

# Supporting Information

## Nucleation and Conversion Transformations of the Transition Metal Polysulfide VS<sub>4</sub> in Lithium Ion Batteries

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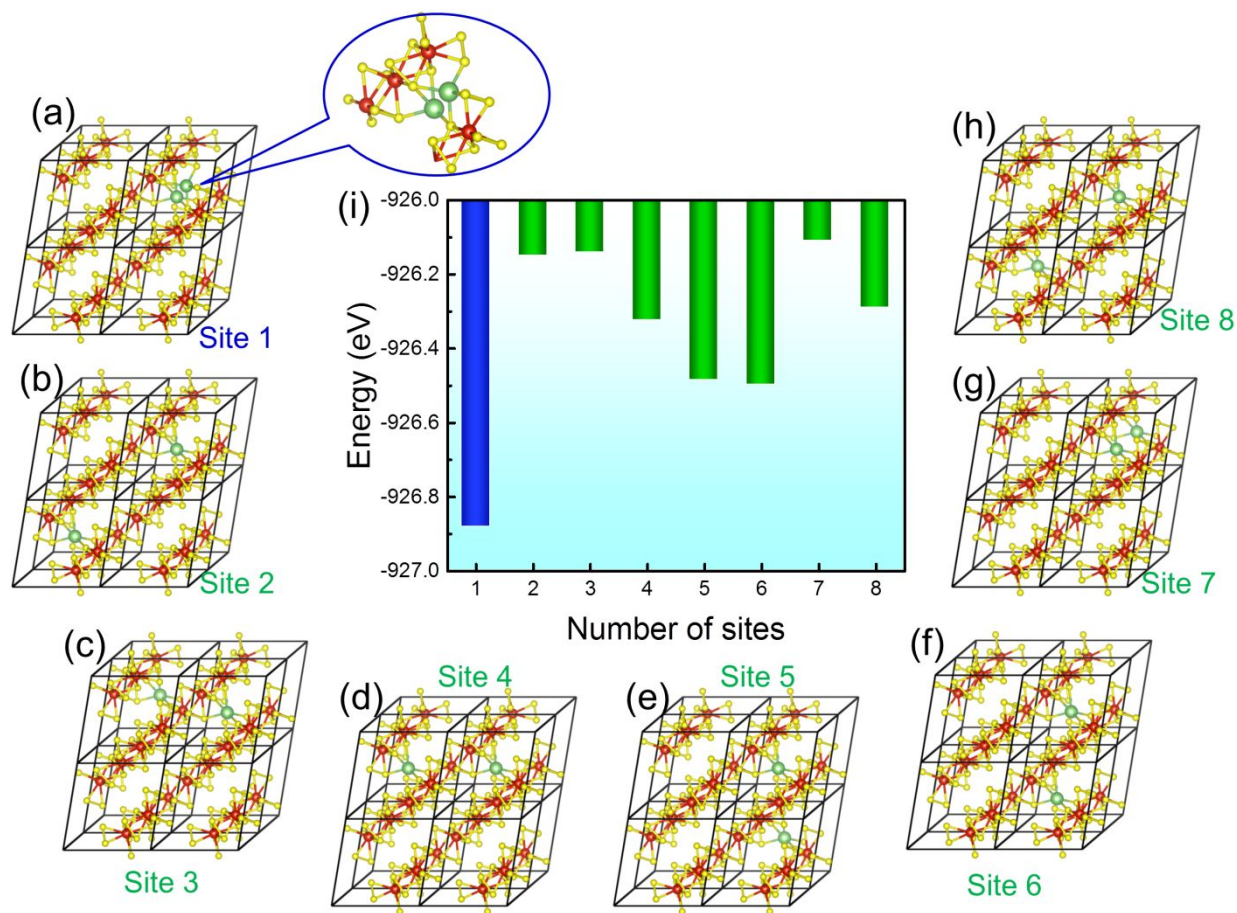
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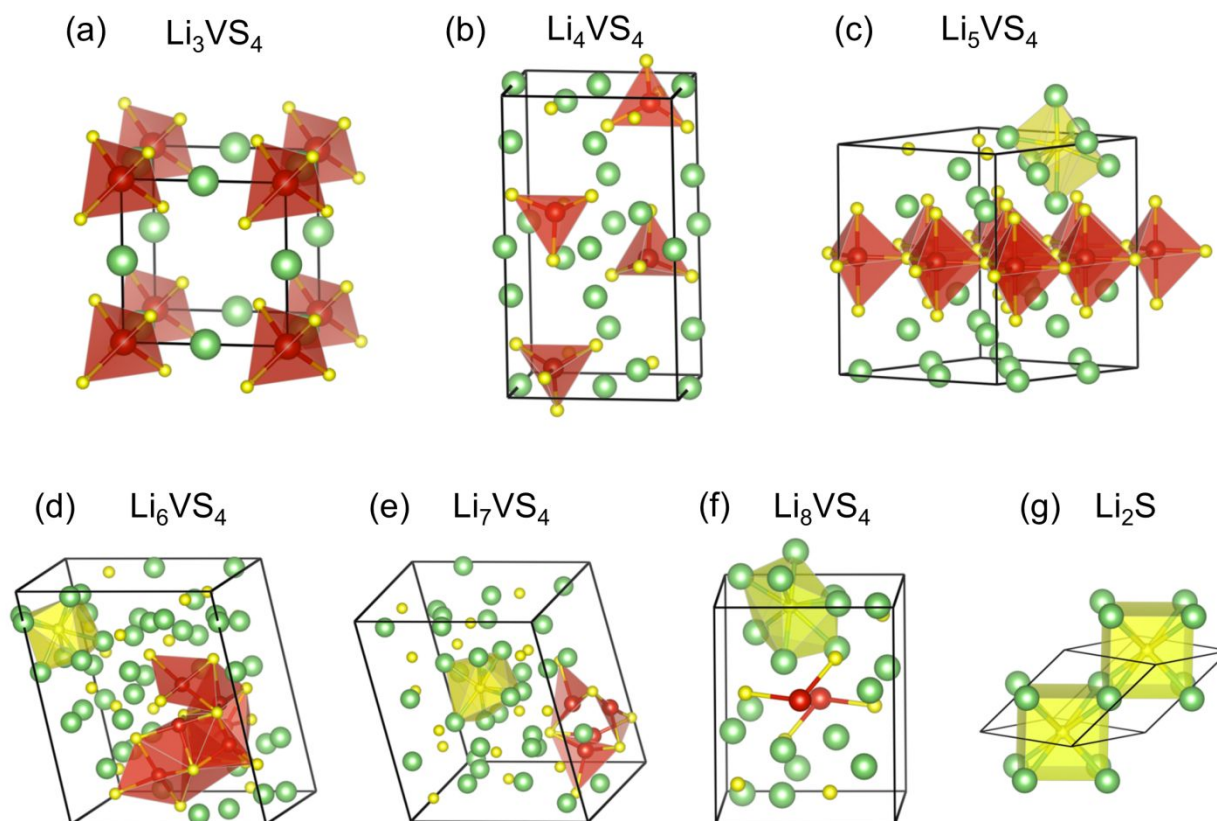
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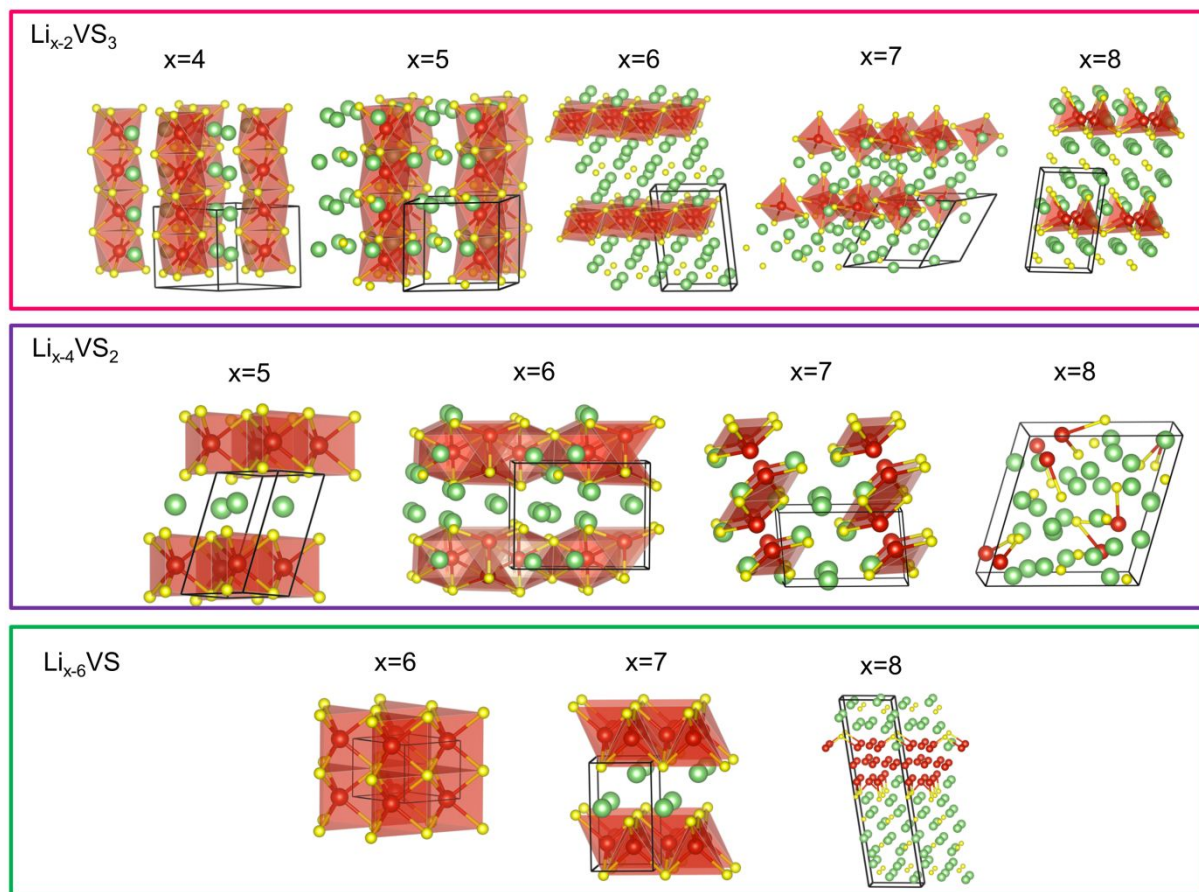
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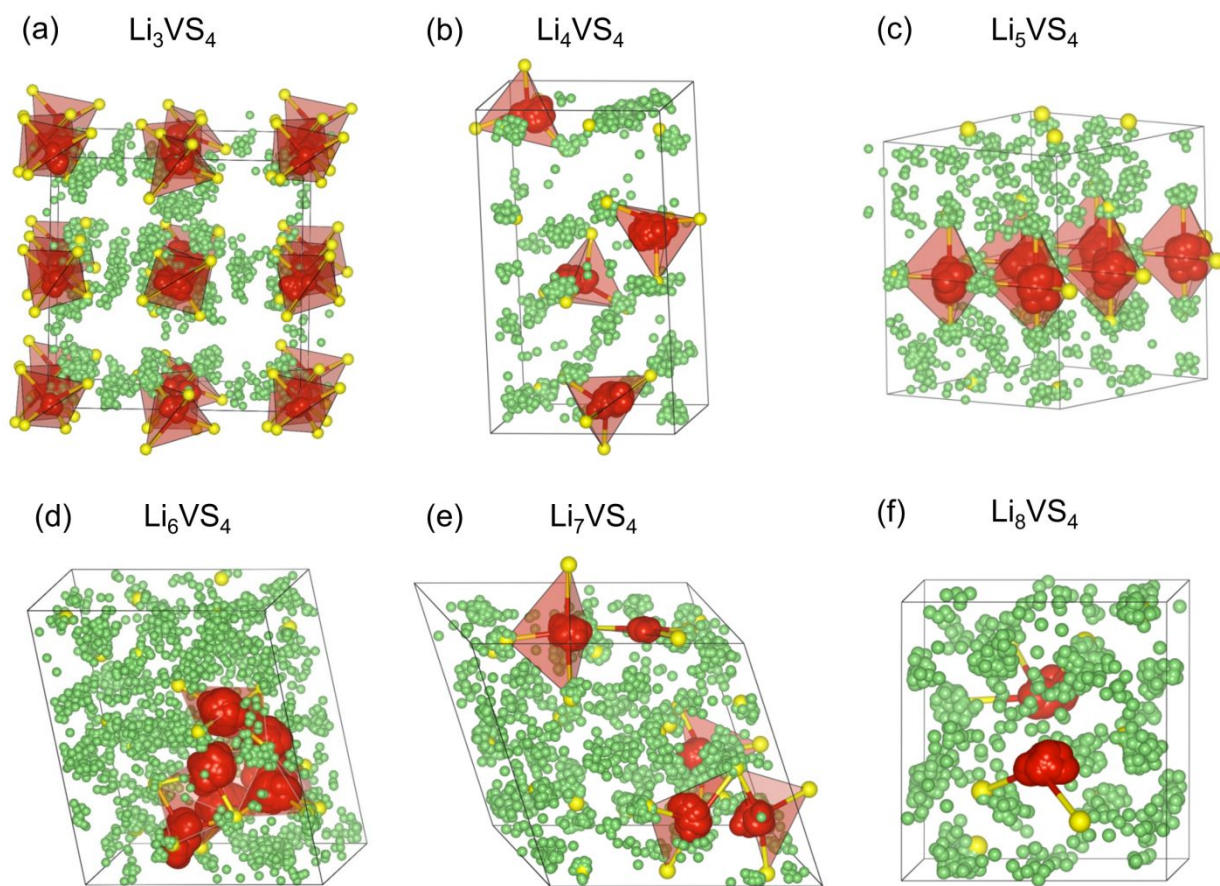
**Fig. S1.** The structures of a  $2 \times 2 \times 2$   $\text{VS}_4$  supercell with two Li atoms inserted in (a) a single unit cell; (b-h) different unit cells. (i) The total energy of these configurations. It showed that the aggregation model (Site 1) was indeed energetically favored.



**Fig. S2.** (a-f) The predicted structures of  $\text{Li}_x\text{VS}_4$  ( $3 \leq x \leq 8$ ). (g) The crystal structure of  $\text{Li}_2\text{S}$ . It can be seen that the  $\text{Li}_8\text{S}$  cluster (yellow) formed in  $\text{Li}_x\text{VS}_4$  ( $5 \leq x \leq 8$ ) was different from the  $\text{Li}_8\text{S}$  unit in  $\text{Li}_2\text{S}$  crystal.



**Fig. S3.** The predicted structures of  $\text{Li}_{x-2}\text{VS}_3$ ,  $\text{Li}_{x-4}\text{VS}_2$  and  $\text{Li}_{x-6}\text{VS}$  ( $3 \leq x \leq 8$ ).



**Fig. S4.** The trace of  $\text{Li}^+$  diffusion in the (a)  $\text{Li}_3\text{VS}_4$ , (b)  $\text{Li}_4\text{VS}_4$ , (c)  $\text{Li}_5\text{VS}_4$ , (d)  $\text{Li}_6\text{VS}_4$ , (e)  $\text{Li}_7\text{VS}_4$ , (f)  $\text{Li}_8\text{VS}_4$  structure at 800K.