

## Supporting Information

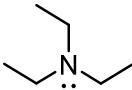
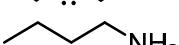
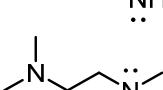
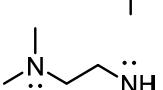
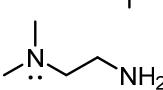
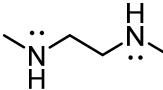
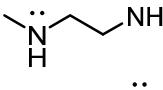
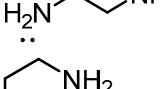
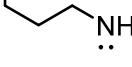
### The Mechanism of Dedoping PEDOT:PSS by Aliphatic Polyamines

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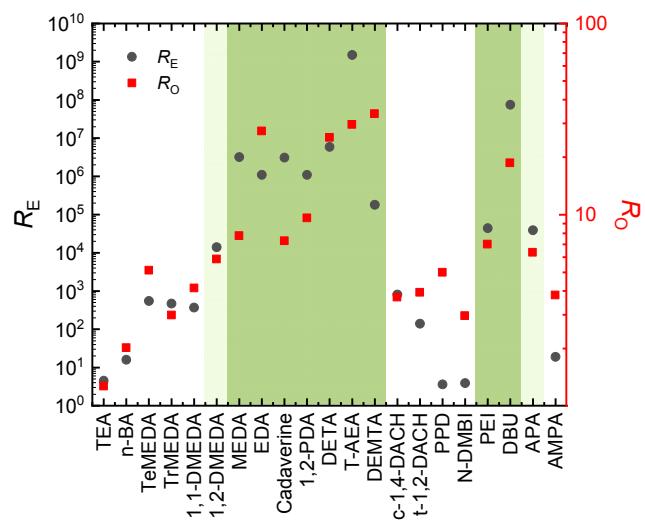
**Table S1.** The electrical ratio ( $R_E$ ) and optical ratio ( $R_O$ ) of dedoping with all tested compounds. For these compounds, their full name is given as well as the acronym and the structural formula. The shade of green denotes the degree of dedoping. The temperature at which the maximum dedoping capacity is obtained is given.

Compound	Acronym	Structural formula	$R_E$	$R_O$	$T$ (°C)
triethylamine	TEA		4.5	1.27	25
<i>n</i> -butylamine	n-BA		$1.6 \times 10^1$	2.02	50
<i>N,N,N',N'</i> -tetramethyl-1,2-ethanediamine	TeMEDA		$5.5 \times 10^2$	5.13	50
<i>N,N,N'</i> -trimethyl-1,2-ethanediamine	TrMEDA		$4.7 \times 10^2$	2.99	25
<i>N,N</i> -dimethyl-1,2-ethanediamine	1,1-DMEDA		$3.7 \times 10^2$	4.14	50
<i>N,N</i> '-dimethyl-1,2-ethanediamine	1,2-DMEDA		$1.4 \times 10^4$	5.87	90
<i>N</i> -methyl-1,2-ethanediamine	MEDA		$3.2 \times 10^6$	7.78	90
1,2-ethanediamine	EDA		$1.1 \times 10^6$	27.4	50-90
1,5-pentanediamine	Cadaverine		$3.1 \times 10^6$	7.32	120

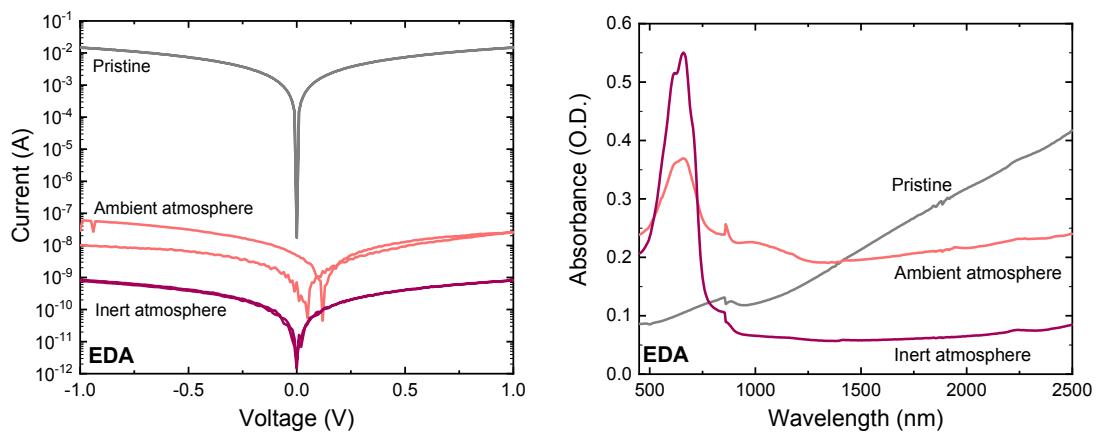
1,2-propanediamine	1,2-PDA		$1.1 \times 10^6$	9.63	25-50
<i>N</i> -(2-aminoethyl)-1,2-ethanediamine	DETA		$5.9 \times 10^6$	25.4	120
<i>N,N</i> -bis(2-aminoethyl)-1,2-ethanediamine	T-AEA		$1.5 \times 10^9$	29.7	120
<i>N</i> -(2-aminoethyl)- <i>N</i> -methyl-1,2-ethanediamine	DEMTA		$1.8 \times 10^5$	33.8	95
cis-1,4-cyclohexanediamine	c-1,4-DACH		$8.1 \times 10^2$	3.71	90
trans-1,2-cyclohexanediamine	t-1,2-DACH		$1.4 \times 10^2$	3.93	90-120
<i>p</i> -phenylenediamine	PPD		3.6	5.01	25
4-(1,3-dimethyl-2,3-dihydro-1 <i>H</i> -benzoimidazol-2-yl)phenyl)dimethylamine	N-DMBI		3.9	2.97	25
branched poly(ethyleneimine)	PEI		$4.4 \times 10^4$	7.03	80
1,8-diazabicyclo(5.4.0)undec-7-ene	DBU		$7.5 \times 10^7$	18.7	90-120
3-(azepan-1-yl)propan-1-amine	APA		$3.9 \times 10^4$	6.36	90-120
3-(azepan-1-yl)- <i>N</i> -methylpropan-1-amine	AMPA		$1.9 \times 10^1$	3.81	90-120

**Table S2.** List of suppliers and purity of the employed amines. The purity is left blank if not specified by the supplier.

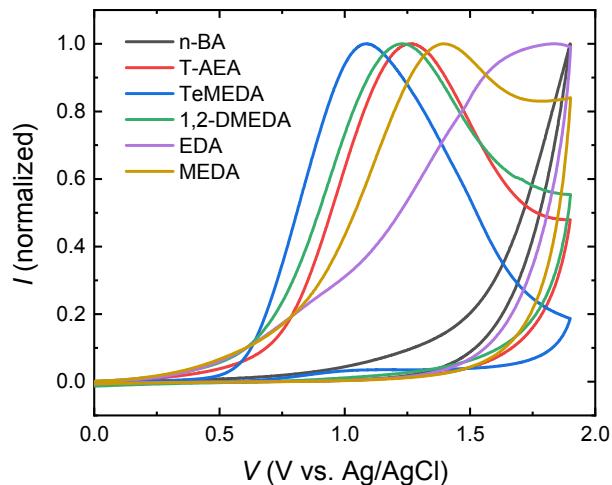
Compound	Supplier	Purity
triethylamine	Sigma Aldrich	>99.5%
<i>n</i> -butylamine	Acros Organics	99+%
<i>N,N,N',N'</i> -tetramethyl-1,2-ethanediamine	Sigma Aldrich	~99%
<i>N,N,N'</i> -trimethyl-1,2-ethanediamine	Acros organics	97%
<i>N,N</i> -dimethyl-1,2-ethanediamine	Aldrich	≥98%
<i>N,N'</i> -dimethyl-1,2-ethanediamine	Aldrich	99%
<i>N</i> -methyl-1,2-ethanediamine	TCI Chem.	
1,2-ethanediamine	Acros Organics	99+%
1,5-pentanediamine	Sigma Aldrich	95%
1,2-propanediamine	Sigma Aldrich	99%
<i>N</i> -(2-aminoethyl)-1,2-ethanediamine	Sigma Aldrich	99%
<i>N,N</i> -bis(2-aminoethyl)-1,2-ethanediamine	Fluka	98%
<i>N</i> -(2-aminoethyl)- <i>N</i> -methyl-1,2-ethanediamine	TCI-EP	
cis-1,4-cyclohexanediamine	TCI	
trans-1,2-cyclohexanediamine	Combi Blocks	98%
<i>p</i> -phenylenediamine	Sigma	
4-(1,3-dimethyl-2,3-dihydro-1H-benzoimidazol-2-yl)phenyl)	Aldrich	98%
dimethylamine		
branched poly(ethyleneimine) $M_w$ 700 (GPC)	Aldrich	
1,8-diazabicyclo(5.4.0)undec-7-ene	Acros Organics	98%



**Figure S1.** The  $R_E$  and  $R_O$  of all tested compounds displayed in a graph for visual aid. Strong and mild dedopants are colored dark-green and light-green respectively.



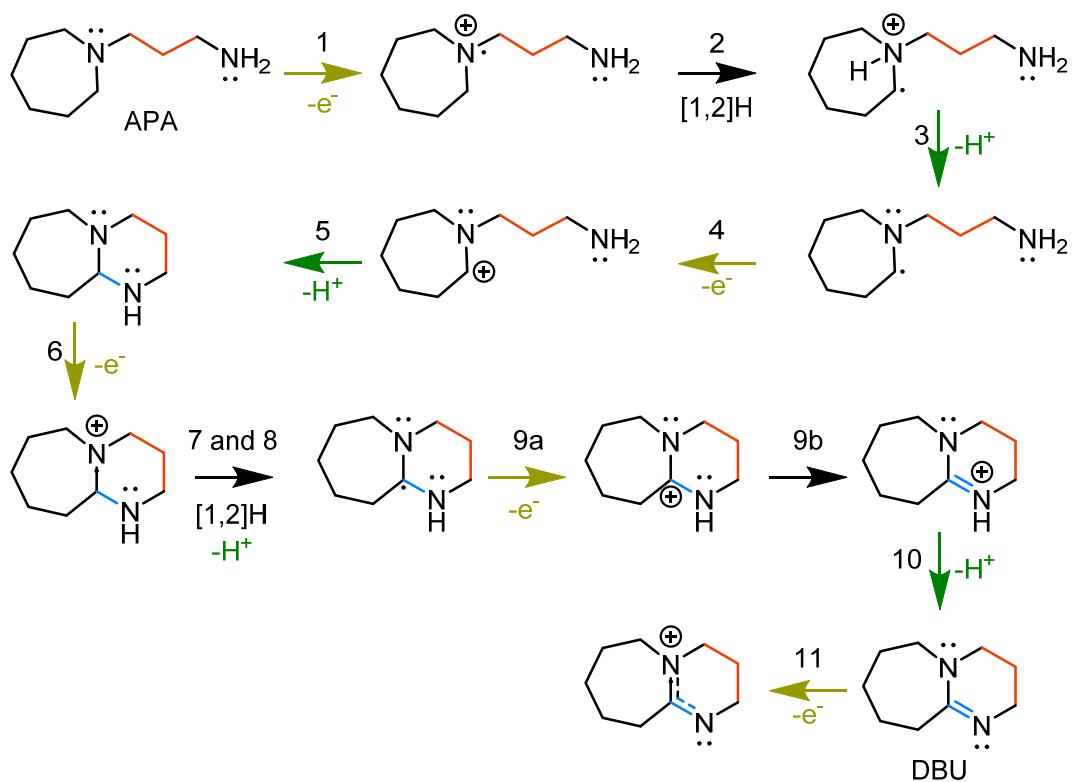
**Figure S2.** (a) Semi-logarithmic  $I$ - $V$  characteristics and (b) UV-vis-NIR spectra of pristine and dedoped PEDOT:PSS layers using EDA as dedopant under ambient and  $N_2$  atmosphere.



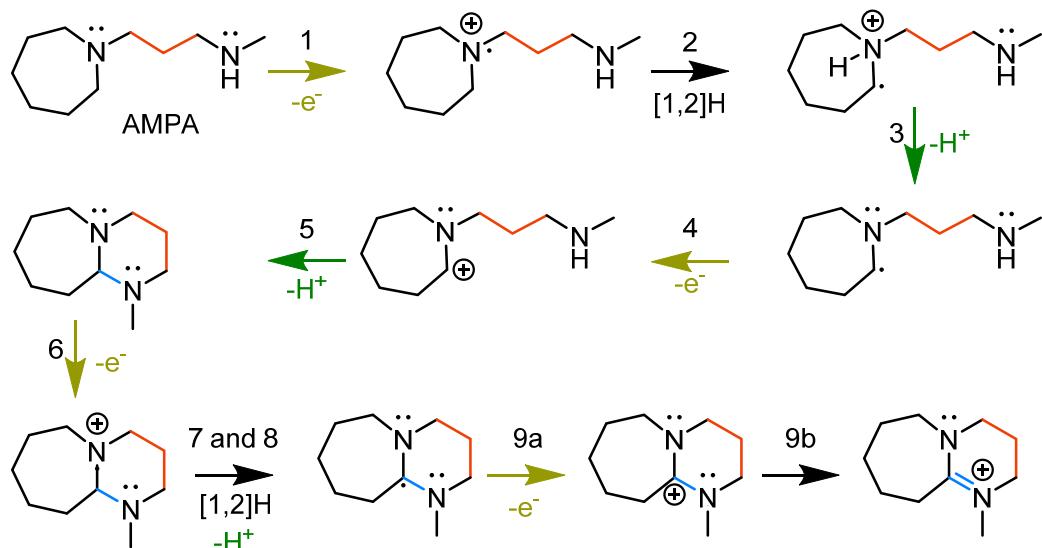
**Figure S3.** Cyclic voltammograms for selected amines.

**Table S3.** Oxidation potentials of selected amines from cyclic voltammetry (Figure S3).

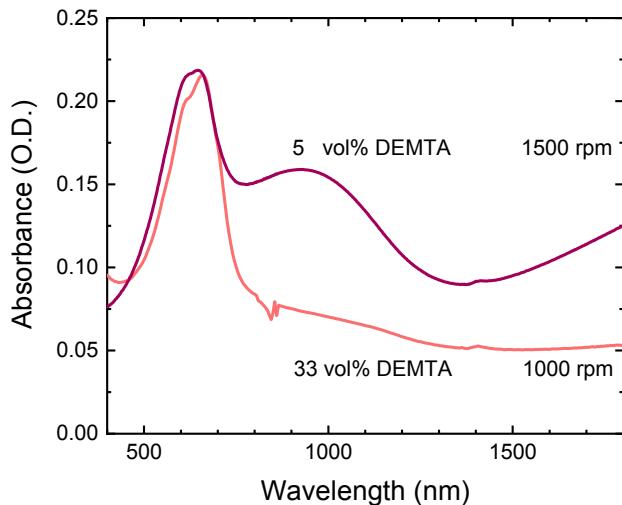
Amine	Oxidation Onset (V)	Peak (V)
n-BA	1.55	N/A
T-AEA	0.70	1.26
TeMEDA	0.61	1.08
1,2-DMEDA	0.68	1.23
EDA	0.84	1.84
MEDA	0.80	1.39



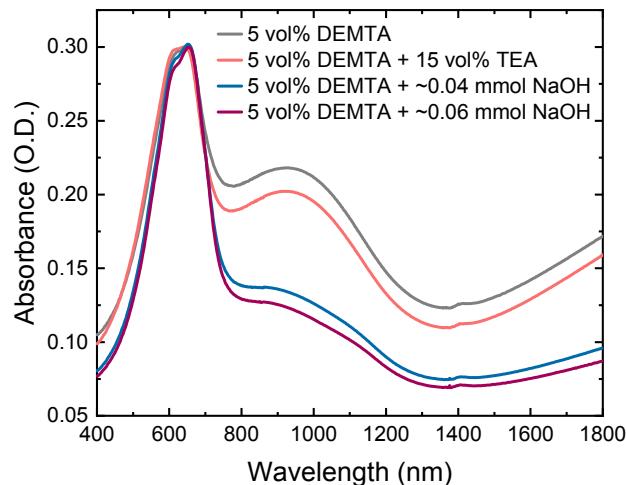
**Figure S4.** Chain of reactions for the oxidation of APA as derived from the mechanism.



**Figure S5.** Chain of reactions for the oxidation of AMPA as derived from the mechanism.



**Figure S6.** The absorbance spectra of layers of PEDOT:PSS spin-coated from a dispersions of aqueous PEDOT:PSS/EG/DEMTA in 0.63:0.03:0.33 and 0.90:0.05:0.05 (v/v/v)) spin coated at 1000 and 1500 rpm respectively.



**Figure S7.** UV-vis-NIR absorption spectra of PEDOT:PSS layers spin-coated from dispersions of PEDOT:PSS with 5 vol% DEMTA (1 mL total volume), without or with additional base added. Spectra were normalized at  $A = 0.3$ .