### **Supporting Information for:**

## Systematic Investigations of Ligand Steric Effects on α-Diimine Nickel Catalyzed Olefin Polymerization and Copolymerization

Yanfeng Gong,<sup>a,b</sup> Shuaikang Li<sup>b</sup> Qi Gong<sup>b</sup> Shaojie Zhang<sup>b,\*</sup>, Binyuan Liu<sup>a,\*</sup>, Shengyu Dai<sup>b,\*</sup> <sup>a</sup> National-Local Joint Engineering Laboratory for Energy Conservation of Chemical Process Integration and Resources Utilization, School of Chemical Engineering and Technology, Hebei University of Technology, No 8 Guangrong Road, 300130 Tianjin, China.

<sup>b</sup>Institutes of Physical Science and Information Technology, School of Computer Science and Technology, Anhui University, Hefei, Anhui, 230601, China

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## 1. Figures, Tables.

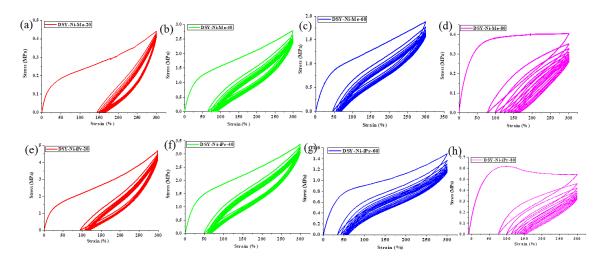


Figure S1. Plots of hysteresis experiments of ten cycles at a strain of 300% for samples generated by 2 at 20°C (a), 40 °C (b), 60 °C (c), 80 °C (d); 3 at 20°C (e), 40 °C (f), 60 °C (g), 80 °C (h).

Ent.	Precat.	T/°C	Strain at break (%)	Stress at break (MPa)	SR (%) <sup>b</sup>
1	2	20	325	4.9	45
2	2	40	529	6.0	63
3	2	60	826	5.5	77
4	2	80	1558	0.3	45
5	3	20	443	7.6	60
6	3	40	567	9.6	77
7	3	60	526	2.1	80
8	3	80	1078	0.5	49
9	4	20	329	20.3	_ <sup>c</sup>
10	4	40	439	13.6	-
11	4	60	408	8.3	46
12	4	80	613	13.6	-
13	5	20	208	7.8	-
14	5	40	361	12.3	-
15	5	60	484	20.0	22
16	5	80	628	22.1	-

Table S1. Mechanical properties.<sup>a</sup>

<sup>*a*</sup>Conditions: Performed at 10 mm/min by means of a Universal Test Machine (UTM2502) at room temperature. <sup>*b*</sup>The strain recovery values (SR) can be calculated by SR =  $100(\epsilon_a - \epsilon_r)/\epsilon_a$ , where  $\epsilon_a$  is the applied strain and  $\epsilon_r$  is the strain in the cycle at zero load after 10th cycle. <sup>*c*</sup>Not determined.

		<u> </u>	
Ent.	Precat.	T/°C	Yield/g
1	1	20	3.03
2	2	20	2.01
3	3	20	1.51
4	4	20	1.32
5	5	20	0.38

Table S2. Effect of Catalyst on Ethylene Polymerization in Short Time.<sup>a</sup>

<sup>*a*</sup>General conditions: Ni = 1.0  $\mu$ mol, Al/Ni = 600, CH<sub>2</sub>Cl<sub>2</sub> = 2 ml, toluene = 40 ml, ethylene = 8 atm, time = 10 min.

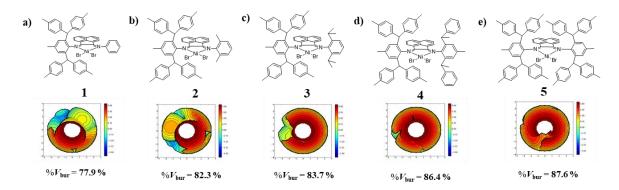


Figure S2. Steric maps for the complexes 1-5: a) 1; b) 2; c) 3; d) 4; e) 5.

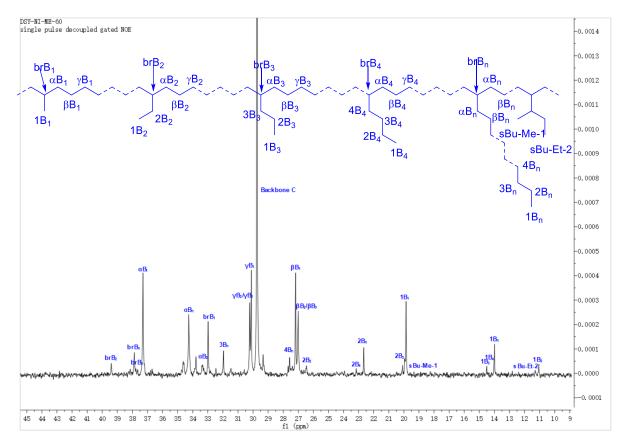


Figure S3. <sup>13</sup>C NMR spectrum of the polymer from table 1, entry 7 (CDCl<sub>2</sub>CDCl<sub>2</sub>, 110 °C).

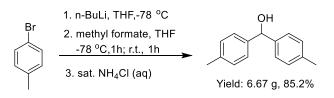
### 2. Experimental sections

### **2.1 General Considerations**

All experiments were carried out under a dry Nitrogen atmosphere using standard Schlenk techniques or in a glove-box. Deuterated solvents used for NMR were dried and distilled prior to use. <sup>1</sup>H, <sup>13</sup>C NMR spectra were recorded by a Bruker Ascend Tm 500 spectrometer at ambient temperature unless otherwise stated. The chemical shifts of the <sup>1</sup>H and <sup>13</sup>C NMR spectra were referenced to the residual solvent; Coupling constants are in Hz. Elemental analysis was performed by the Analytical Center of the University of Science and Technology of China. X-ray Diffraction data were collected at 298(2) K on a Bruker Smart CCD area detector with graphite-monochromated Mo K<sup>a</sup> radiation ( $\lambda = 0.71073$  Å). Molecular weight and molecular weight distribution of the polymers were determined by gel permeation chromatography (GPC) with a PL 210 equipped with one Shodex AT-803S and two Shodex AT-806MS columns at 150 °C using trichlorobenzene as a solvent and calibrated with polystyrene standards.

Stress/strain experiments were performed at 10 mm/min by means of a Universal Test Machine (UTM2502) at room temperature. At least three specimens of each polymer were tested. Polymers were melt-pressed at 30 to 35°C above their melting point to obtain the test specimens. The test specimens had 14-mm gauge length, 2-mm width, and thickness of 0.5 mm.

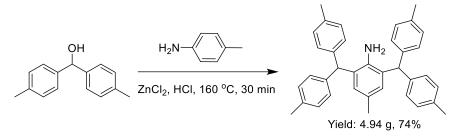
### 2.2 Procedure for the Synthesis of Benzhyldrol.



**Bis(3-methylphenyl)methanol**. 4-methylbromobenzene (12.6 g, 73.8 mmol, 1.0 equiv.) was dissolved in dry THF (75 mL) and cooled to -78 °C under N<sub>2</sub>. *n*-Butyl lithium (46.2 mL, 1.6 M solution in hexane, 73.9 mmol, 1.0 equiv.) was added dropwise and the resulting suspension was stirred at-78 °C for 1 h. Methyl formate (2.22 g, 36.9 mmol, 0.5 equiv.) in THF (15 mL) was then added dropwise over a period of 10 min, and the mixture was stirred at -78 °C for 1 h, then at room temperature for 1 h. Saturated ammonium chloride (aqueous, 100 mL) was added and the mixture was extracted with Et<sub>2</sub>O (3×30 mL). The combined organic layers were

washed with brine (2×30 mL), dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, and concentrated *in vacuo*. The crude product was purified by recrystallization to give a colorless solid (6.67 g, 85.2% yield). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.22 (d, *J* = 8.0 Hz, 4H, aryl-*H*), 7.11 (d, *J* = 8.0 Hz, 4H, aryl-*H*), 5.72 (s, 1H, *CH*(PhMe)<sub>2</sub>), 2.31 (s, 6H, Ar-*CH*<sub>3</sub>) 2.28 (broad peak, 1H, *OH*). <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  141.21, 136.85, 129.02, 126.49, 75.69 (*C*H(PhMe)<sub>2</sub>), 21.09 (Ar-*CH*<sub>3</sub>).

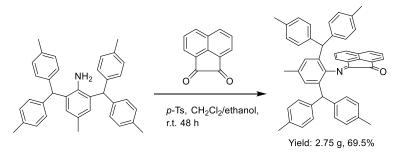
### 2.3 Procedure for the Synthesis of Aniline.



**2,6-Bis(di-p-tolylmethyl)-4-methylaniline.** A mixture of *p*-toluidine (1.44 g, 13.5 mmol, 1.0 equiv.) and bis(p-methylphenyl)methanol (5.71 g, 27.0 mmol, 2.0 equiv.) was heated to 120 °C. A solution of anhydrous zinc chloride (0.92 g, 6.8 mmol, 0.5 equiv.) in concentrated hydrochloric acid (1.13 mL, 37% in H<sub>2</sub>O, 1.0 equiv.) was added to the mixture (exothermic + intense bubbling), and the temperature was raised to 160 °C. After 30 min at 160 °C, the reaction mixture was cooled to room temperature and dissolved in CH<sub>2</sub>Cl<sub>2</sub> (200 mL). The CH<sub>2</sub>Cl<sub>2</sub> layer was washed with water (3 × 100 mL) and dried over anhydrous magnesium sulfate. The solution was concentrated to 20 mL. The product was crashed out with 200 ml methanol and washed with methanol (3 × 100 mL). The desired aniline was obtained as a white crystalline solid at 74.0 % (4.94 g) yield. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.07 (d, *J* = 7.9 Hz, 8H, aryl-*H*), 6.96 (d, *J* = 8.0 Hz, 8H, aryl-*H*), 6.38 (s, 2H, aryl-*H*), 5.36 (s, 2H, *CH*(PhMe)<sub>2</sub>), 2.02 (s, 3H, Ar-*CH*<sub>3</sub>). <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  140.12, 139.73, 136.06, 129.59, 129.48, 129.38, 128.94, 126.62, 51.76 (*C*H(PhMe)<sub>2</sub>), 21.20 (CH(PhMe)<sub>2</sub>), 21.16 (Ar-*C*H<sub>3</sub>). This compound is knwon.<sup>1</sup>

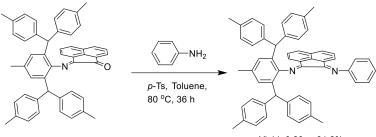
2,6-bis-(sec-phenethyl)-4-methylaniline were synthesized according to the literature.<sup>2</sup>

### 2.4 Procedure for the Synthesis Ligands L1-L5.



**2-((2,6-bis(di-p-tolylmethyl)-4-methylphenyl)imino)acenaphthylen-1-one.** A mixture of 2,6-bis(di-p-tolylmethyl)-4-methylaniline (2.97 g, 6 mmol, 1.0 equiv.) and acenaphthylen-1,2-dione (1.09 g, 6 mmol, 1.0 equiv.) were dissolved in 5 mL ethanol and 100 mL of  $CH_2Cl_2$  containing a catalytic amount of *p*-toluenesulfonic acid and stirred for 48 h at room temperature.

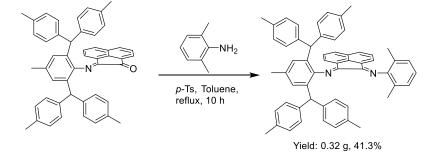
The solvent was evaporated at reduced pressure to give the crude product, which was chromatographed on silica gel with petroleum ether-ethyl acetate (v/v = 30:1). A 2.75 g amount of the product was obtained as red powder in 69.5% yield. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  8.02 (dd, J = 10.8, 7.6 Hz, 2H, aryl-H), 7.76 -7.66 (m, 2H, aryl-H), 7.06-6.99 (m, 5H, aryl-H), 6.96 (t, J = 9.4 Hz, 4H, aryl-H), 6.77 (s, 2H, aryl-H), 6.71 (d, J = 7.9 Hz, 4H, aryl-H), 6.32 (d, J = 7.8 Hz, 4H, aryl-H), 6.03 (d, J = 7.1 Hz, 1H, aryl-H), 5.33 (s, 2H, CH(PhMe)<sub>2</sub>), 2.29 (s, 6H, CH(PhMe)<sub>2</sub>), 2.25 (s, 3H, Ar-CH<sub>3</sub>), 1.67 (s, 6H, CH(PhMe)<sub>2</sub>). <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  190.09 (C=O), 162.56 (C=N), 146.08, 142.38, 140.19, 139.12, 135.60, 134.86, 133.14, 132.77, 131.98, 131.67, 130.24, 129.97, 129.76, 129.47, 129.42, 129.31, 128.93, 128.64, 128.60, 128.43, 127.58, 127.56, 127.33, 126.86, 124.29, 122.22, 121.50, 51.53, 21.65 (Ar-CH<sub>3</sub>), 21.19 (CH(PhMe)<sub>2</sub>), 20.48 (CH(PhMe)<sub>2</sub>). This compound is kwon.<sup>1</sup>



Yield: 0.23 g, 31.2%

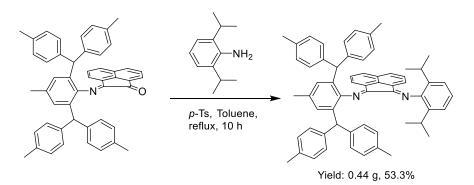
### Acenaphthylene-1-[2,6-bis(di-p-tolylmethyl)-4-methylphenylimino]-2-phenylimine

(L1). A solution of 2-((2,6-bis(di-p-tolylmethyl)-4-methylphenyl)imino)acenaphthylen-1-one (0.66 g, 1 mmol, 1.0 equiv.), aniline (0.19 g, 2 mmol, 2.0 equiv.) and *p*-toluenesulfonic acid (10 mg) in toluene (30 mL) was stirred at 80 °C for 36 h, until there was one main point on the TLC plate. The desired compound was isolated using column chromatography on silica gel with petroleum ether-ethyl acetate (v/v = 40:1). The pure compound was obtained as a yellow solid (31.2%, 0.23 g). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.72 (d, *J* = 8.2 Hz, 1H, aryl-*H*), 7.61 (d, *J* = 8.2 Hz, 1H, aryl-*H*), 7.49 (t, *J* = 7.8 Hz, 2H, aryl-*H*), 7.24 (d, *J* = 4.6 Hz, 1H, aryl-*H*), 7.11-6.92 (m, 12H, aryl-*H*), 6.82-6.75 (m, 6H, aryl-*H*), 6.71 (d, *J* = 7.1 Hz, 1H, aryl-*H*), 6.36 (d, *J* = 7.8 Hz, 4H, aryl-*H*), 6.04 (d, *J* = 7.1 Hz, 1H, aryl-*H*), 5.47 (s, 2H, CH(PhMe)<sub>2</sub>), 2.29 (s, 6H, CH(PhMe)<sub>2</sub>), 2.26 (s, 3H, Ar-CH<sub>3</sub>), 1.68 (s, 6H, CH(PhMe)<sub>2</sub>). <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  163.50 (N=C), 160.99 (N=C), 152.16, 146.75, 140.56, 140.45, 139.29, 135.42, 134.70, 132.48, 132.44, 130.16, 129.88, 129.75, 129.51, 129.48, 129.41, 129.30, 129.25, 128.92, 128.86, 128.70, 128.63, 128.49, 128.35, 128.20, 127.30, 127.04, 126.86, 124.25, 124.22, 123.00, 118.25, 51.57 (CH(PhMe)<sub>2</sub>), 21.67 (Ar-CH<sub>3</sub>), 21.15 (CH(PhMe)<sub>2</sub>), 20.47 (CH(PhMe)<sub>2</sub>). ESI-MS (m/z): calcd for C<sub>55</sub>H<sub>47</sub>N<sub>2</sub>: [M+H]<sup>+</sup>735.3739, found: 735.3751.



## Acenaphthylene-1-[2,6-bis(di-*p*-tolylmethyl)-4-methylphenylimino]-2-(2,6-dimethylphenyl)imine (L2).

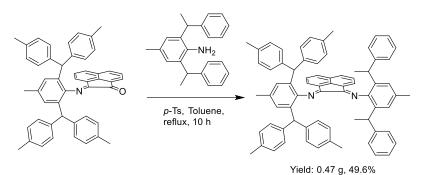
A solution of 2-((2,6-bis(di-p-tolylmethyl)-4-methylphenyl)imino)acenaphthylen-1-one (0.66 g, 1 mmol, 1.0 equiv.), 2,6-dimethylaniline (0.24 g, 2 mmol, 2.0 equiv.) and p-toluenesulfonic acid (10 mg) in toluene (30 mL) was refluxed for 10 h, until there was one main point on the TLC plate. The desired compound was isolated using column chromatography on silica gel with petroleum ether-ethyl acetate (v/v = 40:1). The pure compound was obtained as a yellow solid (41.3%, 0.32 g). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.70 (d, J = 8.2 Hz, 1H, aryl-H), 7.57 (d, J = 8.2 Hz, 1H, aryl-H), 7.24 (d, J = 7.5 Hz, 1H, aryl-H), 7.17 (d, J = 7.5 Hz, 2H, aryl-H), 7.09-6.99 (m, 9H, aryl-H), 6.93-6.88 (m, 1H, aryl-H), 6.79 (dd, J = 9.8, 7.6 Hz, 6H, aryl-H), 6.53 (d, J = 7.1 Hz, 1H, aryl-H), 6.30 (d, J = 7.8 Hz, 4H, aryl-H), 5.92 (d, J = 7.1 Hz, 1H, aryl-H), 5.51 (s, 2H, CH(PhMe)<sub>2</sub>), 2.29 (s, 6H, CH(PhMe)<sub>2</sub>), 2.27 (s, 3H, Ar-CH<sub>3</sub>), 2.23 (s, 6H, aryl-CH<sub>3</sub>), 1.63 (s, 6H, CH(PhMe)<sub>2</sub>). <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ 163.72 (N=C), 161.66 (N=C), 149.53, 146.89, 140.60, 139.87, 139.14, 135.50, 134.71, 132.54, 129.94, 129.87, 129.55, 129.03, 128.90, 128.57, 128.51, 128.47, 128.42, 128.40, 127.01, 126.99, 125.01, 124.59, 123.74, 121.68, 51.65 (CH(PhMe)<sub>2</sub>), 29.84 (Ar-CH<sub>3</sub>), 21.67 (Aryl-CH<sub>3</sub>), 21.13 (CH(PhMe)<sub>2</sub>), 20.42 (CH(PhMe)<sub>2</sub>), 18.29 (aryl-CH<sub>3</sub>). ESI-MS (m/z): calcd for C<sub>57</sub>H<sub>51</sub>N<sub>2</sub>: [M+H]<sup>+</sup>763.4052, found: 763.4066.



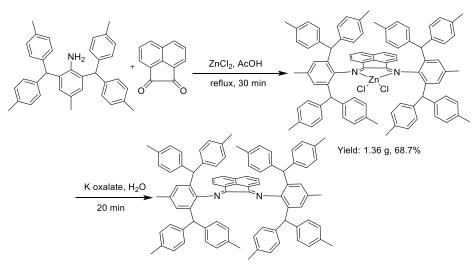
# Acenaphthylene-1-[2,6-bis(di-*p*-tolylmethyl)-4-methylphenylimino]-2-(2,6-diisopropylphenyl)imine (L3).

A solution of 2-((2,6-bis(di-p-tolylmethyl)-4-methylphenyl)imino)acenaphthylen-1-one (0.66 g, 1 mmol, 1.0 equiv.), 2,6-diisopropylaniline (0.35 g, 2 mmol, 2.0 equiv.) and *p*-toluenesulfonic acid (10 mg) in toluene (30 mL) was refluxed for 10 h, until there was one main point on the TLC plate. The solution was evaporated at reduced pressure, and the remaining solution was diluted in methanol (100 mL). The yellow solid was isolated by filtration, followed by recrystallization from dichloromethane and *n*-hexane (53.3%, 0.44 g). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.66 (d, *J* = 8.2 Hz, 1H), 7.50 (d, *J* = 8.2 Hz, 1H), 7.31-7.26 (m, 3H), 7.23-7.19 (m, 1H), 7.03 (dd, *J* = 19.8, 8.1 Hz, 8H), 6.79 (dd, *J* = 14.5, 8.4 Hz, 7H), 6.44 (d, *J* = 7.1 Hz, 1H), 6.25 (d, *J* = 7.8 Hz, 4H), 5.72 (d, *J* = 7.1 Hz, 1H), 5.54 (s, 2H, CH(PhMe)<sub>2</sub>), 3.24-3.13 (m, 2H, CH(CH<sub>3</sub>)<sub>2</sub>), 2.30 (s, 6H, CH(PhMe)<sub>2</sub>), 2.28 (s, 3H, Ar-CH<sub>3</sub>), 1.62 (s, 6H, CH(PhMe)<sub>2</sub>), 1.28 (d, *J* = 6.8 Hz, 6H, CH(CH<sub>3</sub>)<sub>2</sub>), 1.02 (d, *J* = 6.9 Hz, 6H, CH(CH<sub>3</sub>)<sub>2</sub>). <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  163.92 (N=C), 162.32 (N=C), 147.28, 146.96, 140.94, 140.11, 139.88, 138.94, 135.88, 135.54, 135.47, 134.69, 132.59, 132.53, 129.85, 129.68, 129.57, 129.46, 129.26, 129.00, 128.93, 128.81, 128.62, 128.56, 128.38, 128.11, 127.99, 126.96,

126.83, 126.58, 124.65, 124.51, 124.42, 123.64, 123.59, 123.47, 122.83, 51.51 (*C*H(PhMe)<sub>2</sub>), 29.84 (*C*H(CH<sub>3</sub>)<sub>2</sub>), 28.62 (*C*H(CH<sub>3</sub>)<sub>2</sub>), 24.37 (*C*H(*C*H<sub>3</sub>)<sub>2</sub>), 23.84 (*C*H(*C*H<sub>3</sub>)<sub>2</sub>), 23.55 (*C*H(*C*H<sub>3</sub>)<sub>2</sub>), 23.27 (*C*H(*C*H<sub>3</sub>)<sub>2</sub>), 21.66 (Ar-*C*H<sub>3</sub>), 21.12 (*C*H(Ph*Me*)<sub>2</sub>), 20.43 (*C*H(Ph*Me*)<sub>2</sub>). ESI-MS (m/z): calcd for C<sub>61</sub>H<sub>59</sub>N<sub>2</sub>:  $[M+H]^+$  819.4678, found: 819.4694.



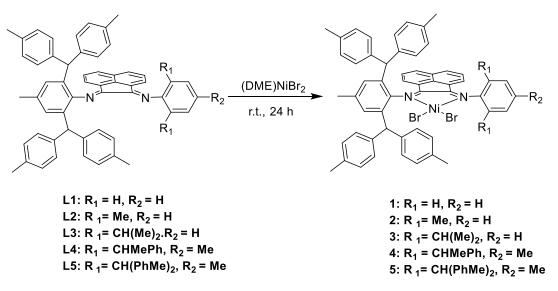
Acenaphthylene-1-[2,6-bis(di-p-tolylmethyl)-4-methylphenylimino]-2-(2,6-bis-(secphenethyl)-4-methylphenyl)imine (L4). Using the same procedure as for the synthesis of L3, L4 was obtained as a yellow powder at 49.6% yield (0.47 g). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$ 7.70-7.53 (m, 2H), 7.18 – 6.65 (m, 28H), 6.55 – 6.16 (m, 5H), 6.12 – 5.83 (m, 1H), 5.68 – 5.46 (m, 2H, CH(PhMe)<sub>2</sub>), 4.34 – 4.24 (m, 2H, CHPhMe), 2.43 – 2.19 (m, 12H, CH(PhMe)<sub>2</sub>+ Ar-CH<sub>3</sub>), 1.88 – 1.54 (m, 12H, CH(PhMe)<sub>2</sub>+ CHPhMe).<sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ 163.77 (C=N), 162.79 (C=N), 162.13 (C=N), 161.60 (C=N), 146.79, 145.71, 145.44, 145.07, 141.11, 140.64, 140.61, 140.08, 139.98, 139.82, 139.24, 135.97, 135.32, 135.26, 134.59, 133.73, 132.49, 132.13, 129.77, 129.47, 129.37, 129.17, 128.91, 128.70, 128.65, 128.50, 128.19, 128.10, 127.98, 127.84, 127.75, 127.67, 127.43, 126.51, 125.52, 125.01, 124.88, 123.60, 51.64 (CH(PhMe)<sub>2</sub>), 51.43 (CH(PhMe)<sub>2</sub>), 51.34 (CH(PhMe)<sub>2</sub>), 51.28 (CH(PhMe)<sub>2</sub>), 51.13 (CH(PhMe)<sub>2</sub>), 40.59 (CHPhMe), 40.40 (CHPhMe), 40.32 (CHPhMe), 40.06 (CHPhMe), 39.73 (CHPhMe), 22.30 (Ar-CH<sub>3</sub>), 22.27 (Ar-CH<sub>3</sub>), 21.86 (CH(PhMe)<sub>2</sub>), 21.73 (CH(PhMe)<sub>2</sub>), 21.66 (CH(PhMe)<sub>2</sub>), 21.56(CH(PhMe)<sub>2</sub>), 21.19 (CHPhMe), 21.09 (CHPhMe), 21.02 (CHPhMe), 20.94 (CHPhMe), 20.37 (CHPhMe), 20.34 (CHPhMe). ESI-MS (m/z): calcd for C<sub>72</sub>H<sub>65</sub>N<sub>2</sub>: [M+H]<sup>+</sup>957.5148, found: 957.5170.



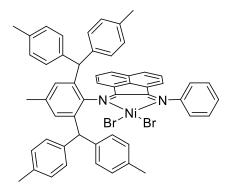
Yield: 0.85 g, 70.1%

Acenaphthylene-bis[2,6-bis(di-p-tolylmethyl)-4-methylphenyl]-1,2-diimine (L4). ZnCl<sub>2</sub> (0.24 g, 1.79 mmol) and acenaphthenequinone (0.29 g, 1.57 mmol) were suspended in glacial acetic acid (6 mL). 2,6-Bis(diphenylmethyl)-4-methylaniline (1.77 g, 3.57 mmol) was added, and the reaction mixture was refluxed under stirring for 30 min. The solution was allowed to cool to room temperature, and a bright orange-red solid precipitated. The solid was separated by filtration and washed with acetic acid  $(3 \times 5mL)$  and diethyl ether  $(5 \times 5mL)$ , to remove remaining acetic acid. Drying under vacuum gave bright orange-red, poorly soluble solid (1.36 g, 68.7%). Then the zinc was removed from the zinc diimine complex. The product of the previous step was suspended in methylene chloride (24 mL), and a solution of potassium oxalate (0.22 g, 1.2 mmol) in water (2 mL) was added. The reaction mixture was stirred vigorously for 20 min. The two phases were separated, and the organic layer was washed with water  $(3 \times 10 \text{ mL})$  and dried with MgSO<sub>4</sub>. After filtration the solvent was removed under vacuum to afford the product as an orange powder (0.85 g, 70.1%), followed by recrystallization from dichloromethane and *n*-hexane. The total yield of two steps is 48.2%. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.55 (d, J = 8.1 Hz, 2H), 6.99 – 6.83 (m, 22H), 6.72 (d, J = 5.6 Hz, 8H), 6.39 (d, J = 5.6 Hz, 8H), 6.12 (d, J = 5.1 Hz, 2H), 5.56 (s, 4H, CH(PhMe)<sub>2</sub>), 2.27 (s, 6H, Ar-CH<sub>3</sub>), 2.25 (s, 12H, CH(PhMe)<sub>2</sub>), 1.88 (s, 12H, CH(PhMe)<sub>2</sub>). <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ 163.53 (C=N), 146.70, 141.17, 140.07, 139.82, 135.25, 134.80, 132.52, 131.84, 129.91, 129.49, 129.21, 128.78, 128.60, 128.55, 127.09, 126.57, 124.49, 50.73 (CH(PhMe)<sub>2</sub>), 21.64 (Ar-CH<sub>3</sub>), 21.09(CH(PhMe)<sub>2</sub>), 20.69 (CH(PhMe)<sub>2</sub>). This compound is known.<sup>1</sup>

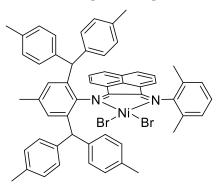
#### 2.5 Procedure for the Synthesis of Nickel Complexes 1-5.



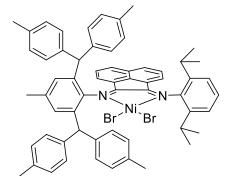
Complexes 1-5 were synthesized by the reaction of 1 equiv. of (DME)NiBr<sub>2</sub> with the corresponding ligands in methylene chloride. The corresponding ligand (0.3 mmol) was added in 10 mL of methylene chloride in a Schlenk tube under a nitrogen atmosphere. A suspension of (DME)NiBr<sub>2</sub> (0.3 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) was added to the above solution. The resulting mixture was stirred at room temperature for 24 hours. The solvent was evaporated under reduced pressure to afford a solid. The product was washed with  $4 \times 5$  mL diethyl ether and dried under vacuum. The single crystal can be obtained by diffusion from layering hexane on to the CH<sub>2</sub>Cl<sub>2</sub> solution at room temperature.



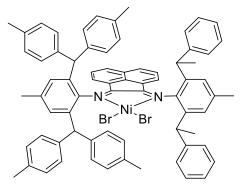
**Complex 1.** Brown solid. Yield: 0.25 g, 86%. Elem. Anal. Calcd for  $C_{55}H_{46}Br_2N_2N_i$ : C, 69.28; H, 4.86; N, 2.94. Found: C, 69.02; H, 4.83; N, 2.99. MALDI-TOF-MS(m/z): calcd for  $C_{55}H_{47}BrN_2N_i$ : [M-Br+H]<sup>+</sup> 872.2276, found:872.8820.



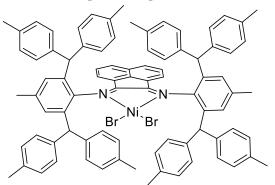
**Complex 2.** Brown solid. Yield: 0.26 g, 89.0%. Elem. Anal. Calcd for  $C_{57}H_{50}Br_2N_2N_1$ : C, 69.75; H, 5.13; N, 2.85. Found: C, 69.53; H, 4.99; N, 2.68. MALDI-TOF-MS(m/z): calcd for  $C_{57}H_{50}BrN_2N_1$ : [M-Br]<sup>+</sup> 901.2490, found: 901.2432.



**Complex 3.** Brown solid. Yield: 0.28 g, 91%. Elem. Anal. Calcd for  $C_{61}H_{58}Br_2N_2N_i$ : C, 70.61; H, 5.63; N, 2.70. Found: C, 70.75; H, 5.73; N, 2.47. MALDI-TOF-MS(m/z): calcd for  $C_{61}H_{58}BrN_2N_i$ : [M-Br]<sup>+</sup> 957.3116, found: 957.3694.



**Complex 4.** Brown solid. Yield 0.31 g, 87%. Elem. Anal. Calcd for  $C_{72}H_{64}Br_2N_2Ni$ : C, 73.55; H, 5.49; N, 2.38. Found: C, 73.40; H, 5.51; N, 2.46. MALDI-TOF-MS(m/z): calcd for  $C_{72}H_{65}BrN_2Ni$ : [M-Br+H]<sup>+</sup>1094.3685, found: 1094.8672.



**Complex 5.** Brown solid. Yield: 0.38 g, 93%. This compound is known.<sup>3</sup>

### 2.6 General in-Situ Activated Polymerization Procedure.

Under an inert atmosphere, a 350 mL glass thick-walled pressure vessel was charged with MAO, 40 mL toluene, and a magnetic stir bar. The vessel was pressurized with 1 atm of ethylene and allowed to equilibrate under constant pressure for 10 minutes with stirring. The nickel complex in 1 mL CH<sub>2</sub>Cl<sub>2</sub> was injected and the ethylene pressure was increased to 8 atm to initiate polymerization and stirred continuously for the desired time. The polymerization was quenched via the addition of MeOH (5 mL) and the polymer was precipitated using excess acidic MeOH (5% HCl in MeOH) and dried in a vacuum oven to constant weight. Polymer branching density was determined by <sup>1</sup>H NMR. B =  $1000 \times 2(I_{CH3})/3(I_{CH2+CH}+I_{CH3})$ . CH<sub>3</sub> (m, 0.77-0.95 ppm); CH<sub>2</sub> and CH (m, ca. 1.0-1.45 ppm).

### 2.7 Copolymerization of Ethylene and UA

In a typical experiment, a 300 mL stainless pressure reactor connected with a high pressure gas line was firstly dried at 90 °C under vacuum for at least 1 h. The reactor was then adjusted to the desired polymerization temperature. 20 mL of toluene with the desired amount Et<sub>2</sub>AlCl was added to the reactor under N<sub>2</sub> atmosphere, then the desired polar monomer and the desired amount of Ni catalyst in 2 mL of CH<sub>2</sub>Cl<sub>2</sub> was injected into the polymerization system via syringe subsequently. With a rapid stirring, the reactor was pressurized and maintained at the desired pressure of ethylene. After 1 h, the pressure reactor was vented and the copolymer was dried under vacuum overnight.

### 3. Spectra Data

### 3.1 <sup>1</sup>H and <sup>13</sup>C NMR of the Synthetic Compounds.

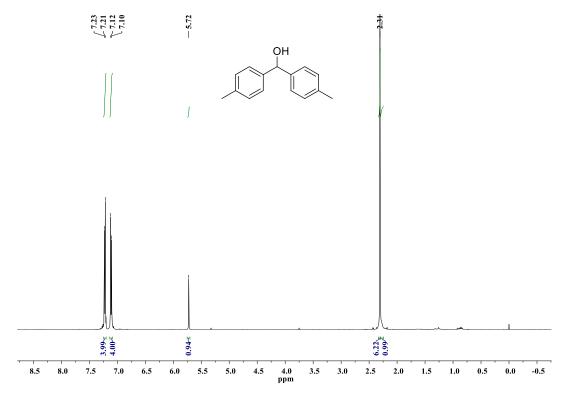


Figure S4. <sup>1</sup>H NMR spectrum of bis(3-methylphenyl)methanol in CDCl<sub>3</sub>

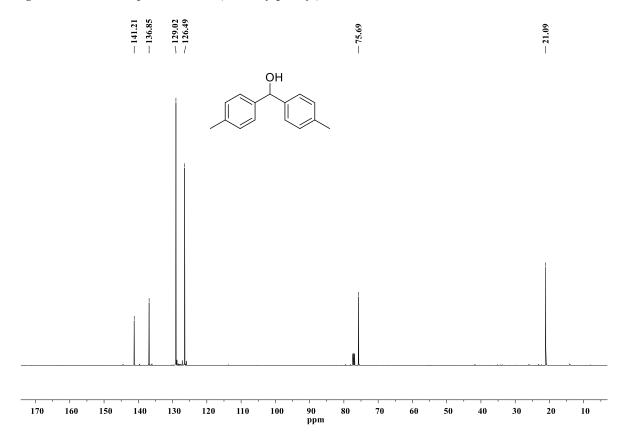


Figure S5. <sup>13</sup>C NMR spectrum of bis(3-methylphenyl)methanol in CDCl<sub>3</sub>.

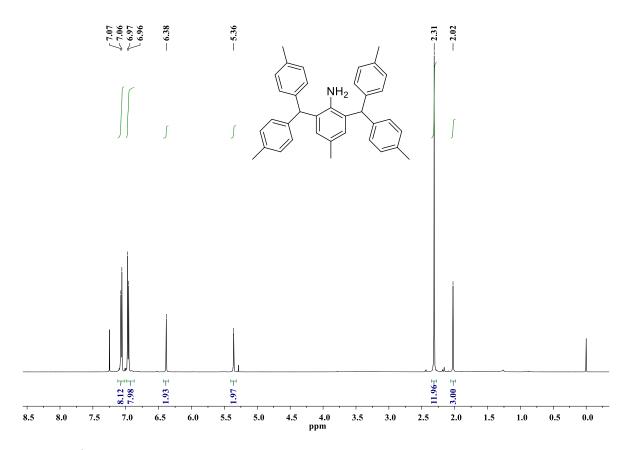


Figure S6. <sup>1</sup>H NMR spectrum of 2,6-Bis(di-p-tolylmethyl)-4-methylaniline in CDCl<sub>3</sub>.

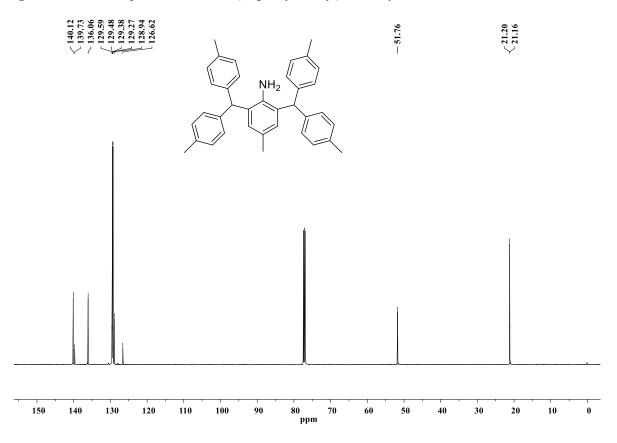
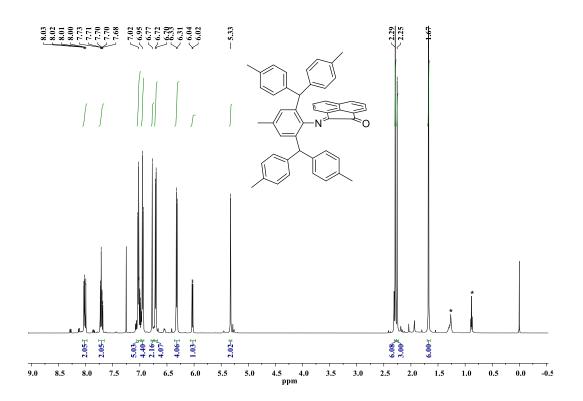
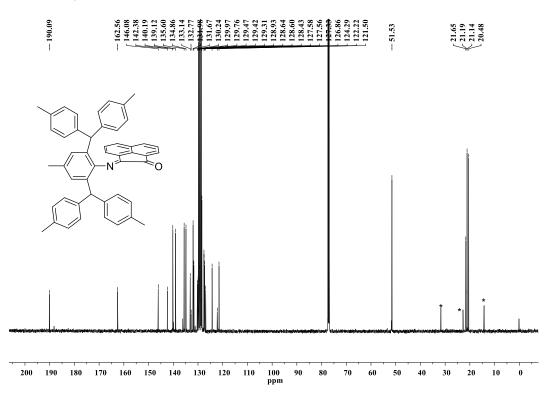


Figure S7. <sup>13</sup>C NMR spectrum of 2,6-Bis(di-p-tolylmethyl)-4-methylaniline in CDCl<sub>3</sub>.



**Figure S8.** <sup>1</sup>H NMR spectrum of **2-((2,6-bis(di-p-tolylmethyl)-4-methylphenyl)imino)acenaphthylen-1-one** in CDCl<sub>3</sub>. \*hexanes



**Figure S9.** <sup>13</sup>C NMR spectrum of **2-((2,6-bis(di-p-tolylmethyl)-4-methylphenyl)imino)acenaphthylen-1-one** in CDCl<sub>3</sub>. \*hexanes

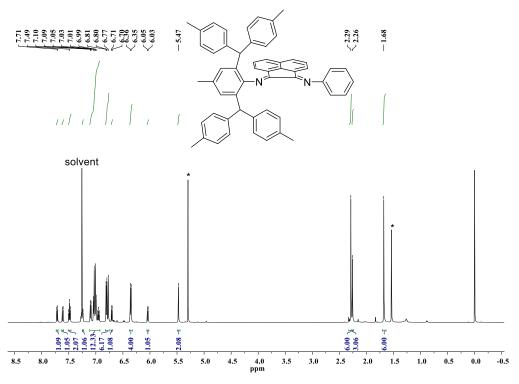


Figure S10. <sup>1</sup>H NMR spectrum of L1 in CDCl<sub>3</sub>.\*DCM, water.

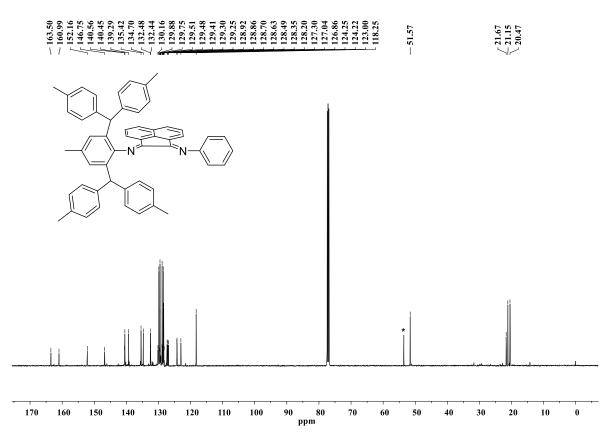


Figure S11. <sup>13</sup>C NMR spectrum of L1 in CDCl<sub>3</sub>. \*DCM.

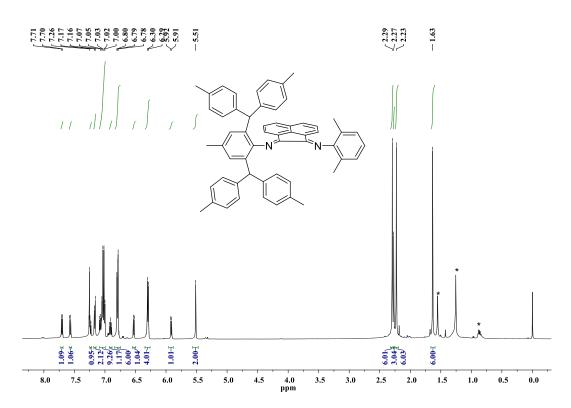


Figure S12. <sup>1</sup>H NMR spectrum of L2 in CDCl<sub>3</sub>. \*hexanes, water.

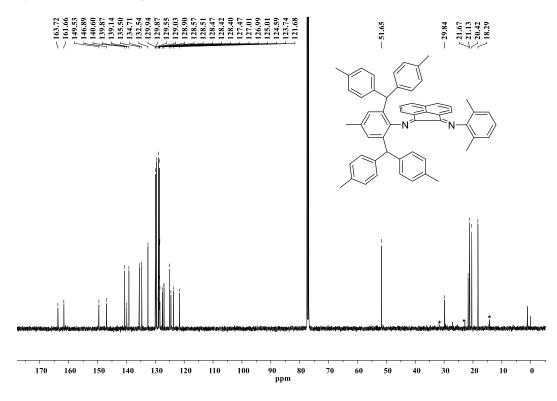


Figure S13. <sup>13</sup>C NMR spectrum of L2 in CDCl<sub>3</sub>. \*hexanes

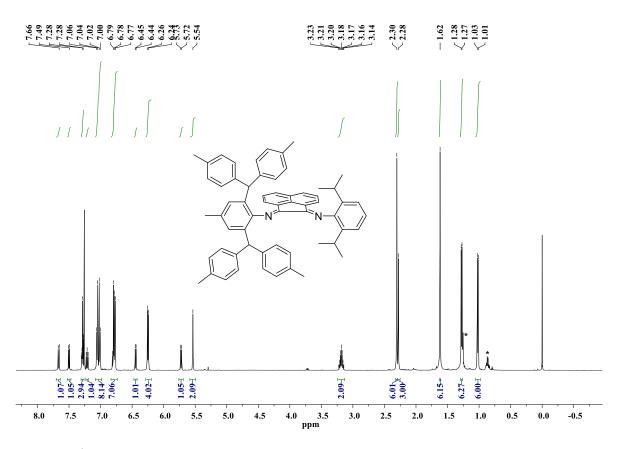


Figure S14. <sup>1</sup>H NMR spectrum of L3 in CDCl<sub>3</sub>. \*hexanes

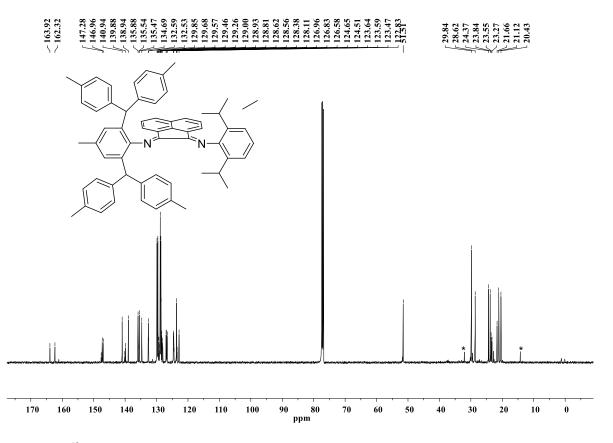


Figure S15. <sup>13</sup>C NMR spectrum of L3 in CDCl<sub>3</sub>. \*hexanes

 $\begin{array}{c} 7.768\\ -7.707\\ -7.707\\ -7.707\\ -7.707\\ -7.707\\ -7.707\\ -7.707\\ -7.707\\ -7.707\\ -7.707\\ -7.707\\ -7.707\\ -7.707\\ -6.84\\ -6.84\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.88\\ -6.$ 

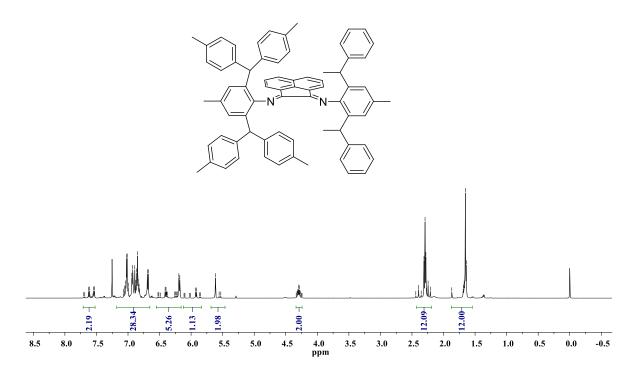
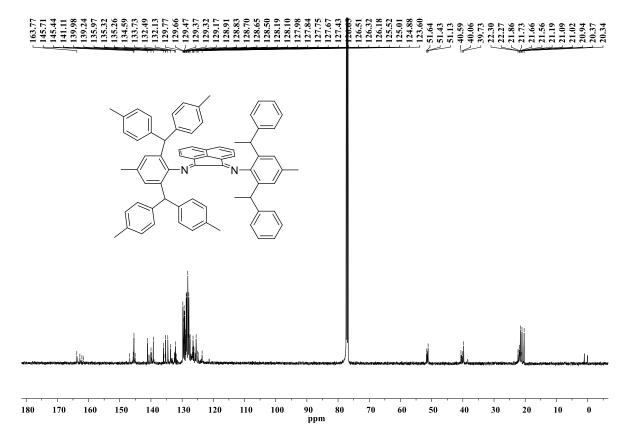
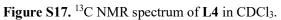


Figure S16. <sup>1</sup>H NMR spectrum of L4 in CDCl<sub>3</sub>.





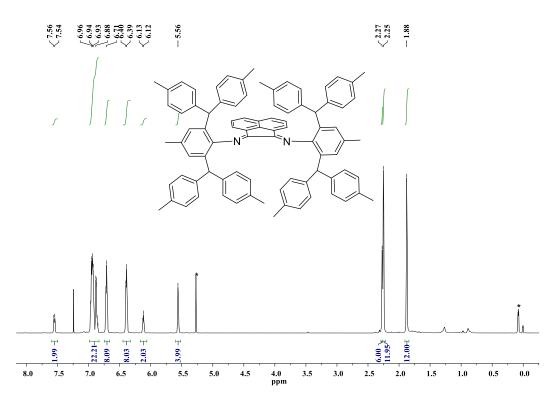


Figure S18. <sup>1</sup>H NMR spectrum of L5 in CDCl<sub>3</sub>. \*DCM, greese.

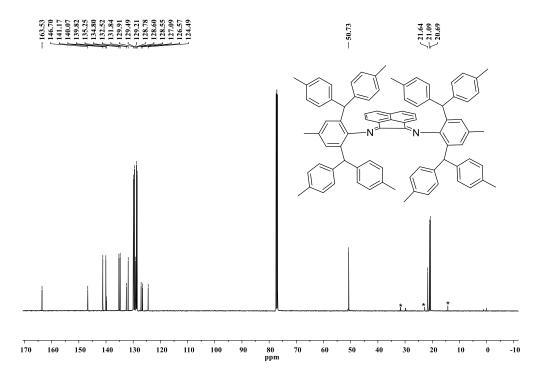


Figure S19. <sup>13</sup>C NMR spectrum of L5 in CDCl<sub>3</sub>. \*hexanes.

### 3.2 ESI-MS of Ligand L1-L4.

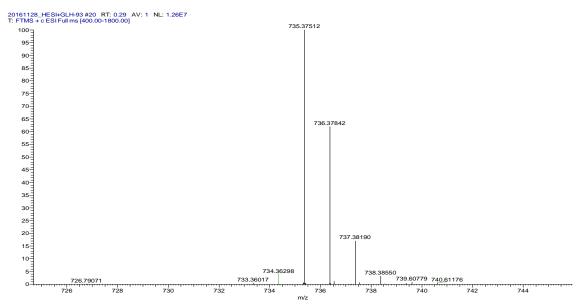


Figure S20. ESI-MS of L1.

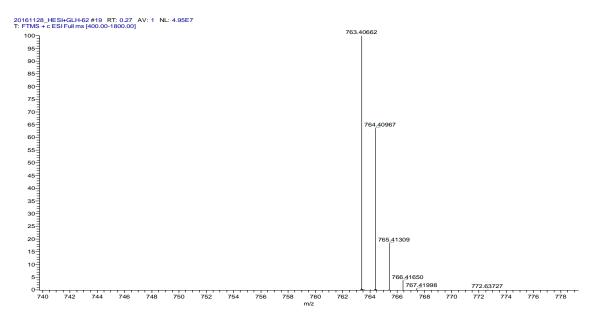
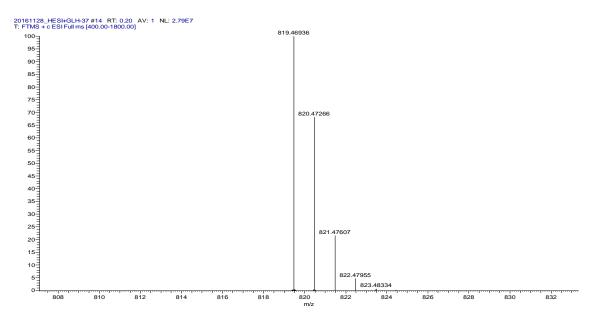


Figure S21. ESI-MS of L2.





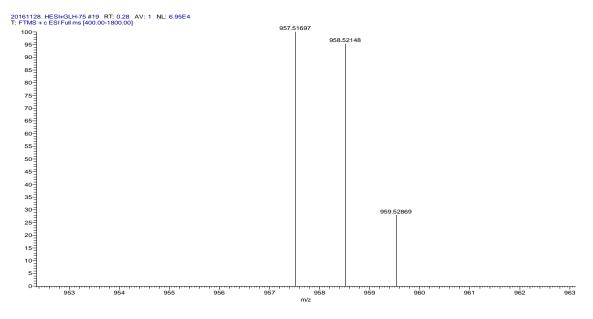


Figure S23. ESI-MS of L4.

## 3.3 MALDI-TOF of Complexes 1-4.

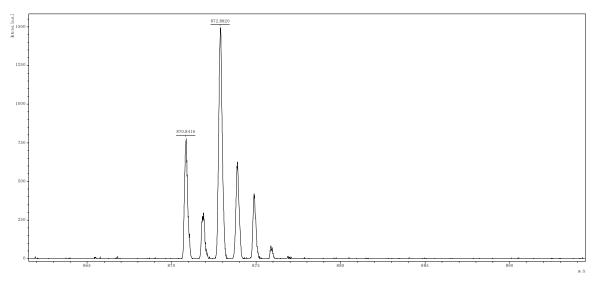


Figure S24. MALDI-TOF-MS of complex 1.

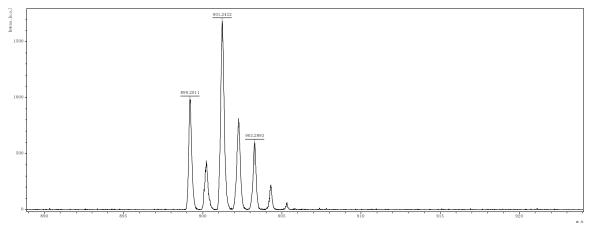


Figure S25. MALDI-TOF-MS of complex 2.

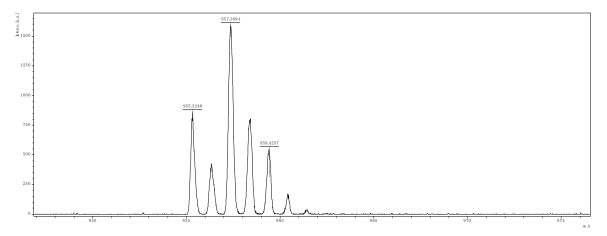


Figure S26. MALDI-TOF-MS of complex 3.

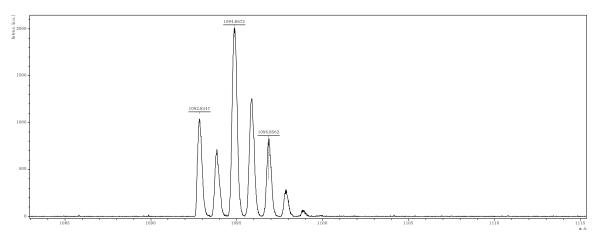
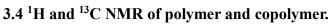


Figure S27. MALDI-TOF-MS of complex 4.



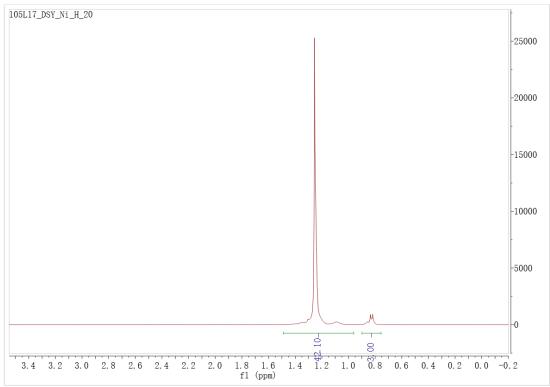


Figure S28. <sup>1</sup>H NMR spectrum of the polymer from table 1, entry 1 (CDCl<sub>2</sub>CDCl<sub>2</sub>, 120 °C).

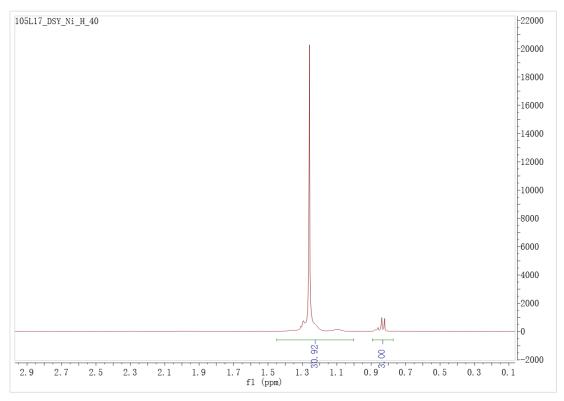


Figure S29. <sup>1</sup>H NMR spectrum of the polymer from table 1, entry 2 (CDCl<sub>2</sub>CDCl<sub>2</sub>, 120 °C).

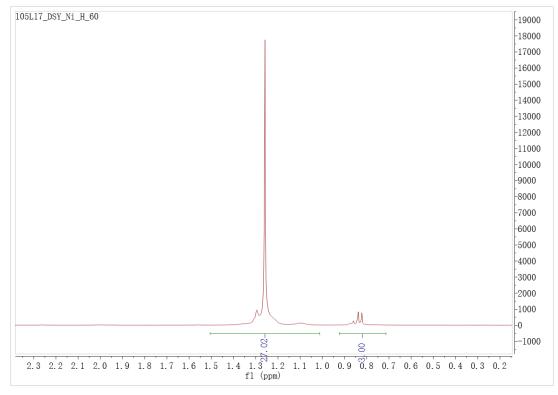


Figure S30. <sup>1</sup>H NMR spectrum of the polymer from table 1, entry 3 (CDCl<sub>2</sub>CDCl<sub>2</sub>, 120 °C).

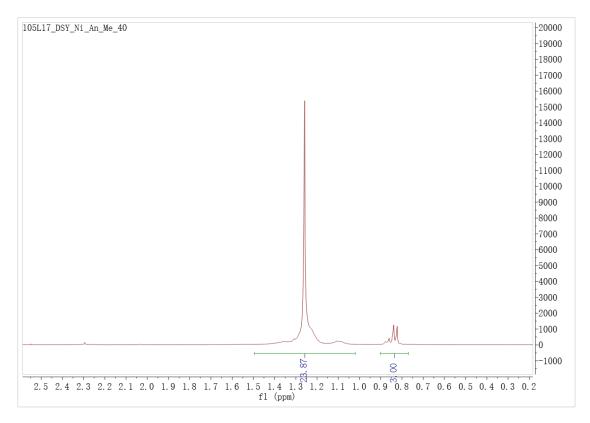


Figure S31. <sup>1</sup>H NMR spectrum of the polymer from table 1, entry 6 (CDCl<sub>2</sub>CDCl<sub>2</sub>, 120 °C).

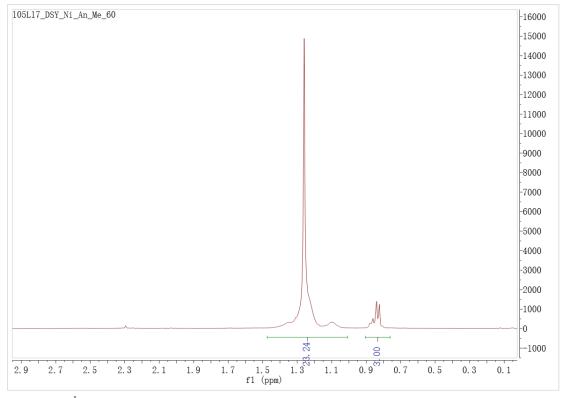


Figure S32. <sup>1</sup>H NMR spectrum of the polymer from table 1, entry 7 (CDCl<sub>2</sub>CDCl<sub>2</sub>, 120 °C).

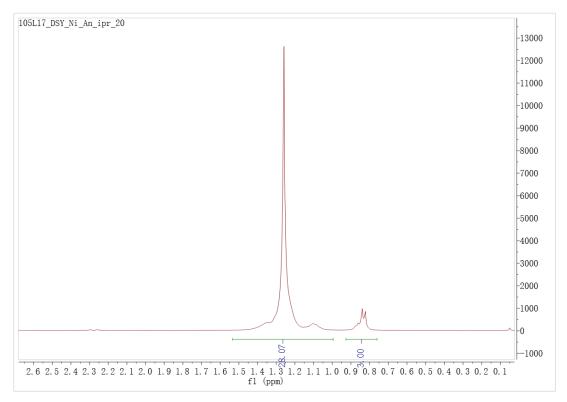


Figure S33. <sup>1</sup>H NMR spectrum of the polymer from table 1, entry 9 (CDCl<sub>2</sub>CDCl<sub>2</sub>, 120 °C).

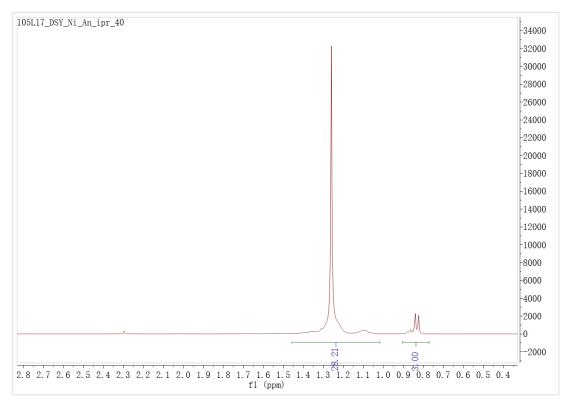


Figure S34. <sup>1</sup>H NMR spectrum of the polymer from table 1, entry 10 (CDCl<sub>2</sub>CDCl<sub>2</sub>, 120 °C).

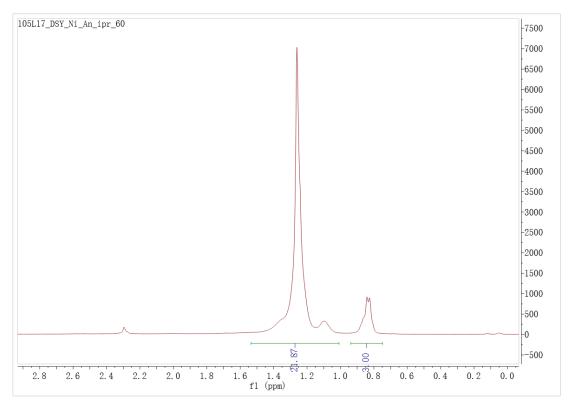


Figure S35. <sup>1</sup>H NMR spectrum of the polymer from table 1, entry 11 (CDCl<sub>2</sub>CDCl<sub>2</sub>, 120 °C).

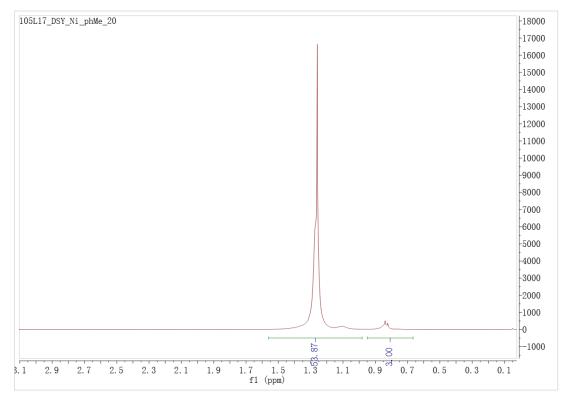


Figure S36. <sup>1</sup>H NMR spectrum of the polymer from table 1, entry 13 (CDCl<sub>2</sub>CDCl<sub>2</sub>, 120 °C).

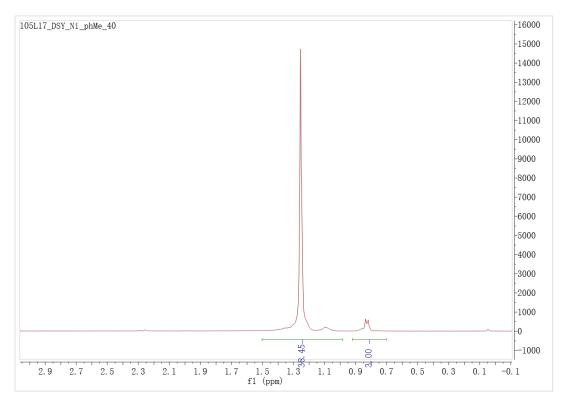


Figure S37. <sup>1</sup>H NMR spectrum of the polymer from table 1, entry 14 (CDCl<sub>2</sub>CDCl<sub>2</sub>, 120 °C).

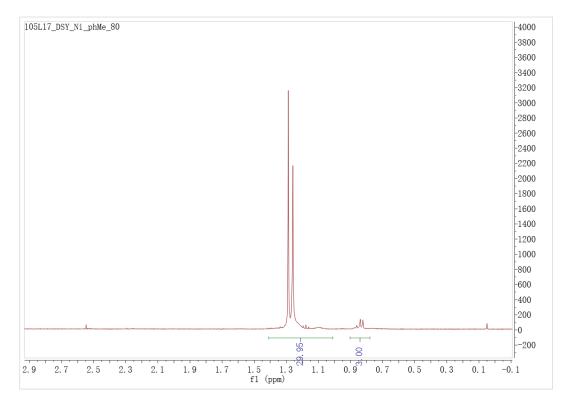


Figure S38. <sup>1</sup>H NMR spectrum of the polymer from table 1, entry 16 (CDCl<sub>2</sub>CDCl<sub>2</sub>, 120 °C).

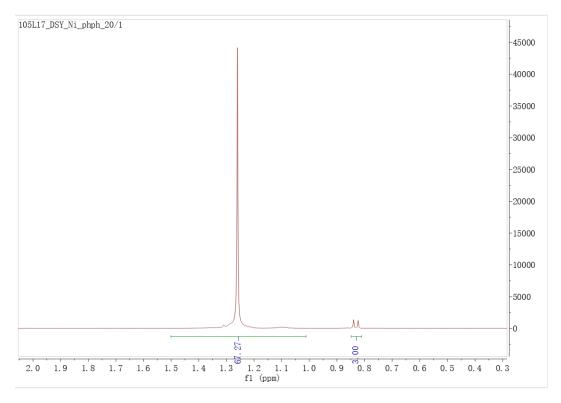


Figure S39. <sup>1</sup>H NMR spectrum of the polymer from table 1, entry 17 (CDCl<sub>2</sub>CDCl<sub>2</sub>, 120 °C).

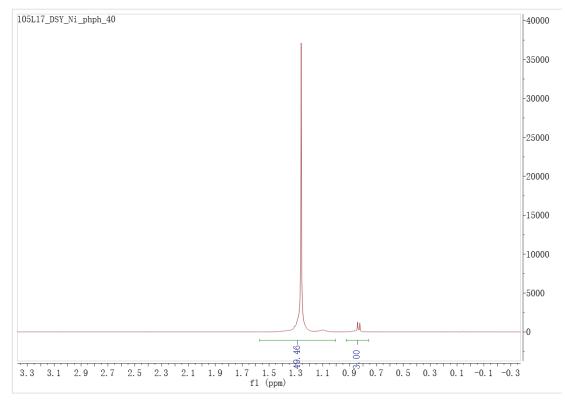


Figure S40. <sup>1</sup>H NMR spectrum of the polymer from table 1, entry 18 (CDCl<sub>2</sub>CDCl<sub>2</sub>, 120 °C).

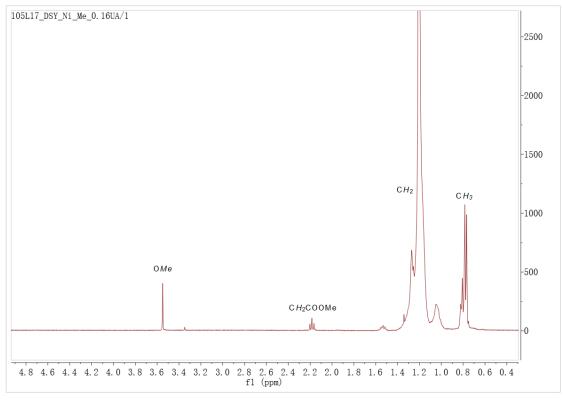


Figure S41. <sup>1</sup>H NMR spectrum of the polymer from table 3, entry 11 (CDCl<sub>2</sub>CDCl<sub>2</sub>, 120 °C).

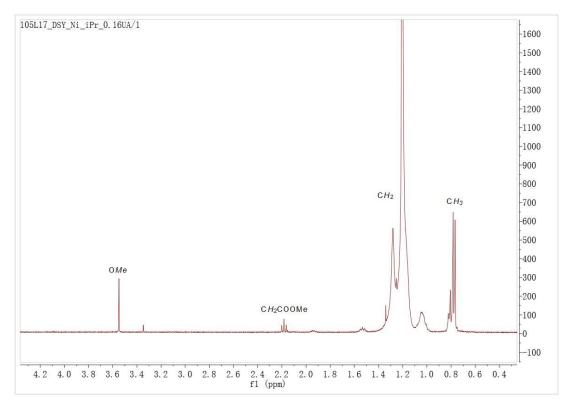


Figure S42. <sup>1</sup>H NMR spectrum of the polymer from table 3, entry 12 (CDCl<sub>2</sub>CDCl<sub>2</sub>, 120 °C).

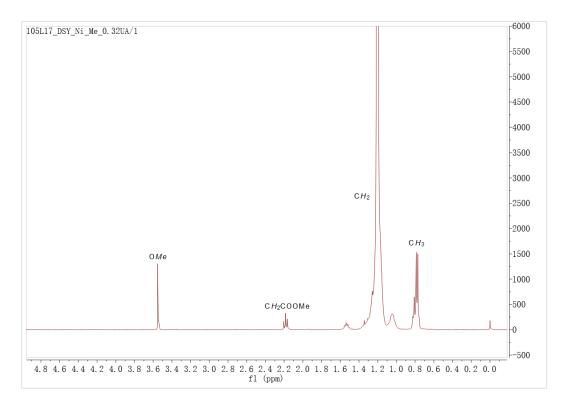


Figure S43. <sup>1</sup>H NMR spectrum of the polymer from table 3, entry 14 (CDCl<sub>2</sub>CDCl<sub>2</sub>, 120 °C).

### 3.5 DSC, GPC of polymer and copolymer.

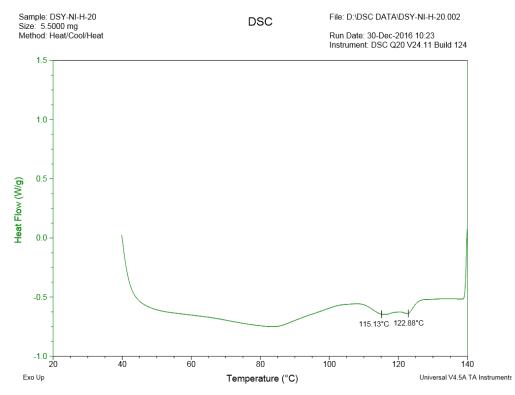


Figure S44. DSC of the polymer from table 1, entry 1.

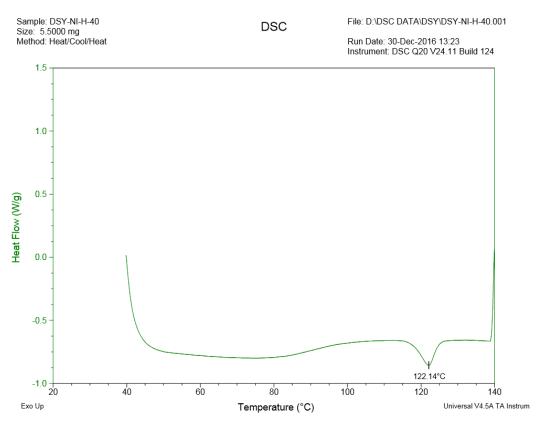


Figure S45. DSC of the polymer from table 1, entry 2.

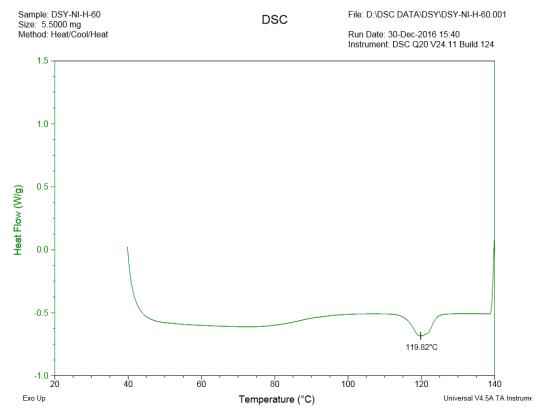


Figure S46. DSC of the polymer from table 1, entry 3.

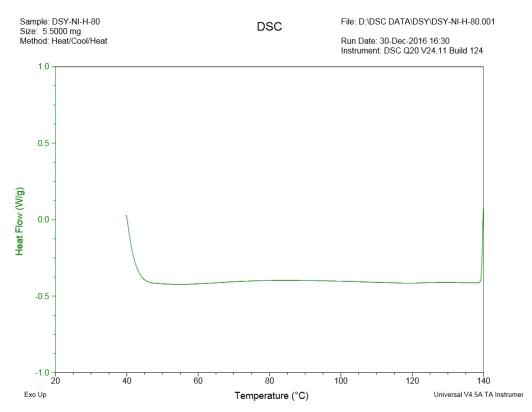


Figure S47. DSC of the polymer from table 1, entry 4.

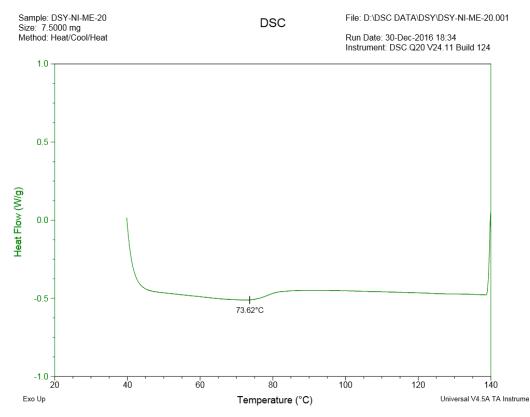


Figure S48. DSC of the polymer from table 1, entry 5.

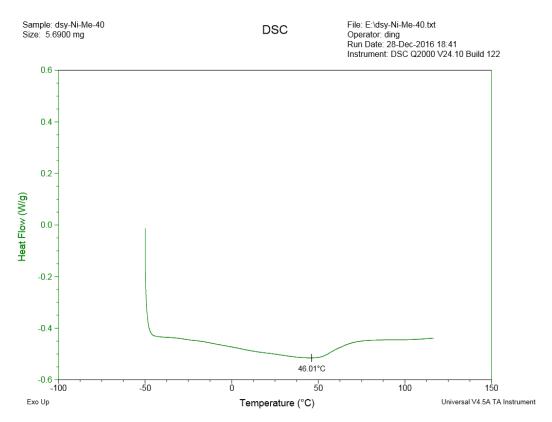


Figure S49. DSC of the polymer from table 1, entry 6.

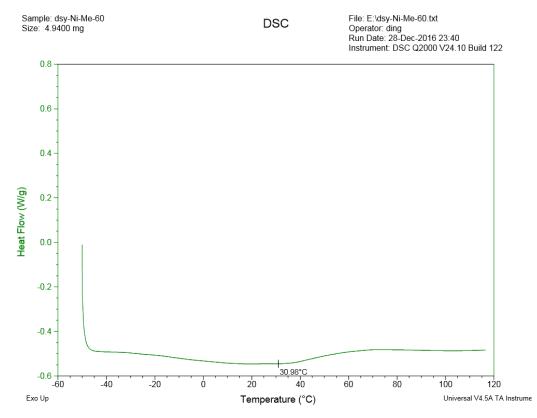


Figure S50. DSC of the polymer from table 1, entry 7.

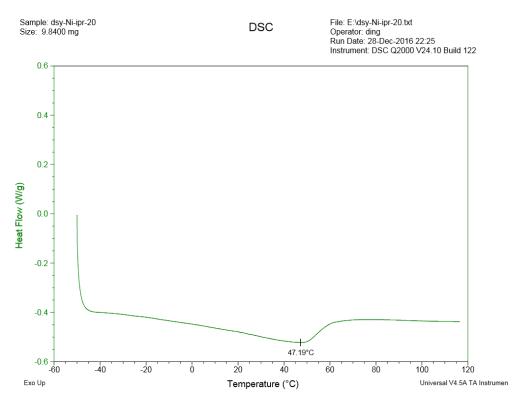


Figure S51. DSC of the polymer from table 1, entry 9.

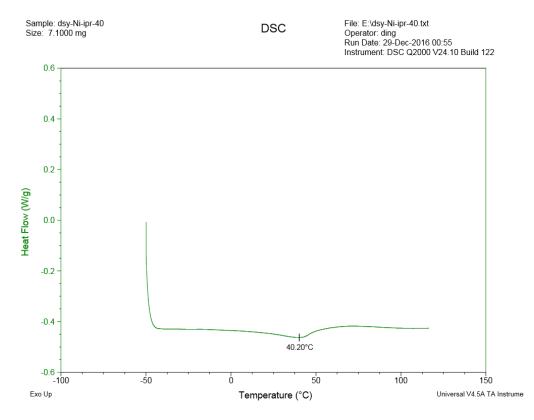


Figure S52. DSC of the polymer from table 1, entry 10.

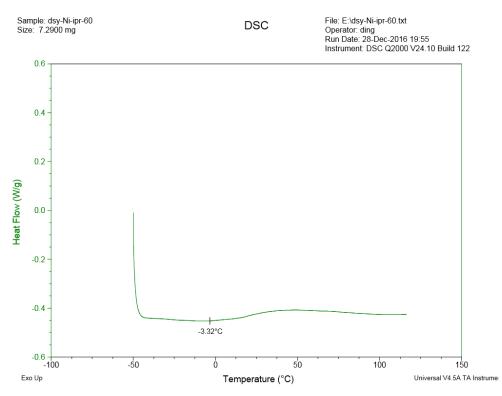


Figure S53. DSC of the polymer from table 1, entry 11.

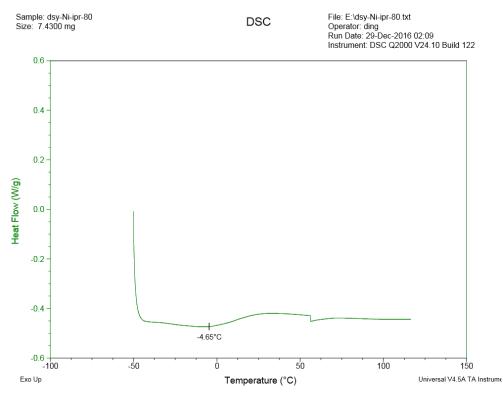


Figure S54. DSC of the polymer from table 1, entry 12.

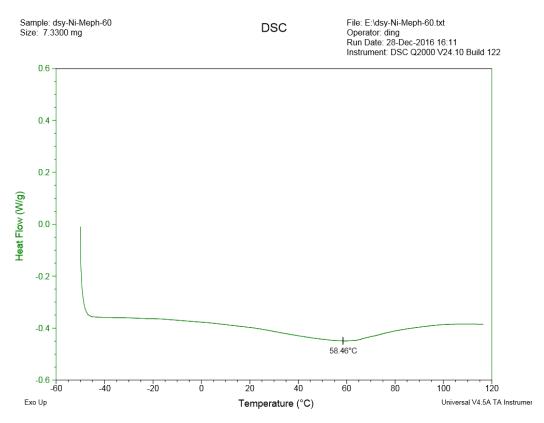


Figure S55. DSC of the polymer from table 1, entry 15.

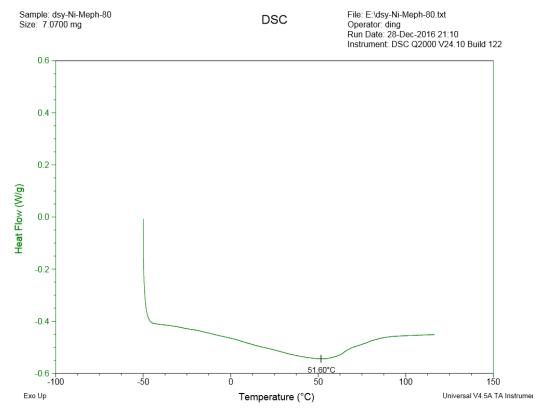


Figure S56. DSC of the polymer from table 1, entry 16.

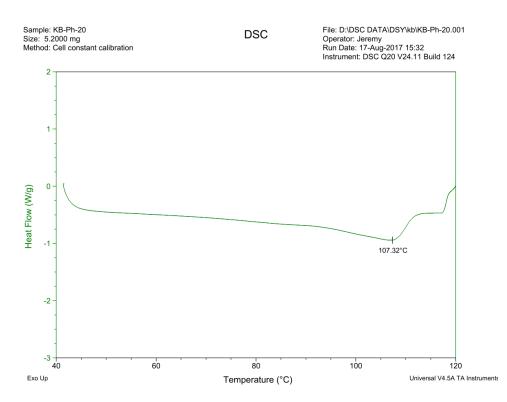


Figure S57. DSC of the polymer from table 1, entry 17.

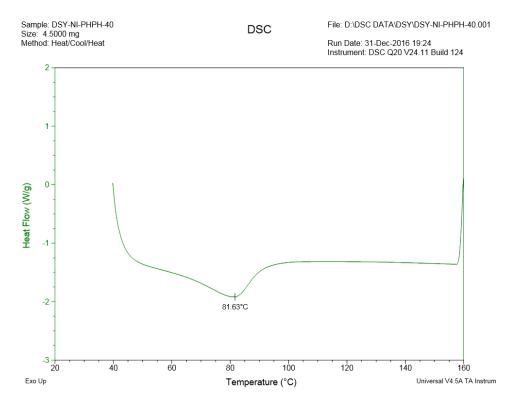


Figure S58. DSC of the polymer from table 1, entry 18.

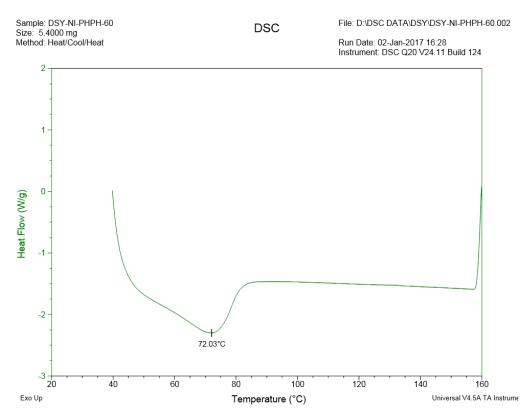


Figure S59. DSC of the polymer from table 1, entry 19.

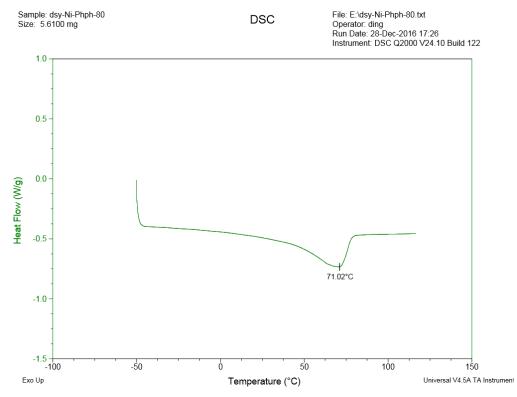
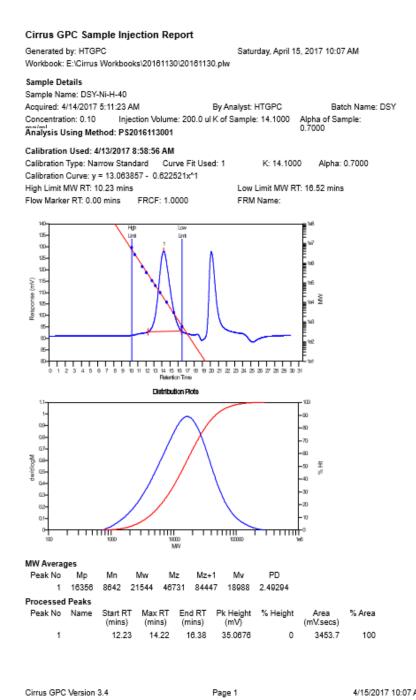


Figure S60. DSC of the polymer from table 1, entry 20.



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Figure S61. GPC of the polymer from table 1, entry 2.

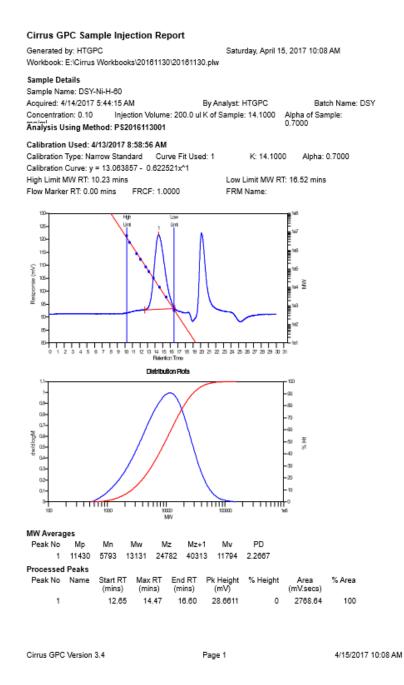


Figure S62. GPC of the polymer from table 1, entry 3.

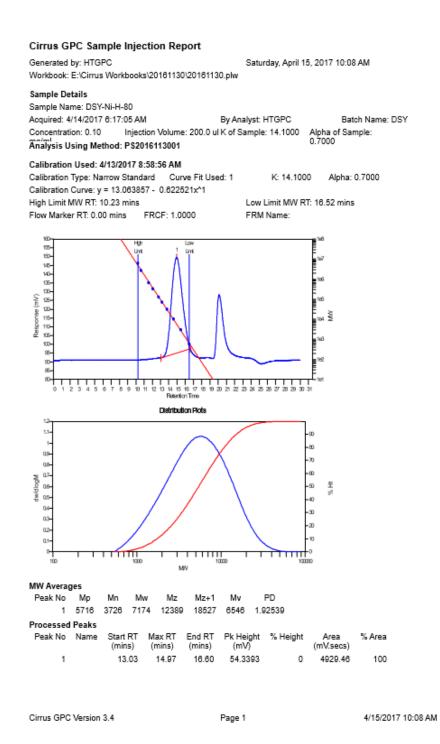
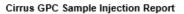


Figure S63. GPC of the polymer from table 1, entry 4.



Generated by: HTGPC Saturday, April 15, 2017 10:00 AM Workbook: E:\Cirrus Workbooks\20161130\20161130.plw

# Sample Details

Sample Name: DSY-Ni-Me-20

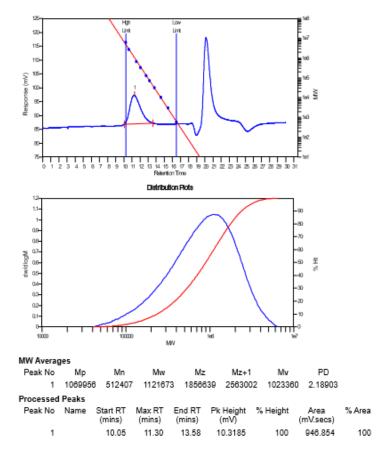
By Analyst: HTGPC Acquired: 4/13/2017 9:57:44 PM Batch Name: DSY Alpha of Sample: 0.7000 Concentration: 0.10 Injection Volume: 200.0 ul K of Sample: 14.1000 Analysis Using Method: PS2016113001

#### Calibration Used: 4/13/2017 8:58:56 AM

Calibration Type: Narrow Standard Curve Fit Used: 1 K: 14.1000 Alpha: 0.7000 Calibration Curve: y = 13.063857 - 0.622521x^1 High Limit MW RT: 10.23 mins Low Limit MW RT: 16.52 mins

Flow Marker RT: 0.00 mins FRCF: 1.0000

FRM Name:



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Figure S64. GPC of the polymer from table 1, entry 5.

4/15/2017 10:00 AM

Generated by: HTGPC Saturday, April 15, 2017 10:00 AM Workbook: E:\Cirrus Workbooks\20161130\20161130.plw

#### Sample Details

Sample Name: DSY-Ni-Me-40 By Analyst: HTGPC Acquired: 4/13/2017 10:30:35 PM Batch Name: DSY Injection Volume: 200.0 ul K of Sample: 14.1000 Alpha of Sample: 0.7000 Concentration: 0.10 Analysis Using Method: PS2016113001

## Calibration Used: 4/13/2017 8:58:56 AM

Calibration Type: Narrow Standard Curve Fit Used: 1 K: 14.1000 Alpha: 0.7000 Calibration Curve: y = 13.063857 - 0.622521x^1 High Limit MW RT: 10.23 mins Low Limit MW RT: 16.52 mins

Flow Marker RT: 0.00 mins FRCF: 1.0000

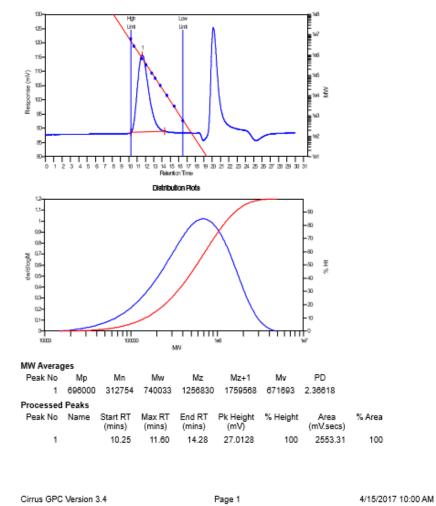


Figure S65. GPC of the polymer from table 1, entry 6.

Generated by: HTGPC Saturday, April 15, 2017 10:01 AM Workbook: E:/Cirrus Workbooks/20161130/20161130.plw

# Sample Details

Sample Name: DSY-Ni-Me-60 Acquired: 4/13/2017 11:05:40 PM By Analyst: HTGPC Batch Name: DSY Alpha of Sample: 0.7000 Concentration: 0.10 Injection Volume: 200.0 ul K of Sample: 14.1000 Analysis Using Method: PS2016113001

## Calibration Used: 4/13/2017 8:58:56 AM

Calibration Type: Narrow Standard Curve Fit Used: 1 K: 14.1000 Alpha: 0.7000 Calibration Curve: y = 13.063857 - 0.622521x^1 High Limit MW RT: 10.23 mins Low Limit MW RT: 16.52 mins

Flow Marker RT: 0.00 mins FRCF: 1.0000

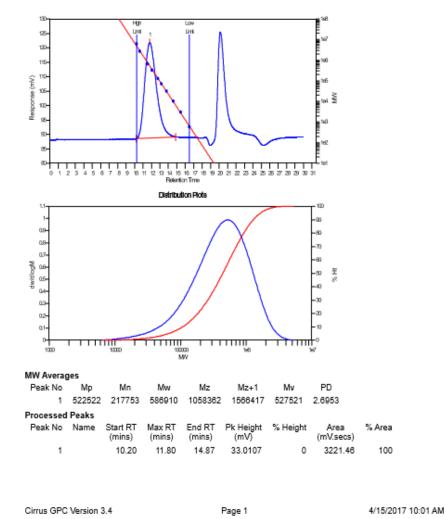


Figure S66. GPC of the polymer from table 1, entry 7.

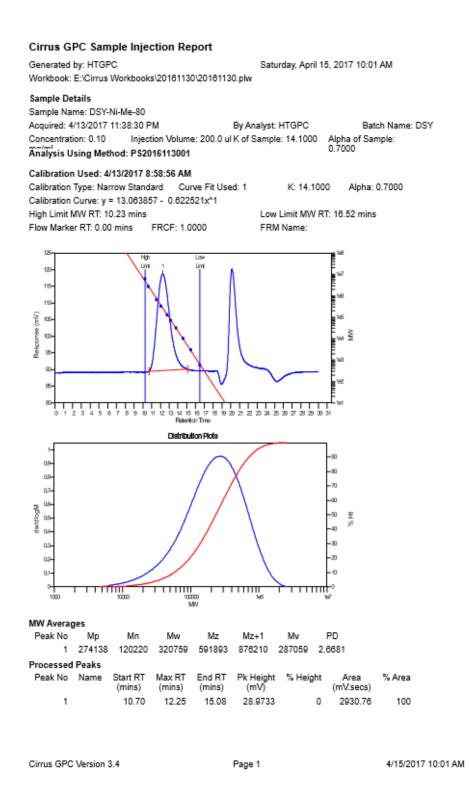


Figure S67. GPC of the polymer from table 1, entry 8.

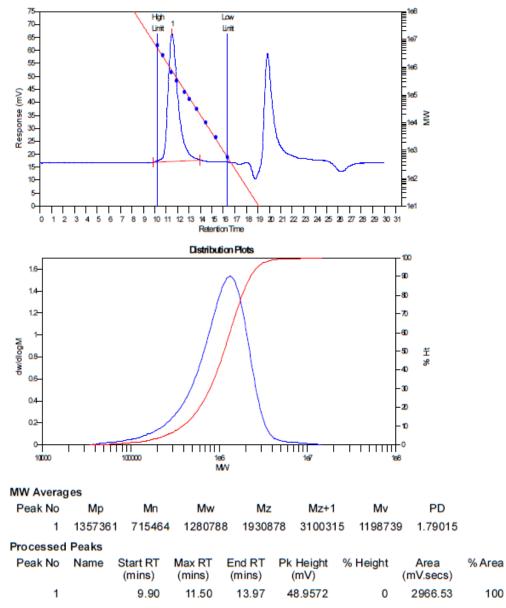


Figure S68. GPC of the polymer from table 1, entry 9.

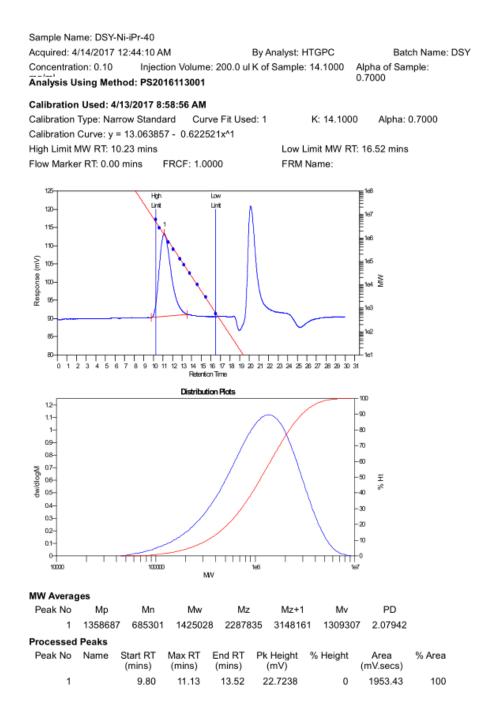


Figure S69. GPC of the polymer from table 1, entry 10.

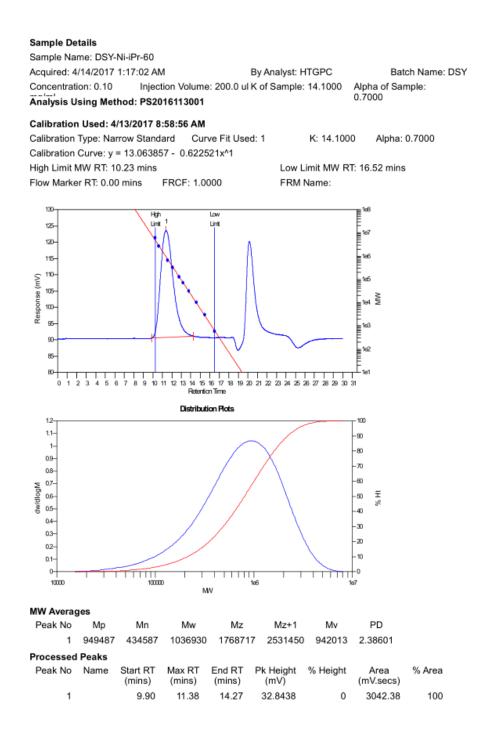


Figure S70. GPC of the polymer from table 1, entry 11.

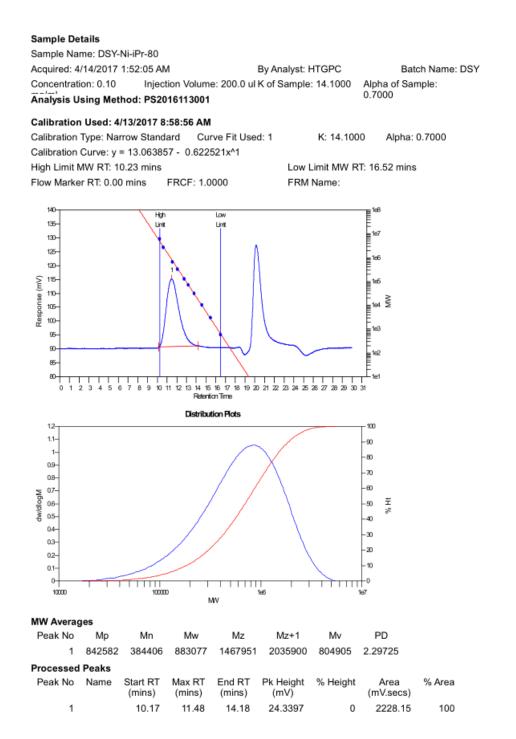


Figure S71. GPC of the polymer from table 1, entry 12.

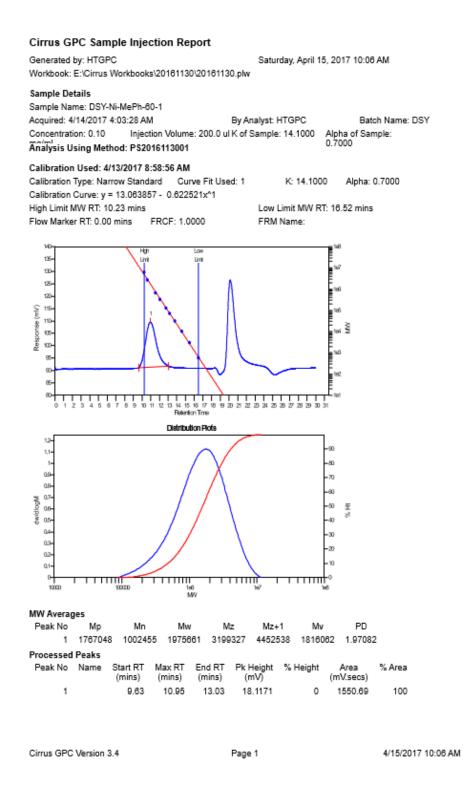


Figure S72. GPC of the polymer from table 1, entry 13.

Generated by: HTGPC Saturday, April 15, 2017 10:05 AM Workbook: E:\Cirrus Workbooks\20161130\20161130.phv

## Sample Details

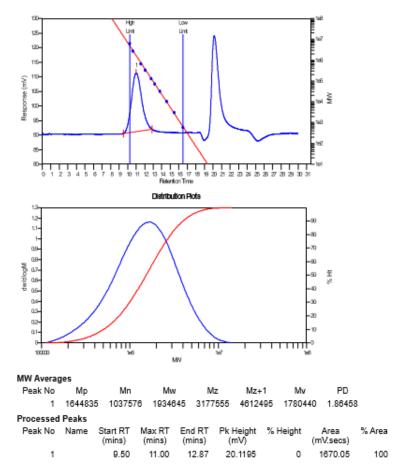
Sample Name: DSY-Ni-MePh-40-1 Acquired: 4/14/2017 2:57:48 AM By Analyst: HTGPC Batch Name: DSY Concentration: 0.10 Injection Volume: 200.0 ul K of Sample: 14.1000 Alpha of Sample: 0.7000 0.7000 Analysis Using Method: PS2016113001

## Calibration Used: 4/13/2017 8:58:56 AM

Calibration Type: Narrow Standard Curve Fit Used: 1 K: 14.1000 Alpha: 0.7000 Calibration Curve: y = 13.063857 - 0.622521x^1 High Limit MW RT: 10.23 mins Low Limit MW RT: 16.52 mins

Flow Marker RT: 0.00 mins FRCF: 1.0000

FRM Name:



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Figure S73. GPC of the polymer from table 1, entry 14.

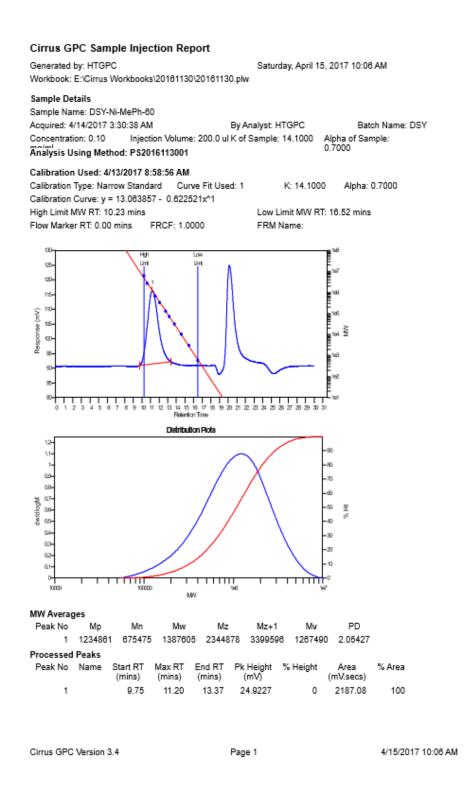


Figure S74. GPC of the polymer from table 1, entry 15.

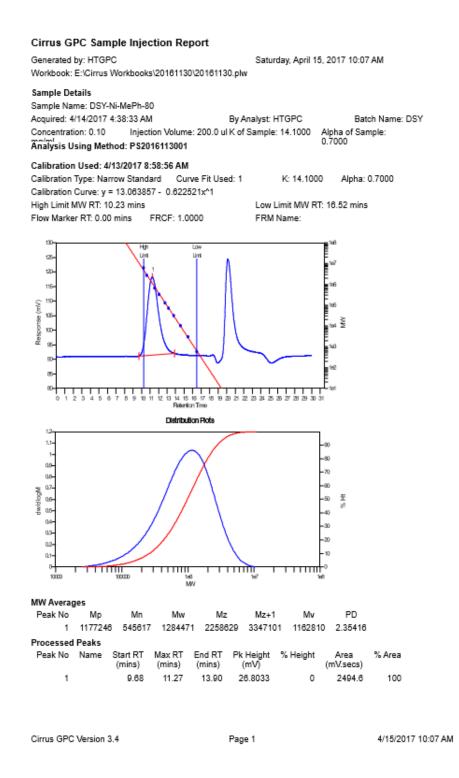


Figure S75. GPC of the polymer from table 1, entry 16.

Generated by: HTGPC Saturday, April 15, 2017 9:35 AM Workbook: E:\Cirrus Workbooks\20161130\20161130.plw

#### Sample Details

Sample Name: DSY-Ni-PhPh-20 Acquired: 4/13/2017 8:19:13 PM By Analyst: HTGPC Batch Name: DSY Concentration: 0.10 Injection Volume: 200.0 ul K of Sample: 14.1000 Alpha of Sample: 0.7000 Analysis Using Method: PS2016113001

## Calibration Used: 4/13/2017 8:58:56 AM

Calibration Type: Narrow Standard Curve Fit Used: 1 K: 14.1000 Alpha: 0.7000 Calibration Curve: y = 13.063857 - 0.622521x^1 High Limit MW RT: 10.23 mins Low Limit MW RT: 16.52 mins

Flow Marker RT: 0.00 mins FRCF: 1.0000

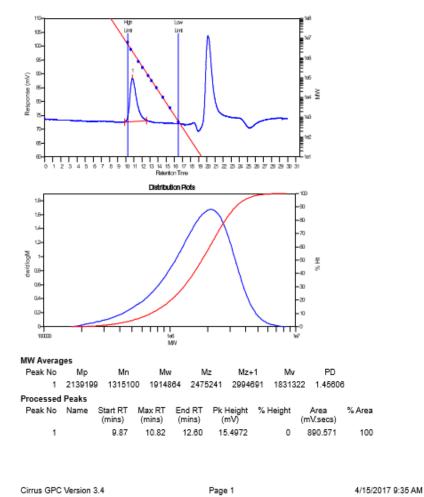


Figure S76. GPC of the polymer from table 1, entry 17.

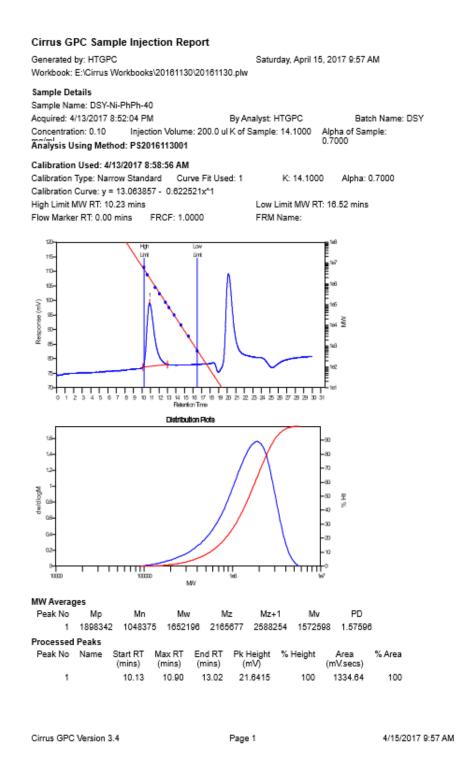


Figure S77. GPC of the polymer from table 1, entry 18.

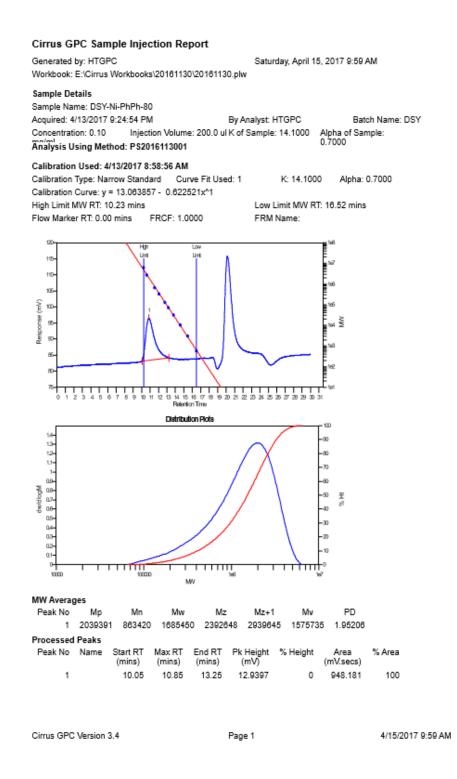


Figure S78. GPC of the polymer from table 1, entry 20.

Generated by: HTGPC Saturday, April 15, 2017 10:09 AM Workbook: E:\Cirrus Workbooks\20161130\20161130.plw

#### Sample Details

Sample Name: DSY-Ni-PhPh-80-5min

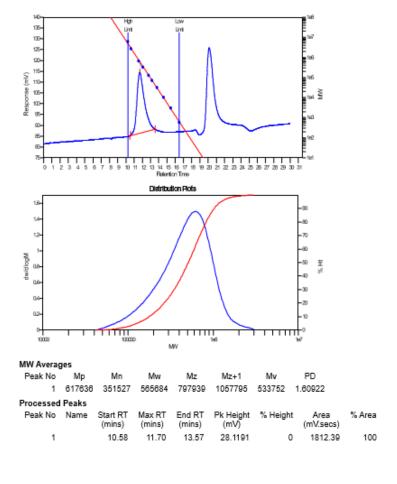
Acquired: 4/14/2017 3:38:36 PM By Analyst: HTGPC Batch Name: DSY Concentration: 0.10 Injection Volume: 200.0 ul K of Sample: 14.1000 Alpha of Sample: 0.7000 Analysis Using Method: PS2016113001

## Calibration Used: 4/13/2017 8:58:56 AM

Calibration Type: Narrow Standard Curve Fit Used: 1 K: 14.1000 Alpha: 0.7000 Calibration Curve: y = 13.063857 - 0.622521x^1 High Limit MW RT: 10.23 mins Low Limit MW RT: 16.52 mins

Flow Marker RT: 0.00 mins FRCF: 1.0000

FRM Name:



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Figure S79. GPC of the polymer from table 2, entry 1.

High Limit MW RT: 10.23 mins

Flow Marker RT: 0.00 mins FRCF: 1.0000

Generated by: HTGPC Workbook: E:/Cirrus Workbooks\20181130\20181130.plw	Saturday, April 15, 2017 10:10 AM
Sample Details Sample Name: DSY-Ni-PhPh-80-10min	
	nalyst: HTGPC Batch Name: DSY Sample: 14.1000 Alpha of Sample: 0.7000
Calibration Used: 4/13/2017 8:58:56 AM Calibration Type: Narrow Standard Curve Fit Used: 1 Calibration Curve: y = 13.063857 - 0.622521x^1	K: 14.1000 Alpha: 0.7000

Low Limit MW RT: 16.52 mins

FRM Name:

包 125 lint t20-115 110 (m/) 981 105-100ŝ 95-50-85-80-1234567891011 12 Distribution Plots Mgolb ž ū 02 10000 1000 Ū, MV. MW Averages Mp PD Peak No Mn Mw Mz Mz+1 Mv 1 1122320 539756 923709 1253551 1509242 871924 1.71135

Processed	Peaks								
Peak No	Name	Start RT (mins)	Max RT (mins)	End RT (mins)	Pk Height (mV)	% Height	Area (mV.secs)	% Area	
1		10.47	11.27	13.30	13.0202	0	841.354	100	

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Figure S80. GPC of the polymer from table 2, entry 2.

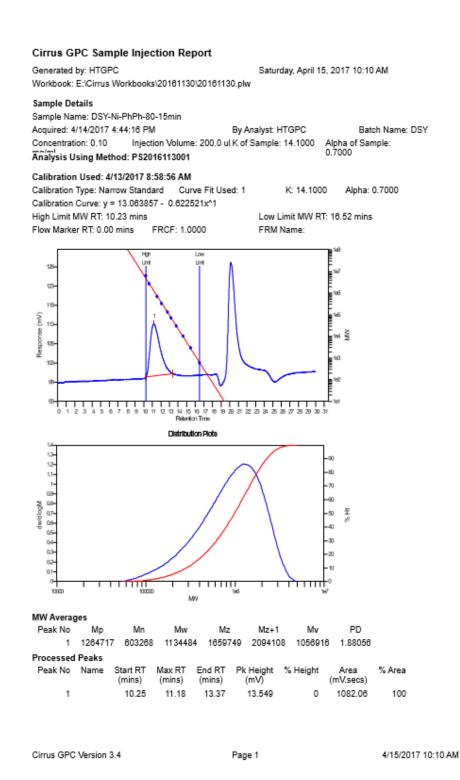


Figure S81. GPC of the polymer from table 2, entry 3.

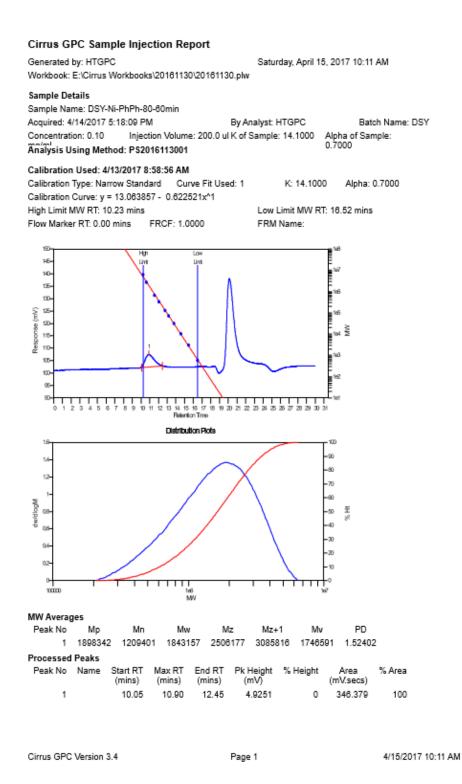


Figure S82. GPC of the polymer from table 2, entry 5.

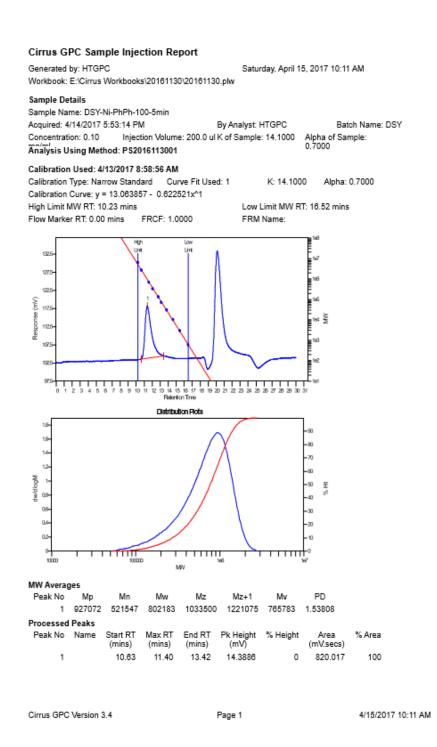


Figure S83. GPC of the polymer from table 2, entry 6.

Generated by: HTGPC Saturday, April 15, 2017 10:12 AM Workbook: E:\Cirrus Workbooks\20161130\20161130.plw

Sample Details

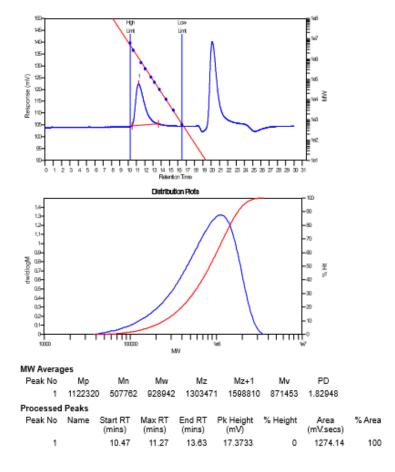
Sample Name: DSY-Ni-PhPh-100-10min Acquired: 4/14/2017 6:26:04 PM By Analyst: HTGPC Batch Name: DSY Alpha of Sample: 0.7000 Concentration: 0.10 Injection Volume: 200.0 ul K of Sample: 14.1000 Analysis Using Method: PS2016113001

## Calibration Used: 4/13/2017 8:58:56 AM

Calibration Type: Narrow Standard Curve Fit Used: 1 K: 14.1000 Alpha: 0.7000 Calibration Curve: y = 13.063857 - 0.622521x^1 High Limit MW RT: 10.23 mins Low Limit MW RT: 16.52 mins

Flow Marker RT: 0.00 mins FRCF: 1.0000

FRM Name:



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Figure S84. GPC of the polymer from table 2, entry 7.

Generated by: HTGPC Saturday, April 15, 2017 10:12 AM Workbook: E:/Cirrus Workbooks\20161130\20161130.plw

## Sample Details

Sample Name: DSY-Ni-PhPh-100-15min Acquired: 4/14/2017 6:58:54 PM By Analyst: HTGPC Batch Name: DSY Concentration: 0.10 Injection Volume: 200.0 ul K of Sample: 14.1000 Alpha of Sample: Concentration: 0.10 Injection Volume: 200.0 ul K of Sample: 14.1000 0.7000

# Calibration Used: 4/13/2017 8:58:56 AM

 Calibration Type: Narrow Standard
 Curve Fit Used: 1
 K: 14.1000
 Alpha: 0.7000

 Calibration Curve: y = 13.063857 - 0.622521x^1
 High Limit MW RT: 10.23 mins
 Low Limit MW RT: 16.52 mins

 Flow Marker RT: 0.00 mins
 FRCF: 1.0000
 FRM Name:

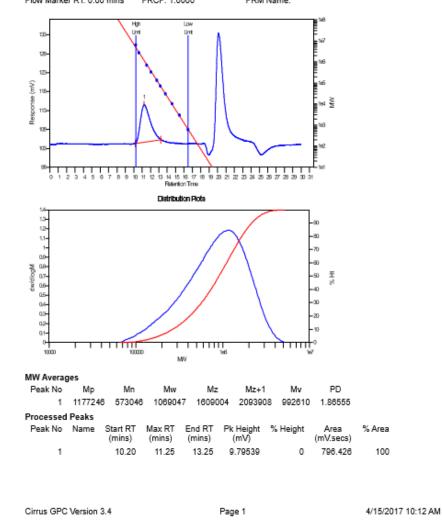


Figure S85. GPC of the polymer from table 2, entry 8.



Generated by: HTGPC Saturday, April 15, 2017 10:13 AM Workbook: E:/Cirrus Workbooks/20161130/20161130.phw

# Sample Details

Sample Name: DSY-Ni-PhPh-100-80min Acquired: 4/14/2017 7:31:46 PM By Analyst: HTGPC Batch Name: DSY Concentration: 0.10 Injection Volume: 200.0 ul K of Sample: 14.1000 Alpha of Sample: Concentration: 0.10 Sample: 14.1000 On Sample: 14.1000 On Sample: 0.7000

#### Calibration Used: 4/13/2017 8:58:56 AM

 Calibration Type: Narrow Standard
 Curve Fit Used: 1
 K: 14.1000
 Alpha: 0.7000

 Calibration Curve: y = 13.063857 - 0.622521x^1
 High Limit MW RT: 10.23 mins
 Low Limit MW RT: 16.52 mins

 Flow Marker RT: 0.00 mins
 FRCF: 1.0000
 FRM Name:

н tφ 127 172 ì 117 Response 112 102 gr 9 20 21 22 23 24 25 25 27 28 29 30 14 15 16 17 18 Baterifico Tone 1 2 3 4 5 6 7 8 9 10 11 12 Distribution Plots an 0.9 0.9 0.7 m 0.6 dwidlogM 0.5 4 -40 Q, 63 30 02 m 0,1 • 10000 ттŋ 1000 W MW Averages Peak No Mp Mn Mz PD Mw Mz+1 Μv 1 1149455 302847 1063686 1892053 2633659 949438 3.51229

Processed	Peaks								
Peak No	Name		Max RT (mins)		Pk Height (mV)	% Height	Area (mV.secs)	% Area	
1		9.90	11.25	14.57	15.6703	0	1586.46	100	

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Figure S86. GPC of the polymer from table 2, entry 10.

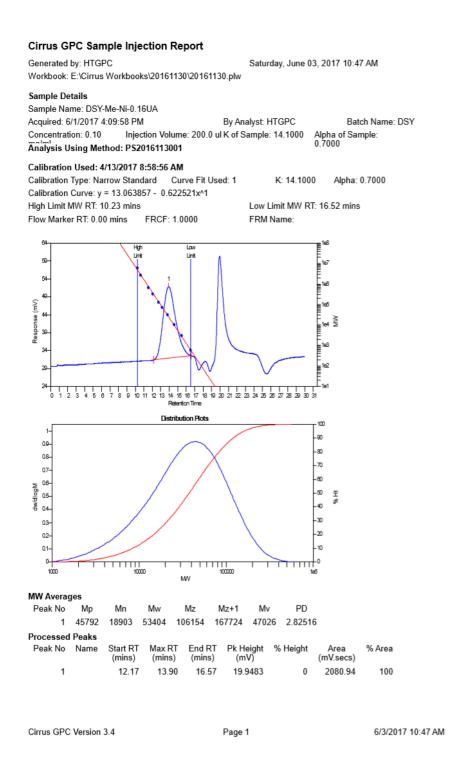


Figure S87. GPC of the polymer from table 3, entry 11.

Generated by: HTGPC Saturday, June 03, 2017 10:47 AM Workbook: E:\Cirrus Workbooks\20161130\20161130.plw

#### Sample Details

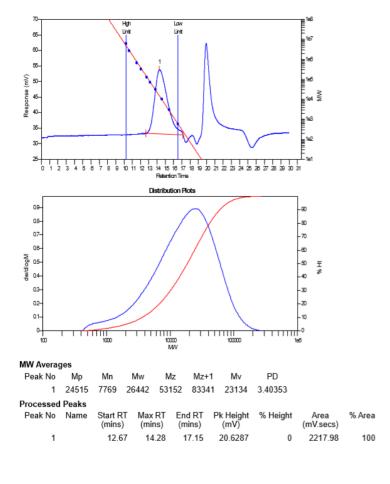
Sample Name: DSY-Me-Ni-0.32UA Acquired: 6/1/2017 4:42:49 PM By Analyst: HTGPC Batch Name: DSY Concentration: 0.10 Injection Volume: 200.0 ul K of Sample: 14.1000 Alpha of Sample: 0.7000 Analysis Using Method: PS2016113001

## Calibration Used: 4/13/2017 8:58:56 AM

Calibration Type: Narrow Standard Curve Fit Used: 1 K: 14.1000 Alpha: 0.7000 Calibration Curve: y = 13.063857 - 0.622521x^1 High Limit MW RT: 10.23 mins Low Limit MW RT: 16.52 mins

Flow Marker RT: 0.00 mins FRCF: 1.0000

FRM Name:



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Figure S88. GPC of the polymer from table 3, entry 14.

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Generated by: HTGPC Saturday, June 03, 2017 10:48 AM Workbook: E:\Cirrus Workbooks\20161130\20161130.plw

#### Sample Details

Sample Name: DSY-PhMe-Ni-0.32UA Acquired: 6/1/2017 5:50:44 PM By Analyst: HTGPC Batch Name: DSY Alpha of Sample: 0.7000 Concentration: 0.10 Injection Volume: 200.0 ul K of Sample: 14.1000 Analysis Using Method: PS2016113001

#### Calibration Used: 4/13/2017 8:58:56 AM

Alpha: 0.7000 Calibration Type: Narrow Standard Curve Fit Used: 1 K: 14.1000 Calibration Curve: y = 13.063857 - 0.622521x^1 High Limit MW RT: 10.23 mins Low Limit MW RT: 16.52 mins



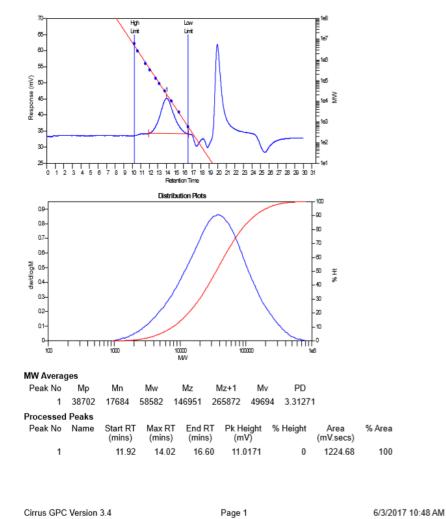


Figure S89. GPC of the polymer from table 3, entry 16.

# 4. References

1, Guo, L. H.; Kong, W. Y.; Xu, Y. J.; Yang, Y. L.; Ma, R.; Cong, L.; Dai, S. Y.; Liu, Z. Large-scale synthesis of novel sterically hindered acenaphthene-based  $\alpha$ -diimine ligands and their application in coordination chemistry. *J. Organomet. Chem.* **2018**, *859*, 58-67.

2, Cherian, A. E.; Domski, G. J.; Rose, J. M.; Lobkovsky, E. B.; Coates, G. W. Acid-catalyzed orthoalkylation of anilines with styrenes: An improved route to chiral anilines with bulky substituents. *Org. Lett.* **2005**, *7*, 5135-5137.

3, Guo, L. H.; Lian, K.; Kong, W.; Xu, S.; Jiang, G.; Dai, S. Y. Synthesis of Various Branched Ultra-High-Molecular-Weight Polyethylenes Using Sterically Hindered Acenaphthene-Based α-Diimine Ni (II) Catalysts. *Organometallics* **2018**, *37*, 2442–2449.

# 5. X-ray Crystallography

CCDC numbers of **1**, **2** and **3** are **1886685**, **1886683** and **1886684** respectively. The data can be obtained free of charge from the Cambridge Crystallographic Data Centre via <u>www.ccdc.cam.ac.uk/data request/cif</u>.

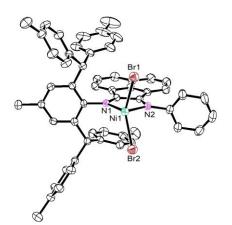


Table S3 Crystal data and structure refinement for 1.		
Identification code	1	
Empirical formula	C57 H52 Br2 Cl4 N2 Ni O	
Formula weight	1141.34	
Temperature/K	298(2) K	
Crystal system	Triclinic	
Space group	P-1	
a/Å	12.5431(11)	
b/Å	14.4530(12)	
c/Å	15.8001(13)	
α/°	83.121(2)	
β/°	76.6860(10)	
γ/°	79.3150(10)	
Volume/Å <sup>3</sup>	2730.1(4)	
Z	2	

$\rho_{calc}g/cm^3$	1.388
$\mu/\text{mm}^{-1}$	2.054
F(000)	1164
Crystal size/mm <sup>3</sup>	0.40 x 0.20 x 0.13
Radiation	MoKa ( $\lambda = 0.71073$ )
20 range for data collection/°	2.36 to 25.02
Index ranges	-14<=h<=14, -17<=k<=13, -18<=l<=18
Reflections collected	13979
Independent reflections	9472 [R(int) = 0.0344]
Data/restraints/parameters	9472 / 0 / 628
Goodness-of-fit on F <sup>2</sup>	1.063
Final R indexes [I>= $2\sigma$ (I)]	R1 = 0.0467, wR2 = 0.0801
Final R indexes [all data]	R1 = 0.1112, wR2 = 0.0866
Largest diff. peak/hole / e Å <sup>-3</sup>	0.573 and -0.619
Radiation 2Θ range for data collection/° Index ranges Reflections collected Independent reflections Data/restraints/parameters Goodness-of-fit on F <sup>2</sup> Final R indexes [I>=2σ (I)] Final R indexes [all data]	MoKa ( $\lambda = 0.71073$ ) 2.36 to 25.02 $-14 \le h \le 14$ , $-17 \le k \le 13$ , $-18 \le l \le 13979$ 9472 [R(int) = 0.0344] 9472 / 0 / 628 1.063 R1 = 0.0467, wR2 = 0.0801 R1 = 0.1112, wR2 = 0.0866

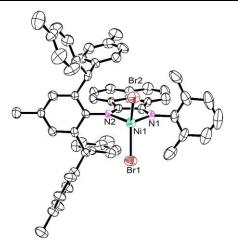


Table S4 Crystal data and structure refinement for 2.		
Identification code	2	
Empirical formula	C57 H50 Br2 N2 Ni	
Formula weight	981.52	
Temperature/K	298(2) K	
Crystal system	Monoclinic	
Space group	P2	
a/Å	10.813(5)	
b/Å	20.216(5)	
c/Å	26.480(5)	
$\alpha/^{\circ}$	90.000(5)	
β/°	100.049(5)	
γ/°	90.000(5)	
Volume/Å <sup>3</sup>	5700(3)	

Z	4
$\rho_{calc}g/cm^3$	1.144
$\mu/\text{mm}^{-1}$	1.775
F(000)	2016
Crystal size/mm <sup>3</sup>	0.30 x 0.20 x 0.20
Radiation	MoKa ( $\lambda = 0.71069$ )
20 range for data collection/°	2.393 to 25.707
Index ranges	-12<=h<=11, -23<=k<=24, -31<=l<=31
Reflections collected	28707
Independent reflections	9978[R(int) = 0.1810]
Data/restraints/parameters	9978 / 0 / 566
Goodness-of-fit on F <sup>2</sup>	0.660
Final R indexes [I>=2 $\sigma$ (I)]	R1 = 0.0760, wR2 = 0.1632
Final R indexes [all data]	R1 = 0.1932, wR2 = 0.1944
Largest diff. peak/hole / e Å <sup>-3</sup>	0.459 and -0.358

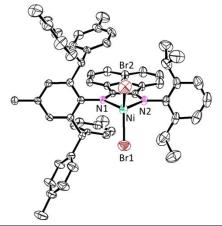


Table S5 Crystal data and structure refinement for 3.		
Identification code	3	
Empirical formula	C61 H58 Br2 N2 Ni	
Formula weight	1037.62	
Temperature/K	298(2) K	
Crystal system	Monoclinic	
Space group	P2(1)/m	
a/Å	11.7600(11)	
b/Å	20.6272(18)	
c/Å	12.0491(12)	
α/°	90.000(5)	
β/°	106.947(2)	
γ/°	90.000(5)	
Volume/Å <sup>3</sup>	2795.9(5)	

Z	2
$\rho_{calc}g/cm^3$	1.233
$\mu/\text{mm}^{-1}$	1.813
F(000)	1072
Crystal size/mm <sup>3</sup>	0.21 x 0.20 x 0.12
Radiation	MoKα ( $\lambda = 0.71073$ )
2@ range for data collection/°	2.35 to 25.02
Index ranges	-13<=h<=13, -21<=k<=24, -14<=l<=14
Reflections collected	14293
Independent reflections	5065[R(int) = 0.0629]
Data/restraints/parameters	9978 / 0 / 333
Goodness-of-fit on F <sup>2</sup>	1.025
Final R indexes [I>=2 $\sigma$ (I)]	R1 = 0.0507, wR2 = 0.0895
Final R indexes [all data]	R1 = 0.1203, wR2 = 0.0976
Largest diff. peak/hole / e Å <sup>-3</sup>	0.452 and -0.397