Supporting Information

P-Chiral Monophosphorus Ligands for Asymmetric

Copper-Catalyzed Allylic Alkylation

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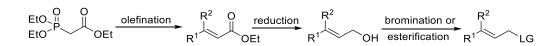
1. General Considerations

Unless otherwise noted, all commercial reagents were purchased from commercial suppliers and used without further purification. All reactions were carried out under nitrogen atmosphere in oven-dried glassware using standard Schlenk techniques. All solvents were purified and dried according to the standard methods prior to use. The ligands, copper complex and allyl substrates were prepared by following the indicated procedures described in the literature. Racemic products were synthesized by reaction of the allyl bromides with the corresponding Grignard reagent at -78 °C in dichloromethane in the presence of copper salt. Flash column chromatography was performed on silica gel (particle size 200–300 mesh, purchased from Canada) and eluted with petroleum ether/ethyl acetate.

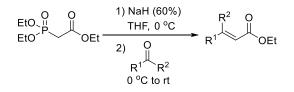
¹H NMR, ¹³C NMR, ³¹P NMR and ¹⁹F NMR spectra were recorded on a Bruker-Ultrashield PLUS400 NMR or a 500 MHz Agilent spectrometer with CDCl₃ as the solvent. ¹H chemical shifts were referenced to CDCl₃ at 7.26 ppm. ¹³C chemical shifts were referenced to CDCl₃ at 77.16 ppm and obtained with ¹H decoupling. ³¹P chemical shifts were referenced to 85% H₃PO₄ in D₂O at 0.0 ppm as external standard and obtained with ¹H decoupling. Multiplicities are abbreviated as follows: singlet (s), doublet (d), triplet (t), quartet (q), doublet-doublet (dd), triplet-doublet (td), quintet (quint), sextet (sextet), septet (septet), multiplet (m), and broad (br). MS was measured on Shimadzu LCMS-2010EV (EI) or Brukerdaltonics APEX III (HR-EI) mass spectrometers. Chiral HPLC analysis were performed on an Agilent 1200 system using chiral column described below in detail. Chiral GC analysis were measured on a Jacsco P-1010 polarimeter. Single crystals for X-ray structure determination were performed on SMART APEX CCD single-crystal diffractometer.

2. Procedure for the Preparation of (E)-Allyl Substrates

The unpurchased (*E*)-allyl substrates were synthesized from the corresponding aldehydes or ketones by a three-step Horner-Wadsworth-Emmonds olefination/DIBAL-H reduction/bromination or esterification sequence.



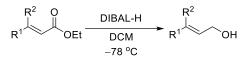
2.1 Procedure for the Horner-Wadsworth-Emmons Olefination



To a stirred suspension of NaH (60% in mineral oil, 1.6 equiv) in dry THF at 0 °C was added triethyl phosphonoacetate (1.6 equiv) dropwise and stirred for a further 1 h. The corresponding aldehydes or ketones (1.0 equiv) was dissolved in dry THF and added dropwise to the reaction mixture. After stirring for 24 h at rt, the solution was quenched with saturated NH₄Cl solution. The aqueous layer was extracted with EA and the combined organic layers were dried over anhydrous sodium sulfate, filtered and concentrated in vacuo to give the crude product as an E: Z mixture. The crude product was purified by column chromatography on silica gel with

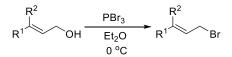
*n*hexane/EA to give the desired (*E*)-allyl esters. ¹H NMR data of all compounds matched with those reports in the literature.¹

2.2 Procedure for the DIBAL-H Reduction



To a stirred solution of the corresponding (*E*)-allyl esters (1.0 equiv) in dry DCM at -78 °C was added dropwise DIBAL-H (1.5 M solution in toluene, 2.2 equiv). After the addition was complete, the solution was warmed to rt slowly and stirred overnight. Upon complete consumption of the starting material (as indicated by TLC), the reaction mixture was cooled to -78 °C and quenched with saturated aqueous NH₄Cl solution. After stirred at rt for 30 min, the reaction mixture was treated with anhydrous sodium sulfate and the resulting suspension was further at rt for 30 min. The suspension was passed through a pad of celite and the organic phased was separated from the filtrate, dried, and concentrated to provide crude (*E*)-allyl alcohols, which was used directly in next step without further purification. ¹H NMR data of all compounds matched with those reported in the literature.²

2.3 Procedure for the Bromination Reaction



To a stirred solution of the corresponding (*E*)-allyl alcohols (1.0 equiv) in anhydrous Et_2O at 0 °C was added dropwise neat PBr₃ (1.05 equiv). After stirring at this temperature for 2 h, the reaction mixture was quenched with cold saturated aqueous NH₄Cl solution and warmed to rt. The organic layer was separated and washed with brine, dried over anhydrous sodium sulfate, filtered, and concentrated in vacuo to give the corresponding (*E*)-allyl bromides in nearly quantitative yield. NMR data of all compounds matched with those reported in the literature.³

(*E*)-3-(4'-(Trifluoromethyl)phenyl)-allylbromide (1b): 92% yield; yellow solid; ¹H NMR (500 MHz, CDCl₃) δ 7.58 (d, *J* = 7.9 Hz, 2H), 7.47 (d, *J* = 7.9 Hz, 2H), 6.67 (d, *J* = 15.6 Hz, 1H), 6.49 (dd, *J* = 15.5, 7.7 Hz, 1H), 4.15 (d, *J* = 7.6 Hz, 2H); ¹³C NMR (126 MHz, CDCl₃) δ 139.3, 132.9, 130.0 (q, *J* = 32.6 Hz), 127.8, 126.9, 125.6 (q, *J* = 3.8 Hz), 125.1, 122.9, 32.4. The ¹H NMR and ¹³C NMR spectra are in agreement with those reported in the literature.^{3a}

(*E*)-3-(4'-Bromophenyl)-allylbromide (1c): 85% yield; grey solid; ¹H NMR (500 MHz, CDCl₃) δ 7.49–7.42 (m, 2H), 7.32–7.20 (m, 2H), 6.58 (d, *J* = 15.6 Hz, 1H), 6.38 (dt, *J* = 15.5, 7.7 Hz, 1H), 4.13 (dd, *J* = 7.7, 0.9 Hz, 2H); ¹³C NMR (126 MHz, CDCl₃) δ 134.7, 133.3, 131.8, 128.2, 126.0, 122.2, 32.9. The ¹H NMR and ¹³C NMR spectra are in agreement with those reported in the literature.^{3a}

(*E*)-3-(4'-Chlorophenyl)-allylbromide (1d): 87% yield; white solid; ¹H NMR (500 MHz, CDCl₃) δ 7.33–7.27 (m, 4H), 6.60 (d, *J* = 15.6 Hz, 1H), 6.37 (dt, *J* = 15.5, 7.8 Hz, 1H), 4.14 (dd, *J* = 7.8, 0.9 Hz, 2H); ¹³C NMR (126 MHz, CDCl₃) δ 134.3, 134.0, 133.2, 128.8, 127.9, 125.8, 33.0. The ¹H NMR and ¹³C NMR spectra are in

agreement with those reported in the literature.^{3a}

(*E*)-3-(4'-Methylphenyl)-allylbromide (1e): 82% yield; grey solid; ¹H NMR (500 MHz, CDCl₃) δ 7.28 (t, *J* = 8.9 Hz, 2H), 7.14 (d, *J* = 7.0 Hz, 2H), 6.62 (d, *J* = 15.5 Hz, 1H), 6.35 (dt, *J* = 15.2, 7.5 Hz, 1H), 4.17 (d, *J* = 7.5 Hz, 2H), 2.35 (s, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 138.3, 134.5, 133.0, 129.3, 126.7, 124.2, 33.8, 21.3. The ¹H NMR and ¹³C NMR spectra are in agreement with those reported in the literature.^{3c}

(*E*)-3-(3'-Bromophenyl)-allylbromide (1f): 89% yield; pale yellow oil; ¹H NMR (500 MHz, CDCl₃) δ 7.53 (s, 1H), 7.39 (d, *J* = 7.9 Hz, 1H), 7.29 (d, *J* = 7.8 Hz, 1H), 7.19 (t, *J* = 7.8 Hz, 1H), 6.56 (d, *J* = 15.6 Hz, 1H), 6.39 (dt, *J* = 15.5, 7.7 Hz, 1H), 4.13 (d, *J* = 7.7 Hz, 2H); ¹³C NMR (126 MHz, CDCl₃) δ 137.9, 132.9, 131.2, 130.1, 129.5, 126.8, 125.4, 122.8, 32.6. The ¹H NMR and ¹³C NMR spectra are in agreement with those reported in the literature.^{3c}

(*E*)-3-(3'-Chlorophenyl)-allylbromide (1g): 92% yield; pale yellow oil; ¹H NMR (500 MHz, CDCl₃) δ 7.37 (s, 1H), 7.25 (s, 3H), 6.58 (d, J = 15.6 Hz, 1H), 6.40 (dt, J = 15.5, 7.7 Hz, 1H), 4.14 (d, J = 7.7 Hz, 2H); ¹³C NMR (126 MHz, CDCl₃) δ 137.6, 134.6, 133.0, 129.9, 128.3, 126.7, 126.6, 124.9, 32.7. The ¹H NMR and ¹³C NMR spectra are in agreement with those reported in the literature.^{3c}

(*E*)-3-(3'-Methylphenyl)-allylbromide (1h): 85% yield; yellow oil; ¹H NMR (500 MHz, CDCl₃) δ 7.30–7.21 (m, 3H), 7.13 (d, *J* = 7.2 Hz, 1H), 6.64 (d, *J* = 15.6 Hz, 1H), 6.42 (dt, *J* = 15.6, 7.8 Hz, 1H), 4.19 (d, *J* = 7.8 Hz, 2H), 2.39 (s, 3H); 13C NMR (126 MHz, CDCl₃) δ 138.2, 135.8, 134.7, 129.2, 128.6, 127.5, 125.0, 124.0, 33.7, 21.4. The ¹H NMR and ¹³C NMR spectra are in agreement with those reported in the literature.^{3c}

(*E*)-3-(Cyclohexyl)-allylbromide (1i): 83% yield; colorless oil; ¹H NMR (500 MHz, CDCl₃) δ 5.71 (dd, *J* = 15.3, 6.4 Hz, 1H), 5.68–5.58 (m, 1H), 3.95 (d, *J* = 7.3 Hz, 2H), 1.99 (ddd, *J* = 11.2, 9.6, 3.2 Hz, 1H), 1.78–1.69 (m, 4H), 1.68–1.61 (m, 1H), 1.31–1.21 (m, 2H), 1.20–1.13 (m, 1H), 1.12–1.01 (m, 2H); ¹³C NMR (126 MHz, CDCl₃) δ 142.1, 123.9, 40.2, 33.4, 32.4, 26.0, 25.6. The ¹H NMR and ¹³C NMR spectra are in agreement with those reported in the literature.^{3a}

(*E*)-3-(2'-Methylphenyl)-allylbromide (11): 80% yield; yellow oil; ¹H NMR (500 MHz, CDCl₃) δ 7.45 (dd, *J* = 8.5, 4.5 Hz, 1H), 7.24–7.06 (m, 3H), 6.88 (d, *J* = 15.4 Hz, 1H), 6.30 (dt, *J* = 15.5, 7.8 Hz, 1H), 4.19 (dd, *J* = 7.8, 0.9 Hz, 2H), 2.37 (s, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 135.8, 134.8, 132.2, 130.4, 128.2, 126.5, 126.2, 125.9, 33.6. The ¹H NMR and ¹³C NMR spectra are in agreement with those reported in the literature.^{3a}

(*E*)-3-(1'-Naphthyl)-allylbromide (1m): 86% yield; grey solid; ¹H NMR (500 MHz, CDCl₃) δ 8.11 (d, *J* = 8.1 Hz, 1H), 7.87 (d, *J* = 7.8 Hz, 1H), 7.82 (d, *J* = 8.2 Hz, 1H), 7.63 (d, *J* = 7.1 Hz, 1H), 7.57–7.50 (m, 2H), 7.47 (t, *J* = 7.4 Hz, 1H), 7.42 (d, *J* = 15.3 Hz, 1H), 6.45 (dtd, *J* = 8.8, 7.7, 1.1 Hz, 1H), 4.28 (d, *J* = 7.7 Hz, 2H); ¹³C NMR (126 MHz, CDCl₃) δ 133.6, 133.4, 131.6, 131.1, 128.7, 128.6, 128.3, 126.3, 125.9, 125.6, 124.3, 123.5, 33.4. The ¹H NMR and ¹³C NMR spectra are in agreement with those reported in the literature.^{3a}

(*E*)-3-(2'-Naphthyl)-allylbromide (1n): 85% yield; white solid; ¹H NMR (500 MHz, CDCl₃) δ 7.81 (t, *J* = 7.4 Hz, 3H), 7.75 (s, 1H), 7.59 (dd, *J* = 8.5, 1.1 Hz, 1H), 7.53–7.40 (m, 2H), 6.81 (d, *J* = 15.6 Hz, 1H), 6.53 (dt, *J* = 15.6, 7.8 Hz, 1H), 4.22 (d, *J* = 7.8 Hz, 2H); ¹³C NMR (126 MHz, CDCl₃) δ 134.7, 133.4, 133.3, 133.2, 128.4, 128.1, 127.7, 127.1, 126.4, 126.3, 125.5, 123.5, 33.6. The ¹H NMR and ¹³C NMR spectra are in agreement with those reported in the literature.^{3c}

(*E*)-(4-Bromobut-2-en-2-yl)benzene (10): 85% yield; pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ 7.43–7.37 (m, 2H), 7.37–7.30 (m, 2H), 7.30–7.23 (m, 1H), 6.08 (ddd, *J* = 8.5, 5.0, 1.3 Hz, 1H), 4.20 (d, *J* = 8.6 Hz, 2H), 2.14 (s, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 142.7, 141.5, 128.3, 127.8, 125.9, 122.8, 29.4, 15.6. The ¹H NMR and ¹³C NMR spectra are in agreement with those reported in the literature.^{3e}

(*E*)-1'-Bromo-4'-(4-bromobut-2-en-2-yl)benzene (1p): 88% yield; pale yellow oil; ¹H NMR (500 MHz, CDCl₃) δ 7.46 (d, *J* = 8.5 Hz, 2H), 7.27 (d, *J* = 8.3 Hz, 3H), 6.08 (t, *J* = 8.5 Hz, 1H), 4.17 (d, *J* = 8.5 Hz, 2H), 2.11 (s, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 141.0, 140.3, 131.4, 127.5, 123.4, 121.8, 28.9, 15.5. The ¹H NMR and ¹³C NMR spectra are in agreement with those reported in the literature.^{3e}

(*E*)-1'-Chloro-4'-(4-bromobut-2-en-2-yl)benzene (1q): 80% yield; pale yellow oil; ¹H NMR (500 MHz, CDCl₃) δ 7.35–7.28 (m, 4H), 6.07 (td, *J* = 8.5, 1.2 Hz, 1H), 4.18 (d, *J* = 8.5 Hz, 2H), 2.12 (s, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 140.6, 140.2, 133.6, 128.5, 127.2, 123.3, 29.0, 15.6. The ¹H NMR and ¹³C NMR spectra are in agreement with those reported in the literature.^{3b}

(*E*)-1'-Trifluoromethyl-4'-(4-bromobut-2-en-2-yl)benzene (1r): 91% yield; white solid; ¹H NMR (500 MHz, CDCl₃) δ 7.59 (d, J = 8.4 Hz, 2H), 7.50 (d, J = 8.1 Hz, 2H), 6.14 (t, J = 8.4 Hz, 1H), 4.18 (d, J = 8.5 Hz, 2H), 2.16 (s, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 145.7, 140.0, 129.8–129.3 (m), 127.4, 126.2, 125.4–125.1 (m), 124.8, 123.0, 28.4, 15.5. The ¹H NMR and ¹³C NMR spectra are in agreement with those reported in the literature.^{3f}

(*E*)-1'-Methyl-4'-(4-Bromobut-2-en-2-yl) benzene (1s): 80% yield; colourless oil; ¹H NMR (500 MHz, CDCl₃) δ 7.33 (d, *J* = 8.1 Hz, 2H), 7.21–7.19 (m, 1H), 7.19–7.15 (m, 2H), 6.09 (dd, *J* = 8.5, 7.5 Hz, 1H), 4.22 (d, *J* = 8.6 Hz, 2H), 2.37 (s, 3H), 2.15 (s, 3H). The ¹H NMR spectra is in agreement with those reported in the literature.^{3b}

(*E*)-1'-Methyl-3'-(4-bromobut-2-en-2-yl)benzene (1t): 82% yield; yellow oil; ¹H NMR (500 MHz, CDCl₃) δ 7.28–7.19 (m, 3H), 7.13 (d, *J* = 7.0 Hz, 1H), 6.10 (t, *J* = 8.5 Hz, 1H), 4.22 (d, *J* = 8.6 Hz, 2H), 2.39 (s, 3H), 2.16 (s, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 142.2, 141.6, 137.9, 128.6, 128.2, 126.7, 123.1, 122.6, 29.6, 21.5, 15.7. The ¹H NMR and ¹³C NMR spectra are in agreement with those reported in the literature.^{3b}

(*E*)-1'-Bromo-3'-(4-bromobut-2-en-2-yl)benzene (1u): 90% yield; pale yellow oil; ¹H NMR (500 MHz, CDCl₃) δ 7.54 (d, *J* = 7.9 Hz, 1H), 7.30–7.24 (m, 1H), 7.13 (ddd, *J* = 15.5, 7.9, 1.5 Hz, 2H), 5.72 (td, *J* = 8.4, 1.3 Hz, 1H), 4.13 (t, *J* = 6.5 Hz, 2H), 2.08 (s, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 144.6, 142.5, 132.8, 129.6, 128.7, 127.3, 126.0, 121.8, 28.2, 17.5. The ¹H NMR and ¹³C NMR spectra are in agreement with those reported in the literature.^{3e}

(*E*)-1'-Chloro-3'-(4-bromobut-2-en-2-yl)benzene (1v): 89% yield; yellow oil; ¹H NMR (500 MHz, CDCl₃) δ 7.38 (s, 1H), 7.29–7.22 (m, 3H), 6.08 (t, *J* = 8.5 Hz, 1H), 4.16 (d, *J* = 8.5 Hz, 2H), 2.11 (s, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 144.0, 140.0, 134.3, 127.8, 126.1, 124.1, 123.9, 28.9, 15.5. The ¹H NMR and ¹³C NMR spectra are in agreement with those reported in the literature.^{3d}

(*E*)-1'-Trifluoromethyl-3'-(4-bromobut-2-en-2-yl)benzene (1w): 90% yield; yellow oil; ¹H NMR (500 MHz, CDCl₃) δ 7.65 (s, 1H), 7.58 (d, *J* = 7.7 Hz, 1H), 7.54 (d, *J* = 7.4 Hz, 1H), 7.46 (t, *J* = 7.8 Hz, 1H), 6.14 (t, *J* = 8.4 Hz, 1H), 4.19 (d, *J* = 8.5 Hz, 2H), 2.17 (s, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 143.0, 140.0, 129.2 (d, *J* = 1.1 Hz), 128.8, 124.4 (q, *J* = 3.4 Hz), 122.7 (q, *J* = 3.8 Hz), 28.4, 15.3. The ¹H NMR and ¹³C NMR spectra are in agreement with those reported in the literature.^{3b}

(*E*)-1'-Bromo-2'-(4-bromobut-2-en-2-yl)benzene (1x): 83% yield; pale yellow oil; ¹H NMR (500 MHz, CDCl₃) δ 7.54 (d, *J* = 7.9 Hz, 1H), 7.29–7.25 (m, 1H), 7.13 (ddd, *J* = 15.5, 7.9, 1.5 Hz, 2H), 5.72 (td, *J* = 8.4, 1.3 Hz, 1H), 4.13 (d, *J* = 8.4 Hz, 2H), 2.08 (s, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 144.7, 142.6, 132.9, 129.6, 128.7, 127.3, 126.0, 121.8, 28.2, 17.5. The ¹H NMR and ¹³C NMR spectra are in agreement with those reported in the literature.^{3b}

(*E*)-1'-(4-Bromobut-2-en-2-yl)naphthalene (1y): 88% yield; pale yellow oil; ¹H NMR (500 MHz, CDCl₃) δ 8.00–7.93 (m, 1H), 7.91–7.86 (m, 1H), 7.81 (t, *J* = 7.6 Hz, 1H), 7.54–7.50 (m, 2H), 7.46 (dd, *J* = 8.1, 7.2 Hz, 1H),

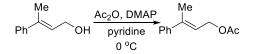
7.29 (dd, J = 7.0, 1.0 Hz, 1H), 5.93 (td, J = 8.5, 1.4 Hz, 1H), 4.28 (d, J = 8.5 Hz, 2H), 2.24 (d, J = 1.2 Hz, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 142.1, 142.0, 133.8, 130.9, 128.4, 127.6, 126.1, 126.0, 125.8, 125.4, 125.3, 124.7, 28.7, 18.8. The ¹H NMR and ¹³C NMR spectra are in agreement with those reported in the literature.^{3e}

(*E*)-2'-(4-Bromobut-2-en-2-yl)naphthalene (1z): 87% yield; white solid; ¹H NMR (500 MHz, CDCl₃) δ 7.85–7.79 (m, 4H), 7.58 (dd, *J* = 8.6, 1.8 Hz, 1H), 7.48 (ddd, *J* = 6.9, 5.7, 3.4 Hz, 2H), 6.26 (td, *J* = 8.5, 1.3 Hz, 1H), 4.27 (d, *J* = 8.5 Hz, 2H), 2.26 (d, *J* = 1.0 Hz, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 141.3, 139.3, 133.3, 132.9, 128.2, 127.9, 127.5, 126.3, 126.1, 124.9, 124.0, 123.3, 29.4, 15.6. The ¹H NMR and ¹³C NMR spectra are in agreement with those reported in the literature.^{3f}

(*E*)-(1-Bromohex-2-en-3-yl)benzene (1x'): 85% yield; colourless oil; ¹H NMR (500 MHz, CDCl₃) δ 7.36 (dd, *J* = 13.9, 6.9 Hz, 5H), 6.00 (t, *J* = 8.6 Hz, 1H), 4.22 (d, *J* = 8.6 Hz, 2H), 2.59 (dd, *J* = 16.3, 8.5 Hz, 2H), 1.42 (m, 2H), 0.91 (t, *J* = 7.2 Hz, 6H). The ¹H NMR spectra is in agreement with those reported in the literature.^{3f}

(*E*)-(1-Bromo-4-methylpent-2-en-3-yl)benzene (1y'): 82% yield; pale yellow oil; ¹H NMR (500 MHz, CDCl₃) δ 7.33–7.27 (m, 3H), 7.18 (dd, *J* = 7.9, 1.6 Hz, 2H), 5.63 (t, *J* = 8.6 Hz, 1H), 4.18 (d, *J* = 8.6 Hz, 2H), 3.15 (dt, *J* = 14.0, 7.0 Hz, 1H), 1.10 (d, *J* = 7.0 Hz, 6H); ¹³C NMR (126 MHz, CDCl₃) δ 152.9, 141.5, 128.2, 127.7, 126.9, 123.9, 29.2, 28.0, 21.6, 21.3. The ¹H NMR and ¹³C NMR spectra are in agreement with those reported in the literature.^{3e}

2.4 Procedure for the Esterification Reaction



Allylic alcohol (1.8 mg, 12.4 mmol) and DMAP (76 mg, 0.62 mmol) were dissolved in pyridine (10 mL) at 0 °C. Acetic anhydride (5 mL) was added slowly at 0 °C. The solution was stirred at rt for 24 h and quenched with saturated aqueous NaHCO₃ solution. It was then diluted with water (10 mL), extracted with CH₂Cl₂ (3×20 mL), dried over anhydrous sodium sulfate and evaporated. The residue was purified by column chromatography on silica gel with *n*hexane/EA to afford the allylic acetate as liquid (85% yield). ¹H NMR (400MHz, CDCl₃) δ 7.27–7.45 (m, 5H), 5.93 (td, *J* = 1.2, 6.9 Hz, 1H), 4.82 (d, *J* = 6.9 Hz, 2H), 2.14 (s, 3H), 2.12 (s, 3H).

To a mixture of pyridine (2.4 mL, 30.0 mmol, 3.0 equiv), the allylic alcohol (1.5 g, 10.0 mmol, 1.0 equiv) and DCM (10 mL) was added benzyl chloroformate (2.8 mL, 20.0 mmol, 2.0 equiv) dropwise at 0 °C. The reaction mixture was allowed to warm to rt and stirred overnight (monitored by TLC). The mixture was then quenched with water (10 mL) and diluted with EA (10 mL). The organic layer was separated, washed sequentially with HCl solution (1M, 20 mL) and brine (20 mL), concentrated, and purified by column chromatography with *n*hexane/EA to give the allylic benzyl carbonate as a colorless oil (89% yield). ¹H NMR (400 MHz, CDCl₃) δ 7.39 (t, *J* = 1.8 Hz, 1H), 7.37 (s, 2H), 7.35 (t, *J* = 2.1 Hz, 2H), 7.34–7.32 (m, 2H), 7.32–7.28 (m, 2H), 7.28–7.25 (m, 1H), 5.90 (dd, *J* = 7.6, 6.4 Hz, 1H), 5.16 (s, 2H), 4.85 (d, *J* = 7.0 Hz, 2H), 2.10 (s, 3H).

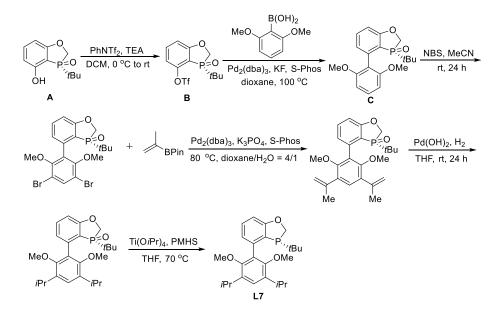
$$\begin{array}{cccc} Me & Et_3N, DMAP & Me & O\\ \hline & (MeO)_2P(O)Cl & Ph & OP'(MeO)_2\\ \hline & DCM, 0 \ ^{\circ}C \ to \ rt & Ph & OP'(MeO)_2\\ \end{array}$$

To a solution of allylic alcohol (1 g, 6.8 mmol, 1.0 equiv) and Et₃N (1.4 ml, 10.1 mmol, 1.5 equiv) in DCM (15 mL), (MeO)₂P(O)Cl (0.9 ml, 8.1 mmol, 1.2 equiv) and DMAP (247 mg, 2.02 mmol, 0.3 equiv) were sequentially added at 0 °C. The reaction mixture was allowed to warm to rt, stirred for another 12 h and then diluted with EA (10 mL) and water (10 mL). The organic layer was separated and the aqueous layer was extracted with EA (10 mL). The combined organic layer was washed with water and brine, dried over anhydrous sodium sulfate, and concentrated to provide crude allylic phosphate as a yellow oil (80% yield). ¹H NMR (500 MHz, CDCl₃) δ 7.41 (d, *J* = 7.3 Hz, 2H), 7.34 (t, *J* = 7.4 Hz, 2H), 7.31–7.26 (m, 1H), 5.95 (td, *J* = 7.0, 1.1 Hz, 1H), 4.83–4.75 (m, 2H), 3.80 (s, 3H), 3.77 (s, 3H), 2.13 (s, 3H).

3. The Synthesis of P-Chiral Monophosphorus Ligands

Ligands L1–L6, L9 and L11 were prepared according to procedures described in our previous reports.⁴⁻⁶ The synthesis of L12–L18 was similar to the procedure for the synthesis of L5 and L6.^{6a} Ligands L7, L8, L10 and L19 were prepared according to procedures described as follows. The compound A was prepared according to a procedure described in our previous reports.⁴

3.1 The Synthesis of L7 and L8

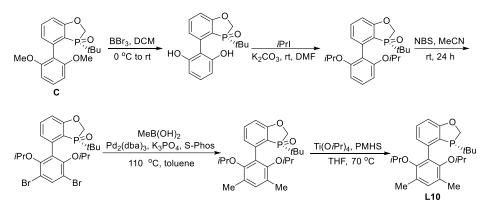


(1) To a solution of compound **A** (1.0 equiv) and triethylamine (4.0 equiv) in DCM at 0 °C was added Tf₂NPh (1.2 equiv). The mixture was stirred at rt for 5 h and then quenched with addition of water. The organic layer was separated, dried over anhydrous sodium sulfate, concentrated, and purified by column chromatography with *n*hexane/EA to give the triflate product as a white solid (94% yield). (2) To a solution of the triflate product (1.0 equiv) and 2, 6-dimethoxyphenylboronic acid (1.5 equiv), Pd₂(dba)₃ (1 mol %), S-Phos (3 mol %) and potassium fluoride (4.0 equiv) was charged degassed dioxane. The mixture was stirred at 100 °C under nitrogen for 10 h and then cooled to rt. The resulting mixture was removed most dioxane under a reduced pressure. And column chromatography was conducted with *n*hexane/EA to provide the coupling product as a white crystalline solid (90% yield). (3) The coupling product (1.0 equiv) and acetonitrile were charged into a flask. NBS (2.5 equiv) was added to the formed solution. The reaction mixture was stirred at rt for 16 h. Water and EA were added to the

reaction mixture. The mixture stirred for 20 min then the organic fraction is separated, washed with 2N HCl, dried over anhydrous sodium sulfate, filtered then concentrated. The crude product is purified by column chromatography with *n*hexane/EA to provide the desired bromide product as a white solid (82% yield). (4) To a solution of the bromide product (1.0 equiv) and isopropenylboronic acid pinacol ester (3.0 equiv), $Pd_2(dba)_3$ (3) mol %), S-Phos (6 mol %) and potassium phosphate (6.0 equiv) was charged degassed dioxane and water. The mixture was stirred at 80 °C under nitrogen for 18 h and then cooled to rt. The resulting mixture was removed most solvent under a reduced pressure. And column chromatography was conducted with *n*hexane/EA to provide the coupling product as a white crystalline solid (76% yield). (5) The hydrogenation of the coupling product was performed in the presence of $Pd(OH)_2$ and H_2 in THF at rt for 16 h. After concentrated, the resulting mixture was directly performed column chromatography with nhexane/EA to provide the oxidation state product as a white crystalline solid (85% yield). (6) The oxaphosphole oxide was added to a Schlenk flask equipped with magnetic stirring bar. THF was charged followed by addition of PMHS and Ti(OiPr)₄ (3.0 equiv). The reaction mixture was heated at 70 °C and monitored by ³¹P NMR spectroscopy. The reaction mixture then was cooled to rt and quenched by dropwise addition of degassed 30 % NaOH at 0 °C, the resulting mixture was further stirred at 60 °C for 0.5 h. The aqueous layer was removed and subsequently extracted under nitrogen atmosphere. The combined organics were dried over anhydrous sodium sulfate then filtered through nitrogen-purged neutral alumina with Et₂O. The solvents were removed under reduced pressure to afford ligand L7 as a colourless oil (80% yield). ¹H NMR (500 MHz, CDCl₃) δ 7.31 (t, J = 7.8 Hz, 1H), 7.13 (s, 1H), 7.03 (dd, J = 7.2, 2.9 Hz, 1H), 6.91 (dd, J = 8.1, 0.5 Hz, 1H), 4.82 (d, J = 12.5, 2.0 Hz, 1H), 4.57 (dd, J = 26.3, 12.5 Hz, 1H), 3.65 (s, 3H), 3.38 (dt, J = 13.8, 6.9 Hz, 1H), 3.32 (s, 3H), 3.25 (dt, J = 13.8, 6.9 Hz, 1H), 1.31 (d, J = 6.9 Hz, 3H), 1.27 (d, J = 7.0 Hz, 1.27 Hz, 3H), 1.23 (d, J = 6.9 Hz, 3H), 1.12 (d, J = 6.9 Hz, 3H), 0.72 (d, J = 12.2 Hz, 9H); ¹³C NMR (126 MHz, CDCl₃) δ 163.7, 153.6, 152.6, 139.7 (d, J = 17.7 Hz), 138.0, 137.3, 130.0, 129.0, 125.0 (d, J = 16.0 Hz), 123.4, 123.3 (d, J = 16.0 Hz), 123.4, 123.4 (d, J = 16.0 Hz), 123.4, 123.4 (d, J = 16.0 Hz), 123.4 (d, J = 16.0 Hz) = 4.1 Hz), 109.7, 70.6, 70.4, 62.3, 61.1, 30.74 (d, J = 19.7 Hz), 29.7, 26.9 (d, J = 14.8 Hz), 26.6 (d, J = 4.7 Hz), 24.4 (d, J = 5.5 Hz), 23.4 (d, J = 5.1 Hz); ³¹P NMR (121 MHz, CDCl₃) δ -4.8; ESI-MS: m/z 415.30 [M+H]⁺; HRMS (ESI) m/z calcd for C₂₅H₃₆O₃P [M+H]⁺: 415.2402, found: 415.2397.

The synthesis of ligand **L8** was followed according to a similar experimental procedure. **L8**: 85% yield; White solid; ¹H NMR (500 MHz, CDCl₃) δ 7.34–7.30 (m, 1H), 7.15 (s, 1H), 7.03 (ddd, J = 7.5, 3.2, 0.8 Hz, 1H), 6.92 (dd, J = 8.1, 0.8 Hz, 1H), 4.83 (dd, J = 12.5, 2.0 Hz, 1H), 4.57 (dd, J = 26.3, 12.5 Hz, 1H), 3.66 (s, 3H), 3.44–3.37 (m, 1H), 3.34 (s, 3H), 3.28 (dd, J = 17.4, 9.9 Hz, 1H), 2.11–1.94 (m, 4H), 1.89–1.77 (m, 4H), 1.74–1.60 (m, 6H), 1.43–1.26 (m, 2H), 0.73 (d, J = 12.2 Hz, 9H); ¹³C NMR (126 MHz, CDCl₃) δ 163.7, 154.4, 153.3, 139.71 (d, J = 17.7 Hz), 135.3, 134.8, 130.0, 128.9, 124.7, 123.3 (d, J = 4.1 Hz), 109.7, 70.7, 70.4, 62.4, 61.2, 38.6 (d, J = 12.0 Hz), 35.4, 34.7, 34.5, 34.2, 30.8 (d, J = 19.8 Hz), 26.9 (d, J = 14.7 Hz), 25.6 (dd, J = 17.3, 12.0 Hz); ³¹P NMR (162 MHz, CDCl₃) δ -6.1; ESI-MS: m/z 467.30 [M+H]⁺; HRMS (ESI) m/z calcd for C₂₉H₄₀O₃P [M+H]⁺: 467.2715, found: 467.2703.

3.2 The Synthesis of L10

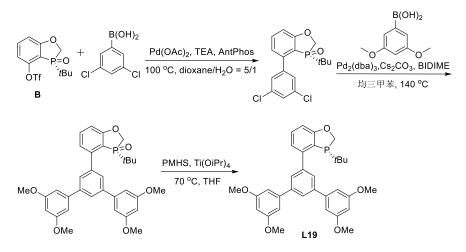


The synthesis of compound \mathbf{B} was followed by the above description.

(1) Compound **B** (1.0 equiv) was placed in a dry Schlenk tube, then DCM was added and stirred to completely dissolve the material. The reaction system was transferred to an ice water bath and BBr₃ (2.5 equiv, 1.0 M in DCM) was slowly added at 0 °C. After the completion of the dropwise addition, the reaction system was warmed to room temperature, stirred overnight and monitored by TLC. The reaction was quenched by adding MeOH. The residue was purified by column chromatography with *n*hexane/EA to give a white crystal product (91% yield). (2) The white crystal product (1.0 equiv) and anhydrous K_2CO_3 (3.0 equiv) were placed in a dry Schlenk tube, DMF was added and stirred to dissolve the materials. iPrI (5.0 equiv) was then added to the reaction system. After complete conversion, the reaction mixture was concentrated and column chromatography was conducted with nhexane/EA to provide the product as a white solid (94% yield). (3) The product (1.0 equiv) and acetonitrile were charged into a flask. NBS (2.5 equiv) was added to the formed solution. The reaction mixture was stirred at rt for 16 h. Water and EA were added to the reaction mixture. The mixture stirred for 20 min then the organic fraction is separated, washed with 2N HCl, dried over anhydrous sodium sulfate, filtered then concentrated. The crude product is purified by column chromatography with *n*hexane/EA to provide the desired bromide product as a white solid (85% yield). (4) To a solution of the bromide product (1.0 equiv) and methyl-boric acid (3.0 equiv), $Pd_2(dba)_3$ (3 mol %), S-Phos (6 mol %) and potassium phosphate (6.0 equiv) was charged dry toluene. The mixture was stirred at 110 °C under nitrogen for 18 h and then cooled to rt. The resulting mixture was removed most solvent under a reduced pressure and filtered over celite. After concentrated, column chromatography was conducted with *n*hexane/EA to provide the coupling product as a white crystalline solid (76% yield). (5) The oxaphosphole oxide was added to a Schlenk flask equipped with magnetic stirbar. THF was charged followed by addition of PMHS and Ti(OiPr)₄ (3.0 equiv). The reaction mixture was heated at 70 °C and monitored by 31 P NMR spectroscopy. The reaction mixture then was cooled to rt and quenched by dropwise addition of degassed 30 % NaOH at 0 °C, the resulting mixture was further stirred at 60 °C for 0.5 h. The aqueous layer was removed and subsequently extracted under nitrogen atmosphere. The combined organics were dried over anhydrous sodium sulfate then filtered through nitrogen-purged neutral alumina with Et₂O. The solvents were removed under reduced pressure to afford ligand L10 as a colourless oil (87% yield). ¹H NMR (500 MHz, CDCl₃) δ 7.30 (t, J = 7.8 Hz, 1H), 7.16 (dd, J = 7.1, 3.6, 1H), 6.97 (s, 1H), 6.90 (dd, J = 8.0, 0.8 Hz, 1H), 4.86 (dd, J = 12.5, 1.9 Hz, 1H), 4.54 (dd, J = 26.4, 12.5 Hz, 1H), 3.88 (dt, J = 12.2, 6.1 Hz, 1H), 3.61 (dq, J = 12.3, 6.1 Hz, 1H), 2.27 (s, 3H), 2.22 (s, 3H), 1.07 (d, J = 6.1 Hz, 3H), 1.03 (d, J = 6.1 Hz, 3H), 0.98 (d, J = 6.1 Hz, 3H), 0.82 (d, J = 6.2 Hz, 3H), 0.78 (d, J = 12.2 Hz, 9H); ¹³C NMR (126 MHz, CDCl₃) δ 163.9, 152.2, 151.8, 140.1, 139.9, 131.8, 130.2,

128.9, 127.3, 126.6, 124.3 (d, J = 4.3 Hz), 109.8, 75.4, 73.9, 70.5, 70.3, 30.8, 30.6, 27.1 (d, J = 14.9 Hz), 22.7 (d, J = 2.4 Hz), 22.4, 22.0, 17.1, 16.8; ³¹P NMR (162 MHz, CDCl₃) δ -4.8; ESI-MS: m/z 415.30 [M+H]⁺; HRMS (ESI) m/z calcd for C₂₅H₃₆O₃P [M+H]⁺: 415.2402, found: 415.2397.

3.3 The Synthesis of L19



(1) To a mixture of compound **B** (1.0 equiv) and 3, 5-dichlorophenylboronic acid (2.0 equiv), $Pd(OAc)_2$ (5 mol %), Antphos (10 mol %) and TEA (3.0 equiv) was charged degassed dioxane/H₂O. The mixture was stirred under nitrogen for 24 h at 100 °C, concentrated, partitioned with water and DCM. The DCM layer was dried over anhydrous sodium sulfate, concentrated, and purified by column chromatography with *n*hexane/EA to provide pure product (55% yield). (2) To a solution of the pure product (1.0 equiv) and 3, 5-dimethoxyphenylboronic acid acid (6.0 equiv), Pd₂(dba)₃ (10 mol %), BI-DIME (20 mol %) and Cs₂CO₃ (8.0 equiv) was charged degassed mesitylene. The mixture was stirred at 140 °C under nitrogen for 24 h and then cooled to rt. The resulting mixture was concentrated and purified by column chromatography with nhexane/EA to provide the oxidation state of L19 (92% yield). (3) To a solution of oxidation state of L19 (1.0 equiv) in THF at rt was added PMHS (3.0 equiv) and Ti(OiPr)₄ (3.0 equiv). The reaction mixture was heated at 70 °C and monitored by ³¹P NMR spectroscopy. The reaction mixture then was cooled to rt and quenched by dropwise addition of degassed 30 % NaOH at 0 °C, the resulting mixture was further stirred at 60 °C for 0.5 h. The aqueous layer was removed and subsequently extracted under nitrogen atmosphere. The combined organics were dried over anhydrous sodium sulfate then filtered through nitrogen-purged neutral alumina with Et₂O. The solvents were removed under reduced pressure to afford ligand L19 as a white solid (75% yield). ¹H NMR (500 MHz, CDCl₃) δ 7.97 (d, J = 1.6 Hz, 2H), 7.74 (t, J = 1.6 Hz, 1H), 7.41–7.37 (m, 1H), 7.13 (dd, J = 6.7, 3.5 Hz, 1H), 6.96 (dd, J = 8.1, 0.8 Hz, 1H), 6.87 (d, J = 2.3Hz, 4H), 6.51 (s, 2H), 4.88 (dd, J = 12.6, 1.9 Hz, 1H), 4.63 (dd, J = 26.0, 12.6 Hz, 1H), 3.88 (s, 12H), 0.69 (d, J = 12.6, 1.9 Hz, 1H), 4.63 (dd, J = 26.0, 12.6 Hz, 1H), 3.88 (s, 12H), 0.69 (d, J = 12.6, 1.9 Hz, 1H), 4.63 (dd, J = 26.0, 12.6 Hz, 1H), 3.88 (s, 12H), 0.69 (d, J = 12.6, 1.9 Hz, 1H), 4.63 (dd, J = 26.0, 12.6 Hz, 1H), 3.88 (s, 12H), 0.69 (d, J = 12.6, 1.9 Hz, 1H), 4.63 (dd, J = 26.0, 12.6 Hz, 1H), 3.88 (s, 12H), 0.69 (d, J = 12.6, 1.9 Hz, 1H), 4.63 (dd, J = 26.0, 12.6 Hz, 1H), 3.88 (s, 12H), 0.69 (d, J = 12.6, 1.9 Hz, 1H), 4.63 (dd, J = 12.1 Hz, 9H); ¹³C NMR (126 MHz, CDCl₃) δ 166.0 (d, J = 19.2 Hz), 161.2, 146.3 (d, J = 5.5 Hz), 142.9, 141.9, 141.5 (d, J = 1.5 Hz), 135.0, 127.6, 125.7, 123.3 (d, J = 7.9 Hz), 113.0 (d, J = 5.2 Hz), 105.6, 99.7, 65.7 65.2, 55.6, 34.3, 33.8, 24.1; ³¹P NMR (162 MHz, CDCl₃) δ -10.6; ESI-MS: m/z 543.20 [M+H]⁺; HRMS (ESI) m/z calcd for C₃₃H₃₆O₅P [M+H]⁺: 543.2300, found: 543.2302.

3.4 Characterization Data of New P-Chiral Monophosphorus Ligands

L12: 85% yield; White solid; ¹H NMR (500 MHz, CDCl₃) δ 7.34 (t, *J* = 7.8 Hz, 1H), 7.29 (s, 1H), 7.22 (s, 2H), 7.15 (s, 2H), 6.94 (dd, *J* = 11.4, 6.0 Hz, 4H), 4.88 (d, *J* = 12.3 Hz, 1H), 4.59 (dd, *J* = 25.5, 12.6 Hz, 1H), 3.36 (s, 2H), 6.94 (dd, *J* = 11.4, 6.0 Hz, 4H), 4.88 (d, *J* = 12.3 Hz, 1H), 4.59 (dd, *J* = 25.5, 12.6 Hz, 1H), 3.36 (s, 2H), 6.94 (dd, *J* = 11.4, 6.0 Hz, 4H), 4.88 (dd, *J* = 12.3 Hz, 1H), 4.59 (dd, *J* = 25.5, 12.6 Hz, 1H), 3.36 (s, 2H), 6.94 (dd, *J* = 11.4, 6.0 Hz, 4H), 4.88 (dd, *J* = 12.3 Hz, 1H), 4.59 (dd, *J* = 25.5, 12.6 Hz, 1H), 3.36 (s, 2H), 6.94 (dd, *J* = 11.4, 6.0 Hz, 4H), 4.88 (dd, *J* = 12.3 Hz, 1H), 4.59 (dd, *J* = 25.5, 12.6 Hz, 1H), 5.36 (s, 2H), 6.94 (dd, *J* = 11.4, 6.0 Hz, 4H), 4.88 (dd, *J* = 12.3 Hz, 1H), 4.59 (dd, *J* = 25.5, 12.6 Hz, 1H), 5.36 (s, 2H), 6.94 (dd, *J* = 12.4 Hz, 1H), 6.94 (dd, *J* = 12.3 Hz, 1H), 5.9 (dd, *J* = 25.5, 12.6 Hz, 1H), 5.9 (dd, *J* = 25.5, 12.6 Hz, 1H), 5.9 (dd, J = 12.3 Hz, 1H), 5.9 (dd, J = 12.5 Hz, 1H), 5.9 (dd, J = 1

3H), 3.26 (s, 3H), 2.35 (d, J = 3.7 Hz, 12H), 0.91 (d, J = 12.1 Hz, 9H); ¹³C NMR (126 MHz, CDCl₃) δ 163.7, 155.3, 154.5, 139.6 (d, J = 18.1 Hz), 138.8, 138.6, 137.7, 137.5, 133.0, 130.8, 130.2 (d, J = 9.2 Hz), 128.5, 127.0, 126.5, 123.1 (d, J = 3.9 Hz), 109.7, 61.2, 60.9, 31.9, 30.9 (d, J = 19.2 Hz), 29.7 (d, J = 4.8 Hz), 29.3, 27.0 (d, J = 14.6 Hz), 22.7, 21.4, 14.1; ³¹P NMR (121 MHz, CDCl₃) δ -11.1; ESI-MS: m/z 539.40 [M+H]⁺; HRMS (ESI) m/z calcd for C₃₅H₄₀O₃P [M+H]⁺: 539.2715, found: 539.2716.

L13: 80% yield; White solid; ¹H NMR (500 MHz, CDCl₃) δ 7.44 (d, J = 1.7 Hz, 2H), 7.43 (s, 1H), 7.42 (d, J = 1.6 Hz, 2H), 7.39 (d, J = 1.8 Hz, 2H), 7.35 (d, J = 7.8 Hz, 1H), 6.98–6.96 (m, 1H), 6.94 (d, J = 8.1 Hz, 1H), 4.87 (d, J = 12.5 Hz, 1H), 4.54 (dd, J = 25.6, 12.6 Hz, 1H), 3.33 (s, 3H), 3.32 (s, 3H), 1.35 (s, 18H), 1.34 (s, 18H), 0.93 (d, J = 12.2 Hz, 9H); ¹³C NMR (126 MHz, CDCl₃) δ 163.7, 155.3, 154.6, 150.7, 150.4, 140.2, 140.0, 138.0, 137.7, 132.9, 131.4, 130.3 (d, J = 10.2 Hz), 130.0, 123.6, 123.0 (d, J = 18.9 Hz), 120.7(d, J = 19.2 Hz), 109.7, 70.8, 70.6, 61.1(d, J = 5.4 Hz), 60.9, 34.9 (d, J = 2.0 Hz), 31.5 (d, J = 7.6 Hz), 30.9 (d, J = 19.4 Hz), 27.0 (d, J = 14.7 Hz); ³¹P NMR (162 MHz, CDCl₃) δ -9.3; ESI-MS: m/z 707.45 [M+H]⁺; HRMS (ESI) m/z calcd for C₄₇H₆₄O₃P [M+H]⁺: 707.4593, found: 707.4574.

L15: 83% yield; White solid; ¹H NMR (500 MHz, CDCl₃) δ 7.36–7.32 (m, 2H), 6.97–6.88 (m, 2H), 6.78 (d, *J* = 2.3 Hz, 2H), 6.70 (d, *J* = 2.3 Hz, 2H), 6.45 (t, *J* = 2.3 Hz, 2H), 4.87 (dd, *J* = 12.5, 1.6 Hz, 1H), 4.56 (dd, *J* = 25.5, 12.6 Hz, 1H), 3.82 (s, 6H), 3.80 (s, 6H), 3.42 (s, 3H), 3.30 (s, 3H), 0.89 (d, *J* = 12.1 Hz, 9H); ¹³C NMR (126 MHz, CDCl₃) δ 163.7, 160.6 (d, *J* = 19.6 Hz), 155.6, 154.9, 140.7 (d, *J* = 15.7 Hz), 139.4 (d, *J* = 18.2 Hz), 132.5, 130.5, 130.4 (d, *J* = 1.2 Hz), 130.3, 129.7, 125.0 (d, *J* = 15.7 Hz), 122.8 (d, *J* = 3.9 Hz), 109.8, 107.3, 106.7, 99.3 (d, *J*=3.3 Hz), 70.8, 70.6, 61.2, 61.0 (d, *J*=4.2 Hz), 55.4 (d, *J* = 9.4 Hz), 30.9 (d, *J* = 19.4 Hz), 26.9 (d, *J* = 14.7 Hz); ³¹P NMR (162 MHz, CDCl₃) δ -8.6; ESI-MS: m/z 603.40 [M+H]⁺; HRMS (ESI) m/z calcd for C₃₅H₄₀O₇P [M+H]⁺: 603.2512, found: 603.2513.

L16: 86% yield; White solid; ¹H NMR (500 MHz, CDCl₃) δ 7.58 (d, *J* = 8.5 Hz, 2H), 7.48 (d, *J* = 8.6 Hz, 2H), 7.33 (t, *J* = 7.8 Hz, 1H), 7.30 (s, 1H), 6.94 (t, *J* = 8.3 Hz, 6H), 4.88 (d, *J* = 12.6 Hz, 1H), 4.60 (dd, *J* = 25.6, 12.5 Hz, 1H), 3.84 (d, *J* = 2.3 Hz, 6H), 3.36 (s, 3H), 3.19 (s, 3H), 0.89 (d, *J* = 12.1 Hz, 9H); ¹³C NMR (126 MHz, CDCl₃) δ 163.6, 158.7 (d, *J* = 5.7 Hz), 155.1, 154.3, 139.5 (d, *J* = 18.1 Hz), 132.4, 131.2, 130.9, 130.5 (d, *J* = 7.6 Hz), 130.20 (d, *J* = 18.4 Hz), 129.8 (d, *J* = 16.2 Hz), 125.0 (d, *J* = 15.2 Hz), 123.0 (d, *J* = 4.0 Hz), 113.8, 113.6, 109.7, 70.7, 70.5, 61.0, 60.6 (d, *J* = 2.7 Hz), 55.2, 30.9 (d, *J* = 19.3 Hz), 29.7, 27.0 (d, *J* = 14.7 Hz); ³¹P NMR (121 MHz, CDCl₃) δ -6.9; ESI-MS: m/z 543.30 [M+H]⁺; HRMS (ESI) m/z calcd for C₃₃H₃₆O₅P [M+H]⁺: 543.2300, found: 543.2302.

L17: 82% yield; White solid; ¹H NMR (500 MHz, CDCl₃) δ 7.37 (t, *J* = 14.6, 6.8 Hz, 1H), 7.35 (s, 1H), 6.95 (dd, *J* = 11.8, 5.3 Hz, 2H), 6.83 (s, 2H), 6.79 (s, 2H), 4.89 (dd, *J* = 12.6, 1.5 Hz, 1H), 4.56 (dd, *J* = 25.5, 12.6 Hz, 1H), 3.91 (s, 3H), 3.90 (s, 3H), 3.89 (s, 6H), 3.87 (s, 6H), 3.41 (s, 3H), 3.35 (s, 3H), 0.92 (d, *J* = 12.1 Hz, 9H); ¹³C NMR (126 MHz, CDCl₃) δ 163.8, 155.3, 154.8, 153.1 (d, *J* = 16.7 Hz), 139.5 (d, *J* = 18.3 Hz), 137.2 (d, *J* = 2.1 Hz), 134.2 (d, *J* = 13.2 Hz), 132.2, 130.5, (d, *J* = 5.1 Hz), 129.7, 122.8, 110.0, 106.4, 105.8, 61.2, 61.0 (d, *J* = 6.1 Hz), 56.2 (d, *J* = 16.1 Hz), 31.0 (d, *J* = 19.1 Hz), 27.0 (d, *J* = 14.6 Hz); ³¹P NMR (162 MHz, CDCl₃) δ -9.1; ESI-MS: m/z 663.30 [M+H]⁺; HRMS (ESI) m/z calcd for C₃₇H₄₄O₉P [M+H]⁺: 663.2723, found: 663.2723.

L18: 83% yield; White solid; ¹H NMR (500 MHz, CDCl₃) δ 7.30 (d, J = 7.8 Hz, 1H), 7.22 (dd, J = 14.2, 5.9 Hz, 2H), 7.08 (dd, J = 7.1, 2.7 Hz, 1H), 6.91 (s, 1H), 6.88 (d, J = 8.0 Hz, 1H), 6.62 (dd, J = 15.0, 7.4 Hz, 4H), 4.83 (dd, J = 12.4, 2.0 Hz, 1H), 4.58 (d, J = 14.2 Hz, 1H), 3.82 (s, 3H), 3.76 (s, 3H), 3.65 (s, 3H), 3.36 (s, 3H), 3.20 (s, 3H), 0.88 (d, J = 12.0 Hz, 9H); ¹³C NMR (126 MHz, CDCl₃) δ 163.7, 158.5 (d, J = 18.8 Hz), 158.1 (d, J = 4.3 Hz), 156.3, 155.0, 139.7 (d, J = 17.1 Hz), 135.7, 130.0, 128.6, 128.3, 128.1, 124.1 (d, J = 4.1 Hz), 123.0, 122.1, 117.4 (d, J = 12.1 Hz), 109.3, 105.3, 104.3 (d, J = 12.8 Hz), 103.9, 60.6 (d, J = 6.4 Hz), 56.0, 55.7 (d, J = 2.7 Hz), 30.8 (d, J = 17.4 Hz), 27.3 (d, J = 14.3 Hz); ³¹P NMR (162 MHz, CDCl₃) δ -6.4; ESI-MS: m/z 603.30 [M+H]⁺; HRMS (ESI) m/z calcd for C₃₅H₄₀O₇P [M+H]⁺: 603.2512, found: 603.2510.

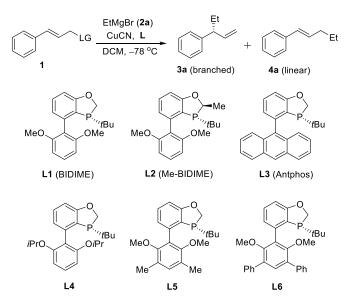
4. Optimization of Reaction Conditions

Table S1. Cu-catalyzed asymmetric allylic alkylation of cinnamyl chloride (1) with EtMgBr (2a) by employing different copper salts and (R)-BIDIME.^{*a*}

	CI EtMgBr (2a) [Cu] (R)-BIDIME DCM, −78 °C	Et + 3a (branched) 4a (linear)	Et
Entry	5 mol % [Cu]	6 mol % L	$3a:4a^b$
1			23/77
2	CuI	(R)-BIDIME	7/93
3	CuBr	(R)-BIDIME	5/95
4	CuCl	(R)-BIDIME	0/100
5	CuBr.SMe ₂	(R)-BIDIME	7/93
6	[Cu(MeCN)4]PF6	(R)-BIDIME	10/90
7	Cu(OAc) ₂	(R)-BIDIME	9/91
8	CuOAc	(R)-BIDIME	0/100
9	Cu(OTf) ₂	(R)-BIDIME	9/91
10	(CuOTf)2·toluene	(R)-BIDIME	9/91
11	(CuOTf)2·benzene	(R)-BIDIME	9/91
12	CuTc	(R)-BIDIME	0/100
13	CuCN	(R)-BIDIME	93/7
14^c	CuCN	(R)-BIDIME	80/20
15 ^d	CuCN	(R)-BIDIME	96/4

^{*a*}Unless otherwise specified, the reactions were performed under nitrogen in DCM at -78 °C in the presence of 5 mol % [Cu] and 6 mol % **L** with cinnamyl chloride (**1**, 0.25 mmol, 1.0 equiv) and EtMgBr (3 M solution in diethyl ether, 0.5 mmol, 2.0 equiv). Slow addition of EtMgBr over 0.5 h. ^{*b*}All reactions gave full conversion. The ratios of **3a**/4a were determined by ¹H NMR spectroscopy. ^{*c*}The reaction was performed under nitrogen in DCM at -78 °C in the presence of 2.5 mol % CuCN and 3 mol % **L** with cinnamyl chloride (**1**, 0.25 mmol, 1.0 equiv) and EtMgBr (3 M solution in diethyl ether, 0.5 mmol, 2.0 equiv). ^{*d*}The reaction was performed under nitrogen in DCM at -78 °C in the presence of 10 mol % CuCN and 12 mol % **L** with cinnamyl chloride (**1**, 0.25 mmol, 1.0 equiv) and EtMgBr (3 M solution in diethyl ether, 0.5 mmol, 2.0 equiv).

Table S2. The preliminary investigation of different ligands for the copper catalyzed asymmetric allylic alkylation.^{*a*}



Entry	LG	5 mol % [Cu]	6 mol % L	3a:4a ^b	$er (3a)^c$
1	Cl	CuCN	L1	93:7	53:47
2	Br	CuCN	L1	62:38	56:44
3	Cl	CuCN	L2	97:3	51:49
4	Br	CuCN	L2	60:40	53:47
5	Cl	CuCN	L3	91:9	53:47
6	Br	CuCN	L3	53:47	54:46
7	Cl	CuCN	L4	98:2	52:48
8	Br	CuCN	L4	28:72	53:47
9	Cl	CuCN	L5	93:7	64:36
10	Br	CuCN	L5	60:40	86:14
11	Cl	CuCN	L6	96:4	56:44
12	Br	CuCN	L6	72:28	85:15

^{*a*}Unless otherwise specified, the reactions were performed under nitrogen in DCM at -78 °C in the presence of **5** mol % [Cu] and 6 mol % **L** with allyl substrates (**1**, 0.25 mmol, 1.0 equiv) and EtMgBr (3 M solution in diethyl ether, 0.5 mmol, 2.0 equiv). Slow addition of EtMgBr over 0.5 h. ^{*b*}All reactions gave full conversion. The ratios of **3a/4a** were determined by ¹H NMR spectroscopy. ^{*c*}The *er* values of **3a** were determined by chiral HPLC analysis.

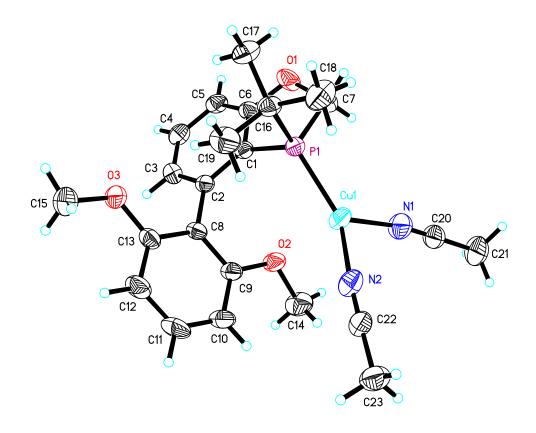
Table S3. The further optimization of reaction conditions for the copper catalyzed asymmetric allylic alkylation.^a

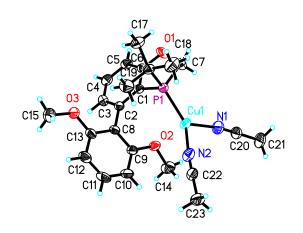
		R 1	LG [Cu], L15 solvent, T 3a (branched	+ Linear)		
Entry	R	LG	[Cu]	L	3a:4a ^b	$er (3a)^c$
1	H	Br	5 mol % CuCN	6 mol % L15	76:24	93:7
2^d	Н	Br	5 mol % CuCN	6 mol % L15	79:21	54:46
2 3 ^e	н	Br	5 mol % CuCN	6 mol % L15	87:13	53:47
4^{f}	Н	Br	5 mol % CuCN	6 mol % L15	67:33	54:46
5^g	н	Br	5 mol % CuCN	6 mol % L15	33:67	70:30
6	Н	Br	5 mol % CuTc	6 mol % L15	80:20	96:4
7	Н	Br	5 mol % [Cu(CNMe) ₄]PF ₆	6 mol % L15	70:30	94:6
8	Н	Br	5 mol % CuCl	6 mol % L15	76:24	93:7
9	Н	Br	5 mol % CuBr	6 mol % L15	71:29	93:7
10	Н	Br	5 mol % CuBr·SMe ₂	6 mol % L15	83:17	81:19
11	Н	Br	5 mol % CuI	6 mol % L15	80:20	94:6
12^{h}	Н	Br	5 mol % CuTc	6 mol % L15	74:26	93:7
13 ⁱ	Н	Br	5 mol % CuTc	6 mol % L15	71:29	86:14
14	Н	Br	2.5 mol % CuTc	3 mol % L15	80:20	96:4
15	Н	Br	1 mol % CuTc	1.2 mol % L15	82:18	95:5
16	Н	Br	0.5 mol%CuTc	0.6 mol % L15	85:15	97:3
17	Н	Br	0.1 mol%CuTc	0.12 mol % L15	60:40	96:4
18	Н	Cl	0.5 mol%CuTc	0.6 mol % L15	40:60	77:23
19	Н	Cl	0.5 mol % CuCN	0.6 mol % L15	88:12	61:39
20	Me	Br	5 mol % CuCN	6 mol % L15	74:26	68:32
21	Me	Br	5 mol % CuTc	6 mol % L15	74:26	95:5
22	Me	Br	2.5 mol % CuTc	3 mol % L15	67:33	82:18
23	Me	Br	1 mol % CuTc	1.2 mol % L15	69:31	79:21
24^{j}	Me	OAc	5 mol % CuTc	6 mol % L15		
25	Me	OCbz	5 mol % CuTc	6 mol % L15	14:86	42:58
26	Me	OPO(OMe) ₂	5 mol % CuTc	6 mol % L15	25:75	44:56

^aUnless otherwise specified, the reactions were performed under nitrogen in DCM at -78 °C in the presence of [Cu] and L15 with allyl bromide and EtMgBr (3 M solution in diethyl ether, 2.0 equiv). Slow addition of EtMgBr over 0.5 h. ^bAll reactions gave full conversion. The ratios of 3a/4a were determined by ¹H NMR spectroscopy. ^cThe er values of 3a were determined by chiral HPLC analysis. ^dThe reaction was performed in Et₂O. "The reaction was performed in MTBE. ^fThe reaction was performed in THF. ^gEtMgBr (1 M solution in THF). ^{*h*}The reaction was performed at $-40 \,^{\circ}$ C. ^{*i*}The reaction was performed at $-10 \,^{\circ}$ C. ^{*i*}No reaction.

5. X-Ray Crystal Data of Copper Complex

CD 1906544 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/conts/retrieving.html (or from the Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB21EZ, UK; fax: (+44)1223-336-033; or deposit@ccdc.cam.ac.uk).







Crystal data and structure refinement for mo_d8v17470_0m.

Identification code	mo_d8v17470_0m	
Empirical formula	C23 H29 Cu F6 N2 O3 P2	
Formula weight	620.96	
Temperature	296(2) K	
Wavelength	0.71073 Å	
Crystal system	Orthorhombic	
Space group	P 21 21 21	
Unit cell dimensions	a = 8.5246(2) Å	= 90°.
	b = 10.5054(3) Å	= 90°.
	c = 30.1709(7) Å	= 90°.
Volume	2701.93(12) Å ³	
Ζ	4	
Density (calculated)	1.527 Mg/m ³	
Absorption coefficient	0.996 mm ⁻¹	
F(000)	1272	
Crystal size	0.200 x 0.160 x 0.130 mm ³	
Theta range for data collection	2.053 to 25.499°.	
Index ranges	-9<=h<=10, -12<=k<=8, -33<=l<=36	
Reflections collected	14429	
Independent reflections	4984 [R(int) = 0.0255]	
Completeness to theta = 25.242°	99.0 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.7456 and 0.5766	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	4984 / 64 / 386	
Goodness-of-fit on F ²	1.078	
Final R indices [I>2sigma(I)]	R1 = 0.0440, wR2 = 0.1179	
R indices (all data)	R1 = 0.0523, wR2 = 0.1261	
Absolute structure parameter	-0.004(6)	
Extinction coefficient	0.018(3)	
Largest diff. peak and hole	0.258 and -0.602 e.	

6. References

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7. NMR Spectra and HPLC chromatographs

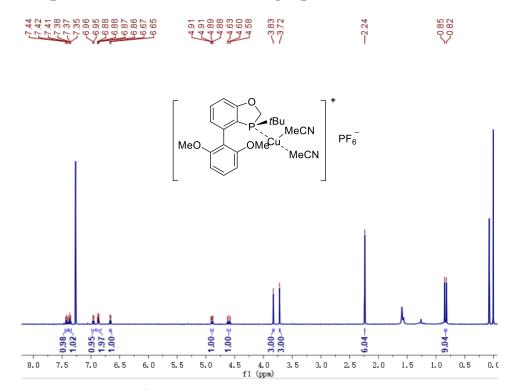


Figure S1: ¹H NMR (500 MHz, CDCl₃) spectrum of copper complex

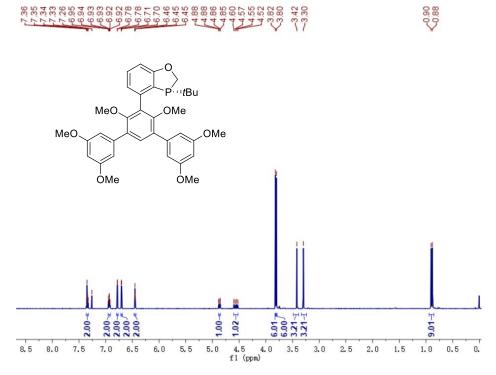
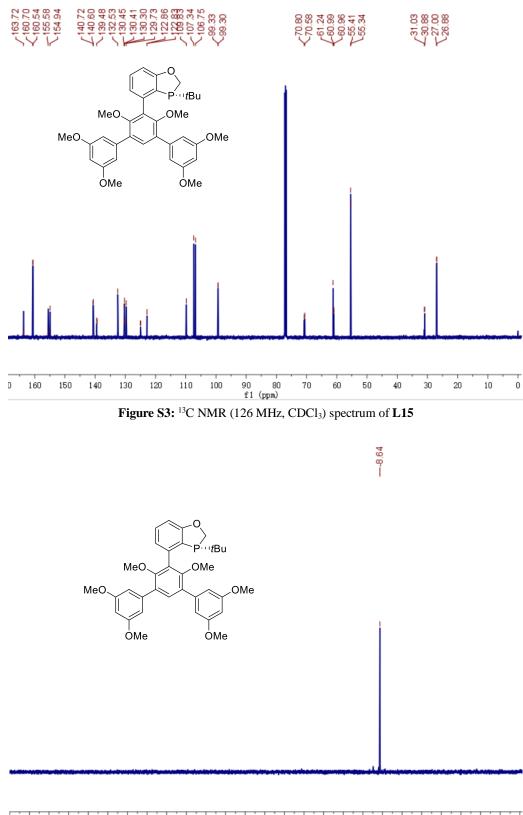


Figure S2: ¹H NMR (500 MHz, CDCl₃) spectrum of L15



170 150 130 110 90 80 70 60 50 40 30 20 10 0 -10 -30 -50 -70 f1 (ppm)

Figure S4: ³¹P NMR (162 MHz, CDCl₃) spectrum of L15

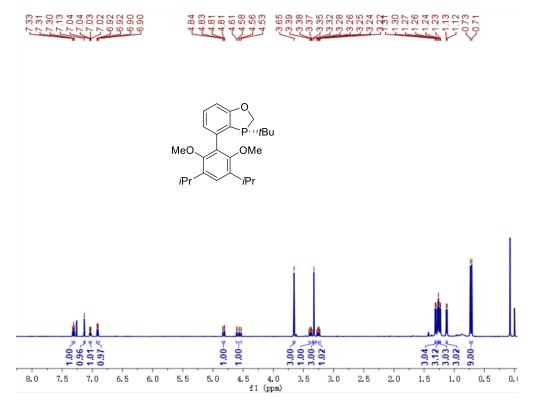


Figure S5: ¹H NMR (500 MHz, CDCl₃) spectrum of L7

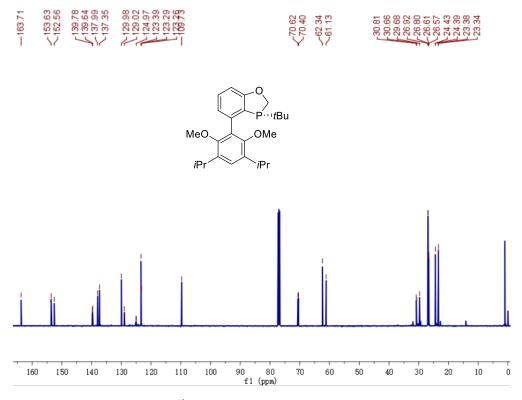
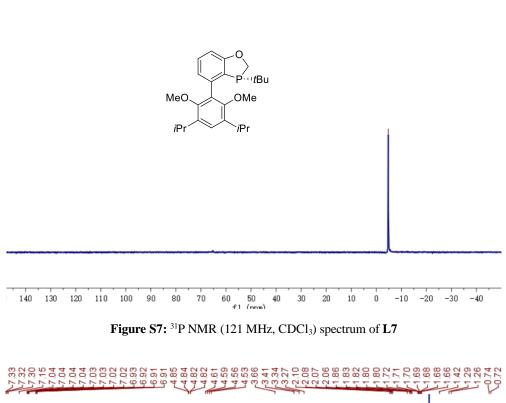


Figure S6: ¹³C NMR (126 MHz, CDCl₃) spectrum of L7



---4.78

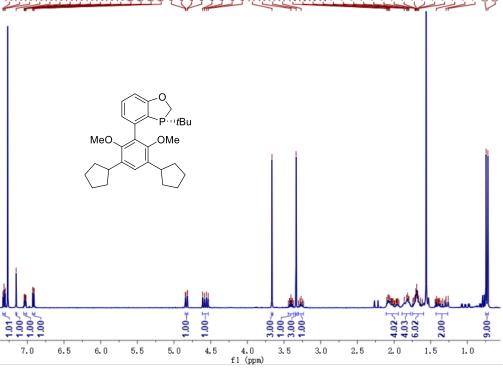


Figure S8: ¹H NMR (500 MHz, CDCl₃) spectrum of L8

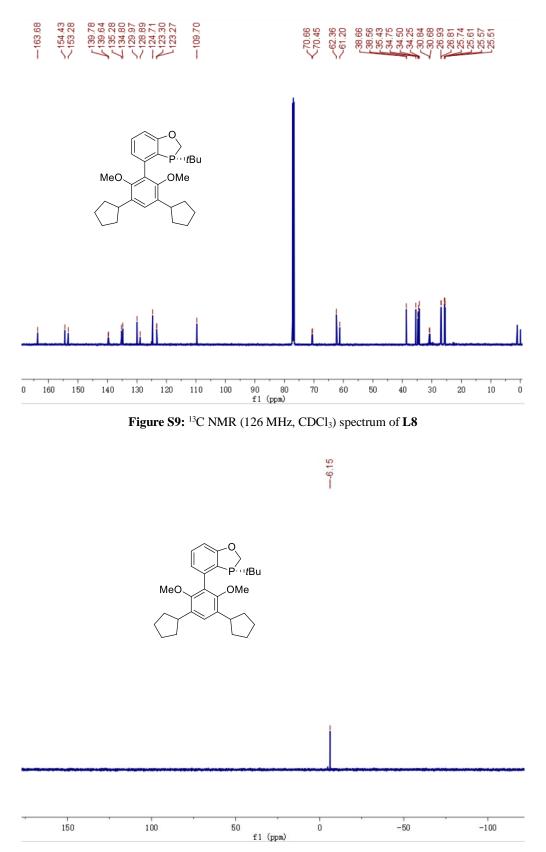


Figure S10: ³¹P NMR (162 MHz, CDCl₃) spectrum of L8



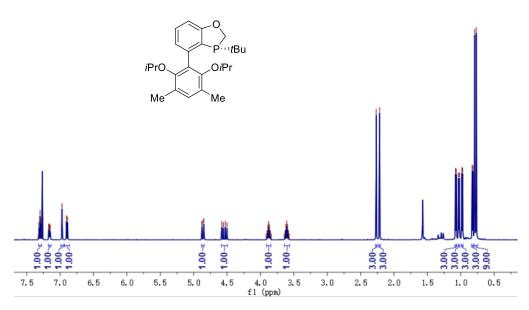


Figure S11: ¹H NMR (500 MHz, CDCl₃) spectrum of L10

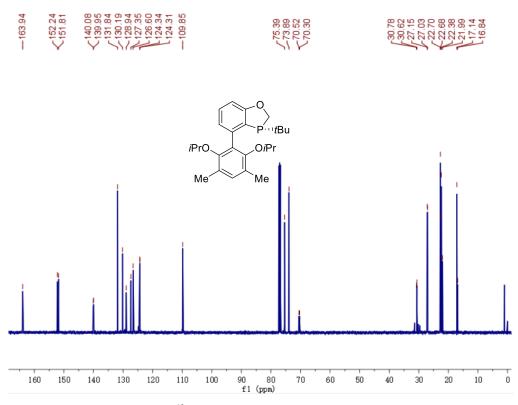


Figure S12: ¹³C NMR (126 MHz, CDCl₃) spectrum of L10

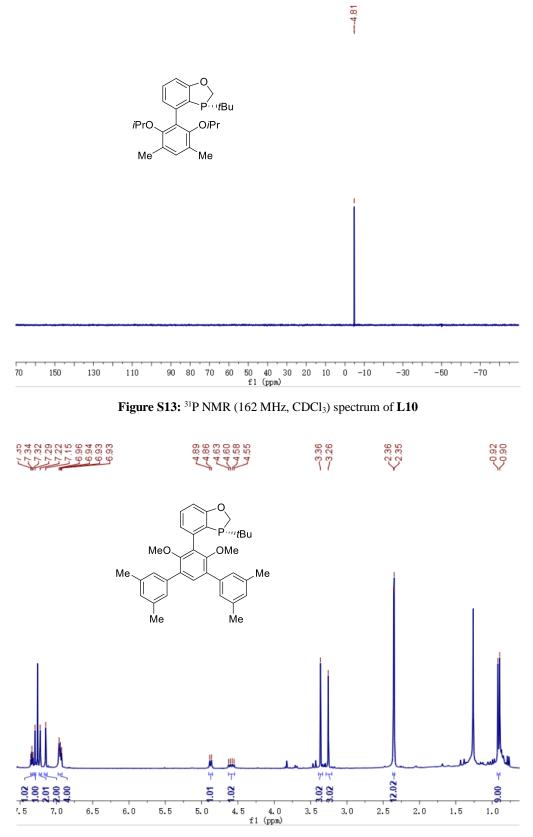
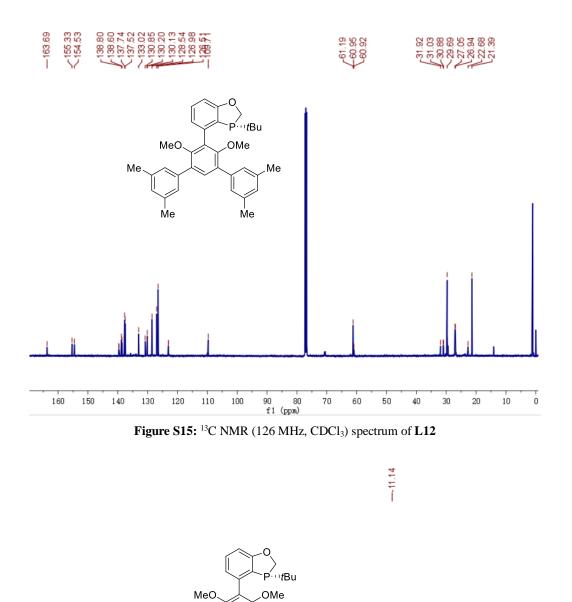


Figure S14: ¹H NMR (500 MHz, CDCl₃) spectrum of L12



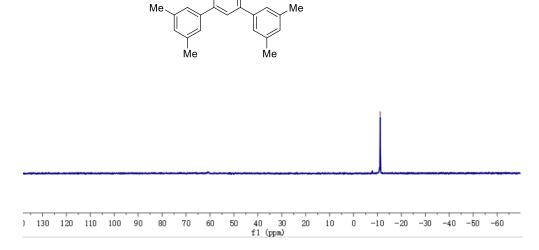


Figure S16: ³¹P NMR (121 MHz, CDCl₃) spectrum of L12

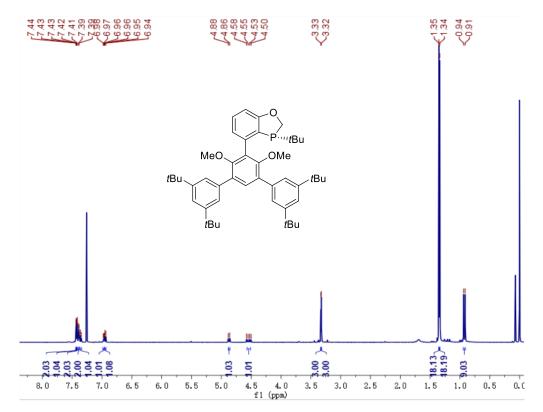


Figure S17: ¹H NMR (500 MHz, CDCl₃) spectrum of L13

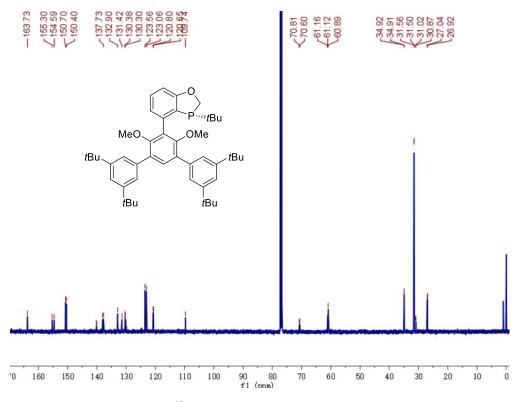


Figure S18: ¹³C NMR (126 MHz, CDCl₃) spectrum of L13

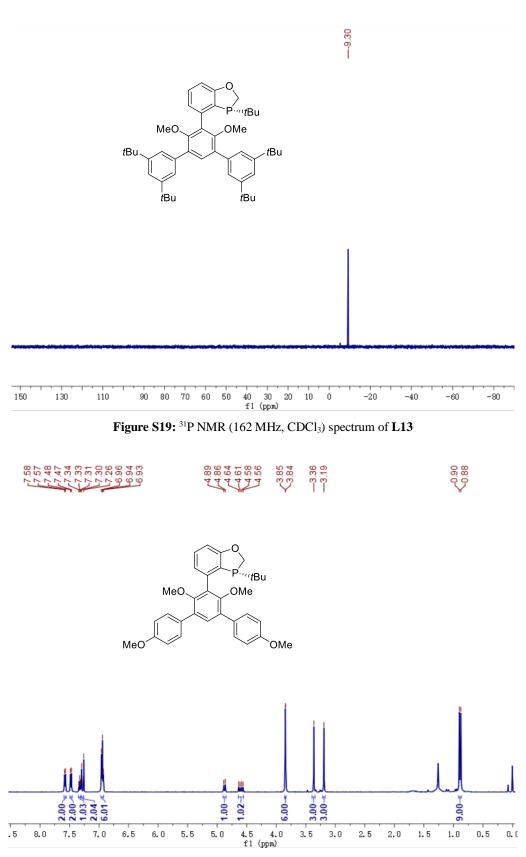
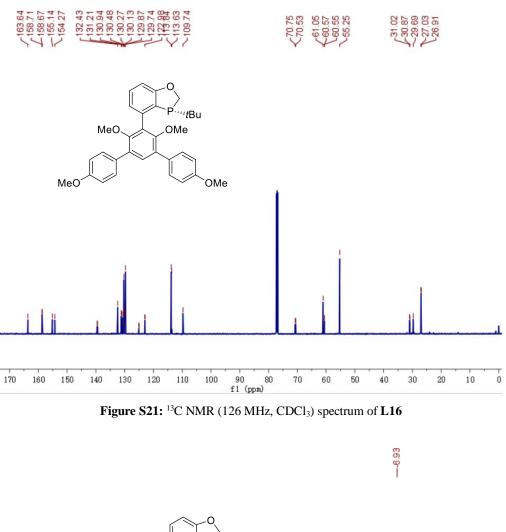


Figure S20: ¹H NMR (500 MHz, CDCl₃) spectrum of L16



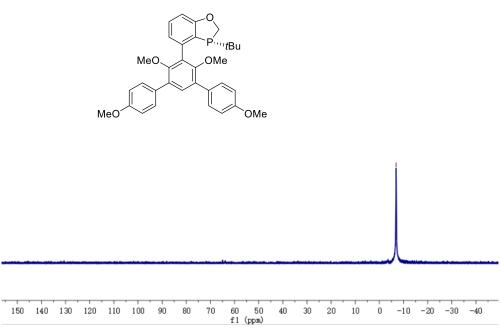
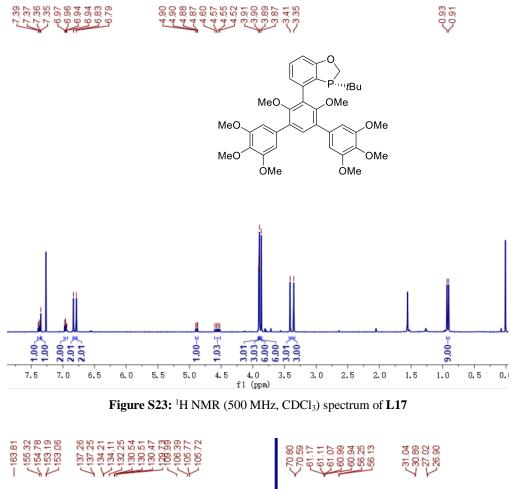


Figure S22: ³¹P NMR (121 MHz, CDCl₃) spectrum of L16



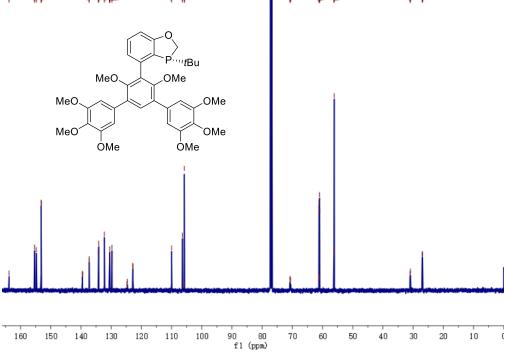


Figure S24: ¹³C NMR (126 MHz, CDCl₃) spectrum of L17

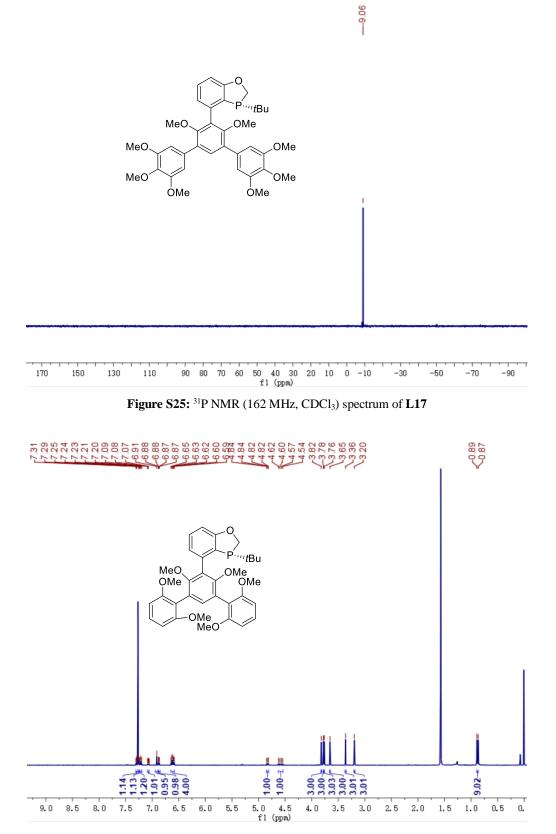


Figure S26: ¹H NMR (500 MHz, CDCl₃) spectrum of L18

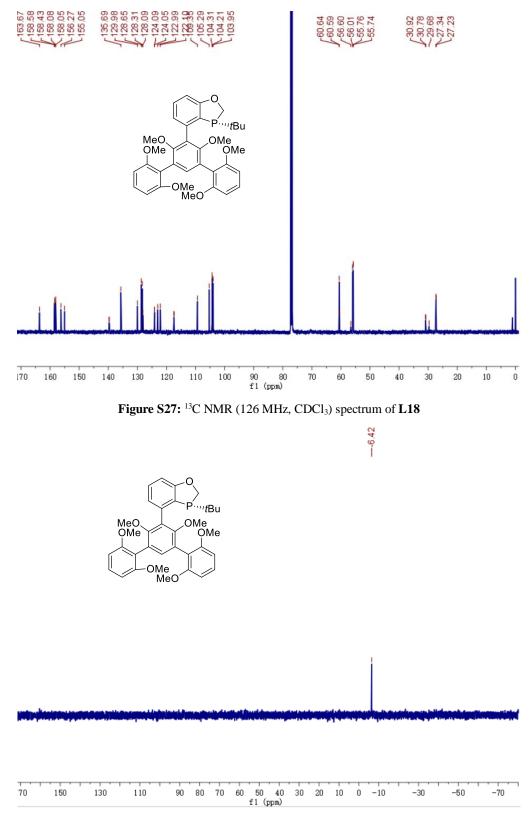


Figure S28: ³¹P NMR (162 MHz, CDCl₃) spectrum of L18



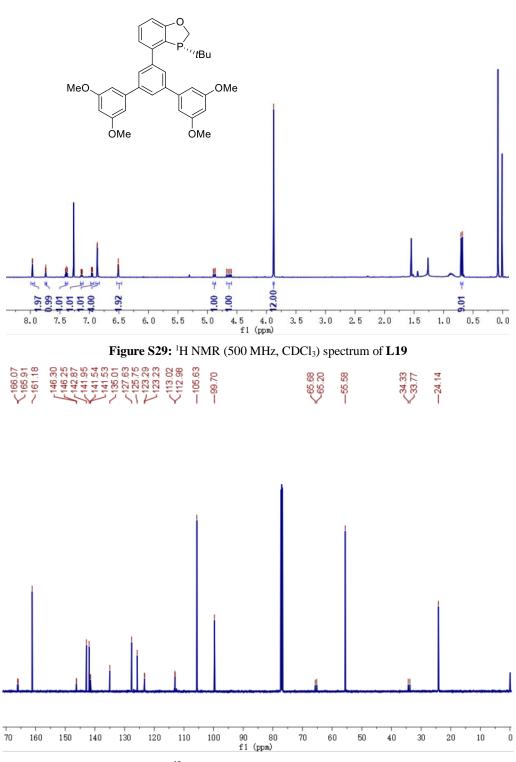


Figure S30: ¹³C NMR (126 MHz, CDCl₃) spectrum of L19

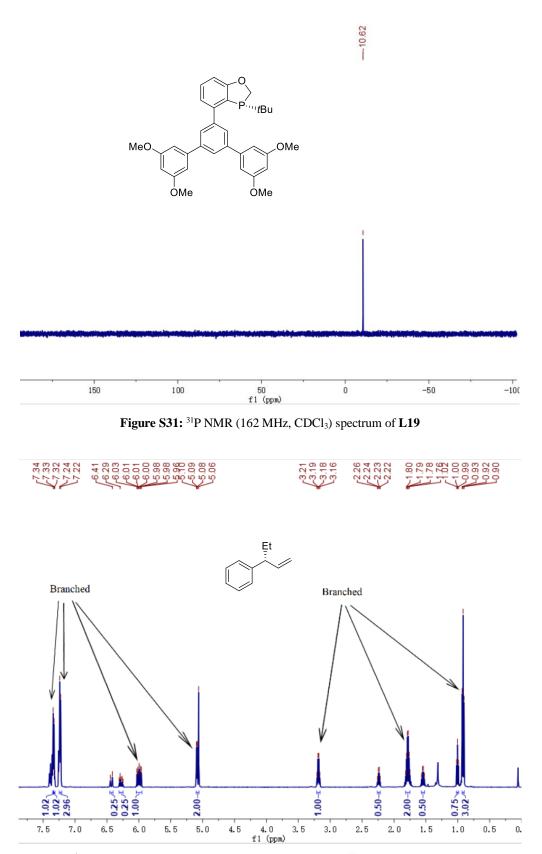


Figure S32: ¹H NMR (500 MHz, CDCl₃) spectrum of 3aa and 4aa (The left peaks are linear products.)

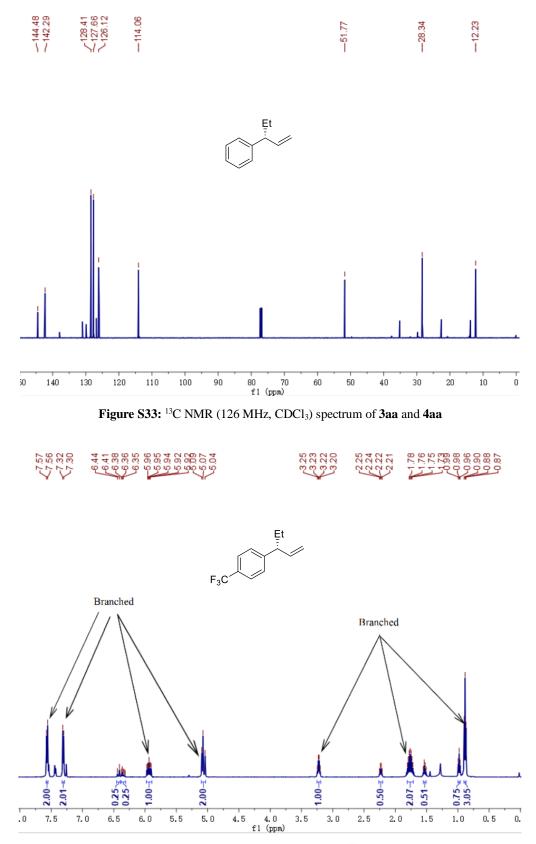


Figure S34: ¹H NMR (500 MHz, CDCl₃) spectrum of 3ba and 4ba (The left peaks are linear products.)

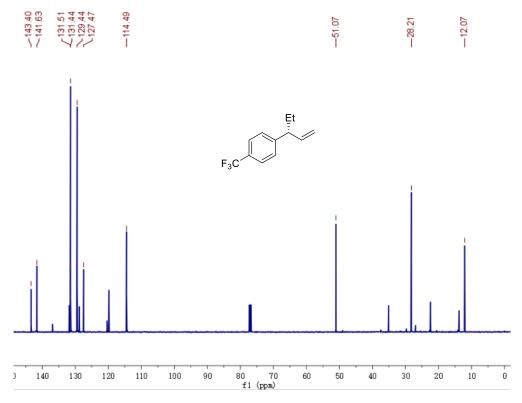


Figure S35: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3ba and 4ba

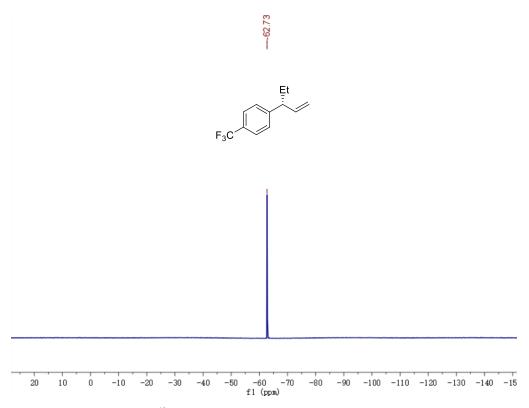
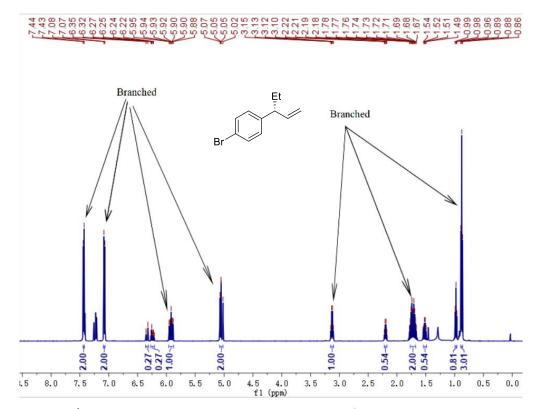
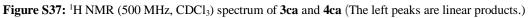


Figure S36: ¹⁹F NMR (282 MHz, CDCl₃) spectrum of 3ba and 4ba





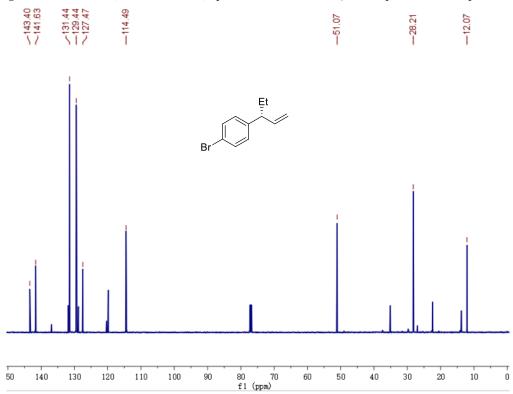


Figure S38: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3ca and 4ca

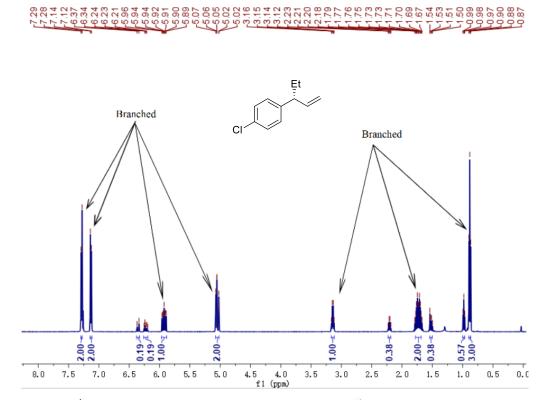


Figure S39: ¹H NMR (500 MHz, CDCl₃) spectrum of 3da and 4da (The left peaks are linear products.)

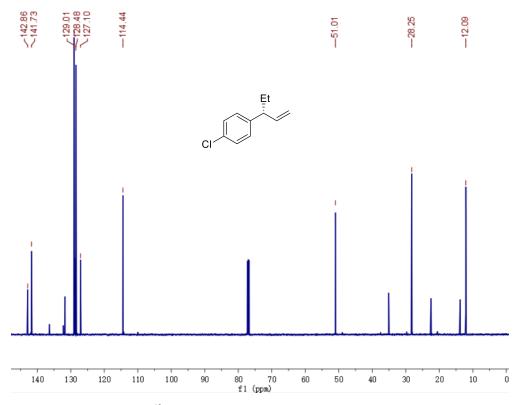


Figure S40: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3da and 4da



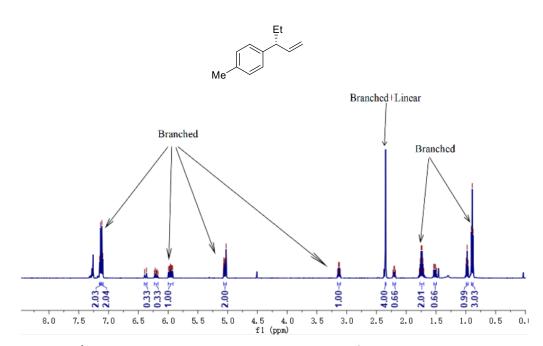


Figure S41: ¹H NMR (500 MHz, CDCl₃) spectrum of 3ea and 4ea (The left peaks are linear products.)

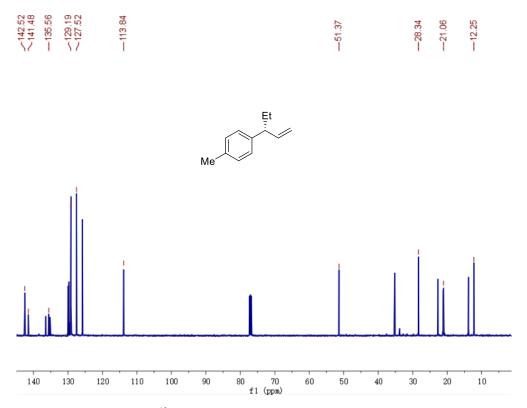


Figure S42: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3ea and 4ea



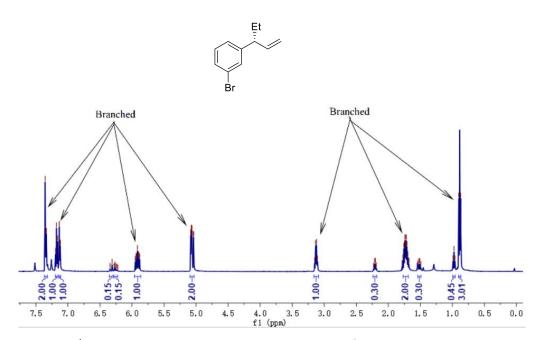


Figure S43: ¹H NMR (500 MHz, CDCl₃) spectrum of 3fa and 4fa (The left peaks are linear products.)

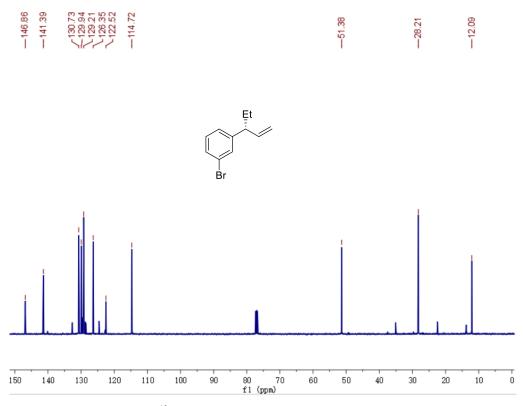


Figure S44: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3fa and 4fa

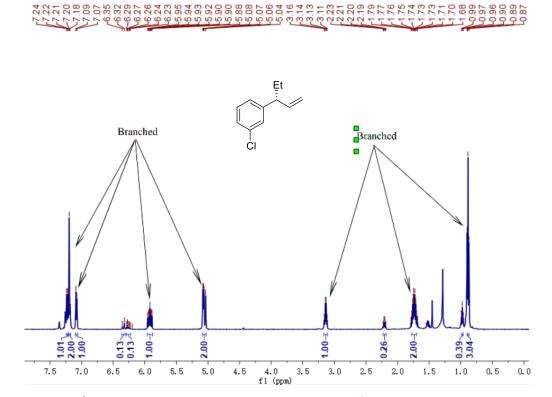


Figure S45: ¹H NMR (500 MHz, CDCl₃) spectrum of 3ga and 4ga (The left peaks are linear products.)

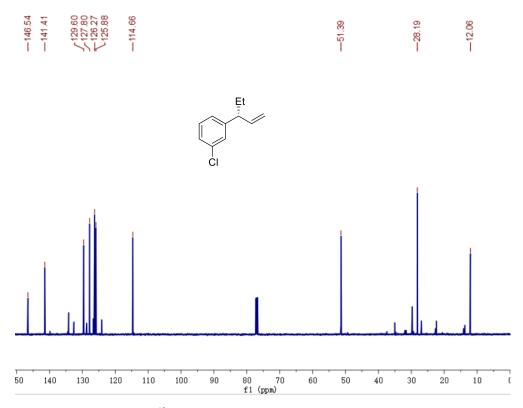


Figure S46: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3ga and 4ga

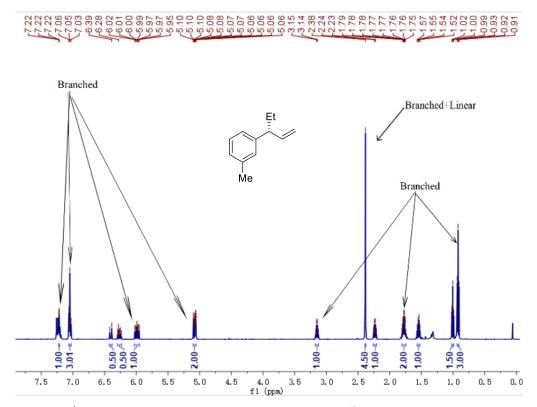


Figure S47: ¹H NMR (500 MHz, CDCl₃) spectrum of **3ha** and **4ha** (The left peaks are linear products.)

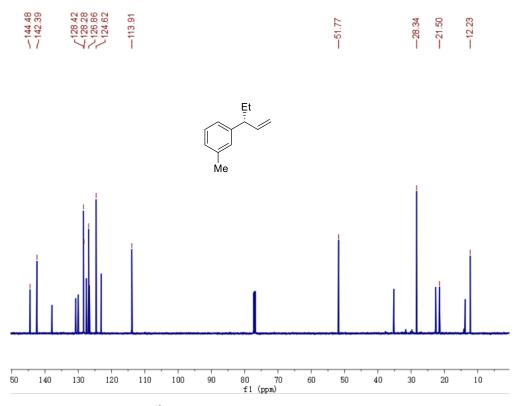


Figure S48: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3ha and 4ha

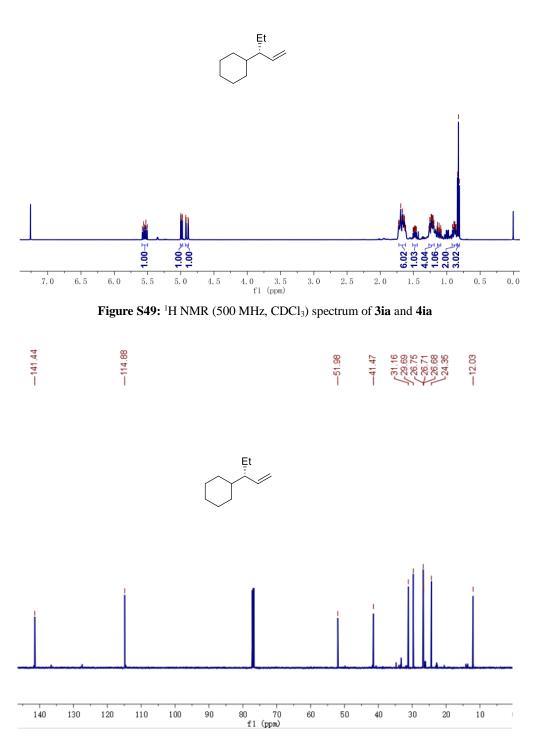
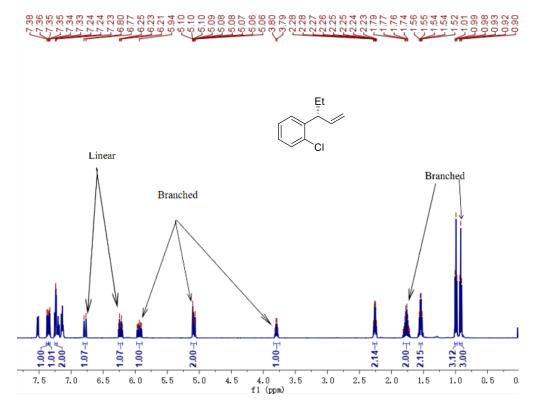
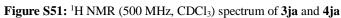


Figure S50: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3ia and 4ia





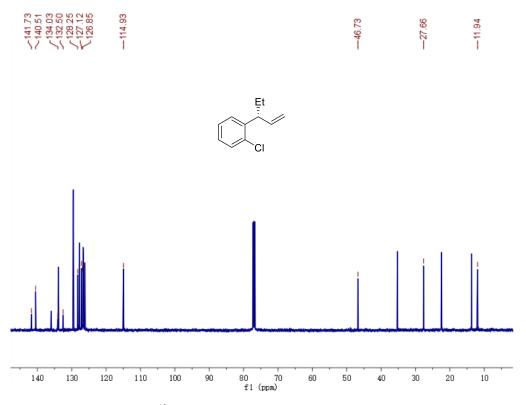


Figure S52: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3ja and 4ja

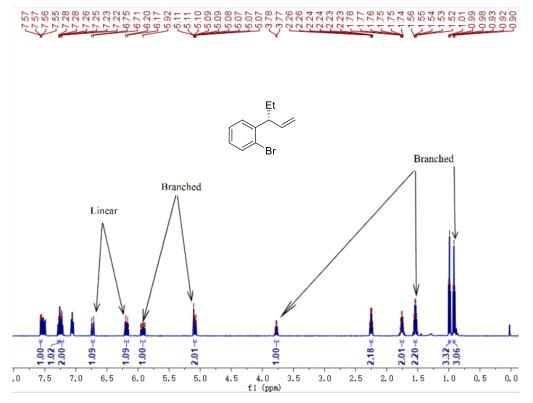


Figure S53: ¹H NMR (500 MHz, CDCl₃) spectrum of 3ka and 4ka

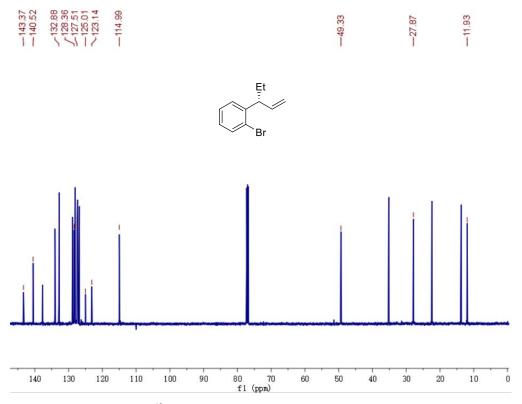
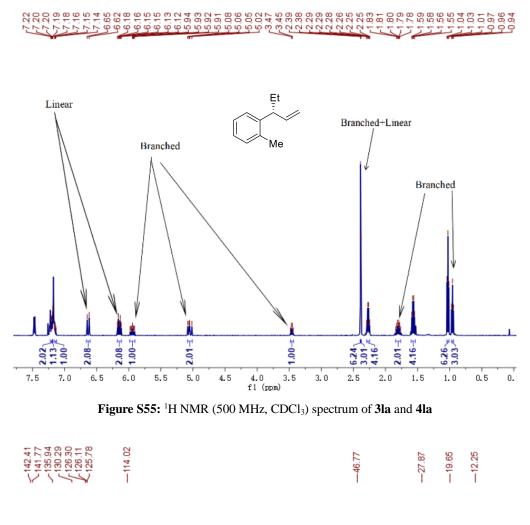


Figure S54: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3ka and 4ka



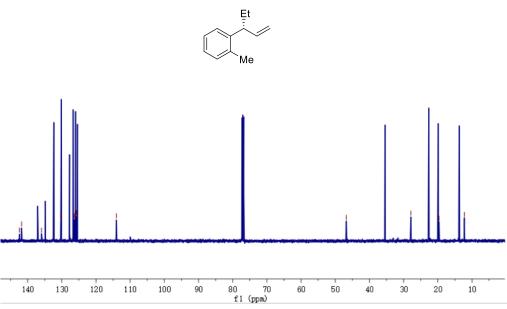


Figure S56: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3la and 4la

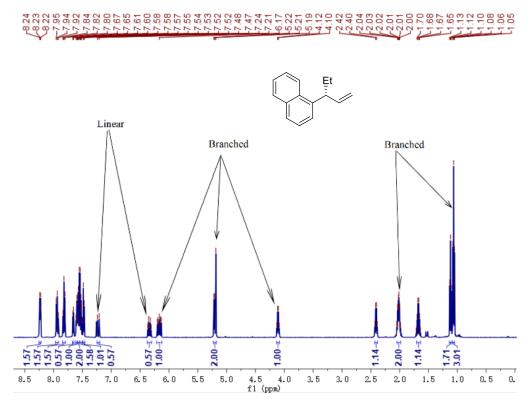


Figure S57: ¹H NMR (500 MHz, CDCl₃) spectrum of 3ma and 4ma

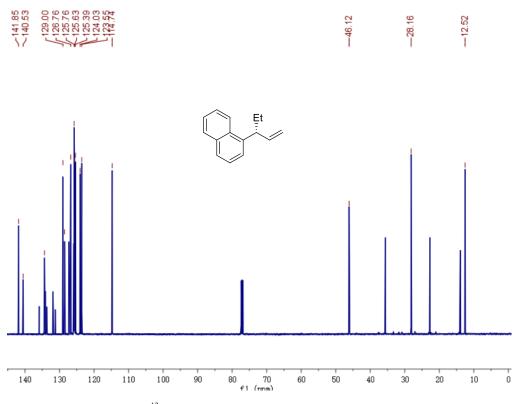


Figure S58: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3ma and 4ma

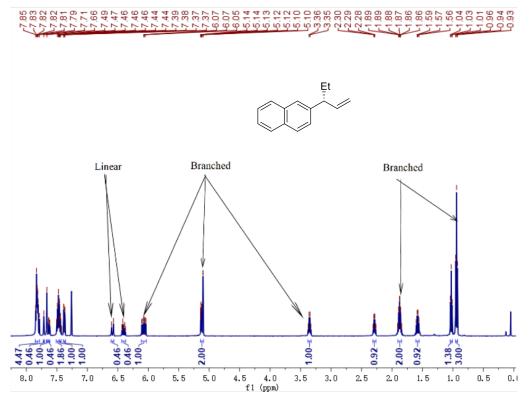


Figure S59: ¹H NMR (500 MHz, CDCl₃) spectrum of 3na and 4na

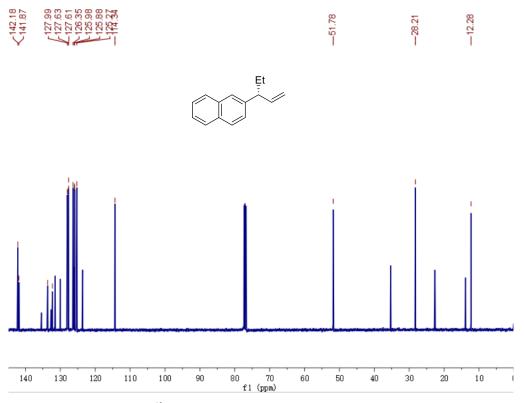


Figure S60: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3na and 4na

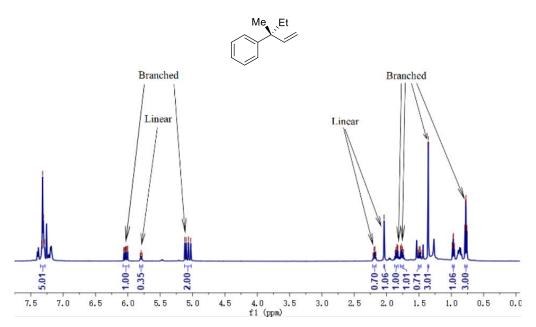


Figure S61: ¹H NMR (500 MHz, CDCl₃) spectrum of 30a and 40a

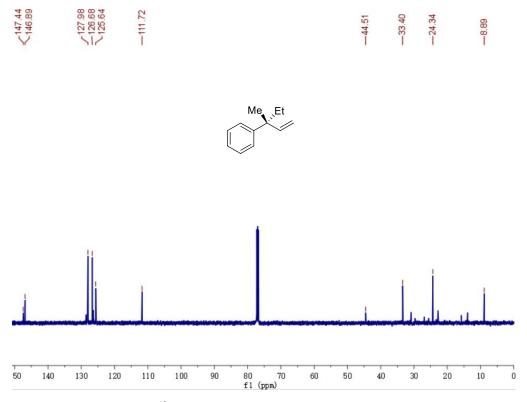


Figure S62: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3oa and 4oa

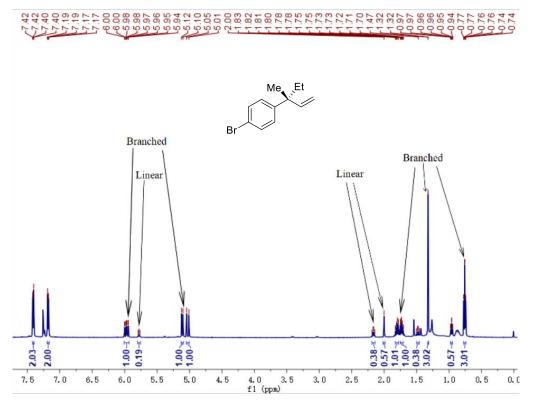


Figure S63: ¹H NMR (500 MHz, CDCl₃) spectrum of 3pa and 4pa

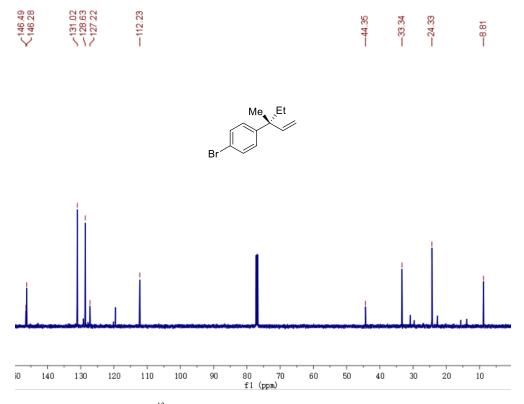


Figure S64: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3pa and 4pa

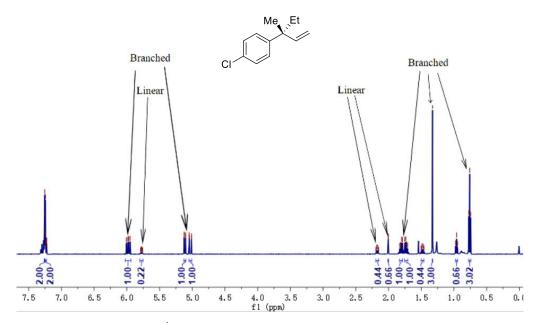
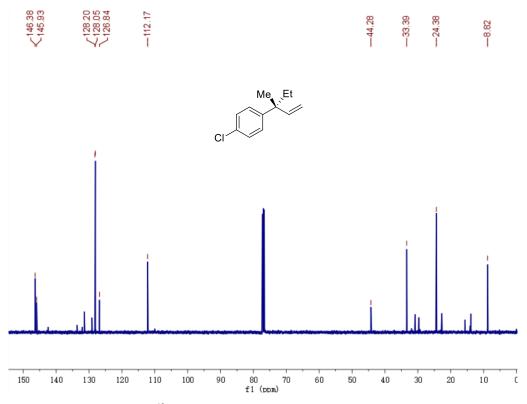
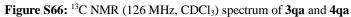


Figure S65: ¹H NMR (500 MHz, CDCl₃) spectrum of 3qa and 4qa





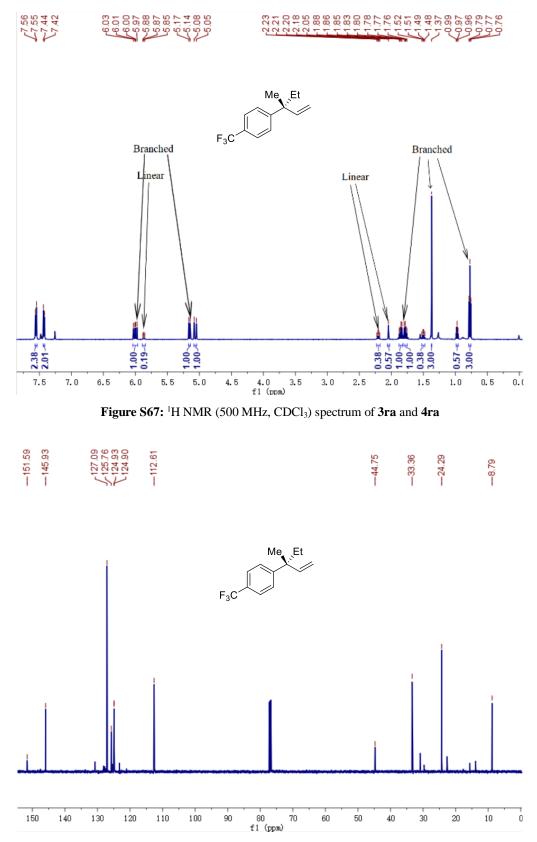
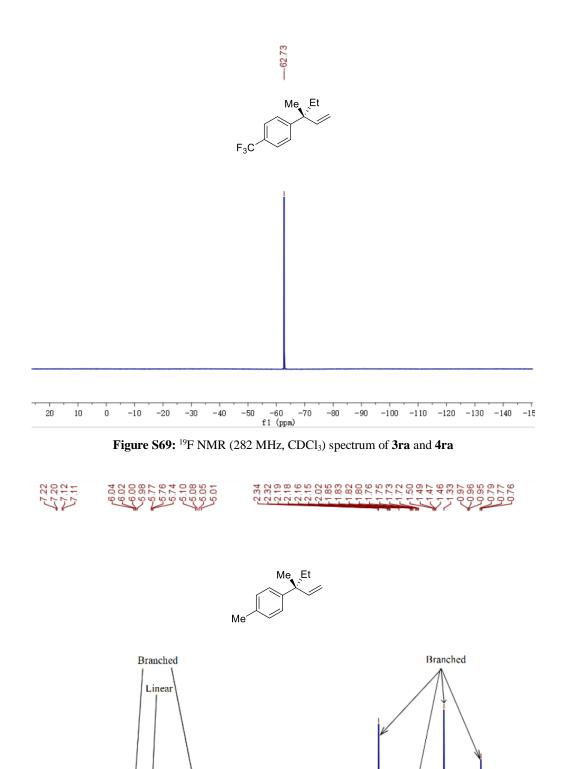
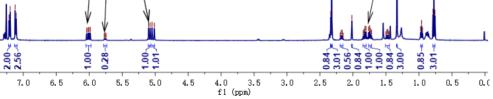
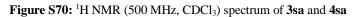


Figure S68: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3ra and 4ra





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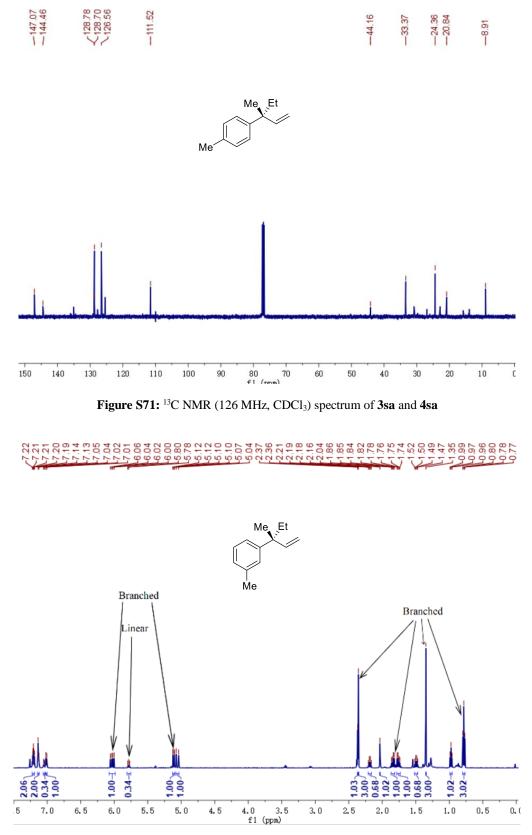


Figure S72: ¹H NMR (500 MHz, CDCl₃) spectrum of 3ta and 4ta

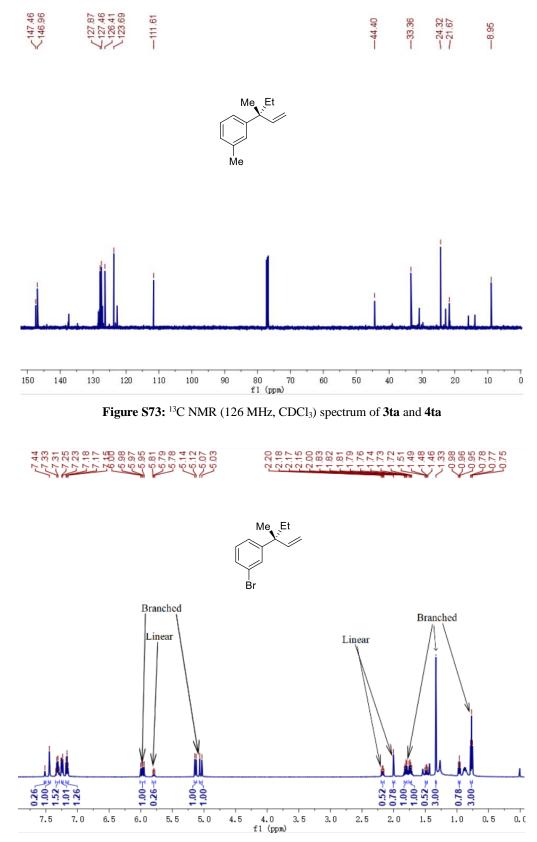


Figure S74: ¹H NMR (500 MHz, CDCl₃) spectrum of 3ua and 4ua

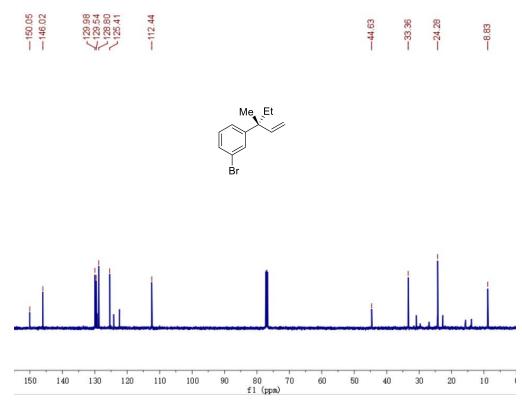


Figure S75: $^{\rm 13}C$ NMR (126 MHz, CDCl₃) spectrum of 3ua and 4ua

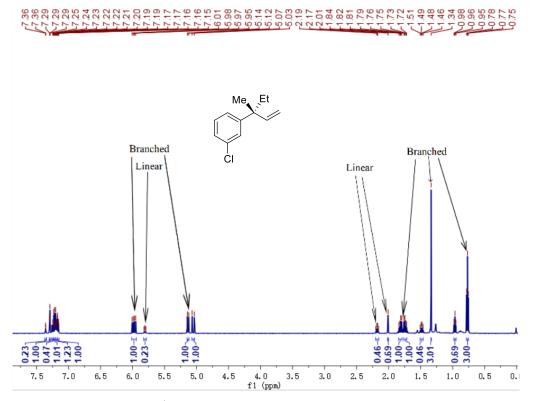


Figure S76: ¹H NMR (500 MHz, CDCl₃) spectrum of 3va and 4va

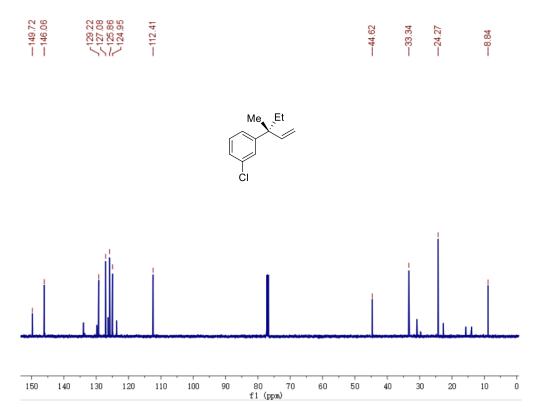


Figure S77: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3va and 4va

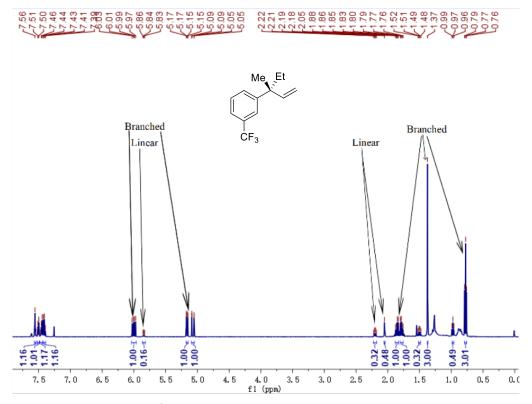


Figure S78: ¹H NMR (500 MHz, CDCl₃) spectrum of 3wa and 4wa

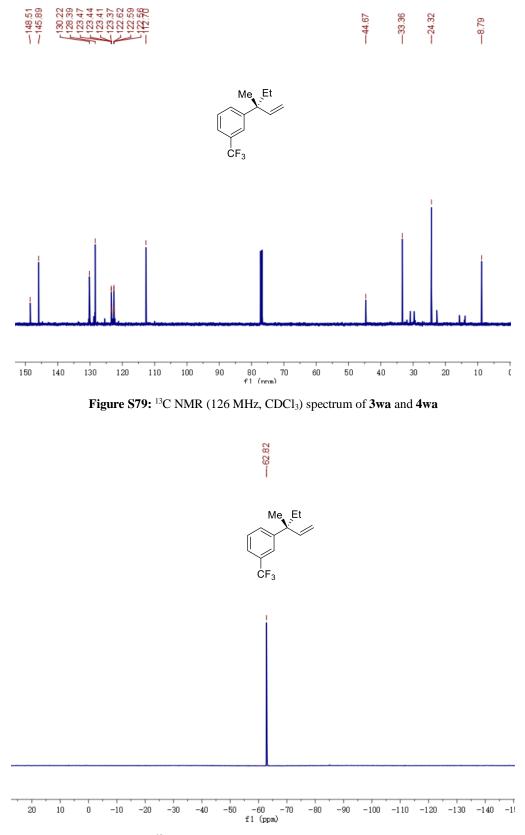


Figure S80: ¹⁹F NMR (282 MHz, CDCl₃) spectrum of 3wa and 4wa



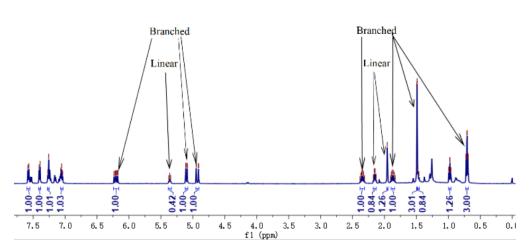


Figure S81: ¹H NMR (500 MHz, CDCl₃) spectrum of 3xa and 4xa

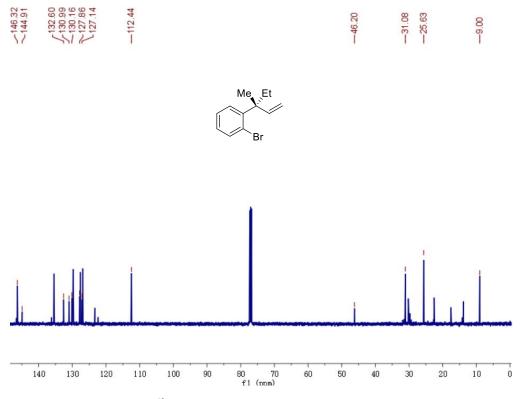


Figure S82: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3xa and 4xa



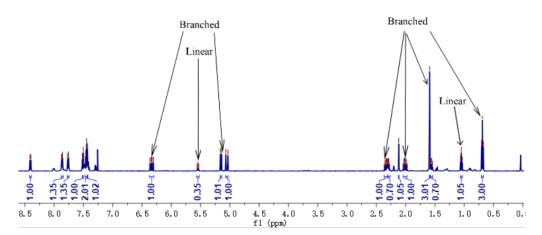


Figure S83: ¹H NMR (500 MHz, CDCl₃) spectrum of 3ya and 4ya

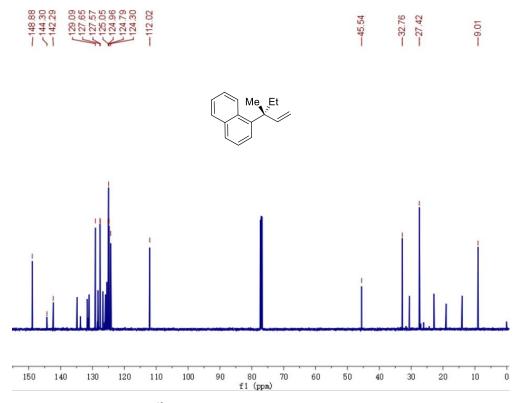
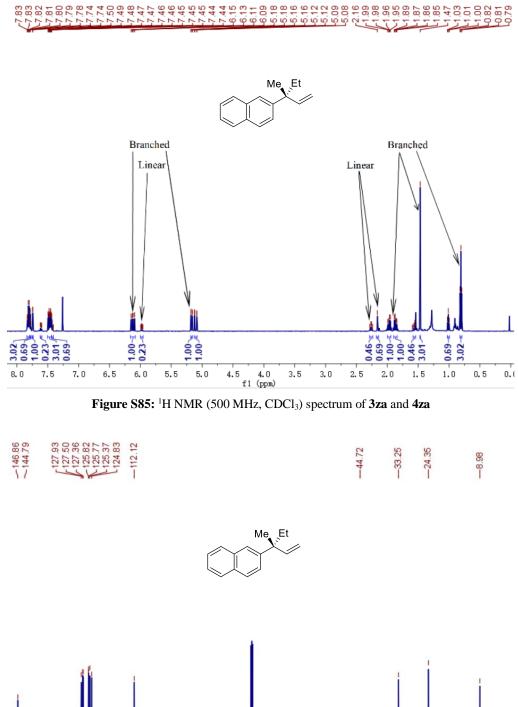
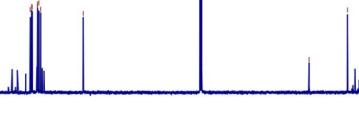


Figure S84: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3ya and 4ya





140 130 120 110 100 90 80 70 60 50 40 30 20 10 f1 (ppm)

Figure S86: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3za and 4za

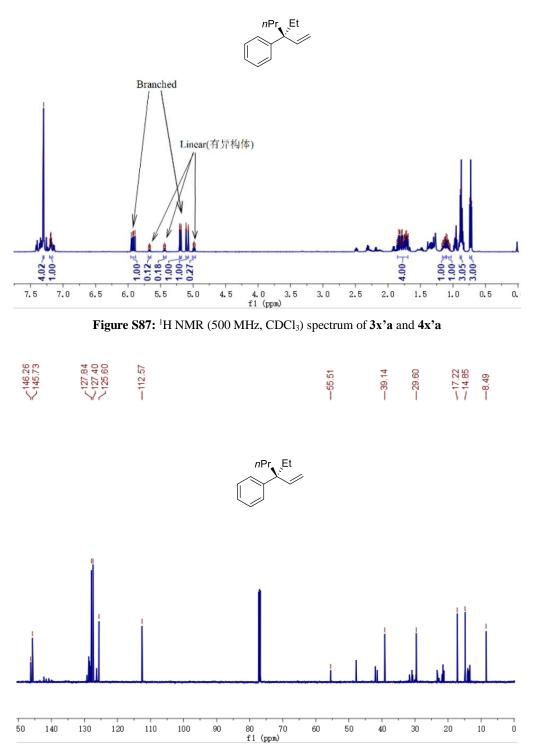


Figure S88: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3x'a and 4x'a





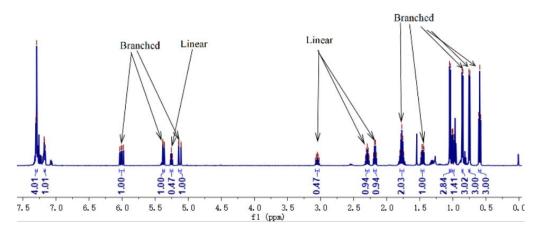


Figure S89: ¹H NMR (500 MHz, CDCl₃) spectrum of 3y'a and 4y'a

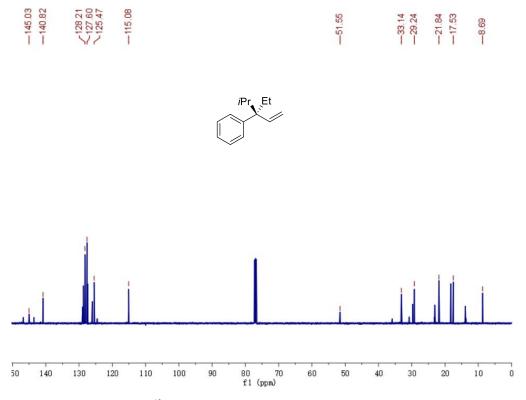


Figure S90: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3y'a and 4y'a

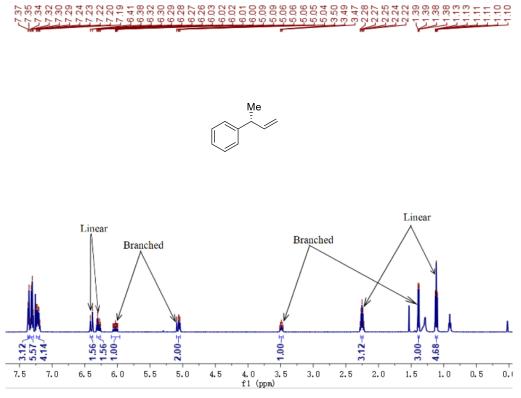


Figure S91: ¹H NMR (500 MHz, CDCl₃) spectrum of 3ab and 4ab

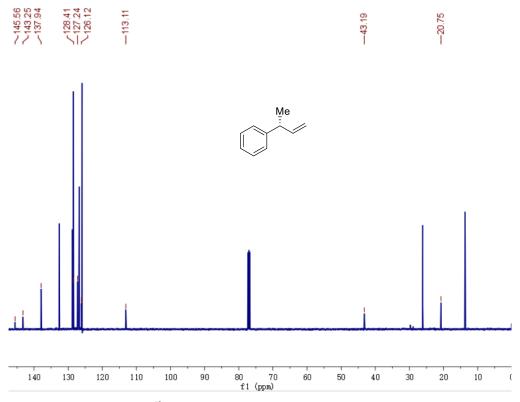


Figure S92: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3ab and 4ab



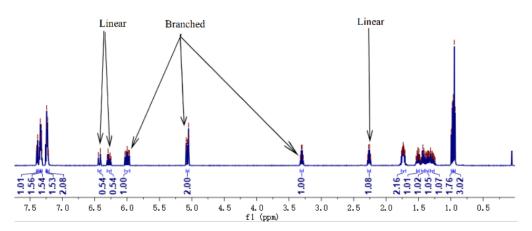


Figure S93: ¹H NMR (500 MHz, CDCl₃) spectrum of 3ac and 4ac

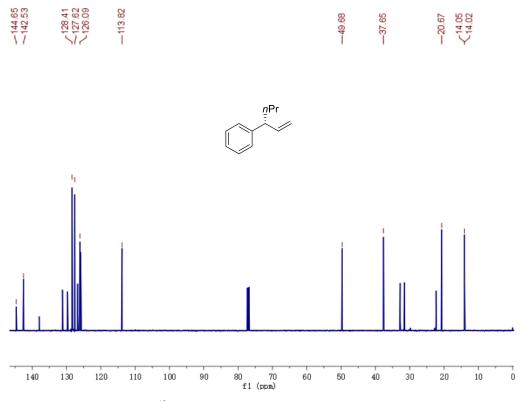
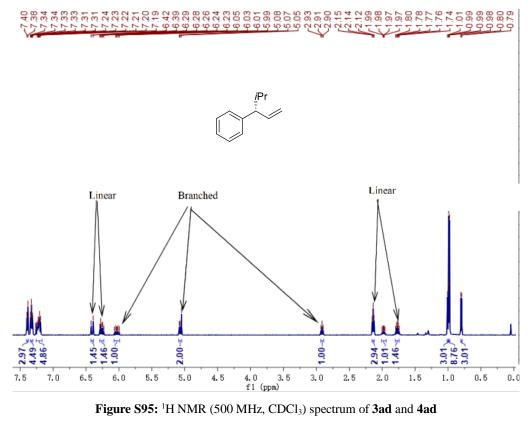


Figure S94: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3ac and 4ac





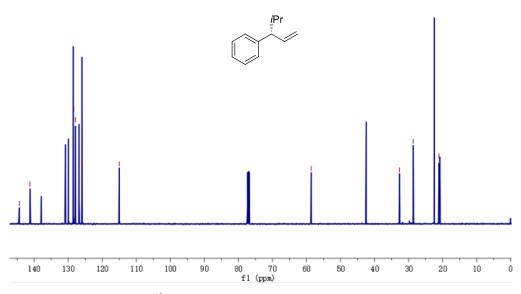


Figure S96: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3ad and 4ad

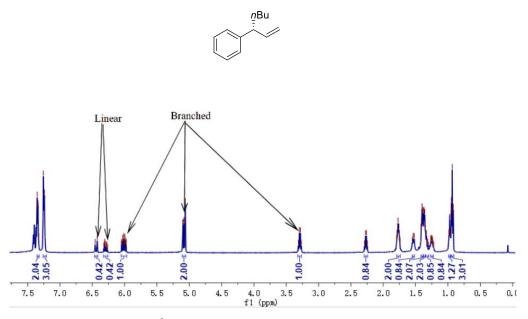


Figure S97: ¹H NMR (500 MHz, CDCl₃) spectrum of 3ae and 4ae

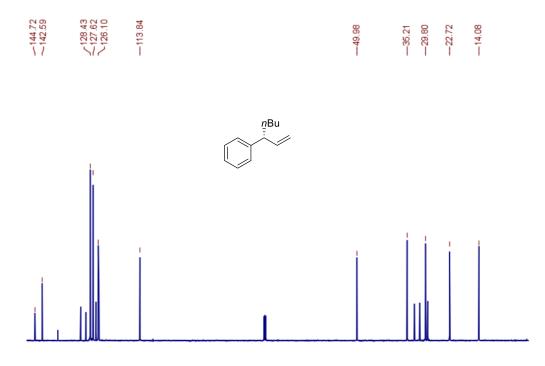


Figure S98: ¹³C NMR (126 MHz, CDCl₃) spectrum of **3ae** and **4ae**

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80 70 f1 (תסס)

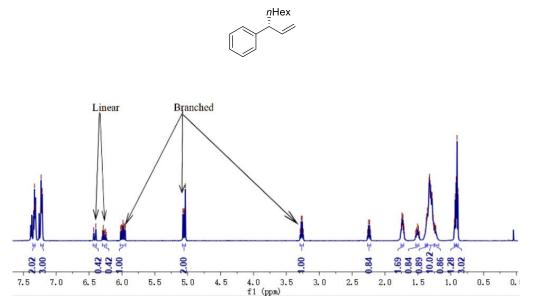


Figure S99: ¹H NMR (500 MHz, CDCl₃) spectrum of 3af and 4af

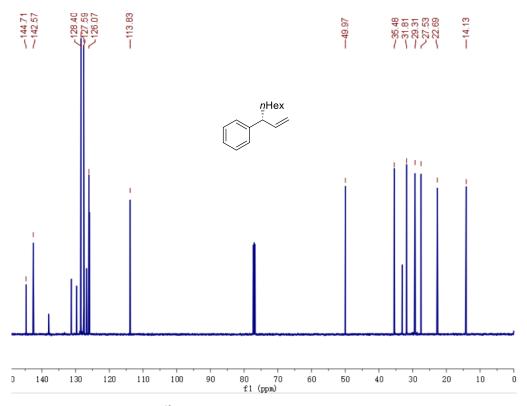


Figure S100: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3af and 4af

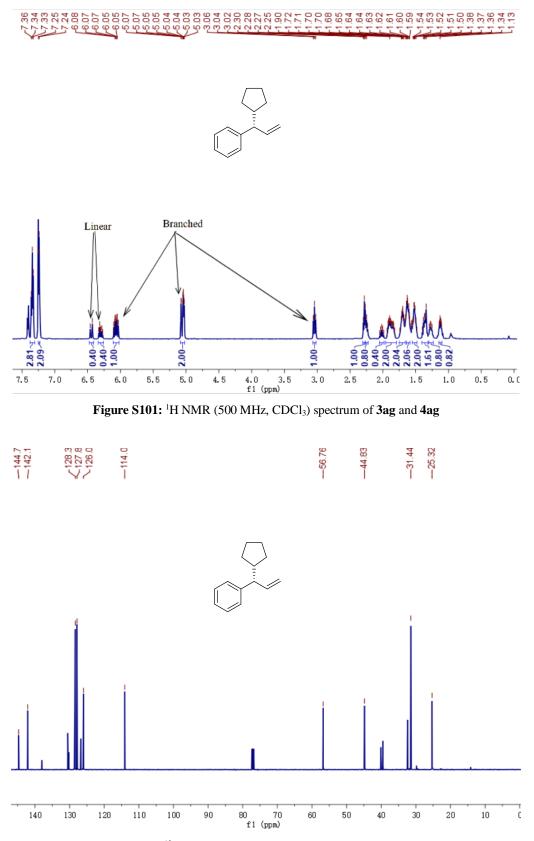


Figure S102: ¹³C NMR (126 MHz, CDCl₃) spectrum of 3ag and 4ag

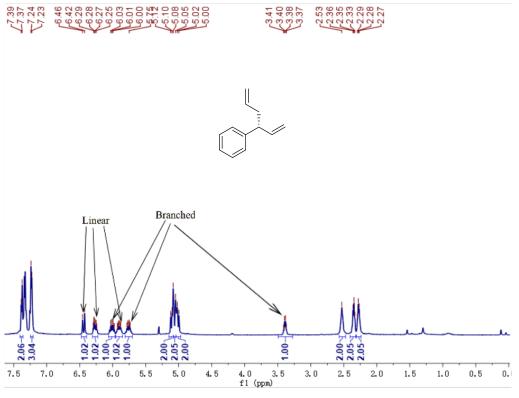


Figure S103: ¹H NMR (500 MHz, CDCl₃) spectrum of 3ah and 4ah

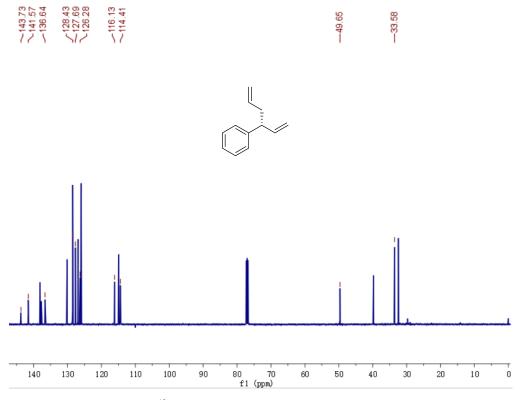


Figure S104: $^{\rm 13}C$ NMR (126 MHz, CDCl₃) spectrum of 3ah and 4ah

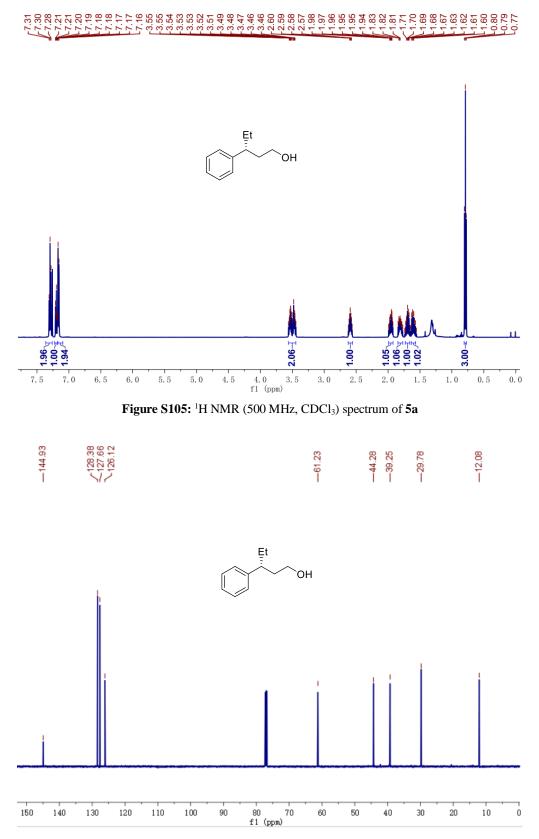


Figure S106: ¹³C NMR (126MHz, CDCl₃) spectrum of 5a

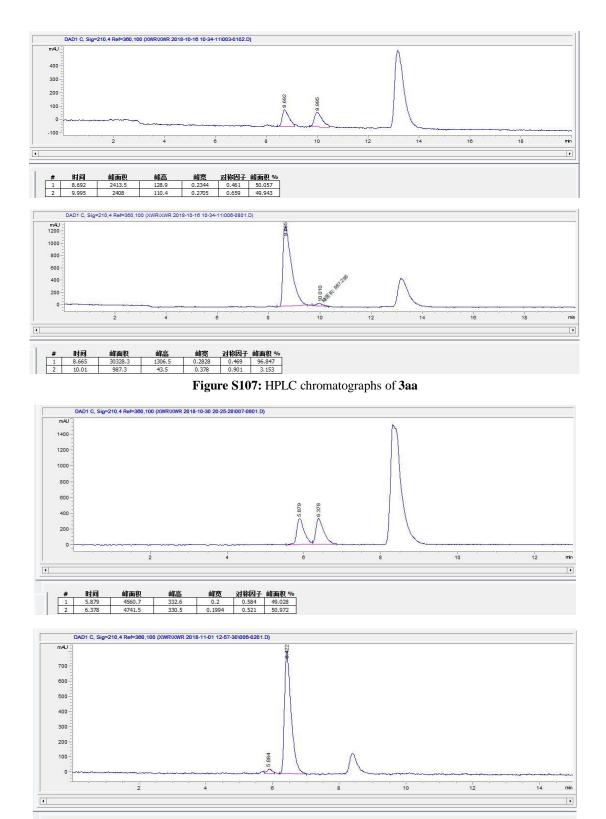
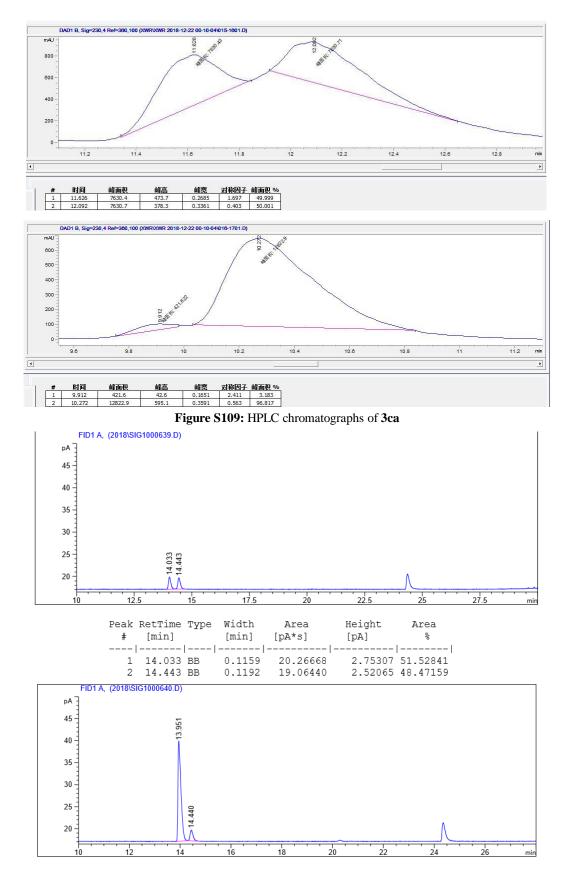




Figure S108: HPLC chromatographs of 3ba



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Peak	RetTime	Туре	Width	Area	Height	Area
#	[min]		[min]	[pA*s]	[pA]	8
1	13.951	BB	0.1245	186.31029	22.52155	90.25405
2	14.440	BB	0.1283	20.11844	2.43716	9.74595

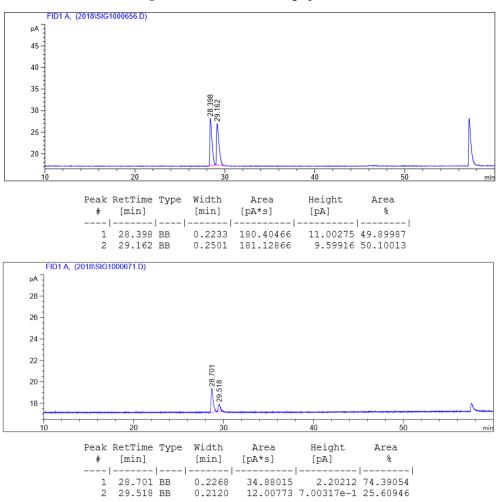
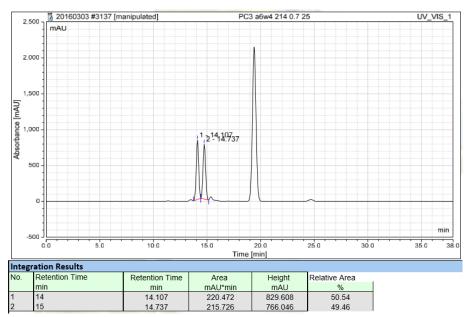


Figure S110: HPLC chromatographs of 3da

Figure S111: HPLC chromatographs of 3ea



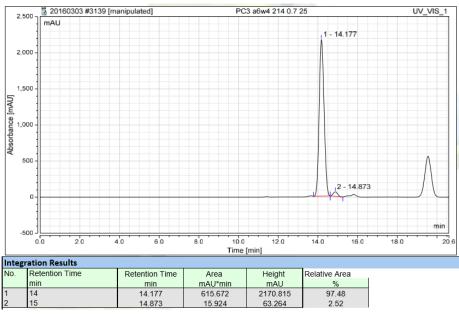
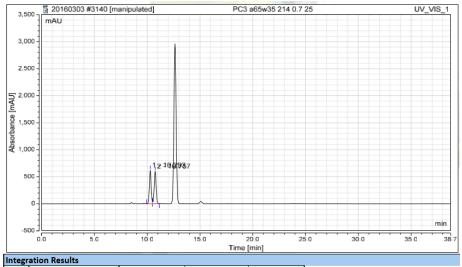


Figure S112: HPLC chromatographs of 3fa



No.	Retention Time	Retention Time Area		Height	Relative Area		
	min	min	mAU*min	mAU	%		
1	10	10.293	119.357	622.507	48.93		
2	11	10.757	124.580	600.002	51.07		

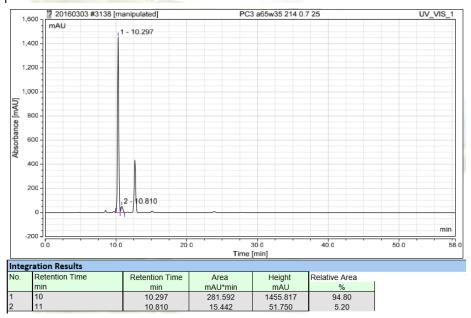
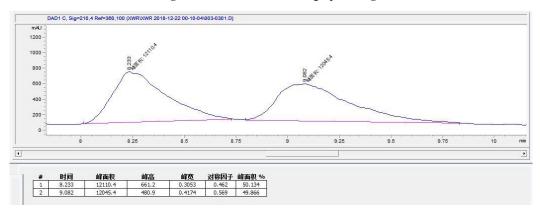
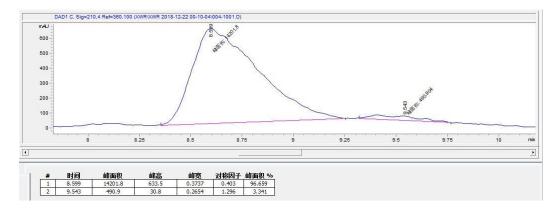
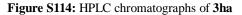
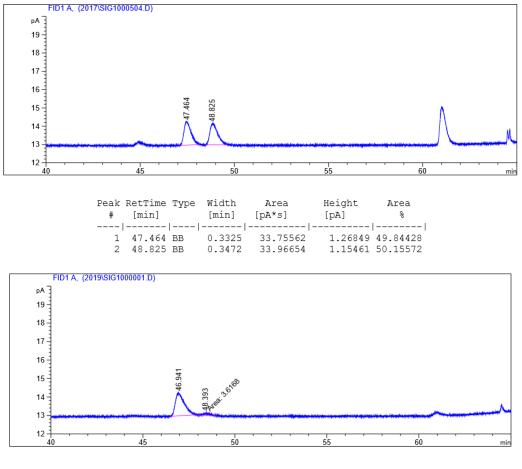


Figure S113: HPLC chromatographs of 3ga



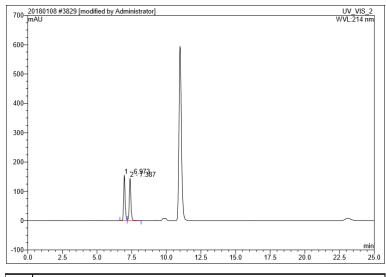






Peak	RetTime	Туре	Width	Area	Height	Area
#	[min]		[min]	[pA*s]	[pA]	90
1	46.941	BB	0.4021	42.47622	1.24477	92.15326
2	48.393	MM	0.3889	3.61680	1.55002e-1	7.84674

Figure S115: HPLC chromatographs of 3ia



No.	Ret.Time	Peak Name	Height	Area	Rel.Area	Amount	Туре	
	min		mAU	mAU*min	%			
1	6.97	n.a.	155.163	19.479	49.66	n.a.	BM	
2	7.39	n.a.	144.625	19.746	50.34	n.a.	MB	

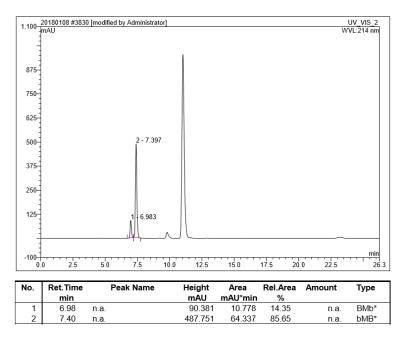
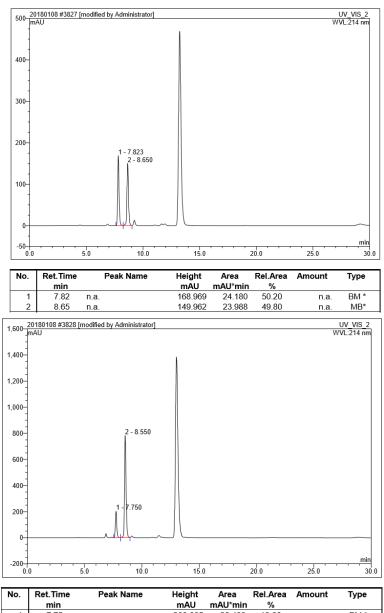
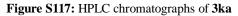
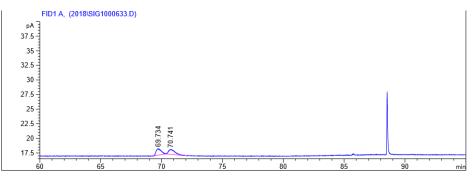


Figure S116: HPLC chromatographs of 3ja



	min		mAU	mAU*min	%		
1	7.75	n.a.	203.005	28.489	18.66	n.a.	BM *
2	8.55	n.a.	782.960	124.199	81.34	n.a.	MB*
Total:			985.965	152.688	100.00	0.000	
							,





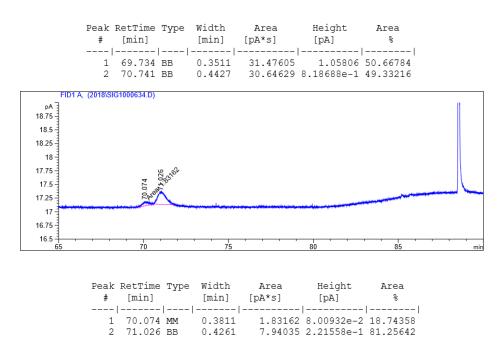


Figure S118: HPLC chromatographs of 3la

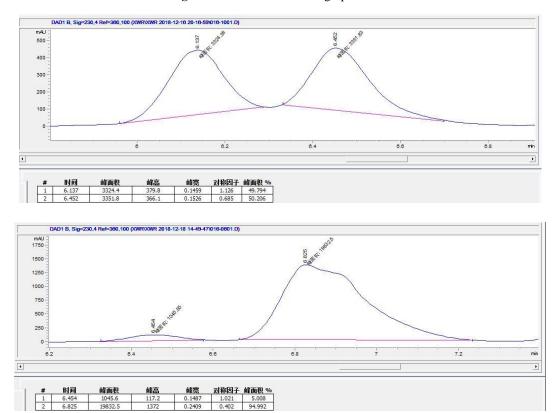


Figure S119: HPLC chromatographs of 3ma

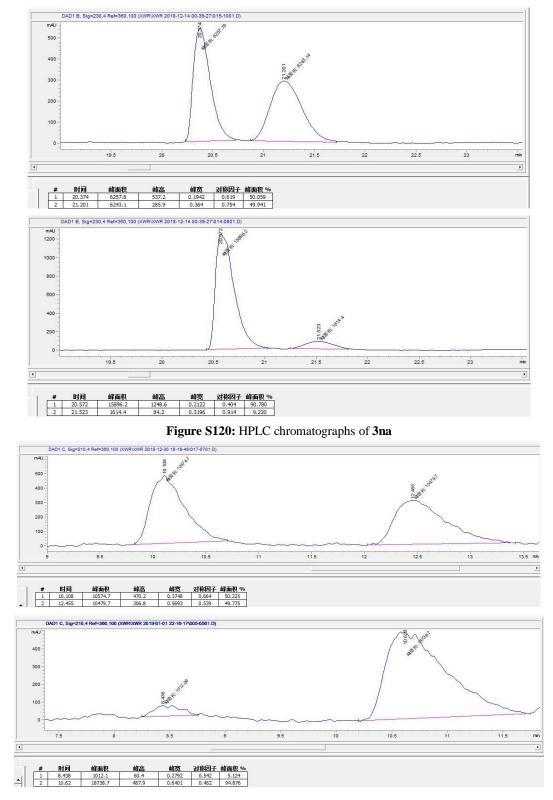
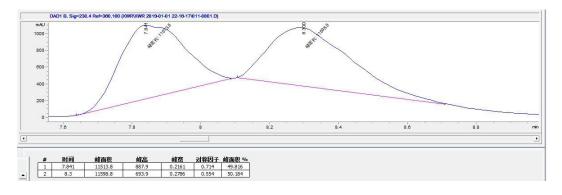
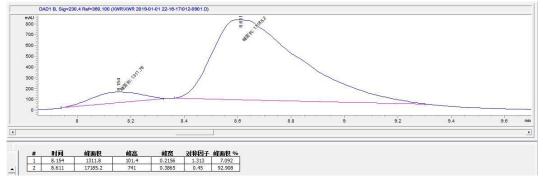
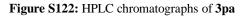


Figure S121: HPLC chromatographs of 30a







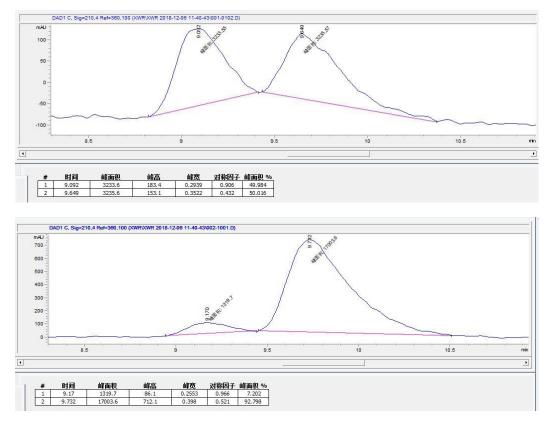
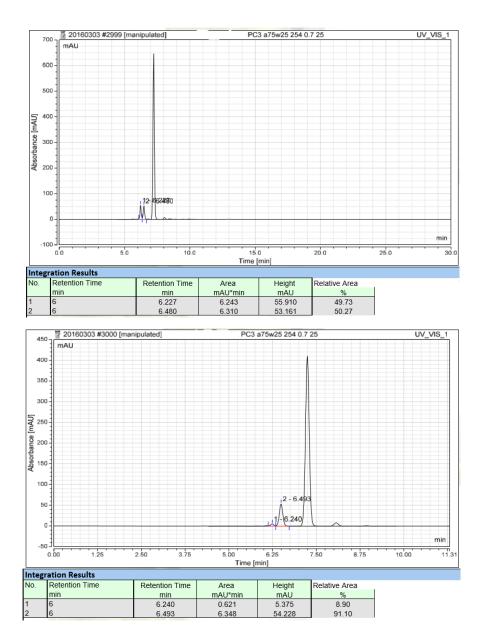
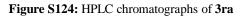
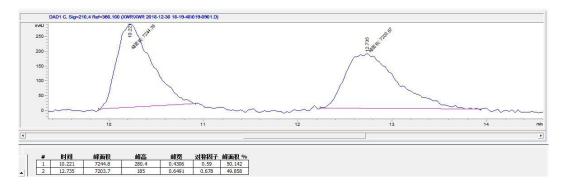


Figure S123: HPLC chromatographs of 3qa







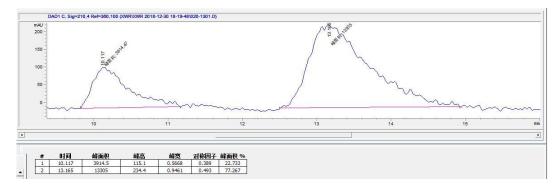
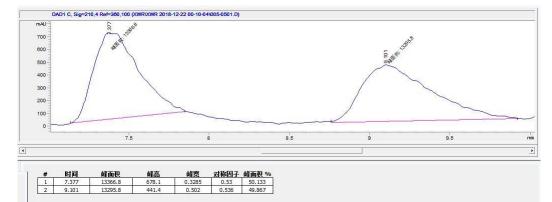


Figure S125: HPLC chromatographs of 3sa



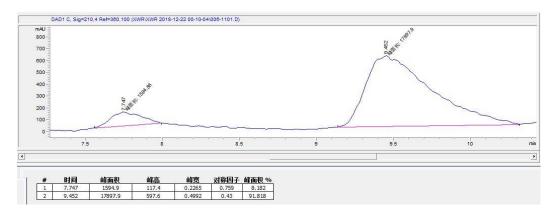
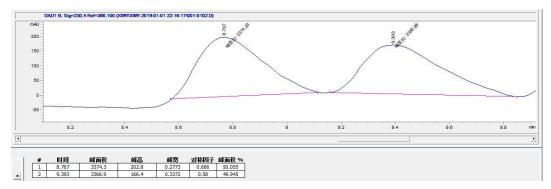
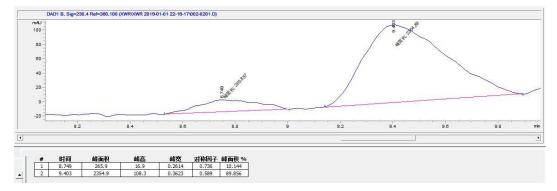
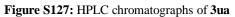


Figure S126: HPLC chromatographs of 3ta







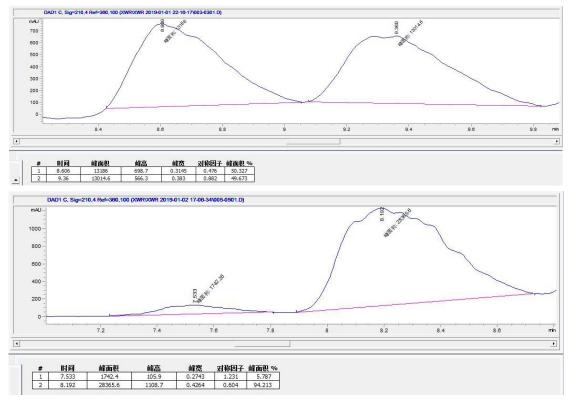


Figure S128: HPLC chromatographs of 3va

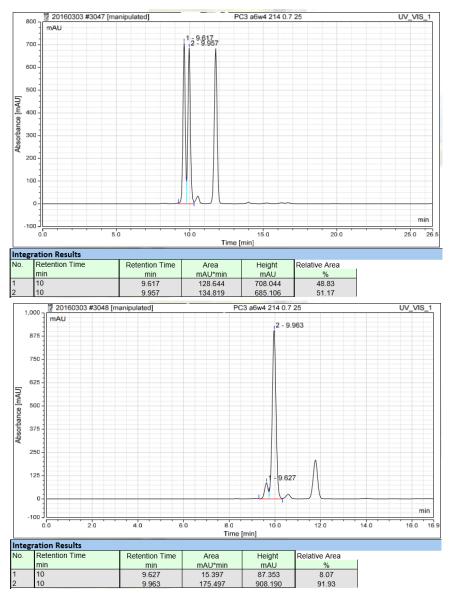


Figure S129: HPLC chromatographs of 3wa

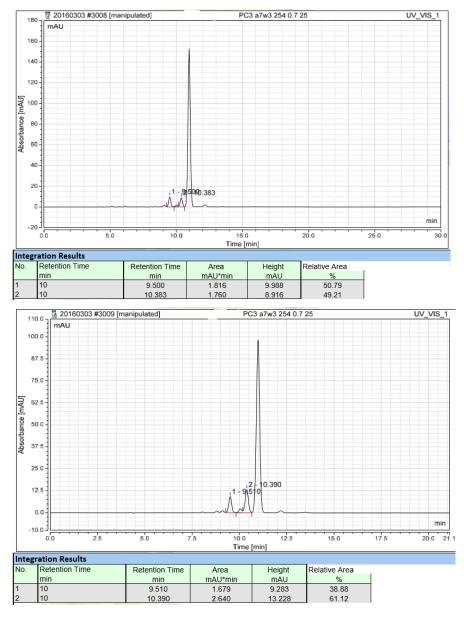
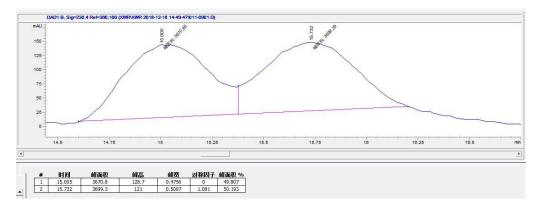
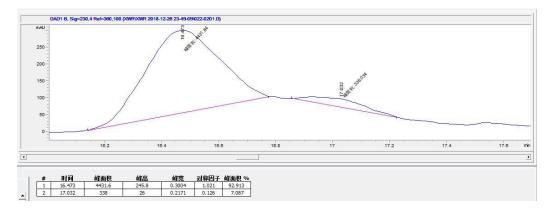
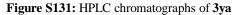


Figure S130: HPLC chromatographs of 3xa







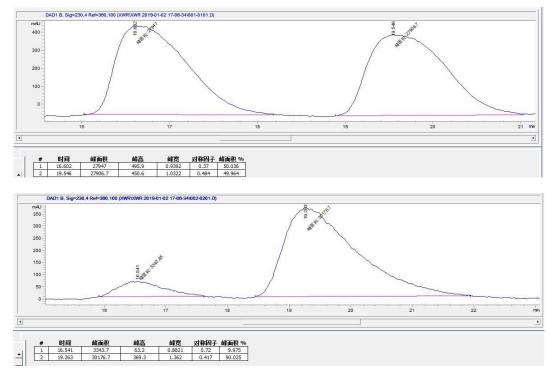
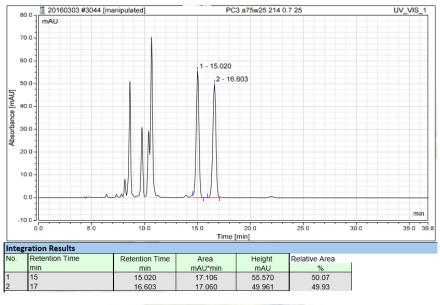


Figure S132: HPLC chromatographs of 3za



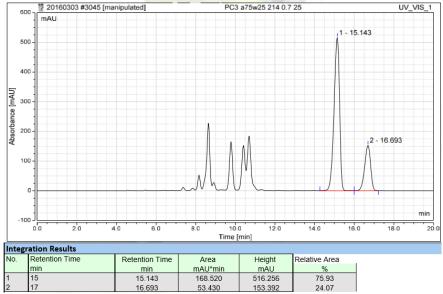
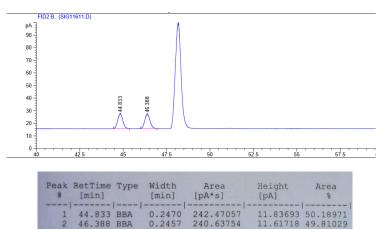
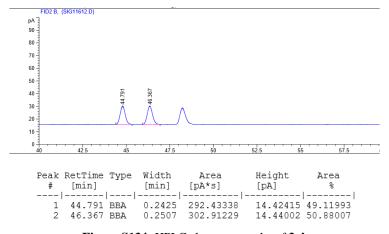
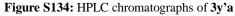
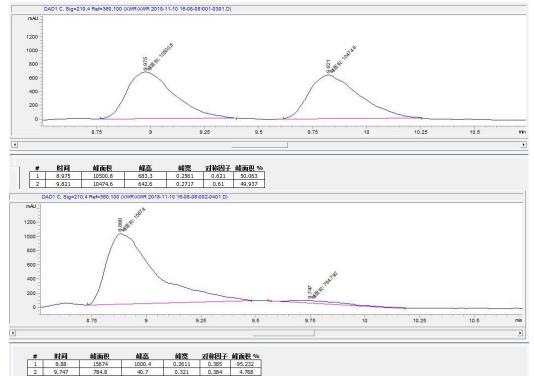


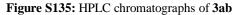
Figure S133: HPLC chromatographs of 3x'a

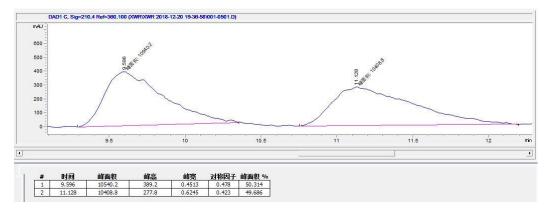












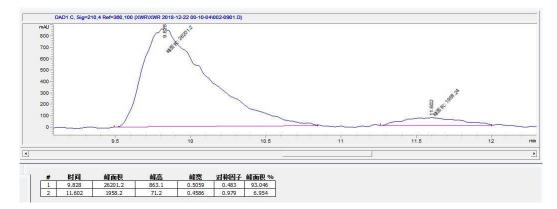
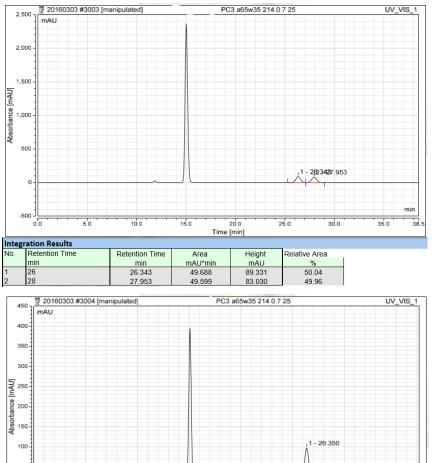
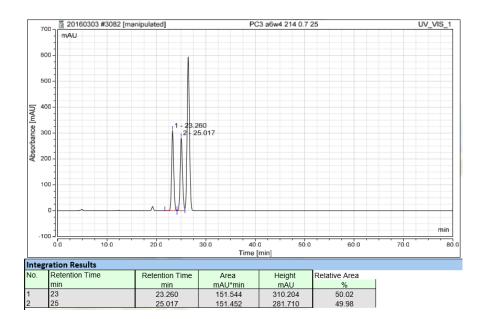


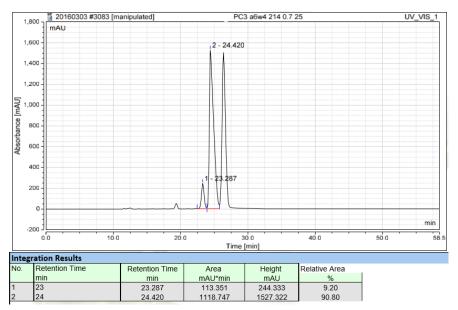
Figure S136: HPLC chromatographs of 3ac

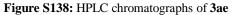


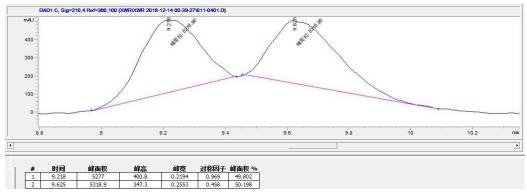
50 2 - 28.013 0min -50 -5.0 10.0 15.0 20.0 Time [min] 25.0 30.0 0.0 35.0 37.2 Integration Results Area mAU*min 54.881 3.508 Height mAU 97.872 Retention Time Retention Time Relative Area No min 26.350 28.013 min % 93.99 26 28 5.804 6.01

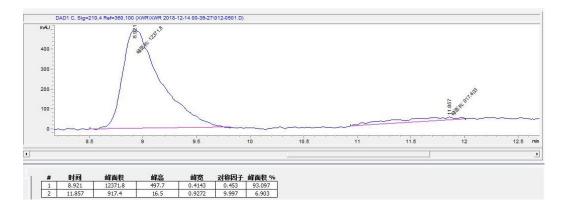
Figure S137: HPLC chromatographs of 3ad



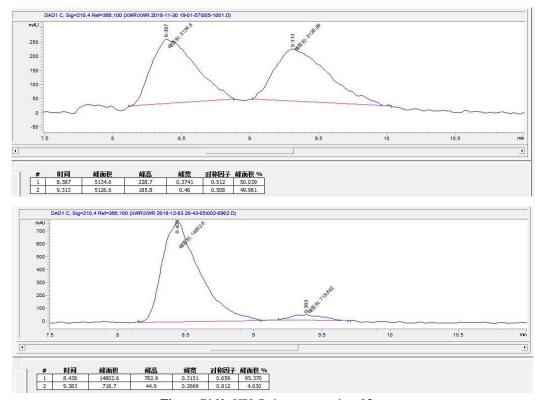


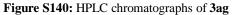


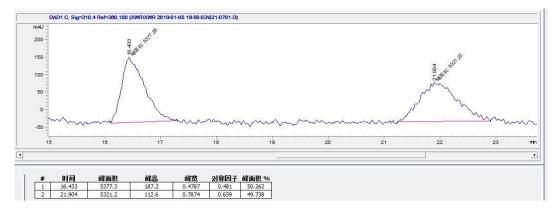


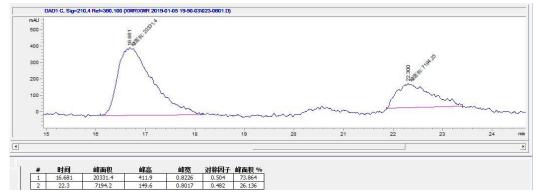


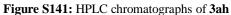


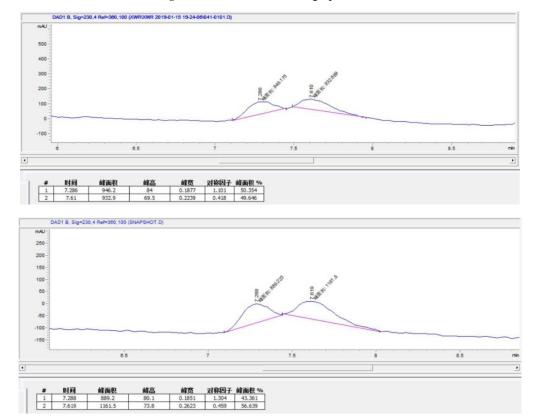


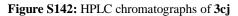


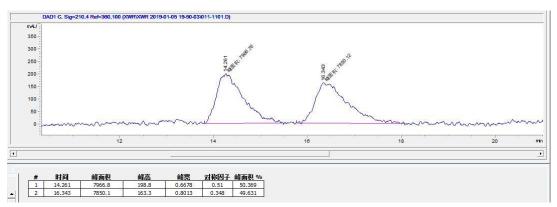












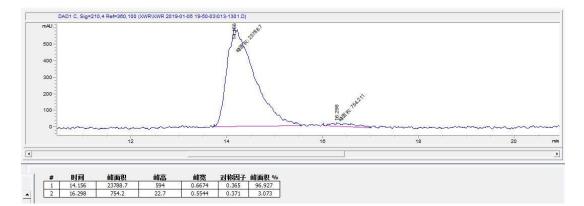
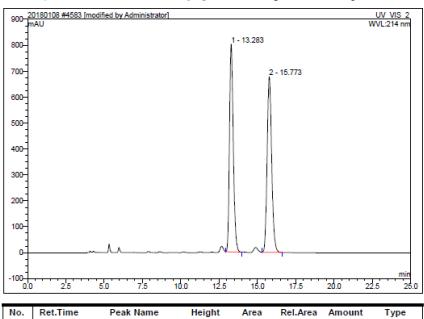


Figure S143: HPLC chromatographs of 3aa of gram-scale experiment



NO.	Ret. I ime	Peak Name	Height	Area	Rel.Area	Amount	Type
	min		mAU	mAU*min	%		
1	13.28	n.a.	801.713	229.791	49.63	n.a.	BMB*
2	15.77	n.a.	677.572	233.213	50.37	n.a.	BMB*
Total:			1479.286	463.004	100.00	0.000	

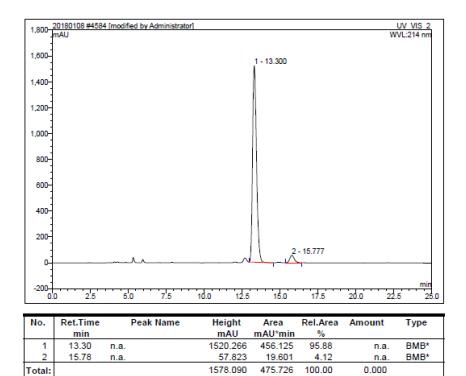


Figure S144: HPLC chromatographs of 5a