

Supporting Information for:

Selective Synthesis of Site-Differentiated Fe₄S₄ and Fe₆S₆ Clusters

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NMR Spectra

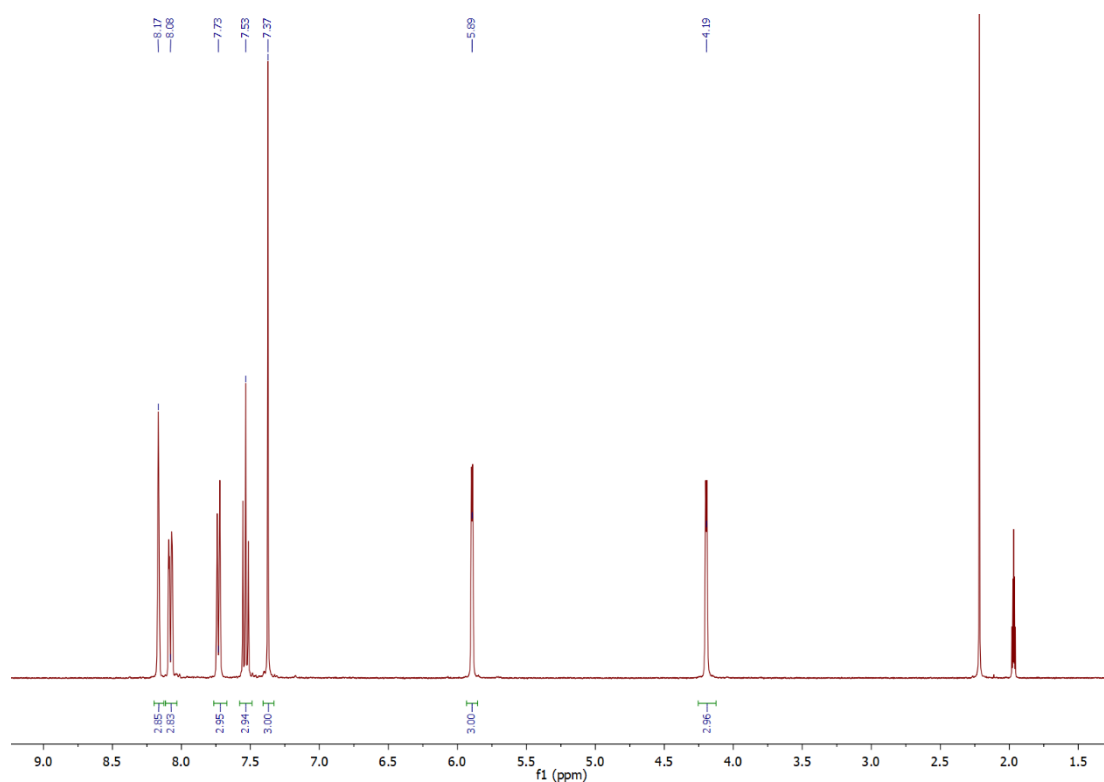


Figure S1. ¹H NMR spectrum of **3** recorded in CD₃CN at 400 MHz.

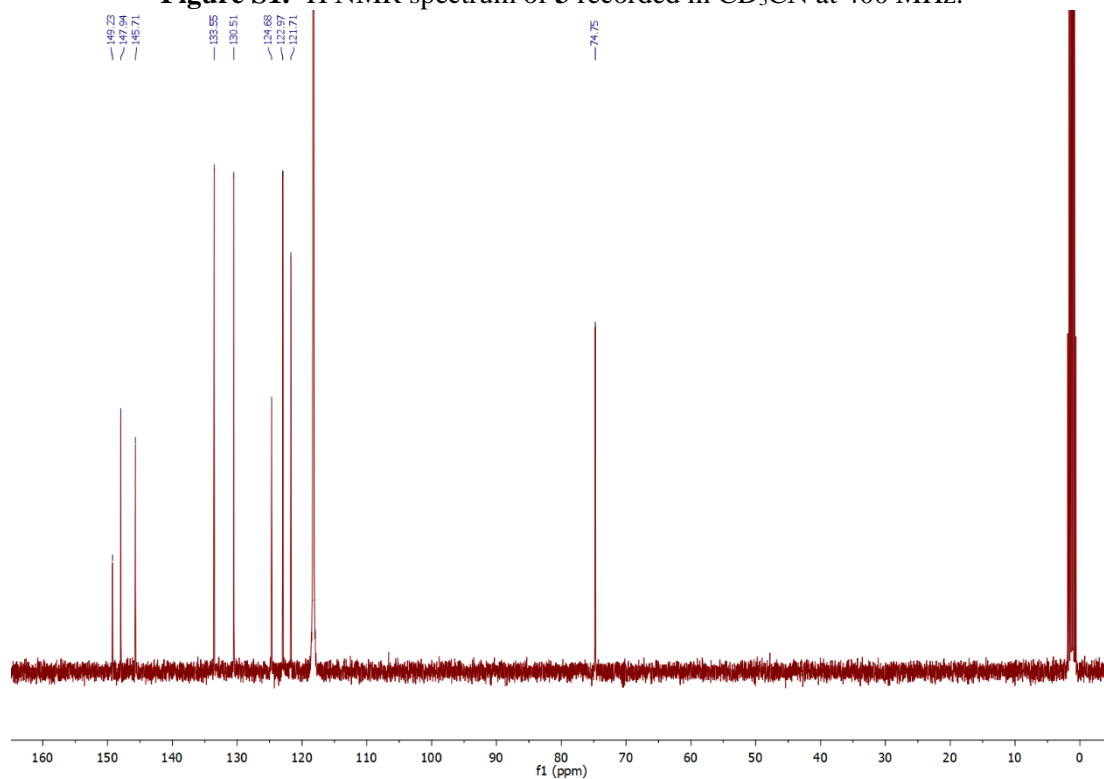


Figure S2. ¹³C{¹H} NMR spectrum of **3** recorded in CD₃CN at 100.5 MHz.

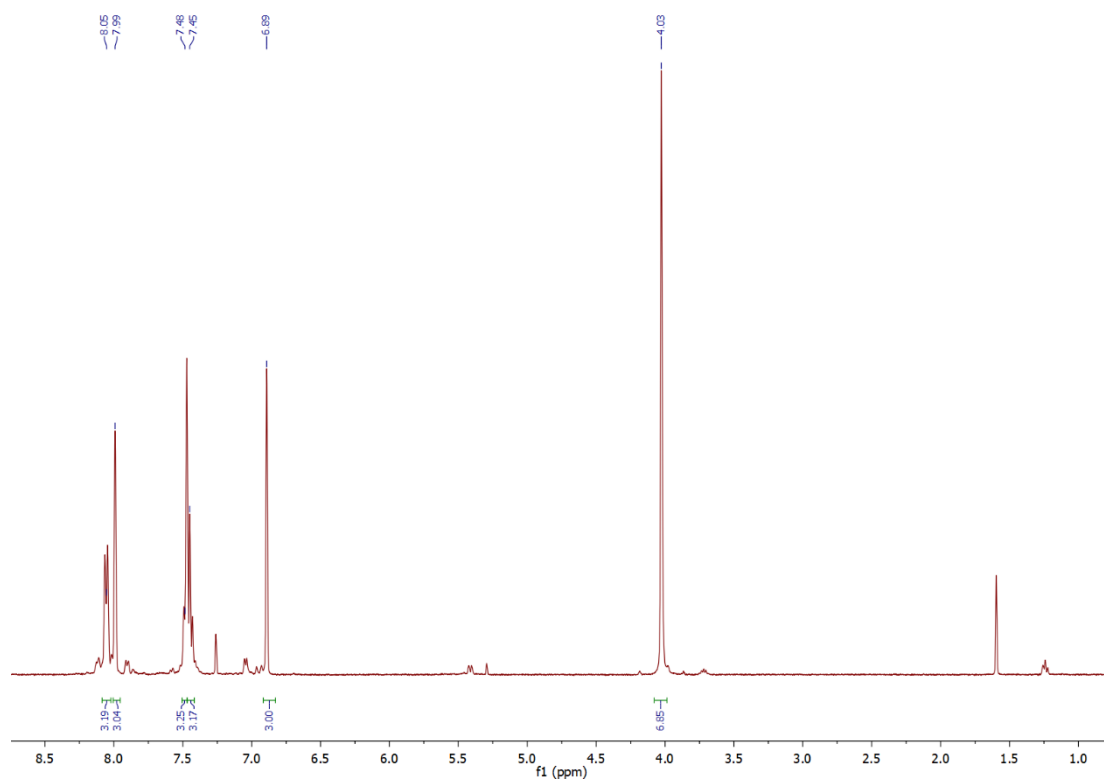


Figure S3. ¹H NMR spectrum of **4** recorded in CDCl₃ at 400 MHz.

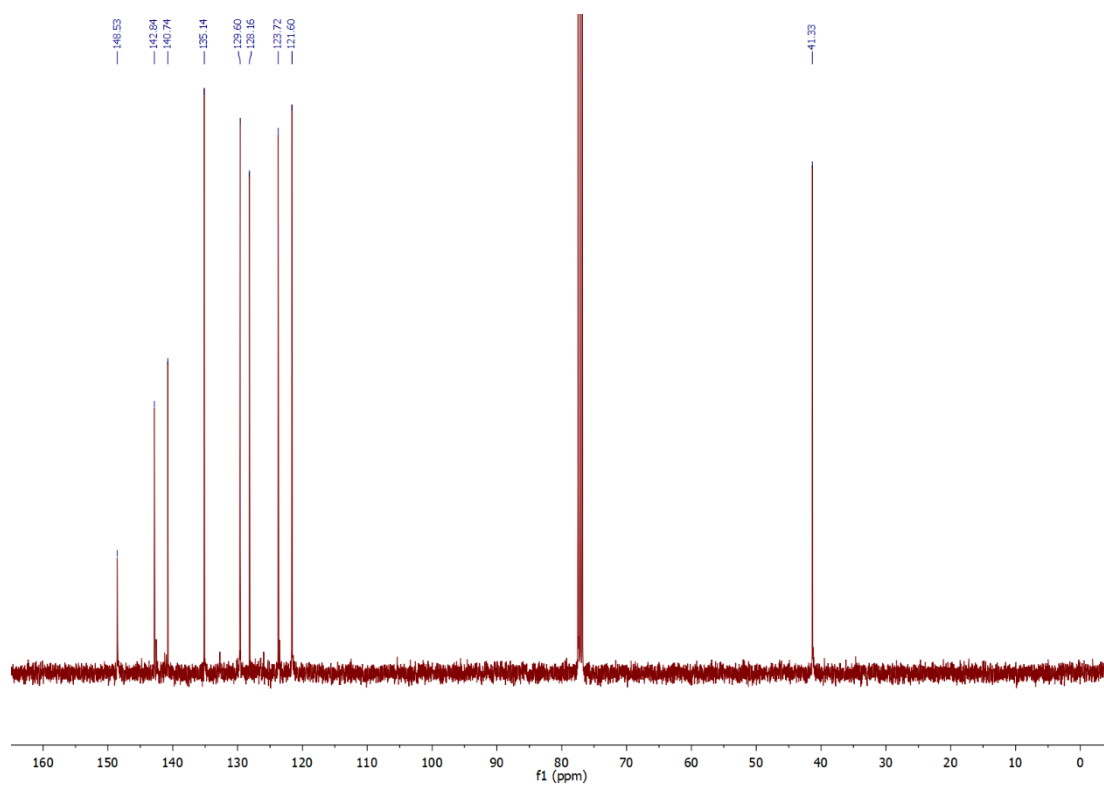


Figure S4. ¹³C{¹H} NMR spectrum of **4** recorded in CDCl₃ at 100.5 MHz.

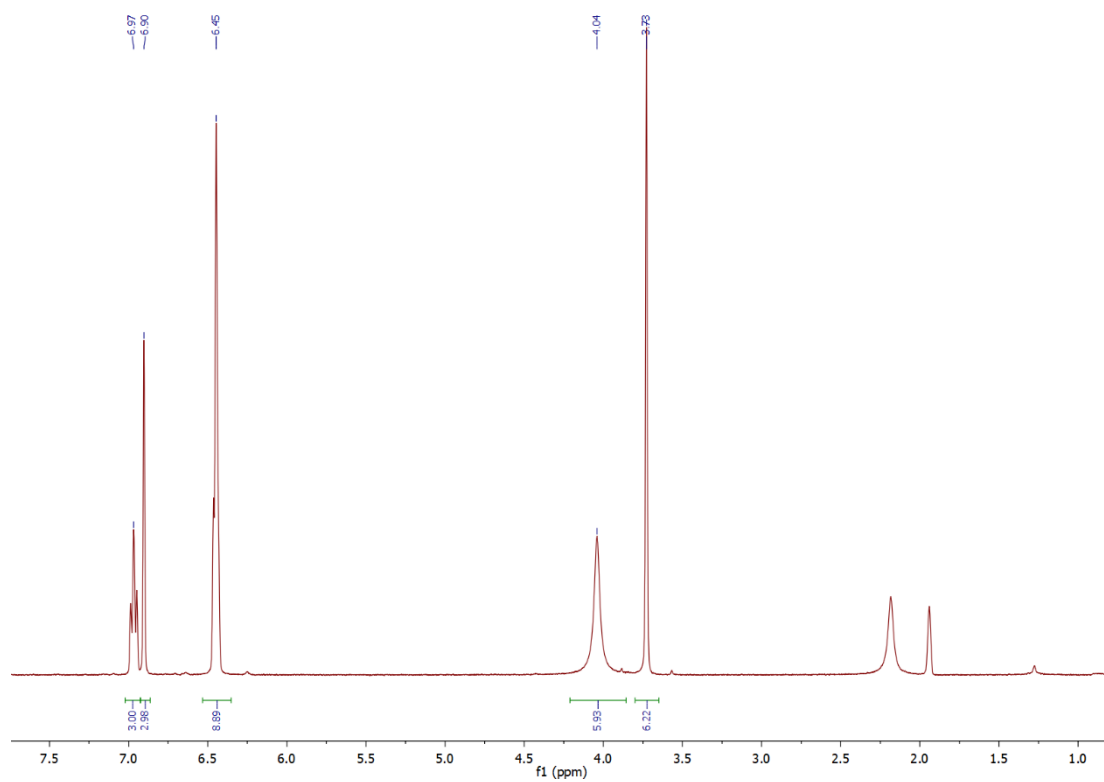


Figure S5. ¹H NMR spectrum of **5** recorded in CD₃CN at 400 MHz.

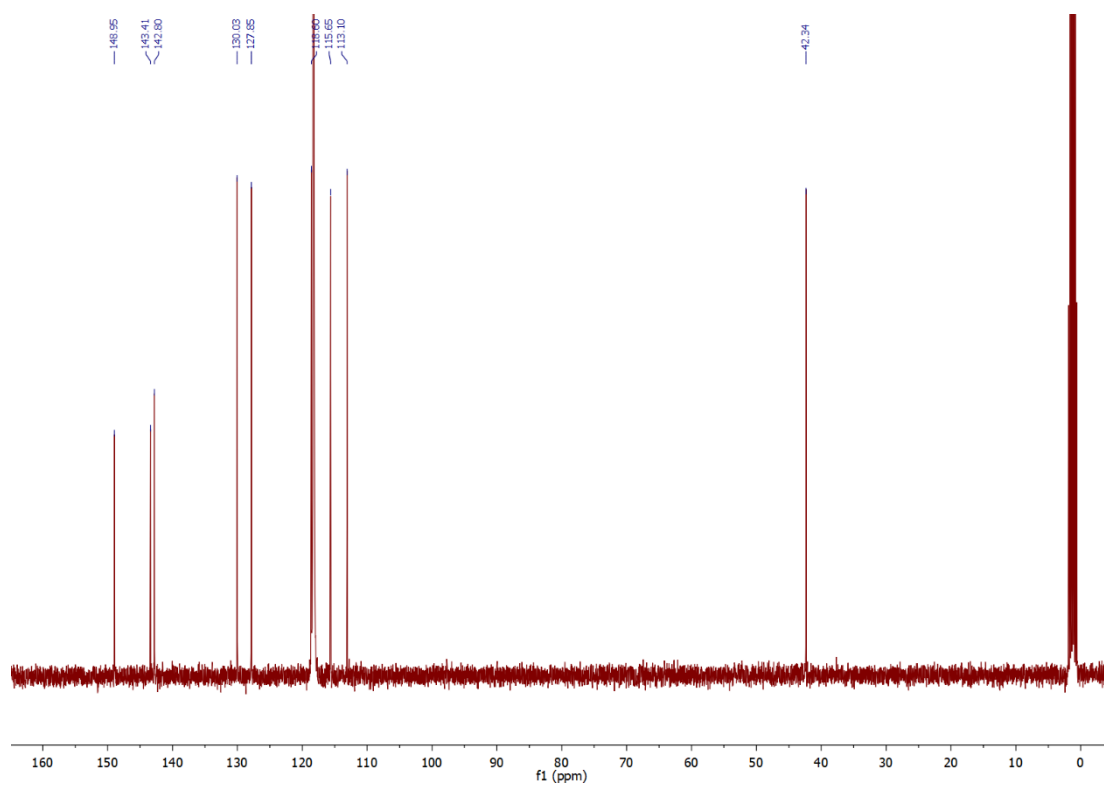


Figure S6. ¹³C{¹H} NMR spectrum of **5** recorded in CD₃CN at 100.5 MHz.

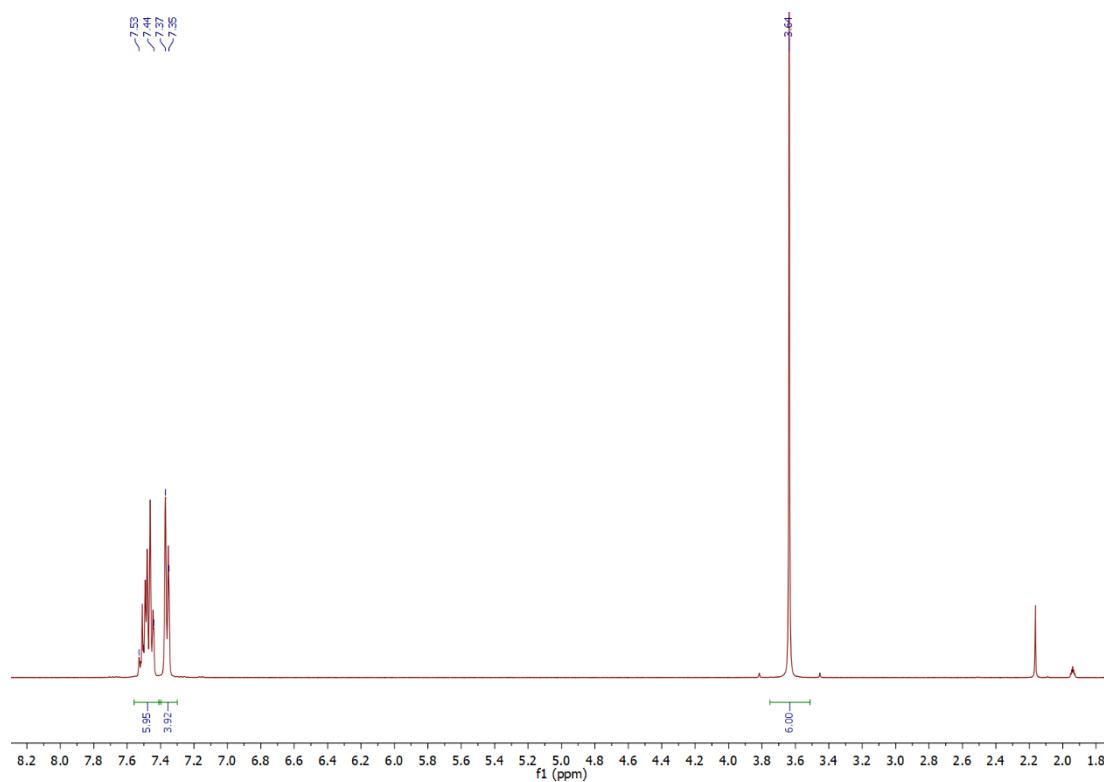


Figure S7. ¹H NMR spectrum of **6** recorded in CD₃CN at 400 MHz.

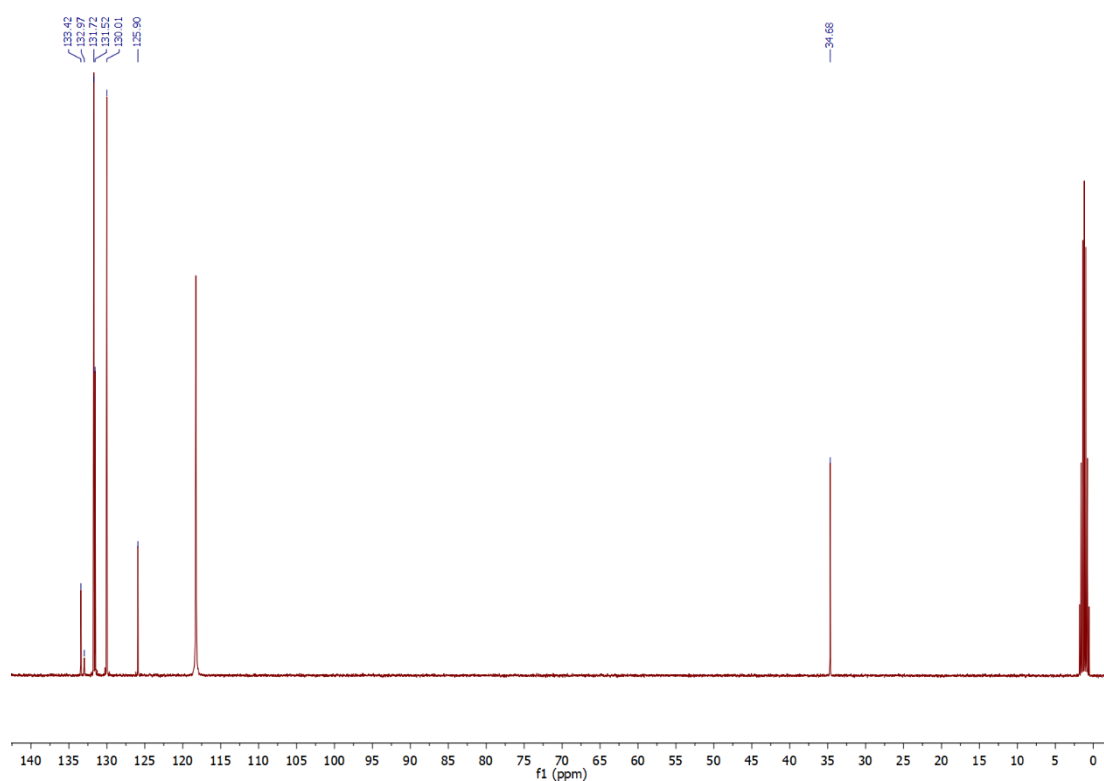


Figure S8. ¹³C{¹H} NMR spectrum of **6** recorded in CD₃CN at 100.5 MHz.

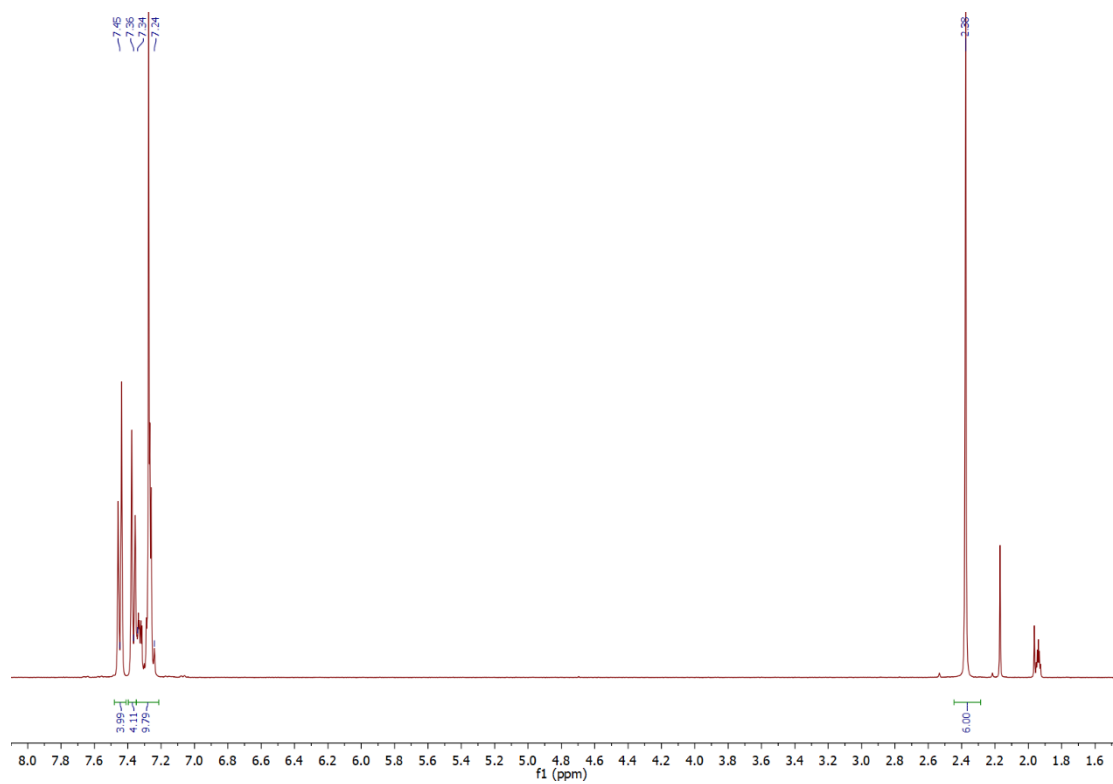


Figure S9. ¹H NMR spectrum of **7** recorded in CD₃CN at 400 MHz.

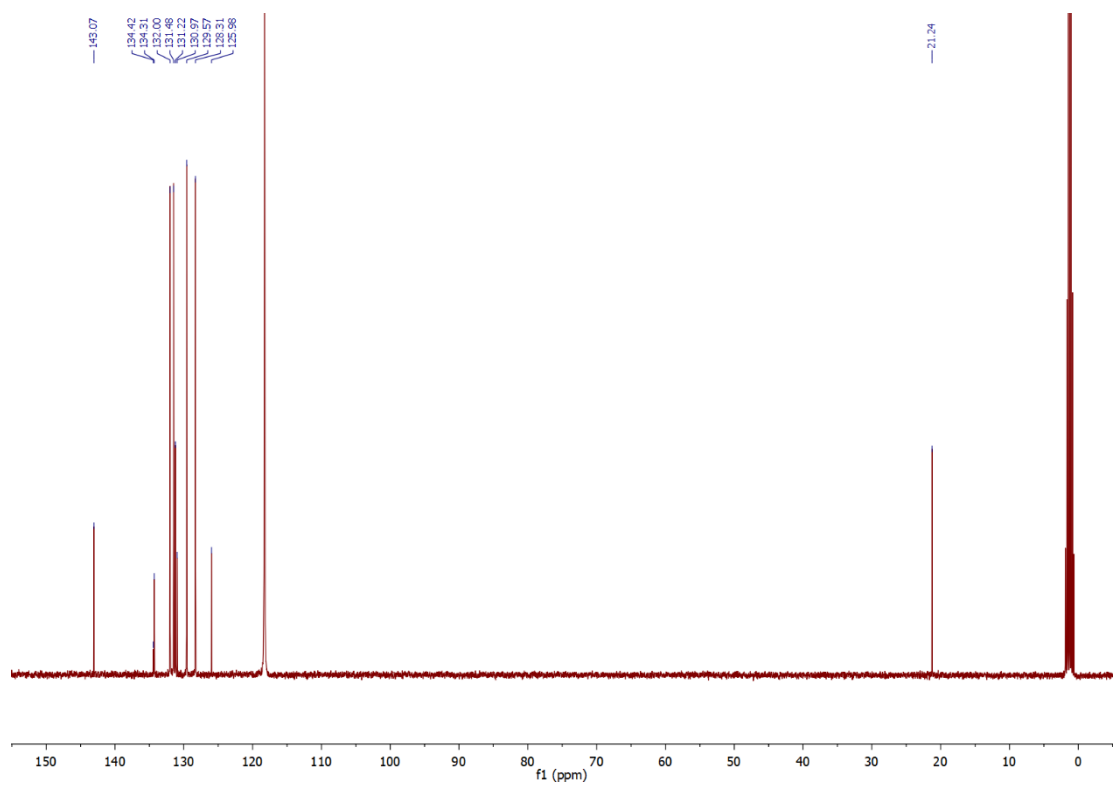


Figure S10. ¹³C{¹H} NMR spectrum of **7** recorded in CD₃CN at 100.5 MHz.

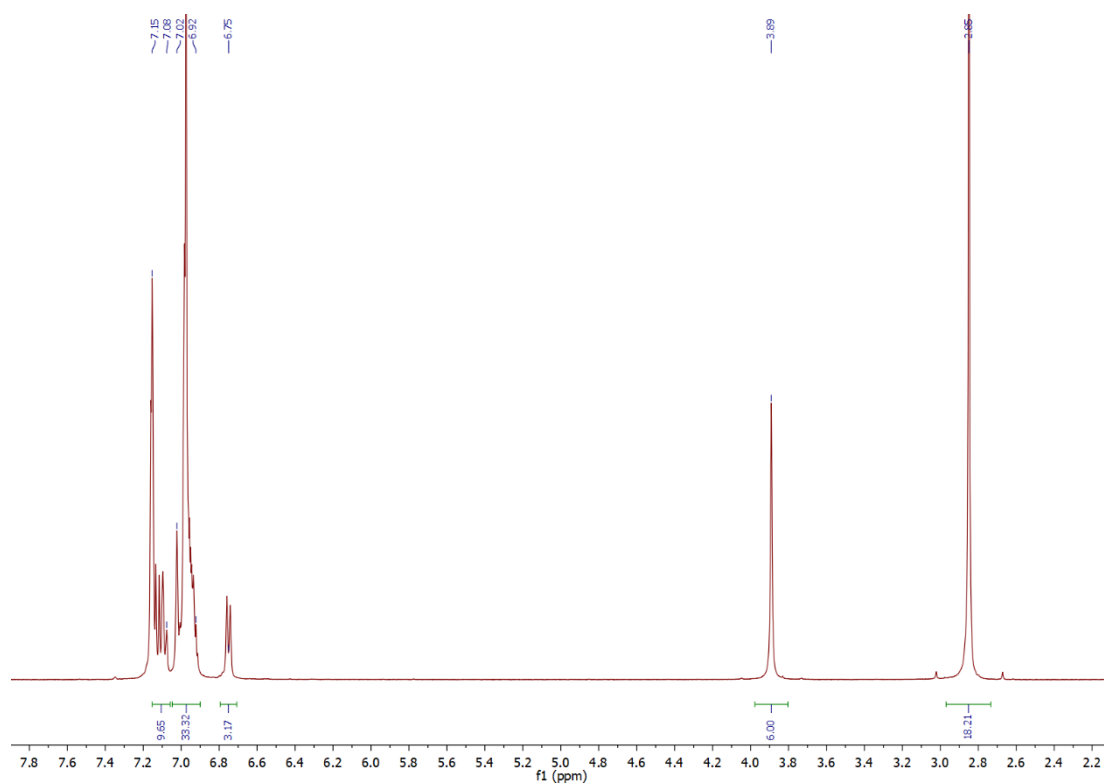


Figure S11. ¹H NMR spectrum of L(NIm^{Me})₃ (**8**) recorded in C₆D₆ at 400 MHz.

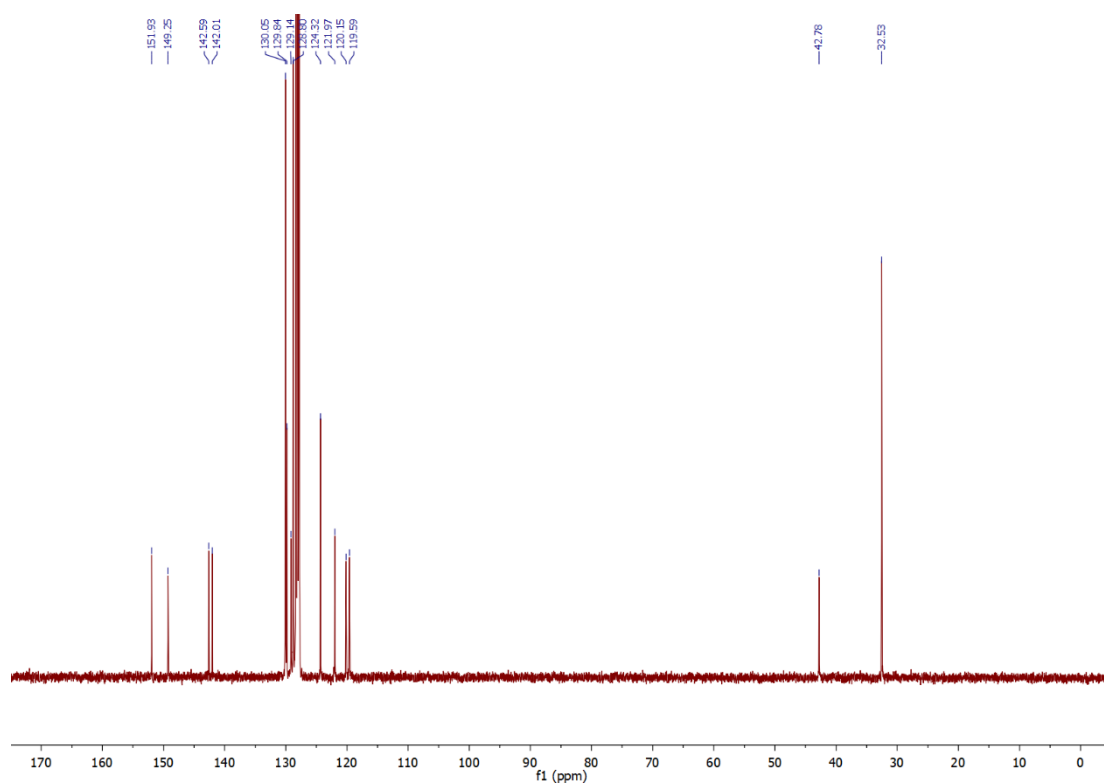


Figure S12. ¹³C{¹H} NMR spectrum of L(NIm^{Me})₃ (**8**) recorded in C₆D₆ at 100.5 MHz.

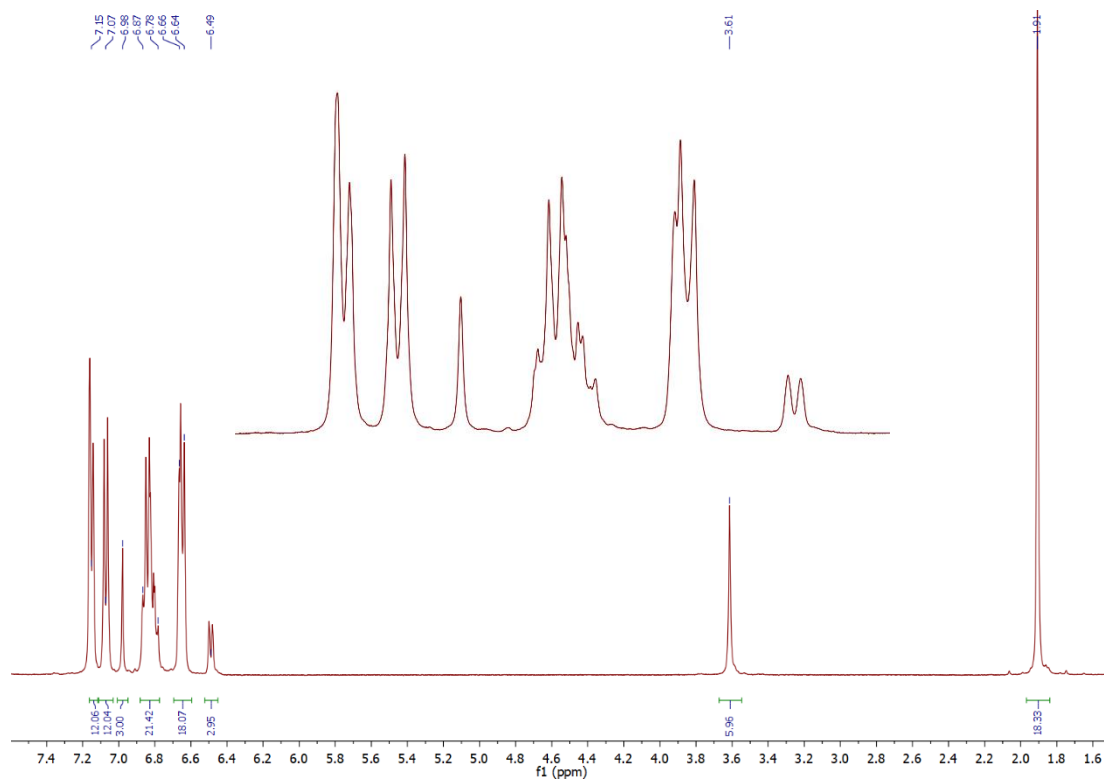


Figure S13. ¹H NMR spectrum of L(NIm^{Tol})₃ (**9**) recorded in C₆D₆ at 400 MHz (aromatic region inset).

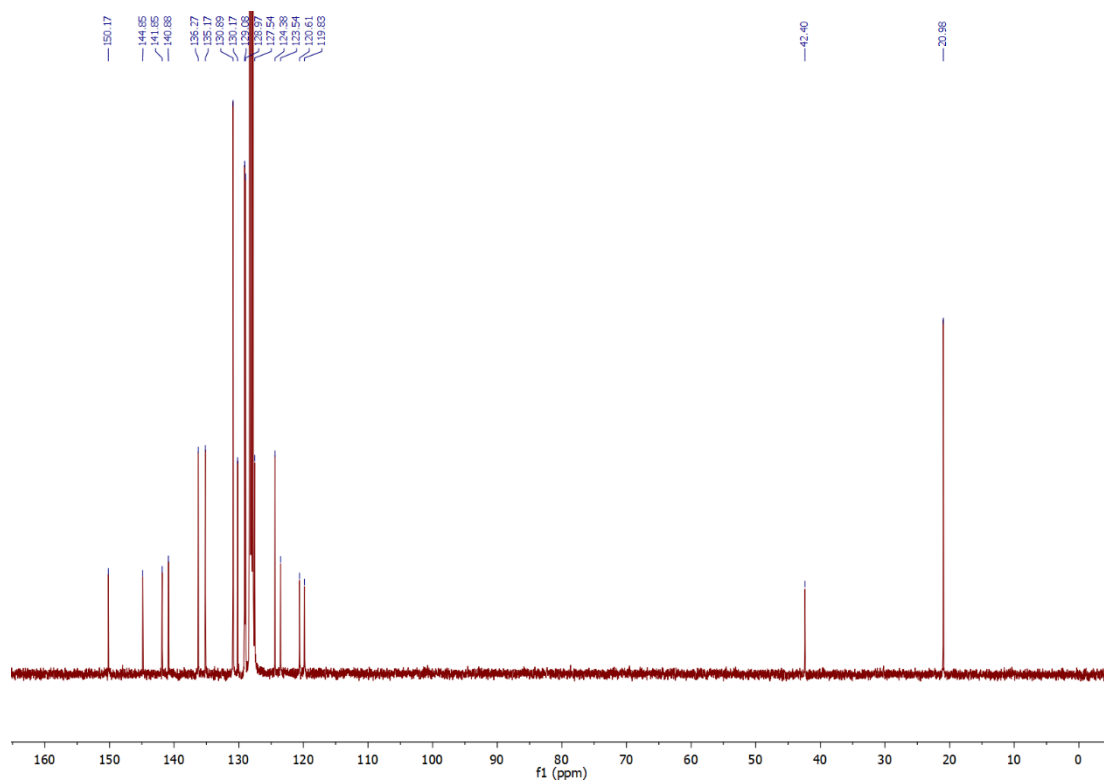


Figure S14. ¹³C{¹H} NMR spectrum of L(NIm^{Tol})₃ (**9**) recorded in C₆D₆ at 100.5 MHz.

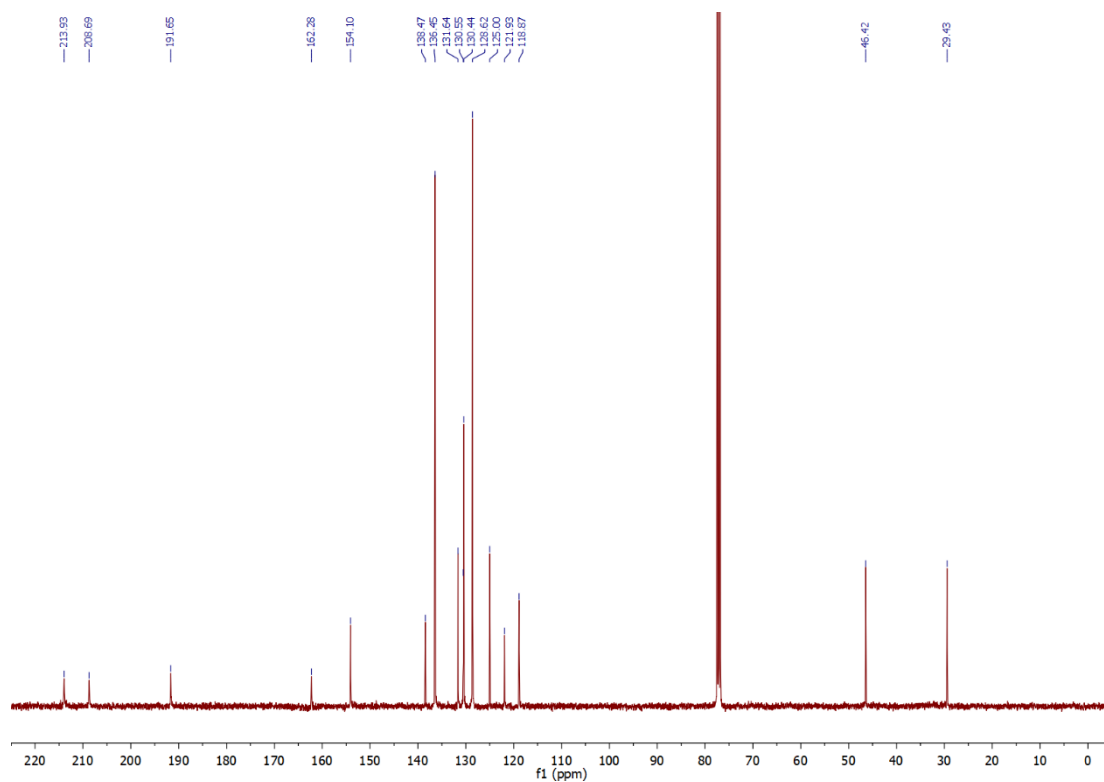


Figure S15. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of $\text{L}(\text{NIm}^{\text{Me}})_3\text{Fe}_6\text{S}_6\text{Cl}_3$ (**11**) recorded in CDCl_3 at 100.5 MHz.

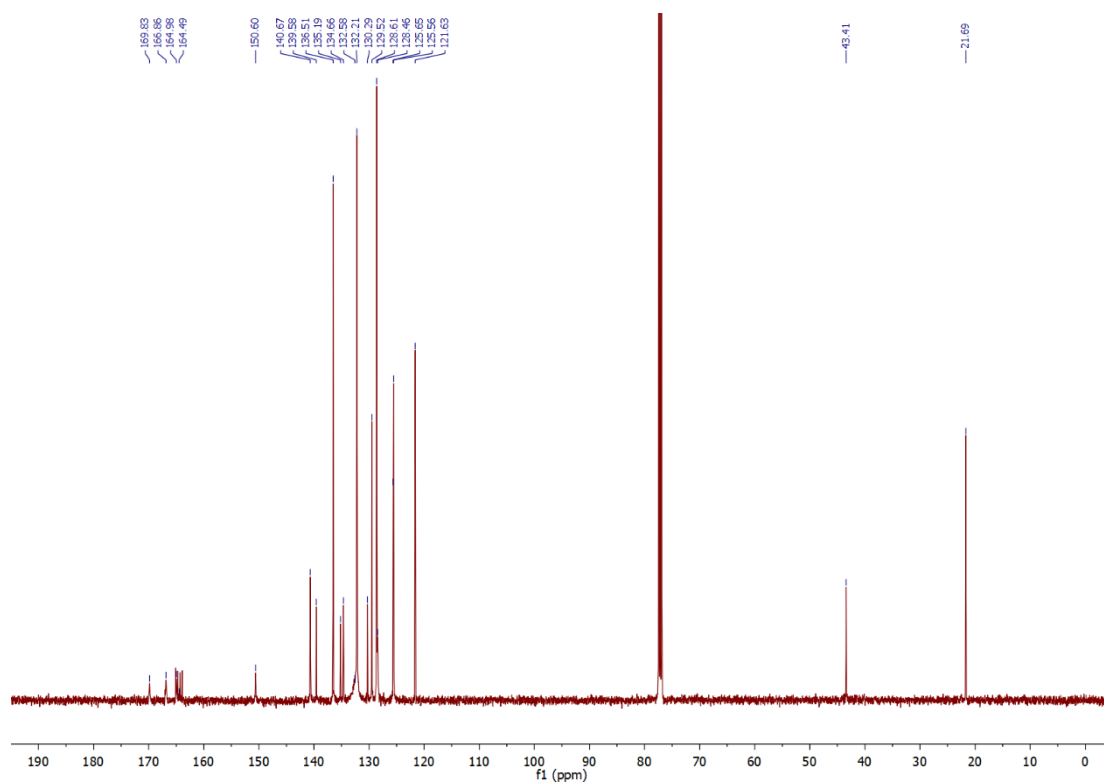


Figure S16. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of $[(\text{L}(\text{NIm}^{\text{Tol}})_3)\text{Fe}_4\text{S}_4\text{Cl}][\text{BPh}_4]$ (**12**) recorded in CDCl_3 at 100.5 MHz.

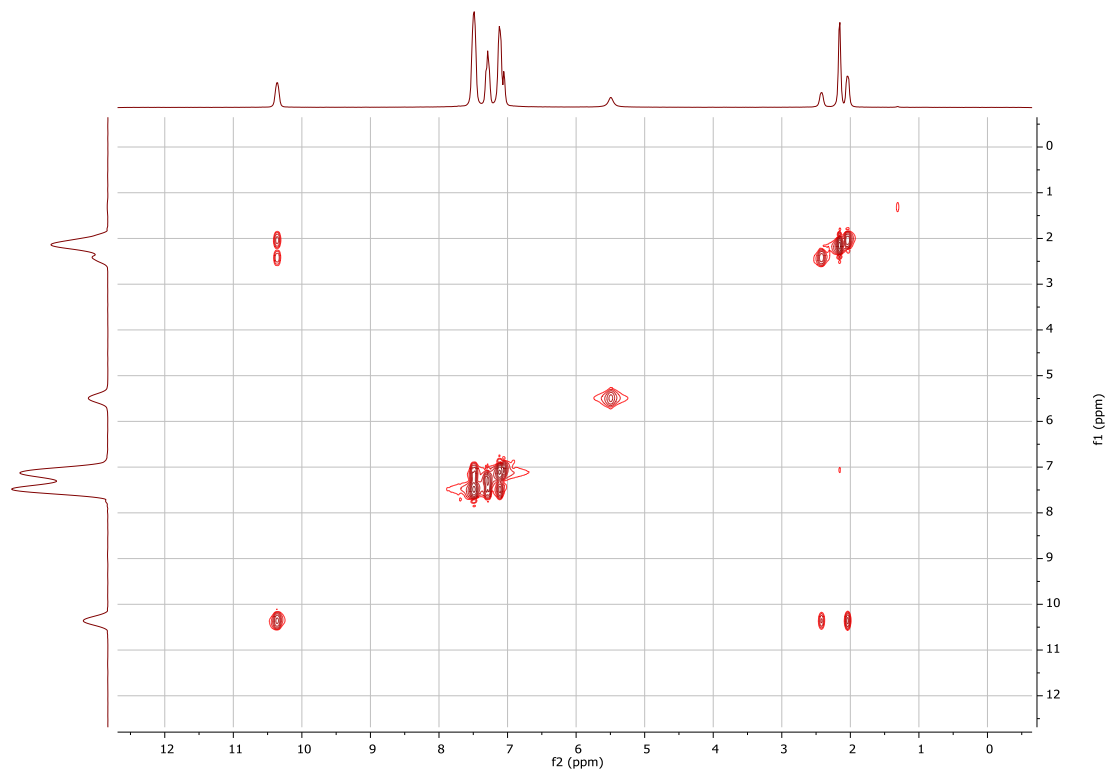


Figure S17. ^1H - ^1H COSY spectrum of $(\text{L}(\text{NIm}^{\text{Me}})_3)\text{Fe}_6\text{S}_6\text{Cl}_3$ (**11**).

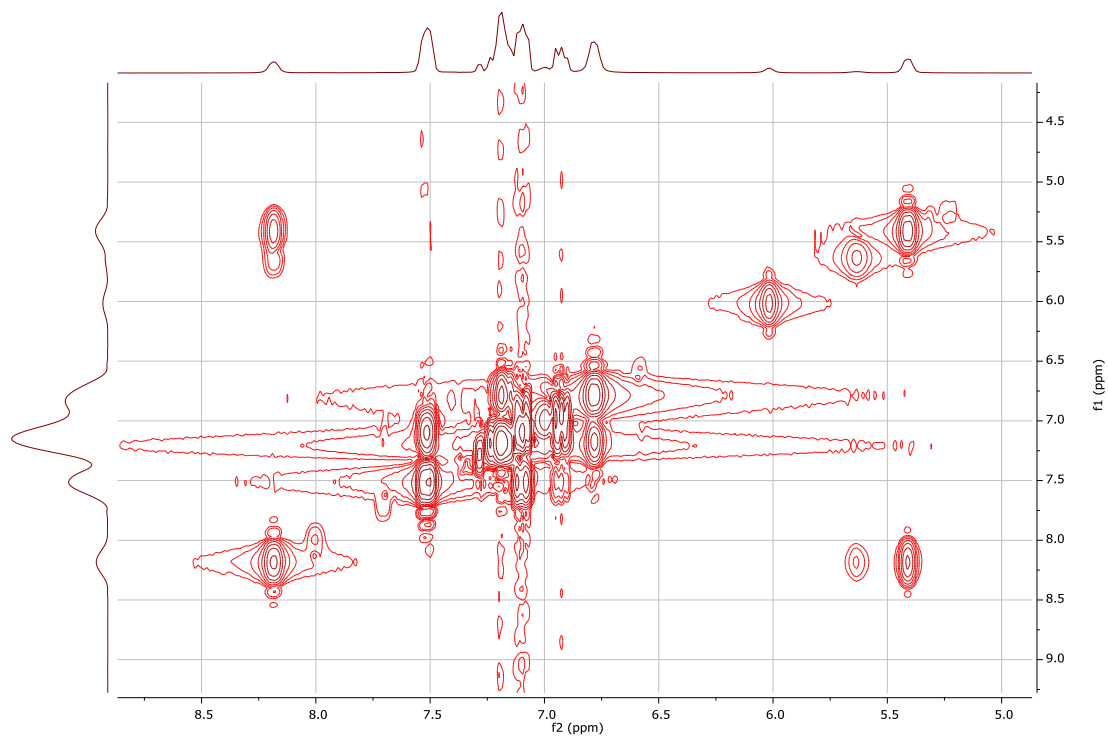


Figure S18. ^1H - ^1H COSY spectrum of $[(\text{L}(\text{NIm}^{\text{Tol}})_3)\text{Fe}_4\text{S}_4\text{Cl}][\text{BPh}_4]$ (**12**) in the aromatic region.

FTIR Spectra

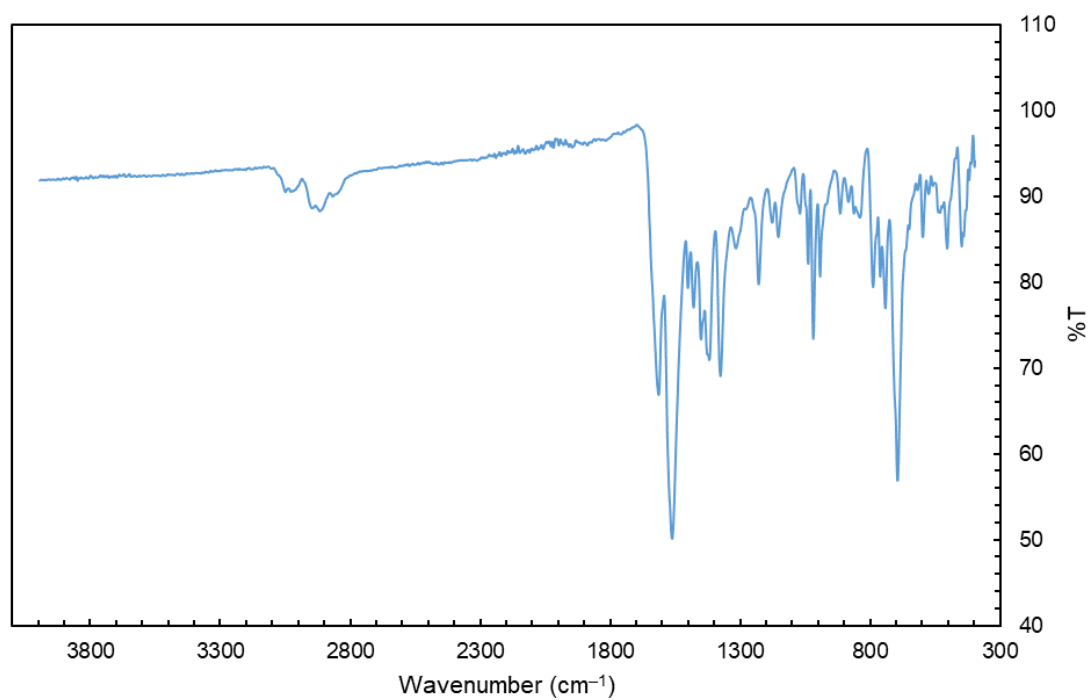


Figure S19. FTIR spectrum of L(NIm^{Me})₃ (**8**).

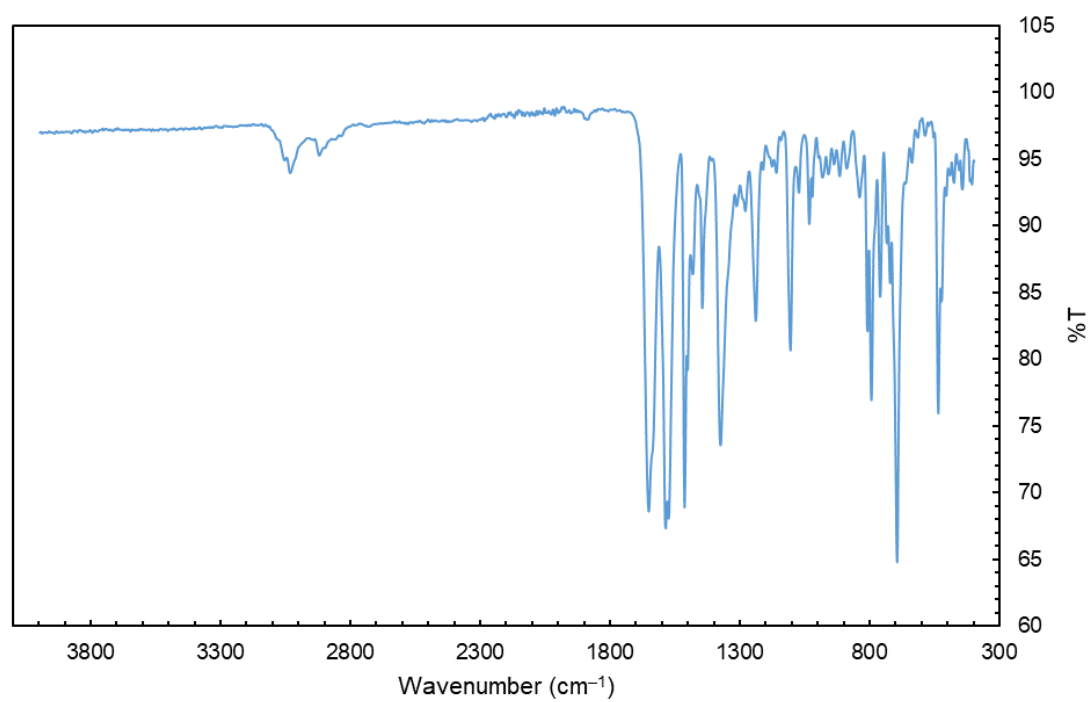


Figure S20. FTIR spectrum of L(NIm^{Tol})₃ (**9**).

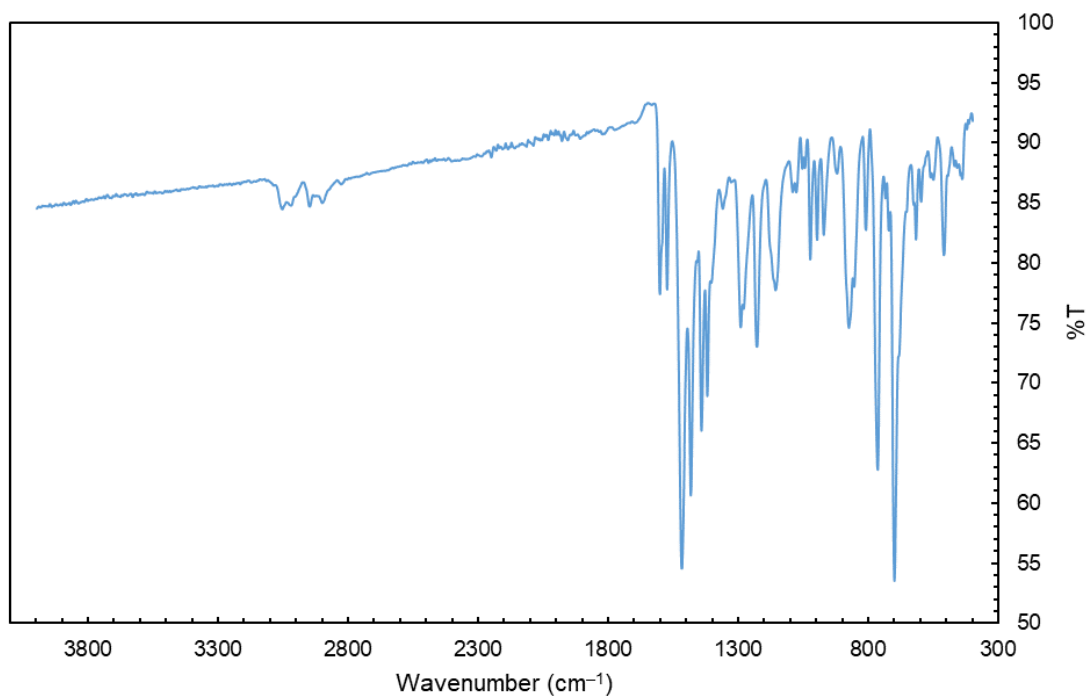


Figure S21. FTIR spectrum of $L(NIm^{Me})_3Fe_6S_6Cl_3$ (**11**).

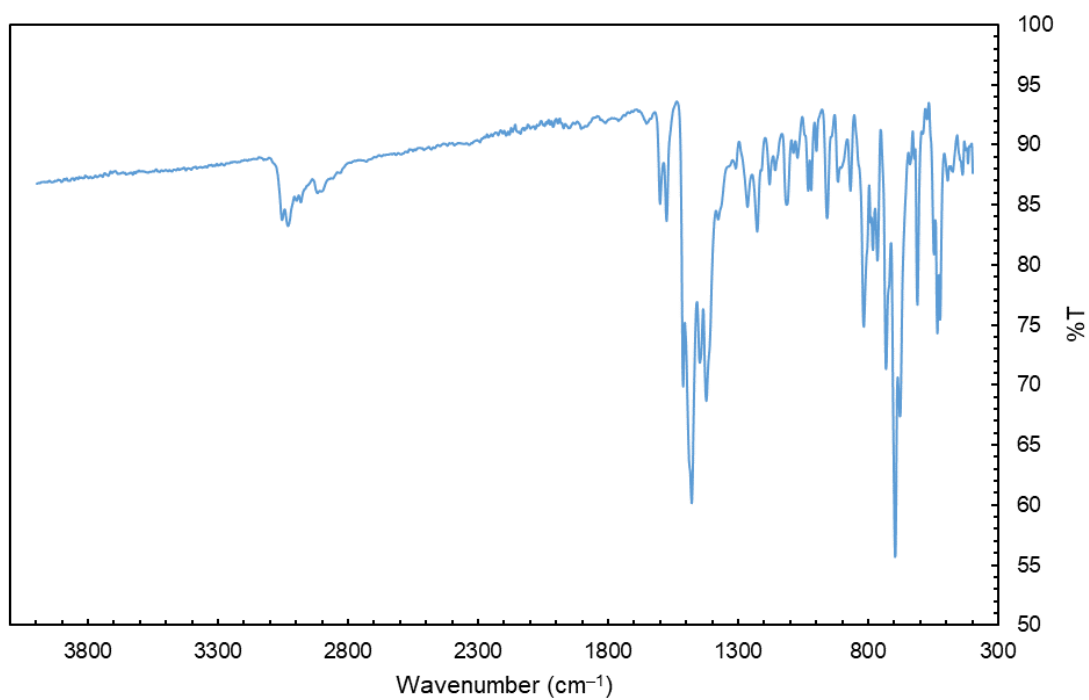


Figure S22. FTIR spectrum of $[(L(NIm^{Tol})_3)Fe_4S_4Cl][BPh_4]$ (**12**).

Additional Spectroscopy

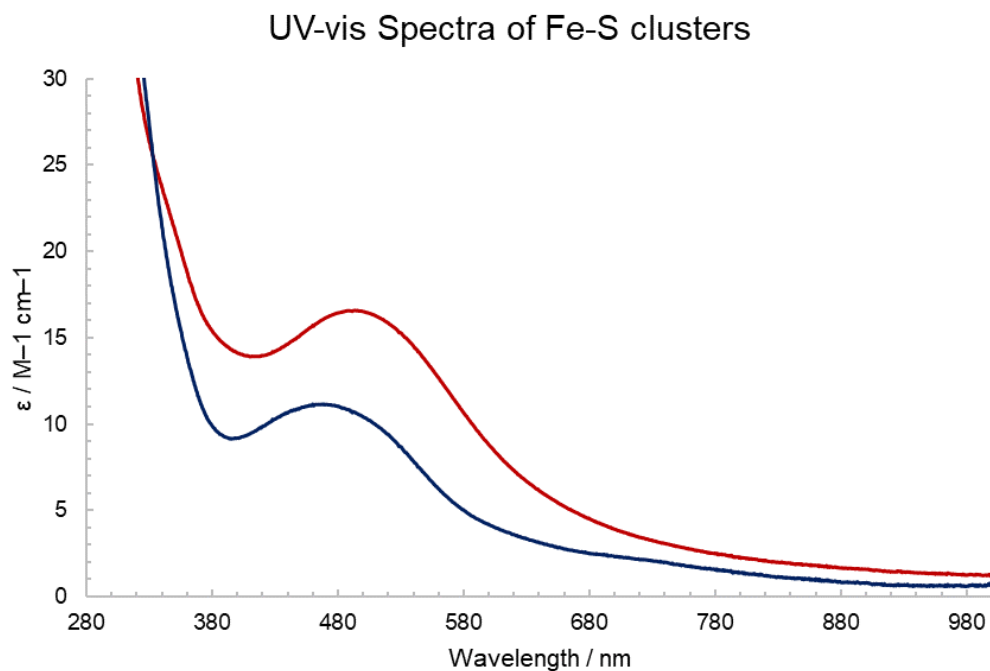


Figure S23. UV-Vis spectra of $(\text{L}(\text{NIm}^{\text{Me}})_3)\text{Fe}_6\text{S}_6\text{Cl}_3$ (**11**; red trace) and $[(\text{L}(\text{NIm}^{\text{Me}})_3)\text{Fe}_4\text{S}_4\text{Cl}_4][\text{BPh}_4]$ (**12**; blue trace) (*ca.* 1mM in THF).

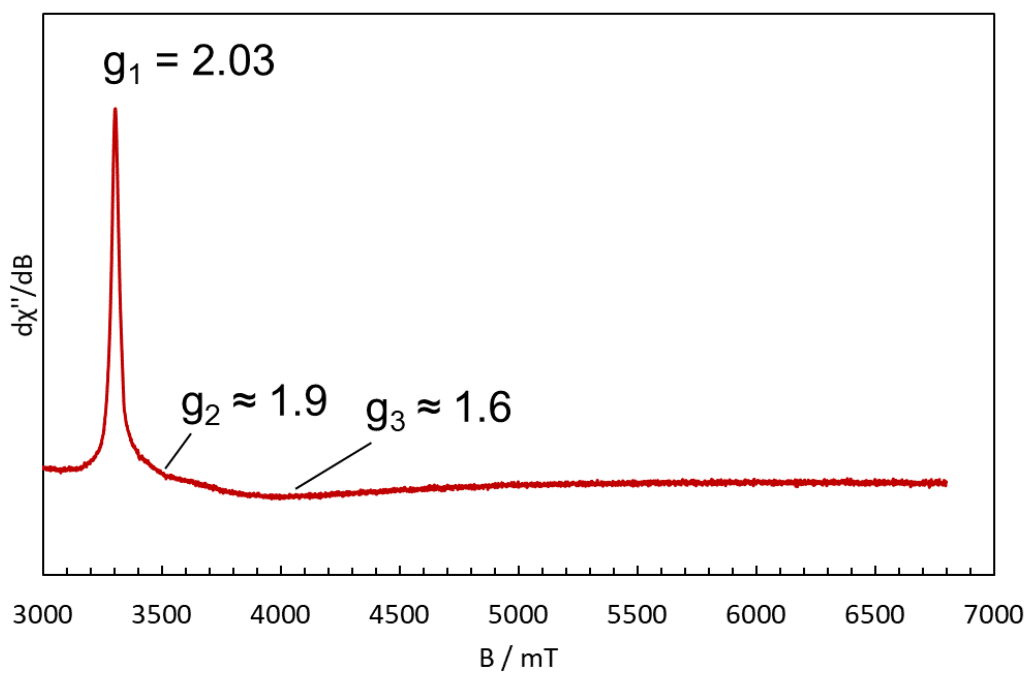


Figure S24. X-band EPR spectrum of $(\text{L}(\text{NIm}^{\text{Me}})_3)\text{Fe}_6\text{S}_6\text{Cl}_3$ (**11**) in a 2-MeTHF glass at 8 K. MW frequency = 9.374 MHz.

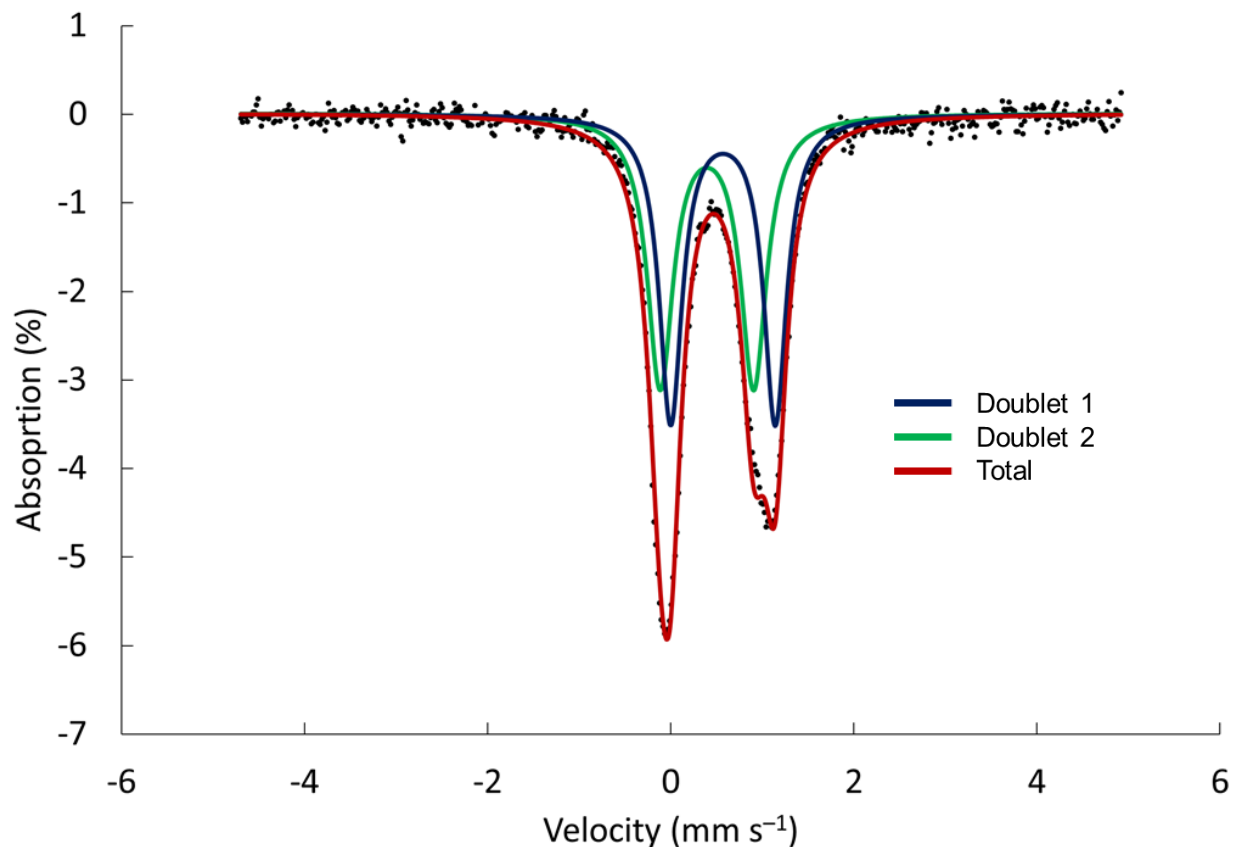


Figure S25. Zero-field ^{57}Fe Mössbauer spectrum of solid $(\text{L}(\text{NIm}^{\text{Me}})_3)\text{Fe}_6\text{S}_6\text{Cl}_3$ (**11**) at 90 K with alternative simulations. Black circles represent experimental data, solid lines are simulations. Simulation parameters: Doublet 1: $\delta = 0.57 \text{ mm s}^{-1}$, $\Delta E_Q = 1.14 \text{ mm s}^{-1}$, $\Gamma = 0.30 \text{ mm s}^{-1}$, relative area = 0.5; Doublet 2: $\delta = 0.39 \text{ mm s}^{-1}$, $\Delta E_Q = 1.02 \text{ mm s}^{-1}$, $\Gamma = 0.34 \text{ mm s}^{-1}$, relative area = 0.5. We disfavor this simulation as the isomer shift values differ significantly from those reported for $[\text{Fe}_6\text{S}_6\text{X}_6]^{3-}$ (X = halide, RS^- , RO^-) clusters.¹

¹H NMR Spectra of Reaction Mixtures

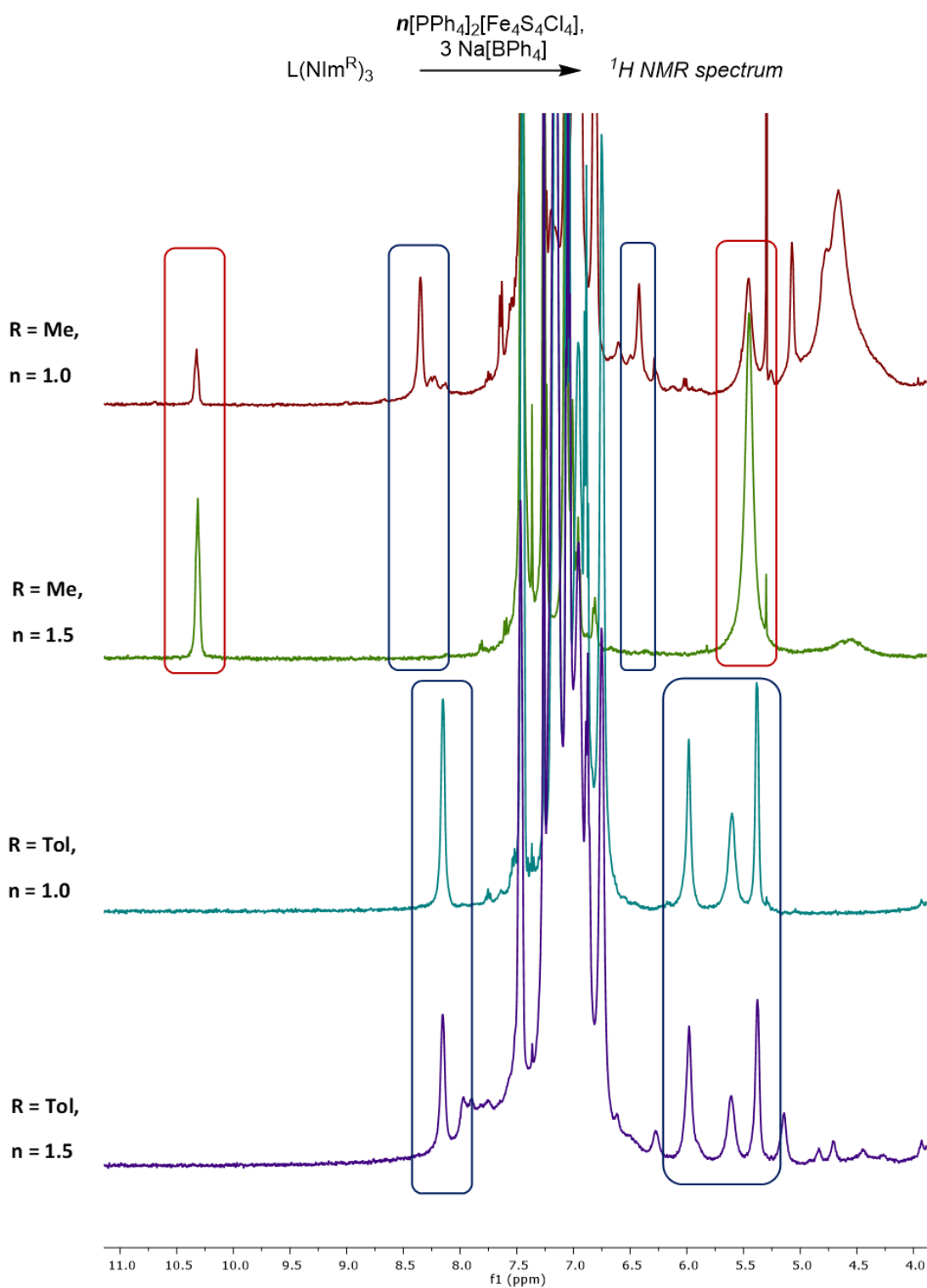


Figure S26. ¹H NMR spectra in CDCl₃ of the crude mixtures from the reactions between L(NIm^R)₃ and different stoichiometries of [PPh₄]₂[Fe₄S₄Cl₄] and Na[BPh₄]. **Blue boxes** indicate peaks assigned to $[(\text{L(NIm}^{\text{R}})_3\text{Fe}_4\text{S}_4\text{Cl}_4)]^+$ clusters; **red boxes** peaks assigned to $(\text{L(NIm}^{\text{R}})_3\text{Fe}_6\text{S}_6\text{Cl}_3)$ clusters.

References

1. Kanatzidis, M. G.; Hagen, W. R.; Dunham, W. R.; Lester, R. K.; Coucouvanis, D. Metastable Fe/S Clusters. The Synthesis, Electronic Structure, and Transformations of the $[\text{Fe}_6\text{S}_6\text{Cl}_6]^{3-}$, $[\text{Fe}_6\text{S}_6\text{Br}_6]^{3-}$, $[\text{Fe}_6\text{S}_6\text{I}_6]^{3-}$, $[\text{Fe}_6\text{S}_6(\text{RS})_6]^{3-}$, $[\text{Fe}_6\text{S}_6(\text{RO})_6]^{3-}$ Clusters and the Structure of $[(\text{C}_2\text{H}_5)_4\text{N}]_3[\text{Fe}_6\text{S}_6\text{Cl}_6]$. *J Am Chem Soc* **1985**, 107 (4), 953-961.