## Probing the Self-Assembly and Nonlinear Friction Behavior of Confined Gold Nano-Particles

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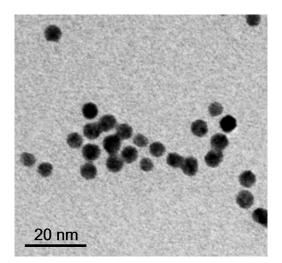


Figure S1. Typical transmission electron microscopy images of Au NPs purchased from Cytodiagnostics Inc.

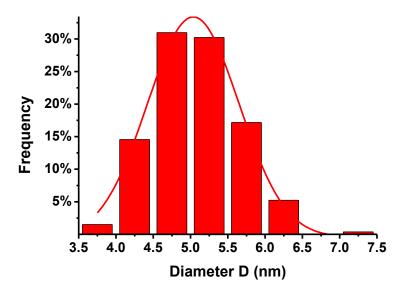
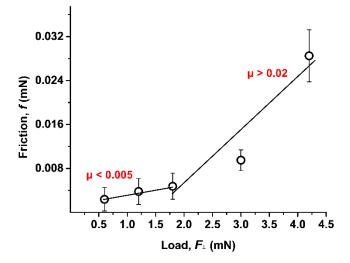
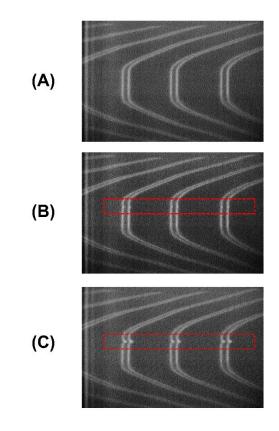


Figure S2. Histograms and the fitted Gaussian distributions of measured size of Au NPs  $(\sim 270 \text{ particles})$  used in this work.



**Figure S3.** Friction force (*f*) as a function of normal load ( $F_{\perp}$ ) for two mica surfaces injecting Au nanoparticles, the shear velocity v = 140 nm/s



**Figure S4.** The change of FECO images during continuous sliding ( $F \perp > 0.6 \text{ mN}$ ), the deformation shown in C proved the damage of mica substrate at the sliding zone,

The equations used for calculating Debye length ( $\lambda_D$ ) are shown in Equation S1-S2 as follows,<sup>1</sup>

$$\lambda_{\rm D} = \sqrt{\frac{\varepsilon_r \varepsilon_0 k_{\rm B} T}{N_{\rm A} q_0^2 I}} \tag{S1}$$

$$I = \frac{1}{2} \sum_{i=1}^{n} M_i z_i^2$$
 (S2)

where  $\varepsilon_r$  is the relative permittivity of the aqueous medium,  $\varepsilon_0$  is the vacuum permittivity, *T* is the temperature in Kelvin,  $k_B$  is the Boltzmann constant, N<sub>A</sub> is the Avogadro's number,  $q_0$  is the elementary electric charge, *n* is number of ion species in the solution, *I* represents the ionic strength of the electrolytic buffer solution, and  $M_i$  and  $z_i$  represent the ion concentration and the number of ion charge, respectively.

## References

 Israelachvili, J. N., *Intermolecular and Surface Forces*. Academic press: 2011; p 704.