SUPPORTING INFORMATION

Mimicking Enzymes: Asymmetric Induction Inside a Carbamate-Based Steroidal Cleft

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SYNTHETIC PROCEDURES

General Considerations

All commercially available reagents and solvents were used without further purification unless otherwise stated. THF was distilled from sodium and bezophenone, DCM was distilled from anhydrous P_2O_5 , Hexane and Toluene were distilled from sodium ribbons. Triethyl amine was distilled from CaH₂ and it was stored over molecular sieves and nitrogen atmosphere. Phenyl isocyanate was distilled under reduced pressure before use.

Flash chromatography of reaction products was carried out using Silica 60A, particle size 230-400 micron (Merk). Analytical thin layer chromatography (TLC) was performed on DC-Alufolien Kieselgel 60F₂₅₄ 0.2 mm plates (Merk) and compounds were visualised by UV fluorescence or 5% phosphomolybdic acid in methanol.

¹H NMR, ¹³C NMR and ¹⁹F NMR spectra were recorded on a Bruker AC-300 or a Bruker AV-400 spectrometer, using deuterated solvents and were referenced internally to the residual solvent peak ($\delta_{\rm H} = 7.26$ ppm, $\delta_{\rm C} = 77.36$ ppm) signal.^[1] Coupling constants (*J*-values) are given in hertzs (Hz). The DEPT 135 technique was used to assign methylene (*C*H₂) signals. Chemical shifts are reported as follows: value (number of protons, description of absortion, coupling constant(s) where applicable, assignment). NMR spectra assignation was aided by comparison with literature values for similar compounds. Only clear identifiable peaks have been assigned.

Accurate mass of new compounds was measured in a high-resolution mass spectrometer (IMPACT II, BRUKER) with a quadrupole and a Time-Of-Flight (TOF) tube as analyzers, and a conventional Electrospray Ion Source (ESI). The equipment uses N₂ at the nebulization (2.4 Bar), and as drying gas (250 °C, 6.0 L/min). Mass spectra were acquired in *full scan* mode (4 eV) and positive ion polarity.

High pressure liquid chromatography (HPLC) analysis was performed on a HewlettPackard 1100 Series instrument equipped with a quaternary pump, using a Chiralcel AD-H column (250 x 4.6 mm) or Chiralcel OD–H column (250 x 4.6 mm). UV detection was monitored at 220 nm or at 210 nm.

^[1] Gottlieb, H. E.; Kotlyar, V.; Nudelman, A. J. Org. Chem. 1997, 62, 7512.

Nitroalkenes 3a-j.

Nitroalkenes **3a-f** and **3i-j**, were purchased from Aldrich. They were used without further purification. Other nitroalkenes (**3g** and **3h**) were prepared according to the following scheme, reproducing literature procedures:^{[2], [3]}



Standard procedure for the synthesis of enantio-enriched Michael adducts 8a-j (SP1)

Aromatic nitroalkene **3a-j** (0.2 mmol), and steroid **13** (0.03 mmol) were dissolved in dry toluene (1.6 mL), at -78 °C, before dimethyl malonate **2** (0.6 mmol) and dry triethyl amine (0.2 mmol) were added. The reaction mixtures were stirred for 72 h (unless otherwise stated), and then quenched with 2 mL NH₄Cl (aq. sat.). Organic materials were extracted with DCM (2 x 10 mL), dried (Na₂SO₄) and the volatiles were removed under vacuum. Crude products were filtered through a plug of silica gel (elution with Hex/EtOAC, 3:1) to render the desired adducts **4a-j** in analytically pure form.

Standard procedure for the synthesis of enantio-enriched Michael adduct 8a at 2 mmol scale

Nitrostyrene **3a** (298 mg, 2 mmol), and steroid **13** (234 mg, 0.3 mmol) were dissolved in dry toluene (16 mL), at -78 °C, inside a Schlenk flask under nitrogen atmosphere, before dimethyl malonate **2** (793 mg, 0.69 mL, 6 mmol) and dry triethyl amine (202 mg, 0.28 mL, 2 mmol) were added. The reaction mixture was stirred for 72 h, and then quenched with 20 mL NH₄Cl (aq. sat.). Organic materials were extracted with DCM (2 x 30 mL), dried (Na₂SO₄) and the volatiles were removed under vacuum. The crude product was filtered through a plug of silica gel (elution with Hex/EtOAC, 3:1) to render the desired adducts **4a** (533 mg, 95% yield) in analytically pure form and 91% *ee* (as determined by HPLC).

Standard procedure for the synthesis of racemic Michael adducts 4a-j.

^[2] Phukan, M.; Borah, K. J.; Borah, R. Synth. Commun. 2008, 38, 3068–3073.

^[3] Ferraro, A.; Bernardi, L.; Fochi, M. Adv. Synth. Catal. 2016, 358, 1561–1565.

Racemic products **4a-j**, necessary for HLPC analyses, were prepared according to a published synthetic procedure: Saidi, M. R.; Azizi, N.; Akbari, E.; Ebrahimi, F. *J. Mol. Cat. A* **2008**, *292*, 44–48.

SPECTROSCOPIC CHARACTERIATION AND HPLC CHROMATOGRAMS OF MICHAEL ADDUCTS 4a-j

(S)-Dimethyl 2-(2-nitro-1-phenylethyl)malonate (4a)^[4]



Prepared according to **SP1**. Purified by flash chromatography (Hexane/EtOAc, 3:1). Obtained as a white solid, 94% isolated yield (53 mg), in 91% *ee*; ¹H NMR (300 MHz, CDCl₃): δ (ppm) = 7.36–7.21 (5H, m, Ar*H*), 4.96–4.84 (2H, m, C*H*₂), 4.28–4.21 (1H, m, C*H*), 3.86 (1H, d, *J* = 9.1 Hz, C*H*), 3.76 (3H, s, OC*H*₃), 3.56 (3H, s, OC*H*₃); ¹³C NMR (75 MHz, CDCl₃): δ (ppm) = 168.2 (*C*O₂), 167.6 (*C*O₂), 136.4 (Ar*C*), 129.4 (Ar*C*H), 128.8 (Ar*C*H), 128.2 (Ar*C*H), 77.6 (*C*H₂) 55.1 (*C*H), 53.4 (*C*H₃), 53.2 (*C*H₃), 43.3 (*C*H).

^[4] H. Li, Y. Wang, L. Tang, L. Deng, J. Am. Chem. Soc. 2004, 126, 9906–9907.



Product **4a** was obtained in a maximum of 91% *ee*. The optical purity was determined by HPLC on a chiralcel AD-H column (hexane/2-propanol 90:10), flow rate 1.0 mL/min, $\lambda = 220$ nm, 25 °C.





Prepared according to **SP1**. Purified by flash chromatography (Hexane/EtOAc, 3:1). Obtained as a white solid, 84% isolated yield (50 mg), in 90% *ee*.; ¹H NMR (300 MHz, CDCl₃): δ (ppm) = 7.24–7.18 (2H, m, Ar*H*), 7.01 (2H, t, *J* = 8.6 Hz, Ar*H*), 4.94–4.79 (2H, m, C*H*₂), 4.27–4.19 (1H, m, C*H*), 3.83 (1H, d, *J* = 9.1 Hz, C*H*), 3.76 (3H, s, OC*H*₃), 3.57 (3H, s, OC*H*₃); ¹³C NMR (75 MHz, CDCl₃): δ (ppm) = 168.0 (*C*O₂), 167.4 (*C*O₂), 162.8 (d, *J* = 247.9 Hz, ArC), 132.2 (d, *J* = 3.7 Hz), 130.0 (d, *J* = 8.2 Hz), 116.4 (d, *J* = 21.7 Hz), 77.7 (*C*H₂), 55.0, 53.4, 53.3, 42.6 (*C*H); ¹⁹F NMR (282 MHz, CDCl₃): δ (ppm) = -113.1 (Ar*F*).

^[5] Watanabe, M.; Ikagawa, A.; Wang, H.; Murata, K.; Ikariya, T. *J. Am. Chem. Soc.* **2004**, *126*, 11148–11149.







¹⁹F NMR (282 MHz, CDCl₃):



Product **4b** was obtained in a maximum of 90% *ee*. The optical purity was determined by HPLC on a chiralcel AD-H column (hexane/2-propanol 90:10), flow rate 1.0 mL/min, $\lambda = 220$ nm, 25 °C.







Prepared according to **SP1**. Purified by flash chromatography (Hexane/EtOAc, 3:1). Obtained as a white solid, 99% isolated yield (62 mg), in 89% *ee*; ¹H NMR (300 MHz, CDCl₃): δ (ppm) = 7.30 (2H, d, J = 8.4 Hz, Ar*H*), 7.18 (2H, d, J = 8.4 Hz, Ar*H*), 4.94–4.80 (2H, m, C*H*₂), 4.26–4.18 (1H, m, C*H*), 3.82 (1H, d, J = 9.0 Hz, C*H*), 3.76 (3H, s, OC*H*₃), 3.59 (3H, s, OC*H*₃); ¹³C NMR (75 MHz, CDCl₃): δ (ppm) = 167.9 (*C*O₂), 167.4 (*C*O₂), 135.0 (Ar*C*), 134.8 (Ar*C*), 129.6 (Ar*C*H), 129.6 (Ar*C*H), 77.5 (*C*H₂) 54.8, 53.5, 53.3, 42.7.



Product **4c** was obtained in a maximum of 89% *ee*. The optical purity was determined by HPLC on a chiralcel AD-H column (hexane/2-propanol 90:10), flow rate 1.0 mL/min, $\lambda = 220$ nm, 25 °C.





Prepared according to **SP1**. Purified by flash chromatography (Hexane/EtOAc, 3:1). Obtained as a white solid, 80% isolated yield (58 mg), in 90% *ee*; ¹H NMR (300 MHz, CDCl₃): δ (ppm) = 7.46 (2H, d, J = 8.5 Hz, Ar*H*), 7.12 (2H, d, J = 8.5 Hz, Ar*H*), 4.94–4.80 (2H, m, C*H*₂), 4.25–4.17 (1H, m, C*H*), 3.82 (1H, d, J = 9.0 Hz, C*H*), 3.76 (3H, s, OC*H*₃), 3.59 (3H, s, OC*H*₃); ¹³C NMR (75 MHz, CDCl₃): δ (ppm) = 167.9 (*C*O₂), 167.4 (*C*O₂), 135.5 (Ar*C*), 132.6 (Ar*C*H), 129.9 (Ar*C*H), 122.9 (Ar*C*), 77.4 (*C*H₂), 54.8, 53.5, 53.3, 42.7 (*C*H).



Product **4d** was obtained in a maximum of 90% *ee*. The optical purity was determined by HPLC on a chiralcel AD-H column (hexane/2-propanol 90:10), flow rate 1.0 mL/min, $\lambda = 220$ nm, 25 °C.





Prepared according to **SP1**. Purified by flash chromatography (Hexane/EtOAc, 3:1). Obtained as a white solid, 76% isolated yield (47 mg), in 80% *ee*; ¹H NMR (300 MHz, CDCl₃): δ (ppm) = 7.14 (2H, d, J = 8.7 Hz, Ar*H*), 6.83 (2H, d, J = 8.7 Hz, Ar*H*), 4.92–4.78 (2H, m, C*H*₂), 4.22–4.15 (1H, m, C*H*), 3.82 (1H, d, *J* = 9.2 Hz, C*H*), 3.77 (3H, s, OC*H*₃), 3.75 (3H, s, OC*H*₃), 3.56 (3H, s, OC*H*₃); ¹³C NMR (75 MHz, CDCl₃): δ (ppm) = 168.2 (*C*O₂), 167.6 (*C*O₂), 159.8 (Ar*C*), 129.3 (Ar*C*H), 128.2 (Ar*C*), 114.7 (Ar*C*H), 78.0 (*C*H₂) 55.5, 55.2, 53.3, 53.2, 42.6 (*C*H).



Product **4e** was obtained in a maximum of 80% *ee*. The optical purity was determined by HPLC on a chiralcel AD-H column (hexane/2-propanol 90:10), flow rate 1.0 mL/min, $\lambda = 220$ nm, 25 °C.





Prepared according to **SP1**. Purified by flash chromatography (Hexane/EtOAc, 3:1). Obtained as a white solid, 65% isolated yield (38 mg), in 85% *ee*; ¹H NMR (300 MHz, CDCl₃): δ (ppm) = 7.11 (4H, m, Ar*H*), 4.94–4.81 (2H, m, C*H*₂), 4.24–4.16 (1H, m, C*H*), 3.84 (1H, d, *J* = 9.1 Hz, C*H*), 3.76 (3H, s, OC*H*₃), 3.57 (3H, s, OC*H*₃), 2.30 (3H, s, C*H*₃); ¹³C NMR (75 MHz, CDCl₃): δ (ppm) = 168.2 (*C*O₂), 167.6 (*C*O₂), 138.5 (Ar*C*), 133.3 (Ar*C*), 130.1 (Ar*C*H), 128.0 (Ar*C*H), 77.9 (*C*H₂) 55.2, 53.3, 53.2, 42.9 (*C*H), 21.4 (*C*H₃).





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Product **4f** was obtained in a maximum of 80% *ee*. The optical purity was determined by HPLC on a chiralcel AD-H column (hexane/2-propanol 90:10), flow rate 1.0 mL/min, $\lambda = 210$ nm, 25 °C.





Prepared according to **SP1**. Purified by flash chromatography (Hexane/EtOAc, 3:1). Obtained as a white solid, 98% isolated yield (64 mg), in 82% *ee*; ¹H NMR (300 MHz, CDCl₃): δ (ppm) = 8.20 (2H, d, J = 8.7 Hz, Ar*H*), 7.45 (2H, d, J = 8.7 Hz, Ar*H*), 5.00–4.88 (2H, m, C*H*₂), 4.41–4.33 (1H, m, C*H*), 3.88 (1H, d, J = 8.8 Hz, C*H*), 3.78 (3H, s, OC*H*₃), 3.61 (3H, s, OC*H*₃); ¹³C NMR (75 MHz, CDCl₃): δ (ppm) = 167.6 (*C*O₂), 167.1 (*C*O₂), 148.2 (Ar*C*), 143.9 (Ar*C*), 129.5 (Ar*C*H), 124.5 (Ar*C*H), 110.3 (*C*H₂), 54.5, 53.7, 53.5, 42.8 (*C*H).

^[6] Li, X.- J.; Liu, K.; Ma, H.; Nie, J.; Ma, J.- A. Synlett 2008, 3242–3246.

¹H NMR (300 MHz, CDCl₃):



Product **4g** was obtained in a maximum of 82% *ee*. The optical purity was determined by HPLC on a chiralcel AD-H column (hexane/2-propanol 80:20), flow rate 1.0 mL/min, $\lambda = 210$ nm, 25 °C.



(S)-Dimethyl 2-(2-nitro-1-(4-phenylphenyl)-ethyl)malonate (4h)^[7]



Prepared according to **SP1**. Purified by flash chromatography (Hexane/EtOAc, 3:1). Obtained as a white solid, 99% isolated yield (71 mg), in 85% *ee*; ¹H NMR (300 MHz, CDCl₃): δ (ppm) = 7.58–7.54 (4H, m, Ar*H*), 7.46–7.41 (2H, m, Ar*H*), 7.37–7.30 (3H, m, Ar*H*), 5.01–4.88 (2H, m, C*H*₂), 4.35–4.27 (1H, m, C*H*), 3.92 (1H, d, *J* = 9.0 Hz, C*H*), 3.78 (3H, s, OC*H*₃), 3.59 (3H, s, OC*H*₃); ¹³C NMR (75 MHz, CDCl₃): δ (ppm) = 168.1 (*C*O₂), 167.5 (*C*O₂), 141.4 (Ar*C*), 140.4 (Ar*C*), 135.4 (Ar*C*), 129.1 (Ar*C*H), 128.6 (Ar*C*H), 127.9 (Ar*C*H), 127.8 (Ar*C*H), 127.3 (Ar*C*H), 77.6 (*C*H₂), 55.0 (*C*H₃), 53.3 (*C*H₃), 53.2 (*C*H₃), 42.9 (*C*H).

^[7] Yang, L.; Zhao, L.; Zhou, Z.; He, C.; Sun, H.; Duan, C. Dalton Trans. 2017, 46, 4086–4092.



¹³C NMR (75 MHz, CDCl₃):



Product **4h** was obtained in a maximum of 85% *ee*. The optical purity was determined by HPLC on a chiralcel AD-H column (hexane/2-propanol 85:15), flow rate 1.0 mL/min, $\lambda = 220$ nm, 25 °C.



(S)-Dimethyl 2-(2-nitro-1-(3-chlorophenyl)-ethyl)malonate (4i)[8]



Prepared according to **SP1**. Purified by flash chromatography (Hexane/EtOAc, 3:1). Obtained as a white solid, 96% isolated yield (61 mg), in 86% *ee*; ¹H NMR (300 MHz, CDCl₃): δ (ppm) = 7.28–7.26 (2H, m, Ar*H*), 7.24 (1H, broad, s, Ar*H*), 7.16–7.11 (1H, m, Ar*H*), 4.96–4.82 (2H, m, C*H*₂), 4.26–4.19 (1H, m, C*H*), 3.83 (1H, d, *J* = 8.8 Hz, C*H*), 3.77 (3H, s, OC*H*₃), 3.61 (3H, s, OC*H*₃); ¹³C NMR (75 MHz, CDCl₃): δ (ppm) = 167.9 (*C*O₂), 167.3 (*C*O₂), 138.6 (Ar*C*), 135.2 (Ar*C*), 130.6 (Ar*C*H), 129.0 (Ar*C*H), 128.5 (Ar*C*H), 126.4 (Ar*C*H), 77.3 (*C*H₂) 54.8, 53.5, 53.3, 42.8 (*C*H).

^[8] García–García, P.; Zagdoun, A.; Copéret, C.; Lesage, A.; Díaz, U.; Corma, A. *Chem. Sci.* **2013**, *4*, 2006–2012.



Product **4i** was obtained in a maximum of 86% *ee*. The optical purity was determined by HPLC on a chiralcel AD-H column (hexane/2-propanol 90:10), flow rate 1.0 mL/min, $\lambda = 220$ nm, 25 °C.



(R)-Dimethyl 2-(1-(furan-2-yl)-2-nitroethyl)malonate (4j)^[4]



Prepared according to **SP1**. Purified by flash chromatography (Hexane/EtOAc, 3:1). Obtained as a white solid, 71% isolated yield (38 mg), in 73% *ee*; ¹H NMR (300 MHz, CDCl₃): δ (ppm) = 7.35–7.34 (1H, m, Ar*H*), 6.29 (1H, dd, *J* = 3.3, 1.9 Hz, Ar*H*), 6.22 (1H, d, *J* = 3.3 Hz, Ar*H*), 4.96–4.84 (2H, m, C*H*₂), 4.42–4.35 (1H, m, C*H*), 3.94 (1H, d, *J* = 7.8 Hz, C*H*), 3.76 (3H, s, OC*H*₃), 3.69 (3H, s, OC*H*₃); ¹³C NMR (75 MHz, CDCl₃): δ (ppm) = 167.8 (*C*O₂), 167.5 (*C*O₂), 149.7 (Ar*C*), 143.2 (Ar*C*H), 110.9 (Ar*C*H), 108.8 (Ar*C*H), 75.6 (*C*H₂), 53.4, 53.4, 53.0, 37.2 (*C*H).



Product **4j** was obtained in a maximum of 71% *ee*. The optical purity was determined by HPLC on a chiralcel OD-H column (hexane/2-propanol 90:10), flow rate 1.0 mL/min, $\lambda = 220$ nm, 25 °C.


SYNTHETIC PROCEDURES AND SPECTROSCOPIC CHARACTERIZATION OF CATALYSTS 5-17



Catalyst 9 (methyl cholate).^[9]



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cholic acid (1)

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^[9] Fieser, L. F.; Rajagopalan, S. J. Am. Chem. Soc. 1950, 72, 5530–5536.

^[10] del Amo, V.; Siracusa, L.; Markidis, T.; Baragaña, B.; Bhattarai, K.- M.; Galobardes, M.; Naredo, G.; Pérez-Payán, M. N.; Davis, A. P. *Org. Biomol. Chem.* **2004**, *2*, 3320–3328.

Catalyst 7.



Methyl cholate, 5 (300 mg, 0.71 mmol) and DMAP (8.6 mg, 0.07 mmol) were dissolved in dry THF (5 mL) and dry NEt₃ (0.15 mL, 109 mg, 1.07 mmol). To the former solution was added benzoyl chloride (0.11 mL, 130 mg, 0.92 mmol), dissolved in 2 mL of dry THF. The resulting mixture was refluxed for 24 h before the solvent and volatiles were evacuated and the resulting crude material was extracted from EtOAc (15 mL), washed with 0.1 M HCl (aq., sat., 10 mL) and dried (MgSO₄). Flash chromatography (Hex/EtOAc, 3:1) afforded diol 7 (204 mg, 55% yield) as a white solid. ¹H-NMR (400 MHz, CDCl₃): δ (ppm) = 8.03 (2H, d, J = 7.4 Hz, ArCH), 7.53 (1H, t, J = 7.1 Hz, ArCH), 7.41 (2H, t, J = 7.5 Hz, ArCH), 4.83 (1H, m, CHOR), 4.01 (1H, s, CHOH), 3.88 (1H, m, CHOH), 2.48 (1H, q, J = 12.6 Hz, CH), 2.40–2.21 (3H, m, CH + CH₂), 0.99 (3H, d, J = 5.7 Hz, CH₃), 0.94 (3H, s, CH₃), 0.71 (3H, s, CH₃); ¹³C-NMR (100 MHz, CDCl₃): δ $(ppm) = 175.5 (CO_2CH_3), 116.5 (PhCO_2), 133.0 (ArCH), 131.2 (ArC), 129.9 (ArCH),$ 128.5 (ArCH), 75.2 (CHOR), 73.3 (CHOH), 68.6 (CHOH), 51.9 (CO₂CH₃), 47.6 (CH), 46.9 (C), 42.4 (CH), 41.6 (CH), 39.8 (CH), 35.6 (CH₂), 35.5 (CH), 35.2 (CH₂), 35.1 (C), 34.8 (CH₂), 31.4 (CH₂), 31.2 (CH₂), 28.7 (CH₂), 27.8 (CH₂), 27.1 (CH), 23.5 (CH₂), 22.8 (CH_3) , 17.7 (CH_3) , 12.9 (CH_3) ; MS (ESI+): m/z (%) = 1075 (40) [M+M+Na]+, 549 (100) [M+Na]⁺; HRMS (ESI+): *m/z* calcd. for [C₃₂H₄₆O₆Na]⁺ 549.3192, found 549.3187.





¹³C NMR (75 MHz, CDCl₃):



Mass Spectrum SmartFormula Report

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Analysis Info Analysis Name

Sample Name

Method

Acquisition Date 16/04/2018 9:32:45

Operator Demo User

Instrument impact II 1825265.10101

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Intens										+MS,	0.7-2.4mi	ו #42 -	·141
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0.8-													
0.6-													
0.4-								1075 <mark>,</mark> 6	484				
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0.0-	200	h	400		600	800	. 10	000	,	1200	140	00	m/z
Meas.	m/z #	lon Fo	ormula	m/z	err [ppm]	mSigma	# mSigma	Score	rdb	e [−] Conf	N-Rule	Add	luct
549.3	3194 1	C32H4	6NaO6	549.3187	-1.4	67.9	1	100.00	10.0	even	ok	M+N	la
1075.0	6484 1	C64H9	2NaO12	1075.6481	-0.3	9.4	2	100.00	19.0	even	ok	2M+	

Catalyst 8.



Methyl cholate, 5 (300 mg, 0.71 mmol) and DMAP (8.6 mg, 0.07 mmol) were dissolved in dry THF (5 mL) and dry NEt₃ (0.15 mL, 109 mg, 1.07 mmol). To the former solution was added pivaloyl chloride (0.11 mL, 111 mg, 0.92 mmol), dissolved in 2 mL of dry THF. The resulting mixture was refluxed for 24 h before the solvent and volatiles were evacuated and the resulting crude material was extracted from EtOAc (15 mL), washed with 0.1 M HCl (aq., sat., 10 mL) and dried (MgSO4). Flash chromatography (Hex/EtOAc, 3:1) afforded diol 8 (220 mg, 61% yield) as a white solid. 1H-NMR (400 MHz, CDCl₃): δ (ppm) = 4.56–4.48 (1H, m, CHOR), 3.98 (1H, m, CHOH), 3.84 (1H, m, CHOH), 3.66 (3H, s, CO_2CH_3), 2.40–2.19 (4H, m, $CH + CH_2$), 1.15 (9H, s, $C(CH_3)_3$), 0.96 (3H, d, J = 6.2 Hz, CH₃), 0.89 (3H, s, CH₃), 0.69 (3H, s, CH₃); ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) = 178.6 (*t*Bu*C*O₂), 175.1 (*C*O₂CH₃), 74.3 (*C*HOR), 73.4 (*C*HOH), 68.7 (CHOH), 51.9 (CO₂CH₃), 47.5 (CH), 46.9 (C), 42.4 (CH), 41.6 (CH), 39.7 (CH), 38.9 (C), 35.6 (CH), 35.4 (CH₂), 35.2 (CH₂), 35.0 (C), 34.9 (CH₂), 31.3 (CH₂), 31.2 (CH₂), 28.7 (CH₂), 27.8 (CH₂), 27.5 (C(CH₃)₃), 27.0 (CH), 27.0 (CH₂), 23.5 (CH₂), 22.8 (CH₃), 17.7 (CH_3), 12.8 (CH_3); MS (ESI+): m/z (%) = 545 (58) [M+K]⁺, 529 (48) [M+Na]⁺; HRMS (ESI+): *m/z* calcd. for [C₃₀H₅₀O₆Na]⁺ 529.3505, found 529.3500.

¹H NMR (400 MHz, CDCl₃):







Mass Spectrum SmartFormula Report

Analysis Info

D:\Data\Pablo\30042018_MNF123.d Analysis Name Method MS 50-1000 orgánica.m 30042018_MNF123 Sample Name

Acquisition Date 30/04/2018 15:04:23

Operator Demo User Instrument impact II

1825265.10101



Catalyst 9.[11]



Catalyst 10.[12]



^[11] Wilson, C. P.; Webb, S. J. Chem. Commun. 2008, 4007–4009.

^[12] Zhang, Q.; Ma, X.; Ward, A.; Hong, W.- X.; Jaakola, V.- P.; Stevens, R. C.; Finn, M. G.; Chang, G. Angew. Chem. Int. Ed. 2007, 46, 7023–7025.

Catalyst 11.[13]







^[13] Májer, F.; Sharma, R.; Mullins, C.; Keogh, L.; Phipps, S.; Duggan, S.; Kelleher, D.; Keely,

S.; Long, A.; Radics, G.; Wang, J.; Gilmer, J. F. Bioorg. Med. Chem. 2014, 22, 256–268.

Catalyst 13.



Methyl cholate, 5 (800 mg, 1.90 mmol) was dissolved in dry DCM (8 mL) inside a Slenck flask, under nitrogen atmosphere. Two drops of freshly distilled TMSCI were added, followed by phenyl isocyanate (1.02 g, 0.93 mL, 8.55 mmol). The resulting mixture was refluxed for 24 h. before the solvent and volatiles were removed under vacuum, with the aid of a vacuum trap. To the resulting crude material 20 mL of DCM were added. Insoluble stuff was removed by filtration. The liquors were concentrated to dryness and the resulting white foam was purified by flash chromatography (DCM/1% EtOAc to DCM/2% EtOAc) to afford triscarbamate 13 (1.06 g, 72% yield) as a white solid. ¹H-NMR (400 MHz, CDCl₃): δ (ppm) = 7.44–7.39 (4H, m, Ar*H*), 7.34–7.22 (8H, m, ArH), 7.09-6.99 (3H, m, ArH), 6.80 (1H, broad s, NH), 6.66 (1H, broad s, NH), 6.54 (1H, broad s, NH), 5.11 (1H, s, CHOR), 4.96 (1H, s, CHOR), 4.52 (1H, m, CHOR), 3.61 (3H, s, CO₂CH₃), 2.34–2.27 (1H, m, CH), 2.22–1.05 (23H, m, CH₂s + CHs), 0.94 (3H, s, CH₃), 0.87 (3H, d, J = 6.5 Hz, CH₃), 0.76 (3H, s, CH₃); ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) = 174.9 (*C*O₂CH₃), 153.4 (O*C*(O)NH), 153.3 (2 x O*C*(O)NH), 138.4 (Ar*C*), 138.3 (ArC), 138.3 (ArC), 129.4 (ArCH), 129.4 (ArCH), 129.3 (ArCH), 123.8 (ArCH), 123.8 (ArCH), 123.7 (ArCH), 119.3 (broad C, 2 x ArCH), 119.0 (ArCH), 76.9 (CHOR), 75.5 (CHOR), 72.5 (CHOR), 51.8 (CO₂CH₃), 47.8 (CH), 45.7 (C), 43.9 (CH), 41.1 (CH), 38.3 (CH), 35.3 (CH₂), 35.0 (CH), 35.0 (CH₂), 34.7 (C), 31.8 (CH₂), 31.3 (CH₂), 31.1 (CH₂), 29.4 (CH), 27.5 (CH₂), 27.4 (CH₂), 26.1 (CH₂), 23.3 (CH₂), 21.8 (CH₃), 17.9 (CH₃), 12.6 (CH_3) ; MS (ESI+): m/z (%) = 818 (24) [M+K]⁺, 802 (100) [M+Na]⁺, 780 (4) [M+H]⁺; HRMS (ESI+): *m/z* calcd. for [C₄₆H₅₇N₃O₈Na]⁺ 802.4043, found 802.4038.

Catalyst **13** was lyophilized twice prior to use, as it includes water and EtOAc inside its cavity.



¹³C NMR (100 MHz, CDCl₃):



Mass Spectrum SmartFormula Report

Analysis Info Acquisition Date 17/04/2018 9:32:03 Analysis Name D:\Data\Pablo\116042018_VAS_1008.d MS 50-1000 orgánica.m Method Operator Demo User Sample Name 116042018_VAS_1008 Instrument impact II 1825265.10101 Comment Acquisition Parameter Source Type ESI Ion Polarity Positive Set Nebulizer 2.4 Bar Focus Scan Begin Scan End Active 50 m/z 1200 m/z Set Capillary Set End Plate Offset 4500 V -500 V 2000 V 250 °C 6.0 l/min Set Dry Heater Set Dry Gas Set Divert Valve Set APCI Heater Set Charging Voltage Waste Set Corona 0 nA 0 °C Intens. x10⁶ +MS, 1.4-3.2min #84-186 802.4034 1.25 1.00-0.75 235.0839 0.50 360.3234 0.25 587.5488 102.1275 0.00 600 1000 200 400 800 m/ż Meas. m/z # Ion Formula m/z err [ppm] mSigma # mSigma Score rdb e⁻ Conf N-Rule Adduct 780.4219 1 C46H58N3O8 780.4218 -0.0 8.3 2 100.00 20.0 even ok M+H C46H57N3NaO8 2 2 802.4034 1 802.4038 0.5 9.6 100.00 20.0 even ok M+Na

0.1

7.2

100.00

20.0

even

ok M+K

818.3776

1

C46H57KN3O8

818.3777

Catalyst 14.^[14]



^[14] Fang, L.; Chan, W.- H.; He, Y.- B.; Kwong, D. W. J.; Lee, A. W. M. *J. Org. Chem.* **2005**, *70*, 7640–7646.

Catalyst 15.



Diol 10 (66 mg, 0.182 mmol) was dissolved in dry DCM (1 mL) inside a Slenck flask, under nitrogen atmosphere. One drop of freshly distilled TMSCI was added, followed by phenyl isocyanate (48 mg, 44 μ L, 0.40 mmol). The resulting mixture was refluxed for 24 h. before the solvent and volatiles were removed under vacuum, with the aid of a vacuum trap. The resulting white foam was purified by flash chromatography (Hexane/DCM, 1:1) to afford biscarbamate 15 (63 mg, 58% yield) as a white solid. ¹H-NMR (400 MHz, CDCl₃): δ (ppm) = 7.47–7.44 (4H, m, ArH), 7.36–7.32 (4H, m, ArH), 7.09 (2H, q, J = 6.5 Hz, ArH), 6.65 (1H, s, NH), 6.56 (1H, s, NH), 5.15 (1H, s, CHOR), 4.94 (1H, s, CHOR), 0.93 (3H, s, CH₃), 0.88 (3H, d, J = 6.5 Hz, CH₃), 0.83 (3H, t, J = 6.9 Hz, CH₃), 0.78 (3H, s, CH₃); ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) = 153.5 (2 x OC(O)NH), 138.5 (2 x ArC), 129.3 (ArCH), 123.7 (ArCH), 119.1 (broad C, 2 x ArCH), 77.2 (CHOR), 72.9 (CHOR), 48.3 (CH), 45.6 (C), 43.9 (CH), 43.0 (CH), 38.4 (CH₂), 38.3 (CH), 37.5 (CH₂), 35.4 (C), 35.3 (CH), 32.2 (CH₂), 30.0 (CH₂), 29.3 (CH), 27.8 (CH₂), 27.6 (CH₂), 26.1 (CH₂), 23.6 (CH₃), 23.3 (CH₂), 21.7 (CH₂), 19.5 (CH₂), 18.2 (CH_3) , 14.8 (CH_3) , 12.6 (CH_3) ; MS (ESI+): m/z (%) = 639 (11) $[M+K]^+$, 623 (100) [M+Na]⁺; HRMS (ESI+): *m/z* calcd. for [C₃₈H₅₂N₂O₄Na]⁺ 623.3825, found 623.3811.



Mass Spectrum SmartFormula Report

Analysis Info

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Acquisition Date 10/3/2018 1:45:04 PM

Operator Demo Us Instrument impact II Demo User

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Acquisition F	Param	nete	er										
Source Type Focus Scan Begin Scan End		E A 50	SI ctive) m/z 500 m/z	lon Pola Set Capi Set End Set Char Set Corc	rity Ilary Plate ging na	Offset Voltage	Posi 4500 -500 e 2000 0 nA	tive V V V	S S S S	et Neb et Dry et Dry et Dive et APC	ulizer Heater Gas rt Valve I Heater	2.4 Ba 250 °(6.0 l/r Waste 0 °C	ar C nin Ə
Intens x10 ⁴											+MS, 3	3.3-4.6min	#193-264
-							623,3	820					
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2-				614.5718	3	619.52	272						
	603.38	72	607.4698							e	35.5012 6	39 ,3 559	
600		60	5 610	615		62	0	625	630		635	640	m/z
Meas.	m/z	#	Ion Formula	m/z	err [ppm]	mSigma	# mSigma	Score	rdb	e⁻ Conf	N-Rule	Adduct
601.3	3992	1	C38H53N2O4	601.4000		1.3	502.5	1	100.00	14.0	even	ok	M+H
623.0	3820	1	C38H52N2NaO4	623.3819		-0.1	47.0	1	100.00	14.0	even	ok	M+Na
639.0	3559	1	C38H52KN2O4	639.3559		-0.1	10.8	1	100.00	14.0	even	ok	M+K

Catalyst 16.^[15]



^[15] Zhao, Z.- G.; Liu, X.- L.; Chen, Y.; Shi, Z.- C. J. Chem. Res. 2010, 34, 481–484.

Catalyst 17.



Methyl chenodeoxycholate, 12 (240 mg, 0.59 mmol) was dissolved in dry DCM (2.5 mL) inside a Slenck flask, under nitrogen atmosphere. One drop of freshly distilled TMSCI was added, followed by phenyl isocyanate (155 mg, 0.14 mL, 1.30 mmol). The resulting mixture was refluxed for 24 h. before the solvent and volatiles were removed under vacuum, with the aid of a vacuum trap. The resulting white foam was purified by flash chromatography (DCM/1% EtOAc) to afford biscarbamate 17 (215 mg, 56% yield) as a white solid. ¹H-NMR (400 MHz, CDCl₃): δ (ppm) = 7.45 (2H, d, J = 7.9 Hz, ArH), 7.37 (2H, d, J = 8.0 Hz, ArH), 7.33–7.26 (4H, m, ArH), 7.05 (2H, q, J = 7.1 Hz, ArH), 6.75 (1H, broad s, NH), 6.58 (1H, broad s, NH), 4.89 (1H, s, CHOR), 4.57 (1H, m, CHOR), 3.65 (3H, s, CO₂CH₃), 2.39–2.32 (1H, m, CH), 2.26–2.18 (1H, m, CH), 0.93 $(3H, s, CH_3)$, 0.92 $(3H, d, J = 6.5 Hz, CH_3)$, 0.65 $(3H, s, CH_3)$; ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) = 175.0 (CO₂CH₃), 153.7 (OC(O)NH), 153.5 (OC(O)NH), 138.7 (ArC), 138.3 (ArC), 129.4 (ArCH), 129.2 (ArCH), 123.6 (ArCH), 123.4 (ArCH), 118.9 (ArCH), 118.8 (ArCH), 75.7 (CHOR), 72.4 (CHOR), 56.0 (CO₂CH₃), 51.8 (CH), 50.7 (CH), 42.9 (C), 40.9 (CH), 39.6 (CH₂), 38.3 (CH), 35.5 (CH), 35.4 (CH₂), 35.2 (CH₂), 34.9 (C), 34.3 (CH), 32.0 (CH₂), 31.3 (CH₂), 31.3 (CH₂), 28.3 (CH₂), 27.5 (CH₂), 23.8 (CH₂), 22.8 (*C*H₃), 20.9 (*C*H₂), 18.6 (*C*H₃), 12.0 (*C*H₃); MS (ESI+): *m/z* (%) = 683 (12) [M+K]⁺, 667 (100) [M+Na]⁺, 645 (4) [M+H]⁺; HRMS (ESI+): *m/z* calcd. for [C₃₉H₅₂N₂O₆Na]⁺ 667.3723, found 667.3718.





Mass Spectrum SmartFormula Report

Analysis Info

D:\Data\03102018_VAS_1048_P.d Analysis Name MS 50-1000 orgánica.m Method Sample Name 03102018_VAS_1048_P

Acquisition Date 10/3/2018 1:37:37 PM

Demo User Operator Instrument impact II

1825265.10101

Comment



Initial screening:

From the very first attempts, under any experimental conditions, triscarbamate **13** proved to be superior respect to structures **5-12**, **14-17**. Initially the Michael-type addition reaction was attempted employing DCM as the solvent.

Table SI_1. Initial screening of potential steroid-based catalysts **5-11**, **13**, **14**, **16**, in the Michaeltype addition of dimethyl malonate, **2**, to nitrostyrene, **3a**.^[a]

<>> NO ₂	NEt ₃ (0.1 equiv	.) MeO ₂ C	CO ₂ Me
Ph ² 2 3a	DCM (dry) –78°C, 72h	Ph	^{··} NO ₂ 4a
oid conv	ersion (%) ^[b]	<i>ee</i> (%) ^[c]	-
	25	rac	-
	14	rac	
	17	rac	
	21	rac	
	23	rac	
	14	rac	
	23	rac	
	78	54	
	77	10	•
	56	3	
	Ph 3a	$ \begin{array}{c} \text{Ph} & \text{NO}_2 \\ 3a & \begin{array}{r} \text{NEt}_3 (0.1 \text{ equiv} \\ \text{DCM (dry)} \\ -78^\circ \text{C}, 72h \\ \end{array} $ $ \begin{array}{c} \text{id} & \text{conversion (\%)}^{[b]} \\ \hline 25 \\ 14 \\ 17 \\ 21 \\ 23 \\ 14 \\ 23 \\ \hline 78 \\ 77 \\ 56 \\ \end{array} $	$\frac{\text{NEt}_{3} (0.1 \text{ equiv.})}{\text{3a}} \xrightarrow{\text{NEt}_{3} (0.1 \text{ equiv.})} \text{MeO}_{2}C} \xrightarrow{\text{MeO}_{2}C} \frac{1}{\text{DCM (dry)}} \xrightarrow{\text{Ph}_{-78^{\circ}C, 72h}} \text{Ph}_{-78^{\circ}C, 72h}$

[a] General reaction conditions: nitrostyrene **3a** (0.2 mmol), and steroid (0.02 mmol) were dissolved in dry DCM (0.4 mL), at -78 °C, before dimethyl malonate **2** (0.6 mmol) and dry NEt₃ (0.02 mmol) were added. The reaction mixtures were stirred for 72 h before quenching with NH₄Cl (aq., sat.).

[b] Conversion of limiting reagent **3a** into product **4a**, as determined by ¹H NMR spectroscopy on crude reaction mixtures. Resonances of **4a** were integrated and quantified against a signal of the corresponding steroid.

[c] Enantiomeric excess of product **4a** as determined by chiral HPLC from crude reaction mixtures.

Table SI_2. Screening of solvents in the Michael-type addition of dimethyl malonate, **2**, to nitrostyrene, **3a** catalyzed by steroidal triscarbamate **13**.^[a]

MeO _o C	CO₀Me + ◇	13 (10 r NEt ₃ (0.1	nol%) equiv.) N	∕leO₂C CO₂Me
2	3; 3;	a solvent (dry –78°C,	/) (0.5 M) 48h	Ph ^{··} NO ₂ 4a
	solvent	conversion (%) ^[b]	<i>ee</i> (%) ^[c]	
	toluene	40	70	
	hexane	10	14	
	CHCl₃	10	30	
	THF	4	rac	
	Et ₂ O	52	12	
	<i>p</i> -xilene ^[d]	>99	25	
	Benzene ^[d]	>99	32	
	Chlorobenzene ^[e]	>99	36	

[a] General reaction conditions: nitrostyrene **3a** (0.2 mmol), and steroid **13** (0.02 mmol) were dissolved in dry solvent (0.4 mL), at -78 °C, before dimethyl malonate **2** (0.6 mmol) and dry NEt₃ (0.02 mmol) were added. The reaction mixtures were stirred for 48 h before quenching with NH₄Cl (aq., sat.).

[b] Conversion of limiting reagent **3a** into product **4a**, as determined by ¹H NMR spectroscopy on crude reaction mixtures. Resonances of **4a** were integrated and quantified against a signal of the steroid.

[c] Enantiomeric excess of product **4a** as determined by chiral HPLC from crude reaction mixtures.

[d] Reaction carried out at 0 °C.

[e] Reaction carried out at -20 °C.

Table SI_3. Screening of different bases in the Michael-type addition of dimethyl malonate, **2**, to nitrostyrene, **3a** catalyzed by steroidal triscarbamate **13**.^[a]

MeO _a C	CO₂Me ± ∧	13 (10 m base (0.1 ∧ .NO₂	nol%) equiv.) Me	O ₂ C CO ₂ Me
2	Ph ²	3a toluene (dry) -78°C,) (0.5 M) 48h	Ph ^{··} NO ₂ 4a
	base	conversion (%) ^[b]	<i>ee</i> (%) ^[c]	
	DIPEA	21	43	
	2,6-lutidine	0	-	
	K ₂ CO ₃	55	rac	
	DBU	91	27	
	Pyridine	0	-	
	NEt ₃	40	70	

[a] General reaction conditions: nitrostyrene **3a** (0.2 mmol), and steroid **13** (0.02 mmol) were dissolved in dry toluene (0.4 mL), at -78 °C, before dimethyl malonate **2** (0.6 mmol) and dry base (0.02 mmol) were added. The reaction mixtures were stirred for 48 h before quenching with NH_4CI (aq., sat.).

[b] Conversion of limiting reagent **3a** into product **4a**, as determined by ¹H NMR spectroscopy on crude reaction mixtures. Resonances of **4a** were integrated and quantified against a signal of the steroid.

[c] Enantiomeric excess of product **4a** as determined by chiral HPLC from crude reaction mixtures.

Table SI_4. Screening of reaction concentration and amount of base employed in the Michaeltype addition of dimethyl malonate, **2**, to nitrostyrene, **3a** catalyzed by steroidal triscarbamate **13**.^[a]

M-0.0 00 M-	NO	13 (10 mol%) NEt ₃ (XX equiv.)	MeO ₂ C <u>CO</u> 2Me
меО ₂ С СО ₂ ме 2	+ Ph 3a	toluene (dry) (XX M) –78°C, 48h	Ph ⁻ NO ₂ 4a
[] (M) ^[b]	NEt ₃ (mol%) ^[c]	conversion (%) ^[d]	$ee~(\%)^{[e]}$
0.5	20	71	75
0.25	10	45	79
0.25 ^[f]	20	92	80
0.25[1]	30	>99	78
0.1251	100	99	83

[a] General reaction conditions: nitrostyrene **3a** (0.2 mmol), and steroid **13** (0.02 mmol) were dissolved in dry toluene (stated amount), at -78 °C, before dimethyl malonate **2** (0.6 mmol) and dry NEt₃ (stated amount) were added. The reaction mixtures were stirred for 48 h before quenching with NH₄Cl (aq., sat.).

[b] Concentration of nitrostyrene 3a (limiting reagent) in dry toluene.

[c] Amount of NEt₃ used in the reaction, referred as %mol respect to nitrostyrene 3a.

[d] Conversion of limiting reagent **3a** into product **4a**, as determined by ¹H NMR spectroscopy on crude reaction mixtures. Resonances of **4a** were integrated and quantified against a signal of the steroid.

[e] Enantiomeric excess of product **4a** as determined by chiral HPLC from crude reaction mixtures.

[f] The reactions mixtures were stirred for 72 h.

Table SI_5. Study on the amount of steroid **13** employed in the Michael-type addition of dimethyl malonate, **2**, to nitrostyrene, **3a**.^[a]

MeO C			13 (ⅩⅩ n NEt ₃ (1 e	nol%) equiv.) N	∕leO₂C CO₂Me
2 vie	200 ₂ ivie	* Ph 3a	toluene (dry) –78°C,	(0.125 M) 72h	Ph NO ₂
-	Catalyst 1	17 (mol%) ^[b]	conversion (%) ^[c]	ee (%) ^{[d}	
-		10	97	83	
		15	99	91	
		20	99	90	

[a] General reaction conditions: nitrostyrene **3a** (0.2 mmol), and steroid **13** (stated amount) were dissolved in dry toluene (1.6 mL), at -78 °C, before dimethyl malonate **2** (0.6 mmol) and dry NEt₃ (0.2 mmol) were added. The reaction mixtures were stirred for 72 h before quenching with NH₄Cl (aq., sat.).

[b] Amount of steroid 13 used in the reaction, referred as %mol respect to nitrostyrene 3a.

[c] Conversion of limiting reagent **3a** into product **4a**, as determined by ¹H NMR spectroscopy on crude reaction mixtures. Resonances of **4a** were integrated and quantified against a signal of the steroid.

[d] Enantiomeric excess of product **4a** as determined by chiral HPLC from crude reaction mixtures.

Table SI_6. Re-screening of different bases, under the finest reaction conditions, in the Michael-type addition of dimethyl malonate, **2**, to nitrostyrene, **3a** catalyzed by steroidal triscarbamate **13**.^[a]

MaO C		In NO ₂	13 (15 m) base (1 e)	ol%) quiv.)	MeO ₂ C CO ₂ Me
2 vie		Ph 3a	toluene (dry) (–78°C, 7	0.125 M) 72h	Ph NO ₂ 4a
	base	e con	version (%) ^[b]	<i>ee</i> (%) ^{[d}	2]
	DIPE	A	24	70	
	Pyridir	ne	0	-	
	NEt ₃	i	99	91	

[a] General reaction conditions: nitrostyrene **3a** (0.2 mmol), and steroid **13** (0.03 mmol) were dissolved in dry toluene (1.6 mL), at -78 °C, before dimethyl malonate **2** (0.6 mmol) and dry base (0.2 mmol) were added. The reaction mixtures were stirred for 72 h before quenching with NH₄Cl (aq., sat.).

[b] Conversion of limiting reagent **3a** into product **4a**, as determined by ¹H NMR spectroscopy on crude reaction mixtures. Resonances of **4a** were integrated and quantified against a signal of the steoid **13**.

[c] Enantiomeric excess of product **4a** as determined by chiral HPLC from crude reaction mixtures.

MeO₂C.	.CO₂Me +	/NO ₂	<mark>5-17</mark> (15 mol%) NEt ₃ (1 equiv.)	MeC	0₂C _CO₂Me
	2	Ph'	Toluene (dry) –78°C, 72h		Ph ^{NO} ₂ 4a
	steroid	conversio	n (%) ^b	<i>ee</i> (%) ^c	
	5	5		-	
	6	7		-	
	7	6		-	
	8	6		-	
	9	5		-	
	10	4		-	
	11	8		-	
	12	0		-	
	13	99 (94	4)	91	
	14	5		-	
	15	11		-	
	16	4		-	
	17	14		-	

Table SI_7. Re-screening of the steroid-based catalysts **5-17**, under our finest reaction conditions, in the Michael-type addition of dimethyl malonate, **2**, to nitrostyrene, **3a**.^{*a*}

[a] General reaction conditions: nitrostyrene **3a** (0.2 mmol), and steroid **5-17** (0.03 mmol) were dissolved in dry toluene (1.6 mL), at -78 °C, before dimethyl malonate **2** (0.6 mmol) and dry triethyl amine (0.2 mmol) were added. The reaction mixtures were stirred for 72 h before quenching with NH₄Cl (aq., sat.).

[b] Conversion of limiting reagent **3a** into product **4a**, as determined by ¹H NMR spectroscopy on crude reaction mixtures. Resonances of **4a** were integrated and quantified against a signal of the steroid. In brackets is given the yield of isolated product **4a** in analytically pure form.

[c] Enantiomeric excess of analytically pure product 4a as determined by chiral HPLC.

NMR STUDY AND DETERMINATION OF ASSOCIATION CONSTANTS

General considerations

Association constants for the complexes were determined by means of the ¹H NMR titration method, using 400 MHz ¹H NMR spectroscopy at a temperature of 298 K. All solutions were made up in toluene- d_8 using volumetric flasks with an accuracy of \pm 0.02 mL. Samples were equilibrated at 298 K prior to use. The corresponding variation in chemical shift, with respect to guest concentration, was monitored and the association constant was determined by fitting this experimental data to the following equation, using the "Solver" tool implemented on Microsoft Excel, using non-linear curve-fittings:

$$\delta_{obs} = \delta_G + \frac{\Delta\delta}{2G_0} \left[K_d + H_0 + G_0 - \sqrt{\left(K_d + H_0 + S_0\right)^2 - 4H_0G_0} \right]$$

This allows the calculation of the dissociation constant, K_d , where δ_{obs} is the chemical shift at concentration G_0 of the guest molecule, H_0 is the concentration of the host molecule, and $\Delta\delta$ is the maximum change in the chemical shift of the bound species. The value of the association constant, K_a is obtained from the best-fit value of K_d :

$$K_d = \frac{1}{K_a}$$

Determination of binding constant for the complex [13·3a]

The association constant, between steroid **13** and nitrostyrene, **3a**, was determined by titration as 2-3 M⁻¹, using 400 MHZ ¹H NMR spectroscopy, at 298 K in toluene-*d*₈. Since **13** and **3a** are inert when mixed together we were able to assign any chemical shift changes as a consequence of the association between complementary recognition sites on **13** and **3a**. Considering the chemical shift manifested by the N-H motifs of the C3, C7 and C12 carbamates upon formation of the complex (Figure SI_1), it is assumed that nitrostyrene incorporates within the cavity of the steroid according to the general model represented in Figure SI_2 (see below).

Figure SI_1. Stacked representation of NMR spectra (toluene- d_8 , 298 K) for steroid **13** (50 mM) with increasing amounts of nitrostyrene **3a**, as stated.



Figure SI_2. Binding model for the formation of complex [13·3a].



For the binding study, the concentration of **3a** (guest) was varied between zero and 168 mM. Host **13** was maintained at 50 mM throughout the study. The observed change in chemical shift of the singlet at $\delta = 6.53$ ppm, or $\delta = 6.19$ ppm, from compound **13** was monitored as the concentration of guest varied (Figure SI_1). From the data presented in Figures SI_3 and SI_4, and making use of the equations presented above, the association constant K_a was calculated as 2–3 M⁻¹.

Figure SI_3. Saturation plot showing the change in chemical shift of the signal at δ = 6.535 ppm arising from compound **13** as the concentration of **3a** increases. The closed circles represent the experimentally determined points and the solid line represents the best-fit value.



Figure SI_4. Saturation plot showing the change in chemical shift of the signal at $\delta = 6.196$ ppm arising from compound **13** as the concentration of **3a** increases. The closed circles represent the experimentally determined points and the solid line represents the best-fit value.



Study of the lack of association between steroid 13 and reaction product 4a

The absence of association between steroid **13** and reaction product **4a** was determined by 400 MHZ ¹H NMR spectroscopy, at 298 K in toluene- d_8 . Since **13** and **4a** are inert when mixed together any chemical shift changes would be a consequence of the association between these two compounds. A 50 mM stock solution of steroid **13** in toluene- d_8 was prepared and a ¹H NMR spectrum registered. Subsequently, an equimolar amount of enantioenriched product **4a** (91% *ee*, prepared according to **SP1**)

was added to the previous sample and it was then analysed by ¹H NMR. No appreciable changes in chemical shift of the singlets at δ = 6.69 ppm and δ = 6.53 ppm from compound **13** (assignable to the NHs of the carbamate units borne on C7 and C12) was monitored (Figure SI_5). A negligible change was appreciated for the singlet at δ = 6.20 ppm (NH of carbamate at C3).

Figure SI_5. Stacked representation of NMR spectra (toluene- d_8 , 298 K) for steroid **13** (50 mM) and for a sample containing equimolar amounts of **13** and reaction product **4a**.



PROPOSED REACTION MECHANISM

Based on the experimental and theoretical findings, discussed later on, the following mechanism is put forward for asymmetric reaction between dimethyl malonate, **2**, and nitrostyrene, **3a**, catalyzed by the cavity steroidal triscarbamate **13**.

Figure SI_6. Proposed mechanism for the addition reaction between dimethyl malonate, 2, and nitrostyrene, 3a, catalyzed by the cavity of steroid 13. H-bonds are represented by green dashed lines.



[13•2'•3a]

QUANTUM CHEMICAL STUDY

Computational Details

All the chemical structures were optimized in the gas phase using the Minnesota functional M06-2X in combination with the cc-pvdz basis set as implemented in the Gaussian 09 program. Considering the low dielectric constant of the solvent employed (toluene) the effect of the solvent in the studied species was considered to be reasonably small. Hence, all the calculations were performed in the gas phase. All stationary states were characterized through their vibrational spectra. All critical points, except for the transition states, exhibit positive vibration armonic frequencies. Transition states were characterized as a first order saddle points showing only one negative frequency. The reaction energy profiles of the C–C bond-forming step within the studied reactions were computed using the Gaussian 09 program, after the optimization of the involved critical points using the calculation setup previously described. The binding energies have been properly corrected for Basis Set Superposition Errors (BSSE). All the thermodynamic properties were computed at both 195 and 298 K employing the geometry computed at room temperature.

General notes regarding the computational study

Considering the experimental procedure, the Michael-type addition reaction was considered to proceed *via* the initial formation of nitrostyrene-triscarbamate supramolecular species [**3a**•**13**], which then leads to a pre-reactive complex with the methyl malonate anion, [**2**'•**3a**•**13**]. This pre-reactive complex affords the corresponding addition product **4a** by the reaction of the styrene derivative and the nucleophile.

Selection of the functional and basis set

With the aim of validating the proposed calculation setup, a preliminary analysis was performed to test the validity and applicability of a set of functionals and basis sets on the studied systems. For that purpose, the structures of the cyclopentane perhydrophenanthrene starting material (cholic acid, **1**) obtained with a combination of different basis set (cc-pVDZ and 6-31G*) and functionals (B3LYP and M06-2X) employing the Gaussian 09 program, were compared with the experimental X-ray

structure (Figure SI_7).^[16] For all four possible combinations the total discrepancy between the experimental and the theoretical findings was found to be below 2.0% (Table SI_8). The minimum deviation in bond lengths was obtained using the M06-2X/cc-pVDZ combination (Table SI_8, entry 1). Therefore, this setup was considered to be optimal for our study.

Table SI_8. Error percentage in bond angles and bond lengths, as calculated employing different basis set and functionals with respect to the experimental values obtained from the X-ray structure of cholic acid, **1**.

entry	basis set	functional	angle (%)	length (%)	
1	cc-pVDZ	M06-2x	1.64%	0.88%	
2	6-31G*	M06-2x	1.29%	0.89%	
3	cc-pVDZ	B3LYP	1.16%	1.15%	
4	6-31G*	B3LYP	1.80%	1.21%	

X-ray structure of cholic acid.^[16]

XYZ coordinates (in Ångströms):

0	25.31227	3.04519	8.57345
0	23.43473	-0.96611	9.53248
0	19.28024	1.22889	8.77338
0	32.56856	2.33413	11.62547
0	31.22962	0.63377	11.22718
С	21.29945	2.73758	11.60517
С	20.69312	2.65333	10.19474
С	20.97109	0.16926	10.22754
С	21.61805	0.25814	11.60829
С	22.56754	-0.90428	11.83634
С	23.78728	-0.77057	10.91167
С	24.51022	0.56112	11.20375
С	23.52568	1.71581	10.85700
С	22.30978	1.59679	11.80041
С	24.30413	3.03591	10.85700
С	25.53190	2.98490	9.93546
С	26.47827	1.95309	10.46027
С	25.57400	0.50315	10.15101
С	26.68438	-0.50547	10.42122
С	27.99704	0.13294	9.86518
С	27.67772	1.64007	9.54654
С	28.95829	2.47788	9.74178
С	28.68784	3.94947	9.37785

[16] Nonappa; Lathinen, M.; Ikonen, S.; Kolehmainen, E.; Kauppinen, R. *Cryst. Growth Des.* **2009**, *9*, 4710–4719. Structure deposited in the Cambridge Crystallographic Data Centre (number: 989890).

С	30.16471	1.97860	8.93270
С	31.54155	2.31867	9.56216
С	31.77841	1.66171	10.81952
С	22.77020	1.70036	13.27800
С	26.95936	2.19037	11.88944
С	19.96464	1.30618	10.04636

Figure SI_7. Balls and sticks representation of the x-ray structure of cholic acid, rendered with Jmol.^[17] C atoms are colored grey. O atoms are colored red. H atoms have omitted for clarity.



^[17] Jmol: an open-source Java viewer for chemical structures in 3D. http://www.jmol.org/
Optimized structure of catalyst 13.

Considering the rigidity of the steroidal core, the x-ray resolved crystalline structure of cholic acid was employed as a starting point for the initial guess of the energetically minimized structure of the tripodal catalyst **13**, utilizing the M06-2X/cc-pVDZ set. The resulting minimized structure is represented in Figure SI_8.

Figure SI_8. Ball and stick representation of the optimized structure (M06-2X/cc-pVDZ) of triscarbamate **13** in the gas phase. C atoms are represented in grey, O in red, N in blue, H in white. H atoms borne on the carbamate units are represented in orange. The optimization of the structure was done with Gaussian09 code and rendered with Jmol.^[17]



Computational study for the formation of the supramolecular complex [13·3a]

In agreement with the NMR experiments presented above (Figures SI_1 to SI_4), DFT calculations (M06-2X/cc-pVDZ, run with the code of Gausian09) revealed that nitrostyrene **3a** docks inside the cavity of steroid **13** adopting two minimum energy conformations, differing on the orientation adopted by **3a**. Both supramolecular complexes [**13·3a**] were named after complex I and complex II (Figure SI_9). They are held together by H-bond contacts (between the nitro group of **3a**, and the N-Hs of the C3 and C7 cabamate units in complex I, and the C7 and C12 appendages in the case of complex II) and π - π stacking interactions. The net binding energies (Δ G) of the steroid **17** with nitrostyrene **3a** account for -6.75 *kcal mol*⁻¹ and -3.90 *kcal mol*⁻¹, for complex I and II respectively (at a temperature of 195 K).

Figure SI_9. Representations of the optimized structures of supramolecular complexes [**3a-13**] **I** and **II** (M06-2X/cc-pVDZ), rendered with Jmol. A) complex **I**. B) complex **II**. The steroidal backbone is coloured green (H atoms are omitted for clarity). C atoms are colored grey, O atoms are coloured red, N atoms are coloured blue, H atoms are coloured white. Carbamate H atoms are coloured orange. H-bonds are represented by pink dashed lines.



 ΔG values for the formation of complexes I and II [13·3a] were computed at different temperatures. From these figures equilibrium constants (K) for the process 13 + 3a $\leftarrow == \rightarrow [13\cdot3a]$ were inferred

$$K = e^{\frac{-\Delta G}{RT}}$$

and represented against T (Figure SI_10).

Figure SI_10. Computed equilibrium constants (K) at different temperatures, ranging from 195 K to 298 K, for the binding process between nitrostyrene **3a** and steroidal triscarbamate **13**. Points represent real computed values. Solid lines represent the best-fit values (blue for complex **I** [**3a**•**13**]; green for complex **II** [**3a**•**13**]).



Reaction energy profiles for the C–C bond-forming step

With the aim of studying the effect of the triscarbamate **13** in the Michael-type addition reaction, and rationalizing the stereopreference observed in the formation of product (*S*)-**4a**, the C–C bond-forming step was characterized and studied through computational calculations (Figures SI_11 and SI_12).

Figure SI_11. Representation of the reaction energy profile (PES along the C–C bond formation) of the non-catalyzed reaction. Activation energies are given at the DFT (M06-2X/cc-pVDZ) and HF/cc-pvdz level of theory.



Reaction coordinate

Figure SI_12. Representation of the reaction energy profile (PES along the C–C bond formation) of the catalyzed reaction. The activation energies are given at the DFT (M06-2X/cc-pVDZ) level of theory.



The calculations of the transition states are referred to the reaction between the anionic nucleophile **2**' and the electrophile **3a** leading to the formation of the corresponding addition product. The protonation of the final product, resulting in the formation of neutral species, is not accounted computationally. As it can be observed from the diagrams, a significant decrease in the activation energy of the catalyzed reaction was computed, proving the catalytic nature of the steroidal triscarbamate **13**. The corresponding activation energies are computed with respect to a reference pre-reactive complex.

Evaluation of the effect of catalyst 13 in the electrophilicity of nitrostyrene 3a

To assess the effect of the H-bond interactions existing in the supramolecular complex **II** [**3a·13**] in the electrophilic character of nitrostyrene **3a**, a frontier molecular orbital analysis was performed. This analysis revealed that, upon complexation, a significant increase in the reactivity of nitrostyrene (represented by the electrophillicity index, Figure SI_13) towards a nucleophilic attack is produced, which supports the catalytic activity of the catalyst. Hence, the catalytic effect of the triscarbamate **13** can be attributed to both, the increase in the reactivity of the electrophile and the preorganization phenomenon experienced inside its cavity.

Table SI_9. HOMO and LUMO orbital energies, electronic chemical potencial (μ), chemical hardness (η), and electrophiliicity index (ω),^[18] calculated for nitrostyrene **3a**, and for nitrostyrene **3a** within the supramolecular complex **II** [**3a-13**]. All values are given in eV.

	HOMO	LUMO	μ	η	ω
3a	-8.360	-1.638	-5.000	3.36	3.717
3a, in complex II [3a•13]	-7.067	-2.167	-4.617	2.45	4.350

Figure SI_13. Representation of the LUMO orbital for the nitrostyrene-triscarbamate supramolecular complex **II** [**3a·13**]. Isosurface value: 0.07 au. Rendered with the Avogadro program.^[19]



^[18] Chattaraj, P. K.; Sarkar, U.; Roy, D. R. Chem. Rev. 2006, 106, 2065–2091.

^{[19] (}a) Avogadro: an open-source molecular builder and visualization tool. http://avogadro.cc.

⁽b) Hanwell, M. D.; Curtis, D. E.; Lonie, D. C.; Vandermeersch, T.; Zurek, E.; Hutchison, G. R. *J. Cheminform.* **2012**, 4–17.

Insertion of MeOH inside the cavity of steroid 13

According to calculations, MeOH (one, two, or three units) can be incorporated to the cavity of triscarbamate **13**, as represented in Figures SI_14 to 16. It docks within the steroid by means of multiple H-bond contacts.

Figure SI_14. Representation of the supramolecular complex [**MeOH-13**], optimized with Gausian09 at the DFT (M06-2X/cc-pVDZ) level of theory. Rendered with the Jmol program.



Figure SI_15. Representation of the supramolecular complex [**2 MeOH·13**], optimized with Gausian09 at the DFT (M06-2X/cc-pVDZ) level of theory. Rendered with the Jmol program.



Figure SI_16. Representation of the supramolecular complex [**3 MeOH·13**], optimized with Gausian09 at the DFT (M06-2X/cc-pVDZ) level of theory. Rendered with the Jmol program.



Calculation of energies and thermodynamic parameters

Table SI_10. Electronic energies (E_0), enthalpies (H), entropy (S) and Gibbs free energies (G) at 298 K and 195 K (-78 °C) for the systems involved in our study. All values are given in *au*.

system	Eo	H (195 K)	S (195 K)	G (298 K)	G (195 K)
13	-2552.620429	-2551.718164	0.000413752	-2551.737243	-2551.798908
[3a•13] complex I	-3066.662374	-3066.662374	0.000441950	-3065.641421	-3066.662374
[3a·13] complex II	-3066.647996	-3065.615406	0.000468512	-3065.635836	-3065.706837
TS	-3562.240524	-3561.099979	0.000499962	-3561.117489	-3561.197546
[3a•13] complex II + 2'	-3562.267809	-3561.118617	0.000484270	-3561.130376	-3561.213122
3a	-513.9884249	-513.8596769	0.000206739	-513.885917	-513.8862829
4a (neutral)	-1010.132619	-1009.873548	0.000206739	-1009.905789	-1009.913893
4a (anionic)	-1009.557171	-1009,311916	0.000200502	-1009.343958	-1009.351044
2 (neutral)	-496.0506259	-495.9227629	0.000142463	-495.950963	-495.9505649
2' (anionic)	-495.5290465	-495.4153625	0.000137117	-495.44366	-495.4421195
[3a·13] complex II + 4a (anionic)	-3562.28973	-3561,138603	0.000490910	-3561.15207	-3561.234404
[3a·13] complex II + 4a (neutral)	-3562.804353	-3561.64107	0.000494156	-3561.654054	-3561.737505

Table SI_11. Interaction energies (E_{int}), BSSE, interaction enthalpies (ΔH_{int}), interaction entropies (ΔS_{int}), free Gibbs interaction energies (ΔG_{int}), calculated for different systems. All the interaction quantities are calculated at 195 K. All values are given in *kcal/mol*, except for ΔS_{int} , which are expressed in *cal/mol K*. The interaction magnitudes are computed as the difference between those of given systems.

system A	system B	System B	Eint	BSSE	ΔH_{int}	ΔS_{int}	ΔG_{int}
3a	13	[3a·13] complex II	-24.56	-9.68	-23.57	-51.19	-13.58
3a	13	[3a·13] complex I	-33.58	-12.13	-32.11	-67.86	-18.87
4a anionic	13	[3a•13] complex II + 4a (anionic)	-69.93	-21.38	-68.10	-77.40	-52,99
4a neutral	13	[3a•13] complex II + 4a (neutral)	-33.46	-12.58	-32.24	-79.28	-16.77
2'	[3a·13] complex II	[3a·13] complex II + 2 '	-56.96	-14.21	-55.12	-76.15	-40.26
4a neutral	17	[3a•13] complex I + 2 '	-81.52	-16.72	-78.70	-127.34	-53.85

XYZ coordinate (in Ångströms) for the optimized structures

Dimethyl malonate (2)

c	2 06022	1 04440	Q 17125
C	-2.90022	-1.04449	-0.1/100
0	-1.71326	-0.52561	-0.63331
С	-1.30612	0.58767	-0.01090
0	-1.94641	1.17010	0.82463
С	0.06294	1.00103	-0.49921
С	1.11790	0.07142	0.06741
0	2.31823	0.34117	-0.46355
0	0.92107	-0.78286	0.89213
С	3.38687	-0.46527	0.03596
Н	-3.15054	-1.93544	-0.76743
Н	-3.75378	-0.30746	-0.31202
Н	-2.89078	-1.29687	0.88882
Н	0.26966	2.02072	-0.16656
Н	0.11330	0.97019	-1.59091
Н	4.28266	-0.12008	-0.47752
Н	3.19861	-1.51880	-0.18098
н	3.48566	-0.33692	1.11595

Dimethyl malonate anion (2')

С	-3.57138	0.24237	0.00002
0	-2.30032	0.83503	-0.00002
С	-1.24358	-0.10231	-0.00009
0	-1.53066	-1.29176	-0.00003
С	0.00000	0.57049	-0.00002
С	1.24358	-0.10231	0.00006
0	2.30032	0.83503	0.00002
0	1.53066	-1.29176	0.00003
С	3.57138	0.24237	0.00001
Н	-4.30156	1.06583	0.00007
Н	-3.73104	-0.39610	0.88537
Н	-3.73112	-0.39604	-0.88537
Н	0.00000	1.65669	-0.00002
Н	4.30156	1.06583	-0.00003
Н	3.73107	-0.39610	-0.88534
Н	3.73110	-0.39604	0.88540

Nitrostyrene (3a)

С	0.62027	0.44777	-0.00010
С	1.59580	-0.46257	-0.00009
Ν	2.99059	-0.03032	-0.00013
0	3.23955	1.15774	0.00016
0	3.81463	-0.92300	0.00022
С	-0.81943	0.17630	-0.00006
Н	1.51190	-1.54562	-0.00004
С	-1.34898	-1.12382	-0.00007
С	-1.70105	1.26524	-0.00002
С	-3.07901	1.06540	0.00003
С	-3.59278	-0.22917	0.00002
С	-2.72346	-1.32290	-0.00003

н	0.93561	1.49450	-0.00014
Н	-0.68187	-1.98652	-0.00011
Н	-3.12264	-2.33747	-0.00003
Н	-4.67126	-0.39004	0.00005
Н	-3.75177	1.92315	0.00006
Н	-1.29572	2.27881	-0.00001

Triscarbamate 13

С	-1.87781	-0.27536	-3.64864
С	-2.66634	-0.68404	-2.40733
С	-3.50053	0.49996	-1.94473
0	-4.21350	0.19756	-0.73125
С	-2.61816	1.69541	-1.64068
С	-1.77861	2.10719	-2.86212
С	-0.90679	3.33540	-2.56871
С	0.30689	3.04112	-1.69145
0	-0.09702	2.72445	-0.33941
С	1.11548	1.84857	-2.19793
С	0.22224	0.61555	-2.46406
С	-0.94225	0.93452	-3.44435
С	-0.41615	1.31836	-4.83623
С	1.06124	-0.59512	-2.89548
С	2.23910	-0.90506	-1.96621
0	1.73676	-1.31387	-0.67518
С	3.13416	0.32146	-1.74180
С	3.84769	0.66227	-3.05769
С	2.23354	1.46887	-1.23507
С	3.21774	2.55206	-0.80165
С	4.41665	1.75097	-0.24205
С	4.16717	0.25296	-0.58222
С	5.46989	-0.53279	-0.79947
С	5.21333	-1.98964	-1.18979
С	6.33065	-0.45077	0.47078
С	7.73210	-1.05085	0.31427
С	8.59987	-0.77065	1.51497
0	9.65516	-0.19130	1.50509
0	8.03900	-1.25124	2.64294
C	8.79757	-1.02068	3.82793
Н	-2.59900	-0.00445	-4.44007
Н	-1.31130	-1.13183	-4.04234
Н	-3.32519	-1.53223	-2.63640
Н	-1.99432	-0.99046	-1.58755
Н	-4.24141	0.74406	-2.72292
C	-5.33207	-0.54770	-0.88336
н	-1.97763	1.43415	-0.78691
н	-3.24253	2.54595	-1.32362
н	-2.49692	2.40021	-3.64958
н	-1.50096	4.13246	-2.09844
н	-0.52462	3.74965	-3.514/1
н	0.94216	3.93691	-1.63/64
	-0.49539	3.//1//	0.41/88
н	T.20803	2.18000	-3.14494
н	-0.2246/	0.35629	-1.48864
П	0.32348	2.13038	-4.81669
н	-1.25246	1.64933	-5.4/129

Н	0.04711	0.45454	-5.33590
Н	1.46299	-0.44615	-3.90963
Н	0.44288	-1.50256	-2.93740
Н	2.79574	-1.75179	-2.38394
Н	4.55620	1.49223	-2.92146
Н	3.15236	0.95015	-3.85753
Н	4.42371	-0.20353	-3.41852
н	1.75050	1.08266	-0.32210
н	2,78388	3,23526	-0.05875
Н	3,51884	3,16582	-1.66583
Н	4.52231	1.89396	0.84177
н	5.36288	2,09029	-0.69350
н	3,64621	-0.21558	0.26954
н	6 03800	-0 04325	-1 61259
н	6 14090	-2 57855	-1 19656
н	1 78069	-2.07664	-2 19464
н	4.78005	-2.07004	-0 17875
LI LI	6 44216	0 60308	0.476506
	0.44J10 E 91040	0.00598	1 20092
п u	J.01049 7 67797	-0.93722	0 10609
п	/.0//8/	-2.14320	0.19698
н	8.24816	-0.63089	-0.55934
н	8.22506	-1.4/21/	4.64433
н	8.93043	0.05667	3.99626
H	9.78920	-1.48580	3.74616
C	1.36342	-2.61062	-0.5/182
0	1.40361	-3.411/9	-1.4/64/
N	0.94590	-2.84151	0./1045
С	0.47469	-4.04473	1.26391
Н	1.01455	-2.04522	1.33158
С	0.16226	-4.03772	2.62998
С	0.30467	-5.21941	0.52116
С	-0.17720	-6.36137	1.15848
С	-0.49090	-6.35870	2.51582
С	-0.31715	-5.18608	3.24902
Н	-0.55490	-5.16131	4.31321
Н	0.29926	-3.12054	3.20802
Н	0.54945	-5.22886	-0.53697
Н	-0.30795	-7.27240	0.57264
Н	-0.86741	-7.26076	2.99803
0	-5.73846	-0.96785	-1.94216
Ν	-5.90655	-0.73023	0.34529
С	-7.07802	-1.44233	0.65163
Н	-5.43060	-0.27014	1.11127
С	-7.50532	-1.42840	1.98632
С	-7.81559	-2.15560	-0.30206
С	-8.96172	-2.84007	0.09769
С	-9.38965	-2.82978	1.42336
С	-8.65132	-2.11698	2.36685
Н	-8.96687	-2.09400	3.41077
Н	-6.93065	-0.87011	2.72914
Н	-7.48978	-2.16588	-1.33802
Н	-9.52868	-3.39312	-0.65263
Н	-10.28783	-3.37200	1.71886
0	-0.43006	4.92995	0.07731
Ν	-0.97577	3.29164	1.60619
С	-1.46919	4.02628	2.69821

-0.93623	2.28523	1.70796
-1.60657	5.42009	2.68941
-2.11452	6.05922	3.81900
-2.48787	5.34272	4.95392
-2.34784	3.95570	4.95549
-1.84276	3.30145	3.83804
-1.31736	5.98598	1.80867
-2.21874	7.14501	3.80339
-2.88408	5.85862	5.82853
-2.63223	3.37376	5.83300
-1.73379	2.21427	3.84305
	-0.93623 -1.60657 -2.11452 -2.48787 -2.34784 -1.84276 -1.31736 -2.21874 -2.88408 -2.63223 -1.73379	-0.936232.28523-1.606575.42009-2.114526.05922-2.487875.34272-2.347843.95570-1.842763.30145-1.317365.98598-2.218747.14501-2.884085.85862-2.632233.37376-1.733792.21427

Supramolecular complex I [3a·13]

С	-0.92025	0.74156	-3.89590
С	-1.85259	0.45687	-2.72710
С	-2.47484	1.76529	-2.26479
0	-3.27755	1.56018	-1.08387
С	-1.40606	2.76722	-1.86682
С	-0.35945	3.00840	-2.96969
С	0.73638	3.97517	-2.48576
С	1.77371	3.34109	-1.55725
0	1.17770	2.96525	-0.30162
С	2.34410	2.04395	-2.13464
С	1.17428	1.08162	-2.44808
С	0.23857	1.68925	-3.53329
С	0.99057	1.97418	-4.84245
С	1.61740	-0.35479	-2.74763
С	2.64051	-0.90330	-1.74965
0	2.10115	-0.95922	-0.40702
С	3.86525	0.02264	-1.65483
С	4.56887	0.08703	-3.01776
С	3.34511	1.40019	-1.18529
С	4.61134	2.18329	-0.84437
С	5.57619	1.10475	-0.29798
С	4.90344	-0.27747	-0.53692
С	5.92584	-1.40190	-0.76727
С	5.26021	-2.74716	-1.06018
С	6.85506	-1.50374	0.45199
С	8.01045	-2.49570	0.27829
С	9.01163	-2.39200	1.40096
0	10.18604	-2.15554	1.28119
0	8.42320	-2.59314	2.59687
С	9.30092	-2.49173	3.71597
Н	-1.50288	1.21558	-4.70530
Н	-0.52652	-0.19577	-4.31593
Н	-2.64109	-0.25483	-3.00926
Н	-1.28247	0.02923	-1.88713
Н	-3.12994	2.15855	-3.05862
С	-4.48613	0.98956	-1.28061
Н	-0.93312	2.38005	-0.95475
Н	-1.87979	3.72478	-1.59849
Н	-0.88333	3.50225	-3.80847
Н	0.28669	4.84014	-1.97473
Н	1.28599	4.37941	-3.35005
Н	2.57250	4.06844	-1.35070

С	0.86026	3.96372	0.55747
Н	2.87234	2.32079	-3.06142
Н	0.59951	1.02055	-1.50897
Н	1.77394	2.73645	-4.74365
Н	0.28182	2.33217	-5.60482
Н	1.45884	1.05899	-5.23760
н	2.06215	-0.44253	-3.75107
н	0.74968	-1.02652	-2.73535
н	2.90112	-1.93072	-2,03578
Н	5,51960	0.63534	-2.94310
Н	3,96416	0.58418	-3.78790
н	4.80525	-0.92438	-3,38440
н	2,81273	1,21060	-0.24128
н	4 42050	2 98147	-0 11424
н	5 02423	2.50147	-1 74686
н	5 78746	1 25830	0 76862
н	6 54733	1 1/678	-0 81731
ц	1 3251 <i>1</i>	-0 5/283	0.01751
ц	6 55319	-1 12532	-1 63/89
н Ц	5 00381	-3 56/02	-1 09952
	1 75212	-3.50452	-1.00032
п Ц	4.75512	-2.74750	-2.03373 0 20102
п	4.51521	-2.90/45	-0.20405
	7.20050	1 70472	1 24125
п	0.20080	-1./84/3	1.34135
п	7.03082	-3.53113	0.20532
	8.00099	-2.3102/	-0.05012
п	0,75000	-2.08108	4.59997
п	9./5008	-1.49035	3.70348
H C	10.10/81	-3.23304	3.64038
	1.1/13/		-0.11546
U N	0.01008	-2.59/5/	-0.92156
N C	0.90004	-1.0/024	1 24319
с u	-0.10292	-2.51/11	1 71546
	1.50647	-1.0/459	1./1540
C	-0.081/4	-1.892/5	3.0000/
C	-0.0002	-3.74000	1.445/4
C	-1./135/	-4.30400	2.0//88
C	-2.31930	-3.66429	3.15886
C	-1./8580	-2.46418	3.63047
н	-2.23949	-1.95/60	4.48312
п	-0.28047	-0.93488	3.35320
н	-0.13967	-4.240/0	0.59670
н	-2.11510	-5.24882	1./0806
н	-3.19331	-4.10509	3.0388/
U	-4.93201	0.6/261	-2.35847
N	-5.10020	0.84537	-0.06163
C	-6.22060	0.052/6	0.22348
H	-4.52433	1.12118	0.72986
C	-6.49866	-0.19131	1.5//93
C C	-7.04089	-0.51309	-0./6186
C	-8.11041	-1.3199/	-0.3/63/
	-8.38483	-1.2/291	1.96/03
	-/.5/188	-0.99/31 1 17452	1.94399
п	-/.//135 E 0E020	-1.1/455 0 35633	2.00108
	-2.02025	0.2003Z	2.54205
п	-0.07921	-9.32009	-1.91001

Н	-8.74124	-1.75857	-1.15114
Н	-9.22834	-2.20225	1.25100
0	1.26205	5.10084	0.45593
Ν	0.03419	3.45201	1.51582
С	-0.52366	4.12985	2.61237
Н	-0.25458	2.48545	1.37175
С	-0.13851	5.41895	3.00036
С	-0.73953	6.00015	4.11587
С	-1.71250	5.32458	4.84916
С	-2.09185	4.04222	4.45259
С	-1.50554	3.44550	3.34246
Н	0.61647	5.95207	2.42913
Н	-0.43500	7.00496	4.41271
Н	-2.17414	5.79341	5.71837
Н	-2.85757	3.49919	5.00856
Н	-1.81372	2.44930	3.01633
С	-3.42255	-1.75645	0.50057
С	-2.26090	-1.12826	0.27616
Ν	-1.93012	0.06690	1.00318
0	-0.82078	0.54459	0.78997
0	-2.72871	0.54724	1.79004
С	-3.85925	-2.96340	-0.19319
Н	-1.46873	-1.43821	-0.39917
С	-5.17117	-3.40874	0.02103
С	-3.01410	-3.68320	-1.05642
С	-3.48511	-4.82629	-1.68902
С	-4.79717	-5.26092	-1.47605
С	-5.63942	-4.55098	-0.62278
Н	-4.10083	-1.33955	1.24761
Н	-5.82830	-2.84304	0.68680
Н	-6.66473	-4.88371	-0.45722
Н	-5.16012	-6.15837	-1.97922
Н	-2.82657	-5.38723	-2.35295
Н	-1.98061	-3.36351	-1.21301

Supramolecular complex II [3a·13]

С	-1.94689	0.02786	-3.74375
С	-2.72410	-0.21326	-2.45849
С	-3.38556	1.08543	-2.02616
0	-4.08052	0.95236	-0.75434
С	-2.33741	2.18289	-1.81865
С	-1.50693	2.41635	-3.08618
С	-0.48049	3.54547	-2.90128
С	0.74057	3.13618	-2.06601
0	0.37777	3.02914	-0.68586
С	1.34030	1.81915	-2.50176
С	0.27944	0.70133	-2.62148
С	-0.84290	1.11664	-3.61335
С	-0.31117	1.35105	-5.02522
С	0.93329	-0.64753	-2.95464
С	2.13869	-1.01761	-2.08505
0	1.69482	-1.26977	-0.73842
С	3.18643	0.08810	-2.02351
С	3.85983	0.22242	-3.39810
С	2.46673	1.36538	-1.58592

С	3.58354	2.35996	-1.28906
С	4.72557	1.47863	-0.74960
С	4.27540	0.00035	-0.91313
С	5.45569	-0.97207	-1.08580
С	4.97633	-2.42269	-1.13805
С	6.44689	-0.77430	0.05715
С	7,67165	-1.69125	-0.02146
C	8.74360	-1.27175	0.95611
0	9.88557	-0.97839	0.68481
0	8,26214	-1.22367	2.21765
Ĉ	9,21522	-0.81201	3,20152
н	-2 64752	0 36375	-4 52572
н	-1.50549	-0.90668	-4, 12803
н	-3 50340	-0 99235	-2 611/15
н	-2 05774	-0 57918	-1 65701
н	_/ 1183/	1 37877	-2 78554
C	-4.11004	0 3/1322	-0 79525
L L	-1 70000	1 80/52	-0.05730
н Ц	-2 84654	2 110/5	-1 51799
	2.04004	2 74694	-1.51/88
	-2.22500	2.74004	-2.00010
п	-0.94012	4.41920	-2.42051
	-0.11255	2.07101	-2.0/214
H C	1.50592	3.93542	-2.15290
C II	0.46824	4.20044	0.02213
н	1./8693	2.01222	-3.50223
н	-0.15848	0.59363	-1.6260/
н	0.50050	2.08/69	-5.08162
н	-1.116/2	1./116/	-5.68046
н	0.0591/	0.408/3	-5.4/491
н	1.2//10	-0.65009	-4.0106/
н	0.2005/	-1.4/33/	-2.8/361
н	2.53250	-1.95667	-2.4/094
н	4.66298	0.95/68	-3.38822
н	3.14834	0.51921	-4.18364
н	4.29292	-0./3666	-3./0/13
н	1.99149	1.11903	-0.60886
н	3.28139	3.14624	-0.5/652
н	3.89/32	2.88180	-2.2113/
Н	4.97189	1.71111	0.28688
н	5.65394	1.65033	-1.33294
Н	3./4586	-0.28544	0.00969
Н	5.97945	-0.71990	-2.03144
Н	5.81708	-3.12547	-1.25743
Н	4.28833	-2.61454	-1.98026
Н	4.45191	-2.69533	-0.20393
Н	6.79863	0.26871	0.07253
Н	5.94094	-0.93297	1.02429
Н	7.38735	-2.73551	0.22645
Н	8.12949	-1.67794	-1.01065
Н	8.66825	-0.77665	4.15615
Н	9.61467	0.17820	2.95729
Н	10.03026	-1.53043	3.27381
C	1.69013	-2.53822	-0.31159
0	1.58853	-3.50853	-1.02379
N	1.76346	-2.55610	1.07516
С	1.33293	-3.63514	1.88395

Н	1.72861	-1.63850	1.49164
С	0.80123	-3.29977	3.13850
С	1.42018	-4.96295	1.47085
С	0.88282	-5.96881	2.32013
С	0.32189	-5.64211	3.55245
С	0.29484	-4.31706	3.95850
Н	-0.17042	-4.03196	4.91247
Н	0.71589	-2.25823	3.44115
Н	1.82178	-5.21419	0.50174
Н	0.88044	-7.00216	1,96340
Н	-0.11660	-6.43643	4,18936
0	-5.74360	-0.22236	-1.76423
N	-5.87223	0.47637	0.44482
C	-7,00718	-0.22047	0,90709
н	-5.32861	0.99616	1,12482
C	-7 11605	-0 37957	2 34264
c	-7 99657	-0 74069	0 02426
C	-9 09090	-1 42654	0 57893
c	-9 20080	-1 58973	2 00555
c	-8 21377	-1 0/92/	2.00555
н	-8 29224	-1 18591	3 99396
н	-6 32781	-0 00623	3 03969
н	-7 89883	-0 61185	-1 07357
н	-9 84043	-1 84417	-0 10897
н	-10 05788	-2 12353	2 42319
0	0 94267	5 23951	-0 44502
N	-0 02995	3 98366	1 29332
C	-0 08436	4 93763	2 32953
н	-0.35194	3,05434	1,51785
C	0.33565	6 27079	2 20014
c	0.18900	7,13376	3,29562
c	-0.35124	6.68197	4,49716
c	-0.75027	5.35312	4,61800
C	-0.61150	4.47687	3,53729
н	0.76795	6.63841	1,27360
н	0.55185	8.18028	3,18972
н	-0.42181	7.37435	5,34395
н	-1.16760	4.98589	5,55939
Н	-0.92417	3.42722	3.63745
C	-1.57285	-1.76650	0.72860
C	-1.88069	-0.48283	0.99859
N	-0.92756	0.35279	1.68092
0	0.23379	-0.01142	1.83872
0	-1.35695	1.45728	2.07013
Č	-2.48861	-2.68883	0.05977
Н	-2.80171	0.03091	0.78887
C	-3.87777	-2.47172	0.09755
C	-1.97806	-3.78308	-0.64151
C	-2.84711	-4.63839	-1.34319
С	-4.22353	-4.40168	-1.31893
С	-4.74078	-3.31307	-0.59967
н	-0.56365	-2.11491	0.96432
н	-4.29333	-1.64883	0.69750
н	-5.81739	-3.10810	-0.58656
Н	-4.89470	-5.05957	-1.87681
н	-2.43901	-5.46671	-1.91476

Н	-0.89661	-3.96109	-0.68596

TS for the reaction between 2' and 3a without catalyst 13

С	-0.57957	1.62667	-0.41614
0	-1.78811	2.07262	0.13342
С	-2.81660	2.30703	-0.79583
Н	-2.83528	1.53881	-1.58128
Н	-2.69619	3.29312	-1.28345
Н	-3.76386	2.29548	-0.23540
0	-0.55135	1.26085	-1.58563
С	0.44234	1.65828	0.56431
С	1.78805	1.33228	0.25950
0	2.26987	0.89620	-0.78262
0	2.61843	1.53736	1.35498
С	3.95898	1.15501	1.13024
Н	4.49138	1.34699	2.07303
Н	4.41658	1.74134	0.31658
Н	4.03236	0.08939	0.86401
С	0.10680	-1.37807	0.36178
Н	0.20680	2.05461	1.54788
С	-1.34623	-1.18423	0.36854
С	-1.99938	-1.01611	1.59646
С	-2.11501	-1.18301	-0.80469
С	-3.49697	-1.03722	-0.74111
С	-4.13656	-0.87460	0.48830
С	-3.37980	-0.85809	1.65954
Н	-3.86625	-0.71569	2.62607
Н	-1.40033	-0.99027	2.50876
Н	-4.08112	-1.03622	-1.66319
Н	-5.22052	-0.75172	0.53164
Н	-1.62206	-1.28930	-1.77039
С	0.88498	-1.42931	-0.72243
Н	0.59425	-1.53851	1.32528
Ν	2.28751	-1.76682	-0.57257
0	2.90962	-1.97155	-1.59491
0	2.75742	-1.88181	0.54849
Н	0.62289	-1.18297	-1.74994

TS for the reaction between 2' and 3a inside steroidal triscarbamate 13

С	0.96673	-3.88964	-2.51018
С	1.96835	-2.82774	-2.04175
С	2.72074	-3.36130	-0.82676
0	3.67384	-2.36597	-0.33931
С	1.75503	-3.68247	0.30477
С	0.69375	-4.72516	-0.12826
С	-0.27085	-5.06225	1.03153
С	-1.36795	-4.02360	1.27966
0	-0.76262	-2.80576	1.77552
С	-2.13022	-3.67935	-0.01527
С	-1.13448	-3.22206	-1.12629
С	-0.07394	-4.34230	-1.44542
С	-0.74900	-5.61146	-2.02325
С	-1.86641	-2.68797	-2.38172
С	-3.00496	-1.69154	-2.10231

0	-2.42042	-0.46613	-1.56472
С	-4.01721	-2.25570	-1.07682
С	-4.73174	-3.48378	-1.69208
С	-3.21483	-2.61760	0.20827
С	-4.30153	-2.88402	1.25900
С	-5.39585	-1.83785	0.92579
С	-5.07360	-1.24971	-0.49354
С	-6.36305	-0.93863	-1.30146
С	-6.11616	-0.47794	-2.74800
С	-7.19797	0.11787	-0.53828
С	-8.63435	0.29908	-1.06981
С	-9.43978	1.27225	-0.24025
0	-10.43577	1.01184	0.40476
0	-8.89913	2.51990	-0.29285
С	-9.56997	3.51658	0.49553
Н	1.53710	-4.78582	-2.81987
Н	0.44994	-3.54486	-3.41988
Н	2.67796	-2.60077	-2.85521
н	1.45989	-1.89065	-1.77019
н	3.30636	-4.24657	-1.11789
С	4.97165	-2.52448	-0.73868
H	1.28411	-2.75271	0.64046
н	2.31657	-4.09034	1.16381
Н	1.24916	-5.65538	-0.35819
Н	0.30052	-5.21145	1,96337
Н	-0.77638	-6.02170	0.82507
н	-2.05263	-4.39088	2.05651
C	-1.19861	-2.27511	2,95846
H	-2.62429	-4.61699	-0.32303
н	-0.59446	-2.36274	-0.70644
Н	-1.53199	-6.02829	-1.37310
Н	0.00158	-6.40523	-2.17854
Н	-1.21102	-5.41248	-3.00430
Н	-2.29341	-3.51579	-2.97120
Н	-1.15079	-2.18353	-3.04822
Н	-3,48439	-1.42179	-3.04942
Н	-5.52905	-3.85848	-1.03347
н	-4.05099	-4.32535	-1.87797
н	-5,20076	-3,23304	-2.65590
н	-2.69910	-1.69039	0.50080
н	-3,93206	-2.78908	2,28829
н	-4.69615	-3.91114	1,15860
н	-5.41180	-1.03809	1.68244
н	-6.39838	-2,29834	0.94221
н	-4.54075	-0.29619	-0.34971
н	-6.96880	-1.86546	-1.33728
н	-7 06348	-0 21815	-3 24984
н	-5 64758	-1 26088	-3 36071
н	-5 / 5687	0 /0163	-2 79808
н	-7 26672	-0 16142	0 52468
н	-6 67586	1 08068	-0 57172
н	-8 615//	A 67380	-2 10700
н	_Q 1722Q	-0 65707	-1 06157
н	- J. 17009 _ 0 00071	_0.05/92 / ///01	0 2/767
н	-9 57760	3 23611	1 56150
н	-10 61200	3 61602	0 1676E
	- 10.01730	5.04035	0.10203

С	-2.51137	0.66178	-2.32970
0	-3.04458	0.70225	-3.43244
Ν	-1.94145	1.70930	-1.66021
С	-1.91270	3.05689	-2.07049
Н	-1.45926	1.49652	-0.77803
С	-1.31913	3.98498	-1.18914
С	-2.45030	3.51642	-3.28945
С	-2.39072	4.87836	-3.60252
С	-1.80539	5.80063	-2.72902
С	-1.26933	5.33931	-1.52031
H	-0.80383	6.03991	-0.82058
Н	-0.89061	3,62680	-0.25293
Н	-2.90218	2.80229	-3,97286
н	-2.81262	5.21854	-4.55350
Н	-1.76511	6.86295	-2,98602
0	5.36384	-3.44174	-1.44688
N	5,73191	-1.50555	-0.22402
C	7.11856	-1.31804	-0.39381
н	5.25523	-0.81500	0.36606
c	7.68631	-0.16468	0.18820
C	7.94732	-2.21016	-1.10336
C	9.31563	-1.94214	-1.21441
C	9,88268	-0.80245	-0.63452
C	9,05380	0.08268	0.06643
н	9 47341	0.00200	0.00045
н	7 03960	0.52960	0.72588
н	7 50842	-3 09507	-1 55731
н	9 94676	-2 64495	-1 76761
н	10 95441	-0 60480	-0 72815
0	-1 87504	-2 88623	3 77764
N	-0.74196	-0.99139	3,03176
C	-0.85875	-0.09443	4,11249
н	-0 28581	-0 65701	2 17038
C	-1 54911	-0 39915	5 30242
c c	-1 59769	0.55515	6 33107
c	-0 97032	1 79238	6 2011/
c	-0 28709	2 08999	5 01523
c	-0.23364	1 16080	3 97619
с ц	-2 03300	-1 36806	5 10103
	-2.05500	-1.J0000 0 30078	7 25126
н	-1 01150	2 52135	7 01617
н Ц	-1.01155 0 22457	2.02100	/ 97729
н	0.22457	1 /196/	3 05800
C C	2 51664	2 31622	-0 55/21
c c	2.016/1	0 00533	-0.55421
N	0 96751	0.71754	0.44217
	0.80751	1 63969	0.10000
0	0.03107	1.03000	0.510//
C C	2 56211	2 67022	-1 5200/
c c	V 232Vð 2.202TT	2.0/323	-1.02601
c	2 50100	3 00/6/	-1.90001
C C	J.J7499 1 55510	J.JJ404 1 30056	-2.03/39
c c	4.33340 5 51/12	4.30000 3 16100	-2.7/403 _2 /1000
C C	5.51410 5 50076	3.40190 3 15/01	-2.41908
с н	J. 500/0	2.13491 0 7/05/	-2.91021
	4.JJ400 6 JE101	U./4704 1 /0150	- 2 3/60F
11	0.23104	1.43132	-2.24005

Н	6.27003	3.76304	-4.15018
Н	4.55639	5.40502	-3.35901
Н	2.85913	4.71786	-1.67988
С	1.98881	6.01947	1.98101
0	2.83345	5.01744	1.40912
С	2.73421	3.76873	2.00607
0	1.98060	3.58247	2.94791
С	3.61326	2.83668	1.30408
С	3.88994	1.48613	1.75364
0	2.94213	0.95185	2.55132
0	4.85385	0.81589	1.35360
С	3.12506	-0.42468	2.92473
Н	2.17359	6.93457	1.39864
Н	2.23024	6.19216	3.04414
Н	0.92442	5.73605	1.91746
Н	3.04110	-1.08280	2.04871
Н	2.31704	-0.64016	3.63435
Н	4.10778	-0.57401	3.40026
Н	4.45244	3.30091	0.79083
Н	1.77039	3.08598	-0.36954
Н	2.63435	0.11295	-0.66856

Preorganized reagents within the catalyst: [2'·3a·13]

С	0.65882	-0.61566	-4.54057
С	1.46585	0.19920	-3.53493
С	2.58000	-0.64573	-2.93167
0	3.15946	0.13032	-1.86885
С	2.03681	-1.94583	-2.35802
С	1.21152	-2.74970	-3.37366
С	0.70875	-4.06984	-2.77781
С	-0.37517	-3.88213	-1.71691
0	0.16247	-3.26730	-0.53747
С	-1.50322	-2.96493	-2.19783
С	-0.98672	-1.65331	-2.83994
С	0.04977	-1.91358	-3.96907
С	-0.58279	-2.65456	-5.15672
С	-2.17095	-0.78811	-3.28895
С	-3.22072	-0.51388	-2.20712
0	-2.69551	0.39004	-1.21545
С	-3.68675	-1.79029	-1.49273
С	-4.59522	-2.56022	-2.46185
С	-2.44325	-2.59312	-1.05742
С	-3.00934	-3.71000	-0.18241
С	-4.23260	-3.05418	0.50337
С	-4.38075	-1.63423	-0.10838
С	-5.81633	-1.09070	-0.04830
С	-5.90542	0.35427	-0.54369
С	-6.34995	-1.19647	1.38952
С	-7.77862	-0.66910	1.56131
С	-8.34340	-0.98924	2.92049
0	-9.36351	-1.59148	3.14391
0	-7.55241	-0.50973	3.90136
С	-8.00986	-0.78128	5.22171
Н	1.32052	-0.90987	-5.37575
Н	-0.12851	0.01206	-4.98520

Н	1.89118	1.09353	-4.01543
Н	0.80296	0.55628	-2.73075
Н	3.35931	-0.84963	-3.68370
С	4.31850	-0.36994	-1.33104
Н	1.42975	-1.67699	-1.48260
Н	2.87698	-2.56082	-2.00047
н	1.88452	-3.00316	-4.21454
н	1.54034	-4.63846	-2.33738
н	0.28426	-4.70201	-3.57434
Н	-0.78248	-4.86400	-1.42979
C	1.04876	-4.00143	0.17639
н	-2.06160	-3.54757	-2,95271
н	-0.47901	-1.10704	-2.02721
н	-1 15231	-3 54659	-4 86197
н	0 20520	-2 97907	-5 85502
н	-1 26237	_1 00200	-5 71565
н	-2 68775	-1 27256	-/ 13300
и Ц	-2.00775	-1.27250 0 103/7	-3 65206
	-1.85101	0.19547	2 60055
п u	-4.03007	-0.00/10	-2.09955
п	-4.98559	-3.48204	-2.00/30
	-4.07090	-2.84318	-3.38500
п	-5.458/5	-1.9410/	-2./5238
н	-1.8/158	-1.92134	-0.39591
н	-2.26611	-4.07414	0.538/8
н	-3.31655	-4.5/214	-0./9/1/
н	-4.10//2	-2.999/2	1.59308
н	-5.14851	-3.64055	0.32004
Н	-3.74606	-0.94669	0.47559
Н	-6.45957	-1.72793	-0.68427
Н	-6.93649	0.73529	-0.51102
Н	-5.56594	0.46631	-1.58051
Н	-5.28032	1.01167	0.08418
Н	-6.33058	-2.24914	1.71038
Н	-5.67949	-0.64622	2.07073
Н	-7.79994	0.42533	1.45097
Н	-8.45975	-1.10594	0.81839
Н	-7.26659	-0.34213	5.89463
Н	-8.08966	-1.86448	5.38691
Н	-8.99808	-0.33050	5.38735
С	-2.99476	1.69814	-1.39889
0	-3.43518	2.16130	-2.42848
Ν	-2.74915	2.37160	-0.23615
С	-2.76975	3.75840	-0.03535
Н	-2.34932	1.80647	0.51490
С	-2.47702	4.20986	1.26169
С	-3.03449	4.69016	-1.04741
С	-2.99077	6.05119	-0.74796
С	-2.69358	6.50458	0.53605
С	-2.43755	5.57094	1.54141
Н	-2.19415	5.89974	2.55253
Н	-2.26039	3.47762	2.04377
Н	-3.25923	4.34275	-2.05177
Н	-3.18791	6.76980	-1.54537
Н	-2.65903	7.57273	0.75218
0	4.99228	-1.20183	-1.90409
Ν	4.54598	0.21069	-0.13100

С	5.63614	-0.02287	0.71781
Н	3.85539	0.89382	0.21830
С	5.61453	0.63935	1.95763
С	6.71038	-0.86892	0.40673
С	7.74936	-1.02337	1.32462
С	7.74392	-0.35307	2.54581
С	6,66416	0.47795	2.85380
H	6.62933	1.00529	3,80884
Н	4.76206	1.27669	2.19437
Н	6.71655	-1.40289	-0.53903
н	8.57781	-1.68808	1.07215
н	8 56424	-0 48145	3 25345
0	1,17064	-5.20337	0.05552
N	1 7504	-3 18159	1 01003
C	2 89/195	-3 55277	1 7/6/8
L L	1 55/12	_2 18120	0 02251
C II	2 20102	-2.10123	2 06949
c	1 26600	-4.00141 E 1EE7/	2.00040
C C	4.30000	-3.133/4	2./05/4
C C	2.22524	-4.15469	2.1000
C	4.90/10	-2.81413	2.86/49
C	3.74842	-2.51946	2.15661
н	2.54257	-5.68238	1.74784
н	4.60024	-6.19418	3.02908
н	6.13/89	-4.36438	3./3/41
н	5.56930	-1.99609	3.15645
Н	3.49120	-1.48444	1.91513
C	0.44790	2.65919	-0.02634
С	0.64504	1.33835	-0.14471
Ν	-0.14892	0.43217	0.63516
0	-0.94264	0.86344	1.46712
0	-0.01781	-0.75965	0.40826
С	1.08049	3.67844	-0.86197
Н	1.38272	0.82458	-0.75277
С	2.11202	3.37763	-1.76728
С	0.60095	4.99252	-0.77956
С	1.11527	5.98658	-1.60967
С	2.11712	5.67514	-2.52678
С	2.61680	4.37104	-2.59747
Н	-0.27208	3.00789	0.71273
Н	2.53559	2.37284	-1.79571
Н	3.42006	4.13100	-3.29569
Н	2.52022	6.44928	-3.18227
Н	0.72670	7.00363	-1.53891
Н	-0.19126	5.22776	-0.06512
С	0.87530	-0.63457	3.74944
0	1.21110	0.74107	3.74432
С	1.93780	1.16084	2.66099
0	2.28985	0.33158	1.80425
С	2.16271	2.55445	2.66652
С	2.93876	3.18767	1.66646
0	2.99082	4.55018	1.84312
0	3.54298	2.68638	0.70967
С	3.82689	5.23237	0.92851
Н	0.32350	-0.80263	4.68294
Н	1.77248	-1.27163	3.72864
Н	0.23404	-0.89491	2.89266

Н	1.75472	3.15286	3.47547
Н	3.80927	6.28682	1.23351
Н	3.46116	5.13520	-0.10371
Н	4.85615	4.84391	0.96420

Reaction product 4a (neutral)

С	-0.15418	-1.57482	0.81468
0	-0.85090	-1.35775	1.93867
С	-1.86070	-2.33217	2.21054
Н	-2.59207	-2.35019	1.39069
Н	-2.33362	-2.01783	3.14599
Н	-1.41253	-3.32855	2.31576
0	-0.27784	-2.54373	0.11589
С	0.78113	-0.40828	0.54332
С	1.63327	-0.74874	-0.66765
0	1.47490	-0.34007	-1.79187
0	2.60316	-1.58800	-0.31463
С	3.46213	-1.99351	-1.38119
Н	2.88306	-2.50900	-2.15839
Н	4.19592	-2.66849	-0.93127
Н	3.95514	-1.11589	-1.81966
С	-0.03956	0.90660	0.46543
Н	1.47049	-0.34586	1.39833
С	-1.39334	0.77211	-0.21379
С	-1.52516	0.42229	-1.56291
С	-2.54838	1.00002	0.54228
С	-3.81244	0.89169	-0.03359
С	-3.93631	0.54779	-1.37860
С	-2.79108	0.31296	-2.13703
Н	-2.87954	0.03630	-3.18842
Н	-0.63464	0.21834	-2.15757
Н	-4.70159	1.07646	0.57094
Н	-4.92372	0.45871	-1.83346
Н	-2.45286	1.26013	1.59892
С	0.74879	2.06609	-0.15591
Н	-0.24138	1.18663	1.50939
Ν	2.00482	2.25955	0.63452
0	2.95903	1.56833	0.33427
0	1.96840	3.04373	1.55509
Н	0.18422	2.99919	-0.07053
Н	1.05049	1.85095	-1.18503

Reaction product 4a (anionic form)

c	-0 00887	-1 72040	0 65570
C	-0.09887	-1.72940	0.05575
0	-0.62467	-1.65434	1.89978
С	-1.57256	-2.66772	2.19672
Н	-2.42937	-2.60415	1.50962
Н	-1.89308	-2.48765	3.22912
Н	-1.12223	-3.66567	2.10048
0	-0.36318	-2.63436	-0.09925
С	0.78631	-0.54927	0.36760
С	1.38942	-0.74374	-1.00395
0	1.05882	-0.22826	-2.04022
0	2.39931	-1.63973	-0.94035

С	2.98682	-1.94579	-2.19422
Н	2.23889	-2.37265	-2.87783
Н	3.77918	-2.67432	-1.98867
Н	3.40597	-1.04152	-2.65761
С	0.06469	0.82495	0.59828
Н	1.60940	-0.51720	1.10209
С	-1.29309	0.90326	-0.07157
С	-1.45618	1.25552	-1.41803
С	-2.44353	0.61413	0.67457
С	-3.71420	0.66246	0.10126
С	-3.86037	1.01037	-1.23974
С	-2.72452	1.30896	-1.99265
Н	-2.82589	1.58472	-3.04445
Н	-0.57132	1.46614	-2.01635
Н	-4.59286	0.43303	0.70817
Н	-4.85199	1.05305	-1.69489
Н	-2.32918	0.35223	1.72884
С	0.97027	1.96388	0.27593
Н	-0.09889	0.81983	1.68822
Ν	2.04020	2.15784	1.06139
0	2.24770	1.34784	2.02622
0	2.82715	3.11427	0.85713
Н	0.83374	2.66034	-0.54284

Supramolecular complex II [3a·13] + product 4a (neutral)

С	1.03989	-3.49042	-3.20309
С	1.97747	-2.53830	-2.47167
С	2.59271	-3.27976	-1.29529
0	3.48116	-2.41734	-0.55460
С	1.52669	-3.78133	-0.33796
С	0.51650	-4.69411	-1.05955
С	-0.57399	-5.21457	-0.11058
С	-1.65399	-4.19444	0.24642
0	-1.09836	-3.22345	1.15604
С	-2.20549	-3.45715	-0.97989
С	-1.06333	-2.90318	-1.85913
С	-0.10126	-4.03302	-2.32240
С	-0.82742	-5.09361	-3.16429
С	-1.57871	-2.03304	-3.01501
С	-2.58048	-0.96287	-2.58344
0	-1.88959	-0.02125	-1.72896
С	-3.74795	-1.56547	-1.79249
С	-4.56479	-2.47093	-2.72684
С	-3.15644	-2.33198	-0.58779
С	-4.38551	-2.67799	0.25180
С	-5.29044	-1.43545	0.08971
С	-4.69097	-0.57218	-1.05959
С	-5.77352	0.15857	-1.87286
С	-5.20802	0.93612	-3.06178
С	-6.56175	1.09030	-0.93916
С	-7.78982	1.73251	-1.59262
С	-8.63063	2.48916	-0.59636
0	-9.79651	2.29742	-0.36391
0	-7.90647	3.43818	0.03075
C	-8.63212	4.19710	0.99486

Н	1.63267	-4.35335	-3.55432
Н	0.63075	-3.01952	-4.10846
Н	2.77114	-2.18494	-3.14303
Н	1.42947	-1.65683	-2.09484
Н	3.19001	-4.12033	-1.68297
С	4.68220	-2.20339	-1.14270
Н	1.02668	-2.92146	0.12850
Н	2.00339	-4.34181	0.48121
Н	1.08525	-5,57388	-1.41318
Н	-0.12268	-5.58599	0.82238
Н	-1.07868	-6.07445	-0.57727
Н	-2,47474	-4.69814	0.77270
C	-1.63700	-3.16471	2,40483
Н	-2.77597	-4.21188	-1.54661
Н	-0.49079	-2.22750	-1.19976
Н	-1.68020	-5.55830	-2.65255
Н	-0.12443	-5.89918	-3.42753
Н	-1.19604	-4.66533	-4.10887
н	-2.05245	-2.65265	-3.79195
н	-0.74066	-1.51241	-3,50217
н	-2.92125	-0.41316	-3,46629
н	-5.47802	-2.82985	-2,23094
н	-4.00266	-3,35296	-3,06282
н	-4 87770	-1 91715	-3 62496
н	-2 56420	-1 58940	-0 02474
н	-4 14088	-2 89642	1 29833
н	-4.87472	-3.57796	-0.15547
н	-5.34146	-0.85941	1.02352
н	-6.32470	-1.72762	-0.15347
н	-4.04193	0,20103	-0.61210
н	-6.48377	-0.59860	-2.25504
Н	-5.98526	1.53647	-3.55538
Н	-4.79432	0.27104	-3.83076
Н	-4.40297	1.62030	-2.74763
Н	-6.89913	0.52210	-0.05865
Н	-5.89441	1.88474	-0.56521
Н	-7.48494	2.44756	-2.37143
н	-8.43965	0.97512	-2.05085
Н	-7.91772	4.91225	1.41443
Н	-9.03158	3.54072	1.77984
н	-9.47148	4.72171	0.51863
C	-1.60873	1.18911	-2.25828
0	-1.80990	1.51416	-3.40480
N	-1.05492	1.97486	-1.27766
C	-0.80470	3,35590	-1.36767
H	-1.04028	1,56128	-0.34770
C	-0.69632	4.06394	-0.16232
C	-0.62326	4,02804	-2.58288
C	-0.33531	5.39242	-2,56903
C	-0.21387	6.09774	-1.37222
C	-0.39599	5.42201	-0.16523
H	-0.30569	5.95085	0.78474
Н	-0.83529	3.53206	0.78107
Н	-0.70958	3.48371	-3.51920
Н	-0.19770	5.90830	-3.52045
Н	0.01616	7.16326	-1.37931

0	5.02524	-2.70153	-2.19062
Ν	5.41872	-1.34617	-0.37070
С	6.61886	-0.71161	-0.73924
Н	4.98413	-1.04272	0.49855
С	7.14611	0.22043	0.16706
С	7.27433	-0.93114	-1.95852
С	8.43077	-0.20804	-2.25073
С	8.94924	0.72803	-1.35818
С	8.29786	0.93408	-0.14219
Н	8.68659	1.66007	0.57335
Н	6.62358	0.38460	1.11023
Н	6.87818	-1.65573	-2.66312
Н	8.93163	-0.38609	-3.20351
Н	9.85177	1.28743	-1.60418
0	-2.52777	-3.88122	2.80674
Ν	-1.01142	-2.17497	3.09494
С	-1.27203	-1.73941	4.40277
Н	-0.25776	-1.70363	2.60170
С	-2.32222	-2.21649	5.19647
С	-2.49701	-1.68932	6.47618
С	-1.65091	-0.70248	6.97835
С	-0.60391	-0.23851	6.18145
С	-0.41250	-0.75196	4.90452
Н	-2.98262	-2.98911	4.81266
Н	-3.31722	-2.06425	7.09054
Н	-1.80194	-0.30299	7.98151
Н	0.08166	0.52544	6.55270
Н	0.41140	-0.39100	4.28653
С	2.22410	2.04406	0.31937
С	1.88697	0.58142	-0.00008
Ν	0.97752	-0.00453	1.03711
0	-0.08884	0.55502	1.24182
0	1.35077	-0.98917	1.62994
С	3.16464	2.58286	-0.74515
С	4.36162	1.92776	-1.05241
С	2.84490	3.76725	-1.41588
С	3.71402	4.28376	-2.37645
С	4.90979	3.62895	-2.67082
С	5.23532	2.44794	-2.00497
Н	4.62135	1.00171	-0.54424
Н	6.16859	1.92051	-2.21675
Н	5.58791	4.03843	-3.42054
Н	3.44893	5.20413	-2.89889
Н	1.90965	4.28573	-1.19193
C	-0.40955	2.79227	3.57433
0	0.69401	3.00995	2.68119
С	1.69689	2.14596	2.82540
0	1.76185	1.31590	3.69713
С	2.76135	2.32363	1.74466
С	3.94556	1.49144	2.19945
0	4.58031	2.10229	3.19285
0	4.25370	0.40316	1.77056
С	5.57352	1.32282	3.86643
Н	-1.14251	3.56875	3.33331
Н	-0.07693	2.88029	4.61647
Н	-0.82431	1.78871	3.40759

Н	5.14210	0.36498	4.18406	
Н	5.87978	1.91959	4.73005	
Н	6.43159	1.13969	3.20625	
н	3.05875	3, 38223	1.77112	
ц	1 20081	2 61732	0 25164	
	1 20607	2.01/JZ 0 EE126	0.23104	
	1.50007	0.55150	-0.95270	
Н	2./4689	-0.09102	-0.04489	
Supran	alagular gampla	v II [20.12] u proc	luct 12 (anionia form	^
Supran		x ii [3a i 3] + pioc	iuci 4a (antonic torn	v
С	1.05688	-2.48029	-3.72103	
С	1.92630	-1.56560	-2.85931	
c	2,79013	-2,44402	-1.97466	
0	2 63844	_1 58005	-1 19/07	
C C	1 01026	2 20571	1 06660	
	1.91930	-3.295/1	-1.00000	
C	0.96999	-4.19047	-1.88361	
C	0.09055	-5.06648	-0.97578	
C	-1.03510	-4.30213	-0.28390	
0	-0.44941	-3.37749	0.63474	
С	-1.87616	-3.49201	-1.28186	
С	-0.97922	-2.59284	-2.16662	
С	0.11714	-3.40202	-2.91488	
С	-0.49914	-4.38649	-3.92363	
С	-1.83822	-1.74977	-3.11833	
С	-2.93313	-0.93125	-2.43143	
0	-2.33657	0.11407	-1.65453	
C	-3,80126	-1.78458	-1.50270	
C	-4 69455	-2 66774	-2 38725	
c	-2 88681	-2.60774	-0 56278	
c c	-2.00001	2.00255	-0.30278	
C C	-3.0///2	-3.24/95 3 10FFC	0.41121	
C C	-4.99450	-2.10000	0.30100	
C	-4.66010	-1.04582	-0.43762	
C	-5.883//	-0.23556	-0.89363	
C	-5.48019	1.00399	-1.69664	
C	-6.72176	0.17581	0.32791	
С	-7.96858	0.99578	-0.02647	
С	-8.88273	1.16828	1.15696	
0	-10.03284	0.81302	1.22873	
0	-8.24846	1.78334	2.17618	
С	-9.03726	1.96224	3.34651	
Н	1.72029	-3.12757	-4.32404	
Н	0.47663	-1.88881	-4.44383	
Н	2.55853	-0.92203	-3.48959	
Н	1.30628	-0.91256	-2.22185	
Н	3.44822	-3.07580	-2.59206	
С	4.77852	-2.13393	-0.72423	
Н	1.36811	-2.61128	-0.40694	
Н	2.55950	-3.93757	-0.44092	
н	1 61077	-4 87469	-2 47241	
Ц	0 71170	-5 55001	-0 2061/	
	0.711/5	- J. JJ004 E 0717E	1 57000	
н Ц	-0.5/005	-7.0/1/2	0 JOCOE	
п С	-1.0070/ 0.00775	-4.99000 0,01170	U.20000 1 01070	
	-0.90//5	-3.311/8	1.912/0	
н	-2.41864	-4.22439	-1.90//2	
Н	-0.47798	-1.89911	-1.46814	
Н	-1.27576	-5.02674	-3.48193	

Н	0.28444	-5.04529	-4.33094
Н	-0.95056	-3.85198	-4.77332
Н	-2.32252	-2.39634	-3.86851
Н	-1.22241	-1.03068	-3.67855
Н	-3.53802	-0.46628	-3.21922
Н	-5.39574	-3.26712	-1.78898
Н	-4.10992	-3.36305	-3.00640
Н	-5.29084	-2.04182	-3.06977
Н	-2.29347	-1.85223	-0.00914
Н	-3.42233	-3.51680	1.37149
Н	-4.27843	-4.17953	-0.02316
Н	-5.05206	-1.80201	1.58956
Н	-5.98248	-2.61786	0.33096
н	-3.97628	-0.34246	0.06631
Н	-6.52136	-0.88355	-1.52456
Н	-6.35766	1.56939	-2.04390
н	-4.88851	0.76433	-2.58770
н	-4.86836	1.67541	-1.07025
н	-7.04534	-0.72419	0.87234
н	-6.09370	0.75718	1.02351
н	-7.68208	2,00034	-0.37104
Н	-8,55292	0.50757	-0.81828
Н	-8.38392	2,44814	4.07839
Н	-9.39059	0.99392	3,72643
Н	-9.91241	2,59069	3,12992
C	-2.14323	1.29292	-2.28997
0	-2.45696	1.50128	-3,44643
N	-1.58119	2.17630	-1.42002
C	-1.28948	3.51845	-1.68158
н	-1,27638	1.80073	-0.49590
C	-0.94636	4.30920	-0.57139
Ċ	-1.29632	4.10211	-2.95736
c	-0.96228	5.44948	-3,09628
C	-0.61563	6.23294	-1,99609
C	-0 60764	5 64728	-0 72821
н	-0.32893	6,22939	0.15231
н	-0 91965	3 83762	0.13231
н	-1 55953	3 49774	-3 82065
н	-0 96996	5 88979	-1 09511
н	-0.35/29	7 28/11	-2 12/63
0	5 17028	-3 24216	-1 01907
N	5 11/020	-1 239/0	0 09631
C	6 6/681	-1 /2291	0.00001
с ц	1 90711	-0 38117	0.75025
C	7 10632	-0.36436	1 53722
c	7.10052	-2 58073	0 62013
c	8 61180	-2.56075	1 20080
c	0 1077	-2.03301	2 00/06
c	9.10272	-1.00772 0.45901	2.09490
с н	8 823NJ 0.27272	-0.43091 0 27750	2.20/JL 2 82/02
	6.03342 6 10560	0.57750 0 EDTOD	2.02473 1 21357
п Ц	0.4000 7 07070	0.32/92 _3 AREAA	1.0125/
n L	1.U/J/J 0 J/JE1	-2.540244	1 20001
п	J. 24201	-2.20290 1 60400	1.20244
	10.00488 _1 /555	-1,00499 _1,00499	2.0209/
U N	-1.40000	-4.232/5	2.489/0
IN	-0.63023	-2.06943	2.3//85

С	-0.76028	-1.59144	3.68144
Н	-0.24441	-1.41915	1.65377
С	-1.34327	-2.31061	4.73419
С	-1.40600	-1.73183	6.00221
С	-0.89893	-0.45669	6.24402
С	-0.31960	0.25275	5.18894
С	-0.25486	-0.30177	3.91800
н	-1.73322	-3.30853	4.54888
н	-1.86262	-2.29811	6.81683
н	-0.95120	-0.02198	7.24340
н	0.10483	1.24624	5.34579
н	0.19753	0.25610	3.09418
С	2.03230	2.45416	-0.29356
С	1.77096	0.98959	-0.17308
N	0.63141	0.56598	0.31440
0	-0.22805	1.38607	0.83232
0	0.33681	-0.68360	0.28279
C	2.84324	2.80541	-1.52932
c	3,90021	2,01127	-1,99028
c	2 50596	3 95881	-2 24776
c	3 21780	4 32164	-3 39022
c	4 27767	3 53256	-3 83425
c c	4.27707	2 37/83	-3 13264
с ц	4.01100	1 00/00	-1 16/15
н Ц	4.10001	1 7/125	-2 /7705
н Ц	1 83110	2 81000	-1 73061
Ц	2 97978	5 21806	-3 9/205
ц	1 65700	1 56547	-1 91956
C II	-0 34566	2 50807	2 07076
0	0.70518	3 80203	2 15527
C C	1 70374	2 91250	2.13327
0	1 85850	2.91290	3 02/15
C C	2 63562	3 10/55	0 98312
c	3 98399	2 55455	1 38906
0	1 561/15	3 362/18	2 28739
0	4.50145	1 53531	0 99613
C C	5 78535	2 88850	2 8/062
L L	_1 11/192	1 26075	2.04002
n u	-1.11402	4.20975	2.905/7
п u	0.02000	2.20000	4.11207
n U	-0./2191	2.50095	2.0/241
	5.05540 C 09722	1.91105	2.21917
n u	0.00/22 6 FE101	00000CC	2.2012Z
n u	0.JJAJC	2./9389 1 19000	2.00/92
n u	2./4430 1 AFE07	4.10070 2 01221	7/120.0 5000C A
л 11	T.42201	2.94234	-0.5988/
н	2.44112	0.21314	-0.52425

Supramolecular complex [MeOH·13]

С	1.71981	0.92921	-3.69861
С	2.48284	1.04715	-2.38422
С	3.16542	-0.27873	-2.08375
0	3.78255	-0.25473	-0.78540
С	2.16912	-1.42684	-2.05199
С	1.33177	-1.51973	-3.34149
С	0.31428	-2.66896	-3.29318

С	-0.88490	-2.41253	-2.38195
0	-0.48197	-2.50447	-0.98909
С	-1.51619	-1.03292	-2.59128
С	-0.46052	0.09353	-2.64908
С	0.64491	-0.17581	-3.70154
С	0.07061	-0.23268	-5.12383
С	-1.12403	1,46837	-2.78594
C	-2.12835	1.73218	-1.65863
0	-1.44797	1.62196	-0.38794
C	-3.23613	0.66612	-1.66241
c	-4.05783	0.77857	-2.95486
c	-2 54899	-0 71072	-1 51499
C	-3 70113	-1 67968	-1 27627
C	-4 68333	-0 85223	-0 /1261
C C	-4.00555	-0.05225	-0.41201
c	-5 22012	1 64220	-0.36451
C C	- 1 84674	2 09712	-0.30431
C C	-4.84074 6 12120	1 /2071	-0.40349
C C	7 11025	1.42071	1 01065
C C	-7.44035	2.19501	
	-0.20045	1 20400	2.2/2/0
0	-9.25811	1.28400	2.33020
0	-7.51939	2.20002	3.30340
C	-8.14422	1.94/03	4.60605
н	2.43996	0.69197	-4.50188
н	1.2/495	1.89670	-3.9/344
н	3.23916	1.84012	-2.44837
н	1.81262	1.29706	-1.54386
Н	3.95085	-0.45885	-2.83581
C	4.92297	0.46644	-0.69067
Н	1.53084	-1.28231	-1.16823
п	2.70534	-2.3/023	-1.89434
п	2.03/15	-1./4502	-4.16189
п	0.79607	-3.00831	-2.98362
п	-0.081/2	-2.84/84	-4.30469
H	-1.63/53	-3.196//	-2.53960
C	-0.30098	-3.//361	-0.51154
н	-2.03120	-1.09586	-3.5652/
н	0.03315	0.09849	-1.66180
н	-0.74464	-0.95919	-5.24238
Н	0.86423	-0.50///	-5.83606
н	-0.31234	0.75297	-5.42867
н	-1.64186	1.56641	-3./5169
н	-0.36913	2.26357	-2./525/
н	-2.52409	2.75140	-1./4/63
н	-4.95494	0.14441	-2.90/6/
н	-3.49336	0.48012	-3.84822
Н	-4.39984	1.81336	-3.11046
н	-2.00432	-0.63330	-0.56244
н	-3.3/363	-2.60286	-0.//664
н	-4.10135	-1.9/688	-2.23222
н	-4./2318	-1.23446	0.61684
	-2./09/0	-0.91849	-0.00/24
н	-3.55848	0.80009	0.45252
п	-0.03235	1.44233	-1.20285
п 		3.80364	-0.23253
Н	-4.48996	3.33049	-1.4/468

Н	-4.01756	3.26377	0.23908
Н	-6.34453	0.35761	1.06497
н	-5.49037	1.72347	1.79586
н	-7.27464	3.27924	1.00425
н	-8.09791	1,93225	0.16521
Н	-7.47727	2.33332	5.38323
н	-8.27192	0.86077	4,70880
н	-9,13238	2,42222	4,67152
C	-0 42318	2 50044	-0 18834
0	-0 3/0/2	3 57/82	-0 73871
N	0.15782	1 95205	0.79871
C	1 67375	2 /19/053	1 1/285
н	A 2/209	0 99751	0 99635
C	2 31881	1 80867	2 1853/
c	2.01001	3 62/17	0 581/6
c c	2.27101	1 05325	1 06670
c	1 15201	3 38357	2 10/00
c	4.13201	2,25063	2.10400
L L	1 02102	1 70709	2.00452
	4.03423	0.02100	2 61105
п	1.04545	0.92190	2.01105
п Ц	2 07262	4.15004	-0.22292
п		4.95052	0.01340
	5.12547	1 02100	2.404/0
N	5.43029	0 41909	-1.01084
C N	6 42207	0.41090	1 17206
с u	0.42297	1.15952	1 24209
C II	4.74447	0 80526	2 524200
c	7 16965	2 00571	2.52472
c	9 16931	2.09571	1 1/220
c	8 110651	2.80170	2 /0128
c	7 70320	1 61185	3 17808
с ц	7 90587	1 /0980	1 23073
н	6 13555	0 13565	3 06324
н Ц	6 95736	2 28106	-0 57442
н	8 7/106	2.28190	-0.57442 0 50130
н Ц	0.74100	2 12211	3 00001
0	-0 50061	-1 76622	_1 12157
N	0.339901	-4.70022	-1.13137
C	0.24005	-3.70077 1 77001	1 60402
с ц	0.52075	-4.77884	1 087/5
C	0.44050	-6 12/17	1 26160
c	0.54155	-7 10871	2 20105
c	1 11780	-6 78133	3 16950
c	1 20553	-5 /3886	3 80180
c	0 00851	-1 11396	2 87824
L L	-0.02617	-6 28785	2.07024
н Ц	-0.02017	-0.38785	1 02972
н Ц	1 2/716	-7 56403	1.92873
н	1 666/15	-5 1597/	4.19204 1 78897
н	1 12700	-3 20120	+,/0092 2 12700
C	-1 3576/	-0 92978	1 92201
0	-0 12010	-0 8/632	1 25178
н	-1 12040	-0 92720	7 07601
н	-1,88531	-1.87537	1,70183
н	-2 00523	-0 07919	1 71970
••	2.00525		

H -0.29946 -1	.06584
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0.31834

Supramolecular complex [2 MeOH·13]

С	2.22284	0.23063	-3.81473
С	2.91056	0.73469	-2.54956
С	3.77119	-0.37731	-1.96973
0	4.35000	0.01366	-0.70962
С	2.95323	-1.62439	-1.68938
С	2.21381	-2.12059	-2.94359
С	1.40091	-3.39120	-2.66757
С	0.13240	-3.15183	-1.85345
0	0.45841	-2.81863	-0.48063
С	-0.71284	-2.00151	-2.40086
С	0.11888	-0.73137	-2.69306
С	1.34341	-1.01904	-3.60671
С	0.90169	-1.48089	-5.00418
С	-0.78583	0.38890	-3.22087
С	-2.00892	0.68420	-2.34723
0	-1.58602	1.24612	-1.09592
С	-2.82881	-0.57417	-2.02908
С	-3.54909	-1.01947	-3.30779
С	-1.86250	-1.63958	-1.46859
С	-2.77918	-2.75624	-0.97199
С	-4.04350	-2.00274	-0.49353
С	-3.82600	-0.50349	-0.83490
C	-5.13043	0.29145	-0.98985
C	-4.86397	1.79022	-1.14176
C	-6.03862	0.03038	0.22264
C	-7.36977	0.78892	0.18262
C	-8.30234	0.31907	1.26927
0	-9.38548	-0.18163	1,10737
0	-7.76224	0.51575	2,48929
C	-8.57256	0.07326	3,57579
Н	3.00151	-0.03406	-4.55211
н	1.63598	1.03772	-4.27664
н	3,54218	1,60375	-2,77731
н	2.16976	1.04827	-1.79560
н	4,59443	-0.59728	-2.66818
C	5,42903	0.83085	-0.78532
н	2 25298	-1 39586	-0 87347
н	3 61397	-2 42184	-1 31363
н	2 99421	-2 38724	-3 67964
н	2 01395	-4 14777	-2 15675
н	1 08818	-3 84002	-3 62329
н	-0 /6163	-1 07580	-1 82104
C	0.40105	-3 85720	0 30197
н	-1 13770	-2 37867	-3 34686
Ц	0 50700	-0 38329	_1 71822
н	0.12530	_2 31325	-1 08120
н	1 77200	-2.51325	-4.90420 _5 50071
н	T.11022	-0 651205	-5.500/4
н	-1 1/206	0.00420 0 12700	- 1 22160
Ц	-1.14720 -0.000	1 32/15	-4.230060
Ц	-0.20072	1 /26/2	-005050 -0 27015
н	-2.00910	-1 20027	-2.07213
	エ・エノエノ ノ	T.0700/	J•16/71

Н	-2.85129	-1.28601	-4.11405
Н	-4.18891	-0.20665	-3.68502
Н	-1.42204	-1.16093	-0.57588
Н	-2.31066	-3.34793	-0.17354
Н	-3.01851	-3.45308	-1.79141
Н	-4.22098	-2.14529	0.58135
Н	-4.94209	-2.38097	-1.00687
Н	-3.26675	-0.03829	-0.00468
н	-5.66633	-0.07289	-1.88644
н	-5.79760	2.35995	-1.24822
н	-4.25404	2.02443	-2.02259
н	-4.33174	2.17413	-0.25382
Н	-6.26168	-1.04561	0.28721
н	-5,49928	0.29836	1.14741
Н	-7.20641	1.86685	0.32643
Н	-7.88466	0.63575	-0.77536
Н	-8.01317	0.31364	4,48537
Н	-8.75307	-1.00814	3.50594
Н	-9.54185	0.58989	3,56906
C	-1.59303	2,60669	-1.01603
0	-1.65516	3.33602	-1.98092
N	-1.50932	2.95257	0.29625
C	-1.40148	4.24450	0.83458
Н	-1.59817	2.17279	0.95698
C	-1.29720	4.33322	2,23046
c	-1.38937	5.41166	0.06086
c	-1.27256	6.64661	0.69695
C	-1.17084	6.74156	2.08304
С	-1.18501	5.57397	2.84664
Н	-1.10607	5.62707	3.93347
Н	-1.30392	3.41249	2.81857
Н	-1.47429	5.34253	-1.02017
Н	-1.26387	7.55186	0.08813
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0	5.90811	1.23489	-1.81878
Ν	5.85977	1.10275	0.48349
С	6.93847	1.91581	0.87328
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С	7.18274	2.03392	2.24799
С	7.75519	2.59909	-0.03606
С	8.79763	3.38728	0.44807
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Н	6.54443	1.49923	2.95542
Н	7.57049	2.50779	-1.10241
Н	9.42896	3.91695	-0.26679
Н	9.86306	4.13045	2.17445
0	0.87439	-5.00861	-0.06423
Ν	1.29124	-3.36728	1.50454
С	1.76235	-4.10410	2.60650
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2.05863	-3.37705	3.76777
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3.11062	-5.91553	5.77698
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1.90586	-2.29520	3.77419
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0.52735	-0.54284	1.12763
0.65991	1.27200	0.10298
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-1.05801	0.06606	1.81827
	2.05863 1.72016 2.57927 3.11062 2.76714 1.90586 1.19353 2.23850 0.52735 0.65991 1.17431 0.42125 -2.75885 -1.71834 -2.39340 -3.26300 -3.50050 -1.05801	2.05863 -3.37705 1.72016 -6.05905 2.57927 -7.20490 3.11062 -5.91553 2.76714 -3.44254 1.90586 -2.29520 1.19353 0.69529 2.23850 0.53077 0.52735 -0.54284 0.65991 1.27200 1.17431 1.26704 0.42125 -1.03296 -2.75885 0.14262 -1.71834 0.73769 -2.39340 -0.26506 -3.26300 0.92346 -1.05801 0.06606

Supramolecular complex [3 MeOH·13]

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Н	-0.28932	-1.75510	-5,52348
Н	-1.79293	-0.79258	-4.19461
Н	-0.82492	0.57153	-3.62993
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Н	-2.44351	-3.47148	0.51851
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Н	-4.23125	-2.09834	1.25865
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Н	-3.52879	-0.15015	0.14294
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Н	8.10799	2.92936	3.00466
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Н	9.31308	4.31184	1.31002
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Ν	1.43662	-2.84803	1.75237
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Н	1.18412	-1.86098	1.69073
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Н	2.34755	0.86757	3.33838
Н	4.03398	0.94839	3.95174
Н	3.53940	1.93579	2.54865
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С	-2.15202	0.78271	3.16667
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Н	-1.24105	1.07840	3.71674
Н	-2.41898	-0.24697	3.45754
Н	-2.97058	1.44805	3.47247
Н	-1.25416	0.33301	1.48350