

Supporting Information

Pt and PtNi nanoparticle-supported multi-walled carbon nanotube electrocatalysts prepared by one-pot pyrolytic synthesis with an ionic liquid

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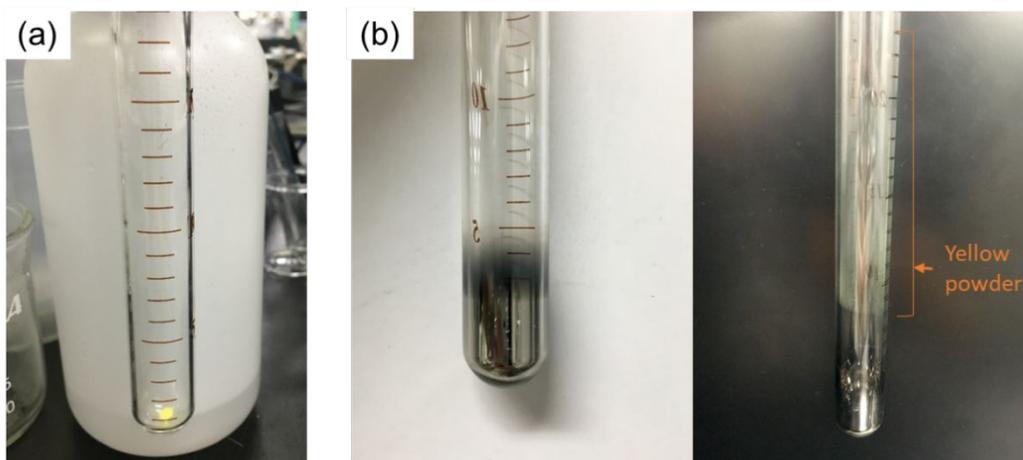


Figure S1. Photographs of a test tube containing $\text{Pt}(\text{acac})_2$ (a) before and (b) after heating. The test tube was heated at 10 K min^{-1} from room temperature to 473 K .

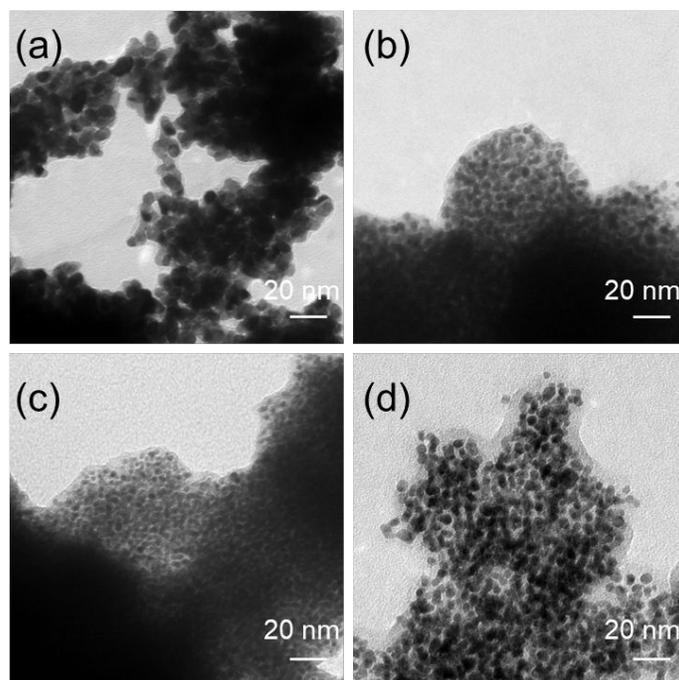


Figure S2. TEM images of Pt and PtNi nanoparticles prepared under the same conditions for specimens (a) **1**, (b) **2**, (c) **3** and (d) **4** but not containing MWCNTs.

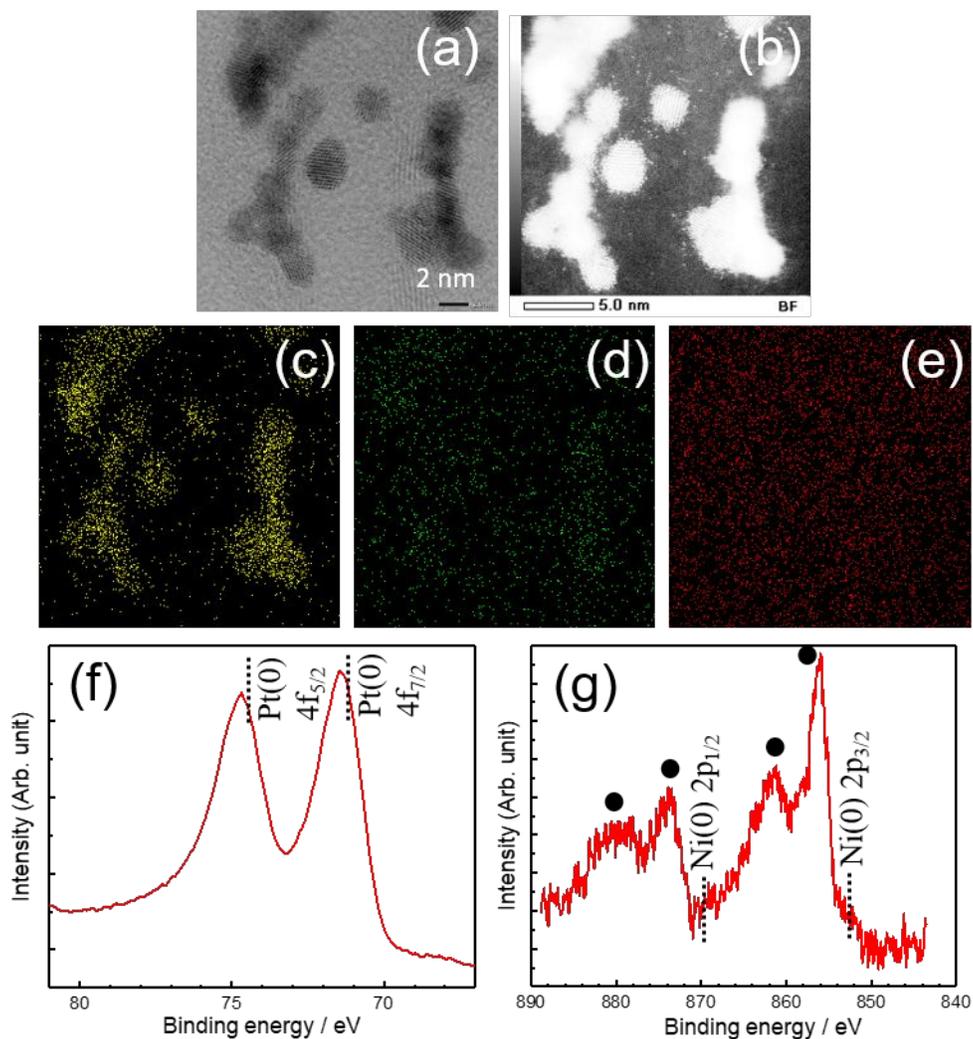


Figure S3. (a) High resolution TEM image, (b) HAADF-STEM image, (c-e) EDS mappings, and (f, g) XPS spectra of specimen **3**. The elements are (c, f) Pt, (d, g) Ni, and (e) C. The characterization of XPS spectra was conducted using Ref. S1. The filled circles, ●, shown in (g) are the XPS spectra related to Ni_2O_3 .

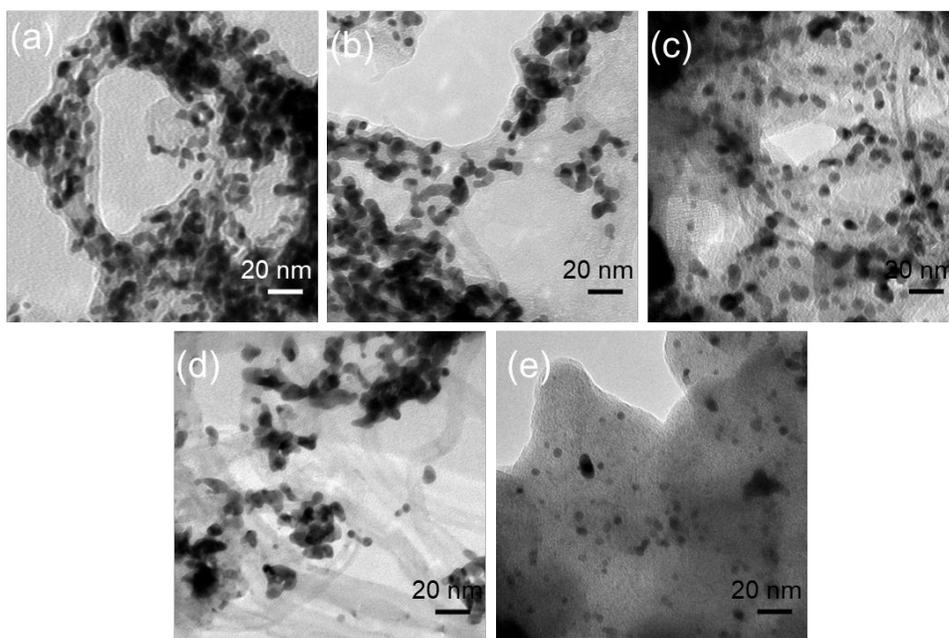


Figure S4. TEM images of specimens 1-5 after 15000th potential cycle test. The specimens are (a) 1, (b) 2, (c) 3, (d) 4, and (e) 5.

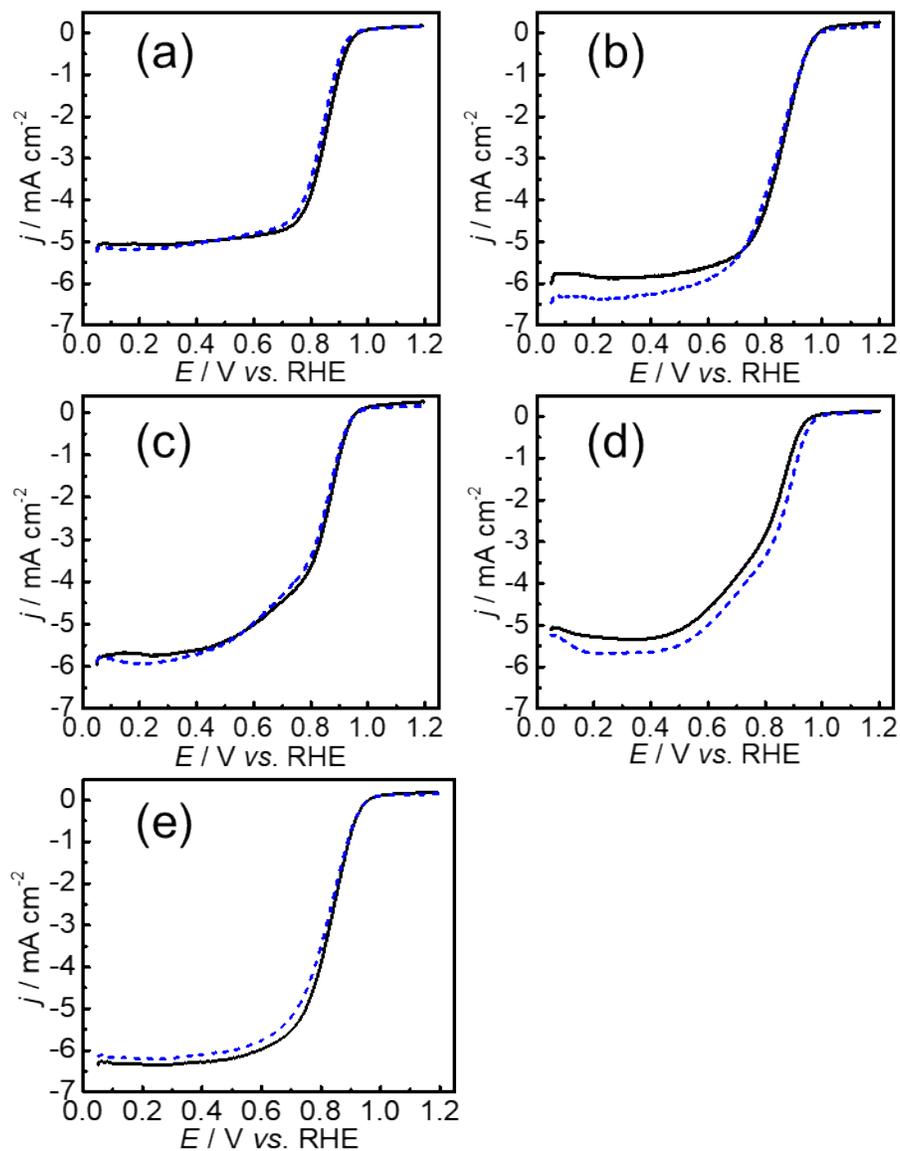


Figure S5. RDE-LSVs recorded at glassy carbon electrodes with the specimens in an O_2 -saturated 0.1 M HClO_4 aqueous solution (—) before and (- - -) after the 15000th potential cycle tests. The rotating speeds were 1600 rpm. The scan rates were 10 mV s^{-1} . The specimens are (a) **1**, (b) **2**, (c) **3**, (d) **4**, and (e) **5**.

Table S1. Summary of Pt and PtNi alloy nanoparticles obtained by different pyrolytic conditions

Metal precursor concentration / mM			Agitation time / h	Mean particle size / nm	Composition of nanoparticle ^a / at%			
Pt(acac) ₂	Ni[Tf ₂ N] ₂	Ni(acac) ₂			Without MWCNTs		With MWCNTs ^c	
			Pt	Ni	Pt	Ni		
5		-	6	6.9 (1.2) ^b	100	-	100	-
15	5	-	4	4.0 (0.7) ^b	91	9	89	11
5	5	-	4	3.0 (0.5) ^b	73	27	72	28
5	-	5	4	4.9 (0.9) ^b	38	62	42	58

^a The composition of the resulting nanoparticles was calculated from EDS measurement.

^b The values in parentheses are standard deviations.

^c These values show the composition of the specimens 1-4 determined from EDS measurement.

Reference

(S1) *Handbook of X-ray photoelectron spectroscopy*; Chastain, J., King, Jr. R. C., Eds.; Physical Electronics, Inc., Minnesota, USA, 1995.