## Supporting Information

Alkyne-alkyne photo-cross-linking on the flipping-out field<br>Kazumitsu Onizuka,*,t,申 Kei Ishida, ${ }^{\dagger, \neq}$ Eriko Mano, ${ }^{\dagger}$ Fumi Nagatsugi*,t, ${ }^{\text {T }}$<br>${ }^{\dagger}$ Institute of Multidisciplinary Research for Advanced Materials, Tohoku University, 2-1-1 Katahira, Aoba-ku, Sendai, Miyagi 980-8577, Japan.<br>${ }^{\text { }}$ Department of Chemistry, Graduate School of Science, Tohoku University, Aoba-ku, Sendai 980-8578, Japan.

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## Materials and methods

The general chemicals were purchased from FUJIFILM Wako Pure Chemical, the Tokyo Chemical Industry, Kanto Chemical or Aldrich. The target DNAs and RNAs were purchased from JBioS (Japan). The ${ }^{1} \mathrm{H}$ NMR spectra ( 400 MHz ) were recorded by a Bruker 400 spectrometer. The ${ }^{1} \mathrm{H}$ NMR spectra ( 600 MHz ) and ${ }^{13} \mathrm{C}$ NMR spectra ( 150 MHz ) were recorded by a Bruker AVANCE III 600 spectrometer. The high resolution electrospray mass analysis was performed by a Bruker MicrOTOFQ II. The HPLC purification was performed by a JASCO HPLC System (PU-2089Plus, UV-2075Plus, FP-2015Plus and CO-2065Plus). MALDI-TOF MS measurements were performed by a Bruker Autoflex speed instrument using a 3-hydroxypicolinic acid/diammonium hydrogen citrate matrix.

## Synthesis of compound 3

To a solution of 3-bromo-5-methyl-2-pyridone ( $582 \mathrm{mg}, 3.10 \mathrm{mmol}$ ) in $\mathrm{CH}_{3} \mathrm{CN}$ $(2.0 \mathrm{~mL})$, was added $N, O$-bis(trimethylsilyl)acetamide ( $760 \mu \mathrm{~L}, 3.10 \mathrm{mmol}$ ) and the mixture was stirred at room temperature. After 30 min , to the mixture were added $\mathrm{CH}_{3} \mathrm{CN}(21.5 \mathrm{~mL})$ and 2-deoxy-3,5-di-O-p-toluoyl-ribofuranosyl chloride $(1.0 \mathrm{~g}, 2.58 \mathrm{mmol})$ and cooled to $0{ }^{\circ} \mathrm{C}$. A solution of $\mathrm{SnCl}_{4}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1 \mathrm{M}, 650$ $\mu \mathrm{L}, 0.65 \mathrm{mmol}$ ) was added, and the mixture was stirred at room temperature. After 5 h , the mixture was diluted with EtOAc ( 230 mL ). The organic phase was washed with saturated aqueous $\mathrm{NaHCO}_{3}(90 \mathrm{~mL} \times 3)$ and brine ( 90 mL ). The organic phase was dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated under reduced pressure. The residue was purified by silica gel column chromatography (hexane/ $\mathrm{EtOAc}=4 / 1$ ) to give the compound $\mathbf{3}$ ( $\beta$-anomer, 520 $\mathrm{mg}, 37 \%)$ as a white foam. The $\alpha$-anomer ( $486 \mathrm{mg}, 35 \%$ ) was obtained as a white foam.
$\boldsymbol{\beta}$-anomer: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 600 \mathrm{MHz}\right) \delta(\mathrm{ppm}) 7.95(2 \mathrm{H}, \mathrm{d}, J=8.4 \mathrm{~Hz}), 7.87(2 \mathrm{H}, \mathrm{d}, J=8.4 \mathrm{~Hz})$, $7.55(1 \mathrm{H}, \mathrm{d}, J=2.4 \mathrm{~Hz}), 7.44(1 \mathrm{H}, \mathrm{dd}, J=1.2,2.4 \mathrm{~Hz}), 7.27(2 \mathrm{H}, \mathrm{m}), 7.22(2 \mathrm{H}, \mathrm{m}), 6.56(1 \mathrm{H}, \mathrm{dd}, J=$ $5.4,8.4 \mathrm{~Hz}), 5.62(1 \mathrm{H}, \mathrm{ddd}, J=1.8,1.8,6.6 \mathrm{~Hz}), 4.83(1 \mathrm{H}, \mathrm{dd}, J=3.0,12 \mathrm{~Hz}), 4.65(1 \mathrm{H}, \mathrm{dd}, J=3.6$, $12 \mathrm{~Hz}), 4.62(1 \mathrm{H}, \mathrm{m}), 3.02(1 \mathrm{H}, \mathrm{ddd}, J=1.8,5.4,14.4 \mathrm{~Hz}), 2.43(3 \mathrm{H}, \mathrm{s}), 2.41(3 \mathrm{H}, \mathrm{s}), 2.24(1 \mathrm{H}$, ddd, $J=6.6,8.4,14.4 \mathrm{~Hz}), 1.84(3 \mathrm{H}, \mathrm{s}) .{ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{CDCl}_{3}, 150 \mathrm{MHz}\right) \delta(\mathrm{ppm}) 166.2,166.1,157.4,144.5$, $144.4,143.9,129.5,129.4,129.3,128.6,126.5,126.4,115.6,115.5,87.4,83.6,75.1,64.1,39.3,21.8$, 21.7, 17.0. ESI-HRMS $(\mathrm{m} / \mathrm{z}):[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{27} \mathrm{H}_{27} \mathrm{BrNO}_{6}{ }^{+}, 540.1016$, found 540.1019.
$\boldsymbol{\alpha}$-anomer: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}, 600 \mathrm{MHz}\right) \delta(\mathrm{ppm}) 7.95(2 \mathrm{H}, \mathrm{d}, J=8.4 \mathrm{~Hz}), 7.67(1 \mathrm{H}, \mathrm{d}, J=2.4 \mathrm{~Hz})$, $7.65(2 \mathrm{H}, \mathrm{d}, J=7.8 \mathrm{~Hz}), 7.47(1 \mathrm{H}, \mathrm{m}), 7.27(2 \mathrm{H}, \mathrm{d}, J=7.8 \mathrm{~Hz}), 7.17(2 \mathrm{H}, \mathrm{d}, J=7.8 \mathrm{~Hz}), 6.47(1 \mathrm{H}$, dd, $J=1.2,7.2 \mathrm{~Hz}), 5.60(1 \mathrm{H}, \mathrm{d}, J=6.0 \mathrm{~Hz}), 4.95(1 \mathrm{H}, \mathrm{dd}, J=4.2,4.2 \mathrm{~Hz}), 4.55(2 \mathrm{H}, \mathrm{m}), 3.02(1 \mathrm{H}$, ddd, $J=6.6,6.6,15.6 \mathrm{~Hz}), 2.59(1 \mathrm{H}, \mathrm{d}, J=15.6 \mathrm{~Hz}), 2.42(3 \mathrm{H}, \mathrm{s}), 2.39(3 \mathrm{H}, \mathrm{s}), 2.05(3 \mathrm{H}, \mathrm{s}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 150 \mathrm{MHz}\right) \delta(\mathrm{ppm}) 166.1,165.4,157.4,144.4,144.2,143.8,129.7,129.5,129.3$, 129.1, 126.5, 126.3, 115.6, 114.6, 89.6, 85.8, 74.8, 64.1, 39.1, 21.7, 17.3. ESI-HRMS $(\mathrm{m} / \mathrm{z}):[\mathrm{M}+\mathrm{H}]^{+}$ calcd for $\mathrm{C}_{27} \mathrm{H}_{27} \mathrm{BrNO}_{6}{ }^{+}, 540.1016$, found 540.1022.

## Synthesis of compound 4



To a solution of $\mathbf{3}(636 \mathrm{mg}, 1.17 \mathrm{mmol})$ in $\mathrm{MeOH}(5.8 \mathrm{~mL})$, was added a solution of MeONa in $\mathrm{MeOH}(581 \mu \mathrm{~L}, 2.83 \mathrm{mmol})$ and the mixture was stirred at room temperature. After 1 h , the mixture was concentrated under reduced pressure. The residue was purified by silica gel column chromatography $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}=9 / 1\right)$ to give the compound $\mathbf{4}(352 \mathrm{mg}, 98 \%)$ as a white solid.
${ }^{1} \mathrm{H}$ NMR (MeOD- $\left.d_{4}, 600 \mathrm{MHz}\right): \delta(\mathrm{ppm}) 7.97(1 \mathrm{H}, \mathrm{s}), 7.82(1 \mathrm{H}, \mathrm{d}, J=2.4 \mathrm{~Hz}), 6.39(1 \mathrm{H}, \mathrm{t}, J=6.0$ $\mathrm{Hz}), 4.36(1 \mathrm{H}, \mathrm{m}), 3.98(1 \mathrm{H}, \mathrm{dd}, J=3.6,7.2 \mathrm{~Hz}), 3.83(1 \mathrm{H}, \mathrm{dd}, J=3.6,12 \mathrm{~Hz}), 3.75(1 \mathrm{H}, \mathrm{dd}, J=4.2$,
$12 \mathrm{~Hz}), 2.48(1 \mathrm{H}$, ddd, $J=4.2,6.0,13.2 \mathrm{~Hz}), 2.12(3 \mathrm{H}, \mathrm{s}), 2.08(1 \mathrm{H}$, ddd, $J=6.0,7.2,13.2 \mathrm{~Hz}){ }^{13} \mathrm{C}$ NMR (MeOD- $d_{4}, 150 \mathrm{MHz}$ ): $\delta(\mathrm{ppm}) 159.2,146.1,131.8,117.9,115.2,89.3,88.4,71.5,62.4,42.6$, 17.0. ESI-HRMS $(\mathrm{m} / \mathrm{z}):[\mathrm{M}+\mathrm{Na}]^{+}$calcd for $\mathrm{C}_{11} \mathrm{H}_{14} \mathrm{BrNNaO}_{4}{ }^{+}, 325.9998$, found 325.9996.

## Synthesis of compound 5



Compound 4 ( $245 \mathrm{mg}, 0.81 \mathrm{mmol}$ ) was co-evaporated with acetonitrile and pyridine, then dissolved in pyridine ( 4.1 mL ). To the solution was added DMTrCl ( $548 \mathrm{mg}, 1.62 \mathrm{mmol}$ ), and the mixture was stirred at room temperature for 1 h . The reaction mixture was then diluted with EtOAc ( 150 mL ), washed with water $(90 \mathrm{~mL})$ and brine ( 90 mL ). The organic layer was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, then concentrated under reduced pressure. The crude was purified by silica gel column chromatography (Hexane:Ethyl acetate $=3: 1 \rightarrow 2: 1 \rightarrow 1: 3$ ) to give 5 as a white foam ( $463 \mathrm{mg}, 95 \%$ ).
${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right): \delta(\mathrm{ppm}) 7.79(1 \mathrm{H}, \mathrm{s}), 7.60(1 \mathrm{H}, \mathrm{d}, J=2.4 \mathrm{~Hz}), 7.40(2 \mathrm{H}, \mathrm{m}), 7.31-7.27$ $(6 \mathrm{H}, \mathrm{m}), 7.22(1 \mathrm{H}, \mathrm{m}), 6.82(4 \mathrm{H}, \mathrm{dd}, J=3.6,14.4 \mathrm{~Hz}), 6.57(1 \mathrm{H}, \mathrm{t}, J=6.0 \mathrm{~Hz}), 4.51(1 \mathrm{H}, \mathrm{m}), 4.21$ $(1 \mathrm{H}, \mathrm{dt}, J=3.6,6.6 \mathrm{~Hz}), 3.79(6 \mathrm{H}, \mathrm{s}), 3.51(1 \mathrm{H}, \mathrm{dd}, J=3.6,10.2 \mathrm{~Hz}), 3.37-3.35(2 \mathrm{H}, \mathrm{m}), 2.79(1 \mathrm{H}$, ddd, $J=3.6,6.6,13.8 \mathrm{~Hz}), 2.24(1 \mathrm{H}, \mathrm{ddd}, J=6.6,7.2,13.8 \mathrm{~Hz}), 1.75(3 \mathrm{H}, \mathrm{s}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(\mathrm{CDCl}_{3}, 150\right.$ $\mathrm{MHz}): \delta(\mathrm{ppm}) 158.6,157.7,144.5,144.1,135.6,135.5,130.1,130.0,129.8,128.1,127.9,127.0$, 115.6, 115.2, 113.2, 87.3, 86.8, 72.2, 63.4, 55.2, 42.3, 16.9. ESI-HRMS $(\mathrm{m} / \mathrm{z}):[\mathrm{M}+\mathrm{Na}]^{+}$calcd for $\mathrm{C}_{32} \mathrm{H}_{32} \mathrm{BrNNaO}_{6}{ }^{+}, 628.1305$, found 628.1302 .

## Synthesis of compound 6-Ph



To a solution of $5(33.7 \mathrm{mg}, 0.056 \mathrm{mmol})$ in DMF $(100 \mu \mathrm{~L})$ were added ethynylbenzene ( $6.8 \mathrm{mg}, 0.067 \mathrm{mmol}$ ), trimethylamine ( $140 \mu \mathrm{~L}, 1.0 \mathrm{mmol}$ ), $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}(13 \mathrm{mg}, 0.011 \mathrm{mmol})$ and $\mathrm{CuI}(2.0 \mathrm{mg}, 0.011 \mathrm{mmol})$. The reaction mixture was stirred at $80^{\circ} \mathrm{C}$ for 4 h and then allowed to cool at room temperature. The reaction mixture was diluted with EtOAc and filtered through Celite. The filtrate was concentrated under reduced pressure and the residue was diluted with water $(15 \mathrm{~mL})$. The mixture was extract with EtOAc ( $15 \mathrm{~mL} \times 3$ ) and the organic phases were washed with brine ( 15 mL ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, then concentrated under reduced pressure. The crude was purified by silica gel column chromatography (Hexane:Ethyl acetate $=1: 1$ ) to give $\mathbf{6}(\mathbf{P h})$ as a white foam $(24.9 \mathrm{mg}, 71 \%)$.
${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 600 \mathrm{MHz}\right): \delta(\mathrm{ppm}) 7.79(1 \mathrm{H}, \mathrm{s}), 7.56(2 \mathrm{H}, \mathrm{dd}, J=7.8,1.8 \mathrm{~Hz}), 7.48(1 \mathrm{H}, \mathrm{d}, J=$ $2.4 \mathrm{~Hz}), 7.41(2 \mathrm{H}, \mathrm{m}), 7.31-7.26(9 \mathrm{H}, \mathrm{m}), 7.21(1 \mathrm{H}, \mathrm{m}), 6.81(4 \mathrm{H}, \mathrm{dd}, J=9.0,2.4 \mathrm{~Hz}), 6.61(1 \mathrm{H}, \mathrm{t}, J$ $=6.0 \mathrm{~Hz}), 4.51(1 \mathrm{H}, \mathrm{m}), 4.23(1 \mathrm{H}, \mathrm{m}), 3.78(6 \mathrm{H}, \mathrm{s}), 3.49(1 \mathrm{H}, \mathrm{dd}, J=10.2,3.6 \mathrm{~Hz}), 3.33(1 \mathrm{H}, \mathrm{dd}, J=$ $10.2,3.6 \mathrm{~Hz}), 2.82(1 \mathrm{H}, \mathrm{ddd}, J=13.2,6.0,3.6 \mathrm{~Hz}), 2.23(1 \mathrm{H}, \mathrm{ddd}, J=13.2,7.2,6.0 \mathrm{~Hz}), 1.78(3 \mathrm{H}, \mathrm{s})$. ${ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}, 150 \mathrm{MHz}$ ): $\delta(\mathrm{ppm}) 160.7$, 158.7, 145.0, 144.7, 135.8, 135.8, 131.9, 130.6, 130.2, $128.4,128.3,128.3,128.1,127.1,123.3,115.3,115.1,113.4,113.2,94.9,86.9,86.8,86.7,85.2,72.3$, 63.6, 55.4, 42.5, 17.2. ESI-HRMS $(\mathrm{m} / \mathrm{z}):[\mathrm{M}+\mathrm{Na}]^{+}$calcd for $\mathrm{C}_{40} \mathrm{H}_{37} \mathrm{NNaO}_{6}{ }^{+}, 650.2513$, found 650.2515 .

## Synthesis of compound 6-An



To a solution of $5(65 \mathrm{mg}, 0.11 \mathrm{mmol})$ in DMF ( 1.1 mL ) were added 9ethynylanthracene ( $55 \mathrm{mg}, 0.28 \mathrm{mmol}$ ), trimethylamine ( $154 \mu \mathrm{~L}, 1.1$ $\mathrm{mmol}), \mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}(25 \mathrm{mg}, 0.022 \mathrm{mmol})$ and $\mathrm{CuI}(4.2 \mathrm{mg}, 0.022 \mathrm{mmol})$. The reaction mixture was stirred at $80^{\circ} \mathrm{C}$ for 1.5 h and then allowed to cool at room temperature. The reaction mixture was diluted with EtOAc and filtered through Celite. The filtrate was concentrated under reduced pressure and the residue was diluted with water $(45 \mathrm{~mL})$. The mixture was extract with EtOAc ( $45 \mathrm{~mL} \times 3$ ) and the organic phases were washed with brine ( 45 mL ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, then concentrated under reduced pressure. The crude was purified
by silica gel column chromatography (Hexane:Ethyl acetate $=1: 1$ ) to give $\mathbf{6}(\mathbf{A n})$ as a red brown foam ( $74.8 \mathrm{mg}, 96 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{CDCl}_{3}, 600 \mathrm{MHz}\right): \delta(\mathrm{ppm}) 8.78(2 \mathrm{H}, \mathrm{dd}, J=1.2,9.0 \mathrm{~Hz}), 8.41(1 \mathrm{H}, \mathrm{s}), 7.99(2 \mathrm{H}, \mathrm{d}, J=$ $8.4 \mathrm{~Hz}), 7.82(1 \mathrm{H}, \mathrm{m}), 7.70(1 \mathrm{H}, \mathrm{d}, J=2.4 \mathrm{~Hz}), 7.59(2 \mathrm{H}, \mathrm{t}, J=7.2 \mathrm{~Hz}), 7.47(2 \mathrm{H}, \mathrm{t}, J=7.2 \mathrm{~Hz})$, $7.42(2 \mathrm{H}, \mathrm{d}, J=7.2 \mathrm{~Hz}), 7.32-7.22(7 \mathrm{H}, \mathrm{m}), 6.82(4 \mathrm{H}, \mathrm{m}), 6.68(1 \mathrm{H}, \mathrm{t}, J=6.6 \mathrm{~Hz}), 4.55(1 \mathrm{H}, \mathrm{m})$, $4.20(1 \mathrm{H}, \mathrm{dt}, J=3.6,6.6 \mathrm{~Hz}), 3.77(6 \mathrm{H}, \mathrm{s}), 3.53(1 \mathrm{H}, \mathrm{dd}, J=3.6,10.8 \mathrm{~Hz}), 3.36(1 \mathrm{H}, \mathrm{dd}, J=3.6$, $10.8), 2.83(1 \mathrm{H}, \mathrm{ddd}, J=4.2,6.0,13.2 \mathrm{~Hz}), 2.70(1 \mathrm{H}, \mathrm{br}), 2.34(1 \mathrm{H}, \mathrm{ddd}, J=6.0,7.2,13.2 \mathrm{~Hz}), 1.85$ $(3 \mathrm{H}, \mathrm{s}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 150 \mathrm{MHz}\right): \delta(\mathrm{ppm}) 160.7,158.8,144.7,144.5,135.7,132.8,131.3,130.6$, $130.2,128.7,128.3,128.1,128.0,127.3,127.2,126.9,125.9,117.5,115.8,115.3,113.4,97.0,92.1$, 87.0, 86.6, 86.4, 72.3, 63.5, 55.4, 42.5, 17.3. ESI-HRMS ( $\mathrm{m} / \mathrm{z}$ ): $[\mathrm{M}+\mathrm{Na}]^{+}$calcd for $\mathrm{C}_{48} \mathrm{H}_{41} \mathrm{NNaO}_{6}{ }^{+}$, 750.2826 , found 750.2822 .

## Synthesis of compound 7-Ph



Compound 6a ( $24.9 \mathrm{mg}, 0.040 \mathrm{mmol}$ ) was co-evaporated with acetonitrile and dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.5 \mathrm{~mL})$. To the solution were added DIPEA ( $42 \mu \mathrm{~L}$, 0.24 mmol ) and 2-cyanoethyl $\mathrm{N}, \mathrm{N}$-diisopropylchlorophosphoramidite ( $23 \mu \mathrm{~L}$, $0.10 \mathrm{mmol})$, and the mixture was stirred at room temperature for 1 h . The reaction mixture was then quenched with $\mathrm{MeOH}(50 \mu \mathrm{~L})$ and saturated aqueous $\mathrm{NaHCO}_{3}(11 \mathrm{~mL})$, and extracted three times with EtOAc ( 15 mL ). The combined organic layers were washed with brine ( 10 mL ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated under reduced pressure. The residue was purified by silica gel column chromatography (Hexane/Ethylacetate $=3: 1$ ) to give $\mathbf{7 ( P h})$ as a white solid ( $18.5 \mathrm{mg}, 56 \%$ ).
${ }^{31} \mathrm{P}$ NMR $\left(\mathrm{CDCl}_{3}, 162 \mathrm{MHz}\right): \delta(\mathrm{ppm})$ 149.2, 148.4. ESI-HRMS $(\mathrm{m} / \mathrm{z}):[\mathrm{M}+\mathrm{Na}]^{+}$calcd for $\mathrm{C}_{49} \mathrm{H}_{54} \mathrm{~N}_{3} \mathrm{NaO}_{7} \mathrm{P}^{+}, 850.3592$, found 850.3590 .

## Synthesis of compound 7-An



Compound 6b ( $70.6 \mathrm{mg}, 0.096 \mathrm{mmol}$ ) was co-evaporated with acetonitrile and toluene, then dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1.2 \mathrm{~mL})$. To the solution were added DIPEA ( $100 \quad \mu \mathrm{~L}, \quad 0.58 \mathrm{mmol})$ and 2-cyanoethyl $N, N-$ diisopropylchlorophosphoramidite ( $54 \mu \mathrm{~L}, 0.24 \mathrm{mmol}$ ), and the mixture was stirred at room temperature for 1 h . The reaction mixture was then quenched with $\mathrm{MeOH}(50 \mu \mathrm{~L})$ and saturated aqueous $\mathrm{NaHCO}_{3}(15 \mathrm{~mL})$, and extracted with EtOAc ( 20 mL ). The combined organic layers were washed with brine ( 15 mL ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated under reduced pressure. The residue was purified by silica gel column chromatography (Hexane/ Ethylacetate $=3: 1$ ) to give $7(\mathbf{A n})$ as a white solid ( $45.1 \mathrm{mg}, 51 \%$ ).
${ }^{31} \mathrm{P}$ NMR ( $\left.\mathrm{CDCl}_{3}, 162 \mathrm{MHz}\right): \delta(\mathrm{ppm})$ 149.1, 148.4. ESI-HRMS $(\mathrm{m} / \mathrm{z}):[\mathrm{M}+\mathrm{Na}]^{+}$calcd for $\mathrm{C}_{57} \mathrm{H}_{58} \mathrm{~N}_{3} \mathrm{NaO}_{7} \mathrm{P}^{+}, 928.4085$, found 928.4065 .

## Synthesis of compound 8-Ph

To a solution of $\mathbf{6 a}(21 \mathrm{mg}, 0.033 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(550 \mu \mathrm{~L})$, was added TFA $(5.0 \mu \mathrm{~L}, 0.066 \mathrm{mmol})$ at $0^{\circ} \mathrm{C}$ and the mixture was stirred. After 20 min , the reaction mixture was quenched with saturated aqueous $\mathrm{NaHCO}_{3}(15 \mathrm{~mL})$ and diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(100 \mathrm{~mL})$. The organic phase was washed with saturated aqueous $\mathrm{NaHCO}_{3}$ (30 mL×2), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated under reduced pressure. The residue was purified by silica gel column chromatography (Hexane/ Ethylacetate $=1: 2 \rightarrow \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}=$ $9: 1)$ to give $\mathbf{8 ( \mathbf { P h }})$ as a white solid ( $8.2 \mathrm{mg}, 77 \%$ ).
${ }^{1} \mathrm{H}$ NMR (MeOD- $\left.d_{4}, 400 \mathrm{MHz}\right): \delta(\mathrm{ppm}) 8.00(1 \mathrm{H}, \mathrm{s}), 7.67(1 \mathrm{H}, \mathrm{d}, J=2.4 \mathrm{~Hz}), 7.57-7.55(2 \mathrm{H}, \mathrm{m})$, $7.37-7.35(3 \mathrm{H}, \mathrm{m}), 6.46(1 \mathrm{H}, \mathrm{t}, J=6.4 \mathrm{~Hz}), 4.39(1 \mathrm{H}, \mathrm{ddd}, J=6.0,4.0,4.0 \mathrm{~Hz}), 4.01(1 \mathrm{H}, \mathrm{dt}, J=7.6$, $4.0 \mathrm{~Hz}), 3.86(1 \mathrm{H}, \mathrm{dd}, J=12.0,4.0 \mathrm{~Hz}), 3.76(1 \mathrm{H}, \mathrm{dd}, J=12.0,4.0), 2.52(1 \mathrm{H}, \mathrm{ddd}, J=13.2,6.4,4.0$ $\mathrm{Hz}), 2.15(3 \mathrm{H}, \mathrm{s}), 2.13(1 \mathrm{H}$, ddd, $J=13.2,6.8,6.4 \mathrm{~Hz}) .{ }^{13} \mathrm{C}$ NMR (MeOD- $\left.d_{4}, 150 \mathrm{MHz}\right): \delta(\mathrm{ppm})$ 162.3, 146.8, 132.7, 132.6, 129.7, 129.5, 124.4, 117.6, 115.6, 95.4, 89.3, 87.8, 85.6, 71.7, 62.5, 42.7, 17.2. ESI-HRMS $(\mathrm{m} / \mathrm{z}):[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{19} \mathrm{H}_{20} \mathrm{NO}_{4}{ }^{+}, 326.1387$, found 326.1389.

Synthesis of compound 8-An


To a solution of $\mathbf{6 b}(21 \mathrm{mg}, 0.028 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(500 \mu \mathrm{~L})$, was added TFA $(4.3 \mu \mathrm{~L}, 0.056 \mathrm{mmol})$ at $0{ }^{\circ} \mathrm{C}$ and the mixture was stirred. After 1 h , the reaction mixture was quenched with saturated aqueous $\mathrm{NaHCO}_{3}(15 \mathrm{~mL})$ and diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(100 \mathrm{~mL})$. The organic phase was washed with saturated aqueous $\mathrm{NaHCO}_{3}(30 \mathrm{~mL} \times 2)$, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated under reduced pressure. The residue was purified by silica gel column chromatography (Hexane/ Ethylacetate $=1: 2 \rightarrow \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}=9: 1$ ) to give $\mathbf{8 ( A n )}$ as a white solid ( $7.2 \mathrm{mg}, 60 \%$ ).
${ }^{1} \mathrm{H}$ NMR (MeOD- $\left.d_{4}, 600 \mathrm{MHz}\right): \delta(\mathrm{ppm}) 8.73(2 \mathrm{H}, \mathrm{dd}, J=8.4,1.2 \mathrm{~Hz}), 8.68(1 \mathrm{H}, \mathrm{s}), 8.16(2 \mathrm{H}, \mathrm{d}, J$ $=8.4 \mathrm{~Hz}), 7.98(1 \mathrm{H}, \mathrm{d}, J=2.4 \mathrm{~Hz}), 7.96(1 \mathrm{H}, \mathrm{dd}, J=1.2,2.4 \mathrm{~Hz}), 7.71-7.68(2 \mathrm{H}, \mathrm{m}), 7.63-7.60(2 \mathrm{H}$, m), $6.47(1 \mathrm{H}, \mathrm{t}, J=6.6 \mathrm{~Hz}), 4.30(1 \mathrm{H}, \mathrm{m}), 3.91(1 \mathrm{H}, \mathrm{dt}, J=7.8,3.6 \mathrm{~Hz}), 3.70(1 \mathrm{H}, \mathrm{dd}, J=3.6,12.0$ $\mathrm{Hz}), 3.63(1 \mathrm{H}, \mathrm{dd}, J=4.2,12.0 \mathrm{~Hz}), 2.37(1 \mathrm{H}, \mathrm{ddd}, J=13.2,6.6,3.6 \mathrm{~Hz}), 2.15(3 \mathrm{H}, \mathrm{s}), 2.07(1 \mathrm{H}, J=$ 13.2, 6.6, 6.0 Hz ). ${ }^{13} \mathrm{C}$ NMR (MeOD- $d_{4}, 150 \mathrm{MHz}$,): $\delta(\mathrm{ppm}) 162.3,146.3,133.8,133.1,132.8$, 132.7, 130.0, 129.8, 129.2, 128.0, 126.9, 118.0, 117.7, 116.1, 97.7, 92.5, 89.3, 87.9, 71.6, 62.5, 42.8, 17.3. ESI-HRMS $(\mathrm{m} / \mathrm{z}):[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{27} \mathrm{H}_{24} \mathrm{NO}_{4}{ }^{+}, 426.1700$, found 426.1695 .

## Preparation of DNA oligonucleotides

DNA oligonucleotides were synthesized following standard protocols on a $1.0 \mu \mathrm{~mol}$ scale on an Applied Biosystems model 392 DNA/RNA synthesizer. Deprotection and cleavage from the CPG support were carried out under a mild condition with $28 \%$ ammonia solution for 4 h at room temperature. Then, the oligonucleotides were purified by HPLC (JASCO HPLC System: PU2089Plus, UV-2075Plus, FP-2015Plus and CO-2065Plus). HPLC purification conditions: C-18 column (Nacalai tesque: COSMOSIL $5 \mathrm{C}_{18}$-MS-II, $10 \times 250 \mathrm{~mm}$ ) by a linear gradient of $10 \%$ $40 \% / 20 \mathrm{~min}$ acetonitrile in $\mathrm{H}_{2} \mathrm{O}$ at a flow rate of $4 \mathrm{~mL} / \mathrm{min}$ at $40^{\circ} \mathrm{C}$. Peaks were monitored by UV detector $(\lambda=254 \mathrm{~nm})$. The concentration of the ODNs was determined by UV absorption at 260 nm .

## Fluorescence measurement with 2-aminopurine-containing ODN

A mixture $(50 \mu \mathrm{~L})$ of the duplex $(5.0 \mu \mathrm{M})$ in MES buffer ( $50 \mathrm{mM}, \mathrm{pH} 7.0$ ) containing $\mathrm{NaCl}(100$ mM ) was transferred to a quartz cell with a $3-\mathrm{mm}$ path length. The emission spectra were obtained with an excitation wavelength at 310 nm at $25^{\circ} \mathrm{C}$. The fluorescence measurement was performed by a FP-6500 (JASCO Corporation) with a temperature controller.

## Melting temperature ( $T_{\mathrm{m}}$ ) measurement (UV)

A mixture ( $325 \mu \mathrm{~L}$ ) of the duplex $(4.0 \mu \mathrm{M})$ in MES buffer ( $50 \mathrm{mM}, \mathrm{pH} 7.0$ ) containing $\mathrm{NaCl}(100$ mM ) was transferred to a microquartz cell with a $1-\mathrm{cm}$ path length. The melting temperature was then measured under UV absorption at 260 nm from 15 to $90{ }^{\circ} \mathrm{C}$ at the rate of $1{ }^{\circ} \mathrm{C} / \mathrm{min}$. The measurements were carried out three times per each sample and averaged for obtaining the final value. The melting temperature measurement was performed by a DU-800 (Beckman-coulter) with a temperature controller.

Melting temperature ( $\boldsymbol{T}_{\mathrm{m}}$ ) measurement (CD)

A mixture $(100 \mu \mathrm{~L})$ of the duplex $(4.0 \mu \mathrm{M})$ in phosphate buffer $(20 \mathrm{mM}, \mathrm{pH} 7.0)$ containing NaCl $(20 \mathrm{mM})$ was transferred to a micro quartz cell with a $1-\mathrm{cm}$ path length. For the melting temperature measurement, the ellipticity at 285 nm was recorded from $20^{\circ} \mathrm{C}$ to $85^{\circ} \mathrm{C}$ at an interval of $1^{\circ} \mathrm{C}$, with temperature increase at a rate of $1^{\circ} \mathrm{C} / \mathrm{min}$. The measurements were repeated three times. CD spectra were recorded on a J-720WI (JASCO Co., Hachioji, Japan) equipped with a Peltier temperature controller.

## Photo-crosslinking reaction (PAGE analysis)

A mixture $(10 \mu \mathrm{~L})$ of the duplex $(1.0 \mu \mathrm{M})$ in MES buffer ( $50 \mathrm{mM}, \mathrm{pH} 7.0$ ) containing $\mathrm{NaCl}(100$ mM ) in a PCR tube (BIO-BIK) was exposed to 360 nm light ( 300 W Xe lamp, $4.2 \mathrm{~mW} / \mathrm{cm}^{2}$, SM25 Xe , Bunkoukeiki Co.) at room temperature. The distance between the lamp and sample was 5 cm . Aliquots of the reaction mixture was collected at various point of time and quenched by addition of loading buffer ( $95 \%$ formamide, 50 mM EDTA pH 8.0). PAGE was performed on a $20 \%$ polyacrylamide gel electrophoresis with 1 X TBE and 7.5 M urea at 300 V for 40 min . The gel was stained by a SYBR gold and ODNs were visualized with FLA-5100 (Fujifilm Co., Tokyo, Japan).

## Photo-crosslinking reaction (HPLC analysis)

A mixture $(100 \mu \mathrm{~L})$ of the duplex $(5.0 \mu \mathrm{M})$ in MES buffer $(50 \mathrm{mM}, \mathrm{pH} 7.0)$ containing $\mathrm{NaCl}(100$ mM ) in a PCR tube (BIO-BIK) was exposed to 360 nm or 440 nm light ( 300 W Xe lamp, 4.2 or 8.4 $\mathrm{mW} / \mathrm{cm}^{2}, \mathrm{SM}-25 \mathrm{Xe}$, Bunkoukeiki Co.) at room temperature. The distance between the lamp and sample was 5 cm . After 1 or 10 min , the reaction mixture was analyzed by HPLC. HPLC analysis conditions: C-18 column (CAPCELL PAK C ${ }_{18}$ MGII, Shiseido, $4.6 \times 250 \mathrm{~mm}$ ) by a linear gradient of $5 \%-30 \% / 20 \mathrm{~min}$ and $30 \%-100 \% / 25 \mathrm{~min}$ acetonitrile in $\mathrm{H}_{2} \mathrm{O}$ at a flow rate of $1 \mathrm{~mL} / \mathrm{min}$ at $40{ }^{\circ} \mathrm{C}$. Peaks were monitored by UV detector $(\lambda=254 \mathrm{~nm})$.

Table S1. MALDI-TOF mass data of ODNs synthesized in this study

| Entry | X or Z | aCalcd. | bFound |
| :--- | :---: | :---: | :---: |
| ODN1 | $\mathrm{X}=\mathrm{Ph}$ | 3109.9 | 3110.1 |
| ODN1 | $\mathrm{X}=\mathrm{An}$ | 3209.9 | 3212.0 |
| ODN3 | $\mathrm{Z}=\mathrm{Ph}$ | 3084.6 | 3081.2 |
| ODN3 | $\mathrm{Z}=\mathrm{An}$ | 3184.6 | 3185.2 |

${ }^{\mathrm{a}}[\mathrm{M}-\mathrm{H}]^{-},{ }^{\mathrm{b}}$ All data were collected in negative mode.


Scheme S1. Synthesis of diol compounds (8).


Figure S1. Calculation of the molar extinction coefficients at $260 \mathrm{~nm}\left(\varepsilon_{260}\right)$ of Ph and An. The UV spectra and titration graphs of (A) 8-Ph and (B) 8-An. The DMSO solution $(200 \mu \mathrm{M})$ of the nucleoside was titrated to $\mathrm{H}_{2} \mathrm{O}$ ( 325 $\mu \mathrm{L})$. The molar extinction coefficients were calculated by the average of three individual titrations.

ODN1: $5^{\prime}-\mathrm{d}(G C G C P h G C C A G)-3^{\prime}$
ODN2: $3^{\prime}-\mathrm{d}($ CGCG N CGGTC)-5'

$$
\mathbf{N}=\mathrm{A}, \mathrm{G}, \mathrm{C}, \mathrm{~T}
$$



Figure S2. $T_{\mathrm{m}}$ measurement of Ph . (A) UV-melting curves of ODN1 $(\mathrm{X}=\mathrm{Ph})$-ODN2 duplex. (B) $T_{\mathrm{m}}$ values of ODN1 $(\mathrm{X}=\mathrm{dT}$ or Ph$)$-ODN2 duplex. The $T_{\mathrm{m}}$ values were measured using duplex $(4.0 \mu \mathrm{M})$ in MES buffer ( 50 $\mathrm{mM}, \mathrm{pH} 7.0)$ containing $\mathrm{NaCl}(100 \mathrm{mM})$.

ODN1: $5^{\prime}-\mathrm{d}(\mathrm{GCGCAnGCCAG})-3^{\prime}$
ODN2: $3^{\prime}-\mathrm{d}($ CGCG N CGGTC)-5'

$$
\mathrm{N}=\mathrm{A}, \mathrm{G}, \mathrm{C}, \mathrm{~T}
$$



Figure S3. UV-melting curves of ODN1 $(\mathrm{X}=\mathrm{An})$-ODN2 (solid line) or ODN1 $(\mathrm{X}=\mathrm{dT})$-ODN2 (dotted line) duplex. The $T_{\mathrm{m}}$ measurements were performed using duplex ( $4.0 \mu \mathrm{M}$ ) in MES buffer ( $50 \mathrm{mM}, \mathrm{pH} 7.0$ ) containing $\mathrm{NaCl}(100 \mathrm{mM})$.

ODN1: 5'-d(GCGCAnGCCAG)-3'
ODN2: $3^{\prime}-\mathrm{d}(\mathrm{CGCG}$ N CGGTC)-5'

$$
\mathrm{N}=\mathrm{A}, \mathrm{G}, \mathrm{C}, \mathrm{~T}
$$

A

B An-dA

C

D $\boldsymbol{T}_{\mathrm{m}}$ Value (CD)

| $\mathrm{An}-\mathrm{dA}$ | $52^{\circ} \mathrm{C}$ |
| :---: | :---: |
| $\mathrm{An}-\mathrm{dG}$ | $53^{\circ} \mathrm{C}$ |
| $\mathrm{An}-\mathrm{dC}$ | $47^{\circ} \mathrm{C}$ |
| $\mathrm{An}-\mathrm{dT}$ | $49^{\circ} \mathrm{C}$ |
| ${ }^{* d T}-\mathrm{dA}$ | $49^{\circ} \mathrm{C}$ |

Figure S4. $T_{\mathrm{m}}$ measurement of An by CD. (A) CD spectra of ODN1(X = An or dT)-ODN2 duplex. (B) CD spectra change of ODN1(An)-ODN2(dA) duplex. The CD spectra were recorded from $20^{\circ} \mathrm{C}$ to $80^{\circ} \mathrm{C}$ at an interval of $4^{\circ} \mathrm{C}$. (C) CD-melting curve of ODN1(An)-ODN2(dA) duplex was recorded at 285 nm from $20^{\circ} \mathrm{C}$ to $85^{\circ} \mathrm{C}$ at an interval of $1^{\circ} \mathrm{C}$, with temperature increase at a rate of $1^{\circ} \mathrm{C} / \mathrm{min}$. The $T_{\mathrm{m}}$ measurements were performed using duplex $(4.0 \mu \mathrm{M})$ in phosphate buffer ( $20 \mathrm{mM}, \mathrm{pH} 7.0$ ) containing $\mathrm{NaCl}(20 \mathrm{mM})$. (D) The $T_{\mathrm{m}}$ values of ODN1(An)-ODN2 or ODN1(dT)-ODN2(dA) duplex. *The $T_{\mathrm{m}}$ measurement of ODN1(dT)-ODN2(dA) duplex was recorded at 270 nm .

## ODN1: $5^{\prime}-\mathrm{d}(G C G C X G C C A G)-3^{\prime} \quad \mathbf{X}=\mathrm{Ph}$ or An ODN3: $3^{\prime}-\mathrm{d}(C G C Z A C G G T C)-5^{\prime} \quad \mathbf{Z}=\mathrm{Ph}$ or An



Figure S5. HPLC profiles of photo-crosslinking reaction after 1-min (top) and 10-min (bottom) photoirradiation ( 360 nm ). (A) ODN1(An)-ODN3(An), (B) ODN1(Ph)-ODN3(Ph).

Table S2. MALDI-TOF mass data of photo-reaction products in this study

| Entry | ODN | X or Z | aCalcd. | bFound |
| :---: | :---: | :---: | :---: | :---: |
| 1 | ODN1-ODN3 | $\mathrm{X}=\mathrm{An}, \mathrm{Y}=\mathrm{An}$ | 6427.5 | 6428.4 |
| 2 | ODN1-ODN3 | $\mathrm{X}=\mathrm{An}, \mathrm{Y}=\mathrm{Ph}$ | 6327.5 | 6327.6 |
| 3 | ODN1-ODN3 | $\mathrm{X}=\mathrm{Ph}, \mathrm{Y}=\mathrm{An}$ | 6327.2 | 6327.2 |
| 4 | ODN1-ODN3 | $\mathrm{X}=\mathrm{Ph}, \mathrm{Y}=\mathrm{Ph}$ | 6227.2 | 6227.4 |
| 5 | ODN1-ODN3 | $\mathrm{X}=\mathrm{Ph}, \mathrm{Y}=\mathrm{Ph}\left(\mathrm{non}-\mathrm{O}_{2}\right.$ ) | 6194.2 | 6194.9 |
| 6 | ODN1 | $\mathrm{X}=$ abasic sugar | 2917.8 | 2918.0 |
| 7 | ODN3 | $\mathrm{Z}=$ abasic sugar | 2893.5 | 2893.5 |
| 8 | ODN1 | $\mathrm{X}=\mathrm{Ph}($ dione $)$ | 3141.9 | 3143.7 |
| 9 | ODN3 | $\mathrm{Z}=\mathrm{Ph}($ dione $)$ | 3116.6 | 3116.4 |
| 10 | ODN3 | $\mathrm{Z}=\mathrm{An}($ dione $)$ | 3216.6 | 3216.8 |

[^0]

Figure S6. MALDI-TOF mass spectrum of the ODN1(Ph)-ODN3(Ph) crosslinked product.


Figure S7. The proposed structure and mechanism of the non- $\mathrm{O}_{2}$ addition crosslinked product.
A: ODN1(An)
B: ODN3(An)
C: ODN1(Ph)
D: ODN3(Ph)








Figure S8. HPLC profiles of the single strand ODN1 or ODN3 before (top) and after (bottom) photoirradiation $(360 \mathrm{~nm}, 10 \mathrm{~min})$. The photo irradiation was carried out with ODN $(1.0 \mu \mathrm{M})$ in MES buffer $(50 \mathrm{mM}, \mathrm{pH} 7.0)$ containing $\mathrm{NaCl}(100 \mathrm{mM})$. (A) ODN1(An), (B) ODN3(An), (C) ODN1(Ph), (D) ODN3(Ph).


Figure S9. The proposed mechanism of the abasic sugar production.


Figure S10. The proposed mechanism of 1,2-dione by-product production.


Figure S11. Photo-crosslinking reaction at 440 nm . The reaction was carried out with $\operatorname{ODN1}(\mathbf{X}=\mathbf{P h}$ or $\mathbf{A n})$ $(1.0 \mu \mathbf{M})$ and $\operatorname{ODN3}(\mathbf{Z}=\mathbf{P h}$ or $\mathbf{A n})(1.0 \mu \mathrm{M})$ in MES buffer $(50 \mathrm{mM}, \mathrm{pH} 7.0)$ containing $\mathrm{NaCl}(100 \mathrm{mM})$ under 440 nm photoirradiation ( 300 W Xenon lamp, $8.4 \mathrm{~mW} / \mathrm{cm}^{2}$ ) for 1 min .


Figure S12. UV-melting curves of ODN1-ODN3 duplex before (dashed line) and after (solid line) photoirradiation [ $365 \mathrm{~nm}, 6 \mathrm{~W}-$ SLUV-6 (AS ONE), 10 min ]. The $T_{\mathrm{m}}$ measurements were performed using duplex $(4.0 \mu \mathrm{M})$ in MES buffer ( 50 mM , pH 7.0 ) containing $\mathrm{NaCl}(100 \mathrm{mM})$. (A) ODN1(An)-ODN3(An), (B) ODN1(An)-ODN3(Ph), (C) ODN1(Ph)-ODN3(An), (D) ODN1(Ph)-ODN3(Ph).

## NMR Data

Compound $\mathbf{3}$ ( $\beta$-anomer)
${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 600 \mathrm{MHz}\right)$


NOESY-NMR $\left(\mathrm{CDCl}_{3}, 600 \mathrm{MHz}\right)$




2' (a)
${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 150 \mathrm{MHz}\right)$


Compound 3( $\alpha$-anomer)
${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 600 \mathrm{MHz}\right)$


NOESY-NMR ( $\left.\mathrm{CDCl}_{3}, 600 \mathrm{MHz}\right)$

${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 150 \mathrm{MHz}\right)$


Compound 4
${ }^{1} \mathrm{H}$ NMR (MeOD- $\left.d_{4}, 600 \mathrm{MHz}\right)$

${ }^{13} \mathrm{C}$ NMR (MeOD- $d_{4}, 150 \mathrm{MHz}$ )


Compound 5
${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right)$

${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 150 \mathrm{MHz}\right)$


Compound 6(Ph)
${ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{CDCl}_{3}, 600 \mathrm{MHz}\right)$




${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 150 \mathrm{MHz}\right)$


Compound 6(An)
${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 600 \mathrm{MHz}\right)$

${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 150 \mathrm{MHz}\right)$


Compound 7(Ph)
${ }^{31} \mathrm{P}$ NMR ( $\left.\mathrm{CDCl}_{3}, 162 \mathrm{MHz}\right)$

$[\lll 148.36$
$\begin{array}{lllllllllllllll}260 & 240 & 220 & 200 & 180 & 160 & 140 & 120 & 100 & 80 & 60 & 40 & 20 & 0 & p p m\end{array}$

Compound 7(An)
${ }^{31} \mathrm{P}$ NMR ( $\left.\mathrm{CDCl}_{3}, 162 \mathrm{MHz}\right)$


Compound 8(Ph)
${ }^{1} \mathrm{H}$ NMR (MeOD- $d_{4}, 400 \mathrm{MHz}$ )

${ }^{13} \mathrm{C}$ NMR (MeOD- $d_{4}, 150 \mathrm{MHz}$ )



Compound 8(An)
${ }^{1} \mathrm{H}$ NMR (MeOD- $d_{4}, 400 \mathrm{MHz}$ )

${ }^{13} \mathrm{C}$ NMR (MeOD- $d_{4}, 150 \mathrm{MHz}$ )



[^0]:    ${ }^{\mathrm{a}}[\mathrm{M}-\mathrm{H}]^{-},{ }^{\mathrm{b}}$ All data were collected in negative mode.

