

Supporting information

Valence Electron and Chemical State Analysis of Be₁₂M (M = Ti, V) Beryllides by Soft X-ray Emission Spectroscopy

Keisuke Mukai,¹ Ryuta Kasada,^{2*} Kiyohiro Yabuuchi,¹ Satoshi Konishi,¹ Jae-Hwan Kim,
³ Masaru Nakamichi³

¹ Institute of Advanced Energy, Kyoto University, Gokasho, Uji, Kyoto 611-0011, Japan

² Institute for Material Research, Tohoku University, Sendai 980-8577, Japan

³ Fusion Energy Research and Development Directorate, National Institutes for Quantum and Radiological Science and Technology, 2-166 Obuchi, Omotedate, Rokkasho, Aomori, 039-3212, Japan

* Corresponding author: r-kasada@imr.tohoku.ac.jp

Table S1 Lattice parameters of Be, Be₁₂Ti, Be₁₂V, and BeO obtained by the structural optimization in the DFT calculations compared with the experimental ones.¹⁻⁴ The calculated lattice parameters agreed well with the experimental ones.

	Calculation		Experimental	
	<i>a</i> (Å)	<i>c</i> (Å)	<i>a</i> (Å)	<i>c</i> (Å)
Be	2.257	3.587	2.287 ^a	3.583 ^a
Be ₁₂ Ti	7.430	4.235	7.35 ^b	4.19 ^b
Be ₁₂ V	7.215	4.147	7.278 ^c	4.212 ^c
BeO	2.705	4.395	2.698 ^d	4.378 ^d

^a Reference 1

^b Reference 2

^c Reference 3

^d Reference 4

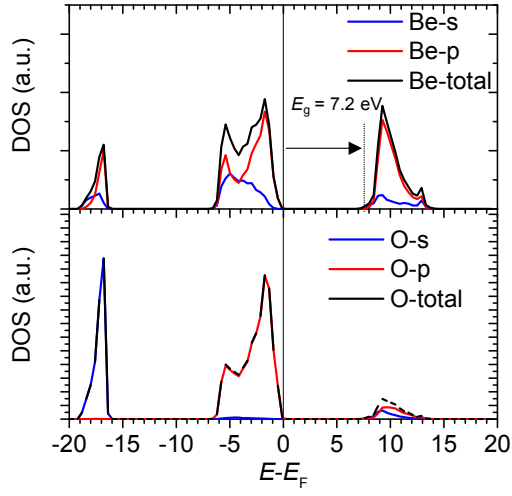


Fig. S1 Partial density of states (DOSs) of BeO plotted with the energy referenced to Fermi energy E_F . Black, blue, red, and green solid lines represent DOSs of total, s , and p states. The band gap energy of BeO was estimated to be 7.2 eV, which were comparable with the previous calculated values of 7.0 and 7.54 eV.^{5,6} The calculated band gap was underestimated from the experimental band gap of BeO (10.6 eV).⁷ The partial DOS Be-2 p convoluted with the Gaussian function is compared with the experimental Be-K α spectrum from the BeO specimen in Fig. 6.

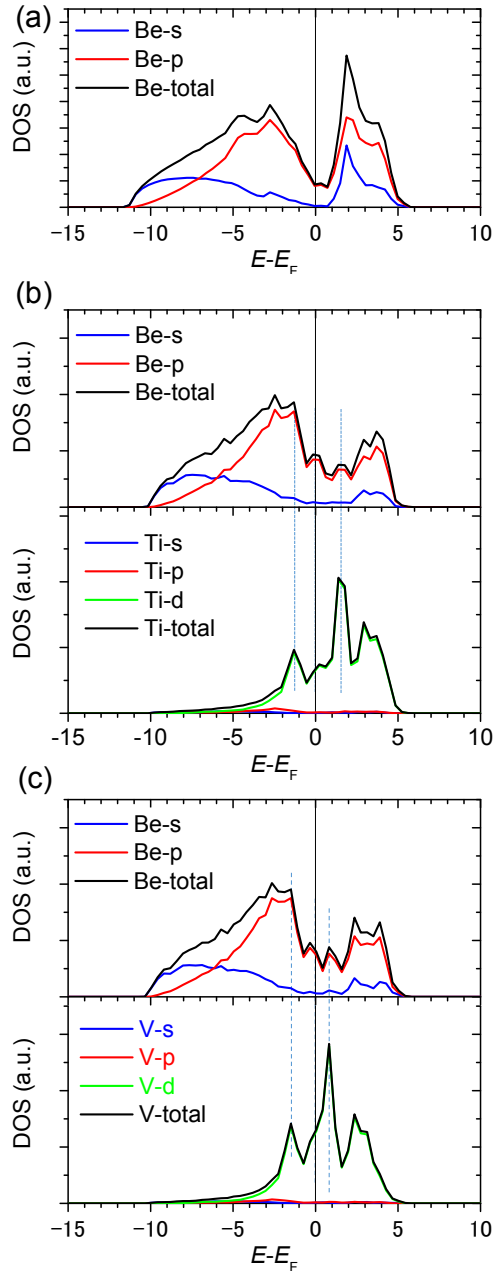


Fig. S2 Density of states in metallic Be (a), Be_{12}Ti (b), and Be_{12}V (c) with the energy referenced to the Fermi levels E_F . Black, blue, red, and green solid lines represent total, s , p , and d states. The DOSs of Be with the energy referenced with the Be-1s centroid are shown in Fig. 3. The valence electron structure of Be in Be_{12}Ti and Be_{12}V were altered from that of Be in metallic Be. The hybridized peaks of O $2p$ and Ti/V $3d$ states shown by blue dashed lines near the Fermi levels ($E - E_F = 0$) were seen in Be_{12}Ti and Be_{12}V .

References

1. Schwarzenberger, D. R. Accurate determination of the lattice parameters of beryllium. *Philos. Mag.* **1959**, *4*, 1242–1246.
2. Gillam, E.; Rooksby, H. P.; Brownlee, L. D. Structural relationships in beryllium–titanium alloys. *Acta Crystallogr.* **1964**, *17*, 762–763.
3. Von Batchelder, F.W.; Raeuchle, R.F. The structure of a new series of MBe_{12} compounds. *Acta Crystallogr.* **1957**, *10*, 648–649.
4. Hazen, R. M.; Finger, L. W. High-pressure and high-temperature crystal chemistry of beryllium oxide. *J. Appl. Phys.* **1986**, *59*, 3728.
5. Chang, K.J.; Froyen, S.; Cohen, M. The electronic band structures for zincblende and wurtzite BeO. *J. Phys. C: Solid State Phys.* **1983**, *16*, 3475.
6. Xu, Y. N.; Ching, W. Y. Electronic, optical, and structural properties of some wurtzite crystals. *Phys. Rev. B*, **1993**, *48*, 4335.
7. Roessler, D. M.; Walker, W. C.; Loh, E. Electronic spectrum of crystalline beryllium oxide. *J. Phys. Chem. Solids* **1969**, *30*, 157–167.