

Supporting Information

Effect of ceramic-target crystallinity on metal-to-insulator transition of epitaxial rare-earth nickelate films grown by pulsed laser deposition

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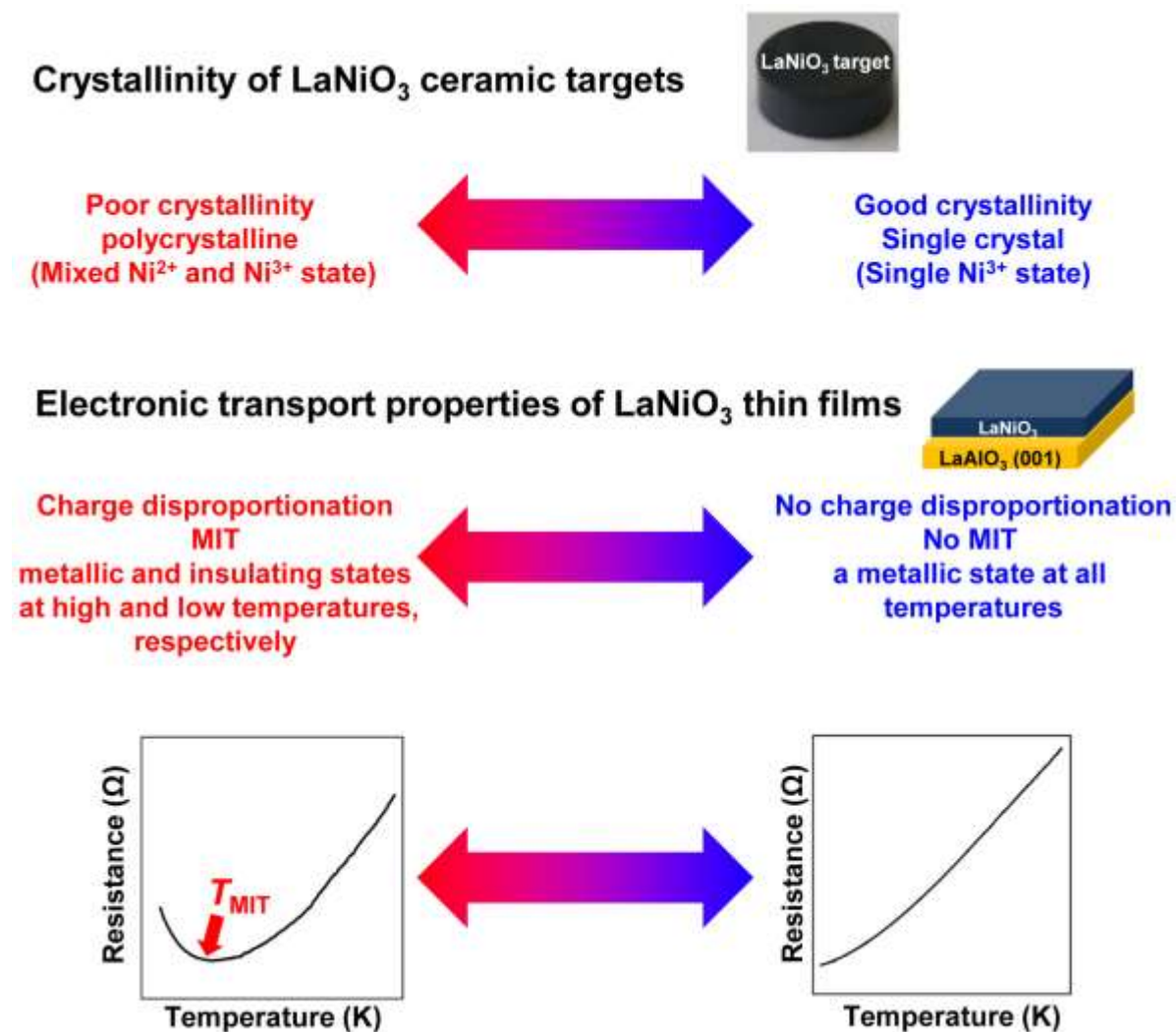
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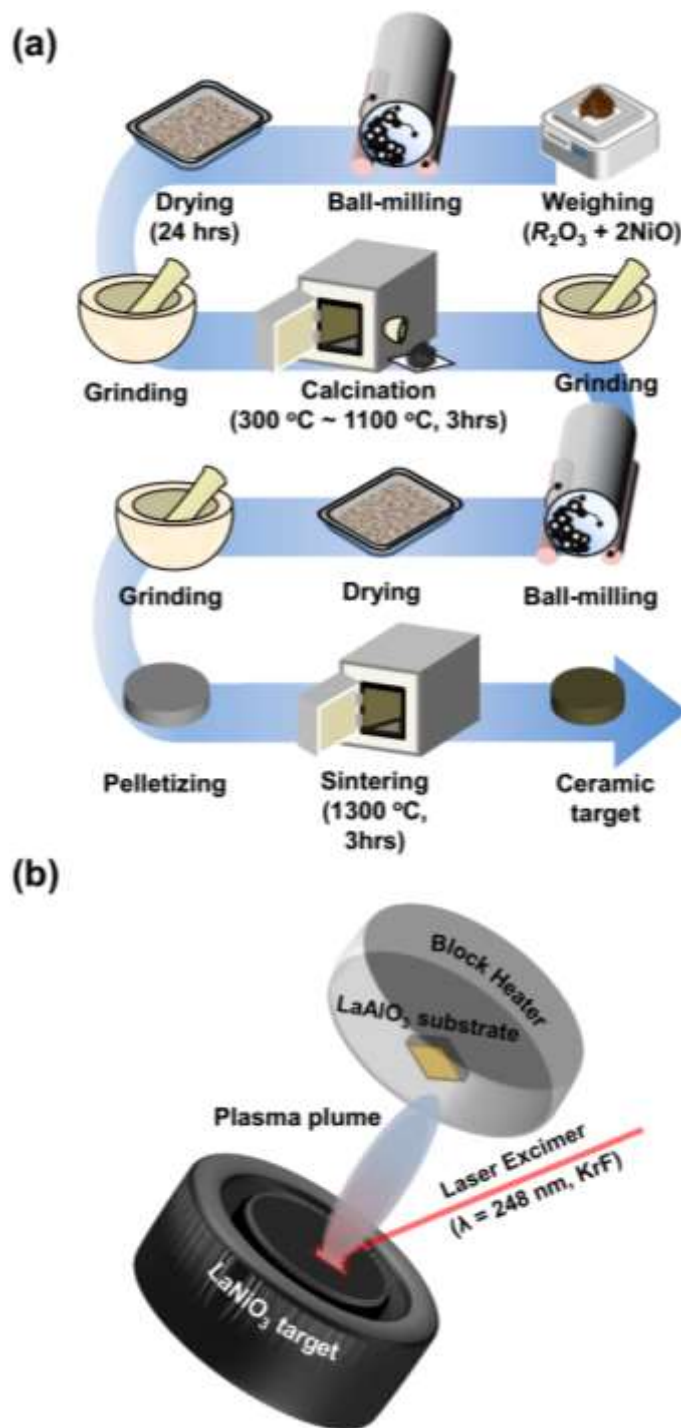
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Hypothesis of our study



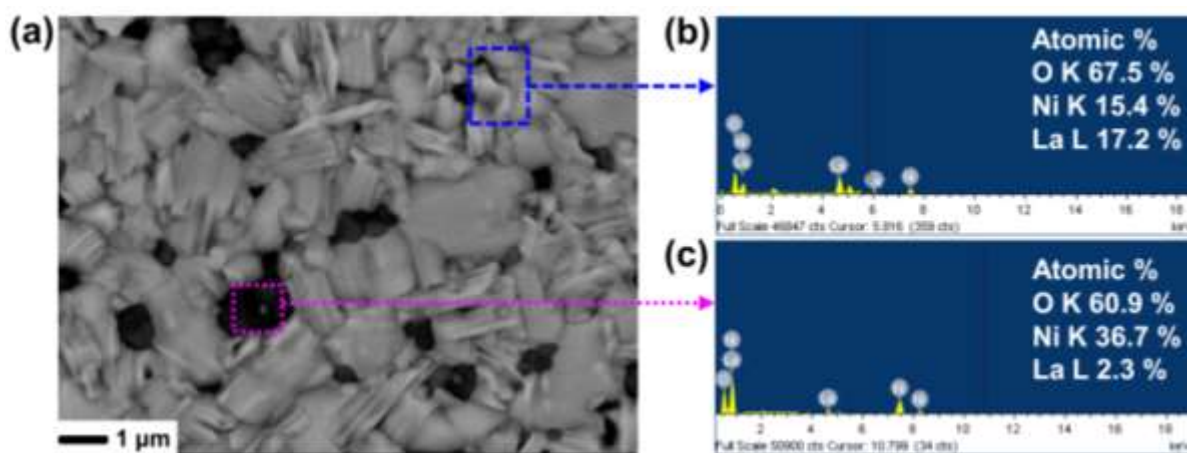
Supplementary Fig. S1. A schematic diagram of tuning of the metal-to-insulator phase transitions in LaNiO_3 thin films via the controlled crystallinity of ceramic targets in pulsed laser deposition (PLD).

Schematic illustrations of ceramic-target synthesis and thin-film growth in LaNiO_3



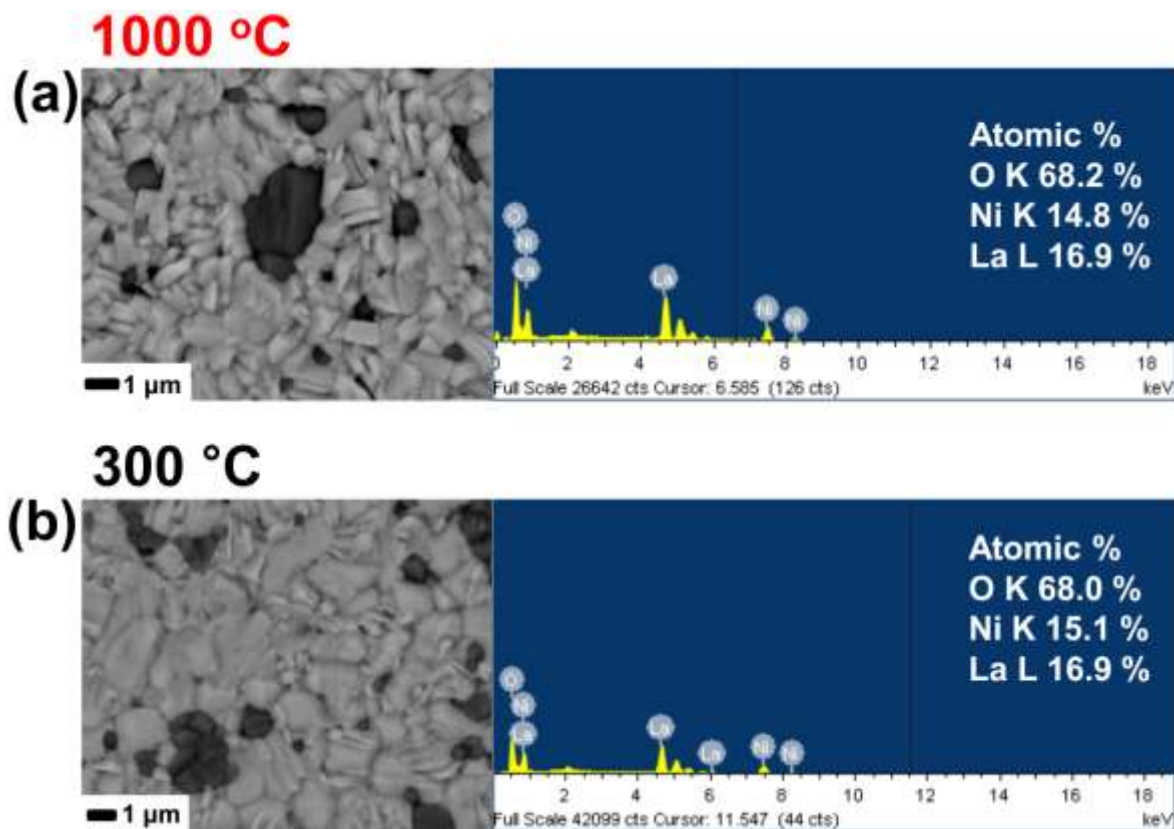
Supplementary Fig. S2. (a) A schematic diagram of the powder calcination and the ceramic-target synthesis in the solid-state-reaction process. (b) A schematic illustration of the pulsed laser deposition (PLD) growth of epitaxial LaNiO_3 (001) films.

Local elemental analyses of the as-sintered LaNiO_3 ceramic target using energy dispersive spectroscopy (EDS)



Supplementary Fig. S3. (a) The back-scattered surface image of a LaNiO_3 ceramic target measured by field emission-scanning electron microscope (FE-SEM). The local energy dispersive spectroscopy (EDS) of (b) the gray region [marked by the blue dashed rectangle], and (c) the black region [marked by the pink dotted square] in the back-scattered SEM image of (a). In the black region, the Ni content is exceedingly excessive, whereas the La content is very deficient.

The overall stoichiometry analyses on LaNiO_3 target surfaces (sintered after the calcination at 300 and 1000 °C)



Supplementary Fig. S4. The back-scattered FE-SEM images and the corresponding EDS spectra of the as-sintered LaNiO_3 target surfaces after the high-temperature and low-temperature calcination at (a) 1000 and (b) 300 °C, respectively. The overall La and Ni contents between these two LaNiO_3 targets (calcined at 300 and 1000 °C) are identical within the measurement error.

Thickness estimation of epitaxial LaNiO₃ (001) films

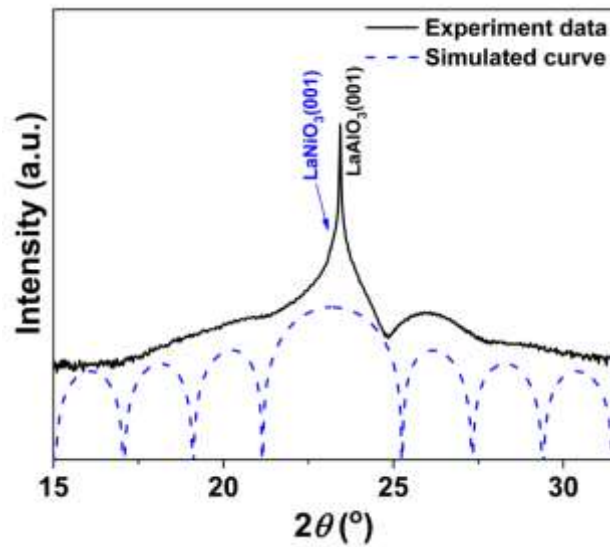
We estimated the film thickness of epitaxial LaNiO₃ (001) films by simulating thickness fringes around the (001) Bragg peak in their X-ray diffraction (XRD) θ - 2θ spectra, as shown in the Supplementary Figure S5. For the simulation of the measured XRD intensity (I) in the thickness fringes, we used the following formulas:

$$I(q_z) \propto \frac{\sin^2(q_z Nd/2)}{\sin^2(q_z d/2)} \quad (1)$$

$$q_z = L - Q_z \quad (2)$$

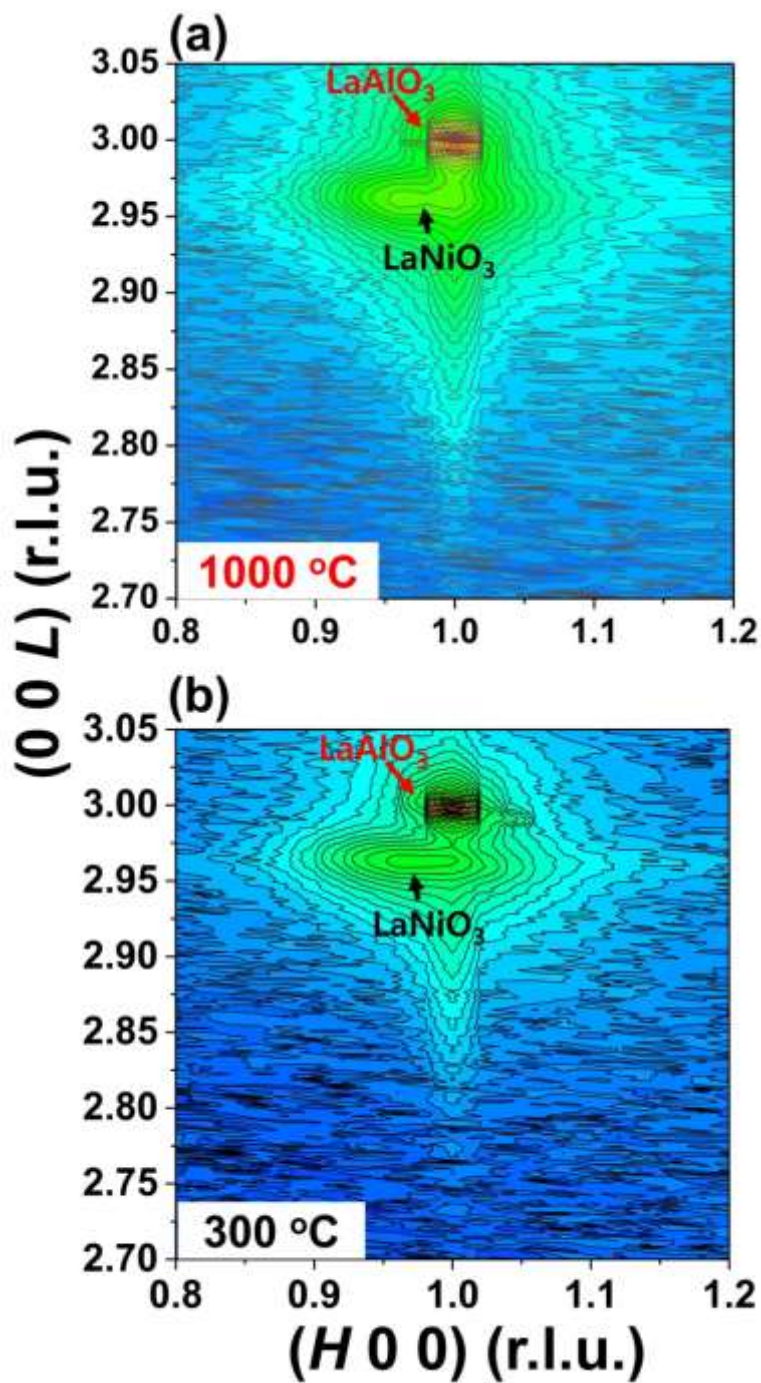
$$Q_z = \frac{4\pi}{\lambda} \sin \theta \quad (3)$$

where N represents the film thickness (the number of unit cells), d is the lattice d -spacing in the normal direction to the diffraction plane, λ is the X-ray wavelength (Cu K α_1 = 1.5406 Å), and $L = \frac{4\pi}{\lambda} \sin \theta_B$ (Here, θ_B is the diffraction angle at the Bragg's condition). By comparing the experimental XRD data (the black solid curve) with the simulated thickness fringe (the blue dashed curve), we determined that the estimated thickness of an ultrathin LaNiO₃ (001) film is 10 unit cells, that is, ~3.84 nm.



Supplementary Fig. S5. Estimating the film thickness in an ultrathin LaNiO₃ (001) film. The measured XRD intensity (the black solid curve) around the (001) Bragg peak is compared with the simulated intensity (the blue dashed curve) of the oscillating thickness fringes.

High-resolution reciprocal space mappings (RSMs) of the LaNiO_3 thin films

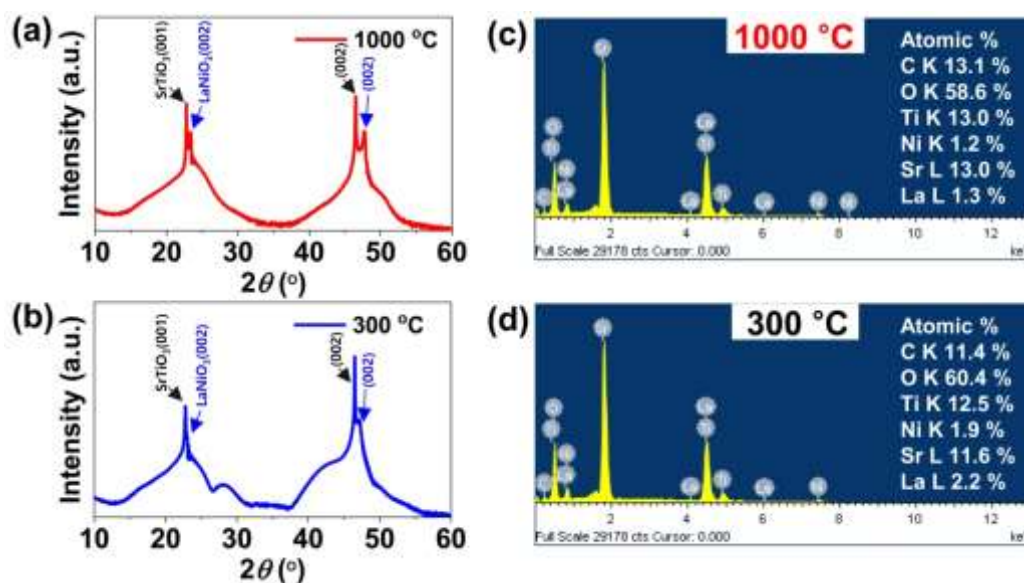


Supplementary Fig. S6. The RSMs of two epitaxial LaNiO_3 (001) films, synthesized from as-sintered LaNiO_3 ceramic targets after (a) high-temperature (1000 °C) and (b) low-temperature (300 °C) calcination, around the (103) Bragg peaks of the underlying LaAlO_3 substrates in pseudocubic notation. In (a) [(b)], the measured out-of-plane and in-plane lattice constants are about 3.834 (3.834) and 3.789 (3.789) Å, respectively.

The surface stoichiometry analyses of the as-grown LaNiO₃ thin films fabricated from different LaNiO₃ ceramic targets sintered after high-temperature (1000 °C) and low-temperature (300 °C) calcination

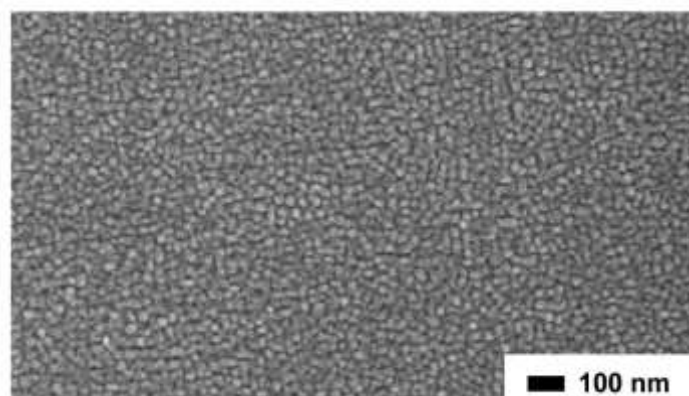
To examine the chemical stoichiometry of the as-grown LaNiO₃ thin films using EDS, we epitaxially grew two LaNiO₃ films on SrTiO₃ (001) substrates with two different LaNiO₃ ceramic targets. In LaNiO₃ films grown on LaAlO₃ (001) substrates, it is very difficult for us to estimate the accurate La content inside the LaNiO₃ film layer, because the incorporated La atoms in the LaAlO₃ substrates generate strong background signals in the raw EDS spectra. To avoid the strong background signals arising from the underlying substrate, we fabricated LaNiO₃ films on SrTiO₃ substrates composed of different cations (i.e., Sr and Ti elements).

The supplementary figure S7 shows the measured EDS spectra of epitaxial LaNiO₃ films grown on SrTiO₃ (001) substrates. Here, two LaNiO₃/SrTiO₃ (001) films were fabricated from two LaNiO₃ ceramic targets sintered after the high-temperature (1000 °C) and low-temperature (300 °C) calcination. However, the measured EDS intensity of the La and Ni atoms was very weak due to the dominant signals of the Sr and Ti atoms inside the SrTiO₃ substrate. Nevertheless, it is worthy of note that the atomic percentages of the La and Ni elements are nearly identical to 1.25 (2.19) and 1.24 (1.92) % in the LaNiO₃/SrTiO₃ (001) film deposited with the LaNiO₃ target after the high-temperature (low-temperature) calcination. This indicates that the La and Ni contents in the as-grown LaNiO₃ films are stoichiometric with no La or Ni vacancies.



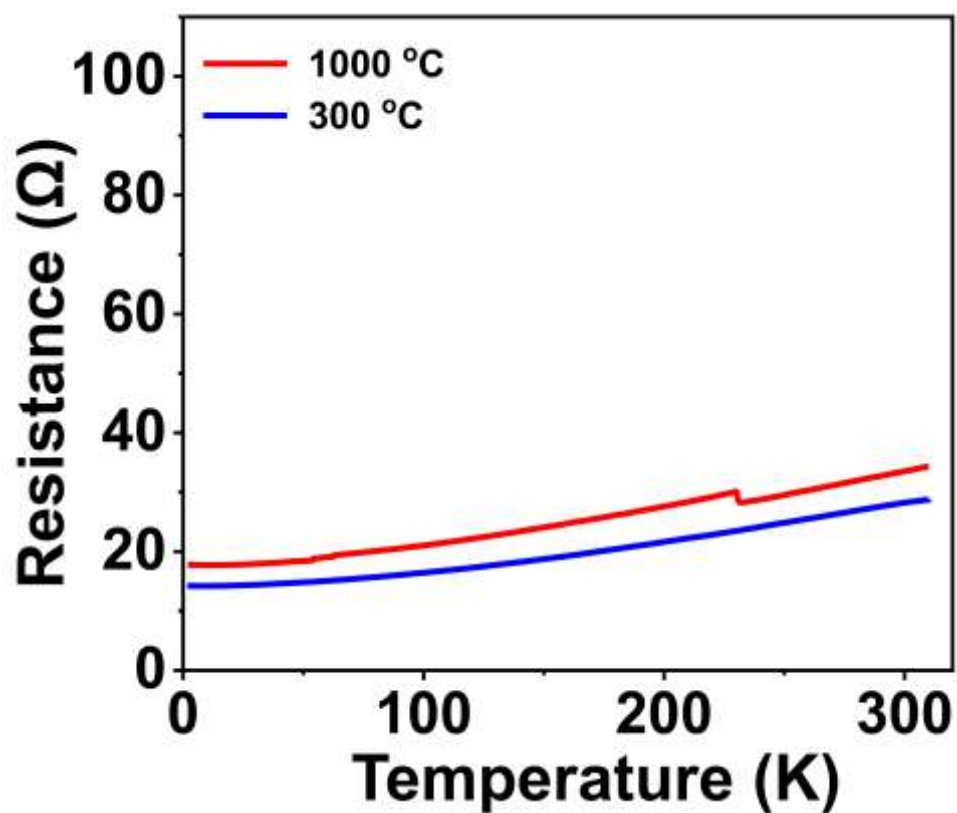
Supplementary Fig. S7. XRD θ -2 θ scans of the as-grown LaNiO₃/SrTiO₃ (001) films fabricated from two LaNiO₃ ceramic targets sintered after the (a) high-temperature (1000 °C) and (b) low-temperature (300 °C) calcination. (c,d) The corresponding EDS spectra of the as-deposited LaNiO₃/SrTiO₃ (001) film surfaces in (a) and (b), respectively.

The surface morphology of the LaNiO₃ (001) thin film



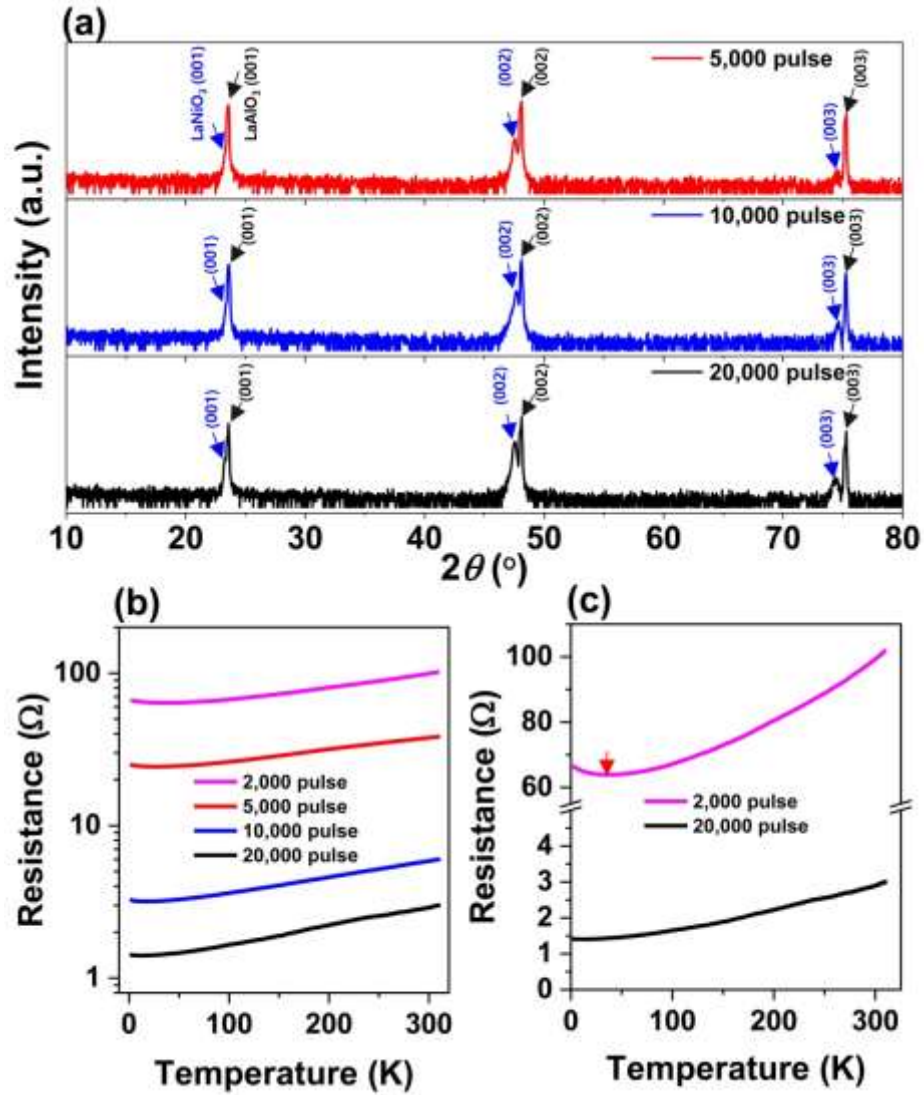
Supplementary Fig. S8. The FE-SEM image of the surface topography of the as-grown LaNiO₃ (001) thin film.

The electrical transport measurements of LaNiO_3 (001) thin films



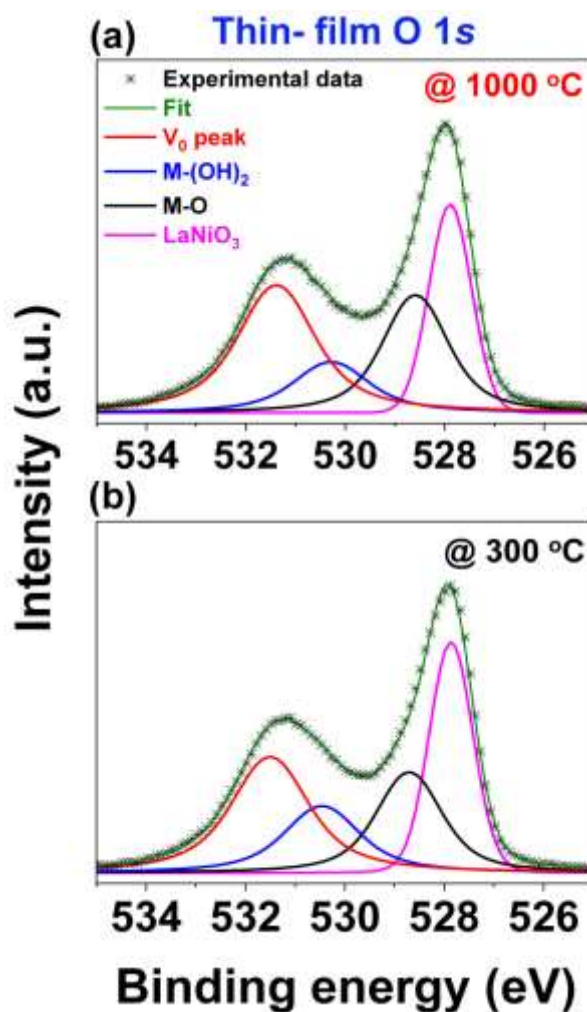
Supplementary Fig. S9. The temperature-dependent resistance of two LaNiO_3 (~10 unit cells) (001) films which are twice thicker than LaNiO_3 (~5 unit cells) (001) films used in Fig. 4(d).

Thickness-dependent transport behaviors of LaNiO₃ (001) thin films



Supplementary Fig. S10. (a) XRD θ - 2θ scans of epitaxial LaNiO₃/LaAlO₃ (001) films with different thickness. Here, the film thickness is manipulated by the number (N_{pulse}) of the applied laser pulses. (b) The thickness dependence of electrical transport properties of the as-grown LaNiO₃/LaAlO₃ (001) films. (c) A comparison of the transport properties between two LaNiO₃ films with different thickness. The thinner LaNiO₃ film ($N_{\text{pulse}} = 2000$ pulses) exhibits a metal-to-insulator transition (a red arrow), while the thicker LaNiO₃ film ($N_{\text{pulse}} = 20000$ pulses) is metallic at all temperatures.

The x-ray photoelectron spectroscopy (XPS) analysis of LaNiO₃ (001) thin films



Supplementary Fig. S11. XPS spectra at the O 1s edge of two as-grown LaNiO₃ (001) films [the same samples used in Figs. 5(a) and 5(b)] fabricated from two different LaNiO₃ ceramic targets sintered after (a) high-temperature (1000 °C) and (b) low-temperature (300 °C) calcination. The red, blue, black, and magenta solid lines represent the fitted curves of the oxygen vacancy (V_O) peak, the metal hydrate (M-(OH)₂) peak, the metal oxide (M-O) peak, and the lattice oxygen peak in LaNiO₃, respectively.