

Supporting information

Multiwalled carbon nanotubes - poly(3-octylthiophene-2,5-diyl) nanocomposite transducer for ion-selective electrodes. Raman spectroscopy insight into the transducer / membrane interface

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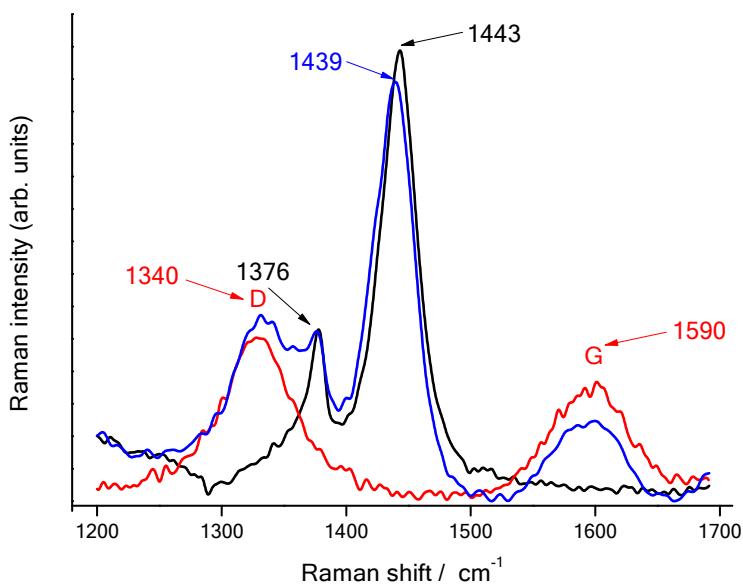


Figure S1. Raman spectra ($\lambda_{\text{exc}}=632.8$ nm) of MWCNT (red curve); POT (black curve) and MWCNT dispersed with POT – composite material (blue curve).

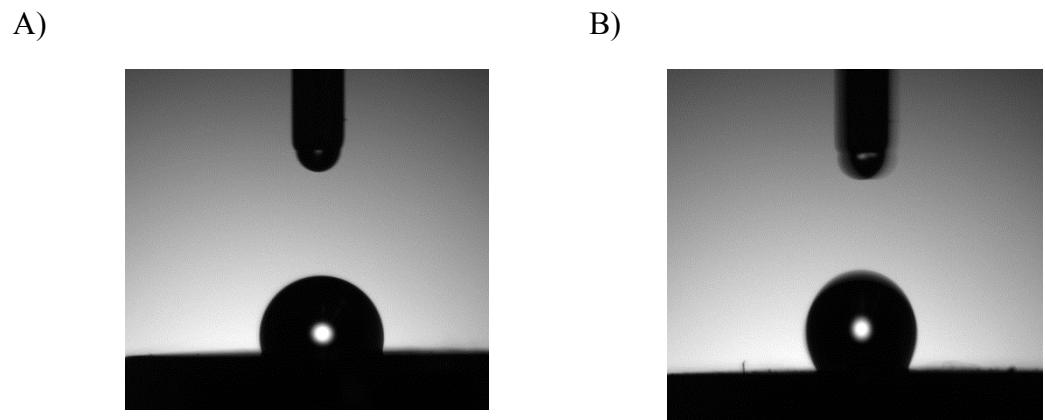


Figure S2. Contact angle measurements of A) GC electrode modified with POT layer; B) - GC electrode modified with POT-MWCNTs layer.

Table S1.

Comparison of electrochemical and analytical parameters offered by different by all-solid-state sensors of different arrangements.

	MWCNT s-POT This work	CW¹	CF-G¹	PEDOT¹	PANI¹	POT¹	GRAPHENE²	CB³	MWCNT s(CMC)⁴	[TbuPPC o-(II)]/[TbuPCCo(III)(PPh₃)] with MWCNT s⁵
Mg²⁺	-3.7 ± 0.4	-3.5±0.3	-3.7±0.3	-3.6±0.2	-3.6±0.3	-3.4±0.4	-4.5±0.3	-5.5±0.2	-4.0±0.2	-3.7±0.4
Ca²⁺	-3.6 ± 0.4	-3.5±0.3	-3.7±0.3	-3.5±0.3	-3.6±0.3	-3.4±0.4	-3.6±0.2	-5.4±0.2	-3.9±0.6	-3.2±0.3
H⁺	-4.6 ± 0.3	-4.5±0.2	-4.6±0.1	-4.2±0.3	-4.6±0.3	-4.5±0.3	-	-	-	-4.3±0.4
Na⁺	-3.5 ± 0.1	-3.4±0.2	-3.5±0.2	-3.1±0.2	-3.5±0.1	-3.1±0.1	-4.5±0.3	-3.5±0.2	-3.4 ± 0.1	-3.7±0.1
slope±SD (mV/dec)	56.3±0.4	61.2±0.7	60.0±1.8	55.2±1.7	59.2±1.2	59.8±0.5	59.2±0.1	59.1±0.2	57.3±0.6	55.7±0.7
linear range	10⁻¹-10⁻⁶	10 ⁻¹ -10 ⁻⁶	10 ⁻¹ -10 ⁻⁷	10 ⁻¹ -10 ⁻⁶	10 ⁻¹ -10 ⁻⁶	10 ⁻¹ -10 ⁻⁷	10 ⁻¹ -10 ^{-4.5}	-	10 ⁻¹ -10 ⁻⁶	10 ⁻¹ -10 ⁻⁷
detection limit [M]	10^{-6.8}	10 ^{-6.5}	10 ^{-7.1}	10 ^{-6.3}	10 ^{-6.6}	10 ^{-6.8}	10 ⁻⁵	10 ^{-6.4}	10 ^{-6.4}	10 ^{-6.9}
resistance [Ω]	10⁵	8.8x10 ⁶	3.0x10 ⁶	1.8x10 ⁶	1.1x10 ⁶	2.5x10 ⁶	-	1.2x10 ⁶	1x10 ⁶	6.2x10 ⁶
capacitance [F]	2x10⁻⁴	2x10 ⁻⁶	3x10 ⁻⁵	9x10 ⁻⁵	2x10 ⁻⁵	1x10 ⁻⁴	-	5.1x10 ⁻⁵	-	1.5x10 ⁻⁵

MWCNTs-POT - multiwalled carbon nanotubes- poly(octylthiophene) nanocomposite, **CW** - coated wire, **CF-G** - carboxy-functionalized graphene, **PEDOT** - poly(3,4-ethylenedioxothiophene), **POT** - regioregular poly(octylthiophene), **CB** - carbon black, **MWCNTs(CMC)** - multi-walled carbon nanotubes dispersion in sodium carboxymethylcellulose, **[TbuPPCo(II)]/[TbuPCCo(III)(PPh₃)] with MWCNTs** - 5,10,15,20-tetrakis(4-tert-butylphenyl)porphyrin and 5,10,15-tris(4-tert-butylphenyl)corrole Co complexes with multi-walled carbon nanotubes

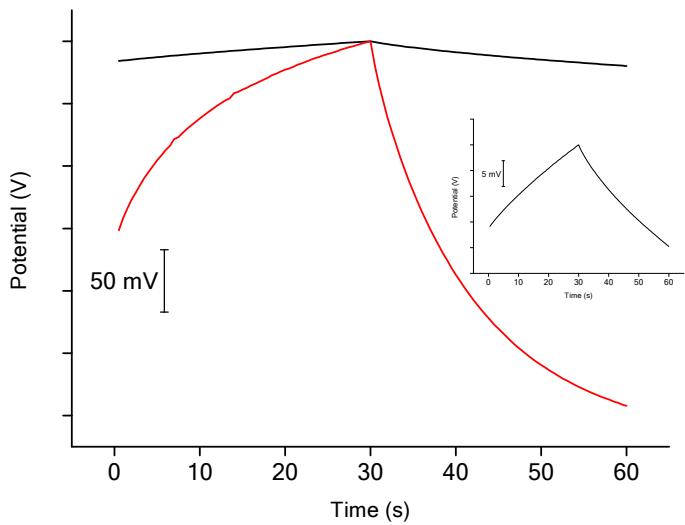


Figure S3. Results of chronopotentiometric experiments performed for glassy carbon substrate electrode covered with (red line) POT or (black line) POT-MWCNTs layer recorded in 10^{-3} M KCl, applying cathodic / anodic current of 10 nA. The experimental curves were shifted to give the same value at the moment of polarization change. Inset – magnified trace for POT-MWCNTs composite scale bar is 5 mV.

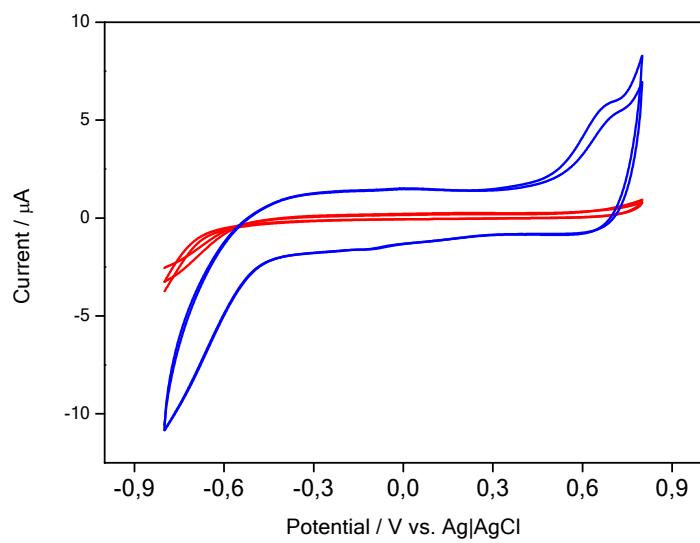


Figure S4. Cyclic voltammograms recorded for (red line) POT or (blue line) POT-MWCNTs layer coated on glassy carbon supporting electrode in 10^{-3} M KCl, scan rate 50 mV/s.

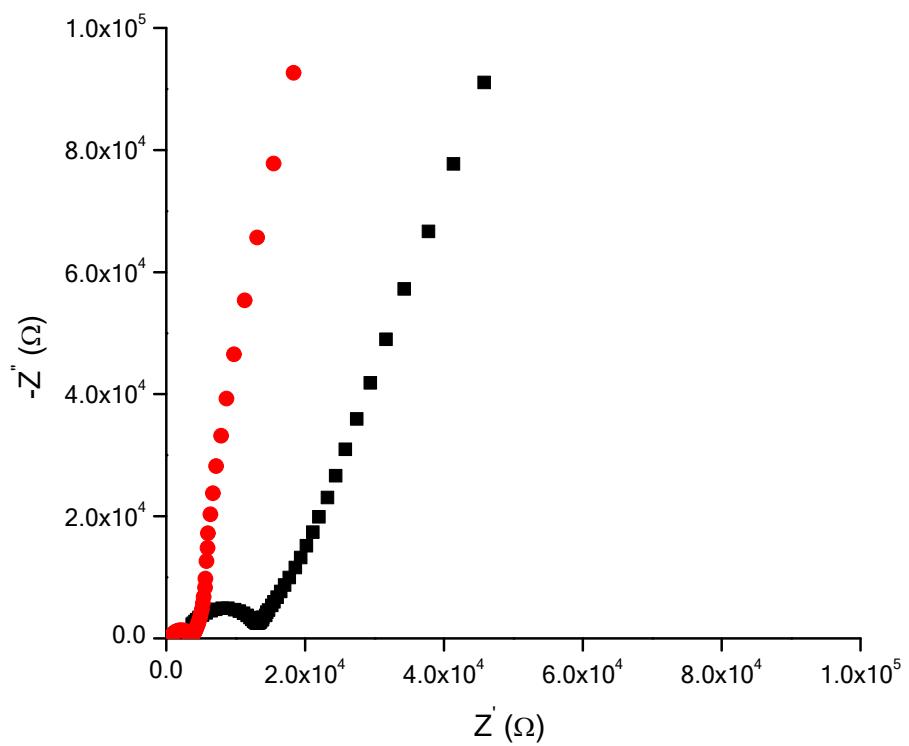


Figure S5. EIS spectra of (■) POT and (●) POT-MWCNTs layer recorded in 10^{-3} M KCl solution, in the frequency range from 0.01 to 10^5 Hz, using an amplitude of 50 mV at a potential of 0.5V.

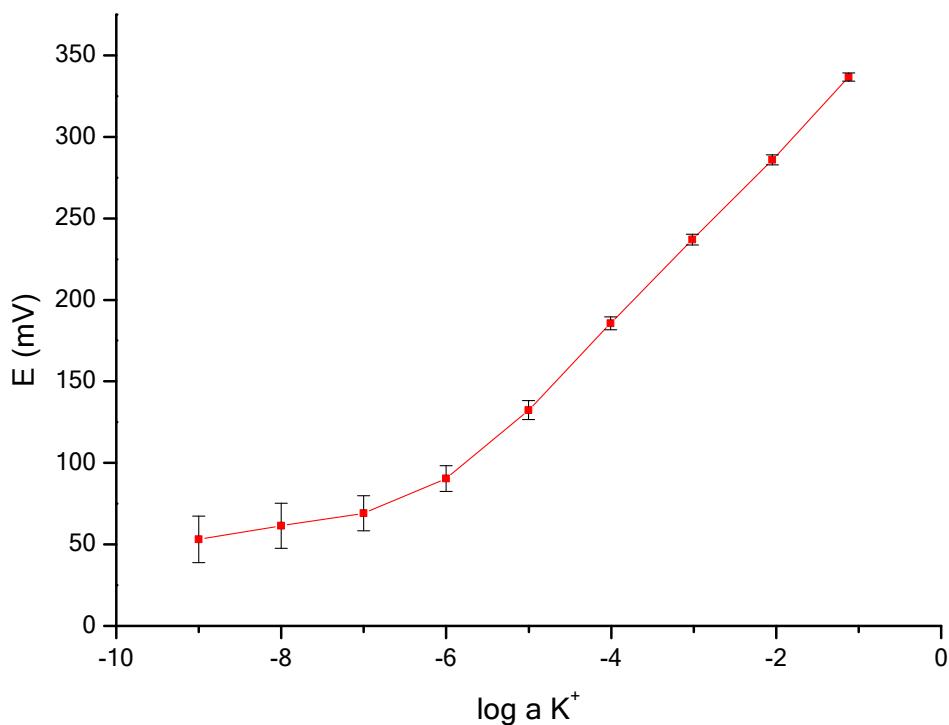


Figure S6. Potentiometric response of potassium-selective electrodes with POT-MWCNTs composite transducer recorded in KCl solutions mean potential values \pm SD recorded over 6 calibrations performed during one week.

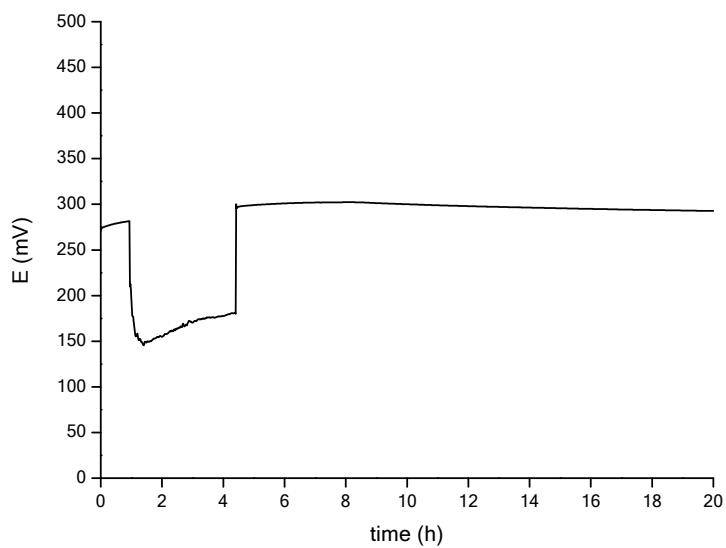
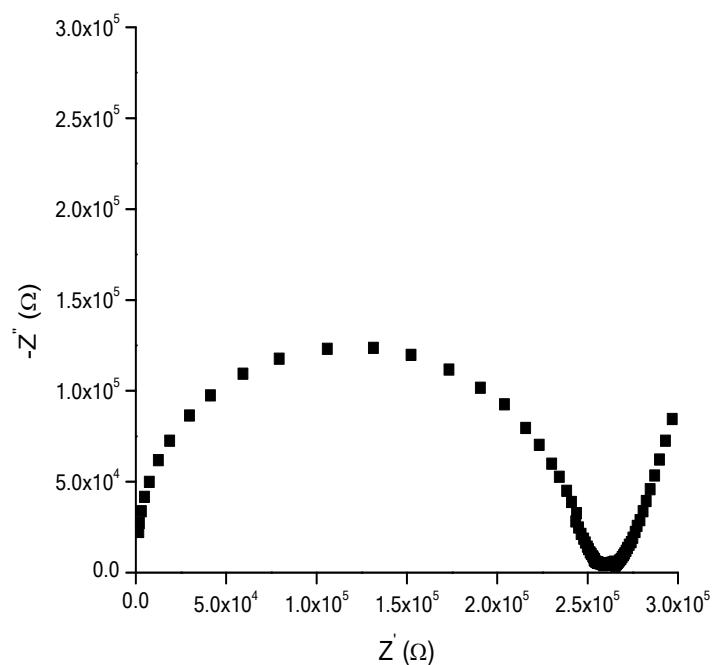


Figure S7. Results of water layer test experiment performed for all-solid-state sensor with POT-MWCNTs transducer recorded in 10^{-3} M KCl solution, 10^{-3} M NaCl and again in 10^{-3} M KCl solution.

A)



B)

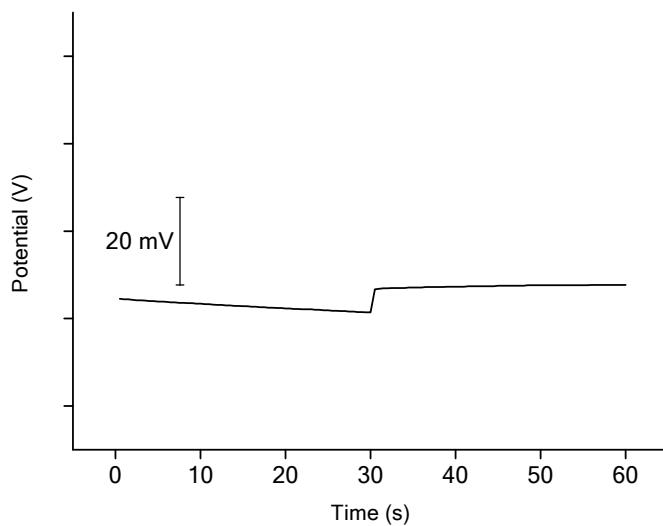


Figure S8. A) EIS spectrum recorded for all-solid-state potassium selective electrode with POT-MWCNTs transducer recorded in 0.1 M KCl solution, in the frequency range from 0.01 to 10^5 Hz, using an amplitude of 50 mV at a potential of 0.5V. B) Chronopotentiometric experiments results performed for all-solid-state potassium selective electrode with POT-MWCNTs transducer recorded in 0.1 M KCl solution using current $1 \cdot 10^{-8}$ A.

References

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