

# Supporting Information

## Soft and Hard Interactions between Polystyrene Nanoplastics and Human Serum Albumin Protein Corona

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## Relevant properties

**Table S1.** Physical properties of HSA, PS<sub>s</sub><sup>-</sup>, and PS<sub>L</sub><sup>-</sup>.

Name	Formula	Molecular weight / g mol <sup>-1</sup>	Density / g cm <sup>-3</sup>	Isoelectric point	SLD <sub>n</sub> / 10 <sup>-6</sup> Å <sup>-2</sup>
HSA	C <sub>2916</sub> H <sub>4624</sub> N <sub>760</sub> O <sub>889</sub> S <sub>41</sub>	66500	1.360	5.5 <sup>a</sup>	1.84-3.14 <sup>b,1</sup>
PS <sub>s</sub> <sup>-</sup>	C <sub>8</sub> H <sub>8</sub> (monomer)	104	1.040	<2	1.30
PS <sub>L</sub> <sup>-</sup>	C <sub>8</sub> H <sub>8</sub> (monomer)	104	1.040	<2	1.30
H <sub>2</sub> O	H <sub>2</sub> O	18	0.997	NA	-0.56
D <sub>2</sub> O	D <sub>2</sub> O	20	1.100	NA	6.25

<sup>a</sup> based on a theoretical calculation (SI Figure S9)

<sup>b</sup> The lower limit of the SLD<sub>n</sub> occurs when there is no hydrogen-deuterium (H/D) exchange for exchangeable hydrogens, occurs while the upper limit assumes the 100% H/D exchange.

## Circular dichroism (CD) spectroscopy

**Table S2.** Secondary structure of HSA resolved by spectra deconvolution of CD spectra collected at pH 7.4.

Sample	$\alpha$ -helix	$\beta$ -sheet	Random coil
Native HSA	51% $\pm$ 2%	21% $\pm$ 2%	28% $\pm$ 1%
With PS <sub>s</sub> <sup>-</sup> (0.1 mg mL <sup>-1</sup> )	52% $\pm$ 2%	20% $\pm$ 3%	29% $\pm$ 2%
With PS <sub>L</sub> <sup>-</sup> (0.1 mg mL <sup>-1</sup> )	52% $\pm$ 1%	21% $\pm$ 2%	28% $\pm$ 2%

**Table S3.** Secondary structure of HSA resolved by spectra deconvolution of CD spectra collected at pH 7.4.

Sample	$\alpha$ -helix	$\beta$ -sheet	Random coil
Native HSA	51% $\pm$ 2%	20% $\pm$ 1%	29% $\pm$ 2%
With PS <sub>s</sub> <sup>-</sup> (0.1 mg mL <sup>-1</sup> )	46% $\pm$ 2%	12% $\pm$ 2%	42% $\pm$ 3%
With PS <sub>L</sub> <sup>-</sup> (0.1 mg mL <sup>-1</sup> )	52% $\pm$ 1%	19% $\pm$ 3%	29% $\pm$ 2%

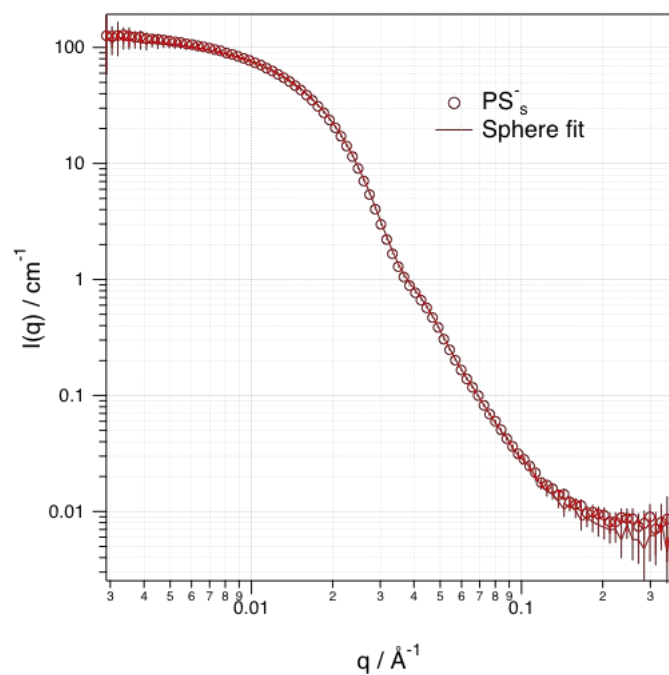
### Zeta potential values

**Table S4.** Zeta potential values of PS particles, HSA, and PS/HSA complexes at pH 7.4 and 5.0.

Sample	pH 7.4	pH 5.0
HSA	-12.2 $\pm$ 3.2 mV	-4.4 $\pm$ 4.5 mV
PS <sub>s</sub> <sup>-</sup>	-50.2 $\pm$ 12.0 mV	-45.3 $\pm$ 6.8 mV
PS <sub>L</sub> <sup>-</sup>	-58.2 $\pm$ 7.0 mV	-46.8 $\pm$ 11.6 mV
HSA/ PS <sub>s</sub> <sup>-</sup>	-15.5 $\pm$ 3.0 mV	-15.8 $\pm$ 9.8 mV
HSA/ PS <sub>L</sub> <sup>-</sup>	-43.4 $\pm$ 7.1 mV	-9.0 $\pm$ 6.8 mV

## Small angle neutron scattering (SANS)

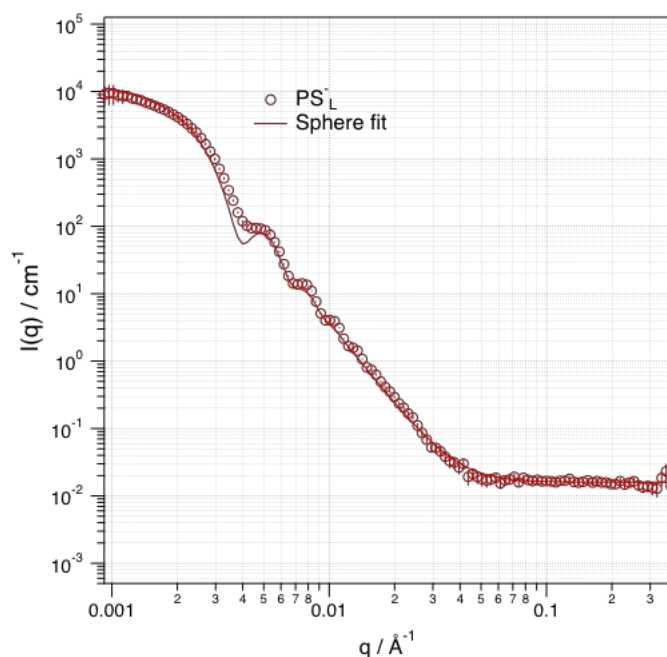
### Polystyrene nanoparticles



**Figure S1.** SANS curve of  $\text{PS}_s$  at  $9.0 \text{ mg mL}^{-1}$  in 100% d-buffer, fitted with sphere model

**Table S5.** Fitting parameters for SANS curve shown in Figure S1

Scale	Background / $\text{cm}^{-1}$	$\text{SLD}_n / 10^{-6} \text{\AA}^{-2}$	$\text{SLD}_n \text{ solvent} / 10^{-6} \text{\AA}^{-2}$	Radius / $\text{\AA}$
$0.00681 \pm 0.000150$	$0.00683 \pm 0.000300$	1.34	6.35	$115 \pm 0.100$

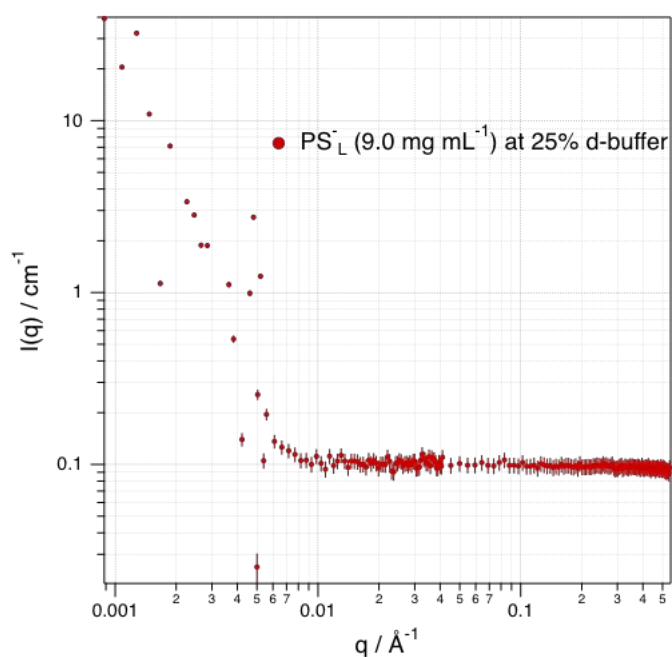


**Figure S2.** SANS curve for  $\text{PS}_L$  at  $9.0 \text{ mg mL}^{-1}$  in 100% d-buffer (pH 7.4), fitted with a spherical model.

**Table S6.** Fitting parameters for SANS curve shown in Figure S2.

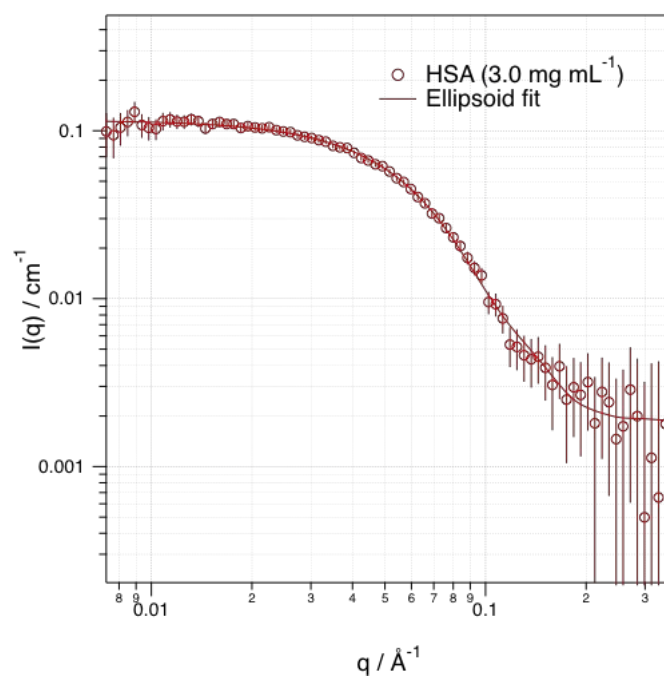
Scale	Background / $\text{cm}^{-1}$	$\text{SLD}_n / 10^{-6} \text{ \AA}^{-2}$	$\text{SLD}_n \text{ solvent} / 10^{-6} \text{ \AA}^{-2}$	Radius / $\text{\AA}$
$0.000820 \pm 0.00000650$	$0.0160 \pm 0.000420$	1.34	6.35	$1187 \pm 1.8^a$

<sup>a</sup> The optimised polydispersity was 2%



**Figure S3.** SANS curve of  $\text{PS}_L$  at  $9.0 \text{ mg mL}^{-1}$  in 25% d-buffer.

## Human serum albumin (HSA)

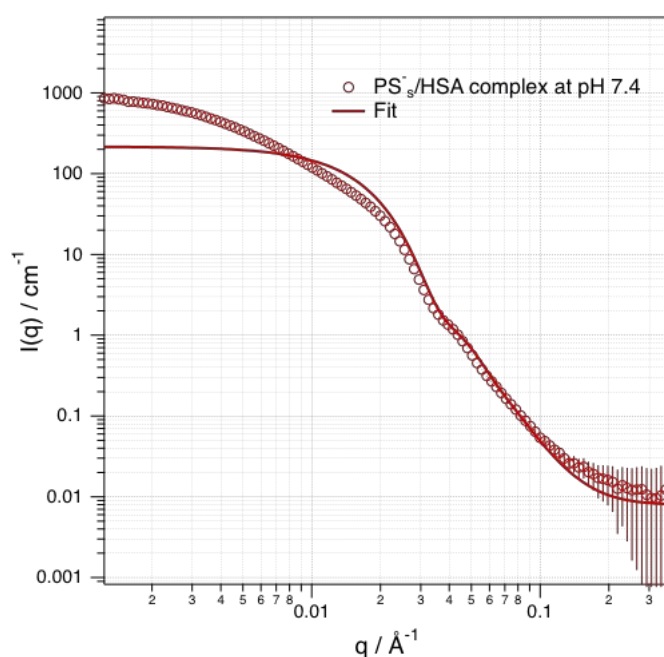


**Figure S4.** SANS curve for HSA at 3.0 mg mL<sup>-1</sup> in 100% d-buffer (pH 7.4), fitted with ellipsoidal model.

**Table S7.** Fitting parameters for SANS curve shown in Figure S4.

SLD <sub>n</sub> / 10 <sup>-6</sup> Å <sup>-2</sup>	SLD <sub>n</sub> solvent / 10 <sup>-6</sup> Å <sup>-2</sup>	Radius polar / Å	Radius equatorial / Å
2.40	6.35	15.6 ± 1.90	43.7 ± 0.700

## PS/HSA complex

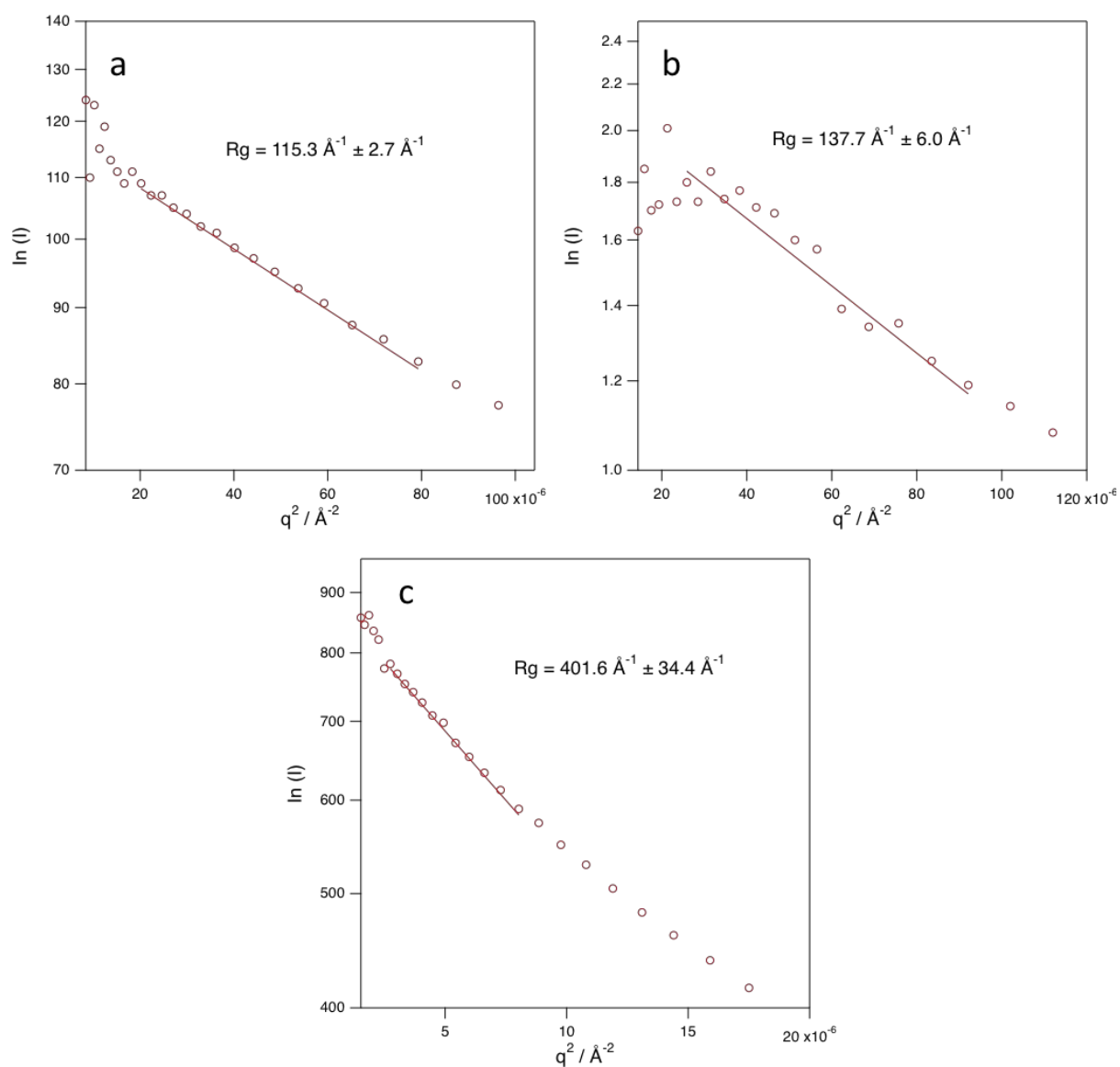


**Figure S5.** SANS curve for PS<sub>s</sub> (9.0 mg mL<sup>-1</sup>)/HSA (3.0 mg mL<sup>-1</sup>) complex in 100% d-buffer (pH 7.4). The scattering intensity was fitted with a combination of spherical and ellipsoidal models.

**Table S8.** Fitting parameters for SANS curve shown in Figure S5. The radius of PS<sub>s</sub>, and dimensions of HSA were fixed with the values that were obtained in single components (Table S5 and Table S7).

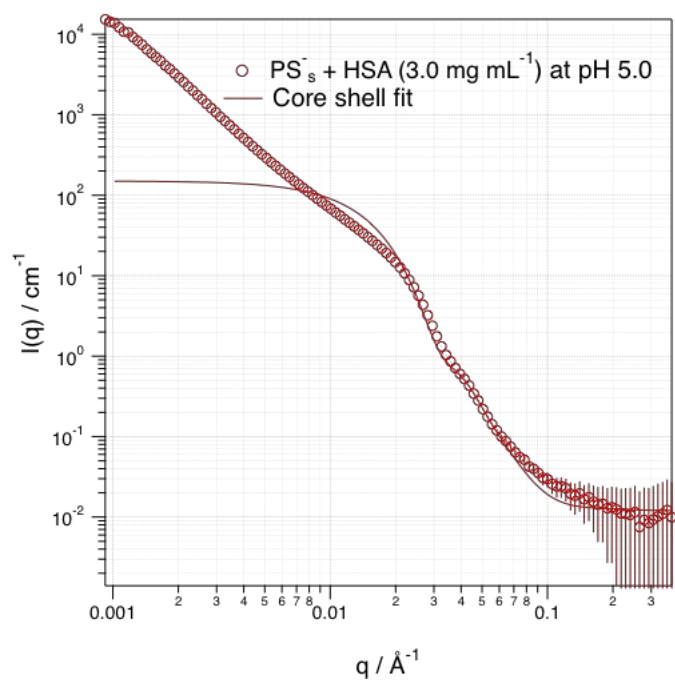
Radius / Å	Radius polar / Å	Radius equatorial / Å	SLD <sub>n</sub> solvent / 10 <sup>-6</sup> Å <sup>-2</sup>	SLD <sub>n</sub> PS / 10 <sup>-6</sup> Å <sup>-2</sup>	SLD <sub>n</sub> HSA / 10 <sup>-6</sup> Å <sup>-2</sup>
135 <sup>a</sup>	15.6	43.7	6.35	1.34	2.40

<sup>a</sup> The optimised polydispersity was 17%



**Figure S6.** Guinier plots for (a) PS<sub>s</sub> (9.0 mg mL<sup>-1</sup>) in 100% d-buffer, (b) PS<sub>s</sub>(9.0 mg mL<sup>-1</sup>)/HSA (3.0 mg mL<sup>-1</sup>) complex in 25% d-buffer, and (c) PS<sub>s</sub>(9.0 mg mL<sup>-1</sup>)/HSA (3.0 mg mL<sup>-1</sup>) complex in 100% d-buffer.





**Figure S7.** SANS curve for PS-s/HSA complex in 100% d-buffer (pH 5.0), fitted with core shell model.

**Table S9.** Fitting parameters for Figure S7

Radius / Å	Thickness / Å	SLD <sub>n</sub> solvent / 10 <sup>-6</sup> Å <sup>-2</sup>	SLD <sub>n</sub> core / 10 <sup>-6</sup> Å <sup>-2</sup>	SLD <sub>n</sub> shell / 10 <sup>-6</sup> Å <sup>-2</sup>
135 <sup>a</sup>	15.6 ± 1.90	6.35	1.34	2.40

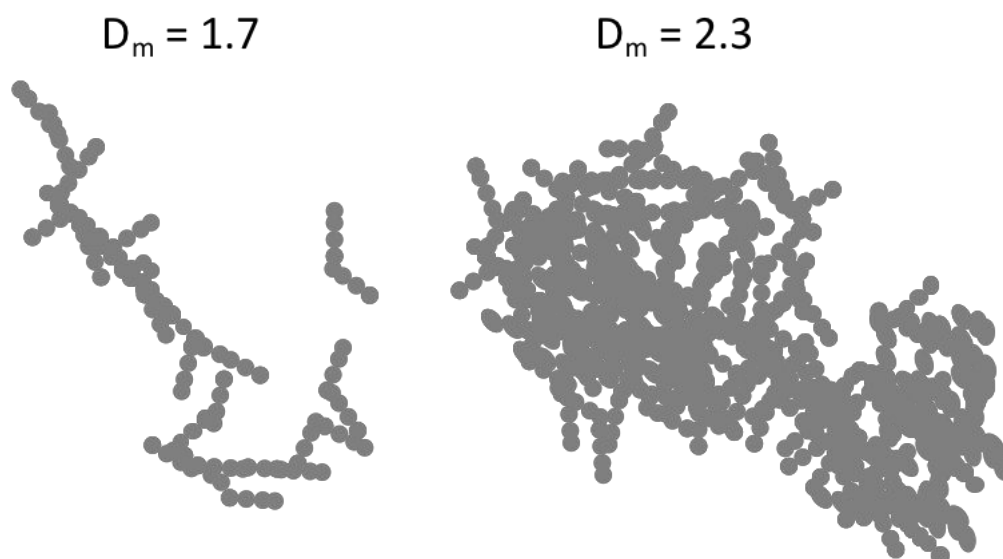
**Table S10.** Fitting parameters for Figure 9, PS<sub>L</sub>/HSA complex at pH 7.4.

Radius / Å	Radius polar / Å	Radius equatorial / Å	SLD <sub>n</sub> solvent / 10 <sup>-6</sup> Å <sup>-2</sup>	SLD <sub>n</sub> PS / 10 <sup>-6</sup> Å <sup>-2</sup>	SLD <sub>n</sub> HSA / 10 <sup>-6</sup> Å <sup>-2</sup>
1187 <sup>a</sup>	15.6	43.7	6.35	1.34	2.40
The optimised polydispersity was fitted to be 2%					

**Table S11.** Fitting parameters for Figure 9, PS<sub>L</sub>/HSA complex at pH 5.0.

Radius / Å	Radius polar / Å	Radius equatorial / Å	SLD <sub>n</sub> solvent / 10 <sup>-6</sup> Å <sup>-2</sup>	SLD <sub>n</sub> PS / 10 <sup>-6</sup> Å <sup>-2</sup>	SLD <sub>n</sub> HSA / 10 <sup>-6</sup> Å <sup>-2</sup>	Perturb	Stickiness
1187 <sup>a</sup>	15.6	43.7	6.35	1.34	2.40	0.0620 ± 0.00085	0.140 ± 0.00015
The optimised polydispersity was fitted to be 2%							

## Representation of fractal dimension in colloidal system



**Figure S8.** Graphical representation of fractals with different fractal dimensions.

## Theoretical calculation of HSA net charge

The net charge of human serum albumin (HSA) was calculated with the following equation:

$$\text{net charge (positive charge)} = \frac{10^{-pK_a}}{10^{-pK_a} + 10^{-pH}} \quad \text{Equation S1}$$

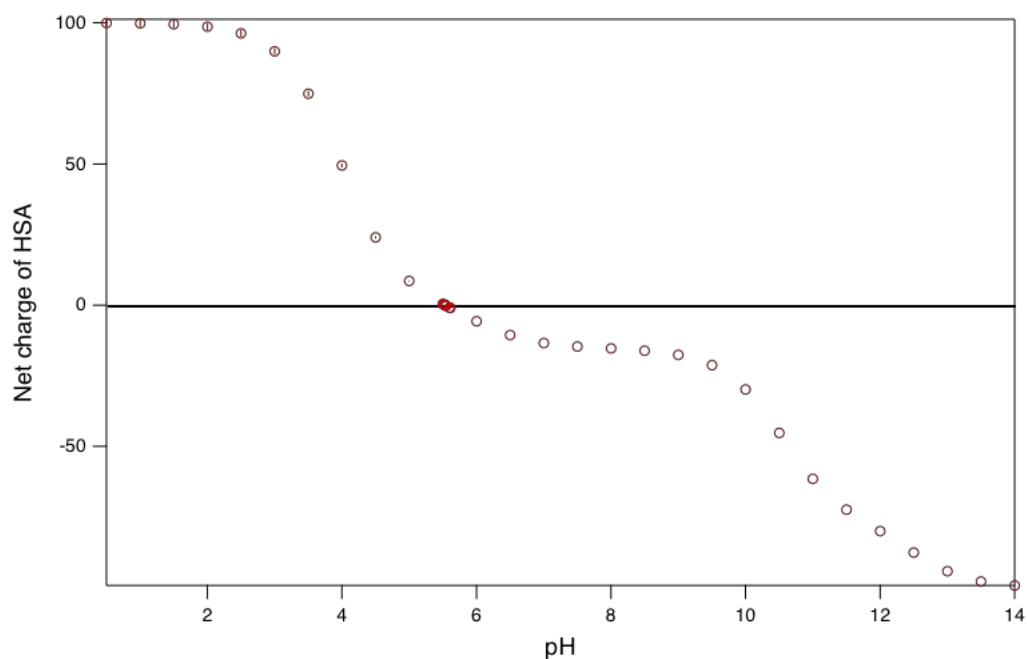
$$\text{net charge (negative charge)} = \frac{10^{-pH}}{10^{-pH} + 10^{-pK_a}} \quad \text{Equation S2}$$

The number and the pKa values of each amino acid used in the calculation is found in Table S11.

**Table S12.** Relevant properties of HSA for calculating net charges at various pH.

Amino acid	pKa of side group <sup>2</sup>	Number <sup>3</sup>
Arginine	12.8	24
Histidine	6.0	16
Lysine	10.1	59
Aspartic acid	3.7	36
Glutamic acid	4.3	62
Cysteine*	8.2	1

\*One cysteine that does not form disulfide bridge is used in this calculation.<sup>3</sup>



**Figure S9.** Theoretically calculated net charge of HSA as a function of pH.

## References

1. Berts, I.; Fragneto, G.; Porcar, L.; Hellsing, M. S.; Rennie, A. R., Controlling adsorption of albumin with hyaluronan on silica surfaces and sulfonated latex particles. *J. Colloid Interface Sci.* **2017**, *504*, 315-324.
2. Lide, D., DR 1991-1992. *Handbook of chemistry and physics. 72nd edition*, CRC Press, Boca Raton, FL, 17.
3. Sugio, S.; Kashima, A.; Mochizuki, S.; Noda, M.; Kobayashi, K., Crystal structure of human serum albumin at 2.5 Å resolution. *Protein Eng.* **1999**, *12* (6), 439-446.