Electronic Supplementary Information (ESI)

Improved Transport Properties and Novel Li Diffusion Dynamics in van der Waals C₂N/Graphene Heterostructure as Anode Materials for Lithium Ion Battery: A First Principles Investigation[†]

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BoltzTraP method

The electrical transport properties were calculated using BoltzTrap code based on a semi-classic Boltzmann transport theory [1].

Under the approximation, the electrical conductivity and electric thermal conductivity as function of temperature T and chemical potential μ can be calculated from the following equations:

$$\sigma_{ij}(T,\mu) = \frac{1}{\Omega} \int g_{ij}(\xi) \left[-\frac{\partial f_{\mu}(T,\xi)}{\partial \xi} \right] d\xi$$
(1)

Here e, Ω and ξ ε represent the electron charge, reciprocal space volume, and band energy, respectively. $\frac{\partial f_{\mu}(T,\xi)}{\partial \xi}$ is the differential of Fermi function. $g_{ij}(\xi)$

denotes transport distribution function, shown as

$$g_{ij}(\xi) = \frac{e^2}{N} \sum_{n,\vec{k}} \tau_{n,\vec{k}} v_i\left(n,\vec{k}\right) v_j\left(n,\vec{k}\right) \frac{\delta\left(\xi - \xi_{n,\vec{k}}\right)}{d\xi}$$
(2)

where N is the k-points number in the entire BZ. $\xi_{n,\vec{k}}$ are the energy for n-th

band at wave vector \vec{k} , $v_i(n, \vec{k})$ is the corresponding velocity as defined as

 $v_i\left(n,\vec{k}\right) = \left(\frac{1}{h}\right)^{\partial \xi_{n,\vec{k}}} / \partial \vec{k}$ The relaxation time is treated as independence with n and

 \vec{k} namely as a constant $(\tau_{n,\vec{k}} = \tau)$. The electrical transport parameters were calculated based on the eigenvalues from PBE functional. To accurately predict the transport parameters, much denser k-mesh (20×20×1) was used for the integrations in 1st BZ. [56] Madsen, G. K. H.; Singh, D. J. BoltzTraP. a code for calculating band-structure dependent quantities. Comp. Phys. Comm. 2006, 175, 67-71.

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Fig. S1. The energy barrier of lithium diffusion pathways along the difference sites (top and side views) of 2D-C₂N.



Fig. S2. The energy barrier of lithium diffusion pathways along the difference sites (top and side views) of Graphene.



Fig. S3 The optimized structures and chosen adsorption sites (top of C_2N) of AA stacking for C_2N/G heterostructures. The charge density difference, $\Delta\rho$ (the top and side views) of Li-middle adsorption of C_2N/G heterostructure of AA stacking. The adsorbed in CN1(a, d),CN2(b, e) and CN3(c, f).



Fig. S4 The charge density difference, $\Delta \rho$ (the view) of Li-middle adsorption of 2D-C₂N/graphene heterostructure of AA stacking. The adsorbed GCN1 (a, e), GCN2 (b, f) and GCN3 3(c, g).



Fig. S5 The optimized structures and chosen adsorption sites (bottom of graphene) of AA stacking for monolayer C_2N/G heterostructures. The charge density difference, (bottom and side views) of Li-middle adsorption of C_2N/G heterostructure of AA stacking. The adsorbed in G1 (a, e),G2 (b, f), G3 (c, g) and G4 (d, h).

		spin non-polarized	spin polarized				
AA	site	Adsorption energy	Adsorption energy				
	CN1	0.1551	0.1404				
Un	CN2	-0.0563	-0.076				
Op	CN3	-2.9637	-2.9803				
	CN4	-2.9805	-3.0003				
	GCN1	-0.3078	-0.3277				
Middle	GCN2	-0.4940	-0.5057				
	GCN3	-2.9640	-2.9807				
	G1	0.3111	0.2901				
Down	G2	0.3041	0.2834				
	G3	0.3025	0.2810				

Table S1 The calculated adsorption energy (E_{ad} , eV) C₂N/G heterostructure using the spin non-polarized methods and spin polarized.

		C ₂ N/G hete	(Fraphene	C ₂ N				
AA	site	Adsorption	۸D	site	Adsorption	sita	Adsorption	gita	Adsorption
		energy	AD		energy	Site	energy	site	energy
Up	CN1	0.1551		CN1	-3.0713	CN1	0.1475	G1	0.5301
	CN2	-0.0563	I.I.a	CN2	-0.1667	CN2	-0.0858	G2	0.5301
	CN3	-2.9637	Ор	CN3	0.0639	CN3	-2.9027	G3	0.5300
	CN4	-2.9805							
	GCN1	-0.3078		GCN1	-3.0889				
Middle	GCN2	-0.4940	Middle	GCN2	-0.4682				
	GCN3	-2.9640		GCN3	-0.2459				
Down	G1	0.3111		D ₁	0.2138	CN1	0.1475	G1	0.5301
	G2	0.3041	Down	D ₂	0.1913	CN2	-0.0857	G2	0.5301
	G3	0.3025		D ₃	0.2236	CN3	-2.9188	G3	0.5301

Table S2The calculated adsorption energy (E_{ad} , eV) value at difference sites

		Up ₁ -U	Up ₁ -U	Up ₂ -U	Mid ₁ -M	Mid ₁ -M	Mid ₂ -M	Down ₁ -Do	Down ₁ -Do	Down ₃ -Do	Up ₁₃ -U	Mid ₁₃ -M
		p ₂	p ₃	p ₃	id ₂	id ₃	id ₃	wn ₂	wn ₃	wn ₄	p ₂₃	id ₂₃
C ₂ N/G	AA	0.14	0.13	0.06	0.17	0.02	0.06	0.30	0.30	0.30	0.14	0.22
e	AB	2.96	3.23	0.36	2.61	2.82	0.03	0.30	0.30	0.30	2.28	1.53
C ₂ N		0.14	0.10	0.06								
Graphene		0.31	0.31	0.31								
VS ₂ /G heterostructur e	[28]					0.2	0.2					
MoS ₂ /G heterostructur e	[44]				0.17	0.23	0.10					
Si/G heterostructur e	[45]	0.36			0.37			0.40				
G/blue-phosp horus heterostructur e	[46]				0.23	0.22						
P/G heterostructur e	[35]	0.09			0.12	0.79		0.25				
G/metal sulfide(MoS2/ G)	[49]				0.18							
G/metal sulfide(SnS ₂ / G)	[49]				0.21							

Table S3 Calculated diffusion barriers (E_a , eV) along different paths.

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	atom	GCN1	GCN2	GCN3		GCN1	GCN2	GCN3		atom	GCN1	GCN2	GCN3		GCN1	GCN2	GCN3
						ΔQ	ΔQ	ΔQ							ΔQ	ΔQ	ΔQ
	C1	3.9481	3.944	3.961	4	-0.0519	-0.056	-0.039		C25	3.5093	3.4848	3.471	4	-0.4907	-0.5152	-0.529
	C2	3.9356	3.9459	3.9505	4	-0.0644	-0.0541	-0.0495]	C26	3.5499	3.4657	3.4766	4	-0.4501	-0.5343	-0.5234
	C3	4.0644	3.9402	3.9547	4	0.0644	-0.0598	-0.0453		C27	3.5607	3.5819	3.5457	4	-0.4393	-0.4181	-0.4543
	C4	4.034	4.0443	4.0476	4	0.034	0.0443	0.0476		C28	3.4552	3.5039	3.4958	4	-0.5448	-0.4961	-0.5042
	C5	4.0554	4.0451	4.0544	4	0.0554	0.0451	0.0544		C29	3.4693	3.4346	3.5662	4	-0.5307	-0.5654	-0.4338
	C6	4.1103	4.0536	4.0563	4	0.1103	0.0536	0.0563		C30	3.4507	3.5225	3.5105	4	-0.5493	-0.4775	-0.4895
	C7	4.0529	3.9482	3.956	4	0.0529	-0.0518	-0.044		C31	3.4406	3.4959	3.4682	4	-0.5594	-0.5041	-0.5318
	C8	3.9966	4.0002	3.9556	4	-0.0034	0.0002	-0.0444		C32	3.4612	3.4522	3.4858	4	-0.5388	-0.5478	-0.5142
	С9	3.9466	4.0016	3.9566	4	-0.0534	0.0016	-0.0434		C33	3.476	3.5897	3.4746	4	-0.524	-0.4103	-0.5254
	C10	4.0878	4.0414	4.056	4	0.0878	0.0414	0.056	up	C34	3.5659	3.5035	3.5086	4	-0.4341	-0.4965	-0.4914
	C11	4.0954	4.1043	4.0555	4	0.0954	0.1043	0.0555		C35	3.5604	3.4471	3.5179	4	-0.4396	-0.5529	-0.4821
down	C12	4.0336	4.085	4.0486	4	0.0336	0.085	0.0486		C36	3.559	3.5208	3.4586	4	-0.441	-0.4792	-0.5414
	C13	3.9552	3.947	3.9568	4	-0.0448	-0.053	-0.0432		C _{sum}					-5.9418	-5.9974	-6.0205
	C14	3.9472	3.9391	3.9529	4	-0.0528	-0.0609	-0.0471		N1	6.0817	6.1143	6.1971	5	1.0817	1.1143	1.1971
	C15	3.9435	3.9376	3.944	4	-0.0565	-0.0624	-0.056		N2	6.0843	6.1418	6.1075	5	1.0843	1.1418	1.1075
	C16	3.9711	4.0457	4.0472	4	-0.0289	0.0457	0.0472		N3	6.1176	6.1209	6.0804	5	1.1176	1.1209	1.0804
	C17	4.0499	4.0439	4.0534	4	0.0499	0.0439	0.0534		N4	6.0775	6.0822	6.186	5	1.0775	1.0822	1.186
	C18	3.9693	3.9836	4.046	4	-0.0307	-0.0164	0.046		N5	6.1312	6.1272	6.1293	5	1.1312	1.1272	1.1293
	C19	4.0077	4.0477	3.9476	4	0.0077	0.0477	-0.0524		N6	6.104	6.0746	6.157	5	1.104	1.0746	1.157
	C20	3.9522	3.9369	3.9531	4	-0.0478	-0.0631	-0.0469		N _{sum}					6.5963	6.6610	6.8573
	C21	3.9475	3.9483	3.9548	4	-0.0525	-0.0517	-0.0452		Up _{sum}					0.6545	0.6636	0.8368
	C22	4.0549	4.1116	4.0573	4	0.0549	0.1116	0.0573									
	C23	3.9836	4.0453	4.0519	4	-0.0164	0.0453	0.0519									
	C24	4.0571	4.0521	4.056	4	0.0571	0.0521	0.056									
	Down _{sun}					0.1999	0.1926	0.0738	mid	Li1	0.1457	0.1438	0.0896	1	-0.8543	-0.8562	-0.9104

Table S4 Bader charge and charger transfer for lithium adsorbed at the middle of

 C_2N/G heterostructure with AA stacking.