Supplement 3

The measurements of viscosity, η and of the specific conductance, Λ_o , are shown for CaCO₃ supersaturated solutions containing different MEG concentrations, at constant ionic strength and temperature. The total solution volume in which measurements were done was 50ml. For viscosity measurement, a Coutte viscometer was used and for the limiting conductance (Λ_o =Conductivity x Volume of solution) measurements, a conductivity meter with the appropriate cell (Conductivity meter, Metrohm 660) was also used. The results of the measurements and Walden's product calculated are shown in Table S3.1

Table S3.1. Viscosity, specific conductance and Walden's constant for different MEG- water mixtures; ionic strength, I=0.1 M, 25°C

% v/v MEG	Limiting Conductance. Λο/ (S·cm²·mol⁻¹)	Viscosity. η/(mPa·s)	Walden product, $W=\eta\Lambda_0$ $(S\cdot cm^2\cdot mol^{-1})\cdot (mPa\cdot s)$
10	199.86	5.58	1115.63
20	145.00	7.26	1053.20
30	120.14	8.73	1048.66
40	94.90	9.45	896.94
50	79.18	9.66	764.88
60	53.09	10.99	583.20
70	38.98	11.43	445.56
80	28.74	14.73	423.46
90	18.98	17.56	333.28

The values of the dielectric constant for the MEG-Water solutions are summarized in Table S3.2.

Table S3.2. Dielectric constant of water-MEG mixtures; 25°C (from Ref.41: Akerlöf G. Dielectric constants of some organic solvent-water mixtures at various temperatures, *J. Am. Chem. Soc.*, **1932**, 54, 4125–4139.)

%v/v MEG	3
0	80.37
10	74.60
20	68.70
30	62.63
40	56.49
50	50.38
60	44.67
70	39.14
80	33.89
90	44.90