

## *Supporting Information*

# Poly(ionic liquid)-Modified Magnetic Janus Particles for Dye Degradation

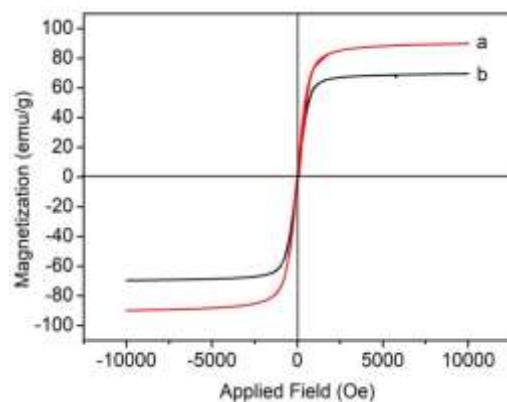
Ruotong Zhao<sup>a,b</sup>, Tianhao Han<sup>b</sup>, Dayin Sun<sup>b</sup>, Liyan Huang<sup>a</sup>, Fuxin Liang<sup>b,c\*</sup>, Zhengping Liu<sup>a\*</sup>

<sup>a</sup> BNU Key Lab of Environmentally Friendly and Functional Polymer Materials, Beijing Key Laboratory of Energy Conversion and Storage Materials, College of Chemistry, Beijing Normal University, Beijing 100875, China.

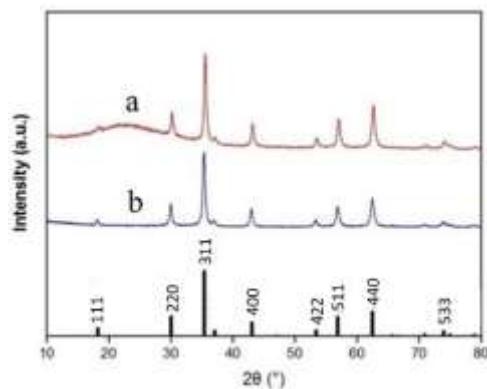
<sup>b</sup> State Key Laboratory of Polymer Physics and Chemistry, Institute of Chemistry, Chinese Academy of Sciences, Beijing 100190, China.

<sup>c</sup> Institute of Polymer Science and Engineering, Department of Chemical Engineering, Tsinghua University, Beijing 100084, China.

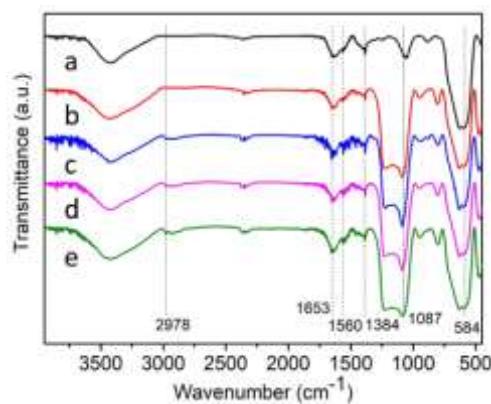
Corresponding author: lzp@bnu.edu.cn; liangfuxin@tsinghua.edu.cn



**Figure S1.** The magnetic hysteresis loops of representative magnetic particles: (a)  $\text{Fe}_3\text{O}_4$  particles; (b)  $\text{Fe}_3\text{O}_4@\text{SiO}_2$  particles.

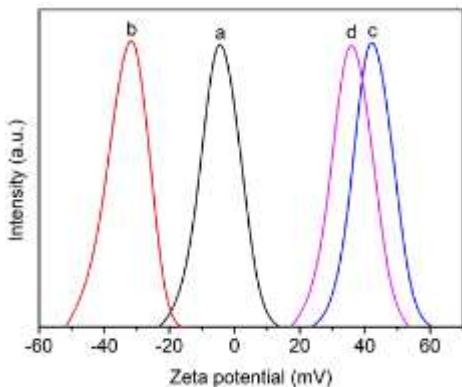


**Figure S2.** X-ray powder diffraction spectra of representative magnetic particles: (a)  $\text{Fe}_3\text{O}_4$  particles; (b)  $\text{Fe}_3\text{O}_4@\text{SiO}_2$  particles.

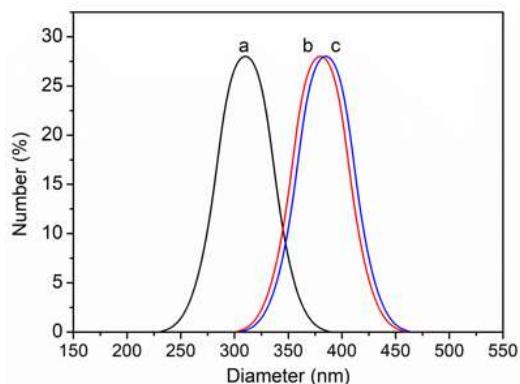


**Figure S3.** FT-IR spectra of representative magnetic particles: (a)  $\text{Fe}_3\text{O}_4$  particles; (b)  $\text{Fe}_3\text{O}_4@\text{SiO}_2$  particles; (c) amino modified  $\text{Fe}_3\text{O}_4@\text{SiO}_2$  particles; (d) phenyl modified  $\text{Fe}_3\text{O}_4@\text{SiO}_2$  particles; (e) initiator based

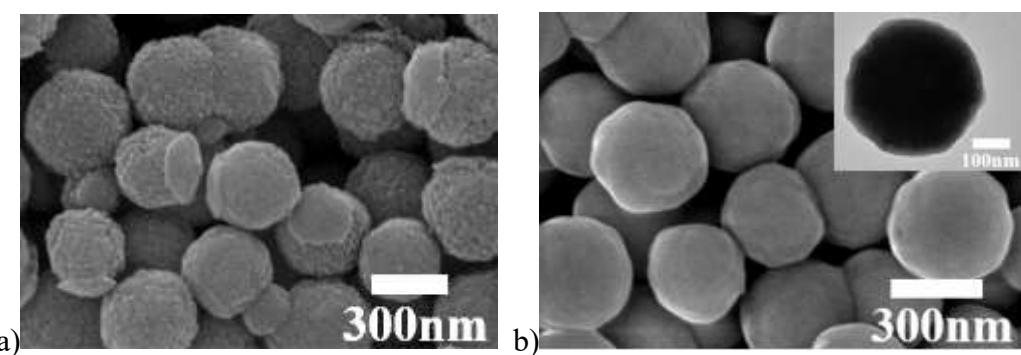
$\text{Fe}_3\text{O}_4@\text{SiO}_2$  Janus particles.



**Figure S4.** Zeta potential of representative magnetic particles ( $T = 300 \text{ K}$ ,  $\text{pH} = 7.0$ ): (a)  $\text{Fe}_3\text{O}_4$  particles; (b)  $\text{Fe}_3\text{O}_4@\text{SiO}_2$  particles; (c) amino modified  $\text{Fe}_3\text{O}_4@\text{SiO}_2$  particles; (d) phenyl modified  $\text{Fe}_3\text{O}_4@\text{SiO}_2$  particles.



**Figure S5.** Particle size analysis of representative magnetic particles: (a)  $\text{Fe}_3\text{O}_4$  particles; (b)  $\text{Fe}_3\text{O}_4@\text{SiO}_2$  particles; (c) phenyl modified  $\text{Fe}_3\text{O}_4@\text{SiO}_2$  particles.



**Figure S6.** Representative Janus particles: (a) SEM image of phenyl modified  $\text{Fe}_3\text{O}_4@\text{SiO}_2$  Janus particles after selective etching for 1 h. (b) SEM and inset TEM images of the  $\text{Fe}_3\text{O}_4@\text{SiO}_2$  Janus particles with

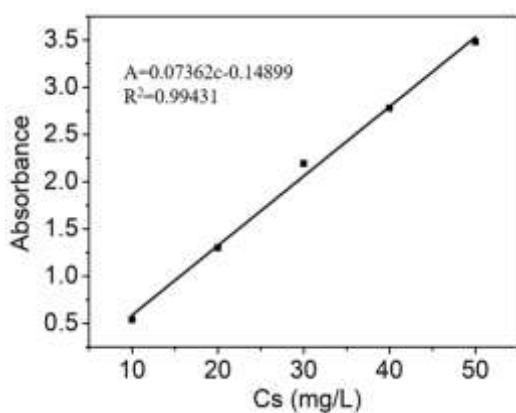
amino groups on one side and phenyl groups on the other side.

**Table S1.** Elemental analysis of the PILs modified Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub> Janus particles by XPS measurement.

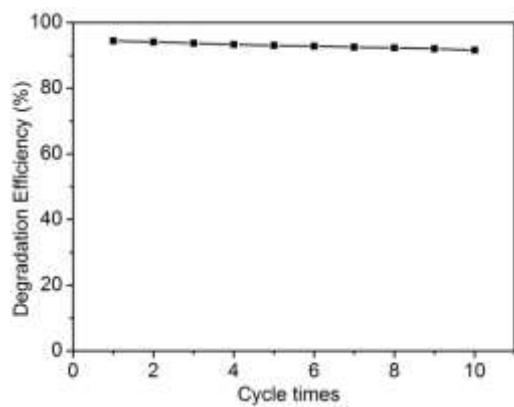
NO.	Atomic (%)						
	C	O	Si	N	Br	F	W
1	71.8	21.3	5.0	1.5	0.3	0	0
2	70.3	20.9	5.6	1.4	0	1.8	0
3	68.9	20.9	5.0	1.7	0	0	3.5

**Table S2.** The absorbance of different concentrations of methyl orange samples.

NO.	Cs (mg/L)	Absorbance
1	10	0.5407
2	20	1.3012
3	30	2.1931
4	40	2.7817
5	50	3.4815



**Figure S7.** The standard working curve of methyl orange.



**Figure S8.** Catalyst recycling of the  $\text{PW}_{12}\text{O}_{40}^{3-}$  based PILs modified  $\text{Fe}_3\text{O}_4@\text{SiO}_2$  Janus particles as solid stabilizers.