# Benzotriazole as an efficient ligand in Cu-catalyzed Glaser reaction

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**1.** <sup>1</sup>H and <sup>13</sup>C NMR Spectrum of Symmetric dialkyne (**2a-u**).



Figure S1: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of compound 2a



Figure S2: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 2a



Figure S3: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of compound **2b** 



Figure S4: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 2b



Figure S5: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of compound 2c



Figure S6: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 2c



Figure S7: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of compound 2d



Figure S8: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 2d



Figure S9: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of compound 2e



Figure S10: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 2e



Figure S11: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of compound 2f







Figure S13: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of compound 2g



Figure S14: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 2g



Figure S15: <sup>1</sup>HNMR (500 MHz, CDCl<sub>3</sub>) of compound 2h



Figure S16: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 2h



Figure S17: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of compound 2i



Figure S18: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 2i



Figure S19: <sup>1</sup>HNMR (500 MHz, CDCl<sub>3</sub>) of compound 2j



Figure S20: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 2j



Figure S21: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of compound 2k



Figure S22: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 2k



Figure S23: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of compound 21



Figure S24: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 21



Figure S25: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of compound 2m



Figure S26: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 2m



Figure S27: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of compound 2n





Figure S28: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 2n



Figure S29: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of compound 20



Figure S30: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 20



Figure S31: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of compound 2p



Figure S32: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 2p



Figure S33: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of compound 2q



Figure 34: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 2q

5.0 4.0





Figure S36: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 2r



Figure S37: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of compound 2s



Figure S38: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 2s



Figure S39: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of compound 2t



Figure S40: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 2t



Figure S41: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of compound 2u



Figure S42: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 2u

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**2**<sup>.1</sup>H and <sup>13</sup>C NMR Spectrum of Unsymmetric dialkyne (**3a-e**).

Figure S43: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of compound 3a



Figure S44: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 3a



Figure S45: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of compound 3b



Figure S46: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 3b



Figure S47: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of compound 3c



Figure S48: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 3c



Figure S49: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of compound 3d



Figure S50: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 3d



Figure S51: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of compound 3e



Figure S52: <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) of compound 3e

# **3.** Single Crystal X-Ray data and structure of compounds 2a. Data Collection and Refinement

Data of compound **2a** (Figure S53) was collected on Bruker Apex II CCD-Diffractometer using graphite monochromated MoK $\alpha$  radiation ( $\lambda = 0.71073$  Å). The structures were solved by the Patterson method as the compound is containing Iodine (I<sup>131</sup>), the heavy metal and then refined on  $F^2$  by the full matrix least-squares technique with the SHELX-86 set of software using the WinGX (version 2014.1) program package. All non-hydrogen atoms were refined anisotropically and hydrogen atoms were treated as riding atoms using SHELX default parameters. The process has been validated through the IUCR site (International Union of Crystallography) and no A level error was found. Hence the cystal solved is validated. Further information on the crystal structure (excluding structure factors) has been given in table 1(Supporting Information) and also deposited in the Cambridge Crystallographic Data Centre as supplementary publication number 1856228. Copies of the data can be obtained free of charge upon application to CCDC, 12 Union Road, Cambridge CB2 1EZ, UK (fax: (+44) 1223-336-033. e-mail: <u>deposit@ccdc.cam.ac.uk</u>) or via internet.

Compound 2a						
<b>Empirical Formula</b>	C <sub>16</sub> H <sub>10</sub>					
Formula Weight	202.24					
Crystal System	'Monoclinic''					
Space group	<i>'P 2(2)/1'</i>					
<i>a</i> (Å)	6.5476(5)					
<i>b</i> (Å)	5.9893(4)					
<i>c</i> (Å)	14.4785(12)					
α, β, γ (°)	90.00, 103.07, 90.00					
$V(Å^3)$	553.08(7)					
Ζ	4					
Density (calc)	1.214					

 Table S1. Crystallographic refinement data for compound 2a

F(000)	212
$\mu$ (mm <sup>-1</sup> )	0.069
Crystal Size [mm]	0.40 x 0.36 x 0.29
Temperature (K)	100(2)
Radiation	ΜοΚα 0.71073
θ Min-Max [°]	5.10, 33.21
h, k, l	-9:10; -9:6; -22:22
Tot.,UniqData, R(int)	16338, 2046, 0.0710
Obs. data $[I > 2.0 \sigma(I)]$	0.0685-0.1839
Nref, Npar	2123, 73
R1, wR2, S	0.0685(1750), 0.1961(2046), 0.852
CCDC	1856228



Figure S53. Molecular structure of 2a. Thermal ellipsoids of carbon is set at 40 % probability.

Number	Atom1	Atom2	Atom3	Angle
1	C2	C1	C1	179.4(1)
2	C1	C2	C3	178.4(1)
3	C2	C3	C4	120.3(1)
4	C2	C3	C8	120.1(1)
5	C4	C3	C8	119.6(1)
6	C3	C4	H4	120.0
7	C3	C4	C5	119.9(1)
8	H4	C4	C5	120.1
9	C4	C5	Н5	119.8
10	C4	C5	C6	120.3(1)
11	Н5	C5	C6	119.9
12	C5	C6	H6	119.9
13	C5	C6	C7	120.1(1)
14	H6	C6	C7	120.0
15	C6	C7	H7	119.8
16	C6	C7	C8	120.3(1)
17	H7	C7	C8	119.8

18	C3	C8	C7	119.8(1)
19	C3	C8	H8	120.1
20	C7	C8	H8	120.1
21	C1	C1	C2	179.4(1)
22	C1	C2	C3	178.4(1)
23	C2	C3	C4	120.3(1)
24	C2	C3	C8	120.1(1)
25	C4	C3	C8	119.6(1)
26	C3	C4	H4	120.0
27	C3	C4	C5	119.9(1)
28	H4	C4	C5	120.1
29	C4	C5	Н5	119.8
30	C4	C5	C6	120.3(1)
31	Н5	C5	C6	119.9
32	C5	C6	H6	119.9
33	C5	C6	C7	120.1(1)
34	H6	C6	C7	120.0
35	C6	C7	H7	119.8
36	C6	C7	C8	120.3(1)

37	H7	C7	C8	119.8
38	C3	C8	C7	119.8(1)
39	C3	C8	H8	120.1
				120.1
40	C7	C8	H8	120.1

### Bond lengths of compound 2a

Number	Atom 1	Atom 2	Cyclicity	Length
1	C1	C2	acyclic	1.212(2)
2	C1	C1	acyclic	1.367(2)
3	C2	C3	acyclic	1.431(2)
4	C3	C4	cyclic	1.400(2)
5	C3	C8	cyclic	1.404(2)
6	C4	H4	acyclic	0.950 1
7	C4	C5	cyclic	1.392(2)
8	C5	Н5	acyclic	0.950 1
9	C5	C6	cyclic	1.389(2)
10	C6	H6	acyclic	0.950 1
11	C6	C7	cyclic	1.389(2)
12	C7	H7	acyclic	0.950 1
13	C7	C8	cyclic	1.391(2)
14	C8	H8	acyclic	0.950 1
15	C1	C2	acyclic	1.212(2)

16	C2	C3	acyclic	1.431(2)
17	C3	C4	cyclic	1.400(2)
18	C3	C8	cyclic	1.404(2)
19	C4	H4	acyclic	0.950 1
20	C4	C5	cyclic	1.392(2)
21	C5	Н5	acyclic	0.950 1
22	C5	C6	cyclic	1.389(2)
23	C6	Н6	acyclic	0.950 1
24	C6	C7	cyclic	1.389(2)
25	C7	H7	acyclic	0.950 1
26	C7	C8	cyclic	1.391(2)
27	C8	H8	acyclic	0.950 1