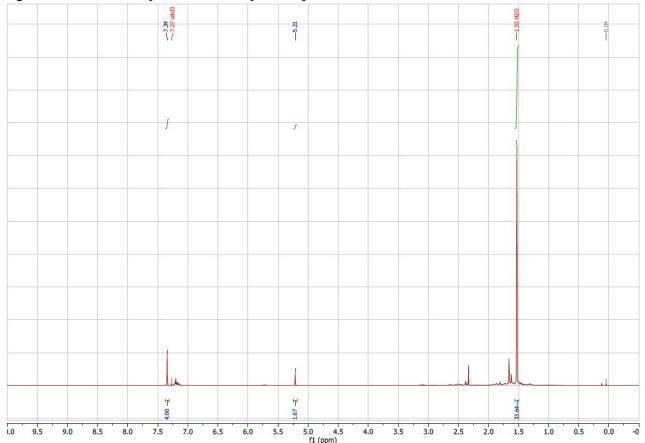


Figure S25. ¹H NMR spectrum of compound **4jj**.

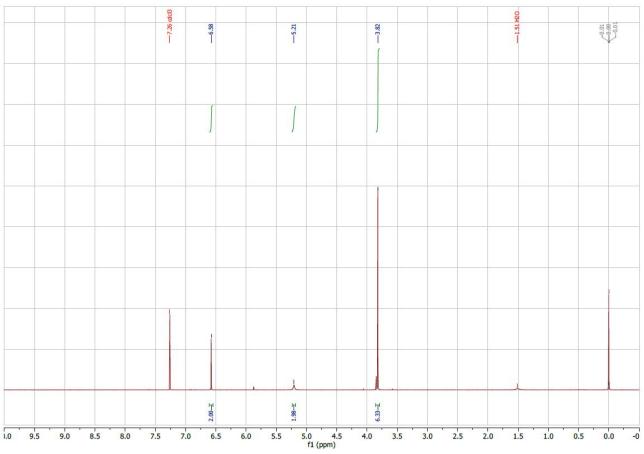


Figure S26. ¹H NMR spectrum of compound **4k**.

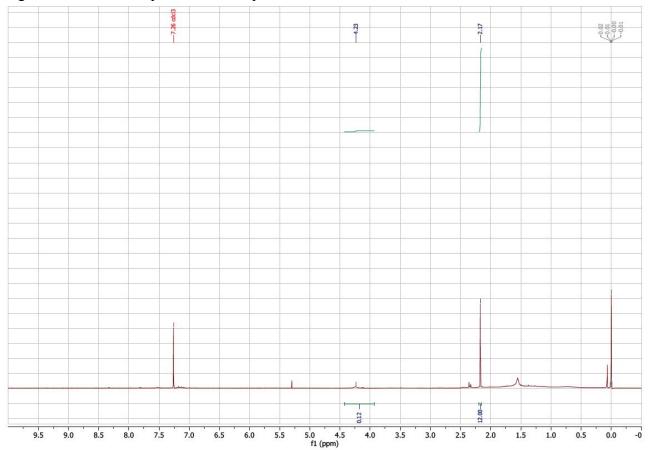


Figure S27. ¹H NMR spectrum of compound **41**.

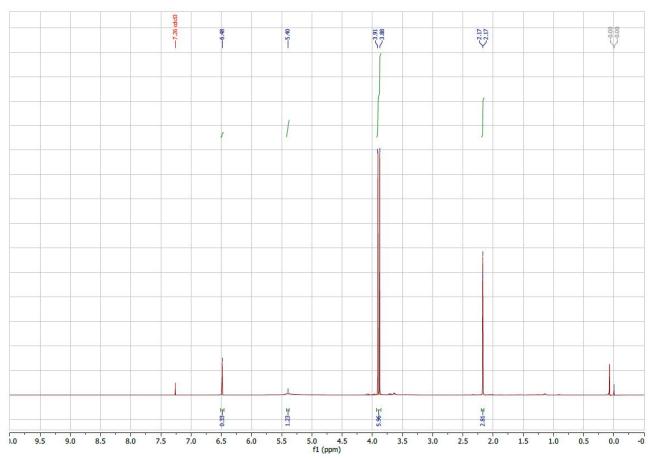


Figure S28. ¹H NMR spectrum of compound **4CoQ**₀.