Construction of Type III-C Rotaxane-Branched Dendrimers and Their Anion-Induced Dimension Modulation Feature

Xu-Qing Wang[†], Wei-Jian Li[†], Wei Wang^{†*}, Jin Wen[§], Ying Zhang[‡], Hongwei Tan[‡], Hai-Bo Yang^{†*}

[†]Shanghai Key Laboratory of Green Chemistry and Chemical Processes, School of Chemistry and Molecular Engineering, Chang-Kung Chuang Institute, East China Normal University, 3663 N. Zhongshan Road, Shanghai 200062, P. R. China § Institute of Organic Chemistry and Biochemistry, Academy of Sciences of the Czech Republic, 16610 Prague 6, Czech Republic

[‡]Department of Chemistry, Beijing Normal University, Beijing, 100050, P.R. China

*To whom correspondence should be addressed: E-mail: <u>weiwang@stu.ecnu.edu.cn</u> (W. Wang); <u>hbyang@chem.ecnu.edu.cn</u> (H.-B. Yang)

Table of Contents (62 Pages)

Section A. Materials and general methods						
Section	on B. Synthesis and characterization of switchable building block [2]rotaxane	S4				
Section	on C. Synthesis and characterization of Type III-C rotaxane-branched dendrimers	S15				
1.	Synthesis of Type III-C rotaxane-branched dendrimers G1-G4	S15				
2	NMR, MS and GPC spectra of Type III-C rotaxane-branched dendrimers G1-G4	S20				
3.	2-D DOSY and DLS spectra of Type III-C rotaxane-branched dendrimers $G1$ - $G4$	S29				
4.	TEM images for Type III-C rotaxane-branched dendrimers G1-G4	S33				
Section	on D. Anion-induced dimension modulation of Type III-C rotaxane-branched	S35				
dend	rimers.					
Section E. Synthesis of the model rotaxane-branched dendrimer G1-a						
Section F. Supplementary data						
Section	on G. References	S62				

Section A. Materials and general methods

All reagents were commercially available and used as supplied without further purification, compounds S_1 , S_2 , S_3 and S_4 were prepared according to the published procedures.^{S1-S3} Deuterated solvents were purchased from Cambridge Isotope Laboratory (Andover, MA).

All solvents were dried according to standard procedures and all of them were degassed under N₂ for 30 minutes before use. All air-sensitive reactions were carried out under inert N₂ atmosphere. ¹H NMR, ¹³C NMR and ³¹P NMR spectra were recorded on Bruker 300 MHz Spectrometer (¹H: 300 MHz; ³¹P: 122 MHz), Bruker 400 MHz Spectrometer (¹H: 400 MHz; ¹³C: 101 MHz, ³¹P: 162 MHz) and Bruker 500 MHz Spectrometer (¹H: 500 MHz; ¹³C: 126 MHz, ³¹P: 202 MHz) at 298 K. The ¹H and ¹³C NMR chemical shifts are reported relative to residual solvent signals, and ³¹P {¹H} NMR chemical shifts are referenced to an external unlocked sample of 85% H₃PO₄ (δ 0.0). 2D NMR spectra (¹H-¹H COSY, ROESY and DOSY) were recorded on Bruker 500 MHz Spectrometer (¹H: 500 MHz) at 298 K. The MALDI MS experiments were carried out on a Bruker UltrafleXtreme MALDI TOF/TOF Mass Spectrometer (Bruker Daltonics, Billerica, MA), equipped with smartbeam-II laser. All spectra were measured in positive reflectron or linear mode. All the AFM images were obtained on a Dimension FastScan (Bruker), using ScanAsyst mode under ambient condition.

General procedure of preparing TEM samples: the solution of Type III-C rotaxane-branched dendrimers **G1-G4** in THF ($c = 10^{-5}$ M) were prepared, the TEM samples were deposited on copper grids, followed by a slow evaporation in air at room temperature. TEM measurements were performed under a Tecnai G2 20 TWIN device.

General procedure of preparing AFM samples: the solution of Type III-C rotaxane-branched dendrimers **G1-G4** in THF ($c = 10^{-7}$ M) were prepared, the samples were prepared by drop casting method using mica sheet as substrate. The measurement was under ambient condition using ScanAsyst mode.

Section B. Synthesis and characterization of switchable building block [2]rotaxane.

Scheme S1: The synthesis route of macrocycle 1.



Scheme S2: The synthesis route of [2]rotaxane 3.



Synthesis of 1: Mixing compound S_3 (2 g, 2.05 mmol) and S_1 (1.4 g, 5.11 mmol) in acetonitrile (predried by Na₂SO₄, 100 mL), then K₂CO₃ (2.26 g, 16.36 mmol) was added into the reaction flask. The resultant suspension was refluxed at 85 °C overnight. After cooling to room temperature, the reaction mixture was filtered and the filtrate was concentrated in vacuum. The resultant residue was

purified by column chromatography (SiO₂; PE/DCM) to yield a white solid **1** (1.37g, 49.3%). Mp: 76 °C. ¹H NMR (400 MHz, CDCl₃): δ 7.40-7.43 (d, J = 12.0 Hz, 4H), 6.77-6.82 (m, 14H), 3.62-4.02 (m, 42H), 1.95- 1.98 (m, 8H), 1.14 (s, 42H). ¹³C NMR (101 MHz, CDCl₃): δ 159.00, 150.83, 150.81, 150.76, 150.69, 149.92, 133.44, 128.30, 128.25, 128.22, 128.09, 115.69, 114.94, 114.24, 114.19, 114.05, 113.91, 107.14, 88.60, 67.86, 67.50, 55.78, 55.75, 29.71, 29.63, 26.33, 26.12, 18.66, 11.34. MS (ESI-TOF): Calculated for [**1**+H]⁺: 1379.75; Found: 1380.13.

Synthesis of S₅: Mixing compound S₄ (9.96 g, 27.28 mmol) and commercial 3, 5-dimethoxyphenol (4.21 g, 27.28 mmol) in acetonitrile (pre-dried by Na₂SO₄, 200 mL), then K₂CO₃ (15.07 g, 109.12 mmol) was added into the reaction flask. The resultant suspension was refluxed at 85 °C overnight. After cooling to room temperature, the reaction mixture was filtered and the filtrate was concentrated in vacuum. The resultant residue was purified by column chromatography (SiO₂; PE/DCM) to yield a white solid S₅ (8.46 g, 71%). ¹H NMR (300 MHz, CDCl₃): δ 7.82-7.85 (m, 2H), 7.69-7.72 (m, 2H), 6.07 (s, 3H), 3.88-3.92 (t, *J* = 6.0 Hz, 2H), 3.76 (s, 6H), 3.65-3.70 (t, *J* = 7.5 Hz, 2H), 1.67-1.79 (m, 4H), 1.29-1.42 (m, 12H). ¹³C NMR (101 MHz, CDCl₃): δ 168.41, 161.45, 161.03, 133.77, 132.17, 123.10, 93.37, 92.83, 68.01, 55.28, 55.26, 38.02, 29.40, 29.35, 29.28, 29.18, 29.10, 28.54, 26.80, 25.98. HRMS (EI-TOF): Calculated for [S₅]⁺: 439.2359; Found: 439.2353.

Synthesis of 2: A CHCl₃ (20 mL) solution of triphosgene (425 mg, 1.43 mmol) was added into a Schlenk flask, the Schlenk flask was evacuated and back-filled with N₂ three times. Then a CHCl₃ (20 mL) solution of 4-ethynylbenzenamine (508 mg, 4.34 mmol) and triethylamine (2.4 mL) were added into the reaction flask and stirred at 0 °C. After 4 h, adding the CHCl₃ (20 mL) solution of S₆ (1.34 g, 4.34 mmol) into the reaction flask, then the reaction mixture was allowed to warm to room temperature and stirred for 18 h. The solution was concentrated and the residue purified through chromatography (SiO₂; PE/acetone) to afford compound 2 as a pale-yellow solid (1.72 g, 87%). Mp: 111 °C. ¹H NMR (400 MHz, acetone-*d*₆): δ 8.03 (s, 1H), 7.49-7.51 (d, *J* = 6.0 Hz, 2H), 7.33-7.35 (d, *J* = 6.0 Hz, 2H), 6.07 (s, 3H), 5.83 (t, 1H), 3.92-3.95 (t, 2H), 3.74 (s, 6H), 3.49 (s, 1H), 3.17-3.22 (m, 2H), 1.72-1.75 (m, 2H), 1.29-1.52 (m, 14H). ¹³C NMR (126 MHz, acetone-*d*₆): δ 162.68, 162.17, 155.78, 142.58, 133.41, 118.47, 115.52, 94.14, 93.60, 84.68, 77.52, 68.58, 55.60, 40.50, 31.08, 27.68, 26.87. HRMS (EI-TOF): Calculated for [**2**]⁺: 452.2675; Found: 452.2679.

Synthesis of [2]rotaxane 3: A Schlenk flask was charged with macrocycle component 1 (1.5 g, 1.09 mmol), thread component 2 (98 mg, 0.22 mmol) and $Pt(PEt_3)_2I_2$ (596 mg, 0.87 mmol). The Schlenk flask was then evacuated and back-filled with N₂ three times. Next, the mixture solvent of degassed CHCl₃ and i-Pr₂NH (v/v, 6/3 mL) was added via syringe. The resultant solution was stirred for two

hours under -10 °C. Then a catalytic amount of CuI was added to the mixture under an inert atmosphere. The reaction mixture was allowed to warm to room temperature and stirred overnight. The solution was concentrated and the residue was purified by column chromatography (SiO₂; PE/DCM). A white solid **3** (400 mg, 77%) was obtained. Mp: 86 °C. ¹H NMR (300 MHz, CD₂Cl₂): δ 7.21-7.41 (m, 9H), 6.81-7.00 (m, 14H), 6.09 (s, 3H), 3.71-4.07 (m, 50H), 2.81 (t, 1H), 2.18-2.28 (m, 12H), 1.96-2.07 (m, 8H), 1.61-1.68 (m, 4H), 1.14-1.23 (m, 60H), 0.89-1.08 (m, 6H), 0.59 (m, 2H), -0.18 (m, 2H), -1.70 (m, 2H), -1.97 (m, 2H). ³¹P NMR (122 MHz, CD₂Cl₂): δ 8.91. ¹³C NMR (101 MHz, CD₂Cl₂): δ 162.11, 161.55, 159.76, 159.65, 154.36, 151.32, 151.12, 150.76, 150.70, 150.28, 150.21, 138.54, 133.81, 133.77, 131.42, 129.54, 129.20, 129.09, 129.03, 128.80, 128.74, 122.56, 119.33, 116.01, 115.89, 115.43, 114.82, 114.76, 113.66, 113.54, 107.65, 107.58, 100.17, 93.81, 92.90, 89.04, 88.94, 68.74, 68.51, 68.32, 68.21, 57.30, 56.85, 56.75, 55.70, 40.10, 31.45, 30.84, 30.48, 30.27, 29.86, 29.54, 27.13, 27.04, 26.86, 26.75, 26.53, 25.31, 18.90, 17.28, 17.11, 16.93, 11.83, 8.53. HRMS (MALDI-TOF-MS): Calculated for [**3**+H]⁺: 2389.0668; Found: 2389.0732.



Fig. S1 ¹H NMR spectrum (CD₂Cl₂, 298 K, 300 MHz) of [2]rotaxane 3.



Fig. S2 31 P NMR spectrum (CD₂Cl₂, 298 K, 122 MHz) of [2]rotaxane 3.



Fig. S3 ¹³C NMR spectrum (CD₂Cl₂, 298 K, 101 MHz) of [2]rotaxane 3.



Fig. S4 HRMS (MALDI-TOF-MS) spectrum of [2]rotaxane 3.



Fig. S5 2D ¹H-¹H COSY spectrum (THF-*d*₈, 298 K, 500 MHz) of [2]rotaxane **3**.



Fig. S6 2D ¹H-¹H ROESY spectrum (THF-*d*₈, 298 K, 500 MHz) of [2]rotaxane 3.



Fig. S7 Stacked ¹H NMR spectra (THF- d_8 , 298 K, 500 MHz) of anion-induced switching behavior of [2]rotaxane **3**. a) **3**; the mixture of **3** and TBAA, for each rotaxane unit: b) TBAA (1 equiv); c) TBAA (2 equiv); d) TBAA (3 equiv); e) TBAA (4 equiv); f) TBAA (5 equiv); and the mixture obtained after adding NaPF₆ to the solution in f), for each rotaxane unit: g) NaPF₆ (5 equiv); h) NaPF₆ (6 equiv); i) NaPF₆ (8 equiv); j) NaPF₆ (10 equiv).



Fig. S8 Partial ¹H NMR spectra (THF- d_8 , 298 K, 500 MHz) of anion-induced switching behavior of [2]rotaxane **3**. a) **3**; the mixture of **3** and TBAA, for each rotaxane unit: b) TBAA (1 equiv); c) TBAA (2 equiv); d) TBAA (3 equiv); e) TBAA (4 equiv); f) TBAA (5 equiv); and the mixture obtained after adding NaPF₆ to the solution in f), for each rotaxane unit: g) NaPF₆ (5 equiv); h) NaPF₆ (6 equiv); i) NaPF₆ (8 equiv); j) NaPF₆ (10 equiv).



Fig. S9 ¹H NMR spectra (THF- d_8 , 298 K, 500 MHz) of anion-induced switching behavior of [2]rotaxane 3. a) [2]rotaxane 3; b) 3 with the addition of TBAA (5 eq.); c) the mixture after adding NaPF₆(10 eq.) into the solution in b).



Fig. S10 (a) The expanded region of the ¹H NMR spectra (THF- d_8 , 298 K, 500 MHz) of [2]rotaxane **3** (Host) with the addition of acetate anion (Guest); (b) The 1H NMR titration isotherm of [2]rotaxane **3** with the addition of acetate anion (TBAA) recorded at 500 MHz in THF- d_8 at 298 K.



Fig. S11 2D ¹H-¹H COSY spectrum (THF- d_8 , 298 K, 500 MHz) of [2]rotaxane **3** with the addition of TBAA (5 eq.).



Fig. S12 2D ¹H-¹H COSY spectrum (THF- d_8 , 298 K, 500 MHz) of the mixture after adding NaPF₆ (10 eq.) into the solution of [2]rotaxane **3** and TBAA.



Fig. S13 Theoretically calculated structures of (a) [2]rotaxane **3**; (b) anion-induced switching motion of macrocycle component in [2]rotaxane **3**.

Section C. Synthesis and characterization of Type III-C rotaxane-branched dendrimers.

1. Synthesis of the first-generation Type III-C rotaxane-branched dendrimer G1.

Scheme S3: Synthesis route of Type III-C rotaxane-branched dendrimer G1.



Synthesis of the first-generation Type III-C rotaxane-branched dendrimer G1: A mixture of 1, 3, 5-triethynylbenzene (3.8 mg, 0.025 mmol) and [2]rotaxane 3 (200 mg, 0.084 mmol) were added in a Schlenk flask, the Schlenk flask was then evacuated and back-filled with N₂ three times. Next, degassed diethylamine (10 mL) and a catalytic amount of CuI was added under an inert atmosphere. The reaction was stirred overnight at room temperature. The solvent was evaporated and the residue was purified by column chromatography (SiO2; PE/DCM) and preparative gel permeation chromatography (GPC) to yield a yellow solid G1 (124 mg, 70%). ¹H NMR (400 MHz, CD₂Cl₂): δ 7.23-7.43 (m, 27H), 6.84-7.01 (m, 45H), 6.10-6.11 (m, 9H), 3.78-4.07 (m, 150H), 2.91 (t, 3H), 2.20-2.23 (m, 36H), 1.97-2.08 (m, 24H), 1.58-1.65 (m, 6H), 1.23-1.31 (m, 54H), 1.00-1.15 (m, 150H), 0.60 (m, 6H), -0.14 (m, 6H), -1.61 (m, 6H), -1.87 (m, 6H). ³¹P NMR (122 MHz, CD₂Cl₂): δ 11.64. ¹³C NMR (101 MHz, CD₂Cl₂): δ 162.12, 161.57, 159.79, 159.66, 154.51, 151.31, 151.12, 150.79, 150.30, 150.22, 138.12, 133.82, 133.79, 131.59, 129.51, 129.17, 129.07, 128.75, 119.37, 116.01, 115.87, 115.40, 114.85, 114.77, 113.55, 107.72, 107.61, 93.83, 92.88, 89.04, 88.90, 68.75, 68.52, 68.33, 68.22, 57.25, 56.81, 55.70, 55.67, 40.14, 31.44, 30.85, 30.45, 30.30, 30.14, 29.84, 29.51, 27.15, 27.06, 26.87, 26.66, 26.55, 25.40, 18.93, 17.07, 16.89, 16.72, 11.85, 8.63. MALDI-TOF-MS: Calculated for $[G1 + Na]^+$: m/z = 6,953.5 and $[G1 + H + Na]^{2+}$: m/z = 3,477.2; Found: *m*/*z* = 6952.7 and *m*/*z* = 3476.5.

2. Synthesis of the second-generation Type III-C rotaxane-branched dendrimer G2.

Scheme S4: Synthesis route of Type III-C rotaxane-branched dendrimer G2.



Synthesis of Type III-C rotaxane-branched dendrimer G2: A reaction flask was charged with a THF solution of G1 (100 mg, 0.0144 mmol) and then a solution of Bu₄NF·3H₂O (54 mg, 0.173 mmol) in THF was added dropwise into the reaction flask. The reaction mixture was stirred at room temperature for 2h, the obtained residue was washed by water, then dried with Na₂SO₄ and concentrated. The residue was further purified by column chromatography (SiO₂; DCM) and preparative gel permeation chromatography (GPC) to afford a pale-yellow solid G1-YNE. Then the obtained G1-YNE (60mg, 0.01 mmol) and [2]rotaxane 3 (160 mg, 0.066 mmol) were added in a Schlenk flask, the Schlenk flask was then evacuated and back-filled with N₂ three times. Next, degassed diethylamine (10 mL) and a catalytic amount of CuI were added under an inert atmosphere. The reaction was stirred overnight at room temperature. The solvent was evaporated and the residue was purified by column chromatography (SiO₂; DCM/EA) and preparative gel permeation chromatography (GPC) to yield a pale-yellow solid G2 (140 mg, 52%). Mp: 110 °C. ¹H NMR (500 MHz, CD₂Cl₂): δ 6.78-7.41 (m, 210H), 6.09-6.12 (m, 27H), 3.77-4.07 (m, 450H), 2.84-2.97 (m, 9H), 1.95-2.23 (m, 180H), 1.62-1.64 (m, 18H), 1.14-1.29 (m, 450H), 1.00-1.01 (m, 36H), 0.59 (m, 18H), -0.16 (m, 18H), -1.91-(-1.51) (m, 36H). ³¹P NMR (122 MHz, CD₂Cl₂): δ 11.68. ¹³C NMR (126 MHz, CD_2Cl_2): δ 162.06, 161.55, 161.51, 159.74, 159.61, 157.30, 157.23, 154.48, 154.40, 151.25, 151.05, 151.02, 150.73, 150.71, 150.66, 150.23, 150.15, 138.05, 133.79, 133.75, 132.21, 131.54, 129.47, 129.12, 129.02, 128.96, 128.74, 128.69, 123.13, 119.30, 115.94, 115.80, 114.79, 114.71, 114.59, 114.50, 113.57, 113.45, 107.65, 107.54, 93.73, 92.82, 89.00, 88.85, 68.68, 68.51, 68.47, 68.27, 68.16, 68.09, 57.25, 57.22, 56.81, 56.77, 56.69, 55.76, 55.67, 40.07, 31.43, 30.82, 30.45, 30.41, 30.26, 29.82, 29.78, 29.50, 29.39, 27.09, 27.00, 26.94, 26.82, 26.70, 26.66, 26.62, 26.49, 25.30, 18.88, 18.86, 16.95, 16.81, 16.67, 11.79, 11.78, 8.60, 8.58, 1.18. MALDI-TOF-MS: Calculated for [G2 + Na]⁺: m/z = 19,592.7 and [G2 + 2Na]²⁺: m/z = 9,807.9; Found: m/z = 19,592.5 and m/z = 9,809.0.

3. Synthesis of the third-generation Type III-C rotaxane-branched dendrimer G3. Scheme S5: Synthesis route of Type III-C rotaxane-branched dendrimer G3.



Synthesis of Type III-C rotaxane-branched dendrimer G3: A reaction flask was charged with a THF solution of G2 (107 mg, 0.0055 mmol) and then a THF solution of Bu₄NF·3H₂O (41 mg, 0.13 mmol) was added dropwise into the reaction flask, the reaction mixture was stirred at room temperature for 2h. The obtained residue was washed by water, then dried with Na₂SO₄ and concentrated. The residue was further purified by column chromatography (SiO₂; DCM) and preparative gel permeation chromatography (GPC) to afford a pale-yellow solid G2-YNE. Then the obtained G2-YNE (65mg, 0.0037 mmol) and [2]rotaxane 3 (115 mg, 0.048 mmol) were added in a Schlenk flask, the Schlenk flask was then evacuated and back-filled with N₂ three times. Next, the mixture solvent of degassed Et₂NH and CH₂Cl₂ (v/v, 5/5 mL) was added via syringe. Subsequently, a catalytic amount of CuI was added under an inert atmosphere. The reaction was stirred overnight at room temperature. The solvent was evaporated and the residue was purified by column chromatography (SiO₂; DCM/EA) and preparative gel permeation chromatography (500 MHz, CD₂Cl₂): δ 6.77-7.40 (m, 486H), 6.08-6.10 (m, 63H), 3.75-4.06 (m, 1050H), 2.83-2.97 (m, 21H), 2.16-2.20 (m, 252H), 1.95-2.05 (m, 168H),

1.00-1.27 (m, 1092H), 0.58 (m, 42H), -0.17 (m, 42H), -1.93-(-1.52) (m, 84H). ³¹P NMR (122 MHz, CD₂Cl₂): δ 11.86. ¹³C NMR (126 MHz, CD₂Cl₂): δ 162.04, 161.54, 161.50, 159.72, 159.60, 157.39, 154.50, 154.39, 151.23, 151.03, 150.71, 150.20, 150.13, 138.18, 133.77, 133.73, 132.21, 131.54, 130.23, 129.44, 129.09, 129.00, 128.67, 122.89, 119.26, 115.92, 115.78, 115.32, 114.77, 114.69, 114.59, 114.50, 113.57, 113.44, 107.63, 107.52, 93.72, 92.79, 88.98, 88.84, 68.66, 68.50, 68.46, 68.26, 68.15, 57.21, 56.78, 56.74, 56.67, 55.75, 55.65, 40.06, 32.32, 31.69, 31.40, 30.81, 30.42, 30.34, 30.25, 30.18, 30.09, 29.91, 29.78, 29.72, 29.48, 29.37, 27.56, 27.07, 26.98, 26.80, 26.63, 26.48, 25.31, 23.09, 18.86, 16.95, 16.81, 16.67, 14.29, 11.77, 8.57, 1.16.

4. Synthesis of the fourth-generation Type III-C rotaxane-branched dendrimer G4.

Scheme S6: Synthesis route of Type III-C rotaxane-branched dendrimer G4.



Synthesis of Type III-C rotaxane-branched dendrimer G4: A reaction flask was charged with a THF solution of G3 (33 mg, 0.74 μ mol) and then a THF solution of Bu₄NF·3H₂O (11 mg, 0.0355 mmol) was added dropwise into the reaction flask, the reaction mixture was stirred at room

temperature for 2h. The obtained residue was washed by water, then dried with Na₂SO₄ and concentrated. The residue was further purified by column chromatography (SiO₂; DCM/EA) and preparative gel permeation chromatography (GPC) to afford a pale-yellow solid G3-YNE. Then the obtained G3-YNE (15 mg, 0.37 µmol) and [2]rotaxane 3 (25 mg, 0.0105 mmol) were added in a Schlenk flask, the Schlenk flask was then evacuated and back-filled with N2 three times. Next, the mixture solvent of degassed Et₂NH and CH₂Cl₂ (v/v, 3/3 mL) was added via syringe. Subsequently, a catalytic amount of CuI was added under an inert atmosphere. The reaction was stirred overnight at room temperature. The solvent was evaporated and the residue was purified by column chromatography (SiO₂; DCM/EA) and preparative gel permeation chromatography (GPC) to yield a pale-yellow solid G4 (20 mg, 29%). ¹H NMR (400 MHz, CD₂Cl₂): δ 6.82-7.40 (m, 1038H), 6.06-6.08 (m, 135H), 3.75-4.06 (m, 2250H), 3.02 (m, 90H), 2.82 (m, 45H), 1.94-2.19 (m, 900H), 0.97-1.35 (m, 2178H), 0.58 (m, 90H), -0.18 (m, 90H), -1.67 (m, 90H), -1.94 (m, 90H). ³¹P NMR (122 MHz, CD₂Cl₂): δ 11.66. ¹³C NMR (126 MHz, CD₂Cl₂): δ 162.05, 161.54, 161.50, 159.72, 159.61, 151.26, 151.04, 150.72, 150.21, 150.15, 133.78, 133.74, 132.26, 131.58, 129.48, 129.12, 129.02, 128.68, 119.29, 115.93, 115.79, 114.78, 114.70, 114.63, 114.53, 113.48, 107.63, 107.52, 93.74, 93.72, 92.82, 89.00, 88.86, 68.69, 68.51, 68.46, 68.27, 68.16, 57.26, 56.78, 56.71, 55.76, 55.66, 40.06, 31.43, 30.81, 30.45, 30.24, 30.09, 29.81, 29.37, 27.08, 26.98, 26.81, 26.70, 26.48, 23.10, 18.87, 18.84, 16.98, 16.84, 16.71, 14.29, 11.78, 11.77, 8.59, 1.17.



Fig. S14 ¹H NMR spectrum (THF-*d*₈, 298 K, 400 MHz) of G1.



Fig. S15 ³¹P NMR spectrum (THF-*d*₈, 298 K, 122 MHz) of G1.



Fig. S16 13 C NMR spectrum (CD₂Cl₂, 298 K, 101 MHz) of G1.



Fig. S17 ¹H NMR spectrum (CD₂Cl₂, 298 K, 500 MHz) of G2.



Fig. S18 31 P NMR spectrum (CD₂Cl₂, 298 K, 122 MHz) of G2.



Fig. S19 13 C NMR spectrum (CD₂Cl₂, 298 K, 126 MHz) of G2.



Fig. S20 ¹H NMR spectrum (CD₂Cl₂, 298 K, 500 MHz) of G3.



Fig. S21 ³¹P NMR spectrum (CD₂Cl₂, 298 K, 122 MHz) of G3.



Fig. S22 ¹³C NMR spectrum (CD₂Cl₂, 298 K, 126 MHz) of G3.



Fig. S23 ¹H NMR spectrum (CD₂Cl₂, 298 K, 400 MHz) of G4.



Fig. S24 ³¹P NMR spectrum (CD₂Cl₂, 298 K, 122 MHz) of G4.



Fig. S25 ¹³C NMR spectrum (CD₂Cl₂, 298 K, 126 MHz) of G4.



Fig. S26 ³¹P NMR spectra (CD₂Cl₂, 298 K, 122 MHz) of Type III-C rotaxane-branched dendrimer **G1** (*top*) and [2]rotaxane **3** (*bottom*).



Fig. S27 Stacked ³¹P NMR spectra (CD₂Cl₂, 298 K, 122 MHz) of Type III-C rotaxane-branched dendrimers: a) G1; b) G2; c) G3; d) G4.



Fig. S28 GPC spectrum of Type III-C rotaxane-branched dendrimer G1.



Fig. S29 GPC spectrum of Type III-C rotaxane-branched dendrimer G2.



Fig. S30 GPC spectrum of Type III-C rotaxane-branched dendrimer G3.



Fig. S31 GPC spectrum of Type III-C rotaxane-branched dendrimer G4.



Fig. S32 2-D DOSY spectrum (THF- d_8 , 298 K, 500 MHz) of [2]rotaxane 3. The diffusion coefficient D is $(14.13 \pm 0.14) \times 10^{-10} \text{ m}^2\text{s}^{-1}$.



Fig. S33 2-D DOSY spectrum (THF- d_8 , 298 K, 500 MHz) of G1.The diffusion coefficient D is $(10.96\pm0.24)\times 10^{-10} \text{ m}^2\text{s}^{-1}$.



Fig. S34 2-D DOSY spectrum (THF- d_8 , 298 K, 500 MHz) of G2. The diffusion coefficient D is $(5.13\pm0.12) \times 10^{-10} \text{ m}^2\text{s}^{-1}$.



Fig. S35 2-D DOSY spectrum (THF- d_8 , 298 K, 500 MHz) of G3. The diffusion coefficient D is $(2.75\pm0.16) \times 10^{-10} \text{ m}^2\text{s}^{-1}$.



Fig. S36 2-D DOSY spectrum (THF- d_8 , 298 K, 500 MHz) of G4. The diffusion coefficient D is $(1.66\pm0.15) \times 10^{-10} \text{ m}^2\text{s}^{-1}$.



Fig. S37 DLS spectra of Type III-C rotaxane-branched dendrimer G1: (a) 2.21 nm; (b) 2.02 nm; (c) 2.09 nm. The average hydrodynamic size was 2.11 ± 0.10 nm.



Fig. S38 DLS spectra of Type III-C rotaxane-branched dendrimer G2: (a) 3.53 nm; (b) 3.75 nm; (c) 3.74 nm. The average hydrodynamic size was 3.67 ± 0.14 nm.



Fig. S39 DLS spectra of Type III-C rotaxane-branched dendrimer G3: (a) 5.34 nm; (b) 5.29 nm; (c) 5.44 nm. The average hydrodynamic size was 5.36 ± 0.08 nm.



Fig. S40 DLS spectra of Type III-C rotaxane-branched dendrimer G4: For (a) 7.15 nm; (b) 7.20 nm; (c) 7.02 nm. The average hydrodynamic size was 7.12 ± 0.10 nm.



Fig. S41 TEM images of Type III-C rotaxane-branched dendrimer G1. The size range is 1.70 ± 0.20 nm.



Fig. S42 TEM images of Type III-C rotaxane-branched dendrimer G2. The size range is 2.65 ± 0.22 nm.



Fig. S43 TEM images of Type III-C rotaxane-branched dendrimer G3. The size range is 3.60 ± 0.30 nm.



Fig. S44 TEM images of Type III-C rotaxane-branched dendrimer G4. The size range is 4.65 ± 0.30 nm.

Section D. Anion-induced dimension modulation of Type III-C rotaxane-branched dendrimers.

Scheme S7 Cartoon representation of anion-induced switching motion of Type III-C rotaxane-branched dendrimer G1.



Fig. S45 ¹H NMR spectra (THF- d_8 , 298 K, 500 MHz) of anion-induced switching behavior of Type III-C rotaxane-branched dendrimer G1. a) G1; b) G1 with the addition of TBAA (5 eq. for each urea unit); c) the mixture after adding NaPF₆ (10 eq. for each urea unit) into the solution in b).

Scheme S8 Cartoon representation of anion-induced switching motion of Type III-C rotaxane-branched dendrimer G2.



Fig. S46 ¹H NMR spectra (THF- d_8 , 298 K, 500 MHz) of anion-induced switching behavior of Type III-C rotaxane-branched dendrimer G2. a) G2; b) G2 with the addition of TBAA (5 eq. for each urea unit); c) the mixture after adding NaPF₆ (10 eq. for each urea unit) into the solution in b).



Scheme S9 Cartoon representation of anion-induced switching motion of Type III-C rotaxane-branched dendrimer G3.

Fig. S47 ¹H NMR spectra (THF- d_8 , 298 K, 500 MHz) of anion-induced switching behavior of Type III-C rotaxane-branched dendrimer G3. a) G3; b) G3 with the addition of TBAA (5 eq. for each urea unit); the mixture after adding NaPF₆ into the solution in b): for c) NaPF₆ (10 eq. for each urea unit); d) NaPF₆ (12 eq. for each urea unit).





Fig. S48 ¹H NMR spectra (THF- d_8 , 298 K, 500 MHz) of anion-induced switching behavior of Type III-C rotaxane-branched dendrimer G4. a) G4; b) G4 with the addition of TBAA (5 eq. for each urea unit); the mixture after adding NaPF₆ into the solution in b): for c) NaPF₆ (12 eq. for each urea unit); d) NaPF₆ (14 eq. for each urea unit).



Fig. S49 2-D DOSY spectrum (THF- d_8 , 298 K, 500 MHz) of Type III-C rotaxane-branched dendrimer G1 with the addition of TBAA (5 eq. for each urea unit). The diffusion coefficient *D* is $(8.91\pm0.20) \times 10^{-10} \text{ m}^2\text{s}^{-1}$.



Fig. S50 2-D DOSY spectrum (THF- d_8 , 298 K, 500 MHz) of the mixture after adding NaPF₆ into the solution of Type III-C rotaxane-branched dendrimer G1 and TBAA.The diffusion coefficient D is $(10.23 \pm 0.25) \times 10^{-10} \text{ m}^2\text{s}^{-1}$.



Fig. S51 2-D DOSY spectrum (THF- d_8 , 298 K, 500 MHz) of Type III-C rotaxane-branched dendrimer G2 with the addition of TBAA (5 eq. for each urea unit). The diffusion coefficient D is $(3.98 \pm 0.10) \times 10^{-10} \text{ m}^2\text{s}^{-1}$.



Fig. S52 2-D DOSY spectrum (THF- d_8 , 298 K, 500 MHz) of the mixture after adding NaPF₆ into the solution of Type III-C rotaxane-branched dendrimer G2 and TBAA.The diffusion coefficient D is $(4.90\pm0.20) \times 10^{-10} \text{ m}^2\text{s}^{-1}$.



Fig. S53 2-D DOSY spectrum (THF- d_8 , 298 K, 500 MHz) of Type III-C rotaxane-branched dendrimer G3 with the addition of TBAA (5 eq. for each urea unit). The diffusion coefficient D is $(2.04\pm0.15) \times 10^{-10} \text{ m}^2\text{s}^{-1}$.



Fig. S54 2-D DOSY spectrum (THF- d_8 , 298 K, 500 MHz) of the mixture after adding NaPF₆ into the solution of Type III-C rotaxane-branched dendrimer G3 and TBAA. The diffusion coefficient D is $(2.51\pm0.15) \times 10^{-10} \text{ m}^2\text{s}^{-1}$.



Fig. S55 2-D DOSY spectrum (THF- d_8 , 298 K, 500 MHz) of Type III-C rotaxane-branched dendrimer G4 with the addition of TBAA (5 eq. for each urea unit). The diffusion coefficient *D* is $(1.20\pm0.08) \times 10^{-10} \text{ m}^2\text{s}^{-1}$.



Fig. S56 2-D DOSY spectrum (THF- d_8 , 298 K, 500 MHz) of the mixture after adding NaPF₆ into the solution of Type III-C rotaxane-branched dendrimer G4 and TBAA. The diffusion coefficient D is $(1.58\pm0.10) \times 10^{-10} \text{ m}^2\text{s}^{-1}$.



Fig. S57 DLS spectra of Type III-C rotaxane-branched dendrimer G1 with the addition of TBAA (5 eq. for each urea unit): (a) 2.51 nm; (b) 2.55 nm; (c) 2.75 nm. The average hydrodynamic size was 2.60 ± 0.15 nm.



Fig. S58 DLS spectra of the mixture after adding NaPF₆ into the solution of Type III-C rotaxane-branched dendrimer G1 and TBAA: (a) 2.16 nm; (b) 2.34 nm; (c) 2.05 nm. The average hydrodynamic size was 2.18 ± 0.16 nm.



Fig. S59 DLS spectra of Type III-C rotaxane-branched dendrimer G2 with the addition of TBAA (5 eq. for each urea unit): (a) 4.57 nm; (b) 4.88 nm; (c) 4.51 nm. The average hydrodynamic size was 4.65 ± 0.23 nm.



Fig. S60 DLS spectra of the mixture after adding NaPF₆ into the solution of Type III-C rotaxane-branched dendrimer G2 and TBAA: (a) 3.79 nm; (b) 4.07 nm; (c) 4.02 nm. The average hydrodynamic size was 3.97 ± 0.18 nm.



Fig. S61 DLS spectra of Type III-C rotaxane-branched dendrimer G3 with the addition of TBAA (5 eq. for each urea unit): (a) 7.14 nm; (b) 7.02 nm; (c) 6.96 nm. The average hydrodynamic size was 7.04 ± 0.10 nm.



Fig. S62 DLS spectra of the mixture after adding NaPF₆ into the solution of Type III-C rotaxane-branched dendrimer G3 and TBAA: (a) 5.43nm; (b) 5.62 nm; (c) 5.74 nm. The average hydrodynamic size was 5.60 ± 0.17 nm.



Fig. S63 DLS spectra of Type III-C rotaxane-branched dendrimer G4 with the addition of TBAA (5 eq. for each urea unit): (a) 9.69 nm; (b) 9.70 nm; (c) 9.79 nm. The average hydrodynamic size was 9.73 ± 0.05 nm.



Fig. S64 DLS spectra of the mixture after adding NaPF₆ into the solution of Type III-C rotaxane-branched dendrimer G4 and TBAA: (a) 7.15 nm; (b) 7.15 nm; (c) 7.20 nm. The average hydrodynamic size was 7.17 ± 0.03 nm.



Fig. S65 AFM image of Type III-C rotaxane-branched dendrimer G1 with the addition of TBAA (5 eq. for each urea unit) (*left*) and the height range is 2.11 ± 0.23 nm (*right*).



Fig. S66 AFM image of Type III-C rotaxane-branched dendrimer G2 with the addition of TBAA (5 eq. for each urea unit) (*left*) and the height range is 3.05 ± 0.19 nm (*right*).



Fig. S67 AFM image of Type III-C rotaxane-branched dendrimer G3 with the addition of TBAA (5 eq. for each urea unit) (*left*) and the height range is 4.29 ± 0.16 nm (*right*).



Fig. S68 AFM image of Type III-C rotaxane-branched dendrimer G4 with the addition of TBAA (5 eq. for each urea unit) (*left*) and the height range is 6.47 ± 0.28 nm (*right*).



Fig. S69 Optimized structures of the Type III-C rotaxane-branched dendrimer **G1** before *(left)* and after *(right)* the addition of TBAA as stimulus. According to the simulations, the swelling ratio of **G1** is 17% after the addition of TBAA.

Probe radius / Å	40	50	60	70	80
$S_{G2}/Å^2$	14180.5	14501.8	14789.8	15044.2	15257.8
$S_{G2-S}/Å^2$	22740.9	23481.8	24100.3	24563.4	25002.2
$(S_{G2-S}/S_{G2})^{-1/2}$	1.27	1.27	1.28	1.28	1.28

Table S1. Surface Areas of G2 and G2-S Calculated by Different Probe Radii.

Table S2. Volumes of G2 and G2-S Calculated by Different Probe Radii.

Probe radius / Å	40	50	60	70	80
$V_{G2}/Å^3$	72296.9	77602.0	82017.9	85604.3	88588.1
$V_{\text{G2-S}}/\text{\AA}^3$	101103.5	115600.0	128814.9	138435.8	147402.1
$(V_{G2-S}/V_{G2})^{-1/2}$	1.18	1.22	1.25	1.27	1.29

Section E. Synthesis of the model rotaxane-branched dendrimer G1-a.

Scheme S11: The synthesis route of (a) model [2]rotaxane 3-a without urea moiety in the axle component and (b) model first-generation rotaxane-branched dendrimer G1-a from corresponding building block [2]rotaxane 3-a.



Synthesis of [2]rotaxane 3-a: A Schlenk flask was charged with S_7 (100 mg, 0.23 mmol), macrocycle component 1 (1.57 g, 1.14 mmol) and Pt(PEt₃)₂I₂ (630 mg, 0.92 mmol). The Schlenk flask was then evacuated and back-filled with N₂ three times. Next, the mixture solvent of degassed CHCl₃ and i-Pr₂NH (v/v, 6/3 mL) was added via syringe. The resultant solution was stirred for one hour under -10 °C. Then CuI (4 mg) was added to the mixture under an inert atmosphere and the mixture was allowed to warm to room temperature and stirred overnight. The solution was concentrated and the residue was purified by column chromatography (SiO₂; PE/DCM). A pale yellow solid **3-a** (360 mg, 66%) was obtained. ¹H NMR (500 MHz, CD₂Cl₂): δ 7.38-7.40 (m, 4H), 7.21-7.22 (m, 2H), 6.75-6.90 (m, 16H), 6.06 (s, 3H), 3.70-4.01 (m, 48H), 3.38-3.42 (m, 2H),

3.18-3.23 (m, 2H), 2.20-2.24 (m, 12H), 1.90-2.01 (m, 8H), 1.13-1.20 (m, 60H), 0.77-0.80 (m, 2H), 0.66 (m, 4H), 0.44-0.48 (m, 4H), 0.33 (m, 2H), 0.06 (m, 2H), -0.07 (m, 2H), -0.25 (m, 2H), -0.50 (m, 2H). ³¹P NMR (202 MHz, CD₂Cl₂): δ 8.86. ¹³C NMR (126 MHz, CD₂Cl₂): δ 171.29, 162.04, 161.65, 159.66, 157.93, 150.80, 150.78, 150.76, 150.75, 150.72, 150.70, 150.23, 150.16, 133.79, 133.78, 132.01, 128.80, 128.71, 128.53, 128.51, 128.46, 128.39, 120.89, 115.93, 115.89, 114.74, 114.72, 114.53, 114.43, 113.81, 113.77, 113.60, 113.52, 113.46, 107.60, 107.57, 99.91, 93.87, 92.27, 88.98, 88.95, 87.13, 87.02, 68.68, 68.67, 68.30, 68.21, 68.19, 64.58, 31.11, 30.68, 30.64, 29.73, 29.62, 29.42, 29.36, 29.23, 27.08, 27.06, 26.74, 25.07, 24.67, 21.15, 19.56, 18.89, 17.19, 17.05, 16.91, 14.97, 13.93, 11.80, 8.52, 8.26. MALDI-TOF-MS: Calculated for [**3-a** + H]⁺: *m/z* = 2375.07; Found: *m/z* = 2375.1.

Synthesis of the first-generation model rotaxane-branched dendrimer G1-a: A mixture of model compound [2]rotaxane **3-a** (110 mg, 0.046 mmol) and 1, 3, 5-triethynylbenzene (2.1 mg, 0.014 mmol) were added in a Schlenk flask, the Schlenk flask was then evacuated and back-filled with N2 three times. Next, degassed diethylamine (8.0 mL) and a catalytic amount of CuI were added under an inert atmosphere. The reaction was stirred overnight at room temperature. The solvent was evaporated and the residue was purified by column chromatography (SiO₂; PE/DCM) and preparative gel permeation chromatography (GPC) to yield a pale-yellow solid G1-a (40 mg, 42%). ¹H NMR (500 MHz, CD₂Cl₂): δ 7.38-7.41 (m, 12H), 7.19-7.21 (m, 6H), 6.80-6.94 (m, 45H), 6.74-6.75 (m, 6H), 6.06 (s, 9H), 3.70-4.02 (m, 144H), 3.38-3.43 (m, 6H), 3.19-3.24 (m, 6H), 2.15-2.21 (m, 36H), 1.91-2.02 (m, 24H), 1.13-1.27 (m, 180H), 0.78-0.81 (m, 6H), 0.65-0.68 (m, 12H), 0.44-0.47 (m, 12H), 0.31 (m, 6H), 0.05-0.06 (m, 6H), -0.06 (m, 6H), -0.26 (m, 6H), -0.50 (m, 6H). ³¹P NMR (202 MHz, CD₂Cl₂): δ 11.66. ¹³C NMR (126 MHz, CD₂Cl₂): δ 171.32, 171.22, 162.03, 161.64, 159.67, 159.65, 150.78, 150.75, 150.71, 150.19, 150.16, 133.78, 133.77, 132.18, 128.77, 128.70, 128.51, 128.48, 128.44, 128.38, 115.89, 115.88, 114.75, 114.70, 114.53, 114.41, 113.80, 113.74, 113.57, 113.51, 107.57, 93.85, 92.27, 88.95, 80.43, 68.67, 68.26, 68.20, 68.16, 64.58, 60.64, 55.91, 55.85, 55.79, 55.68, 55.65, 55.62, 42.31, 31.10, 31.09, 30.66, 30.63, 29.73, 29.61, 29.41, 29.34, 29.24, 27.46, 27.04, 26.71, 25.41, 25.09, 24.65, 21.18, 21.14, 19.53, 18.87, 18.85, 16.96, 16.82, 16.68, 14.41, 13.89, 12.01, 11.78, 11.56, 8.59. MALDI-TOF-MS: Calculated for [G1-a + H]⁺: m/z = 6895.1; Found: m/z = 6898.7.



Fig. S70 ¹H NMR spectra (THF- d_8 , 298 K, 500 MHz) of the model compound [2]rotaxane 3-a *(bottom)*; model complex 3-a with the addition of TBAA (5 eq.) *(top)*.



Fig. S71 ¹H NMR spectra (THF- d_8 , 298 K, 500 MHz) of the model dendrimer G1-a *(bottom)*; model complex G1-a with the addition of TBAA (15 eq.) *(top)*.



Fig. S72 2-D DOSY spectrum (THF- d_8 , 298 K, 500 MHz) of the model rotaxane-branched dendrimer G1-a. The diffusion coefficient D is $(1.55\pm0.14) \times 10^{-9} \text{ m}^2 \text{ s}^{-1}$.



Fig. S73 2-D DOSY spectrum (THF- d_8 , 298 K, 500 MHz) of the model rotaxane-branched dendrimer G1-a with the addition of TBAA (5 eq. for each urea unit). The diffusion coefficient *D* is $(1.48\pm0.12) \times 10^{-9} \text{ m}^2 \text{ s}^{-1}$.



Fig. S74 DLS spectra of model rotaxane-branched dendrimer G1-a: (a) 1.63 nm; (b) 1.55 nm; (c) 1.43 nm. The average hydrodynamic size was 1.54 ± 0.11 nm.



Fig. S75 DLS spectra of model rotaxane-branched dendrimer G1-a with the addition of TBAA (15 eq.): (a) 1.58 nm; (b) 1.35 nm; (c) 1.64 nm. The average hydrodynamic size was 1.52 ± 0.17 nm.



Fig. S76 ¹H NMR spectrum (CDCl₃, 298 K, 400 MHz) of macrocycle 1.



Fig. S77 ¹³C NMR spectrum (CDCl₃, 298 K, 101 MHz) of macrocycle 1.



Fig. S78 ESI-MS spectrum of macrocycle 1.



Fig. S79 ¹H NMR spectrum (CDCl₃, 298 K, 300 MHz) of S₅.



Fig. S80 ¹³C NMR spectrum (CDCl₃, 298 K, 101 MHz) of S₅.

Elemental Composition Report							Page 1	
Single Ma Tolerance = Element pr	ss Analysis = 5.0 mDa / D ediction: Off	BE: min = -1	.5, max = 50	0.0				
Monoisotopi 19 formula(e Elements Us C: 0-28 H LWJ-0506-2 20170511007	c Mass, Odd and) evaluated with 1 sed: 1: 0-35 N: 0-2	Even Electror results withir O: 0-5	lons limits (up to	50 best isotop	bic matches for	each mass)		TOF MS EI
100				439	.2353			1.41e+00
-	+	420.000	439.100	439.200	439.300	439.400	439.500	439.600 m
0 438.800	438.900	439.000						
0 438.800 Minimum: Maximum:	438.900	5.0	10.0	-1.5 50.0				
0 438.800 Minimum: Maximum: Mass	438.900 Calc. Mass	439.000 5.0 mDa	10.0 PPM	-1.5 50.0 DBE	i-FIT	Formula		

Fig. S81 HRMS (EI-TOF) spectrum of S_5 .



Fig. S82 ¹H NMR spectrum (acetone-*d*₆, 298 K, 400 MHz) of thread component 2.



Fig. S83 13 C NMR spectrum (acetone- d_6 , 298 K, 126 MHz) of thread component 2.

Elemental Composition Report							
Single Mass Analysis Tolerance = 5.0 mDa / DBE: min = -1.5, max = 50.0 Element prediction: Off							
Monoisotopic Mass, Odd and Even Electron Ions 26 formula(e) evaluated with 1 results within limits (up to 50 best isotopic matches for each mass) Elements Used: C: 0-30 H: 0-40 N: 0-2 O: 0-5 WXQ-C-AXLE 20171121001 313 (10.418) Cm (289:313) Tol							
452.2679	1.90e+002						
%							
0 451.900 452.000 452.100 452.200 452.300 452.400 452.500 452.600	452.700 m/z						
Minimum: -1.5 Maximum: 5.0 10.0 50.0							
Mass Calc. Mass mDa PPM DBE i-FIT Formula							
452.2679 452.2675 0.4 0.9 11.0 5546112.5 C27 H36 N2 O4							

Fig. S84 HRMS (EI-TOF) spectrum of thread component 2.



Fig. S85 ¹H NMR spectrum (CD₂Cl₂, 298 K, 500 MHz) of model [2]rotaxane 3-a.



Fig. S86 ³¹P NMR spectrum (CD₂Cl₂, 298 K, 202 MHz) of model [2]rotaxane 3-a.



Fig. S87 ¹³C NMR spectrum (CD₂Cl₂, 298 K, 126 MHz) of model [2]rotaxane 3-a.



Fig. S88 MALDI-TOF-MS spectrum of model [2]rotaxane 3-a.



Fig. S89 ¹H NMR spectrum (CD₂Cl₂, 298 K, 500 MHz) of model rotaxane-branched dendrimer G1-a.



Fig. S90 ³¹P NMR spectrum (CD₂Cl₂, 298 K, 202 MHz) of model rotaxane-branched dendrimer G1-a.



Fig. S91 ¹³C NMR spectrum (CD₂Cl₂, 298 K, 126 MHz) of model rotaxane-branched dendrimer G1-a.



Fig. S92 MALDI-TOF-MS spectrum of model rotaxane-branched dendrimer G1-a.

Section G. References

- 1. R. Reuter, H. A. Wegner, Chem. Commun., 2013, 49, 146-148.
- Q. Lin, Y.-Q. Fan, P.-P. Mao, L. Liu, J. Liu, Y.-M. Zhang, H. Yao, T.-B. Wei, *Chem. Eur. J.*, 2018, 24, 777 –783.
- 3. X. F. Kong, Z. Q. He, Y. N. Zhang, L. P. Mu, C. J. Liang, B. Chen, X. P. Jing, A. N. Cammidge, *Org. Lett.*, **2011**, *13*, 764–767.