#### Electronic Supplementary Information

# Aprotic vs protic ionic liquids for lignocellulosic biomass pre-

# treatment: Anion effects, enzymatic hydrolysis, solid state NMR, distillation and recycle

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## **TABLES**

**Table S1:** List of physical properties for five different ILs used in this study.

Ionic liquid	Melting	pH at 100	Density (g	Viscosity at 25	T <sub>d (max)</sub>
	point (°C)	g l <sup>-1</sup>	cm <sup>-3</sup> )	°C (mPa s)	(°C)
[Emim][OAc]	< 25	6.5	1.027 <sup>a,b</sup>	162 <sup>4</sup>	245
[Emim]Cl	82 ± 5	5.0	1.112 <sup>a,c</sup>	-	307
[Eim][OAc]	< 25	7.0	1.026 <sup>b</sup>	4.2	193
[Eim][HCOO]	< 25	5.1	1.092 <sup>b</sup>	6.9	190
[Eim]Cl	69 ± 5	5.5	1.128 <sup>c</sup>	-	276

<sup>&</sup>lt;sup>a</sup> data from Sigma-Aldrich, <sup>b</sup> at 25 °C, <sup>c</sup> at 80 °C, T<sub>d (max)</sub> is the temperatures at maximum rate of thermal decomposition

**Table S2:** Maximum solubility of wood flour and percentage of recovered solid from different ILs.

Ionic liquid	Maximum dissolution (wt%)	Recovered solid (wt%)
[Emim][OAc]	5.5 (± 0.5)	80
[Emim]Cl	2.5 (± 0.5)	48
[Eim]Cl	3.5 (± 0.5)	58
[Eim][OAc]	No dissolution	95
	(only extract lignin)	
[Eim][HCOO]	No dissolution	92
	(only extract lignin)	

Where, [Emim] = 1-ethyl-3-methylimidazolium, [Eim] = 1-ethylimidazolium,

[OAc] = acetate and [HCOO] = formate.

**Table S3:** Maximum conversion efficiency from different ILs-treated wood after 96 h enzymatic hydrolysis.

Ionic liquid	Maximum enzymatic
	conversion (%)
Untreated wood	15.7 (± 0.6)
[Emim][OAc]	86.7 (± 9.8)
[Emim]Cl	56.2 (± 0.6)
[Eim]Cl	74.6 (± 2.3)
[Eim][OAc]	12.7 (± 1.5)

Where, [Emim] = 1-ethyl-3-methylimidazolium, [Eim] = 1-ethylimidazolium and [OAc] = acetate.

**Table S4:** <sup>13</sup>C sNMR chemical shift assigned for various moieties of pine wood. All values and assignments are adapted from reference [1].

Moiety / Functional group	Chemical shift region (ppm)
-COO/ CH₃COO	180-165
Aromatic C-O (C3, 4 of lignin)	160-141
Aromatic C-C (C1 of lignin)	141-125.8
Aromatic C-H (C2, 5, 6 of lignin)	125.8-108.5
C1 of cellulose	108.5-93.5
C4 of cellulose (crystalline)	93.5-86.6
C4 of cellulose (amorphous)	86.6-79.5
C2, 3, 5 of cellulose	79.5-68.0
C6 of cellulose	68.0-58.9
OCH <sub>3</sub> of lignin	58.9-50.8
CH₃ of hemicellulose	24.1-18.6

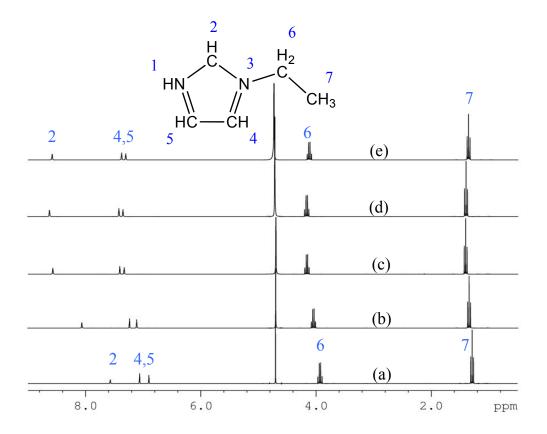
## **FIGURES**

**Figure S1:** Structure of five imidazolium-based protic ([Eim][OAc], [Eim][HCOO], [Eim]Cl) and aprotic ([Emim][OAc], [Emim]Cl) ionic liquids used for wood processing in this study.

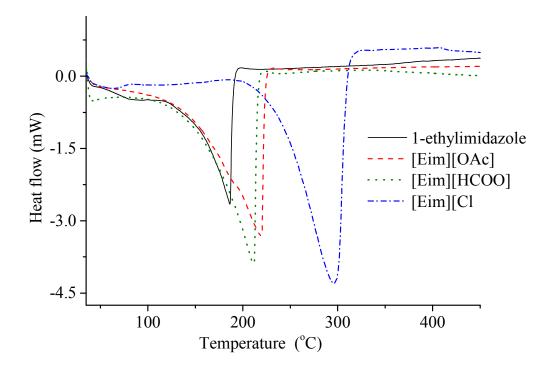
$$[Emim][OAc] \qquad [Emim]Cl$$

$$[Eim][OAc] \qquad [Eim][HCOO] \qquad [Eim]Cl$$

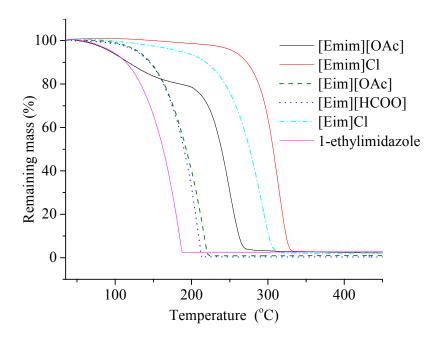
**Figure S2:** <sup>1</sup>H NMR spectra for various [Eim]Cl in presence of excess 1-ethylimidazole (Eim) and HCl acid where the ratio of Eim and HCl was maintained as 1:0 (a), 1:0.5 (b), 1:1 (c), 1:1.5 (d) and 1:2 (e)



**Figure S3:** DSC curves for imidazolium-based protic ILs, [Eim][OAc], [Eim][HCOO], [Eim]Cl; their precursor 1-ethylimidazole and aprotic [Emim][OAc], [Emim]Cl ionic liquids.

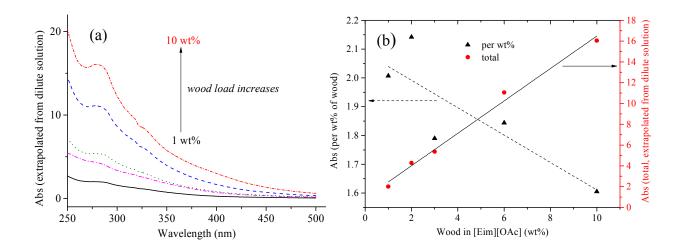


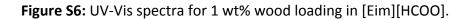
**Figure S4:** TGA curves for imidazolium-based protic ILs, [Eim][OAc], [Eim][HCOO], [Eim]Cl and their precursor, 1-ethylimidazole.

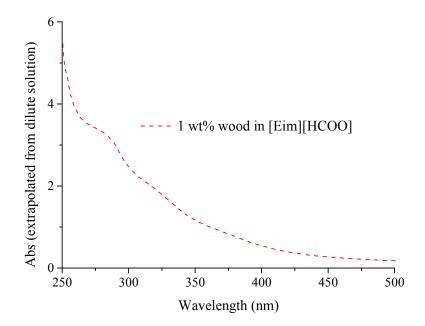


Please note [Eim][OAc], [Eim][HCOO] and [Eim]Cl were measured as synthesised, with water content <1wt%; the deliquescent, commercial [Emim]Cl was stored in a glovebox, and thus also had a water content of <1wt%; the commercial [Emim][OAc] was taken from the bottle directly (also <1wt% water content), but the significant mass loss before 200 °C is associated with water loss, whereby the water must have been uptaken by the sample from the atmosphere prior to measurement.

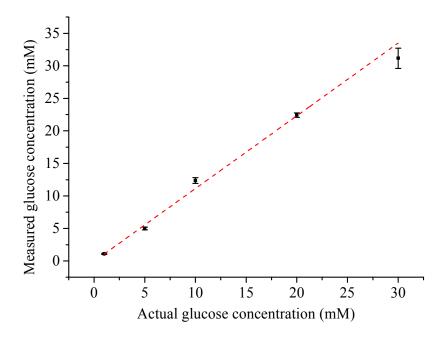
**Figure S5:** (a) UV-Vis spectra (b) plot of absorbance *vs* wood load for different amount of wood loading in [Eim][OAc] where 1 g of IL and different amount of wood flour were taken in round bottom flask at 115 °C for 18 h stirring at 700 rpm with reflux set-up.



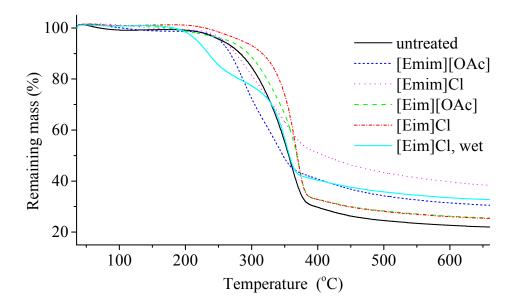




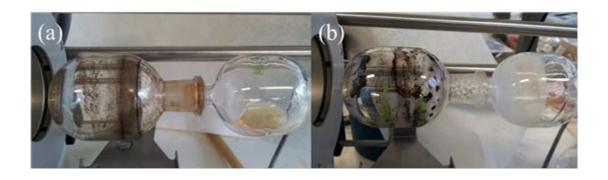
**Figure S7:** Calibration of commercial glucose sensor (Accuchek Active, used in this study) over its detection range (0.6 to 33.1 mM).



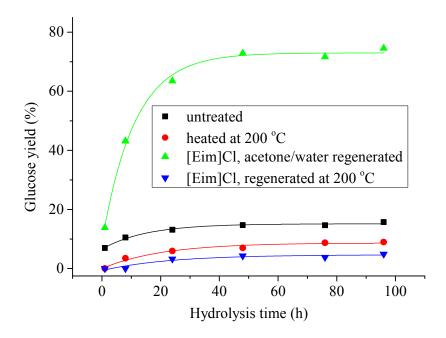
**Figure S8:** Thermogravimetric analysis of untreated wood and regenerated solid from different ionic liquids; [Emim][OAc], [Emim]Cl, [Eim][OAc], and [Eim]Cl.



**Figure S9:** Photograph of contaminated [Eim]Cl distillation in Kugelrohr apparatus after 1<sup>st</sup> (a) and 2<sup>nd</sup> (b) distillation at 200 °C.



**Figure S10:** Enzymatic conversion of untreated wood, wood heated at 200 °C, regenerated solid from [Eim]Cl using acetone-water mixture, and regenerated solid from [Eim]Cl obtained after distillation in Kugelrohr apparatus.



#### **REFERENCES**

[1] Holtman, K. M.; Chen, N.; Chappell, M. A.; Kadla, J. F.; Xu, L.; Mao, J. *J. Agric. Food Chem.* **2010**, *58*, 9882.