

# A Robust Multifunctional Eu<sub>6</sub>-Cluster Based Framework for Gas Separation and Recognition of Small Molecules and Heavy Metal Ions

Falu Hu,<sup>†‡</sup> Zhengyi Di,<sup>†</sup> Mingyan Wu,<sup>†</sup> Maochun Hong<sup>\*⊥</sup> and Jing Li<sup>\*‡†</sup>

<sup>†</sup> Hoffmann Institute of Advanced Materials, Shenzhen Polytechnic, 7098 Liuxian Blvd, Nanshan District, Shenzhen, 518055, China.

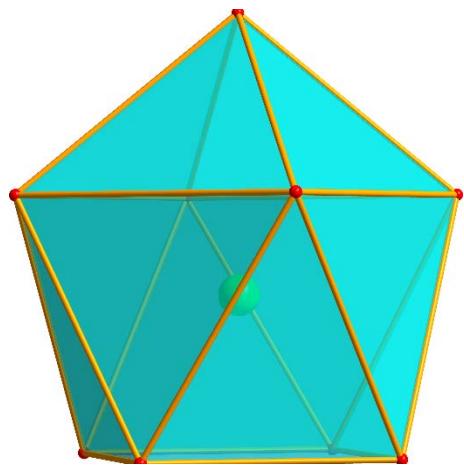
<sup>‡</sup> Department of Chemistry and Chemical Biology, Rutgers University, 610 Taylor Road, Piscataway, NJ, 08854, USA.

<sup>⊥</sup> State Key Laboratory of Structure Chemistry, Fujian Institute of Research on the Structure of Matter, Chinese Academy of Sciences, Fuzhou, 350002, China.

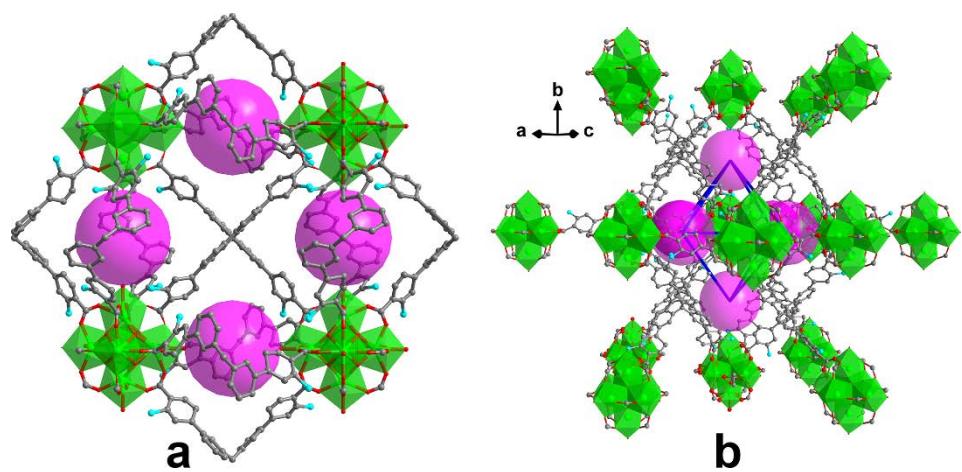
## Supporting Information

**Table S1.** Crystal data of Eu-SPFF (CCDC number: 1912826)

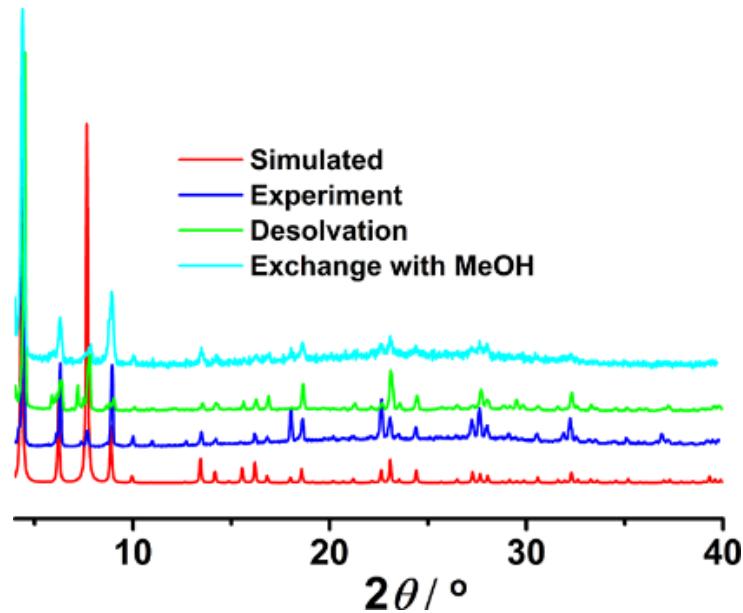
Identification code	<b>Eu-SPFF</b>
Empirical formula	C <sub>159</sub> H <sub>116</sub> Eu <sub>6</sub> F <sub>12</sub> O <sub>38</sub>
Formula weight	3774.27
Temperature	100.01(10) K
Wavelength	1.54184 Å
Crystal system	Cubic
Space group	<i>Pm-3m</i>
Unit cell dimensions	a = 19.4694(2) Å
Volume	7380.0(2) Å <sup>3</sup>
Z	1
Density (calculated)	0.849 g/cm <sup>3</sup>
Absorption coefficient	0.397 mm <sup>-1</sup>
F(000)	1860
Theta range for data collection	4.5280 to 69.1140°.
Completeness to theta = 67.684°	98 %
Max. and min. transmission	1.00000 and 0.847
Goodness-of-fit on F <sup>2</sup>	1.391
Final R indices [I>2sigma(I)]	R1 = 0.0940, wR2 = 0.2826
R indices (all data)	R1 = 0.0984, wR2 = 0.2909
Largest diff. peak and hole	4.485 and -5.215 e.Å <sup>-3</sup>



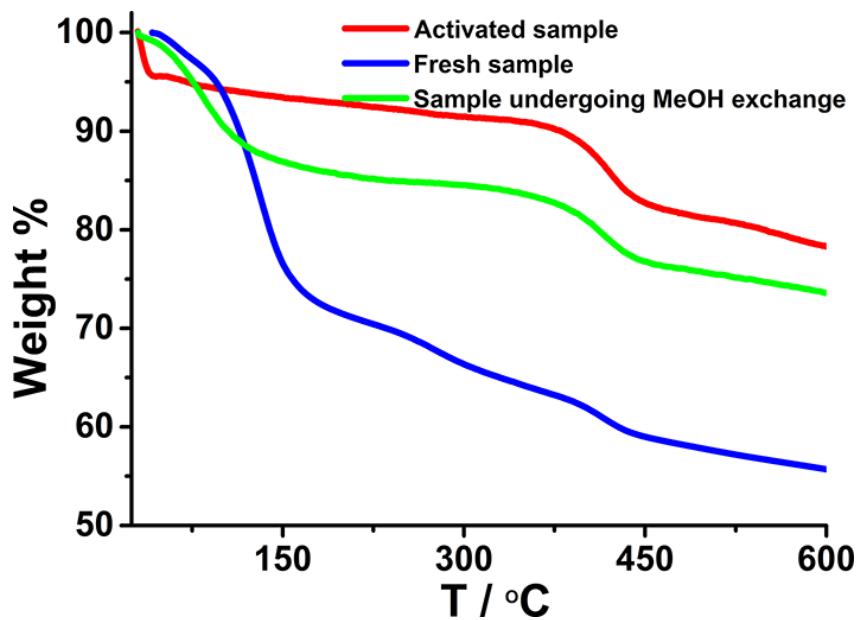
**Figure S1.** The mono-capped square antiprism coordination geometry of Eu(III).



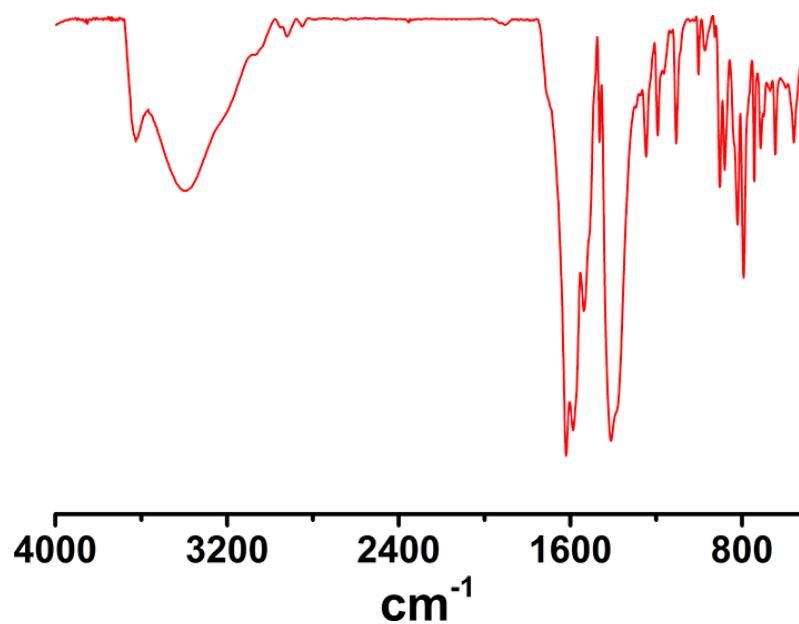
**Figure S2.** Coordination environment of (a) ligand and (b) Eu<sub>6</sub> cluster.



**Figure S3.** The PXRD patterns of Eu-SPFF (from bottom to top: simulated, as-synthesized, after exchange with MeOH and after desolvation).



**Figure S4.** The TGA curves of fresh, activated and MeOH exchanged Eu-SPFF.



**Figure S5.** The FT-IR spectrum of **Eu-SPFF**.

## The IAST Selectivity Calculations

Based on ideal adsorbed solution theory (IAST), the adsorption isotherm is fitted by the single-site Langmuir-Freundlich equation:

$$N = A_1 b_1 P^{c1} / (1 + b_1 P^{c1})$$

N: molar loading of species, mmol/g

A: saturation capacity of species, mmol/g

b: constant

c: constant

The adsorption selectivities,  $S$ , for binary mixtures  $C_2H_6 / CH_4$ ,  $C_2H_4 / CH_4$ ,  $C_2H_2 / CH_4$ ,  $C_3H_8 / CH_4$ ,  $C_3H_6 / CH_4$ ,  $C_3H_4 / CH_4$ ,  $n\text{-}C_4H_{10} / CH_4$ ,  $i\text{-}C_4H_{10} / CH_4$ ,  $l\text{-}C_4H_8 / CH_4$  and  $t\text{-}C_4H_8 / CH_4$  (15:85).

$$S = (x_1/x_2)(y_2/y_1)$$

$S$ : adsorption selectivity

$x_i$ : the mole fractions of component in the adsorbed i ( $i = 1, 2$ )

$y_i$ : the mole fractions of component in the bulk phases i ( $i = 1, 2$ )

The IAST calculations were carried out for equimolar binary gas-phase mixtures of  $C_2H_6 / CH_4$ ,  $C_2H_4 / CH_4$ ,  $C_2H_2 / CH_4$ ,  $C_3H_8 / CH_4$ ,  $C_3H_6 / CH_4$ ,  $C_3H_4 / CH_4$ ,  $n\text{-}C_4H_{10} / CH_4$ ,  $i\text{-}C_4H_{10} / CH_4$ ,  $l\text{-}C_4H_8 / CH_4$  and  $t\text{-}C_4H_8 / CH_4$  (15:85).





6. Wang, J.; Krishna, R.; Yang, T.; Deng, S., Nitrogen-rich microporous carbons for highly selective separation of light hydrocarbons. *J. Mater. Chem. A* **2016**, *4*, 13957-13966.
7. Han, G.; Wang, K.; Peng, Y.; Zhang, Y.; Huang, H.; Zhong, C., Enhancing Higher Hydrocarbons Capture for Natural Gas Upgrading by Tuning van der Waals Interactions in fcu-Type Zr-MOFs. *Ind. Eng. Chem. Res.* **2017**, *56*, 14633-14641.
8. Gao, S.; Morris, C. G.; Lu, Z.; Yan, Y.; Godfrey, H. G. W.; Murray, C.; Tang, C. C.; Thomas, K. M.; Yang, S.; Schröder, M., Selective Hysteretic Sorption of Light Hydrocarbons in a Flexible Metal–Organic Framework Material. *Chem. Mater.* **2016**, *28*, 2331-2340.
9. Meng, S.; Ma, H.; Jiang, L.; Ren, H.; Zhu, G., A facile approach to prepare porphyrinic porous aromatic frameworks for small hydrocarbon separation. *J. Mater. Chem. A* **2014**, *2*, 14536-14541.
10. Jia, J.; Wang, L.; Sun, F.; Jing, X.; Bian, Z.; Gao, L.; Krishna, R.; Zhu, G., The adsorption and simulated separation of light hydrocarbons in isoreticular metal-organic frameworks based on dendritic ligands with different aliphatic side chains. *Chem. Eur. J.* **2014**, *20*, 9073-9080.
11. Shi, R.; Lv, D.; Chen, Y.; Wu, H.; Liu, B.; Xia, Q.; Li, Z., Highly selective adsorption separation of light hydrocarbons with a porphyrinic zirconium metal-organic framework PCN-224. *Sep. Purif. Technol.* **2018**, *207*, 262-268.
12. Zhang, Y.; Xiao, H.; Zhou, X.; Wang, X.; Li, Z., Selective Adsorption Performances of UiO-67 for Separation of Light Hydrocarbons C<sub>1</sub>, C<sub>2</sub>, and C<sub>3</sub>. *Ind. Eng. Chem. Res.* **2017**, *56*, 8689-8696.
13. He, Y. P.; Tan, Y. X.; Zhang, J., Tuning a layer to a pillared-layer metal-organic framework for adsorption and separation of light hydrocarbons. *Chem. Commun.* **2013**, *49*, 11323-11325.
14. Chen, C. X.; Zheng, S. P.; Wei, Z. W.; Cao, C. C.; Wang, H. P.; Wang, D.; Jiang, J.; Fenske, D.; Su, C. Y., A Robust Metal-Organic Framework Combining Open Metal Sites and Polar Groups for Methane Purification and CO<sub>2</sub> /Fluorocarbon Capture. *Chem. Eur. J.* **2017**, *23*, 4060-4064.
15. Fan, W.; Wang, Y.; Xiao, Z.; Huang, Z.; Dai, F.; Wang, R.; Sun, D., Two-dimensional cobalt metal-organic frameworks for efficient C<sub>3</sub>H<sub>6</sub> /CH<sub>4</sub> and C<sub>3</sub>H<sub>8</sub> /CH<sub>4</sub> hydrocarbon separation. *Chinese Chem. Lett.* **2018**, *29*, 865-868.
16. He, Y.; Xiang, S.; Zhang, Z.; Xiong, S.; Fronczek, F. R.; Krishna, R.; O'Keeffe, M.; Chen, B., A microporous lanthanide-tricarboxylate framework with the potential for purification of natural gas. *Chem. Commun.* **2012**, *48*, 10856-10858.
17. Xue, D. X.; Cadiau, A.; Weselinski, L. J.; Jiang, H.; Bhatt, P. M.; Shkurenko, A.; Wojtas, L.; Chen, Z.; Belmabkhout, Y.; Adil, K.; Eddaoudi, M., Topology meets MOF chemistry for pore-aperture fine tuning: ftw-MOF platform for energy-efficient separations via adsorption kinetics or molecular sieving. *Chem. Commun.* **2018**, *54*, 6404-6407.
18. Ling, Y.; Jiao, J.; Zhang, M.; Liu, H.; Bai, D.; Feng, Y.; He, Y., A porous lanthanide metal–organic framework based on a flexible cyclotriphosphazene-functionalized hexacarboxylate exhibiting selective gas adsorption. *Cryst. Eng. Comm.* **2016**, *18*,

- 6254-6261.
- 19. Wang, D.; Liu, Z.; Xu, L.; Li, C.; Zhao, D.; Ge, G.; Wang, Z.; Lin, J., A heterometallic metal-organic framework based on multi-nuclear clusters exhibiting high stability and selective gas adsorption. *Dalton. Trans.* **2018**, *48*, 278-284.
  - 20. Bai, D.; Wang, Y.; He, M.; Gao, X.; He, Y., Structural diversities and gas adsorption properties of a family of rod-packing lanthanide-organic frameworks based on cyclotriphosphazene-functionalized hexacarboxylate derivatives. *Inorg. Chem. Front.* **2018**, *5*, 2227-2237.
  - 21. Fu, H. R.; Wang, F.; Zhang, J., A stable zinc-4-carboxypyrazole framework with high uptake and selectivity of light hydrocarbons. *Dalton. Trans.* **2015**, *44*, 2893-2896.