# Ruthenium(II) Catalysis/Noncovalent Interaction Synergy for Cross-Dehydrogenative Coupling of Arene Carboxylic Acids 

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## General information

$\left[\mathrm{Ru}(p \text {-cymene }) \mathrm{Cl}_{2}\right]_{2}$ and CuO were purchased from Alfa Aesar company. Benzoic acids and DBU were purchased from Avra chemicals and Spectrochem. All reactions were monitored by thin layer chromatography (TLC) on Merck 60 F 254 precoated silica plates and visualized using a UV lamp ( 366 or 254 nm ) or by use of potassium permanganate, $5 \mathrm{~g} \mathrm{~K}_{2} \mathrm{CO}_{3}$, / 100 mL water. Products were isolated by column chromatography (Merck silica gel $100-200 \mu \mathrm{~m}$ ). ${ }^{13} \mathrm{C}$ and ${ }^{1} \mathrm{H}$ NMR spectra were recorded on a Bruker 400 or Bruker 500 MHz spectrometers. Chemical shift values ( $\delta$ ) are reported in ppm and calibrated to the residual solvent peak $\mathrm{CDCl}_{3}$ $\delta=7.2600 \mathrm{ppm}$ for ${ }^{1} \mathrm{H}, \delta=77.16$ for ${ }^{13} \mathrm{C}$; or calibrated to tetramethylsilane ( $\delta=0.00$ ). All NMR spectra were recorded at ambient temperature ( 290 K ) unless otherwise noted. ${ }^{1} \mathrm{H}$ NMR spectra are reported as follows: chemical shift (multiplicity, coupling constant, integration). The following abbreviations are used to indicate multiplicities: $s$, singlet; $d$, doublet; $t$, triplet; q, quartet; quint, quintet; sext, sextet; sept, septet; m, multiplet; dd, doublet of doublet; dt, doublet of triplet; dq, doublet of quartet; td , triplet of doublet; tt , triplet of triplet; dq , doublet of quartet; br, broad. Mass spectra were recorded by electron spray ionization (ESI) method on a Q-TOF Micro with lock spray source.

Typical ruthenium(II)-catalyzed homo-dimerization of benzoic acids (isolated as ester):


Procedure: To an oven dried screw cap reaction tube ( $10 \times 1.5 \mathrm{~cm}$ ), corresponding benzoic acid 1 ( $0.3 \mathrm{mmol}, 1.0$ equiv), $\left[\mathrm{Ru}\left(p \text {-cymene) } \mathrm{Cl}_{2}\right]_{2}(2.5 \mathrm{~mol} \%), \mathrm{CuO}(1.0\right.$ equiv), and DBU (1.0 equiv) were taken. Then dry dioxane ( 0.4 mL ) was added in it and the mixture was stirred at $110^{\circ} \mathrm{C}$ for 24 h under air. After completion (monitored by TLC), $\mathrm{K}_{2} \mathrm{CO}_{3}$ ( 3.0 equiv) and iodomethane ( 3.0 equiv) were added to the reaction mixture and stirred for 4 h at room temperature. Then the solvent was evaporated under reduced pressure. In order to get pure 2,2'-biaryl acid methyl ester 2, the resulting residue was purified by column chromatography on silica gel with a gradient eluent of hexane and ethyl acetate.

Typical ruthenium(II)-catalyzed cross-dimerization of benzoic acids (isolated as ester):


Procedure: To an oven dried screw cap reaction tube ( $10 \times 1.5 \mathrm{~cm}$ ), corresponding benzoic acid ( $\mathbf{A r}^{1}$ ) ( $0.15 \mathrm{mmol}, 1.0$ equiv), benzoic acid ( $\mathrm{Ar}^{2}$ ) (4.0 equiv), $[\mathrm{Ru}(p$ cymene) $\left.\mathrm{Cl}_{2}\right]_{2}$ ( $10.0 \mathrm{~mol} \%$ ), CuO ( 1.5 equiv), and DBU ( 5.0 equiv) were taken. Then dry dioxane $(0.5 \mathrm{~mL})$ was added in it and the mixture was stirred at $110^{\circ} \mathrm{C}$ for 24 h under air. After completion (monitored by TLC), $\mathrm{K}_{2} \mathrm{CO}_{3}$ (4.0 equiv) and iodomethane (8.0 equiv) were added to the reaction mixture and stirred for 4 h at room temperature. Then the solvent was evaporated under reduced pressure. In order to get pure $2,2^{\prime}-$ biaryl acid methyl ester 3, the resulting residue was purified by column chromatography on silica gel with a gradient eluent of hexane and ethyl acetate.

Table for the $\mathbf{A r}^{1} / \mathbf{A r}^{2}$ acid ratio:

| Compound | $\mathbf{A r}^{1}$ (1.0 equiv) | $\mathbf{A r}^{2}$ (4.0 equiv) |
| :---: | :--- | :--- |
| $\mathbf{3 a}$ | 2-Methoxybenzoic acid | 2-Fluorobenzoic acid |
| $\mathbf{3 b}$ | 2,4-Dimethylbenzoic acid | 3-Acetylbenzoic acid |
| 3c | 4-Methoxybenzoic acid | 2-Fluorobenzoic acid |
| 3d | 1-Naphthoic acid | 2-Benzoylbenzoic acid |
| 3e | 2-Methoxybenzoic acid | 1-Naphthoic acid |
| $\mathbf{3 f}$ | 4-Methoxybenzoic acid | 1-Naphthoic acid |

## Typical synthesis of biphenyl-tethered lactones (4a-e):



Procedure: A solution of corresponding 2, $2^{\prime}$-biaryl acid methyl ester 2 ( $0.15 \mathrm{mmol}, 1$ equiv) in THF ( 1.0 mL ) was slowly added to a suspension of $\mathrm{LiAlH}_{4}$ ( 3.0 equiv) in THF ( 2.0 mL ). After stirring for 2 h at room temperature, the mixture was hydrolyzed by careful addition of water ( 25 mL ) and acidified with 2 NHCl . The organic solvent was evaporated, and the aqueous residue was diluted with water ( 5 mL ) and extracted with dichloromethane. The organic extract was washed with water and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. After evaporation of the solvent, $\left[\mathrm{RuCl}_{2}(p \text {-cymene })\right]_{2}$ (10.0 $\mathrm{mol} \%$ ) and $\mathrm{Cs}_{2} \mathrm{CO}_{3}(20.0 \mathrm{~mol} \%)$ were added to the crude product. Next, Toluene ( 1.5 mL ) was added in the reaction vessel and the reaction mixture was stirred at $100^{\circ} \mathrm{C}$ under oxygen atmosphere (using an $\mathrm{O}_{2}$ balloon) for 3 h . After cooling to room temperature, the solvent was evaporated under reduced pressure. In order to get pure product 4 , the resulting residue was purified by column chromatography on silica gel with a gradient eluent of hexane and ethyl acetate.

## Typical synthesis of biphenyl-tethered ether (6):



Procedure: A solution of 2,2'-biaryl acid methyl ester 2c ( $0.15 \mathrm{mmol}, 1$ equiv) in THF (1.0 mL ) was slowly added to a suspension of $\mathrm{LiAlH}_{4}$ ( 3.0 equiv) in THF ( 2.0 mL ). After stirring for 2 h at room temperature, the mixture was hydrolyzed by careful addition of water ( 25 mL ) and acidified with 2 N HCl . The organic solvent was evaporated, and the aqueous residue was diluted with water $(5 \mathrm{~mL})$ and extracted with dichloromethane. The combined organic extract was washed with water and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. After evaporation of the solvent conc. sulphuric acid ( 0.1 equiv) and 0.5 mL dioxane were added to the crude product and the reaction mixture was stirred at $60^{\circ} \mathrm{C}$ for 24 h . Then solvent was evaporated under reduced pressure and $\mathrm{DCM}(10 \mathrm{~mL})$ was added. The mixture was carefully washed with sodium bicarbonate solution. The solvent was evaporated under reduced pressure and resulting residue was purified by column chromatography on silica gel with a gradient eluent of hexane and ethyl acetate.

## Typical synthesis of $\mathbf{6 H}$-benzo[c]chromen-6-one (7):



Procedure: 2, 2'-Biaryl acid methyl ester $\mathbf{2 b}$ ( $0.15 \mathrm{mmol}, 1.0$ equiv) and KOH (3.0 equiv) were taken in a 25 mL round bottom flask and 5 mL MeOH :water (4:1) was added in it. The mixture was refluxed until the consumption of the ester (monitored by TLC). Volatiles were removed under vacuo and the resulting solution was neutralized with 2 N HCl . The acid was extracted using ethyl acetate, dried with $\mathrm{MgSO}_{4}$, and evaporated to get the acid $\mathbf{2} \mathbf{b}^{\prime}$. The crude acid $\mathbf{2} \mathbf{b}^{\prime}$ was taken in a screw cap reaction tube $(10 \times 1.5 \mathrm{~cm})$. Then $\mathrm{AgNO}_{3}$ ( $40 \mathrm{~mol} \%$ ), $\left(\mathrm{NH}_{4}\right)_{2} \mathrm{~S}_{2} \mathrm{O}_{8}$ (5.0 equiv), KOAc ( 5.0 equiv), dichloroethane (DCE, 1 mL ), and $\mathrm{H}_{2} \mathrm{O}(1 \mathrm{~mL})$ were sequentially added in it. The reaction mixture was stirred at room temperature for 36 h . After completion, it was diluted with $\mathrm{H}_{2} \mathrm{O}$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$
$(3 \times 20 \mathrm{~mL})$. The organic extract was dried over $\mathrm{MgSO}_{4}$ and concentrated in vacuo. Purification of the crude product by flash chromatography on silica gel using ethyl acetate/hexane system afforded the desired product 7 .

## Mechanistic studies:

(a) H/D-Scrambling study:


1c


Dioxane, $110^{\circ} \mathrm{C}, 14 \mathrm{~h}$ $\mathrm{D}_{2} \mathrm{O}$ (15.0 equiv)
ii) $\mathrm{K}_{2} \mathrm{CO}_{3}$, Mel, MeCN, rt


1c'

Procedure: To an oven dried screw cap reaction tube $(10 \times 1.5 \mathrm{~cm})$, benzoic acid 1c ( $0.3 \mathrm{mmol}, 1$ equiv), $\left[\mathrm{Ru}(p \text {-cymene }) \mathrm{Cl}_{2}\right]_{2}(2.5 \mathrm{~mol} \%), \mathrm{CuO}$ ( 1.0 equiv), and $\mathrm{DBU}(1.0$ equiv) were taken. Then dry dioxane ( 0.4 mL ) and $\mathrm{D}_{2} \mathrm{O}$ ( 15.0 equiv) were added and the mixture was stirred at $110{ }^{\circ} \mathrm{C}$ for 14 h under air. Then, a spoon of sodium sulfate was added in the reaction mixture. The reaction mixture was diluted with DCM and transferred to a 25 mL round-bottom flask. Next, the volatiles were evaporated under reduced pressure and $\mathrm{K}_{2} \mathrm{CO}_{3}$ (3.0 equiv) and iodomethane ( 3.0 equiv) were added to the reaction mixture followed by the addition of acetonitrile $(2.0 \mathrm{~mL})$. The resulting solution was stirred at room temperature for 4 h . Then the solvent was evaporated under reduced pressure. In order to get pure product $\mathbf{1} \mathbf{c}^{\prime}$, the resulting residue was purified by column chromatography on silica gel with a gradient eluent of hexane and ethyl acetate. The deuterium scrambling was observed through ${ }^{1} \mathrm{H}$ NMR spectroscopy.

(b) Radical scavengers study:

ii) $\mathrm{K}_{2} \mathrm{CO}_{3}$, Mel, MeCN, rt

Radical scavengers: TEMPO $=45 \% ; B H T=54 \% ; 1,1$-diphenylethene $=41 \%$
Procedure: To an oven dried screw cap reaction tube $(10 \times 1.5 \mathrm{~cm})$ benzoic acid 1a ( $0.3 \mathrm{mmol}, 1.0$ equiv), $\left[\mathrm{Ru}\left(p \text {-cymene) } \mathrm{Cl}_{2}\right]_{2}\right.$ ( $2.5 \mathrm{~mol} \%$ ), CuO ( 1.0 equiv), DBU ( 1.0 equiv), and radical scavenger ( 3.0 equiv) were taken. Then dry dioxane ( 0.4 mL ) were added into the mixture and it was stirred at $110^{\circ} \mathrm{C}$ for 24 h under air. After completion (monitored by TLC), $\mathrm{K}_{2} \mathrm{CO}_{3}$ ( 3.0 equiv) and iodomethane ( 3.0 equiv) were added to the reaction mixture and the resulting solution was stirred at room temperature for 4 h. Then the solvent was evaporated under reduced pressure. In order to get pure 2,2'biaryl acid methyl ester $\mathbf{2 a}$, the resulting residue was purified by column chromatography on silica gel with a gradient eluent of hexane and ethyl acetate.

## (c) Studying the role of oxygen:



Procedure: To an oven dried reaction tube $(10 \times 1.5 \mathrm{~cm})$ benzoic acid 1a $(0.3 \mathrm{mmol}$, 1 equiv), $\left[\mathrm{Ru}(p \text {-cymene }) \mathrm{Cl}_{2}\right]_{2}$ ( $2.5 \mathrm{~mol} \%$ ), CuO ( 1.0 equiv), and DBU ( 1.0 equiv) were taken. Then the reaction tube was evacuated under vacuum and back-filled with nitrogen gas (three-times). Then dry dioxane ( 0.4 mL ) were added into the mixture and it was stirred at $110^{\circ} \mathrm{C}$ for 24 h under air. After completion (monitored by TLC), $\mathrm{K}_{2} \mathrm{CO}_{3}$ (3.0 equiv) and iodomethane ( 3.0 equiv) were added to the reaction mixture and the resulting solution was stirred at room temperature for 4 h . Then the solvent was evaporated under reduced pressure. In order to get pure $2,2^{\prime}$-biaryl acid methyl ester 2a, the resulting residue was purified by column chromatography on silica gel with a gradient eluent of hexane and ethyl acetate.
(d) Use of strong oxidant:


Procedure: To an oven dried screw cap reaction tube ( $10 \times 1.5 \mathrm{~cm}$ ) corresponding benzoic acid 1 ( $0.3 \mathrm{mmol}, 1$ equiv), $\left[\mathrm{Ru}\left(p \text {-cymene) } \mathrm{Cl}_{2}\right]_{2}\right.$ ( $5.0 \mathrm{~mol} \%$ ), $\mathrm{AgNO}_{3}(1.5$ equiv), $\mathrm{Na}_{2} \mathrm{CO}_{3}$ ( 2.0 equiv), and $\mathrm{K}_{2} \mathrm{~S}_{2} \mathrm{O}_{8}$ ( 2.0 equiv) were taken. Then dry acetonitrile $(0.5 \mathrm{~mL})$ was added into the mixture and it was stirred at $100^{\circ} \mathrm{C}$ for 24 h under air. After completion (monitored by TLC), $\mathrm{K}_{2} \mathrm{CO}_{3}$ (3.0 equiv) and iodomethane (3.0 equiv) were added to the reaction mixture and the resulting solution was stirred at room temperature for 4 h . Then the solvent was evaporated under reduced pressure. In order to get pure product $\mathbf{8}$, the resulting residue was purified by column chromatography on silica gel with a gradient eluent of hexane and ethyl acetate.

## (e) Role of $\mathrm{DBUH}^{+}$:



Procedure: To an oven dried screw cap reaction tube ( $10 \times 1.5 \mathrm{~cm}$ ) sodium benzoate ( $9,0.3 \mathrm{mmol}, 1$ equiv), $\left[\mathrm{Ru}\left(p \text {-cymene) } \mathrm{Cl}_{2}\right]_{2}\right.$ ( $2.5 \mathrm{~mol} \%$ ), and CuO ( 1.0 equiv) were taken. Then dry dioxane $(0.4 \mathrm{~mL})$ was added into the mixture and it was stirred at 110 ${ }^{\circ} \mathrm{C}$ for 24 h under air. After completion (monitored by TLC), $\mathrm{K}_{2} \mathrm{CO}_{3}$ (3.0 equiv) and iodomethane ( 3.0 equiv) were added to the reaction mixture and the resulting solution was stirred at room temperature for 4 h . Same experiment was also performed in the presence of DBU (1.0 equiv). In both of the cases, the formation of $2,2^{\prime}$-biaryl acid methyl esters was not observed, indicating the decisive role of $\mathrm{DBUH}^{+}$in the transformation.

## (f) ESI-HRMS analysis of reaction mixture:

To an oven dried screw cap reaction tube $(10 \times 1.5 \mathrm{~cm})\left[\mathrm{Ru}(p \text {-cymene }) \mathrm{Cl}_{2}\right]_{2}(0.016$ mmol ), benzoic acid 1a ( 2.0 equiv.), and DBU ( 2.0 equiv.) were taken. Then dry dioxane $(0.3 \mathrm{~mL})$ was added in it and the mixture was stirred at $110^{\circ} \mathrm{C}$ under air. After 90 minutes, the crude reaction mixture was taken through a syringe and ESIHRMS was recorded.


Detected in ESI-HRMS
$[\mathrm{MH}]^{+}=659.2423$ (expected) 659.2435 (observed)





Dimethyl 4,4'-dimethyl-[1,1'-biphenyl]-2,2'-dicarboxylate (2a): yield = 32.3 $\mathrm{mg}(72 \%)$, colourless liquid; ${ }^{1} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.90(\mathrm{~d}, J=8.0 \mathrm{~Hz}$, $2 \mathrm{H}), 7.21(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.00(\mathrm{~s}, 2 \mathrm{H}), 3.61(\mathrm{~s}, 6 \mathrm{H}), 2.40(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathbf{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 167.46,143.77,142.09,131.02,130.04,127.91,126.62$, $51.78,21.64 \mathrm{ppm}$.
Dimethyl [1,1'-biphenyl]-2,2'-dicarboxylate (2b): yield $=23.4 \mathrm{mg}$ (58\%), colourless liquid; ${ }^{1} \mathbf{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.00(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.63$ $7.49(\mathrm{~m}, 2 \mathrm{H}), 7.47-7.38(\mathrm{~m}, 2 \mathrm{H}), 7.21(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.61(\mathrm{~s}, 6 \mathrm{H}) \mathrm{ppm} ;$ ${ }^{13}$ C NMR $\left(101 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 167.55,143.41,131.60,130.33,129.98,129.49$, 127.30, 51.94 ppm .

Dimethyl 3,3'-dimethyl-[1,1'-biphenyl]-2,2'-dicarboxylate (2c): yield $=27$ $\mathrm{mg}(60 \%)$, colourless liquid; ${ }^{1} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.37-7.25(\mathrm{~m}, 2 \mathrm{H})$, $7.25-7.17(\mathrm{~m}, 2 \mathrm{H}), 7.16-7.04(\mathrm{~m}, 2 \mathrm{H}), 3.55(\mathrm{~s}, 6 \mathrm{H}), 2.40(\mathrm{~s}, 6 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathbf{C}$ NMR (101 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 169.64,139.26,135.85,133.12,129.59,129.02$, 127.17, 51.80, 20.18 ppm .

Dimethyl 3,3'-dimethoxy-[1,1'-biphenyl]-2,2'-dicarboxylate (2d): yield = $33.6 \mathrm{mg}(68 \%)$, white solid; ${ }^{1} \mathbf{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.36-7.31(\mathrm{~m}, 2 \mathrm{H})$, $6.94-6.90(\mathrm{~m}, 2 \mathrm{H}), 6.90-6.87(\mathrm{~m}, 2 \mathrm{H}), 3.86(\mathrm{~s}, 6 \mathrm{H}), 3.59(\mathrm{~s}, 6 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathbf{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 167.85,156.60,139.32,130.20,123.32,121.89$, 110.48, 56.09, 52.12 ppm ; HRMS ( $\mathrm{m} / \mathrm{z}$ ) $\mathrm{C}_{18} \mathrm{H}_{18} \mathrm{O}_{6} \mathrm{~K}^{+} 369.0735$.

Dimethyl 3,3'-difluoro-[1,1'-biphenyl]-2,2'-dicarboxylate (2e): yield $=33 \mathrm{mg}$ ( $72 \%$ ), colourless liquid; ${ }^{1} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.48-7.37(\mathrm{~m}, 2 \mathrm{H})$, $7.19-7.13(\mathrm{~m}, 2 \mathrm{H}), 7.06(\mathrm{dd}, J=7.7,0.5 \mathrm{~Hz}, 2 \mathrm{H}), 3.66(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathbf{C}$ NMR (101 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 165.15,160.05(\mathrm{~d}, J=253.6 \mathrm{~Hz}), 140.67,131.52(\mathrm{~d}, J=9.3 \mathrm{~Hz})$, $125.68(\mathrm{~d}, J=3.4 \mathrm{~Hz}), 121.11(\mathrm{~d}, J=15.5 \mathrm{~Hz}), 115.84(\mathrm{~d}, J=21.9 \mathrm{~Hz}), 52.52$ ppm; HRMS (m/z) $\mathrm{C}_{16} \mathrm{H}_{12} \mathrm{~F}_{2} \mathrm{O}_{4} \mathrm{Na}^{+} 329.0591$.
Dimethyl 3,3'-dibromo-[1,1'-biphenyl]-2,2'-dicarboxylate (2f): yield $=41.7$ $\mathrm{mg}(65 \%)$, white solid; ${ }^{1} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.53(\mathrm{dd}, J=7.3,1.8 \mathrm{~Hz}$, $2 \mathrm{H}), 7.26-7.14(\mathrm{~m}, 4 \mathrm{H}), 3.57(\mathrm{~s}, 6 \mathrm{H}) \mathrm{ppm}$; ${ }^{13} \mathbf{C}$ NMR $\left(101 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $167.29,138.74,135.47,132.63,130.22,128.70,119.88,52.56$ ppm; HRMS $(\mathrm{m} / \mathrm{z}) \mathrm{C}_{16} \mathrm{H}_{12} \mathrm{Br}_{2} \mathrm{O}_{4} \mathrm{~K}^{+} 464.8726$.
Dimethyl 3,3'-dibenzoyl-[1,1'-biphenyl]-2,2'-dicarboxylate (2g): yield = 57.4 $\operatorname{mg}(80 \%)$, white solid; ${ }^{1} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.08-8.01(\mathrm{~m}, 2 \mathrm{H}), 7.79$ $-7.70(\mathrm{~m}, 3 \mathrm{H}), 7.69-7.60(\mathrm{~m}, 2 \mathrm{H}), 7.59-7.50(\mathrm{~m}, 4 \mathrm{H}), 7.47-7.36(\mathrm{~m}, 5 \mathrm{H})$, $3.60(\mathrm{~s}, 6 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathbf{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 197.15,166.47,141.74$, $137.22,133.17,132.51,130.16,129.72,129.31,129.27,128.60,127.85,52.26$ ppm.
Dimethyl [1, $1^{\prime}: 3^{\prime}, 1^{\prime \prime}: 3^{\prime \prime}, 1^{\prime \prime \prime}$-quaterphenyl]-2', $\mathbf{2}^{\prime \prime}$-dicarboxylate (2h): yield = $39.9 \mathrm{mg}(63 \%)$, white solid; ${ }^{1} \mathbf{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.43-7.35(\mathrm{~m}, 2 \mathrm{H})$, $7.35-7.20(\mathrm{~m}, 14 \mathrm{H}), 3.25(\mathrm{~s}, 6 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathbf{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 168.29$, $139.78,139.60,138.02,131.85,128.38,128.06,127.89,127.45,127.41,126.60$, $50.78 \mathrm{ppm} ;$ HRMS $(\mathrm{m} / \mathrm{z}) \mathrm{C}_{28} \mathrm{H}_{22} \mathrm{O}_{4} \mathrm{Na}^{+} 445.1411$.
Dimethyl-4,4'"-dichloro-[1,1':3',1':3',1'"-quaterphenyl]-2',2'-
dicarboxylate (2i): yield $=44.8 \mathrm{mg}(61 \%)$, white solid; ${ }^{1} \mathbf{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 7.47(\mathrm{t}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.39-7.30(\mathrm{~m}, 12 \mathrm{H}), 3.35(\mathrm{~s}, 6 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathbf{C}$ NMR (101 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 169.01,139.41,139.19,139.05,133.86,132.77$, $129.82,129.34,129.22,128.69,128.66,51.90 \mathrm{ppm}$.

Dimethyl 4,4'-diacetyl-[1,1'-biphenyl]-2,2'-dicarboxylate (2j): yield = 29.7 $\operatorname{mg}(56 \%)$, colourless liquid; ${ }^{1} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.62(\mathrm{~d}, J=1.4 \mathrm{~Hz}$, 2 H ), 8.15 (dd, $J=8.0,1.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.29(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 3.68(\mathrm{~s}, 6 \mathrm{H}), 2.68$ (s, 6H) ppm; ${ }^{13}$ C NMR (101 MHz, $\mathrm{CDCl}_{3}$ ) $\delta$ 196.92, 166.28, 147.35, 136.37, 131.27, 130.38, 130.27, 129.43, 52.34, 26.82 ppm ; HRMS (m/z) $\mathrm{C}_{20} \mathrm{H}_{18} \mathrm{O}_{6} \mathrm{Na}^{+}$ 377.0990 .


Dimethyl 5,5'-dimethoxy-[1,1'-biphenyl]-2,2'-dicarboxylate (2k): yield = $32.7 \mathrm{mg}(66 \%)$, colourless liquid; ${ }^{1} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.02(\mathrm{~d}, J=8.8$ $\mathrm{Hz}, 2 \mathrm{H}), 6.91$ (dd, $J=8.8,2.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.69(\mathrm{~d}, J=2.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.84(\mathrm{~s}, 6 \mathrm{H})$, $3.61(\mathrm{~s}, 6 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathbf{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 166.79,161.99,146.05$, 132.19, 121.57, 115.56, 112.38, 55.52, 51.70 ppm.

Dimethyl 5,5'-bis(benzyloxy)-[1,1'-biphenyl]-2,2'-dicarboxylate (21): yield = $46.2 \mathrm{mg}(64 \%)$, colourless liquid; ${ }^{1} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.03$ (d, $J=8.7$ $\mathrm{Hz}, 2 \mathrm{H}), 7.59-7.32(\mathrm{~m}, 10 \mathrm{H}), 7.00(\mathrm{dd}, J=8.8,2.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.80(\mathrm{~d}, J=2.6$ $\mathrm{Hz}, 2 \mathrm{H}), 5.10(\mathrm{~s}, 4 \mathrm{H}), 3.62(\mathrm{~s}, 6 \mathrm{H}) \mathrm{ppm}$; ${ }^{13}$ C NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 165.78, 160.16, 144.98, 135.38, 131.19, 127.74, 127.28, 126.71, 120.88, 115.36, 112.29, 69.17, 50.68 ppm.

Dimethyl 5,5'-bis(methoxymethoxy)-[1,1'-biphenyl]-2,2'-dicarboxylate (2m): yield $=22.2 \mathrm{mg}(38 \%)$, colourless liquid; ${ }^{1} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 7.99 (d, $J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.05$ (dd, $J=8.7,2.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.82$ (d, $J=2.5 \mathrm{~Hz}, 2 \mathrm{H})$, $5.24-5.18(\mathrm{~m}, 4 \mathrm{H}), 3.61(\mathrm{~s}, 6 \mathrm{H}), 3.49(\mathrm{~s}, 6 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathbf{C} \mathbf{N M R}\left(101 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta 166.82,159.69,145.67,132.08,122.75,117.78,114.29,94.39,56.44,51.72$ ppm; HRMS (m/z) $\mathrm{C}_{20} \mathrm{H}_{22} \mathrm{O}_{8} \mathrm{Na}^{+}$413.1201.
Dimethyl 5,5'-dichloro-[1,1'-biphenyl]-2,2'-dicarboxylate (2n): yield $=25.3$ $\mathrm{mg}(50 \%)$, colourless liquid; ${ }^{1} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.98(\mathrm{~d}, J=8.4 \mathrm{~Hz}$, $2 \mathrm{H}), 7.42(\mathrm{dd}, J=8.4,2.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.20(\mathrm{~d}, J=2.1 \mathrm{~Hz}, 2 \mathrm{H}), 3.65(\mathrm{~s}, 6 \mathrm{H}) \mathrm{ppm} ;$ ${ }^{13}$ C NMR (101 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 166.24,143.91,138.07,131.55,130.14,127.92$, 127.67, 52.19 ppm .

Dimethyl 5,5'-dibromo-[1,1'-biphenyl]-2,2'-dicarboxylate (20): yield $=37.8$ $\mathrm{mg}(59 \%)$, white solid; ${ }^{1} \mathbf{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.90(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H})$, $7.58(\mathrm{dd}, J=8.4,2.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.36(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 2 \mathrm{H}), 3.65(\mathrm{~s}, 6 \mathrm{H}) \mathrm{ppm}$; ${ }^{13} \mathbf{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 166.37,143.85,133.02,131.59,130.94,128.16$, 126.59, 52.21 ppm .

Dimethyl 3,3',5,5'-tetramethyl-[1,1'-biphenyl]-2,2'-dicarboxylate (2p): yield $=25.9 \mathrm{mg}(53 \%)$, colourless liquid; ${ }^{1} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.00(\mathrm{~s}, 2 \mathrm{H})$, $6.88(\mathrm{~s}, 2 \mathrm{H}), 3.54(\mathrm{~s}, 6 \mathrm{H}), 2.37(\mathrm{~s}, 6 \mathrm{H}), 2.31(\mathrm{~s}, 6 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathbf{C}$ NMR ( 101 MHz , $\mathrm{CDCl}_{3}$ ) $\delta 169.80,139.99,139.11,136.00,130.32,129.97,127.95,51.66,21.29$, 20.27 ppm ; HRMS (m/z) $\mathrm{C}_{20} \mathrm{H}_{22} \mathrm{O}_{4} \mathrm{Na}^{+} 349.1409$.

Dimethyl 3,3',4,4'-tetramethyl-[1,1'-biphenyl]-2,2'-dicarboxylate (2q): yield $=23.0 \mathrm{mg}(47 \%)$, colourless liquid; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.15(\mathrm{~d}, J=$ $7.8 \mathrm{~Hz}, 2 \mathrm{H}), 6.99(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 3.57(\mathrm{~s}, 6 \mathrm{H}), 2.30(\mathrm{~s}, 6 \mathrm{H}), 2.25(\mathrm{~s}, 6 \mathrm{H})$ ppm; ${ }^{13} \mathrm{C}$ NMR $\left(101 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 170.39,136.39,136.23,134.08,133.53$, 130.26, 127.04, 51.84, 20.22, 17.01 ppm .

Dimethyl [2,2'-binaphthalene]-1,1'-dicarboxylate (2r): yield $=46.6 \mathrm{mg}$ ( $84 \%$ ), colourless liquid; ${ }^{1} \mathbf{H}$ NMR ( $\left.400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.10(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H})$, $8.02-7.86(\mathrm{~m}, 4 \mathrm{H}), 7.65-7.53(\mathrm{~m}, 4 \mathrm{H}), 7.49(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.61(\mathrm{~s}, 6 \mathrm{H})$ ppm; ${ }^{13} \mathbf{C}$ NMR $\left(101 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 169.14,137.75,132.63,130.10,129.77$, $128.32,127.64,127.39,126.65,125.57,52.19 \mathrm{ppm} ;$ HRMS (m/z) $\mathrm{C}_{24} \mathrm{H}_{18} \mathrm{O}_{4} \mathrm{Na}^{+}$ 393.1109.

Dimethyl 3-fluoro-3'-methoxy-[1,1'-biphenyl]-2,2'-dicarboxylate (3a): yield $=25.3 \mathrm{mg}(53 \%)$, colourless liquid; ${ }^{1} \mathbf{H} \mathbf{~ N M R ~ ( ~} 400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.46-7.30$ $(\mathrm{m}, 2 \mathrm{H}), 7.19-7.04(\mathrm{~m}, 2 \mathrm{H}), 6.96(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.85(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H})$, $3.88(\mathrm{~s}, 3 \mathrm{H}), 3.66(\mathrm{~s}, 3 \mathrm{H}), 3.59(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathbf{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ $167.60,165.33,159.82(\mathrm{~d}, J=250.0 \mathrm{~Hz}), 156.63,141.05,138.92,134.24,131.24$ $(\mathrm{d}, J=9.1 \mathrm{~Hz}), 130.38,125.77(\mathrm{~d}, J=3.4 \mathrm{~Hz}), 122.90,121.81,115.50(\mathrm{~d}, J=$ 21.8 Hz ), 110.73, 56.14, 52.41, 52.15 ppm ; HRMS (m/z) $\mathrm{C}_{17} \mathrm{H}_{15} \mathrm{FO}_{5} \mathrm{Na}^{+}$ 341.0790 .

Dimethyl 4'-acetyl-3,5-dimethyl-[1,1'-biphenyl]-2,2'-dicarboxylate (3b): yield $=31.1 \mathrm{mg}(61 \%)$, colourless liquid; ${ }^{\mathbf{1}} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.49$ (d, $J=1.9 \mathrm{~Hz}, 1 \mathrm{H}), 8.08(\mathrm{dd}, J=8.0,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.35(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.07$ $(\mathrm{s}, 1 \mathrm{H}), 6.86(\mathrm{~s}, 1 \mathrm{H}), 3.70(\mathrm{~s}, 3 \mathrm{H}), 3.48(\mathrm{~s}, 3 \mathrm{H}), 2.67(\mathrm{~s}, 3 \mathrm{H}), 2.39(\mathrm{~s}, 3 \mathrm{H}), 2.36$ (s, 3H) ppm; ${ }^{13}$ C NMR (101 MHz, $\mathrm{CDCl}_{3}$ ) $\delta$ 197.12, 169.32, 167.00, 147.11, 52.34, 51.70, 26.82, 21.39, 20.38 ppm ; HRMS (m/z) C $\mathrm{C}_{20} \mathrm{H}_{20} \mathrm{O}_{5} \mathrm{Na}^{+}$363.1196.

Dimethyl 3-fluoro-5'-methoxy-[1,1'-biphenyl]-2,2'-dicarboxylate (3c): yield $=23.9 \mathrm{mg}(50 \%)$, colourless liquid; ${ }^{1} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.98(\mathrm{~d}, J=$ $8.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.43$ (dd, $J=5.0,2.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.22-6.89(\mathrm{~m}, 3 \mathrm{H}), 6.73(\mathrm{~s}, 1 \mathrm{H})$, $3.84(\mathrm{~s}, 3 \mathrm{H}), 3.76-3.55(\mathrm{~m}, 6 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 166.87$, $165.46,162.00,159.82(\mathrm{~d}, J=252.3 \mathrm{~Hz}), 143.40,142.88,132.61,131.19$ (d, $J$ $=9.2 \mathrm{~Hz}), 125.18,122.36,120.79(\mathrm{~d}, J=15.3 \mathrm{~Hz}), 116.15,114.97(\mathrm{~d}, J=22.0$ $\mathrm{Hz}), 113.38,55.64,52.33,51.88 \mathrm{ppm}$; HRMS (m/z) $\mathrm{C}_{17} \mathrm{H}_{15} \mathrm{FO}_{5} \mathrm{Na}^{+} 341.0788$.
Methyl 2-(3-benzoyl-2-(methoxycarbonyl)phenyl)-1-naphthoate (3d): yield $=49.0 \mathrm{mg}(77 \%)$, colourless liquid; ${ }^{1} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.06-7.99$ $(\mathrm{m}, 1 \mathrm{H}), 7.97-7.92(\mathrm{~m}, 1 \mathrm{H}), 7.92-7.88(\mathrm{~m}, 1 \mathrm{H}), 7.86-7.79(\mathrm{~m}, 2 \mathrm{H}), 7.62-$ $7.41(\mathrm{~m}, 9 \mathrm{H}), 3.66(\mathrm{~s}, 3 \mathrm{H}), 3.25(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathbf{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ $196.51,169.00,167.34,141.24,139.86,137.10,137.06,133.32,132.75,132.65$, $131.87,130.01,129.86,129.80,129.77,128.64,128.38,128.33,127.68,127.29$, 126.71, 125.56, 52.16, $52.09 \mathrm{ppm} ;$ HRMS (m/z) $\mathrm{C}_{27} \mathrm{H}_{20} \mathrm{O}_{5} \mathrm{Na}^{+} 447.1194$.

Methyl 2-(3-methoxy-2-(methoxycarbonyl)phenyl)-1-naphthoate (3e): yield $=29.4 \mathrm{mg}(56 \%)$, colourless liquid; ${ }^{1} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.06-7.97$ (m, 1H), $7.92-7.84(\mathrm{~m}, 2 \mathrm{H}), 7.61-7.50(\mathrm{~m}, 2 \mathrm{H}), 7.44-7.36(\mathrm{~m}, 2 \mathrm{H}), 7.00-$ $6.90(\mathrm{~m}, 2 \mathrm{H}), 3.90(\mathrm{~s}, 3 \mathrm{H}), 3.68(\mathrm{~s}, 3 \mathrm{H}), 3.52(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathbf{C}$ NMR ( 101 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 169.28,167.81,156.78,140.48,136.68,132.68,130.36,130.27$, $130.09,129.61,128.30,127.56,127.20,126.62,125.59,123.35,122.19,110.59$, 56.20, $52.21,52.15 \mathrm{ppm}$; HRMS (m/z) $\mathrm{C}_{21} \mathrm{H}_{18} \mathrm{O}_{5} \mathrm{Na}^{+}$373.1043.

Methyl 2-(5-methoxy-2-(methoxycarbonyl)phenyl)-1-naphthoate (3f): yield $=30.5 \mathrm{mg}(58 \%)$, colourless liquid; ${ }^{1} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.10-7.86$ $(\mathrm{m}, 4 \mathrm{H}), 7.60-7.47(\mathrm{~m}, 2 \mathrm{H}), 7.37(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.95(\mathrm{dd}, J=8.8,2.6 \mathrm{~Hz}$, $1 \mathrm{H}), 6.80(\mathrm{~d}, J=2.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.83(\mathrm{~s}, 3 \mathrm{H}), 3.61(\mathrm{~s}, 3 \mathrm{H}), 3.57(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathbf{C}$ NMR (101 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 169.37,167.02,161.92,144.38,138.92,132.69$, $132.49,129.89,129.29,129.03,128.34,127.42,127.24,126.36,125.42,122.30$, 116.03, 113.50, 55.65, 52.09, 51.83 ppm ; HRMS ( $\mathrm{m} / \mathrm{z}$ ) $\mathrm{C}_{21} \mathrm{H}_{18} \mathrm{O}_{5} \mathrm{Na}^{+} 373.1041$.

2,10-dimethyldibenzo[c,e]oxepin-5(7H)-one (4a): yield $=22.1 \mathrm{mg}$ (62\%), colourless liquid; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.88(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.45$ (s, 1H), 7.41 (s, 1H), $7.35-7.30(\mathrm{~m}, 2 \mathrm{H}), 7.25-7.20(\mathrm{~m}, 1 \mathrm{H}), 5.05-4.87(\mathrm{~m}$, 2H), 2.49 ( $\mathrm{s}, 3 \mathrm{H}$ ), 2.45 ( $\mathrm{s}, 3 \mathrm{H}$ ) ppm; ${ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 170.70$, $143.21,140.12,139.12,137.54,132.29,132.23,129.33$ (4C), 128.59, 128.14, 69.07, 21.75, 21.58 ppm.

Dibenzo[c,e]oxepin-5(7H)-one (4b): yield $=21.4 \mathrm{mg}(68 \%)$, colourless liquid; ${ }^{1} \mathbf{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.04-7.94(\mathrm{~m}, 1 \mathrm{H}), 7.72-7.59(\mathrm{~m}, 3 \mathrm{H}), 7.58-$ $7.49(\mathrm{~m}, 2 \mathrm{H}), 7.49-7.39(\mathrm{~m}, 2 \mathrm{H}), 5.03(\mathrm{~d}, J=23.5 \mathrm{~Hz}, 2 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathbf{C}$ NMR $\left(101 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 170.42,139.13,137.42,134.97,132.73,132.10,130.81$, $130.29,128.83$ (2C), 128.72 (2C), 128.59, 69.34 ppm .
4,8-dimethyldibenzo[c,e]oxepin-5(7H)-one (4c): yield $=25.7 \mathrm{mg}$ ( $72 \%$ ), colourless liquid; ${ }^{1} \mathbf{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.88(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.49$ $7.38(\mathrm{~m}, 2 \mathrm{H}), 7.37-7.28(\mathrm{~m}, 2 \mathrm{H}), 7.24-7.19(\mathrm{~m}, 1 \mathrm{H}), 5.10-4.81(\mathrm{~m}, 2 \mathrm{H}), 2.49$ $(\mathrm{s}, 3 \mathrm{H}), 2.46(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathbf{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 170.68,143.21$, $140.14,139.17,137.57,132.33,132.26,129.35$ (2C), 129.31 (2C), 128.61, 128.20, 69.09, 21.76, 21.59 ppm ; HRMS (m/z) C $\mathrm{C}_{16} \mathrm{H}_{14} \mathrm{O}_{2} \mathrm{Na}^{+}$261.0894.

4,8-dimethoxydibenzo[c,e]oxepin-5(7H)-one (4d): yield $=23.1 \mathrm{mg}(57 \%)$, colourless liquid; ${ }^{1} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.52(\mathrm{t}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.43$ (t, $J=8.1 \mathrm{~Hz}, 1 \mathrm{H}$ ), $7.22(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.14(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.05(\mathrm{~d}, J$ $=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.95(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 5.62(\mathrm{~d}, J=12.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.68(\mathrm{~d}, J=$ $12.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.94(\mathrm{~s}, 3 \mathrm{H}), 3.89(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathbf{C}$ NMR ( $\left.101 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $167.23,158.42,156.67,140.75,138.71,132.12,130.65,124.33,120.72$ (2C), 111.49, 110.82, 60.89, 56.51, 56.06 ppm ; HRMS (m/z) C C ${ }_{16} \mathrm{H}_{14} \mathrm{O}_{4} \mathrm{Na}^{+}$293.0790.








2,10-dimethoxydibenzo[c,e]oxepin-5(7H)-one (4e): yield $=28.3 \mathrm{mg}(70 \%)$, colourless liquid; ${ }^{1} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.96(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.37$ (d, $J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.18-7.01(\mathrm{~m}, 3 \mathrm{H}), 6.95(\mathrm{dd}, J=8.5,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.16-$ $4.80(\mathrm{~m}, 2 \mathrm{H}), 3.91(\mathrm{~d}, J=17.6 \mathrm{~Hz}, 6 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathbf{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ $170.39,162.79,160.96,140.55,139.49,134.57,130.08,127.74,123.49,114.51$, $114.23,113.84,113.80,68.76,55.76,55.70 \mathrm{ppm} ;$ HRMS (m/z) $\mathrm{C}_{16} \mathrm{H}_{14} \mathrm{O}_{4} \mathrm{~K}^{+}$ 309.0525 .

4,8-dimethyl-5,7-dihydrodibenzo[c,e]oxepine (6): yield $=21.5 \mathrm{mg}$ (64\%), colourless liquid; ${ }^{1} \mathbf{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.41-7.34(\mathrm{~m}, 2 \mathrm{H}), 7.34-$ $7.28(\mathrm{~m}, 2 \mathrm{H}), 7.24-7.19(\mathrm{~m}, 2 \mathrm{H}), 4.32(\mathrm{~s}, 4 \mathrm{H}), 2.46(\mathrm{~s}, 6 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathbf{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 141.36,138.77,132.60,129.76,128.97,128.25,67.33$, 21.52 ppm ; HRMS (m/z) $\mathrm{C}_{16} \mathrm{H}_{17} \mathrm{O}^{+}$225.1291.
$\mathbf{6 H}$-benzo[c]chromen-6-one (7): yield $=26.5 \mathrm{mg}(90 \%)$, white solid; ${ }^{1} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.39(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.11(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.05(\mathrm{~d}$, $J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.82(\mathrm{t}, J=7.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.58(\mathrm{t}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.47(\mathrm{t}, J=7.2$ $\mathrm{Hz}, 1 \mathrm{H}), 7.41-7.29(\mathrm{~m}, 2 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathbf{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 161.28$, $151.43,134.95,134.89,130.69,130.56,128.99,124.67,122.88,121.80,121.40$, $118.17,117.90 \mathrm{ppm}$.
Methyl 4-methyl-2-((4-methylbenzoyl)oxy)benzoate (8a): yield $=17.4 \mathrm{mg}$ ( $41 \%$ ), colourless liquid; ${ }^{1} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.10(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H})$, $7.96(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.31(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.15(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.04$ $(\mathrm{s}, 1 \mathrm{H}), 3.71(\mathrm{~s}, 3 \mathrm{H}), 2.45(\mathrm{~s}, 3 \mathrm{H}), 2.42(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathbf{C} \mathbf{N M R}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta 165.71,165.22,150.98,145.24,144.42,131.98,130.41,129.42,126.98$, 124.65, 120.64, 52.15, 21.90, 21.59 ppm ; HRMS (m/z) C $\mathrm{C}_{17} \mathrm{H}_{16} \mathrm{O}_{4} \mathrm{Na}^{+} 307.0931$.

Methyl 2-methyl-6-((2-methylbenzoyl)oxy)benzoate (8b): yield $=14.5 \mathrm{mg}$ (34\%), colourless liquid; ${ }^{1} \mathbf{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.12$ (dd, $J=8.1,1.2$ $\mathrm{Hz}, 1 \mathrm{H}), 7.48(\mathrm{td}, J=7.6,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.39(\mathrm{t}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.32(\mathrm{t}, J=7.2$ $\mathrm{Hz}, 2 \mathrm{H}), 7.15(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.09(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.75(\mathrm{~s}, 3 \mathrm{H}), 2.66(\mathrm{~s}$, $3 \mathrm{H}), 2.44(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm}$; ${ }^{13} \mathbf{C}$ NMR ( $125 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 167.19,165.55,148.78$, $141.66,138.48,132.95,132.09,131.34,130.90,128.25,126.65,126.08,120.79$, 52.29, 21.95, 20.18 ppm.

Methyl 2-(benzoyloxy)benzoate (8c): yield $=12.7 \mathrm{mg}(33 \%)$, colourless liquid; ${ }^{1} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.23(\mathrm{dd}, J=8.0,1.5 \mathrm{~Hz}, 2 \mathrm{H}), 8.08(\mathrm{dd}, J=7.9$, $1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.69-7.58(\mathrm{~m}, 2 \mathrm{H}), 7.53(\mathrm{t}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.42-7.33(\mathrm{~m}, 1 \mathrm{H})$, $7.24(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.74(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathbf{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $165.44,165.09,150.80,133.89,133.59,131.96,130.29,129.51,128.61,126.12$, $124.00,123.49,52.23 \mathrm{ppm}$.
methyl 4-(tert-butyl)-2-((4-(tert-butyl)benzoyl)oxy)benzoate (8d): yield = $21.0 \mathrm{mg}(38 \%)$, colourless liquid; ${ }^{\mathbf{1}} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.15(\mathrm{~d}, J=$ $8.3 \mathrm{~Hz}, 2 \mathrm{H}), 8.00(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.53(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.36$ (dd, $J=8.3$, $1.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.19(\mathrm{~d}, J=1.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.73(\mathrm{~s}, 3 \mathrm{H}), 1.37(\mathrm{~s}, 9 \mathrm{H}), 1.34(\mathrm{~s}, 9 \mathrm{H})$ $\mathrm{ppm} ;{ }^{13} \mathbf{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 165.67,165.21,158.38,157.32,150.95$, $131.74,130.26,126.98,125.72,123.28,121.08,120.55,52.19,35.33,35.29$, 31.28, 31.11 ppm ; HRMS (m/z) $\mathrm{C}_{23} \mathrm{H}_{28} \mathrm{O}_{4} \mathrm{Na}^{+}$391.1875.

## Computational details:

Density functional theory calculations were performed employing the functional M06. ${ }^{1}$ In all the calculations, Pople's basis set $6-31+G(d, p)$ was used for oxygen and nitrogen, while for carbon, hydrogen and ruthenium Ahlrichs basis set def2-SVP ${ }^{2,3}$ with effective core potential ${ }^{4}$ employed for ruthenium as the parameters provided in the EMSL website. ${ }^{5}$ During the optimization of geometries, Grimme's D3 dispersion correction ${ }^{6}$ was also employed. The absence of imaginary frequencies while performing analytical frequency calculations on the optimized geometries confirmed the obtained geometries are local minima (stationary points) for the intermediates. For transition states, the corresponding analytical frequency calculations on the optimized geometries result in only one imaginary frequency which show that the stationary points are saddle points. Further, the imaginary frequencies of the transition states show that the vibrations correspond to C-H-O stretching which is the reaction coordinate for the desired transition states. Unless otherwise stated, the reported energies in the manuscript are corrected enthalpies in $\mathrm{kcal} /$ mole obtained by adding electronic energy and thermal correction to enthalpy. All the calculations were performed in gas at 298.15 K using Linux version of Gaussian 16.

## References:

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5. https://bse.pnl.gov/bse/portal
6. Grimme, S.; Antony, J.; Ehrlich, S.; Krieg, H. A Consistent and Accurate ab initio Parametrization of Density Functional Dispersion Correction (DFT-D) for the 94 Elements H-Pu. J. Chem. Phys. 2010, 132, 154104.

## Cartesian coordinates



Electronic Energy =-1323.39623772 Hartree
Sum of Electronic and Thermal Enthalpies $=-1322.94593$ Hartree
Dispersion Correction $=-0.013197983$ Hartree

01
$\begin{array}{llll}\mathrm{Ru} & -0.93613940 & 0.57367276 & -0.11741750\end{array}$
$\begin{array}{lllll}\mathrm{O} & -1.02802023 & -0.77510242 & -1.71528969\end{array}$
$\begin{array}{llll}\text { C } & -1.87455694 & -1.03179991 & 0.75828092\end{array}$
$\begin{array}{llll}\mathrm{O} & 0.95653387 & -0.42082964 & 0.34380458\end{array}$
$\begin{array}{llll}\text { C } & 0.08482549 & 2.55309012 & 0.15887790\end{array}$
$\begin{array}{llll}\text { C } & -0.80293222 & 2.26714085 & 1.25005785\end{array}$
$\begin{array}{llll}\text { C } & -2.15584844 & 1.93395360 & 1.00983879\end{array}$
$\begin{array}{llll}\text { C } & -2.66908489 & 1.81858317 & -0.31905415\end{array}$
$\begin{array}{llll}\text { C } & -1.75398984 & 2.08942267 & -1.38174391\end{array}$
$\begin{array}{llll}\text { C } & -0.42008444 & 2.53331722 & -1.15860647\end{array}$
C
$\begin{array}{lll}1.53672880 & 2.81867635 & 0.47721138\end{array}$
$\begin{array}{lllll}\mathrm{C} & 1.67886770 & 4.22710339 & 1.04999983\end{array}$
$\begin{array}{lllll}\text { C } & 2.48427162 & 2.61528815 & -0.69259501\end{array}$
$\begin{array}{llll}\text { C } & -4.06074749 & 1.34902874 & -0.59183835\end{array}$
$\begin{array}{lllll}\mathrm{O} & -1.75827858 & -2.77976554 & -2.38981493\end{array}$
$\begin{array}{llll}\text { C } & -1.65843515 & -1.93367980 & -1.51978323\end{array}$
$\begin{array}{lllll}\text { C } & -2.16086337 & -2.08019154 & -0.12989286\end{array}$
$\begin{array}{llll}\mathrm{O} & 1.50288486 & -1.07000177 & -1.74146839\end{array}$
$\begin{array}{llll}\text { C } & 1.77075470 & -0.86354968 & -0.48328296\end{array}$

| C | 3.16261610 | -1.15284078 | -0.07432421 |
| :---: | :---: | :---: | :---: |
| C | 3.54930610 | -0.91236892 | 1.24923267 |
| C | -2.82901089 | -3.23273487 | 0.28694587 |
| C | -3.21932985 | -3.35703270 | 1.61594289 |
| C | -2.92242705 | -2.33303550 | 2.51803322 |
| C | -2.25705521 | -1.18007110 | 2.09581202 |
| C | 4.85919471 | -1.15523414 | 1.64385682 |
| C | 5.78641875 | -1.63805827 | 0.71991431 |
| C | 5.40362569 | -1.88085419 | -0.59844215 |
| C | 4.09336132 | -1.64044237 | -0.99827462 |
| H | 1.80726183 | 2.09173187 | 1.26894899 |
| H | 0.49864204 | $-0.95613484$ | -1.91036781 |
| H | -0.41022551 | 2.25143782 | 2.27223133 |
| H | -2.79741227 | 1.64744656 | 1.85003117 |
| H | -2.07922639 | 1.88809276 | -2.40846899 |
| H | 0.23943957 | 2.69924033 | -2.01307537 |
| H | 2.72136426 | 4.42633973 | 1.34542020 |
| H | 1.39250857 | 4.98036995 | 0.29570217 |
| H | 1.04017633 | 4.38164754 | 1.93417920 |
| H | 2.34190461 | 3.38801109 | -1.46779147 |
| H | 3.52936396 | 2.68862504 | -0.35243066 |
| H | 2.36035990 | 1.63141415 | -1.17394020 |
| H | -4.08931368 | 0.74092336 | -1.51028975 |
| H | -4.43383083 | 0.72423927 | 0.23429897 |
| H | -4.75488392 | 2.19590862 | -0.72307718 |
| H | -3.02286591 | -4.01969057 | -0.44933626 |
| H | -3.74546097 | -4.25343757 | 1.95684591 |
| H | -3.21269568 | $-2.43467883$ | 3.56928835 |
| H | -2.02548908 | -0.40088324 | 2.83306061 |

$\begin{array}{llll}\mathrm{H} & 5.16139164 & -0.97026474 & 2.67826088\end{array}$
$\begin{array}{lllll}\mathrm{H} & 6.81739965 & -1.82913326 & 1.03180819\end{array}$
$\begin{array}{llll}\mathrm{H} & 6.13150259 & -2.26247336 & -1.31947341\end{array}$
$\begin{array}{llll}\mathrm{H} & 3.77730768 & -1.82822189 & -2.02717621\end{array}$
H
$2.80432495-0.53877942 \quad 1.95686924$


Electronic Energy =-1784.94457889 Hartree
Sum of Electronic and Thermal Enthalpies $=-1784.236405$ Hartree
Dispersion Correction $=-0.026768825$ Hartree

01
$\mathrm{Ru} \quad-2.07149675-0.27268652-0.32147530$
$\begin{array}{lllll}\mathrm{O} & -0.74284800 & -1.51945671 & -1.35821784\end{array}$
$\begin{array}{llll}\text { C } & -1.40960952 & -1.49335187 & 1.18087794\end{array}$
$\begin{array}{llll}\text { O } & -0.46952992 & 1.09022862 & -0.02904713\end{array}$
$\begin{array}{llll}\text { C } & -3.38952332 & 1.44087745 & -0.08416184\end{array}$
$\begin{array}{lllll}C & -3.89557054 & 0.34303650 & 0.65542309\end{array}$
$\begin{array}{llll}\text { C } & -4.07461580 & -0.92497100 & 0.02069954\end{array}$
$\begin{array}{llll}\text { C } & -3.76548195 & -1.10356236 & -1.35377588\end{array}$
$\begin{array}{llll}\text { C } & -3.38464452 & 0.05622531 & -2.12033468\end{array}$
$\begin{array}{lllll}\text { C } & -3.22227628 & 1.29903592 & -1.50772221\end{array}$
C
$-3.05188095 \quad 2.76536188 \quad 0.55638432$
$\begin{array}{llll}\text { C } & -2.60173449 & 2.65644861 & 2.00334702\end{array}$
$\begin{array}{llll}\text { C } & -4.23438330 & 3.71808721 & 0.40894473\end{array}$
$\begin{array}{lllll}\text { C } & -3.80563450 & -2.45474152 & -1.98946135\end{array}$

O

C
C
0.58940082 -3.29077461 -1.21403540
-0.23400132 $-2.50805203-0.69724088$
$-0.62512780-2.568085630 .72775389$
1.253026890 .065168761 .02695025
$0.66030624 \quad 1.04901080 \quad 0.57137150$
$\begin{array}{lll}1.33624931 & 2.40207656 & 0.64401911\end{array}$
$\begin{array}{llll}1.03094361 & 3.40790768 & -0.27921690\end{array}$
-0.14306966 $-3.54614008 \quad 1.59742233$
-0.42748354 $-3.45918733 \quad 2.95664738$
$\begin{array}{llll}-1.18503861 & -2.38447962 & 3.42780242\end{array}$
$-1.67657607 \quad-1.41457358 \quad 2.55190957$
$\begin{array}{lll}1.74968911 & 4.60144501 & -0.28557622\end{array}$
$2.760583534 .81309021 \quad 0.65150022$
$3.04471823 \quad 3.82950821 \quad 1.59950988$
$2.33984146 \quad 2.62886671 \quad 1.59038380$
$-2.19775980 \quad 3.16724497-0.02170761$
-4.04782834 $0.42974948 \quad 1.73481776$
-4.34155888 -1.79825866 0.62571080
-3.10985781 $-0.07383695-3.17248936$
-2.811118392 .13948811 -2.07783526
$-2.24656523 \quad 3.63501034 \quad 2.36433885$
-3.42694666 $2.34891921 \quad 2.66990399$
$-1.776553381 .934437542 .11322082$
$-5.10941657 \quad 3.33782640 \quad 0.96511879$
-3.99097781 4.716741670 .80686753
$-4.538016493 .83654470-0.64437162$
-2.98968410 $-2.56135485-2.72087949$
-3.68018990 -3.24543166 -1.23328566
$-4.76177785 \quad-2.62794602 \quad-2.51201134$
$\begin{array}{llll}\mathrm{H} & 0.46332339 & -4.36123989 & 1.18738332\end{array}$
$\begin{array}{llll}\mathrm{H} & -0.05738821 & -4.21887898 & 3.65171184\end{array}$
$\begin{array}{llll}\mathrm{H} & -1.39825279 & -2.30062277 & 4.49923217\end{array}$
$\begin{array}{llll}\mathrm{H} & -2.25857910 & -0.57841714 & 2.95922542\end{array}$
$\begin{array}{llll}\text { C } & 1.63384729 & 0.04255630 & -2.62678550\end{array}$
$0.90126347 \quad 0.48052812-1.92306855$
$1.07158188-0.20965958-3.53778924$
$4.00173932-0.17670496-1.10689971$
$3.55598595-2.54896366-0.65268078$
$\begin{array}{lllll}\text { C } & 3.66227916 & 1.12432587 & -1.68746356\end{array}$
$\begin{array}{lllll}\mathrm{H} & 2.41795405 & 1.96724134 & -3.21607465\end{array}$
$\begin{array}{llll}\mathrm{H} & 3.38823806 & 0.57969282 & -3.75220470\end{array}$
$\begin{array}{lllll}\text { C } & 5.07204627 & -0.18263059 & -0.10354461\end{array}$
C
$3.28725628-2.55566969 \quad 0.85463722$
$\begin{array}{lllll}\mathrm{H} & 2.93507044 & -3.31828095 & -1.13224441\end{array}$
$\begin{array}{lllll}\mathrm{H} & 4.61408400 & -2.79282581 & -0.86099914\end{array}$
$3.16844941 \quad 1.74181866-0.91387977$
$4.60854760 \quad 1.62706355-1.94722165$
$4.59525279-0.416167441 .32203617$
$\begin{array}{llll}\mathrm{H} & 5.55169573 & 0.80471352 & -0.18217795\end{array}$
$\begin{array}{llll}\mathrm{H} & 5.84336839 & -0.92233653 & -0.39067383\end{array}$
C $\quad 4.36919154-1.877189161 .67584357$
$\begin{array}{llll}\mathrm{H} & 2.30685027 & -2.08359114 & 1.04740739\end{array}$
$\begin{array}{llll}\mathrm{H} & 3.19025581 & -3.60720837 & 1.17030184\end{array}$
H
H
H
4.10925015 -1.95268106 2.74498327

H
5.32397462 -2.42897247 1.55377765


Electronic Energy =-1323.35889952 Hartree
Sum of Electronic and Thermal Enthalpies $=-1322.908405$ Hartree
Dispersion Correction $=-0.014901179$ Hartree

01
$\begin{array}{llll}\mathrm{Ru} & -0.53987500 & -0.53392734 & 0.11419815\end{array}$
O
$-0.44770459 \quad 0.47646351 \quad 2.03094144$
$\begin{array}{llll}\text { C } & 0.04775165 & 1.36115597 & -0.41658373\end{array}$
$\begin{array}{llll}\text { C } & 1.50158886 & -0.87896873 & 0.49301310\end{array}$
$\begin{array}{lllll}\text { C } & -2.78011861 & -0.68965689 & -0.11551727\end{array}$
$\begin{array}{llll}\text { C } & -2.29576134 & -1.71544188 & 0.75452215\end{array}$
$\begin{array}{lllll}\text { C } & -1.30948195 & -2.64800547 & 0.36799585\end{array}$
C $\quad-0.65888575 \quad-2.49978393-0.89099986$
$\begin{array}{lllll}\text { C } & -1.08931440 & -1.45919287 & -1.76179985\end{array}$
$\begin{array}{llll}\text { C } & -2.14315618 & -0.57622757 & -1.37233006\end{array}$
$\begin{array}{llll}C & -3.85911382 & 0.25133548 & 0.36275045\end{array}$
$\begin{array}{llll}\text { C } & -5.23328806 & -0.33858592 & 0.05922750\end{array}$
$\begin{array}{llll}\text { C } & -3.72955941 & 1.66159356 & -0.18980355\end{array}$
$0.45691324-3.41120922-1.28388813$
$\begin{array}{llll}\mathrm{O} & 0.29389463 & 2.23596581 & 3.11655327\end{array}$
$\begin{array}{llll}\text { C } & 0.05412532 & 1.60661026 & 1.96816819\end{array}$
$\begin{array}{lll}0.32710377 & 2.21768792 & 0.67823916\end{array}$
$\begin{array}{llll}\mathrm{O} & 3.42488280 & 0.95646589 & -1.94437768\end{array}$
$\begin{array}{llll}\mathrm{C} & 2.68507967 & 0.06114204 & -1.59140873\end{array}$
$\begin{array}{llll}\text { C } & 2.66477964 & -0.53972682 & -0.23434570\end{array}$
$\begin{array}{lllll}\mathrm{O} & 1.86340227 & -0.54189108 & -2.48385081\end{array}$
C
$\begin{array}{lllll}\text { C } & 0.90023145 & 4.06245776 & -0.76559081\end{array}$
$\begin{array}{lllll}\text { C } & 0.58332342 & 3.25355255 & -1.86388362\end{array}$
$\begin{array}{lllll}\text { C } & 0.17011686 & 1.93742079 & -1.69228320\end{array}$
$\begin{array}{lllll}\text { C } & 1.72797841 & -1.47748939 & 1.75002261\end{array}$
$\begin{array}{lllll}C & 3.00117738 & -1.72628500 & 2.25616898\end{array}$
$\begin{array}{llll}\text { C } & 4.12919984 & -1.35401218 & 1.52813486\end{array}$
$\begin{array}{llll}\text { C } & 3.95106021 & -0.74997059 & 0.29354864\end{array}$
$\begin{array}{llll}\mathrm{H} & -3.74216628 & 0.31159355 & 1.46337074\end{array}$
$\begin{array}{lllll}\mathrm{H} & 0.72255834 & 3.08876796 & 2.94527068\end{array}$
$\begin{array}{llll}\mathrm{H} & -2.68562846 & -1.75132085 & 1.77913913\end{array}$
H
H
$-0.95803950-3.40448800 \quad 1.07566816$
-0.58201416 -1.32168716 -2.71971269
-2.407688430 .25010895 -2.04024311
$-6.03964462 \quad 0.298713130 .45757678$
$-5.38077862-0.42764534-1.03165896$
-5.34892416 -1.34561490 0.49222913
-3.90364566 1.69158788 -1.27968577
-4.47943476 2.324696630 .26996986
$-2.72991207 \quad 2.08520939 \quad 0.00415508$

| H | 0.05800678 | -4.29611304 | -1.80788143 |
| :--- | :--- | :--- | :--- |
| H | 1.16287078 | -2.90006937 | -1.95425052 |
| H | 1.01396811 | -3.75991415 | -0.40090937 |
| H | 0.98641489 | 4.18623373 | 1.37048630 |
| H | 1.24111713 | 5.09011475 | -0.91238871 |
| H | 0.67255178 | 3.66424159 | -2.87478125 |
| H | -0.05396191 | 1.33844696 | -2.58220729 |
| H | 0.86908095 | -1.74792993 | 2.37551883 |
| H | 3.10920337 | -2.20338039 | 3.23594224 |
| H | 5.13507056 | -1.52942659 | 1.91995271 |
| H | 4.81461732 | -0.43598891 | -0.30093334 |
| H | 2.02381839 | -0.06958756 | -3.31998047 |



Electronic Energy =-1784.91751645 Hartree
Sum of Electronic and Thermal Enthalpies $=-1784.209577$ Hartree
Dispersion Correction $=-0.028412112$ Hartree

01
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$\begin{array}{llll}\mathrm{O} & -0.59251524 & 1.04625053 & 1.32994414\end{array}$
$\begin{array}{lllll}\text { C } & -1.22141146 & 0.77692029 & -1.23232512\end{array}$

C
$0.11932134-1.44854490 \quad 0.20343063$
$\begin{array}{lllll}\text { C } & -3.77461019 & 0.24457655 & 0.93260664\end{array}$
$\begin{array}{llll}\text { C } & -3.14959341 & -0.47164719 & 2.00472461\end{array}$
$\begin{array}{lllll}\text { C } & -2.69930316 & -1.80295486 & 1.87731240\end{array}$

| C | -2.70698978 | -2.42097920 | 0.59118531 |
| :---: | :---: | :---: | :---: |
| C | -3.28798101 | $-1.72202617$ | -0.50387563 |
| C | -3.81645633 | -0.40865756 | -0.31929728 |
| C | -4.24645479 | 1.66011554 | 1.16529175 |
| C | -5.55883010 | 1.65231204 | 1.94460216 |
| C | -4.37585253 | 2.48363261 | -0.10547630 |
| C | -2.12477191 | -3.78431431 | 0.40048501 |
| O | 0.46488004 | 2.99079458 | 1.06326423 |
| C | -0.23403561 | 2.04898982 | 0.61359635 |
| C | -0.62130067 | 1.98265302 | -0.81348349 |
| O | 1.08206666 | -1.68077784 | -3.22132818 |
| C | 0.36128573 | $-1.99692788$ | -2.29451446 |
| C | 0.80161133 | $-2.05483944$ | -0.87619621 |
| O | -0.89739339 | -2.42578467 | -2.51049829 |
| C | -0.37053364 | 3.04101364 | $-1.69146505$ |
| C | -0.70737703 | 2.91887717 | -3.03473632 |
| C | -1.29271114 | 1.72945753 | -3.47964277 |
| C | -1.54812315 | 0.68240093 | $-2.59474441$ |
| C | 0.75472925 | -1.59724754 | 1.45914057 |
| C | 1.91970221 | $-2.33501131$ | 1.64785598 |
| C | 2.55910808 | -2.93228409 | 0.55930507 |
| C | 2.00512564 | -2.76173785 | -0.70082775 |
| H | -3.46999276 | 2.13660952 | 1.79802576 |
| H | -2.99041176 | 0.05528513 | 2.95374987 |
| H | -2.21892101 | -2.30817719 | 2.72086768 |
| H | -3.29469528 | -2.18346562 | -1.49454169 |
| H | -4.18357444 | 0.13103469 | -1.19795838 |
| H | -5.88611067 | 2.67718963 | 2.18550675 |
| H | -6.35670974 | 1.17401368 | 1.34929653 |


| H | -5.47189516 | 1.09314693 | 2.89069021 |
| :---: | :---: | :---: | :---: |
| H | -5.18284324 | 2.10009357 | -0.75480310 |
| H | -4.63254617 | 3.52608345 | 0.14142421 |
| H | -3.44082393 | 2.49651373 | -0.69029383 |
| H | -2.88316502 | -4.55857596 | 0.60610544 |
| H | -1.76787993 | -3.91650101 | -0.63158284 |
| H | -1.27478923 | -3.95392260 | 1.08056713 |
| H | 0.08542688 | 3.95508708 | -1.29430834 |
| H | -0.52324247 | 3.74004328 | -3.73382742 |
| H | -1.56311132 | 1.62133143 | -4.53582629 |
| H | -2.03184306 | -0.22388799 | -2.97888344 |
| H | 0.29222463 | -1.12457284 | 2.33490661 |
| H | 2.32901379 | -2.45494200 | 2.65981322 |
| H | 3.47197088 | -3.52246375 | 0.69217765 |
| H | 2.49153731 | -3.19531911 | $-1.58203468$ |
| H | -1.03034559 | -2.34566633 | -3.47183563 |
| H | 1.79580910 | 2.22418243 | 1.60464719 |
| N | 2.66179896 | 1.63640431 | 1.74911215 |
| C | 2.81896467 | 0.95803502 | 3.01930119 |
| C | 3.45289131 | 1.44933805 | 0.71445916 |
| C | 4.26883185 | 0.58259463 | 3.22324106 |
| H | 2.16334340 | 0.06790289 | 3.02339483 |
| H | 2.45312902 | 1.62837732 | 3.81137036 |
| N | 4.53736104 | 0.66703690 | 0.78570863 |
| C | 3.08135800 | 2.09209145 | -0.58121226 |
| C | 4.80341995 | -0.11664880 | 1.99193384 |
| H | 4.37974457 | -0.07218112 | 4.10096354 |
| H | 4.86668638 | 1.49091031 | 3.41755411 |
| C | 5.36404805 | 0.38600068 | -0.39471943 |

$\begin{array}{llll}\text { C } & 2.59917947 & 1.04799336 & -1.59777592\end{array}$
$\begin{array}{llll}\mathrm{H} & 2.28217411 & 2.81952301 & -0.37240081\end{array}$
$\begin{array}{llll}\mathrm{H} & 3.94320366 & 2.65684066 & -0.97935186\end{array}$
$\begin{array}{llll}\mathrm{H} & 4.34951284 & -1.11973102 & 1.87209220\end{array}$
$\begin{array}{llll}\mathrm{H} & 5.89348544 & -0.25512987 & 2.06797306\end{array}$
$\begin{array}{llll}\text { C } & 4.67232347 & -0.44546396 & -1.46163161\end{array}$
$\begin{array}{llll}\mathrm{H} & 6.25248214 & -0.14059995 & -0.01481951\end{array}$
$\begin{array}{lllll}\mathrm{H} & 5.73201753 & 1.33939040 & -0.81574160\end{array}$
$\begin{array}{llll}\text { C } & 3.71147252 & 0.33436737 & -2.34490742\end{array}$
$\begin{array}{lllll}\mathrm{H} & 1.95616151 & 0.31182062 & -1.07703772\end{array}$
$\begin{array}{llll}\mathrm{H} & 1.92943337 & 1.54806382 & -2.31772998\end{array}$
$\begin{array}{llll}\mathrm{H} & 4.14167569 & -1.27855877 & -0.96436336\end{array}$
$\begin{array}{llll}\mathrm{H} & 5.45262502 & -0.90721024 & -2.09106335\end{array}$
$\begin{array}{llll}\mathrm{H} & 3.25176015 & -0.35239677 & -3.07511199\end{array}$
$\begin{array}{llll}\mathrm{H} & 4.28901210 & 1.07874397 & -2.92941583\end{array}$


TS 1
Electronic Energy =-1323.30651951 Hartree
Sum of Electronic and Thermal Enthalpies $=-1322.860757$ Hartree
Dispersion Correction $=-0.014833813$ Hartree

01
$\begin{array}{llll}\mathrm{Ru} & -0.02187072 & -0.52363567 & -0.00740733\end{array}$
$\begin{array}{llll}\mathrm{O} & -0.59973103 & 0.36617040 & 1.89681966\end{array}$

C
1.88220898 -0.06221241 0.74340952

O
1.6058
$1.30031625-1.82834463$

| C | -0.69178234 | 1.63244494 | -0.58784808 |
| :---: | :---: | :---: | :---: |
| H | 0.46696599 | 1.26733316 | -0.76923376 |
| O | -1.28813604 | 2.14033530 | 2.98702742 |
| O | 3.79120652 | 0.93472490 | -2.15981086 |
| C | -1.03778267 | 1.51484983 | 1.84341396 |
| C | -1.33828397 | 2.14875134 | 0.55762646 |
| C | 2.78172211 | 0.96537009 | -1.44913690 |
| C | 2.91833778 | 0.53442218 | 0.00444785 |
| C | -1.82339131 | -1.90934444 | -0.13295056 |
| C | -0.68668786 | -2.48151271 | 0.54827614 |
| C | 0.57668833 | -2.60135552 | -0.07162578 |
| C | 0.81002006 | -2.02929582 | $-1.36297682$ |
| C | -0.28816018 | $-1.36973920$ | -1.97408032 |
| C | -1.60615491 | $-1.38511871$ | -1.41065294 |
| C | -3.14126610 | $-1.82525986$ | 0.59450441 |
| C | -4.04764659 | -0.71445248 | 0.09171930 |
| C | -3.84247560 | -3.18014875 | 0.53960998 |
| C | 2.14945923 | -2.09171739 | -2.01457674 |
| H | -2.88961690 | -1.61037196 | 1.65246676 |
| H | 1.42062800 | -3.01249053 | 0.49168792 |
| H | -0.80541779 | $-2.80933610$ | 1.58838398 |
| H | -0.10377468 | -0.82155390 | $-2.90369500$ |
| H | -2.40426217 | -0.84263987 | -1.92520518 |
| H | -4.40867311 | -0.91756650 | -0.93135053 |
| H | -3.53870735 | 0.26508122 | 0.08255114 |
| H | -4.93856764 | -0.62746155 | 0.73296320 |
| H | -4.09715129 | -3.44014878 | $-0.50241246$ |
| H | -3.20797833 | -3.98795600 | 0.93866880 |
| H | -4.77696985 | -3.16481716 | 1.12286120 |


| H | 2.23818689 | -3.02831179 | -2.59048535 |
| :--- | :--- | :--- | :--- |
| H | 2.95968281 | -2.07105845 | -1.27006399 |
| H | 2.30616038 | -1.24673071 | -2.70140041 |
| C | -2.36302899 | 3.09925132 | 0.44627360 |
| C | -2.76896739 | 3.54411318 | -0.80442058 |
| C | -2.13778208 | 3.04560786 | -1.94624076 |
| C | -1.10499243 | 2.12125837 | -1.83640905 |
| C | 4.17740008 | 0.70523796 | 0.59945896 |
| C | 4.42111173 | 0.31282220 | 1.90872852 |
| C | 3.40106893 | -0.29462962 | 2.63867668 |
| C | 2.15064116 | -0.48658711 | 2.05442934 |
| H | -2.89293433 | 3.45819171 | 1.33675723 |
| H | -3.57986266 | 4.27168822 | -0.89137167 |
| H | -2.44582033 | 3.39840244 | -2.93469614 |
| H | -0.55534178 | 1.78932508 | -2.72190949 |
| H | 4.96540564 | 1.14879325 | -0.01658149 |
| H | 5.40809009 | 0.46527663 | 2.35607824 |
| H | 3.57621964 | -0.62781085 | 3.66704686 |
| H | 1.36728507 | -0.97390624 | 2.64801111 |
| H | -1.51607363 | 3.06913219 | 2.82338180 |



TS 2
Electronic Energy =-1784.89611675 Hartree
Sum of Electronic and Thermal Enthalpies $=-1784.189305$ Hartree
Dispersion Correction $=-0.028362797$ Hartree

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Ru
O
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C
C
C
C
H

H

H

H

H
H
$-1.719259650 .600513470 .14633983$
$-0.78316488-0.10005645-1.65231593$
$-0.30267040 \quad 2.09927348 \quad-0.06417180$
$0.17696952 \quad 0.777285892 .55441565$
-0.36870082 $-1.55992788 \quad 0.72368456$
$-0.05592113-0.50171950 \quad 0.84465728$
$0.33559193-1.70321012$-2.73250622
$2.28587876 \quad 0.69743628 \quad 1.75396838$
-0.29199681 -1.27957785 -1.74450338
$-0.42124953-2.15701044-0.53608118$
$1.12042009 \quad 1.16583412 \quad 1.81899910$
$\begin{array}{lll}0.79541317 & 2.24627399 & 0.80455133\end{array}$
$\begin{array}{lll}-3.82480514 & -0.27235282 & -0.28771979\end{array}$
-3.56939647 1.07665242 -0.73437356
-3.20426993 $2.10856156 \quad 0.17504744$
$-2.88146045 \quad 1.80869006 \quad 1.53428576$
$-2.94761577 \quad 0.439425281 .89806776$
$-3.51373101-0.564206451 .03919536$
-4.29253599 -1.29065091-1.29611012
-4.21808206 -2.72472690 -0.80500296
$-5.71173089-0.94997633-1.74669141$
$-2.36049354 \quad 2.84374526 \quad 2.47386424$
-3.61680947 -1.18300370 -2.16953451
-3.04018852 $3.12443052-0.19887447$
$-3.68746550 \quad 1.31709468-1.79801034$
$-2.54636847 \quad 0.14400309 \quad 2.87310502$
-3.54519131 -1.597473051 .39548784
-4.91288977 -2.89481553 0.03619673

H

| -3.20437645 | -3.00265823 | -0.47415036 |
| :--- | :--- | :--- |
| -4.50729528 | -3.41828363 | -1.60991588 |

$$
-6.41453357-1.03142756-0.89932644
$$

$$
\begin{array}{lll}
-5.79069842 & 0.07284120 & -2.14812982
\end{array}
$$

-6.04915605 -1.64486806 -2.53209568

$$
\begin{array}{lll}
-3.15726310 & 3.19920539 & 3.14816881
\end{array}
$$

$$
\begin{array}{lll}
-1.96132838 & 3.71015573 & 1.92396527
\end{array}
$$

$$
\begin{array}{lll}
-1.54019305 & 2.41624649 & 3.07259869
\end{array}
$$

$$
\begin{array}{llll}
-0.55456261 & -3.54395205 & -0.62962887
\end{array}
$$

$$
\begin{array}{lll}
-0.68977294 & -4.30354280 & 0.53080579
\end{array}
$$

$$
\begin{array}{lll}
-0.65323894 & -3.69139266 & 1.78835632
\end{array}
$$

$$
\begin{array}{lll}
-0.47183545 & -2.31402419 & 1.89099660
\end{array}
$$

$$
\begin{array}{lll}
1.64905380 & 3.34918910 & 0.66948185
\end{array}
$$

$$
\begin{array}{lll}
1.44104800 & 4.30638538 & -0.31981655
\end{array}
$$

$$
\begin{array}{lll}
0.39618450 & 4.13156064 & -1.22759219
\end{array}
$$

$$
\begin{array}{lll}
-0.45484775 & 3.03331870 & -1.09988104
\end{array}
$$

-0.56505848 -4.00841514 -1.62054383

$$
\begin{array}{lll}
-0.81941238 & -5.38763517 & 0.45861609
\end{array}
$$

$$
\begin{array}{lll}
-0.74400726 & -4.30093565 & 2.69236350
\end{array}
$$

$$
\begin{array}{lll}
-0.37829796 & -1.80002006 & 2.85315552
\end{array}
$$

$$
\begin{array}{lll}
2.48808901 & 3.44486447 & 1.36863356
\end{array}
$$

$$
\begin{array}{lll}
2.09914849 & 5.17849761 & -0.39292299
\end{array}
$$

$$
\begin{array}{lll}
0.23873960 & 4.85311289 & -2.03617605
\end{array}
$$

$$
\begin{array}{lll}
-1.25286986 & 2.89706213 & -1.84142096
\end{array}
$$

$$
\begin{array}{lll}
1.86697417 & -0.79752645 & -2.36215218
\end{array}
$$

$$
2.60546435-0.21042828-1.93919969
$$

$$
3.48968817 \quad-0.79497810 \quad-1.15096861
$$

$$
4.45202824-0.10345706-0.54719831
$$

$$
4.41369122 \quad 1.36330986 \quad-0.57572210
$$

$\begin{array}{llll}\text { C } & 3.87695461 & 1.85413684 & -1.90226484\end{array}$
$\begin{array}{lllll}\text { C } & 2.53046191 & 1.22444942 & -2.16735213\end{array}$
$\begin{array}{lllll}C & 3.34685296 & -2.26371119 & -0.91776552\end{array}$
$\begin{array}{llll}\text { C } & 2.95355902 & -2.57255802 & 0.53214664\end{array}$
$\begin{array}{llll}\text { C } & 4.11326286 & -2.59458073 & 1.51254662\end{array}$
$\begin{array}{lllll}\text { C } & 4.81796397 & -1.25629203 & 1.66275568\end{array}$
$\begin{array}{llll}\text { C } & 5.41139126 & -0.74029813 & 0.36302942\end{array}$
$\begin{array}{llll}\mathrm{H} & 5.43921948 & 1.72514168 & -0.40738779\end{array}$
$\begin{array}{llll}\mathrm{H} & 3.77576296 & 1.68934950 & 0.26909926\end{array}$
$\begin{array}{llll}\mathrm{H} & 4.58804480 & 1.60320488 & -2.71039587\end{array}$
$\begin{array}{llll}\mathrm{H} & 3.77310681 & 2.94960690 & -1.87782100\end{array}$
$\begin{array}{llll}\mathrm{H} & 2.18626153 & 1.38557381 & -3.20020992\end{array}$
$\begin{array}{llll}\mathrm{H} & 1.75895876 & 1.64435301 & -1.50062644\end{array}$
$\begin{array}{lllll}\mathrm{H} & 4.28855814 & -2.77716811 & -1.18558363\end{array}$
$\mathrm{H} \quad 2.57293783-2.63440311$-1.60926206
$\begin{array}{lllll}\mathrm{H} & 2.43713820 & -3.54724748 & 0.54268280\end{array}$
$\begin{array}{llll}\mathrm{H} & 2.21538099 & -1.82153774 & 0.86803402\end{array}$
H
H
$3.73714750-2.91766704 \quad 2.49769572$
H
$5.64154595-1.35742753 \quad 2.39035565$
H
$4.11418503-0.50202879 \quad 2.06022537$
$\begin{array}{llll}\mathrm{H} & 5.93488425 & -1.54916215 & -0.18111124\end{array}$
H
$6.17228676 \quad 0.02671748 \quad 0.56922285$

NMR Spectra of Synthesized Compounds


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$\begin{array}{llllllllllll}210 & 200 & 190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 & 100 \\ \mathrm{fl}(\mathrm{ppm})\end{array}$









21

$\left.\begin{array}{lllllllllllllllllllllllll}210 & 200 & 190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 & 100 \\ f 1(\mathrm{ppm})\end{array}\right)$


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3d






3f


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| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 170 | 160 | 150 | 140 | 130 | 120 | 110 | 100 |  |  | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 |



## 



4b






[^2]

4 e

$\begin{array}{llllllllllllllllllllllllllllll}210 & 200 & 190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 & 100 & 90 & 80 & 70 & 60 & 50 & 40 & 30 & 20 & 10 & 0 & -10\end{array}$

$\begin{array}{llllllllllll}210 & 200 & 190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 & 100 \\ f 1\end{array}$

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7


Nos


7
$\begin{array}{llllllllllllllllllllllllll}210 & 200 & 190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 & 100 & 90 & 80 & 70 & 60 & 50 & 40 & 30 & 20 & 10 & 0 & -10\end{array}$
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8a


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8b


## 



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8c


| 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| :00 | 190 | 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 | 100 | 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 |

## $\underbrace{\text { No }}$


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$\begin{array}{lllllllllllllllllllllllll}00 & 190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 & 100 & 90 & 80 & 70 & 60 & 50 & 40 & 30 & 20 & 10 & 0 & -1\end{array}$


[^0]:    

[^1]:    $\begin{array}{llllllllllllllllllllllllllll}210 & 200 & 190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 & 100 & 90 & 80 & 70 & 60 & 50 & 40 & 30 & 20 & 10 & 0 & -10\end{array}$

[^2]:    

[^3]:    

