

SUPPORTING INFORMATION

March 7, 2019

Revised M11 Exchange–Correlation Functional for Electronic Excitation Energies and Ground–State Properties

Pragya Verma,^{1,*} Ying Wang,^{2,3} Soumen Ghosh,¹ Xiao He,^{2,4,*} and Donald G. Truhlar^{1,*}

¹*Department of Chemistry, Chemical Theory Center, Nanoporous Materials Genome Center, and Minnesota Supercomputing Institute, University of Minnesota, Minneapolis, Minnesota, 55455-0431, USA*

²*Shanghai Engineering Research Center of Molecular Therapeutics and New Drug Development, School of Chemistry and Molecular Engineering, East China Normal University, Shanghai, 200062, China*

³*The National and Local Joint Engineering Laboratory of Animal Peptide Drug Development, College of Life Sciences, Hunan Normal University, Changsha, 410006, China*

⁴*NYU-ECNU Center for Computational Chemistry at NYU Shanghai, Shanghai, 200062, China*

*E-mail: verma045@umn.edu (P.V.), xiaohe@phy.ecnu.edu.cn (X.H.), and truhlar@umn.edu (D.G.T.)

TABLE OF CONTENTS

Item	Description^{a,b}	Page
Databases used for training		
Table S1	Reference and calculated values of the SR-MGM-BE8 database	S-4
Table S2	Reference and calculated values of the SR-MGN-BE107 database	S-5
Table S3	Reference and calculated values of the SR-TM-BE15 database	S-8
Table S4	Reference and calculated values of the MR-MGM-BE4 database	S-9
Table S5	Reference and calculated values of the MR-MGN-BE17 database	S-10
Figure S1	Structure of S ₄	S-10
Table S6	Reference and calculated values of the MR-TM-BE12 database	S-11
Table S7	Reference and calculated values of the MR-TMD-BE3 database	S-12
Table S8	Reference and calculated values of the HTBH38/18 database	S-13
Table S9	Reference and calculated values of the NHTBH38/18 database	S-14
Table S10	Reference and calculated values of the NCCE30/18 database	S-15
Table S11	Reference and calculated values of the NGD21/18 database	S-16
Table S12	Reference and calculated values of the S6x6 database	S-17
Table S13	Reference and calculated values of the IP23 database	S-18
Table S14	Reference and calculated values of the EA13/03 database	S-19
Table S15	Reference and calculated values of the PA8 database	S-20
Table S16	Reference and calculated values of the 2pIsoE4 database	S-21
Table S17	Reference and calculated values of the 4pIsoE4 database	S-22
Table S18	Reference and calculated values of the IsoL6/11 database	S-23
Table S19	Reference and calculated values of the π TC13 database	S-24
Table S20	Reference and calculated values of the AE17 database	S-25
Table S21	Reference and calculated values of the HC7/11 database	S-26
Table S22	Reference and calculated values of the SMAE3/19 database	S-27
Table S23	Reference and calculated values of the DC9/19 database	S-28
Table S24	Reference and calculated values of the ABDE13 database	S-29
Table S25	Reference and calculated values of the DGL6 database	S-30
Table S26	Reference and calculated values of the DGH4 database	S-31
Table S27	Reference and calculated values of the 3dEE8 database	S-32
Table S28	Reference and calculated values of the 4dAEE5 database	S-33
Table S29	Reference and calculated values of the pAEE5 database	S-34
Table S30	Reference and calculated values of the EE23 database	S-35
Table S31	Reference and calculated values of the LRCTEE9 database	S-36
Databases used for testing		
Table S32	Reference and calculated values of the Al2X6 database	S-37
Table S33	Reference and calculated values of the BHDIV10 database	S-38
Table S34	Reference and calculated values of the BHPERI26 database	S-39
Table S35	Reference and calculated values of the BHROT27 database	S-40
Table S36	Reference and calculated values of the DIPCS10 database	S-41
Table S37	Reference and calculated values of the HeavySB11 database	S-42

Table S38	Reference and calculated values of the PX13 database	S-43
Table S39	Reference and calculated values of the SIE4x4 database	S-44
Table S40	Reference and calculated values of the YBDE18 database	S-45
Table S41	Reference and calculated values of the S66x8 database	S-46
Table S42	Reference and calculated values of the TMBH22 database	S-53
Table S43	Reference and calculated values of the WCCR10/18 database	S-54
Table S44	Reference and calculated values of the ASNC2 database	S-55
Table S45	Reference and calculated values of the DM79 database	S-56
Table S46	Reference and calculated values of the PEC4 database	S-58
Table S47	Reference and calculated values of the TSG48 database	S-61
Table S48	Reference and calculated values of the MGBL193 database	S-63
Table S49	Reference and calculated values of the TMDBL10 database	S-69
Table S50	Reference and calculated values of the NaCl database	S-70
Table S51	Reference and calculated values of the EEA11 database	S-71
Table S52	Reference and calculated values of the AEE15 database	S-72
Table S53	Reference and calculated values of the EEArO-T5 database	S-73
Table S54	Reference and calculated values of the EE69 database	S-74
Table S55	Reference and calculated values of the EER5 database	S-76
Table S56	Reference and calculated values of the LRCTEE2 database	S-77
Table S57	Geometries (\AA), total electronic energies (E in hartrees), charges, and spin multiplicities of selected systems in the SR-MGN-BE107 database	S-78
Reference		S-79

^aThe calculated values are reported using the PBE, B3LYP, M11, and revM11 density functionals.

^bMost of the tables report MUE or MUEPB. MUE denotes mean unsigned error. MUEPB denotes mean unsigned error per bond.

Table S1. Atomization energies (kcal/mol) of the SR-MGM-BE8 database

System	PBE	B3LYP	M11	revM11	Ref.
AlCl ₃	303.9	291.9	308.6	301.9	309.9
AlF ₃	425.8	415.8	438.6	426.8	429.6
KOH	84.3	80.0	92.4	86.0	85.0
NaO	68.5	64.8	70.2	66.1	65.2
LiCl	109.4	109.0	110.8	109.2	113.9
AlCl	119.8	116.2	118.5	116.6	121.6
ZnSe	30.5	22.6	39.8	20.6	25.2
ZnCl	52.5	46.9	64.3	54.7	53.5
MUEPB ^a	2.5	4.4	5.9	2.6	0.0

^aMUEPB denotes mean unsigned error per bond. It was computed as follows. First, each entry is used to compute the average bond energy in the given molecule by dividing the atomization energy by the number of bonds (for this purpose single, double, and triple bonds all count as just one bond). Then we compute the error in the average bond energy of the molecule by subtracting the reference value from the density functional value. Then we calculate the MUEPB by averaging the absolute values of these errors over all the rows of the table. Thus the MUEPB is the MUE of the average bond energies.

Table S2. Atomization energies (kcal/mol) of the SR-MGN-BE107 database

System	PBE	B3LYP	M11	revM11	Ref.
C ₂ H ₆	96.9	91.7	97.8	96.8	97.4
iPr-CH ₃	89.6	85.0	92.6	92.7	95.0
C ₂ H ₆ O	86.8	82.1	90.3	91.8	89.8
iPr-OCH ₃	83.7	79.7	89.3	91.3	91.5
Et-H	104.9	106.4	108.6	108.0	108.9
Et-CH ₃	93.1	88.2	94.9	94.5	95.9
Et-OCH ₃	86.3	82.1	90.6	92.5	95.3
Et-OH	99.6	93.3	99.7	101.3	100.3
tBu-H	97.3	99.6	102.2	102.7	103.9
tBu-CH ₃	86.2	81.6	90.9	91.2	93.7
tBu-OCH ₃	80.1	76.2	87.9	89.7	89.3
tBu-OH	97.0	91.1	99.2	101.2	115.0
CH (² Π)	84.5	85.3	82.9	84.4	84.2
CH ₂ (³ B ₁)	194.4	192.0	187.7	192.4	190.7
CH ₂ (¹ A ₁)	178.6	180.5	179.9	183.3	181.4
CH ₃ (² A ["] ₂)	309.9	309.8	307.4	309.5	307.8
CH ₄	420.1	420.9	420.8	421.5	420.3
NH	88.3	87.9	83.3	88.0	83.1
NH ₂	188.2	187.6	186.2	187.3	182.6
NH ₃	301.7	300.7	303.6	303.0	298.0
OH	109.5	107.9	107.4	107.9	107.2
H ₂ O	233.5	230.3	233.6	233.5	232.8
HF	141.2	138.5	142.1	142.2	141.3
SiH ₂ (¹ A ₁)	147.4	153.3	152.8	153.6	151.8
SiH ₂ (³ B ₁)	131.2	132.7	133.2	134.7	131.1
SiH ₃	222.0	228.1	228.0	231.8	227.6
SiH ₄	313.0	323.3	323.2	329.2	324.5
PH ₂	154.4	158.6	155.4	156.2	153.2
PH ₃	239.1	244.7	242.3	243.0	242.3
H ₂ S	181.5	181.5	183.3	180.9	183.4
HCl	105.4	104.1	106.3	104.6	106.7
C ₂ H ₂	414.7	403.0	402.1	403.8	405.4
CH ₂ CH ₂	571.4	563.0	562.1	563.7	563.5
CH ₃ CH ₃	716.8	711.4	712.6	715.7	712.8
HCN	326.0	313.1	313.8	315.5	313.3
HCO	294.7	279.9	278.7	282.9	279.1
H ₂ CO	385.6	373.2	373.7	377.4	374.4
CH ₃ OH	519.5	511.1	514.6	518.2	513.2
NH ₂ NH ₂	452.6	443.6	450.0	452.8	438.6
HOOH	281.0	266.1	269.2	273.1	268.6
Si ₂ (multiplicity = 3)	79.3	70.5	68.2	67.7	75.7
P ₂	121.5	115.8	110.8	105.7	117.6
S ₂	113.6	101.8	103.0	99.5	103.1
Cl ₂	63.2	53.1	57.6	56.3	58.1

C ₃ H ₄ (cyclopropene)	701.4	678.3	684.0	688.6	682.7
CH ₃ COOH	827.6	797.8	803.9	811.6	803.0
CH ₃ COCH ₃	1000.0	974.3	978.1	984.9	978.0
C ₃ H ₆ (cyclopropane)	867.9	849.1	855.9	862.4	853.4
CH ₃ CHCH ₂	873.5	858.3	859.1	862.6	860.6
C ₃ H ₈	1014.9	1003.1	1006.4	1011.8	1006.9
C ₂ H ₅ OCH ₃	1111.1	1090.6	1096.6	1106.2	1095.1
C ₄ H ₁₀ (isobutane)	1313.8	1295.2	1301.5	1309.1	1303.0
C ₄ H ₁₀ (anti-periplanar butane)	1313.0	1294.7	1300.2	1307.9	1301.3
C ₄ H ₈ (cyclobutane)	1167.1	1141.2	1147.6	1157.1	1149.0
C ₄ H ₈ (isobutene)	1175.3	1153.3	1156.4	1162.0	1158.6
C ₅ H ₈ (spiropentane)	1315.2	1275.9	1289.2	1301.3	1284.3
C ₆ H ₆	1409.2	1360.8	1362.4	1367.4	1367.6
CH ₃ CO	603.4	582.0	582.5	588.2	581.6
(CH ₃) ₂ CH	914.2	900.4	901.5	906.8	900.8
(CH ₃) ₃ C	1216.4	1195.6	1199.3	1206.4	1199.3
H ₂ CCO	557.9	533.5	534.2	537.6	532.3
MUEPB ^a	3.4	2.2	0.9	1.5	0.0

^aMUEPB denotes mean unsigned error per bond. It is the MUE of the average bond energies as explained in the footnote to Table S1.

Table S3. Atomization energies (kcal/mol) of the SR-TM-BE15 database

System	PBE	B3LYP	M11	revM11	Ref.
CrCl ₂	176.5	169.5	175.1	172.0	181.1
MnF ₂	254.4	233.6	261.0	243.9	232.3
FeCl ₂	197.4	184.9	204.3	192.8	190.3
CoCl ₂	186.4	170.4	189.8	180.2	182.9
Ag ₂	40.7	35.4	30.2	30.9	38.3
AgH	55.9	54.4	57.2	54.6	54.0
CrCH ₃ ⁺	43.9	35.0	30.5	31.3	28.8
Cu ₂	48.0	41.0	33.5	35.8	47.2
CuAg	44.3	38.2	32.6	31.1	40.7
CuH ₂ O ⁺	43.1	38.8	33.0	39.5	38.8
VCO ⁺	38.2	29.6	41.6	27.9	28.2
Zr ₂	85.0	56.6	60.3	40.5	70.8
Pd(PH ₃) ₂ -C ₆ H ₈	13.3	3.8	14.5	15.9	16.2
Pd(PH ₃) ₂ -C ₁₀ H ₁₂ -b	9.1	-1.5	14.9	16.1	17.3
FeCl	84.8	78.9	95.3	85.4	80.5
MUEPB ^a	5.4	4.4	7.3	5.4	0.0

^aMUEPB denotes mean unsigned error per bond. It is the MUE of the average bond energies as explained in the footnote to Table S1.

Table S4. Atomization energies (kcal/mol) of the MR-MGM-BE4 database

System	PBE	B3LYP	M11	revM11	Ref.
CaO	124.2	104.5	110.6	101.6	96.2
LiO ⁻	61.8	53.4	48.8	58.3	57.6
KO ⁻	37.7	23.4	17.5	41.9	33.1
MgS	55.3	46.9	49.9	44.0	55.7
MUEPB ^a	9.3	7.8	11.2	6.7	0.0

^aMUEPB denotes mean unsigned error per bond. It is the MUE of the average bond energies as explained in the footnote to Table S1.

Table S5. Reaction energies (kcal/mol) of the MR-MGN-BE17 database^a

System	PBE	B3LYP	M11	revM11	Ref.
NF ₃	242.6	205.8	212.6	223.0	204.5
CO ₂	415.6	387.3	389.2	394.2	389.6
SiO (multiplicity = 1)	195.7	187.4	187.7	186.7	192.4
SO ₂	277.1	249.1	250.8	254.9	259.6
CO	268.3	254.7	257.6	260.1	259.4
SO (multiplicity = 1)	139.6	125.4	125.1	127.1	125.7
ClO	79.9	64.9	64.3	65.8	64.8
F ₂	51.2	35.6	34.7	39.9	38.3
N ₂	243.0	229.0	232.0	234.5	228.5
O ₂	142.8	123.0	118.8	127.7	120.4
NO	171.8	154.7	155.1	161.2	152.7
CN	197.0	179.0	176.0	177.0	181.3
B ₂	76.9	60.0	48.3	58.1	67.4
C ₂	93.6	119.2	109.5	127.6	146.9
O ₃ → O ₂ + O	41.2	16.3	14.8	19.2	26.6
S ₄ → 2S ₂	28.4	12.9	8.1	16.2	25.8
Cl ₂ O → Cl ₂ + O	53.0	40.4	40.5	43.9	41.7
MUEPB ^b	14.8	5.1	7.0	5.6	0.0

^aFor the first 14 rows, the reaction is atomization. For the last three rows, the reaction is as shown.

^bMUEPB denotes mean unsigned error per bond. For the first 14 rows the error is computed from the average bond energy of the molecule as described in the footnote to Table S1. For the last three rows it is the error in the reaction as indicated; note that S₄ is a chain molecule (pictured below), not a four-membered ring, and therefore these three reactions each break one bond.

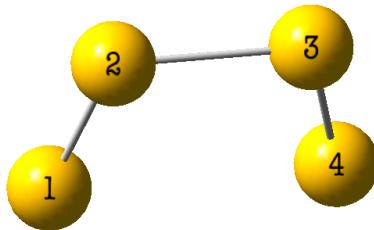
**Figure S1.** Structure of S₄.

Table S6. Reaction energies (kcal/mol) of the MR-TM-BE12 database^a

System	PBE	B3LYP	M11	revM11	Ref.
TiCl	115.6	102.9	110.8	104.8	101.7
VF ₅	629.8	556.2	563.6	549.2	564.2
CrCl	88.9	85.8	90.3	89.2	90.2
CrOF	254.9	226.7	219.4	221.7	247.6
(FeBr ₂) ₂	369.4	340.4	395.7	359.5	366.8
Co(CO) ₄ H	1309.7	1189.7	1185.3	1217.2	1230.1
CuCl	86.4	80.1	89.4	85.4	90.2
VO	182.1	151.1	158.5	151.3	151.0
NiCl	89.5	81.1	90.6	86.6	88.7
VS	128.3	102.4	111.1	98.4	106.9
NiCH ₂ ⁺ → Ni ⁺ + CH ₂	90.8	73.0	50.3	64.7	76.3
Fe(CO) ₅ → Fe + 5CO	193.0	122.2	100.8	115.2	147.4
MUEPB ^b	10.2	4.8	6.9	4.7	0.0

^aFor the first 14 rows, the reaction is atomization. For the last three rows, the reaction is as shown.

^bMUEPB denotes mean unsigned error per bond. For the first ten rows the error is computed from the average bond energy of the molecule as described in the footnote to Table S1. For the eleventh row, it is the error in the reaction as indicated; this reaction breaks one bond. For the final row, the error is divided by five so that it corresponds to the average bond energy of the five metal–ligand bonds that are broken.

Table S7. Atomization energies (kcal/mol) of the MR-TMD-BE3 database

System	PBE	B3LYP	M11	revM11	Ref.
Fe ₂	41.2	9.4	20.9	1.5	26.6
Cr ₂	32.4	-3.3	-136.6	-29.8	36.0
V ₂	102.2	37.4	-7.1	9.0	64.2
MUEPB ^a	18.7	27.8	83.2	48.7	0.0

^aMUEPB denotes mean unsigned error per bond. It is the MUE of the average bond energies as explained in the footnote to Table S1.

Table S8. Barrier heights (kcal/mol) of the HTBH38/18 database

Reaction	PBE	B3LYP	M11	revM11	Ref.
H + HCl → H ₂ + Cl	0.3 -1.2	-0.9 4.4	4.2 6.4	1.8 8.9	6.1 8.0
OH + H ₂ → H ₂ O + H	-5.9 13.5	0.9 13.2	3.8 20.9	5.9 19.1	5.2 21.6
CH ₃ + H ₂ → CH ₄ + H	4.0 9.5	8.9 9.7	10.1 14.2	13.5 13.0	11.9 15.0
OH + CH ₄ → H ₂ O + CH ₃	-5.2 8.8	2.3 13.9	5.0 18.0	4.2 18.0	6.3 19.5
H + H ₂ → H ₂ + H	3.8 3.8	4.3 4.3	8.7 8.7	11.1 11.1	9.7 9.7
OH + NH ₃ → H ₂ O + NH ₂	-11.6 -0.9	-2.3 7.2	4.2 13.2	2.5 12.5	3.4 13.7
HCl + CH ₃ → CH ₄ + Cl	-5.7 -1.8	-1.3 4.8	-0.9 5.3	-0.9 5.7	1.8 6.8
OH + C ₂ H ₆ → H ₂ O + C ₂ H ₅	-8.6 10.6	-0.7 15.5	2.4 20.2	1.9 19.6	3.5 20.4
F + H ₂ → HF + H	-12.4 24.5	-5.6 23.1	-0.9 32.3	-1.5 28.6	1.6 33.8
O + CH ₄ → OH + CH ₃	-0.1 -0.6	7.5 4.4	11.6 5.7	10.4 6.4	14.4 8.9
H + PH ₃ → H ₂ + PH ₂	-1.7 18.4	-0.9 23.2	2.0 24.4	2.4 28.1	2.9 24.7
H + HO → H ₂ + O	3.6 -1.3	4.0 6.3	8.5 10.4	8.2 12.8	10.9 13.2
H + H ₂ S → H ₂ + HS	-1.1 9.5	-0.5 15.9	2.8 16.5	2.1 20.6	3.9 17.2
O + HCl → OH + Cl	-10.1 -6.8	1.4 4.4	9.2 9.5	7.5 10.0	10.4 9.9
CH ₃ + NH ₂ → CH ₄ + NH	0.7 10.8	6.1 17.4	8.3 18.7	7.5 20.2	8.9 22.0
C ₂ H ₅ + NH ₂ → C ₂ H ₆ + NH	2.9 7.8	8.2 14.8	10.1 15.7	8.7 17.5	9.8 19.4
NH ₂ + C ₂ H ₆ → NH ₃ + C ₂ H ₅	1.5 10.1	8.8 15.6	9.9 18.7	9.9 17.6	11.3 17.8
NH ₂ + CH ₄ → NH ₃ + CH ₃	4.5 7.8	11.4 13.5	12.7 16.7	12.4 16.1	13.9 16.9
<i>s-trans cis</i> -C ₅ H ₈ → <i>s-trans cis</i> -C ₅ H ₈	31.4 31.4	39.0 39.0	38.3 38.3	40.1 40.1	39.7 39.7
MUE	9.7	4.5	1.4	1.8	0.0

Table S9. Barrier heights (kcal/mol) of the NHTBH38/18 database

Reaction	PBE	B3LYP	M11	revM11	Ref.
H + N ₂ O → OH + N ₂	10.5 52.8	11.8 73.1	17.2 80.2	16.4 77.5	17.7 82.6
H + FH → HF + H	28.0 28.0	31.8 31.8	39.3 39.3	39.7 39.7	42.1 42.1
H + ClH → HCl + H	10.4 10.4	13.1 13.1	17.7 17.7	16.7 16.7	17.8 17.8
H + FCH ₃ → HF + CH ₃	18.7 41.1	22.0 48.7	28.8 53.9	31.3 54.9	30.5 56.9
H + F ₂ → HF + F	-9.6 80.0	-7.3 95.2	0.2 107.2	-0.1 101.8	1.5 104.8
CH ₃ + FCl → CH ₃ F + Cl	-6.4 42.0	-1.5 51.2	6.0 60.6	7.9 62.4	7.1 59.8
F ⁻ + CH ₃ F → FCH ₃ + F ⁻	-8.3 -8.3	-3.9 -3.9	-1.9 -1.9	-0.4 -0.4	-0.6 -0.6
F ⁻ ⋯CH ₃ F → FCH ₃ ⋯F ⁻	6.7 6.7	10.0 10.0	13.7 13.7	14.8 14.8	13.4 13.4
Cl ⁻ + CH ₃ Cl → ClCH ₃ + Cl ⁻	-3.7 -3.7	-0.5 -0.5	4.1 4.1	6.0 6.0	2.5 2.5
Cl ⁻ ⋯CH ₃ Cl → ClCH ₃ ⋯Cl ⁻	6.9 6.9	9.0 9.0	15.3 15.3	17.0 17.0	13.5 13.5
F ⁻ + CH ₃ Cl → FCH ₃ + Cl ⁻	-19.5 12.1	-16.5 18.0	-12.8 18.5	-10.9 20.2	-12.3 19.8
F ⁻ ⋯CH ₃ Cl → FCH ₃ ⋯Cl ⁻	-1.0 21.0	0.1 26.2	4.4 28.9	5.5 30.6	3.5 29.6
OH ⁻ + CH ₃ F → HOCH ₃ + F ⁻	-10.7 9.6	-5.9 14.5	-4.2 18.3	-2.8 20.8	-2.7 17.6
OH⋯CH ₃ F → HOCH ₃ ⋯F ⁻	3.4 42.7	7.4 45.1	11.5 49.4	12.6 51.9	11.0 47.7
H + N ₂ → HN ₂	5.6 9.2	7.8 11.0	12.2 12.0	11.9 13.0	14.6 10.9
H + CO → HCO	-1.7 24.7	-0.6 24.7	2.2 23.3	1.9 24.6	3.2 22.8
H + C ₂ H ₄ → CH ₃ CH ₂	0.0 40.4	-0.1 41.9	1.8 43.7	1.6 45.6	2.0 42.0
CH ₃ + C ₂ H ₄ → CH ₃ CH ₂ CH ₂	1.6 29.7	6.1 29.4	5.6 33.2	6.9 36.9	6.4 33.0
HCN → HNC	46.0 31.0	47.7 33.8	45.6 33.8	45.9 34.1	48.1 33.0
MUE	8.4	4.5	1.3	2.0	0.0

Table S10. Noncovalent complexation energies (kcal/mol) of the NCCE30/18 database

Reaction	PBE	B3LYP	M11	revM11	Ref.
(NH ₃) ₂	3.34	2.67	3.04	3.48	3.09
(HF) ₂	5.00	4.72	4.85	5.30	4.49
(H ₂ O) ₂	5.44	4.96	5.41	5.79	4.91
NH ₃ ···H ₂ O	7.15	6.32	6.56	7.11	6.38
(HCONH ₂) ₂	14.24	13.06	14.97	14.76	15.41
(HCOOH) ₂	16.33	15.17	17.34	17.32	17.60
C ₂ H ₄ ···F ₂	3.17	1.50	1.17	1.39	1.06
NH ₃ ···F ₂	5.45	2.90	1.57	2.01	1.80
C ₂ H ₂ ···ClF	6.18	3.79	3.80	4.01	3.79
HCN···ClF	5.93	4.57	4.68	5.15	4.80
NH ₃ ···Cl ₂	7.95	5.48	4.22	4.50	4.85
H ₂ O···ClF	7.41	5.69	5.62	5.94	5.20
NH ₃ ···ClF	17.09	12.83	9.97	10.58	11.17
(H ₂ S) ₂	1.80	0.90	1.26	1.59	1.62
(HCl) ₂	2.11	1.32	1.57	1.98	1.91
HCl···H ₂ S	4.16	2.87	2.94	3.33	3.26
CH ₃ Cl···HCl	3.39	2.21	3.29	3.62	3.39
HCN···CH ₃ SH	3.52	2.57	3.64	3.77	3.58
CH ₃ SH···HCl	5.61	3.77	4.58	4.93	4.74
CH ₄ ···Ne	0.27	0.02	0.23	0.26	0.18
C ₆ H ₆ ···Ne	0.35	-0.11	0.96	0.98	0.41
(CH ₄) ₂	0.01	-0.50	0.41	0.79	0.53
CO ₂ ···Ar	0.26	-0.26	0.35	0.66	0.57
(C ₂ H ₂) ₂	1.00	0.44	1.33	1.58	1.36
(C ₂ H ₄) ₂	0.36	-0.48	1.73	2.17	1.44
sandwich (C ₆ H ₆) ₂	-1.56	-2.26	1.57	1.98	1.65
T-shaped (C ₆ H ₆) ₂	0.11	-0.21	2.89	3.17	2.63
parallel-displaced (C ₆ H ₆) ₂	-0.96	-1.82	3.18	3.46	2.59
parallel-displaced (CO ₂) ₂	0.71	0.20	1.23	1.70	1.49
sandwich (C ₅ H ₅ N) ₂	-0.19	-1.69	2.74	2.99	2.89
MUE	1.43	1.25	0.28	0.38	0.0

Table S11. Interaction energies (kcal/mol) of the NGD21/18 database

System^a	PBE	B3LYP	M11	revM11	Ref.
He ₂	0.06	-0.04	0.00	0.02	0.02
Ne ₂	0.12	-0.06	0.07	0.12	0.08
Ar ₂	0.11	-0.22	-0.04	0.21	0.29
Kr ₂	0.09	-0.32	-0.03	0.31	0.40
HeNe	0.09	-0.05	0.03	0.05	0.04
HeAr	0.09	-0.08	-0.01	0.05	0.06
NeAr	0.13	-0.10	0.07	0.15	0.13
HeHe_L_0.3A	0.07	-0.07	-0.07	0.02	0.01
HeHe_R_0.3A	0.04	-0.03	0.02	0.02	0.02
ArAr_L_0.3A	-0.18	-0.57	-0.28	0.13	0.14
ArAr_R_0.3A	0.14	-0.11	0.04	0.14	0.24
NeNe_L_0.3A	0.04	-0.16	-0.03	0.14	0.01
NeNe_R_0.3A	0.10	-0.04	0.08	0.06	0.07
KrKr_L_0.3A	-0.29	-0.79	-0.30	0.19	0.24
KrKr_R_0.3A	0.16	-0.16	0.02	0.22	0.34
HeNe_L_0.3A	0.08	-0.10	-0.04	0.08	0.01
HeNe_R_0.3A	0.06	-0.03	0.04	0.03	0.03
HeAr_L_0.3A	0.06	-0.15	-0.10	0.06	0.02
HeAr_R_0.3A	0.07	-0.05	0.02	0.03	0.05
NeAr_L_0.3A	0.02	-0.25	-0.04	0.17	0.05
NeAr_R_0.3A	0.11	-0.06	0.08	0.08	0.11
MUE	0.10	0.28	0.14	0.04	0.0

^aThe notation _L_0.3A indicates 0.3 Å smaller than the equilibrium distance, and the notation _R_0.3A indicates 0.3 Å bigger than the equilibrium distance.

Table S12. Noncovalent interaction energies (kcal/mol) of the S6x6 database

System	PBE	B3LYP	M11	revM11	Ref.
04_Water-Peptide_0.90	-7.03	-6.35	-8.03	-8.53	-7.63
04_Water-Peptide_0.95	-7.44	-6.84	-8.44	-8.91	-8.06
04_Water-Peptide_1.00	-7.48	-6.92	-8.41	-8.90	-8.08
04_Water-Peptide_1.05	-7.27	-6.76	-8.11	-8.63	-7.84
04_Water-Peptide_1.10	-6.93	-6.45	-7.64	-8.20	-7.46
04_Water-Peptide_1.25	-5.61	-5.18	-5.95	-6.54	-5.99
17_Uracil-Uracil_BP_0.90	-14.39	-13.24	-14.47	-14.34	-15.70
17_Uracil-Uracil_BP_0.95	-15.48	-14.55	-15.88	-15.74	-16.90
17_Uracil-Uracil_BP_1.00	-15.72	-14.94	-16.32	-16.22	-17.18
17_Uracil-Uracil_BP_1.05	-15.40	-14.75	-16.09	-16.09	-16.86
17_Uracil-Uracil_BP_1.10	-14.74	-14.17	-15.42	-15.56	-16.17
17_Uracil-Uracil_BP_1.25	-11.95	-11.51	-12.27	-12.85	-13.15
20_AcOH-AcOH_0.90	-17.57	-16.36	-17.99	-18.18	-17.45
20_AcOH-AcOH_0.95	-18.74	-17.69	-19.33	-19.43	-18.79
20_AcOH-AcOH_1.00	-18.91	-18.00	-19.65	-19.70	-19.09
20_AcOH-AcOH_1.05	-18.43	-17.64	-19.27	-19.33	-18.72
20_AcOH-AcOH_1.10	-17.58	-16.86	-18.42	-18.55	-17.94
20_AcOH-AcOH_1.25	-14.12	-13.56	-14.63	-15.17	-14.56
28_Benzene-Uracil_pi-pi_0.90	4.87	7.98	-3.72	-3.05	-3.53
28_Benzene-Uracil_pi-pi_0.95	1.95	4.40	-5.04	-5.15	-5.22
28_Benzene-Uracil_pi-pi_1.00	0.34	2.32	-5.12	-5.75	-5.71
28_Benzene-Uracil_pi-pi_1.05	-0.52	1.14	-4.65	-5.54	-5.55
28_Benzene-Uracil_pi-pi_1.10	-0.92	0.51	-3.96	-4.91	-5.08
28_Benzene-Uracil_pi-pi_1.25	-1.00	-0.03	-1.97	-2.54	-3.29
29_Pyridine-Uracil_pi-pi_0.90	4.92	8.08	-3.94	-3.00	-3.74
29_Pyridine-Uracil_pi-pi_0.95	1.10	3.43	-5.98	-6.04	-6.17
29_Pyridine-Uracil_pi-pi_1.00	-0.81	0.99	-6.16	-6.87	-6.81
29_Pyridine-Uracil_pi-pi_1.05	-1.70	-0.22	-5.60	-6.58	-6.58
29_Pyridine-Uracil_pi-pi_1.10	-2.04	-0.78	-4.79	-5.77	-5.97
29_Pyridine-Uracil_pi-pi_1.25	-1.84	-1.02	-2.58	-3.07	-3.87
47_Benzene-Benzene_TS_0.90	2.17	3.94	-1.17	-1.21	-1.64
47_Benzene-Benzene_TS_0.95	0.66	2.08	-2.01	-2.34	-2.58
47_Benzene-Benzene_TS_1.00	-0.15	1.04	-2.18	-2.71	-2.87
47_Benzene-Benzene_TS_1.05	-0.55	0.46	-2.05	-2.64	-2.82
47_Benzene-Benzene_TS_1.10	-0.72	0.16	-1.80	-2.37	-2.60
47_Benzene-Benzene_TS_1.25	-0.70	-0.09	-1.03	-1.32	-1.77
MUE	2.61	3.78	0.63	0.52	0.00

Table S13. Ionization potentials (kcal/mol) of the IP23 database

System	PBE	B3LYP	M11	revM11	Ref.
C	266.1	266.3	260.2	265.1	259.7
S	240.9	243.8	241.7	243.9	238.9
SH	239.3	241.4	242.4	243.4	238.9
Cl	298.9	301.1	300.3	302.4	299.1
Cl ₂	256.0	261.2	263.4	266.3	265.3
OH	304.6	306.3	307.2	306.5	299.1
O	324.7	326.8	328.5	329.8	313.9
O ₂	282.5	289.7	284.9	288.1	278.9
P	241.0	238.6	240.9	241.0	241.9
PH	236.0	234.3	234.9	236.8	234.1
PH ₂	229.9	228.8	228.1	231.4	226.3
S ₂	216.7	219.3	220.0	223.2	216.0
Si	188.9	187.0	188.4	189.5	187.9
Cr	170.5	164.6	148.8	159.9	156.0
Cu	193.1	190.2	165.6	183.0	178.2
FeC	186.8	184.6	170.9	178.9	173.7
Mo	172.1	167.1	152.1	164.6	163.7
Pd	200.2	196.4	191.0	194.3	192.2
Rh	180.4	177.7	159.3	172.7	172.1
Ru	178.9	175.3	135.7	170.9	169.9
Zn	221.2	221.7	197.7	216.9	216.6
Co	187.5	186.9	159.7	179.2	181.1
Sc	146.9	150.8	140.0	149.3	151.3
MUE	6.0	5.3	7.9	4.0	0.0

Table S14. Electron affinities (kcal/mol) of the EA13/03 database

System	PBE	B3LYP	M11	revM11	Ref.
C	35.8	31.2	31.0	31.8	29.1
S	49.6	50.6	48.9	50.0	47.9
SH	52.7	52.9	53.2	53.6	53.3
Cl	83.1	83.7	83.5	84.1	83.4
Cl ₂	59.8	64.6	57.0	58.0	55.6
OH	42.3	40.4	41.6	43.4	42.1
O	38.3	36.9	34.2	38.0	33.7
O ₂	8.8	12.0	13.6	13.4	10.8
P	20.2	22.0	18.1	19.7	17.2
PH	24.1	25.1	24.4	24.6	23.2
PH ₂	28.2	28.3	28.7	29.0	29.4
S ₂	36.2	38.6	39.2	41.3	38.5
Si	33.3	30.0	31.3	30.8	31.9
MUE	2.2	2.3	1.0	1.9	0.0

Table S15. Proton affinities (kcal/mol) of the PA8 database

System	PBE	B3LYP	M11	revM11	Ref.
NH ₃	210.9	211.3	211.4	211.0	211.9
H ₂ O	170.4	170.5	170.6	170.6	171.8
C ₂ H ₂	158.9	158.4	159.2	159.0	156.6
SiH ₄	157.0	157.4	154.8	156.9	156.5
PH ₃	190.2	193.1	192.0	191.7	193.1
H ₂ S	174.4	174.8	173.3	171.2	173.7
HCl	138.9	138.0	137.0	135.8	137.1
H ₂	106.0	104.5	106.6	107.3	105.9
MUE	1.3	1.0	1.0	1.4	0.0

Table S16. Isomerization energies (kcal/mol) of the 2pIsoE4 database

System	PBE	B3LYP	M11	revM11	Ref.
C	1.5	0.8	3.3	3.2	3.8
N	64.4	69.8	63.7	55.1	57.1
O	9.4	10.8	9.6	10.6	9.9
F	26.1	24.7	26.7	26.5	26.9
MUE	2.7	4.7	1.9	0.9	0.0

Table S17. Isomerization energies (kcal/mol) of the 4pIsoE4 database

System	PBE	B3LYP	M11	revM11	Ref.
As	37.9	44.7	38.4	32.5	33.0
Br	-5.2	-7.7	-7.7	-6.8	-6.3
Ge	22.7	22.9	24.6	20.6	24.6
Se	22.7	23.0	24.2	22.5	20.8
MUE	2.4	4.2	2.5	1.7	0.0

Table S18. Isomerization energies (kcal/mol) of the IsoL6/11 database

System	PBE	B3LYP	M11	revM11	Ref.
10-	5.3	2.5	5.9	4.8	6.8
13-	31.1	30.3	30.8	28.1	33.5
14-	6.2	3.7	4.5	2.6	5.3
20-	4.9	4.3	3.5	4.6	4.7
3-	7.2	7.3	9.1	12.4	9.8
9-	17.6	18.1	21.5	20.5	21.8
MUE	2.0	2.6	1.1	2.3	0.0

Table S19. Reaction energies (kcal/mol) of the π TC13 database^a

System	PBE	B3LYP	M11	revM11	Ref.
E2-E1	3.1	2.2	0.4	-0.9	-1.4
E4-E3	0.2	-2.4	-7.4	-9.4	-8.8
E6-E5	-1.4	-5.5	-13.4	-16.1	-14.3
P-2	167.9	168.2	169.0	169.3	167.8
P-4	196.4	198.5	194.5	194.0	193.5
P-6	214.3	216.3	210.7	210.1	209.7
P-8	225.7	227.6	220.8	220.0	219.7
P-10	234.0	235.7	227.7	226.7	226.0
SB-2	213.7	215.6	216.5	216.1	214.5
SB-4	228.2	230.0	229.2	228.5	226.2
SB-6	238.3	239.7	237.3	236.3	233.4
SB-8	245.3	246.5	242.7	241.3	238.2
SB-10	250.7	251.5	246.4	244.8	241.0
MUE	5.6	6.0	2.2	1.6	0.0

^aThe first three rows are reaction energies of isomerization reactions, and the next ten rows are proton affinities.

Table S20. Atomic energies (kcal/mol) of the AE17 database

System	PBE	B3LYP	M11	revM11	Ref.
H	-0.5	-0.5	-0.5	-0.5	-0.5
He	-2.9	-2.9	-2.9	-2.9	-2.9
Li	-7.5	-7.5	-7.5	-7.5	-7.5
Be	-14.6	-14.7	-14.7	-14.6	-14.7
B	-24.6	-24.7	-24.7	-24.6	-24.7
C	-37.8	-37.9	-37.8	-37.8	-37.8
N	-54.5	-54.6	-54.6	-54.6	-54.6
O	-75.0	-75.1	-75.1	-75.1	-75.1
F	-99.7	-99.8	-99.8	-99.7	-99.7
Ne	-128.9	-129.0	-129.0	-128.9	-128.9
Na	-162.2	-162.3	-162.3	-162.3	-162.3
Mg	-200.0	-200.1	-200.1	-200.0	-200.1
Al	-242.2	-242.4	-242.4	-242.3	-242.3
Si	-289.2	-289.4	-289.4	-289.4	-289.4
P	-341.1	-341.3	-341.3	-341.3	-341.3
S	-398.0	-398.1	-398.1	-398.1	-398.1
Cl	-460.0	-460.2	-460.2	-460.2	-460.1
MUE	47.3	18.3	9.1	5.0	0.0

Table S21. Energies (kcal/mol) of the HC7/11 database

System	PBE	B3LYP	M11	revM11	Ref.
E22-E1	13.8	-1.1	20.8	32.7	14.3
E31-E1	18.4	1.2	28.9	45.0	25.0
octane iso	-5.2	-8.0	4.5	1.8	1.9
DE (rxn a)	5.8	4.6	7.5	7.3	9.8
DE (rxn b)	8.6	6.8	11.2	10.9	14.8
DE (rxn c)	193.0	163.0	198.8	212.5	194.0
DE (rxn d)	124.9	103.0	129.5	139.9	127.2
MUE	4.0	16.8	3.7	10.9	0.0

Table S22. Atomization energies (kcal/mol) of the SMAE3/19 database

System	PBE	B3LYP	M11	revM11	Ref.
SO ₃	370.3	329.2	332.6	339.3	344.2
H ₂ S ₂	244.5	236.6	241.4	238.6	240.8
H ₂ SO ₄	621.7	578.0	590.8	596.2	602.2
MUEPB ^a	4.4	3.5	2.0	1.1	0.0

^aMUEPB denotes mean unsigned error per bond. It is the MUE of the average bond energies as explained in the footnote to Table S1.

Table S23. Reaction energies (kcal/mol) of the DC9/19 database

Reaction	<i>n</i>	PBE	B3LYP	M11	revM11	Ref.
$\text{HCN}\cdots\text{BF}_3 \rightarrow \text{HCN} + \text{BF}_3$	1	4.3	3.8	6.8	6.4	5.7
$\text{C}_6\text{Cl}_6 + 6\text{HCl} \rightarrow 6\text{Cl}_2 + \text{C}_6\text{H}_6$	1	136.0	127.0	143.7	143.7	152.6
$\text{P}_4 \rightarrow 4\text{P}$	6	308.1	278.5	288.7	281.7	289.9
$\text{SF}_6 \rightarrow \text{S} + 6\text{F}$	6	501.9	448.8	472.9	483.9	477.5
$\text{PF}_5 \rightarrow \text{P} + 5\text{F}$	5	562.0	532.8	552.6	556.6	556.4
$\text{P}_4\text{O}_{10} \rightarrow \text{P}_4 + 5\text{O}_2$	5	618.6	648.7	702.4	687.1	719.7
$\text{C}_6\text{F}_6 \rightarrow 6\text{C} + 6\text{F}$	12	1484.5	1378.4	1395.1	1417.2	1388.1
$\text{Si}(\text{OCH}_3)_4 \rightarrow \text{Si} + 4\text{C} + 4\text{O} + 12\text{H}$	20	2045.6	2003.9	2033.3	2044.6	2023.5
urotropine $\rightarrow 6\text{C} + 4\text{N} + 12\text{H}$	24	2213.9	2129.1	2167.8	2199.3	2151.1
MUEPB ^a		6.5	6.2	1.9	2.7	0.0

^aMUEPB denotes mean unsigned error per bond. In computing the MUEPB, the error in each reaction energy was divided by *n* to convert it to per-bond basis. The value of *n* is the net number of bonds broken (the only exception to this is the $\text{C}_6\text{Cl}_6 + 6\text{HCl}$ reaction for which we choose *n* to be 1). For the P_4O_{10} reaction, P_4O_{10} has 16 bonds (four P=O bonds and 12 P–O bonds), P_4 has six bonds, and five O_2 molecules have a total of five bonds; therefore *n* is 5.

Table S24. Bond dissociation energies (kcal/mol) of the ABDE13 database

Reaction	PBE	B3LYP	M11	revM11	Ref.
propene → methyl + vinyl	105.7	101.4	106.9	106.6	107.2
methyl formate → HC(O)O + methyl	87.7	88.6	104.8	104.0	97.0
butane → propyl + methyl	93.5	88.7	95.9	95.2	96.2
butane → 2 ethyl	89.1	84.7	92.0	92.4	94.4
<i>cis</i> -3-hexene → C ₂ H ₅ CHCH + ethyl	100.0	95.8	102.3	102.8	104.4
<i>cis</i> -3-hexene → C ₂ H ₅ CHCHCH ₂ + methyl	76.3	71.6	78.4	79.4	80.2
1-pentene → CH ₂ CHCH ₂ CH ₂ + methyl	93.6	88.9	96.1	95.3	96.4
1-pentene → CH ₂ CHCH ₂ + ethyl	73.8	69.2	77.2	78.7	80.0
1-pentene → CH ₂ CH + propyl	102.2	98.2	104.3	104.8	106.3
methyl butanoate → CH ₃ OC(O)CH ₂ CH ₂ + methyl	94.8	89.9	97.3	96.3	97.4
methyl butanoate → CH ₃ OC(O)CH ₂ + ethyl	85.1	80.8	90.0	89.9	91.2
methyl butanoate → CH ₃ OC(O) + propyl	92.3	89.2	97.4	97.0	98.7
methyl butanoate → CH ₃ + OC(O)CH ₂ CH ₂ CH ₃	83.5	84.4	101.3	100.4	94.0
MUE	5.1	8.6	2.3	2.1	0.0

Table S25. Bond distances (\AA) of the DGL6 database

System	PBE	B3LYP	M11	revM11	Ref.
Cl_2	2.011	2.016	1.987	1.978	1.988
H_2	0.751	0.743	0.751	0.742	0.741
HF	0.929	0.922	0.923	0.923	0.917
MgS	2.157	2.151	2.134	2.118	2.143
N_2	1.103	1.091	1.084	1.088	1.098
OH	0.983	0.974	0.973	0.974	0.970
MUE	0.013	0.009	0.007	0.009	0.000

Table S26. Bond distances (\AA) of the DGH4 database

System	PBE	B3LYP	M11	revM11	Ref.
NaBr	2.522	2.520	2.535	2.501	2.502
HBr	1.432	1.424	1.422	1.419	1.414
ZnS	2.063	2.077	2.043	2.035	2.046
Ag ₂	2.559	2.584	2.556	2.554	2.530
MUE	0.021	0.028	0.018	0.010	0.000

Table S27. Δ SCF excitation energies (kcal/mol) of the 3dEE8 database

System	PBE	B3LYP	M11	revM11	Ref.
Sc	13.3	20.1	18.9	25.9	32.9
Mn ⁺	26.6	21.9	24.0	21.0	27.1
Fe	21.3	22.4	16.0	23.0	34.2
Ni ⁺	24.2	17.9	35.2	36.6	24.0
Zn	96.8	97.7	79.0	88.8	92.4
Ca ⁺	24.2	27.4	20.9	27.2	39.0
V	-12.2	-0.5	-13.2	-5.8	6.0
Fe ₂	11.3	-3.6	-6.2	1.8	12.2
MUE	9.0	9.4	14.5	9.3	0.0

Table S28. Δ SCF excitation energies (kcal/mol) of the 4dAEE5 database

System	PBE	B3LYP	M11	revM11	Ref.
Mo ⁺	32.9	27.8	41.5	34.2	43.5
Ru ⁺	24.0	28.1	30.5	24.3	26.2
Rh ⁺	16.5	16.5	14.9	12.6	23.3
Pd	15.8	13.8	27.6	17.4	18.8
Y ⁺	-1.4	0.5	-3.3	3.0	2.4
MUE	5.3	6.3	5.8	4.8	0.0

Table S29. Δ SCF excitation energies (kcal/mol) of the pAEE5 database

System	PBE	B3LYP	M11	revM11	Ref.
F	293.5	295.1	301.1	297.3	292.8
Ar	261.4	262.9	272.4	266.8	266.3
C ⁺	114.1	122.6	122.7	125.0	123.0
Al	82.9	88.4	78.5	84.8	83.0
Si ⁺	121.9	127.7	118.9	126.3	126.0
MUE	3.7	2.6	5.3	1.8	0.0

Table S30. Excitation energies (eV) of the EE23 database

Reaction	Transition	PBE	B3LYP	M11	revM11	Ref.
Acetaldehyde	$^1A'' n \rightarrow \pi^*$	4.1	4.2	3.9	3.9	4.3
Acetone	$^1A_2 n \rightarrow \pi^*$	4.2	4.4	4.1	4.0	4.4
Formaldehyde	$^1A_2 n \rightarrow \pi^*$	3.8	3.9	3.5	3.6	4.0
Pyrazine	$^1B_{3u} n \rightarrow \pi^*$	3.5	3.9	4.0	4.0	4.0
Pyridazine	$^1B_1 n \rightarrow \pi^*$	3.1	3.6	3.6	3.7	3.6
Pyridine	$^1B_1 n \rightarrow \pi^*$	4.3	4.8	4.9	4.9	4.7
Pyrimidine	$^1B_1 n \rightarrow \pi^*$	3.8	4.2	4.4	4.5	4.2
s-Tetrazine	$^1B_{3u} n \rightarrow \pi^*$	1.8	2.3	2.3	2.3	2.3
Ethylene	$^1B_{1u} \pi \rightarrow \pi^*$	7.4	7.3	7.3	7.4	8.0
Butadiene	$^1B_u \pi \rightarrow \pi^*$	5.4	5.5	5.7	5.8	6.2
Benzene	$^1B_{2u}, \pi \rightarrow \pi^*$	5.1	5.3	5.5	5.4	4.9
Benzene	$^3B_{1u} \pi \rightarrow \pi^*$	3.9	3.7	3.6	3.5	4.1
Naphthalene	$^1B_{3u} \pi \rightarrow \pi^*$	4.0	4.3	4.7	4.8	4.0
Naphthalene	$^3B_{2u} \pi \rightarrow \pi^*$	2.8	2.7	2.7	2.6	3.1
Furan	$^1B_2 \pi \rightarrow \pi^*$	5.9	5.9	6.0	6.1	6.1
Furan	$^3B_2 \pi \rightarrow \pi^*$	3.9	3.7	3.9	3.7	4.2
Hexatriene	$^1B_u \pi \rightarrow \pi^*$	4.4	4.6	4.9	5.0	4.9
Hexatriene	$^3B_u \pi \rightarrow \pi^*$	2.3	2.1	2.0	2.1	2.7
Water	Singlet, $2p_x \rightarrow 3s$	6.4	6.9	6.9	7.2	7.4
Water	Triplet, $2p_x \rightarrow 3s$	6.0	6.5	6.7	6.8	7.0
pNA	Intramolecular CT, $^1A_1, \pi \rightarrow \pi^*$	3.6	4.0	4.7	4.8	4.3
DMABN	Intramolecular CT, $^1A_1, \pi \rightarrow \pi^*$	4.4	4.7	5.0	5.0	4.6
B-TCNE	Intermolecular CT, $^1A, \pi \rightarrow \pi^*$	1.4	1.9	3.6	3.9	3.6
	MUE (VEE18)	0.4	0.3	0.3	0.4	0.0
	MUE (REE2)	1.0	0.5	0.4	0.2	0.0
	MUE (CTEE3)	1.1	0.7	0.3	0.4	0.0

Table S31. Charge transfer excitation energies (eV) of the LRCTEE9^a database

System	PBE	B3LYP	M11	revM11	Ref.
<i>d</i>(NH₃···HNO₂)					
3.6772	2.21	3.56	6.28	6.93	7.10
6.1133	1.85	3.44	7.16	7.86	8.22
8.5632	1.78	3.49	7.69	8.39	8.75
11.0156	1.75	3.53	8.01	8.70	9.13
13.4708	1.74	3.56	8.22	8.92	9.35
15.9272	1.73	3.58	8.37	9.07	9.51
18.3868	1.73	3.60	8.49	9.18	9.62
23.3034	1.72	3.63	8.64	9.33	9.77
25.763	1.72	3.64	8.70	9.39	9.83
MUE	7.23	5.47	1.08	0.39	0.00

^aThe LRCTEE9 database comprises of $\pi \rightarrow \pi^*$ ${}^1\text{A}_1$ excitations of the NH₃···HNO₂ complex at nine intermonomer distances (*d* in Å).

Table S32. Dimerization energies (kcal/mol) of the AL2X6 database

Reaction	PBE	B3LYP	M11	revM11	Ref.
$\text{Al}_2\text{H}_6 \rightarrow 2\text{AlH}_3$	38.6	33.6	39.9	39.2	38.5
$\text{Al}_2\text{F}_6 \rightarrow 2\text{AlF}_3$	44.2	45.0	55.3	50.9	51.6
$\text{Al}_2\text{Cl}_6 \rightarrow 2\text{AlCl}_3$	26.6	21.8	32.2	31.6	32.5
$\text{Al}_2\text{Me}_4 \rightarrow 2\text{AlMe}_2$	35.9	31.2	39.4	38.3	38.4
$\text{Al}_2\text{Me}_5 \rightarrow \text{AlMe}_2 + \text{AlMe}_3$	27.1	20.9	32.0	31.4	31.2
$\text{Al}_2\text{Me}_6 \rightarrow 2\text{AlMe}_3$	17.5	9.5	23.9	23.7	23.1
MUE	4.3	8.9	1.3	0.5	0.0

Table S33. Barrier heights (kcal/mol) of the BHDIV10 database

Reaction	PBE	B3LYP	M11	revM11	Ref.
ed1 → ts1	15.9	20.7	24.5	27.6	25.7
ed2 → ts2	59.2	62.2	58.9	54.9	56.9
ed3 → ts3	25.7	30.8	37.8	37.5	36.5
ed4 → ts4	83.1	94.9	94.7	95.1	96.2
ed5 → ts5	5.5	12.9	14.7	17.9	15.9
ed6 → ts6	7.4	13.7	15.2	13.1	13.6
ed7 → ts7	23.0	31.1	28.4	27.3	27.5
ed8 → ts8	36.3	48.2	50.9	55.3	50.2
ed9 → ts9	56.4	64.8	65.2	66.3	65.8
ed10 → ts10	63.2	64.5	61.1	62.1	64.9
MUE	8.2	2.8	1.5	1.7	0.0

Table S34. Barrier heights (kcal/mol) of the BHPERI26 database

Reaction	PBE	B3LYP	M11	revM11	Ref.
Cyclobutene → TS1	32.2	33.1	36.5	40.9	35.3
cis-1,3,5-Hexatriene → TS2	25.5	31.2	30.2	32.8	30.8
ortho-xylylene → TS3	23.2	28.3	29.9	31.5	28.1
1,3-Pentadiene → TS4	31.3	38.9	38.4	40.1	39.7
1,3-Cyclopentadiene → TS5	22.8	27.1	24.2	25.1	28.3
1,5-Hexadiene → TS6	28.2	36.0	34.9	39.3	35.8
1,3-Butadiene + Ethylene → TS7	16.5	26.3	21.6	24.8	22.3
1,3-Cyclopentadiene + Ethylene → TS8	14.6	24.2	18.9	21.2	18.0
1,3-Cyclopentadiene → TS11	13.4	23.8	18.1	20.2	14.5
Cis-triscyclopropacyclohexane → TS9	20.2	21.9	31.1	40.7	26.4
13r_1 + 13_c2h4 → 13ts_1a	20.8	28.9	31.7	31.5	27.6
13r_2 + 13_c2h4 → 13ts_2a	14.6	22.5	24.0	24.1	20.0
13r_3 + 13_c2h4 → 13ts_3a	11.0	18.3	18.0	18.4	13.8
13r_4 + 13_c2h4 → 13ts_4a	7.8	15.5	17.8	18.5	11.8
13r_5 + 13_c2h4 → 13ts_5a	4.2	10.3	9.6	10.1	6.5
13r_6 + 13_c2h4 → 13ts_6a	3.4	9.7	7.6	8.6	4.7
13r_7 + 13_c2h4 → 13ts_7a	9.5	16.5	14.0	14.2	13.1
13r_8 + 13_c2h4 → 13ts_8a	3.5	10.3	6.3	6.9	5.9
13r_9 + 13_c2h4 → 13ts_9a	-0.6	4.8	0.9	1.6	0.5
02r + 00r → 02ts	14.4	24.4	19.0	21.3	18.1
03r + 00r → 03ts	12.0	23.2	17.4	19.8	16.6
04r + 00r → 04ts	19.8	29.6	24.1	25.3	22.9
05r + 00r → 06ts	25.5	35.9	30.8	30.9	27.8
07r + 00r → 07ts	18.7	29.0	22.8	24.6	21.3
07r + 00r → 08ts	18.7	28.9	22.8	24.6	21.6
09r + 00r → 09ts	29.1	39.5	33.2	34.1	31.3
MUE	3.9	4.3	2.2	3.6	0.0

Table S35. Barrier heights (kcal/mol) of the BHROT27 database

Reaction	PBE	B3LYP	M11	revM11	Ref.
ethane_st → ethane_ecl	2.5	2.6	2.5	2.6	2.7
tmethane_60 → tmethane_0	6.8	6.9	7.0	7.1	7.0
tmethane_60 → tmethane_120	3.2	3.2	3.9	3.7	3.5
tmethane_180 → tmethane_120	3.4	3.5	3.9	3.8	3.7
methanol_st → methanol_ecl	1.1	1.0	0.9	0.9	1.0
methylamine_st → methylamine_ecl	2.7	2.4	2.1	2.2	2.3
h2o2 → h2o2_trans	1.0	0.9	1.0	1.1	1.0
h2o2 → h2o2_cis	7.1	7.1	7.3	7.5	7.2
h2s2 → h2s2_trans	6.0	5.8	5.7	5.9	5.8
h2s2 → h2s2_cis	8.3	7.9	7.8	8.1	8.0
n2h4_st1 → n2h4_ecl1	1.8	1.6	1.5	1.5	1.6
n2h4_st1 → n2h4_ecl2	8.4	8.7	9.2	9.5	8.4
nh2oh_st1 → nh2oh_ecl	6.9	6.9	7.1	7.3	6.9
nh2oh_st2 → nh2oh_ecl	2.8	2.8	2.9	3.1	2.7
acetamide_RC → acetamide_TS1	18.8	18.6	18.9	18.3	17.2
acetamide_RC → acetamide_TS2	16.1	15.8	16.0	15.5	14.5
biphenyl → biphenyl_TS	1.7	1.9	2.3	2.5	2.1
bifuran_anti → bifuran_TS	4.7	5.0	4.6	4.0	3.9
bifuran_syn → bifuran_TS	3.2	3.2	3.0	2.3	2.1
bithiophene_anti → bithiophene_TS	2.2	2.0	2.2	1.7	1.8
bithiophene_syn → bithiophene_TS	2.0	1.6	1.7	1.3	1.4
butadiene_strans → butadiene_TS	7.5	7.2	5.9	5.5	6.3
butadiene_scis → butadiene_TS	3.8	3.5	3.5	3.0	3.4
ethylthiourea_180 → ethylthiourea_TS1	10.9	11.2	12.4	12.0	10.4
ethylthiourea_0 → ethylthiourea_TS1	11.0	11.4	12.6	12.3	10.2
ethylthiourea_180 → ethylthiourea_TS2	17.4	17.6	19.1	18.6	17.2
ethylthiourea_0 → ethylthiourea_TS2	17.5	17.8	19.3	18.9	17.1
MUE	0.5	0.4	0.7	0.5	0.0

Table S36. Double ionization potentials (kcal/mol) of the DIPCS10 database

System	PBE	B3LYP	M11	revM11	Ref.
$\text{C}_4\text{H}_4 \rightarrow \text{C}_4\text{H}_4^{2+}$	525.3	522.7	525.1	531.5	529.2
$\text{C}_2\text{H}_6 \rightarrow \text{C}_2\text{H}_6^{2+}$	654.3	666.7	665.2	676.3	667.1
$\text{C}_2\text{H}_4 \rightarrow \text{C}_2\text{H}_4^{2+}$	648.8	650.6	648.5	655.5	655.8
$\text{N}_2\text{H}_2 \rightarrow \text{N}_2\text{H}_2^{2+}$	627.1	633.5	628.3	630.6	626.9
$\text{NH}_3 \rightarrow \text{NH}_3^{2+}$	779.0	779.0	775.1	776.8	776.5
$\text{CH}_2\text{O} \rightarrow \text{CH}_2\text{O}^{2+}$	739.2	752.3	751.0	755.0	747.6
$\text{H}_2\text{S} \rightarrow \text{H}_2\text{S}^{2+}$	731.2	733.3	733.1	737.4	733.0
$\text{PH}_3 \rightarrow \text{PH}_3^{2+}$	648.2	649.5	644.9	652.5	649.6
$\text{Mg} \rightarrow \text{Mg}^{2+}$	529.2	535.1	515.2	531.6	522.1
$\text{Be} \rightarrow \text{Be}^{2+}$	634.0	639.5	633.3	643.9	634.8
MUE	4.6	4.4	3.3	4.9	0.0

Table S37. Dissociation energies (kcal/mol) of the HeavySB11 database

Reaction	PBE	B3LYP	M11	revM11	Ref.
$\text{Ge}_2\text{H}_6 \rightarrow 2\text{GeH}_3$	70.0	67.6	76.9	70.7	73.8
$\text{Sn}_2\text{Me}_6 \rightarrow 2\text{SnMe}_3$	55.2	52.9	63.6	56.1	61.7
$\text{Pb}_2\text{Me}_6 \rightarrow 2\text{PbMe}_3$	43.7	40.8	50.9	48.0	52.9
$\text{H}_2\text{S}_2 \rightarrow 2\text{SH}$	70.3	61.9	66.1	65.4	67.9
$\text{H}_2\text{Se}_2 \rightarrow 2\text{SeH}$	60.6	53.2	57.8	55.1	58.4
$\text{Te}_2\text{Me}_2 \rightarrow 2\text{TeMe}$	53.7	46.7	50.4	46.2	52.9
$\text{P}_2\text{Me}_4 \rightarrow 2\text{PMe}_2$	54.8	49.0	55.0	55.3	61.9
$\text{As}_2\text{Me}_4 \rightarrow 2\text{AsMe}_2$	47.2	41.7	50.0	48.2	52.2
$\text{Sb}_2\text{Me}_4 \rightarrow 2\text{SbMe}_2$	41.0	35.8	43.9	37.8	43.8
$\text{Cl}_2 \rightarrow 2\text{Cl}$	65.6	55.4	60.4	58.9	59.7
$\text{Br}_2 \rightarrow 2\text{Br}$	57.8	49.1	53.1	50.6	53.2
MUE	4.6	7.6	2.0	4.2	0.0

Table S38. Proton-exchange barriers (kcal/mol) of the PX13 database

Reaction	PBE	B3LYP	M11	revM11	Ref.
nh3_2 → nh3_2_ts	50.6	57.5	58.4	56.7	59.3
nh3_3 → nh3_3_ts	34.6	44.2	45.4	42.2	46.9
nh3_4 → nh3_4_ts	36.4	45.8	47.4	45.4	48.4
h2o_2 → h2o_2_ts	38.0	46.0	46.5	44.2	48.6
h2o_3 → h2o_3_ts	18.7	26.7	28.2	25.4	29.8
h2o_4 → h2o_4_ts	14.6	23.0	25.0	20.9	26.6
h2o_5 → h2o_5_ts	15.7	25.6	28.5	23.1	30.1
h2o_6 → h2o_6_ts	18.2	29.7	33.2	26.6	35.1
hf_2 → hf_2_ts	31.6	38.8	37.5	37.2	42.3
hf_3 → hf_3_ts	10.7	16.9	15.0	16.4	20.7
hf_4 → hf_4_ts	5.3	10.8	8.2	9.9	14.7
hf_5 → hf_5_ts	4.3	10.2	7.2	9.1	14.6
hf_6 → hf_6_ts	4.6	11.5	7.9	10.3	16.6
MUE	11.6	3.6	3.5	5.1	0.0

Table S39. Interaction energies (kcal/mol) of the SIE4x4 database

Reaction	PBE	B3LYP	M11	revM11	Ref.
$\text{h}_2^+ \text{ } 1.0 \rightarrow \text{h} + \text{h}^+$	68.6	67.6	64.9	70.2	64.4
$\text{h}_2^+ \text{ } 1.25 \rightarrow \text{h} + \text{h}^+$	67.0	65.2	63.7	66.5	58.9
$\text{h}_2^+ \text{ } 1.5 \rightarrow \text{h} + \text{h}^+$	61.0	58.5	56.9	57.6	48.7
$\text{h}_2^+ \text{ } 1.75 \rightarrow \text{h} + \text{h}^+$	54.9	51.6	48.9	48.3	38.3
$\text{he}_2^+ \text{ } 1.0 \rightarrow \text{he} + \text{he}^+$	76.4	76.5	69.5	73.6	56.9
$\text{he}_2^+ \text{ } 1.25 \rightarrow \text{he} + \text{he}^+$	75.7	73.5	66.2	69.5	46.9
$\text{he}_2^+ \text{ } 1.5 \rightarrow \text{he} + \text{he}^+$	69.5	64.9	55.7	59.1	31.3
$\text{he}_2^+ \text{ } 1.75 \rightarrow \text{he} + \text{he}^+$	65.6	59.0	47.4	50.4	19.1
$(\text{nh}_3)_2^+ \text{ } 1.0 \rightarrow \text{nh}_3 + \text{nh}_3^+$	47.2	42.1	39.9	38.7	35.9
$(\text{nh}_3)_2^+ \text{ } 1.25 \rightarrow \text{nh}_3 + \text{nh}_3^+$	43.9	37.5	31.3	29.3	25.9
$(\text{nh}_3)_2^+ \text{ } 1.5 \rightarrow \text{nh}_3 + \text{nh}_3^+$	37.8	29.9	19.3	17.0	13.4
$(\text{nh}_3)_2^+ \text{ } 1.75 \rightarrow \text{nh}_3 + \text{nh}_3^+$	34.7	25.5	11.6	8.4	4.9
$(\text{h}_2\text{o})_2^+ \text{ } 1.0 \rightarrow \text{h}_2\text{o} + \text{h}_2\text{o}^+$	57.3	49.6	43.3	44.8	39.7
$(\text{h}_2\text{o})_2^+ \text{ } 1.25 \rightarrow \text{h}_2\text{o} + \text{h}_2\text{o}^+$	55.4	45.6	34.1	35.4	29.1
$(\text{h}_2\text{o})_2^+ \text{ } 1.5 \rightarrow \text{h}_2\text{o} + \text{h}_2\text{o}^+$	50.6	38.9	23.0	23.6	16.9
$(\text{h}_2\text{o})_2^+ \text{ } 1.75 \rightarrow \text{h}_2\text{o} + \text{h}_2\text{o}^+$	48.5	35.5	16.6	16.1	9.3
MUE	23.4	17.6	9.6	10.6	0.0

Table S40. Bond-dissociation energies (kcal/mol) of the YBDE18 database

Reaction	PBE	B3LYP	M11	revM11	Ref.
f2s-cbh22 → f2s + cbh22	51.5	45.2	53.2	54.5	57.2
f2s-ch2 → f2s + ch2	96.3	87.5	90.4	86.8	91.1
h2s-cbh22 → h2s + cbh22	16.3	11.1	19.7	21.0	21.0
h2s-ch2 → h2s + ch2	43.0	35.2	38.8	35.4	36.4
me2s-cbh22 → me2s + cbh22	31.2	25.2	36.4	37.0	39.0
me2s-ch2 → me2s + ch2	55.9	47.9	52.4	48.9	51.7
nf3-cbh22 → nf3 + cbh22	14.2	7.0	11.6	11.9	12.3
nf3-ch2 → nf3 + ch2	63.7	55.0	54.4	50.4	53.7
nh3-cbh22 → nh3 + cbh22	27.2	23.2	31.9	35.7	32.6
nh3-ch2 → nh3 + ch2	31.7	26.9	31.8	31.0	28.9
nme3-cbh22 → nme3 + cbh22	29.4	24.4	41.2	42.6	42.1
nme3-ch2 → nme3 + ch2	40.2	35.1	45.0	42.8	42.8
pf3-cbh22 → pf3 + cbh22	39.7	36.6	48.6	49.3	52.3
pf3-ch2 → pf3 + ch2	74.6	69.5	76.6	72.7	76.4
ph3-cbh22 → ph3 + cbh22	36.1	32.9	41.6	44.3	44.8
ph3-ch2 → ph3 + ch2	62.0	56.7	60.6	58.5	60.1
pme3-cbh22 → pme3 + cbh22	54.8	50.8	62.7	63.2	66.6
pme3-ch2 → pme3 + ch2	77.6	72.2	77.3	73.9	78.0
MUE	5.9	8.2	1.8	2.1	0.0

11_MeNH ₂ -Peptide	-4.88	-5.17	-5.09	-4.82	-4.44	-2.69	-1.18	-0.41
12_MeNH ₂ -Water	-6.52	-7.02	-7.10	-6.89	-6.51	-5.01	-2.98	-1.21
13_Peptide-MeOH	-5.22	-5.68	-5.75	-5.56	-5.25	-4.13	-2.67	-1.29
14_Peptide-MeNH ₂	-5.85	-6.48	-6.67	-6.57	-6.27	-4.96	-3.10	-1.41
15_Peptide-Peptide	-7.41	-7.98	-8.08	-7.88	-7.49	-5.98	-3.89	-1.67
16_Peptide-Water	-4.49	-4.89	-4.94	-4.79	-4.55	-3.64	-2.44	-1.22
17_Uracil-Uracil_BP	-14.47	-15.88	-16.32	-16.09	-15.42	-12.27	-7.58	-3.17
18_Water-Pyridine	-5.89	-6.37	-6.46	-6.29	-5.95	-4.59	-2.79	-1.19
19_MeOH-Pyridine	-5.71	-6.33	-6.52	-6.41	-6.12	-4.81	-2.92	-1.25
20_AcOH-AcOH	-17.99	-19.33	-19.65	-19.27	-18.42	-14.63	-8.85	-3.52
21_AcNH ₂ -AcNH ₂	-14.17	-15.48	-15.84	-15.56	-14.87	-11.84	-7.51	-3.02
22_AcOH-Uracil	-17.50	-18.92	-19.33	-19.05	-18.30	-14.78	-9.32	-4.06
23_AcNH ₂ -Uracil	-16.80	-18.17	-18.62	-18.41	-17.77	-14.64	-9.67	-4.61
24_Benzene-Benzene_pi-pi	0.18	-1.41	-1.81	-1.67	-1.34	-0.30	0.08	0.06
25_Pyridine-Pyridine_pi-pi	-0.93	-2.54	-2.94	-2.76	-2.39	-1.17	-0.38	-0.12
26_Uracil-Uracil_pi-pi	-7.99	-9.16	-9.02	-8.27	-7.30	-4.56	-2.25	-0.79
27_Benzene-Pyridine_pi-pi	-0.39	-2.10	-2.49	-2.32	-1.91	-0.76	-0.13	-0.03
28_Benzene-Uracil_pi-pi	-3.72	-5.04	-5.12	-4.65	-3.96	-1.97	-0.57	-0.04
29_Pyridine-Uracil_pi-pi	-3.94	-5.98	-6.16	-5.60	-4.79	-2.58	-1.17	-0.39
30_Benzene-Ethene	0.01	-0.66	-0.79	-0.67	-0.48	0.00	0.15	0.09
31_Uracil-Ethene	-2.58	-3.00	-2.94	-2.67	-2.30	-1.32	-0.57	-0.17
32_Uracil-Ethyne	-2.83	-3.40	-3.39	-3.10	-2.73	-1.65	-0.73	-0.21
33_Pyridine-Ethene	-0.59	-1.14	-1.22	-1.08	-0.87	-0.28	0.02	0.04
34_Pentane-Pentane	-3.80	-3.91	-3.55	-3.01	-2.45	-1.09	-0.27	-0.04
35_Neopentane-Pentane	-1.81	-2.10	-1.98	-1.70	-1.38	-0.62	-0.19	-0.03
36_Neopentane-Neopentane	-0.81	-0.98	-0.96	-0.85	-0.71	-0.34	-0.12	-0.01
37_Cyclopentane-Neopentane	-1.57	-1.85	-1.75	-1.52	-1.25	-0.58	-0.18	-0.03
38_Cyclopentane-Cyclopentane	-1.68	-1.98	-1.90	-1.64	-1.32	-0.57	-0.16	-0.03
39_Benzene-Cyclopentane	-2.31	-3.07	-3.10	-2.78	-2.33	-1.13	-0.35	-0.06
40_Benzene-Neopentane	-1.64	-2.22	-2.24	-2.02	-1.72	-0.89	-0.32	-0.08
41_Uracil-Pentane	-4.34	-4.67	-4.37	-3.80	-3.02	-1.24	-0.37	-0.04
42_Uracil-Cyclopentane	-2.90	-3.42	-3.26	-2.84	-2.34	-1.10	-0.35	-0.06
43_Uracil-Neopentane	-2.66	-3.07	-2.93	-2.56	-2.13	-1.07	-0.40	-0.08
44_Ethene-Pentane	-1.85	-1.87	-1.67	-1.40	-1.13	-0.51	-0.17	-0.03
45_Ethyne-Pentane	-1.16	-1.41	-1.35	-1.17	-0.96	-0.46	-0.19	-0.04
46_Peptide-Pentane	-4.46	-4.43	-4.07	-3.57	-3.03	-1.64	-0.52	-0.10
47_Benzene-Benzene_TS	-1.17	-2.01	-2.18	-2.05	-1.80	-1.03	-0.43	-0.12
48_Pyridine-Pyridine_TS	-2.15	-2.78	-2.86	-2.68	-2.40	-1.53	-0.75	-0.28
49_Benzene-Pyridine_TS	-1.76	-2.61	-2.77	-2.61	-2.32	-1.45	-0.69	-0.26
50_Benzene-Ethyne_CH-pi	-2.03	-2.68	-2.77	-2.60	-2.33	-1.52	-0.77	-0.25
51_Ethyne-Ethyne_TS	-0.92	-1.16	-1.21	-1.15	-1.06	-0.76	-0.42	-0.13
52_Benzene-AcOH_OH-pi	-4.14	-4.60	-4.57	-4.28	-3.88	-2.67	-1.42	-0.50
53_Benzene-AcNH ₂ _NH-pi	-3.60	-3.99	-3.98	-3.77	-3.47	-2.48	-1.34	-0.41
54_Benzene-Water_OH-pi	-3.43	-3.65	-3.55	-3.29	-2.98	-2.09	-1.17	-0.43
55_Benzene-MeOH_OH-pi	-3.80	-4.22	-4.19	-3.93	-3.57	-2.46	-1.31	-0.47
56_Benzene-MeNH ₂ _NH-pi	-2.58	-2.98	-2.97	-2.73	-2.39	-1.46	-0.67	-0.20
57_Benzene-Peptide_NH-pi	-3.92	-4.88	-4.95	-4.62	-4.12	-2.68	-1.36	-0.52
58_Pyridine-Pyridine_CH-N	-2.08	-3.07	-3.32	-3.02	-2.61	-1.58	-0.75	-0.21
59_Ethyne-Water_CH-O	-2.64	-2.88	-2.88	-2.76	-2.58	-1.98	-1.24	-0.53
60_Ethyne-AcOH_OH-pi	-4.55	-4.96	-4.93	-4.65	-4.25	-3.00	-1.63	-0.56
61_Pentane-AcOH	-2.63	-2.56	-2.35	-2.06	-1.75	-0.96	-0.29	-0.04

44_Ethene-Pentane	-2.51	-2.69	-2.51	-2.17	-1.77	-0.79	-0.16	-0.01
45_Ethyne-Pentane	-1.74	-2.13	-2.09	-1.85	-1.53	-0.69	-0.16	-0.02
46_Peptide-Pentane	-5.12	-5.45	-5.25	-4.76	-4.12	-2.25	-0.53	-0.07
47_Benzene-Benzene_TS	-1.21	-2.34	-2.71	-2.64	-2.37	-1.32	-0.41	-0.09
48_Pyridine-Pyridine_TS	-2.17	-3.15	-3.45	-3.35	-3.05	-1.88	-0.74	-0.23
49_Benzene-Pyridine_TS	-1.72	-2.89	-3.25	-3.16	-2.86	-1.71	-0.66	-0.22
50_Benzene-Ethyne_CH-pi	-1.99	-2.88	-3.12	-3.00	-2.71	-1.65	-0.70	-0.22
51_Ethyne-Ethyne_TS	-1.13	-1.43	-1.49	-1.42	-1.29	-0.83	-0.38	-0.11
52_Benzene-AcOH_OH-pi	-4.38	-5.00	-5.09	-4.86	-4.47	-3.05	-1.43	-0.44
53_Benzene-AcNH2_NH-pi	-3.82	-4.37	-4.48	-4.33	-4.02	-2.84	-1.38	-0.37
54_Benzene-Water_OH-pi	-3.76	-4.10	-4.06	-3.80	-3.43	-2.28	-1.12	-0.39
55_Benzene-MeOH_OH-pi	-4.10	-4.72	-4.81	-4.59	-4.20	-2.81	-1.29	-0.42
56_Benzene-MeNH2_NH-pi	-2.88	-3.51	-3.63	-3.42	-3.02	-1.74	-0.64	-0.18
57_Benzene-Peptide_NH-pi	-3.81	-5.19	-5.54	-5.32	-4.82	-3.03	-1.31	-0.46
58_Pyridine-Pyridine_CH-N	-2.13	-3.38	-3.90	-3.69	-3.24	-1.87	-0.76	-0.19
59_Ethyne-Water_CH-O	-2.71	-3.04	-3.11	-3.02	-2.84	-2.13	-1.22	-0.48
60_Ethyne-AcOH_OH-pi	-4.79	-5.28	-5.33	-5.13	-4.77	-3.43	-1.72	-0.51
61_Pentane-AcOH	-3.57	-3.62	-3.40	-3.03	-2.60	-1.41	-0.30	-0.02
62_Pentane-AcNH2	-4.16	-4.39	-4.20	-3.77	-3.25	-1.76	-0.52	-0.08
63_Benzene-AcOH	-3.09	-4.05	-4.20	-3.92	-3.44	-1.90	-0.62	-0.15
64_Peptide-Ethene	-3.00	-3.34	-3.31	-3.07	-2.72	-1.64	-0.58	-0.11
65_Pyridine-Ethyne	-3.02	-3.49	-3.65	-3.62	-3.46	-2.68	-1.53	-0.57
66_MeNH2-Pyridine	-3.55	-4.03	-4.12	-3.96	-3.66	-2.51	-1.15	-0.34

MUE (S66x8)

0.30

CCSD(T) reference values

01_Water-Water	-4.57	-4.88	-4.89	-4.72	-4.45	-3.46	-2.11	-0.87
02_Water-MeOH	-5.19	-5.55	-5.57	-5.38	-5.08	-3.94	-2.39	-0.95
03_Water-MeNH2	-6.49	-6.87	-6.88	-6.65	-6.28	-4.92	-2.98	-1.14
04_Water-Peptide	-7.63	-8.06	-8.08	-7.84	-7.46	-5.99	-3.83	-1.44
05_MeOH-MeOH	-5.28	-5.69	-5.75	-5.58	-5.28	-4.14	-2.52	-1.01
06_MeOH-MeNH2	-6.97	-7.47	-7.54	-7.34	-6.97	-5.50	-3.35	-1.27
07_MeOH-Peptide	-7.63	-8.14	-8.22	-8.02	-7.66	-6.18	-3.65	-1.10
08_MeOH-Water	-4.59	-4.95	-5.00	-4.85	-4.59	-3.59	-2.20	-0.91
09_MeNH2-MeOH	-2.81	-3.04	-3.04	-2.90	-2.69	-1.98	-1.10	-0.39
10_MeNH2-MeNH2	-3.70	-4.09	-4.15	-4.01	-3.75	-2.79	-1.31	-0.39
11_MeNH2-Peptide	-4.95	-5.36	-5.41	-5.24	-4.94	-3.21	-1.40	-0.46
12_MeNH2-Water	-6.75	-7.20	-7.25	-7.03	-6.66	-5.22	-3.15	-1.20
13_Peptide-MeOH	-5.67	-6.11	-6.18	-6.03	-5.74	-4.60	-2.95	-1.30
14_Peptide-MeNH2	-6.81	-7.34	-7.45	-7.29	-6.97	-5.61	-3.55	-1.49
15_Peptide-Peptide	-8.02	-8.53	-8.62	-8.45	-8.10	-6.66	-4.42	-1.77
16_Peptide-Water	-4.70	-5.06	-5.12	-4.98	-4.74	-3.79	-2.45	-1.13
17_Uracil-Uracil_BP	-15.70	-16.90	-17.18	-16.86	-16.17	-13.15	-8.35	-3.33
18_Water-Pyridine	-6.39	-6.80	-6.83	-6.62	-6.27	-4.93	-3.01	-1.19
19_MeOH-Pyridine	-6.78	-7.31	-7.40	-7.23	-6.88	-5.48	-3.38	-1.33
20_AcOH-AcOH	-17.45	-18.79	-19.09	-18.72	-17.94	-14.56	-9.21	-3.59
21_AcNH2-AcNH2	-14.93	-16.03	-16.26	-15.94	-15.26	-12.42	-8.00	-3.00
22_AcOH-Uracil	-17.90	-19.19	-19.49	-19.15	-18.41	-15.14	-9.86	-4.15
23_AcNH2-Uracil	-17.70	-18.89	-19.19	-18.90	-18.22	-15.21	-10.23	-4.65
24_Benzene-Benzene_pi-pi	-0.23	-2.07	-2.74	-2.80	-2.58	-1.54	-0.49	-0.06
25_Pyridine-Pyridine_pi-pi	-1.32	-3.17	-3.83	-3.86	-3.58	-2.36	-0.96	-0.24

26_Uracil-Uracil_pi-pi	-7.86	-9.50	-9.82	-9.43	-8.69	-6.07	-3.10	-0.99
27_Benzene-Pyridine_pi-pi	-0.73	-2.69	-3.37	-3.40	-3.13	-1.97	-0.72	-0.15
28_Benzene-Uracil_pi-pi	-3.53	-5.22	-5.71	-5.55	-5.08	-3.29	-1.36	-0.25
29_Pyridine-Uracil_pi-pi	-3.74	-6.17	-6.81	-6.58	-5.97	-3.87	-1.78	-0.53
30_Benzene-Ethene	-0.26	-1.13	-1.41	-1.40	-1.26	-0.69	-0.17	0.01
31_Uracil-Ethene	-2.57	-3.24	-3.38	-3.24	-2.96	-1.99	-0.93	-0.25
32_Uracil-Ethyne	-2.79	-3.56	-3.74	-3.59	-3.29	-2.23	-1.04	-0.27
33_Pyridine-Ethene	-0.89	-1.62	-1.86	-1.83	-1.67	-1.03	-0.37	-0.05
34_Pentane-Pentane	-2.86	-3.62	-3.77	-3.61	-3.30	-2.24	-1.05	-0.27
35_Neopentane-Pentane	-1.85	-2.49	-2.61	-2.48	-2.25	-1.50	-0.70	-0.18
36_Neopentane-Neopentane	-1.45	-1.74	-1.77	-1.68	-1.54	-1.05	-0.50	-0.14
37_Cyclopentane-Neopentane	-1.61	-2.25	-2.41	-2.33	-2.14	-1.47	-0.71	-0.19
38_Cyclopentane-Cyclopentane	-2.24	-2.84	-3.00	-2.86	-2.60	-1.71	-0.79	-0.20
39_Benzene-Cyclopentane	-2.16	-3.24	-3.57	-3.47	-3.17	-2.07	-0.90	-0.19
40_Benzene-Neopentane	-1.86	-2.68	-2.89	-2.80	-2.57	-1.70	-0.77	-0.19
41_Uracil-Pentane	-3.83	-4.69	-4.84	-4.60	-4.08	-2.44	-0.98	-0.21
42_Uracil-Cyclopentane	-3.05	-3.97	-4.13	-3.92	-3.54	-2.29	-1.02	-0.25
43_Uracil-Neopentane	-2.87	-3.60	-3.70	-3.49	-3.15	-2.05	-0.92	-0.23
44_Ethene-Pentane	-1.66	-1.97	-1.99	-1.86	-1.67	-1.09	-0.49	-0.12
45_Ethyne-Pentane	-1.13	-1.63	-1.75	-1.68	-1.52	-0.98	-0.42	-0.10
46_Peptide-Pentane	-3.75	-4.21	-4.24	-4.05	-3.73	-2.63	-1.19	-0.29
47_Benzene-Benzene_TS	-1.64	-2.58	-2.87	-2.82	-2.60	-1.77	-0.83	-0.22
48_Pyridine-Pyridine_TS	-2.53	-3.31	-3.53	-3.45	-3.22	-2.29	-1.18	-0.37
49_Benzene-Pyridine_TS	-2.09	-3.04	-3.32	-3.26	-3.02	-2.10	-1.05	-0.33
50_Benzene-Ethyne_CH-pi	-1.87	-2.65	-2.86	-2.79	-2.58	-1.79	-0.89	-0.27
51_Ethyne-Ethyne_TS	-1.21	-1.47	-1.52	-1.46	-1.35	-0.93	-0.46	-0.14
52_Benzene-AcOH_OH-pi	-3.97	-4.58	-4.70	-4.55	-4.25	-3.13	-1.70	-0.55
53_Benzene-AcNH ₂ _NH-pi	-3.79	-4.27	-4.36	-4.23	-3.97	-2.97	-1.65	-0.48
54_Benzene-Water_OH-pi	-2.80	-3.21	-3.27	-3.13	-2.90	-2.10	-1.16	-0.42
55_Benzene-MeOH_OH-pi	-3.45	-4.05	-4.19	-4.07	-3.80	-2.80	-1.53	-0.52
56_Benzene-MeNH ₂ _NH-pi	-2.51	-3.09	-3.23	-3.12	-2.86	-1.95	-0.94	-0.26
57_Benzene-Peptide_NH-pi	-3.75	-4.96	-5.28	-5.14	-4.78	-3.41	-1.80	-0.62
58_Pyridine-Pyridine_CH-N	-2.80	-3.82	-4.15	-3.89	-3.45	-2.18	-1.01	-0.28
59_Ethyne-Water_CH-O	-2.52	-2.80	-2.85	-2.76	-2.60	-1.98	-1.17	-0.46
60_Ethyne-AcOH_OH-pi	-4.28	-4.78	-4.86	-4.69	-4.39	-3.26	-1.77	-0.56
61_Pentane-AcOH	-2.70	-2.90	-2.88	-2.73	-2.52	-1.79	-0.78	-0.17
62_Pentane-AcNH ₂	-3.15	-3.50	-3.51	-3.32	-3.04	-2.12	-1.04	-0.27
63_Benzene-AcOH	-2.73	-3.60	-3.80	-3.65	-3.34	-2.24	-1.02	-0.26
64_Peptide-Ethene	-2.60	-2.95	-2.99	-2.85	-2.63	-1.86	-0.88	-0.19
65_Pyridine-Ethyne	-3.54	-3.91	-3.99	-3.89	-3.68	-2.84	-1.67	-0.62
66_MeNH ₂ -Pyridine	-3.45	-3.88	-3.97	-3.85	-3.62	-2.71	-1.50	-0.49
MUE (S66x8)					0.00			

Table S42. Barrier heights (kcal/mol) of the TMBH22 database

Reaction	PBE	B3LYP	M11	revM11	Ref.
Mo_W					
reaction 1	15.9	24.4	24.9	23.1	25.0
reaction 2	5.9	6.9	8.6	7.8	8.5
reaction 3a	19.6	18.8	18.6	20.9	18.7
reaction 3b	3.7	17.2	25.2	24.3	13.5
reaction 4a_Mo	-0.6	0.8	1.8	0.9	0.1
reaction_4b_Mo	15.7	16.4	15.0	19.1	14.7
reaction 4a_W	-0.9	0.1	0.5	0.0	-0.6
reaction 4b	17.7	19.4	18.7	23.2	14.7
reaction 5a	5.5	1.7	2.8	4.6	4.3
reaction 5b	28.1	23.6	19.1	21.7	19.8
reaction 6_Mo	11.7	22.8	29.0	30.5	17.4
reaction 6_W	9.0	21.0	27.3	28.7	15.8
reaction 7	5.9	9.8	10.8	7.7	8.6
Zr					
reaction 1	5.3	9.9	3.7	4.3	3.4
reaction 2	6.7	11.7	5.1	5.5	4.1
reaction 3	2.1	6.1	-1.5	-0.2	-2.0
reaction 4	9.4	18.1	11.9	10.3	10.2
Re					
reaction 1a	23.3	28.0	27.1	21.7	24.2
reaction 2a	8.7	13.2	12.8	10.1	11.4
reaction 3a	33.8	36.2	38.7	40.7	37.1
reaction 3b	14.7	17.0	15.0	18.5	17.0
reaction 3c	15.2	20.4	21.9	21.7	20.9
MUE (TMBH22)	3.5	3.1	2.7	3.3	0.0

Table S43. Reaction energies^a (kcal/mol) of the WCCR10/18 database

Reaction	Metal	PBE	B3LYP	M11	revM11	Ref.
Reaction 1	Pt	19.5	20.7	27.5	26.7	25.9
Reaction 2	Cu	43.3	42.5	51.0	54.1	47.6
Reaction 3	Cu	43.7	42.9	51.8	54.8	48.2
Reaction 4	Ru	27.2	22.9	49.4	48.7	30.9
Reaction 5	Pt	34.8	28.4	35.2	39.9	44.5
Reaction 6	Au	63.2	58.5	58.3	65.7	48.2
Reaction 7	Cu	61.2	54.7	49.9	57.9	50.1
Reaction 8	Ag	49.2	44.5	42.2	48.7	44.6
Reaction 9	Pd	35.1	26.3	28.9	34.6	38.7
Reaction 10	Pt	17.6	13.9	19.7	23.5	22.8
MUE		6.8	7.6	6.2	7.0	0.0

^aThe reaction energies presented here differ from those presented in our previous work¹ for the following reasons: (a) A conversion factor of 4.184 is now used consistently in going from kJ/mol to kcal//mol and (b) The WCCR10 database is used instead of its subset WCCR9.

Table S44. Cluster binding energies (kcal/mol) of the ASNC2 database

System	PBE	B3LYP	M11	revM11	Ref.
1A1D	-22.8	-21.1	-26.4	-24.5	-24.5
2A1N	-42.3	-40.1	-49.7	-48.6	-46.0
MUE	2.7	4.7	2.8	1.3	0.0

Table S45. Dipole moments (debye) of the DM79 database

System	PBE	B3LYP	M11	revM11	Ref.
acetyl chloride	2.79	2.85	2.91	2.81	2.72
AlF	1.50	1.56	1.43	1.44	1.53
AsH ₃	0.27	0.25	0.27	0.33	0.22
aziridine	1.63	1.69	1.74	1.73	1.89
benzonitrile	4.63	4.71	4.77	4.71	4.18
BF ₃ NH ₃	6.03	6.08	6.23	6.24	5.90
BH ₂ Cl	0.44	0.61	0.74	0.73	0.75
bromoform	0.85	0.88	0.95	0.90	0.99
bromomethane	1.85	1.93	2.01	1.88	1.82
C ₂ HI	0.22	0.14	0.02	0.03	0.03
CH ₃ I	1.65	1.70	1.82	1.60	1.64
CrO	3.54	4.15	4.77	4.52	3.88
CuF	4.56	5.12	6.04	5.60	5.77
CuO	4.22	4.59	5.67	5.21	4.45
dimethyl sulfoxide	3.78	3.99	4.21	4.17	3.96
ethanol	1.51	1.58	1.64	1.64	1.44
FeO	4.27	5.17	6.72	5.88	4.70
fluoroacetylene	0.48	0.62	0.75	0.77	0.72
fluorosilane	1.26	1.32	1.30	1.39	1.27
formic acid (<i>s-cis</i>)	1.49	1.54	1.59	1.57	1.43
formic acid (<i>s-trans</i>)	3.76	3.92	4.11	4.09	3.79
GeO	3.16	3.46	3.54	3.65	3.28
H ₂ CO	2.21	2.36	2.51	2.51	2.33
H ₂ O	1.87	1.91	1.98	1.96	1.85
H ₂ S	1.02	1.02	1.07	1.04	0.98
HCl	1.13	1.14	1.16	1.16	1.11
HCN	2.95	3.03	3.12	3.11	2.99
HF	1.80	1.84	1.89	1.88	1.83
HfO ₂	7.44	7.99	8.61	8.49	7.92
HfO	3.24	3.55	3.47	3.74	3.43
imidazole	3.67	3.72	3.83	3.82	3.80
ketene	1.45	1.49	1.53	1.48	1.42
LaO	3.88	4.25	3.64	3.82	3.21
LiOH	4.27	4.34	4.49	4.40	4.50
methylamine	1.27	1.29	1.30	1.30	1.31
methylphosphine	1.18	1.15	1.23	1.16	1.10
methylsilane	0.84	0.78	0.92	0.72	0.73
MgO	7.04	7.45	9.17	3.20	6.20
N ₂ O	0.14	0.07	0.21	0.26	0.16
NaCl	8.52	8.71	9.07	8.92	9.00
NH ₃	1.54	1.53	1.55	1.55	1.47
NiH	2.15	2.46	3.58	3.13	2.40
nitrobenzene	4.51	4.69	4.68	4.64	4.22
nitromethane	3.41	3.59	3.77	3.74	3.46

nitrosyl hydride	1.60	1.71	1.83	1.84	1.62
PbO	4.40	4.78	5.10	5.00	4.64
PbS	3.90	4.12	4.31	4.32	3.59
PH ₃	0.60	0.59	0.63	0.65	0.57
phenol	1.25	1.28	1.32	1.32	1.22
propyleneimine (<i>cis</i>)	1.75	1.81	1.85	1.83	1.77
propyleneimine (<i>trans</i>)	1.57	1.63	1.68	1.67	1.57
propane	0.88	0.84	0.84	0.76	0.78
ScF	2.19	2.57	1.61	2.79	1.72
ScO	3.47	3.92	3.40	3.79	4.55
SeO ₂	2.65	2.88	3.06	3.06	2.62
SO ₂	1.57	1.71	1.78	1.78	1.63
TiH	2.57	2.73	2.39	2.44	2.46
TiN	3.83	3.50	2.89	3.13	3.56
TiO	3.26	3.65	3.12	3.50	2.96
VN	3.47	3.23	2.77	2.95	3.07
VO	3.24	3.64	3.39	3.64	3.36
YO	4.38	4.83	4.19	4.64	4.52
ZrO ₂	7.39	7.83	8.23	8.12	7.80
ZrO	3.07	3.32	3.90	3.30	2.55
AgBr	4.65	5.18	6.31	5.84	5.62
AgCl	4.91	5.43	6.41	6.02	6.08
AgI	4.13	4.64	6.02	5.39	4.55
HBr	0.89	0.89	0.91	0.90	0.83
HI	0.43	0.43	0.47	0.43	0.45
KF	8.14	8.43	8.70	8.68	8.59
RbCl	10.41	10.78	11.11	10.97	10.51
RbF	8.62	8.94	9.25	9.19	8.55
SrO	7.80	8.67	9.72	9.61	8.90
TIBr	4.45	4.77	4.95	4.64	4.49
TICl	4.46	4.74	4.86	4.61	4.54
TIF	4.05	4.27	4.39	4.28	4.23
TII	4.31	4.65	5.03	4.52	4.61
CrN	2.95	3.16	3.01	2.98	2.31
NaLi	0.21	0.03	2.28	0.49	0.46
MUE	0.24	0.22	0.39	0.30	0.00

Table S46. Interaction energies (kcal/mol) of the PEC4 database at various intermonomer distances (\AA)

System/distance	PBE	B3LYP	M11	revM11	Ref.
<i>d</i> (Ar–Ar)					
3.0	2.59179	3.13251	2.38954	1.86337	2.25887
3.1	1.66072	2.15023	1.53192	1.00760	1.23823
3.2	1.02262	1.47368	0.97008	0.46076	0.59526
3.3	0.58953	1.01290	0.59901	0.12230	0.19976
3.4	0.30111	0.70109	0.36660	-0.06897	-0.03499
3.5	0.11237	0.48972	0.22433	-0.16525	-0.16651
3.6	-0.00812	0.35030	0.11436	-0.21240	-0.23274
3.7	-0.07995	0.25754	0.05498	-0.21706	-0.25856
3.8	-0.11988	0.19393	0.03161	-0.19772	-0.26014
3.9	-0.13938	0.15305	-0.01022	-0.18312	-0.24789
4.0	-0.14398	0.12629	-0.03990	-0.16283	-0.22840
4.1	-0.13952	0.10569	-0.03224	-0.12953	-0.20578
4.2	-0.13089	0.09091	-0.03892	-0.10697	-0.18250
4.3	-0.11920	0.08138	-0.06390	-0.09694	-0.15998
4.4	-0.10569	0.07336	-0.06848	-0.08044	-0.13899
4.5	-0.09274	0.06463	-0.05067	-0.05766	-0.11989
4.6	-0.08103	0.05744	-0.05284	-0.04764	-0.10277
4.7	-0.06941	0.05219	-0.06347	-0.04489	-0.08758
4.8	-0.05849	0.04721	-0.06378	-0.03868	-0.07422
4.9	-0.04943	0.04098	-0.04668	-0.02562	-0.06251
5.0	-0.04233	0.03510	-0.04202	-0.02010	-0.05227
5.1	-0.03551	0.03064	-0.04384	-0.01905	-0.04335
5.2	-0.02884	0.02736	-0.04479	-0.01886	-0.03556
5.3	-0.02320	0.02375	-0.03505	-0.01358	-0.02878
5.4	-0.01949	0.01962	-0.02754	-0.00921	-0.02286
5.5	-0.01677	0.01595	-0.02649	-0.00836	-0.01769
5.6	-0.01380	0.01339	-0.02432	-0.00777	-0.01317
5.7	-0.01052	0.01187	-0.02154	-0.00750	-0.00921
5.8	-0.00775	0.01036	-0.01635	-0.00511	-0.00574
5.9	-0.00627	0.00844	-0.01417	-0.00372	-0.00269
6.0	-0.00571	0.00645	-0.01336	-0.00356	0.00000
<i>d</i> (Ne–Ne)					
2.5	0.54330	0.67158	0.56224	0.21355	0.50252
2.6	0.23208	0.39840	0.27713	-0.00711	0.22262
2.7	0.05311	0.24222	0.10777	-0.11156	0.06513
2.8	-0.04510	0.15247	0.02360	-0.14492	-0.01942
2.9	-0.09761	0.10259	-0.04824	-0.15577	-0.06118
3.0	-0.11642	0.07568	-0.06083	-0.13408	-0.07841
3.1	-0.12187	0.05894	-0.07667	-0.11489	-0.08204
3.2	-0.11878	0.04890	-0.09669	-0.10084	-0.07855
3.3	-0.10662	0.04303	-0.08003	-0.07200	-0.07170
3.4	-0.09436	0.03803	-0.07679	-0.05643	-0.06362

3.5	-0.08317	0.03349	-0.08859	-0.05241	-0.05546
3.6	-0.06946	0.03001	-0.07306	-0.03633	-0.04780
3.7	-0.05618	0.02723	-0.05754	-0.02400	-0.04090
3.8	-0.04623	0.02379	-0.05740	-0.02249	-0.03485
3.9	-0.03830	0.01991	-0.05637	-0.02170	-0.02960
4.0	-0.03020	0.01668	-0.03955	-0.01265	-0.02511
4.1	-0.02292	0.01490	-0.03287	-0.00930	-0.02128
4.2	-0.01812	0.01247	-0.02915	-0.00827	-0.01802
4.3	-0.01538	0.00930	-0.02548	-0.00856	-0.01526
4.4	-0.01269	0.00687	-0.01865	-0.00580	-0.01291
4.5	-0.00921	0.00639	-0.01489	-0.00388	-0.01092
4.6	-0.00649	0.00606	-0.01218	-0.00284	-0.00922
4.7	-0.00558	0.00461	-0.01012	-0.00314	-0.00777
4.8	-0.00561	0.00270	-0.00807	-0.00302	-0.00654
4.9	-0.00487	0.00194	-0.00638	-0.00225	-0.00547
5.0	-0.00314	0.00240	-0.00493	-0.00123	-0.00456
5.1	-0.00163	0.00281	-0.00377	-0.00073	-0.00377
5.2	-0.00132	0.00227	-0.00296	-0.00090	-0.00309
5.3	-0.00192	0.00111	-0.00246	-0.00115	-0.00250
5.4	-0.00225	0.00035	-0.00217	-0.00109	-0.00199
5.5	-0.00172	0.00046	-0.00185	-0.00069	-0.00154
5.6	-0.00068	0.00102	-0.00129	-0.00021	-0.00115
5.7	0.00009	0.00133	-0.00058	0.00003	-0.00080
5.8	0.00012	0.00107	-0.00025	-0.00007	-0.00050
5.9	-0.00037	0.00047	-0.00044	-0.00032	-0.00023
6.0	-0.00080	0.00000	-0.00073	-0.00045	0.00000
d (Kr–Kr)					
3.0	7.66852	8.89938	7.45673	6.78409	8.02942
3.1	5.31975	6.35525	5.00032	4.41930	5.07071
3.2	3.63595	4.52203	3.29963	2.75393	3.08920
3.3	2.43972	3.21125	2.14538	1.60612	1.77660
3.4	1.59611	2.27780	1.37699	0.83100	0.91974
3.5	1.00233	1.61499	0.86529	0.32292	0.37157
3.6	0.59105	1.14876	0.53235	0.00424	0.03097
3.7	0.31055	0.82139	0.31798	-0.17927	-0.17140
3.8	0.12128	0.59356	0.17321	-0.27617	-0.28282
3.9	-0.00275	0.43606	0.07766	-0.31550	-0.33542
4.0	-0.08072	0.32529	0.03705	-0.31229	-0.35084
4.1	-0.12767	0.24849	0.01349	-0.29084	-0.34356
4.2	-0.15292	0.19647	-0.01475	-0.26444	-0.32317
4.3	-0.16233	0.15900	-0.01775	-0.22596	-0.29591
4.4	-0.16172	0.13101	-0.00816	-0.18357	-0.26578
4.5	-0.15539	0.11193	-0.02849	-0.15649	-0.23525
4.6	-0.14512	0.09821	-0.05595	-0.13674	-0.20580
4.7	-0.13235	0.08567	-0.04946	-0.10733	-0.17826
4.8	-0.11884	0.07476	-0.03696	-0.07998	-0.15302

4.9	-0.10536	0.06707	-0.04971	-0.06915	-0.13021
5.0	-0.09226	0.06100	-0.06885	-0.06488	-0.10979
5.1	-0.08020	0.05418	-0.06532	-0.05196	-0.09164
5.2	-0.06956	0.04701	-0.04977	-0.03591	-0.07558
5.3	-0.05978	0.04155	-0.04934	-0.02966	-0.06141
5.4	-0.05040	0.03763	-0.05509	-0.02886	-0.04893
5.5	-0.04234	0.03381	-0.05730	-0.02750	-0.03796
5.6	-0.03601	0.02910	-0.04608	-0.01994	-0.02832
5.7	-0.03102	0.02456	-0.03860	-0.01433	-0.01983
5.8	-0.02624	0.02126	-0.03805	-0.01273	-0.01237
5.9	-0.02136	0.01907	-0.03728	-0.01219	-0.00580
6.0	-0.01721	0.01711	-0.03449	-0.01167	0.00000
<i>d</i> (Kr–He)					
3.0	0.29664	0.56804	0.48402	0.12273	0.32571
3.1	0.13673	0.39844	0.33164	0.01862	0.16589
3.2	0.03185	0.28335	0.21823	-0.03765	0.06652
3.3	-0.03089	0.20632	0.14795	-0.06069	0.00660
3.4	-0.06602	0.15577	0.08893	-0.06713	-0.02784
3.5	-0.08538	0.12080	0.03716	-0.06818	-0.04605
3.6	-0.09136	0.09632	0.01897	-0.05853	-0.05411
3.7	-0.08926	0.08035	-0.00026	-0.05096	-0.05599
3.8	-0.08499	0.06795	-0.02466	-0.04811	-0.05426
3.9	-0.07809	0.05727	-0.02300	-0.03834	-0.05056
4.0	-0.06818	0.05019	-0.01897	-0.03025	-0.04594
4.1	-0.05829	0.04521	-0.02859	-0.02896	-0.04101
4.2	-0.05101	0.03888	-0.03098	-0.02552	-0.03614
4.3	-0.04472	0.03228	-0.02197	-0.01935	-0.03156
4.4	-0.03696	0.02853	-0.02116	-0.01630	-0.02735
4.5	-0.02871	0.02661	-0.02540	-0.01507	-0.02355
4.6	-0.02310	0.02342	-0.02412	-0.01226	-0.02017
4.7	-0.02061	0.01846	-0.01881	-0.00988	-0.01719
4.8	-0.01840	0.01433	-0.01717	-0.00825	-0.01457
4.9	-0.01435	0.01265	-0.01962	-0.00722	-0.01227
5.0	-0.00952	0.01239	-0.01925	-0.00557	-0.01027
5.1	-0.00633	0.01136	-0.01326	-0.00357	-0.00852
5.2	-0.00591	0.00870	-0.00862	-0.00262	-0.00700
5.3	-0.00680	0.00550	-0.00949	-0.00262	-0.00567
5.4	-0.00673	0.00349	-0.01252	-0.00288	-0.00451
5.5	-0.00477	0.00340	-0.01131	-0.00245	-0.00349
5.6	-0.00190	0.00434	-0.00517	-0.00097	-0.00260
5.7	0.00002	0.00482	-0.00064	0.00029	-0.00182
5.8	-0.00010	0.00390	-0.00145	0.00050	-0.00114
5.9	-0.00175	0.00185	-0.00520	-0.00042	-0.00053
6.0	-0.00334	-0.00006	-0.00686	-0.00158	0.00000

Table S47. Internuclear distances (Å) of the TSG48 database

System	PBE	B3LYP	M11	revM11	Ref.
R1	NA ^a	2.535	2.552	2.548	2.530
	NA	1.325	1.372	1.376	1.341
	NA	1.213	1.189	1.182	1.192
R2	1.469	1.202	1.200	1.218	1.215
	0.804	0.907	0.919	0.892	0.894
	2.272	2.109	2.119	2.110	2.109
R3	1.380	1.380	1.418	1.402	1.426
	2.889	2.792	2.620	2.663	2.578
	1.511	1.415	1.204	1.265	1.160
R4	1.484	1.471	1.428	1.432	1.431
	2.302	2.282	2.215	2.195	2.187
	1.227	1.223	1.217	1.217	1.226
R5	1.493	1.489	1.490	1.482	1.480
	2.986	2.979	2.980	2.964	2.960
	1.493	1.489	1.490	1.482	1.480
R6	NA	1.717	1.754	1.763	1.767
	NA	4.103	3.852	3.830	3.814
	NA	2.387	2.098	2.067	2.047
R7	2.336	2.355	2.312	2.292	2.305
	4.672	4.710	4.624	4.584	4.610
	2.336	2.355	2.312	2.292	2.305
R8	2.552	2.157	2.005	1.995	2.020
	4.354	4.234	4.155	4.123	4.134
	1.878	2.077	2.150	2.128	2.114
R9	1.756	1.771	1.763	1.743	1.758
	3.842	3.821	3.766	3.729	3.745
	2.087	2.051	2.004	1.987	1.988
R10	1.119	1.110	1.106	1.108	1.127

	2.353	2.285	2.207	2.228	2.201
	1.583	1.526	1.461	1.478	1.439
R11	1.339	1.333	1.332	1.333	1.351
	2.242	2.210	2.041	1.996	1.925
	2.941	2.875	2.724	2.686	2.662
R12	1.193	1.180	1.170	1.174	1.187
	1.201	1.190	1.177	1.173	1.183
	1.376	1.388	1.444	1.438	1.387
R13	1.404	1.345	1.296	1.278	1.289
	2.549	2.528	2.509	2.506	2.501
	1.168	1.201	1.253	1.266	1.240
R14	1.276	1.302	1.293	1.274	1.301
	2.745	2.722	2.711	2.703	2.697
	1.468	1.421	1.421	1.435	1.398
R15	0.974	1.288	1.254	1.256	1.246
	2.872	2.349	2.345	2.336	2.337
	2.031	1.093	1.108	1.098	1.103
R16	1.209	1.205	1.247	1.265	1.248
	2.516	2.506	2.495	2.492	2.490
	1.309	1.302	1.259	1.239	1.248
MUE	[0.141] ^b	0.065	0.020	0.019	0.000

^aNA = not available. The saddle points could not be located in these cases.

^bThe MUE for the PBE functional cannot be calculated for all 16 reactions because we could not locate a transition structure for two of the reactions. The value listed is obtained by averaging over 14 transition structures (42 internuclear distances excluding those for R1 and R6). For the other three functionals we include all 16 reactions (48 internuclear distances).

Table S48. Bond lengths (\AA) of the MGBL193 database

System	PBE	B3LYP	M11	revM11	Ref.
aziridine					
<i>r</i> (1,2)	1.4852	1.4800	1.4805	1.4788	1.4777
<i>r</i> (1,3)	1.4776	1.4700	1.4578	1.4527	1.4714
<i>r</i> (1,4)	1.0918	1.0830	1.0847	1.0840	1.0805
<i>r</i> (1,5)	1.0908	1.0819	1.0838	1.0829	1.0791
<i>r</i> (3,8)	1.0230	1.0134	1.0104	1.0134	1.0126
benzene					
<i>r</i> (1,2)	1.3974	1.0819	1.0844	1.3861	1.3919
<i>r</i> (1,7)	1.0909	1.0819	1.0844	1.0836	1.0795
butadiene					
<i>r</i> (1,2)	1.3443	1.3344	1.3263	1.3253	1.3386
<i>r</i> (1,5)	1.0895	1.0808	1.0838	1.0829	1.0793
<i>r</i> (1,6)	1.0920	1.0831	1.0862	1.0852	1.0815
<i>r</i> (2,3)	1.4514	1.4527	1.4614	1.4645	1.4543
<i>r</i> (2,7)	1.0952	1.0858	1.0881	1.0868	1.0839
CCl ₂ F ₂					
<i>r</i> (1,2)	1.7824	1.7794	1.7637	1.7595	1.7519
<i>r</i> (1,3)	1.3447	1.3342	1.3365	1.3304	1.3286
CH ₂ CHF					
<i>r</i> (1,2)	1.3265	1.3172	1.3121	1.3126	1.3234
<i>r</i> (1,3)	1.0875	1.0790	1.0812	1.0806	1.0768
<i>r</i> (1,4)	1.0886	1.0801	1.0826	1.0820	1.0782
<i>r</i> (2,5)	1.0910	1.0811	1.0848	1.0834	1.0784
<i>r</i> (2,6)	1.3561	1.3494	1.3508	1.3437	1.3412
CH ₂ Cl ₂					
<i>r</i> (2,3)	1.7855	1.7862	1.7733	1.7655	1.7642
<i>r</i> (2,4)	1.0916	1.0817	1.0848	1.0836	1.0810
CH ₂ ClF					
<i>r</i> (1,2)	1.0957	1.0855	1.0888	1.0870	1.0842
<i>r</i> (1,4)	1.3738	1.3654	1.3685	1.3605	1.3591
<i>r</i> (1,5)	1.7910	1.7908	1.7763	1.7695	1.7645
CH ₂ F ₂					
<i>r</i> (1,2)	1.3723	1.3633	1.3645	1.3576	1.3533
<i>r</i> (2,4)	1.1004	1.0899	1.0931	1.0907	1.0867
CHClF ₂					
<i>r</i> (1,2)	1.7884	1.7847	1.7684	1.7639	1.7558
<i>r</i> (2,4)	1.3534	1.3435	1.3457	1.3395	1.3363
<i>r</i> (2,5)	1.0969	1.0864	1.0905	1.0879	1.0849
cis-CHFCHCl					
<i>r</i> (1,2)	1.3325	1.3215	1.3154	1.3162	1.3266
<i>r</i> (1,3)	1.0869	1.0774	1.0799	1.0794	1.0795
<i>r</i> (1,4)	1.7264	1.7295	1.7214	1.7182	1.7124
<i>r</i> (2,5)	1.0901	1.0805	1.0838	1.0827	1.0776
<i>r</i> (2,6)	1.3447	1.3379	1.3391	1.3325	1.3306

<i>cis</i> -acrolein					
	<i>r</i> (1,2)	1.4837	1.4817	1.4879	1.4901
	<i>r</i> (1,3)	1.1184	1.1070	1.1079	1.1050
	<i>r</i> (1,4)	1.2205	1.2095	1.2037	1.2102
	<i>r</i> (2,5)	1.0930	1.0837	1.0853	1.0845
	<i>r</i> (2,6)	1.3404	1.3310	1.3234	1.3228
	<i>r</i> (6,7)	1.0917	1.0823	1.0863	1.0845
	<i>r</i> (6,8)	1.0897	1.0811	1.0841	1.0829
<i>cis</i> -formic acid					
	<i>r</i> (1,2)	1.3619	1.3520	1.3471	1.3442
	<i>r</i> (1,4)	0.9751	0.9652	0.9610	0.9641
	<i>r</i> (2,3)	1.2013	1.1910	1.1875	1.1885
	<i>r</i> (2,5)	1.1132	1.1024	1.1051	1.1021
<i>cis</i> -hexatriene					
	<i>r</i> (1,2)	1.3613	1.3489	1.3360	1.3344
	<i>r</i> (2,3)	1.4451	1.4476	1.4581	1.4620
	<i>r</i> (2,8)	1.0943	1.0850	1.0875	1.0866
	<i>r</i> (3,5)	1.3475	1.3370	1.3281	1.3268
	<i>r</i> (3,9)	1.0932	1.0836	1.0857	1.0844
	<i>r</i> (5,11)	1.0921	1.0831	1.0863	1.0852
	<i>r</i> (5,12)	1.0894	1.0807	1.0838	1.0829
<i>cis</i> -methyl formate					
	<i>r</i> (1,2)	1.4485	1.4417	1.4356	1.4279
	<i>r</i> (1,3)	1.3499	1.3382	1.3326	1.3310
	<i>r</i> (2,6)	1.0935	1.0849	1.0876	1.0865
	<i>r</i> (2,7)	1.0971	1.0881	1.0913	1.0900
	<i>r</i> (3,4)	1.2102	1.2002	1.1972	1.1977
	<i>r</i> (3,8)	1.1068	1.0967	1.0995	1.0965
cyclobutene					
	<i>r</i> (1,2)	1.5708	1.5700	1.5648	1.5603
	<i>r</i> (1,4)	1.5162	1.5154	1.5139	1.5130
	<i>r</i> (1,5)	1.1004	1.0914	1.0919	1.0912
	<i>r</i> (3,4)	1.3455	1.3349	1.3284	1.3279
	<i>r</i> (3,9)	1.0911	1.0819	1.0843	1.0832
cyclopropane					
	<i>r</i> (1,2)	1.5088	1.0807	1.5012	1.4975
	<i>r</i> (1,4)	1.0892	1.0807	1.0820	1.0818
dimethyl ether					
	<i>r</i> (1,2)	1.4174	1.4116	1.4082	1.4017
	<i>r</i> (1,4)	1.0968	1.0881	1.0905	1.0889
	<i>r</i> (1,5)	1.1067	1.0971	1.0994	1.0982
dioxirane					
	<i>r</i> (1,2)	1.5094	1.4985	1.4576	1.4523
	<i>r</i> (1,3)	1.3977	1.3879	1.3834	1.3787
	<i>r</i> (3,4)	1.0982	1.0879	1.0900	1.0877
ethene					

	<i>r</i> (1,4)	1.3329	1.3248	1.3203	1.3199	1.3317
	<i>r</i> (4,5)	1.0913	1.0825	1.0851	1.0839	1.0805
ethenol	<i>r</i> (1,2)	1.3346	1.3250	1.3200	1.3196	1.3316
	<i>r</i> (1,4)	1.0868	1.0784	1.0807	1.0804	1.0770
	<i>r</i> (1,5)	1.0892	1.0807	1.0829	1.0825	1.0812
	<i>r</i> (2,3)	1.3738	1.3681	1.3633	1.3612	1.3598
	<i>r</i> (2,6)	1.0939	1.0843	1.0877	1.0863	1.0798
	<i>r</i> (3,7)	0.9691	0.9601	0.9568	0.9596	0.9605
furan	<i>r</i> (1,2)	1.3644	1.3547	1.3504	1.3479	1.3542
	<i>r</i> (1,5)	1.4334	1.4322	1.4312	1.4336	1.4344
	<i>r</i> (1,6)	1.0848	1.0760	1.0784	1.0780	1.0743
	<i>r</i> (2,3)	1.3693	1.3618	1.3532	1.3509	1.3598
	<i>r</i> (2,7)	1.0835	1.0746	1.0782	1.0775	1.0739
glycolaldehyde	<i>r</i> (1,2)	1.5039	1.5031	1.5058	1.5075	1.5014
	<i>r</i> (1,4)	1.2183	1.2065	1.2022	1.2025	1.2083
	<i>r</i> (1,5)	1.1156	1.1055	1.1064	1.1035	1.1011
	<i>r</i> (2,3)	1.4030	1.3997	1.3953	1.3905	1.3970
	<i>r</i> (2,6)	1.1097	1.0992	1.0998	1.0984	1.0964
	<i>r</i> (3,8)	0.9813	0.9690	0.9662	0.9685	0.9618
glyoxylic acid	<i>r</i> (1,2)	1.5384	1.5341	1.5369	1.5391	1.5211
	<i>r</i> (1,5)	1.2070	1.1963	1.1907	1.1916	1.2087
	<i>r</i> (1,6)	1.1147	1.1032	1.1052	1.1020	1.0963
	<i>r</i> (2,3)	1.3470	1.3360	1.3293	1.3273	1.3317
	<i>r</i> (2,4)	1.2151	1.2035	1.1979	1.1987	1.1977
	<i>r</i> (3,7)	0.9793	0.9694	0.9659	0.9691	0.9697
H_2CCCH_2	<i>r</i> (1,2)	1.3078	1.0831	1.2971	1.2984	1.3075
	<i>r</i> (2,4)	1.0920	1.0831	1.0845	1.0838	1.0800
	<i>r</i> (1,3)					
H_2CO	<i>r</i> (1,2)	1.2096	1.2002	1.1969	1.1982	1.2051
	<i>r</i> (1,3)	1.1173	1.1058	1.1076	1.1036	1.1002
H_2O	<i>r</i> (1,2)	0.9703	0.9618	0.9581	0.9610	0.9572
	<i>r</i> (1,3)					
HCCCC	<i>r</i> (1,2)	1.0699	1.0614	1.0658	1.0660	1.0610
	<i>r</i> (2,3)	1.2174	1.2042	1.1967	1.1960	1.2070
	<i>r</i> (3,4)	1.3608	1.3638	1.3776	1.3858	1.3726
HCCH	<i>r</i> (1,2)	1.0703	1.0616	1.0656	1.0658	1.0611
	<i>r</i> (2,3)	1.2072	1.1963	1.1920	1.1920	1.2036
	<i>r</i> (1,3)					
HCN	<i>r</i> (1,2)	1.1583	1.1460	1.1396	1.1414	1.1536
	<i>r</i> (1,3)					

	<i>r</i> (1,3)	1.0749	1.0656	1.0690	1.0685	1.0645
HCO^+	<i>r</i> (1,2)	1.1038	1.0953	1.1009	1.1003	1.0916
	<i>r</i> (2,3)	1.1138	1.1020	1.0967	1.0991	1.1057
	<i>r</i> (1,2)	1.0213	1.0148	1.0193	1.0216	1.0138
HNCCN^+	<i>r</i> (2,3)	1.1511	1.1382	1.1312	1.1321	1.1392
	<i>r</i> (3,4)	1.3631	1.3648	1.3761	1.3827	1.3735
	<i>r</i> (4,5)	1.1711	1.1555	1.1453	1.1465	1.1607
	<i>r</i> (1,2)	1.0056	0.9969	0.9979	1.0007	0.9946
	<i>r</i> (2,3)	1.1759	1.1644	1.1596	1.1610	1.1688
imidazole	<i>r</i> (1,2)	1.3768	1.3670	1.3635	1.3601	1.3627
	<i>r</i> (1,4)	1.3791	1.3750	1.3693	1.3718	1.3794
	<i>r</i> (1,6)	1.0854	1.0763	1.0794	1.0786	1.0752
	<i>r</i> (2,3)	1.3800	1.3763	1.3701	1.3716	1.3743
	<i>r</i> (2,7)	1.0837	1.0747	1.0774	1.0771	1.0764
	<i>r</i> (3,5)	1.3688	1.3627	1.3574	1.3568	1.3616
	<i>r</i> (3,8)	1.0132	1.0046	1.0042	1.0070	1.0011
	<i>r</i> (4,5)	1.3206	1.3100	1.3043	1.3023	1.3103
	<i>r</i> (5,9)	1.0859	1.0772	1.0807	1.0799	1.0770
	<i>r</i> (1,2)	1.2019	1.1901	1.1836	1.1848	1.1896
maleic anhydride	<i>r</i> (1,3)	1.4038	1.3912	1.3824	1.3771	1.3843
	<i>r</i> (1,4)	1.4878	1.4871	1.4952	1.4960	1.4857
	<i>r</i> (4,6)	1.3386	1.3285	1.3196	1.3191	1.3320
	<i>r</i> (4,8)	1.0867	1.0776	1.0806	1.0795	1.0761
	<i>r</i> (1,2)	1.0220	1.0132	1.0102	1.0133	1.0111
oxirane	<i>r</i> (1,3)	1.4376	1.4296	1.4193	1.4109	1.4281
	<i>r</i> (2,3)	1.4693	1.4632	1.4629	1.4632	1.4615
	<i>r</i> (2,4)	1.0939	1.0844	1.0865	1.0853	1.0814
propanal	<i>r</i> (1,2)	1.2145	1.2046	1.2003	1.2011	1.2075
	<i>r</i> (1,3)	1.5075	1.5052	1.5054	1.5052	1.5037
	<i>r</i> (1,5)	1.1240	1.1119	1.1134	1.1099	1.1040
	<i>r</i> (3,4)	1.5348	1.5336	1.5308	1.5286	1.5165
	<i>r</i> (3,6)	1.1051	1.0959	1.0962	1.0951	1.0946
	<i>r</i> (4,8)	1.0974	1.0892	1.0905	1.0899	1.0879
	<i>r</i> (4,9)	1.0984	1.0902	1.0917	1.0911	1.0888
	<i>r</i> (1,2)	1.4967	1.4966	1.4981	1.4984	1.4956
propene	<i>r</i> (1,5)	1.1016	1.0931	1.0936	1.0928	1.0895
	<i>r</i> (1,6)	1.0984	1.0900	1.0915	1.0908	1.0880

<i>r</i> (2,3)	1.3359	1.3271	1.3221	1.3215	1.3326
<i>r</i> (2,7)	1.0954	1.0863	1.0883	1.0872	1.0841
<i>r</i> (3,8)	1.0903	1.0816	1.0843	1.0835	1.0804
<i>r</i> (3,9)	1.0927	1.0837	1.0866	1.0856	1.0818
pyrazole					
<i>r</i> (1,2)	1.4129	1.4098	1.4085	1.4095	1.4090
<i>r</i> (1,4)	1.3394	1.3275	1.3201	1.3183	1.3289
<i>r</i> (1,6)	1.0860	1.0772	1.0801	1.0793	1.0757
<i>r</i> (2,3)	1.3851	1.3764	1.3719	1.3697	1.3765
<i>r</i> (2,7)	1.0839	1.0751	1.0774	1.0771	1.0739
<i>r</i> (3,5)	1.3600	1.3537	1.3485	1.3482	1.3519
<i>r</i> (3,8)	1.0848	1.0759	1.0787	1.0783	1.0745
<i>r</i> (4,5)	1.3501	1.3451	1.3316	1.3330	1.3441
<i>r</i> (5,9)	1.0125	1.0042	1.0047	1.0074	1.0014
pyridazine					
<i>r</i> (1,2)	1.3859	1.3777	1.3710	1.3704	1.3778
<i>r</i> (1,3)	1.3977	1.3924	1.3895	1.3902	1.3926
<i>r</i> (1,7)	1.0904	1.0812	1.0837	1.0826	1.0791
<i>r</i> (3,5)	1.3411	1.3304	1.3218	1.3217	1.3324
<i>r</i> (3,9)	1.0922	1.0826	1.0856	1.0839	1.0804
pyridine					
<i>r</i> (1,2)	1.3949	1.3883	1.3834	1.3834	1.3888
<i>r</i> (1,7)	1.0908	1.0818	1.0846	1.0835	1.0802
<i>r</i> (2,4)	1.3970	1.3905	1.3861	1.3858	1.3907
<i>r</i> (2,8)	1.0904	1.0812	1.0835	1.0826	1.0796
<i>r</i> (4,6)	1.3419	1.3337	1.3275	1.3276	1.3358
<i>r</i> (4,10)	1.0937	1.0841	1.0873	1.0856	1.0818
pyrrole					
<i>r</i> (1,2)	1.4240	1.4214	1.4193	1.4214	1.4228
<i>r</i> (1,6)	1.0851	1.0764	1.0789	1.0787	1.0745
<i>r</i> (2,4)	1.3827	1.3735	1.3692	1.3660	1.3732
<i>r</i> (3,5)	1.3758	1.3713	1.3653	1.3657	1.3694
<i>r</i> (4,9)	1.0842	1.0755	1.0785	1.0783	1.0744
<i>r</i> (5,10)	1.0112	1.0032	1.0033	1.0062	1.0007
SH_3^+					
<i>r</i> (1,2)	1.3713	1.3598	1.3586	1.3559	1.3502
thiophene					
<i>r</i> (1,2)	1.4240	1.4230	1.4223	1.4253	1.4233
<i>r</i> (1,6)	1.0885	1.0797	1.0819	1.0814	1.0792
<i>r</i> (2,4)	1.3736	1.3634	1.3583	1.3552	1.3625
<i>r</i> (4,5)	1.7271	1.7257	1.7150	1.7131	1.7127
<i>r</i> (4,9)	1.0853	1.0765	1.0793	1.0792	1.0772
<i>trans</i> -acrolein					
<i>r</i> (1,2)	1.4718	1.4705	1.4781	1.4799	1.4703
<i>r</i> (1,3)	1.2201	1.2092	1.2023	1.2028	1.2109
<i>r</i> (1,4)	1.1214	1.1092	1.1112	1.1077	1.1044

$r(2,5)$	1.0923	1.0831	1.0852	1.0840	1.0817
$r(2,6)$	1.3409	1.3314	1.3238	1.3233	1.3355
$r(6,7)$	1.0931	1.0838	1.0874	1.0859	1.0826
$r(6,8)$	1.0898	1.0811	1.0841	1.0829	1.0792
<i>trans</i> -formic acid					
$r(1,2)$	1.3551	1.3450	1.3404	1.3375	1.3417
$r(1,4)$	0.9804	0.9703	0.9664	0.9695	0.9656
$r(2,3)$	1.2082	1.1977	1.1943	1.1953	1.1973
$r(2,5)$	1.1062	1.0960	1.0992	1.0960	1.0918
<i>trans</i> -glyoxal					
$r(1,2)$	1.5290	1.5243	1.5296	1.5295	1.5149
$r(1,3)$	1.1169	1.1052	1.1069	1.1036	1.1006
$r(2,5)$	1.2127	1.2019	1.1961	1.1972	1.2051
MUE	0.0103	0.0037	0.0060	0.0057	0.0000

Table S49. Bond lengths (Å) of the TMDBL10 database

System	PBE	B3LYP	M11	revM11	Ref.
Ag ₂	2.560	2.585	2.556	2.555	2.530
Au ₂	2.520	2.545	2.489	2.504	2.472
Cu ₂	2.252	2.280	2.308	2.273	2.219
Fe ₂	2.156	2.421	2.505	2.320	2.020
FeC	1.555	1.609	1.725	1.680	1.596
Ir ₂	2.213	2.304	2.267	2.273	2.270
Ni ₂	2.120	2.295	2.330	2.295	2.155
Os ₂	2.236	2.238	2.247	2.236	2.280
Pd ₂	2.472	2.496	2.484	2.472	2.480
Pt ₂	2.331	2.335	2.416	2.424	2.333
MUE	0.043	0.084	0.104	0.078	0.000

Table S50. Dipole charges^a on Na atom (kcal/mol) of the NaCl database

System	PBE	B3LYP	M11	revM11	Ref.
Na atom	0.462	0.445	0.908	0.000	0.000

^aThe values tabulated are the charges on Na atom in NaCl with an internuclear distance of 10 Å.

Table S51. Excitation energies of atoms (eV) in the EEA11 database

System	states	orbitals	type	PBE	B3LYP	M11	revM11	Ref.
H	^2S - ^2S	1s-2s	Rydberg	8.1	8.5	8.4	9.0	10.2
He	^1S - ^1S	1s ² -1s2s	Rydberg	17.6	18.6	18.0	18.3	20.6
Li	^2S - ^2P	2s-2p	valence	2.0	2.0	1.9	1.8	1.9
Be	^1S - ^1P	2s ² -2s2p	valence	5.0	4.9	5.0	5.0	5.3
B	^2P - ^2S	2p-3s	Rydberg	4.1	4.5	4.3	5.0	5.0
Ne	^1S - ^1P	2p ⁶ -2p ⁵ 3s	Rydberg	15.9	16.6	16.9	16.4	16.7
Na	^2S - ^2P	3s-3p	valence	2.1	2.2	1.9	2.1	2.1
Mg	^1S - ^1P	3s ² -3s3p	valence	4.2	4.3	4.2	4.1	4.4
Al	^2P - ^2S	3p-4s	Rydberg	2.7	2.8	2.9	3.3	3.1
Ar	^1S - ^1P	3p ⁶ -3p ⁵ 4s	Rydberg	11.0	11.4	11.5	12.6	11.7
K	^2S - ^2P	4s-4p	valence	1.6	1.7	1.3	1.6	1.6
MUE				0.8	0.5	0.6	0.5	0.0

Table S52. Adiabatic excitation energies (eV) of the AEE15 database

System	PBE	B3LYP	M11	revM11	Ref.
acetaldehyde	3.6	3.8	3.6	3.6	3.7
anisole	4.4	4.7	5.0	5.0	4.5
benzene	4.9	5.1	5.4	5.3	4.7
benzophenone ketyl radical	1.9	2.3	2.8	2.7	2.3
BF	6.1	6.1	6.1	6.0	6.3
C ₂ H ₂	4.9	4.7	4.5	4.5	5.2
cinnoline	1.8	2.4	2.7	2.7	2.8
CO	7.8	8.0	7.8	7.8	8.1
glyoxal	2.0	2.4	2.4	2.5	2.7
<i>p</i> -diethynylbenzene	3.9	4.2	4.6	4.5	4.3
propynal	2.9	3.2	3.3	3.3	3.2
pyridone lactim	4.2	4.6	4.8	4.8	4.5
quinoline	2.9	3.5	3.8	3.9	4.0
ScO	2.1	2.2	1.9	2.0	2.0
VO	1.5	1.4	0.8	1.1	1.6
MUE	0.4	0.2	0.3	0.3	0.4

Table S53. Charge transfer excitation energies (eV) of the EEAroT5 database

System^a	PBE	B3LYP	M11	revM11	Ref.
Benzene–TCNE	1.5	2.1	3.6	4.0	3.6
Toluene–TCNE	1.4	1.8	3.3	3.7	3.4
Xylene–TCNE	1.1	1.5	3.1	3.5	3.2
Naphthalene–TCNE	0.4	0.9	2.6	3.0	2.6
Anthracene–TCNE	1.1 ^b	1.2 ^b	1.8 ^b	2.1 ^b	1.7 ^c
MUE	1.8	1.4	0.1	0.4	0.0

^aTCNE = tetracyanoethylene

^bThese values are obtained by subtracting 0.32 eV from the gas phase values to obtain solution phase values that can be compared to the reference value.

^cSolution phase value

Table S54. Excitation energies (eV) of the EE69 database

System	State	Type	PBE	B3LYP	M11	revM11	Ref.
acetaldehyde							
	A''	valence	4.14	4.26	3.92	3.91	4.28
	2 A'	Rydberg	5.39	6.17	6.28	7.03	6.82
	3 A'	Rydberg	5.86	6.66	6.94	7.75	7.46
	4 A'	Rydberg	6.18	6.88	7.08	7.96	7.75
	6 A'	Rydberg	6.58	7.31	7.91	8.74	8.43
	7 A'	Rydberg	6.64	7.42	7.94	8.89	8.69
acetone							
	1 A ₂	valence	4.22	4.38	4.09	4.05	4.43
	1 B ₂	Rydberg	4.90	5.73	6.05	6.79	6.36
	2 A ₂	Rydberg	5.76	6.59	6.67	7.66	7.36
	2 A ₁	Rydberg	5.63	6.50	6.92	7.85	7.41
	2 B ₂	Rydberg	5.85	6.67	6.87	7.79	7.49
	3 A ₁	Rydberg	6.16	7.07	7.85	8.49	7.80
	3 B ₂	Rydberg	6.02	6.91	7.37	8.23	8.09
	1 B ₁	Rydberg	6.21	7.12	7.66	8.62	8.17
ethylene							
	1 B _{3u}	Rydberg	6.30	6.49	6.48	7.20	7.11
	1 B _{1u}	valence	7.22	7.31	7.37	7.50	7.65
	1 B _{1g}	Rydberg	6.77	6.99	6.91	7.77	7.80
	1 B _{2g}	Rydberg	6.73	6.98	7.01	7.95	7.90
	2 A _g	Rydberg	7.06	7.32	7.58	8.26	8.28
	2 B _{3u}	Rydberg	7.32	7.64	7.88	8.79	8.62
	3 B _{3u}	Rydberg	7.51	7.81	8.56	9.21	8.90
	4 B _{3u}	Rydberg	8.13	8.35	8.76	9.29	9.08
	3 B _{1g}	Rydberg	7.73	8.08	8.89	9.42	9.20
	2 B _{1u}	Rydberg	7.93	8.08	8.59	9.32	9.33
	5 B _{3u}	Rydberg	9.45	9.63	10.01	10.61	9.51
formaldehyde							
	1 A ₂	valence	3.80	3.92	3.56	3.62	4.00
	1 B ₂	Rydberg	5.74	6.44	6.42	7.17	7.08
	2 B ₂	Rydberg	6.51	7.22	7.16	7.93	7.97
	2 A ₁	Rydberg	6.38	7.17	7.23	8.07	8.14
	2 A ₂	Rydberg	6.64	7.42	7.26	8.23	8.37
	3 B ₂	Rydberg	6.78	7.68	8.35	9.23	8.88
	1 B ₁	valence	8.82	8.96	8.70	8.78	9.00
	3 A ₂	Rydberg	8.00	8.68	8.76	9.65	9.22
	4 B ₂	Rydberg	7.73	8.46	8.78	9.48	9.26
	4 A ₁	Rydberg	7.87	8.57	9.23	9.69	9.58
	5 B ₂	Rydberg	7.86	8.57	8.86	9.58	9.63
isobutene							
	B ₁	Rydberg	5.27	5.60	5.76	6.53	6.17
	A ₁	Rydberg	5.87	6.17	6.27	6.82	6.70

pyrazine	B _{3u}	valence	3.57	3.97	4.02	4.07	3.83
	B _{2u}	valence	5.23	5.33	5.46	5.36	4.81
	B _{2g}	valence	5.14	5.60	5.71	5.74	5.46
	B _{1g}	valence	5.59	6.42	7.11	7.19	6.10
	B _{1u}	valence	6.40	6.46	6.67	6.62	6.51
pyridazine	1 B ₁	valence	3.13	3.59	3.69	3.70	3.60
	2 A ₁	valence	5.46	5.60	5.77	5.69	5.00
	1 A ₂	valence	3.50	4.16	4.53	4.55	5.30
	2 B ₁	valence	5.45	6.12	6.47	6.54	6.00
	2 B ₂	valence	6.32	6.28	6.63	6.84	6.50
pyridine	1 B ₁	valence	4.37	4.82	4.92	4.97	4.59
	1 B ₂	valence	5.33	5.47	5.62	5.56	4.99
	1 A ₂	valence	4.46	5.12	5.50	5.53	5.43
	2 A ₁	valence	6.18	6.26	6.42	6.46	6.38
pyrimidine	1 B ₁	valence	3.79	4.29	4.47	4.52	3.85
	1 A ₂	valence	4.01	4.61	4.91	4.95	4.62
	1 B ₂	valence	5.58	5.73	5.89	5.82	5.12
	2 A ₂	valence	5.12	5.69	5.92	5.96	5.52
	2 B ₁	valence	5.33	5.97	6.27	6.32	5.90
	2 A ₁	valence	6.43	6.53	6.78	6.76	6.70
s-tetrazine	1 B _{3u}	valence	1.84	2.26	2.33	2.34	2.25
	1 A _u	valence	2.81	3.48	3.86	3.90	3.40
	2 A _u	valence	4.63	5.10	5.32	5.38	5.00
	2 B _{3u}	valence	5.63	6.31	6.60	6.71	6.34
t-butadiene	1 B _u	valence	5.40	5.54	5.79	5.88	5.91
	1 B _g	Rydberg	5.39	5.57	5.53	6.30	6.22
	2 A _u	Rydberg	5.67	5.88	5.98	6.82	6.66
	2 B _u	Rydberg	6.23	6.41	6.51	7.17	7.07
	2 B _g	Rydberg	6.18	6.43	6.55	7.48	7.36
	3 A _g	Rydberg	6.54	6.75	7.16	7.80	7.62
	3 B _u	Rydberg	7.02	7.20	7.76	8.44	8.00
MUE			0.97	0.58	0.49	0.30	0.00

Table S55. Charge transfer excitation energies (eV) of the EER5 database

System	PBE	B3LYP	M11	revM11	Ref.
11-Z-cis-retinal					
S ₁	2.1	2.3	2.4	2.4	2.1
S ₂	2.6	3.0	3.8	3.9	3.3
11-Z-cis-7,8-dihydroretinal					
S ₁	1.0	1.5	3.9	4.2	2.6
S ₂	3.2	3.1	3.0	3.1	3.0
S ₃	2.2	3.2	5.3	5.5	4.4
MUE	0.9	0.5	0.6	0.7	0.0

Table S56. Charge transfer excitation energies (eV) of the LRCTEE2 database

System	PBE	B3LYP	M11	revM11	Ref.
C ₂ F ₄ ⋯⋯C ₂ H ₄ (8 Å)	5.2	6.8	11.4	10.8	10.9
NH ₃ ⋯⋯F ₂ (6 Å)	0.1	2.2	6.4	6.9	7.7
MUE	6.7	4.8	0.9	0.4	0.0

Table S57. Geometries (\AA), total electronic energies (E in hartrees), charges, and spin multiplicities of selected systems in the SR-MGN-BE107 database^a

System			
H_2O ($E = -76.427690$)			
Charge, spin multiplicity = 0, 1			
O	0.000000	0.000000	0.117145
H	0.000000	0.756709	-0.468582
H	0.000000	-0.756709	-0.468582
C_2H_2 ($E = -77.292447$)			
Charge, spin multiplicity = 0, 1			
C	0.000000	0.000000	0.601571
C	0.000000	0.000000	-0.601571
H	0.000000	0.000000	-1.664176
H	0.000000	0.000000	1.664176
CH (${}^2\Pi$) ($E = -38.458990$)			
Charge, spin multiplicity = 0, 2			
C	0.000000	0.000000	0.159876
H	0.000000	0.000000	-0.959254
CH_2 (${}^3\text{B}_1$) ($E = -39.131921$)			
Charge, spin multiplicity = 0, 3			
C	0.000000	0.000000	0.105529
H	0.000000	0.990531	-0.316586
H	0.000000	-0.990531	-0.316586

^aThe basis set is MG3S, which for these four systems (or for any system with no atom heavier than Ne) is the same as 6-311+G(2df,2p).

Reference

-
1. Wang, Y.; Verma, P.; Jin, X. S.; Truhlar, D. G.; He, X. Revised M06 Density Functional for Main-Group and Transition-Metal Chemistry. *Proc. Natl. Acad. Sci. USA* **2018**, *115*, 10257–10262.