

Dual Tuning of Ultra-thin α -Co(OH)₂ Nanosheets by Solvent Engineering and Coordination Competition for Efficient Oxygen Evolution

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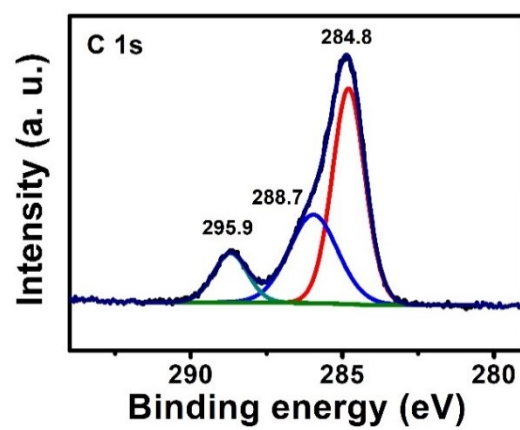


Figure S1. The high-resolution C 1s XPS spectrum

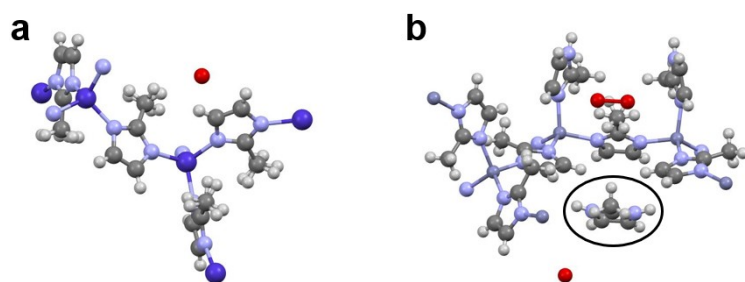


Figure S2. Crystal structures of ZIF-67 (a) and ZIF-L (b).

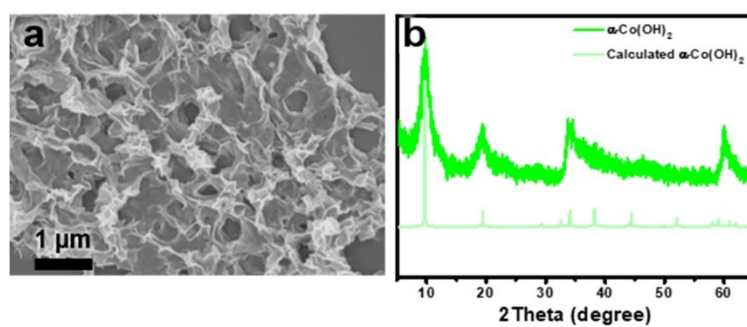


Figure S3. SEM image (a) and the corresponding XRD pattern (b) of the $\alpha\text{-Co(OH)}_2$ obtained by dissolving $\text{Co(NO}_3)_2$ and 2-MI in methanol and water, respectively. The calculated XRD pattern of $\alpha\text{-Co(OH)}_2$ is shown in Figure S3a₁.

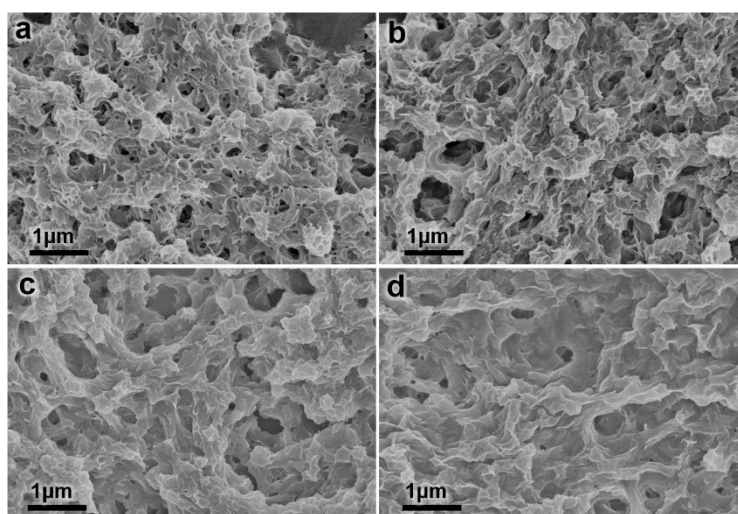


Figure S4. SEM images of the as-prepared products without aging (a) and with different aging time (b: 1h; c: 3h; d: 9h)

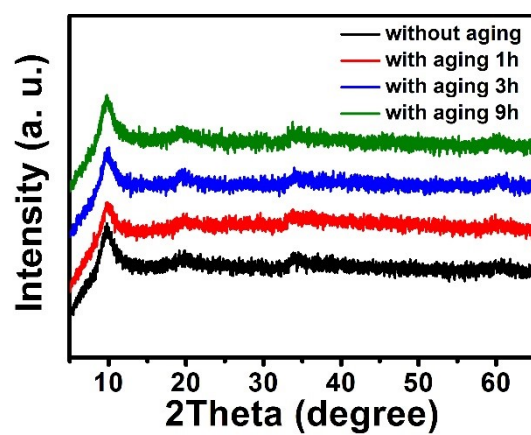


Figure S5. XRD patterns of the as-prepared products without aging and with different aging time

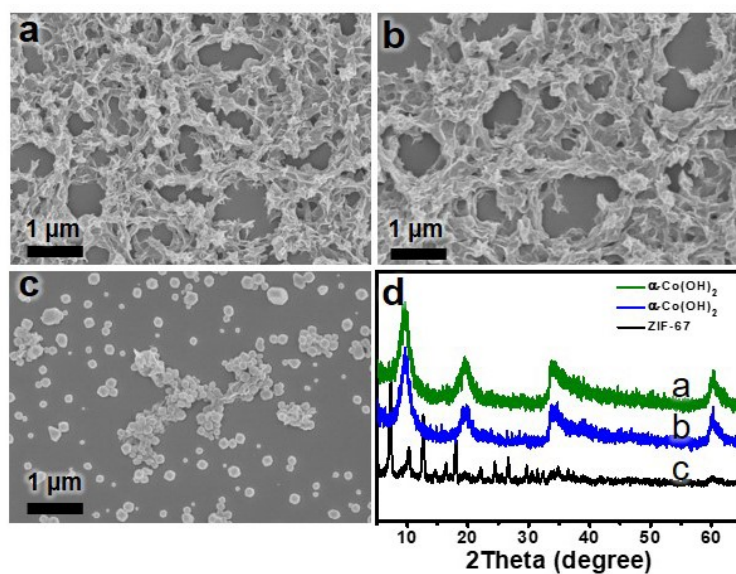


Figure S6. SEM images (a-c) and experimental XRD patterns (d) of as-prepared products with different molar ratio of Co/2-MI (a-1:1, b-1:3, and c-1:4).

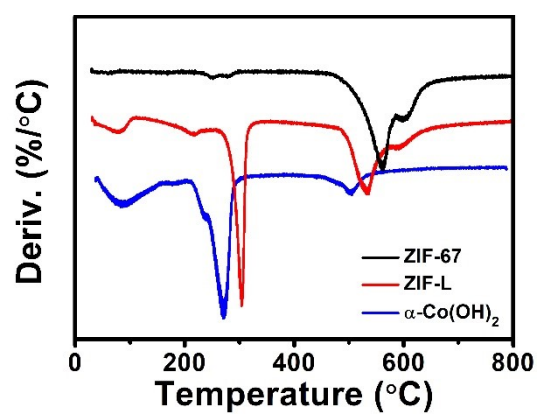


Figure S7. The differential thermal gravity (DTG) curves of the as-prepared ZIF-67, ZIF-L, and α -Co(OH)₂.

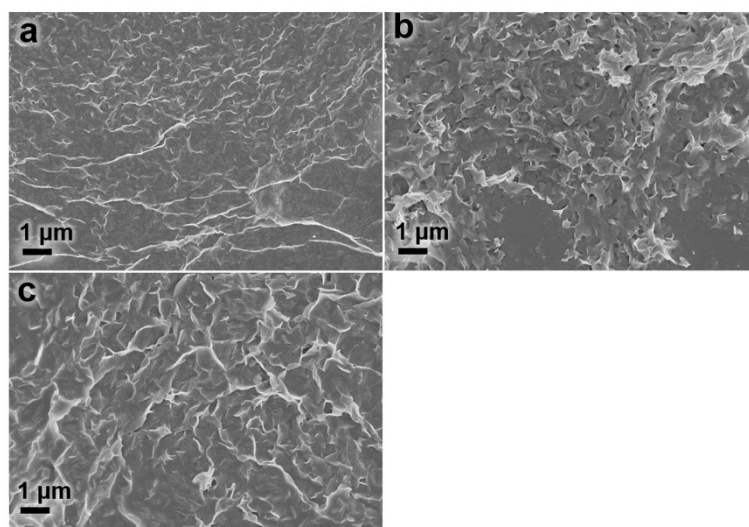


Figure S8. SEM images of the as-prepared products with different imidazole (a: 2-ethylimidazole; b: 2-propylimidazole; c: 2-buthylimidazole)

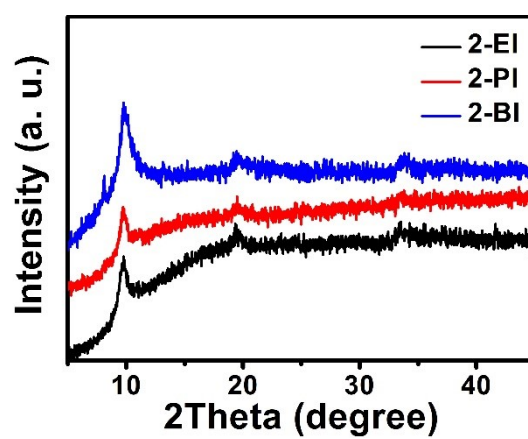


Figure S9. XRD patterns of the as-prepared products with different imidazole (2-ethylimidazole (2-EI); 2-propylimidazole (2-PI); 2-buthylimidazole (2-BI))

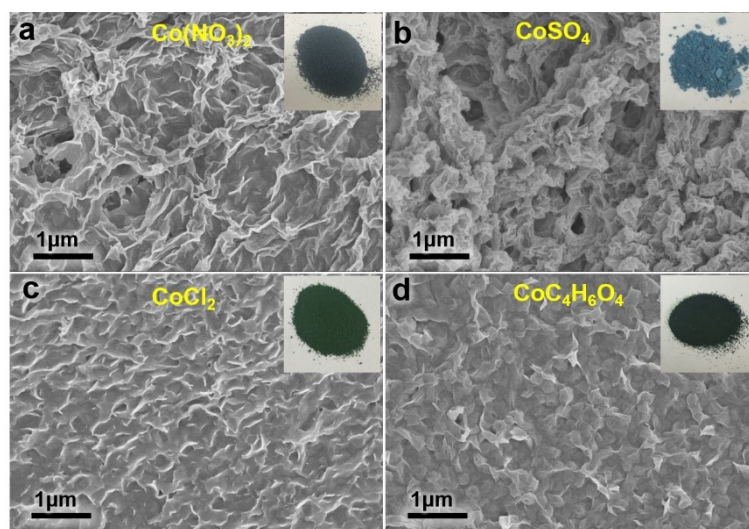


Figure S10. SEM images of the as-prepared products with different Co salts (a: $\text{Co(NO}_3)_2$; b: CoSO_4 ; c: CoCl_2 ; d: $\text{CoC}_4\text{H}_6\text{O}_4$)

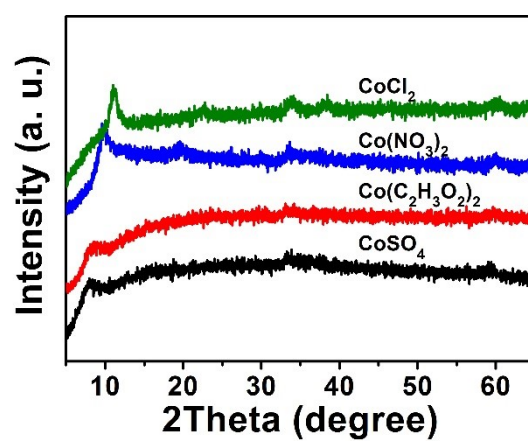


Figure S11. XRD patterns of the as-prepared products with different Co salts

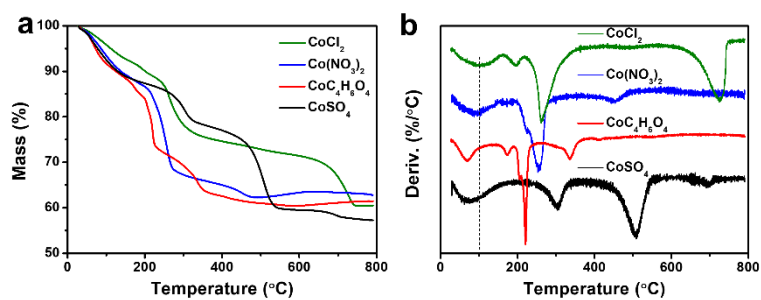


Figure S12. TGA curves (a) and corresponding DTG curves (b) of Co(OH)₂ nanosheets prepared with different salts

The decreased temperature of the water removal from CoCl₂ to Co(NO₃)₂ to CoC₄H₆O₄ to CoSO₄ confirms the increased distance between layers. This is because the water molecules evaporate quickly when located in nanosheets with larger layer distance. The different decomposition temperature for the as-prepared Co(OH)₂ nanosheets further confirms the existence of counterbalance anions between layers.

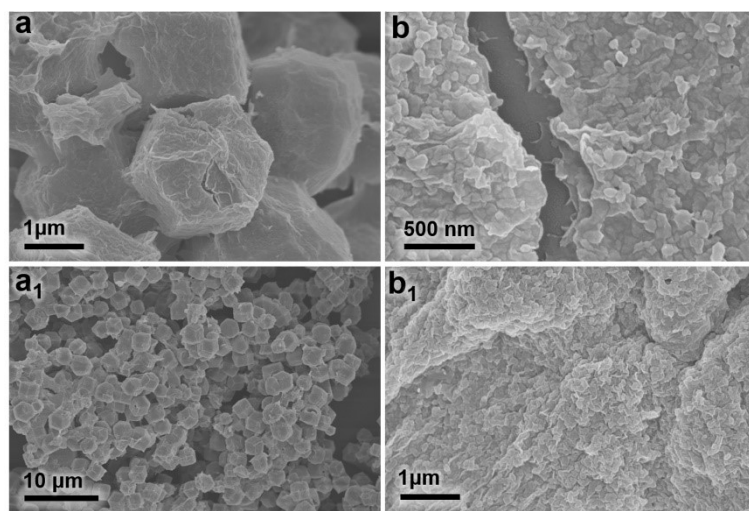


Figure S13. SEM images of the as-prepared materials dispersing the ZIF-67 (a, a₁) and ZIF-L (b, b₁) in water/methanol mixed solvent.

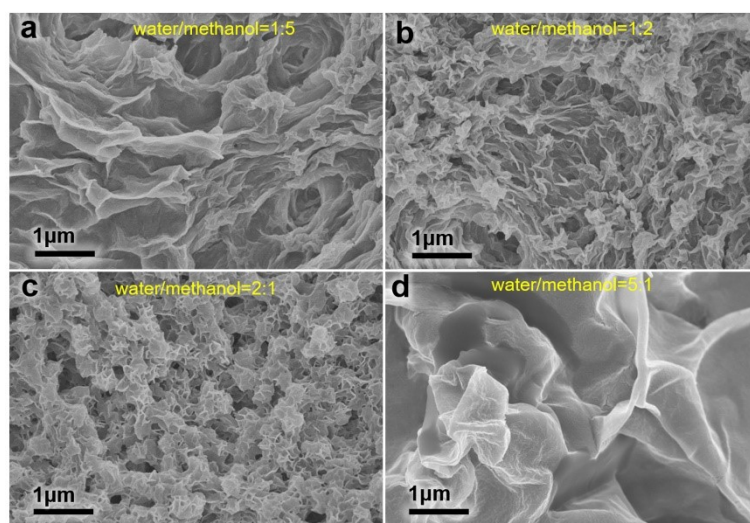


Figure S14. SEM images of the as-prepared products with different amount of solvents

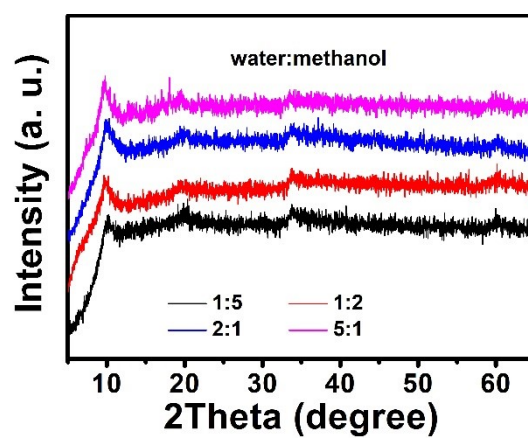


Figure S15. XRD patterns of the as-prepared products with different amount of solvents

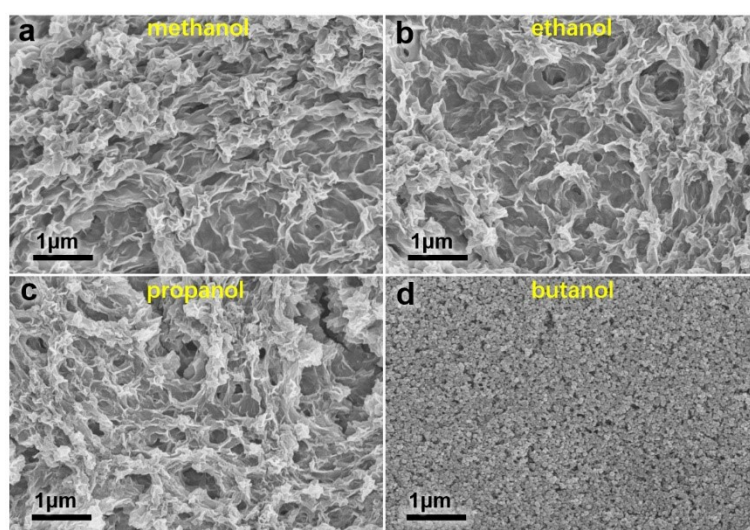


Figure S16. SEM images of the as-prepared products with different solvents (a: methanol; b: ethanol; c: propanol; d: butanol)

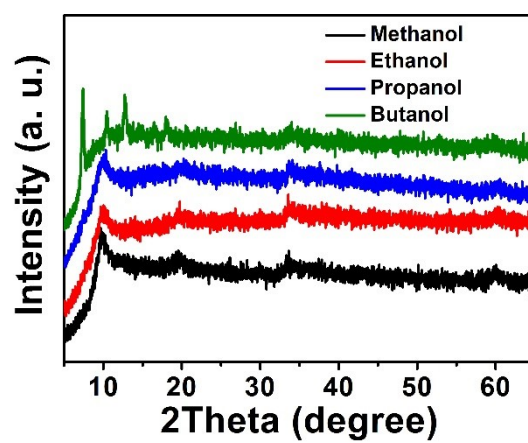


Figure S17. XRD patterns of the as-prepared products with different solvents (methanol, ethanol, propanol, and butanol)

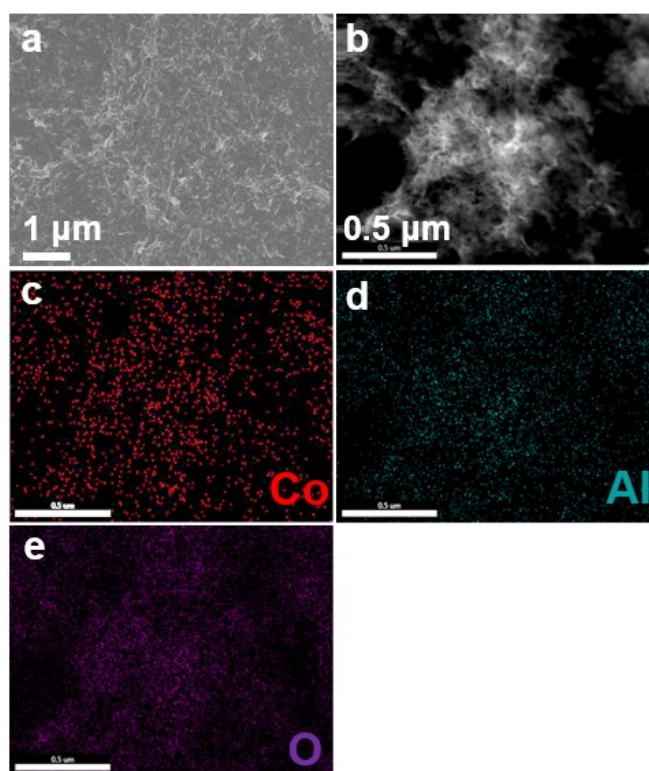


Figure S18. The SEM image (a), HAADF-STEM image (b) and the elemental mapping images (c-e) of the as-prepared Al-Co nanosheets.

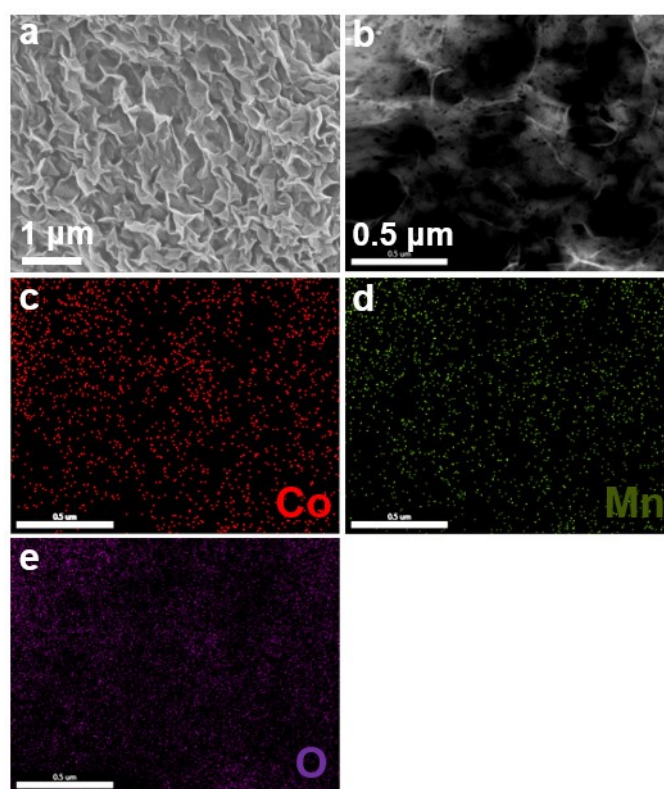


Figure S19. The SEM image (a), HAADF-STEM image (b) and the elemental mapping images (c-e) of the as-prepared Mn-Co nanosheets.

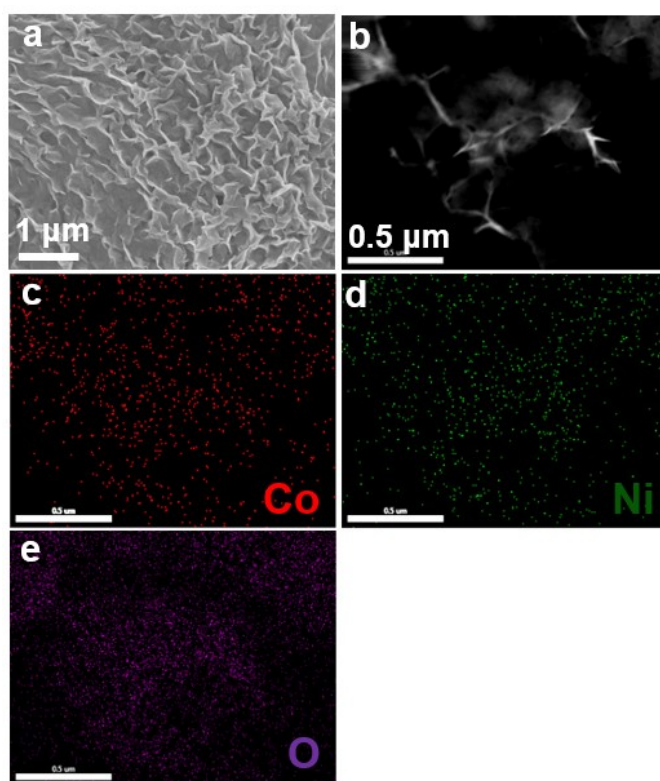


Figure S20. The SEM image (a), HAADF-STEM image (b) and the elemental mapping images (c-e) of the as-prepared Ni-Co nanosheets.

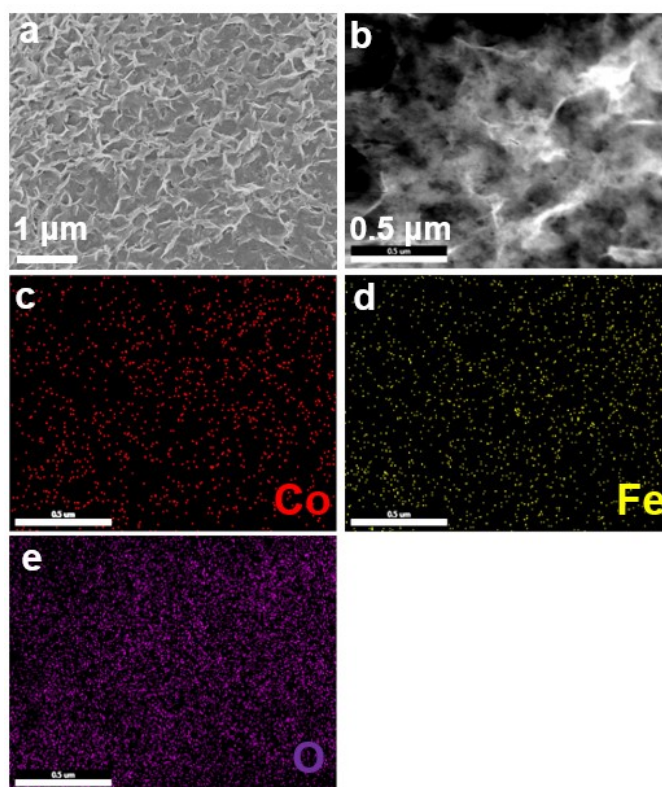


Figure S21. The SEM image (a), HAADF-STEM image (b) and the elemental mapping images (c-e) of the as-prepared Fe-Co nanosheets.

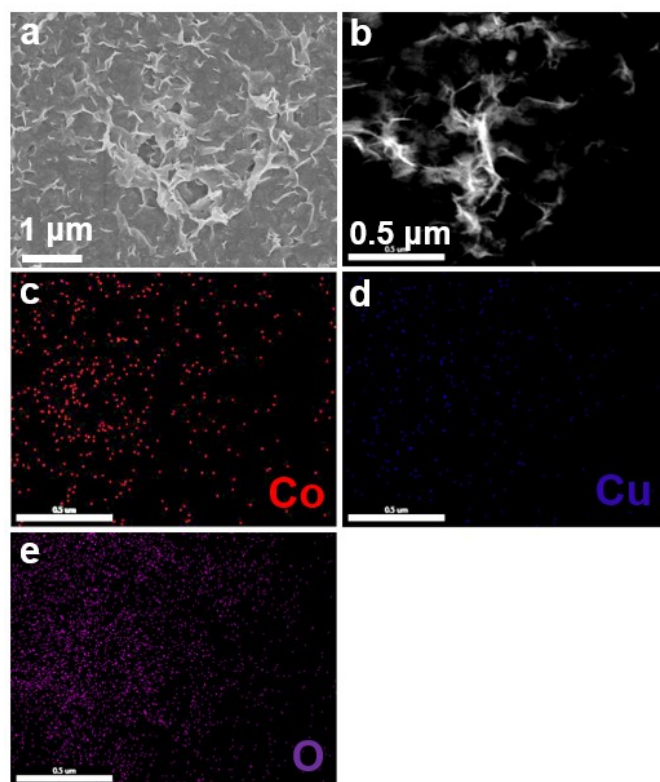


Figure S22. The SEM image (a), HAADF-STEM image (b) and the elemental mapping images (c-e) of the as-prepared Cu-Co nanosheets.

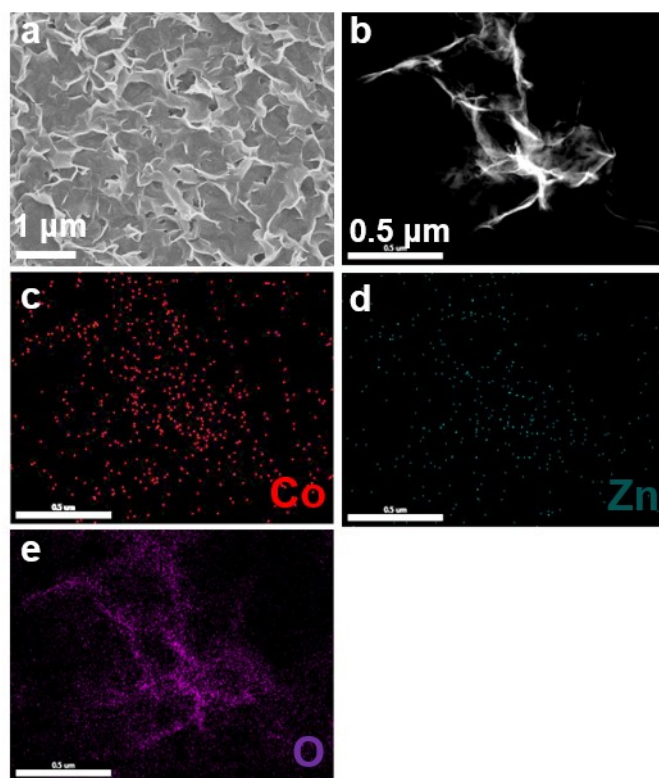


Figure S23. The SEM image (a), HAADF-STEM image (b) and the elemental mapping images (c-e) of the as-prepared Zn-Co nanosheets.

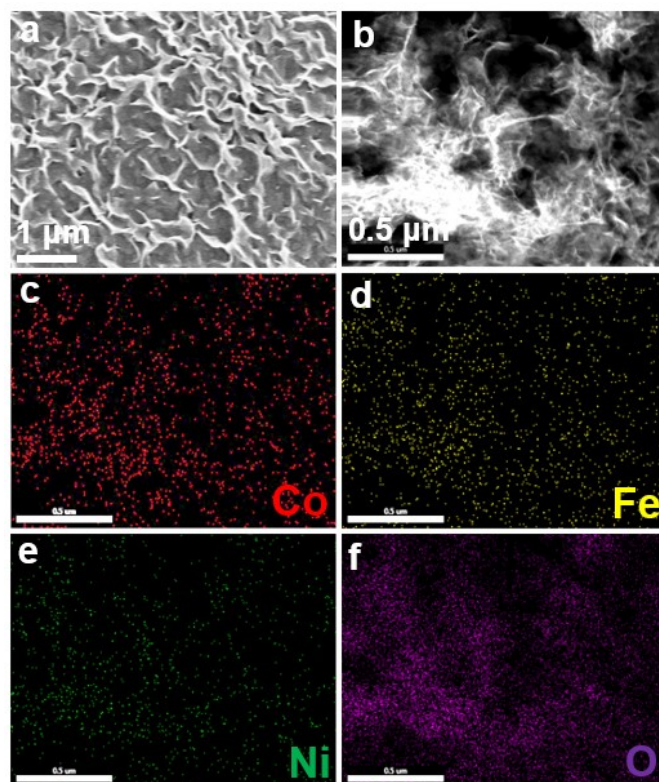


Figure S24. The SEM image (a), HAADF-STEM image (b) and the elemental mapping images (c-f) of the as-prepared Ni-Fe-Co nanosheets.

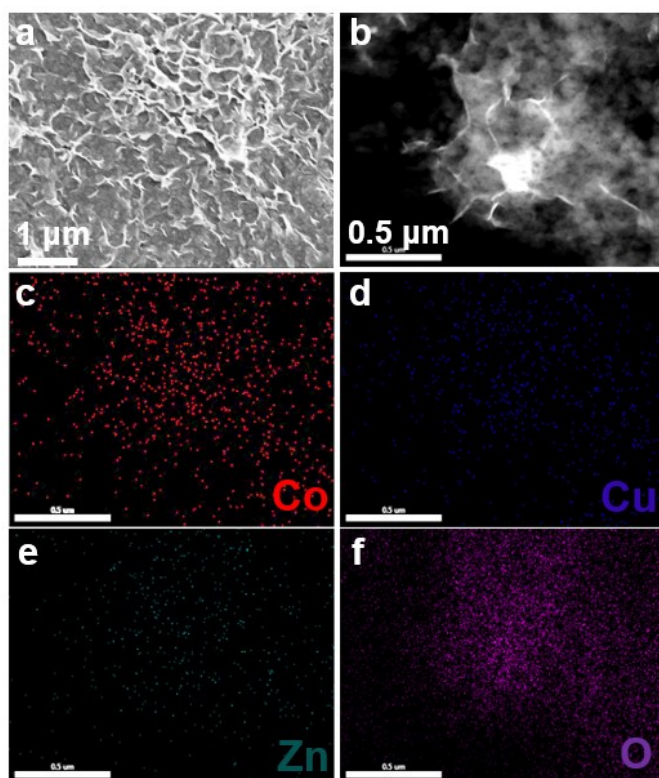


Figure S25. The SEM image (a), HAADF-STEM image (b) and the elemental mapping images (c-f) of the as-prepared Cu-Zn-Co nanosheets.

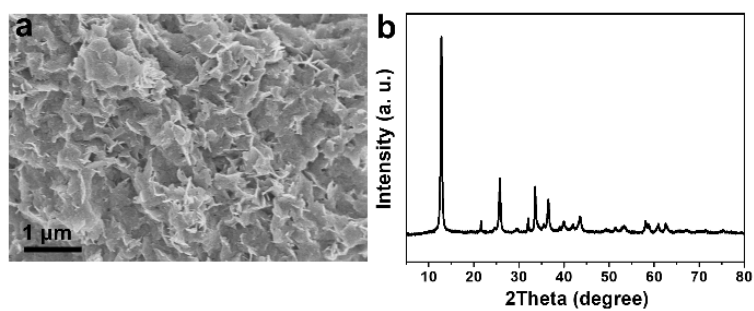


Figure S26. The SEM image and XRD pattern of as-prepared Cu hydroxide nanosheets.

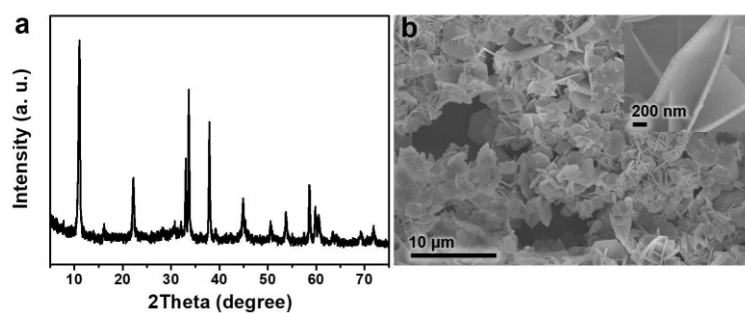


Figure S27. XRD pattern (a) and SEM image (b) of hexagonal α -Co(OH)₂ plates

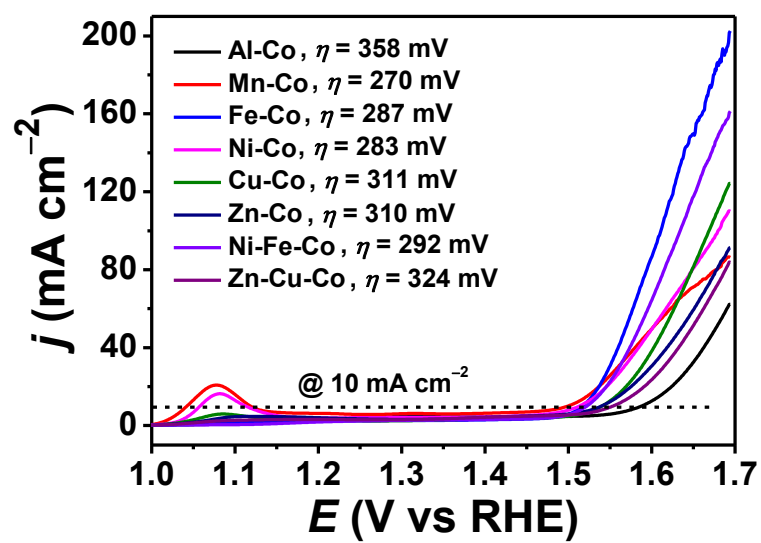


Figure S28. LSV curves of Co-based layered double and triple hydroxides

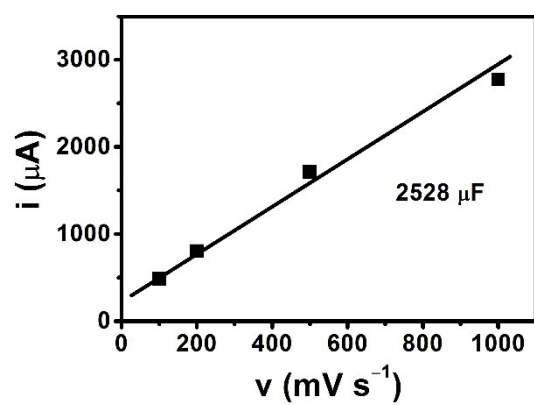


Figure S29. Capacitance currents dependence of scan rates for the $\alpha\text{-Co(OH)}_2$ nanosheets.

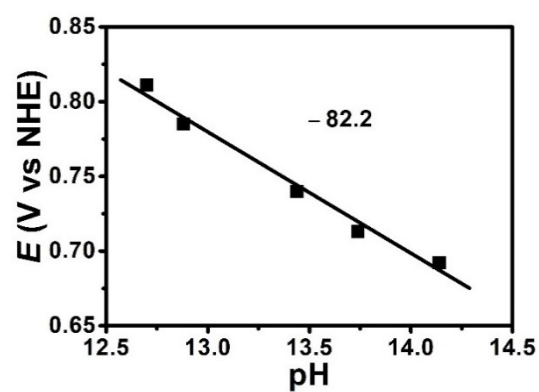


Figure S30. The pH dependence of the potential E at a specific current density (10 mA cm^{-2}) for the $\alpha\text{-Co(OH)}_2$ nanosheets