

Dehydrogenative Synthesis of Quinolines, 2-Aminoquinolines and Quinazolines using Singlet Di-radical Ni(II)-Catalysts

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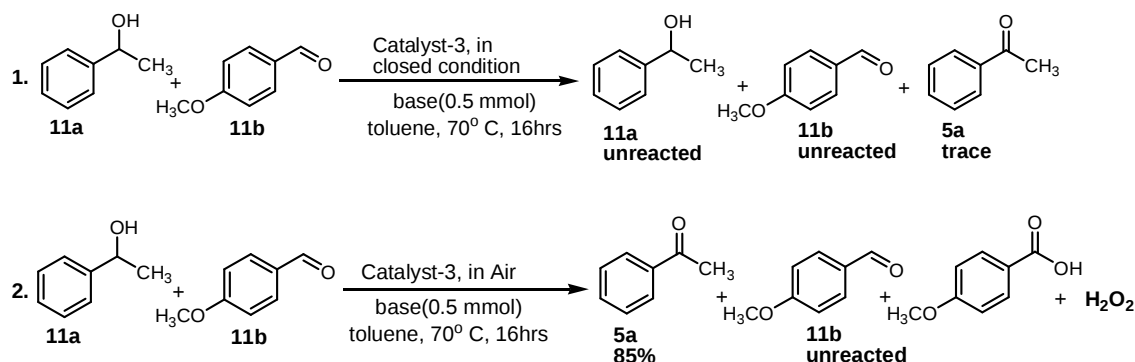
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Table of Contents

Experimental Details	S3-S4
EPR spectra of the reaction mixture containing catalyst 3 and KO ^t Bu recorded at RT.	S4
Figures of ¹ H and ¹³ C NMR spectra	S5-S66
References	S67

Intermolecular Transfer Hydrogenation Experiments using **3** as the catalyst.



Scheme S1. Intermolecular Transfer Hydrogenation Experiments using **3** as the catalyst.

Detection of Hydrogen Peroxide During the Catalytic Alcohol Oxidation Reaction.¹

During catalytic alcohol oxidation reaction H_2O_2 was produced and it was detected spectrophotometrically following the gradual development of the characteristic peak of I_3^- at $\lambda_{\text{max}} = 351 \text{ nm}$, upon reaction with KI. A mixture of 1.0 mmol 1-phenylethanol, 0.5 mmol KO^tBu and catalyst **3** (4 mol%) in a 5.0 ml dry toluene was placed in a 50 ml round-bottom flask fitted with a stir bar and a condenser and stirred in a preheated oil bath at 70°C for 16h. To this mixture an equal volume of water was added and the total reaction mixture was extracted three times with dichloromethane. The separated aqueous layer was acidified with H_2SO_4 to $\text{pH}=2$ to stop further oxidation. Then 1.0 mL of a 10% solution of KI and a few drops of a 3% solution of ammonium molybdate was added to it. Produced hydrogen peroxide oxidizes I^- to I_2 , which reacts with excess I^- to form I_3^- according to the following reactions: (i) $\text{H}_2\text{O}_2 + 2\text{I}^- + 2\text{H}^+ \rightarrow 2\text{H}_2\text{O} + \text{I}_2$, (ii) $\text{I}_2(\text{aq}) + \text{I}^- \rightarrow \text{I}_3^-$. (Fig: S1).

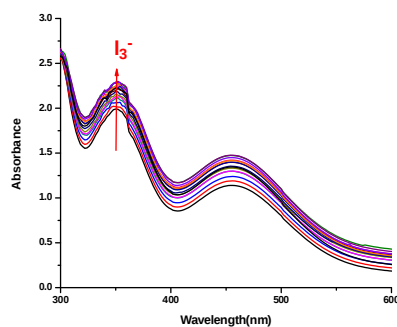
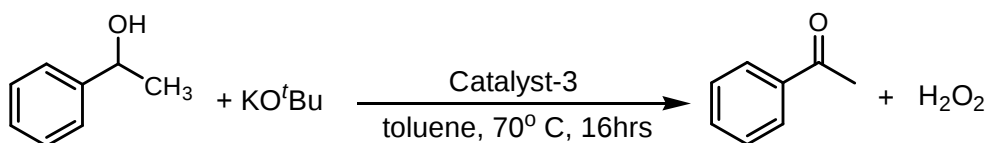


Figure S1: Detection of H_2O_2 . Absorption spectral changes during formation of I_3^- in presence of H_2O_2 .

Quantification of Hydrogen Peroxide During the Catalytic Alcohol Oxidation Reaction.

Hydrogen Peroxide was quantified iodometrically using standard sodium thiosulphate solution. At first (N/20) sodium thiosulphate solution was standardized by standard (N/20) potassium dichromate solution ($S_{\text{dichromate}} = 0.611/0.6129 \text{ (N/20)} = 0.0498 \text{ (N)}$) using 1% starch solution as indicator. Strength of sodium thiosulphate solution was 0.0517 (N).



Scheme S2. Reaction for quantification of H_2O_2 using **3** as the catalyst.

Serial No.	Amount of 1-phenyl ethanol	Amount of KO^tBu	Amount of catalyst 3	Amount of H_2O_2 (w.r.t. alcohol)
1	1.0 mmol	0.5 mmol	4.0 mol%	0.68 equiv.
2	1.0 mmol	1.0 mmol	4.0 mol%	0.48 equiv.
3	0.1 mmol	0.1 mmol	0.1 mmol	0.82 equiv.

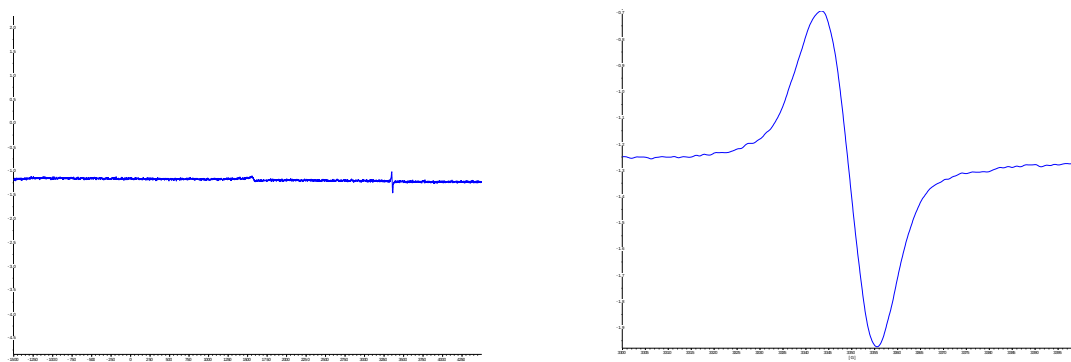


Figure S2. EPR spectra (left full spectrum, right zoom) of the reaction mixture containing catalyst **3** and KO^tBu recorded at RT.

Figures of ^1H and ^{13}C NMR spectra

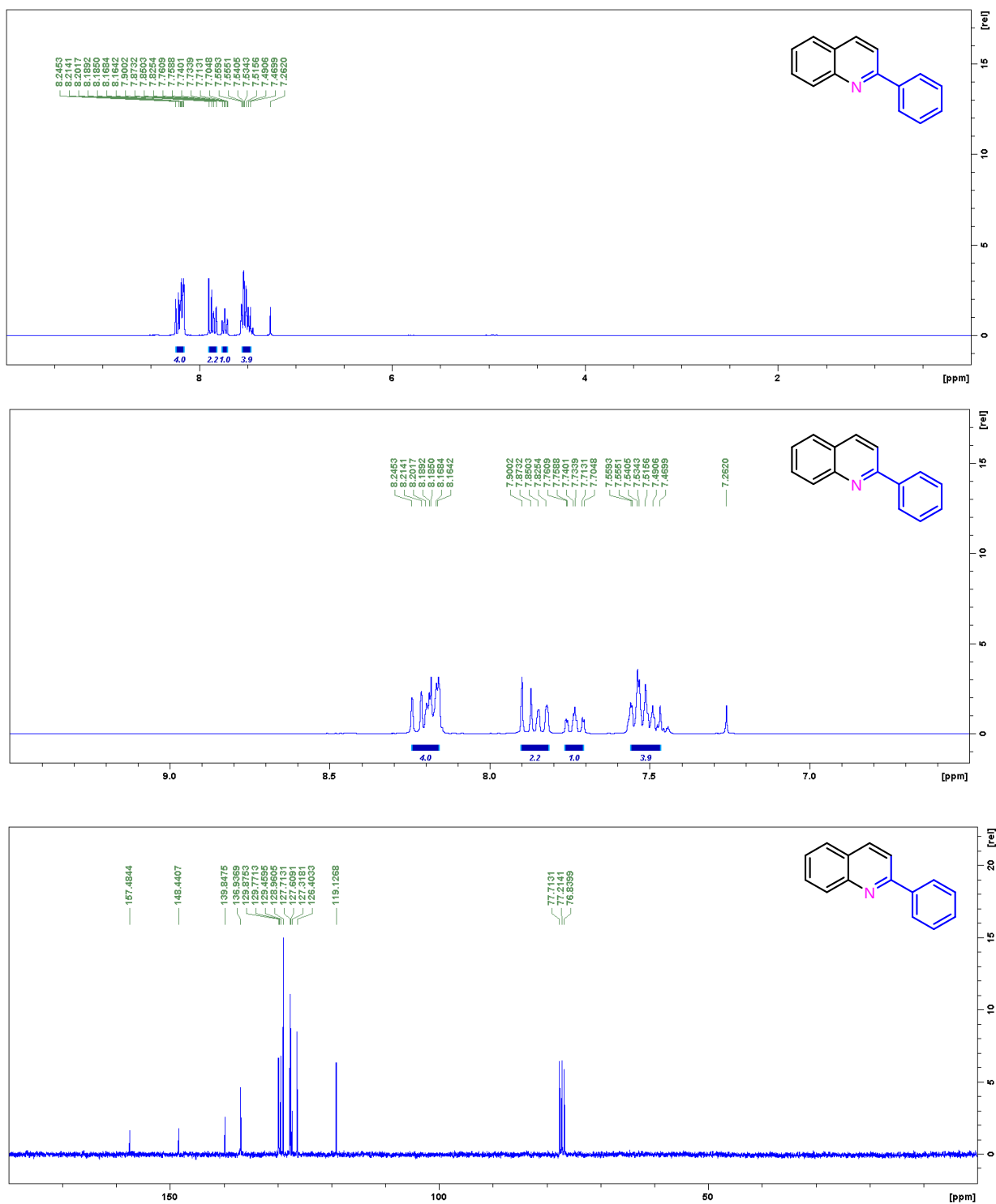


Figure S3. ^1H and ^{13}C NMR spectra of compound **8aa** in CDCl_3 .

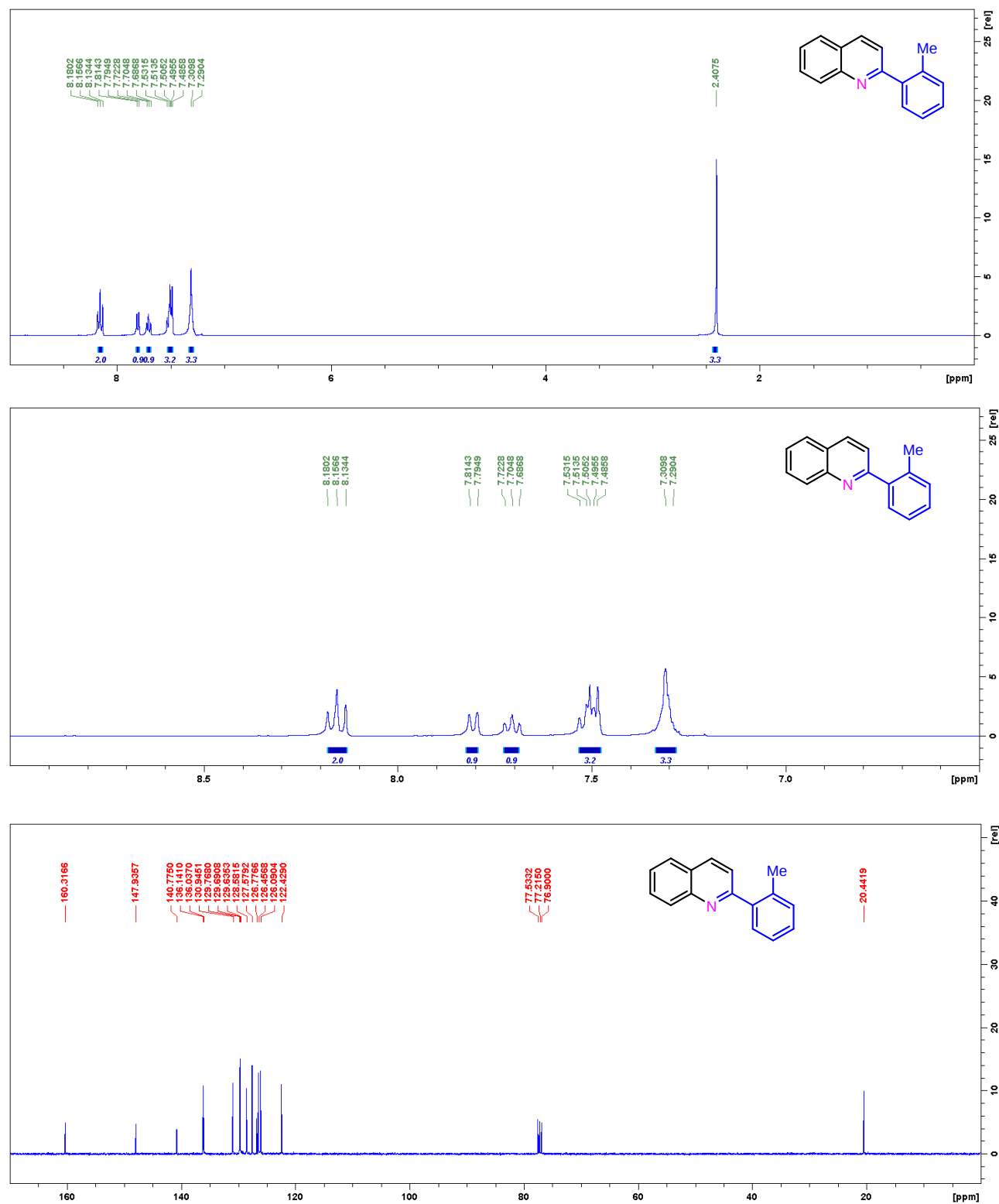


Figure S4. ¹H and ¹³C NMR spectra of compound **8ab** in CDCl₃.

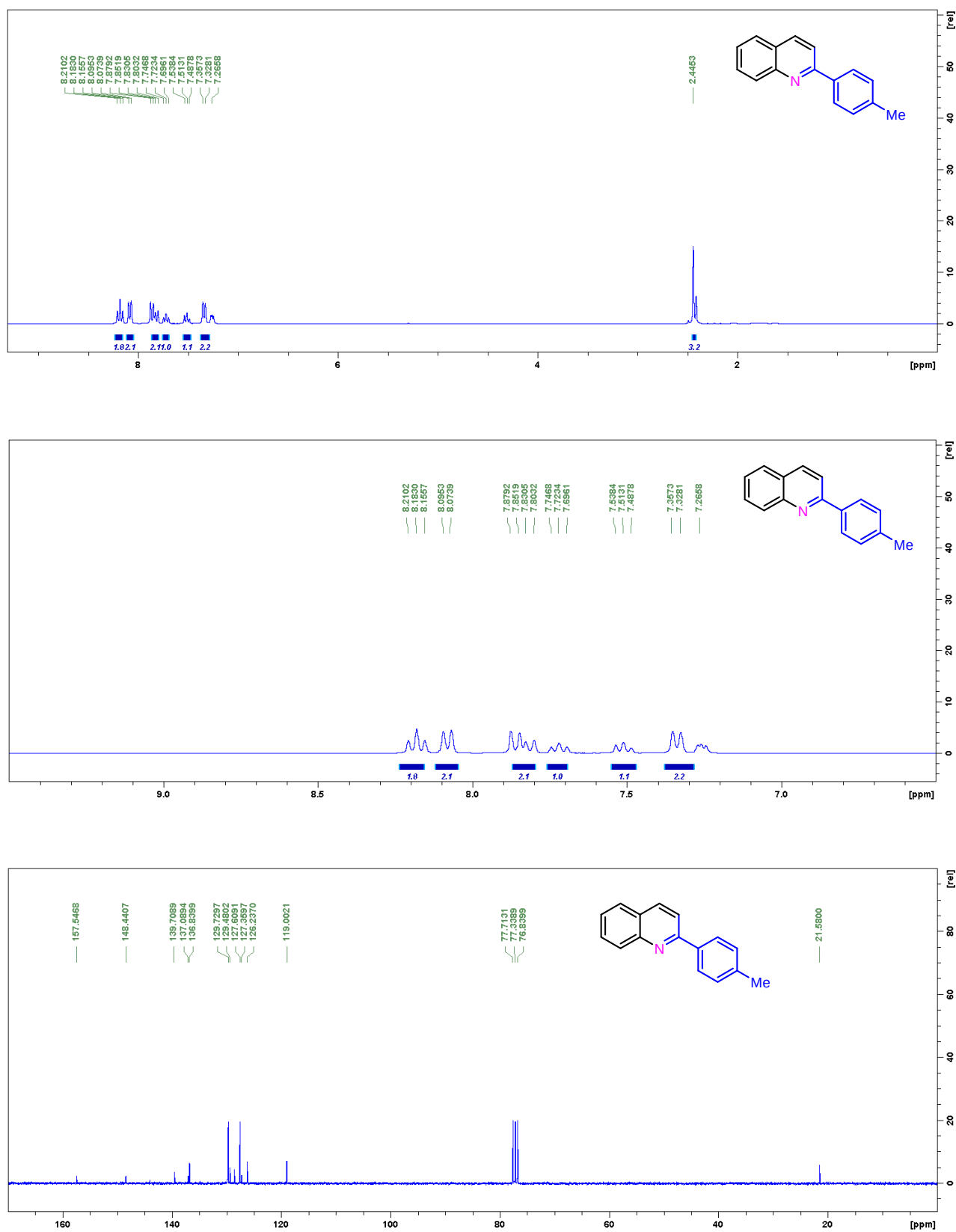


Figure S5. ¹H and ¹³C NMR spectra of compound **8ac** in CDCl₃.

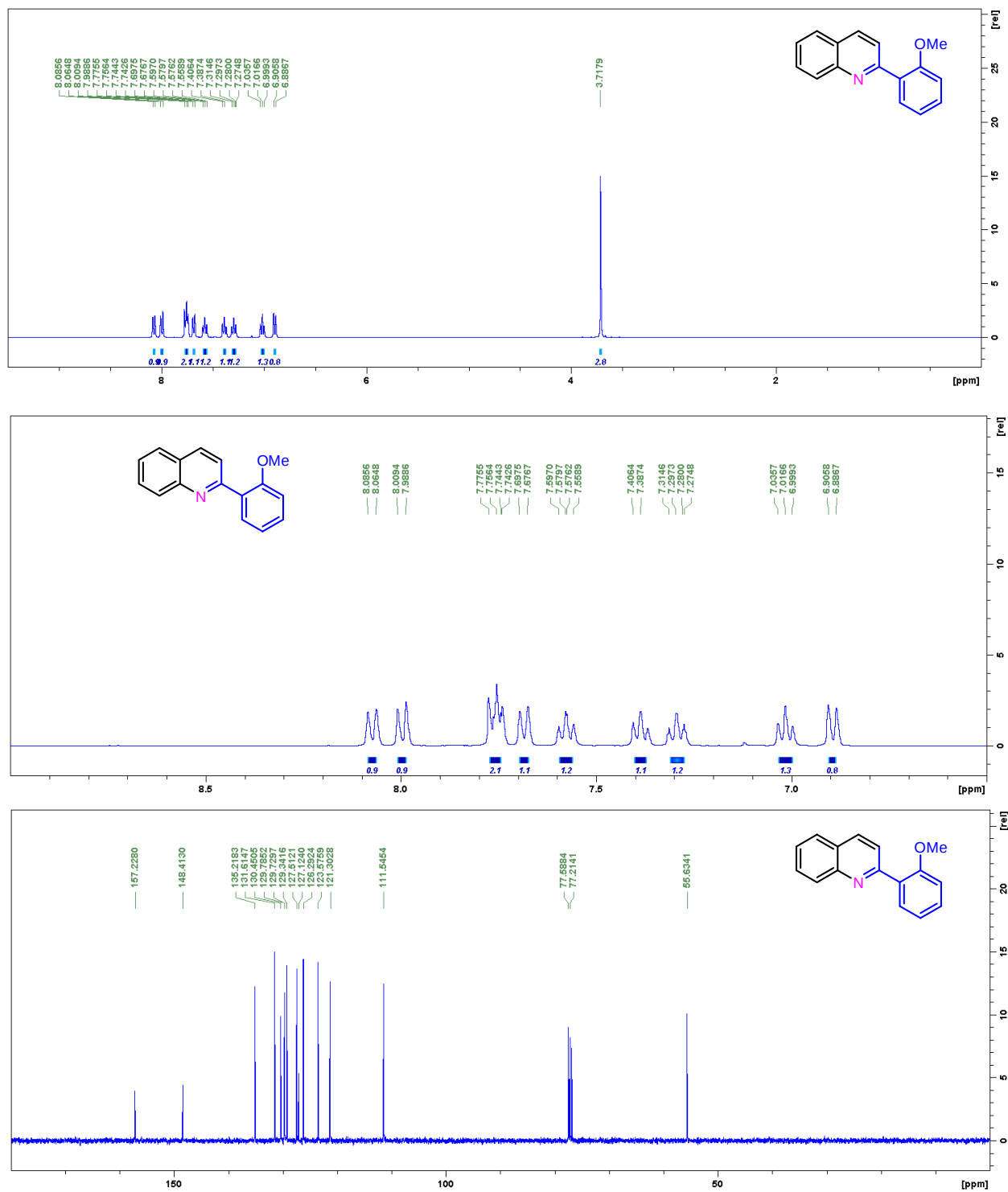


Figure S6. ¹H and ¹³C NMR spectra of compound **8ad** in CDCl₃.

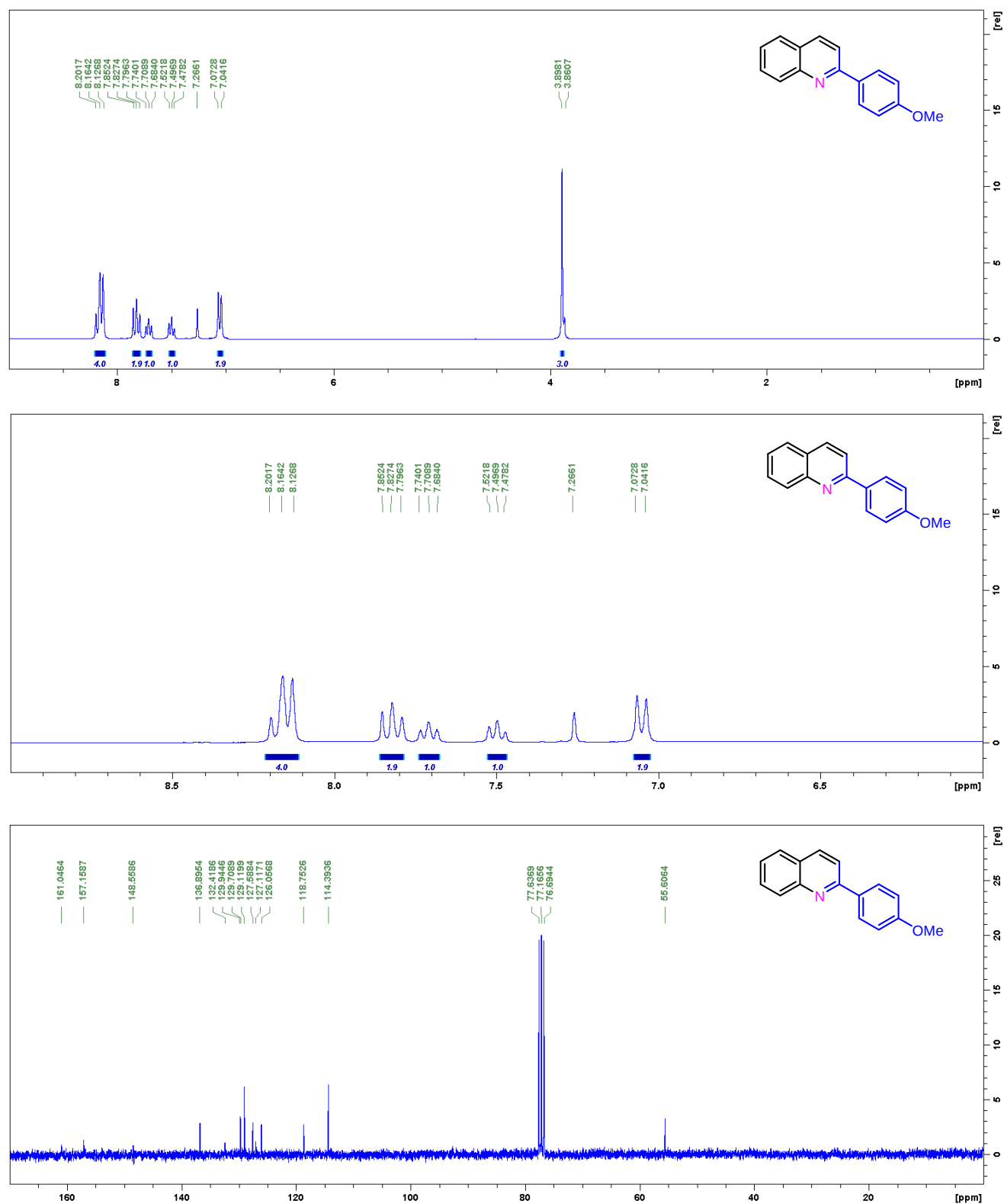


Figure S7. ¹H and ¹³C NMR spectra of compound **8ae** in CDCl₃.

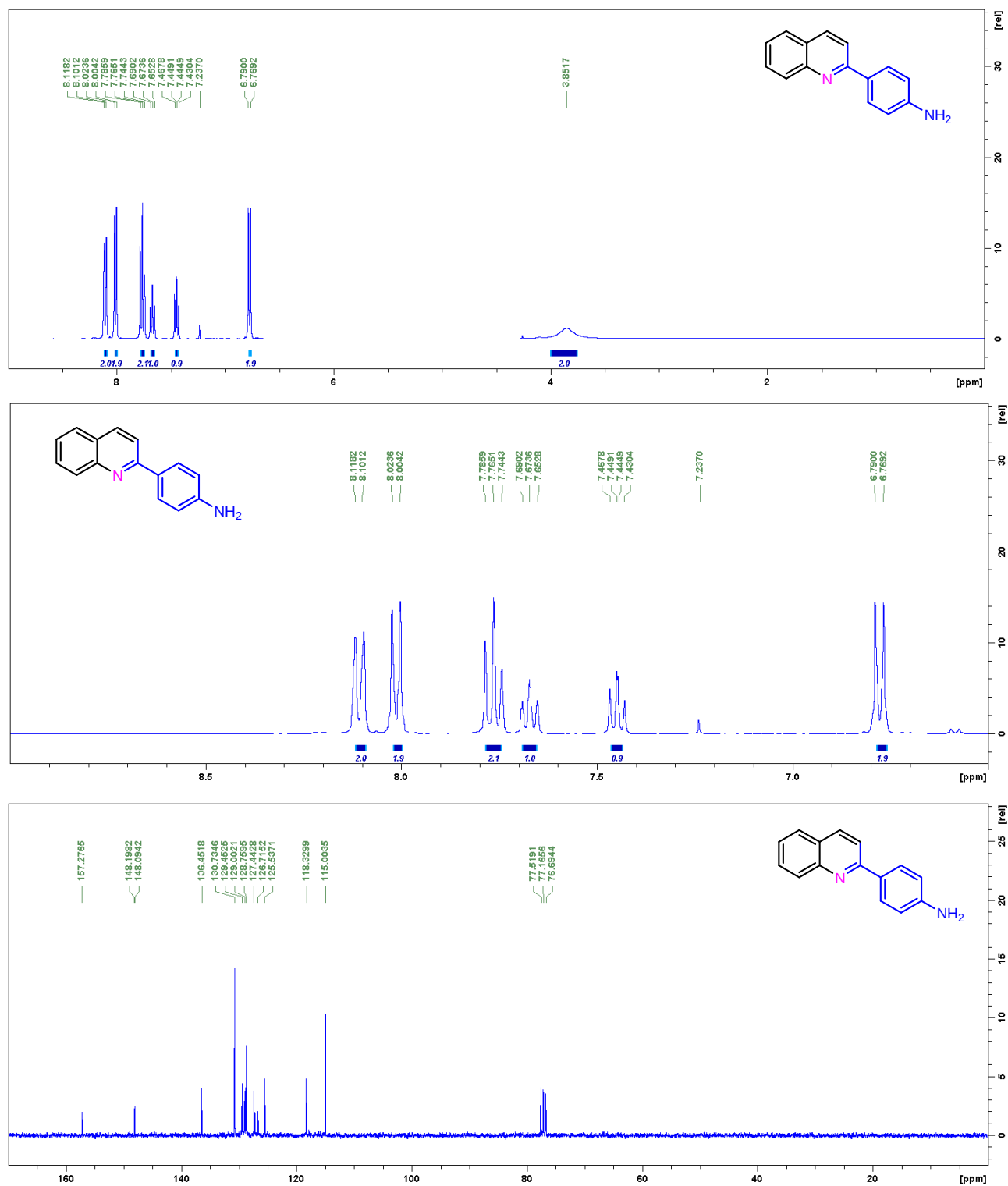
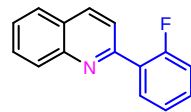
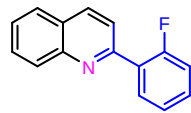
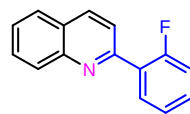


Figure S8. ¹H and ¹³C NMR spectra of compound **8af** in CDCl₃.



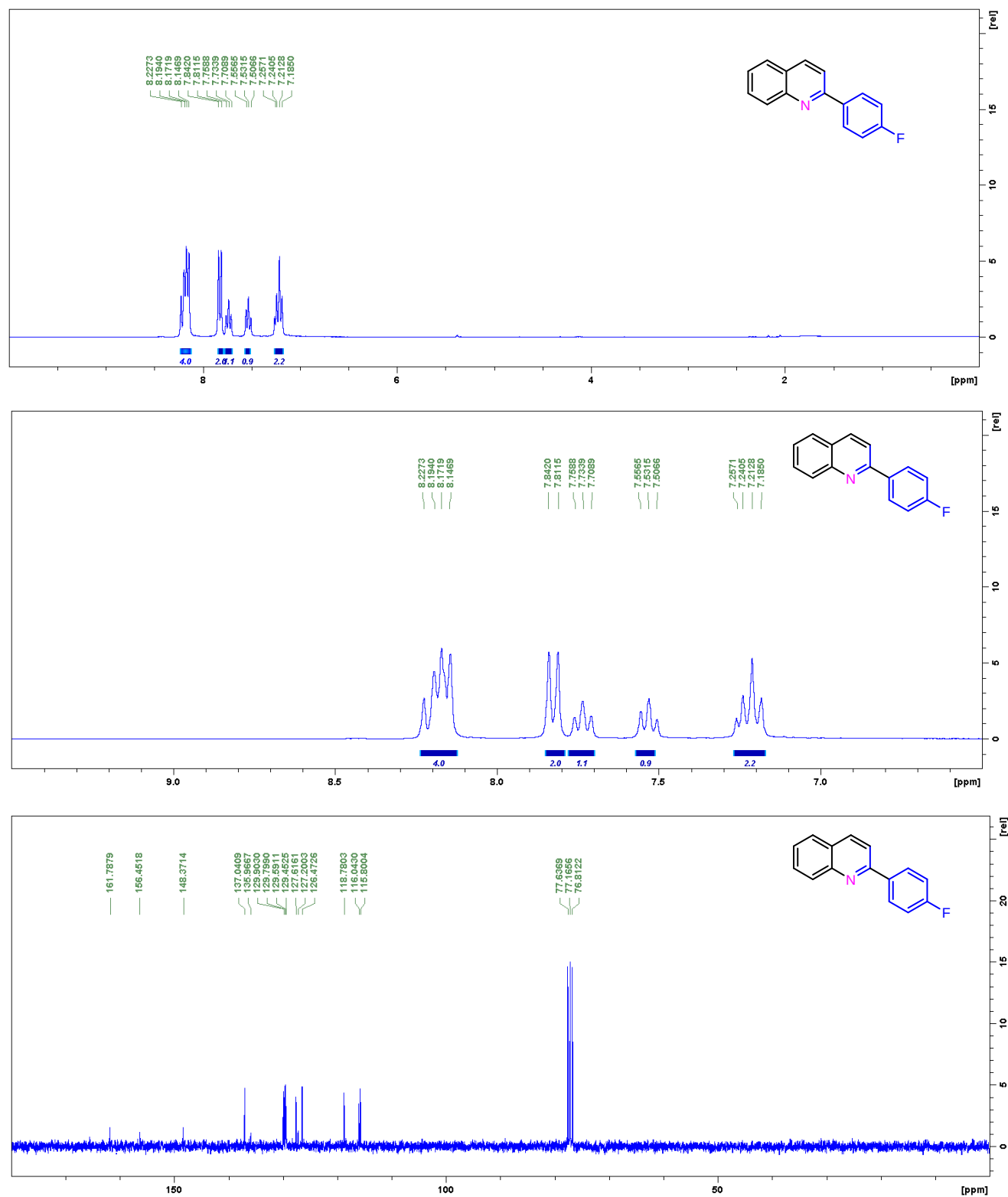


Figure S10. ^1H and ^{13}C NMR spectra of compound **8ah** in CDCl_3 .

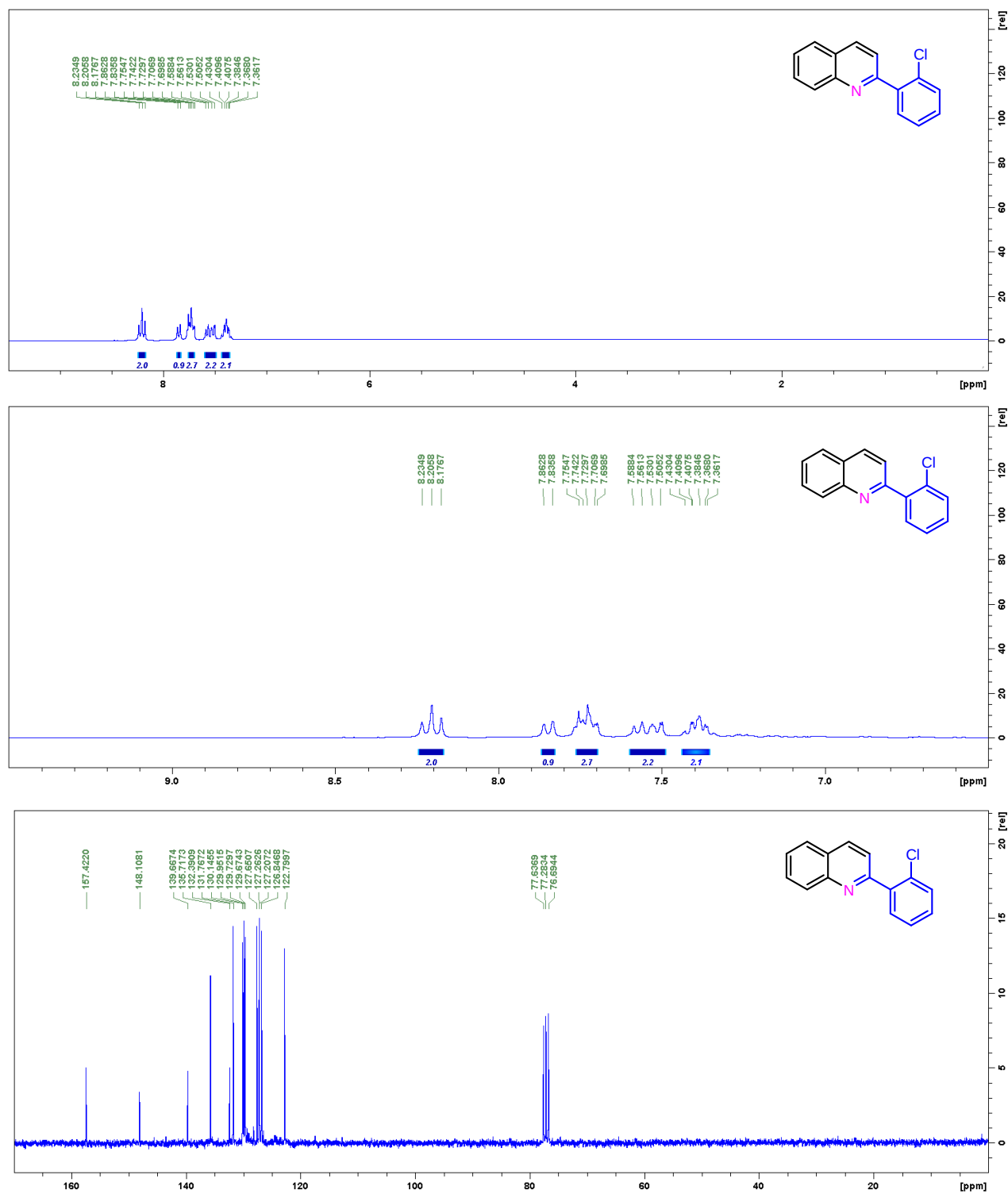


Figure S11. ¹H and ¹³C NMR spectra of compound **8ai** in CDCl₃.

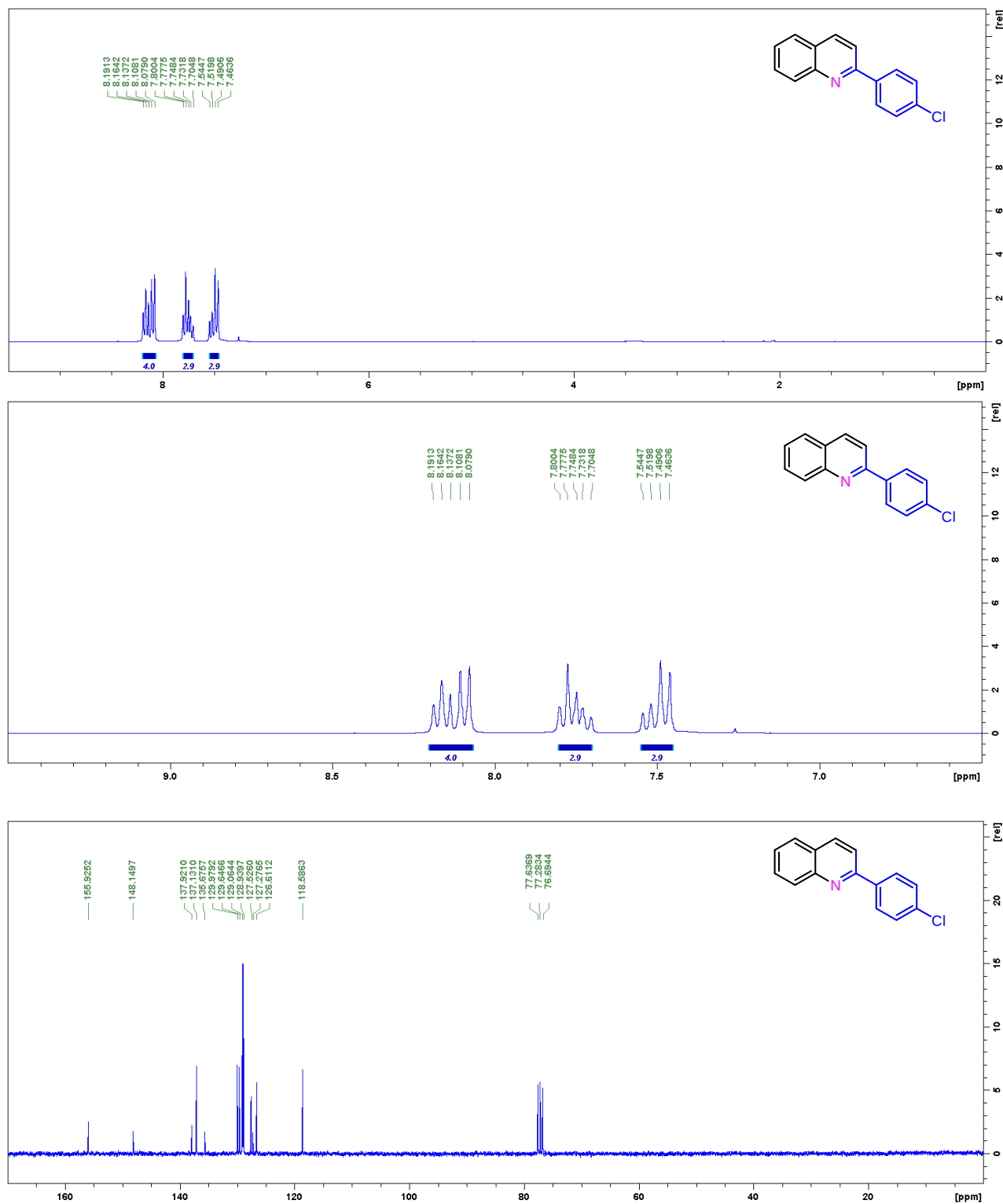


Figure S12. ¹H and ¹³C NMR spectra of compound **8aj** in CDCl₃.

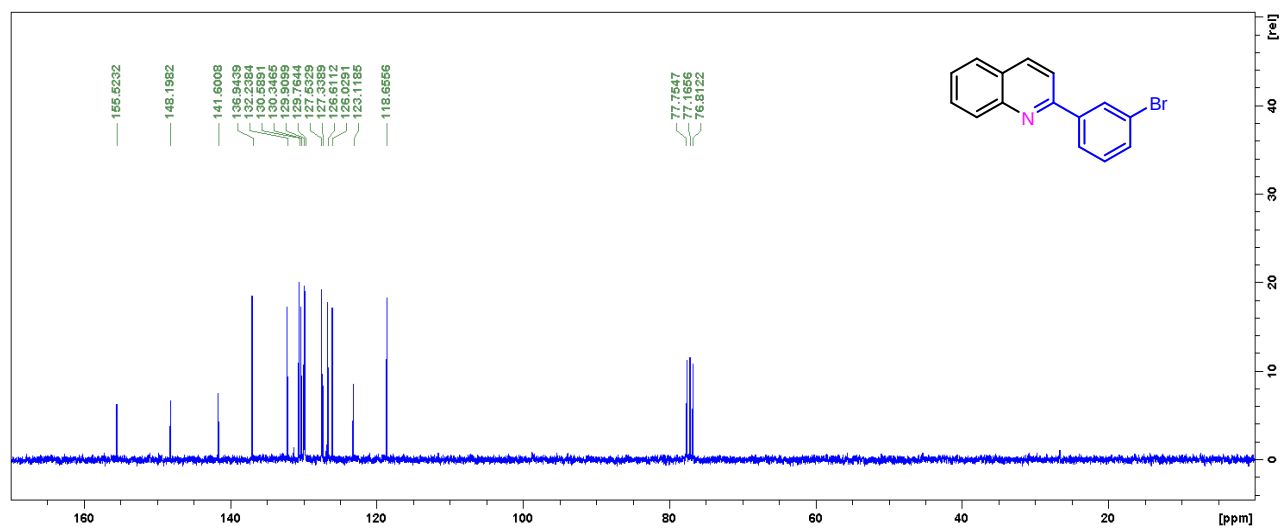
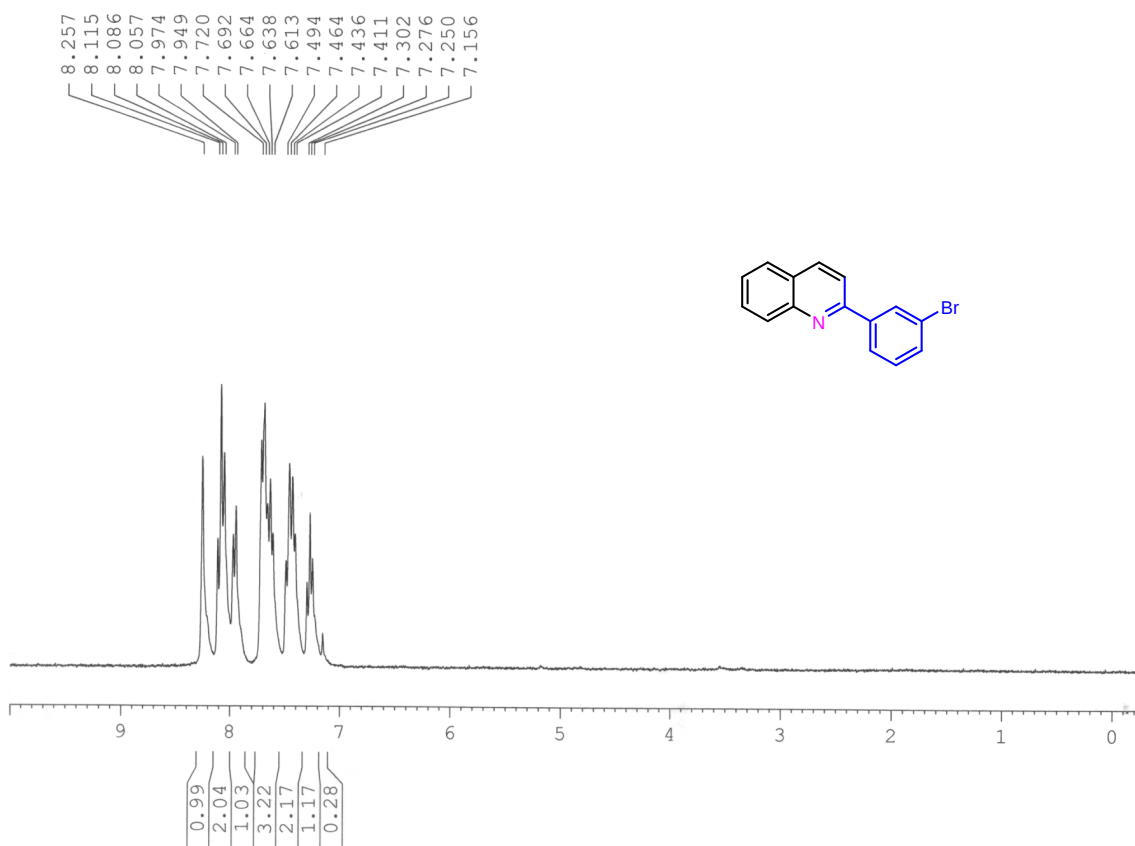


Figure S13. ¹H and ¹³C NMR spectra of compound **8ak** in CDCl₃.

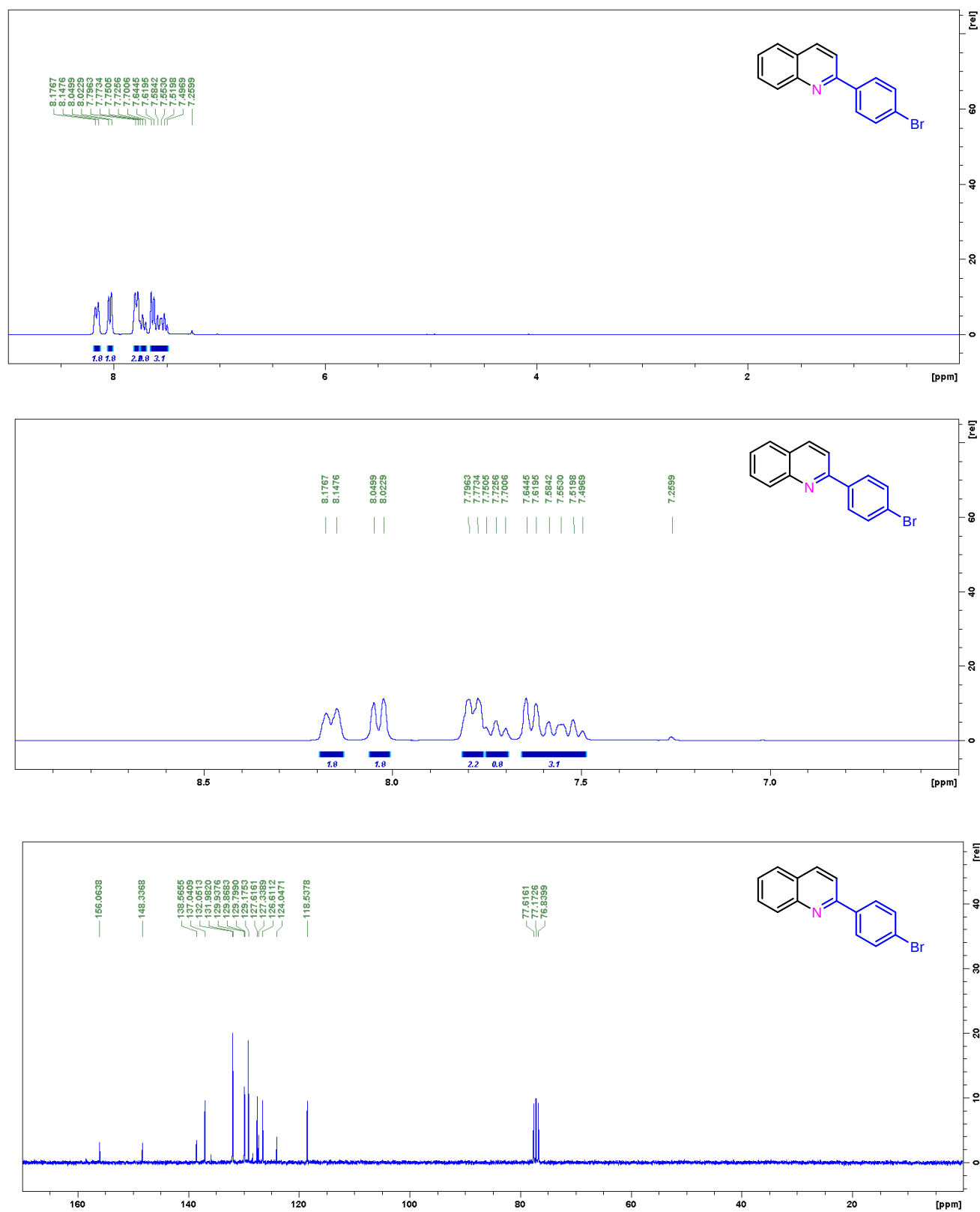


Figure S14. ¹H and ¹³C NMR spectra of compound **8al** in CDCl₃.

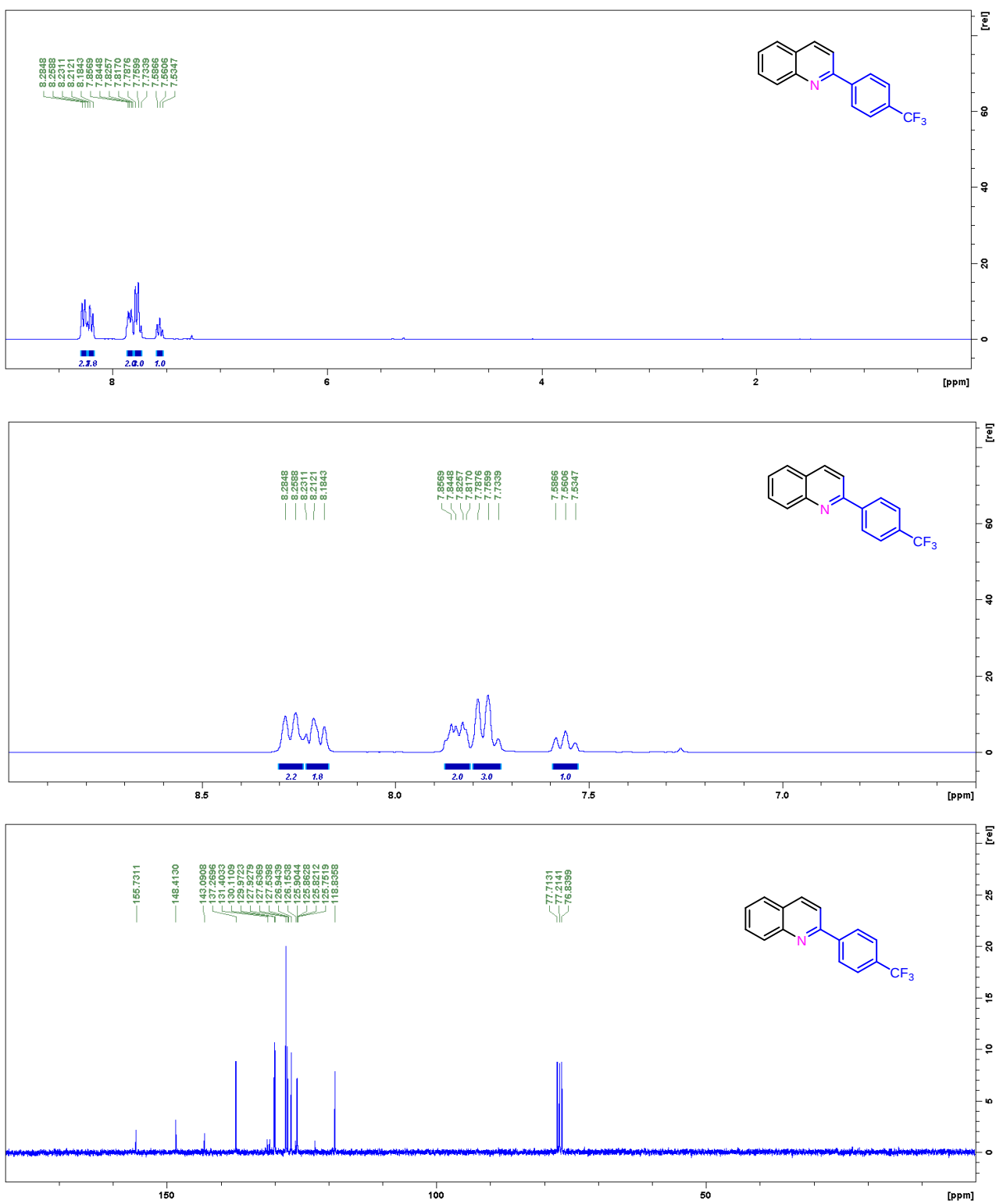


Figure S15. ¹H and ¹³C NMR spectra of compound **8am** in CDCl₃.

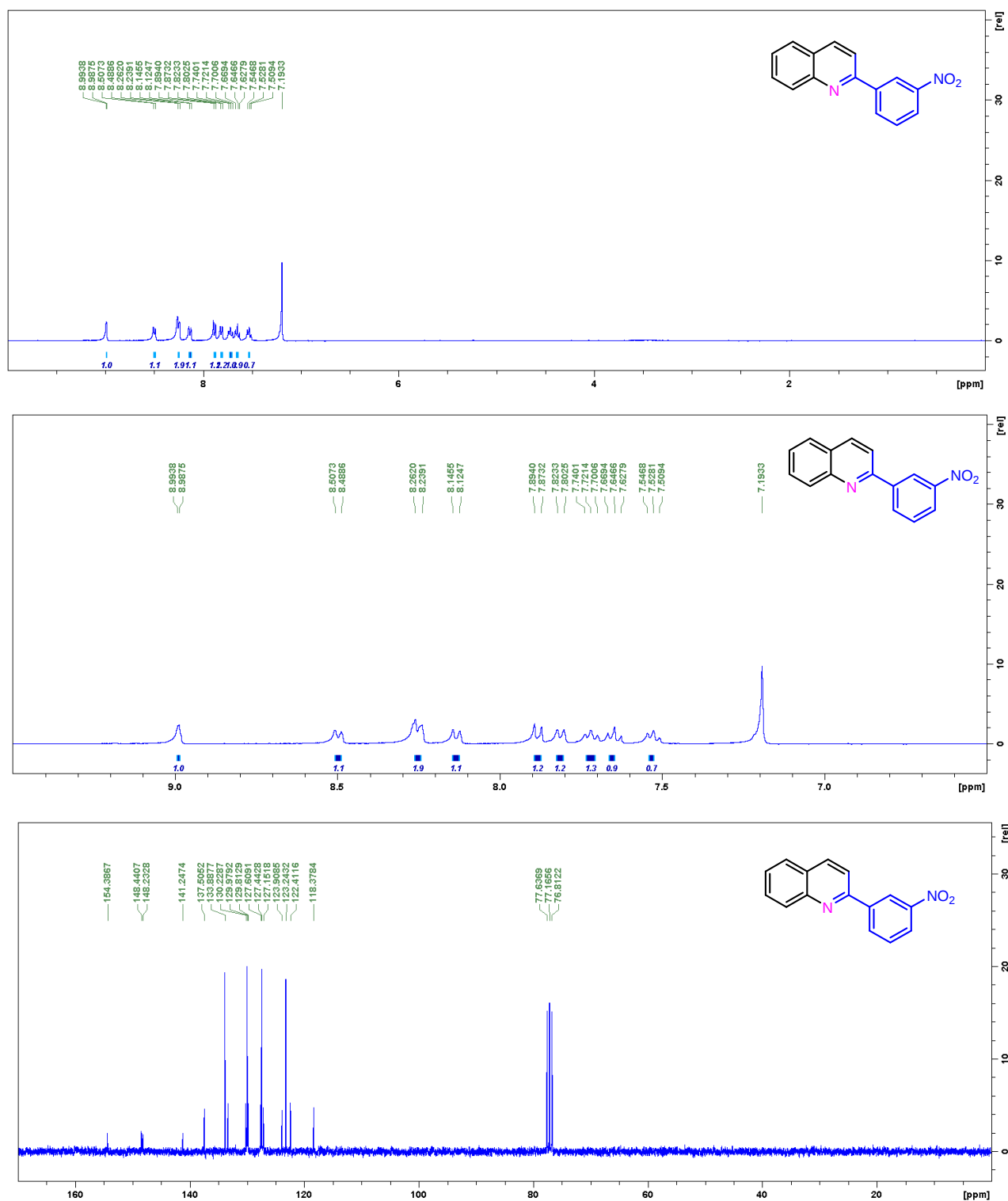


Figure S16. ¹H and ¹³C NMR spectra of compound **8an** in CDCl₃.

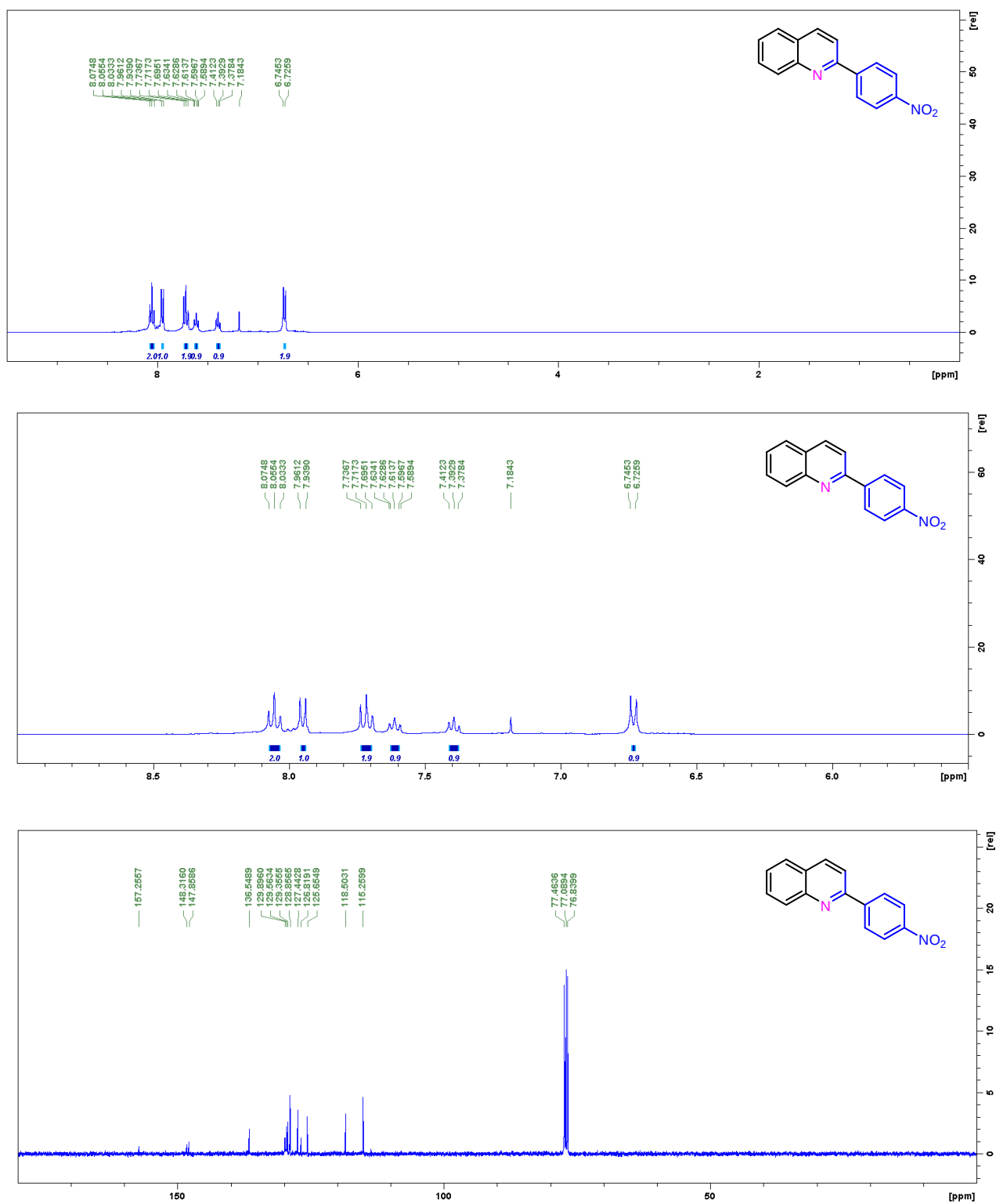


Figure S17. ¹H and ¹³C NMR spectra of compound **8ao** in CDCl₃.

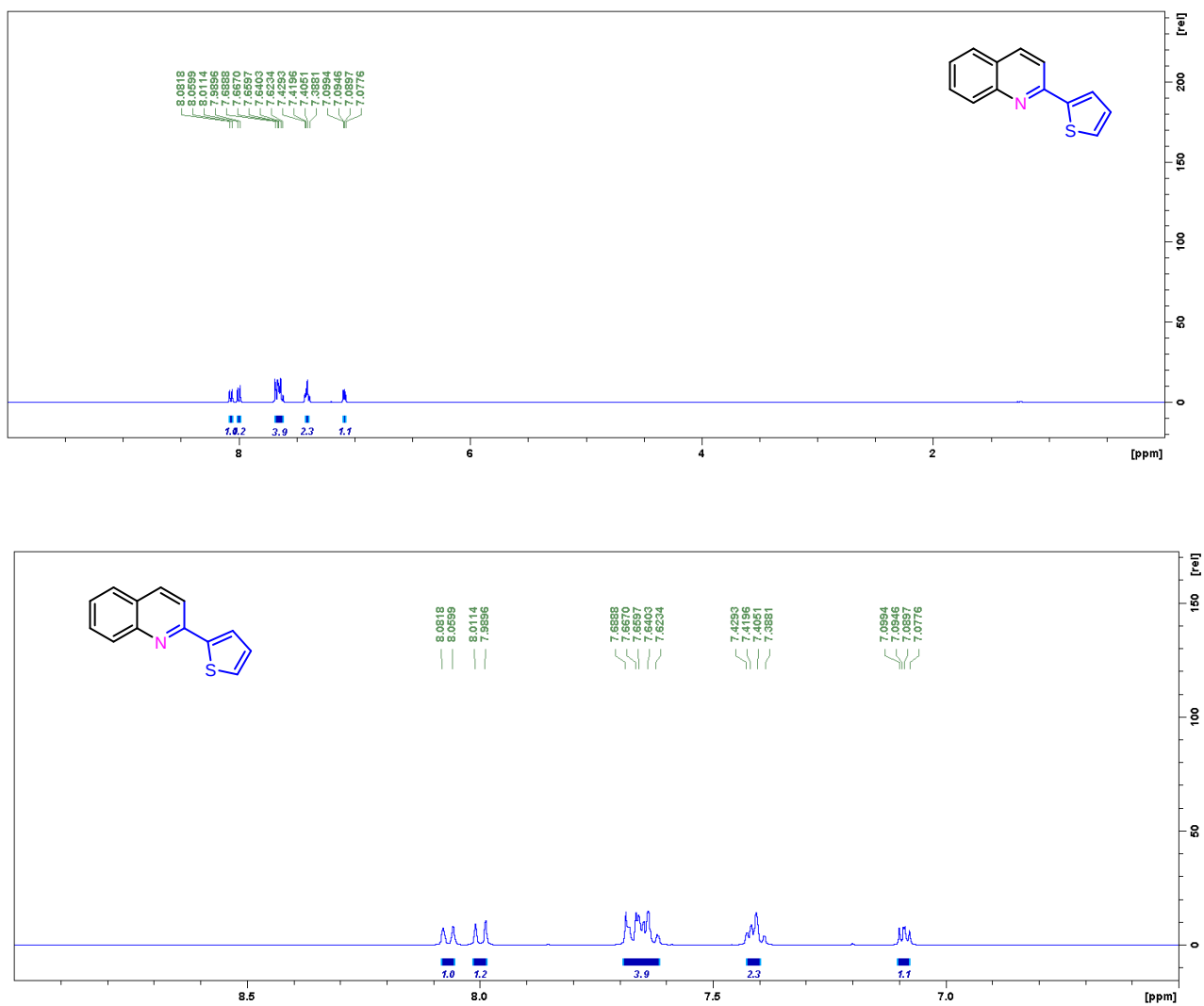


Figure S18. ¹H NMR spectrum of compound **8ap** in CDCl₃.

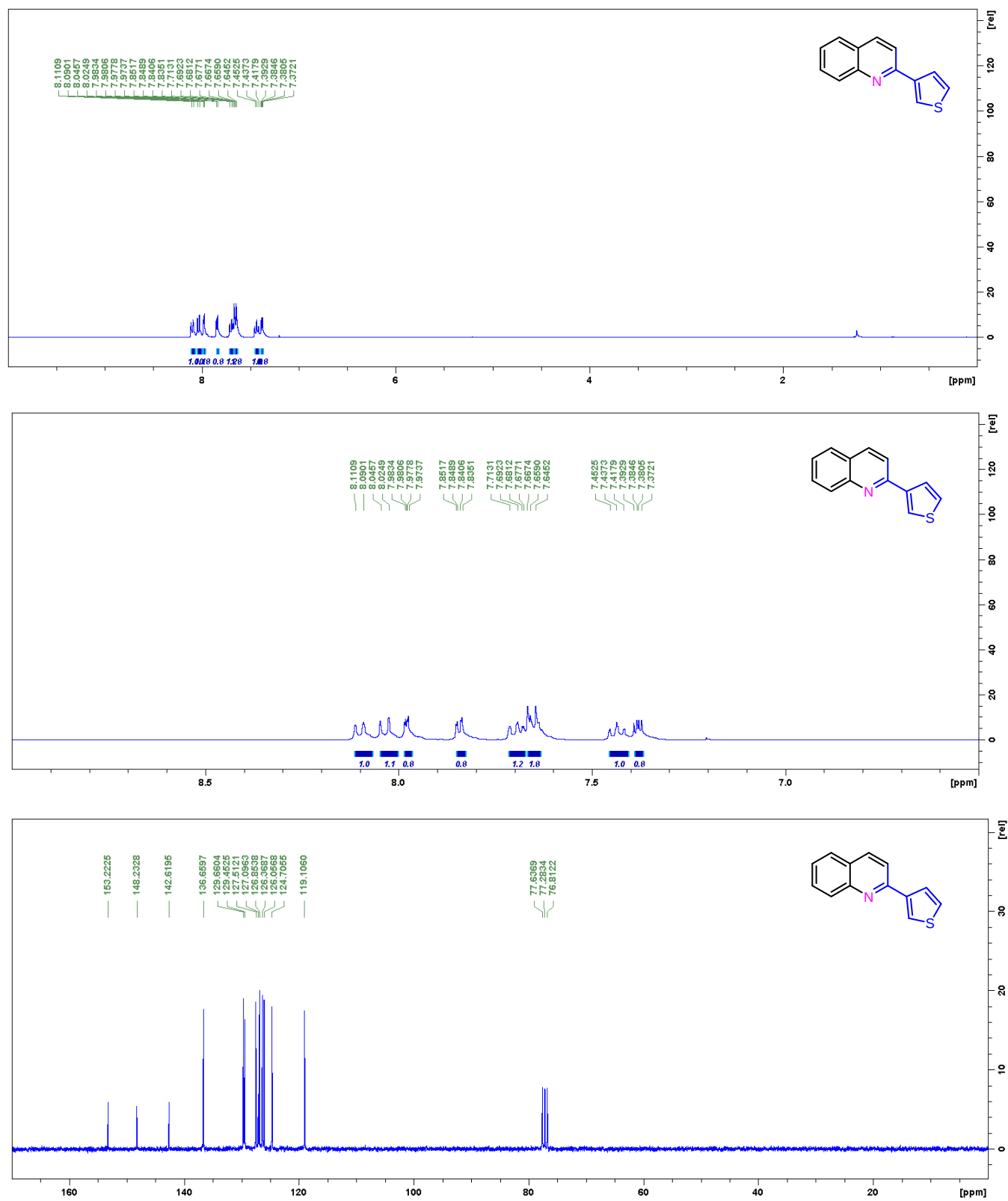


Figure S19. ¹H and ¹³C NMR spectra of compound **8aq** in CDCl₃.

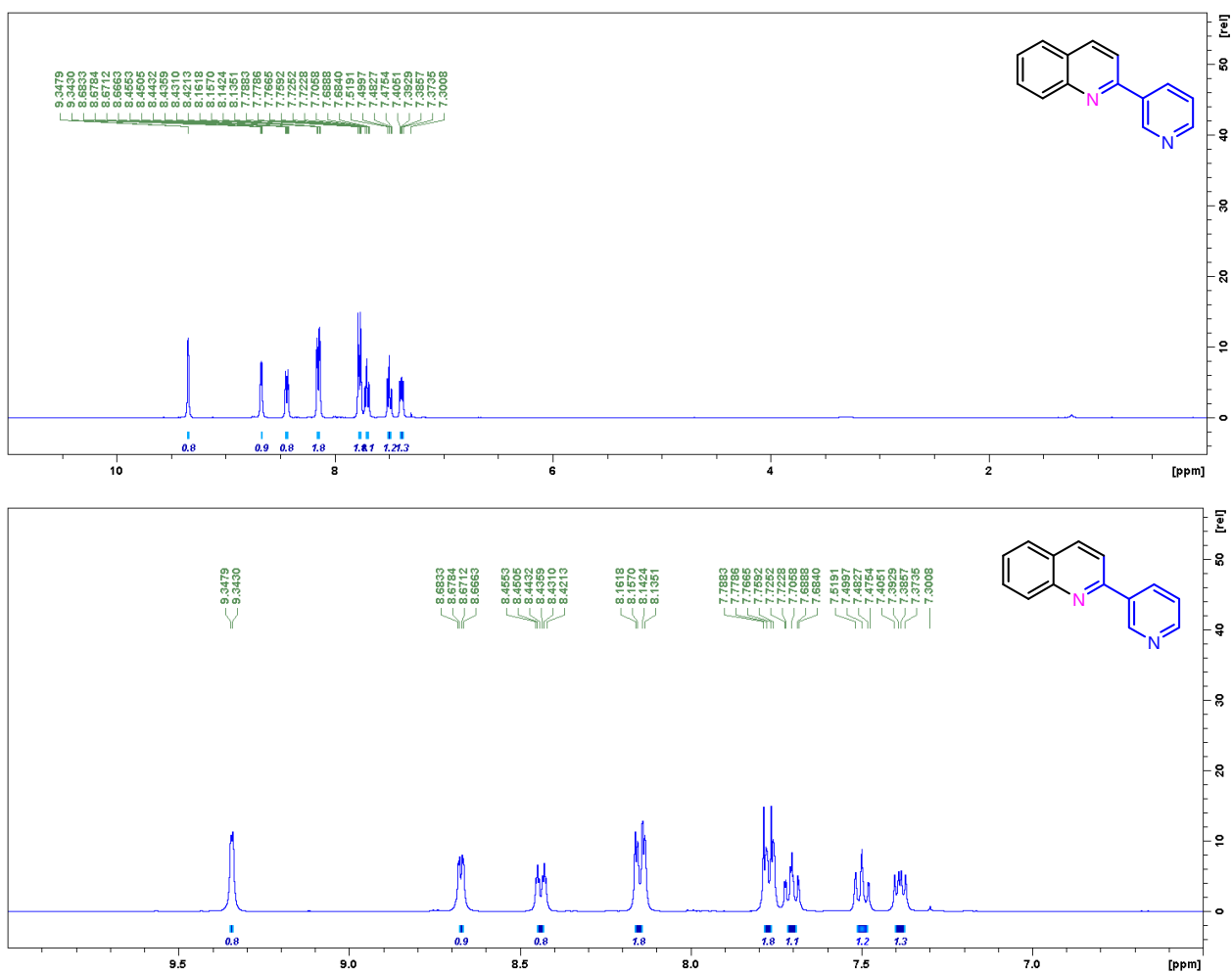


Figure S20. ^1H NMR spectrum of compound **8ar** in CDCl_3 .

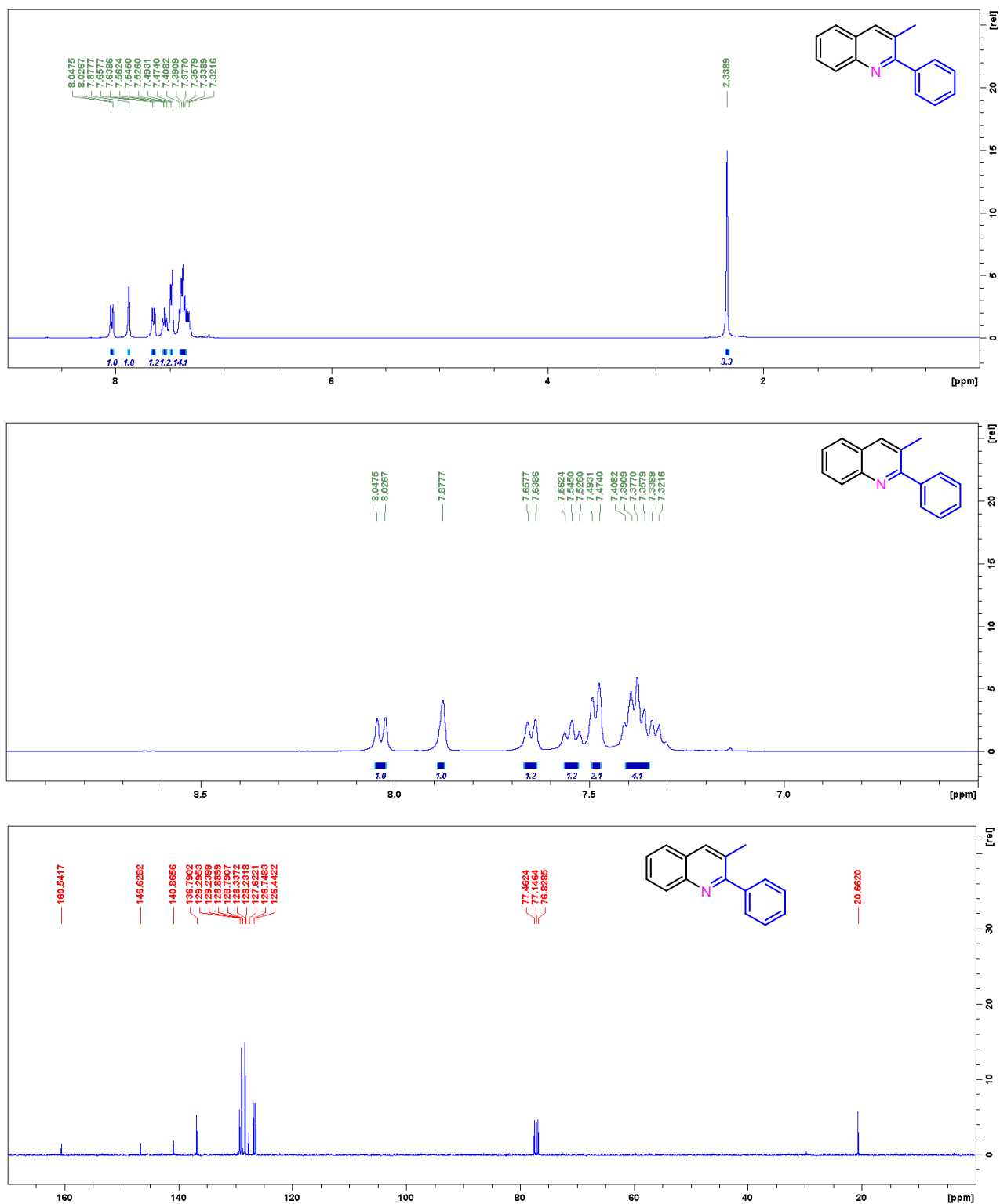


Figure S21. ^1H and ^{13}C NMR spectra of compound **8as** in CDCl_3 .

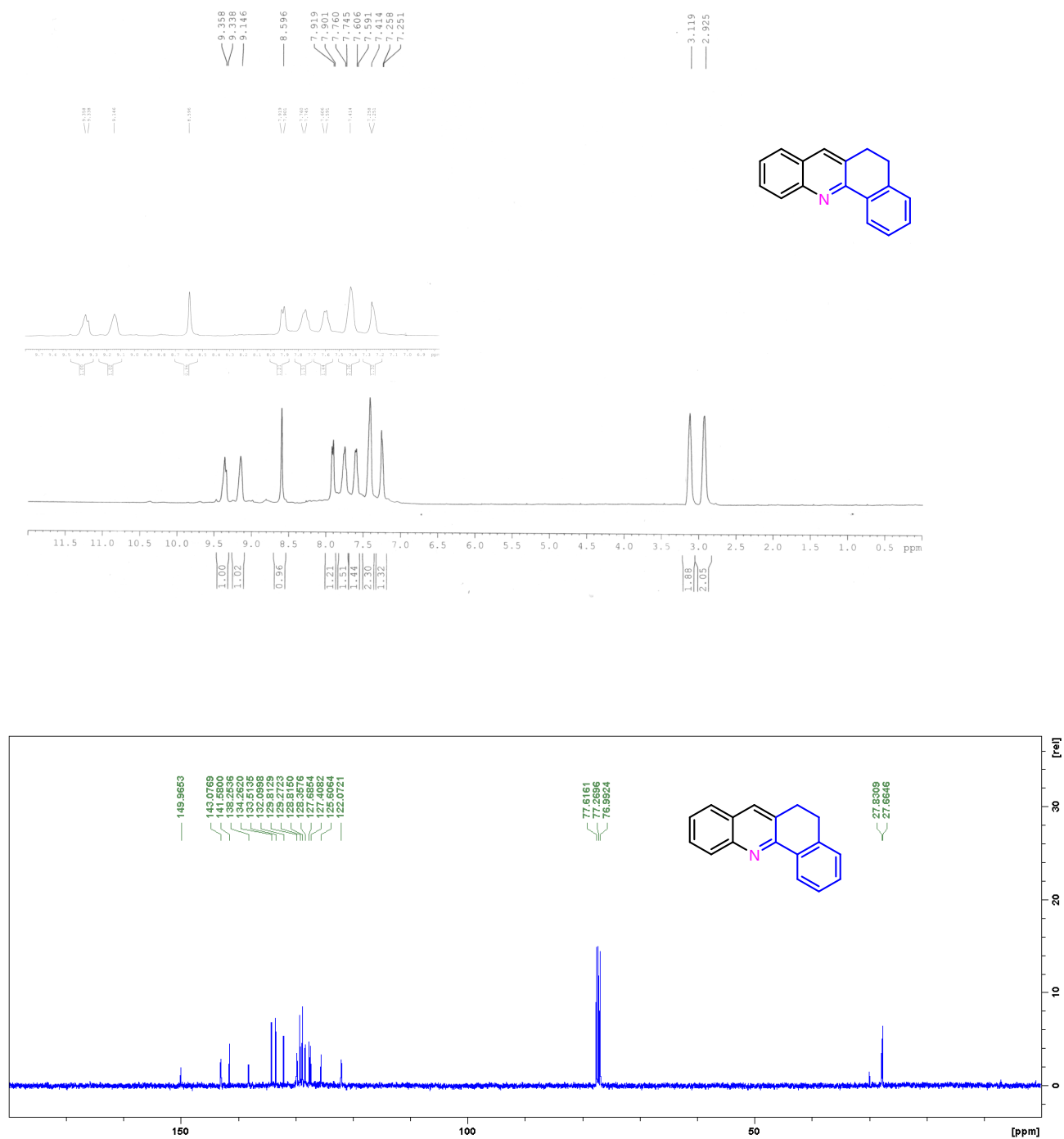


Figure S22. ¹H and ¹³C NMR spectra of compound **8at** in CDCl₃.

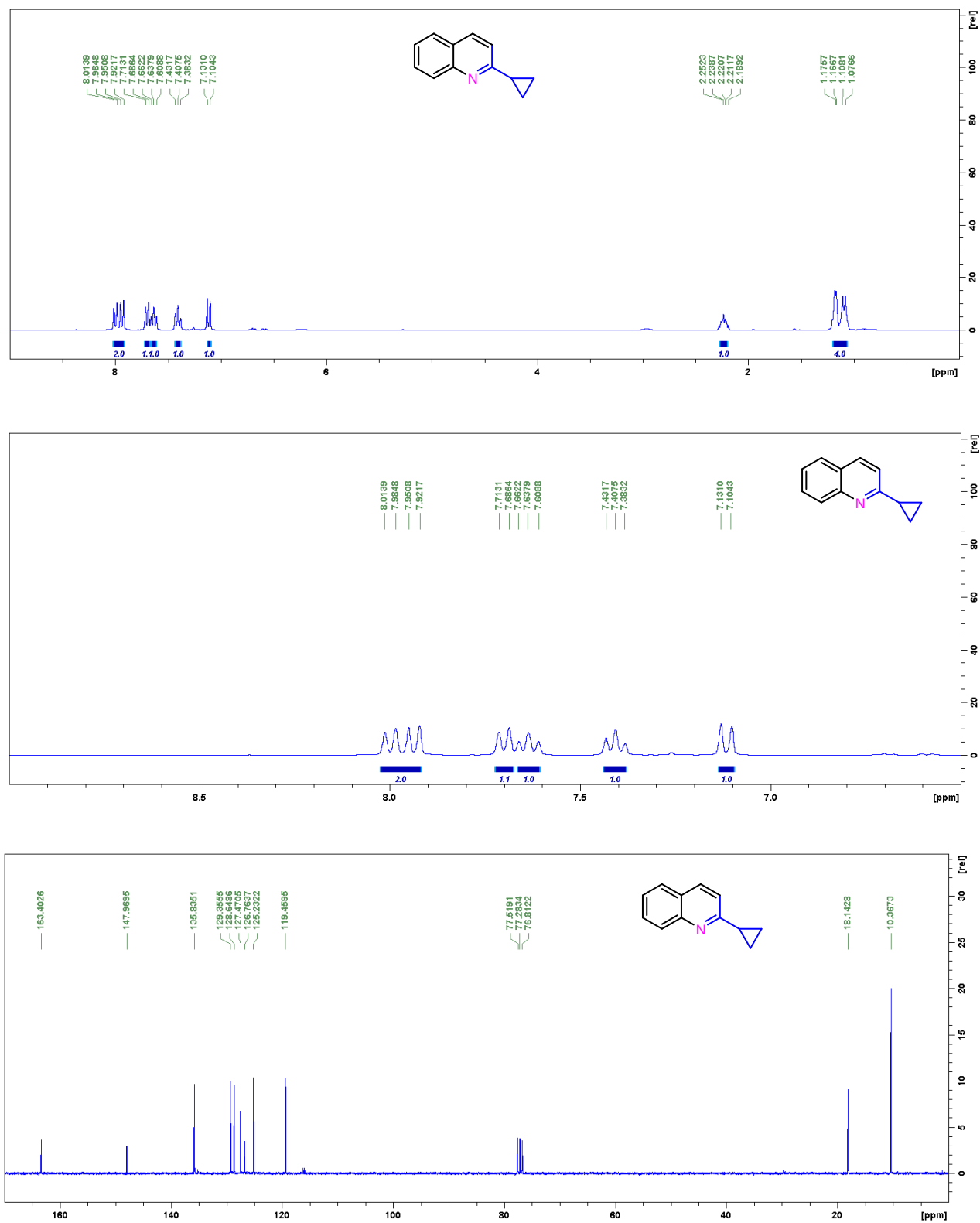


Figure S23. ¹H and ¹³C NMR spectra of compound **8au** in CDCl₃.

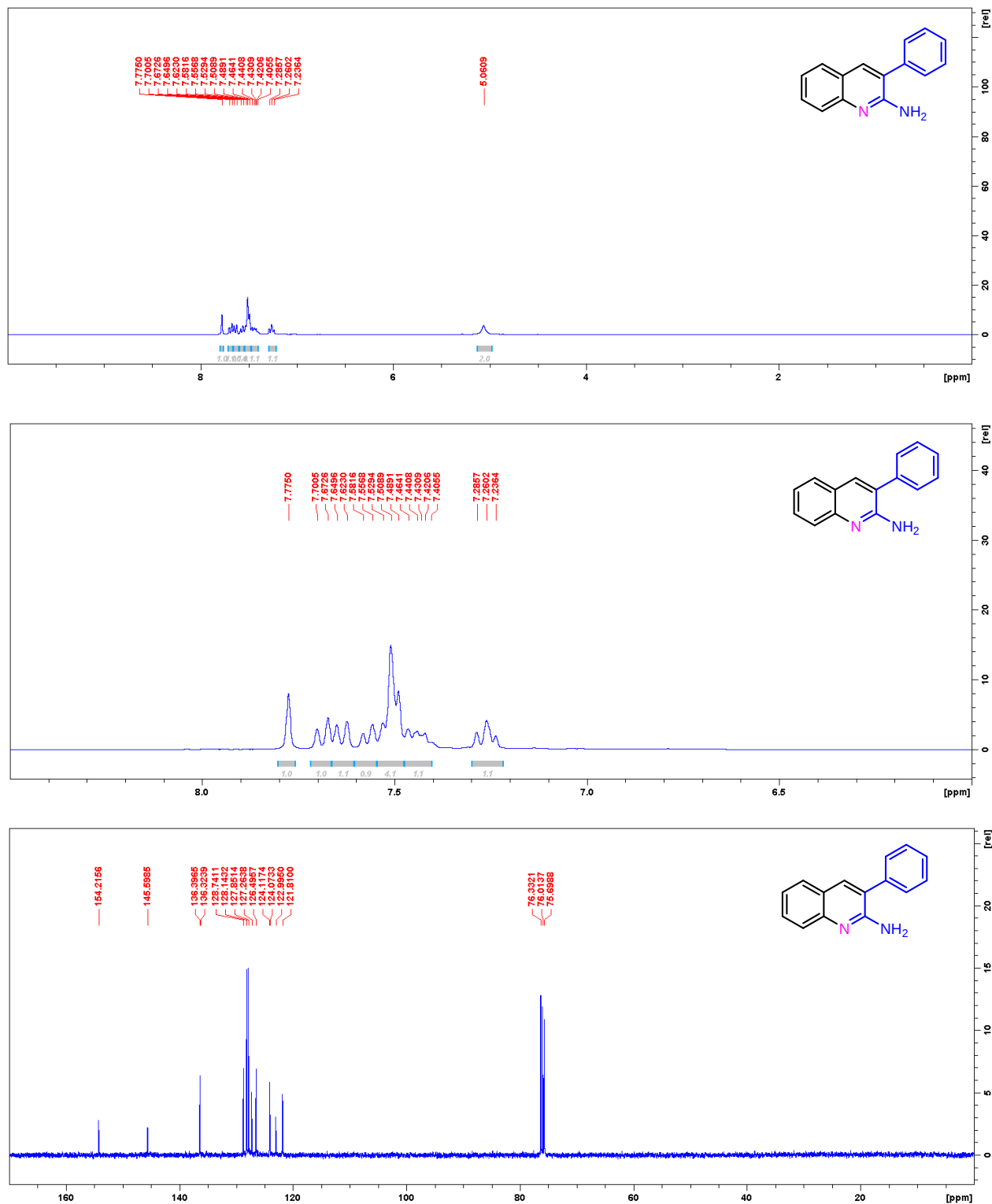


Figure S24. ¹H and ¹³C NMR spectra of compound **9aa** in CDCl₃.

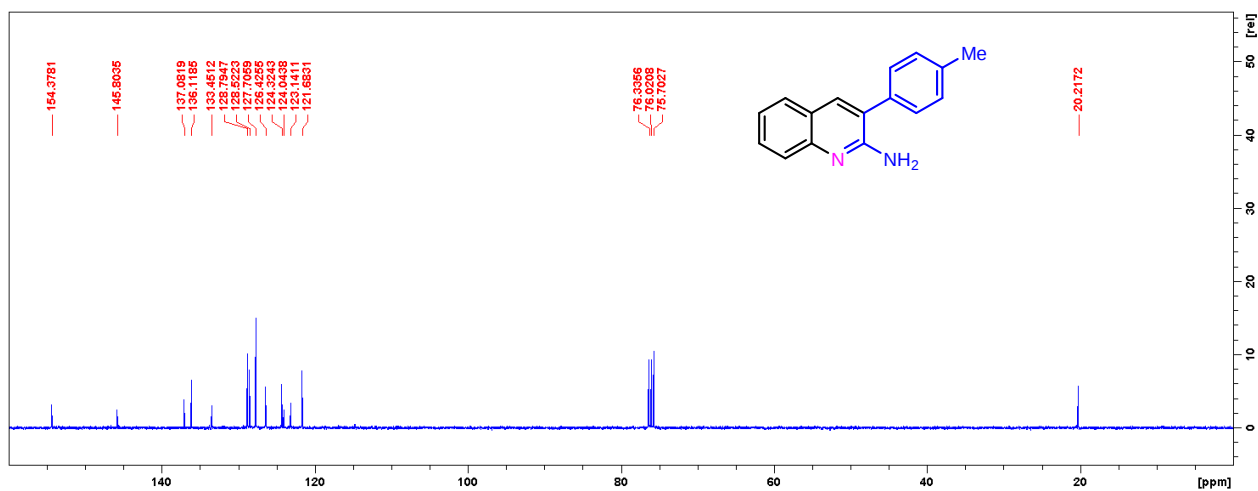
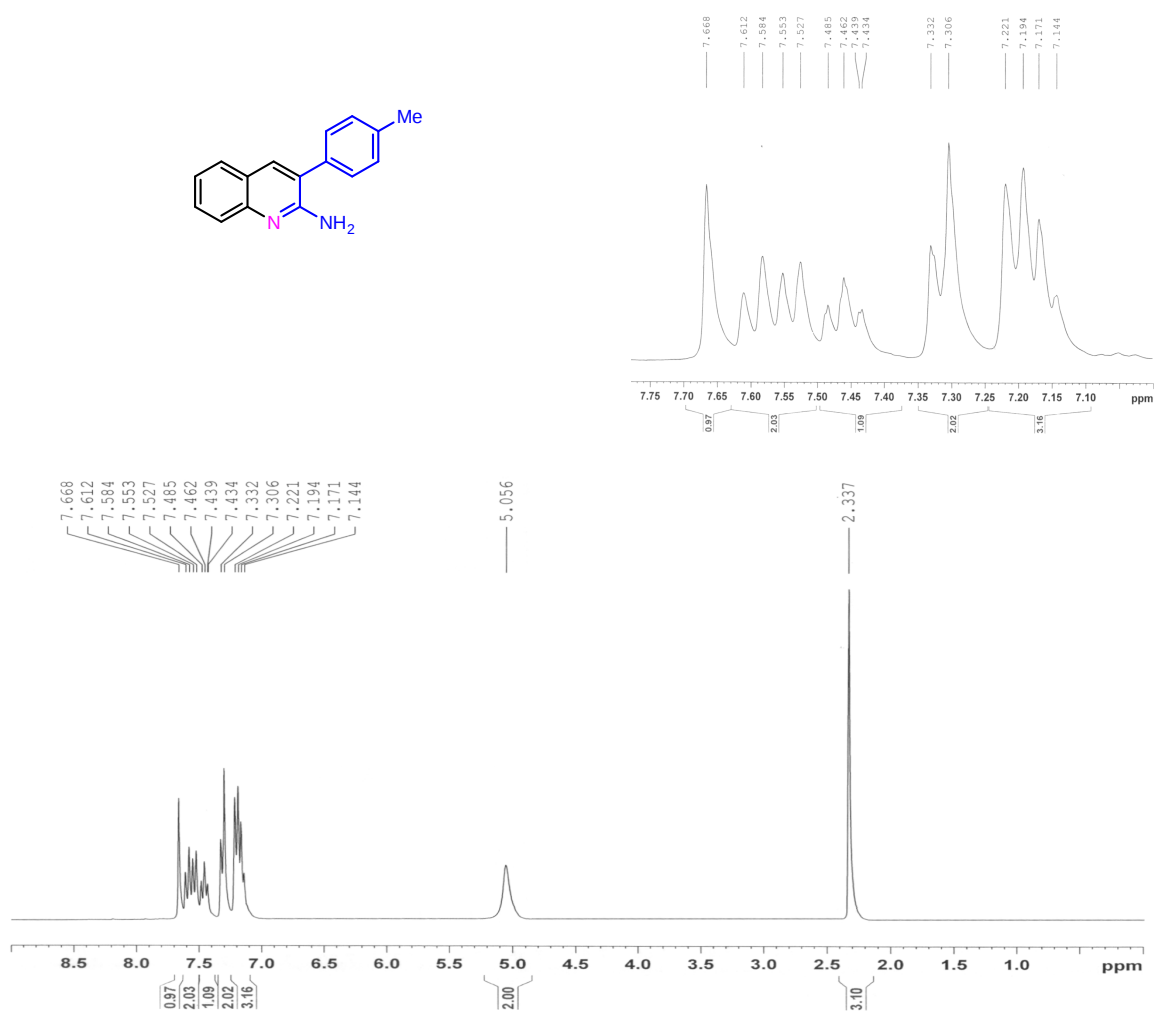


Figure S25. ¹H and ¹³C NMR spectra of compound 9ab in CDCl₃.

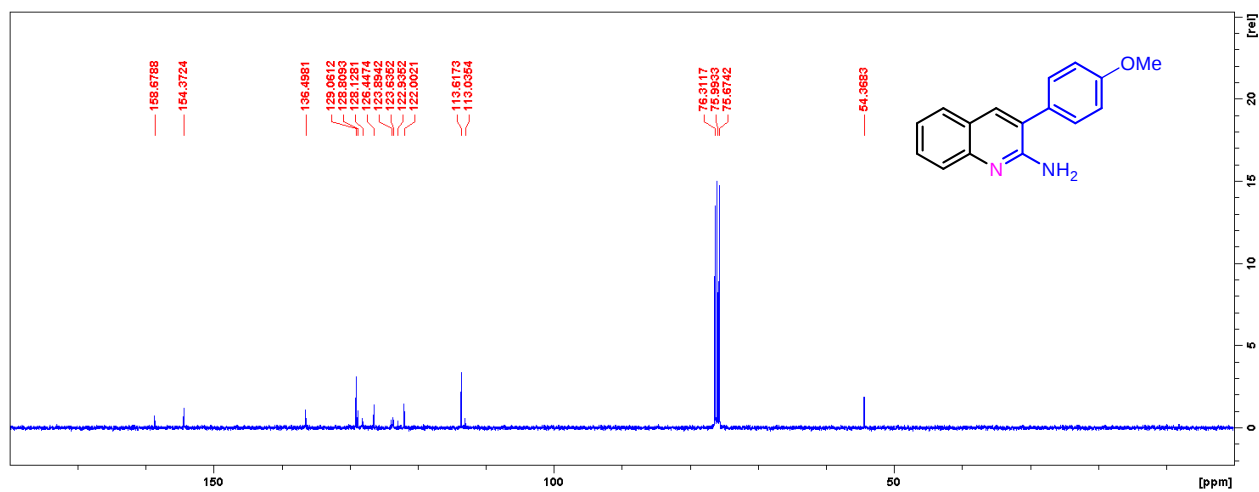
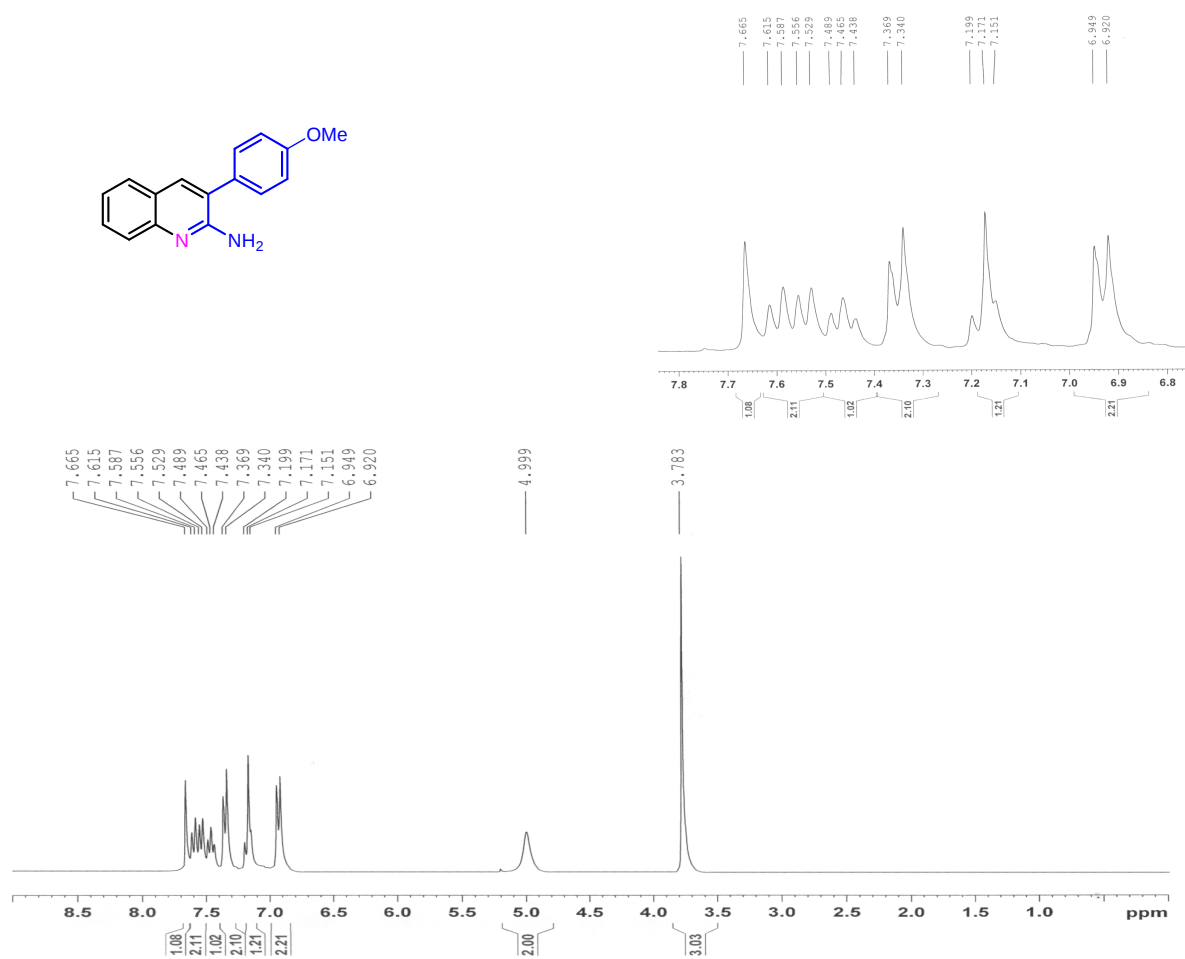


Figure S26. ¹H and ¹³C NMR spectra of compound **9ac** in CDCl₃.

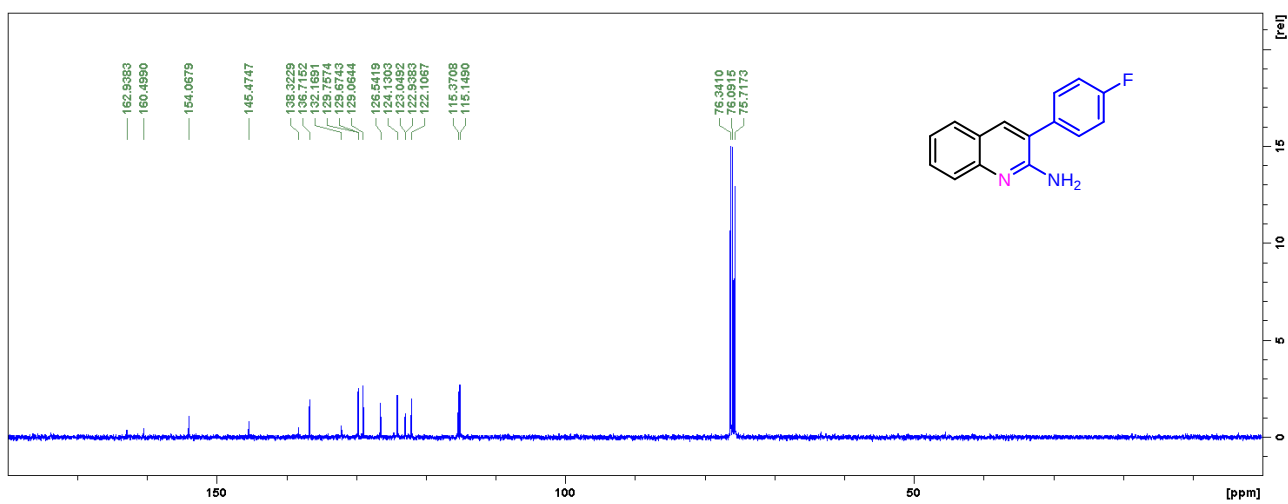
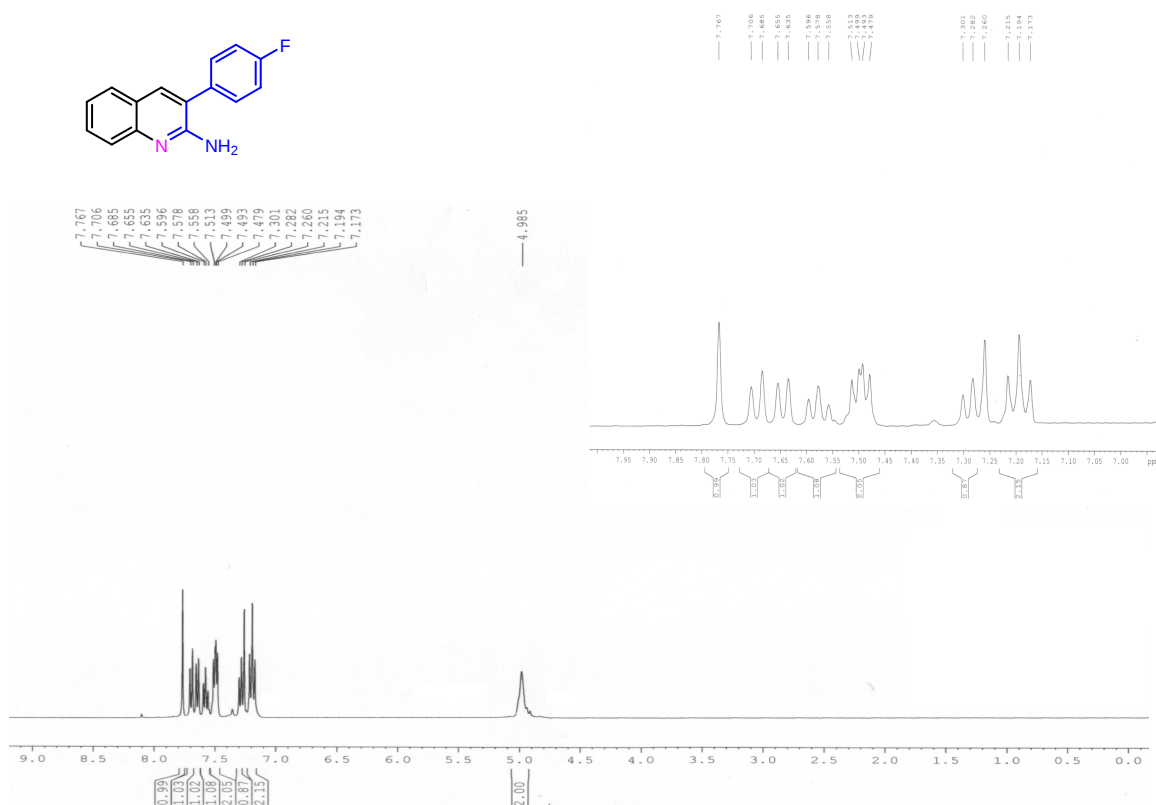


Figure S27. ¹H and ¹³C NMR spectra of compound **9ad** in CDCl₃.

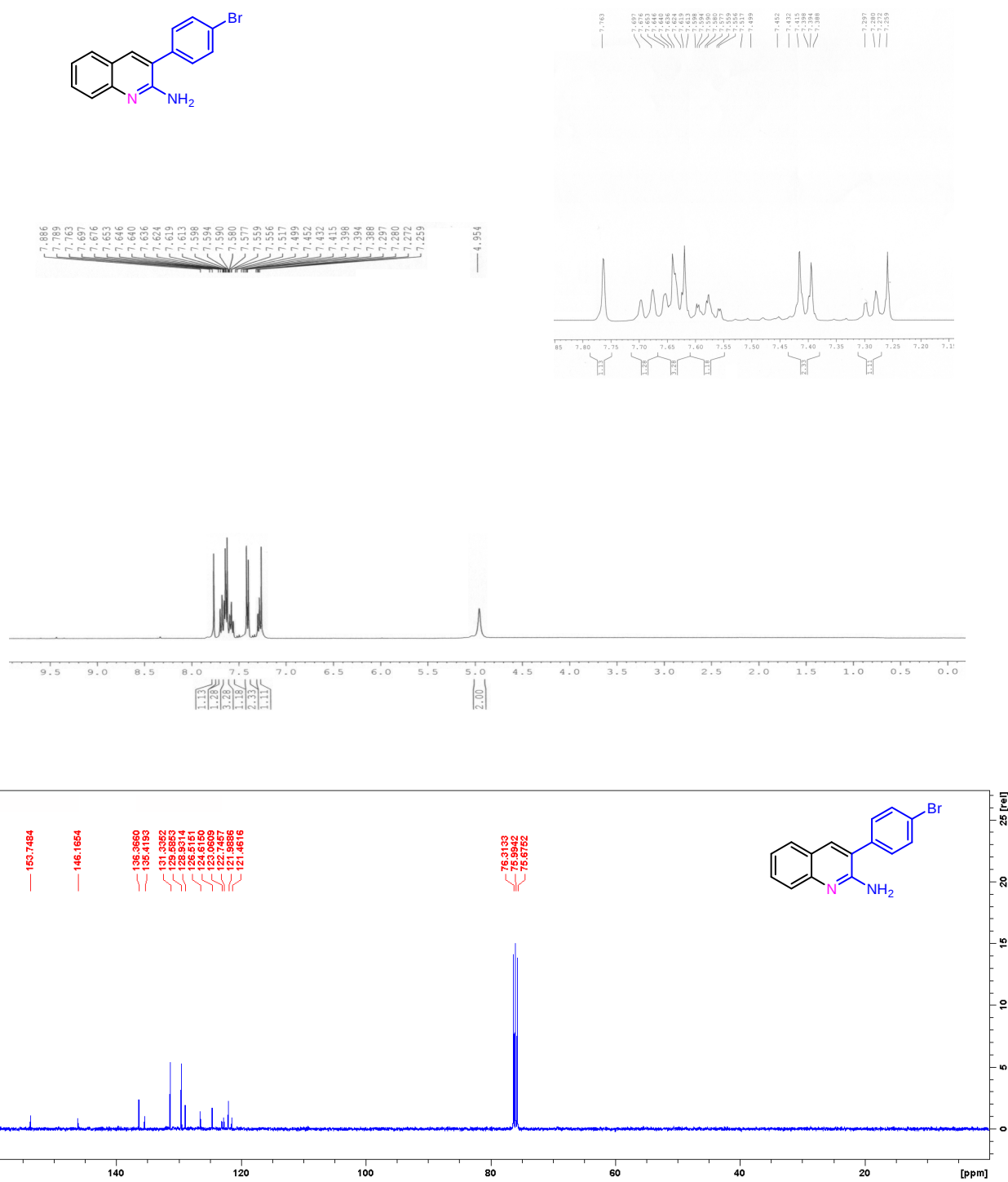


Figure S28. ¹H and ¹³C NMR spectra of compound **9ae** in CDCl₃

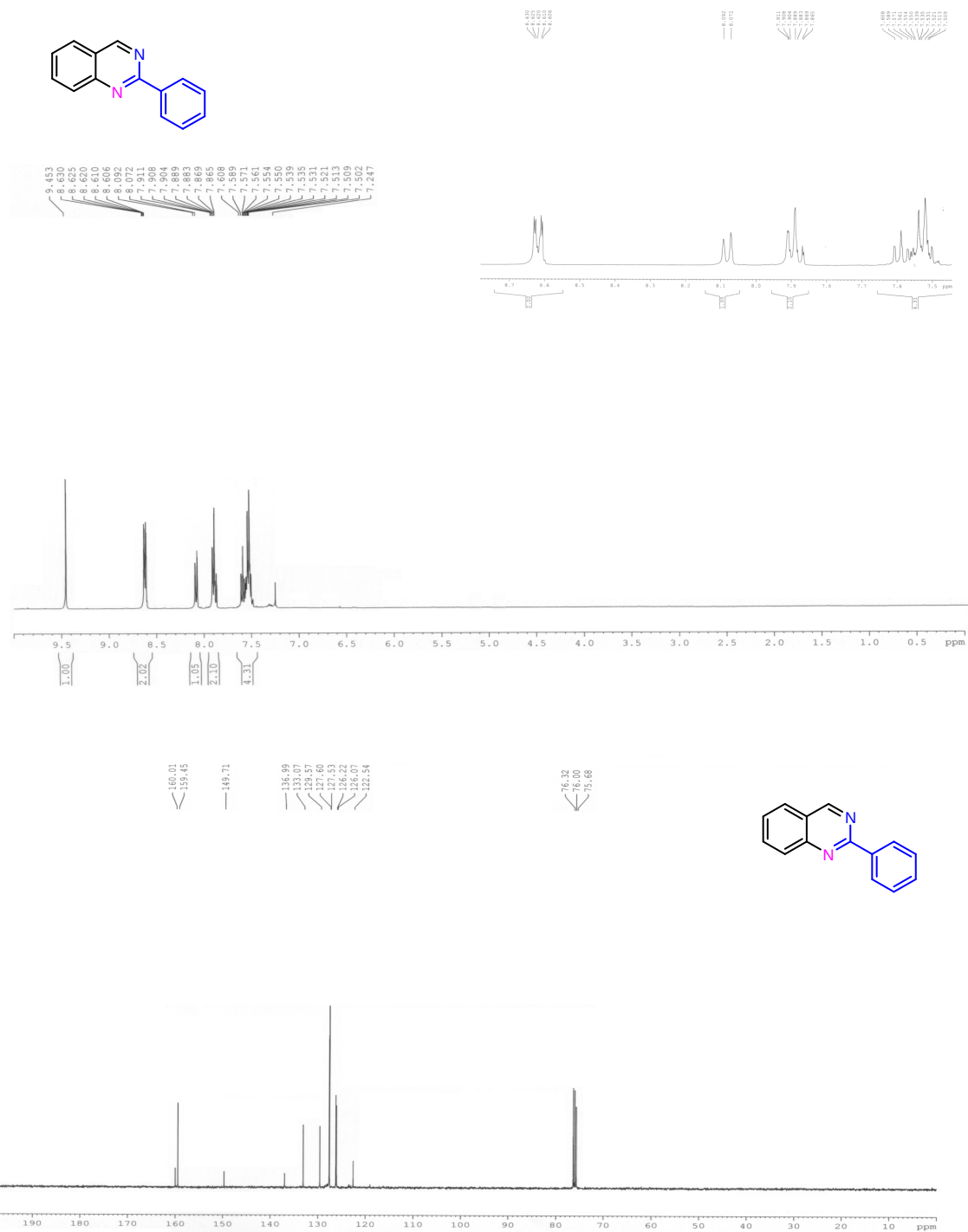


Figure S29. ¹H and ¹³C NMR spectra of compound **10aa** in CDCl₃.

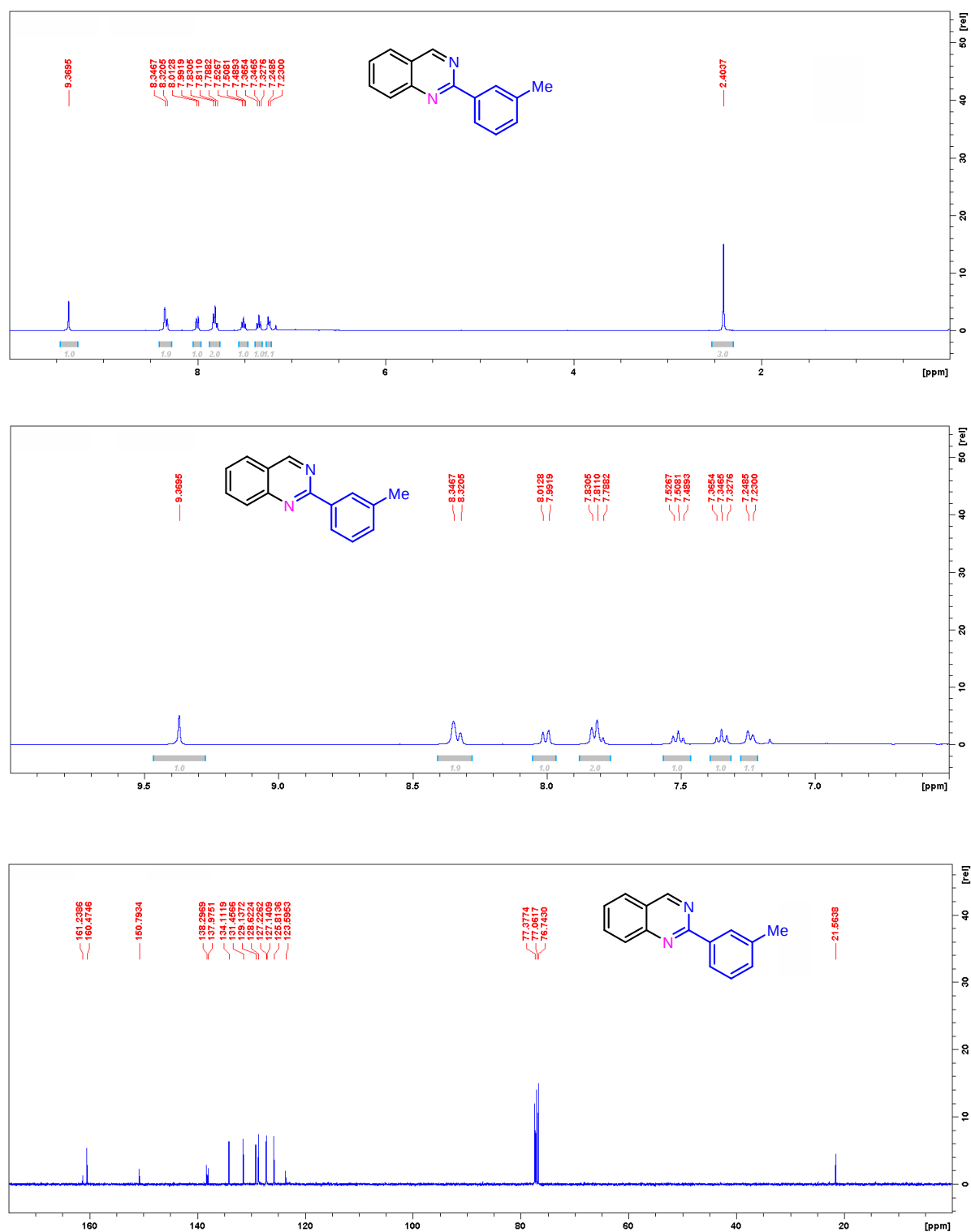


Figure S30. ¹H and ¹³C NMR spectra of compound **10ab** in CDCl₃.

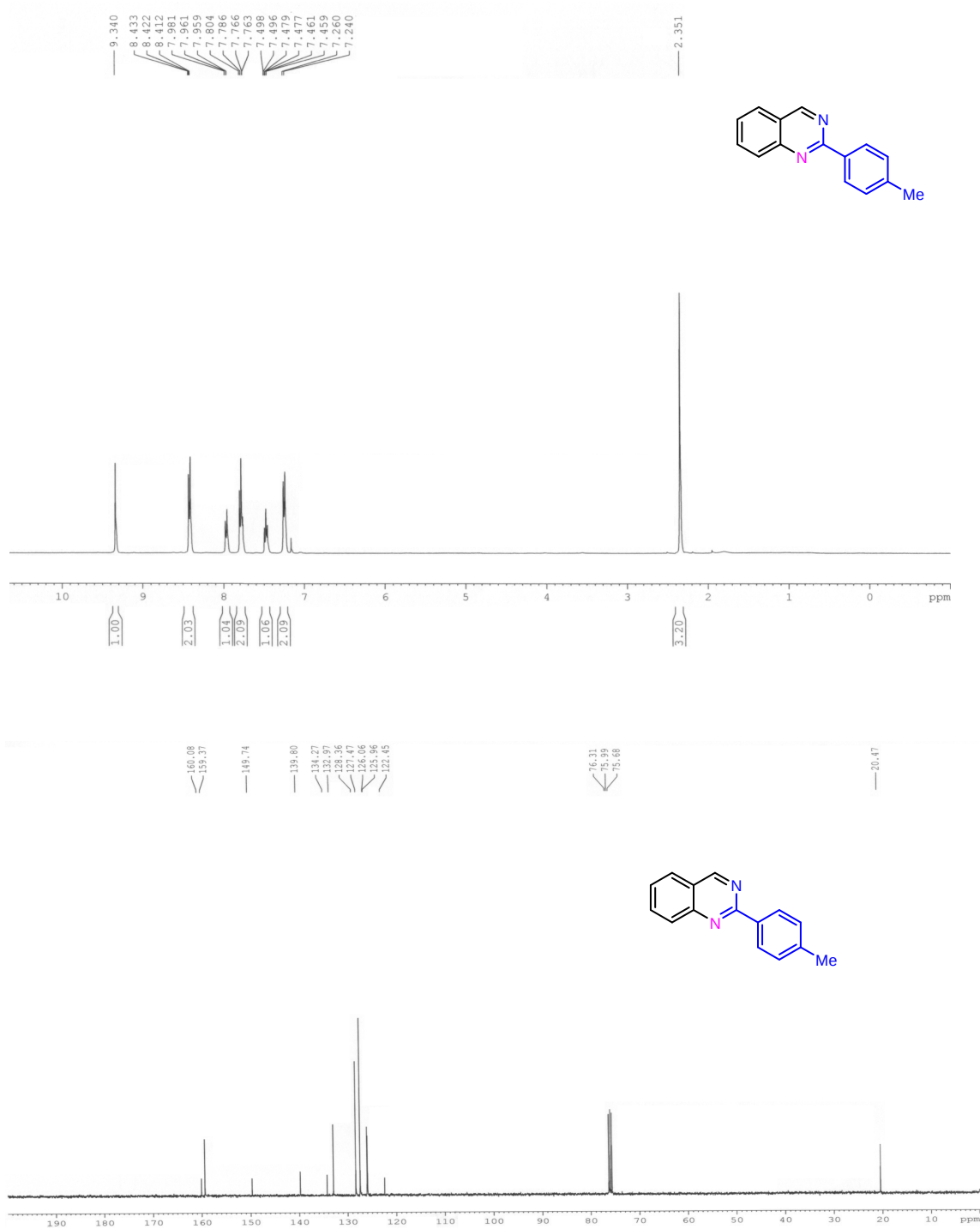


Figure S31. ¹H and ¹³C NMR spectra of compound **10ac** in CDCl₃.

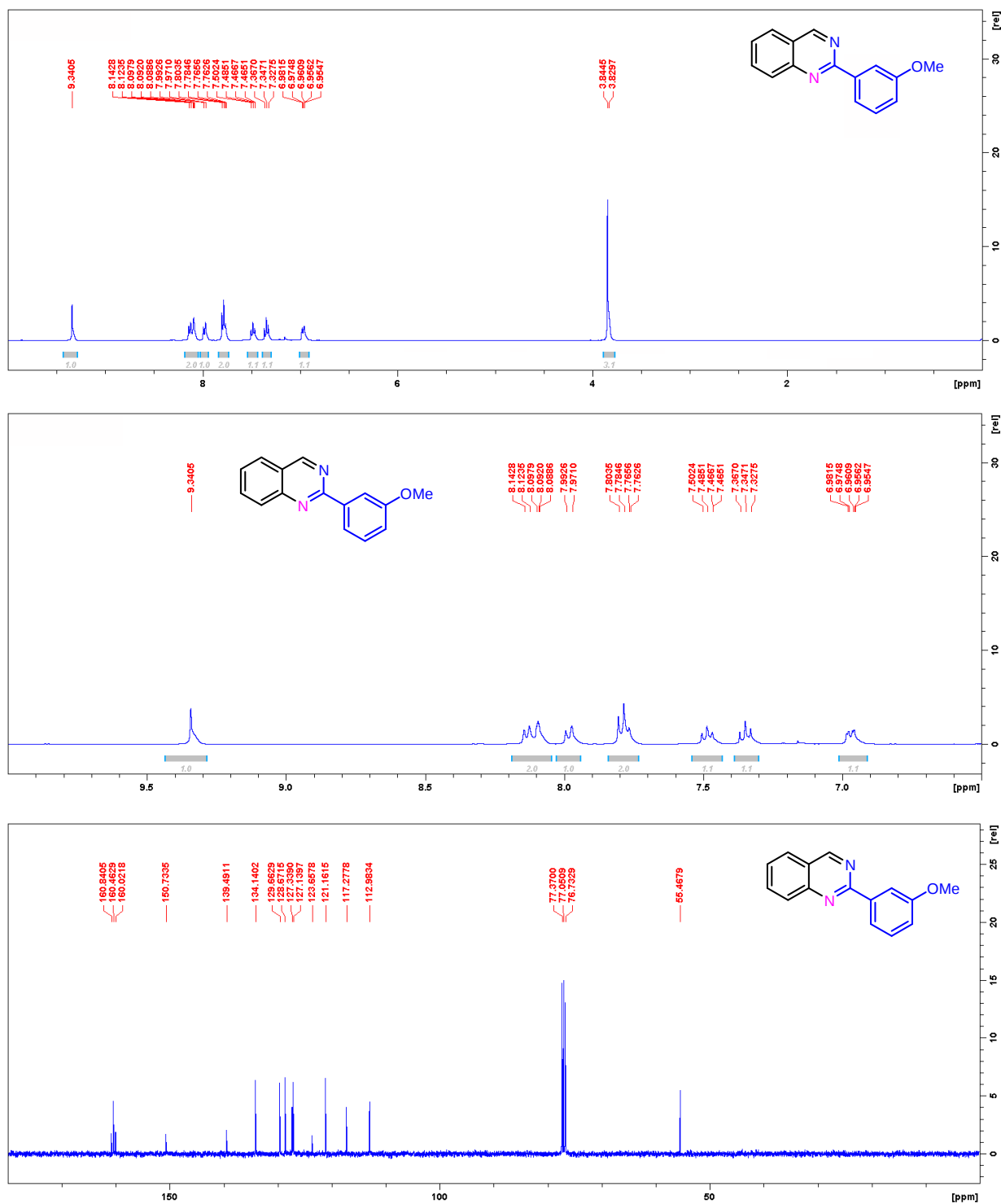


Figure S32. ¹H and ¹³C NMR spectra of compound **10ad** in CDCl₃.

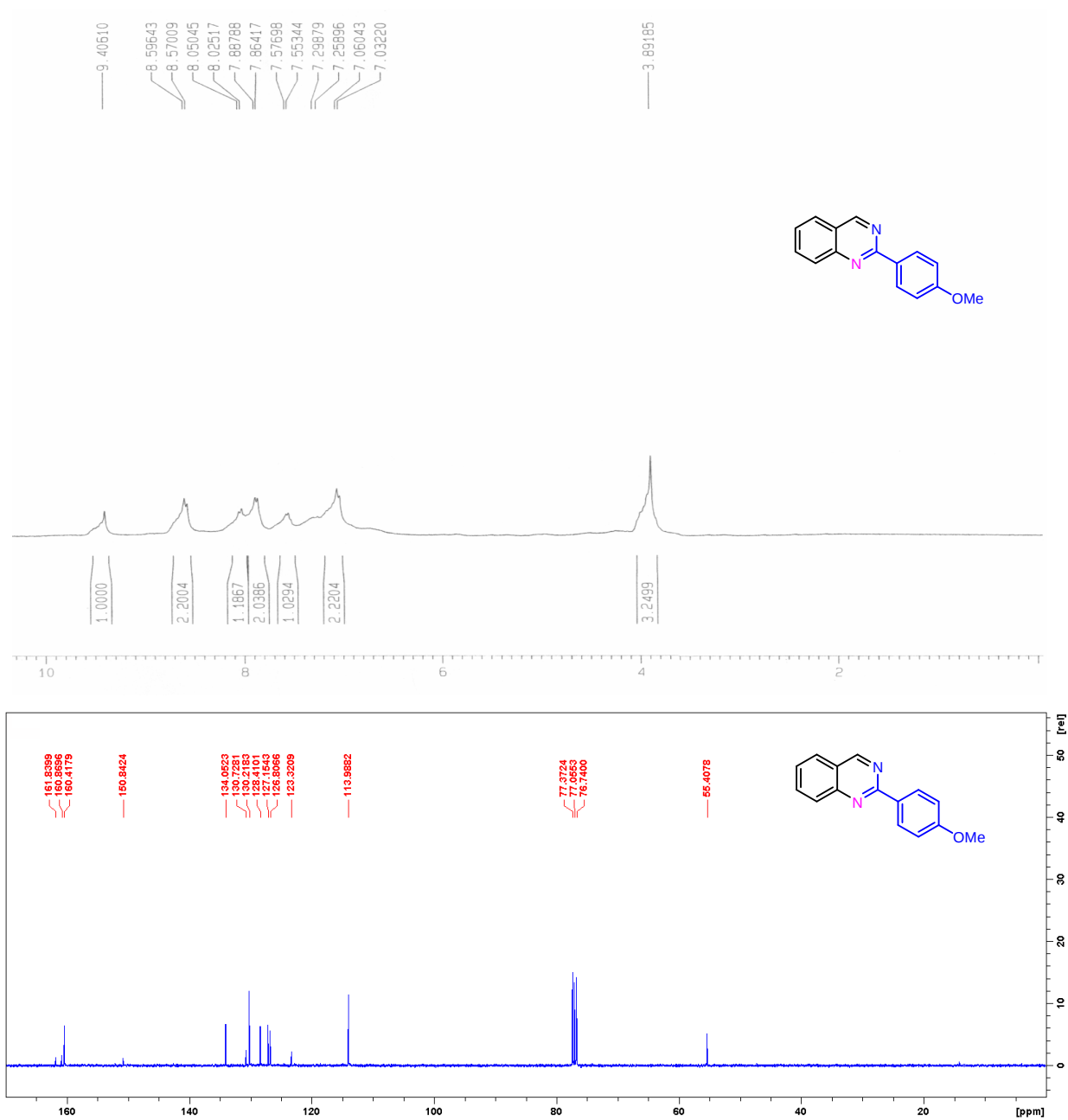


Figure S33. ¹H and ¹³C NMR spectra of compound **10ae** in CDCl₃.

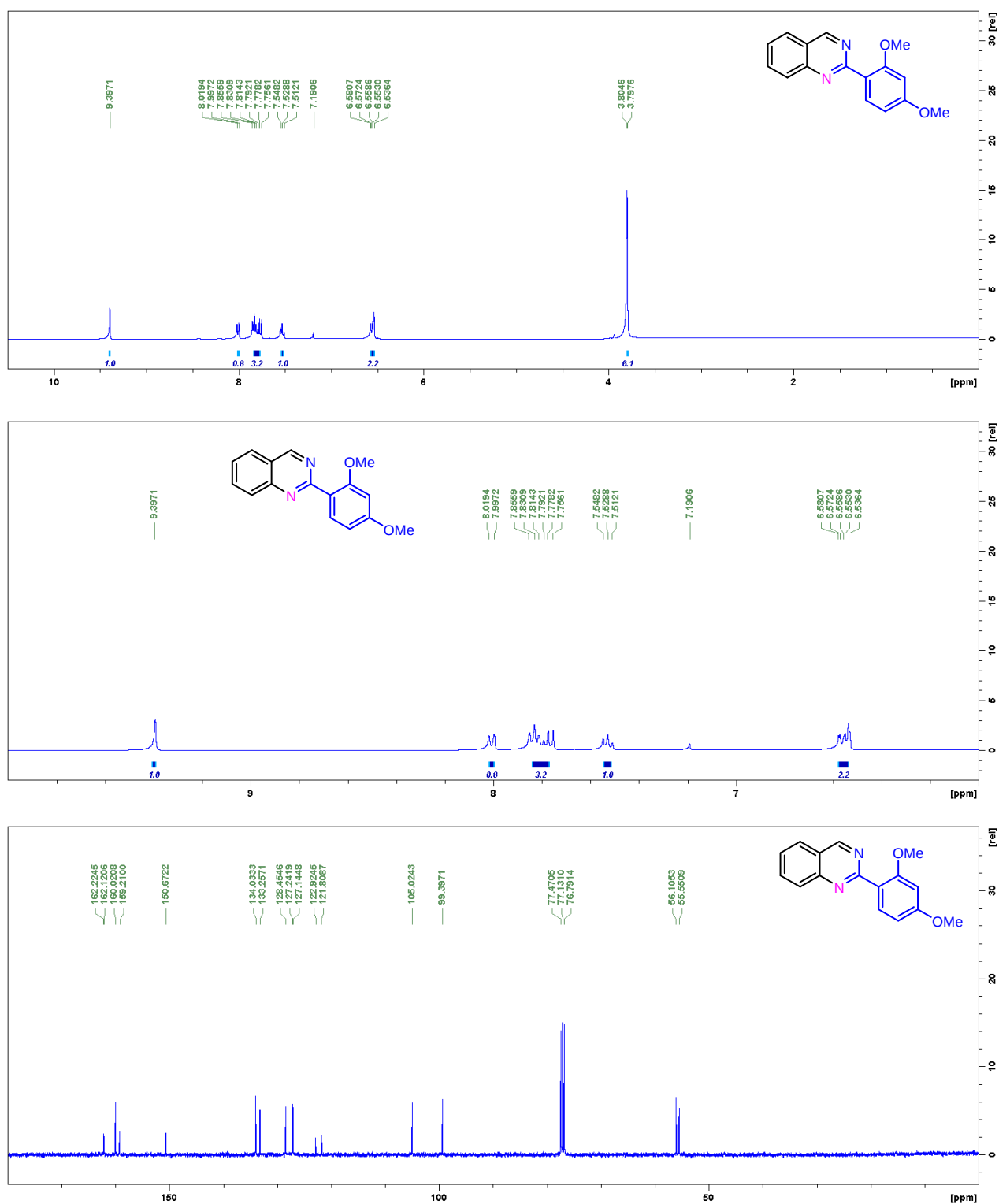


Figure S34. ¹H and ¹³C NMR spectra of compound **10af** in CDCl₃.

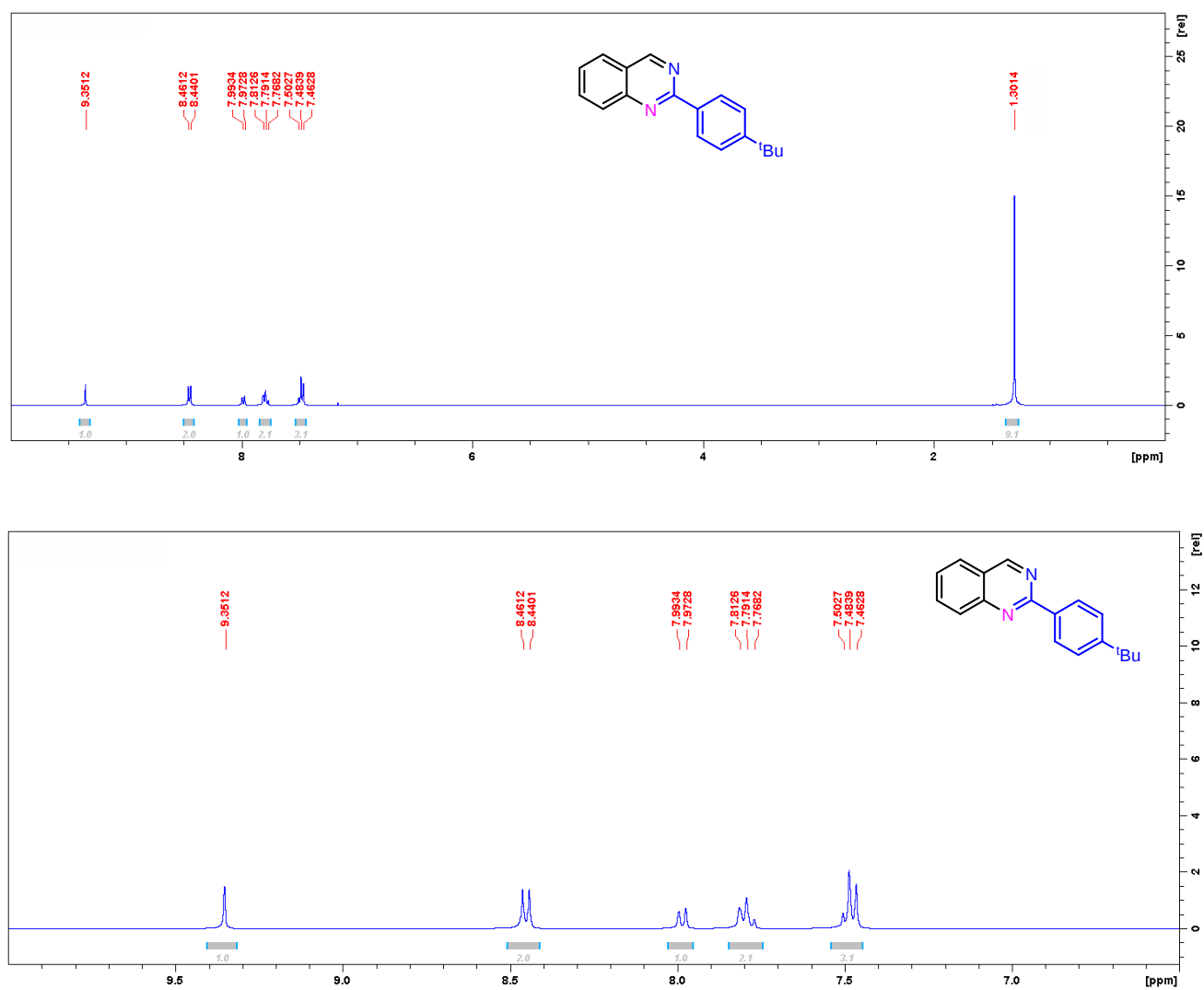


Figure S35. ^1H NMR spectrum of compound **10ag** in CDCl_3 .

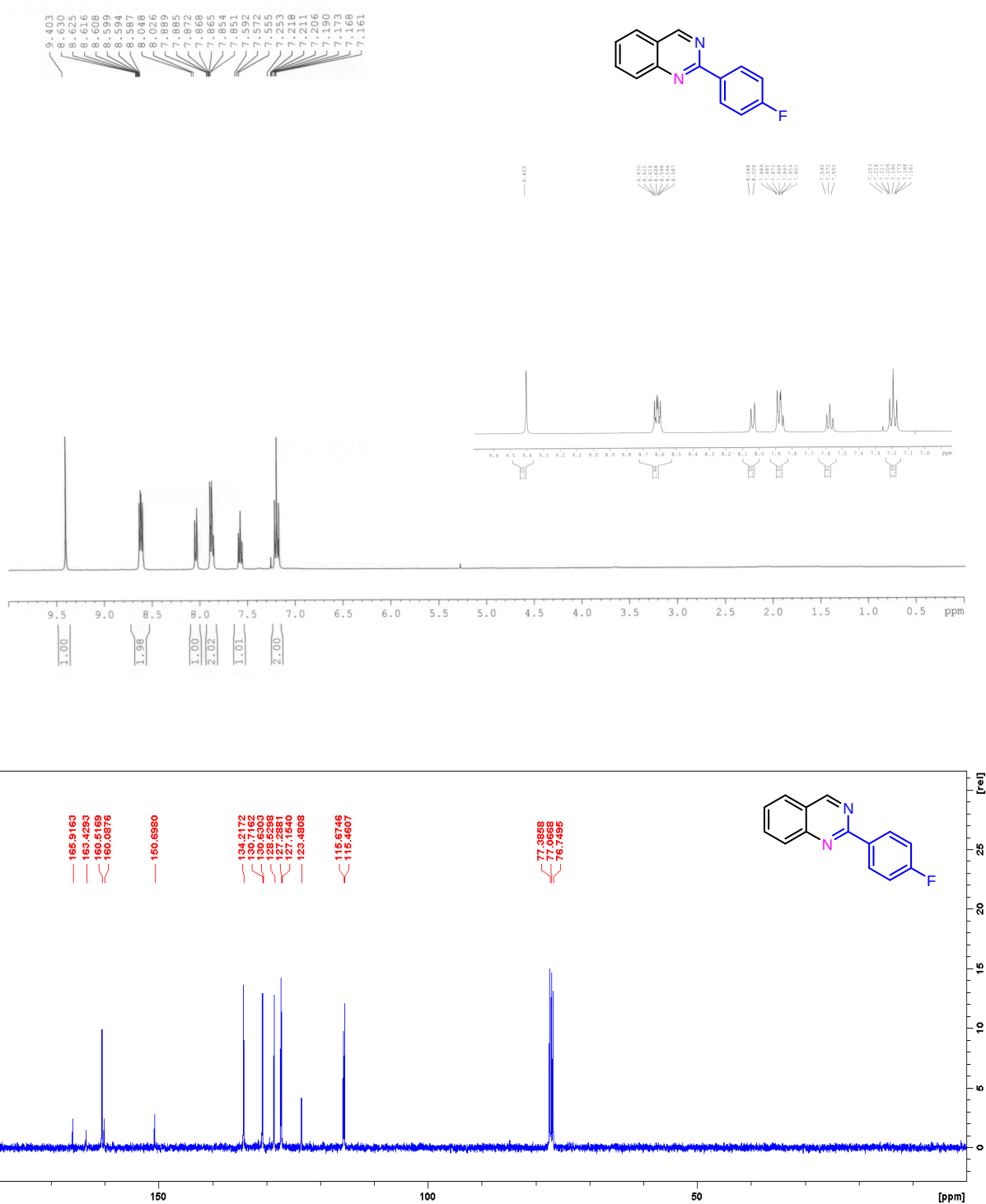


Figure S36. ¹H and ¹³C NMR spectra of compound **10ah** in CDCl₃.

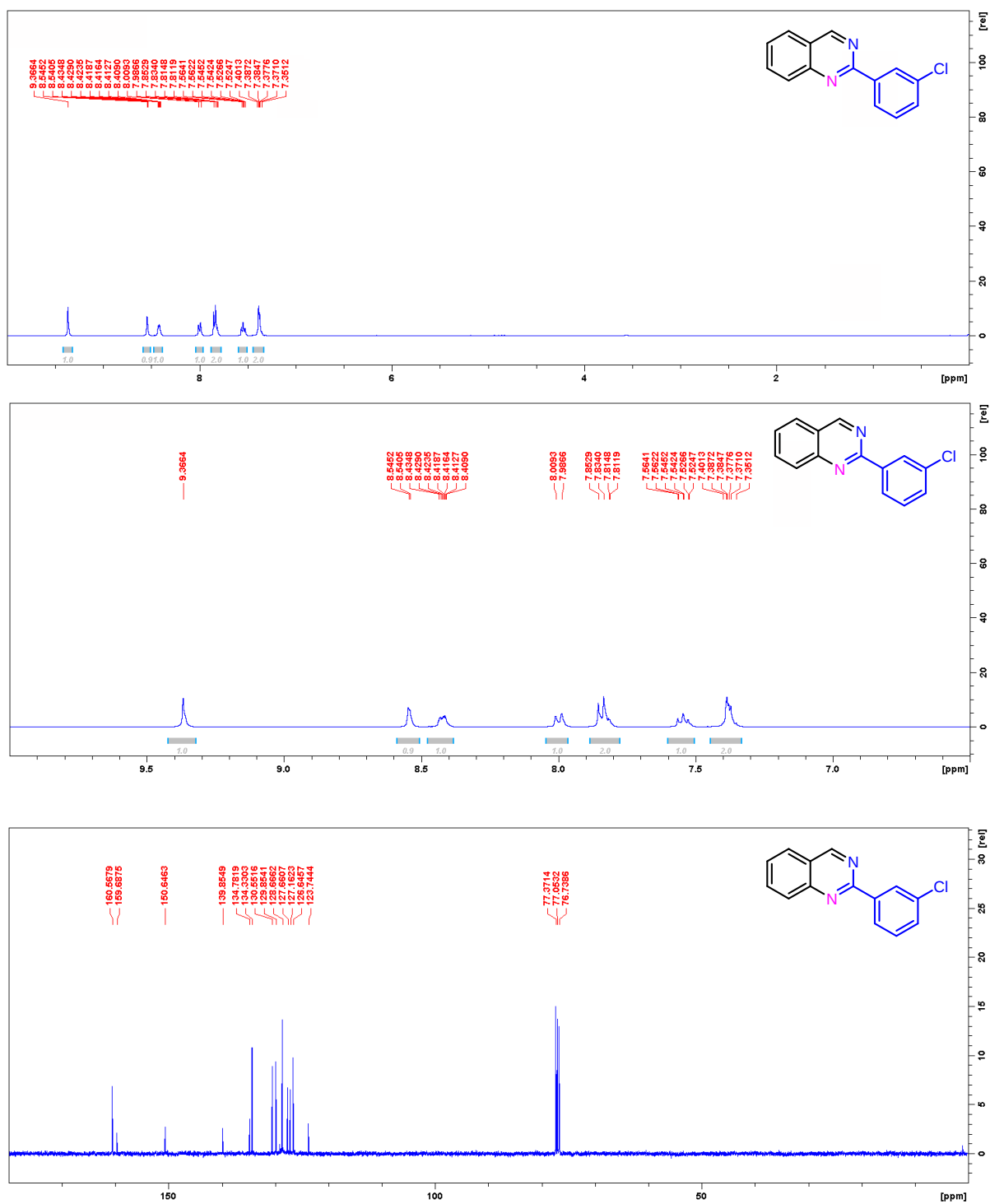


Figure S38. ¹H and ¹³C NMR spectra of compound **10aj** in CDCl₃.

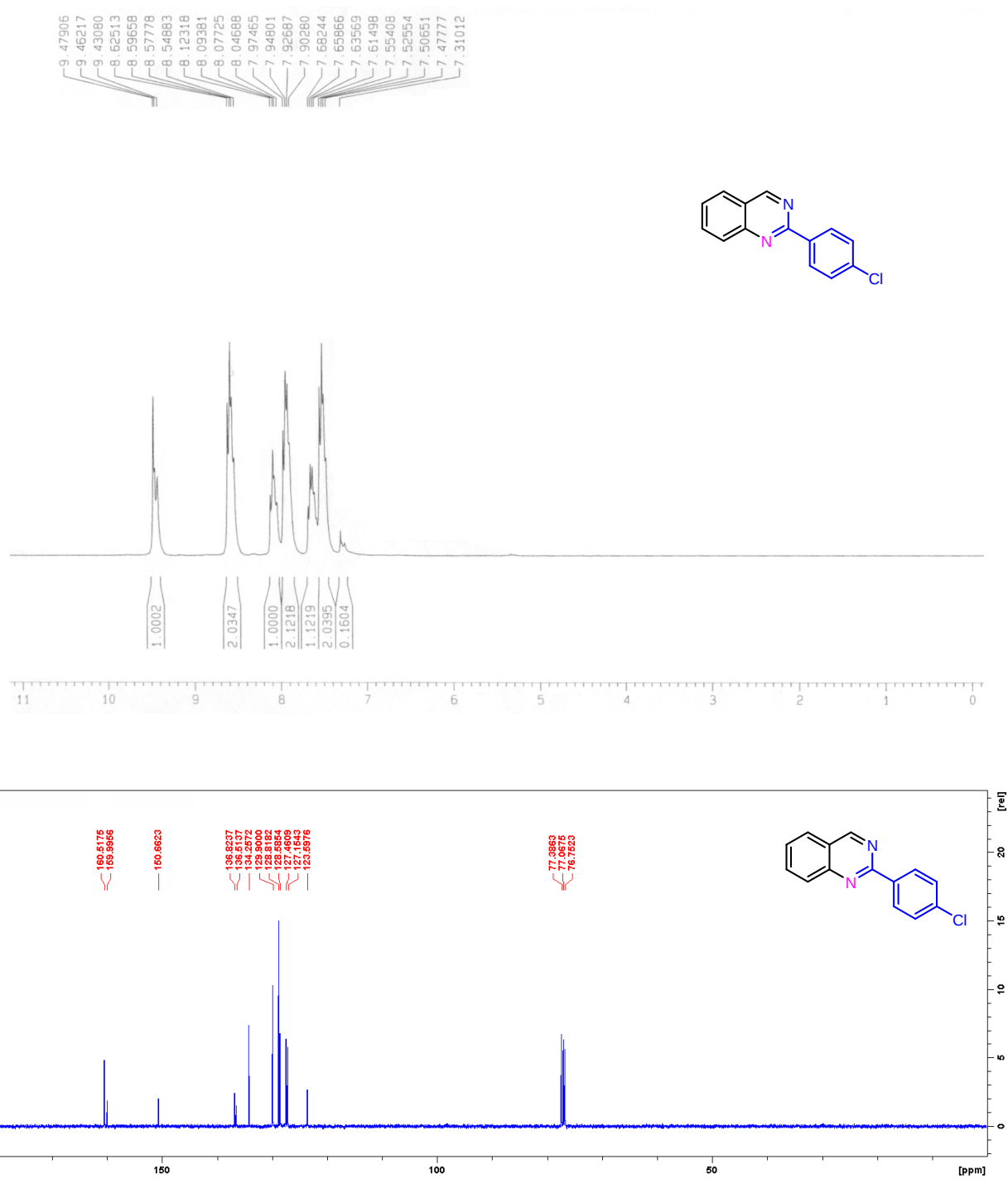


Figure S39. ¹H and ¹³C NMR spectra of compound **10ak** in CDCl₃.

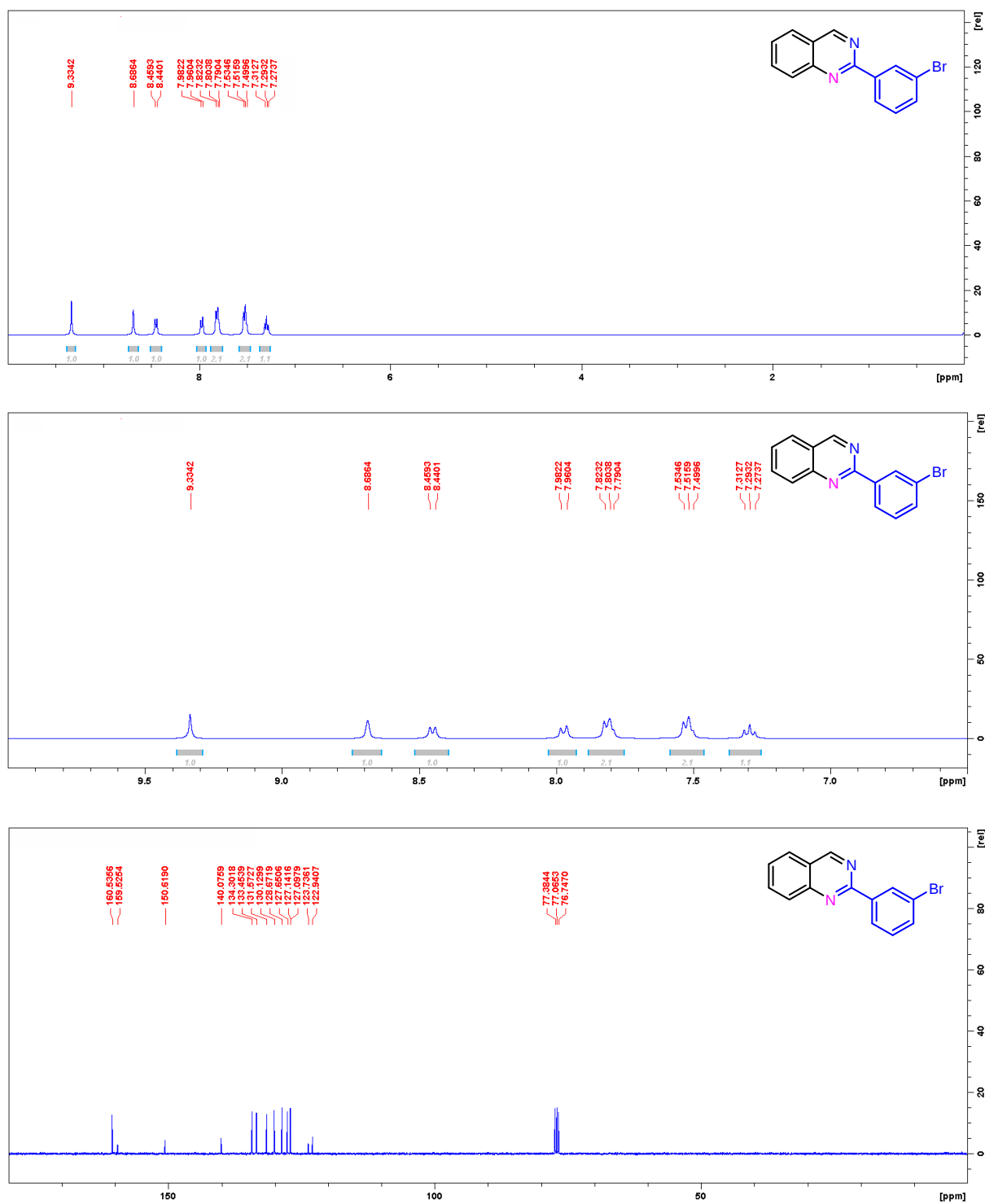


Figure S40. ¹H and ¹³C NMR spectra of compound **10al** in CDCl₃.

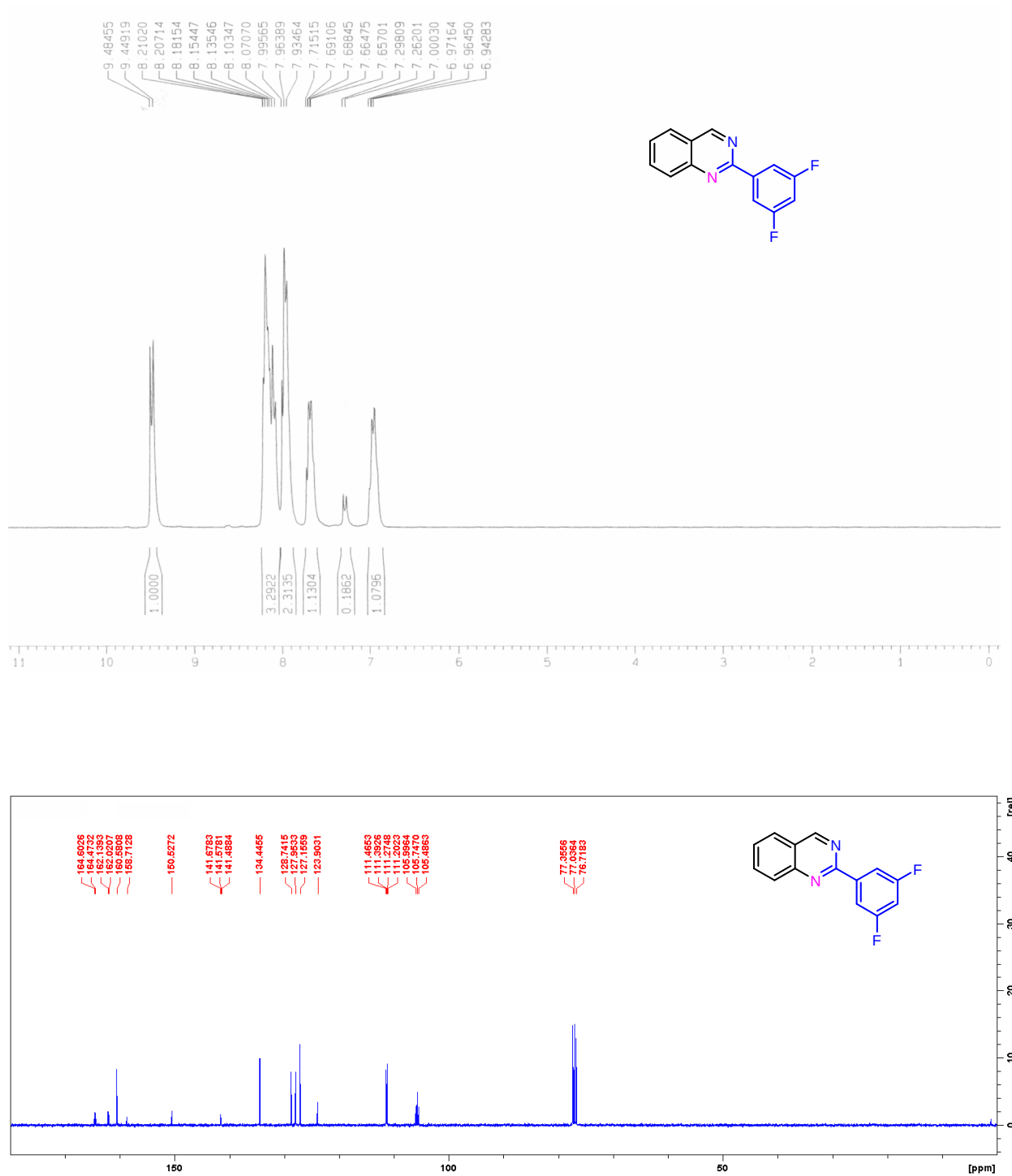


Figure S42. ¹H and ¹³C NMR spectra of compound **10an** in CDCl₃.

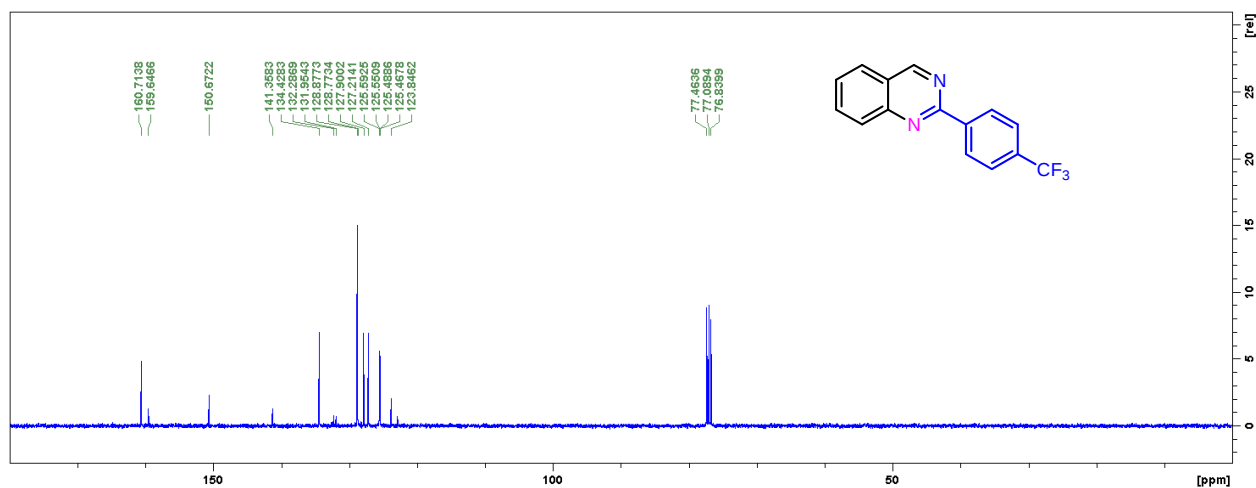
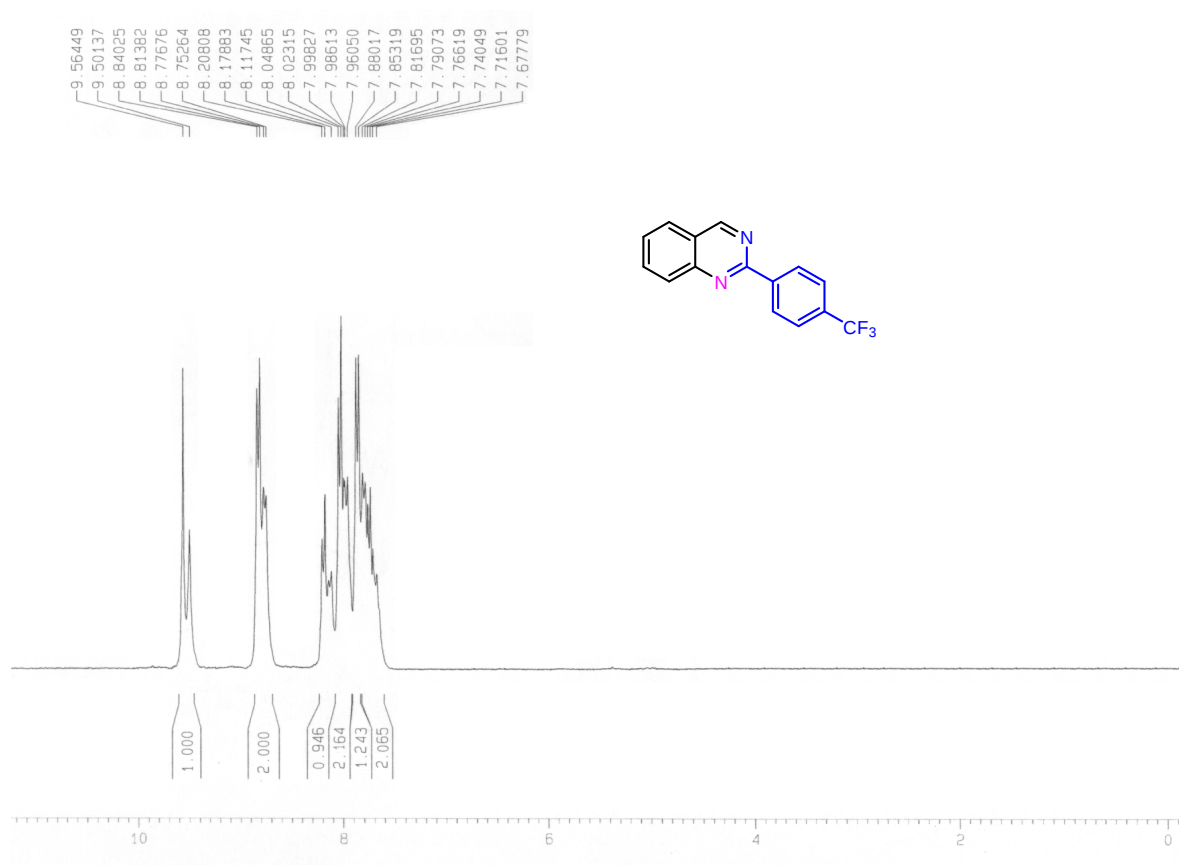


Figure S43. ¹H and ¹³C NMR spectra of compound **10ao** in CDCl₃.

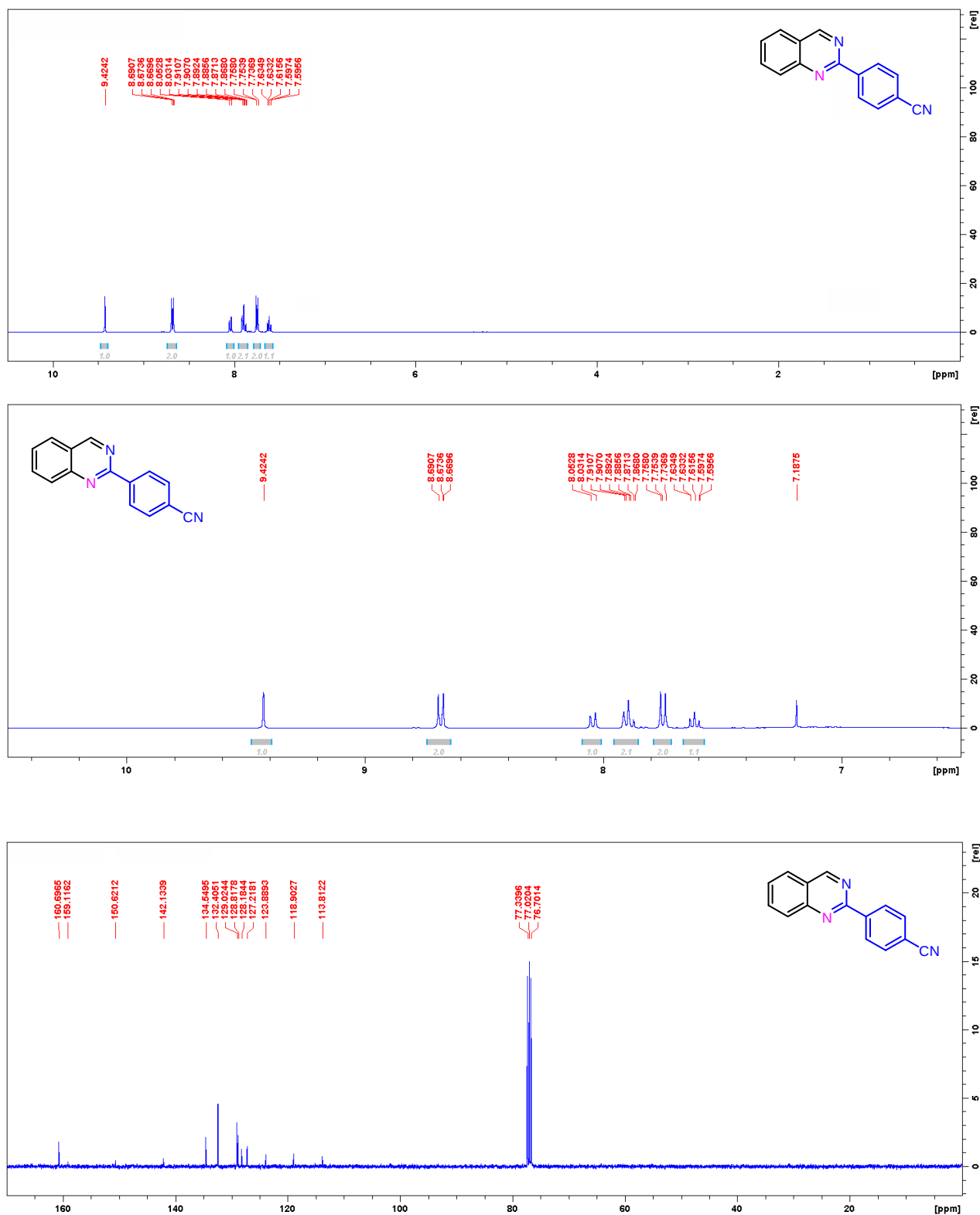
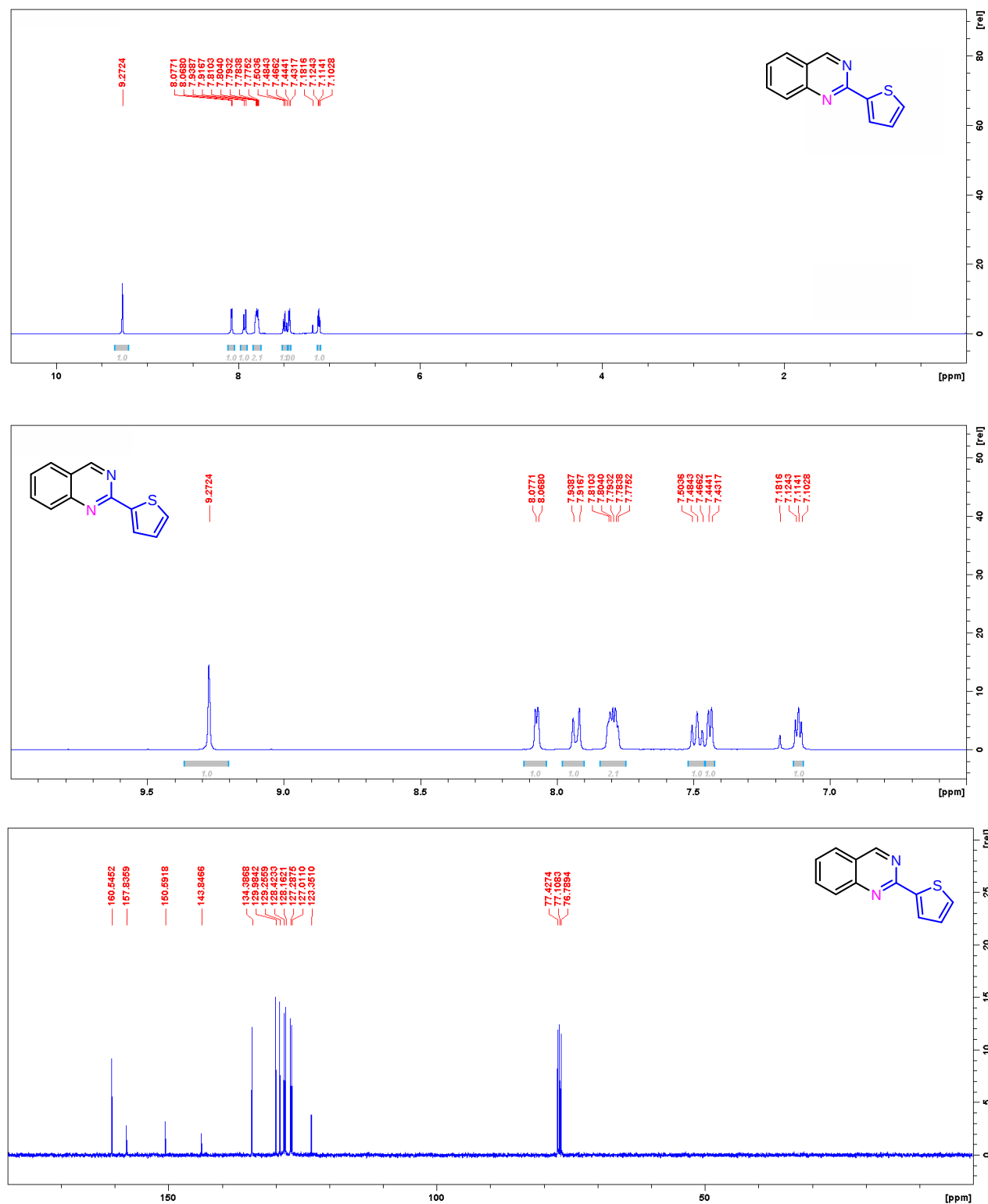


Figure S44. ¹H and ¹³C NMR spectra of compound **10ap** in CDCl₃.



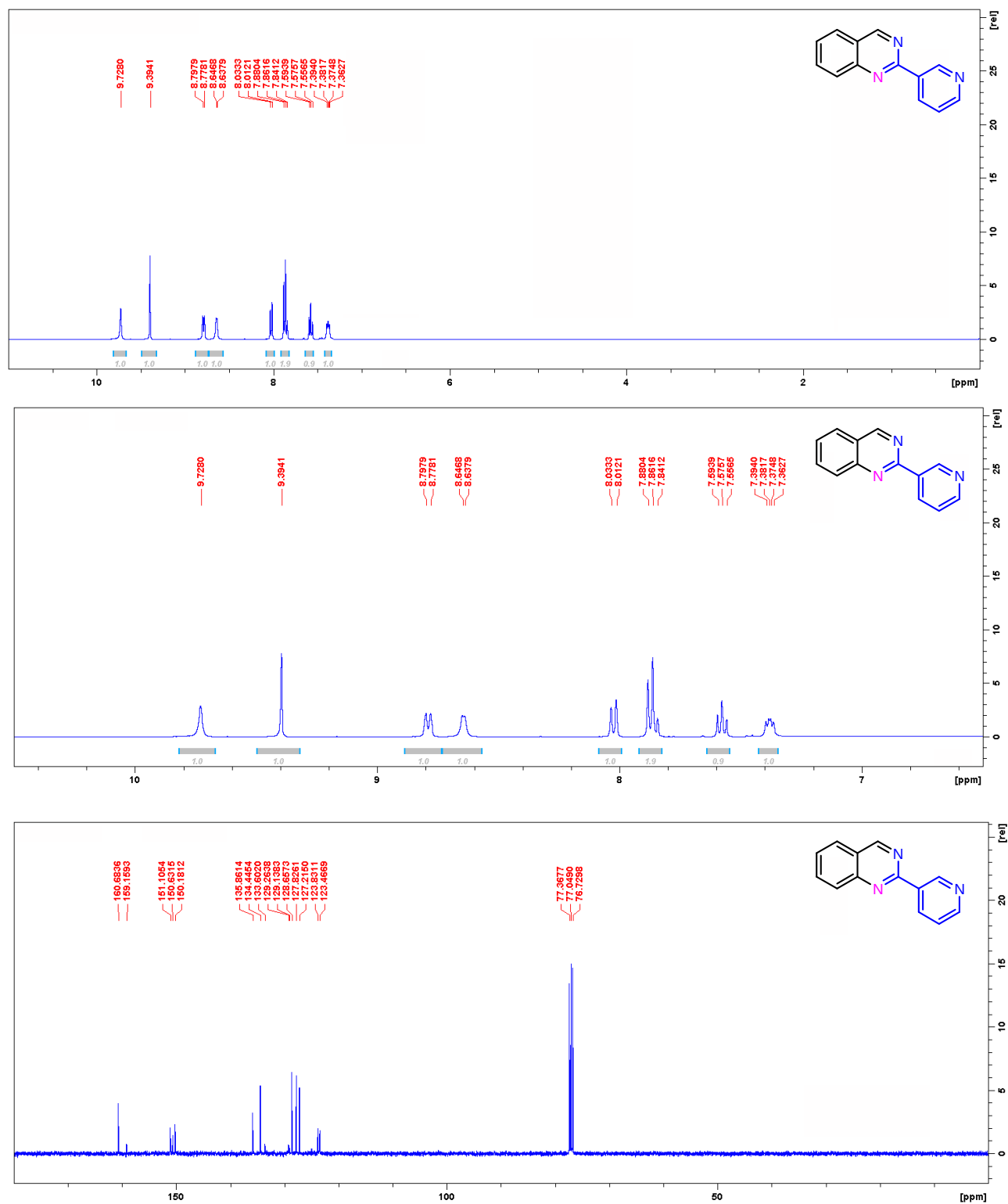


Figure S46. ¹H and ¹³C NMR spectra of compound **10ar** in CDCl₃.

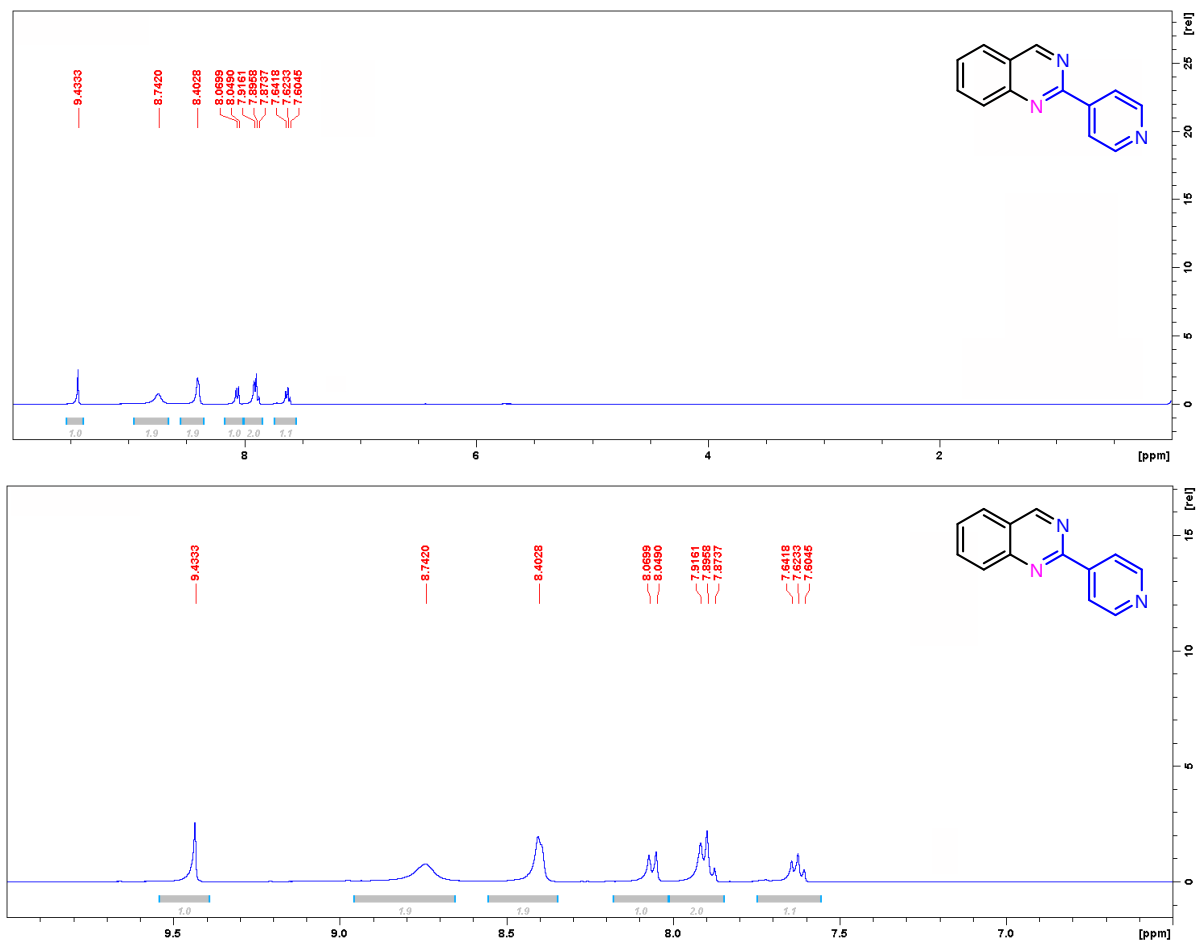


Figure S47. ^1H NMR spectrum of compound **10as** in CDCl_3 .

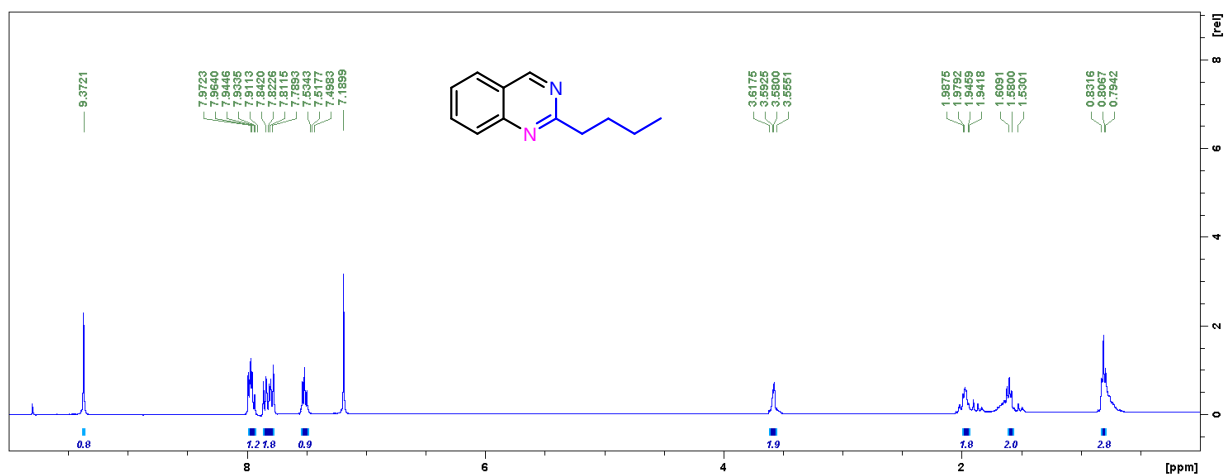


Figure S48. ^1H NMR spectrum of compound **10at** in CDCl_3 .

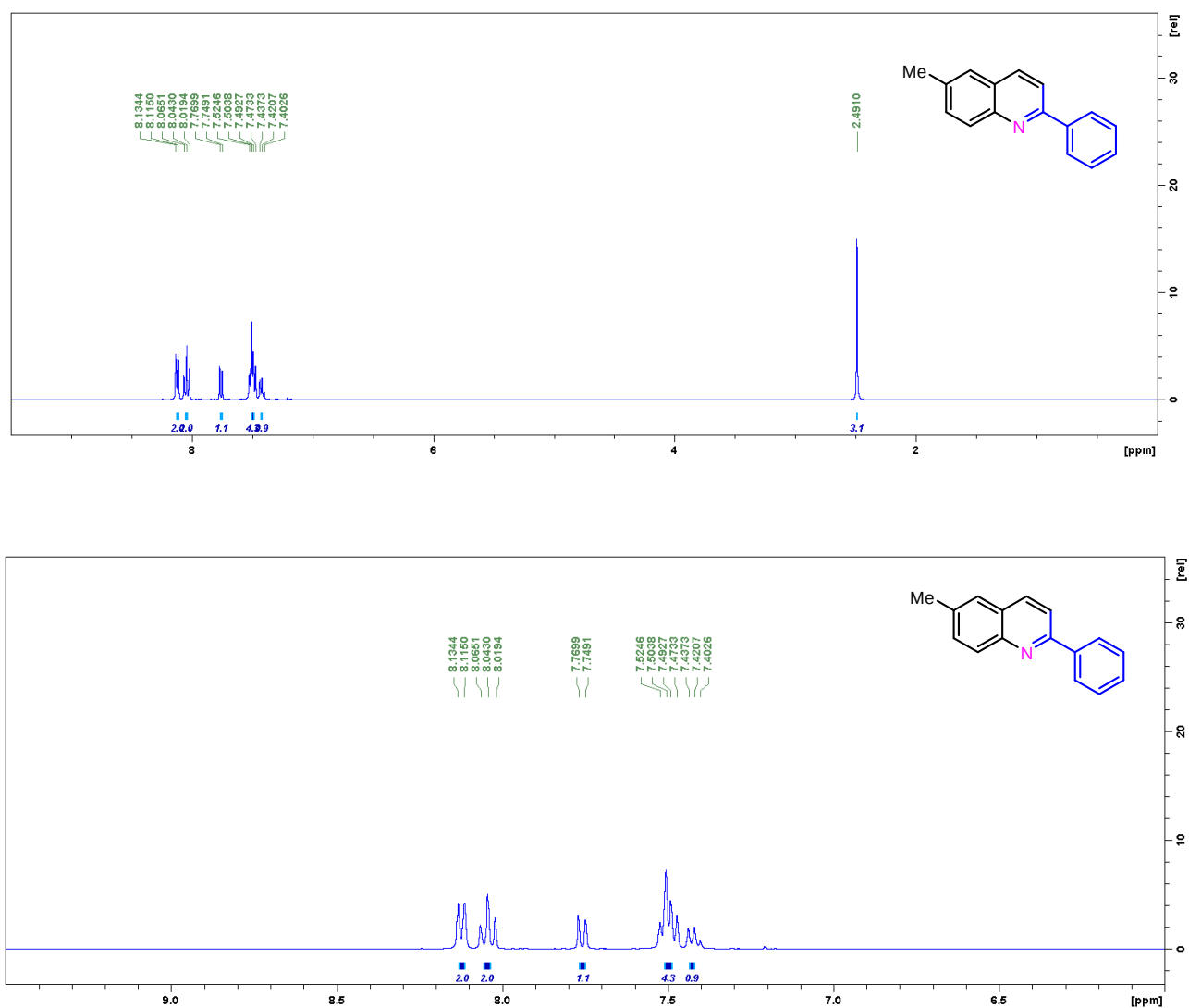


Figure S49. ^1H NMR spectrum of compound **8ba** in CDCl_3 .

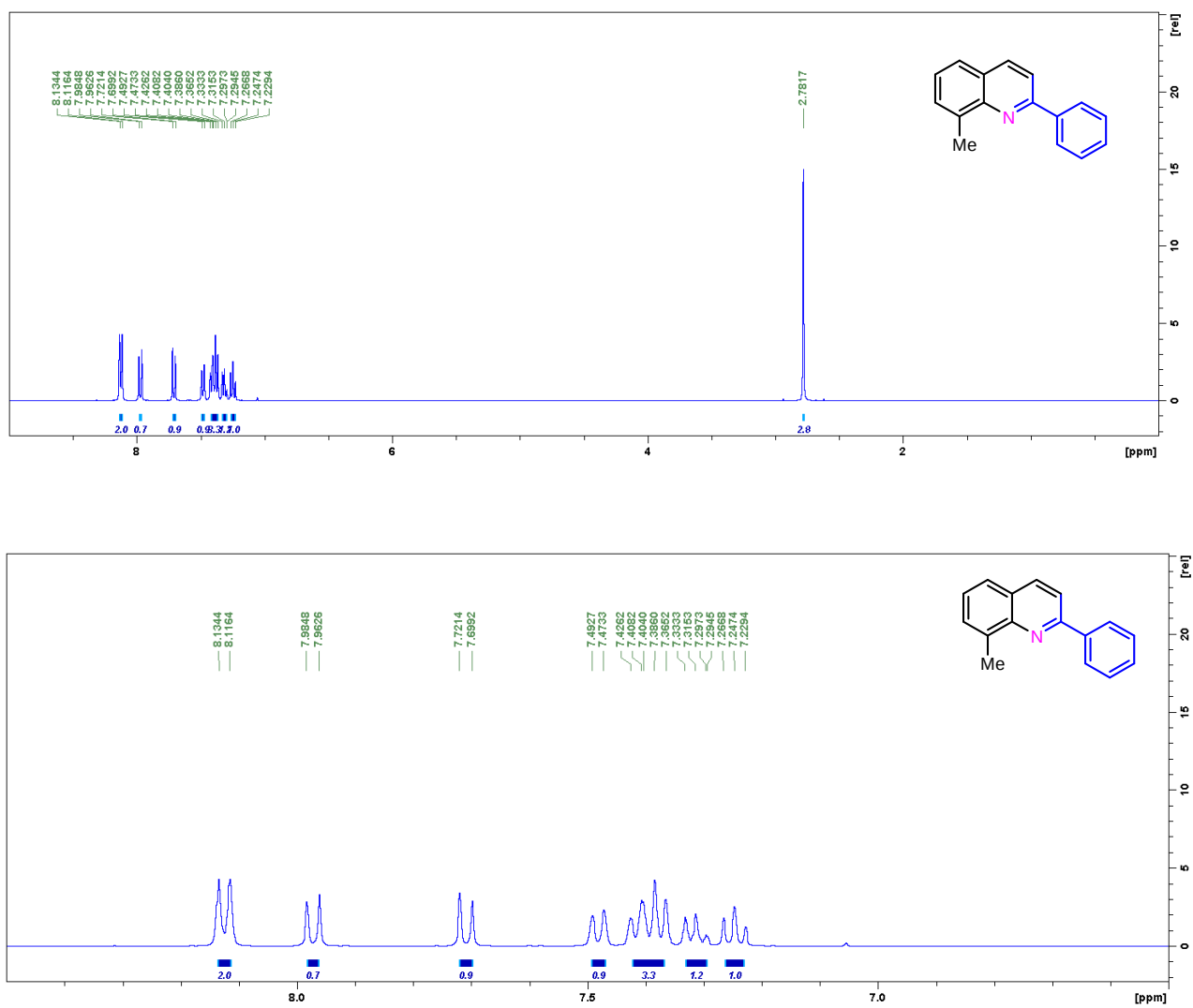


Figure S50. ^1H NMR spectrum of compound **8ca** in CDCl_3 .

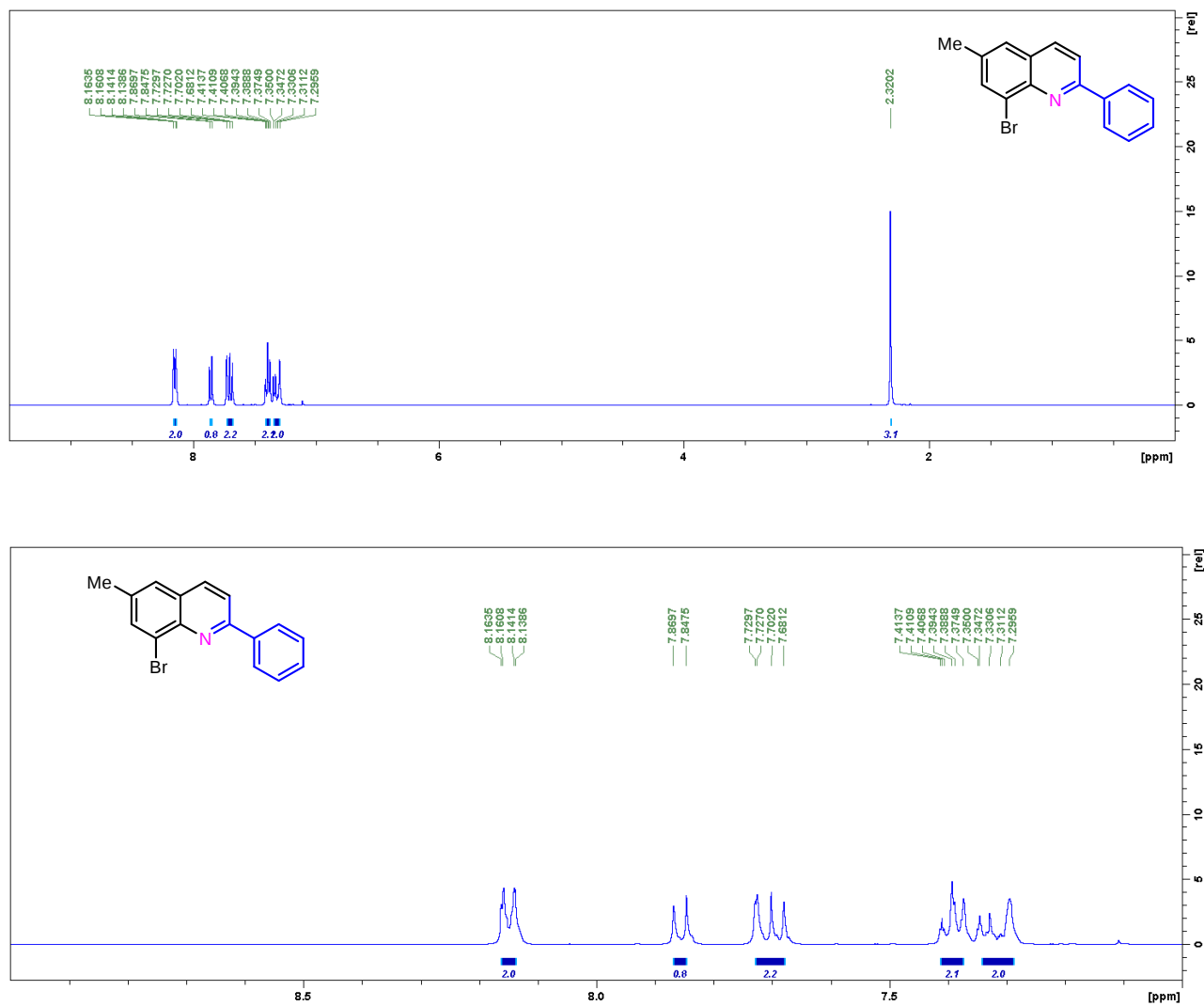


Figure S51. ^1H NMR spectrum of compound **8da** in CDCl_3 .

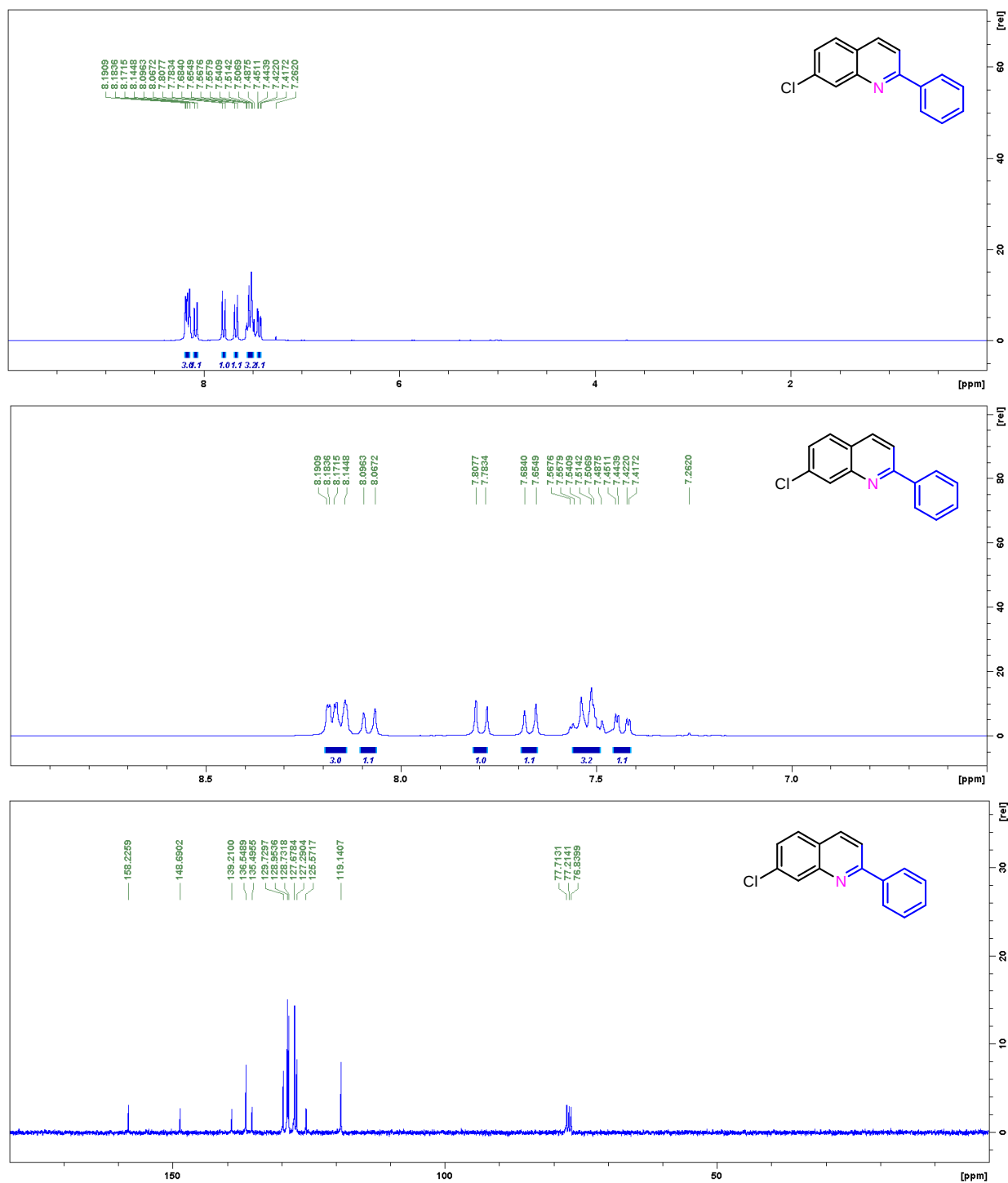


Figure S52. ¹H and ¹³C NMR spectra of compound **8ea** in CDCl₃.

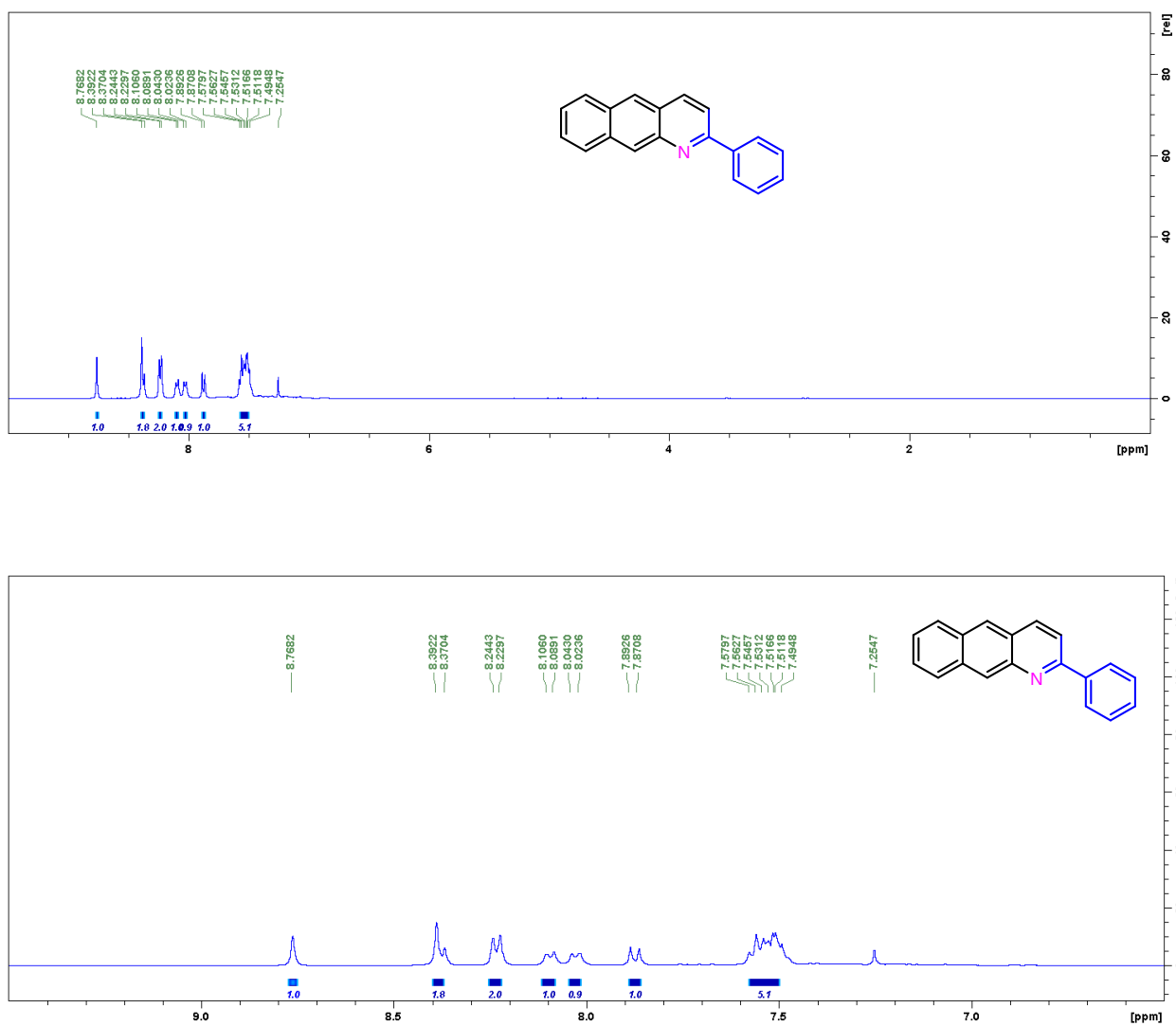


Figure S53. ^1H NMR spectrum of compound **8fa** in CDCl_3 .

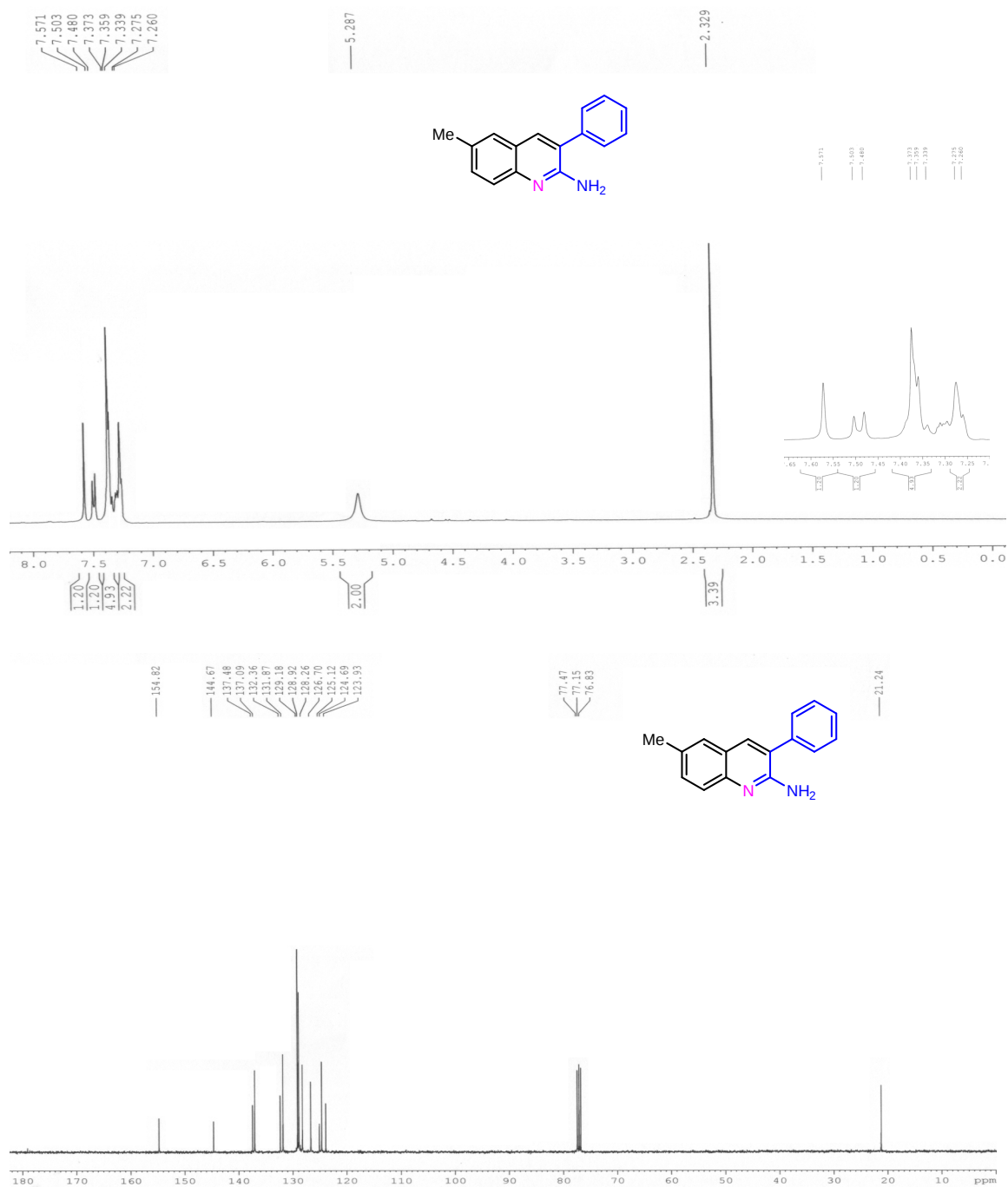


Figure S54. ¹H and ¹³C NMR spectra of compound **9ba** in CDCl₃.

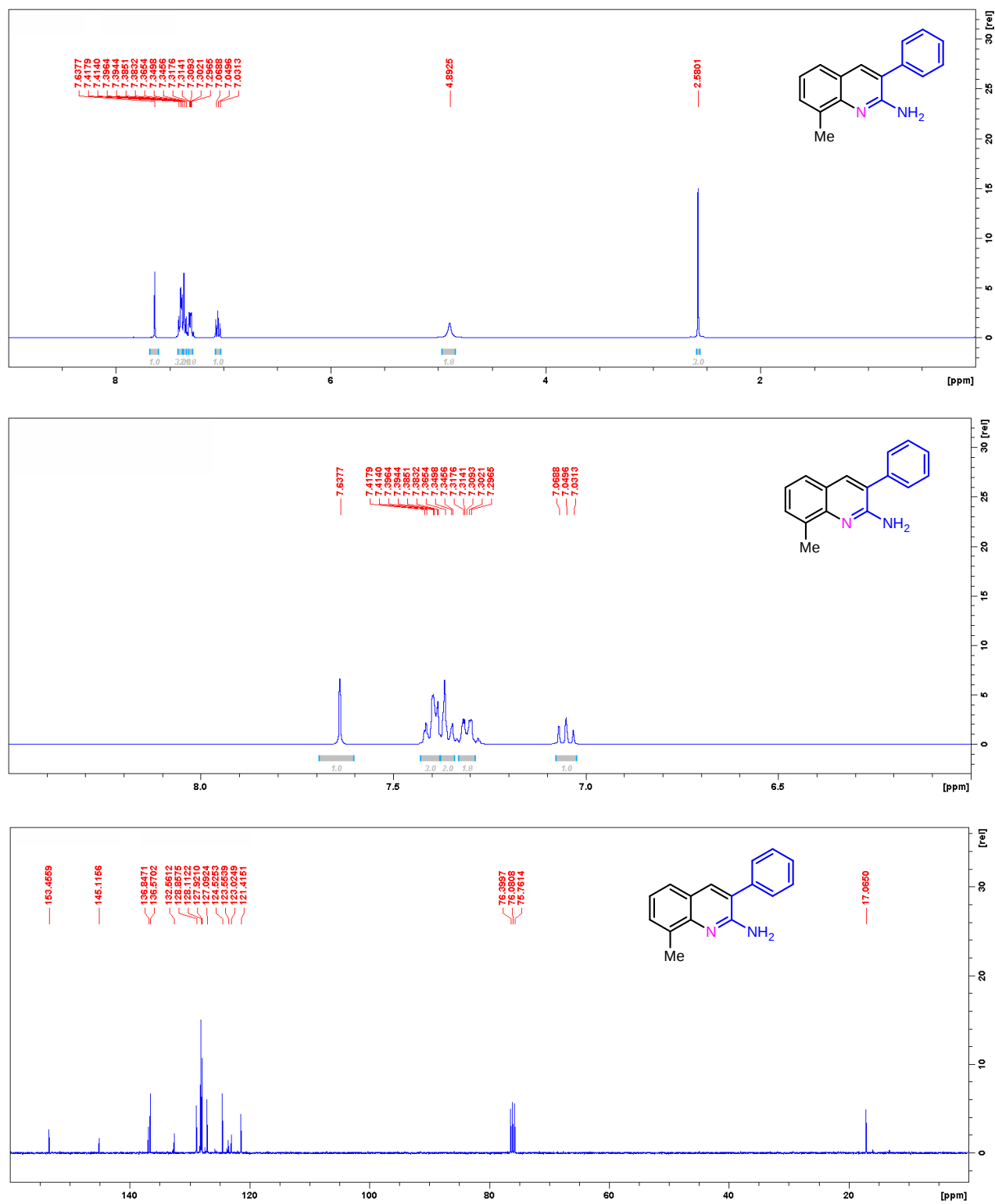


Figure S55. ¹H and ¹³C NMR spectra of compound **9ca** in CDCl₃.

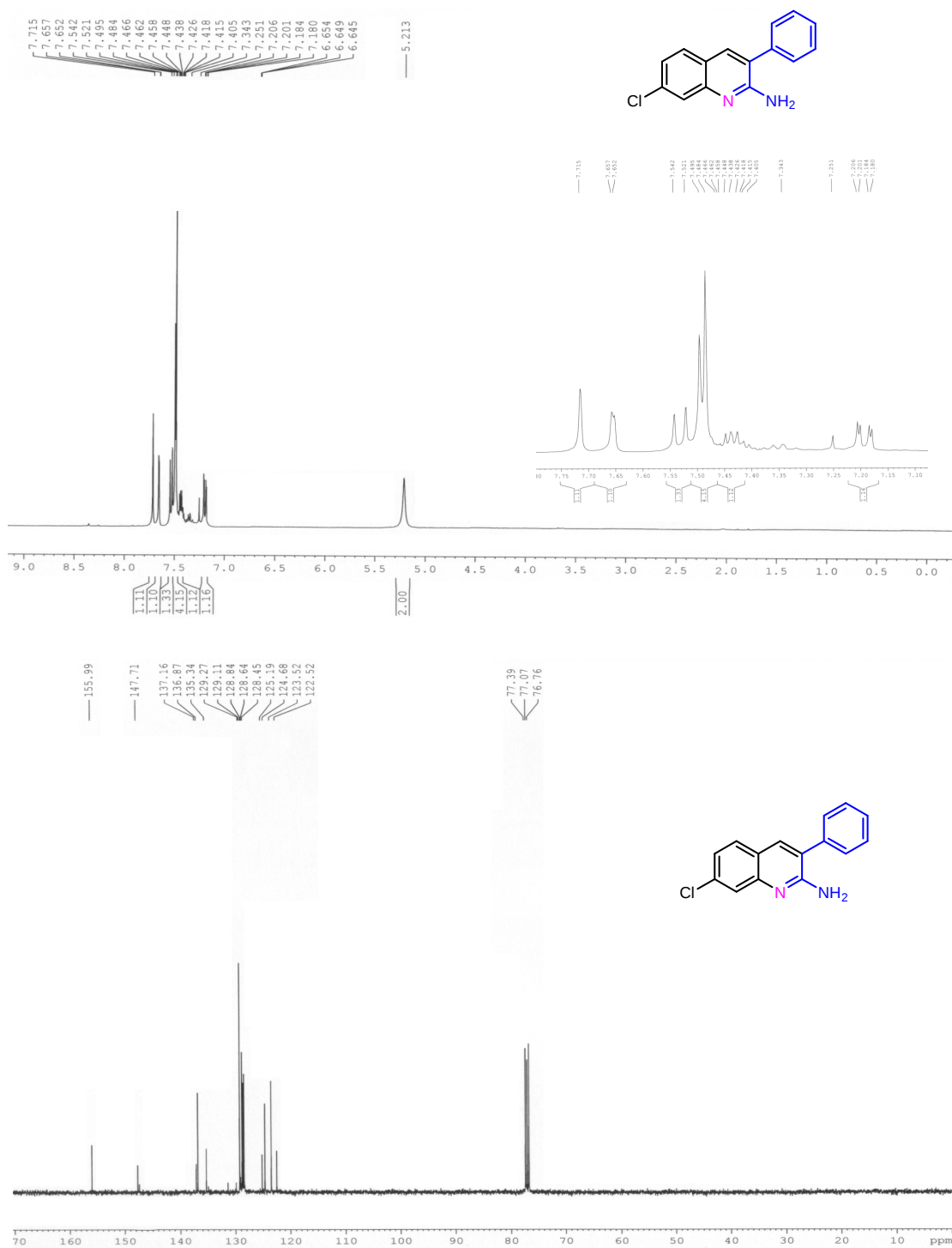


Figure S56. ¹H and ¹³C NMR spectra of compound **9da** in CDCl₃.

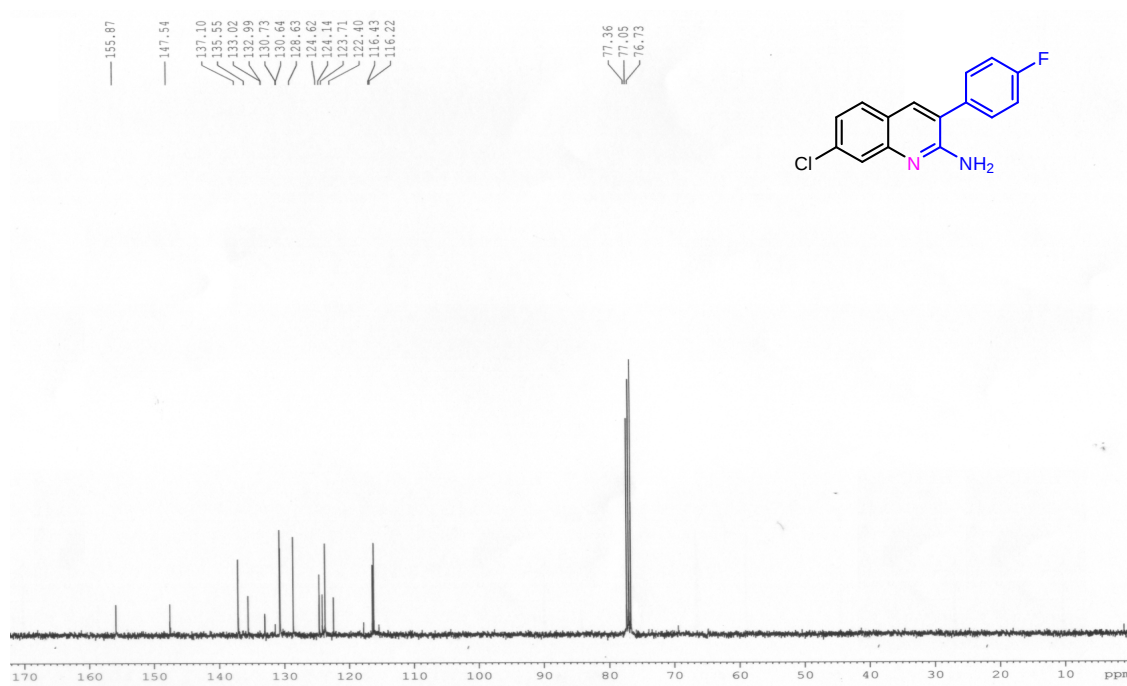
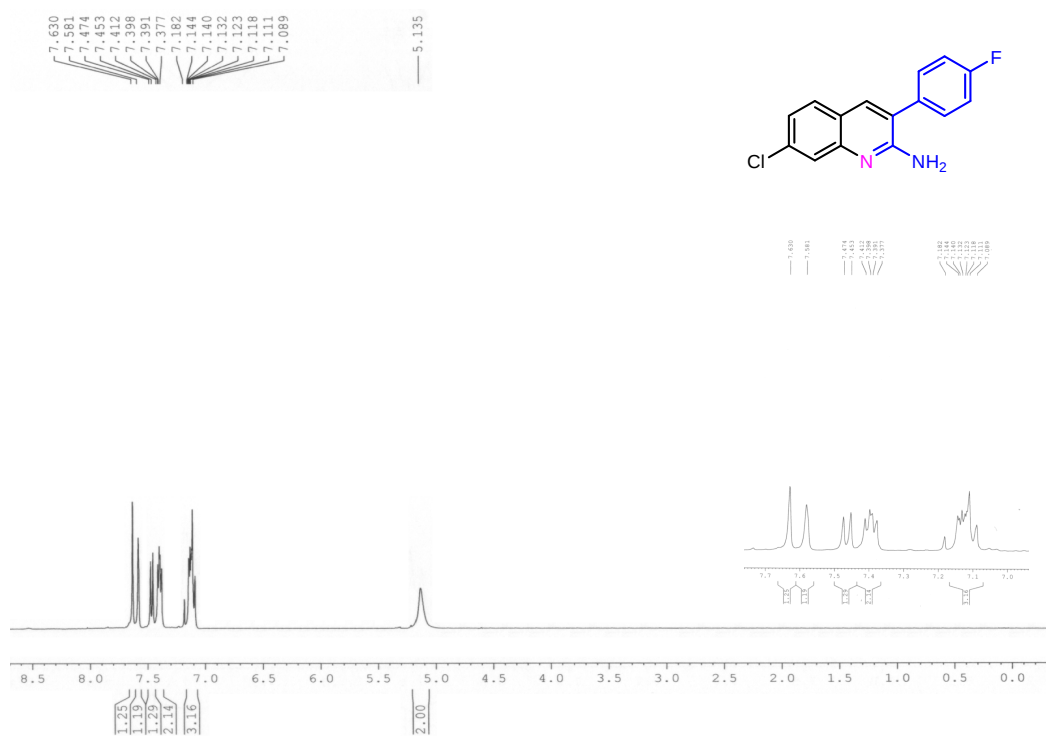


Figure S57. ¹H and ¹³C NMR spectra of compound **9ea** in CDCl₃.

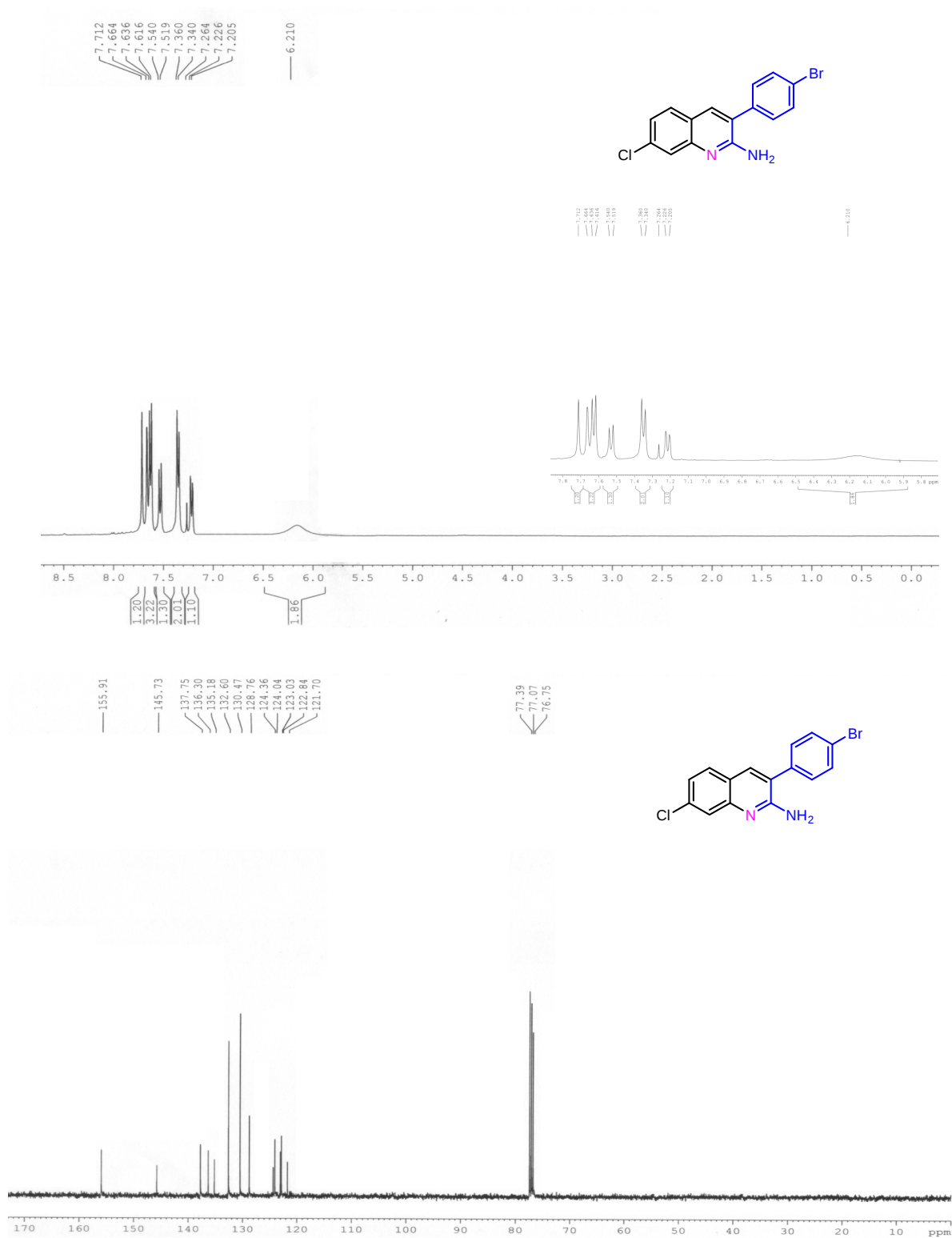


Figure S58. ¹H and ¹³C NMR spectra of compound **9fa** in CDCl₃.

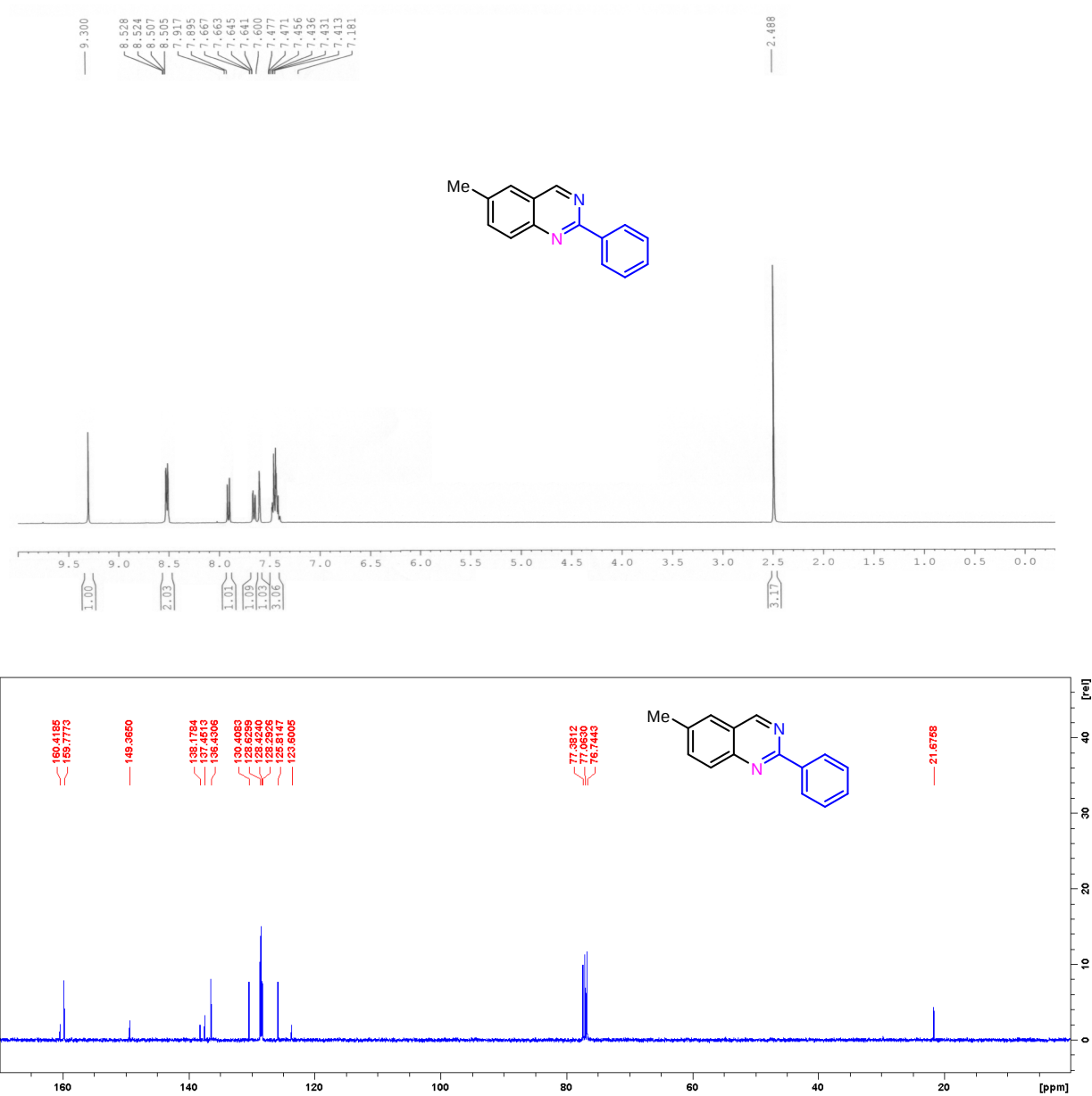


Figure S59. ¹H and ¹³C NMR spectra of compound **10ba** in CDCl₃.

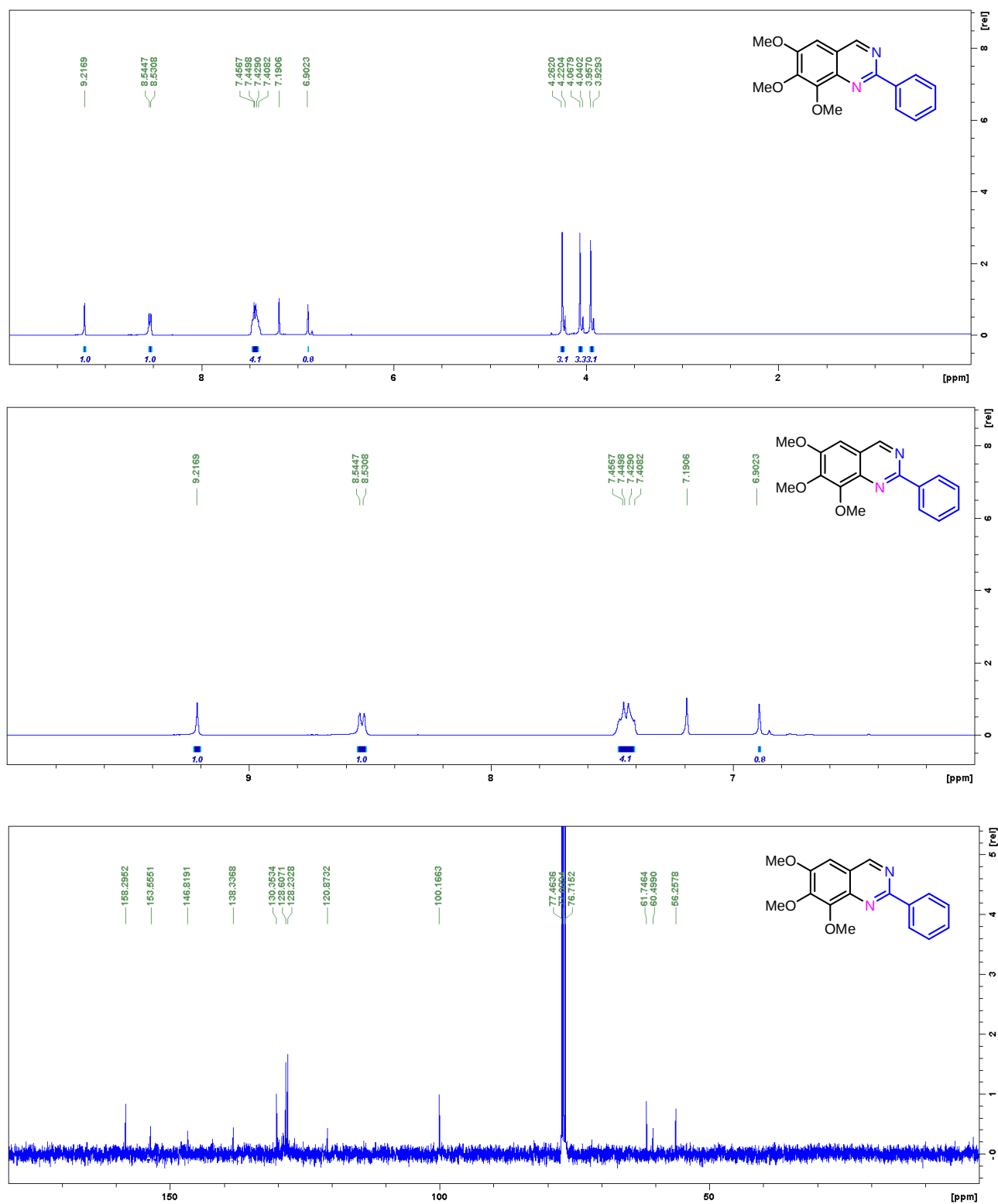


Figure S60. ^1H and ^{13}C NMR spectra of compound **10ca** in CDCl_3 .

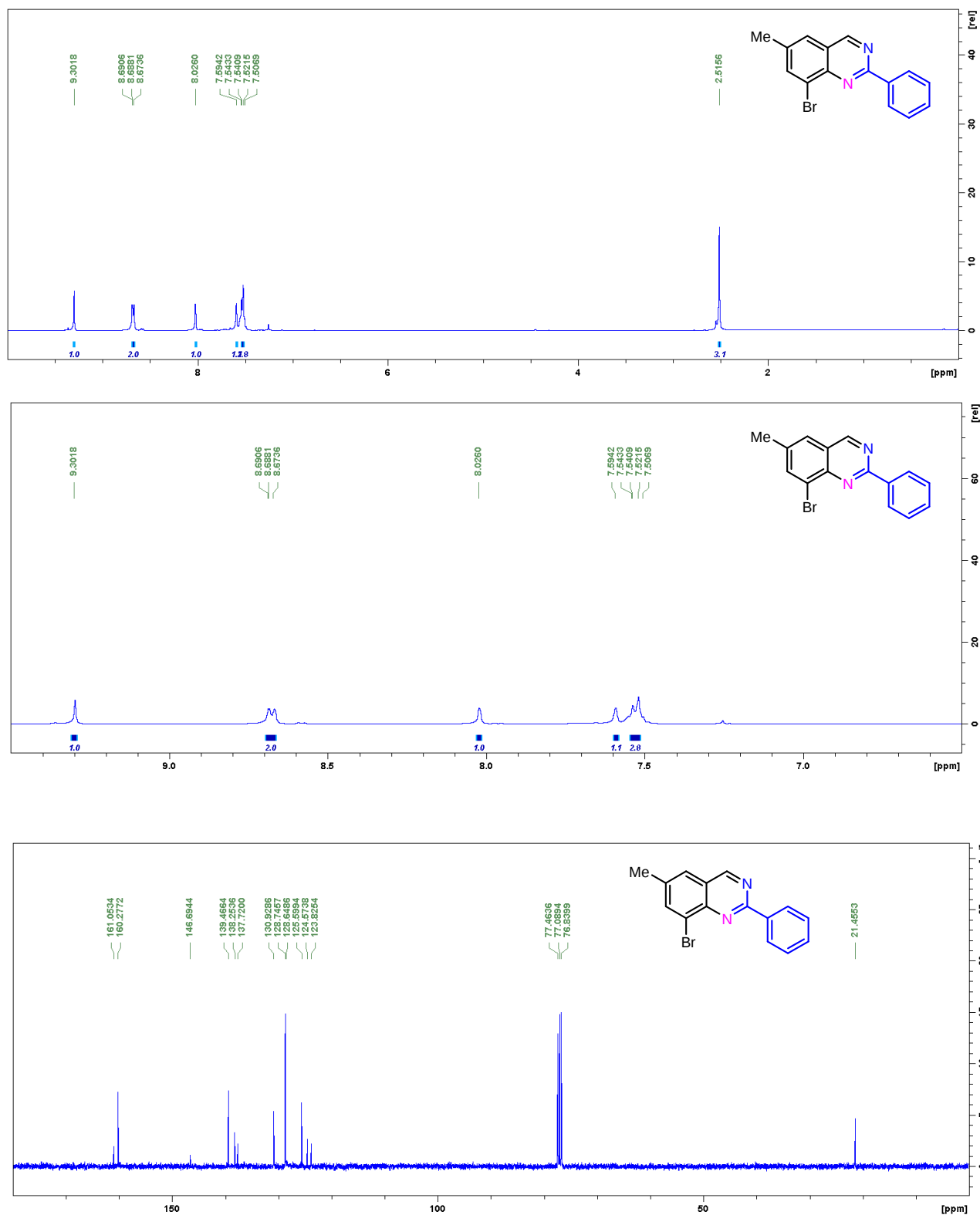


Figure S61. ¹H and ¹³C NMR spectra of compound **10da** in CDCl₃.

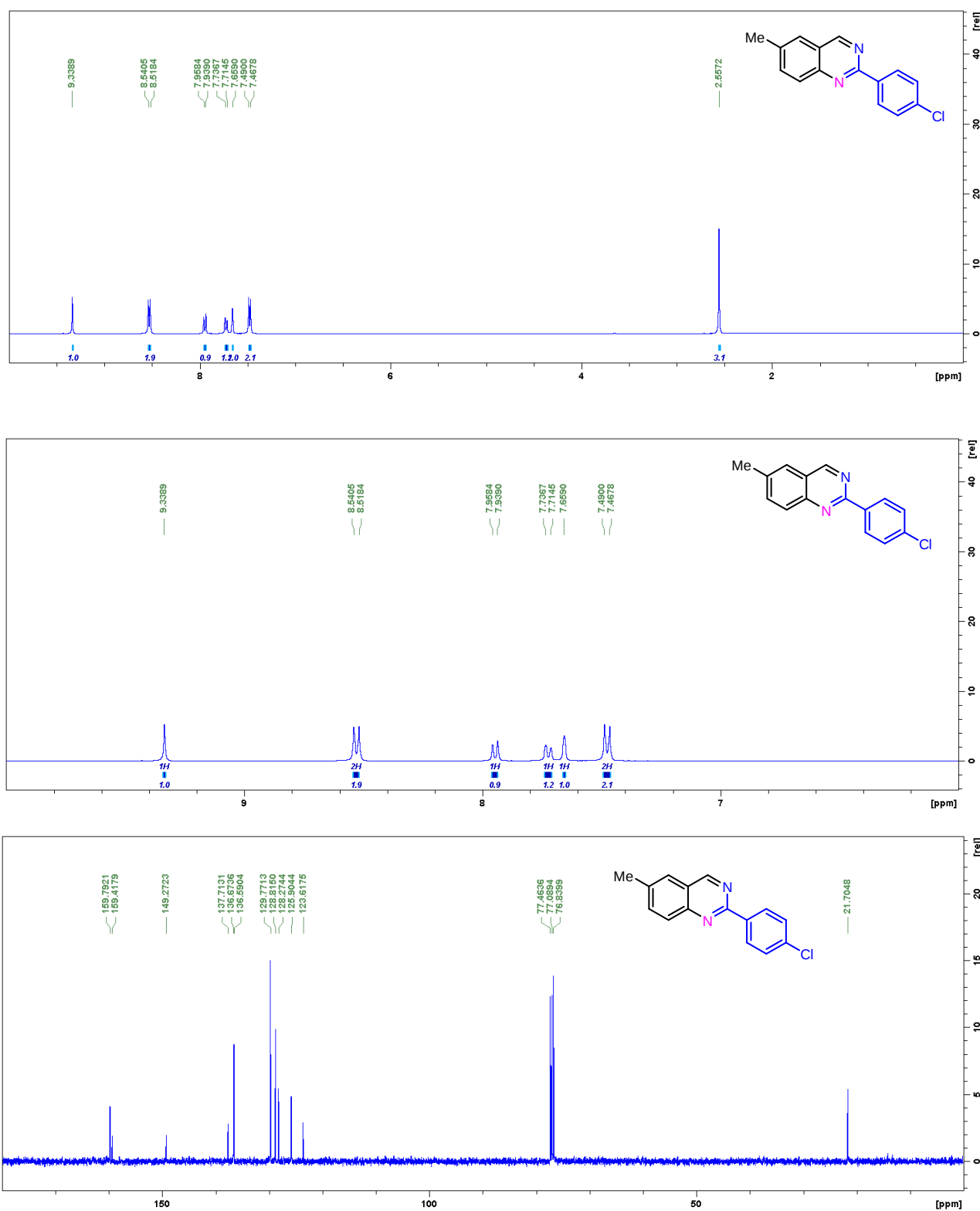


Figure S62. ¹H and ¹³C NMR spectra of compound **10ea** in CDCl₃.

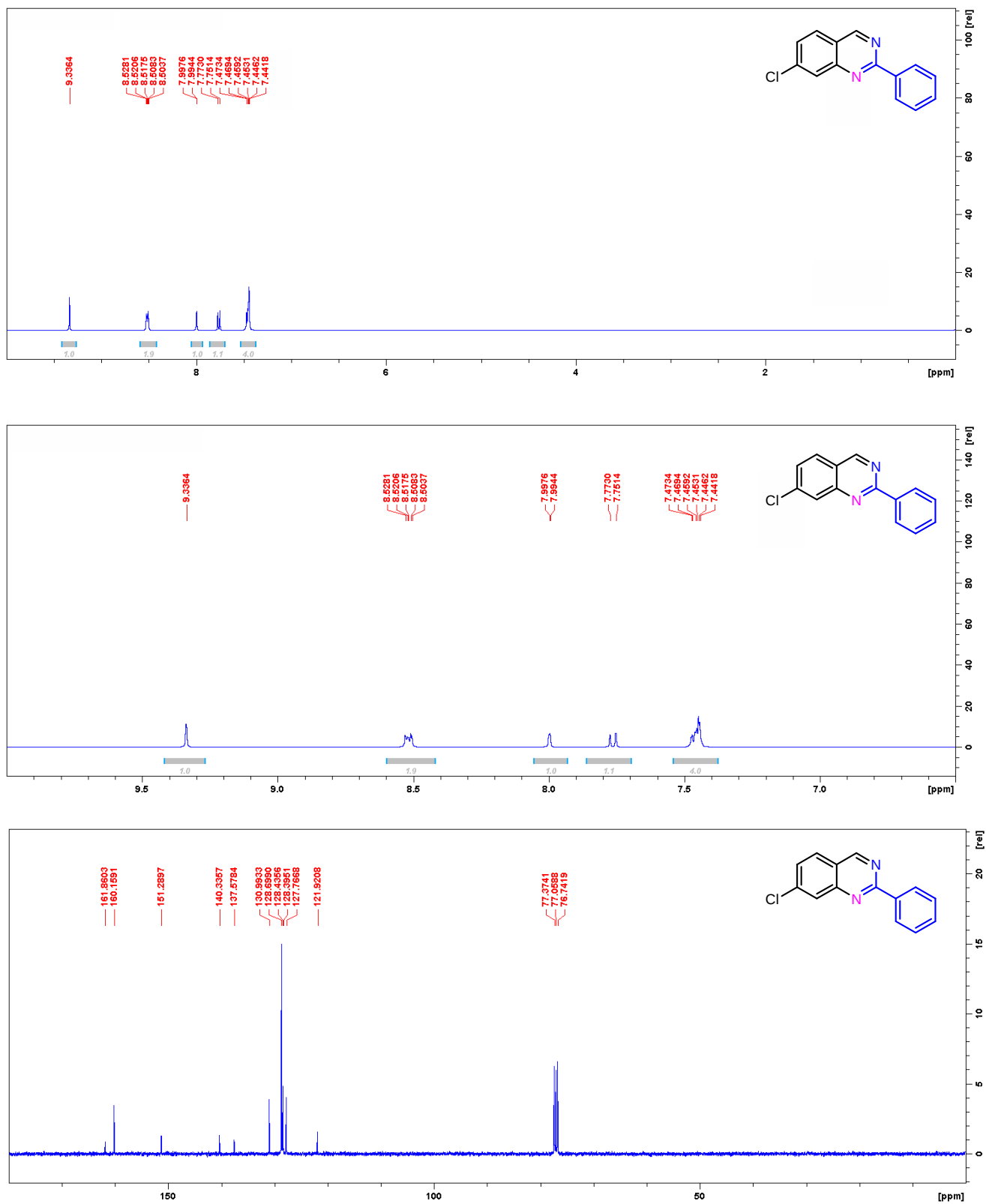


Figure S63. ¹H and ¹³C NMR spectra of compound **10fa** in CDCl₃.

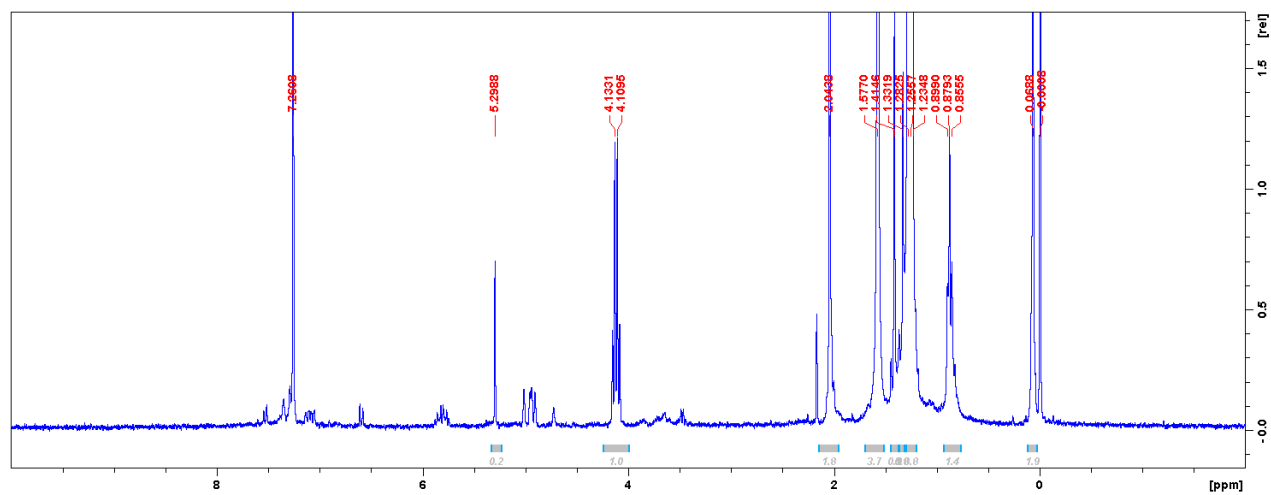


Figure S64. ^1H NMR spectrum of reaction mixture of obtained after the dehydrogenation of cyclobutanol, catalyzed by **3** in CDCl_3 .

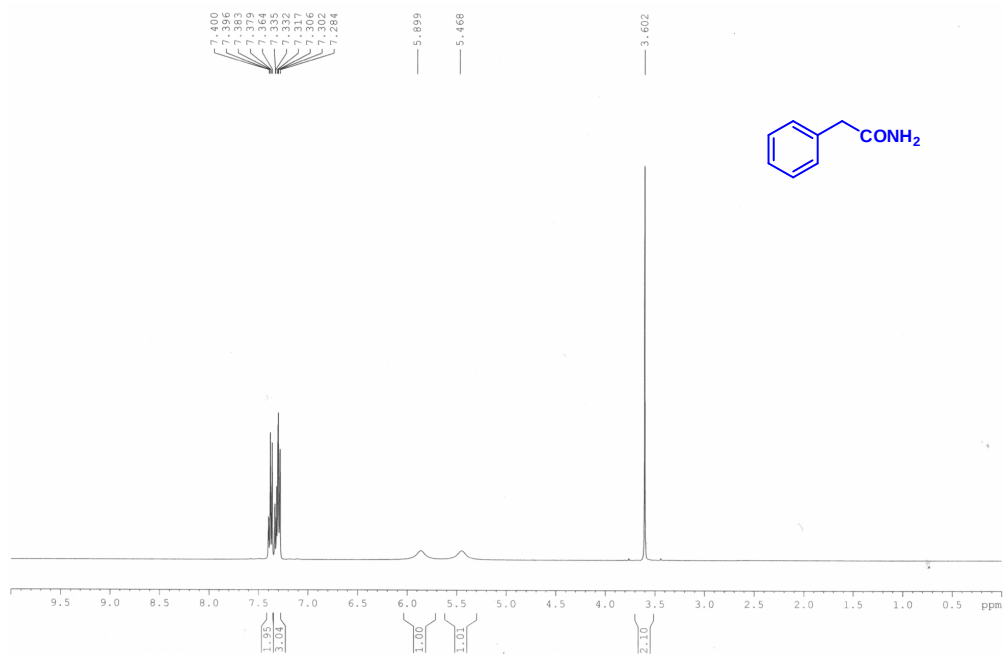


Figure S65. ¹H NMR spectrum of **6a'** in CDCl₃.

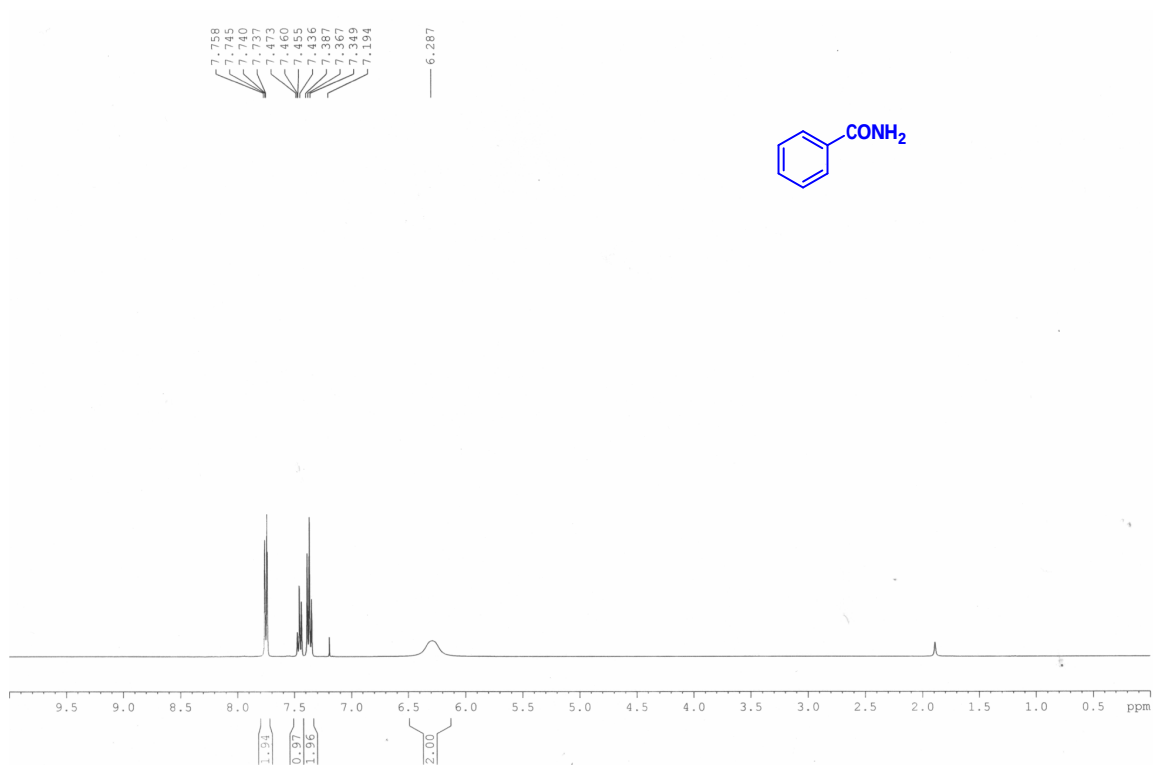


Figure S66. ¹H NMR spectrum of **7a'** in CDCl₃.

References.

- (1) Sinha, S.; Das, S.; Sikari, R.; Parua, S.; Brandaõ, P.; Demeshko, S.; Meyer, F.; Paul, N. D. Redox Noninnocent Azo-Aromatic Pincers and Their Iron Complexes. Isolation, Characterization, and Catalytic Alcohol Oxidation. *Inorg. Chem.* **2017**, *56*, 14084–14100.