

Supporting Information

Understanding Water Adsorption and the Impact on CO₂ Capture in Chemically Stable Covalent Organic Frameworks

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Section S-1: Summary of characteristics of the COFs

Table S1. Summary of characteristics of the COFs studied in this work

COFs	V_p , cm ³ g ⁻¹		BET	Geometric surface area	Pore Diameter, Å	
	Lit	Sim	m ² g ⁻¹	m ² g ⁻¹	Lit	Sim
COF-42	0.31 ¹	1.07	710 ⁵	2644	23.0 ⁵	16.47
COF-43	0.36 ⁵	2.07	620 ⁵	2571	38.0 ⁵	31.71
COF-300	0.72 ⁶	1.19	1360 ⁶	3453	7.2 ⁶	8.73
COF-320	0.81 ²	0.60	-	1694	8.0 ²	7.71
COF-LZU1	0.54 ⁷	1.01	410 ⁷	2129	12.0 ⁷	15.15
TpPa-1	-	0.81	535 ⁸	1639	12.5 ⁸	15.27
TpPa-2	-	0.63	339 ⁸	1569	13.5 ⁸	12.87
TpPa-F4	-	0.51	438 ¹	1115	17.0 ¹	14.55
TpPa-NO2	-	0.54	129 ¹	1254	16.0 ¹	
NPN-1	0.48 ⁹	0.42	-	861	7.3×3.3 ⁹	3.51, 5.01
NPN-2	0.54 ⁹	0.46	-	1104	7.8×3.4 ⁹	3.75, 5.01
NPN-3	0.46 ⁹	0.34	-	915	5.2×5.2 ⁹	4.35, 5.25
TpBD	-	1.12	537 ¹⁰	1710	17.2 ¹⁰	22.11
CTF-1	0.40 ¹¹	0.36	791 ¹¹	905	12.0 ¹¹	8.07
DAAQ-TFP	-	1.10	1280 ¹²	1737	23.0 ¹²	21.63

Table S2. Summary of adsorption heats, inflection points and water uptake in $p/p_0 = 0.1$, 0.3 and 0.9 of the COF at 298 K.

COFs	Adsorption heat	Inflection point	Water uptake, cm ³ g ⁻¹		
	kJ mol ⁻¹	RH	$p/p_0 = 0.1$	$p/p_0 = 0.3$	$p/p_0 = 0.9$
COF-42	55.89	0.71	326	394	1315
COF-43	65.89	0.78	143	238	2620
COF-300	13.57	0.52	0.049	0.149	0.821
COF-320	29.69	0.52	0.245	0.773	17.267
COF-LZU1	19.79	0.45	0.050	0.164	0.537
TpPa-1	48.17	0.26	65.0	894	962
TpPa-2	58.60	0.26	163	699	747
TpPa-F4	35.11	0.58	0.820	2.474	75.3
TpPa-NO₂	60.47	0.13	410	612	653
NPN-1	24.10	0.45	1.579	207	344
NPN-2	27.73	0.19	18.532	393	435
NPN-3	31.48	0.26	20.286	282	363
TpBD	58.12	0.65	171	196	1347
CTF-1	6.44	0.45	0.005	0.015	28.871
DAAQ-TFP	38.86	0.22	226	1253	1348

Table S3. Summary of the adsorption heats of 6 water models at low coverage in TpPa-1 and TpPa-2.

COFs	TpPa-1	TpPa-2
	Heat of adsorption (kJ mol ⁻¹)	Heat of adsorption (kJ mol ⁻¹)
SPC	49.00	59.30
SPC/E	48.17	58.60
TIP4P	52.77	61.03
TIP4P_EW	52.79	64.46
TIP5P	39.13	53.49
TIP5P_EW	39.68	52.00

Section S-2: Structure-Property correlations

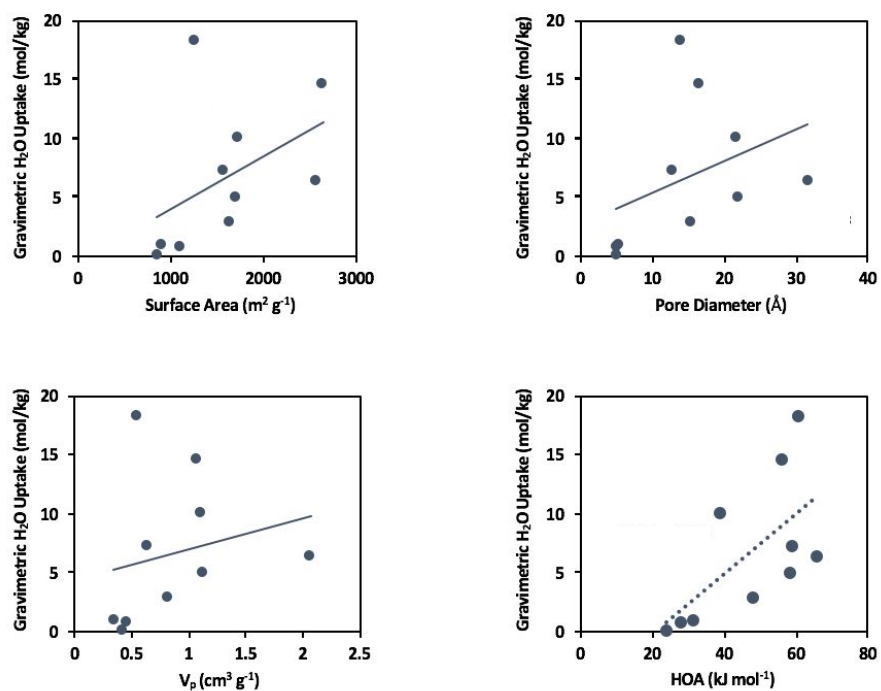


Figure S1. Structure-property correlations between geometric water uptake and the surface area, pore diameter, porosity, heat of adsorption (HOA) of the COFs at RH = 0.1 and 298 K.

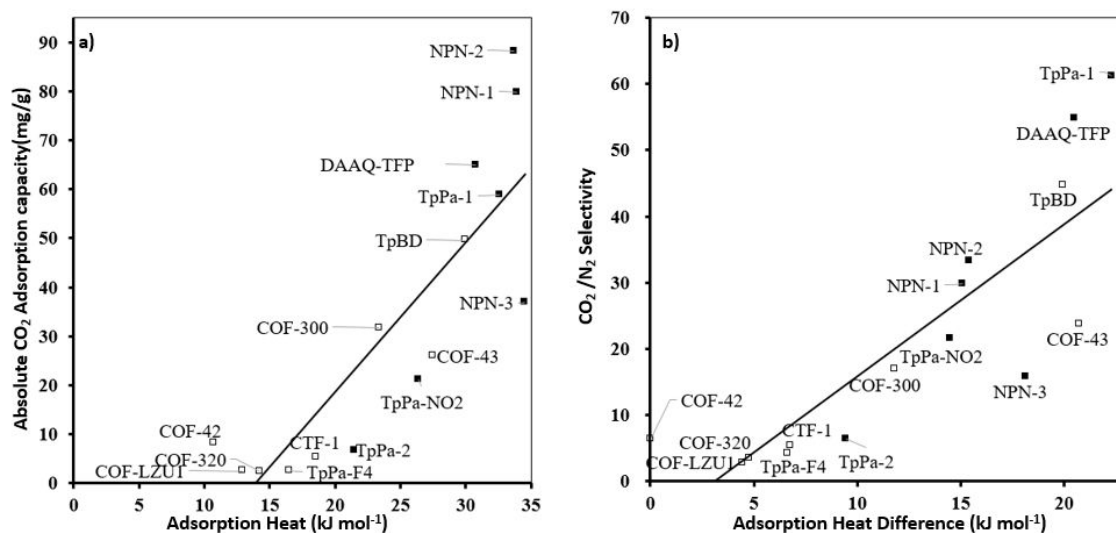


Figure S2. (a) Correlation between absolute CO₂ uptake and its isosteric heat of adsorption in the COFs at 1 bar (partial pressure of CO₂ is 0.15 bar, dry condition). (b) Correlation between CO₂/N₂ selectivity and the adsorption heat difference of CO₂/N₂ in the COFs at 1 bar (partial pressure of CO₂ is 0.15 bar, dry condition).

Section S-3: Model Clusters and atomic partial charges for TpPa-NO₂

TpPa-NO₂

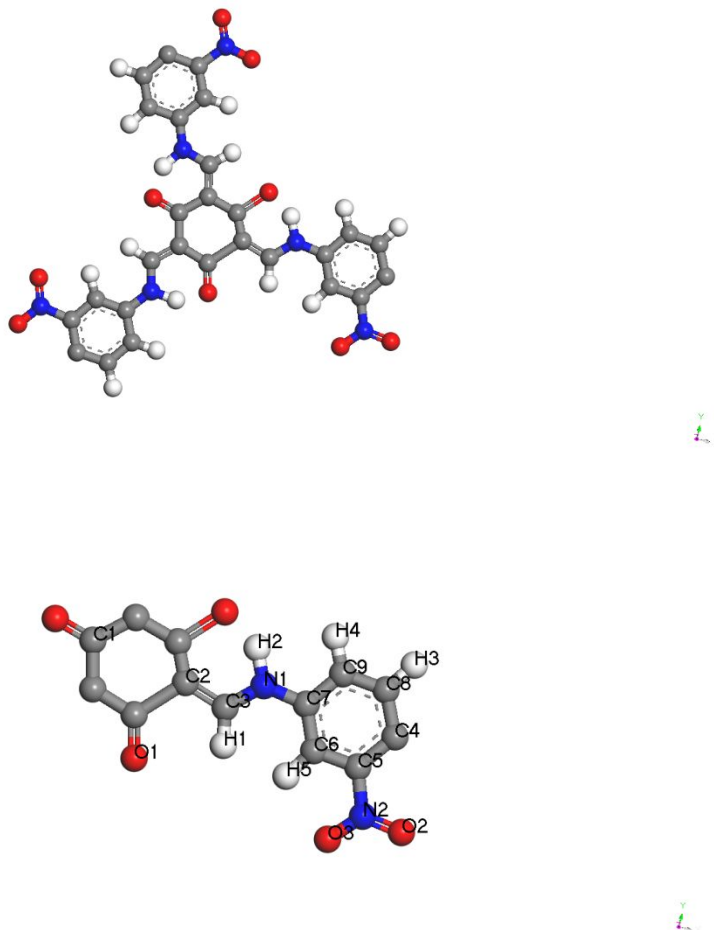


Figure S3. Model cluster used for calculating the partial charges for each atom of TpPa-NO₂.

Table S4. Atomic partial charges calculated using ChelpG method.

Atomic Types	C1	C2	C3	C4	C5	C6	C7	C8	C9	H1
Charge (e)	0.522	-0.347	0.263	0.291	-0.075	-0.260	0.293	-0.209	-0.164	0.071

Atomic Types	H2	H3	H4	H5	N1	N2	O1	O2	O3
Charge (e)	0.371	0.143	0.121	0.171	-0.388	0.721	-0.559	-0.449	-0.449

Section S-4 Simulated water adsorption snapshots in the COFs

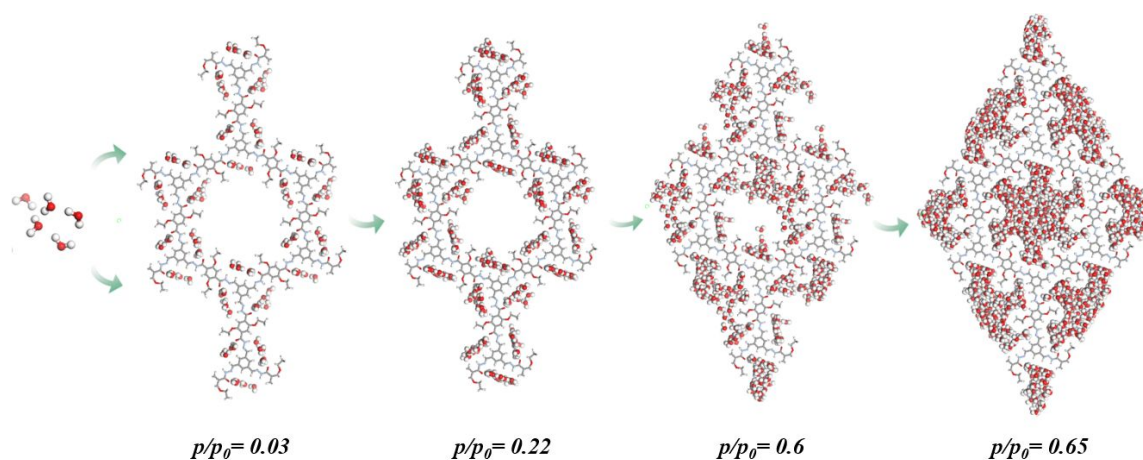


Figure S4. Snapshots of water molecules adsorbed in COF-42 as a function of pressure at 298 K.

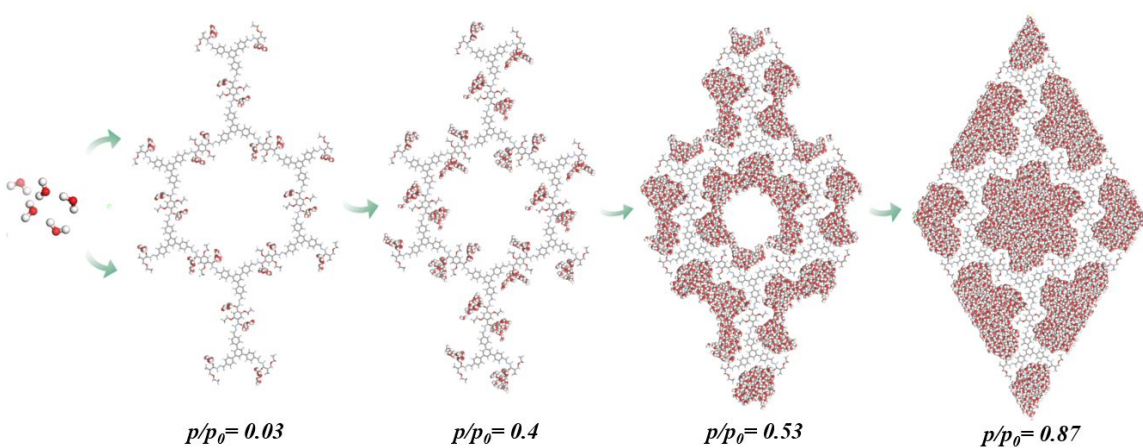


Figure S5. Snapshots of water molecules adsorbed in COF-43 as a function of pressure at 298 K.

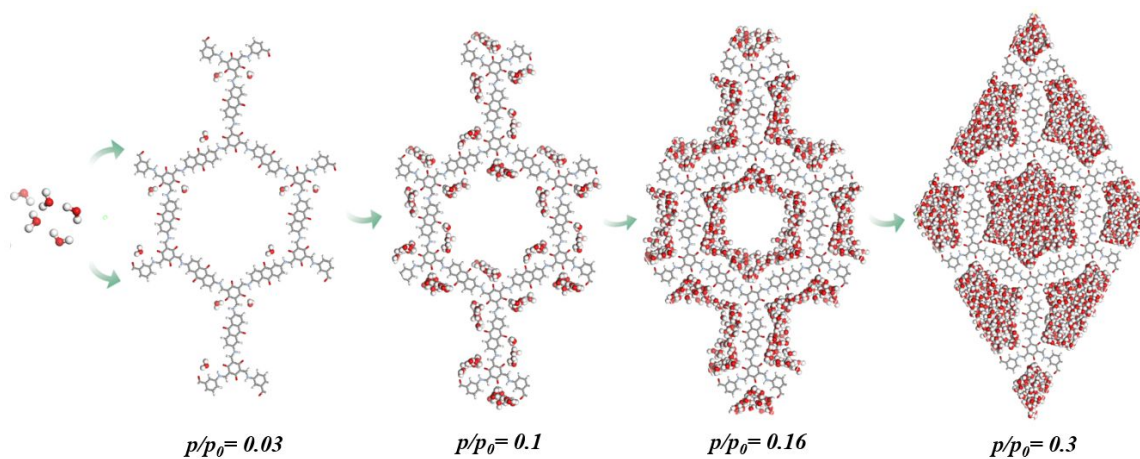


Figure S6. Snapshots of water molecules adsorbed in DAAQ-TFP as a function of pressure at 298 K.

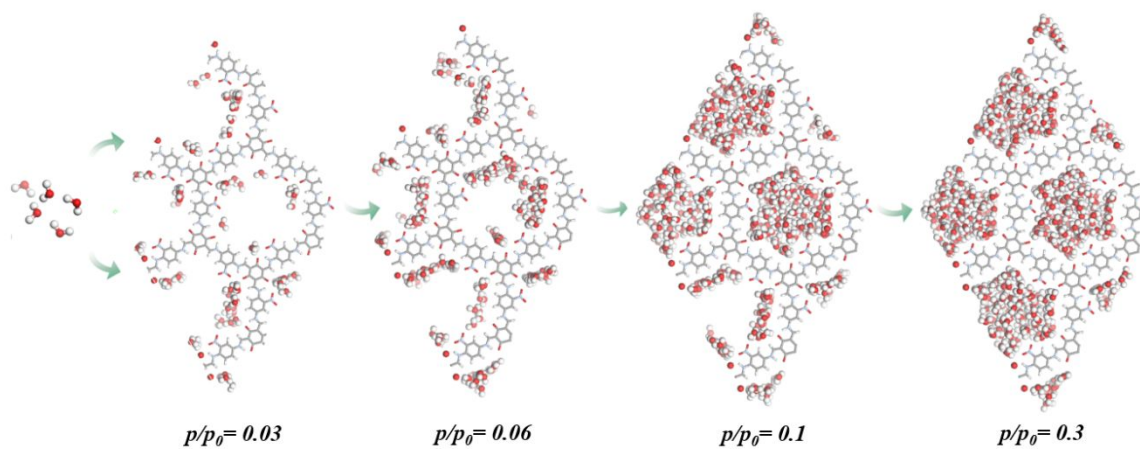


Figure S7. Snapshots of water molecules adsorbed in TpPa-NO₂ as a function of pressure at 298 K.

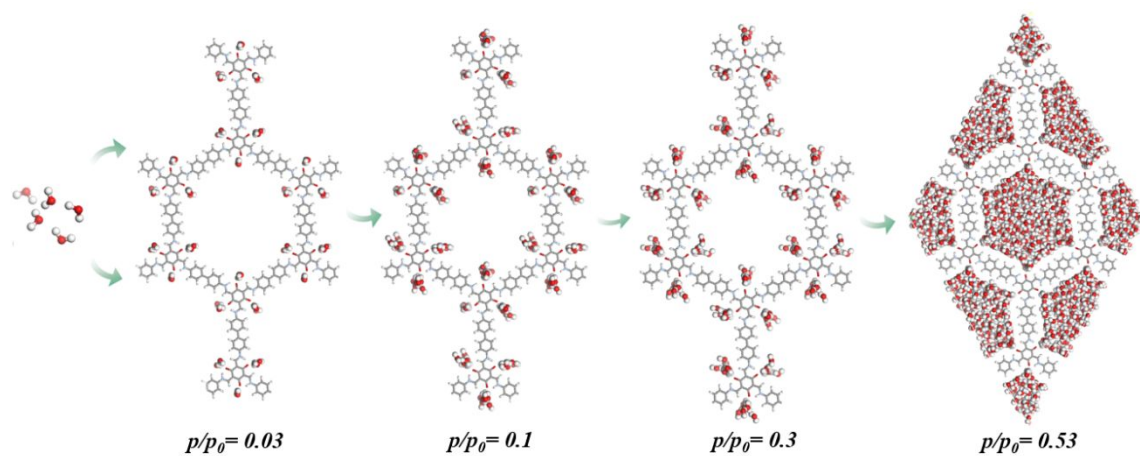


Figure S8. Snapshots of water molecules adsorbed in TpBD as a function of pressure at 298 K.

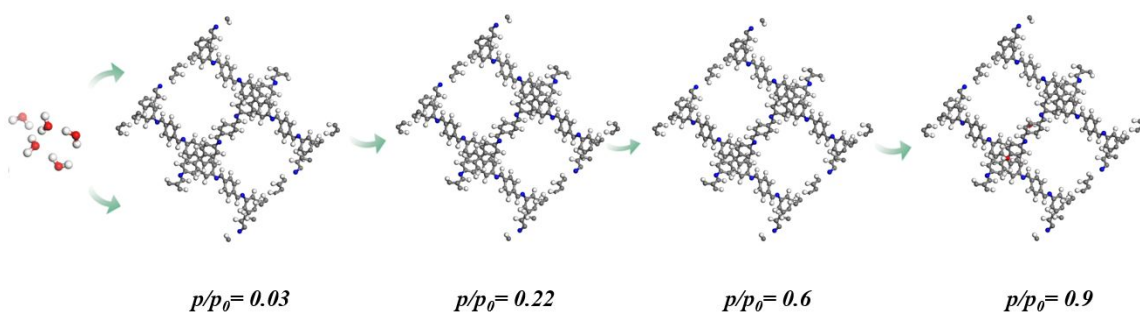


Figure S9. Snapshots of water molecules adsorbed in COF-300 as a function of pressure at 298 K.

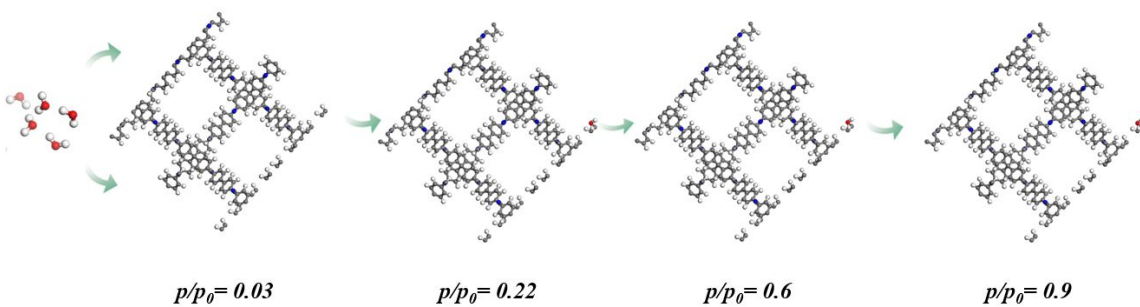


Figure S10. Snapshots of water molecules adsorbed in COF-320 as a function of pressure at 298 K.

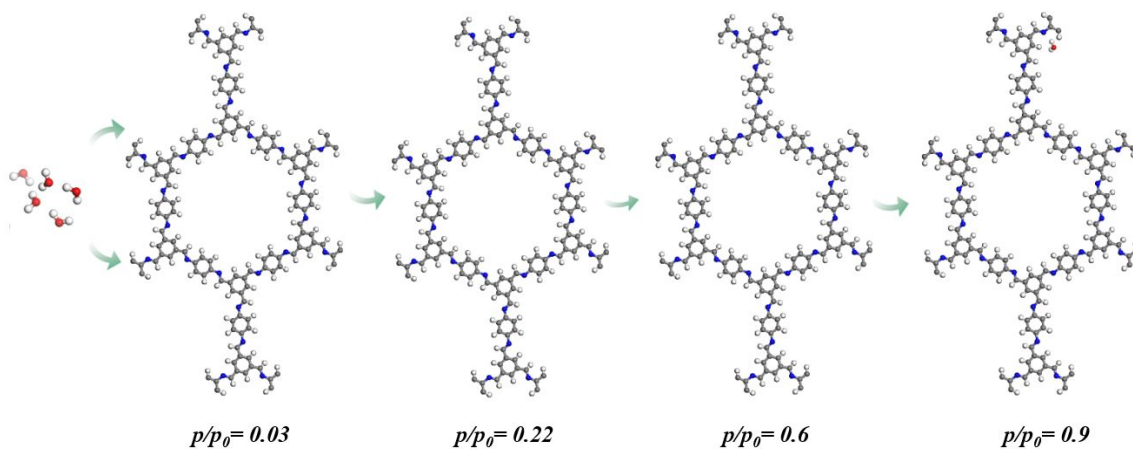


Figure S11. Snapshots of water molecules adsorbed in COF-LZU1 as a function of pressure at 298 K.

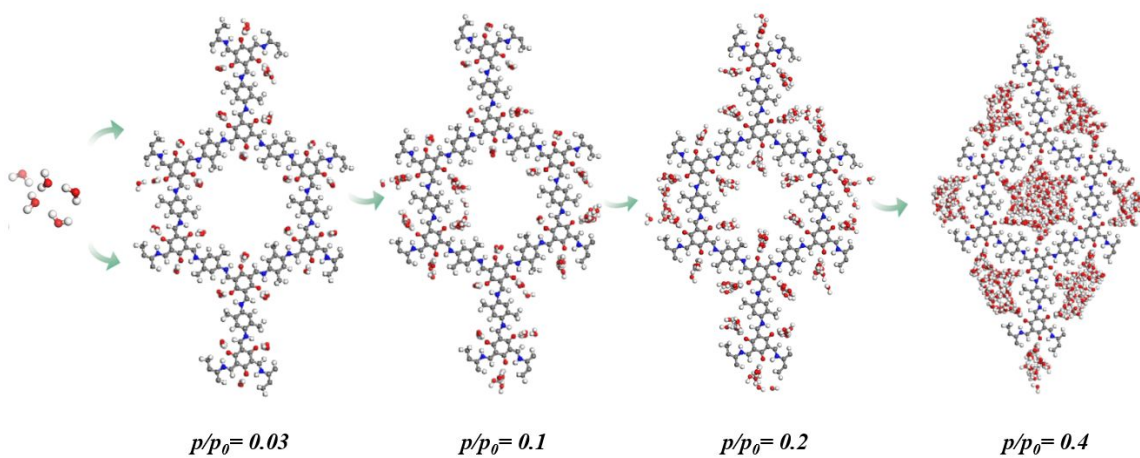


Figure S12. Snapshots of water molecules adsorbed in TpPa-2 as a function of pressure at 298 K.

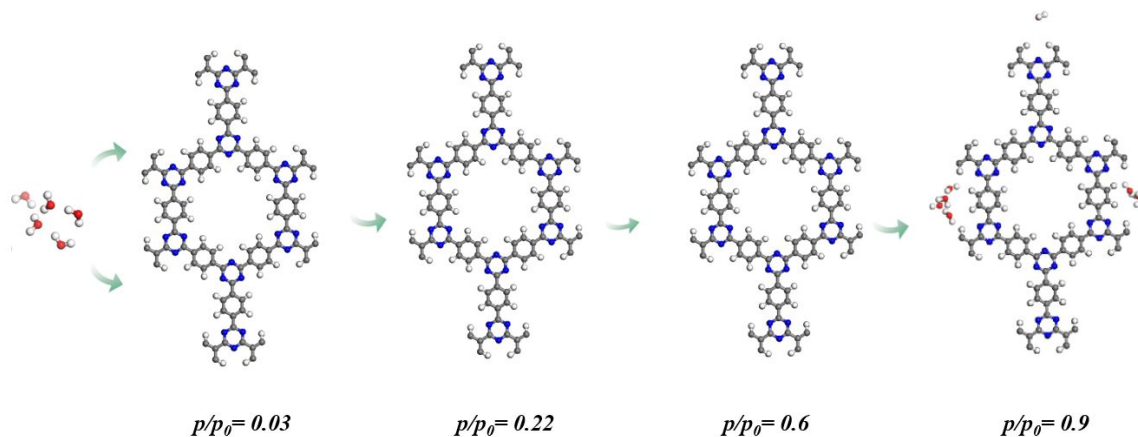


Figure S13. Snapshots of water molecules adsorbed in CTF-1 as a function of pressure at 298 K.

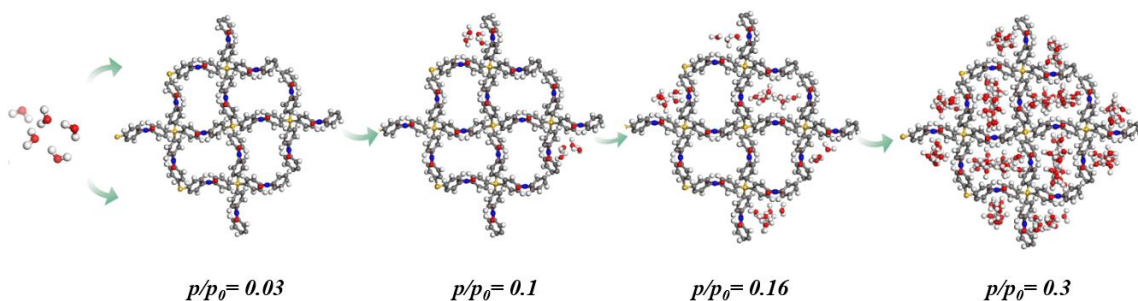


Figure S14. Snapshots of water molecules adsorbed in NPN-2 as a function of pressure at 298 K.

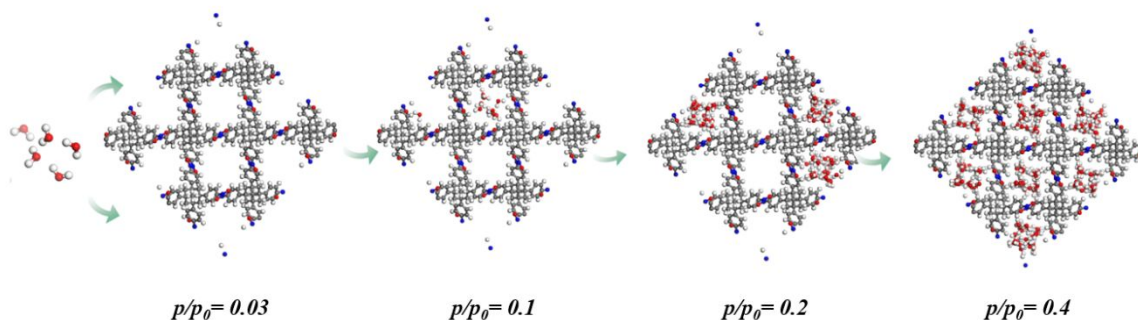


Figure S15. Snapshots of water molecules adsorbed in NPN-3 as a function of pressure at 298 K.

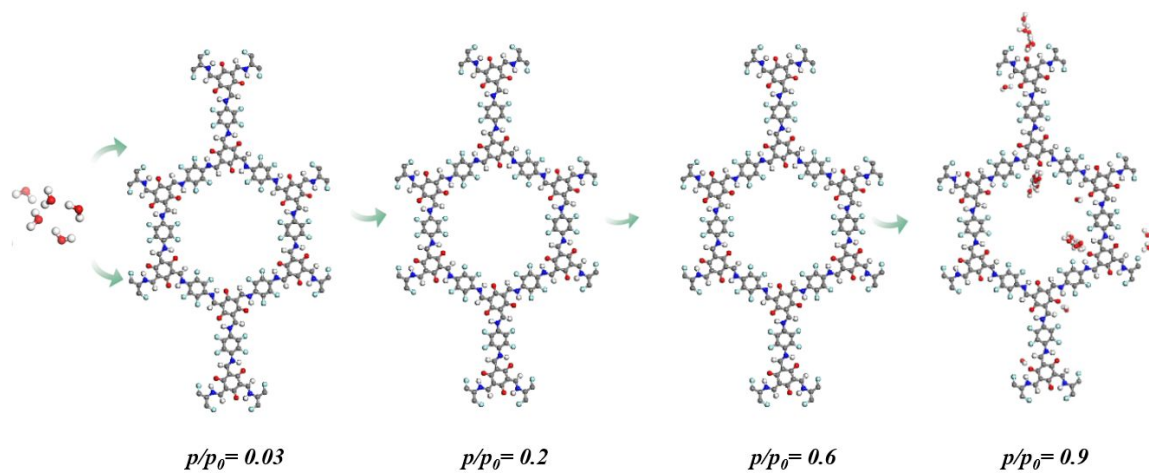


Figure S16. Snapshots of water molecules adsorbed in TpPa-F4 as a function of pressure at 298 K.

Section S-5 Simulated water and CO₂ adsorption density plots in the COFs.

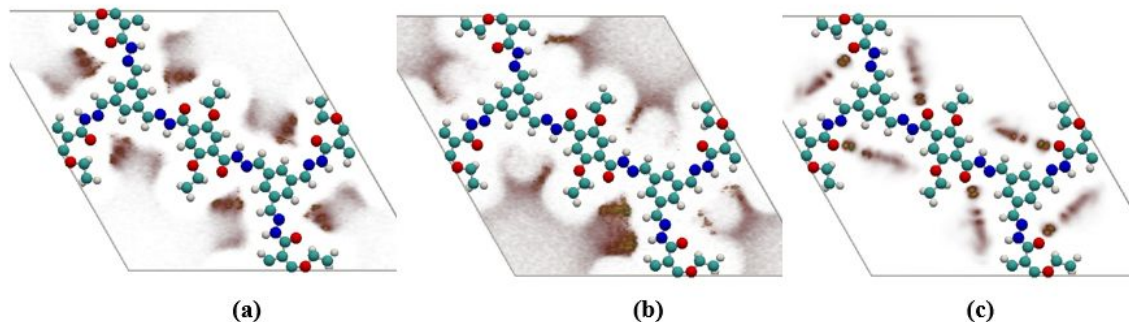


Figure S17. The simulated adsorption density plots of CO₂ (a) adsorption in CO₂/N₂ mixture (15:85) with RH = 0, CO₂ (b) and water (c) adsorption in CO₂/N₂ mixture (15:85) with RH = 0.1 in COF-42 at 298 K and 1 bar. (Color code: N, blue; O, red; C, green; H, white.)

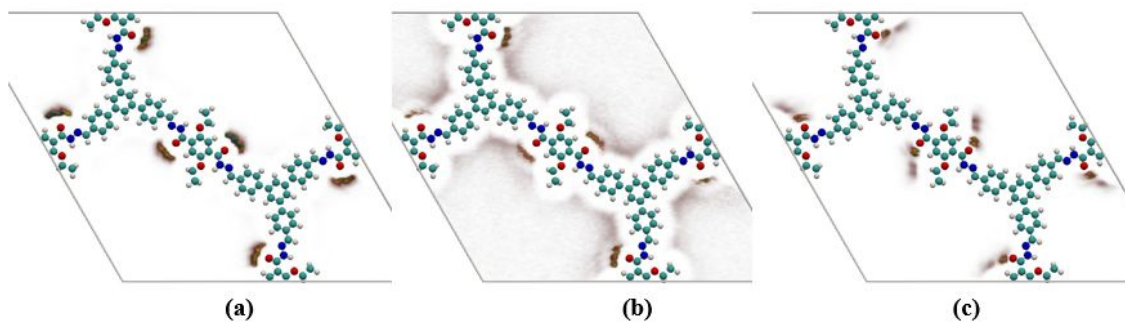


Figure S18. The simulated adsorption density plots of CO₂ (a) adsorption in CO₂/N₂ mixture (15:85) with RH = 0, CO₂ (b) and water (c) adsorption in CO₂/N₂ mixture (15:85) with RH = 0.1 in COF-43 at 298 K and 1 bar. (Color code: N, blue; O, red; C, green; H, white.)

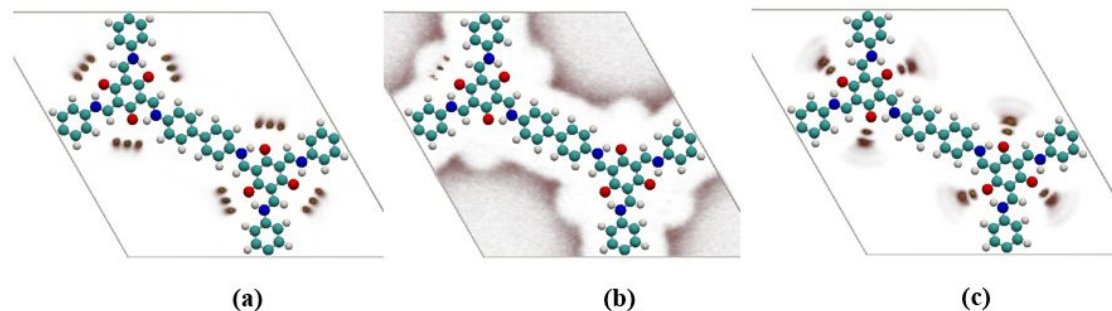


Figure S19. The simulated adsorption density plots of CO₂ (a) adsorption in CO₂/N₂ mixture (15:85) with RH = 0, CO₂ (b) and water (c) adsorption in CO₂/N₂ mixture (15:85) with RH = 0.1 in TpBD at 298 K and 1 bar. (Color code: N, blue; O, red; C, green; H, white.)

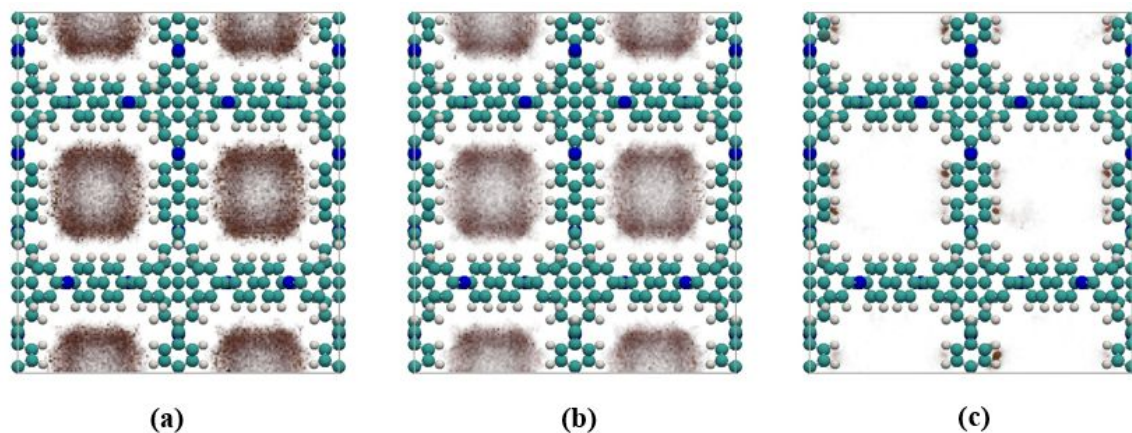


Figure S20. The simulated adsorption density plots of CO₂ (a) adsorption in CO₂/N₂ mixture (15:85) with RH = 0, CO₂ (b) and water (c) adsorption in CO₂/N₂ mixture (15:85) with RH = 0.1 in COF-320 at 298 K and 1 bar. (Color code: N, blue; O, red; C, green; H, white.)

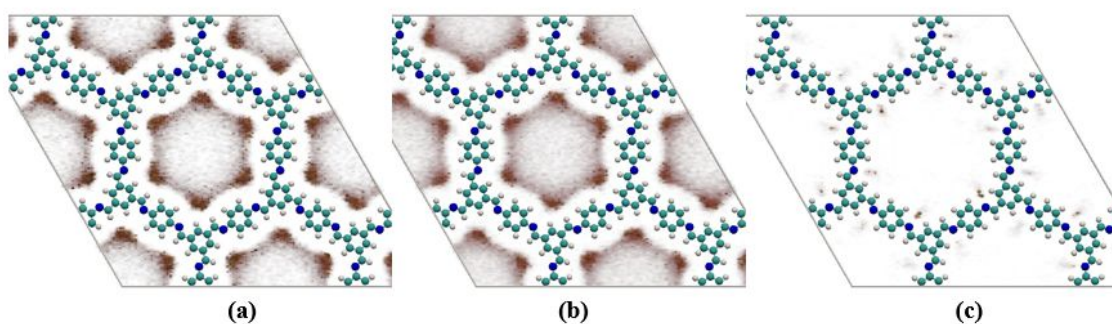


Figure S21. The simulated adsorption density plots of CO₂ (a) adsorption in CO₂/N₂ mixture (15:85) with RH = 0, CO₂ (b) and water (c) adsorption in CO₂/N₂ mixture (15:85) with RH = 0.1 in COF-LZU1 at 298 K and 1 bar. (Color code: N, blue; O, red; C, green; H, white.)

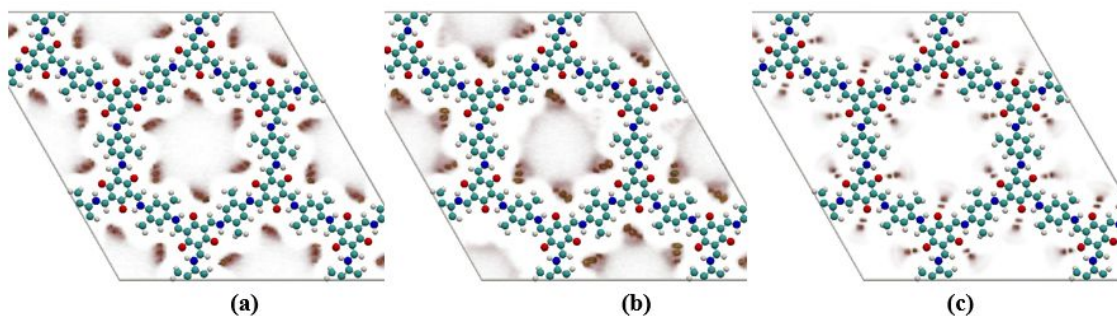


Figure S22. The simulated adsorption density plots of CO₂ (a) adsorption in CO₂/N₂ mixture (15:85) with RH = 0, CO₂ (b) and water (c) adsorption in CO₂/N₂ mixture (15:85) with RH = 0.1 in TpPa-2 at 298 K and 1 bar. (Color code: N, blue; O, red; C, green; H, white.)

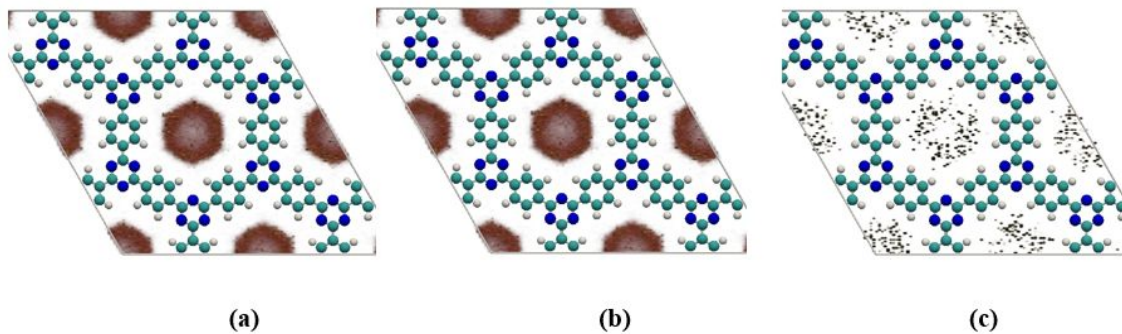


Figure S23. The simulated adsorption density plots of CO₂ (a) adsorption in CO₂/N₂ mixture (15:85) with RH = 0, CO₂ (b) and water (c) adsorption in CO₂/N₂ mixture (15:85) with RH = 0.1 in CTF-1 at 298 K and 1 bar. (Color code: N, blue; O, red; C, green; H, white.)

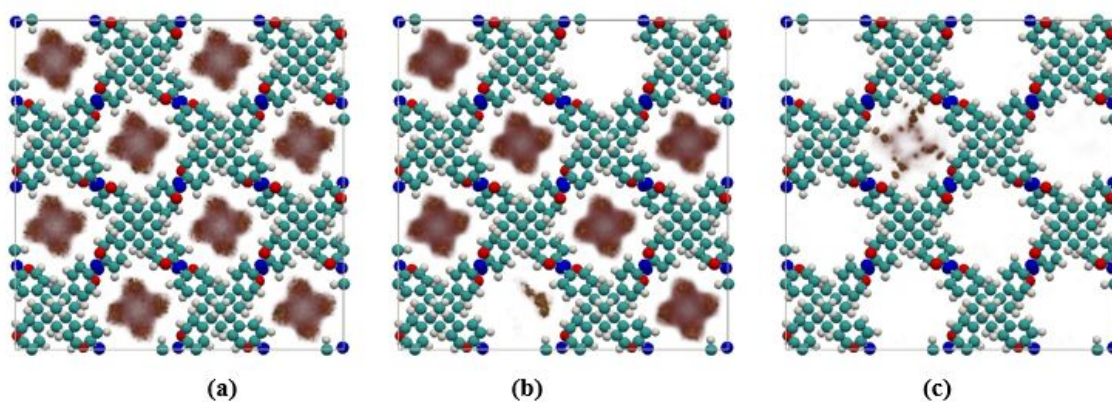


Figure S24. The simulated adsorption density plots of CO₂ (a) adsorption in CO₂/N₂ mixture (15:85) with RH = 0, CO₂ (b) and water (c) adsorption in CO₂/N₂ mixture (15:85) with RH = 0.1 in NPN-3 at 298 K and 1 bar. (Color code: N, blue; O, red; C, green; H, white.)

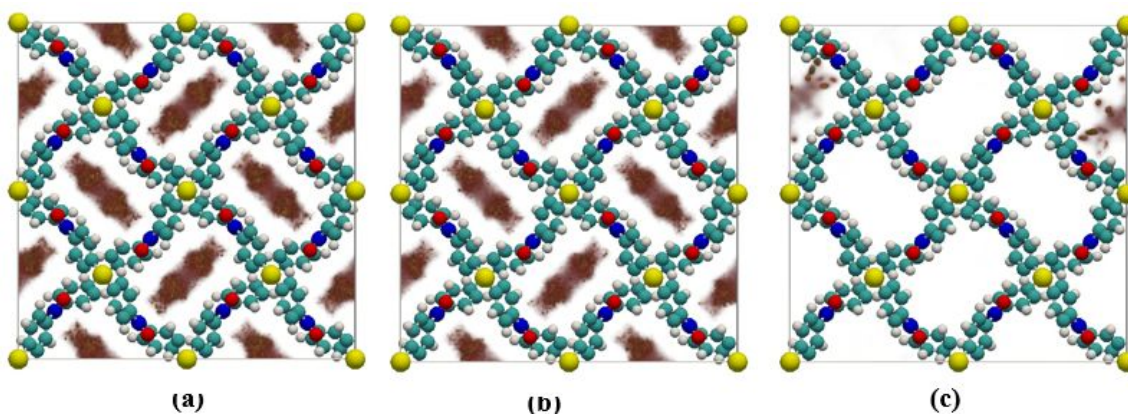


Figure S25. The simulated adsorption density plots of CO₂ **(a)** adsorption in CO₂/N₂ mixture (15:85) with RH = 0, CO₂ **(b)** and water **(c)** adsorption in CO₂/N₂ mixture (15:85) with RH = 0.1 in NPN-2 at 298 K and 1 bar. (Color code: N, blue; O, red; C, green; H, white.)

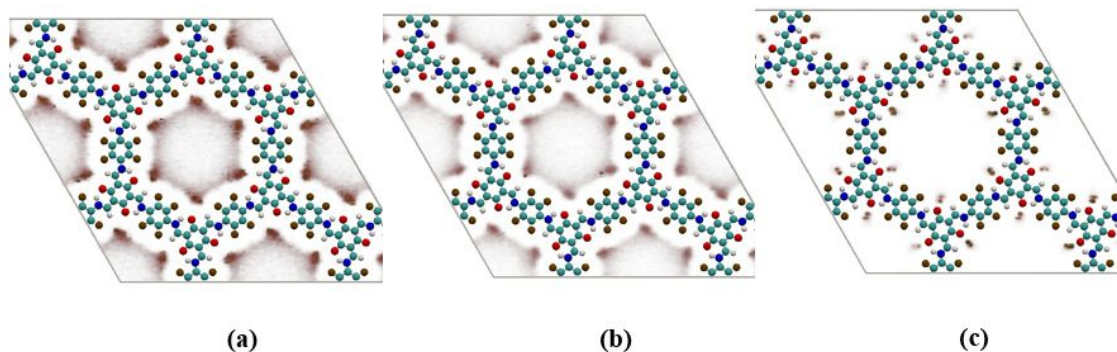


Figure S26. The simulated adsorption density plots of CO₂ **(a)** adsorption in CO₂/N₂ mixture (15:85) with RH = 0, CO₂ **(b)** and water **(c)** adsorption in CO₂/N₂ mixture (15:85) with RH = 0.1 in TpPa-F4 at 298 K and 1 bar. (Color code: N, blue; O, red; C, green; H, white.)

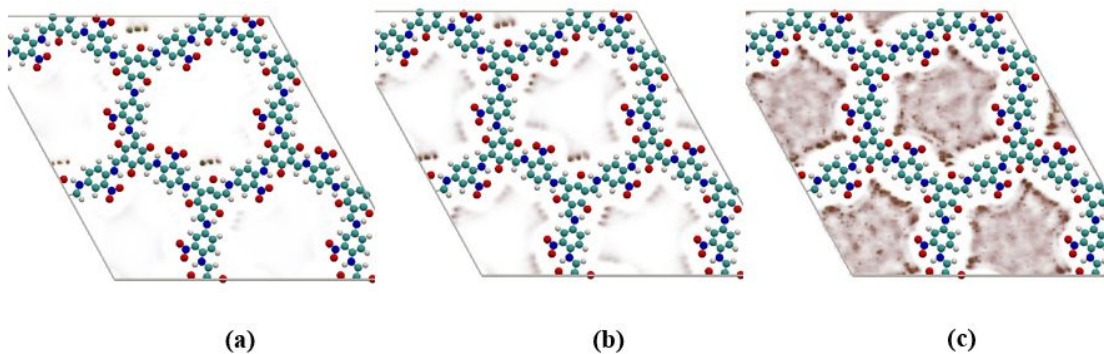


Figure S27. The simulated adsorption density plots of CO₂ **(a)** adsorption in CO₂/N₂ mixture (15:85) with RH = 0, CO₂ **(b)** and water **(c)** adsorption in CO₂/N₂ mixture (15:85) with RH = 0.1 in TpPa-NO₂ at 298 K and 1 bar. (Color code: N, blue; O, red; C, green; H, white.)

Section S-6 Pore Size Distributions of the COFs

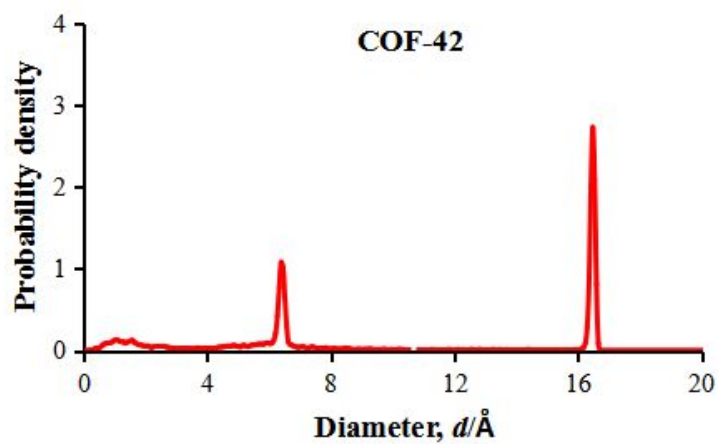


Figure S28. Pore Size distribution for COF-42.

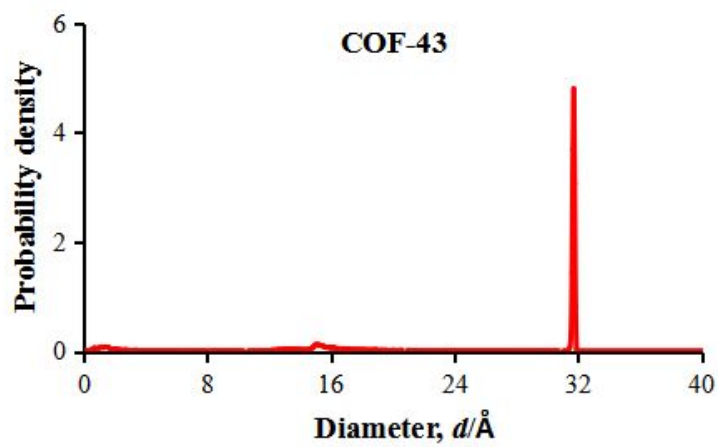


Figure S29. Pore Size distribution for COF-43.

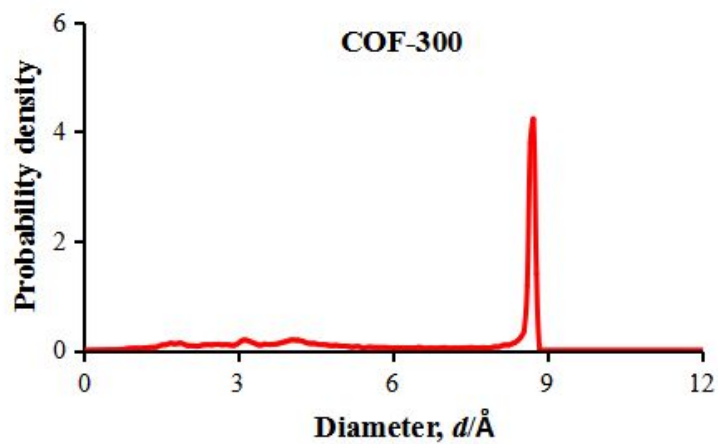


Figure S30. Pore Size distribution for COF-300.

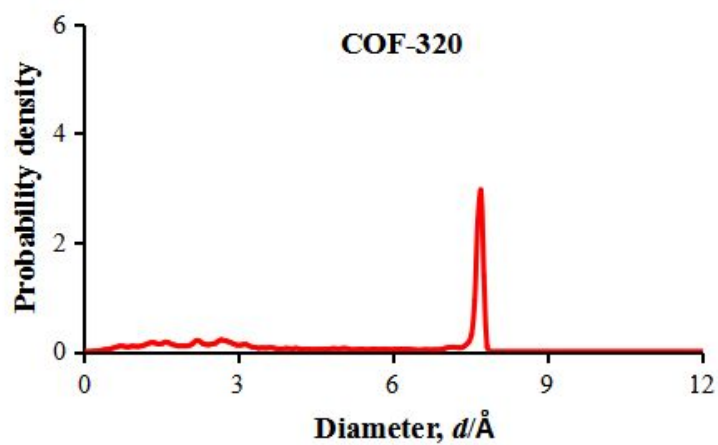


Figure S31. Pore Size distribution for COF-320.

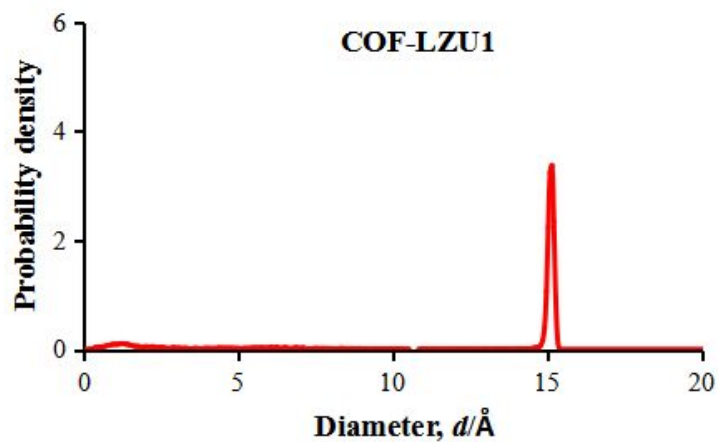


Figure S32. Pore Size distribution for COF-LZU1.

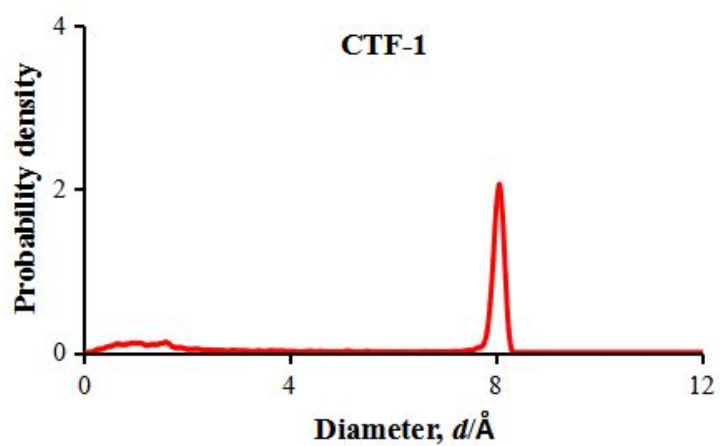


Figure S33. Pore Size distribution for CTF-1.

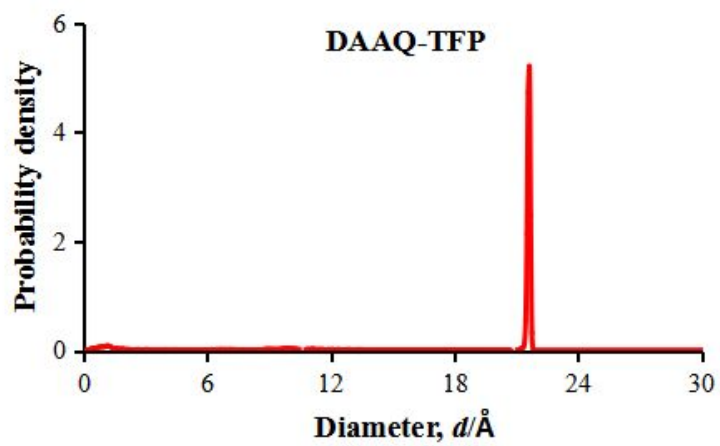


Figure S34. Pore Size distribution for DAAQ-TFP.

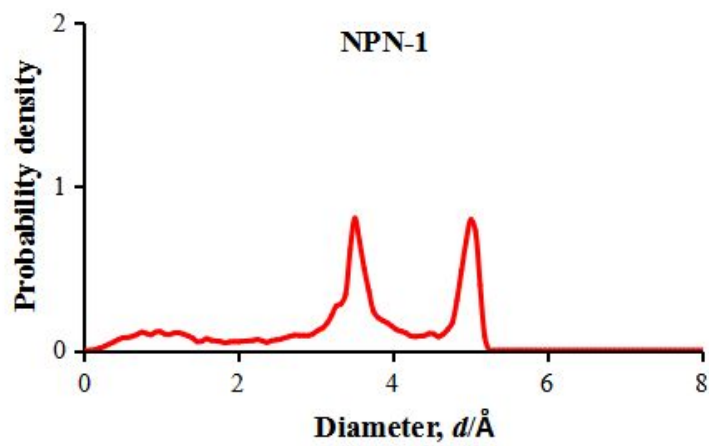


Figure S35. Pore Size distribution for NPN-1.

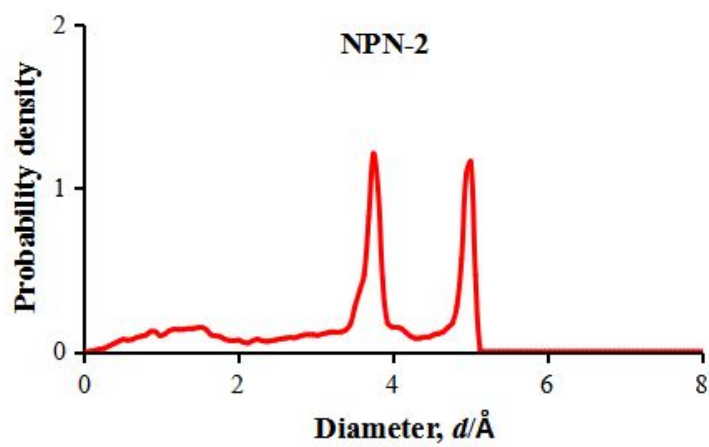


Figure S36. Pore Size distribution for NPN-2.

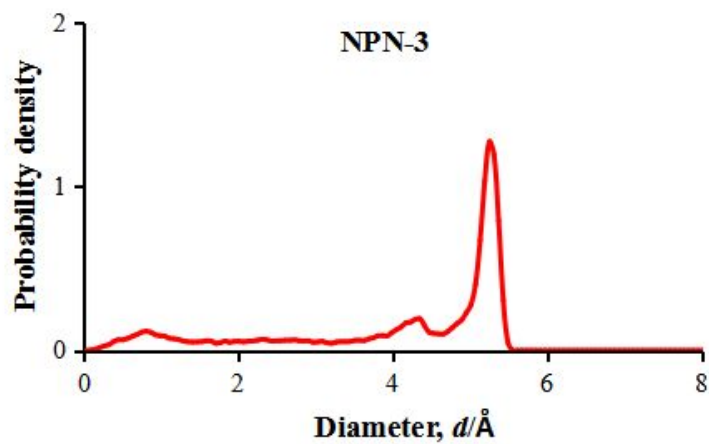


Figure S37. Pore Size distribution for NPN-3.

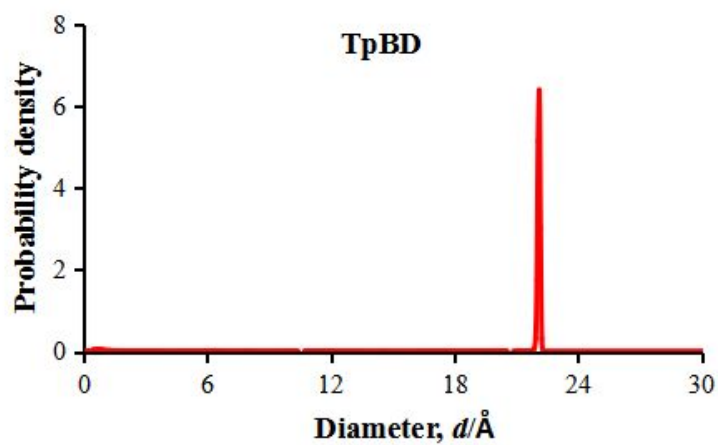


Figure S38. Pore Size distribution for TpBD.

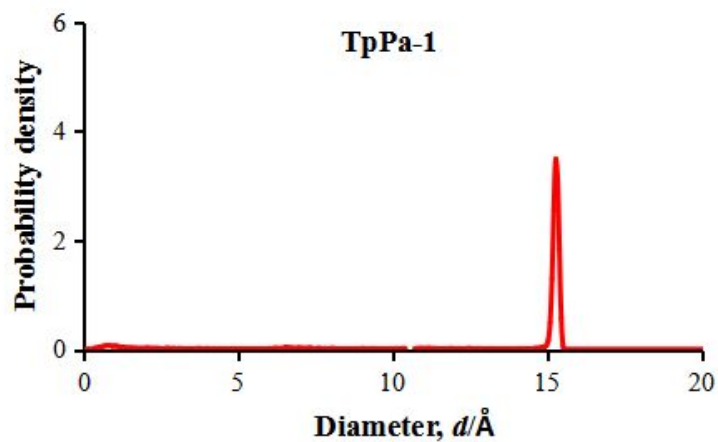


Figure S39. Pore Size distribution for TpPa-1.

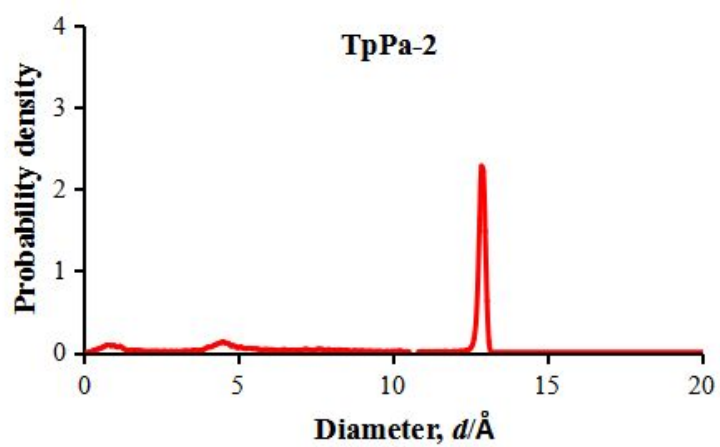


Figure S40. Pore Size distribution for TpPa-2.

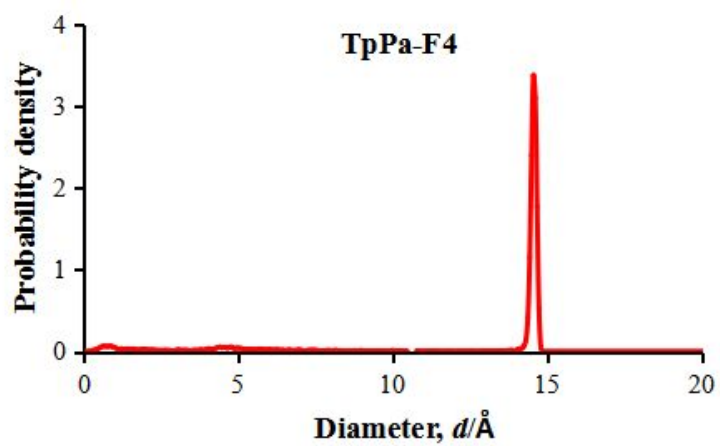


Figure S41. Pore Size distribution for TpPa-F4.

Section S-7 Radial distribution functions (RDF) of the oxygen-oxygen distance between oxygen atoms of the water molecules and the COF framework oxygen atoms

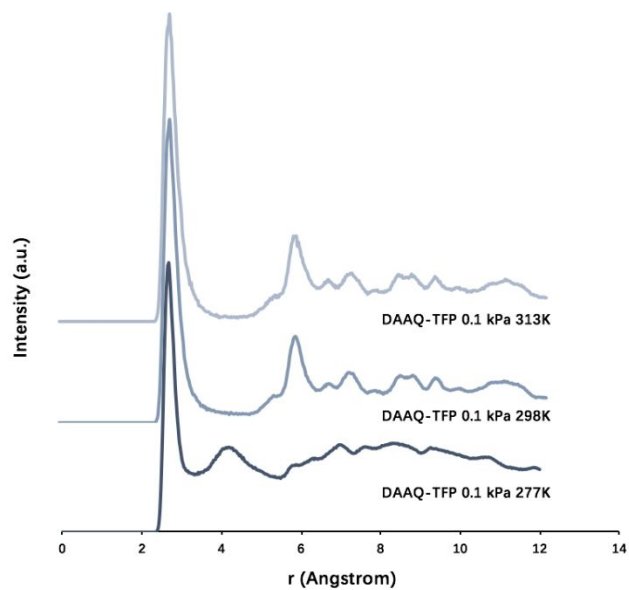


Figure S42. Radial distribution function (RDF) of the oxygen-oxygen distance between oxygen atom of the adsorbed water molecules and the COF framework oxygen atoms in DAAQ-TFP at different temperatures and 0.1 kPa.

Section S-8 Force field parameters used in the GCMC simulations

Table S5. Lennard-Jones parameters for the framework atoms of the COFs.

Atoms	C	O	H	N	F	Si
$\sigma(\text{\AA})$	3.47	3.03	2.85	3.26	3.09	3.80
ε/k_B	47.86	48.16	7.650	38.95	156.0	10.53

ε is the well depth, σ is the diameter. The LJ interaction parameters are from DREIDING force field.

Table S6. Force-Field parameters and geometries of six water models considered in this study

Model	Sites	σ (Å)	ι_1 (Å)	ι_2 (Å)	q_1 (e)	q_2 (e)	θ (°)	φ (°)
SPC	Three	3.1656	1.000	-	+0.4100	-0.8200	109.47	-
SPC/E	Three	3.1656	1.000	-	+0.4238	-0.8476	109.47	-
TIP4P	Four	3.1540	0.957	0.150	+0.5200	-1.0400	104.52	52.26
TIP4P_Ew	Four	3.1643	0.957	0.125	+0.5242	-1.0484	104.52	52.26
TIP5P	Five	3.1200	0.957	0.700	+0.2410	-0.2410	104.52	109.47
TIP5P_Ew	Five	3.0970	0.957	0.700	+0.2410	-0.2410	104.52	109.47

References

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