Supporting Information

Nucleic Acid Functionalized Metal-Organic Frameworks-Based Homogeneous Electrochemical Biosensor for Simultaneous Detection of Multiple Tumor Biomarkers

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Table of Contents

Chemical and materials
ApparatusS-3
Table S1. Sequence of the oligonucleotides used in the experiments
Table S2. miRNA assay performance of our strategy and other methods
Table S3. Recoveries and RSDs of the proposed biosensor in serum samples
Figure S1. SEM and EDS mapping imagesS-7
Figure S2. XPS curves of UIO-66-NH ₂ and TMB@UIOS-8
Figure S3. EDX. curves of UIO-66-NH ₂ and TMB@UIOS-9
Figure S4. FT-IR spectra of TMB, UIO-66-NH ₂ , and TMB@UIOS-10
Figure S5. Stability of MB@UIO and TMB@UIO for different incubation timesS-11
Figure S6. Experimental condition optimizationS-12
Figure S7. Reusability of C _x modified UIO-66-NH ₂ S-13
Figure S8. Analysis of let-7a and miRNA-21 in breast cancer patient serumS-14
References

EXPERIMENTAL SECTION

Chemical and Materials. All DNA and RNA were purchased from Shanghai Sangon Biotechnology Co. Ltd. (Shanghai, China) and TaKaRa Biotechnology Co., Ltd. (Dalian, China), respectively, with their sequences exhibited in Table S1. Zirconium oxide chloride octahydrate (ZrOCl₂ 8H₂O), 2-aminoterephthalic acid, 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide (EDC), N-hydroxysulfosuccinimide (NHS), 3,3',5,5'-tetramethyl-benzidine (TMB), methylene blue (MB), CH₃COOH, Na₂HPO₄, NaH₂PO₄ and other reagents were ordered from Sigma-Aldrich (St. Louis, MO, USA) and used without further treatment.

Apparatus. X-ray diffraction (XRD) patterns were obtained using a Bruker D8 Advance X-ray diffractometer (Bruker, Germany). Transmission electron microscopy (TEM) images were recorded HT7700 (Hitachi, on an microscope Japan). X-ray photoelectron spectroscopy (XPS) was recorded on an ESCALAB 250Xi (Thermo Fisher S cientific, USA). Energy dispersive spectroscopy (EDS) mapping images and spectrum analysis we re conducted using a HITACHI S4800 SEM (Hitachi, Japan).Zeta potential measurement was recorded with a Nano Zetasizer ZS90 (Malvern, U.K.). Differential pulse voltammetric (DPV) measurements were carried out on an Autolab electrochemical workstation (Metrohm, Netherland) employing a three-electrochemical system. The fluorescence measurements were performed on an F-4600 fluorescence spectrometer (Hitachi, Japan). Fourier transform infrared spectra (FT-IR) were recorded using a NICOLET 380 FT-IR spectrometer (Nicolet Thermo, USA).

Probe Name	Sequence (5'-3')				
C _{MB}	COOH-TGA GGT AGT AGG TT				
P _{MB}	AA CTA TAC AA CCT ACT ACC TCA				
let-7a	UGA GGU AGU AGG UUG UAU AGU U				
let-7b	UGA GGU AGU AGG UUG UGU GGU U				
let-7d	AGA GGU AGU AGG UUG CAU AGU U				
Стмв	COOH-TAG CTT ATC AGA CT				
P _{TMB}	TG AAC ATC AG TCT GAT AAG CTA				
miRNA-21	UAG CUU AUC AGA CUG AUG UUG A				
miRNA-16	UAG CAG CAC GUA AAU AUU GGC G				
miRNA-155	UUA AUG CUA AUC GUG AUA GGG GU				

Table S1. Sequences of the oligonucleotides used in the experiments

	E	Signal Molecule	Linear	Detection	D.e	
Method	Enzyme	Labeling	Range	Limit	Ref.	
Fluorescence	Yes	Yes	0.001~10000	1 fM	1	
Thuorescence	105	105	pМ	1 1101	1	
Fluorescence	Yes	Yes	0.01~1000 pM	10 fM	2	
Fluorescence	Yes	No	0.06~12.0 pM	10.8 fM	3	
hemiluminescence Yes		No	0.01~50.0 pM	3.02 fM	4	
Electroche-		N/	0.01.1000.01		_	
miluminescence	Yes	Yes	0.01~1000 fM	3.3 aM	5	
Electrochemistry	No	Yes	0.005~500 pM	1.4 fM	6	
Electrochemistry	Yes	Yes	0.01~5.0 fM	3.2 aM	7	
Electrochemistry	Yes	No	0.01~1.0 pM	4.53 fM	8	
electrochemistry			0.01~10.0 pM	3.6 fM	This	
	No	No	0.02~10.0 pM	8.2 fM	work	

Table S2. miRNA assay performance of our strategy and other methods.

Sample No.	Added (pM)		Mean measured (pM)		Mean recovery ^a (%)		RSD (%)	
	let-7a	miRNA-21	let-7a	miRNA-21	let-7a	miRNA-21	let-7a	miRNA-21
1	0.1	0.1	0.0947	0.0963	94.7	96.3	4.75	4.07
2	1.0	1.0	1.0324	0.9736	103.0	97.4	3.89	3.20
3	3.0	3.0	3.1586	3.1267	105.0	104.0	4.16	4.38

Table S3. Recoveries and RSDs of the MOFs-based biosensor for let-7a and miRNA-21 spiked in human serum samples.

 ${}^{a}Recovery~(\%) = 100 \times (c_{mean~measured} \ / \ c_{added})$

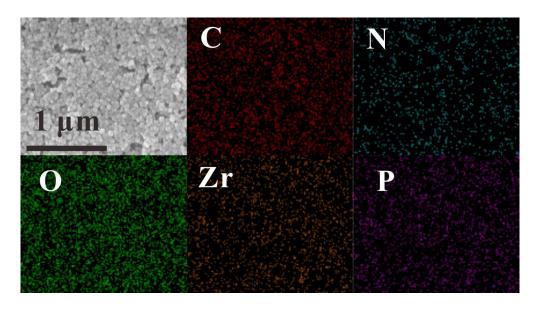


Figure S1. SEM and EDS mapping images of TMB@UIO and the corresponding distribution of C,

N, O, Zr, and P elements.

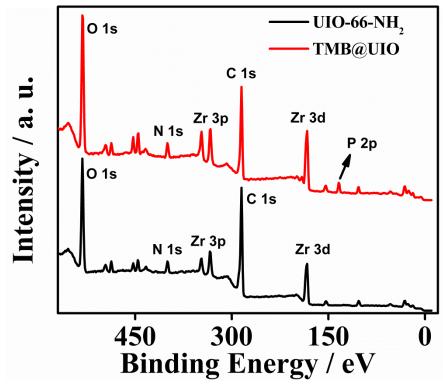


Figure S2. XPS curves of UIO-66-NH₂ and TMB@UIO.

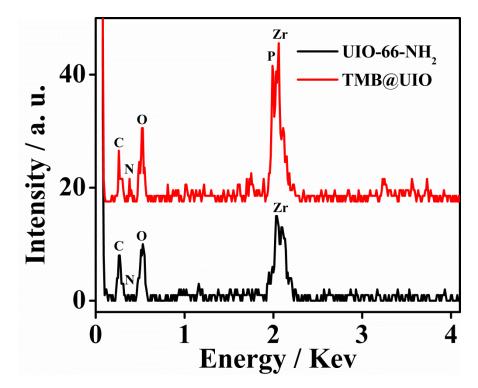


Figure S3. EDX spectra of UIO-66-NH₂ and TMB@UIO.

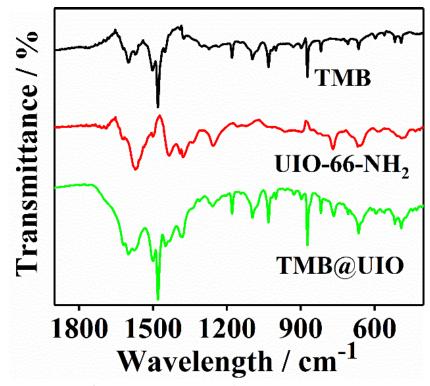


Figure S4. FT-IR spectra of TMB, UIO-66-NH₂, and TMB@UIO.

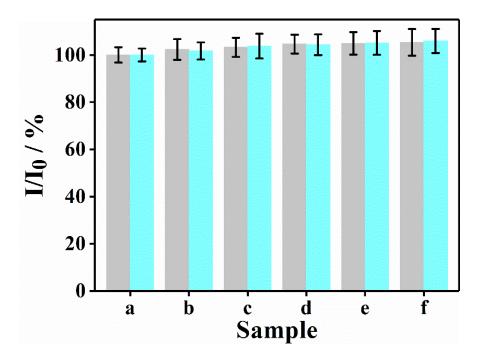


Figure S5. The stability of MB@UIO and TMB@UIO in PBS for different incubation times: a: 0

h, b: 1 h, c: 3 h, d: 5 h, e: 7 h, and f :in diluted human serum sample for 7 h.

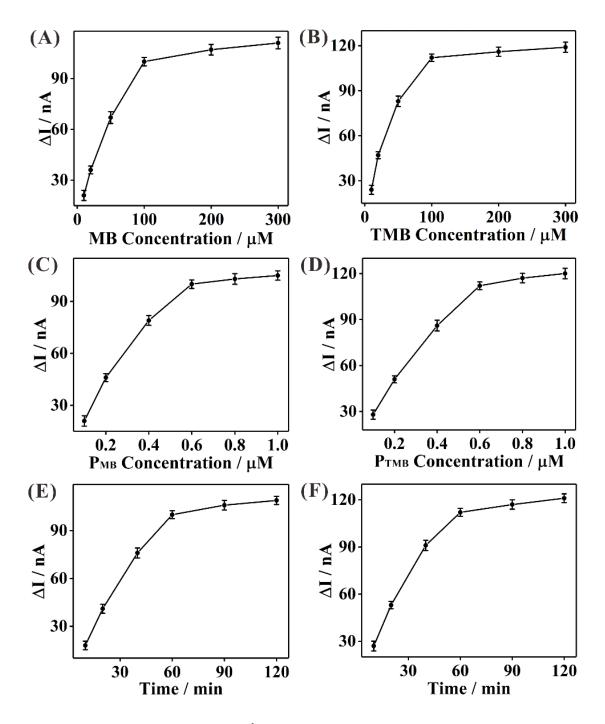


Figure S6. The DPV current changes (Δ I) of the proposed biosensor upon the addition of let-7a with different MB concentrations (A), P_{MB} concentrations (C), and the hybridization time between let-7a and MB@UIO (E). The DPV current changes (Δ I) of the proposed biosensor upon the addition of miRNA-21 with different TMB concentrations (B), P_{TMB} concentrations (D), and the hybridization time between miRNA-21 and TMB@UIO (F).

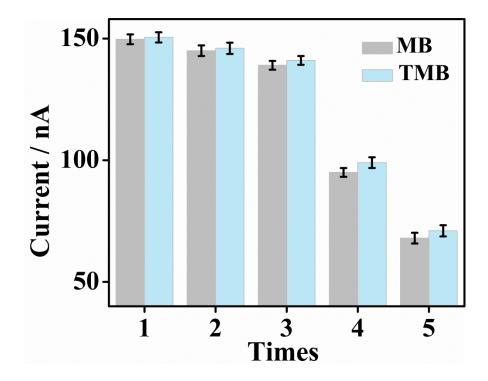


Figure S7. The reusability of C_x modified UIO-66-NH₂ in development of MB@UIO and TMB@UIO biosensors.

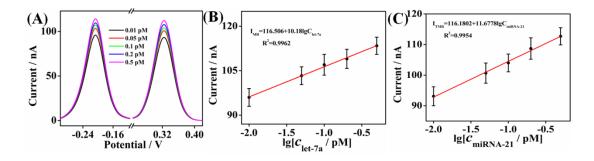


Figure S8. (A) The DPV curves of the proposed biosensor in the serum sample from breast cancer patient with different concentrations of let-7a and miRNA-21. (B) The linear relationship between the DPV current of MB and the logarithm of let-7a concentrations. (C) The linear relationship between the DPV current of TMB and the logarithm of miRNA-21 concentrations.

References

(1) Yin, B. C.; Liu, Y. Q.; Ye, B.-C., Sensitive Detection of MicroRNA in Complex Biological Samples via Enzymatic Signal Amplification Using DNA Polymerase Coupled with Nicking Endonuclease. *Anal. Chem.* **2013**, *85*, 11487–11493.

(2) Duan, R. X.; Zuo, X. L.; Wang, S. T.; Quan, X. Y.; Chen, D. L.; Chen, Z. F.; Jiang, L.; Fan, C. H.; Xia, F., Lab in a Tube: Ultrasensitive Detection of MicroRNAs at the Single-Cell Level and in Breast Cancer Patients Using Quadratic Isothermal Amplification. *J. Am. Chem. Soc.* 2013, *135*, 4604–4607.

(3) Liu, H. Y.; Li, L.; Wang, Q.; Duan, L. L.; Tang, B., Graphene Fluorescence Switch-Based Cooperative Amplification: A Sensitive and Accurate Method to Detection MicroRNA. *Anal. Chem.* **2014**, *86*, 5487–5493.

(4) Ling, K.; Jiang, H.; Huang, X.; Li, Y.; Lin, J.; Li, F.-R., Direct Chemiluminescence Detection of Circulating MicroRNAs in Serum Samples Using a Single-strand Specific Nuclease-Distinguishing Nucleic Acid Hybrid System. *Chem. Commun.* **2018**, *54*, 1909–1912.

(5) Chen, A. Y.; Gui, G. F.; Zhuo, Y.; Chai, Y. Q.; Xiang, Y.; Yuan, R., Signal-off Electrochemiluminescence Biosensor Based on Phi29 DNA Polymerase Mediated Strand Displacement Amplification for MicroRNA Detection. *Anal. Chem.* **2015**, *87*, 6328–6334.

(6) Shi, K.; Dou, B. T.; Yang, C. Y.; Chai, Y. Q.; Yuan, R.; Xiang, Y., DNA-Fueled Molecular Machine Enables Enzyme-Free Target Recycling Amplification for Electronic Detection of MicroRNA from Cancer Cells with Highly Minimized Background Noise. *Anal. Chem.* **2015**, *87*, 8578–8583.

(7) Miao, P.; Jiang, Y.; Zhang, T.; Huang, Y.; Tang, Y., Electrochemical Sensing of Attomolar

MiRNA Combining Cascade Strand Displacement Polymerization and Reductant-Mediated Amplification. *Chem. Commun.* **2018**, *54*, 7366–7369.

(8) Shi, L.; Mu, C.; Gao, T.; Chen, T.; Hei, S.; Yang, J.; Li, G., DNA Nanoflower Blooms in

Nanochannels: a New Strategy for MiRNA Detection. Chem. Commun. 2018, 54, 11391–11394.