

## Synthesis of Cryptophane-223 Type Derivatives with dual functionalization

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S4: <sup>1</sup>H NMR (400 MHz) spectrum of compound **4** recorded in CDCl<sub>3</sub> at 298 K.

S5: <sup>13</sup>C NMR (100.6 MHz) spectrum of compound **4** recorded in CDCl<sub>3</sub> at 298 K.

S6: <sup>1</sup>H NMR (400 MHz) spectrum of CTB **5** recorded in CDCl<sub>3</sub> at 298 K.

S7: <sup>13</sup>C NMR (100.6 MHz) spectrum of CTB **5** recorded in CDCl<sub>3</sub> at 298 K.

S8: <sup>1</sup>H NMR (400 MHz) spectrum of CTB **7** recorded in CDCl<sub>3</sub> at 298 K.

S9: <sup>13</sup>C NMR (100.6 MHz) spectrum of CTB **7** recorded in CDCl<sub>3</sub> at 298 K.

S10: <sup>1</sup>H NMR (400 MHz) spectrum of compound **8** recorded in CDCl<sub>3</sub> at 298 K.

S11: <sup>13</sup>C NMR (100.6 MHz) spectrum of compound **8** recorded in CDCl<sub>3</sub> at 298 K.

S12: <sup>1</sup>H NMR (400 MHz) spectrum of compound **9** recorded in CDCl<sub>3</sub> at 298 K.

S13: <sup>13</sup>C NMR (100.6 MHz) spectrum of compound **9** recorded in CDCl<sub>3</sub> at 298 K.

S14: <sup>1</sup>H NMR (400 MHz) spectrum of compound **10** recorded in CDCl<sub>3</sub> at 298 K.

S15: <sup>13</sup>C NMR (100.6 MHz) spectrum of compound **10** recorded in CDCl<sub>3</sub> at 298 K.

S16: <sup>1</sup>H NMR (400 MHz) spectrum of compound **11** recorded in DMSO-*d*<sub>6</sub> at 298 K.

S17: <sup>13</sup>C NMR (100.6 MHz) spectrum of compound **11** recorded in DMSO-*d*<sub>6</sub> at 298K.

S18:  $^1\text{H}$  NMR (400 MHz) spectrum of compound **14** recorded in  $\text{CDCl}_3$  at 298 K.

S19:  $^{13}\text{C}$  NMR (100.6 MHz) spectrum of compound **14** recorded in  $\text{CDCl}_3$  at 298 K.

S20: HSQC (400 MHz) spectrum of compound **14** recorded in  $\text{CDCl}_3$  at 298 K.

S21:  $^1\text{H}$  NMR (400 MHz) spectrum of compound **15** recorded in  $\text{CDCl}_3$  at 298 K.

S22:  $^{13}\text{C}$  NMR (100.6 MHz) spectrum of compound **15** recorded in  $\text{CDCl}_3$  at 298 K.

S23: HSQC (400 MHz) spectrum of compound **15** recorded in  $\text{CDCl}_3$  at 298 K.

S24:  $^1\text{H}$  NMR (400 MHz) spectrum of compound **18** recorded in  $\text{CDCl}_3$  at 298 K.

S25:  $^{13}\text{C}$  NMR (100.6 MHz) spectrum of compound **18** recorded in  $\text{CDCl}_3$  at 298 K.

S26: HSQC (400 MHz) spectrum of compound **18** recorded in  $\text{CDCl}_3$  at 298 K.

S27:  $^1\text{H}$  NMR (400 MHz) spectrum of compound **19** recorded in  $\text{CDCl}_3$  at 298 K.

S28:  $^{13}\text{C}$  NMR (100.6 MHz) spectrum of compound **19** recorded in  $\text{CDCl}_3$  at 298 K.

S29: HSQC (400 MHz) spectrum of compound **19** recorded in  $\text{CDCl}_3$  at 298 K.

S30:  $^1\text{H}$  NMR (400 MHz) spectrum of compound **21** recorded in  $\text{CDCl}_3$  at 298 K.

S31:  $^{13}\text{C}$  NMR (100.6 MHz) spectrum of compound **21** recorded in  $\text{CDCl}_3$  at 298 K.

S32: HSQC (400 MHz) spectrum of compound **21** recorded in  $\text{CDCl}_3$  at 298 K.

S33:  $^1\text{H}$  NMR (400 MHz) spectrum of compound **2** recorded in  $\text{DMSO}-d_6$  at 298 K.

S34:  $^{13}\text{C}$  NMR (100.6 MHz) spectrum of compound **2** recorded in  $\text{DMSO}-d_6$  at 298 K.

S35: HSQC (400 MHz) spectrum of compound **2** recorded in  $\text{DMSO}-d_6$  at 298 K.

S36: COSY (400 MHz) spectrum of compound **2** recorded in  $\text{DMSO}-d_6$  at 298 K.

S37:  $^1\text{H}$  NMR (400 MHz) spectrum of compound **23** recorded in  $\text{CDCl}_3$  at 298 K.

S38:  $^{13}\text{C}$  NMR (100.6 MHz) spectrum of compound **23** recorded in  $\text{CDCl}_3$  at 298 K.

S39: HSQC (400 MHz) spectrum of compound **23** recorded in  $\text{CDCl}_3$  at 298 K.

S40: ORTEP representation of compound **9** (hydrogen atoms and solvent molecule were omitted for clarity. The displacement ellipsoids were plotted at 30% probability level).

S41: Calorimetric titration of compound **2** in  $\text{H}_2\text{O}$ /TRIS (20 mM; pH = 7.6). The solution host ( $c = 0.08$  mM) was placed into the calorimeter cell (1.4 mL) and 28 successive aliquots (10  $\mu\text{L}$ ) of  $\text{Ni}^{2+}$  solution ( $c = 1.0$  mM) were added at 3 min intervals.

S42: Calorimetric titration of Cryptophane-222 hexacarboxylate in H<sub>2</sub>O/TRIS (20 mM; pH = 7.0). The solution host (c = 0.08 mM) was placed into the calorimeter cell (1.4 mL) and 28 successive aliquots (10 µL) of Ni<sup>2+</sup> solution (c = 1.0 mM) were added at 3 min intervals.

S43: Hyperpolarized <sup>129</sup>Xe spectra of compound **2** in TRIS buffer (20 mM, pH = 7.5). a) in absence of Zn<sup>2+</sup> (green spectrum). b) in presence of 1 equiv. of Zn<sup>2+</sup> (blue spectrum). c) in presence of 5 equiv. of Zn<sup>2+</sup> (red spectrum). Spectra recorded at 25°C.

S44: hyperpolarized <sup>129</sup>Xe NMR spectrum of compound **10** recorded at 298 K in C<sub>2</sub>D<sub>2</sub>Cl<sub>4</sub>.

Table S1: Crystallographic data of cryptophane **9**.

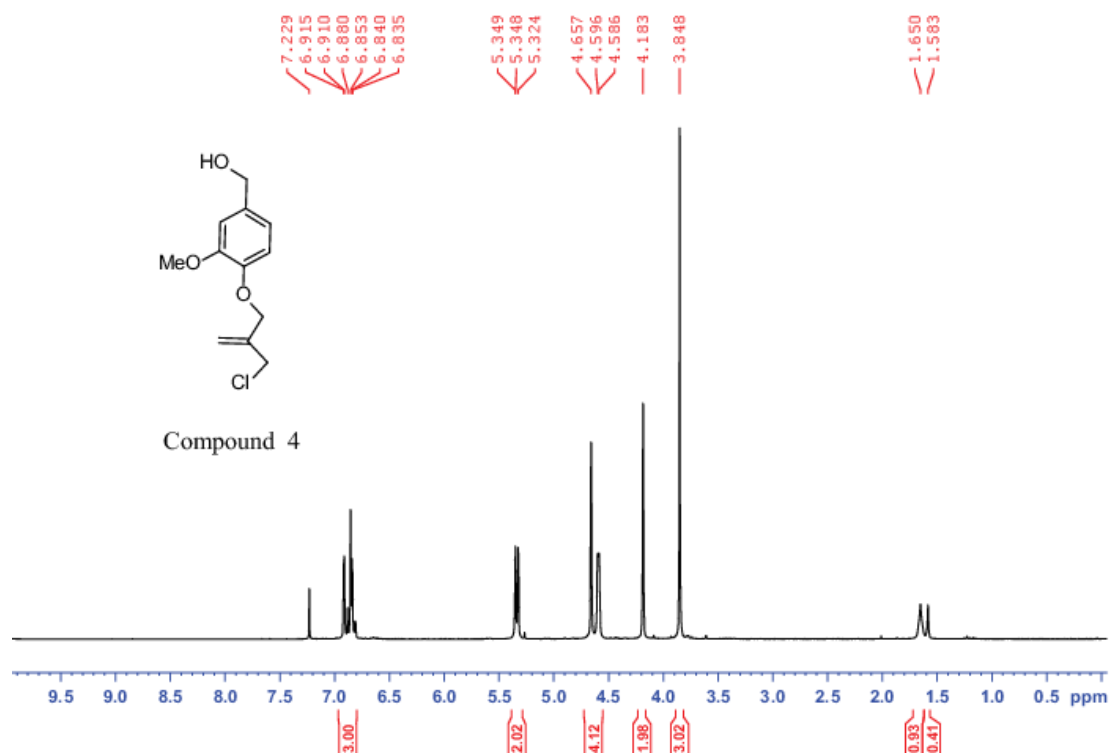


Figure S1: <sup>1</sup>H NMR (400 MHz) spectrum of compound **4** recorded in CDCl<sub>3</sub> at 298 K.

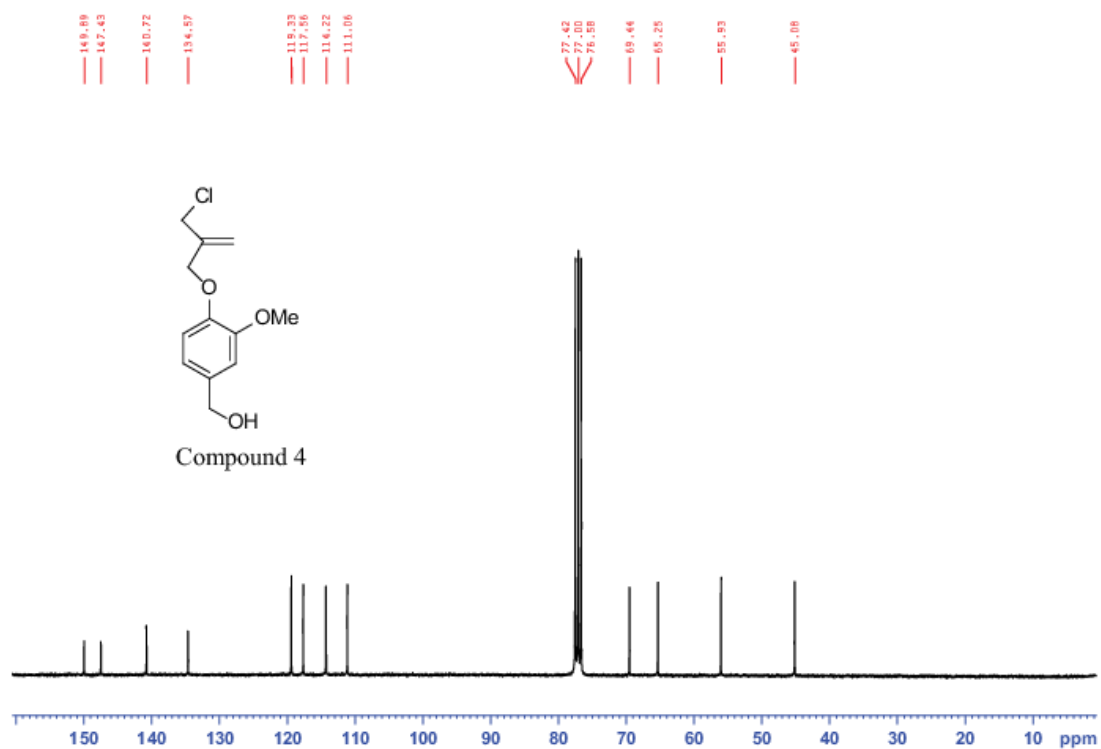


Figure S2: <sup>13</sup>C NMR (100.6 MHz) spectrum of compound **4** recorded in CDCl<sub>3</sub> at 298 K.

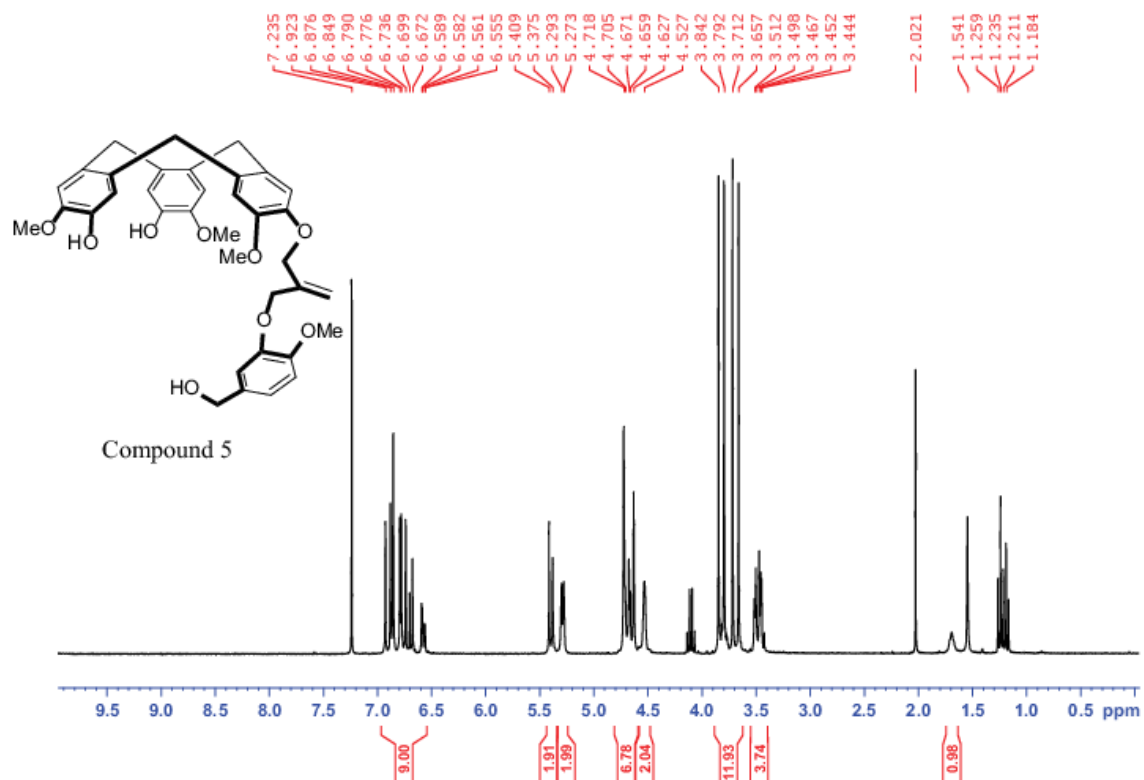


Figure S3:  $^1\text{H}$  NMR (400 MHz) spectrum of CTB **5** recorded in  $\text{CDCl}_3$  at 298 K.

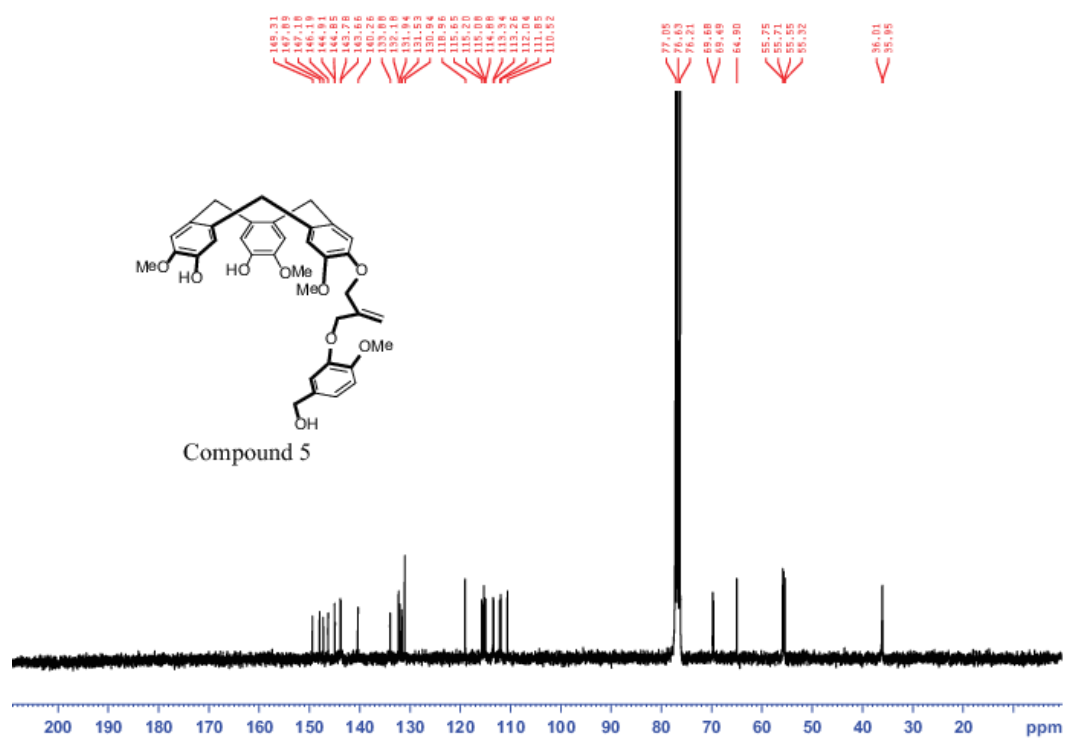


Figure S4:  $^{13}\text{C}$  NMR (100.6 MHz) spectrum of CTB **5** recorded in  $\text{CDCl}_3$  at 298 K.

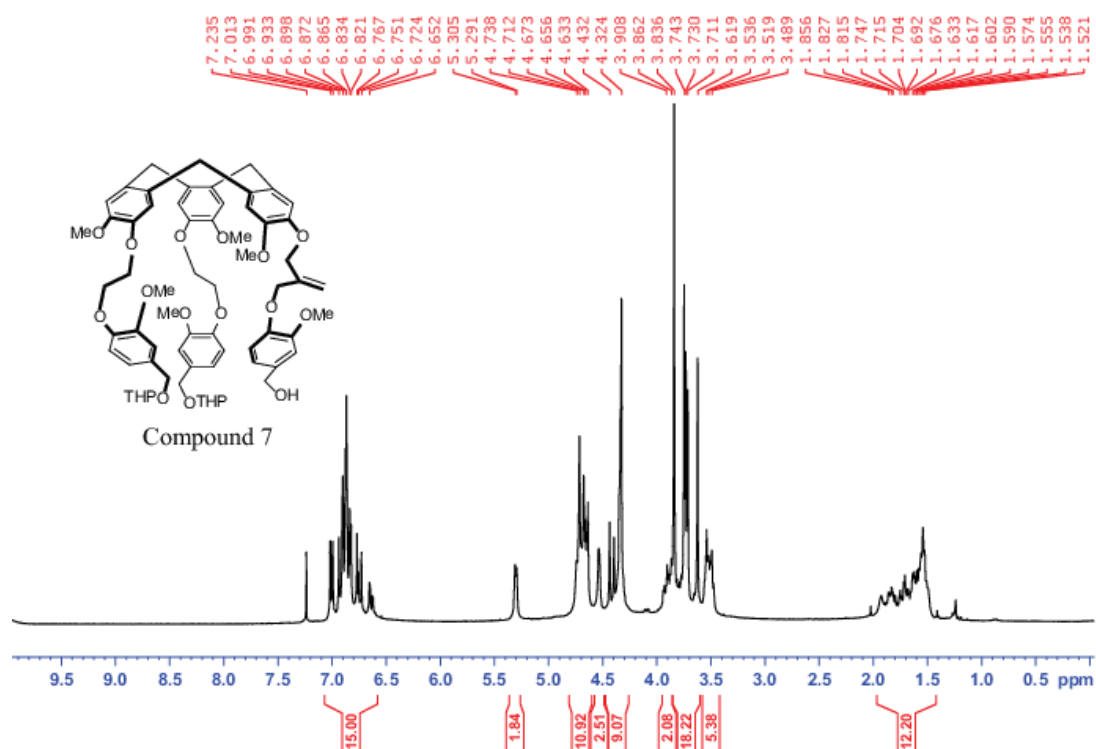


Figure S5:  $^1\text{H}$  NMR (400 MHz) spectrum of CTB **7** recorded in  $\text{CDCl}_3$  at 298 K.



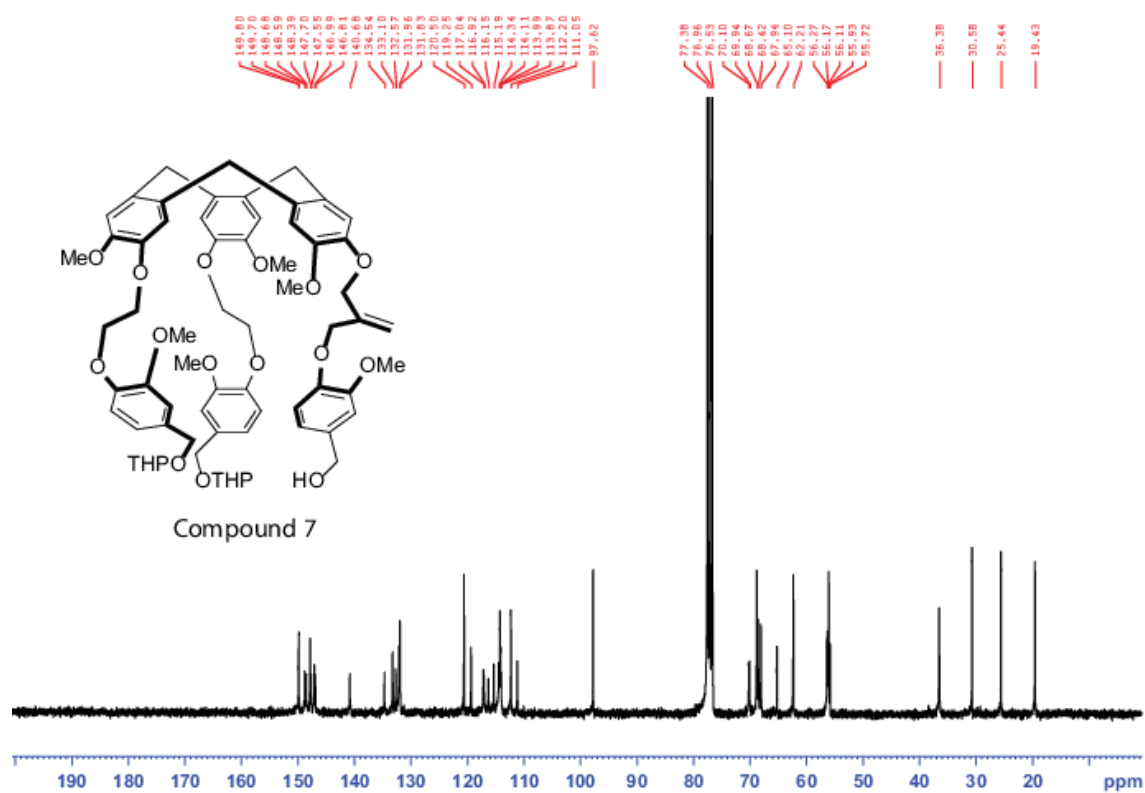


Figure S6:  $^{13}\text{C}$  NMR (100.6 MHz) spectrum of CTB **7** recorded in  $\text{CDCl}_3$  at 298 K.

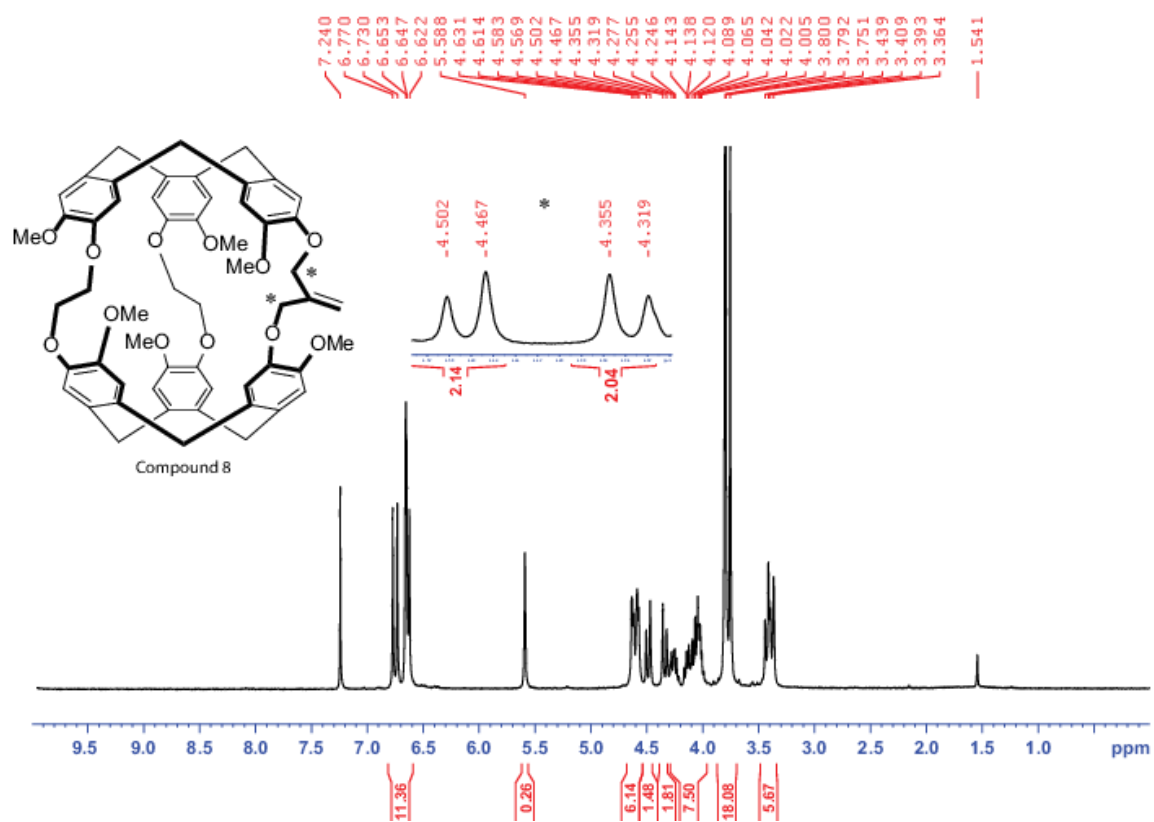


Figure S7:  $^1\text{H}$  NMR (400 MHz) spectrum of compound **8** recorded in  $\text{CDCl}_3$  at 298 K.

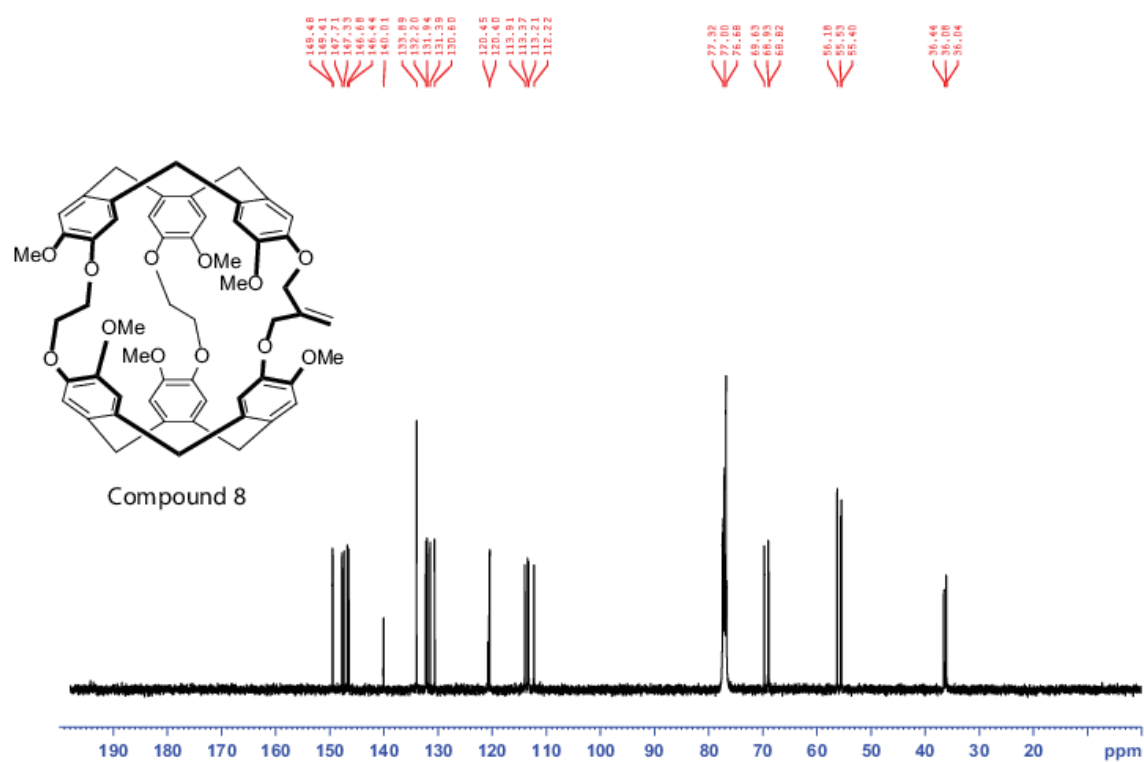


Figure S8:  $^{13}\text{C}$  NMR (100.6 MHz) spectrum of compound **8** recorded in  $\text{CDCl}_3$  at 298 K.

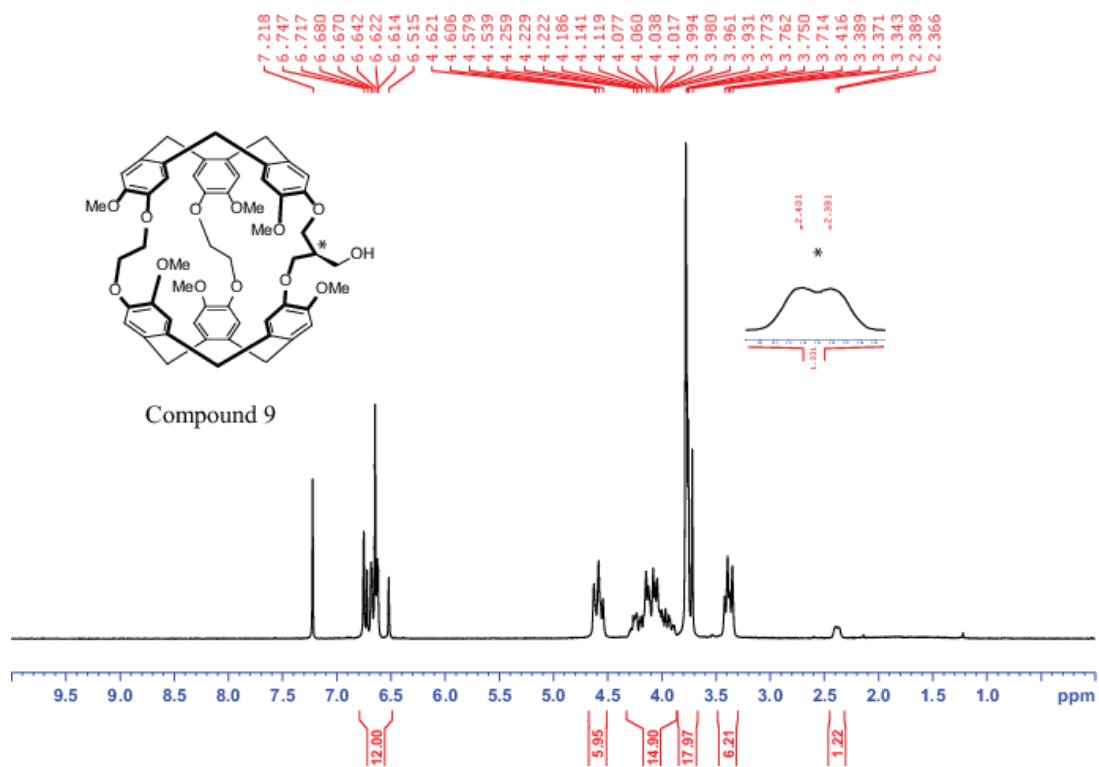


Figure S9:  $^1\text{H}$  NMR (400 MHz) spectrum of compound **9** recorded in  $\text{CDCl}_3$  at 298 K.

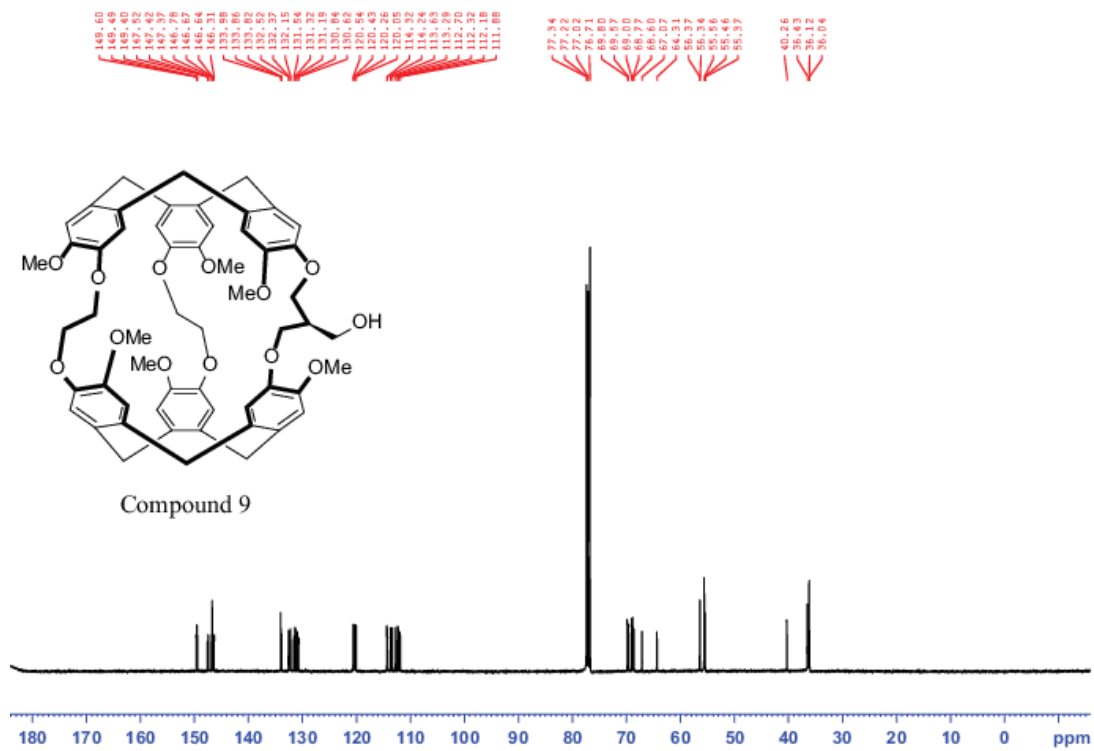


Figure S10:  $^{13}\text{C}$  NMR (100.6 MHz) spectrum of compound **9** recorded in  $\text{CDCl}_3$  at 298 K.

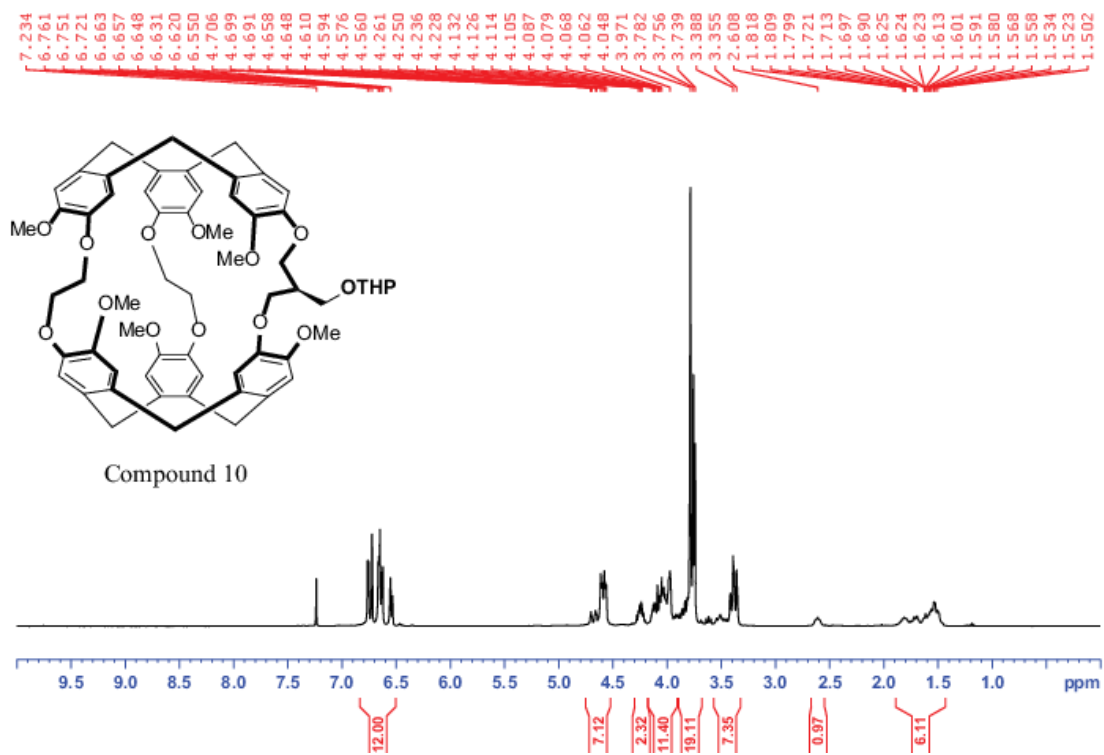


Figure S11: <sup>1</sup>H NMR (400 MHz) spectrum of compound **10** recorded in CDCl<sub>3</sub> at 298 K.

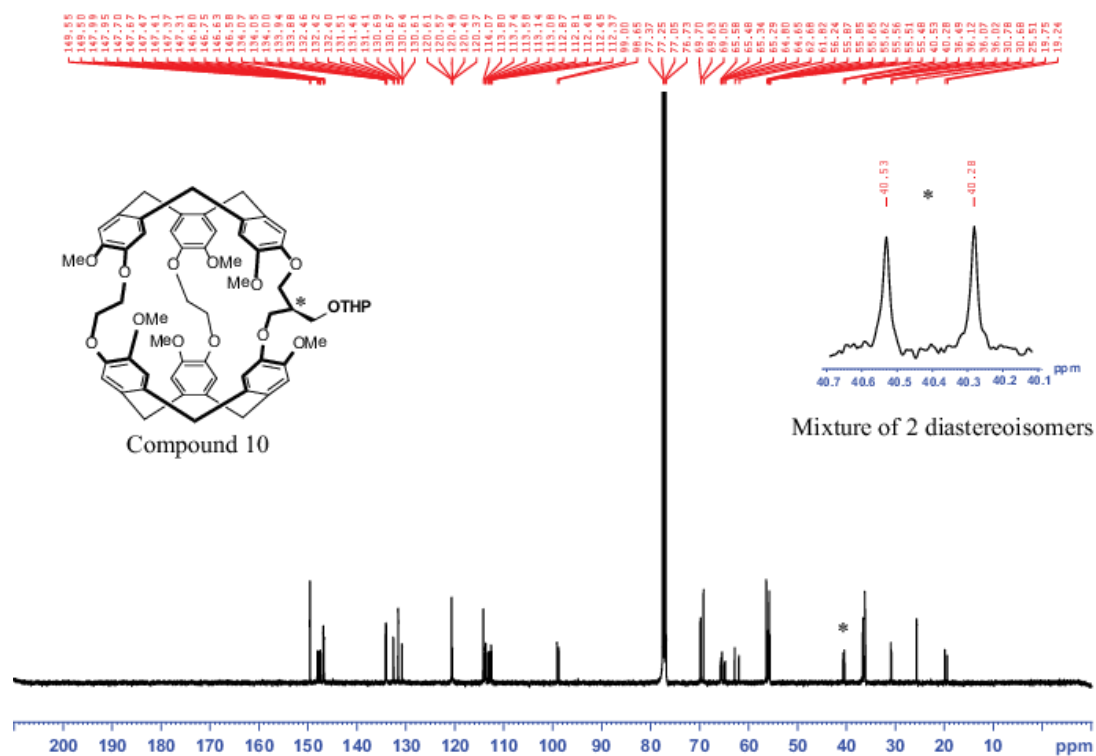


Figure S12:  $^{13}\text{C}$  NMR (100.6 MHz) spectrum of compound **10** recorded in  $\text{CDCl}_3$  at 298 K.

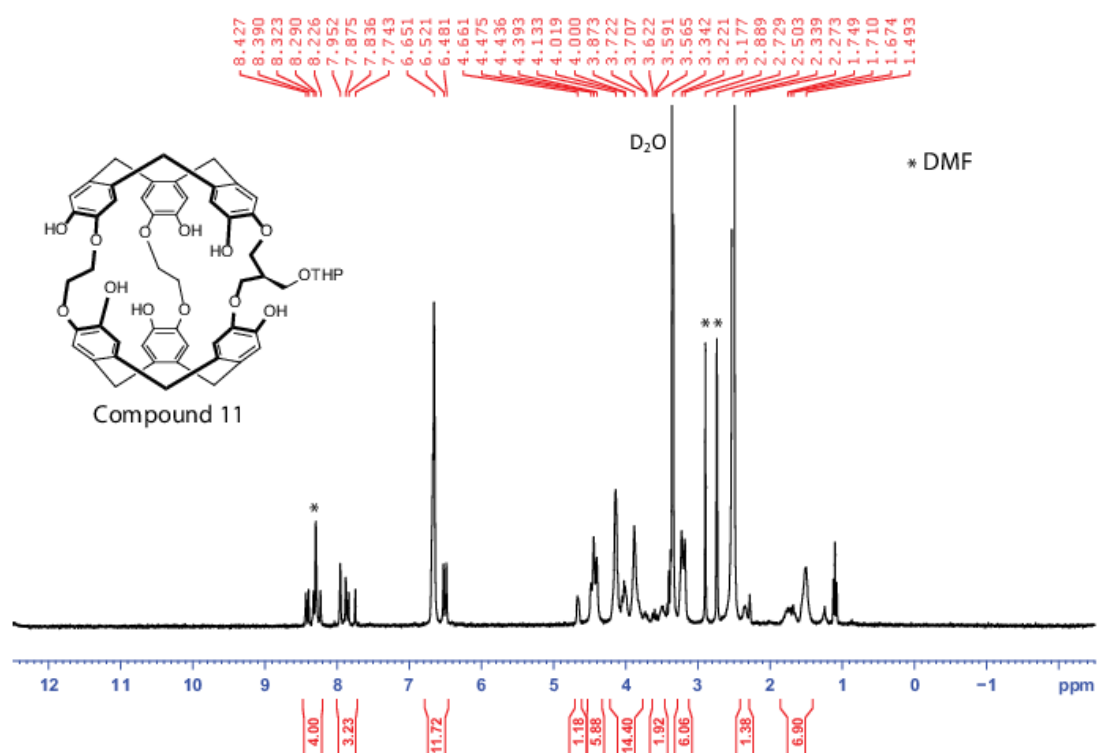


Figure S13:  $^1\text{H}$  NMR (400 MHz) spectrum of compound **11** recorded in  $\text{DMSO}-d_6$  at 298 K.



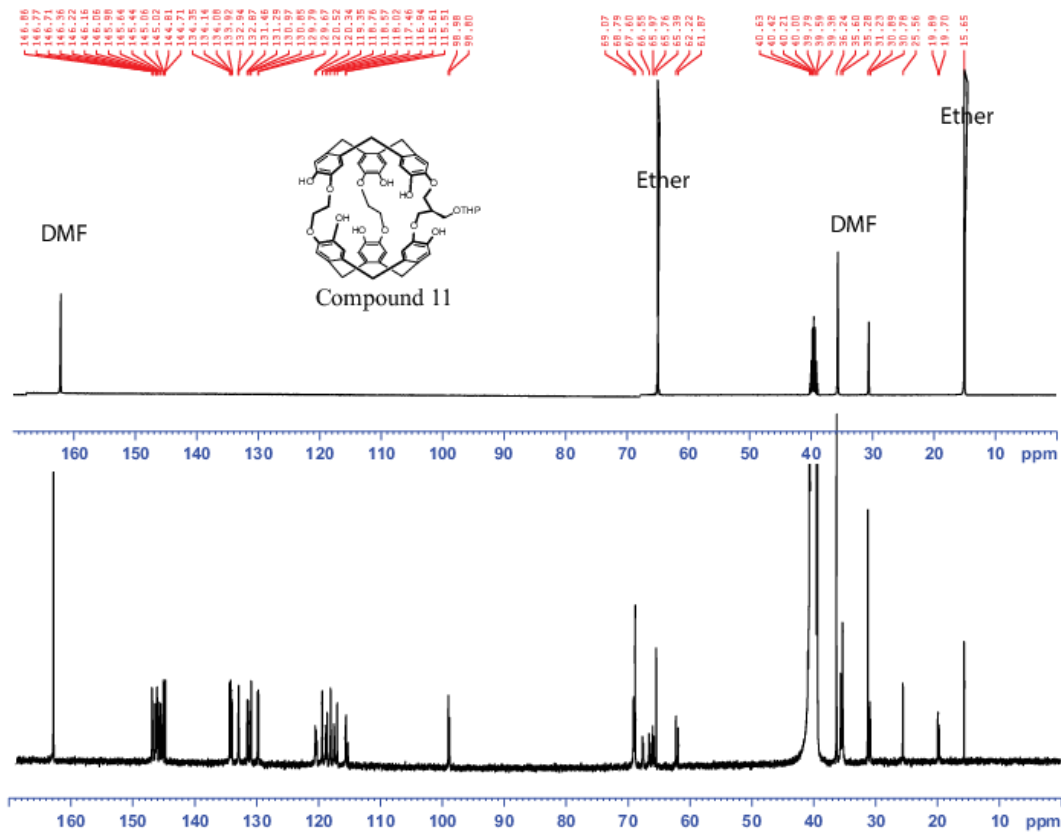


Figure S14:  $^{13}\text{C}$  NMR (100.6 MHz) spectrum of compound **11** recorded in DMSO- $d_6$  at 298K.

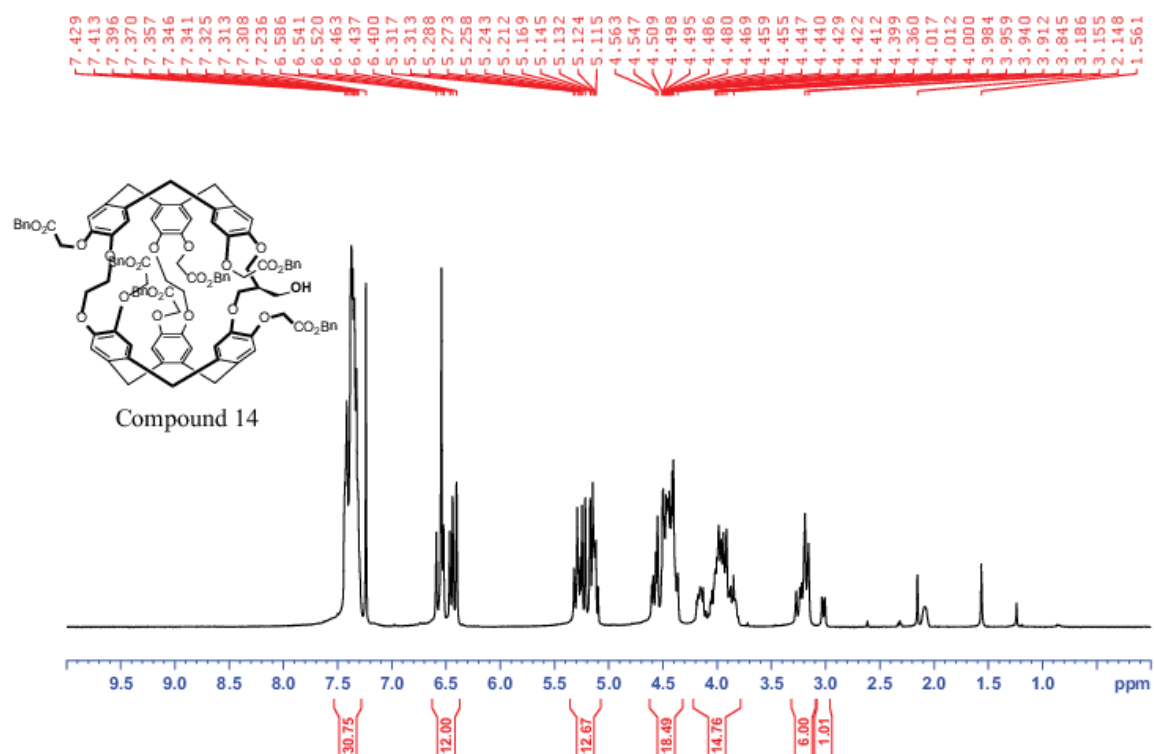


Figure S15: <sup>1</sup>H NMR (400 MHz) spectrum of compound **14** recorded in CDCl<sub>3</sub> at 298 K.

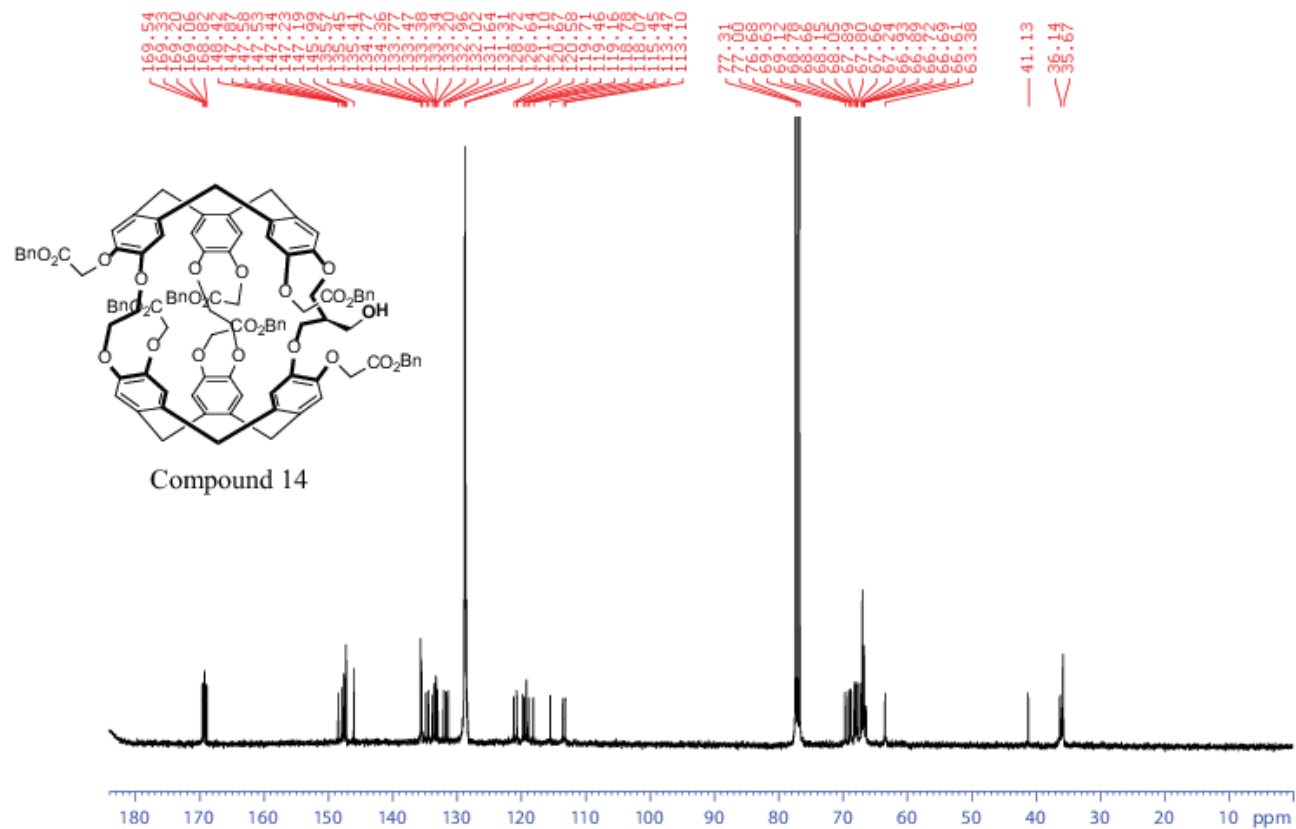


Figure S16:  $^{13}\text{C}$  NMR (100.6 MHz) spectrum of compound **14** recorded in  $\text{CDCl}_3$  at 298 K.

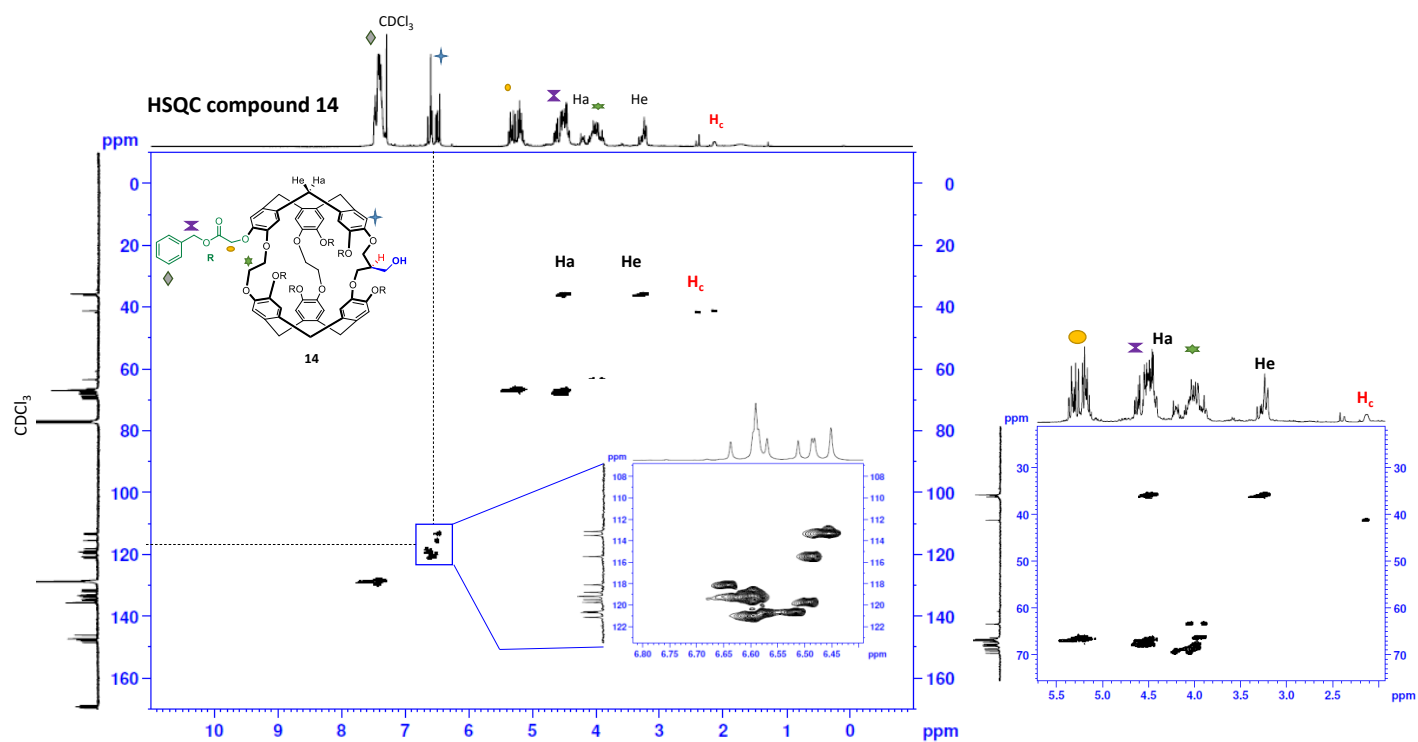


Figure S17: HSQC (400 MHz) spectrum of compound **14** recorded in  $\text{CDCl}_3$  at 298 K.



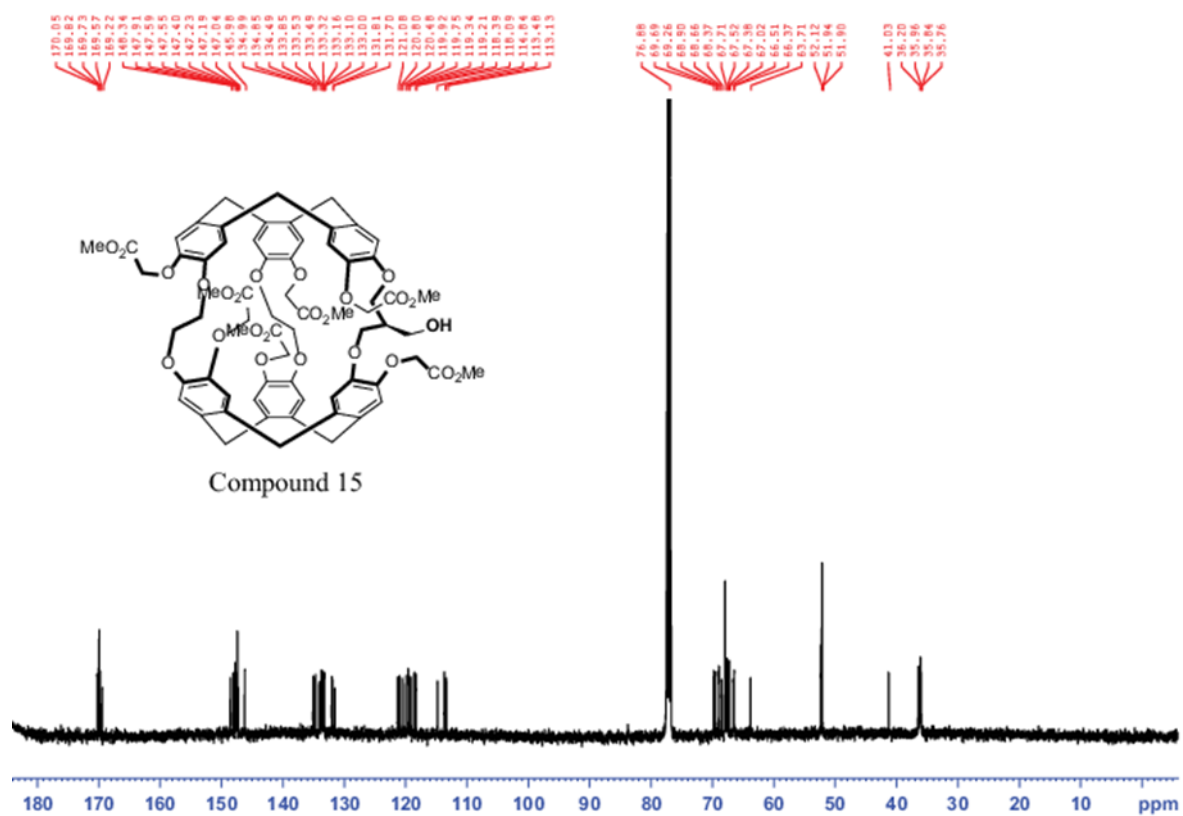


Figure S19:  $^{13}\text{C}$  NMR (100.6 MHz) spectrum of compound **15** recorded in  $\text{CDCl}_3$  at 298 K.

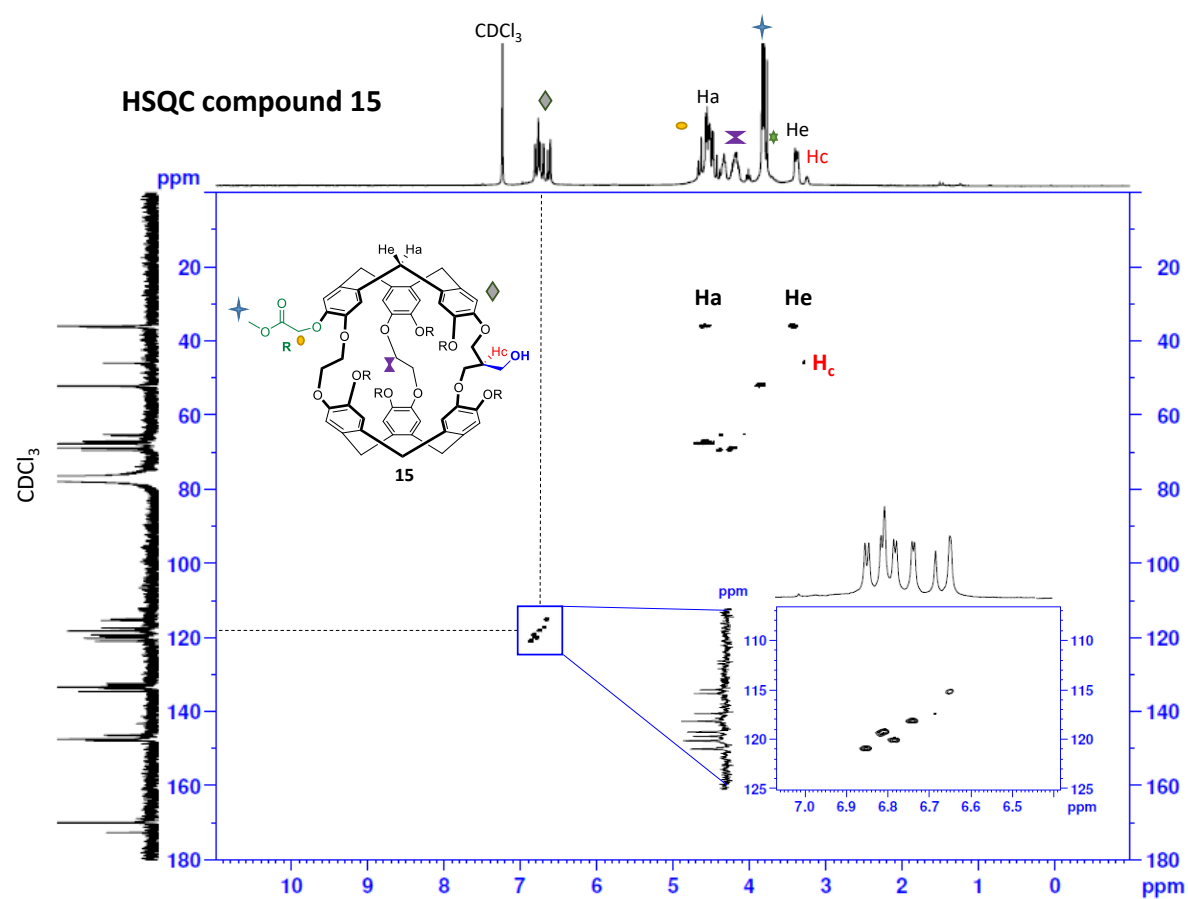


Figure S20: HSQC (400 MHz) spectrum of compound **15** recorded in CDCl<sub>3</sub> at 298 K.

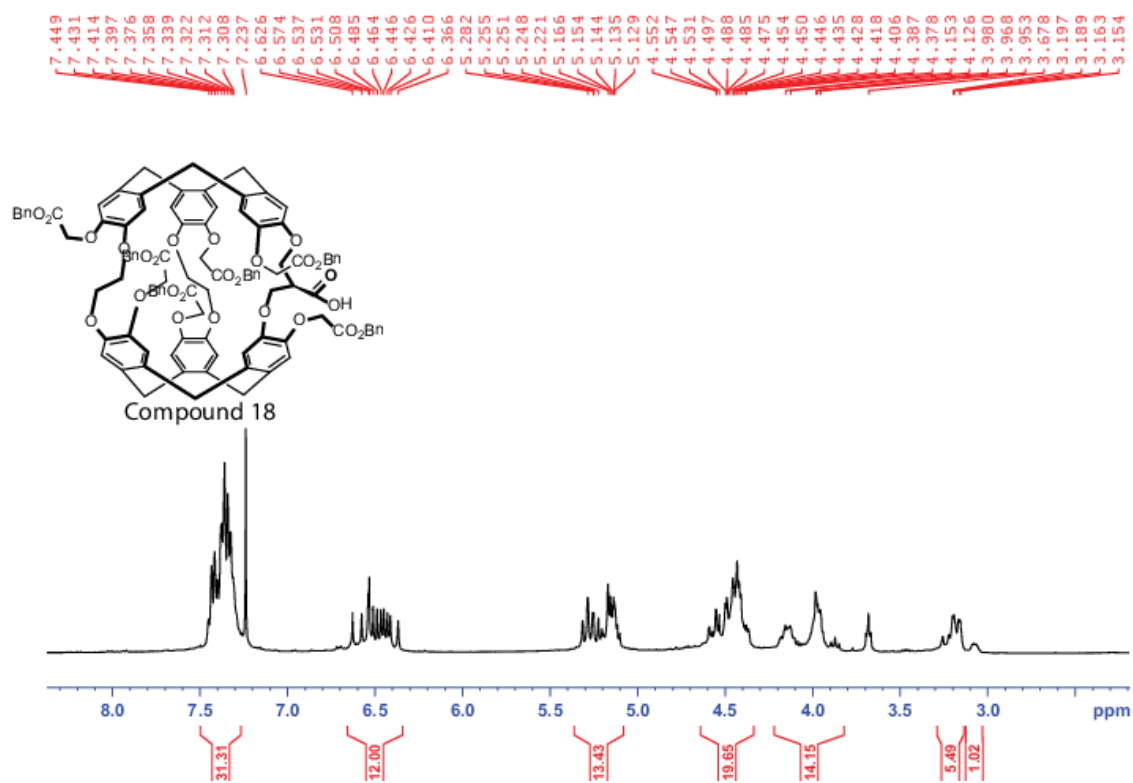


Figure S21:  $^1\text{H}$  NMR (400 MHz) spectrum of compound **18** recorded in  $\text{CDCl}_3$  at 298 K.



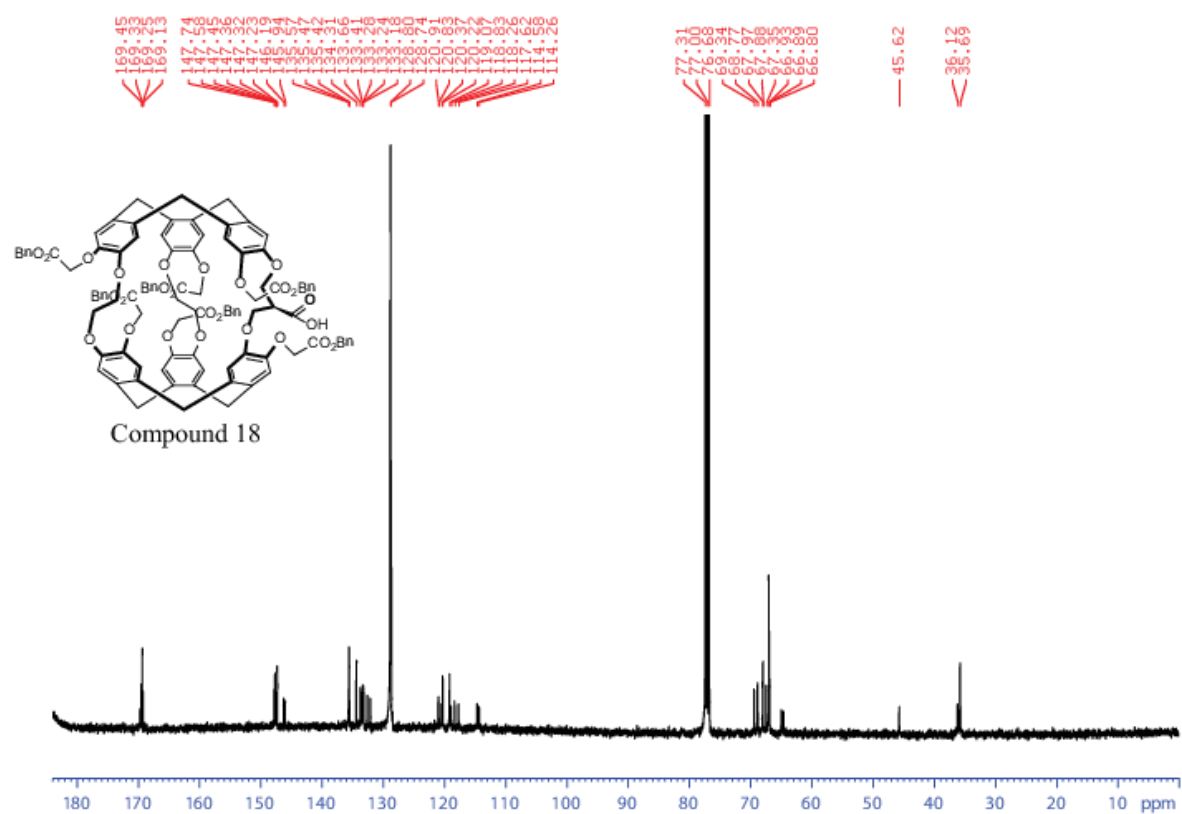


Figure S22:  $^{13}\text{C}$  NMR (100.6 MHz) spectrum of compound **18** recorded in CDCl<sub>3</sub> at 298 K.

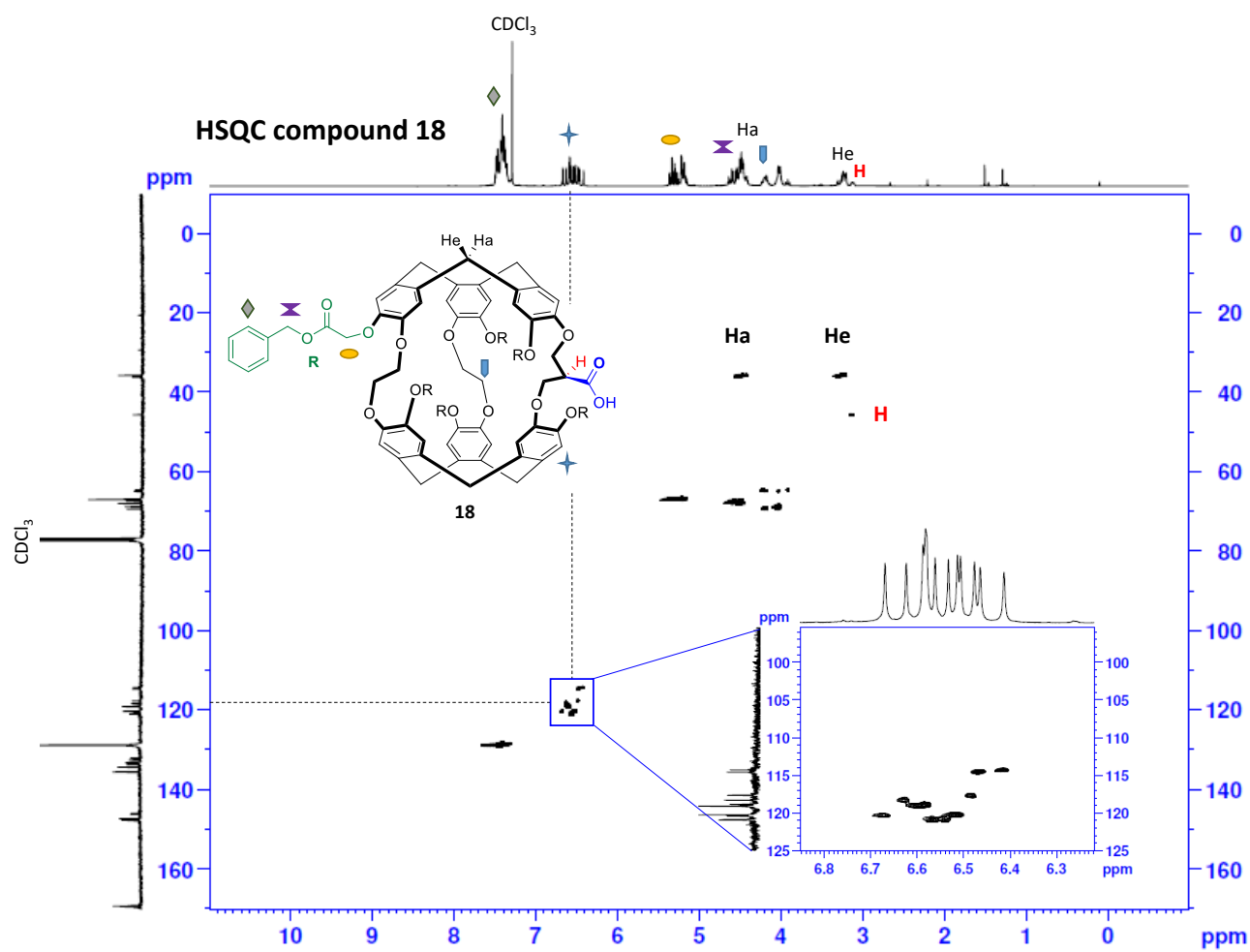


Figure S23: HSQC (400 MHz) spectrum of compound **18** recorded in CDCl<sub>3</sub> at 298 K.

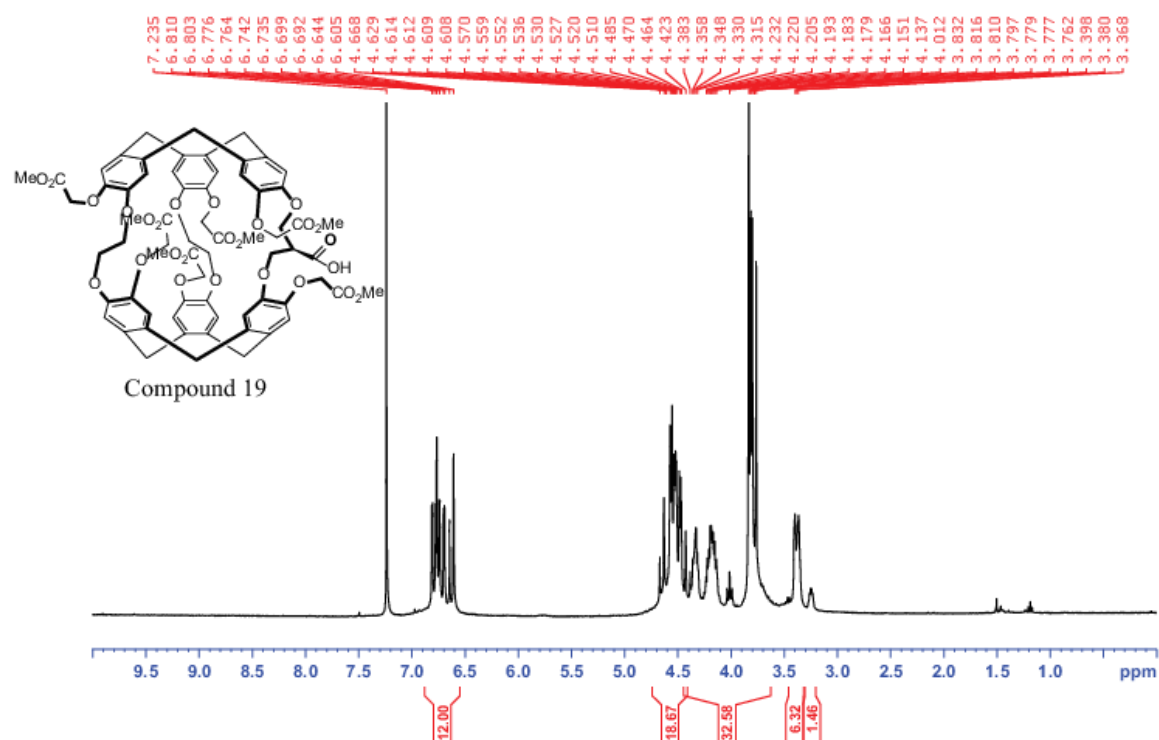


Figure S24:  $^1\text{H}$  NMR (400 MHz) spectrum of compound **19** recorded in  $\text{CDCl}_3$  at 298 K.

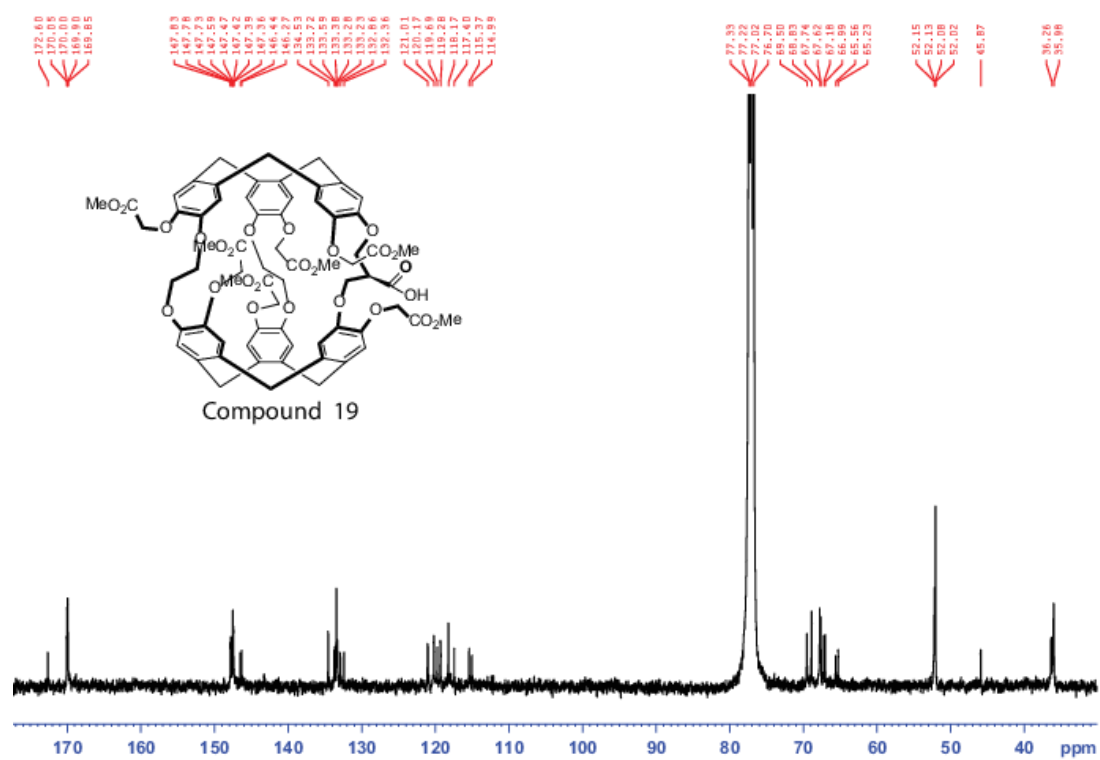


Figure S25:  $^{13}\text{C}$  NMR (100.6 MHz) spectrum of compound **19** recorded in  $\text{CDCl}_3$  at 298 K.

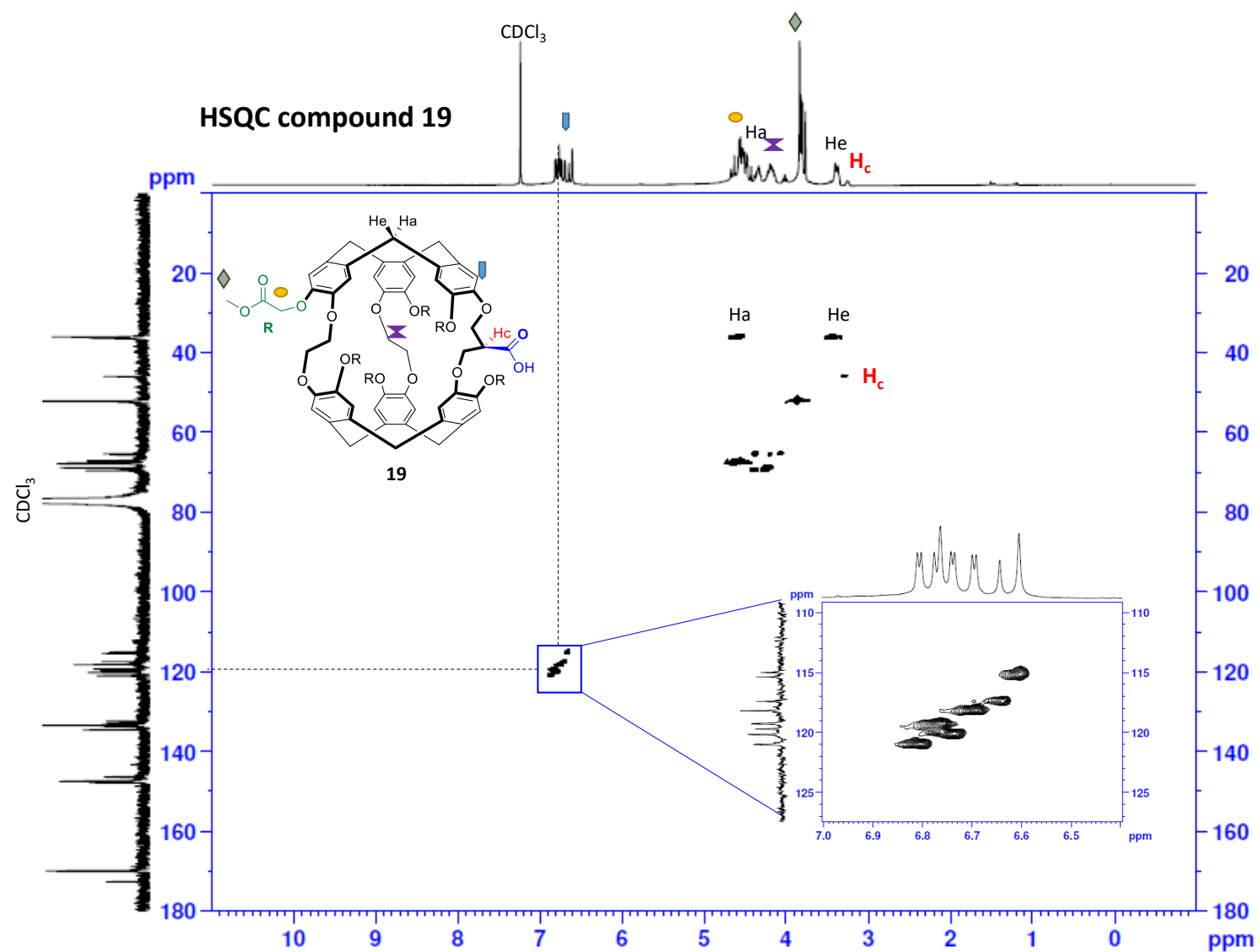


Figure S26: HSQC (400 MHz) spectrum of compound **19** recorded in CDCl<sub>3</sub> at 298 K.

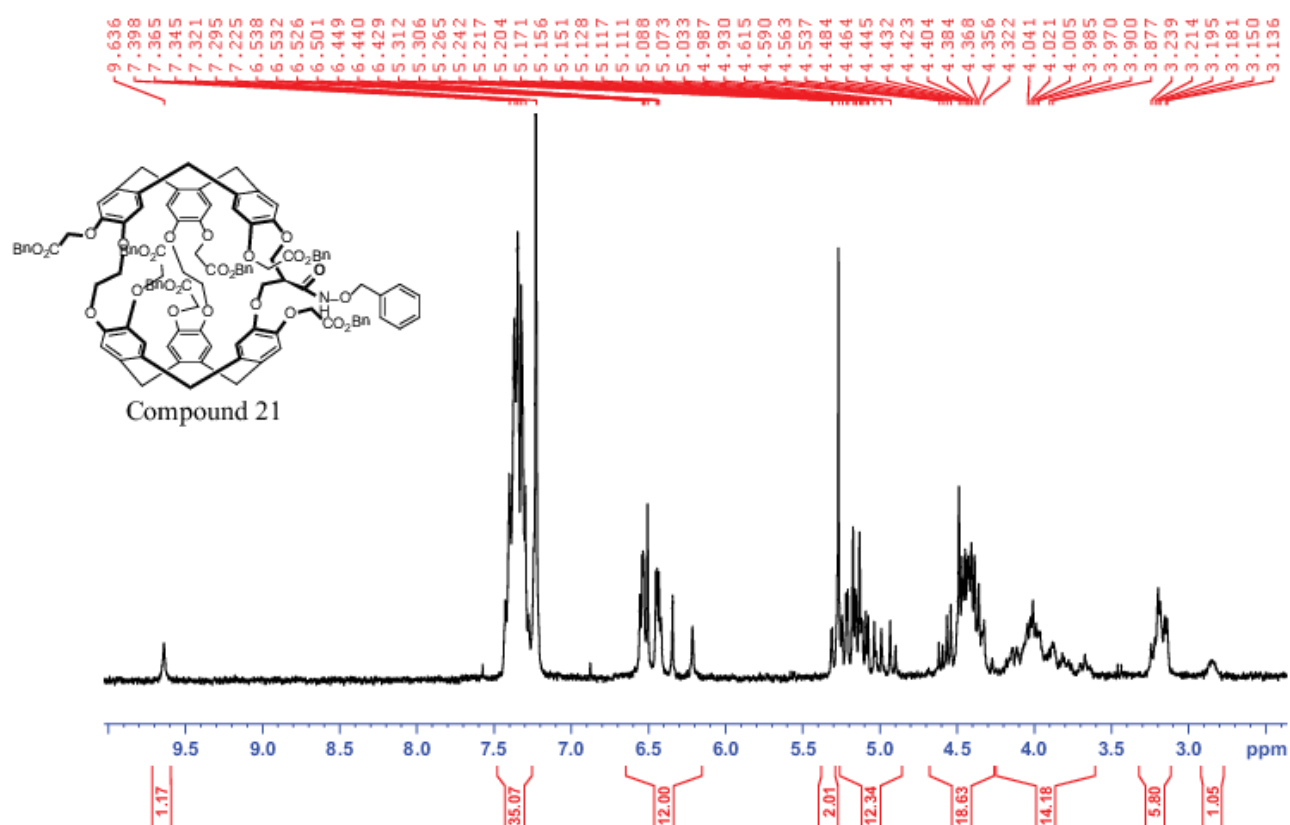


Figure S27:  $^1\text{H}$  NMR (400 MHz) spectrum of compound **21** recorded in  $\text{CDCl}_3$  at 298 K.



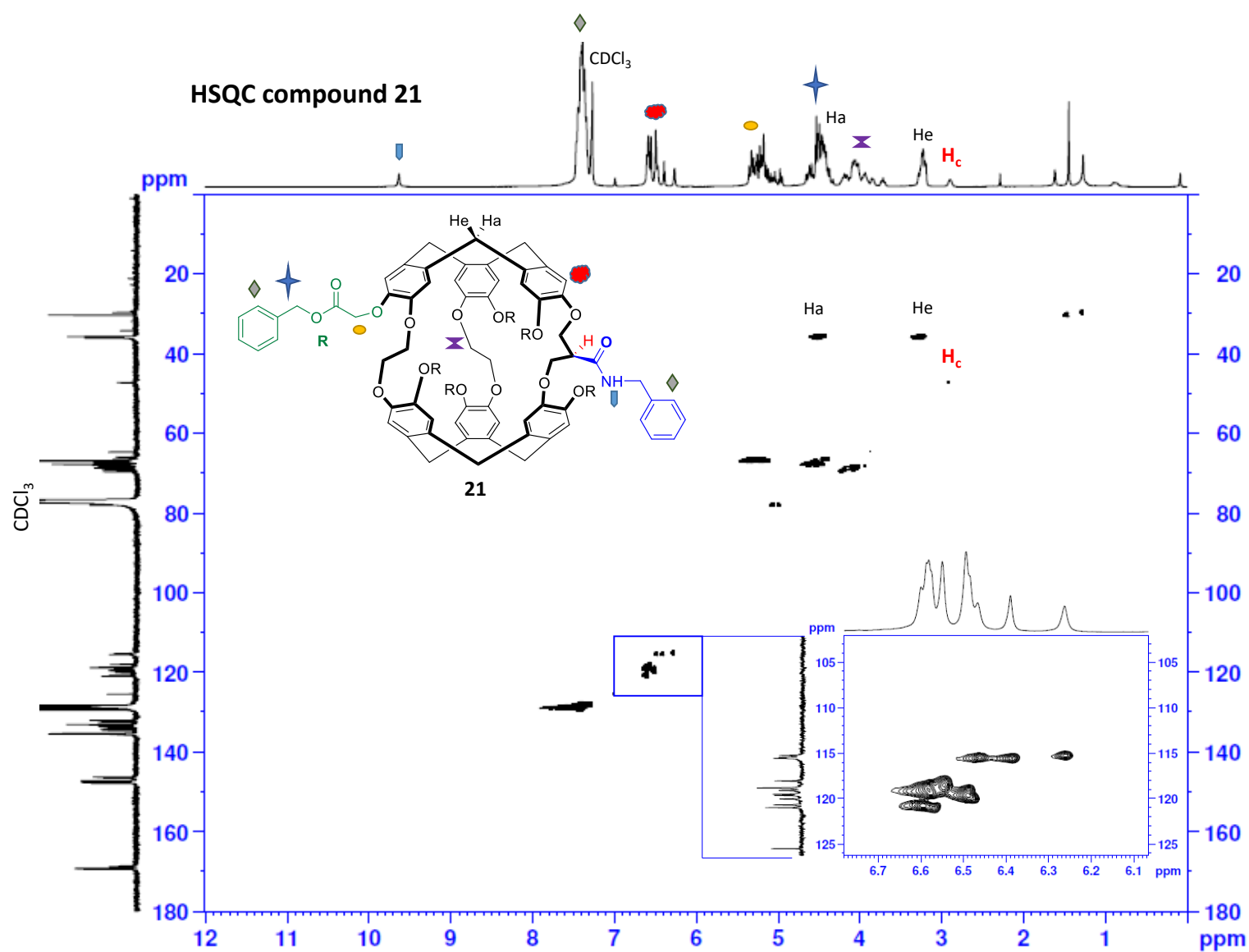


Figure S29: HSQC (400 MHz) spectrum of compound **21** recorded in CDCl<sub>3</sub> at 298 K.



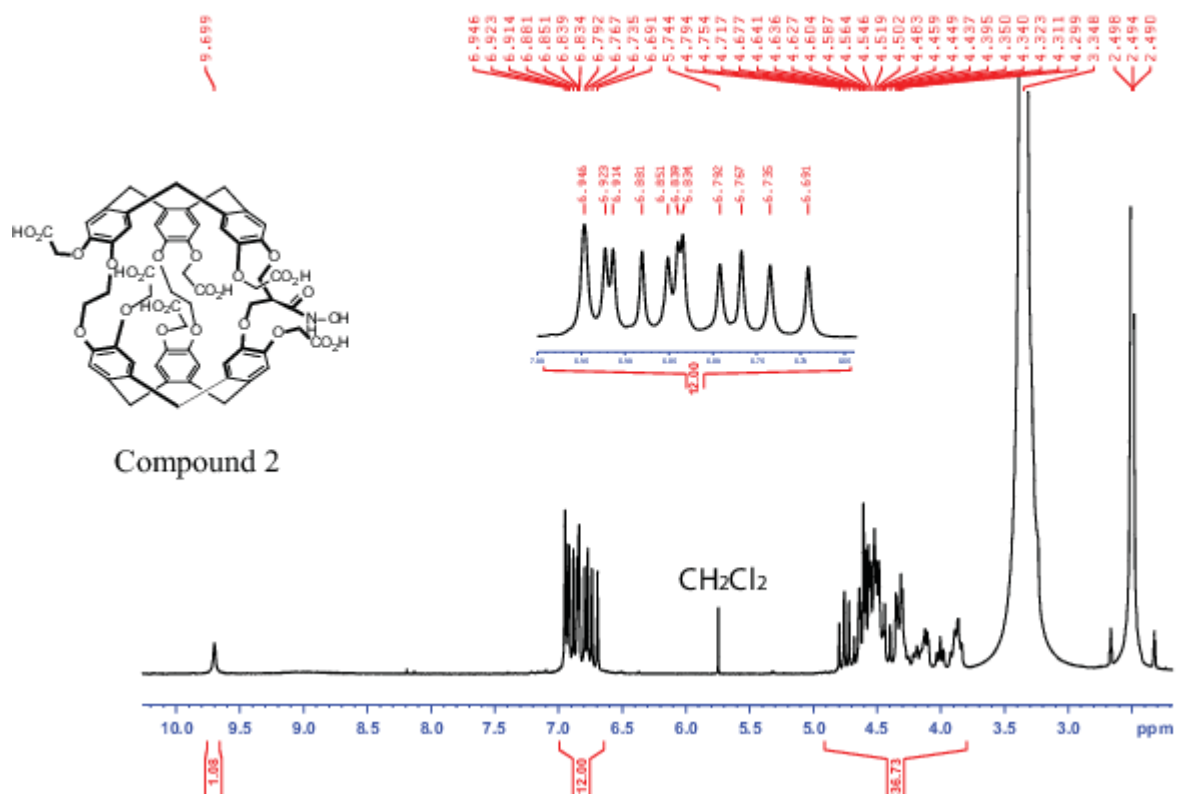


Figure S30: <sup>1</sup>H NMR (400 MHz) spectrum of compound **2** recorded in DMSO-*d*<sub>6</sub> at 298 K.



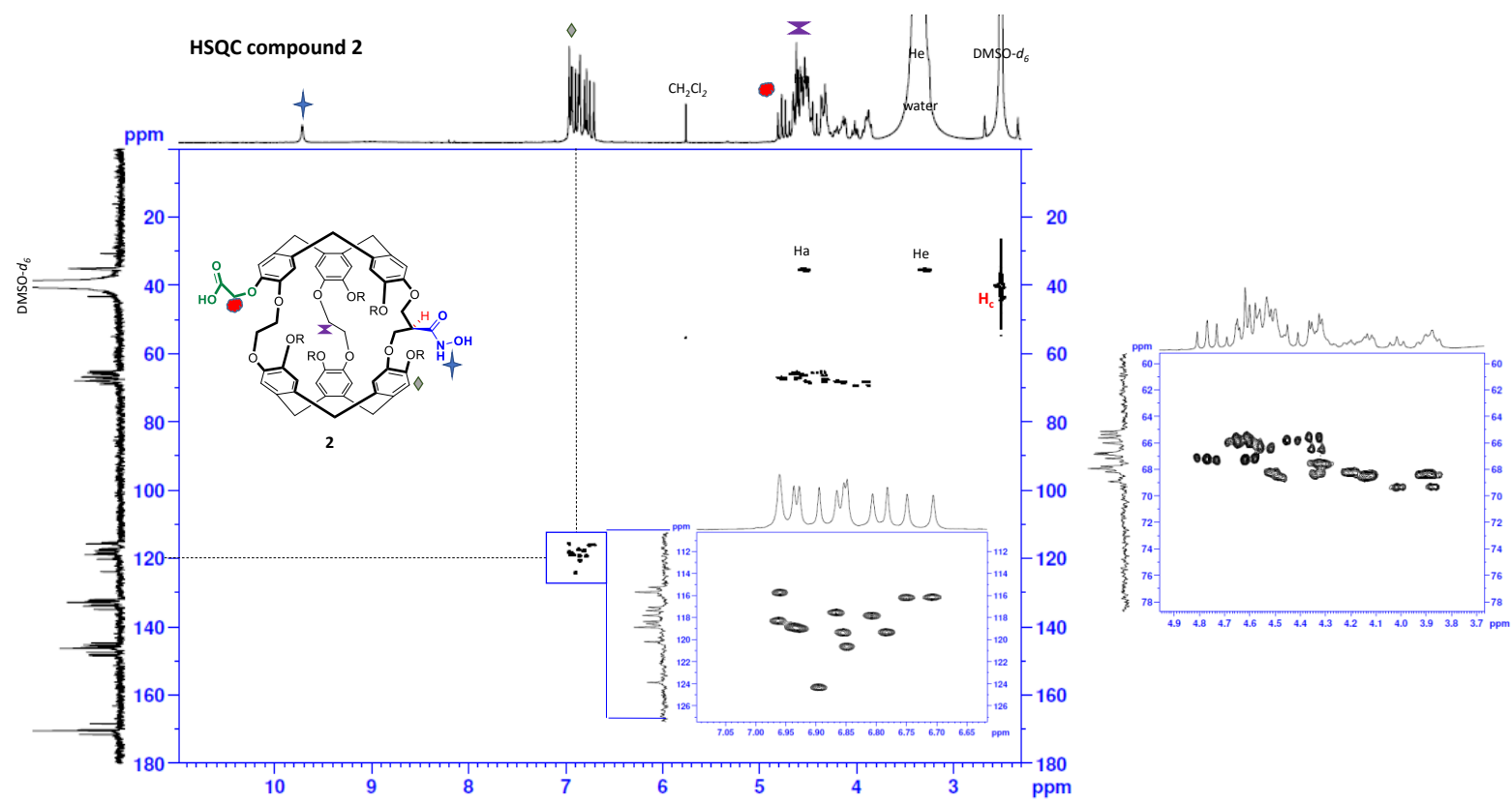


Figure S32: HSQC (400 MHz) spectrum of compound **2** recorded in DMSO-*d*<sub>6</sub> at 298 K.

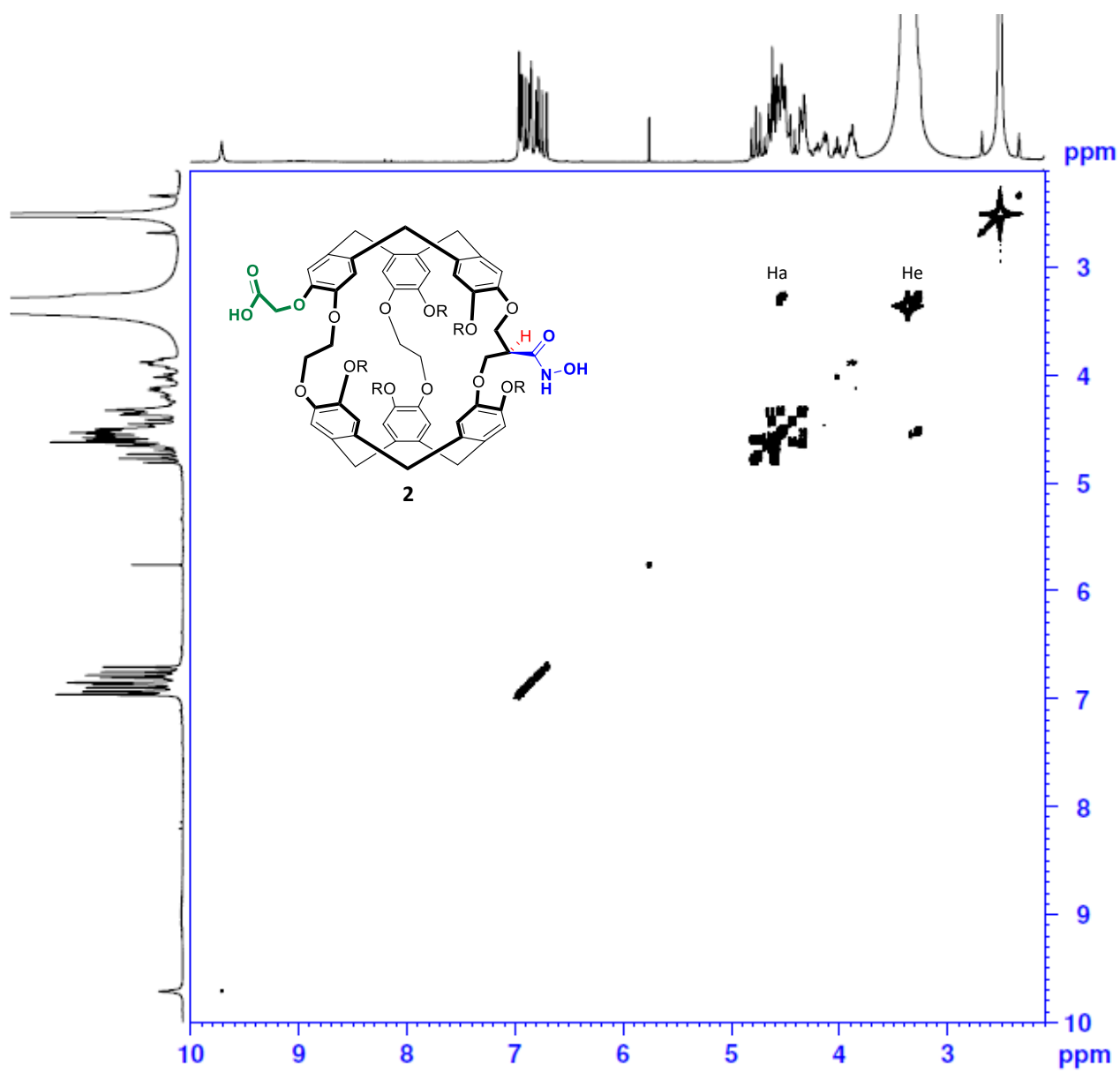


Figure S33: COSY (400 MHz) spectrum of compound **2** recorded in  $\text{DMSO-}d_6$  at 298 K.







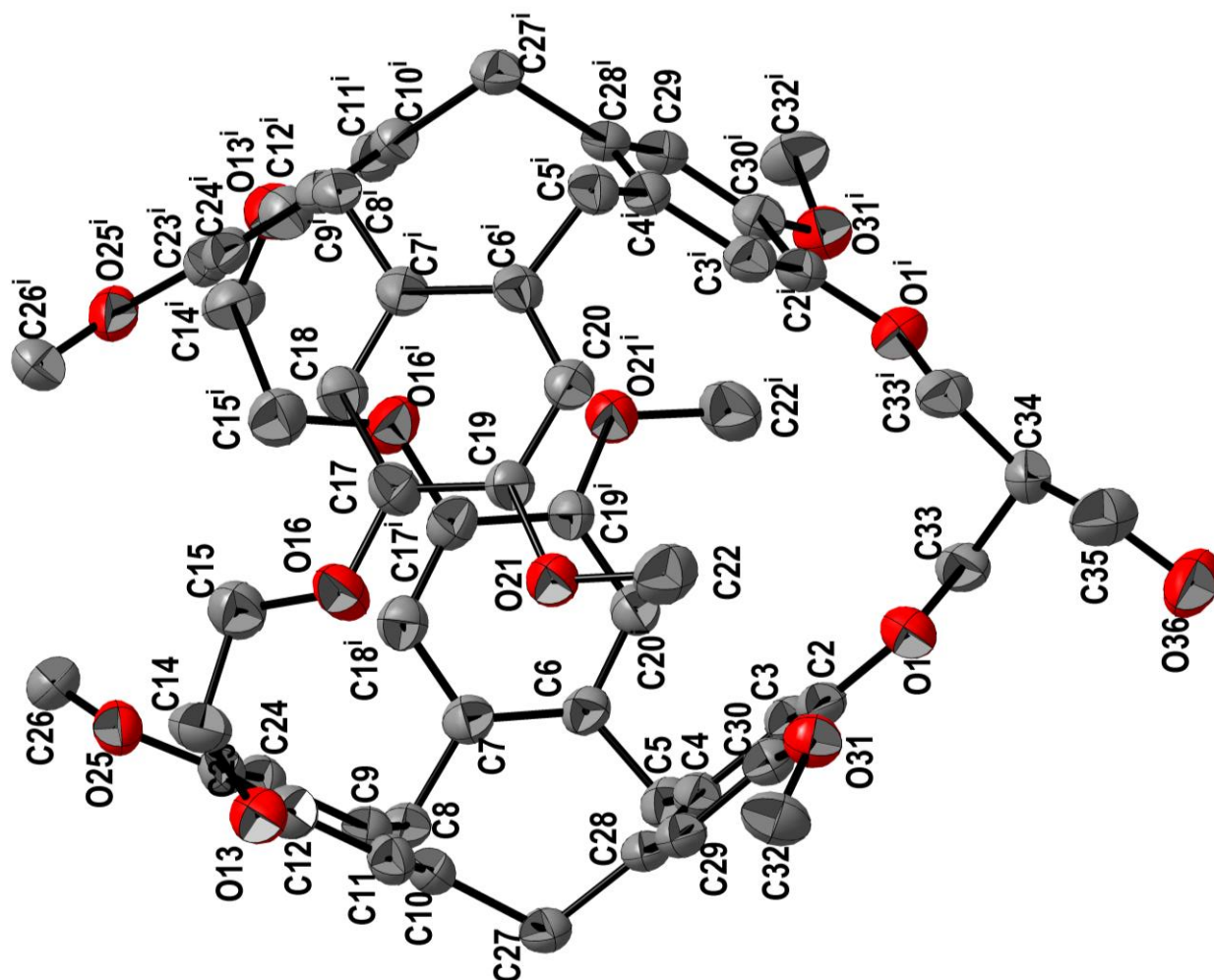


Figure S37: ORTEP representation of compound **9** (hydrogen atoms and solvent molecule were omitted for clarity. The displacement ellipsoids were plotted at 30% probability level).



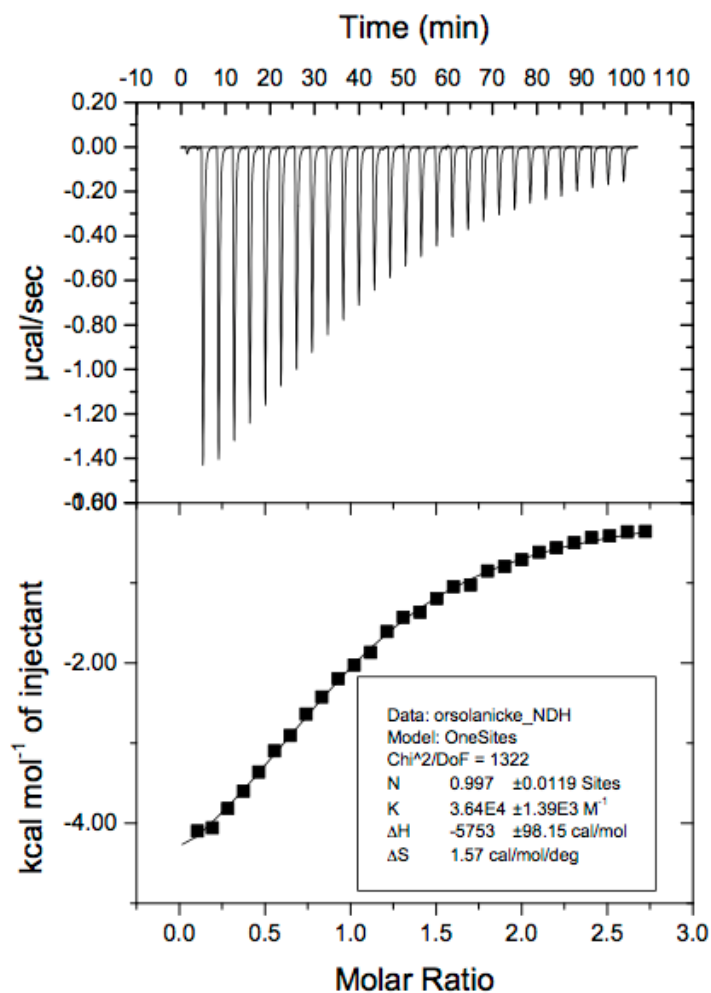


Figure S38: Calorimetric titration of compound **2** in H<sub>2</sub>O/TRIS (20 mM; pH = 7.6). The solution host ( $c = 0.08$  mM) was placed into the calorimeter cell (1.4 mL) and 28 successive aliquots (10  $\mu$ L) of Ni<sup>2+</sup> solution ( $c = 1.0$  mM) were added at 3 min intervals.

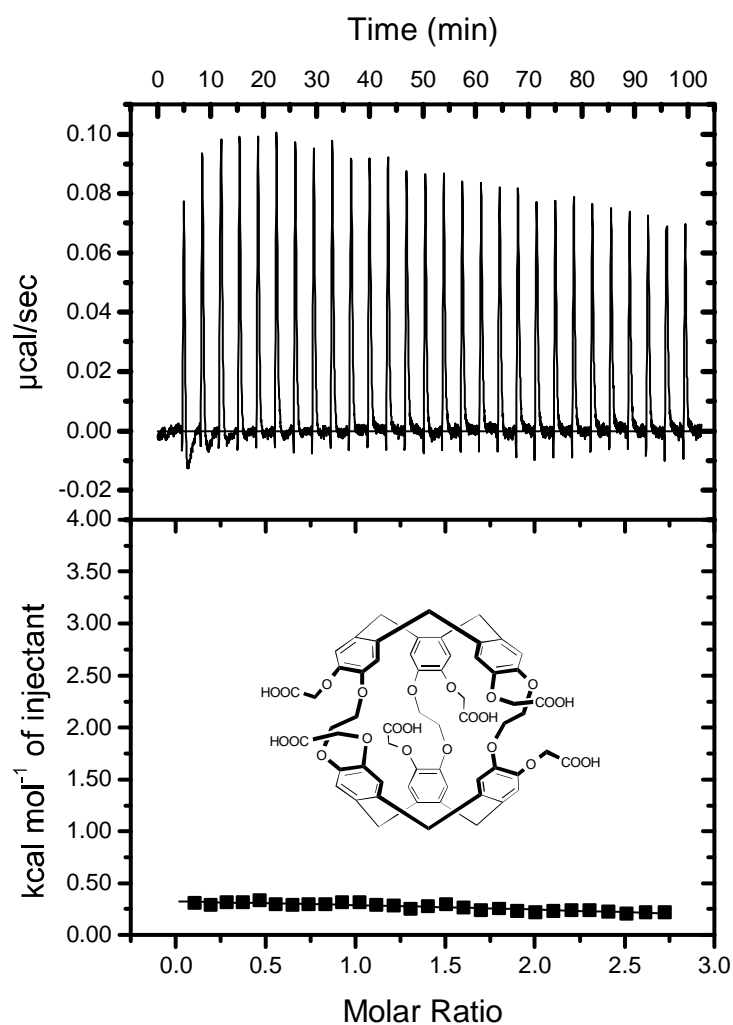


Figure S39: Calorimetric titration of Cryptophane-222 hexacarboxylate in H<sub>2</sub>O/TRIS (20 mM; pH = 7.0). The solution host ( $c = 0.08$  mM) was placed into the calorimeter cell (1.4 mL) and 28 successive aliquots (10  $\mu$ L) of Zn<sup>2+</sup> solution ( $c = 1.0$  mM) were added at 3 min intervals.

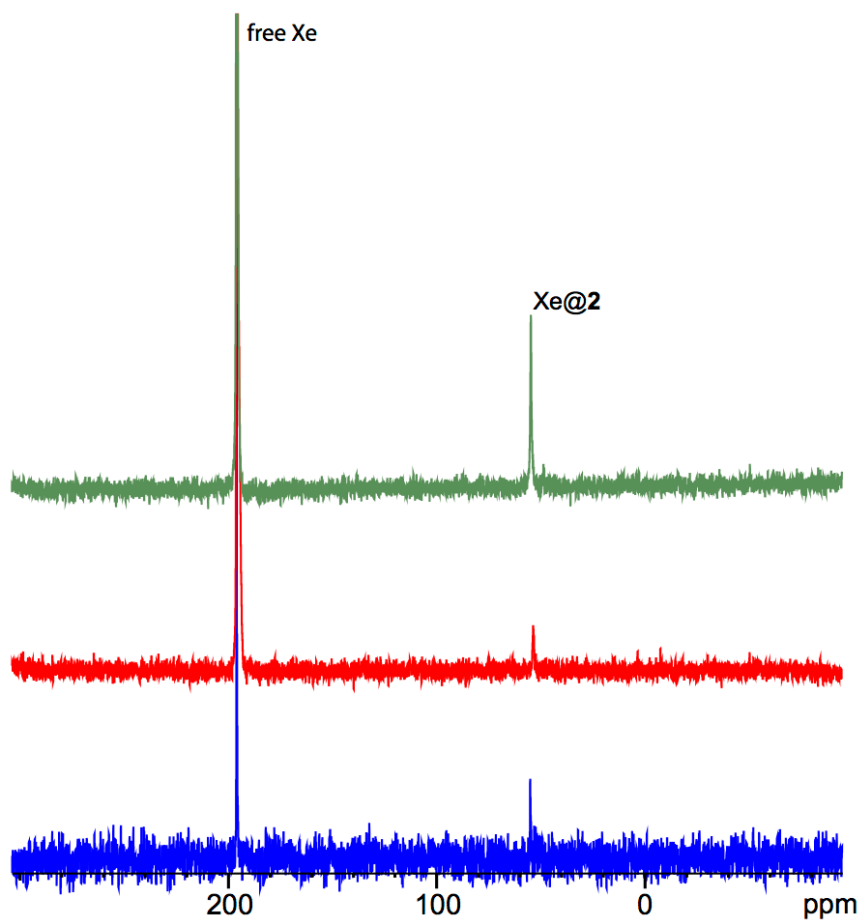


Figure S40: Hyperpolarized  $^{129}\text{Xe}$  spectra (1 scann) of compound **2** in TRIS buffer (20 mM, pH = 7.5). a) in absence of  $\text{Zn}^{2+}$  (green spectrum). b) in presence of 1.0 equiv. of  $\text{Zn}^{2+}$  (blue spectrum). c) in presence of 5.0 equiv. of  $\text{Zn}^{2+}$  (red spectrum). Spectra recorded at 25°C. 0.39 mg of **1** in 1000  $\mu\text{L}$  of TRIS buffer (20 mM)/10%  $\text{D}_2\text{O}$  pH 7.5 ( $c = 0.32$  mM).

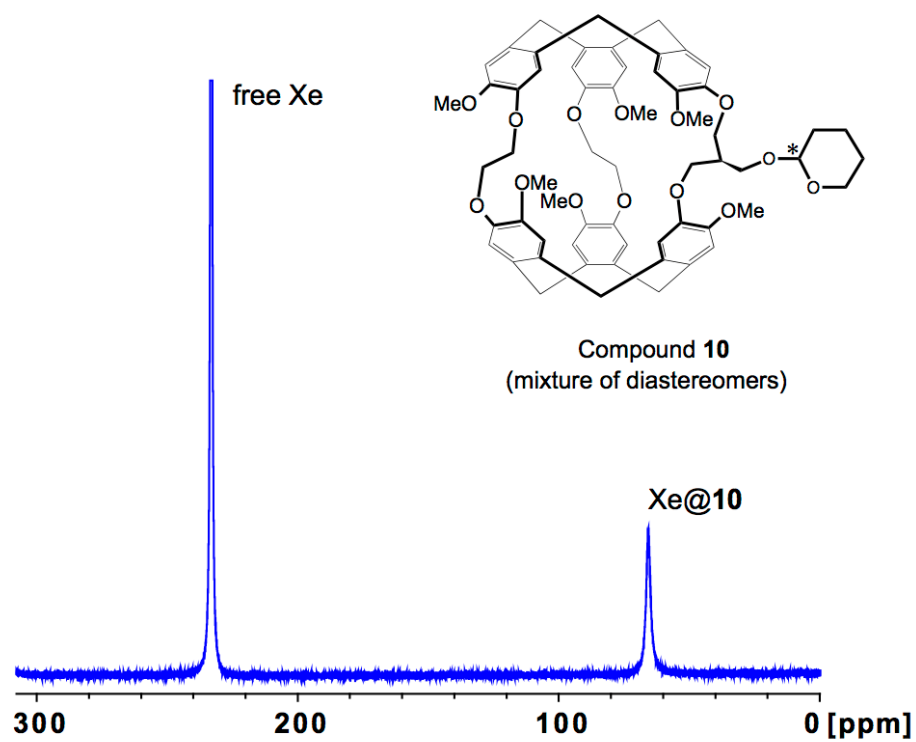


Figure S41: hyperpolarized  $^{129}\text{Xe}$  NMR spectrum of compound **10** recorded at 298 K in  $\text{C}_2\text{D}_2\text{Cl}_4$ . 1.8 mg of cryptophane **10** in 600  $\mu\text{L}$  of 1,1,2,2-tetrachloroethane- $d_2$ .

### Crystal data

$C_{56}H_{58}O_{13}$	$D_x = 0.987 \text{ Mg m}^{-3}$
$M_r = 939.07$	Cu $K\alpha$ radiation, $\lambda = 1.54184 \text{ \AA}$
Cubic, $I2_13$	Cell parameters from 3819 reflections
Hall symbol: I 2b 2c 3	$q = 3.2\text{--}66.9^\circ$
$a = 33.5971 (16) \text{ \AA}$	$m = 0.57 \text{ mm}^{-1}$
$V = 37923 (5) \text{ \AA}^3$	$T = 150 \text{ K}$
$Z = 24$	Plate, colorless
$F(000) = 11952$	$0.40 \times 0.32 \times 0.19 \text{ mm}$

### Data collection

Xcalibur, Atlas, Gemini ultra diffractometer	10442 independent reflections
Radiation source: fine-focus sealed X-ray tube, Enhance Ultra (Cu) X-ray Source	6339 reflections with $I > 2.0\sigma(I)$
Mirror monochromator	$R_{\text{int}} = 0.069$
Detector resolution: $10.4685 \text{ pixels mm}^{-1}$	$q_{\text{max}} = 67.0^\circ$ , $q_{\text{min}} = 3.2^\circ$
w scans	$h = -38\text{--}28$
Absorption correction: analytical <i>CrysAlis PRO</i> 1.171.38.46 (Rigaku Oxford Diffraction, 2015) Analytical numeric absorption correction using a multifaceted crystal model based on expressions derived by R.C. Clark & J.S. Reid. (Clark, R. C. & Reid, J. S. (1995). <i>Acta Cryst. A</i> 51, 887-897) Empirical absorption correction using spherical harmonics, implemented in SCALE3 ABSPACK scaling algorithm.	$k = -40\text{--}10$
$T_{\text{min}} = 0.863$ , $T_{\text{max}} = 0.912$	$l = -29\text{--}26$
25070 measured reflections	

### Refinement

Refinement on $F^2$	Hydrogen site location: difference Fourier map
Least-squares matrix: full	H-atom parameters constrained
$R[F^2 > 2s(F^2)] = 0.078$	Method, part 1, Chebyshev polynomial, (Watkin, 1994, Prince, 1982) $[\text{weight}] = 1.0/[A_0 \cdot T_0(x) + A_1 \cdot T_1(x) \cdots + A_{n-1} \cdot T_{n-1}(x)]$ where $A_i$ are the Chebyshev coefficients listed below and $x = F/F_{\text{max}}$ Method = Robust Weighting (Prince, 1982) $W = [\text{weight}] \cdot [1 - (\Delta F / 6 \cdot \sigma(F))^2]$ $A_i$ are: 347. 529. 246. 53.0
$wR(F^2) = 0.189$	$(D/s)_{\text{max}} = 0.003$
$S = 1.00$	$D\rho_{\text{max}} = 0.53 \text{ e \AA}^{-3}$
645 parameters	Absolute structure: Parsons, Flack & Wagner (2013), 3517 Friedel Pairs
60 restraints	Absolute structure parameter: 0.473 (13)
Primary atom site location: other	

Table S1: Crystallographic data of cryptophane **9**.