

Supporting Information

Switching of Cell Proliferation/Differentiation in Thiol-Maleimide Clickable Microcapsules Triggered by *in Situ* Conjugation of Biomimetic Peptides

Yuichiro Oki,[†] Katsuhisa Kirita,[‡] Seiichi Ohta,[§] Shinsuke Ohba,^{‡,§} Ikki Horiguchi,[⊥]

Yasuyuki Sakai,^{†,‡} and Taichi Ito^{,†,‡,§}*

[†]Department of Chemical System Engineering, The University of Tokyo, 7-3-1 Hongo, Bunkyo-ku, Tokyo 113-8656, Japan

[‡]Department of Bioengineering, The University of Tokyo, 7-3-1 Hongo, Bunkyo-ku, Tokyo 113-8656, Japan

[§]Center for Disease Biology and Integrative Medicine, The University of Tokyo, Hongo 7-3-1, Bunkyo-ku, Tokyo 113-0033, Japan

[⊥]Department of Biotechnology, Graduate School of Engineering, Osaka University, 2-1 Yamada-oka, Suita, Osaka 565-0871, Japan

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Estimation of characteristic diffusion time and Thiele modulus

Supplementary Note 2

Simulation of reaction-diffusion process in the microcapsule

Table S1. The primer sequences used in reverse transcription quantitative polymerase chain reaction (RT-qPCR) analysis.

<u>GAPDH</u>
F: 5' AAATGGTGAAGGTCGGTGTG 3'
R: 5' TGAAGGGGTCGTTGATGG 3'
<u>Col1a1 (Col I)</u>
F: 5' CTGACTGGAAGAGCGGAGAGTAC 3'
R: 5' ACAGACGGCTGAGTAGGGAACA 3'
<u>bFGF</u>
F: 5' GCTATGAAGGAAGATGGACG 3'
R: 5' GCCACATACCAACTGGAGTATTT 3'
<u>TGF-β</u>
F: 5' ATCCTGTCCAAACTAAGGCTCG 3'
R: 5' ACCTCTTTAGCATAGTAGTCCGC 3'

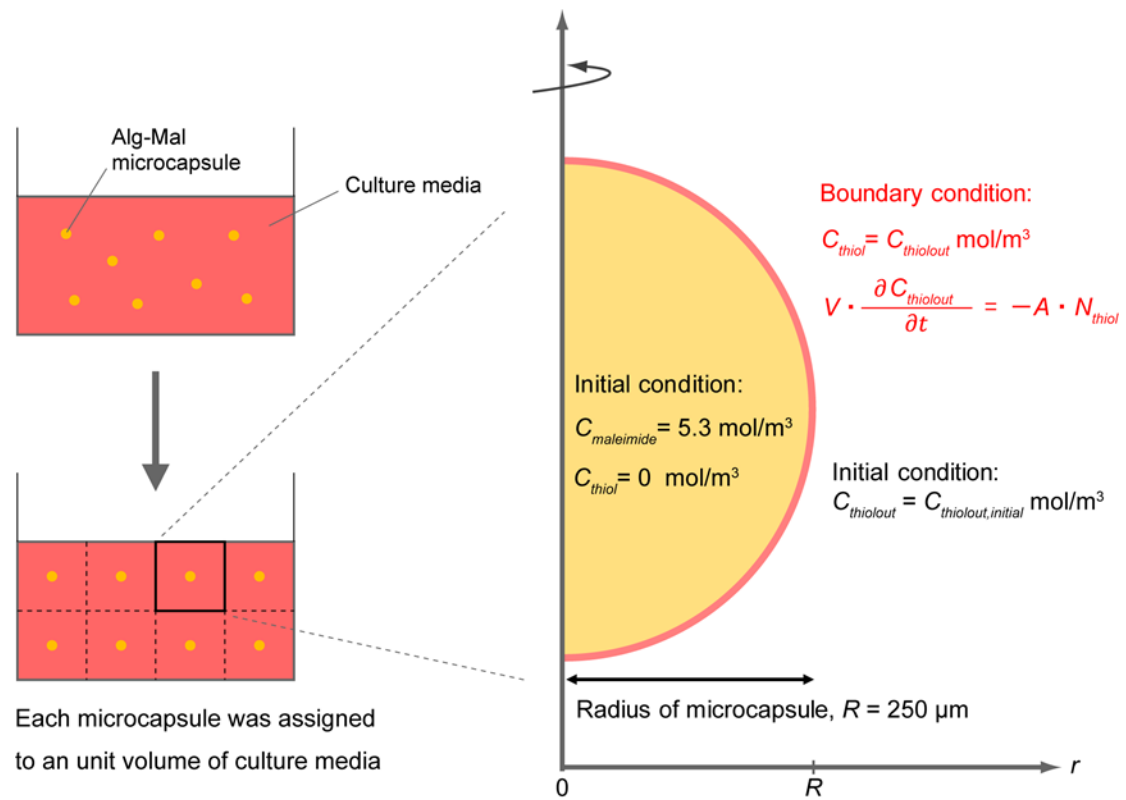


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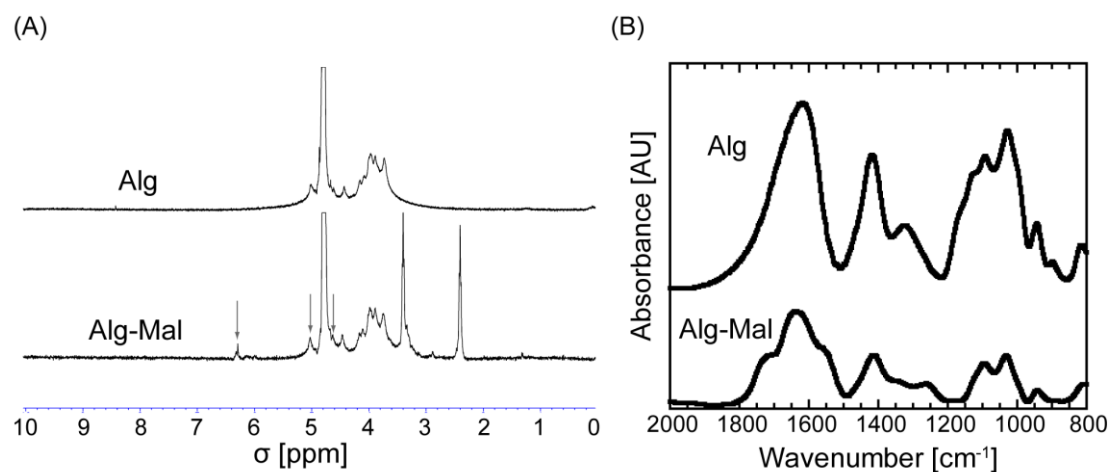


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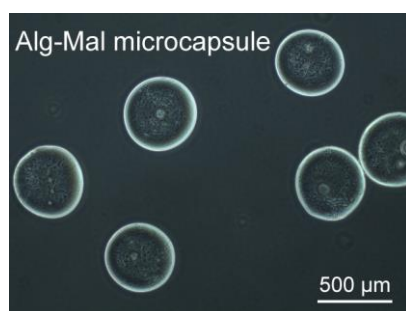


Figure S3. Observation of Alg-Mal microcapsules via optical microscopy.

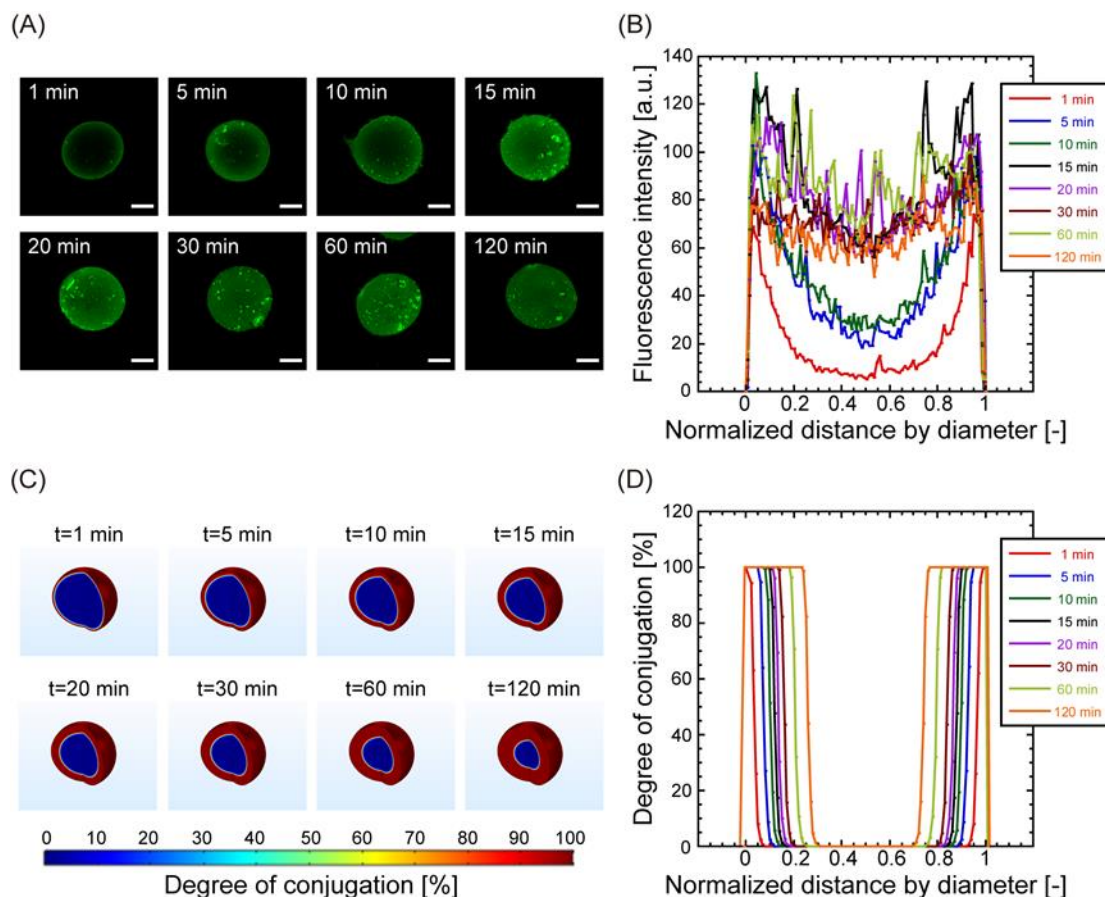


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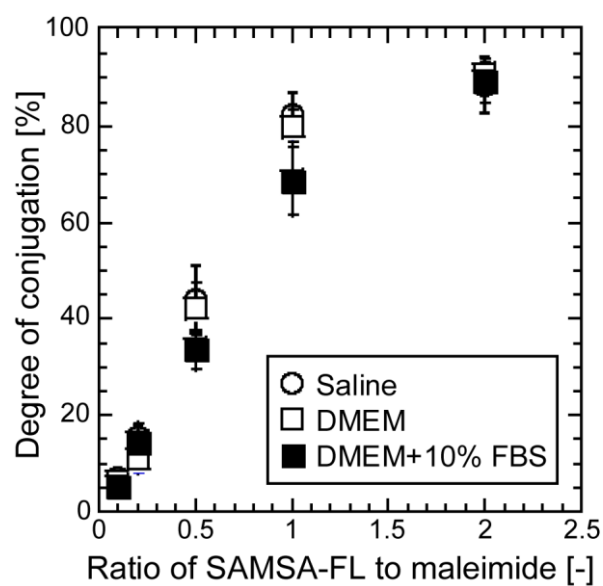


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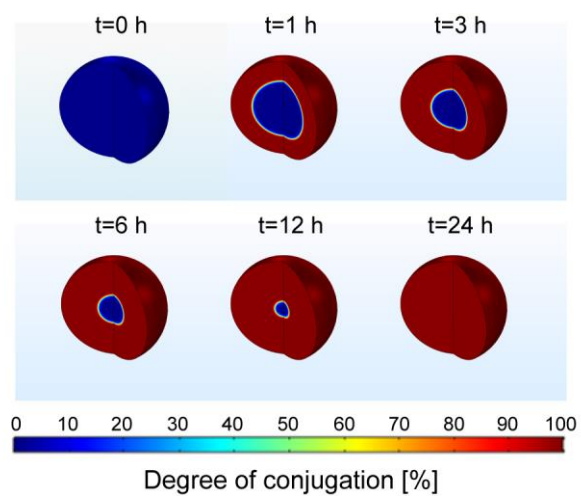


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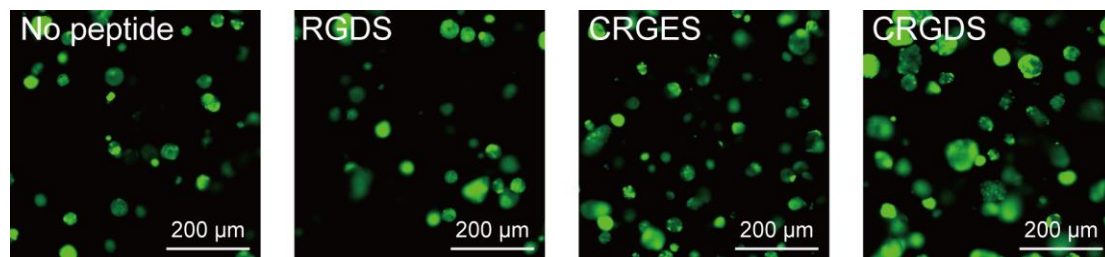


Figure S7. Enlarged view of NIH-3T3 cells encapsulated in Alg-Mal microcapsules 7 days after no addition of any peptide or addition of RGDS, CRGES, or CRGDS. The cells were stained using calcein (green). Results for the sample without peptide addition are also shown as no peptide.

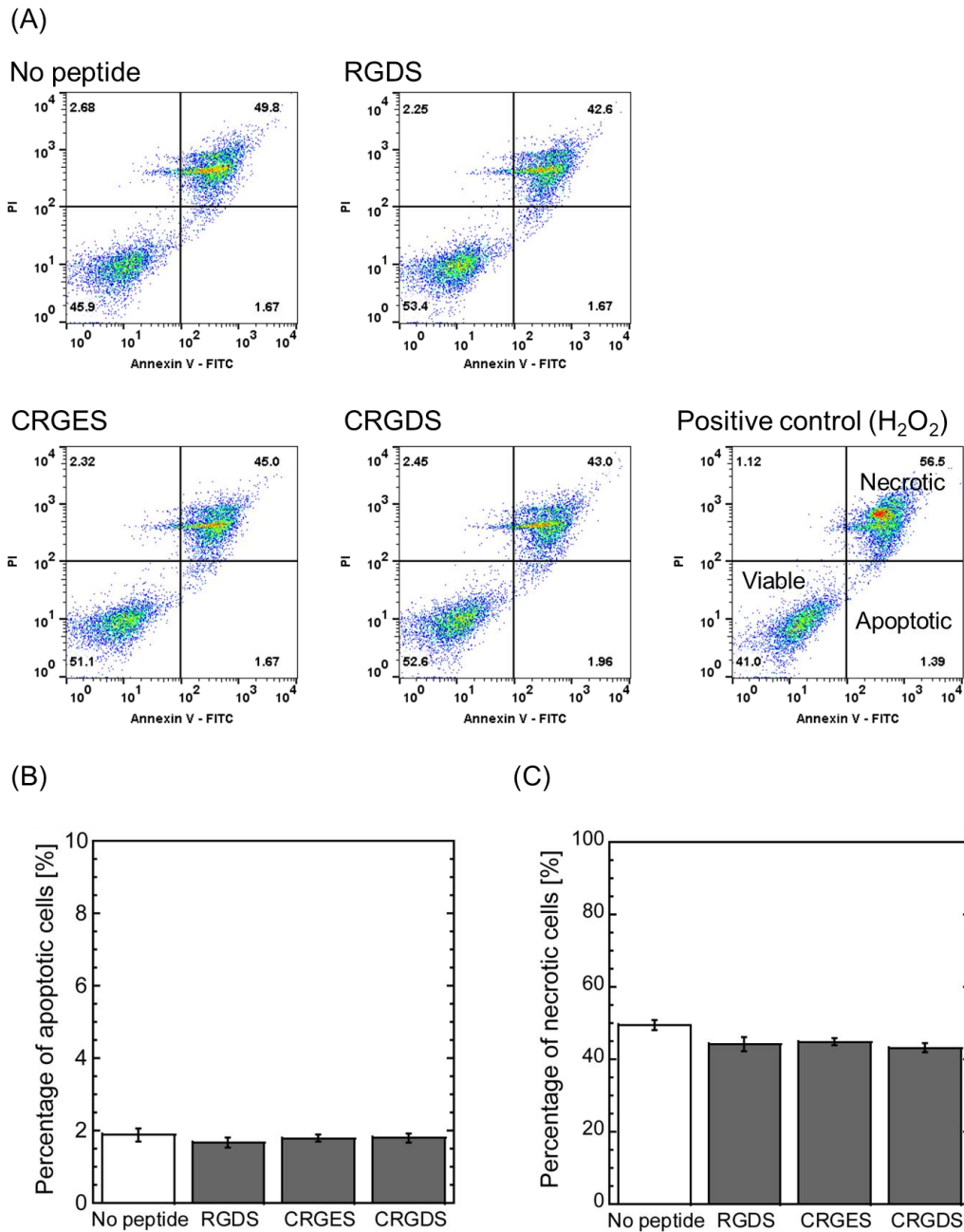


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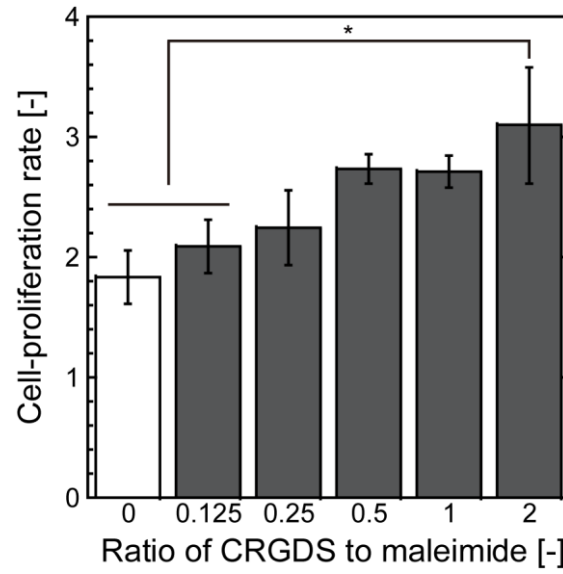


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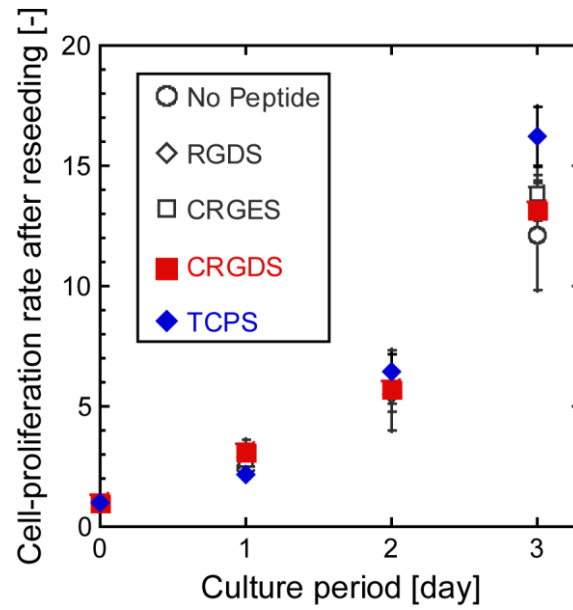


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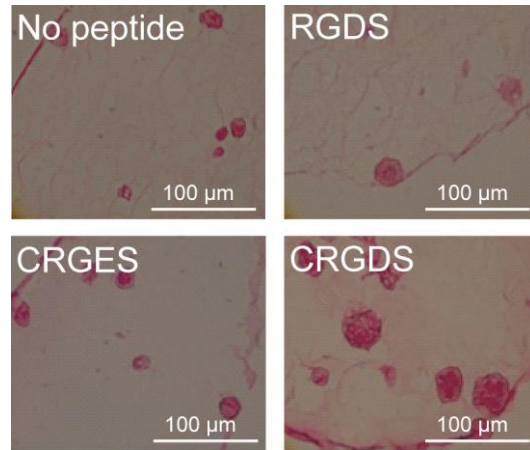


Figure S11. Effect of *in situ* conjugation of various peptides on collagen production, evaluated by sirius red staining.

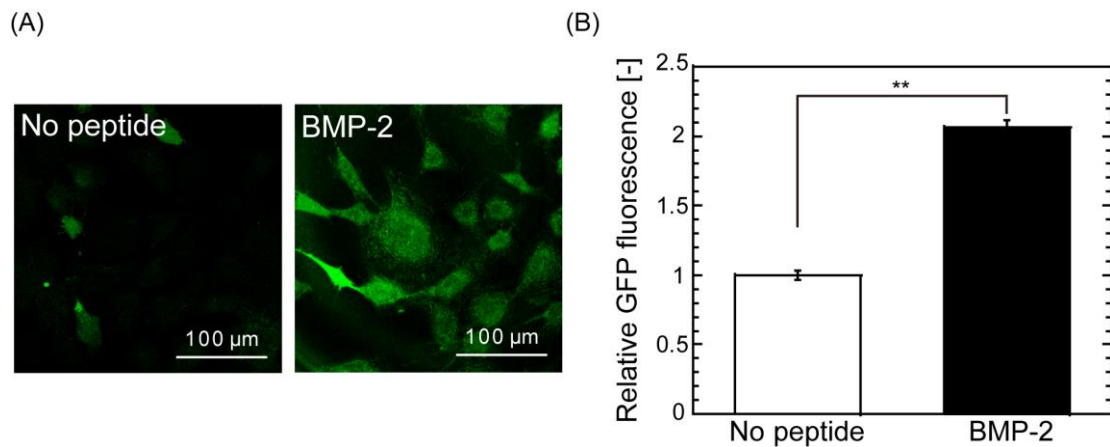


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Supplementary Note 1

Estimation of characteristic diffusion time and Thiele modulus

The characteristic time for the diffusion of SAMSA-FL, τ , was estimated using the following equation based on a previous report [1];

$$\tau = \frac{L^2}{D} \quad (\text{eq. 1})$$

Here, L and D are the diameter of the microcapsules and diffusion coefficient of SAMSA-FL inside the microcapsule, respectively. Because the hydrodynamic radius of fluorescein, which has a similar chemical structure to SAMSA-FL, is reported as ca. 0.5 nm [2], which is much smaller than the alginate hydrogel matrix, it diffuses rapidly through the Alg matrix due to the negligible resistance of the hydrogel network [3]. Therefore, the diffusion coefficient in the microcapsule, D , is considered almost the same as that in water and can thus be estimated as $4.3 \times 10^{-10} \text{ m}^2 \text{ s}^{-1}$ using the reported value for fluorescein [2]. Additionally, according to the microscopic observation of microcapsules shown in Figure 2A, the radius of the microcapsules L was ca. 250 μm . By using these values, τ was estimated as 2.4 min, which was consistent with the results in Figure 2A and B.

The Thiele modulus ϕ was also calculated, according to the following equation based on a previous report [4]:

$$\phi = L \sqrt{\frac{kC_{int}}{D}} \quad (\text{eq. 2})$$

Here, k and C_{int} are the reaction rate constant and initial concentration of SAMSA-FL in the media. The reaction rate constant k between DNP-PEG4-Cys and N-(2-aminoethyl)maleimide was reported as $k = 5.2 \times 10^3 \text{ M}^{-1} \text{ s}^{-1}$ at pH 7.4 [5]. Using this reported value as a representative rate constant, ϕ was calculated as 5.1 and 27 when $C_{int} = 0.035$ and 1 mM.

Supplementary Note 2

Simulation of Reaction-diffusion process in the microcapsule

Model Overview, Assumptions and Geometry

The simulation used an axisymmetric domain. Figure S1 shows the model geometry and boundary and initial conditions of the simulation. By assuming a homogeneous microcapsule distribution, each Alg-Mal microcapsule was assigned to a unit volume of culture media. This unit volume was chosen as a domain for the model calculation. It was assumed that the microcapsule was spherical and that reaction points, i.e., maleimide groups, were equally distributed in the microcapsule. The diameter of the microcapsule was assumed to be 500 μm . During reaction-diffusion, the diameter of the microcapsules and diffusion coefficient of SAMSA-FL were assumed to be constant. Additionally, it was assumed that SAMSA-FL outside the microcapsule was completely mixed and that there was no concentration distribution.

Definition of Governing Equations and Parameters

To simulate the *in situ* conjugation of SAMSA-FL and the immobilized maleimide groups in the microcapsules, the following reaction-diffusion equations were used;

$$\frac{\partial C_{maleimide}}{\partial t} = -kC_{thiol}C_{maleimide} \quad (\text{eq. 3})$$

$$\frac{\partial C_{thiol}}{\partial t} = \nabla \cdot (D \nabla C_{thiol}) - k C_{thiol} C_{maleimide} \quad (\text{eq. 4})$$

$$\frac{\partial C_{thiolout}}{\partial t} = - \frac{A}{V} \cdot N_{thiol} \quad (\text{eq. 5})$$

Here, $C_{maleimide}$ is the concentration of the immobilized maleimide groups. The concentrations of free SAMSA-FL inside and outside the microcapsule are defined as C_{thiol} and $C_{thiolout}$, respectively, while k is the rate constant for the reaction between SAMSA-FL and maleimide groups. The boundary flux of SAMSA-FL was defined as N_{thiol} at the microcapsule surface. Additionally, A and V are the surface area of the microcapsules and unit volume of the media assigned to each microcapsule.

For the boundary conditions, the following equation was applied:

$$C_{thiol} = C_{thiolout} \text{ for all the surface of the spherical microcapsule (eq. 6)}$$

$$\nabla C_{thiol} = 0 \text{ for the center point of the spherical microcapsule (eq. 7);}$$

$k = 5.2 \text{ m}^3 \text{ mol}^{-1} \text{ s}^{-1}$ and $D = 4.3 \times 10^{-10} \text{ m}^2 \text{ s}^{-1}$ were used for the calculation, according to previous reports [2, 5]. $A = 7.9 \times 10^{-7} \text{ m}^2$, $V = 1.0 \times 10^{-8} \text{ m}^3$, $C_{maleimide,initial} = 5.3 \text{ mol m}^{-3}$, $C_{thiol,initial} = 0 \text{ mol m}^{-3}$, and $C_{thiolout,initial} = 3.5 \times 10^{-2}$ or 1.0 mol m^{-3} were used based on the experiments.

Numerical Implementation

The equations were solved using a commercial finite element package, COMSOL Multiphysics version 5.4 (COMSOL Multiphysics Burlington, MA). The COMSOL

module, *Transport of Diluted Species*, was used to solve equations 3–7. The mesh comprised triangular elements with minimum and maximum sizes of 0.01 and 5 μm , respectively. The relative tolerance was set as 0.005. A backward differentiation formula stepping algorithm was used with a maximum time step of 1 s thereafter. It was confirmed that even if the mesh size and time step were smaller, the calculation result would hardly be affected.

Supplementary References

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- (3) Amsden, B., Solute Diffusion within Hydrogels. Mechanisms and Models. *Macromolecules* **1998**, *31*, 8382–8395.
- (4) Magyari, E., Exact analytical solutions of diffusion reaction in spherical porous catalyst. *Chem. Eng. J.* **2010**, *158*, 266–270.

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