

Supporting Information

External Stress-Free Reversible Multiple Shape Memory Polymers

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Table S1. Typical multiple shape memory polymers

Polymers	SME			Switch mechanisms	Ref.
	One-way	Quasi two-way	True two-way		
Non-woven poly(ϵ -caprolactone) (PCL) fiber/epoxy-based copolymer/thermoset system	● ^{a)}	○ ^{b)}	○	Glass transition and melting	(S1)
Epoxy-based foam	●	○	○	Glass transition and melting	(S2)
PCL/poly(cyclohexyl methacrylate) (PCHMA) networks	●	○	○	Glass transition and melting	(S3)
PCL/PCHMA based nanocomposites	●	○	○	Glass transition and melting	S4)
Clay montmorillonite/PCL electrospun microfiber/epoxy composites	●	○	○	Glass transition and melting	(S5)
2-Dimethylamino-ethylmethacrylate (DMAEMA)/methyl-allyl-polyethenoxy-ether (TPEG) copolymer	●	○	○	Glass transition and melting	(S6)
Semicrystalline poly(decamethylene terephthalamide) (PA 10T)	●	○	○	Glass transition and melting	(S7)
Polyurethane (PU)/ethylene vinyl-acetate copolymer (EVA)/poly(vinyl acetate) (PVAc) multilayered films	●	○	○	Glass transition and melting	(S8)
Epoxy/PCL nanoweb	●	○	○	Glass transition and melting	(S9)
Commercial ethylene/1-butenecopolymer (TEPE)/carbon composites	●	○	○	Glass transition and melting	(S10)
Cellulose sisal fibre (SF)/PCL polyurethane networks	●	○	○	Glass transition and melting	(S11)
Poly(ester urethane)	●	○	○	Glass transition and melting	(S12)
Star-shaped polyhedral oligomeric silsesquioxane/poly(ϵ -caprolactone) polyurethane	●	○	○	Glass transition and melting	(S13)
Poly(ester urethane) (PBA3500)	●	○	○	Glass transition and melting	(S14)
PCL fiber/epoxy composites	●	○	○	Glass transition and melting	(S15)
PCL/epoxy	●	○	○	Glass transition	(S16)

				and melting	
Nanocrystalline non-isocyanate polyhydroxyurethanes	●	○	○	Glass transition and melting	(S17)
Poly(ether ether ketone) ionomer/sodium oleate composites	●	○	○	Glass transition and melting((S18)
Crosslinked poly(D,L-lactide) (PDLLA)/ poly(tetramethylene oxide) glycol (PTMEG) network	●	○	○	Glass transition and melting	(S19)
Poly(ethylene terephthalate) (PET) film	●	○	○	Glass transition and melting	(S20)
Poly(L-lactic acid)-b-poly(ethylene-co-butylene)-b-poly(L-lactic acid)	●	○	○	Glass transition and melting	(S21)
PCL/N,N-bis (2-hydroxyethyl) cinnamamide PU	●	○	○	Glass transition and melting	(S22)
4-Hexadecyloxybenzoic acid (HOBA)/shape-memory polyurethanes (SMPUs)	●	○	○	Glass transition and melting	(S23)
4-Hexadecyloxybenzoic acid (HOBA)/N,N-bis(2-hydroxyethyl) iso-nicotinamine (BINA) based PU complex	●	○	○	Glass transition and melting	(S24)
4,4-Azodibenzoic acid (Azoa)/hexamethylenediisocyanate (HDI)/PCL polyurethane networks	●	○	○	Glass transition and melting	(S25)
Lignin-co-poly(ester-amine-amide)	●	○	○	Two glass transitions	(S26)
Thiol-Michael addition based polymer networks	●	○	○	Two glass transitions	(S27)
Copolymer elastomers/clay nanocomposite	●	○	○	Two glass transitions	(S28)
Poly(propylene carbonate)/graphene oxide nanocomposites	●	○	○	Two glass transitions	(S29)
PU/PLA/PTMEG blends	●	○	○	Two glass transitions	(S30)
All-aromatic liquid crystal multiblock copoly(ether imide)s	●	○	○	Two glass transitions	(S31)
Silanized polyurethane/silane-functionalized graphene oxide nanocomposites bilayer	●	○	○	Two glass transitions	(S32)
Organic-inorganic nanocomposite bilayers	●	○	○	Two glass transitions	(S33)
Multicomposite styrene-based shape polymer	●	○	○	Two glass transitions	(S34)
Ethyl cellulose (EC)-g-tetrahydrofurfuryl methacrylate)-g-lauryl methacrylate polymers	●	○	○	Two glass transition	(S35)

PLA/PVAc/graphene nanocomposite blends	●	○	○	Two glass transitions	(S36)
Poly(mannitol sebacate)/electrospun poly(lactic acid) nanofibers composites	●	○	○	Two glass transitions	(S37)
SiO ₂ /epoxy resin nanocomposites	●	○	○	Two glass transitions	(S38)
epoxy bilayer polymer	●	○	○	Two glass transitions	(S39)
Peroxide crosslinked EVA/PCL blends	○	●	○	Two melting transitions	(S40)
Oligo(ω -pentadecalactone)/oligo(ϵ -caprolactone)/oligotetrahydrofuran polymer network	●	○	○	Two melting transitions	(S41)
Star-shaped poly(ω -pentadecalactone) (PPD) and PCL copolymer network	●	○	○	Two melting transitions	(S42)
Star-shaped PPD/PCL copolymer	○	●	○	Two melting transitions	(S43)
PCL/PEG nanocomposites	●	○	○	Two melting transitions	(S44)
Silica-coated nanoparticles/PCL/PEG nanocomposites	●	○	○	Two melting transitions	(S45)
PCL/PEG copolymer network	●	○	○	Two melting transitions	(S46)
PCL-g-PEG network	●	○	○	Two melting transitions	(S47)
4-Octyldecyloxybenzoic acid (OOBA)/PEG-based SMPU	●	○	○	Two melting transitions	(S48)
4-Dodecyloxybenzoic acid/liquid-crystalline (LC) SMPU composites	●	○	○	Two melting transitions	(S49)
4-Hexadecyloxybenzoic acid (HOBA)-PEG based SMPU composites	●	○	○	Two melting transitions	(S50)
Supramolecular semicrystalline polyolefin elastomer blends	●	○	○	Two melting transitions	(S51)
Maleated-polystyrene-b-poly(ethylene-co-butylene)-b-polystyreneblock copolymer	●	○	○	Two melting transitions	(S52)
Crosslinked polyalkenamer based polymer blends	●	○	○	Two melting transitions	(S53)
Polyolefin elastomer/stearic acid composite	●	○	○	Two melting transitions	(S54)
Ionomer (Surlyn 9520)/polycyclooctene crosslinked polymer blends	●	○	○	Two melting transitions	(S55)
Crosslinked polyethylene/PCL blends	○	●	○	Two melting transitions	(S56)
Thermoplastic polyurethane/olefin block copolymer/polycaprolactone blends	●	○	○	Two melting transitions	(S57)

UPy-PTMEG/UPy-star shaped PCL network	●	○	○	Two melting transitions	(S58)
Poly(L-lactide)(PLA)/PCL/graphene nanoplatelets nanocomposite	●	○	○	Two melting transitions	(S59)
Silver nanowires/PCL blends	●	○	○	Two melting transitions	(S60)
Foamed eucommia ulmoides gum/high-density polyethylene (HDPE) composites	●	○	○	Two melting transitions	(S61)
Natural eucommia ulmoides rubber/polybutene-1 composites	●	○	○	Two melting transitions	(S62)
Crosslinked poly(ethylenevinyl acetate)/PCL blends	●	○	○	Two melting transitions	(S63)
Poly(tetramethylene oxide)/poly(p-dioxanone) co-network	●	○	○	Two melting transitions	(S64)
Poly(p-dioxanone)/PEG network	●	○	○	Two melting transitions	(S65)
UPy-functionalized PCL/poly(p-dioxanone) interpenetrating polymer networks (IPNs)	●	○	○	Two melting transitions	(S66)
Crosslinked polycyclooctene/carbon nanotube (CNT)/polyethylene nanocomposites	●	○	○	Two melting transitions	(S67)
Trans-1,4-polyisoprene (TPI)/low density polyethylene (LDPE) blends	●	○	○	Two melting transitions	(S68)
Semicrystalline ethylene-propylene-diene rubber/PCL blends	●	○	○	Two melting transitions	(S69)
PU/poly(methacrylic acid) (PMAA) network	●	○	○	Two melting transitions	(S70)
Crosslinked polyethylene (PE)/polypropylene(PP) blends	●	○	○	Two melting transitions	(S71)
Thermoplastic polyurethane (TPU)/poly(butylene succinate) (PBS)/PCL blends (SLBs) multilayers	●	○	○	Two melting transitions	(S72)
PCL/PTMEG polyurethane	●	○	○	Two melting transitions	(S73)
Cholesteryl isonicotinate/PCL based polyurethane	●	○	○	Two melting transitions	(S74)
4-n-Octyldecyloxybenzoic acid (OOBA)/pyridine-containing polyurethane complex	●	○	○	Two melting transitions	(S75)
Commercial UV curable glassy thermoset (Norland Optical Adhesive 63)	●	○	○	Broad glass transition	(S76)
Cellulose derivative	●	○	○	Broad glass transition	(S77)
α -Amino acid based poly(ester urea)s	●	○	○	Broad glass transition	(S78)
Crosslinked solution-polymerized styrene butadiene rubber	●	○	○	Broad glass transition	(S79)

Polyvinylpyrrolidone/poly(hydroxyethyl methacrylate-co-butyl acrylate) semi-IPNs	●	○	○	Broad glass transition	(S80)
PCL/poly(vinyl chloride) (PVC) blends	●	○	○	Broad glass transition	(S81)
A fluorine-containing difunctional benzoxazine	●	○	○	Broad glass transition	(S82)
Poly(methyl methacrylate) (PMMA)/PCL covalently crosslinked polymer co-network	●	○	○	Broad glass transition	(S83)
Poly(ester urea)s	●	○	○	Broad glass transition	(S84)
Epoxy based photo-curable resin with 3D printing method	●	○	○	Broad glass transition	(S85)
Poly(L-lactide) (PLLA)/PMMA blends	●	○	○	Broad glass transition	(S86)
UPy functionalized n-alkyl acrylate crosslinked network	●	○	○	Broad glass transition	(S87)
Nafion membrane	●	○	○	Broad glass transition	(S88)
HDPE/PEG thermoset polyurethane	●	○	○	Broad glass transition	(S89)
Perfluorosulphonic acid ionomer (PFSA)-Nafion	●	○	○	Broad glass transition	(S90)
Poly(benzoxazole-imide) (PIB)/polyetherimide (PIO) blends	●	○	○	Broad glass transition	(S91)
Carbon nanotube/water-borne epoxy nanocomposites	●	○	○	Broad glass transition	(S92)
ENR (a kind of epoxy)/FeCl ₃ elastomer	●	○	○	Broad glass transition	(S93)
Bisphenol-A cyanate ester-bismaleimide crosslinked networks	●	○	○	Broad glass transition	(S94)
PU/PMMA composites	●	○	○	Broad glass transition	(S95)
Nafion	●	○	○	Broad glass transition	(S96)
Cholic acid-cinnamic acid-PEG crosslinked network	●	○	○	Broad glass transition	(S97)
2-Methoxyethyl acrylate-methylol acrylamide copolymer	●	○	○	Broad glass transition	(S98)
1,3-Adamantanediol-based polyurethanes	●	○	○	Broad glass transition	(S99)
3-Dimethyl (methacryloyloxyethyl)ammonium propane sulfonate-co-acrylic acid polymer	●	○	○	Broad glass transition	(S100)
Pyridine type zwitterionic polyurethane	●	○	○	Broad glass transition	(S101)

BINA/HDI/1,3-propanesultonezwitterionic polyurethanes	●	○	○	Broad glass transition	(S102)
Crosslinked poly[ethylene-co-(vinyl acetate)]	●	○	○	Broad melting transition	(S103)
Ethyl cellulose-g-PCL network	●	○	○	Broad melting transition	(S104)
Ethylene-1-octene copolymer (EOC)/LDPE/HDPE blends	●	○	○	Broad melting transition	(S105)
Paraffin/styrene-b-(ethylene-co-butylene)-b-styrene blends	●	○	○	Broad melting transition	(S106)
Star-shaped PCL based polyurethane	●	○	○	Broad melting transition	(S107)
Poly(vinyl alcohol)-g-polyurethane	●	○	○	Broad melting transition	(S108)
Side-chain liquid crystalline type random terpolymer networks	●	○	○	Glass transition and liquid crystal transition	(S109)
Side-chain liquid crystalline random terpolymers	●	○	○	Glass transition and liquid crystal transition	(S110)
Linear poly(lactic acid) (PLA) based copolymers	●	○	○	Glass transition and liquid crystal transition	(S111)
Liquid crystalline polyurethane networks	●	○	○	Glass transition and liquid crystal transition	(S112)
Side-chain liquid crystalline polyurethane networks	●	○	○	Glass transition and liquid crystal transition	(S113)
Liquid crystalline based poly(4-vinyl pyridine)	●	○	○	Glass transition and liquid crystal transition	(S114)
Polypropylene glycol-epoxy/PCL blends	●	○	○	Melting and glass transition	(S115)
Poly(cyclohexyl methacrylate)/PCL copolymer networks	●	○	○	Melting and glass transition	(S116)
Oligo(ϵ -caprolactone)dimethacrylate/silica coated magnetite nanoparticles nanocomposites	●	○	○	Narrow melting transition	(S117)
Polydopamine-poly(ϵ -caprolactone) network	●	○	○	Glass transition and a broad melting transition	(S118)
Polyurethane-based trilayer laminates	●	○	○	Three melting transitions	(S119)

Olefin block copolymer/ styrene- <i>b</i> -(ethylene-co-butylene)- <i>b</i> -styrene/paraffin blends	●	○	○	Three melting transitions	(S120)
Neat epoxy/ <i>p</i> -aminodiphenylimide-epoxy /multiwalled carbon nanotube (MWCNT)-epoxy multicomposites	●	○	○	Melting and <i>trans/cis</i> photo-isomerization of azobenzene	(S121)
PMMA/PEG semi-interpenetrating networks (semi-IPNs)	●	○	○	Melting and broad glass transition	(S122)
Ethylene-1-octene copolymers (EOC)/HDPE/EOC blends	●	○	○	Multiple melting transitions	(S123)
PE/polycyclooctene blends	●	○	○	Multiple melting transitions	(S124)
PEG-4,4'-diphenylmethanediisocyanate (MDI)-dimethylol propionic acid (DMPA) polyurethane	●	○	○	Hydrogen bond and glass transition	(S125)
Epoxy network	●	○	○	Glass transition and Diels-Alder reaction	(S126)
Zn(Mebip) ₂ (NTf ₂) ₂ /epoxy composites	●	○	○	Glass transition and metal complexes	(S127)
Spherical Fe ₃ O ₄ nanoparticles/MWCNTs/epoxy nanocomposites	●	○	○	Multiple glass transitions	(S128)
Poly(L-lactide)/PTMEG copolymers	●	○	○	Melting and reversible photodimerization of anthracene groups	(S129)
4'-ethoxy-4-(11-hydroxyundecyloxy)-azobenzene (EHAB)/GO nanocomposite films	●	○	○	Melting and <i>trans/cis</i> photo-isomerization of azobenzene	(S130)
This work	○	○	●	Two melting transitions	

a) ●: Applicable. b) ○: Not applicable.

Figure S5. ^1H NMR spectra of UPy.

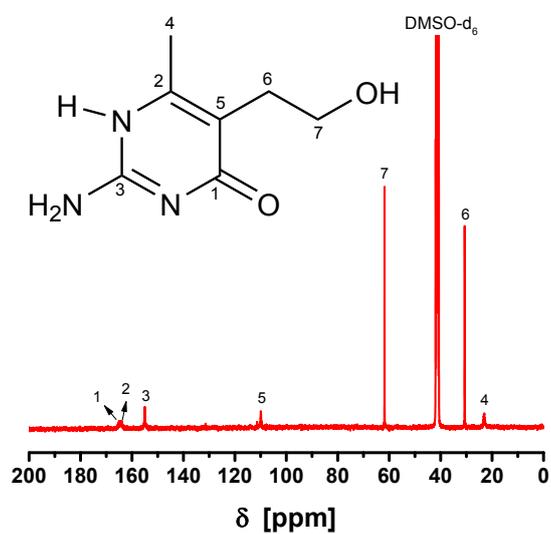


Figure S6. ^{13}C NMR spectrum of UPy.

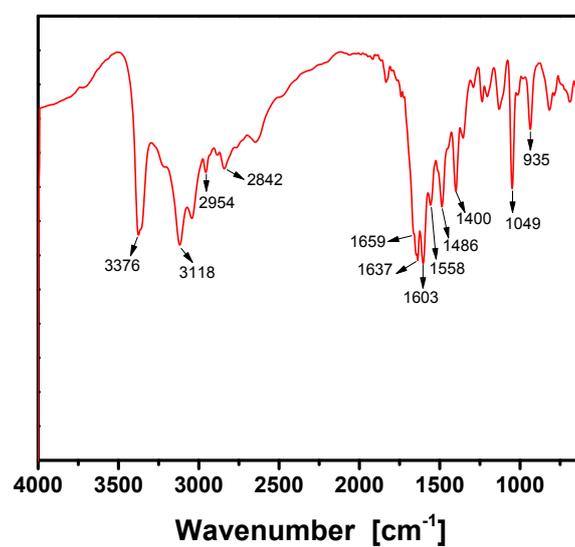


Figure S7. FTIR spectrum of UPy.

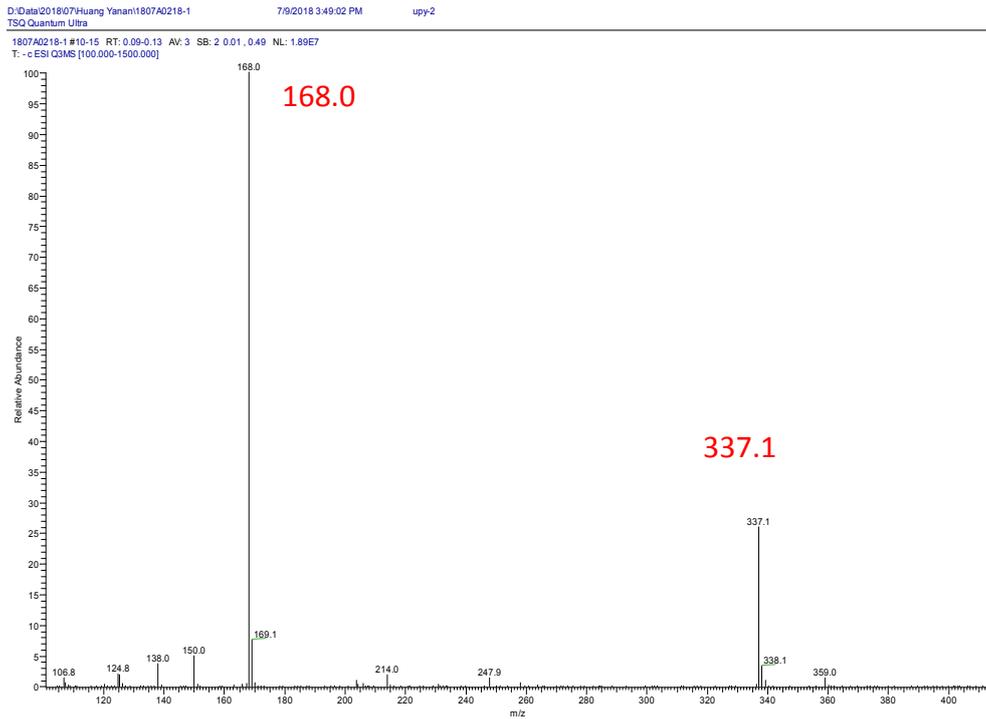


Figure S8. Mass spectrum of UPy.

Table S2. Effect of weight ratio of PCL/PTMEG on reversible strain of the programmed PU_{UPy}

Formulae	PCL (g)	PTMEG (g)	HDI (g)	TMPMP (g)	UPy (g)	Average total reversible strain (%)	Reversible strain due to PCL segment (%)	Reversible strain due to PTMEG segment (%)
1#	3.00	26.10				9.26	-	-
2#	9.00	20.30				12.12	5.46	6.66
3#	15.00	14.50	3.70	1.91	0.81	10.75	8.77	1.98
4#	21.00	8.70				9.22	-	-
5#	27.00	2.90				7.88	-	-

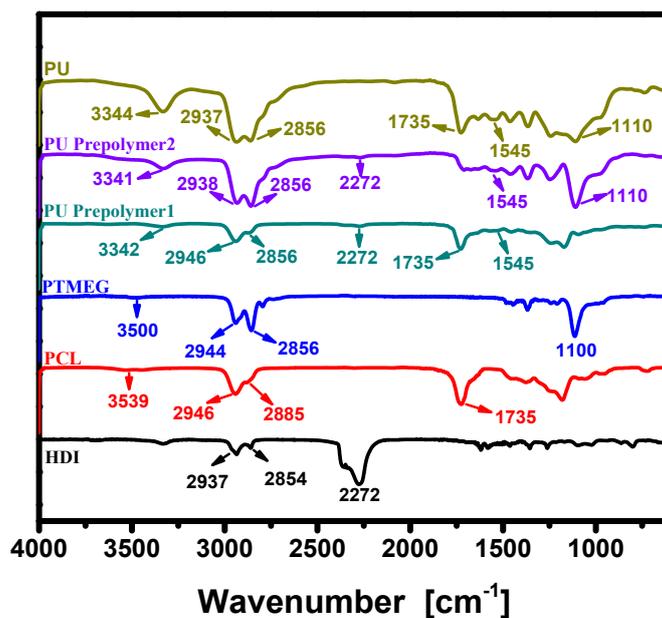


Figure S9. FTIR spectra of HDI, PCL, PTMEG, PU prepolymer 1, PU prepolymer 2, and PU_{UPy}. FTIR (KBr/cm⁻¹): HDI (2931, 2854 and 2272), PCL (3539, 2944, 2856 and 1735), PTMEG (3500, 2944, 2856 and 1100), PU prepolymer 1 (3342, 2946, 2856, 2272, 1735 and 1545), PU prepolymer 2 (3341, 2938, 2856, 2272, 1545 and 1110), and PU_{UPy} (3344, 2937, 2856, 1735, 1545 and 1110).

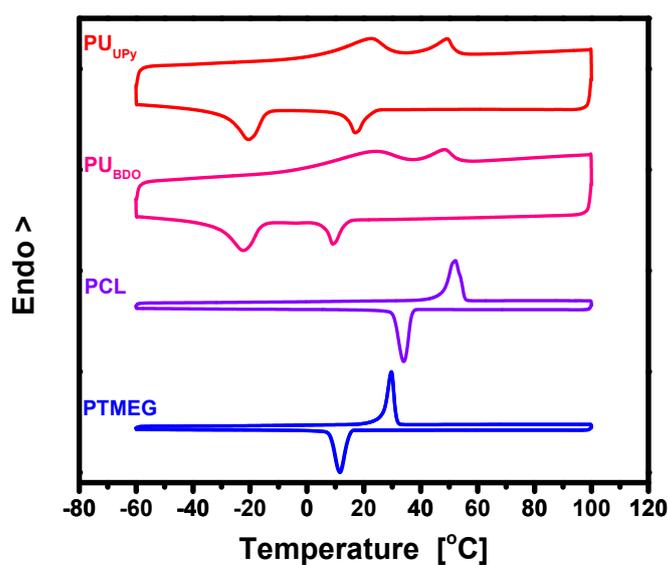


Figure S10. DSC heating and cooling scans of PU_{UPy}, PU_{BDO}, PCL and PTMEG (ramp: 3 °C/min).

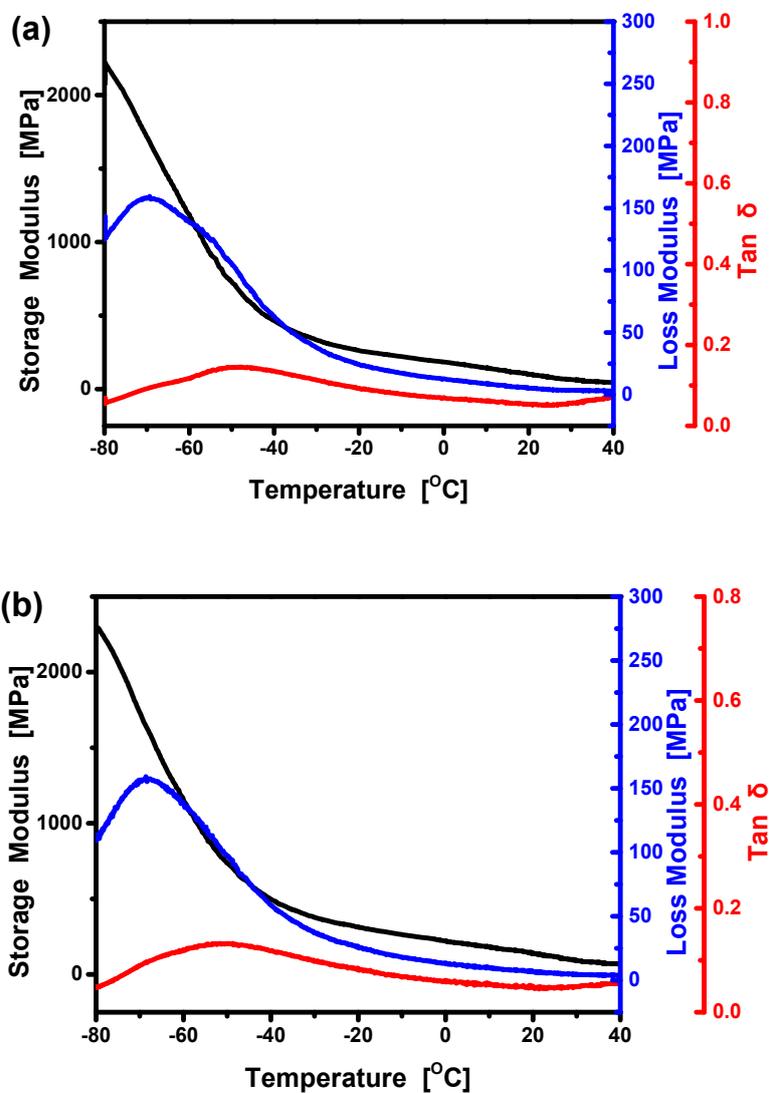


Figure S11. Typical DMA curves of (a) PU_{UPy} and (b) PU_{BDO} (ramp: 3 °C/min; frequency: 1 Hz). Note: The curves from 40 to 120 °C are not displayed because there is not any significant change within this range.

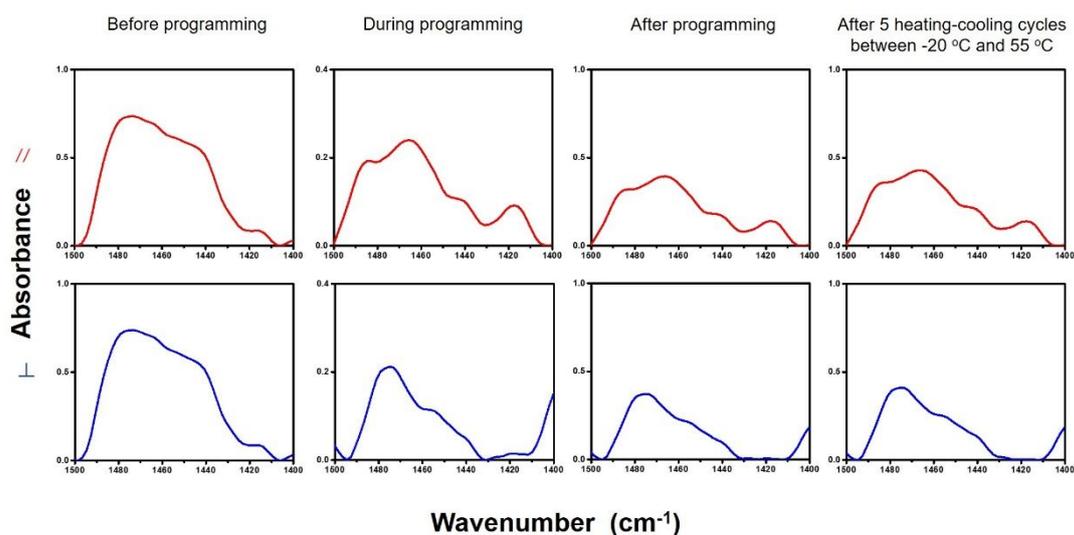


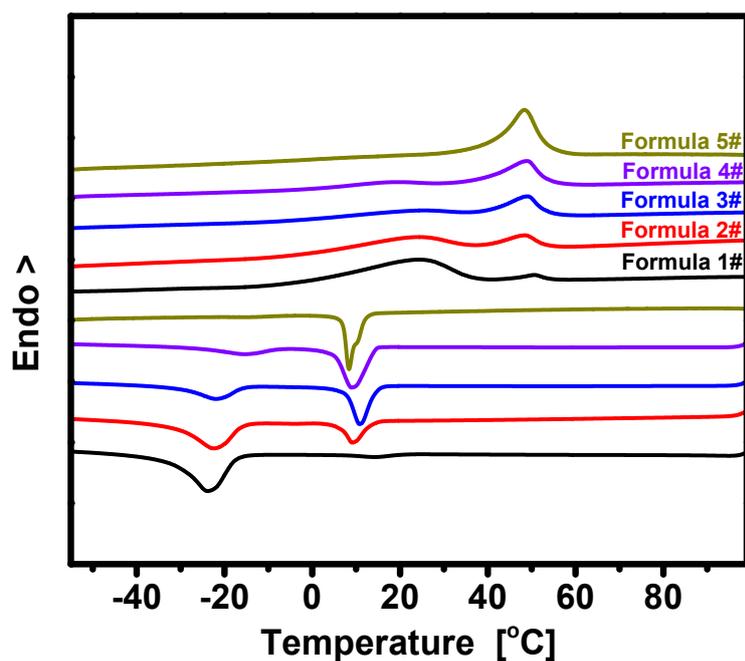
Figure S12. $\delta(-\text{CH}_2-)$ region on the FTIR spectra of the PU_{UPy} specimens.

Note: The specimen measured during programming means that it was deformed to a temporary shape with strain of 400 % (refer to step ④ in **Figure 1**).

Table S3. Characterization of orientation of the PU_{UPy} specimen based on the measurement of FTIR dichroism in **Figure S12**

PU_{UPy}	Before programming	During programming*	After programming	After 5 heating-cooling cycles between -20 °C and 55 °C
$A_{//}$	0.68	0.68	0.67	0.62
A_{\perp}	0.68	0.46	0.53	0.50
R	1.00	1.49	1.26	1.24
$(R-1)/(R+2)$	0	0.14	0.08	0.07

*The specimen was deformed to a temporary shape with strain of 400 % (refer to step ④ in **Figure 1**).



Figurer S13. DSC heating and cooling scans of the programmed PU_{UPy} (ramp: 3 °C/min).

Table S4. Characteristic parameters of the DSC curves of **Figure S13**

Formulae	PCL			PTMEG			Total X_c (%)
	T_m^a (°C)	T_c^b (°C)	X_c^c (%)	T_m (°C)	T_c (°C)	X_c (%)	
PCL	51.14	34.48	61.20	-	-	-	-
PTMEG	-	-	-	29.80	11.60	55.04	-
1#	50.63	14.09	0.99	23.68	-23.89	18.22	19.21
2#	48.42	9.11	3.63	24.13	-22.44	9.08	12.71
3#	49.03	10.70	7.91	25.43	-21.97	4.61	12.52
4#	48.91	8.74	13.48	19.7	-15.32	2.93	16.41
5#	48.45	8.37	24.49	-	-14.62	0.24	24.73

^{a)} T_m : peak endothermic temperature; ^{b)} T_c : peak exothermic temperature; ^{c)} X_c : crystallinity of PTMEG or PCL calculated from $\Delta H_f/\Delta H_f^p$ (where ΔH_f denotes the measured heat of fusion, and ΔH_f^p is the heat of fusion of 100 % crystalline PTMEG)

($172.2 \text{ J g}^{-1} \text{ S131}$) or PCL ($139.5 \text{ J g}^{-1} \text{ S132}$)).

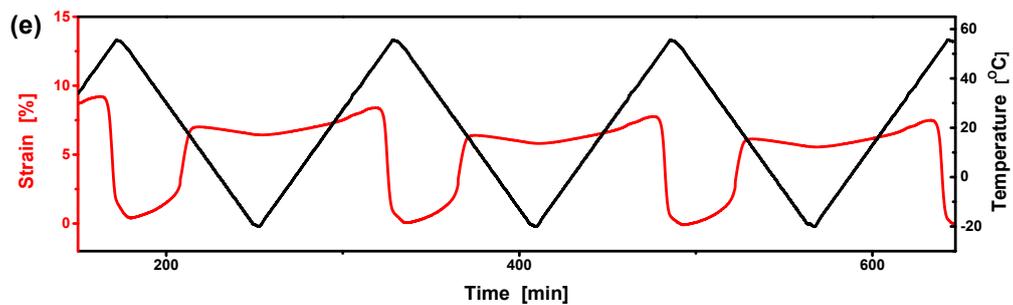
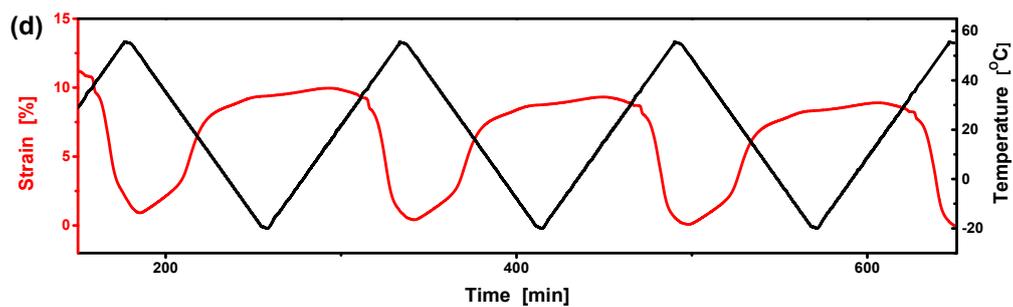
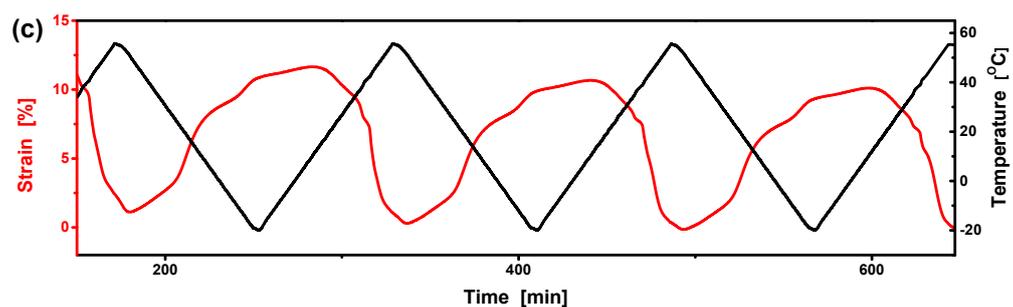
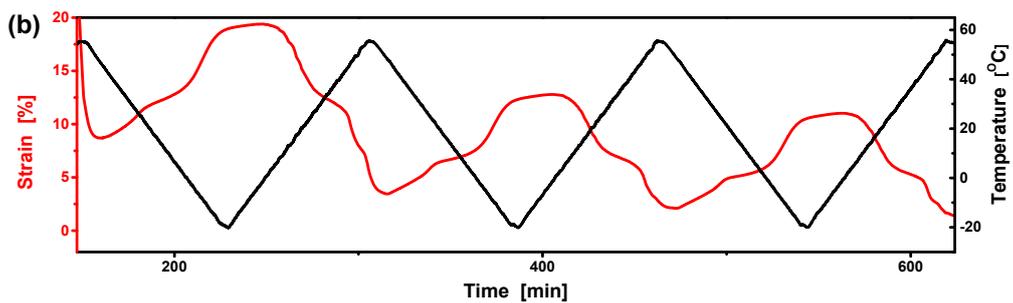
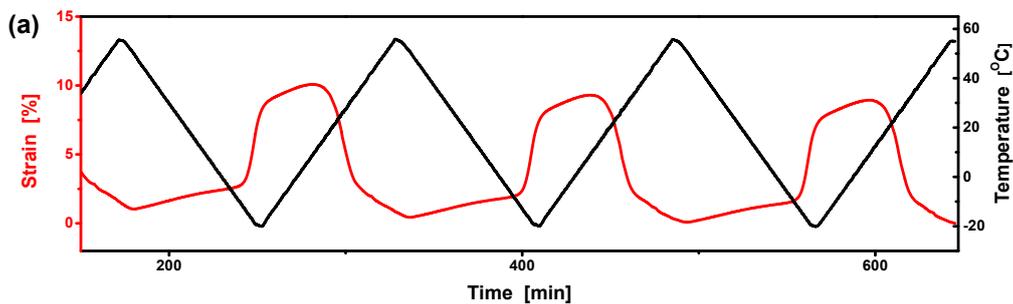


Figure S14. Reversible strains measured by DMA of the programmed PU_{UPy} during repeated heating and cooling cycles between -20 and 55 °C (ramp: 3 °C/min). (a) Formula #1, (b) formula #2, (c) formula #3, (d) formula #4, and (e) formula #5.

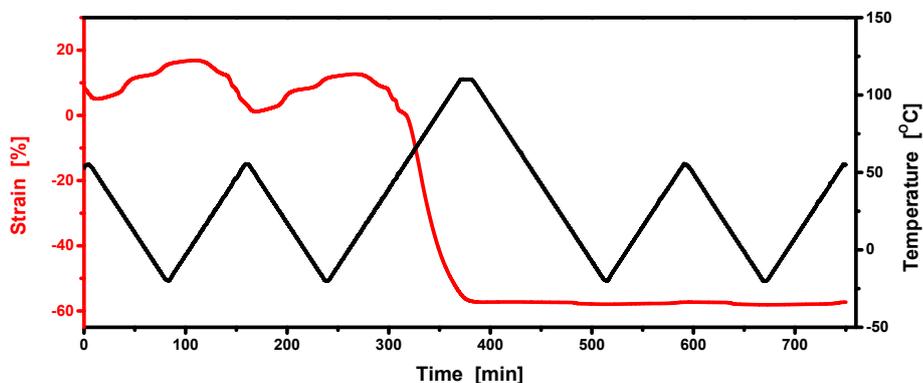


Figure S15. Thermomechanical behavior (measured by DMA) of the programmed PU_{UPy} before and after removing the internal stress provided by the hydrogen bonds.

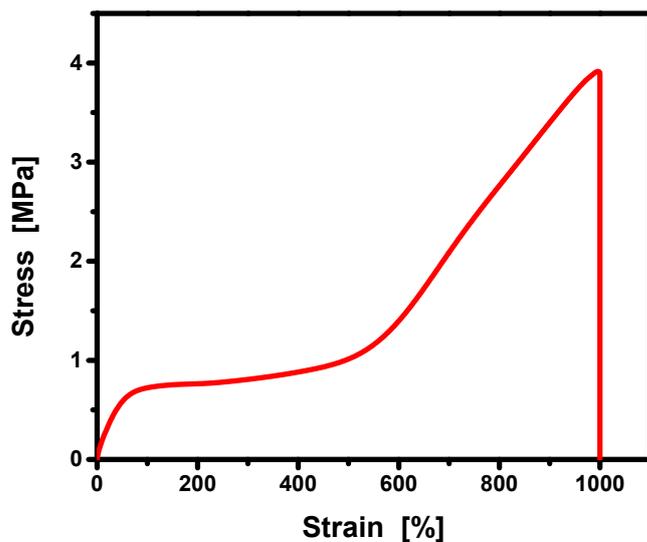


Figure S16. Typical tensile stress-strain curve of the PU_{UPy} measured at 49.1 °C.

°C. Prior to the test, the specimen was firstly immersed in the THF solution of LiBr (500 mL, 20 g/L) for 24 h at room temperature to remove the inter- and intra-macromolecular hydrogen bonds. Then, the specimen was dried at room temperature for 24 h allowing for evaporation of THF. Finally, the specimen was heated to 49.1 °C for 2 min and tested to failure under tension.

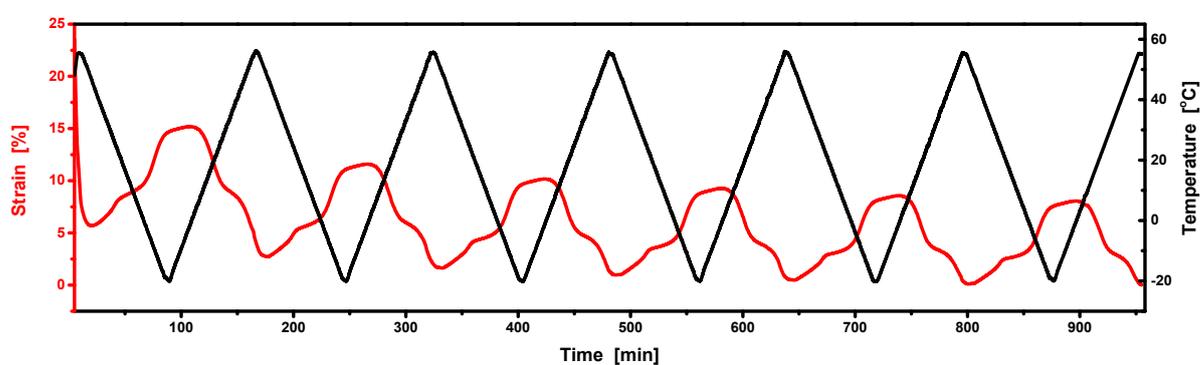


Figure 17. Thermomechanical behavior (measured by DMA) of the reprogrammed PU_{UPy} during repeated heating and cooling cycles between -20 and 55 °C (ramp: 1 °C/min). Note: The specimen used here was firstly programmed and then lost its two-way shape memory effect by heating up to 110 °C (**Figure S15**). Afterwards, it was reprogrammed to obtain the two-way shape memory effect once again. The total average reversible strain of the reprogrammed specimen is estimated to 10.5 % according to data collected during the first three heating/cooling cycles, and the average reversible strain induced by the melting of PCL and PTMEG are 4.8 % and 6.0 %, respectively. The results are close to those of the original programmed version (**Figure 5b**).

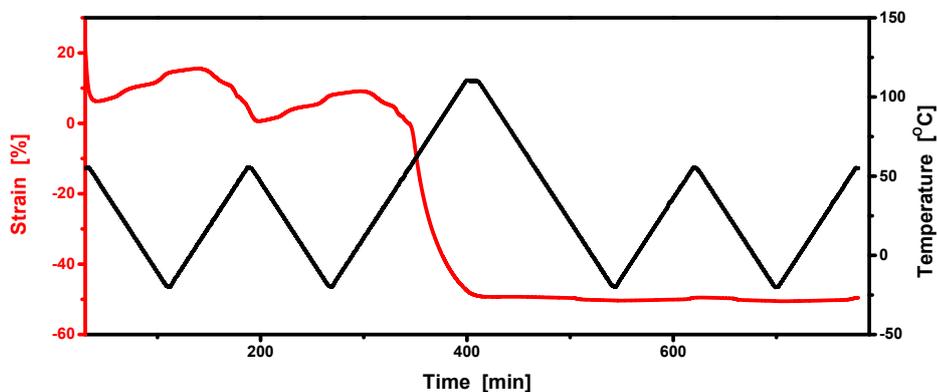


Figure S18. Thermomechanical behavior (measured by DMA) of the programmed PU_{BDO} before and after removing the internal stress provided by the hydrogen bonds.

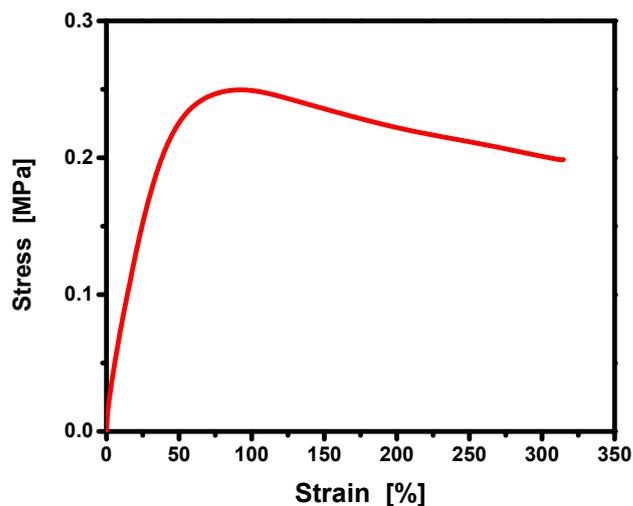


Figure S19. Typical tensile stress-strain curve of the PU_{BDO} measured at 47.7 °C. Prior to the test, the specimen was firstly immersed in the THF solution of

LiBr (500 mL, 20 g/L) for 24 h at room temperature to remove the inter- and intra-macromolecular hydrogen bonds. Then, the specimen was dried at room temperature for 24 h allowing for evaporation of THF. Finally, the specimen was heated to 47.7 °C for 2 min and tested to failure under tension.

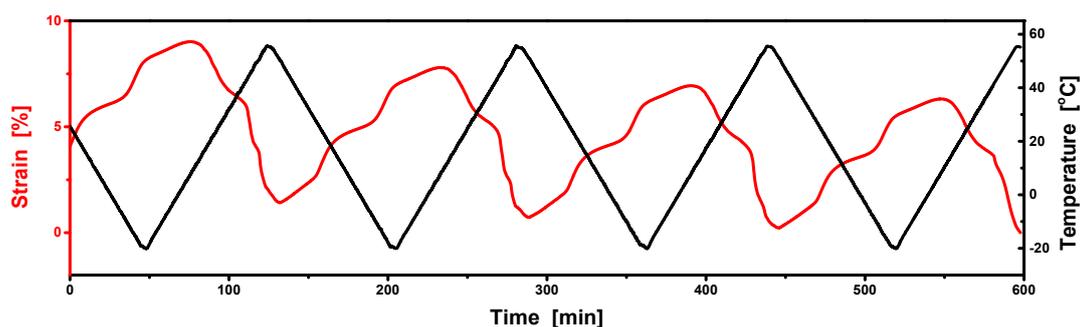


Figure S20. Thermomechanical behavior (measured by DMA) of the programmed PU_{BDO}. The first cycle that was used to remove the thermal history is not displayed for clarity.

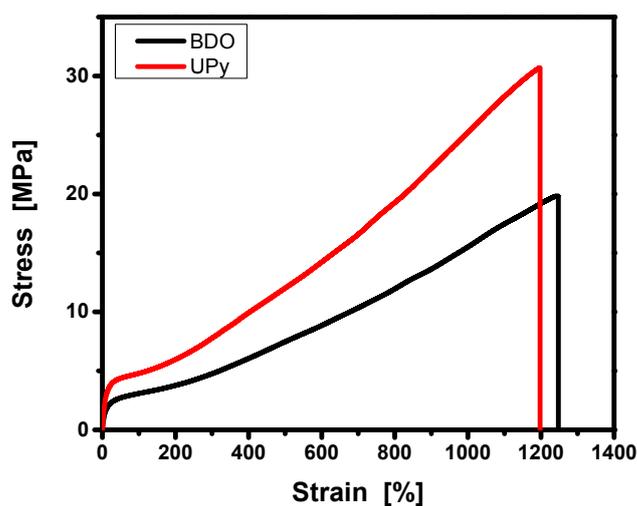


Figure S21. Typical tensile stress-strain curves of the programmed PU_{UPy} and PU_{BDO}.

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