- Supporting Information -

# Ruthenium(II) Catalysis/Noncovalent Interaction Synergy for Cross-Dehydrogenative Coupling of Arene Carboxylic Acids

# Suman Dana,<sup>a</sup> Deepan Chowdhury,<sup>a</sup> Anup Mandal,<sup>a</sup> Francis A. S. Chipem,<sup>b</sup> and Mahiuddin Baidya<sup>a\*</sup>

<sup>a</sup>Department of Chemistry, Indian Institute of Technology Madras, Chennai 600036, India <sup>b</sup>Department of Chemistry, Manipur University, Canchipur 795003, India *mbaidya@iitm.ac.in* 

# Table of Contents

1	General information	S3
2	Typical ruthenium(II)-catalyzed homo-dimerization of benzoic acids (isolated as ester)	S4
3	Typical ruthenium(II)-catalyzed cross-dimerization of benzoic acids (isolated as ester):	S4
4	Typical synthesis of biphenyl-tethered lactones (4a-e)	S5
5	Typical synthesis of biphenyl-tethered ether (6)	S6
6	Typical synthesis of 6 <i>H</i> -benzo[c]chromen-6-one (7)	S6
7	H/D-Scrambling study	S8
8	Radical scavengers study	S9
9	Studying the role of oxygen	S9
10	Use of strong oxidant	S10
11	Role of DBUH <sup>+</sup>	S10
11	ESI-HRMS analysis of reaction mixture	S11
12	NMR spectroscopic data of synthesized compounds	S12-15
13	Computational details	S16
14	Cartesian coordinates	S17-32
15	NMR spectra of synthesized compounds	S33-68

#### **General information**

[Ru(*p*-cymene)Cl<sub>2</sub>]<sub>2</sub> and CuO were purchased from Alfa Aesar company. Benzoic acids and DBU were purchased from Avra chemicals and Spectrochem. All reactions were monitored by thin layer chromatography (TLC) on Merck 60 F 254 precoated silica plates and visualized using a UV lamp (366 or 254 nm) or by use of potassium permanganate, 5 g K<sub>2</sub>CO<sub>3</sub>, / 100 mL water. Products were isolated by column chromatography (Merck silica gel 100-200µm). <sup>13</sup>C and <sup>1</sup>H NMR spectra were recorded on a Bruker 400 or Bruker 500 MHz spectrometers. Chemical shift values ( $\delta$ ) are reported in ppm and calibrated to the residual solvent peak CDCl<sub>3</sub>  $\delta = 7.2600$  ppm for <sup>1</sup>H,  $\delta = 77.16$  for <sup>13</sup>C; or calibrated to tetramethylsilane ( $\delta = 0.00$ ). All NMR spectra were recorded at ambient temperature (290 K) unless otherwise noted. <sup>1</sup>H NMR spectra are reported as follows: chemical shift (multiplicity, coupling constant, integration). The following abbreviations are used to indicate multiplicities: s, singlet; d, doublet; t, triplet; q, quartet; quint, quintet; sext, sextet; sept, septet; m, multiplet; dd, doublet of doublet; dt, doublet of triplet; dq, doublet of quartet; td, triplet of doublet; tt, triplet of triplet; dq, doublet of quartet; br, broad. Mass spectra were recorded by electron spray ionization (ESI) method on a Q-TOF Micro with lock spray source. Typical ruthenium(II)-catalyzed *homo-*dimerization of benzoic acids (*isolated as ester*):



**Procedure:** To an oven dried screw cap reaction tube  $(10 \times 1.5 \text{ cm})$ , corresponding benzoic acid **1** (0.3 mmol, 1.0 equiv), [Ru(*p*-cymene)Cl<sub>2</sub>]<sub>2</sub> (2.5 mol %), CuO (1.0 equiv), and DBU (1.0 equiv) were taken. Then dry dioxane (0.4 mL) was added in it and the mixture was stirred at 110 °C for 24 h under air. After completion (monitored by TLC), K<sub>2</sub>CO<sub>3</sub> (3.0 equiv) and iodomethane (3.0 equiv) were added to the reaction mixture and stirred for 4 h at room temperature. Then the solvent was evaporated under reduced pressure. In order to get pure 2,2'-biaryl acid methyl ester **2**, the resulting residue was purified by column chromatography on silica gel with a gradient eluent of hexane and ethyl acetate.

### Typical ruthenium(II)-catalyzed *cross*-dimerization of benzoic acids (*isolated as ester*):



**Procedure:** To an oven dried screw cap reaction tube  $(10 \times 1.5 \text{ cm})$ , corresponding benzoic acid (**Ar**<sup>1</sup>) (0.15 mmol, 1.0 equiv), benzoic acid (**Ar**<sup>2</sup>) (4.0 equiv), [Ru(*p*-cymene)Cl<sub>2</sub>]<sub>2</sub> (10.0 mol %), CuO (1.5 equiv), and DBU (5.0 equiv) were taken. Then dry dioxane (0.5 mL) was added in it and the mixture was stirred at 110 °C for 24 h under air. After completion (monitored by TLC), K<sub>2</sub>CO<sub>3</sub> (4.0 equiv) and iodomethane (8.0 equiv) were added to the reaction mixture and stirred for 4 h at room temperature. Then the solvent was evaporated under reduced pressure. In order to get pure 2,2'-biaryl acid methyl ester **3**, the resulting residue was purified by column chromatography on silica gel with a gradient eluent of hexane and ethyl acetate.

Compound	<b>Ar</b> <sup>1</sup> (1.0 equiv)	<b>Ar</b> <sup>2</sup> (4.0 equiv)
3a	2-Methoxybenzoic acid	2-Fluorobenzoic acid
3b	2,4-Dimethylbenzoic acid	3-Acetylbenzoic acid
3c	4-Methoxybenzoic acid	2-Fluorobenzoic acid
3d	1-Naphthoic acid	2-Benzoylbenzoic acid
3e	2-Methoxybenzoic acid	1-Naphthoic acid
3f	4-Methoxybenzoic acid	1-Naphthoic acid

Table for the  $Ar^1/Ar^2$  acid ratio:

Typical synthesis of biphenyl-tethered lactones (4a-e):



Procedure: A solution of corresponding 2,2'-biaryl acid methyl ester 2 (0.15 mmol, 1 equiv) in THF (1.0 mL) was slowly added to a suspension of LiAlH<sub>4</sub> (3.0 equiv) in THF (2.0 mL). After stirring for 2 h at room temperature, the mixture was hydrolyzed by careful addition of water (25 mL) and acidified with 2 N HCl. The organic solvent was evaporated, and the aqueous residue was diluted with water (5 mL) and extracted with dichloromethane. The organic washed extract was with water and dried over Na<sub>2</sub>SO<sub>4</sub>. After evaporation of the solvent,  $[RuCl_2(p-cymene)]_2$ (10.0)mol %) and  $Cs_2CO_3$ (20.0 mol %) were added to the crude product. Next, Toluene (1.5 mL) was added in the reaction vessel and the reaction mixture was stirred at 100 °C under oxygen atmosphere (using an O<sub>2</sub> balloon) for 3 h. After cooling to room temperature, the solvent was evaporated under reduced pressure. In order to get pure product 4, the resulting residue was purified by column chromatography on silica gel with a gradient eluent of hexane and ethyl acetate.

Typical synthesis of biphenyl-tethered ether (6):



**Procedure:** A solution of 2,2'-biaryl acid methyl ester **2c** (0.15 mmol, 1 equiv) in THF (1.0 mL) was slowly added to a suspension of LiAlH<sub>4</sub> (3.0 equiv) in THF (2.0 mL). After stirring for 2 h at room temperature, the mixture was hydrolyzed by careful addition of water (25 mL) and acidified with 2 N HC1. The organic solvent was evaporated, and the aqueous residue was diluted with water (5 mL) and extracted with dichloromethane. The combined organic extract was washed with water and dried over Na<sub>2</sub>SO<sub>4</sub>. After evaporation of the solvent conc. sulphuric acid (0.1 equiv) and 0.5 mL dioxane were added to the crude product and the reaction mixture was stirred at 60 °C for 24 h. Then solvent was evaporated under reduced pressure and DCM (10 mL) was added. The mixture was carefully washed with sodium bicarbonate solution. The solvent was evaporated under reduced pressure and pressure and pressure and resulting residue was purified by column chromatography on silica gel with a gradient eluent of hexane and ethyl acetate.

### Typical synthesis of 6*H*-benzo[c]chromen-6-one (7):



**Procedure:** 2,2'-Biaryl acid methyl ester **2b** (0.15 mmol, 1.0 equiv) and KOH (3.0 equiv) were taken in a 25 mL round bottom flask and 5 mL MeOH:water (4:1) was added in it. The mixture was refluxed until the consumption of the ester (monitored by TLC). Volatiles were removed under vacuo and the resulting solution was neutralized with 2N HC1. The acid was extracted using ethyl acetate, dried with MgSO<sub>4</sub>, and evaporated to get the acid **2b'**. The crude acid **2b'** was taken in a screw cap reaction tube ( $10 \times 1.5$  cm). Then AgNO<sub>3</sub> (40 mol %), (NH<sub>4</sub>)<sub>2</sub>S<sub>2</sub>O<sub>8</sub> (5.0 equiv), KOAc (5.0 equiv), dichloroethane (DCE, 1 mL), and H<sub>2</sub>O (1 mL) were sequentially added in it. The reaction mixture was stirred at room temperature for 36 h. After completion, it was diluted with H<sub>2</sub>O and extracted with CH<sub>2</sub>Cl<sub>2</sub>

 $(3 \times 20 \text{ mL})$ . The organic extract was dried over MgSO<sub>4</sub> and concentrated in vacuo. Purification of the crude product by flash chromatography on silica gel using ethyl acetate/hexane system afforded the desired product 7.

Mechanistic studies:

(a) H/D-Scrambling study:



**Procedure:** To an oven dried screw cap reaction tube ( $10 \times 1.5$  cm), benzoic acid **1c** (0.3 mmol, 1 equiv), [Ru(*p*-cymene)Cl<sub>2</sub>]<sub>2</sub> (2.5 mol %), CuO (1.0 equiv), and DBU (1.0 equiv) were taken. Then dry dioxane (0.4 mL) and D<sub>2</sub>O (15.0 equiv) were added and the mixture was stirred at 110 °C for 14 h under air. Then, a spoon of sodium sulfate was added in the reaction mixture. The reaction mixture was diluted with DCM and transferred to a 25 mL round-bottom flask. Next, the volatiles were evaporated under reduced pressure and K<sub>2</sub>CO<sub>3</sub> (3.0 equiv) and iodomethane (3.0 equiv) were added to the reaction mixture followed by the addition of acetonitrile (2.0 mL). The resulting solution was stirred at room temperature for 4 h. Then the solvent was evaporated under reduced pressure. In order to get pure product **1c'**, the resulting residue was purified by column chromatography on silica gel with a gradient eluent of hexane and ethyl acetate. The deuterium scrambling was observed through <sup>1</sup>H NMR spectroscopy.



(b) Radical scavengers study:



Radical scavengers: TEMPO = 45%; BHT = 54%; 1,1-diphenylethene = 41%

**Procedure:** To an oven dried screw cap reaction tube  $(10 \times 1.5 \text{ cm})$  benzoic acid **1a** (0.3 mmol, 1.0 equiv), [Ru(*p*-cymene)Cl<sub>2</sub>]<sub>2</sub> (2.5 mol %), CuO (1.0 equiv), DBU (1.0 equiv), and radical scavenger (3.0 equiv) were taken. Then dry dioxane (0.4 mL) were added into the mixture and it was stirred at 110 °C for 24 h under air. After completion (monitored by TLC), K<sub>2</sub>CO<sub>3</sub> (3.0 equiv) and iodomethane (3.0 equiv) were added to the reaction mixture and the resulting solution was stirred at room temperature for 4 h. Then the solvent was evaporated under reduced pressure. In order to get pure 2,2′-biaryl acid methyl ester **2a**, the resulting residue was purified by column chromatography on silica gel with a gradient eluent of hexane and ethyl acetate.

### (c) Studying the role of oxygen:



**Procedure:** To an oven dried reaction tube  $(10 \times 1.5 \text{ cm})$  benzoic acid **1a** (0.3 mmol, 1 equiv), [Ru(*p*-cymene)Cl<sub>2</sub>]<sub>2</sub> (2.5 mol %), CuO (1.0 equiv), and DBU (1.0 equiv) were taken. Then the reaction tube was evacuated under vacuum and back-filled with nitrogen gas (three-times). Then dry dioxane (0.4 mL) were added into the mixture and it was stirred at 110 °C for 24 h under air. After completion (monitored by TLC), K<sub>2</sub>CO<sub>3</sub> (3.0 equiv) and iodomethane (3.0 equiv) were added to the reaction mixture and the resulting solution was stirred at room temperature for 4 h. Then the solvent was evaporated under reduced pressure. In order to get pure 2,2'-biaryl acid methyl ester **2a**, the resulting residue was purified by column chromatography on silica gel with a gradient eluent of hexane and ethyl acetate.

#### (d) Use of strong oxidant:



**Procedure:** To an oven dried screw cap reaction tube  $(10 \times 1.5 \text{ cm})$  corresponding benzoic acid 1 (0.3 mmol, 1 equiv),  $[\operatorname{Ru}(p\text{-cymene})\operatorname{Cl}_2]_2$  (5.0 mol %), AgNO<sub>3</sub> (1.5 equiv), Na<sub>2</sub>CO<sub>3</sub> (2.0 equiv), and K<sub>2</sub>S<sub>2</sub>O<sub>8</sub> (2.0 equiv) were taken. Then dry acetonitrile (0.5 mL) was added into the mixture and it was stirred at 100 °C for 24 h under air. After completion (monitored by TLC), K<sub>2</sub>CO<sub>3</sub> (3.0 equiv) and iodomethane (3.0 equiv) were added to the reaction mixture and the resulting solution was stirred at room temperature for 4 h. Then the solvent was evaporated under reduced pressure. In order to get pure product **8**, the resulting residue was purified by column chromatography on silica gel with a gradient eluent of hexane and ethyl acetate.

#### (e) Role of DBUH<sup>+</sup>:



**Procedure:** To an oven dried screw cap reaction tube  $(10 \times 1.5 \text{ cm})$  sodium benzoate (9, 0.3 mmol, 1 equiv),  $[\text{Ru}(p\text{-cymene})\text{Cl}_2]_2$  (2.5 mol %), and CuO (1.0 equiv) were taken. Then dry dioxane (0.4 mL) was added into the mixture and it was stirred at 110 °C for 24 h under air. After completion (monitored by TLC), K<sub>2</sub>CO<sub>3</sub> (3.0 equiv) and iodomethane (3.0 equiv) were added to the reaction mixture and the resulting solution was stirred at room temperature for 4 h. Same experiment was also performed in the presence of DBU (1.0 equiv). In both of the cases, the formation of 2,2'-biaryl acid methyl esters was not observed, indicating the decisive role of DBUH<sup>+</sup> in the transformation.

## (f) ESI-HRMS analysis of reaction mixture:

To an oven dried screw cap reaction tube  $(10 \times 1.5 \text{ cm}) [\text{Ru}(p\text{-cymene})\text{Cl}_2]_2 (0.016 \text{ mmol})$ , benzoic acid **1a** (2.0 equiv.), and DBU (2.0 equiv.) were taken. Then dry dioxane (0.3 mL) was added in it and the mixture was stirred at 110 °C under air. After 90 minutes, the crude reaction mixture was taken through a syringe and ESI-HRMS was recorded.



#### NMR spectroscopic data of synthesized compounds:





















**Dimethyl 4,4'-dimethyl-[1,1'-biphenyl]-2,2'-dicarboxylate (2a):** yield = 32.3 mg (72%), colourless liquid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.90 (d, J = 8.0 Hz, 2H), 7.21 (d, J = 8.0 Hz, 2H), 7.00 (s, 2H), 3.61 (s, 6H), 2.40 (s, 6H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  167.46, 143.77, 142.09, 131.02, 130.04, 127.91, 126.62, 51.78, 21.64 ppm.

**Dimethyl [1,1'-biphenyl]-2,2'-dicarboxylate (2b):** yield = 23.4 mg (58%), colourless liquid; <sup>1</sup>**H** NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.00 (d, J = 7.8 Hz, 2H), 7.63 – 7.49 (m, 2H), 7.47 – 7.38 (m, 2H), 7.21 (d, J = 7.6 Hz, 2H), 3.61 (s, 6H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  167.55, 143.41, 131.60, 130.33, 129.98, 129.49, 127.30, 51.94 ppm.

**Dimethyl 3,3'-dimethyl-[1,1'-biphenyl]-2,2'-dicarboxylate (2c):** yield = 27 mg (60%), colourless liquid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.37 – 7.25 (m, 2H), 7.25 – 7.17 (m, 2H), 7.16 – 7.04 (m, 2H), 3.55 (s, 6H), 2.40 (s, 6H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  169.64, 139.26, 135.85, 133.12, 129.59, 129.02, 127.17, 51.80, 20.18 ppm.

**Dimethyl 3,3'-dimethoxy-[1,1'-biphenyl]-2,2'-dicarboxylate (2d):** yield = 33.6 mg (68%), white solid; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.36 – 7.31 (m, 2H), 6.94 – 6.90 (m, 2H), 6.90 – 6.87 (m, 2H), 3.86 (s, 6H), 3.59 (s, 6H) ppm; <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  167.85, 156.60, 139.32, 130.20, 123.32, 121.89, 110.48, 56.09, 52.12 ppm; HRMS (m/z) C<sub>18</sub>H<sub>18</sub>O<sub>6</sub>K<sup>+</sup> 369.0735.

**Dimethyl 3,3'-difluoro-[1,1'-biphenyl]-2,2'-dicarboxylate (2e):** yield = 33 mg (72%), colourless liquid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.48 – 7.37 (m, 2H), 7.19 – 7.13 (m, 2H), 7.06 (dd, J = 7.7, 0.5 Hz, 2H), 3.66 (s, 6H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  165.15, 160.05 (d, J = 253.6 Hz), 140.67, 131.52 (d, J = 9.3 Hz), 125.68 (d, J = 3.4 Hz), 121.11 (d, J = 15.5 Hz), 115.84 (d, J = 21.9 Hz), 52.52 ppm; HRMS (m/z) C<sub>16</sub>H<sub>12</sub>F<sub>2</sub>O<sub>4</sub>Na<sup>+</sup> 329.0591.

**Dimethyl 3,3'-dibromo-[1,1'-biphenyl]-2,2'-dicarboxylate (2f):** yield = 41.7 mg (65%), white solid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.53 (dd, *J* = 7.3, 1.8 Hz, 2H), 7.26 - 7.14 (m, 4H), 3.57 (s, 6H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  167.29, 138.74, 135.47, 132.63, 130.22, 128.70, 119.88, 52.56 ppm; HRMS (m/z) C<sub>16</sub>H<sub>12</sub>Br<sub>2</sub>O<sub>4</sub>K<sup>+</sup>464.8726.

**Dimethyl 3,3'-dibenzoyl-[1,1'-biphenyl]-2,2'-dicarboxylate (2g):** yield = 57.4 mg (80%), white solid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.08 – 8.01 (m, 2H), 7.79 – 7.70 (m, 3H), 7.69 – 7.60 (m, 2H), 7.59 – 7.50 (m, 4H), 7.47 – 7.36 (m, 5H), 3.60 (s, 6H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 197.15, 166.47, 141.74, 137.22, 133.17, 132.51, 130.16, 129.72, 129.31, 129.27, 128.60, 127.85, 52.26 ppm.

**Dimethyl [1,1':3',1'':3'',1'''-quaterphenyl]-2',2''-dicarboxylate (2h):** yield = 39.9 mg (63%), white solid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.43 – 7.35 (m, 2H), 7.35 – 7.20 (m, 14H), 3.25 (s, 6H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  168.29, 139.78, 139.60, 138.02, 131.85, 128.38, 128.06, 127.89, 127.45, 127.41, 126.60, 50.78 ppm; HRMS (m/z) C<sub>28</sub>H<sub>22</sub>O<sub>4</sub>Na<sup>+</sup> 445.1411.

Dimethyl-4,4"'-dichloro-[1,1':3',1"':3",1"'-quaterphenyl]-2',2"-

dicarboxylate (2i): yield = 44.8 mg (61%), white solid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.47 (t, *J* = 7.7 Hz, 2H), 7.39 – 7.30 (m, 12H), 3.35 (s, 6H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  169.01, 139.41, 139.19, 139.05, 133.86, 132.77, 129.82, 129.34, 129.22, 128.69, 128.66, 51.90 ppm.

**Dimethyl 4,4'-diacetyl-[1,1'-biphenyl]-2,2'-dicarboxylate (2j):** yield = 29.7 mg (56%), colourless liquid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.62 (d, J = 1.4 Hz, 2H), 8.15 (dd, J = 8.0, 1.7 Hz, 2H), 7.29 (d, J = 8.0 Hz, 2H), 3.68 (s, 6H), 2.68 (s, 6H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  196.92, 166.28, 147.35, 136.37, 131.27, 130.38, 130.27, 129.43, 52.34, 26.82 ppm; HRMS (m/z) C<sub>20</sub>H<sub>18</sub>O<sub>6</sub>Na<sup>+</sup> 377.0990.





















**Dimethyl 5,5'-dimethoxy-[1,1'-biphenyl]-2,2'-dicarboxylate (2k):** yield = 32.7 mg (66%), colourless liquid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.02 (d, J = 8.8 Hz, 2H), 6.91 (dd, J = 8.8, 2.7 Hz, 2H), 6.69 (d, J = 2.6 Hz, 2H), 3.84 (s, 6H), 3.61 (s, 6H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  166.79, 161.99, 146.05, 132.19, 121.57, 115.56, 112.38, 55.52, 51.70 ppm.

**Dimethyl 5,5'-bis(benzyloxy)-[1,1'-biphenyl]-2,2'-dicarboxylate (2l):** yield = 46.2 mg (64%), colourless liquid; <sup>1</sup>**H NMR** (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.03 (d, J = 8.7 Hz, 2H), 7.59 – 7.32 (m, 10H), 7.00 (dd, J = 8.8, 2.6 Hz, 2H), 6.80 (d, J = 2.6 Hz, 2H), 5.10 (s, 4H), 3.62 (s, 6H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  165.78, 160.16, 144.98, 135.38, 131.19, 127.74, 127.28, 126.71, 120.88, 115.36, 112.29, 69.17, 50.68 ppm.

**Dimethyl 5,5'-bis(methoxymethoxy)-[1,1'-biphenyl]-2,2'-dicarboxylate (2m):** yield = 22.2 mg (38%), colourless liquid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.99 (d, J = 8.7 Hz, 2H), 7.05 (dd, J = 8.7, 2.6 Hz, 2H), 6.82 (d, J = 2.5 Hz, 2H), 5.24 – 5.18 (m, 4H), 3.61 (s, 6H), 3.49 (s, 6H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  166.82, 159.69, 145.67, 132.08, 122.75, 117.78, 114.29, 94.39, 56.44, 51.72 ppm; HRMS (m/z) C<sub>20</sub>H<sub>22</sub>O<sub>8</sub>Na<sup>+</sup>413.1201.

**Dimethyl 5,5'-dichloro-[1,1'-biphenyl]-2,2'-dicarboxylate (2n):** yield = 25.3 mg (50%), colourless liquid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.98 (d, J = 8.4 Hz, 2H), 7.42 (dd, J = 8.4, 2.1 Hz, 2H), 7.20 (d, J = 2.1 Hz, 2H), 3.65 (s, 6H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  166.24, 143.91, 138.07, 131.55, 130.14, 127.92, 127.67, 52.19 ppm.

**Dimethyl 5,5'-dibromo-[1,1'-biphenyl]-2,2'-dicarboxylate (20):** yield = 37.8 mg (59%), white solid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.90 (d, J = 8.4 Hz, 2H), 7.58 (dd, J = 8.4, 2.0 Hz, 2H), 7.36 (d, J = 2.0 Hz, 2H), 3.65 (s, 6H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  166.37, 143.85, 133.02, 131.59, 130.94, 128.16, 126.59, 52.21 ppm.

**Dimethyl 3,3',5,5'-tetramethyl-[1,1'-biphenyl]-2,2'-dicarboxylate (2p):** yield = 25.9 mg (53%), colourless liquid; <sup>1</sup>**H NMR** (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.00 (s, 2H), 6.88 (s, 2H), 3.54 (s, 6H), 2.37 (s, 6H), 2.31 (s, 6H) ppm; <sup>13</sup>**C NMR** (101 MHz, CDCl<sub>3</sub>)  $\delta$  169.80, 139.99, 139.11, 136.00, 130.32, 129.97, 127.95, 51.66, 21.29, 20.27 ppm; **HRMS** (m/z) C<sub>20</sub>H<sub>22</sub>O<sub>4</sub>Na<sup>+</sup> 349.1409.

**Dimethyl 3,3',4,4'-tetramethyl-[1,1'-biphenyl]-2,2'-dicarboxylate (2q):** yield = 23.0 mg (47%), colourless liquid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.15 (d, J = 7.8 Hz, 2H), 6.99 (d, J = 7.8 Hz, 2H), 3.57 (s, 6H), 2.30 (s, 6H), 2.25 (s, 6H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  170.39, 136.39, 136.23, 134.08, 133.53, 130.26, 127.04, 51.84, 20.22, 17.01 ppm.

**Dimethyl** [2,2'-binaphthalene]-1,1'-dicarboxylate (2r): yield = 46.6 mg (84%), colourless liquid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.10 (d, J = 8.3 Hz, 2H), 8.02 – 7.86 (m, 4H), 7.65 – 7.53 (m, 4H), 7.49 (d, J = 8.4 Hz, 2H), 3.61 (s, 6H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  169.14, 137.75, 132.63, 130.10, 129.77, 128.32, 127.64, 127.39, 126.65, 125.57, 52.19 ppm; HRMS (m/z) C<sub>24</sub>H<sub>18</sub>O<sub>4</sub>Na<sup>+</sup> 393.1109.

**Dimethyl 3-fluoro-3'-methoxy-[1,1'-biphenyl]-2,2'-dicarboxylate (3a):** yield = 25.3 mg (53%), colourless liquid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.46 – 7.30 (m, 2H), 7.19 – 7.04 (m, 2H), 6.96 (d, J = 8.3 Hz, 1H), 6.85 (d, J = 7.7 Hz, 1H), 3.88 (s, 3H), 3.66 (s, 3H), 3.59 (s, 3H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  167.60, 165.33, 159.82 (d, J = 250.0 Hz), 156.63, 141.05, 138.92, 134.24, 131.24 (d, J = 9.1 Hz), 130.38, 125.77 (d, J = 3.4 Hz), 122.90, 121.81, 115.50 (d, J = 21.8 Hz), 110.73, 56.14, 52.41, 52.15 ppm; HRMS (m/z) C<sub>17</sub>H<sub>15</sub>FO<sub>5</sub>Na<sup>+</sup> 341.0790.

**Dimethyl** 4'-acetyl-3,5-dimethyl-[1,1'-biphenyl]-2,2'-dicarboxylate (3b): yield = 31.1 mg (61%), colourless liquid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.49 (d, J = 1.9 Hz, 1H), 8.08 (dd, J = 8.0, 1.9 Hz, 1H), 7.35 (d, J = 8.0 Hz, 1H), 7.07 (s, 1H), 6.86 (s, 1H), 3.70 (s, 3H), 3.48 (s, 3H), 2.67 (s, 3H), 2.39 (s, 3H), 2.36 (s, 3H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  197.12, 169.32, 167.00, 147.11,

S13

















139.89, 139.53, 136.19, 136.10, 131.46, 130.92, 130.72, 130.16, 129.25, 127.27, 52.34, 51.70, 26.82, 21.39, 20.38 ppm; **HRMS** (m/z)  $C_{20}H_{20}O_5Na^+$ 363.1196.

**Dimethyl 3-fluoro-5'-methoxy-[1,1'-biphenyl]-2,2'-dicarboxylate (3c):** yield = 23.9 mg (50%), colourless liquid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.98 (d, *J* = 8.3 Hz, 1H), 7.43 (dd, *J* = 5.0, 2.1 Hz, 1H), 7.22 – 6.89 (m, 3H), 6.73 (s, 1H), 3.84 (s, 3H), 3.76 – 3.55 (m, 6H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  166.87, 165.46, 162.00, 159.82 (d, *J* = 252.3 Hz), 143.40, 142.88, 132.61, 131.19 (d, *J* = 9.2 Hz), 125.18, 122.36, 120.79 (d, *J* = 15.3 Hz), 116.15, 114.97 (d, *J* = 22.0 Hz), 113.38, 55.64, 52.33, 51.88 ppm; HRMS (m/z) C<sub>17</sub>H<sub>15</sub>FO<sub>5</sub>Na<sup>+</sup> 341.0788.

**Methyl 2-(3-benzoyl-2-(methoxycarbonyl)phenyl)-1-naphthoate (3d):** yield = 49.0 mg (77%), colourless liquid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.06 – 7.99 (m, 1H), 7.97 – 7.92 (m, 1H), 7.92 – 7.88 (m, 1H), 7.86 – 7.79 (m, 2H), 7.62 – 7.41 (m, 9H), 3.66 (s, 3H), 3.25 (s, 3H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 196.51, 169.00, 167.34, 141.24, 139.86, 137.10, 137.06, 133.32, 132.75, 132.65, 131.87, 130.01, 129.86, 129.80, 129.77, 128.64, 128.38, 128.33, 127.68, 127.29, 126.71, 125.56, 52.16, 52.09 ppm; HRMS (m/z) C<sub>27</sub>H<sub>20</sub>O<sub>5</sub>Na<sup>+</sup>447.1194.

**Methyl 2-(3-methoxy-2-(methoxycarbonyl)phenyl)-1-naphthoate (3e):** yield = 29.4 mg (56%), colourless liquid; <sup>1</sup>**H NMR** (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.06 – 7.97 (m, 1H), 7.92 – 7.84 (m, 2H), 7.61 – 7.50 (m, 2H), 7.44 – 7.36 (m, 2H), 7.00 – 6.90 (m, 2H), 3.90 (s, 3H), 3.68 (s, 3H), 3.52 (s, 3H) ppm; <sup>13</sup>**C NMR** (101 MHz, CDCl<sub>3</sub>)  $\delta$  169.28, 167.81, 156.78, 140.48, 136.68, 132.68, 130.36, 130.27, 130.09, 129.61, 128.30, 127.56, 127.20, 126.62, 125.59, 123.35, 122.19, 110.59, 56.20, 52.21, 52.15 ppm; **HRMS** (m/z) C<sub>21</sub>H<sub>18</sub>O<sub>5</sub>Na<sup>+</sup> 373.1043.

**Methyl 2-(5-methoxy-2-(methoxycarbonyl)phenyl)-1-naphthoate (3f):** yield = 30.5 mg (58%), colourless liquid; <sup>1</sup>**H NMR** (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.10 – 7.86 (m, 4H), 7.60 – 7.47 (m, 2H), 7.37 (d, J = 8.4 Hz, 1H), 6.95 (dd, J = 8.8, 2.6 Hz, 1H), 6.80 (d, J = 2.6 Hz, 1H), 3.83 (s, 3H), 3.61 (s, 3H), 3.57 (s, 3H) ppm; <sup>13</sup>**C NMR** (101 MHz, CDCl<sub>3</sub>)  $\delta$  169.37, 167.02, 161.92, 144.38, 138.92, 132.69, 132.49, 129.89, 129.29, 129.03, 128.34, 127.42, 127.24, 126.36, 125.42, 122.30, 116.03, 113.50, 55.65, 52.09, 51.83 ppm; **HRMS** (m/z) C<sub>21</sub>H<sub>18</sub>O<sub>5</sub>Na<sup>+</sup> 373.1041.

**2,10-dimethyldibenzo**[**c,e**]**oxepin-5(7H)-one (4a):** yield = 22.1 mg (62%), colourless liquid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.88 (d, *J* = 7.9 Hz, 1H), 7.45 (s, 1H), 7.41 (s, 1H), 7.35 – 7.30 (m, 2H), 7.25 – 7.20 (m, 1H), 5.05 – 4.87 (m, 2H), 2.49 (s, 3H), 2.45 (s, 3H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 170.70, 143.21, 140.12, 139.12, 137.54, 132.29, 132.23, 129.33 (4C), 128.59, 128.14, 69.07, 21.75, 21.58 ppm.

**Dibenzo[c,e]oxepin-5(7H)-one (4b):** yield = 21.4 mg (68%), colourless liquid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.04 – 7.94 (m, 1H), 7.72 – 7.59 (m, 3H), 7.58 – 7.49 (m, 2H), 7.49 – 7.39 (m, 2H), 5.03 (d, *J* = 23.5 Hz, 2H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  170.42, 139.13, 137.42, 134.97, 132.73, 132.10, 130.81, 130.29, 128.83 (2C), 128.72 (2C), 128.59, 69.34 ppm.

**4,8-dimethyldibenzo**[c,e]oxepin-5(7H)-one (4c): yield = 25.7 mg (72%), colourless liquid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.88 (d, J = 8.0 Hz, 1H), 7.49 – 7.38 (m, 2H), 7.37 – 7.28 (m, 2H), 7.24 – 7.19 (m, 1H), 5.10 – 4.81 (m, 2H), 2.49 (s, 3H), 2.46 (s, 3H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  170.68, 143.21, 140.14, 139.17, 137.57, 132.33, 132.26, 129.35 (2C), 129.31 (2C), 128.61, 128.20, 69.09, 21.76, 21.59 ppm; HRMS (m/z) C<sub>16</sub>H<sub>14</sub>O<sub>2</sub>Na<sup>+</sup> 261.0894.

**4,8-dimethoxydibenzo[c,e]oxepin-5(7H)-one (4d):** yield = 23.1 mg (57%), colourless liquid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.52 (t, J = 8.1 Hz, 1H), 7.43 (t, J = 8.1 Hz, 1H), 7.22 (d, J = 7.8 Hz, 1H), 7.14 (d, J = 7.8 Hz, 1H), 7.05 (d, J = 8.4 Hz, 1H), 6.95 (d, J = 8.3 Hz, 1H), 5.62 (d, J = 12.2 Hz, 1H), 4.68 (d, J = 12.1 Hz, 1H), 3.94 (s, 3H), 3.89 (s, 3H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  167.23, 158.42, 156.67, 140.75, 138.71, 132.12, 130.65, 124.33, 120.72 (2C), 111.49, 110.82, 60.89, 56.51, 56.06 ppm; HRMS (m/z) C<sub>16</sub>H<sub>14</sub>O<sub>4</sub>Na<sup>+</sup>293.0790.















**2,10-dimethoxydibenzo**[c,e]oxepin-5(7H)-one (4e): yield = 28.3 mg (70%), colourless liquid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.96 (d, J = 8.5 Hz, 1H), 7.37 (d, J = 8.2 Hz, 1H), 7.18 – 7.01 (m, 3H), 6.95 (dd, J = 8.5, 2.0 Hz, 1H), 5.16 – 4.80 (m, 2H), 3.91 (d, J = 17.6 Hz, 6H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  170.39, 162.79, 160.96, 140.55, 139.49, 134.57, 130.08, 127.74, 123.49, 114.51, 114.23, 113.84, 113.80, 68.76, 55.76, 55.70 ppm; HRMS (m/z) C<sub>16</sub>H<sub>14</sub>O<sub>4</sub>K<sup>+</sup> 309.0525.

**4,8-dimethyl-5,7-dihydrodibenzo**[c,e]oxepine (6): yield = 21.5 mg (64%), colourless liquid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.41 – 7.34 (m, 2H), 7.34 – 7.28 (m, 2H), 7.24 – 7.19 (m, 2H), 4.32 (s, 4H), 2.46 (s, 6H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  141.36, 138.77, 132.60, 129.76, 128.97, 128.25, 67.33, 21.52 ppm; HRMS (m/z) C<sub>16</sub>H<sub>17</sub>O<sup>+</sup> 225.1291.

**6H-benzo[c]chromen-6-one (7):** yield = 26.5 mg (90%), white solid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.39 (d, J = 7.8 Hz, 1H), 8.11 (d, J = 8.0 Hz, 1H), 8.05 (d, J = 7.7 Hz, 1H), 7.82 (t, J = 7.3 Hz, 1H), 7.58 (t, J = 7.5 Hz, 1H), 7.47 (t, J = 7.2 Hz, 1H), 7.41 – 7.29 (m, 2H) ppm; <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  161.28, 151.43, 134.95, 134.89, 130.69, 130.56, 128.99, 124.67, 122.88, 121.80, 121.40, 118.17, 117.90 ppm.

**Methyl 4-methyl-2-((4-methylbenzoyl)oxy)benzoate (8a):** yield = 17.4 mg (41%), colourless liquid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.10 (d, J = 8.1 Hz, 2H), 7.96 (d, J = 8.0 Hz, 1H), 7.31 (d, J = 7.9 Hz, 2H), 7.15 (d, J = 8.0 Hz, 1H), 7.04 (s, 1H), 3.71 (s, 3H), 2.45 (s, 3H), 2.42 (s, 3H) ppm; <sup>13</sup>C NMR (100MHz, CDCl<sub>3</sub>)  $\delta$  165.71, 165.22, 150.98, 145.24, 144.42, 131.98, 130.41, 129.42, 126.98, 124.65, 120.64, 52.15, 21.90, 21.59 ppm; HRMS (m/z) C<sub>17</sub>H<sub>16</sub>O<sub>4</sub>Na<sup>+</sup> 307.0931.

**Methyl 2-methyl-6-((2-methylbenzoyl)oxy)benzoate (8b):** yield = 14.5 mg (34%), colourless liquid; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  8.12 (dd, J = 8.1, 1.2 Hz, 1H), 7.48 (td, J = 7.6, 1.2 Hz, 1H), 7.39 (t, J = 7.9 Hz, 1H), 7.32 (t, J = 7.2 Hz, 2H), 7.15 (d, J = 7.8 Hz, 1H), 7.09 (d, J = 8.2 Hz, 1H), 3.75 (s, 3H), 2.66 (s, 3H), 2.44 (s, 3H) ppm; <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$  167.19, 165.55, 148.78, 141.66, 138.48, 132.95, 132.09, 131.34, 130.90, 128.25, 126.65, 126.08, 120.79, 52.29, 21.95, 20.18 ppm.

**Methyl 2-(benzoyloxy)benzoate (8c):** yield = 12.7 mg (33%), colourless liquid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.23 (dd, J = 8.0, 1.5 Hz, 2H), 8.08 (dd, J = 7.9, 1.7 Hz, 1H), 7.69 – 7.58 (m, 2H), 7.53 (t, J = 7.7 Hz, 2H), 7.42 – 7.33 (m, 1H), 7.24 (d, J = 8.0 Hz, 1H), 3.74 (s, 3H) ppm; <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  165.44, 165.09, 150.80, 133.89, 133.59, 131.96, 130.29, 129.51, 128.61, 126.12, 124.00, 123.49, 52.23 ppm.

**methyl 4-(tert-butyl)-2-((4-(tert-butyl)benzoyl)oxy)benzoate (8d):** yield = 21.0 mg (38%), colourless liquid; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.15 (d, J = 8.3 Hz, 2H), 8.00 (d, J = 8.3 Hz, 1H), 7.53 (d, J = 8.3 Hz, 2H), 7.36 (dd, J = 8.3, 1.9 Hz, 1H), 7.19 (d, J = 1.9 Hz, 1H), 3.73 (s, 3H), 1.37 (s, 9H), 1.34 (s, 9H) ppm; <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  165.67, 165.21, 158.38, 157.32, 150.95, 131.74, 130.26, 126.98, 125.72, 123.28, 121.08, 120.55, 52.19, 35.33, 35.29, 31.28, 31.11 ppm; HRMS (m/z) C<sub>23</sub>H<sub>28</sub>O<sub>4</sub>Na<sup>+</sup>391.1875.

# **Computational details:**

Density functional theory calculations were performed employing the functional M06.<sup>1</sup> In all the calculations, Pople's basis set 6-31+G(d,p) was used for oxygen and nitrogen, while for carbon, hydrogen and ruthenium Ahlrichs basis set def2-SVP<sup>2,3</sup> with effective core potential<sup>4</sup> employed for ruthenium as the parameters provided in the EMSL website.<sup>5</sup> During the optimization of geometries, Grimme's D3 dispersion correction<sup>6</sup> was also employed. The absence of imaginary frequencies while performing analytical frequency calculations on the optimized geometries confirmed the obtained geometries are local minima (stationary points) for the intermediates. For transition states, the corresponding analytical frequency calculations on the optimized geometries result in only one imaginary frequencies of the transition states show that the vibrations correspond to C-H-O stretching which is the reaction coordinate for the desired transition states. Unless otherwise stated, the reported energies in the manuscript are corrected enthalpies in kcal/mole obtained by adding electronic energy and thermal correction to enthalpy. All the calculations were performed in gas at 298.15 K using Linux version of Gaussian 16.

# **References:**

- Zhao, Y.; Truhlar, D. G. The M06 Suite of Density Functionals for Main Group Thermochemistry, Thermochemical Kinetics, Noncovalent Interactions, Excited States, and Transition Elements: Two New Functionals and Systematic Testing of Four M06-Class Functionals and 12 Other Functionals. *Theor. Chem. Acc.* 2008, *120*, 215–241.
- 2. Weigend, F.; Ahlrichs, R. Balanced Basis Sets of Split Valence, Triple Zeta Valence and Quadruple Zeta Valence Quality for H to Rn: Design and Assessment of Accuracy. *Phys. Chem. Chem. Phys.* **2005**, *7*, 3297–3305.
- 3. Weigend, F. Accurate Coulomb-Fitting Basis Sets for H to Rn. *Phys. Chem. Chem. Phys.* **2006**, *8*, 1057–1065.
- 4. Andrae, D.; Haeussermann, U.; Dolg, M.; Stoll, H.; Preuss, H. Energy-Adjusted *ab initio* Pseudopotentials for the Second and Third Row Transition Elements. *Theor. Chim. Acta*, **1990**, *77*, 123–141.
- 5. https://bse.pnl.gov/bse/portal
- 6. Grimme, S.; Antony, J.; Ehrlich, S.; Krieg, H. A Consistent and Accurate *ab initio* Parametrization of Density Functional Dispersion Correction (DFT-D) for the 94 Elements H-Pu. *J. Chem. Phys.* **2010**, *132*, 154104.

# **Cartesian coordinates**



Electronic Energy = -1323.39623772 Hartree Sum of Electronic and Thermal Enthalpies = -1322.94593 Hartree Dispersion Correction = -0.013197983 Hartree

Ru	-0.93613940	0.57367276	-0.11741750
0	-1.02802023	-0.77510242	-1.71528969
С	-1.87455694	-1.03179991	0.75828092
0	0.95653387	-0.42082964	0.34380458
С	0.08482549	2.55309012	0.15887790
С	-0.80293222	2.26714085	1.25005785
С	-2.15584844	1.93395360	1.00983879
С	-2.66908489	1.81858317	-0.31905415
С	-1.75398984	2.08942267	-1.38174391
С	-0.42008444	2.53331722	-1.15860647
С	1.53672880	2.81867635	0.47721138
С	1.67886770	4.22710339	1.04999983
С	2.48427162	2.61528815	-0.69259501
С	-4.06074749	1.34902874	-0.59183835
0	-1.75827858	-2.77976554	-2.38981493
С	-1.65843515	-1.93367980	-1.51978323
С	-2.16086337	-2.08019154	-0.12989286
0	1.50288486	-1.07000177	-1.74146839
С	1.77075470	-0.86354968	-0.48328296

С	3.16261610	-1.15284078	-0.07432421
С	3.54930610	-0.91236892	1.24923267
С	-2.82901089	-3.23273487	0.28694587
С	-3.21932985	-3.35703270	1.61594289
С	-2.92242705	-2.33303550	2.51803322
С	-2.25705521	-1.18007110	2.09581202
С	4.85919471	-1.15523414	1.64385682
С	5.78641875	-1.63805827	0.71991431
С	5.40362569	-1.88085419	-0.59844215
С	4.09336132	-1.64044237	-0.99827462
Н	1.80726183	2.09173187	1.26894899
Н	0.49864204	-0.95613484	-1.91036781
Н	-0.41022551	2.25143782	2.27223133
Н	-2.79741227	1.64744656	1.85003117
Н	-2.07922639	1.88809276	-2.40846899
Н	0.23943957	2.69924033	-2.01307537
Н	2.72136426	4.42633973	1.34542020
Н	1.39250857	4.98036995	0.29570217
Н	1.04017633	4.38164754	1.93417920
Н	2.34190461	3.38801109	-1.46779147
Н	3.52936396	2.68862504	-0.35243066
Н	2.36035990	1.63141415	-1.17394020
Н	-4.08931368	0.74092336	-1.51028975
Н	-4.43383083	0.72423927	0.23429897
Н	-4.75488392	2.19590862	-0.72307718
Н	-3.02286591	-4.01969057	-0.44933626
Н	-3.74546097	-4.25343757	1.95684591
Н	-3.21269568	-2.43467883	3.56928835
Н	-2.02548908	-0.40088324	2.83306061

Н	5.16139164	-0.97026474	2.67826088
Н	6.81739965	-1.82913326	1.03180819
Н	6.13150259	-2.26247336	-1.31947341
Н	3.77730768	-1.82822189	-2.02717621
Н	2.80432495	-0.53877942	1.95686924



Electronic Energy = -1784.94457889 Hartree Sum of Electronic and Thermal Enthalpies = -1784.236405 Hartree Dispersion Correction = -0.026768825 Hartree

Ru	-2.07149675	-0.27268652	-0.32147530
0	-0.74284800	-1.51945671	-1.35821784
С	-1.40960952	-1.49335187	1.18087794
0	-0.46952992	1.09022862	-0.02904713
С	-3.38952332	1.44087745	-0.08416184
С	-3.89557054	0.34303650	0.65542309
С	-4.07461580	-0.92497100	0.02069954
С	-3.76548195	-1.10356236	-1.35377588
С	-3.38464452	0.05622531	-2.12033468
С	-3.22227628	1.29903592	-1.50772221
С	-3.05188095	2.76536188	0.55638432
С	-2.60173449	2.65644861	2.00334702
С	-4.23438330	3.71808721	0.40894473
С	-3.80563450	-2.45474152	-1.98946135

0	0.58940082	-3.29077461	-1.21403540
С	-0.23400132	-2.50805203	-0.69724088
С	-0.62512780	-2.56808563	0.72775389
0	1.25302689	0.06516876	1.02695025
С	0.66030624	1.04901080	0.57137150
С	1.33624931	2.40207656	0.64401911
С	1.03094361	3.40790768	-0.27921690
С	-0.14306966	-3.54614008	1.59742233
С	-0.42748354	-3.45918733	2.95664738
С	-1.18503861	-2.38447962	3.42780242
С	-1.67657607	-1.41457358	2.55190957
С	1.74968911	4.60144501	-0.28557622
С	2.76058353	4.81309021	0.65150022
С	3.04471823	3.82950821	1.59950988
С	2.33984146	2.62886671	1.59038380
Н	-2.19775980	3.16724497	-0.02170761
Н	-4.04782834	0.42974948	1.73481776
Н	-4.34155888	-1.79825866	0.62571080
Н	-3.10985781	-0.07383695	-3.17248936
Н	-2.81111839	2.13948811	-2.07783526
Н	-2.24656523	3.63501034	2.36433885
Н	-3.42694666	2.34891921	2.66990399
Н	-1.77655338	1.93443754	2.11322082
Н	-5.10941657	3.33782640	0.96511879
Н	-3.99097781	4.71674167	0.80686753
Н	-4.53801649	3.83654470	-0.64437162
Н	-2.98968410	-2.56135485	-2.72087949
Н	-3.68018990	-3.24543166	-1.23328566
Н	-4.76177785	-2.62794602	-2.51201134

Н	0.46332339	-4.36123989	1.18738332
Н	-0.05738821	-4.21887898	3.65171184
Н	-1.39825279	-2.30062277	4.49923217
Н	-2.25857910	-0.57841714	2.95922542
Н	1.51851478	5.37518659	-1.02422175
Н	3.32278134	5.75181823	0.64928495
Н	3.82275532	4.00137023	2.34961680
Н	2.55462838	1.84040528	2.31763281
Н	0.23321623	3.22526620	-1.00654290
Н	1.52489499	-2.02694154	-2.02893324
Ν	2.15103154	-1.19275627	-2.06561020
С	1.63384729	0.04255630	-2.62678550
С	3.22335778	-1.24424734	-1.30241671
С	2.78834896	0.97570942	-2.91302020
Н	0.90126347	0.48052812	-1.92306855
Н	1.07158188	-0.20965958	-3.53778924
Ν	4.00173932	-0.17670496	-1.10689971
С	3.55598595	-2.54896366	-0.65268078
С	3.66227916	1.12432587	-1.68746356
Н	2.41795405	1.96724134	-3.21607465
Н	3.38823806	0.57969282	-3.75220470
С	5.07204627	-0.18263059	-0.10354461
С	3.28725628	-2.55566969	0.85463722
Н	2.93507044	-3.31828095	-1.13224441
Н	4.61408400	-2.79282581	-0.86099914
Н	3.16844941	1.74181866	-0.91387977
Н	4.60854760	1.62706355	-1.94722165
С	4.59525279	-0.41616744	1.32203617
Н	5.55169573	0.80471352	-0.18217795

S21

Η	5.84336839	-0.92233653	-0.39067383
С	4.36919154	-1.87718916	1.67584357
Н	2.30685027	-2.08359114	1.04740739
Н	3.19025581	-3.60720837	1.17030184
Н	3.65788263	0.14909585	1.47313972
Н	5.35048132	0.01208805	2.00354569
Н	4.10925015	-1.95268106	2.74498327
Н	5.32397462	-2.42897247	1.55377765



Electronic Energy = -1323.35889952 Hartree

Sum of Electronic and Thermal Enthalpies = -1322.908405 Hartree Dispersion Correction = -0.014901179 Hartree

Ru	-0.53987500	-0.53392734	0.11419815
0	-0.44770459	0.47646351	2.03094144
С	0.04775165	1.36115597	-0.41658373
С	1.50158886	-0.87896873	0.49301310
С	-2.78011861	-0.68965689	-0.11551727
С	-2.29576134	-1.71544188	0.75452215
С	-1.30948195	-2.64800547	0.36799585
С	-0.65888575	-2.49978393	-0.89099986
С	-1.08931440	-1.45919287	-1.76179985
С	-2.14315618	-0.57622757	-1.37233006
С	-3.85911382	0.25133548	0.36275045
С	-5.23328806	-0.33858592	0.05922750

С	-3.72955941	1.66159356	-0.18980355
С	0.45691324	-3.41120922	-1.28388813
0	0.29389463	2.23596581	3.11655327
С	0.05412532	1.60661026	1.96816819
С	0.32710377	2.21768792	0.67823916
0	3.42488280	0.95646589	-1.94437768
С	2.68507967	0.06114204	-1.59140873
С	2.66477964	-0.53972682	-0.23434570
0	1.86340227	-0.54189108	-2.48385081
С	0.76179301	3.54163708	0.50939386
С	0.90023145	4.06245776	-0.76559081
С	0.58332342	3.25355255	-1.86388362
С	0.17011686	1.93742079	-1.69228320
С	1.72797841	-1.47748939	1.75002261
С	3.00117738	-1.72628500	2.25616898
С	4.12919984	-1.35401218	1.52813486
С	3.95106021	-0.74997059	0.29354864
Н	-3.74216628	0.31159355	1.46337074
Н	0.72255834	3.08876796	2.94527068
Н	-2.68562846	-1.75132085	1.77913913
Н	-0.95803950	-3.40448800	1.07566816
Н	-0.58201416	-1.32168716	-2.71971269
Н	-2.40768843	0.25010895	-2.04024311
Н	-6.03964462	0.29871313	0.45757678
Н	-5.38077862	-0.42764534	-1.03165896
Н	-5.34892416	-1.34561490	0.49222913
Н	-3.90364566	1.69158788	-1.27968577
Н	-4.47943476	2.32469663	0.26996986
Н	-2.72991207	2.08520939	0.00415508

Н	0.05800678	-4.29611304	-1.80788143
Н	1.16287078	-2.90006937	-1.95425052
Н	1.01396811	-3.75991415	-0.40090937
Н	0.98641489	4.18623373	1.37048630
Н	1.24111713	5.09011475	-0.91238871
Н	0.67255178	3.66424159	-2.87478125
Н	-0.05396191	1.33844696	-2.58220729
Н	0.86908095	-1.74792993	2.37551883
Н	3.10920337	-2.20338039	3.23594224
Н	5.13507056	-1.52942659	1.91995271
Н	4.81461732	-0.43598891	-0.30093334
Н	2.02381839	-0.06958756	-3.31998047



Electronic Energy = -1784.91751645 Hartree Sum of Electronic and Thermal Enthalpies = -1784.209577 Hartree Dispersion Correction = -0.028412112 Hartree

Ru	-1.72254944	-0.45816688	0.33470825
0	-0.59251524	1.04625053	1.32994414
С	-1.22141146	0.77692029	-1.23232512
С	0.11932134	-1.44854490	0.20343063
С	-3.77461019	0.24457655	0.93260664
С	-3.14959341	-0.47164719	2.00472461
С	-2.69930316	-1.80295486	1.87731240

С	-2.70698978	-2.42097920	0.59118531
С	-3.28798101	-1.72202617	-0.50387563
С	-3.81645633	-0.40865756	-0.31929728
С	-4.24645479	1.66011554	1.16529175
С	-5.55883010	1.65231204	1.94460216
С	-4.37585253	2.48363261	-0.10547630
С	-2.12477191	-3.78431431	0.40048501
0	0.46488004	2.99079458	1.06326423
С	-0.23403561	2.04898982	0.61359635
С	-0.62130067	1.98265302	-0.81348349
0	1.08206666	-1.68077784	-3.22132818
С	0.36128573	-1.99692788	-2.29451446
С	0.80161133	-2.05483944	-0.87619621
0	-0.89739339	-2.42578467	-2.51049829
С	-0.37053364	3.04101364	-1.69146505
С	-0.70737703	2.91887717	-3.03473632
С	-1.29271114	1.72945753	-3.47964277
С	-1.54812315	0.68240093	-2.59474441
С	0.75472925	-1.59724754	1.45914057
С	1.91970221	-2.33501131	1.64785598
С	2.55910808	-2.93228409	0.55930507
С	2.00512564	-2.76173785	-0.70082775
Н	-3.46999276	2.13660952	1.79802576
Н	-2.99041176	0.05528513	2.95374987
Н	-2.21892101	-2.30817719	2.72086768
Н	-3.29469528	-2.18346562	-1.49454169
Н	-4.18357444	0.13103469	-1.19795838
Н	-5.88611067	2.67718963	2.18550675
Н	-6.35670974	1.17401368	1.34929653

Н	-5.47189516	1.09314693	2.89069021
Н	-5.18284324	2.10009357	-0.75480310
Н	-4.63254617	3.52608345	0.14142421
Н	-3.44082393	2.49651373	-0.69029383
Н	-2.88316502	-4.55857596	0.60610544
Н	-1.76787993	-3.91650101	-0.63158284
Н	-1.27478923	-3.95392260	1.08056713
Н	0.08542688	3.95508708	-1.29430834
Н	-0.52324247	3.74004328	-3.73382742
Н	-1.56311132	1.62133143	-4.53582629
Н	-2.03184306	-0.22388799	-2.97888344
Н	0.29222463	-1.12457284	2.33490661
Н	2.32901379	-2.45494200	2.65981322
Н	3.47197088	-3.52246375	0.69217765
Н	2.49153731	-3.19531911	-1.58203468
Н	-1.03034559	-2.34566633	-3.47183563
Н	1.79580910	2.22418243	1.60464719
Ν	2.66179896	1.63640431	1.74911215
С	2.81896467	0.95803502	3.01930119
С	3.45289131	1.44933805	0.71445916
С	4.26883185	0.58259463	3.22324106
Н	2.16334340	0.06790289	3.02339483
Н	2.45312902	1.62837732	3.81137036
Ν	4.53736104	0.66703690	0.78570863
С	3.08135800	2.09209145	-0.58121226
С	4.80341995	-0.11664880	1.99193384
Н	4.37974457	-0.07218112	4.10096354
Н	4.86668638	1.49091031	3.41755411
С	5.36404805	0.38600068	-0.39471943

С	2.59917947	1.04799336	-1.59777592
Н	2.28217411	2.81952301	-0.37240081
Н	3.94320366	2.65684066	-0.97935186
Н	4.34951284	-1.11973102	1.87209220
Н	5.89348544	-0.25512987	2.06797306
С	4.67232347	-0.44546396	-1.46163161
Н	6.25248214	-0.14059995	-0.01481951
Н	5.73201753	1.33939040	-0.81574160
С	3.71147252	0.33436737	-2.34490742
Н	1.95616151	0.31182062	-1.07703772
Н	1.92943337	1.54806382	-2.31772998
Н	4.14167569	-1.27855877	-0.96436336
Н	5.45262502	-0.90721024	-2.09106335
Н	3.25176015	-0.35239677	-3.07511199
Н	4.28901210	1.07874397	-2.92941583



Electronic Energy = -1323.30651951 Hartree Sum of Electronic and Thermal Enthalpies = -1322.860757 Hartree Dispersion Correction = -0.014833813 Hartree

Ru	-0.02187072	-0.52363567	-0.00740733
0	-0.59973103	0.36617040	1.89681966
С	1.88220898	-0.06221241	0.74340952
0	1.60582345	1.30031625	-1.82834463

С	-0.69178234	1.63244494	-0.58784808
Н	0.46696599	1.26733316	-0.76923376
0	-1.28813604	2.14033530	2.98702742
0	3.79120652	0.93472490	-2.15981086
С	-1.03778267	1.51484983	1.84341396
С	-1.33828397	2.14875134	0.55762646
С	2.78172211	0.96537009	-1.44913690
С	2.91833778	0.53442218	0.00444785
С	-1.82339131	-1.90934444	-0.13295056
С	-0.68668786	-2.48151271	0.54827614
С	0.57668833	-2.60135552	-0.07162578
С	0.81002006	-2.02929582	-1.36297682
С	-0.28816018	-1.36973920	-1.97408032
С	-1.60615491	-1.38511871	-1.41065294
С	-3.14126610	-1.82525986	0.59450441
С	-4.04764659	-0.71445248	0.09171930
С	-3.84247560	-3.18014875	0.53960998
С	2.14945923	-2.09171739	-2.01457674
Н	-2.88961690	-1.61037196	1.65246676
Н	1.42062800	-3.01249053	0.49168792
Н	-0.80541779	-2.80933610	1.58838398
Н	-0.10377468	-0.82155390	-2.90369500
Н	-2.40426217	-0.84263987	-1.92520518
Н	-4.40867311	-0.91756650	-0.93135053
Н	-3.53870735	0.26508122	0.08255114
Н	-4.93856764	-0.62746155	0.73296320
Н	-4.09715129	-3.44014878	-0.50241246
Н	-3.20797833	-3.98795600	0.93866880
Н	-4.77696985	-3.16481716	1.12286120

Н	2.23818689	-3.02831179	-2.59048535
Н	2.95968281	-2.07105845	-1.27006399
Н	2.30616038	-1.24673071	-2.70140041
С	-2.36302899	3.09925132	0.44627360
С	-2.76896739	3.54411318	-0.80442058
С	-2.13778208	3.04560786	-1.94624076
С	-1.10499243	2.12125837	-1.83640905
С	4.17740008	0.70523796	0.59945896
С	4.42111173	0.31282220	1.90872852
С	3.40106893	-0.29462962	2.63867668
С	2.15064116	-0.48658711	2.05442934
Н	-2.89293433	3.45819171	1.33675723
Н	-3.57986266	4.27168822	-0.89137167
Н	-2.44582033	3.39840244	-2.93469614
Н	-0.55534178	1.78932508	-2.72190949
Н	4.96540564	1.14879325	-0.01658149
Н	5.40809009	0.46527663	2.35607824
Н	3.57621964	-0.62781085	3.66704686
Н	1.36728507	-0.97390624	2.64801111
Н	-1.51607363	3.06913219	2.82338180



Electronic Energy = -1784.89611675 Hartree

Sum of Electronic and Thermal Enthalpies = -1784.189305 Hartree

Dispersion Correction = -0.028362797 Hartree

Ru	-1.71925965	0.60051347	0.14633983
0	-0.78316488	-0.10005645	-1.65231593
С	-0.30267040	2.09927348	-0.06417180
0	0.17696952	0.77728589	2.55441565
С	-0.36870082	-1.55992788	0.72368456
Н	-0.05592113	-0.50171950	0.84465728
0	0.33559193	-1.70321012	-2.73250622
0	2.28587876	0.69743628	1.75396838
С	-0.29199681	-1.27957785	-1.74450338
С	-0.42124953	-2.15701044	-0.53608118
С	1.12042009	1.16583412	1.81899910
С	0.79541317	2.24627399	0.80455133
С	-3.82480514	-0.27235282	-0.28771979
С	-3.56939647	1.07665242	-0.73437356
С	-3.20426993	2.10856156	0.17504744
С	-2.88146045	1.80869006	1.53428576
С	-2.94761577	0.43942528	1.89806776
С	-3.51373101	-0.56420645	1.03919536
С	-4.29253599	-1.29065091	-1.29611012
С	-4.21808206	-2.72472690	-0.80500296
С	-5.71173089	-0.94997633	-1.74669141
С	-2.36049354	2.84374526	2.47386424
Н	-3.61680947	-1.18300370	-2.16953451
Н	-3.04018852	3.12443052	-0.19887447
Н	-3.68746550	1.31709468	-1.79801034
Н	-2.54636847	0.14400309	2.87310502
Н	-3.54519131	-1.59747305	1.39548784
Н	-4.91288977	-2.89481553	0.03619673

Н	-3.20437645	-3.00265823	-0.47415036
Н	-4.50729528	-3.41828363	-1.60991588
Н	-6.41453357	-1.03142756	-0.89932644
Н	-5.79069842	0.07284120	-2.14812982
Н	-6.04915605	-1.64486806	-2.53209568
Н	-3.15726310	3.19920539	3.14816881
Н	-1.96132838	3.71015573	1.92396527
Н	-1.54019305	2.41624649	3.07259869
С	-0.55456261	-3.54395205	-0.62962887
С	-0.68977294	-4.30354280	0.53080579
С	-0.65323894	-3.69139266	1.78835632
С	-0.47183545	-2.31402419	1.89099660
С	1.64905380	3.34918910	0.66948185
С	1.44104800	4.30638538	-0.31981655
С	0.39618450	4.13156064	-1.22759219
С	-0.45484775	3.03331870	-1.09988104
Н	-0.56505848	-4.00841514	-1.62054383
Н	-0.81941238	-5.38763517	0.45861609
Н	-0.74400726	-4.30093565	2.69236350
Н	-0.37829796	-1.80002006	2.85315552
Н	2.48808901	3.44486447	1.36863356
Н	2.09914849	5.17849761	-0.39292299
Н	0.23873960	4.85311289	-2.03617605
Н	-1.25286986	2.89706213	-1.84142096
Н	1.86697417	-0.79752645	-2.36215218
Ν	2.60546435	-0.21042828	-1.93919969
С	3.48968817	-0.79497810	-1.15096861
Ν	4.45202824	-0.10345706	-0.54719831
С	4.41369122	1.36330986	-0.57572210

С	3.87695461	1.85413684	-1.90226484
С	2.53046191	1.22444942	-2.16735213
С	3.34685296	-2.26371119	-0.91776552
С	2.95355902	-2.57255802	0.53214664
С	4.11326286	-2.59458073	1.51254662
С	4.81796397	-1.25629203	1.66275568
С	5.41139126	-0.74029813	0.36302942
Н	5.43921948	1.72514168	-0.40738779
Н	3.77576296	1.68934950	0.26909926
Н	4.58804480	1.60320488	-2.71039587
Н	3.77310681	2.94960690	-1.87782100
Н	2.18626153	1.38557381	-3.20020992
Н	1.75895876	1.64435301	-1.50062644
Н	4.28855814	-2.77716811	-1.18558363
Н	2.57293783	-2.63440311	-1.60926206
Н	2.43713820	-3.54724748	0.54268280
Н	2.21538099	-1.82153774	0.86803402
Н	4.84603384	-3.36535756	1.19571551
Н	3.73714750	-2.91766704	2.49769572
Н	5.64154595	-1.35742753	2.39035565
Н	4.11418503	-0.50202879	2.06022537
Н	5.93488425	-1.54916215	-0.18111124
Н	6.17228676	0.02671748	0.56922285

NMR Spectra of Synthesized Compounds















210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 f1 (ppm)





**S**41







210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 f1 (ppm)















 $\begin{array}{c} < 8,00 \\ < 7,06 \\ 7,06 \\ 7,06 \\ 7,06 \\ 7,06 \\ 7,06 \\ 7,06 \\ 7,06 \\ 7,06 \\ 7,06 \\ 7,06 \\ 7,06 \\ 7,06 \\ 7,06 \\ 7,10 \\ 6,82 \\ 7,10 \\ 6,82 \\ 7,10 \\ 6,82 \\ 7,10 \\ 6,82 \\ 7,10 \\ 6,82 \\ 7,10$ 



210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 f1 (ppm)



S47





210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 f1 (ppm)















#### - 804 - 804 - 801







#### 



S57



S58







#### C 213 C











**S66** 



